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Modeling post-combustion CO₂ capture with amine solvents

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Global warming context

February 1993



February 2000



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1. Introduction

1. Introduction

- Importance of coal for electricity generation in the near-future

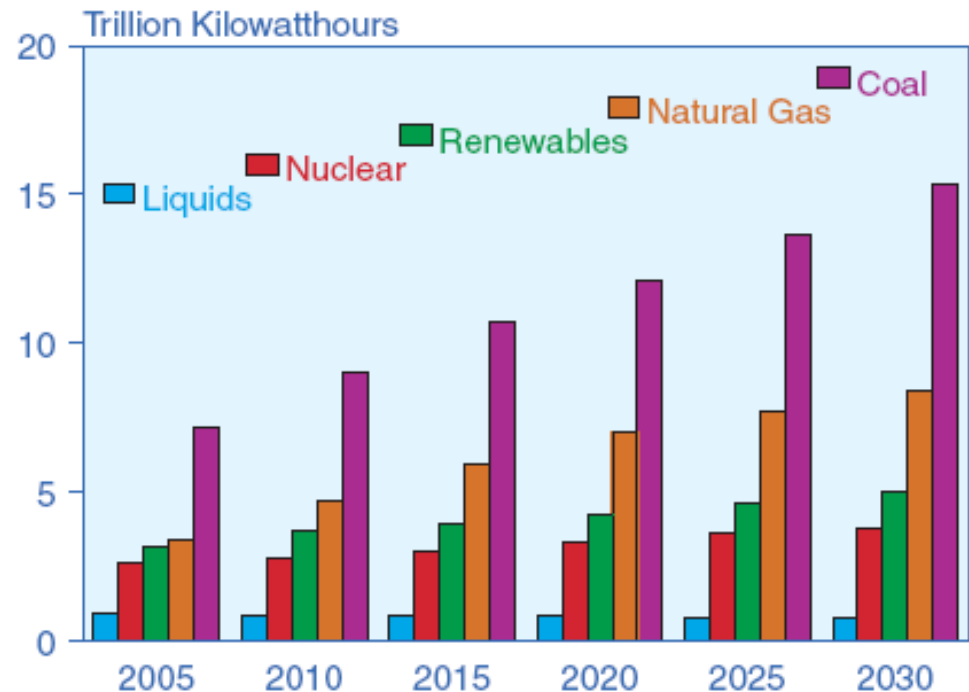
- Environmental concerns

=> Energy efficiency

=> biomass

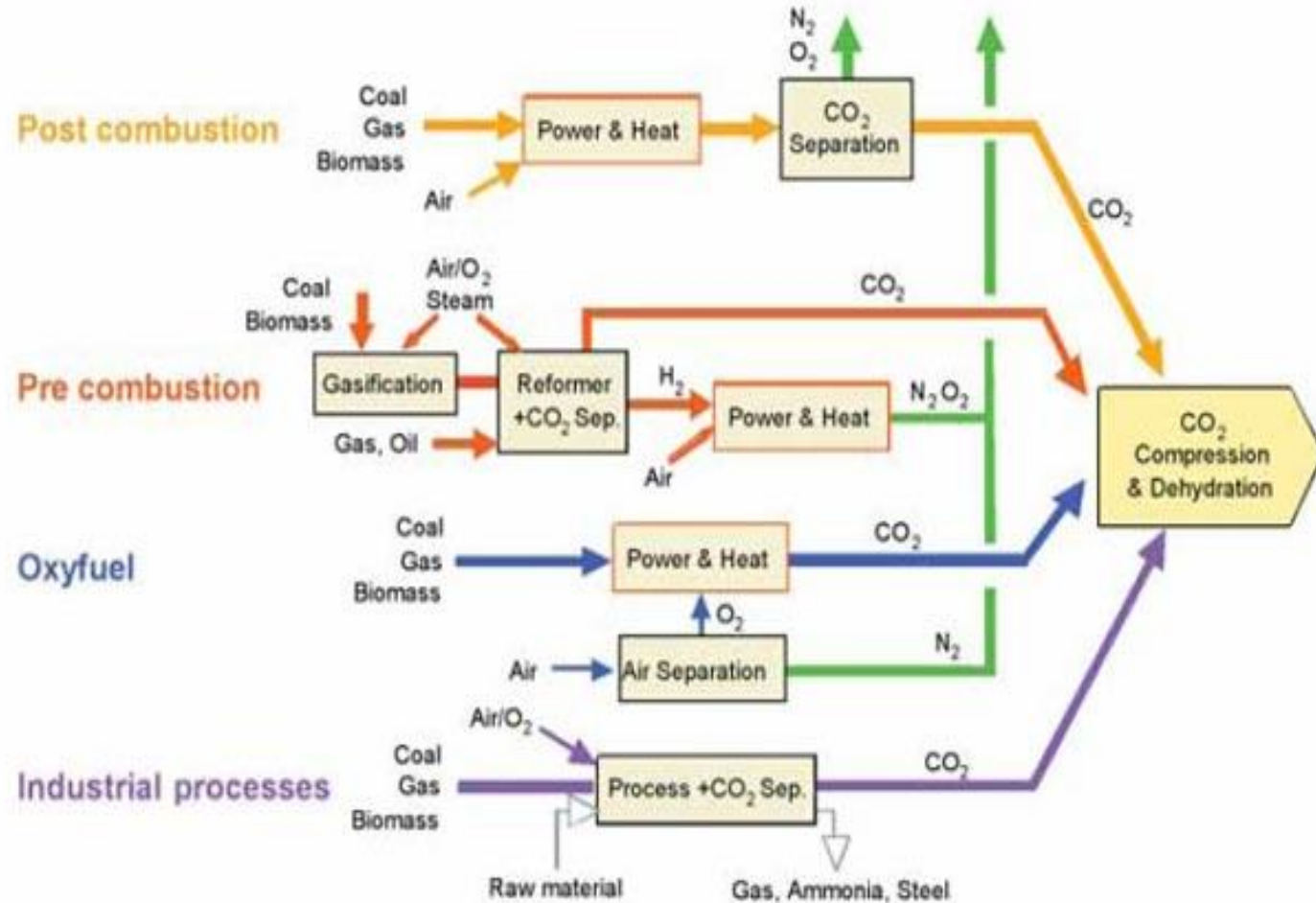
=> Carbon Capture and Storage (CCS)

World Electricity Generation by Fuel, 2005-2030



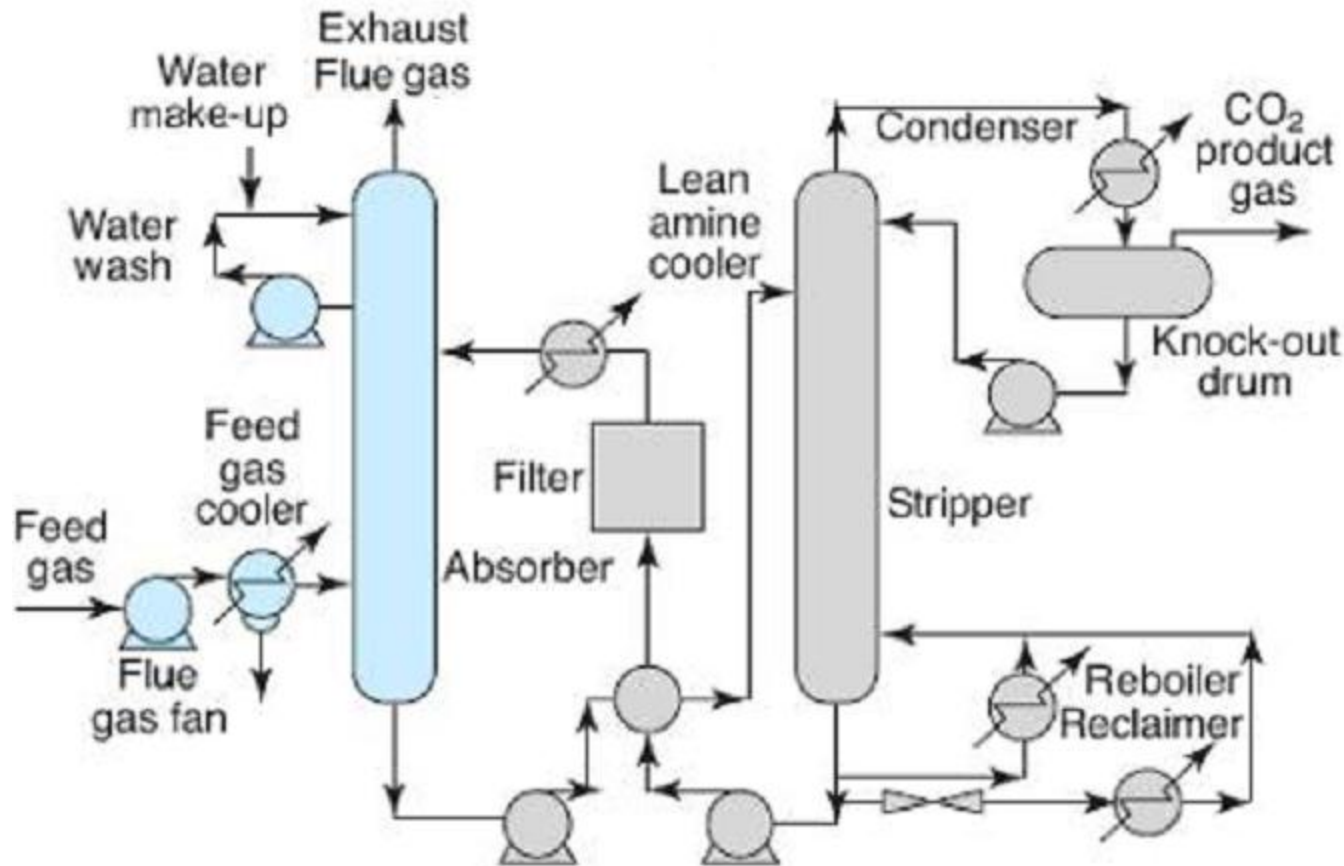
1. Introduction

CO₂-Capture methods



1. Introduction

Post-combustion capture



2. Objectives

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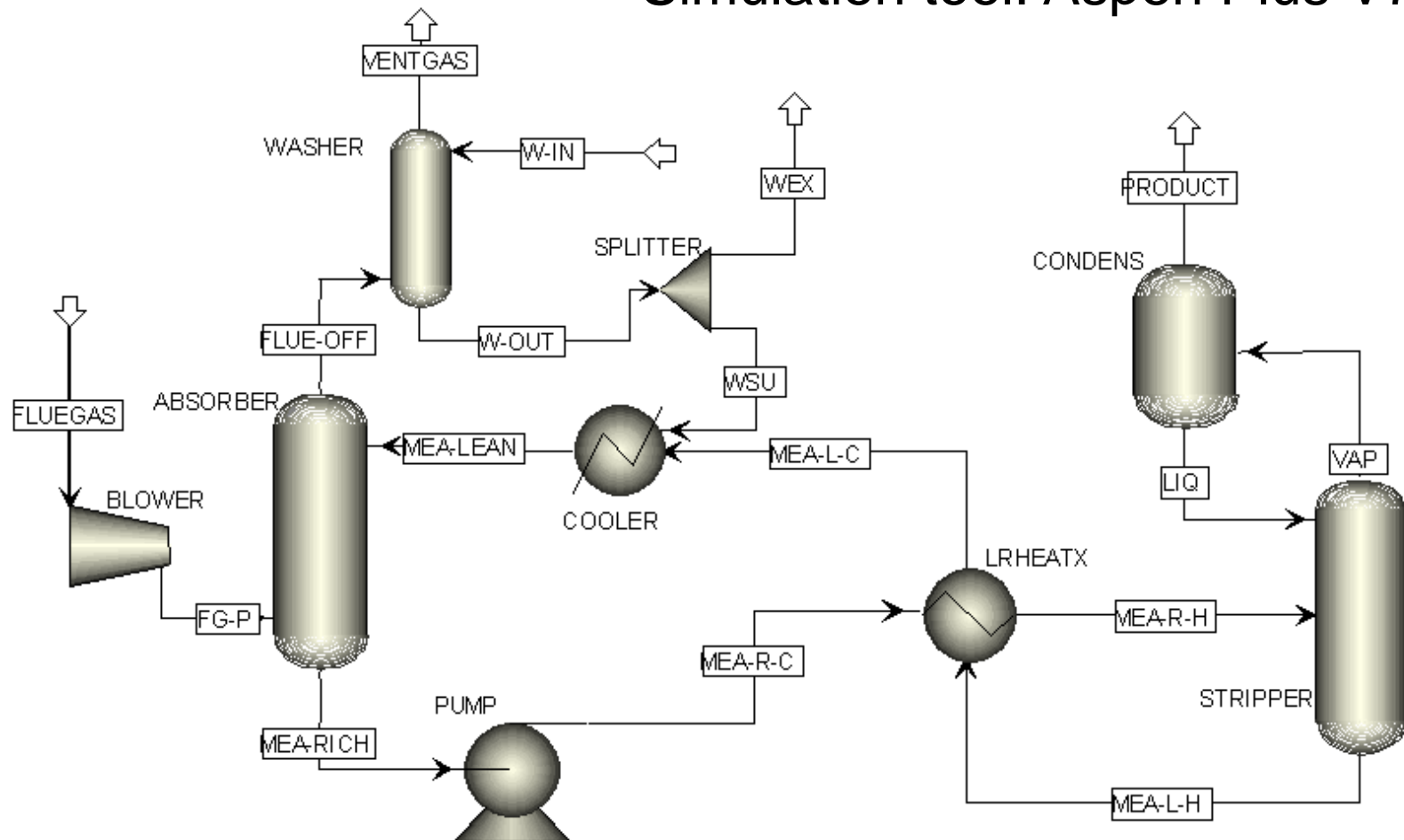
Objectives:

- Development of a simulation model for a pilot capture installation (5000 Nm³ flue gas per hour / 1 ton CO₂ per hour)
- Identification of clue parameters
- Optimization of the developed model
- Simulation of process improvements

3. Model description

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Simulation tool: Aspen Plus V7.0



3. Model description

Model main characteristics

- Flue gas flow: 2500 Nm³/h
- 90% CO₂-recovery rate
- captured CO₂: 566 kg/h
- MEA solvent, concentration = 30 wt-%

3. Model description

Thermodynamical equilibrium model for the columns

=> No mass transfer limitations

=> No reaction kinetics limitations

However, global results are still in good agreement with the experimental results^[1].

⇒ Not a problem as long as internal column profiles are not studied in details!^[2]



[1]: J. Knudsen, P. Vilhelmsen, J. Jensen, O. Biede, First year operation experience with a 1t/h CO₂ absorption pilot plant at Esbjerg coal-fired power plant, 2008

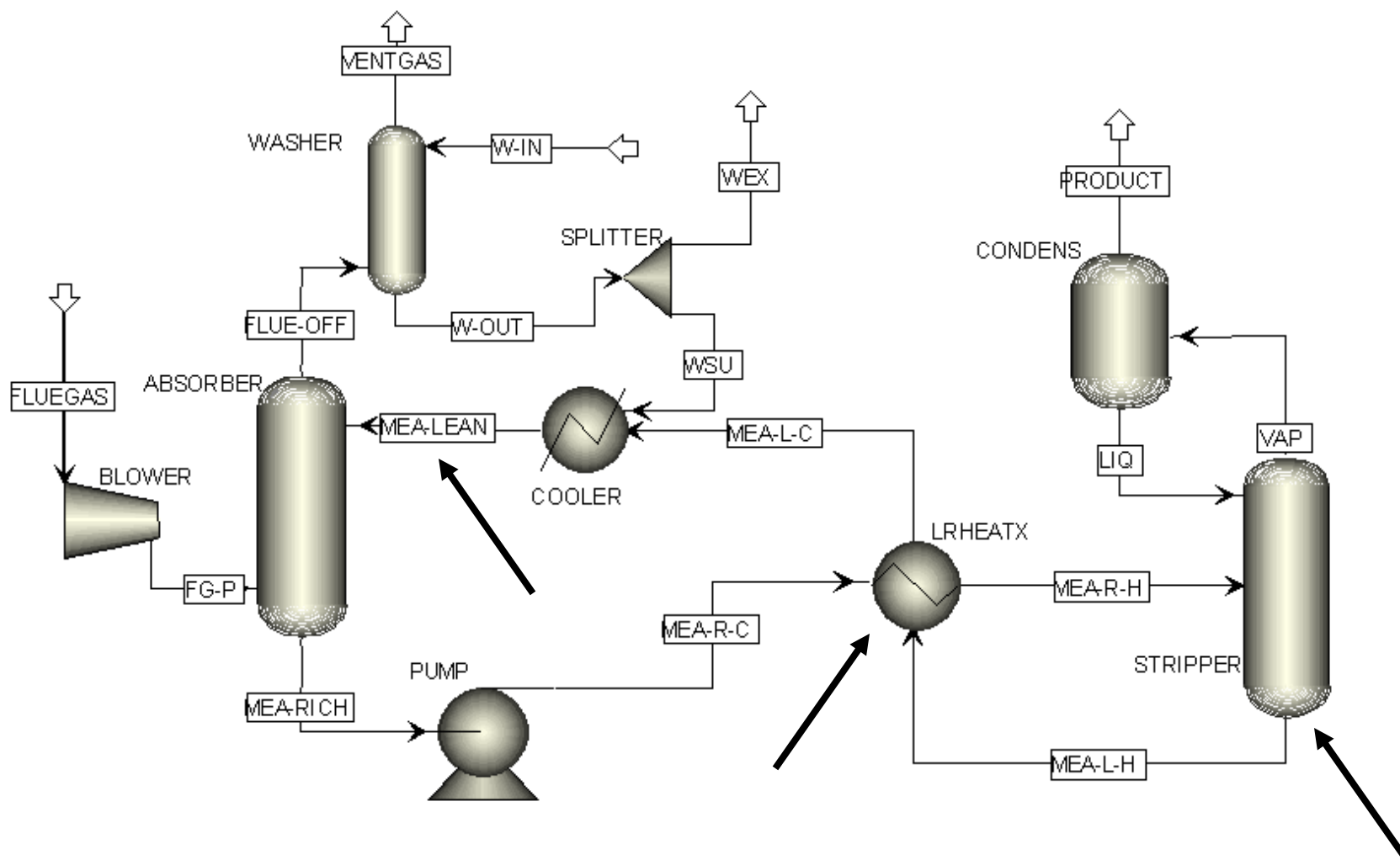
[2]: Abu Zahra M., 2009. Carbon dioxide capture from flue gas, PhD Thesis, TU Delft

4. Simulation results

Optimization of the base case

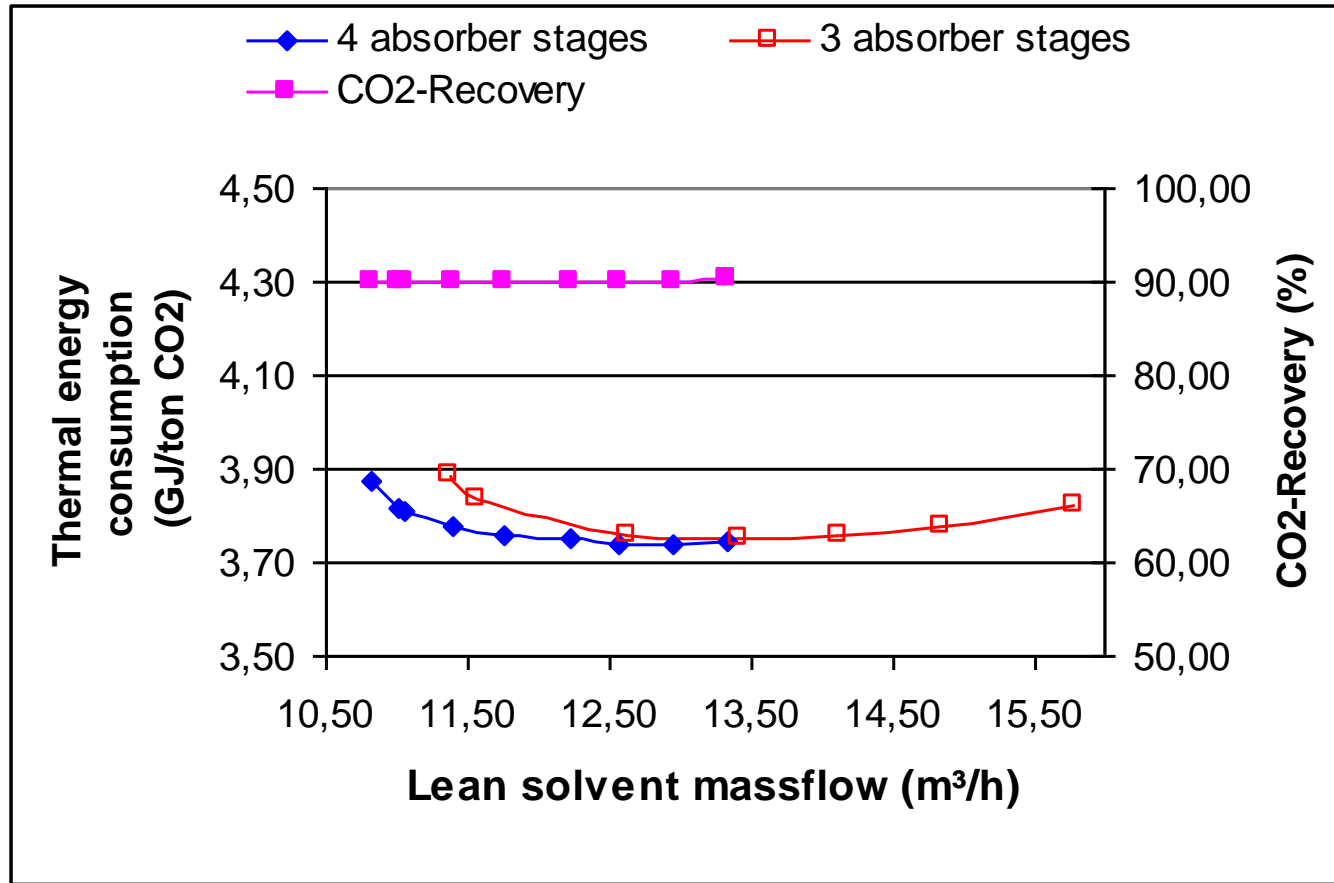
- 4.1 Lean solvent flow
- 4.2 Lean solvent concentration
- 4.3 Lean solvent inlet temperature in the absorber
- 4.4 Stripper pressure
- 4.5 Temperature approach at the lean-rich heat exchanger

4. Simulation results



4. Simulation results

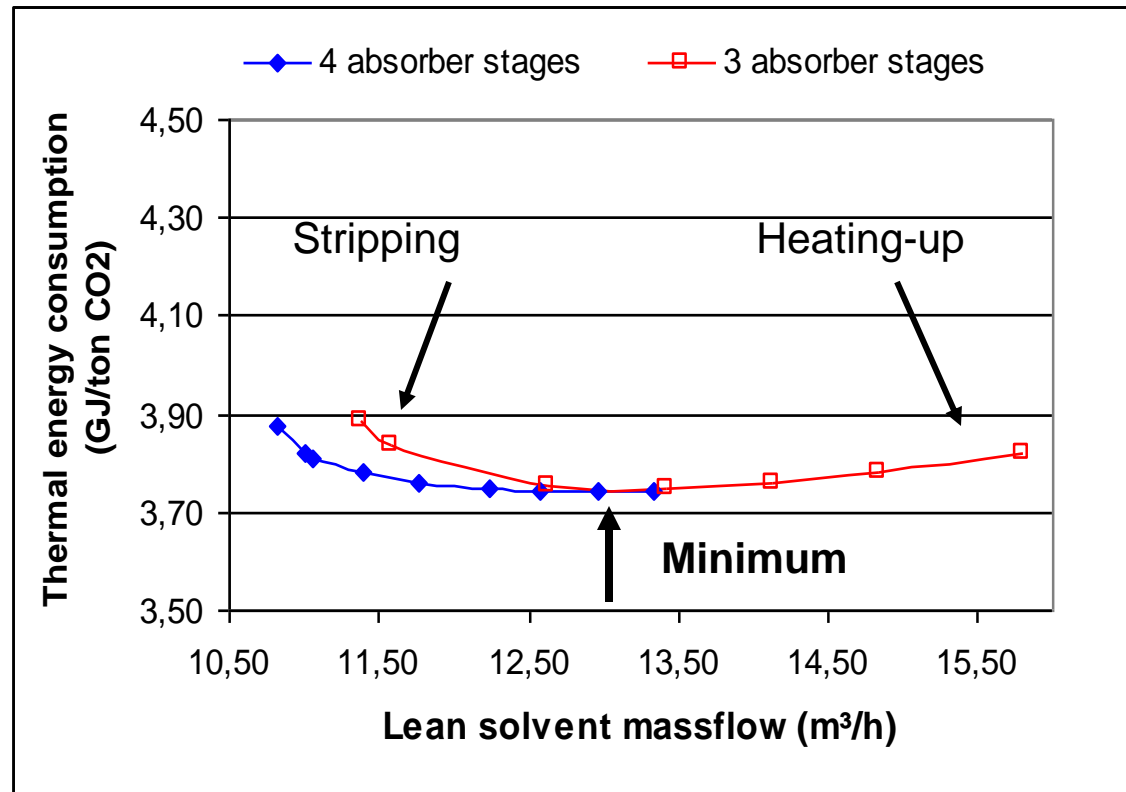
4.1 Optimization of the lean solvent flow



4. Simulation results

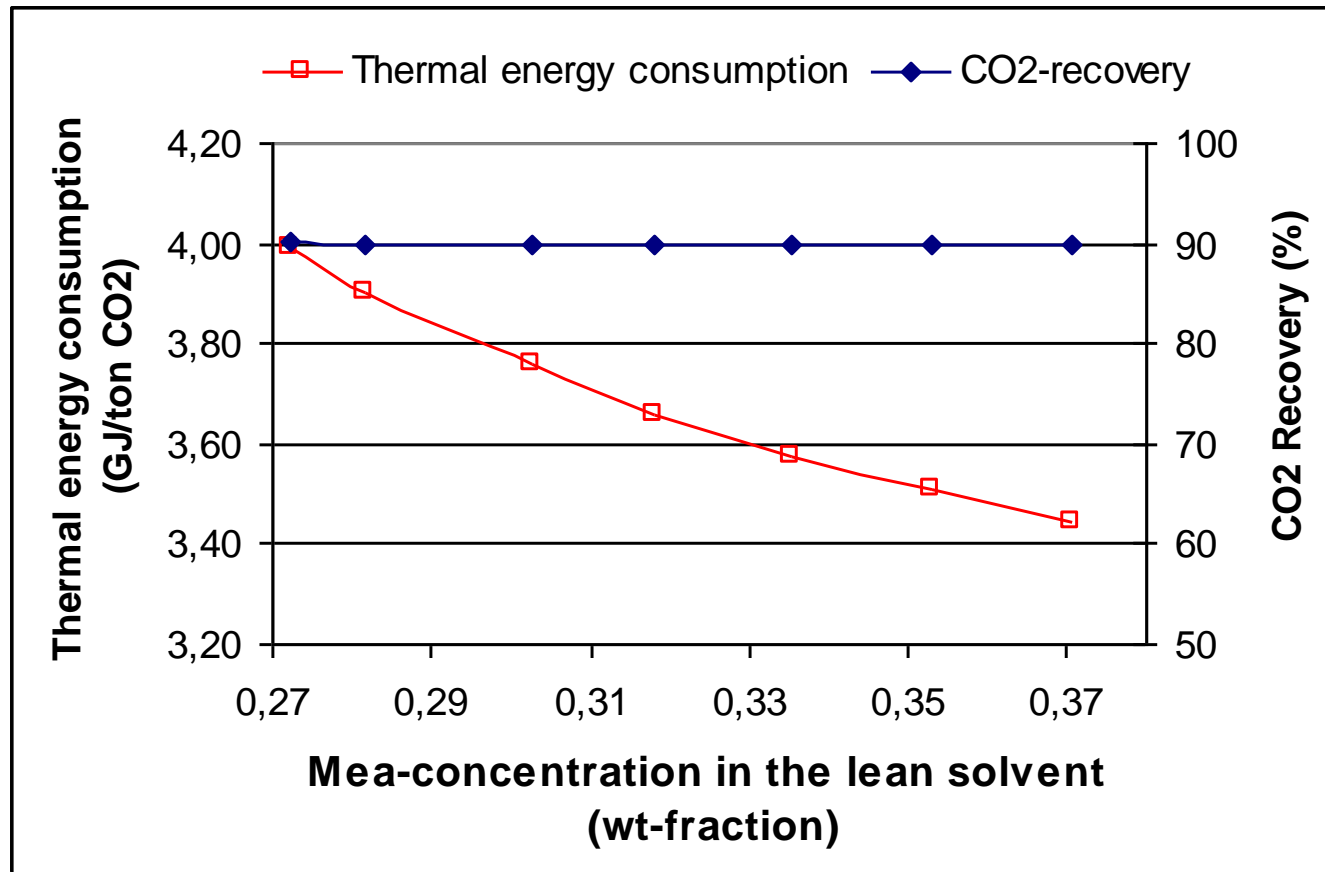
3 contributions to the thermal energy consumption :

- Solvent heating-up
- Desorption enthalpy
- Stripping steam



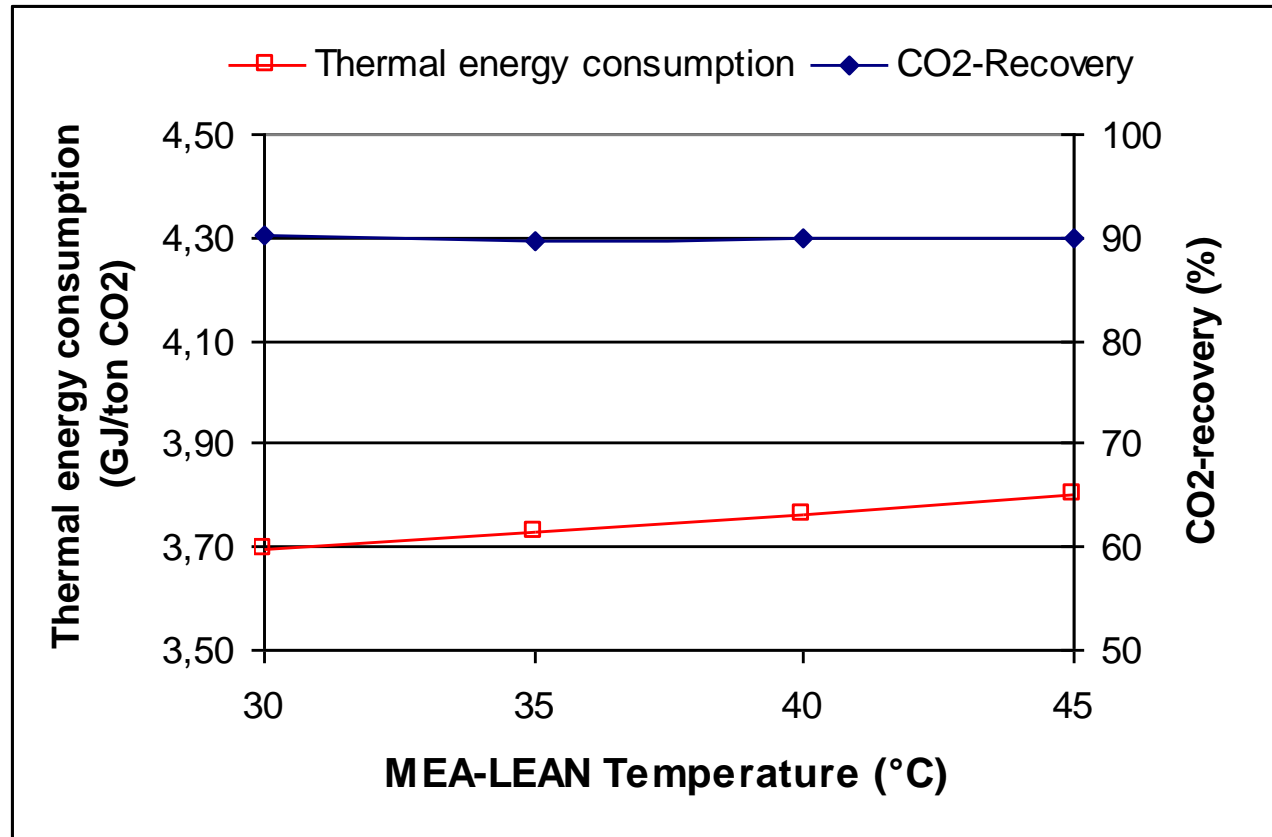
4. Simulation results

4.2 Optimization of the lean solvent concentration



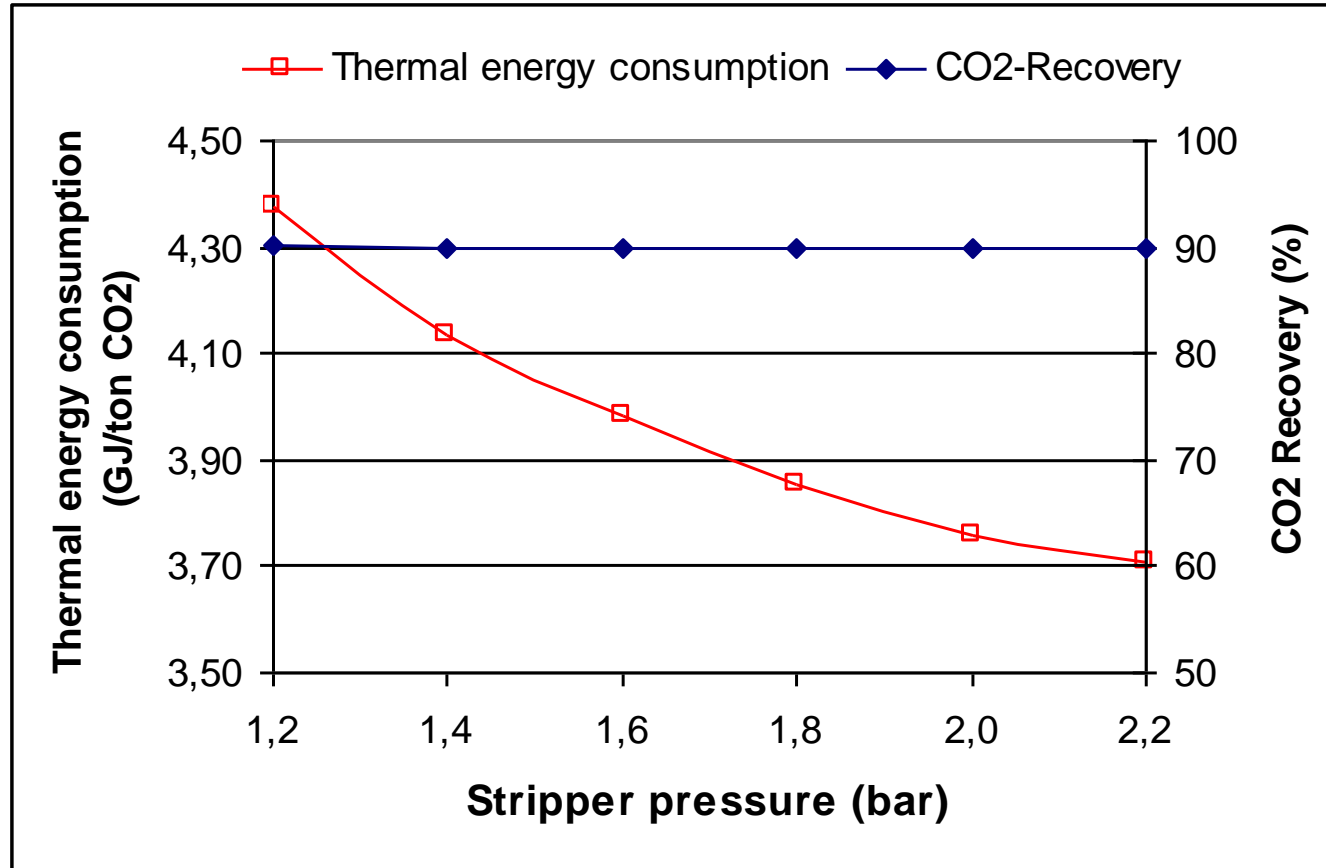
4. Simulation results

4.3 Optimization of the lean-solvent inlet temperature in the absorber



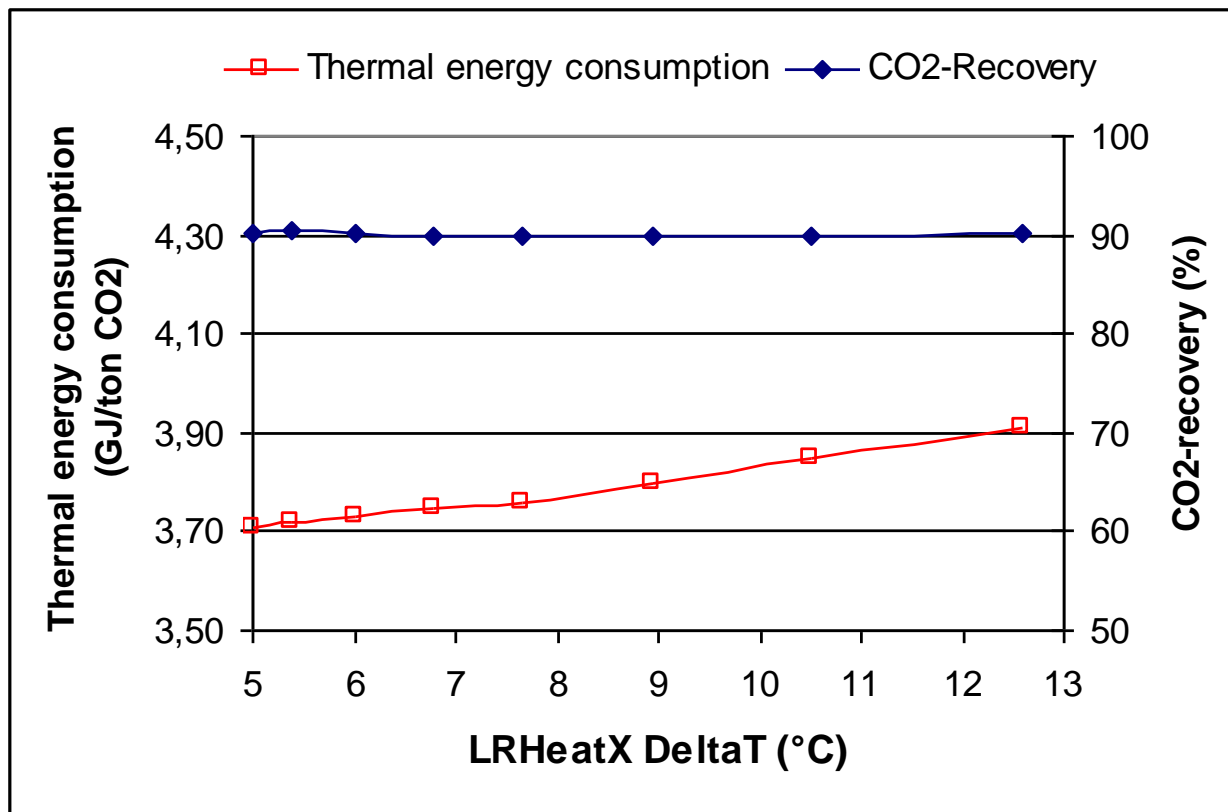
4. Simulation results

4.4 Optimization of the stripper pressure



4. Simulation results

4.5 Optimization of the temperature approach at the lean-rich heat exchanger



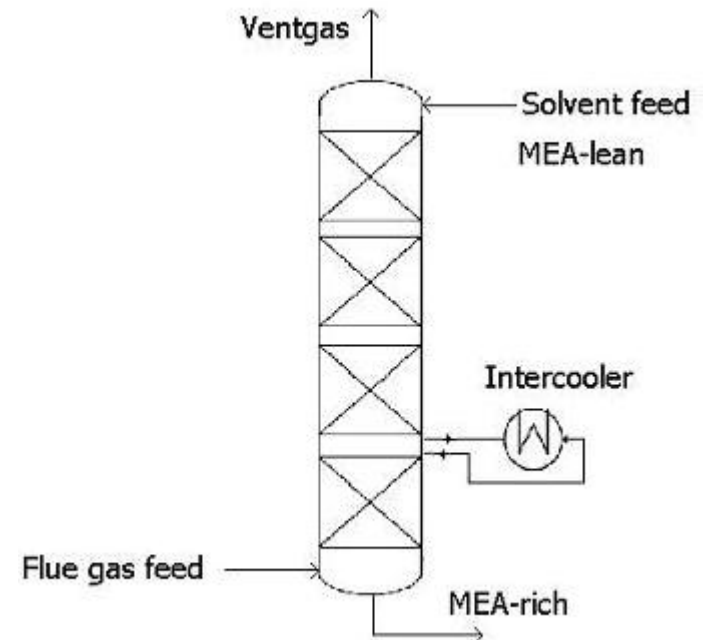
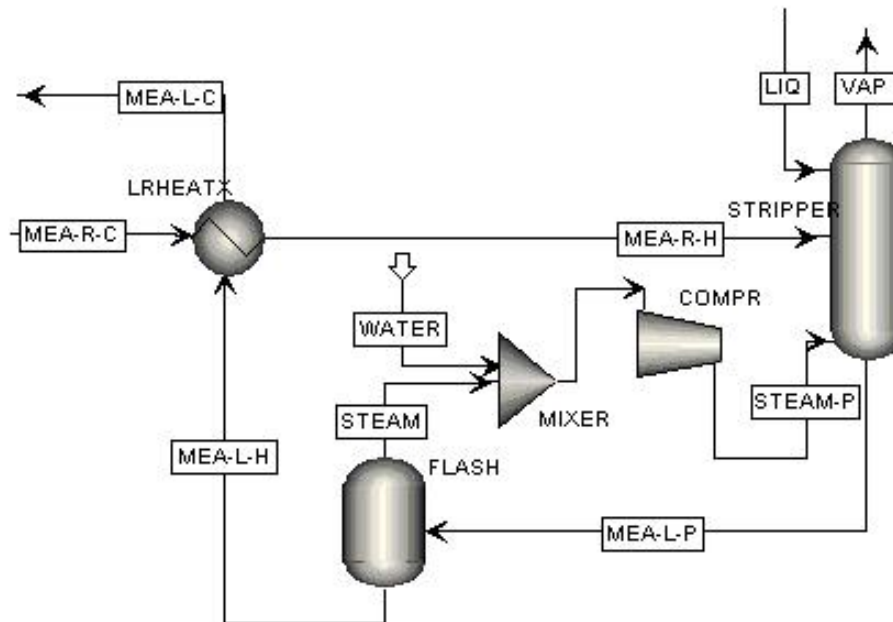
4. Simulation results

Parameter	Best-case value	Reduction of the thermal energy consumption	Disadvantage
MEA inlet flow	11,725 m ³ /h	- 3%	not experimentally confirmed yet
MEA inlet concentration	40 wt-%	- 12,5%	Corrosive behavior
MEA inlet temperature	30 °C	-2,5%	Increase of the cooling water requirement
Stripper pressure	2,2 bar	- 16%	Possibility of solvent degradation
Temp. approach at the L-R heat exchanger	5 °K	- 5%	Increase of the equipment costs

5. Process modifications

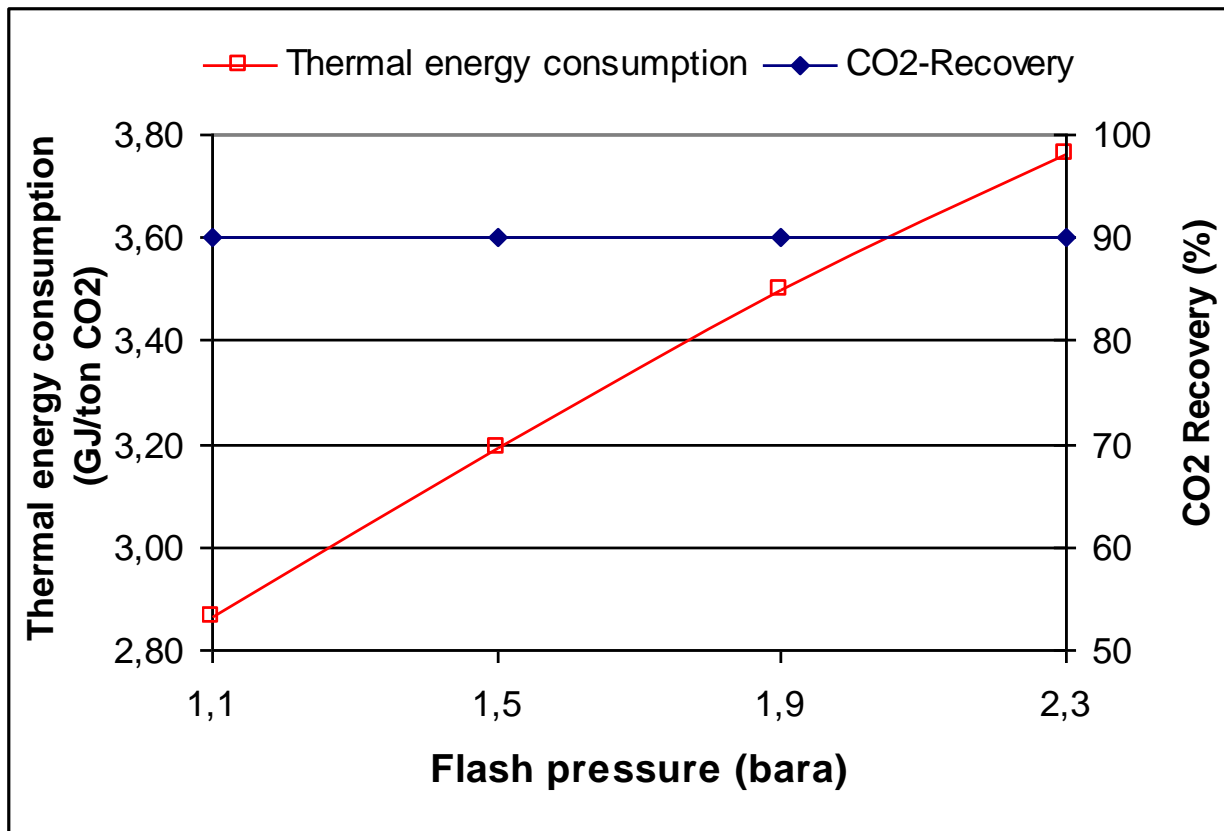
5. Process modifications

- 5.1 Lean vapor compression
- 5.2 Absorber intercooling



5. Process modifications

5.1 Influence of the flash pressure by the LVC

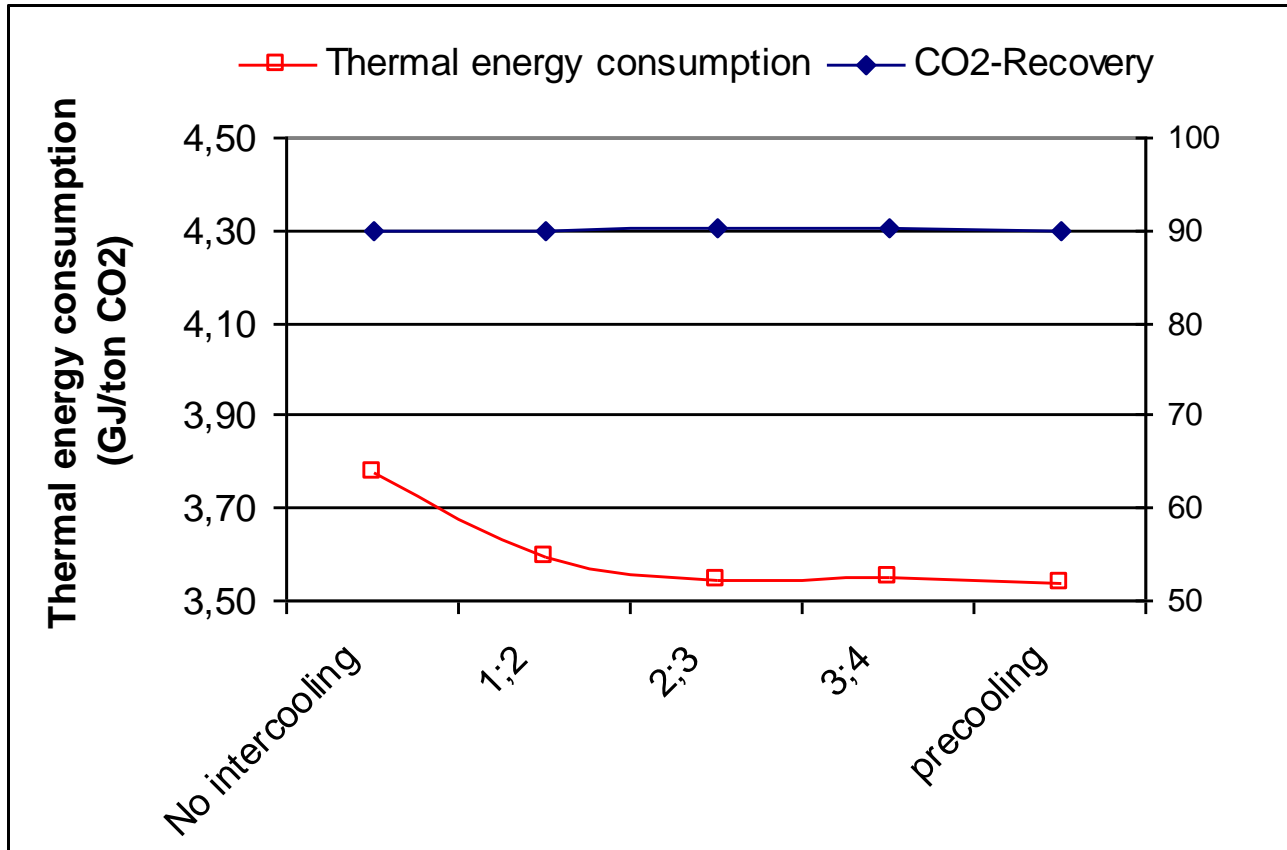


- 25 % Thermal energy consumption

- 18 % Exergy consumption

5. Process modifications

5.2 Influence of intercooler position



- 6 % Thermal
energy consumption

6. Conclusion and perspectives

6. Conclusion and perspectives

- Simulation model => coherent results in comparison with the experiments on the Esbjerg pilot plant
- Reduction of the thermal energy consumption by 29% achievable (IC + LVC)
- Reduction of the process exergy consumption reaches 19,5%
- Still some limitations to overcome

6. Conclusion and perspectives

Consequences of degradation on:

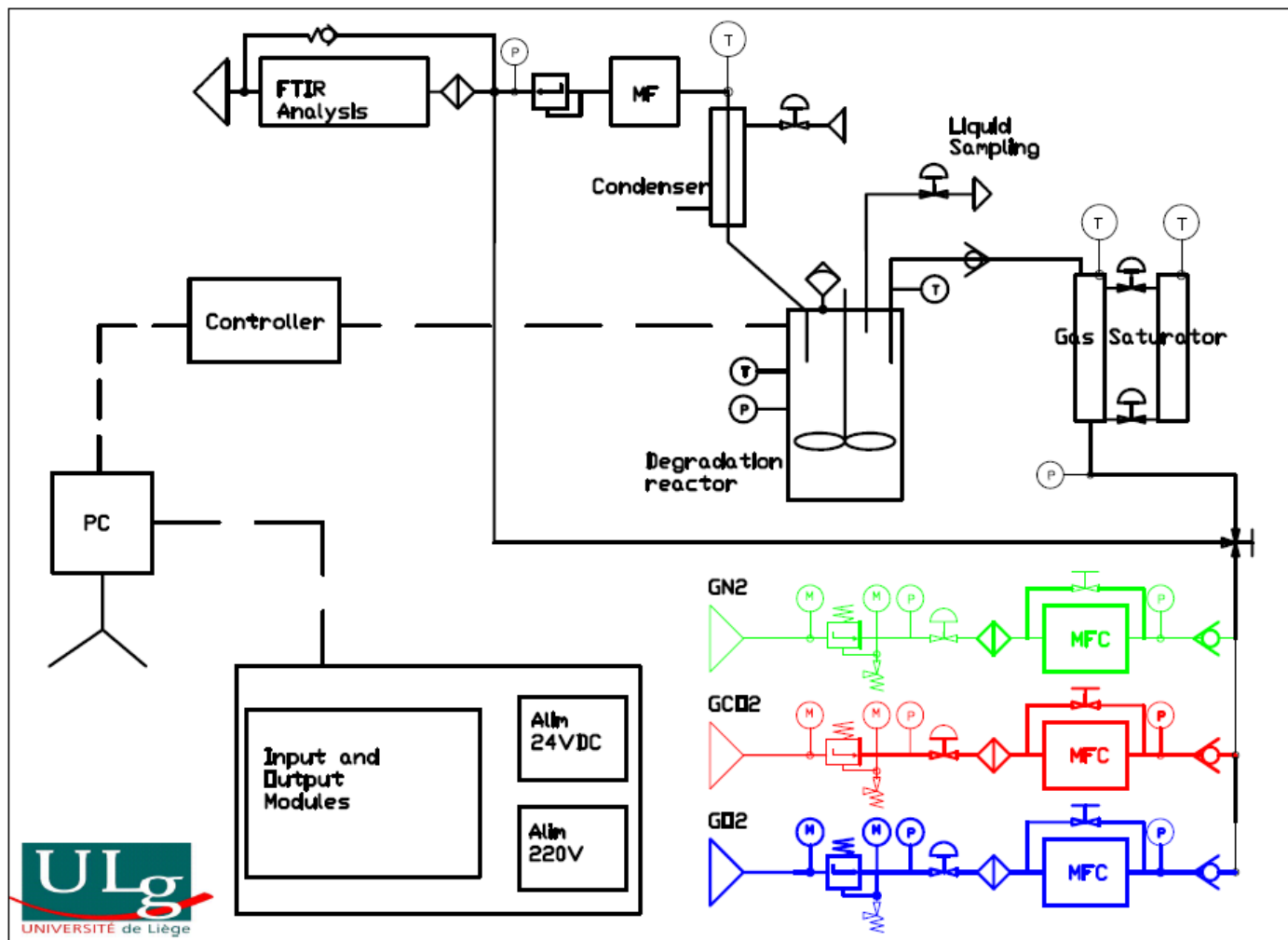
- Process operating costs:
 - Solvent make-up (4,78 M€/year for a 600MWe bituminous coal-fired power plant^[1])
 - Removal and disposal of toxic degradation products
 - 4-10% of the total operating costs!
- Process performance
 - Solvent loading capacity decreases
 - Viscosity increases
- Capital costs
 - Corrosion

6. Conclusion and perspectives

Degradation studies:

- Construction of a test installation for the study of solvent degradation phenomena
- Study of classical solvents
- Study of newly developed solvents and of degradation inhibitors
- Optimal conditions in order to minimize degradation

6. Conclusion and perspectives



6. Conclusion and perspectives

- Results of degradation studies will be used for process simulation purpose.
- Multi-objective analysis will focus on the conciliation of economical and environmental considerations
- Final objective is to propose optimal operating conditions for the CO₂ capture process

Acknowledgements

*Laboratory for Analysis
and Synthesis of
Chemical Systems*



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Thank you for your attention !

