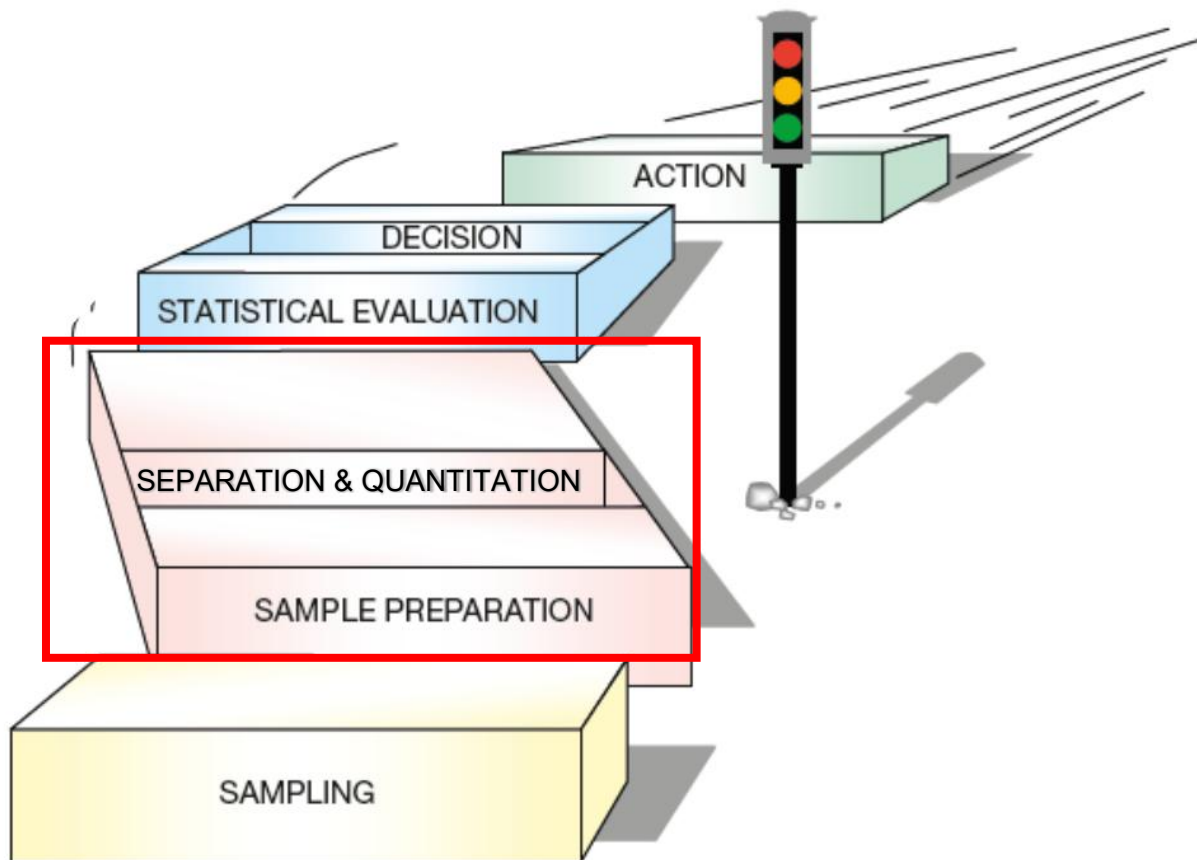


GC×GC an opportunity for simplifying sample preparation or an extra difficulty in food analysis?

Giorgia Purcaro , Damien Eggermont, Aleksandra Gorska

Gembloux Agro Bio-Tech, University of Liège, Belgium

gpurcaro@uliege.be



Sample preparation:

➤ Handle Matrix complexity

- Selectivity (e.g., targeted)
- Representativeness (e.g., untargeted)
- Greenness (e.g., miniaturization)
- Automation
- Sustainability
- ...

Chromatography:

➤ Maximize separation resolution

“reductionist
approach”

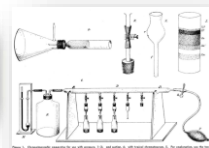
“interactionism” and -omics
sciences”

1800s **Wet chemistry era**
main components



1900s **Instrumental era**
minor and trace components

1906



1934 Introduction of the “Acidimeter” (automatic pHmeter)

1941 **LC**

Gas-liquid Partition Chromatography: the
Micro-estimation of Volatile Fatty Acids
Formic Acid to Dodecanoic Acid

1951 **GC**

151. A NEW FORM OF CHROMATOGRAM
EMPLOYING TWO LIQUID PHASES

1. A THEORY OF CHROMATOGRAPHY

2. APPLICATION TO THE HIGHER

OF THE HIGHER

By A. J. P. MARTIN

From the Wool Industries Research Institute

(Received 5 June 1951)

(Received 5 June 1951)

(Received 5 June 1951)

(Received 5 June 1951)

(Received 5 June 1951)

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1958 **Capillary column**

1959 **GC-MS**

Time-of-Flight Mass Spectrometry and
Gas-Liquid Partition Chromatography

Journal of Chromatographic Science, Vol. 18, October 1980

R. S. GOHIL

Spectroscopy

1980 **LC-GC**

Multidimensional High Performance
Liquid Chromatography*

Journal of Chromatographic Science, Vol. 29, June 1991

Journal of Chromatographic Science, Vol. 29, June 1991

Journal of Chromatographic Science, Vol. 29, June 1991

Journal of Chromatographic Science, Vol. 29, June 1991

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Journal of Chromatographic Science, Vol. 29, June 1991

Journal of Chromatographic Science, Vol. 29, June 1991

Journal of Chromatographic Science, Vol. 29, June 1991

1991 **GC×GC**

Comprehensive LC × GC for Enhanced Headspace Analysis

Wes W. C. Quigley

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

Department of Chemistry, University of

2000 **LC×GC**

Comprehensive multi-dimensional chromatographic studies on the
separation of saturated hydrocarbon ring structures
in petrochemical samples

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

Journal of Chromatography A, 1086 (2005) 13–20

2005 **LC-GC×GC**

R. Edam^a, J. Blomberg^{b,*}, H.-G. Janssen^c, P.J. Schoenmakers^a

^a University of Amsterdam, Van 't Hoff Institute for Molecular Sciences, Amsterdam, The Netherlands

^b Shell Global Solutions, Shell Research & Technology Centre Amsterdam, Amsterdam, The Netherlands

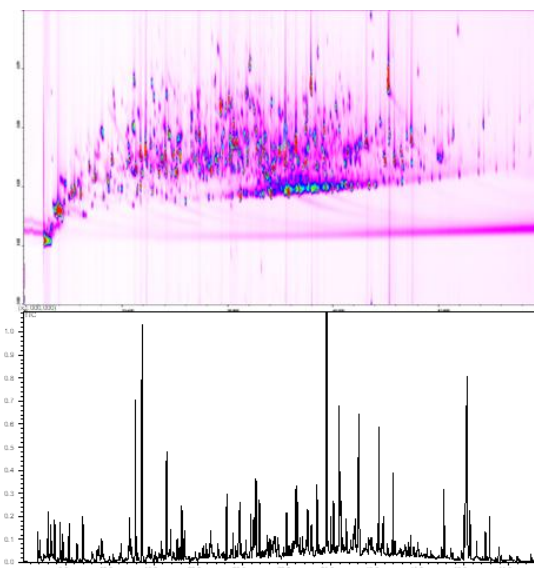
^c Unilever Research and Development, Central Analytical Science, Vlaardingen, The Netherlands

Chromatography:

➤ Maximize separation resolution



Multidimensional Chromatography

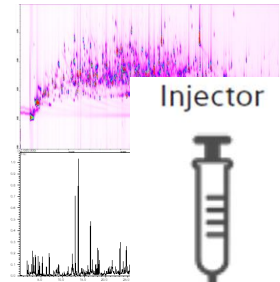


Journal of Chromatographic Science, Vol. 29, June 1991

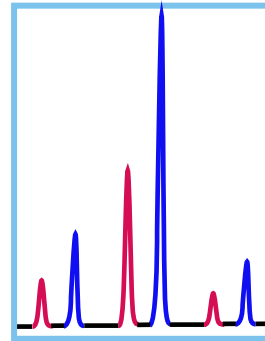
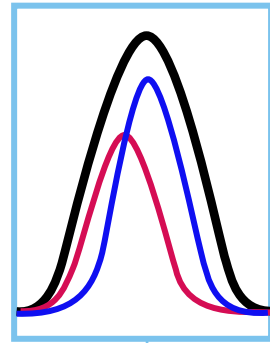
Comprehensive Two-Dimensional Gas Chromatography using an On-Column Thermal Modulator Interface

Zaiyou Liu and John B. Phillips*
Department of Chemistry & Biochemistry, Southern Illinois University, Carbondale, Illinois 62901

1991 GC×GC



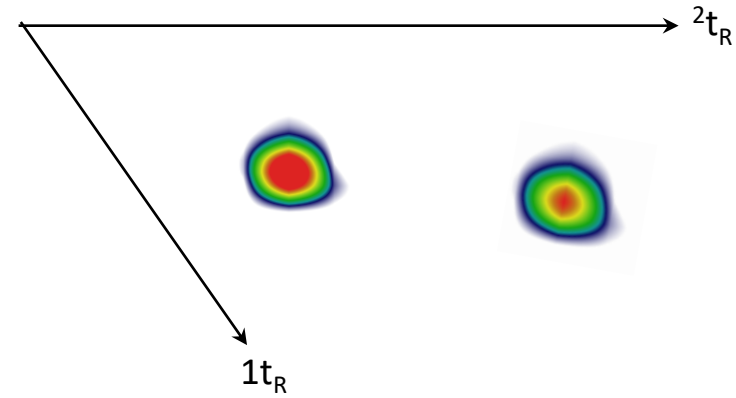
Injector



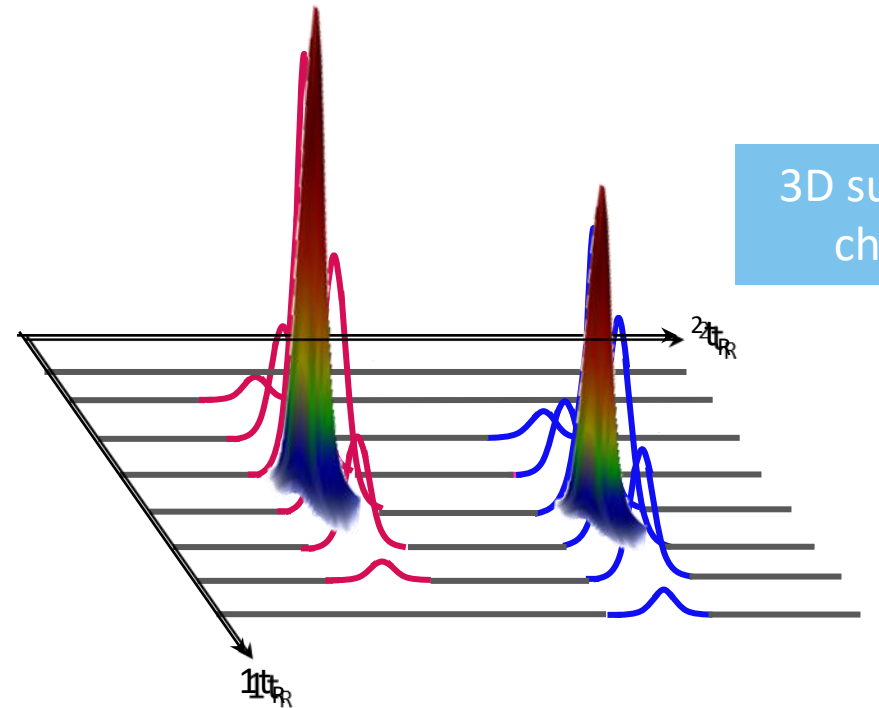
Primary column

The
"beating heart"
of GC×GC

Secondary column



Colour plot



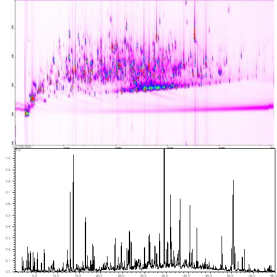
3D surface chart

Journal of Chromatographic Science, Vol. 29, June 1991

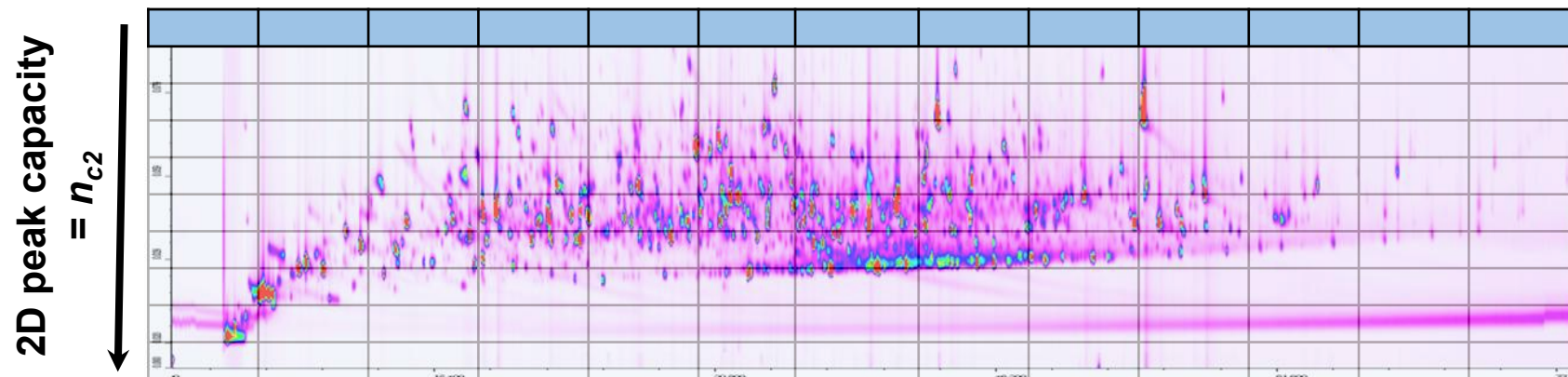
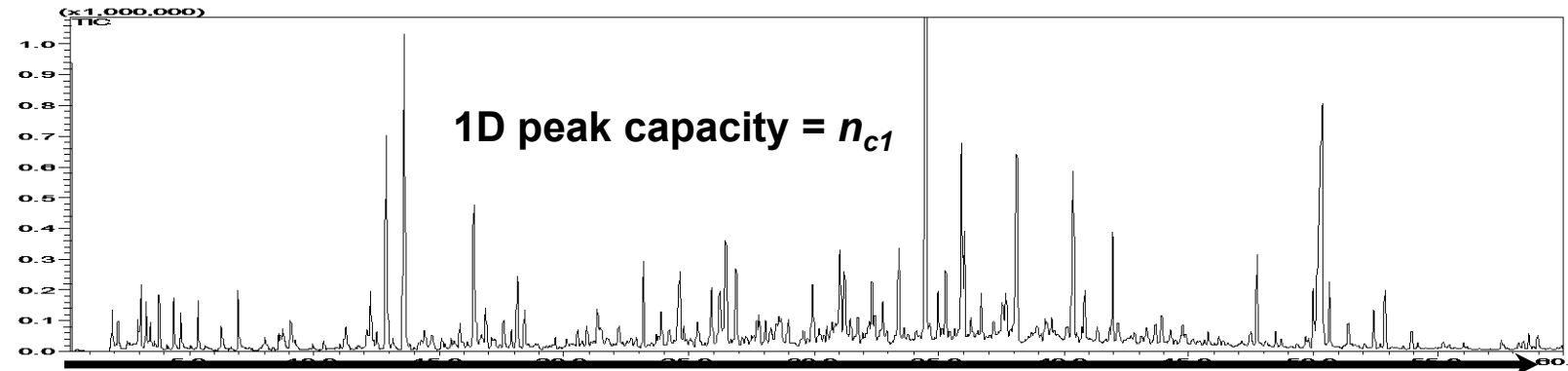
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1991 GC×GC

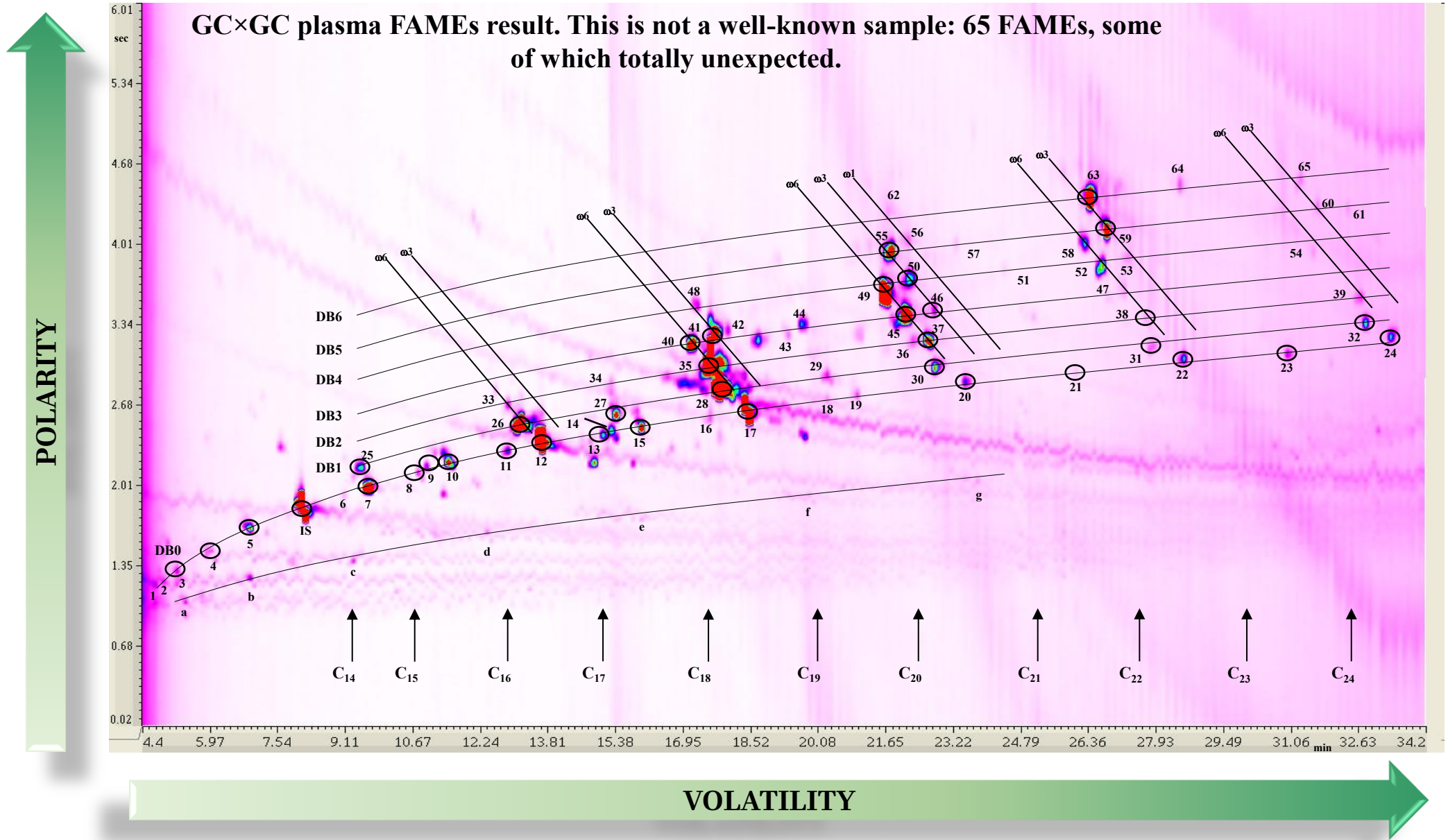


Comprehensive: CC × CC



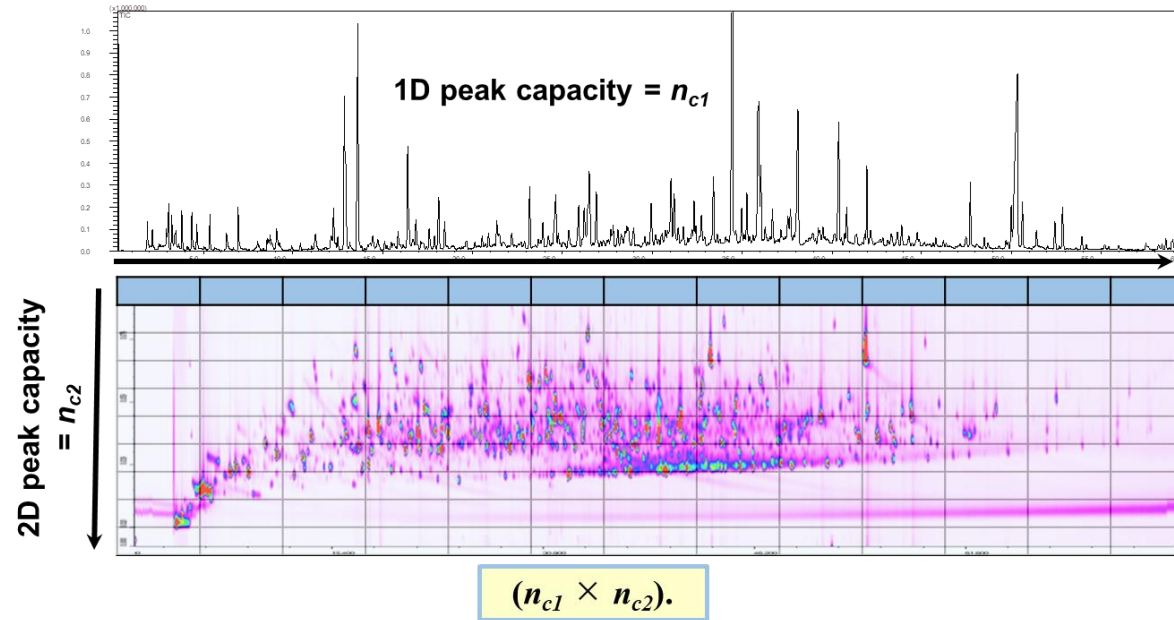
$$(n_{c1} \times n_{c2}).$$

GC×GC structured chromatogram



GC×GC in food analysis

Comprehensive: CC × CC

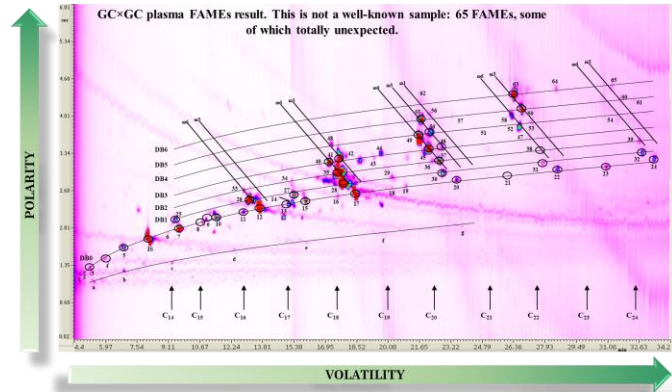


Synergic to Sample
Prep



Unveil and explored sample
multidimensionality





Selectivity of different fibers

Journal of Chromatography A, 1251 (2012) 208–218



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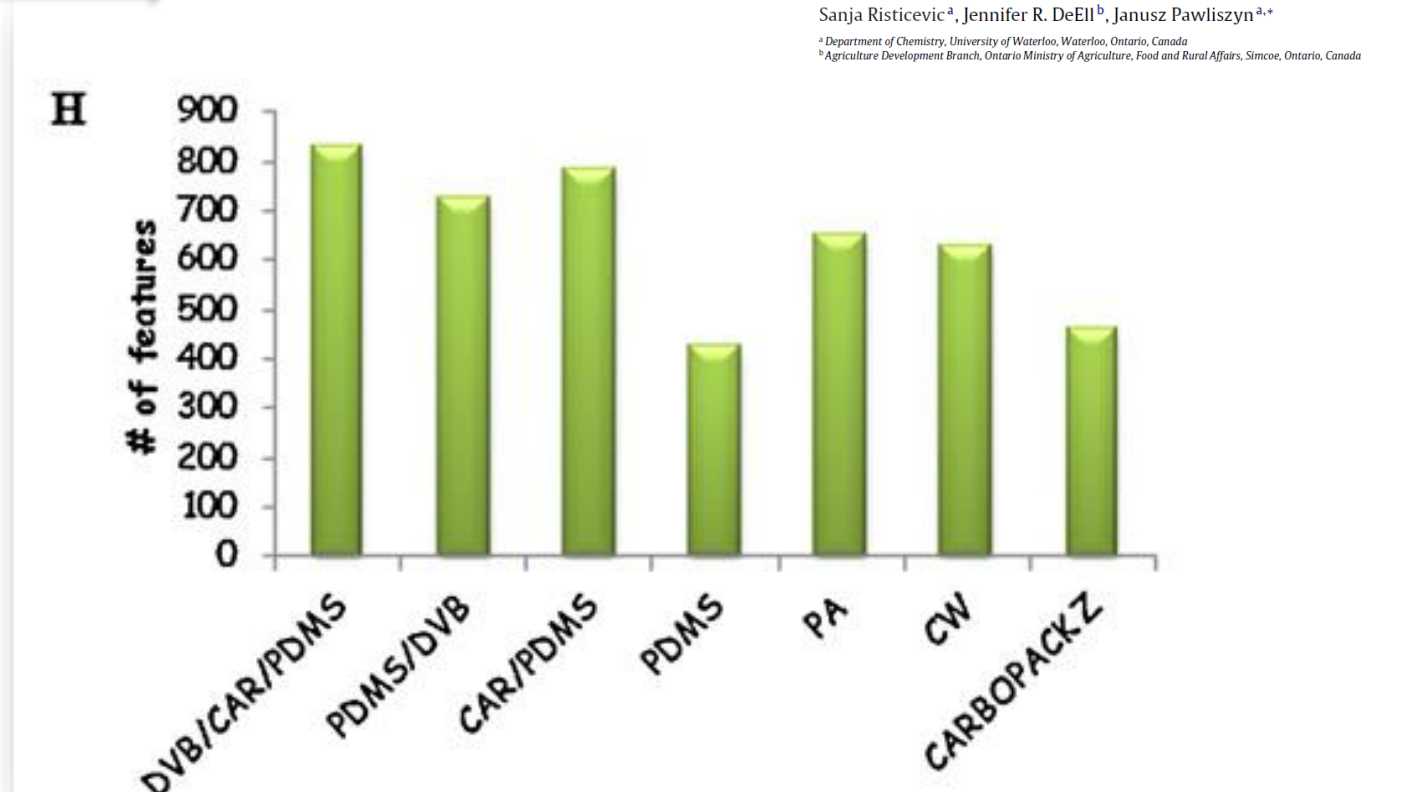


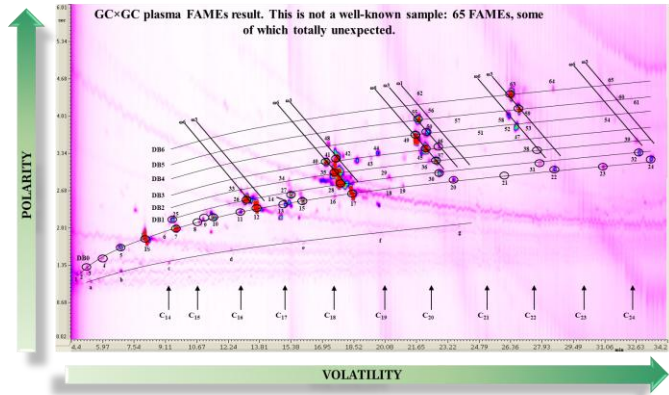
Solid phase microextraction coupled with comprehensive two-dimensional gas chromatography–time-of-flight mass spectrometry for high-resolution metabolite profiling in apples: Implementation of structured separations for optimization of sample preparation procedure in complex samples[☆]

Sanja Risticvic^a, Jennifer R. DeEll^b, Janusz Pawliszyn^{a,*}

^aDepartment of Chemistry, University of Waterloo, Waterloo, Ontario, Canada

^bAgriculture Development Branch, Ontario Ministry of Agriculture, Food and Rural Affairs, Simcoe, Ontario, Canada





Qualitative

Selectivity of different fibers

Journal of Chromatography A, 1251 (2012) 208–218



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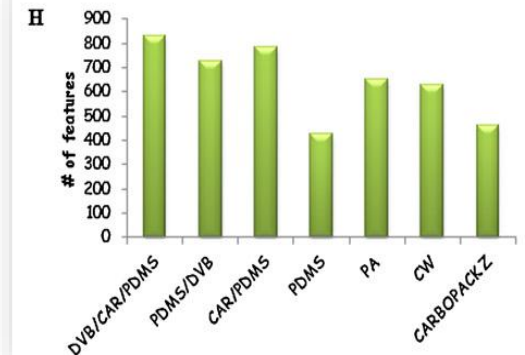
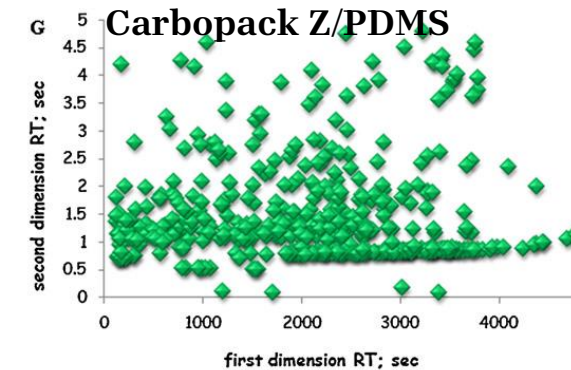
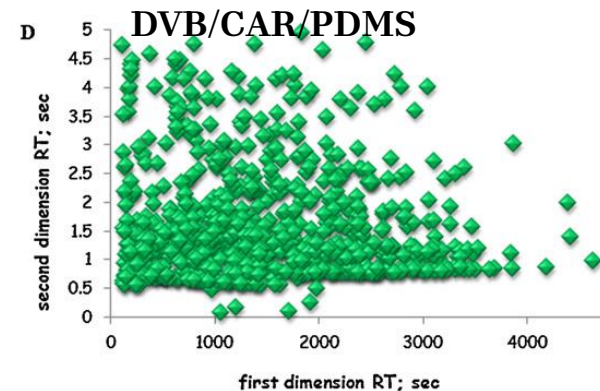
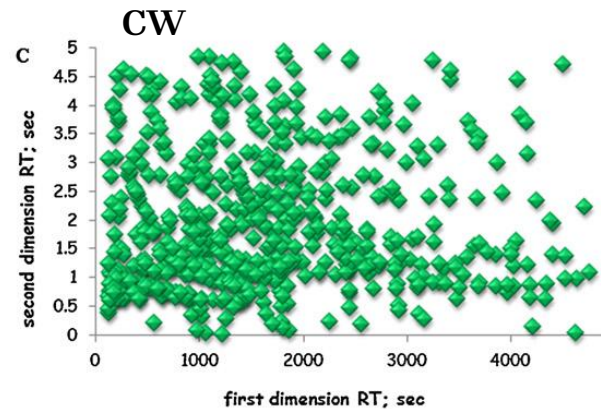
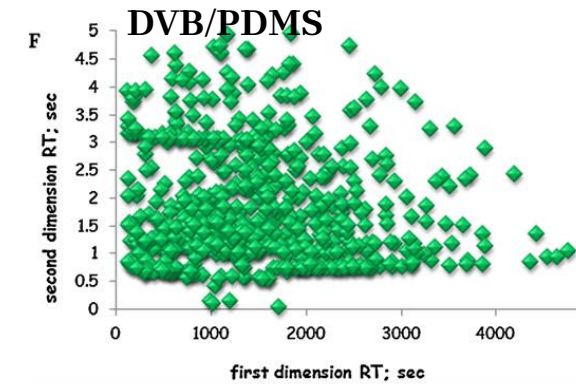
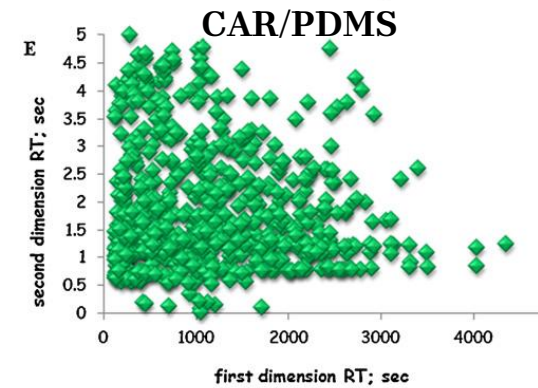
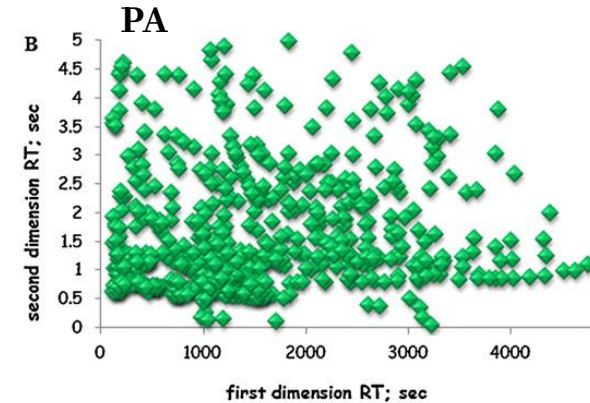
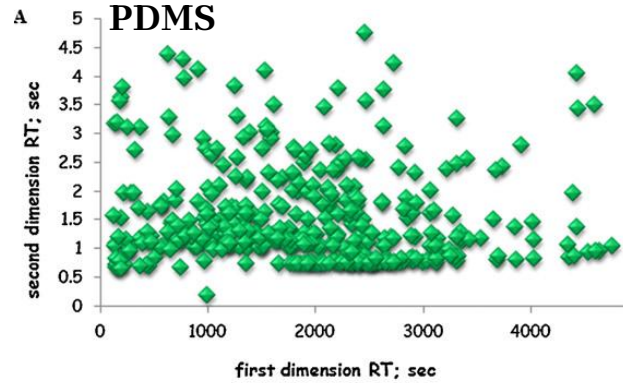
Journal of Chromatography A

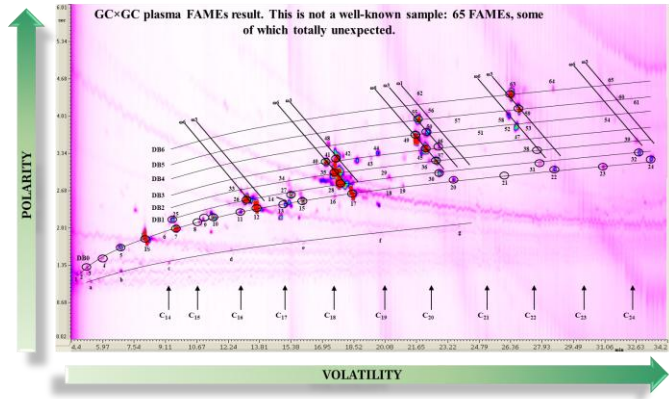
journal homepage: www.elsevier.com/locate/chroma



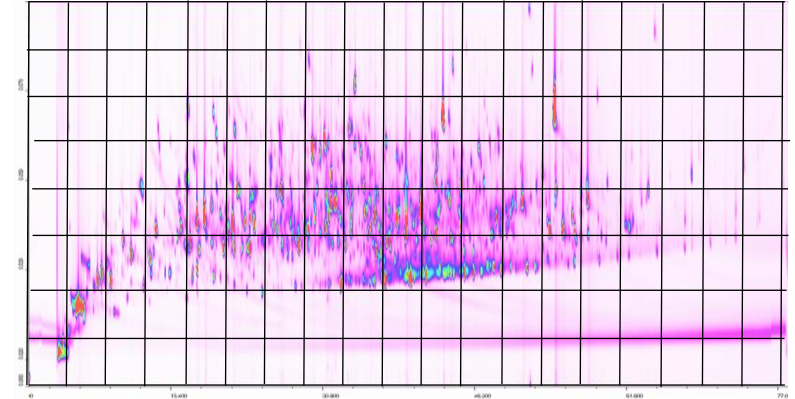
Solid phase microextraction coupled with comprehensive two-dimensional gas chromatography–time-of-flight mass spectrometry for high-resolution metabolite profiling in apples: Implementation of structured separations for optimization of sample preparation procedure in complex samples[☆]

Sanja Risticvic^a, Jennifer R. DeEll^b, Janusz Pawliszyn^{a,*}





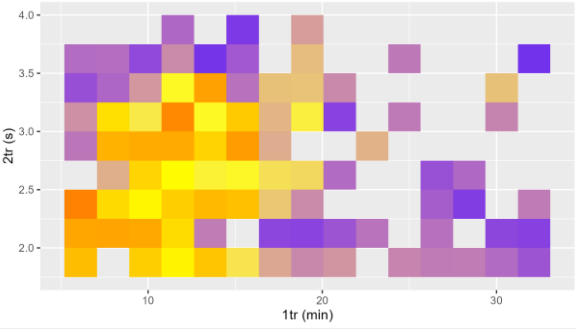
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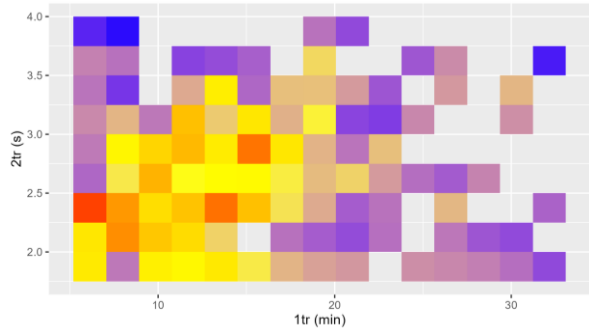
TIME PROFILE

Area

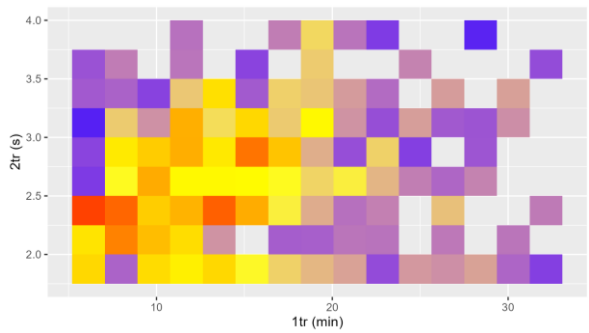
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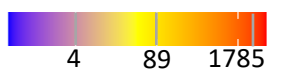
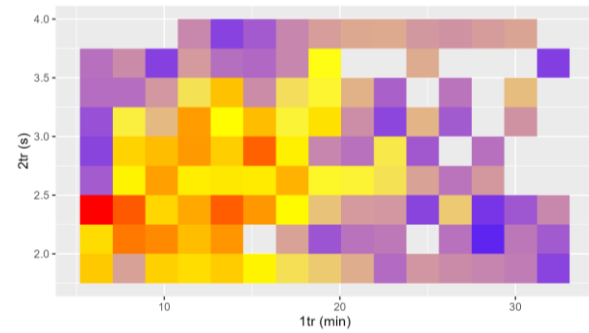
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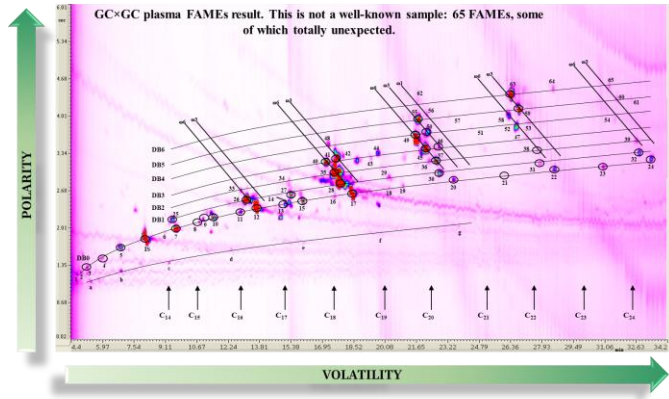
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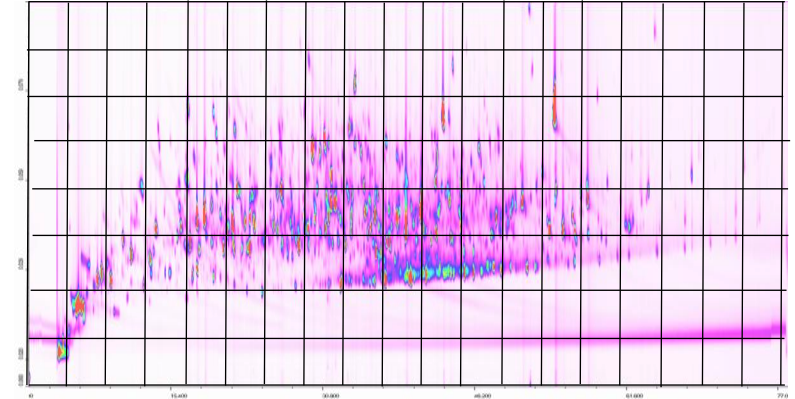
50 min



Chocolate aroma profile

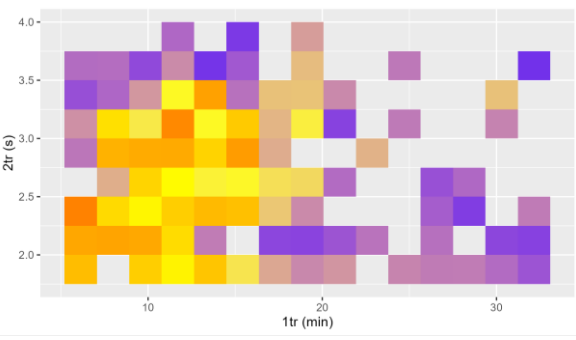


Quantitative

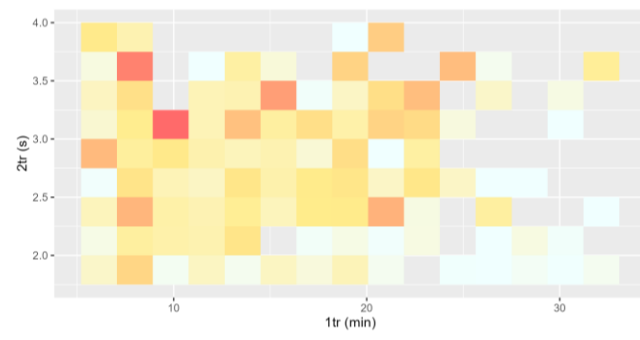


TIME PROFILE

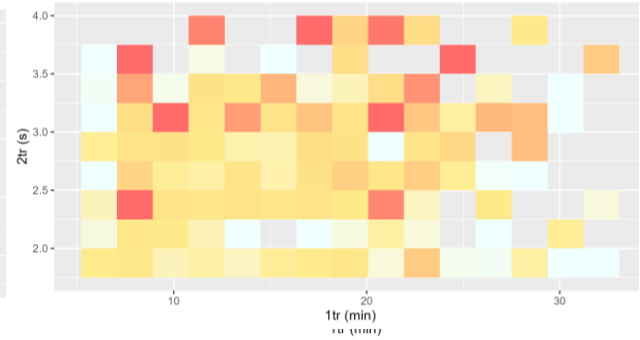
Area
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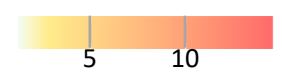
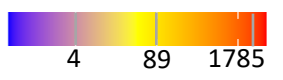
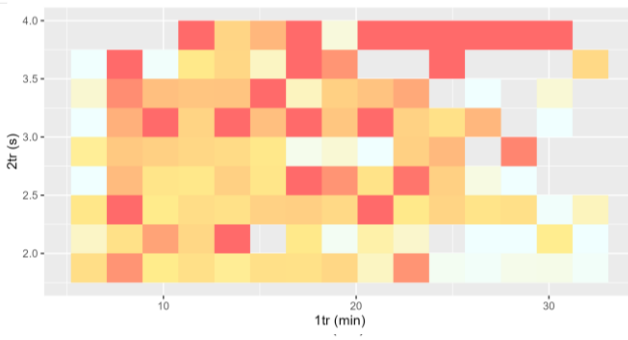
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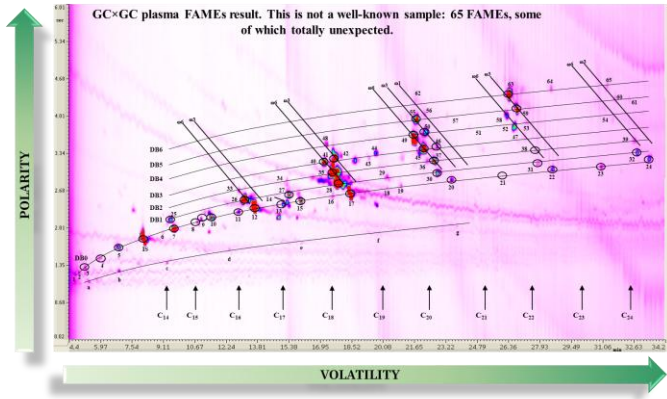
35 min



50 min



“improvement plots”



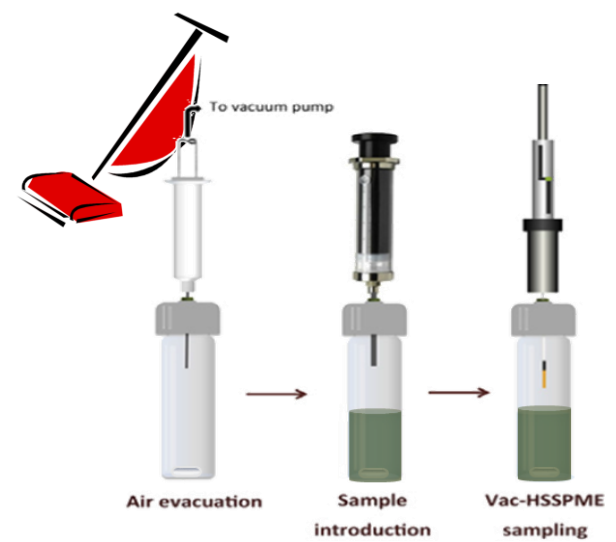
Selectivity of different techniques

HS-SPME

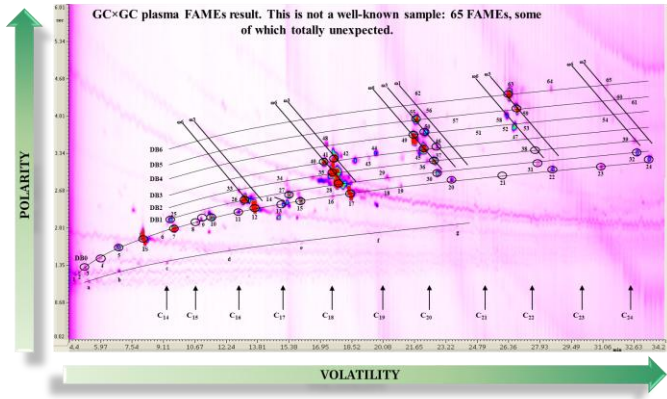
Vac-HS-SPME

HiSorb®

1cm fiber 100µm :
0.6 µL



~10.5mm x 0.8 mm:
~ 63 µL



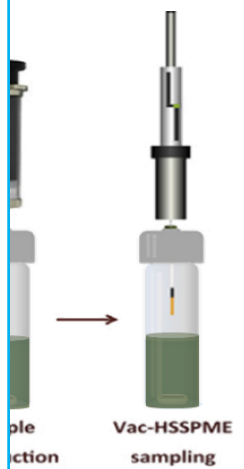
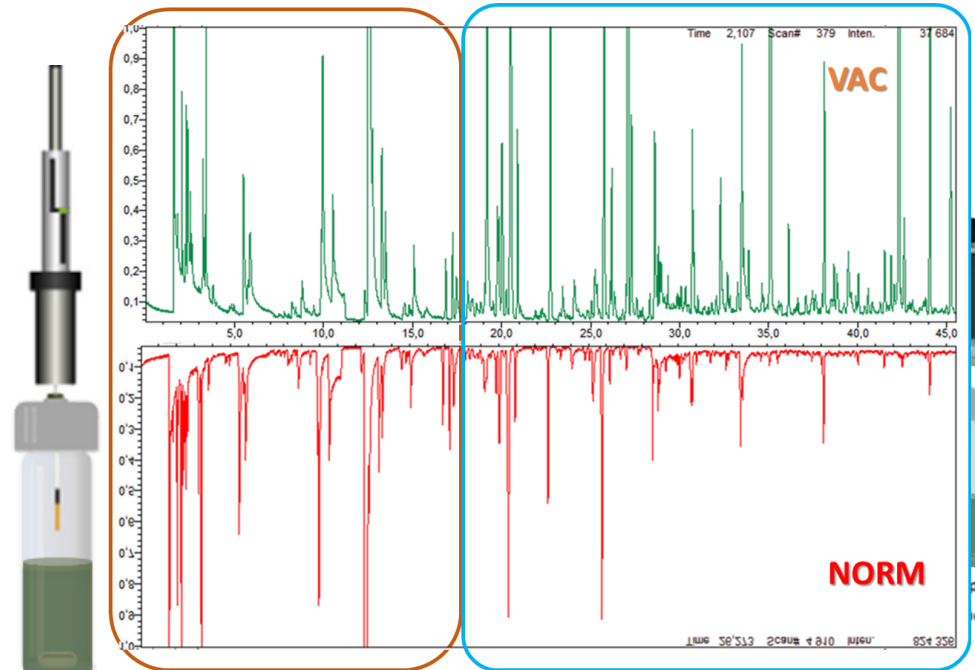
Selectivity of different techniques

HS-SPME

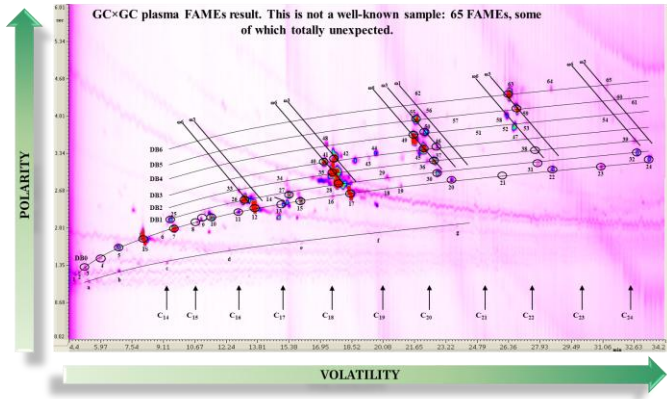
Vac-HS-SPME

HiSorb®

1cm fiber 100µm :
0.6 µL



~10.5mm x 0.8 mm:
~ 63 µL



Selectivity of different techniques

HS-SPME

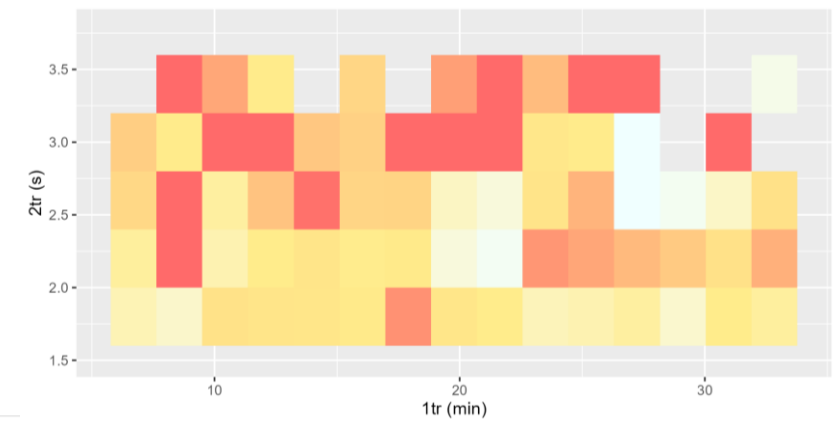
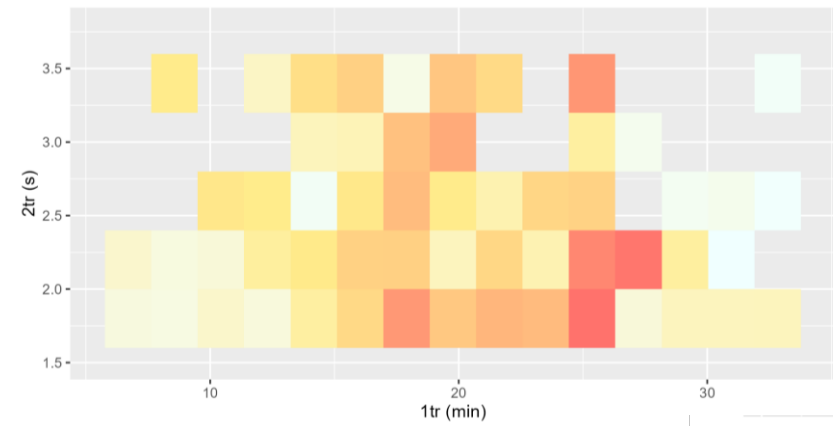
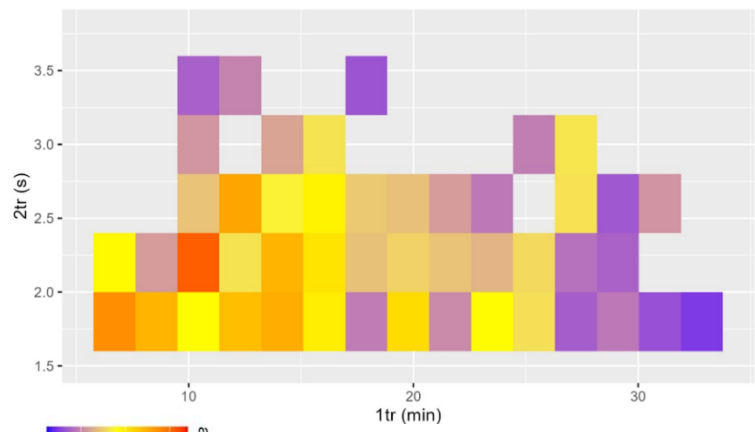
Vac-HS-SPME

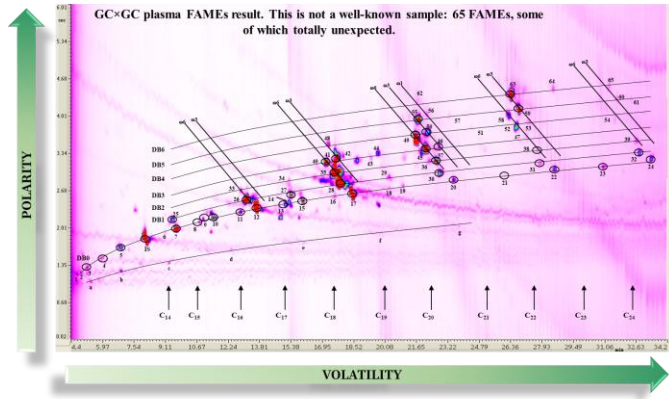
HiSorb®

Vac-HS-SPME/HS-SPME

HiSorb/HS-SPME

R-HS-SPME



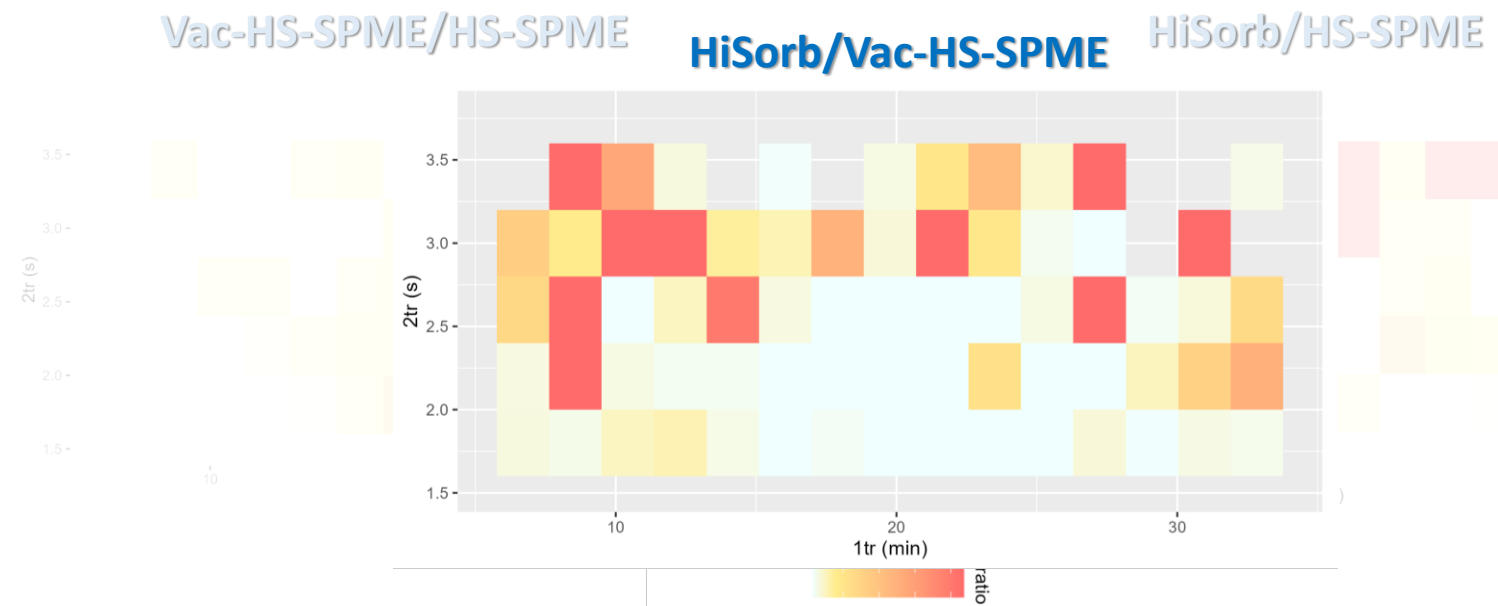
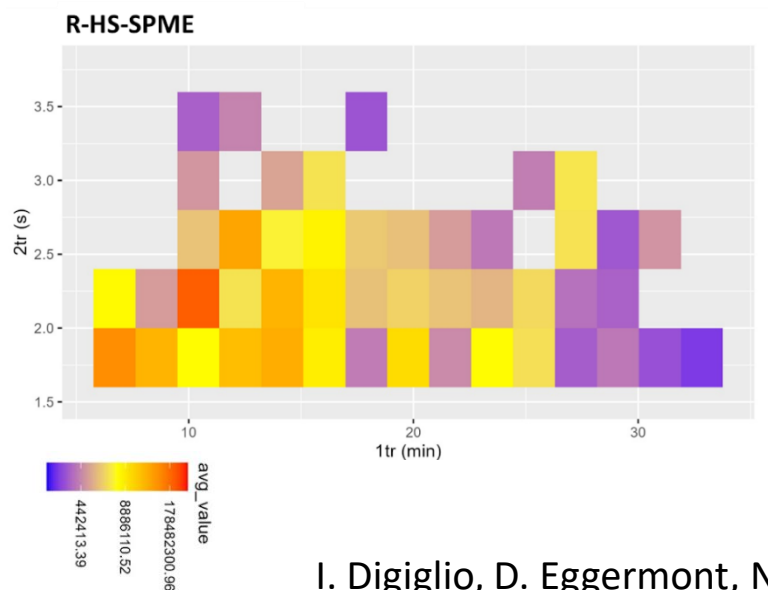


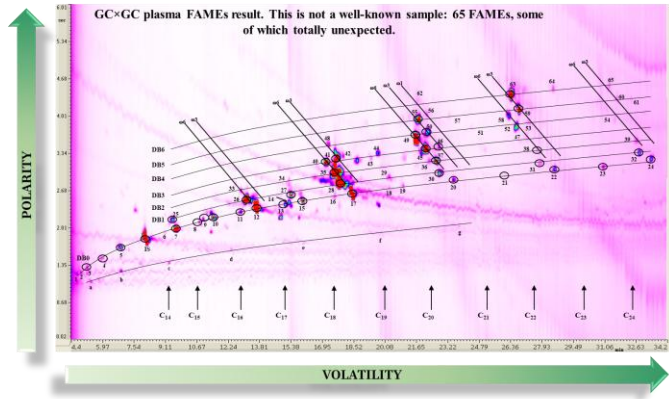
Selectivity of different techniques

HS-SPME

Vacuum-assisted extraction
(Vac)-HS-SPME

HiSorb®





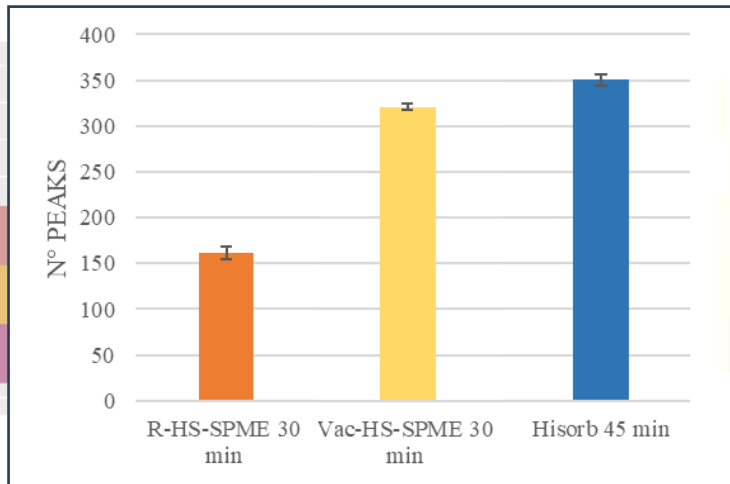
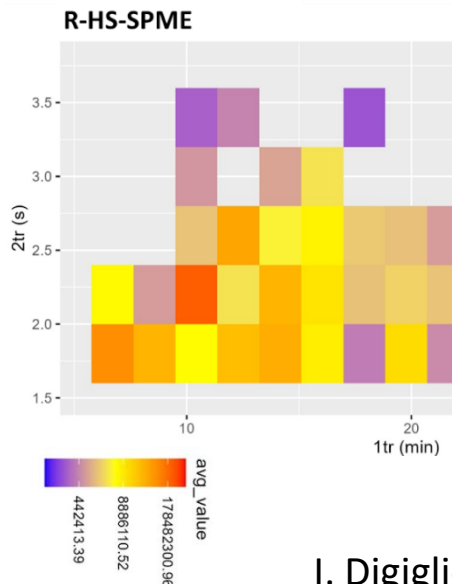
HS-SPME



Selectivity of different techniques

**Vacuum-assisted extraction
(Vac)-HS-SPME**

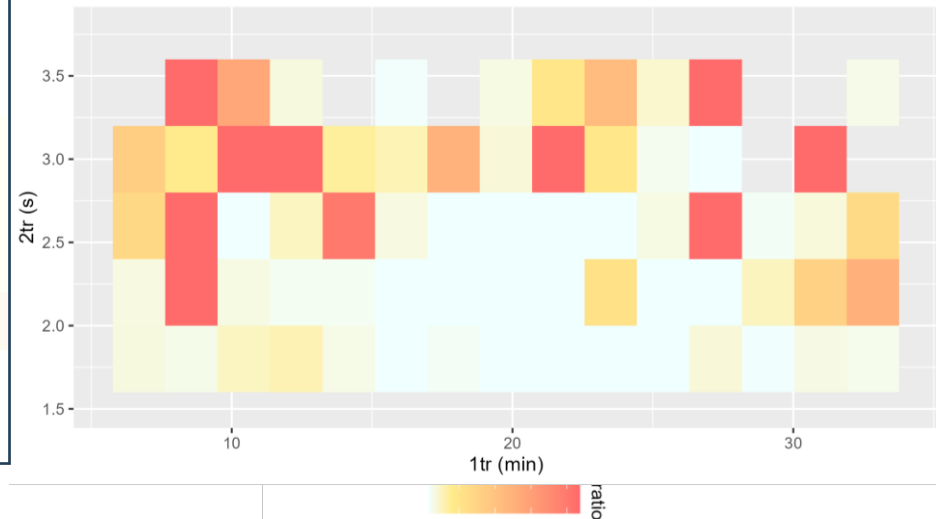
HiSorb®

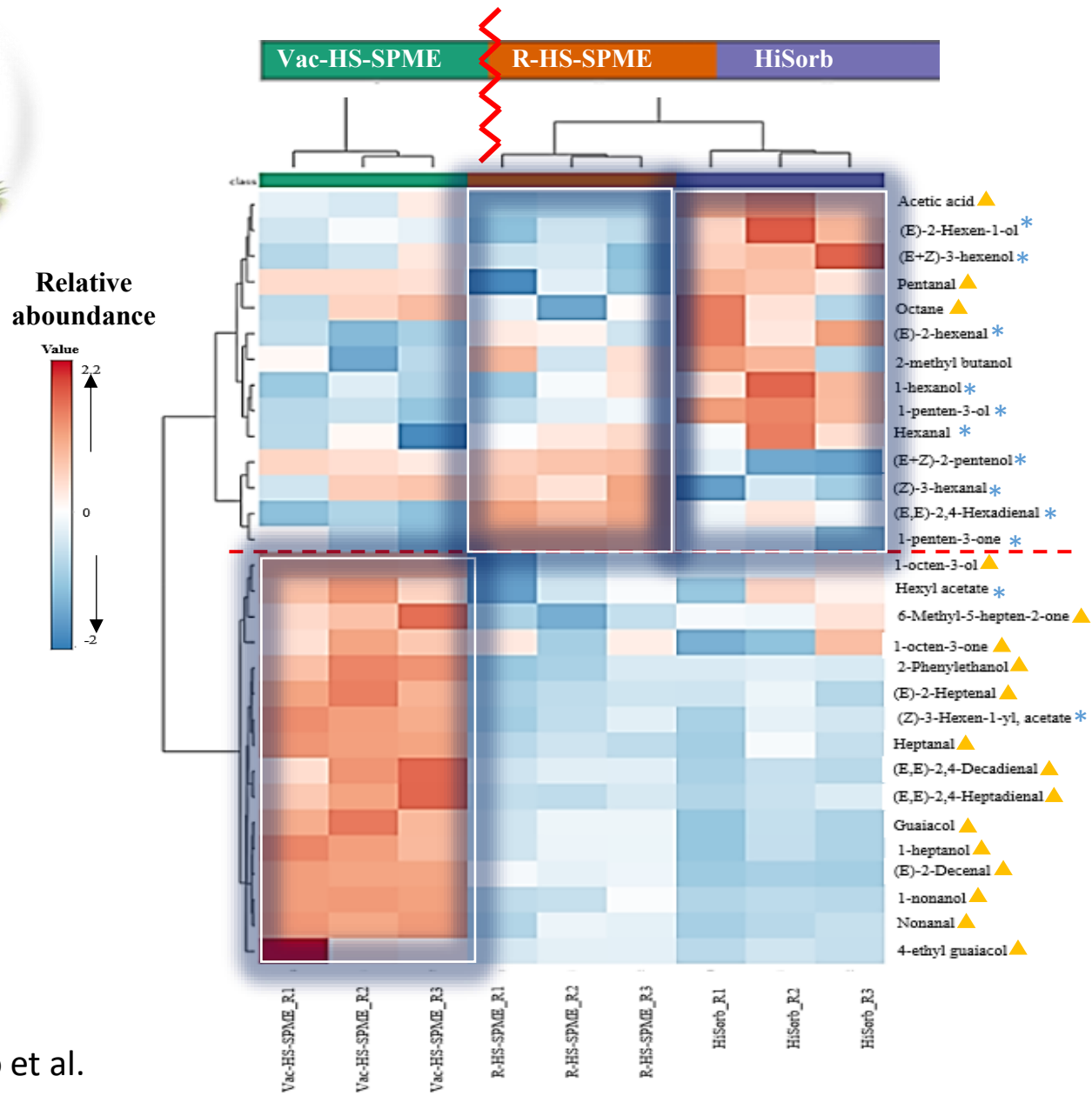


Vac-HS-SPME/HS-SPME

HiSorb/Vac-HS-SPME

HiSorb/HS-SPME





Selectivity of different techniques

Higher volatility



Lipoxygenase (LOX) Pathway *
Series of biochemical reactions that lead to the formation of C5–C6 aldehydes, alcohols, and their corresponding esters, which are responsible for green and fruity aroma notes.

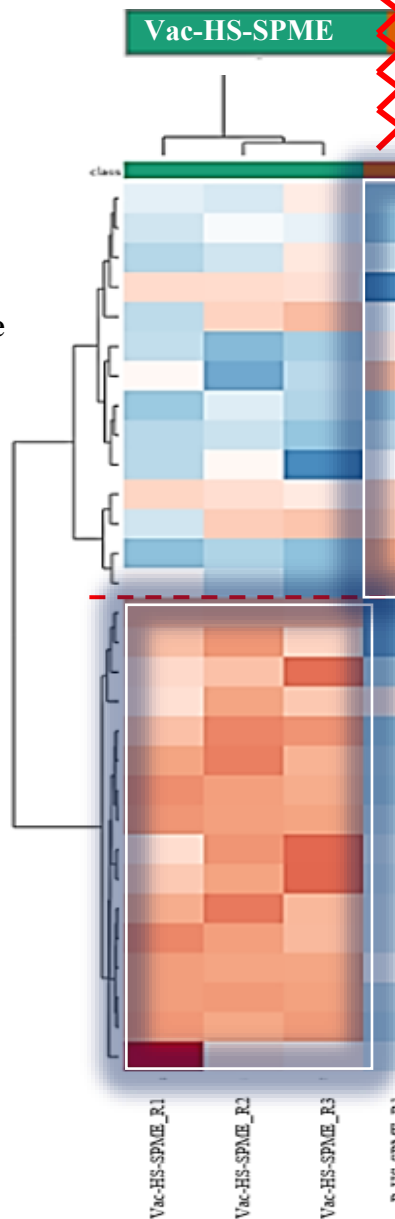
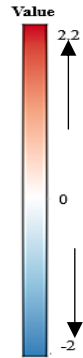
Lower volatility



C7-C12 VOCs ▲
Compounds typically associated with oxidative and microbiological defects, often linked to poor quality or improper preservation of the raw material and inadequate storage conditions.



Relative abundance



Vac-HS-SPME

Enhancing headspace sorbent analysis for target analysis in food by vacuum assisted headspace and/or multi-cumulative trapping

Damien Eggermont, Steven Mascrez, Giorgia Purcaro
Gembloux Agro-Bio Tech, University of Liège, 5030 Gembloux, Belgium

INTRODUCTION

Headspace (HS) sorbent analysis refers to the techniques used to sample HS using a sorbent with varying volumes and surface significant enrichment factor of the analytes of interest. Solid-phase microextraction (SPME) is the most well-known and wide fingerprint of the volatiles in the HS. HS-SPME offers significant advantages by minimizing matrix-related interferences, enhancing its lifespan of both the extraction material and the analytical instrument. Nevertheless, in HS-SPME, the extraction of less volatile compounds is limited by kinetic and thermodynamic limitations. To improve the overall extraction, several strategies can be used. Stirring the sample in while adjusting the extraction temperature impacts both the kinetics and thermodynamics, but it can be critical for the creation of alternatives are the use of reduced pressure, called vacuum-assisted HS extraction (Vac-HS-SPME), which significantly favors the method thus increasing the extraction of the less volatile, and the use of multiple extractions from the same vial, which allows for first extraction then modulate the equilibrium and favors the extraction of the less volatile. The two fractions are desorbed and trapped together in MS or GCxGC-MS.



Vac-HS-SPME



Fig. 1: Scheme of Vac-HS-SPME extraction

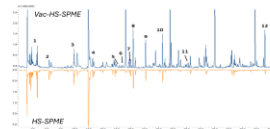


Fig. 2: Chromatogram obtained during Vac-HS-SPME and regular HS-SPME.

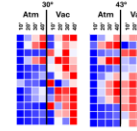


Fig. 3: Heatmaps showing the extraction response obtained using regular and Vac-HS-SPME at 30 and 43 °C for different extraction times.

A special cap, provided by ExtraTECH (Greece) is needed to guarantee the sealing and maintain the vacuum in the vial (Fig.1).

For the most volatile compounds, the same performance can be observed between regular and Vac-HS-SPME (Fig. 2). The kinetics of these compounds is rapid; thus, the effect of using reduced pressure conditions is limited or even non-existent. For the rest of the compounds, the effect of VAC and T° is instead synergistic, significantly improving the extraction efficiency at an earlier sampling time (Fig. 3).

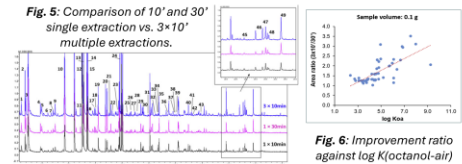
MCT-HS-SPME



Fig. 4: Picture of MCT-HS-SPME-GC(JGC-MS) instrument

In single-vial multi-cumulative trapping (SV-MCT-)HS-SPME extraction, the same vial is extracted several times, and the extracts are collected on a focusing trap that rapidly desorbs the entire shot at once into the GC system (Fig.4).

Extraction yield is higher for MCT extraction compared to an equivalent but single extraction time (Fig. 5). This effect is more pronounced for VOCs with less affinity to air compared to oil (high K_{ow}) (Fig. 6).



MCT-HS-SPME increases the extraction yield of the less volatile compounds.

Vac-MCT-HS-SPME

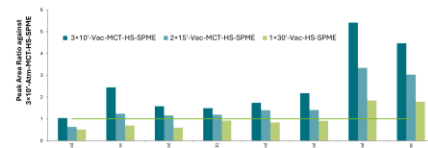


Fig. 7: Ratio of improvement of different Vac-MCT-HS-SPME extractions (modalities in legend) of 0.1 g of olive oil extracted at 43 °C

When Vac and MCT extractions are combined, a synergistic effect is observed, resulting in increased extraction yield, especially for low-volatile compounds (Fig. 7). Notably, long Vac extractions show close to similar performances to Atm-MCT but significantly less than Vac-MCT for the same total time.

APPLICATION: Extra-Virgin / Virgin Olive Oil Classification

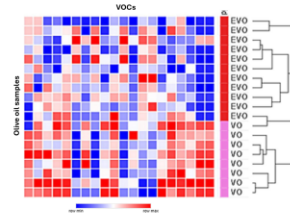


Fig. 8: Heatmap of selected VOCs of EVO/VO samples analysed by 3x10' Vac-MCT-HS-SPME (0.1 g of olive oil extracted at 43 °C)

Eighteen olive oil samples – 10 Extra Virgin Oil (EVO) and 8 Virgin Oil (VO) – were analysed by Vac-MCT-HS-SPME. Hierarchical cluster analysis based on distance of Pearson's correlation of selected VOCs allows to correctly cluster EVO from VO (Fig. 8).

Conclusion

The use of vacuum or multi-cumulative trapping separately increase the extraction of low-volatile compounds. Their combination benefits from the advantages of both techniques to give a significantly enhanced extraction. This synergistic approach has been successfully tested for the classification of extra virgin and virgin olive oils in the context of commercial quality assessment.

REFERENCES

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Mascrez and Purcaro, J Sep Sci 2020;43:1934–1941;
Mascrez et al., Analytica Chimica Acta 1103 (2020) 106e114

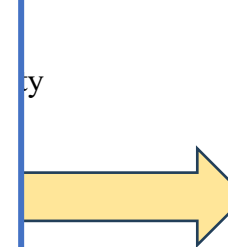
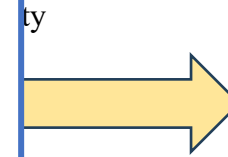
ACKNOWLEDGMENTS

The authors would like to thank Markes, ExtraTECH, Sepsolve, Shimadzu and Supelco for their support. This work is supported by ACESSS (Academic Center of Excellence for Separation Science and Sensing).

of different techniques

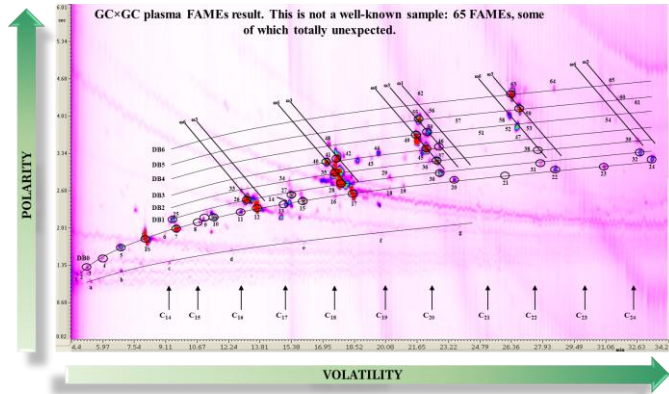
Lipoxygenase (LOX) Pathway *

Series of biochemical reactions that lead to the formation of C5–C6 aldehydes, alcohols, and their corresponding esters, which are responsible for green and fruity aroma notes.

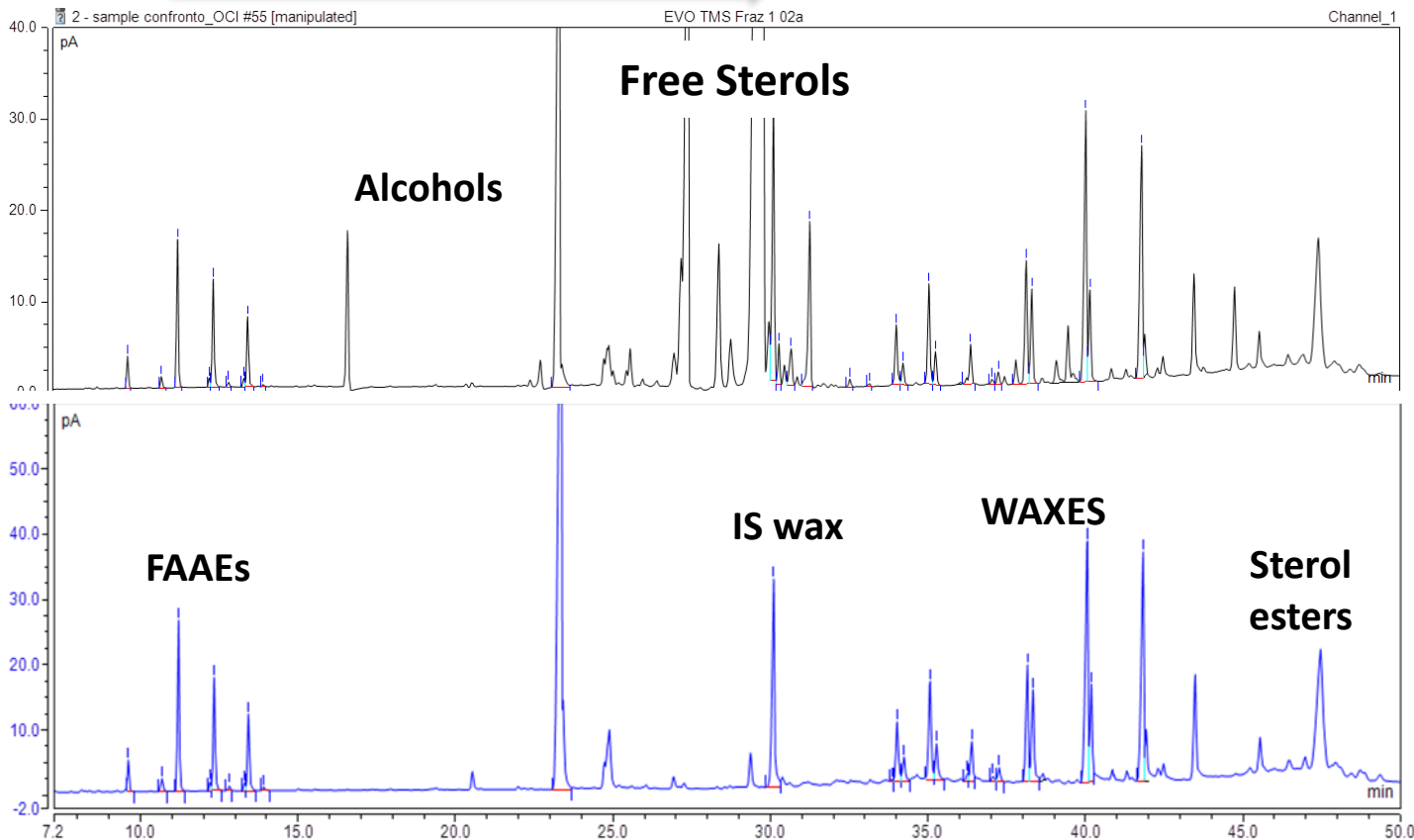


C7-C12 VOCs ▲

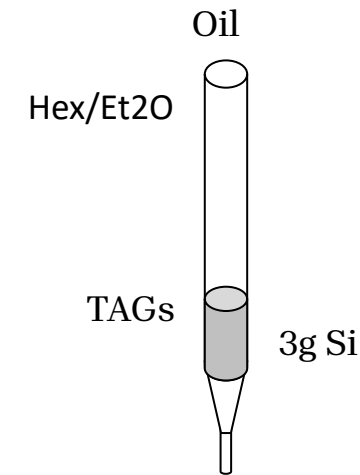
Compounds typically associated with oxidative and microbiological defects, often linked to poor quality or improper preservation of the raw material and inadequate storage conditions.



➤ Merge workflows

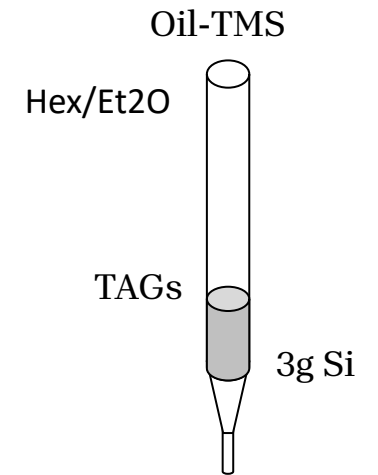


Silylation before loading the oil



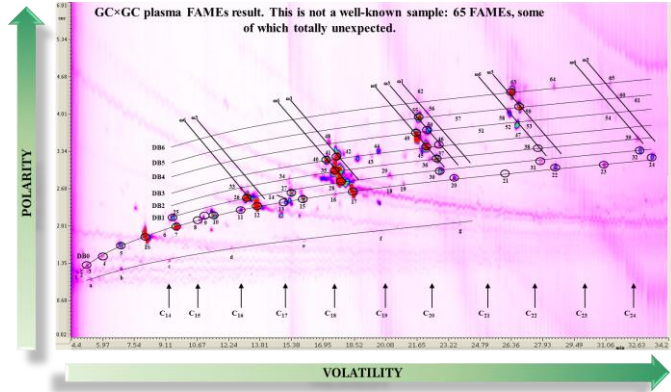
FAAEs + Waxes + Esterified Sterols

Official method

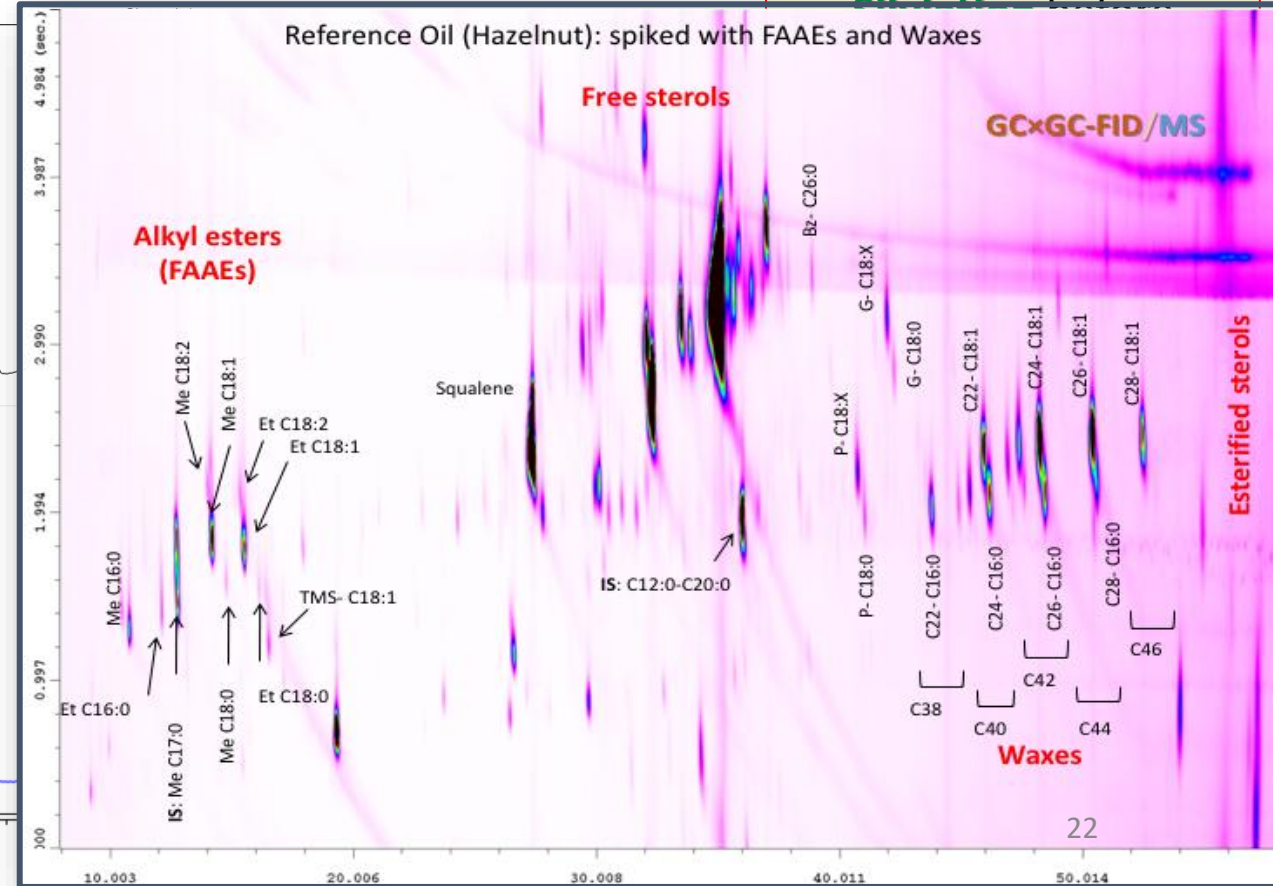
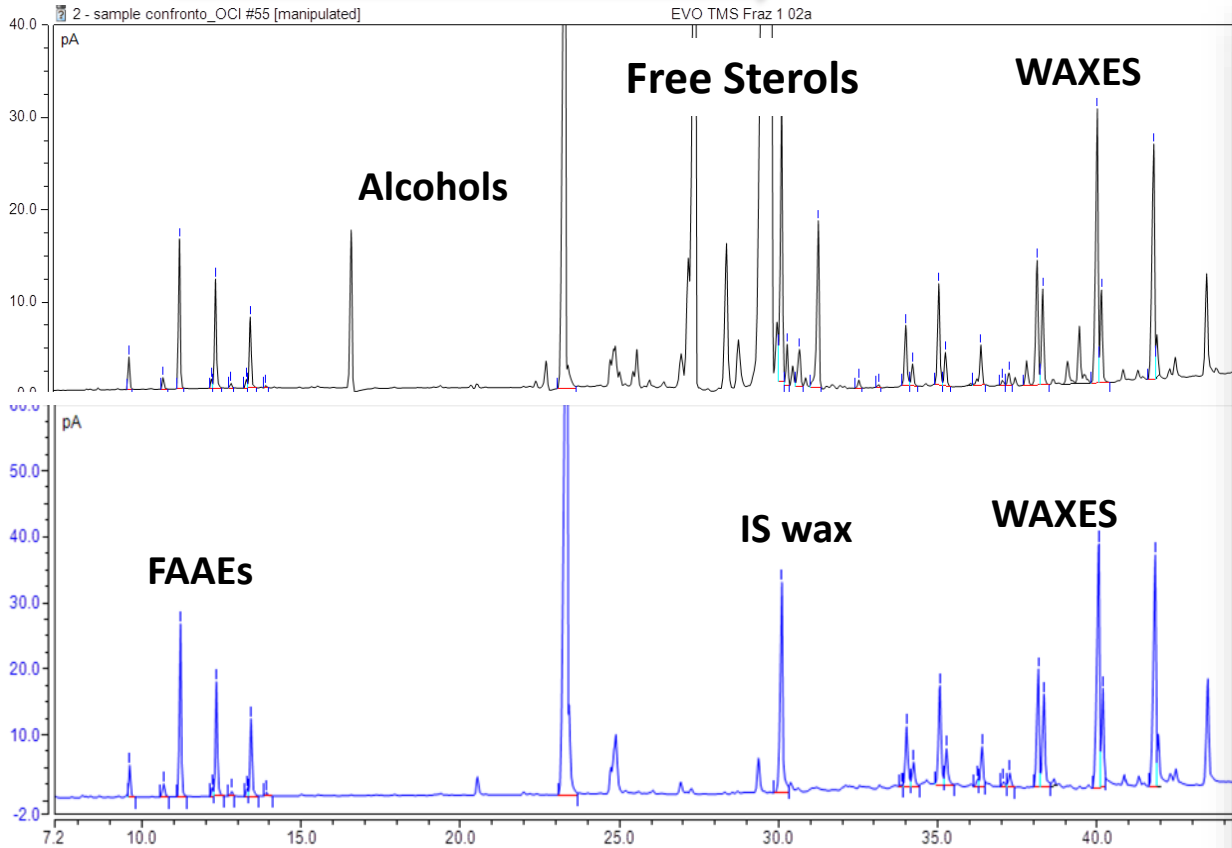


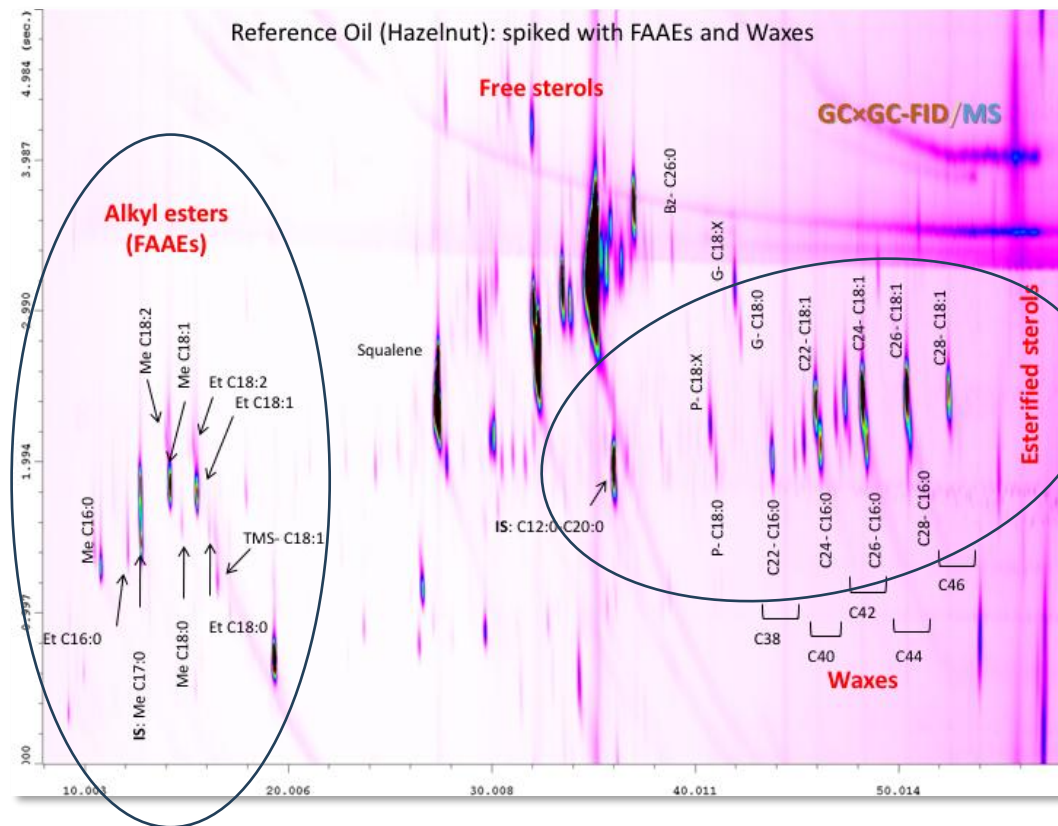
FAAEs + Waxes + Esterified Sterols

Alcohols, Sterols-TMS



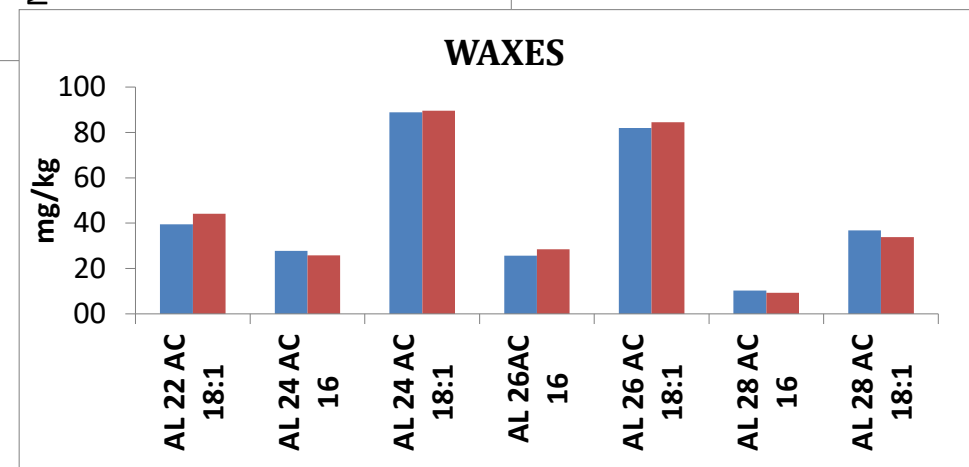
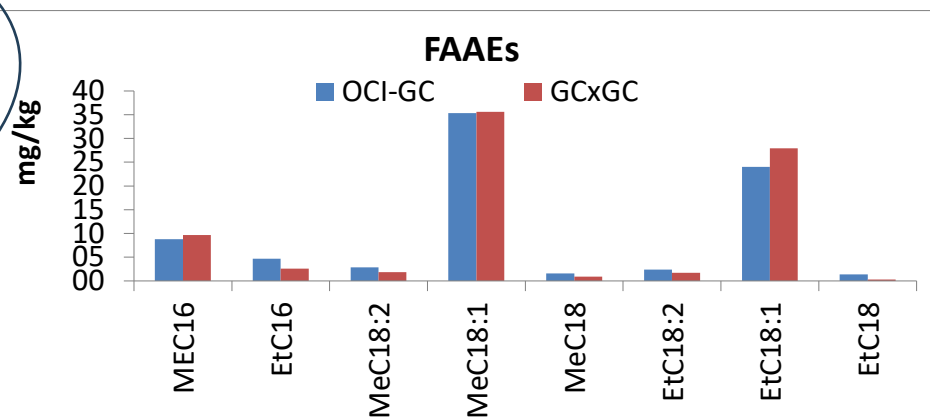
➤ Merge workflows





comparison with the Official method

(FAAEs and Wax) GC×GC-FID/MS

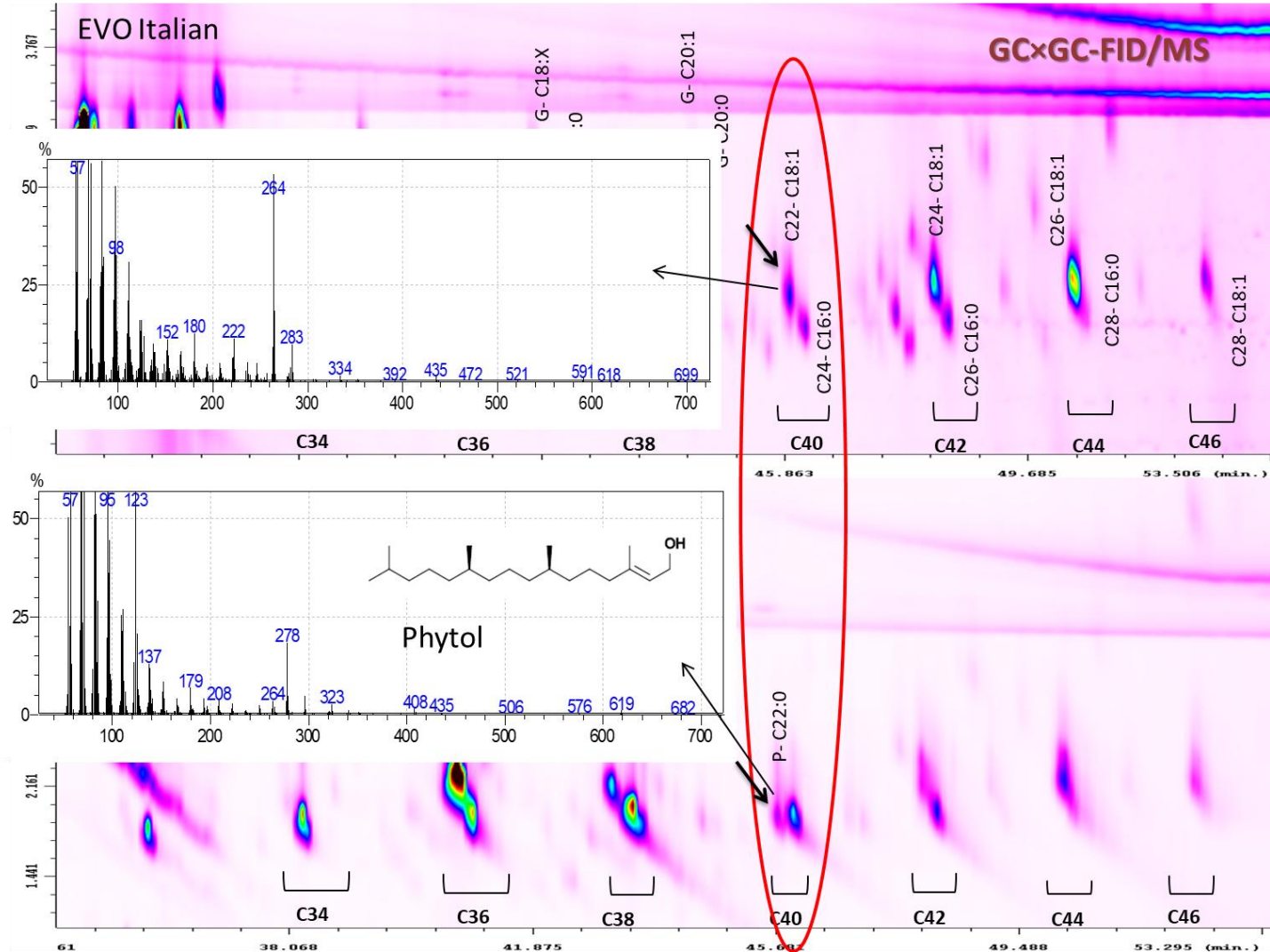
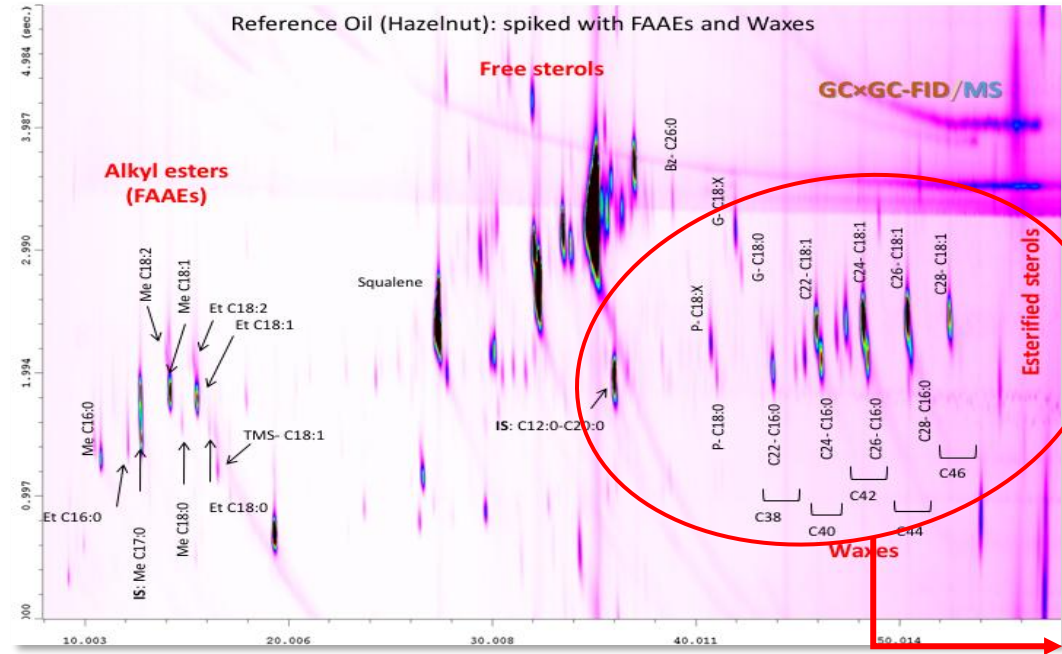


G. Purcaro*, L. Barp, L. Conte, *J. Sep. Sci.* 38 (2015) 2278-2285

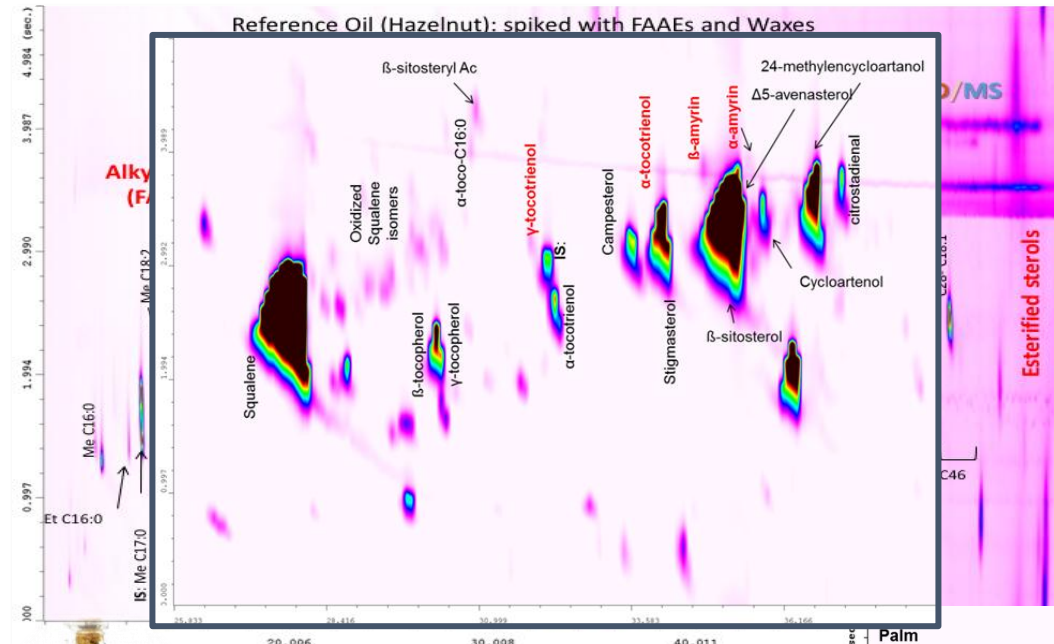
G. Purcaro*, L. Barp, M. Beccaria, L. Conte, *Anal Bioanal Chem* 407 (1) (2015) 309-319.

L. Barp, F. A. Franchina, G. Purcaro*, P. Q. Tranchida, L. Mondello, *Talanta* 165 (2017) 598-603.

Support data interpretation



More information



It was clearly highlighted the presence of the following sterols, not present in EVO

α- and γ-tocotrienol

α- and β-amyirin.

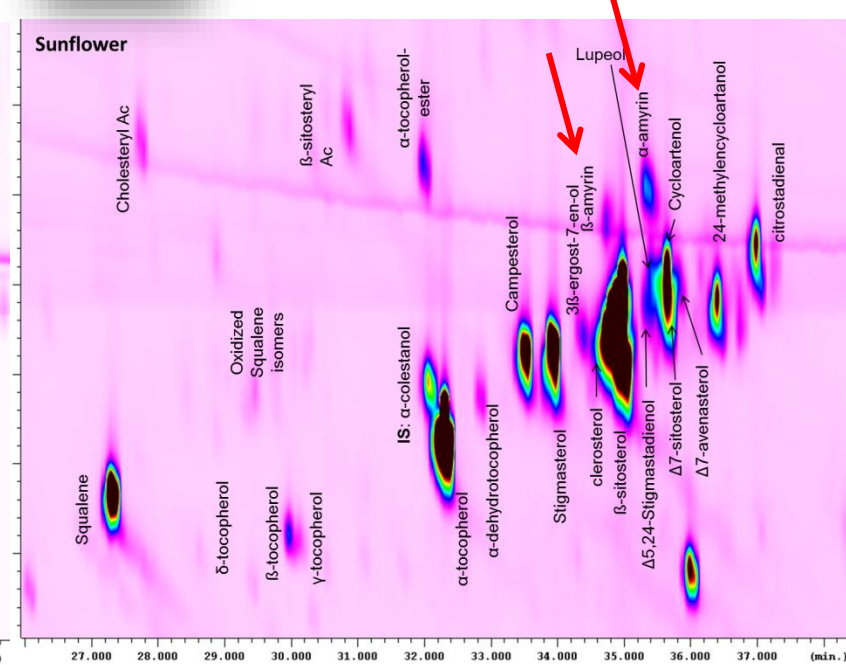
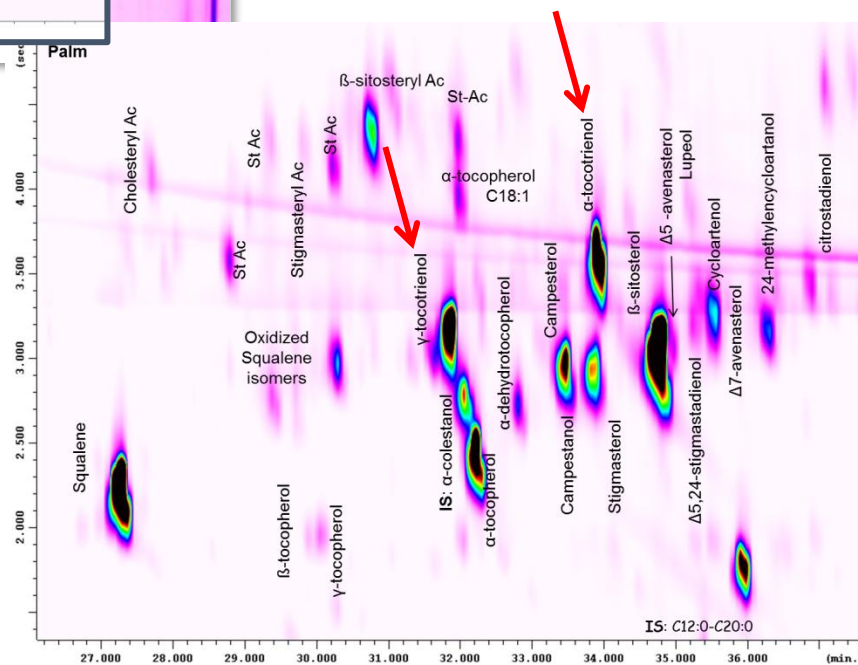


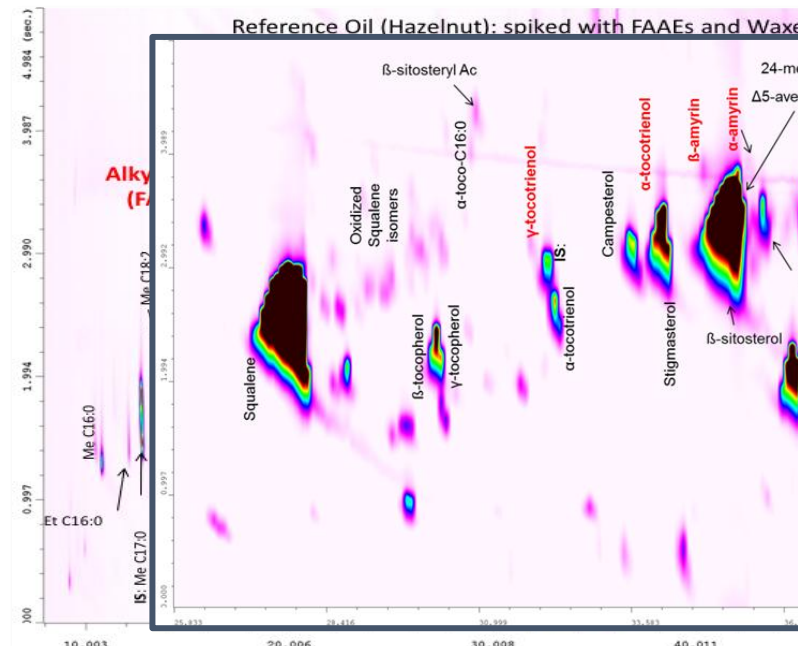
Palm

Sunflower



More information





More information

Development of an Innovative Analytical Method for the Analysis of the Sterol Fraction in Olive Oil to Detect Food Fraud

Donatella Ferrara^{1,2}; Nicola Ruini³; Marco Beccaria³; Chiara E. Cordero¹; Giorgia Purcaro²

¹ University of Turin, Department of science and drug technology, Torino, Italy
² Gembloux Agro-Bio Tech, University of Liège, 5030 Gembloux, Belgium
³ University of Ferrara, Department of Chemical, Pharmaceutical, and Agricultural Sciences

1 INTRODUCTION

Olive oil is one of the most valued edible oils, both for its nutritional profile and its cultural and economic significance. For this reason, it is subject to fraud, though mixing it with cheaper oils. As a consequence, olive oil is subject to strict international regulations to ensure authenticity and quality. Sterol profiles can reveal whether the oil has been adulterated with cheaper vegetable oils and help distinguish between different origins. The International Olive Council (IOC) has established an official method to characterize olive oil and detect potential adulterations. Several key steps: initial saponification followed by liquid-liquid extraction and isolation of the fraction using thin-layer chromatography (TLC) analysis is carried out with gas chromatography coupled with a flame ionization detector (GC-FID). To streamline and accelerate this process, a Microwave-Assisted Saponification and Extraction (MASE) method, with purification achieved through solid-phase extraction (SPE) prior to analysis in 2D-dimensional chromatography (2D GC). This study focuses on improving the greenness of the method by reducing solvent volumes, increasing sample throughput, and enhancing the separation of sterols from interfering compounds (Des A, B, and C).

2 MATERIALS & METHODS

IOC METHOD SAPONIFICATION 5 g of olive oil + 50 mL 2M KOH (EtOH/H ₂ O 80:20 v/v)	LIQUID-LIQUID EXTRACTION 3 times using: 80 mL +70 mL +70 mL of ethyl ether	TLC development in Hexane: ethyl ether 65:35 v/v	GC-FID Rxi 5MS 30 × 0.25 mm i.d. × 0.25 μm. Flow: 1.8 mL/min Ramp: 80°C held 1 min. ramp to 160°C at 20°C/min and to 340°C at 5°C/min. FID: 350°C, air 350 mL/min, H ₂ O 35 mL/min, make-up gas 30 mL/min.	TOTAL TIME 7h 40 min
				PROPOSED METHOD MASE 1 g of olive oil + 10 mL 2M KOH (EtOH/H ₂ O 50/50 v/v) + 10 mL Hexane

3 RESULTS & DISCUSSION

MASE
Temperature and Time were the two variables to optimize for the MASE. We used an inscribed central composite design (Figure 1) to explore the temperature in the range of 100°C to 140°C and the time in the range of 10 minutes to 30 minutes. The model incorporated first-order (linear) and quadratic terms for both time and temperature. A response surface methodology (RSM) was applied to model the relationship between residues (response variables) and the two independent variables. The design of the experiment outcome (Figure 2) shows that increasing temperature and time the response (residue) decrease till a certain point. In this case, the optimal condition were found targeting the same residue as the official method. Under this requirement, the chosen conditions were 20 min and 120 °C.

SPE
To optimize the solid-phase extraction procedure, we started from the method developed by Mascres et al. [2], with the goal of reducing solvent volumes and improving the co-elution of sterols with certain interfering compounds, hereafter referred to as Des A, B, and C. The initial step involved replacing hexane with cyclohexane, monitoring how changes in solvent strength of the washing mixture affected elution, and adjusting accordingly (Figure 3).

GREENNESS
Although the greenness scores may indicate limited greenness improvement, some aspects must be considered for a proper interpretation of the results. In particular, the AgreePrep metric shows limitations, since it assigns the same score to markedly different values (i.e. waste 2 L in IOC vs 80 mL in the proposed method).

4 CONCLUSION

The optimized MASE method offers a faster approach to sample preparation for sterol analysis. In this method, two steps that are performed separately in the official protocol are combined, and the replacement of TLC by SPE leads to reduced solvent consumption and shorter processing time.

REFERENCES & ACKNOWLEDGMENTS

[1] International Olive Council (IOC); Madrid, Spain, 2017. [2] Mascres et al. Foods 2021 Feb 18;10(2):445 [3] Mathiason et al. A Rapid Method To Determine Sterol, Erythritol, and Uvaol Concentrations in Olive Oil. Journal of Agricultural Chemistry, 2013. Authors want to thank Milestone, Shimadzu and Access for their support

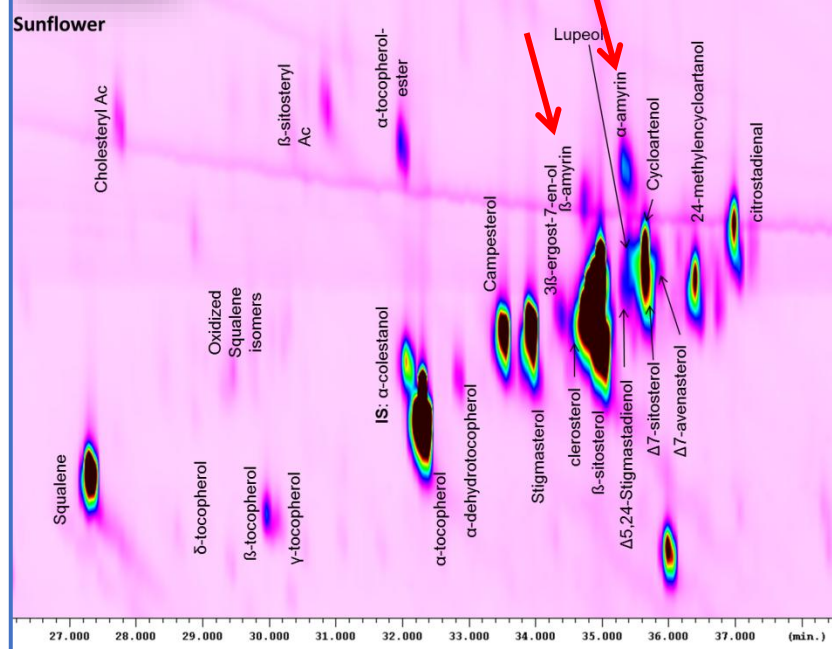
presence of the following sterols, present in EVO

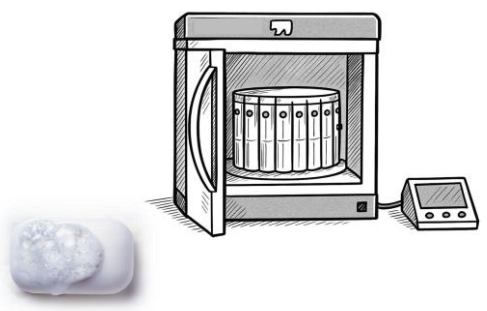
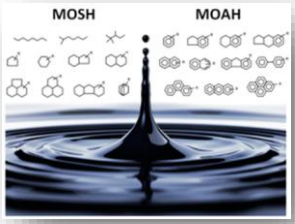


α- and β-amyryrin.

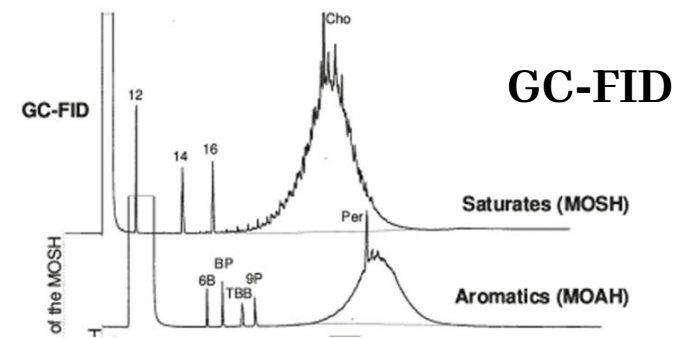
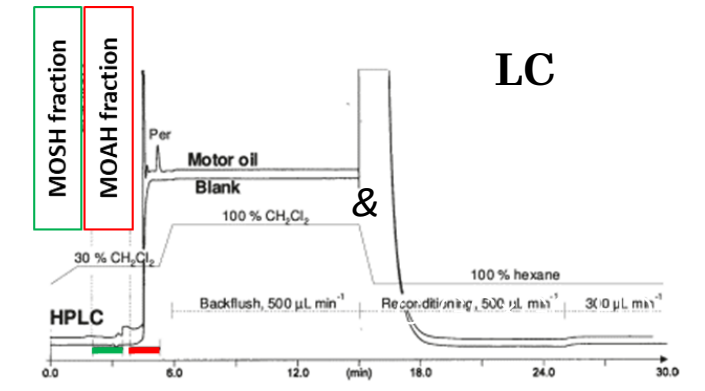
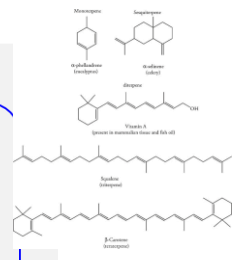
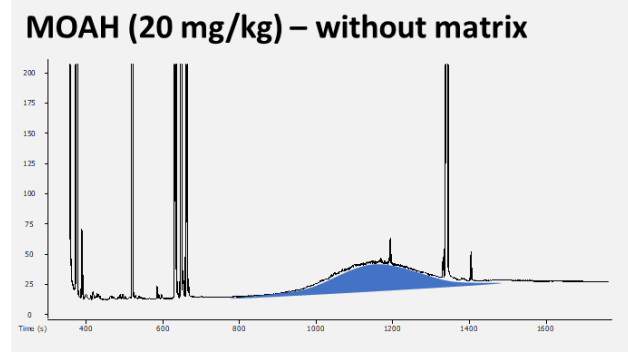
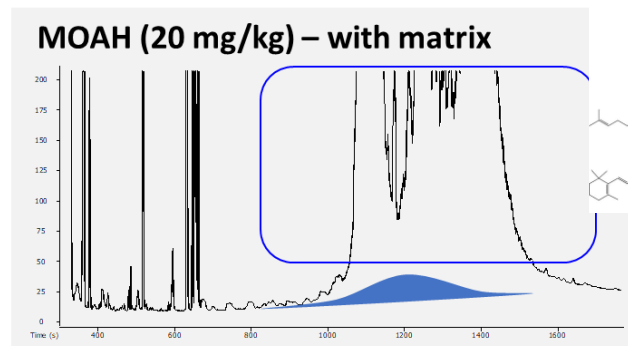


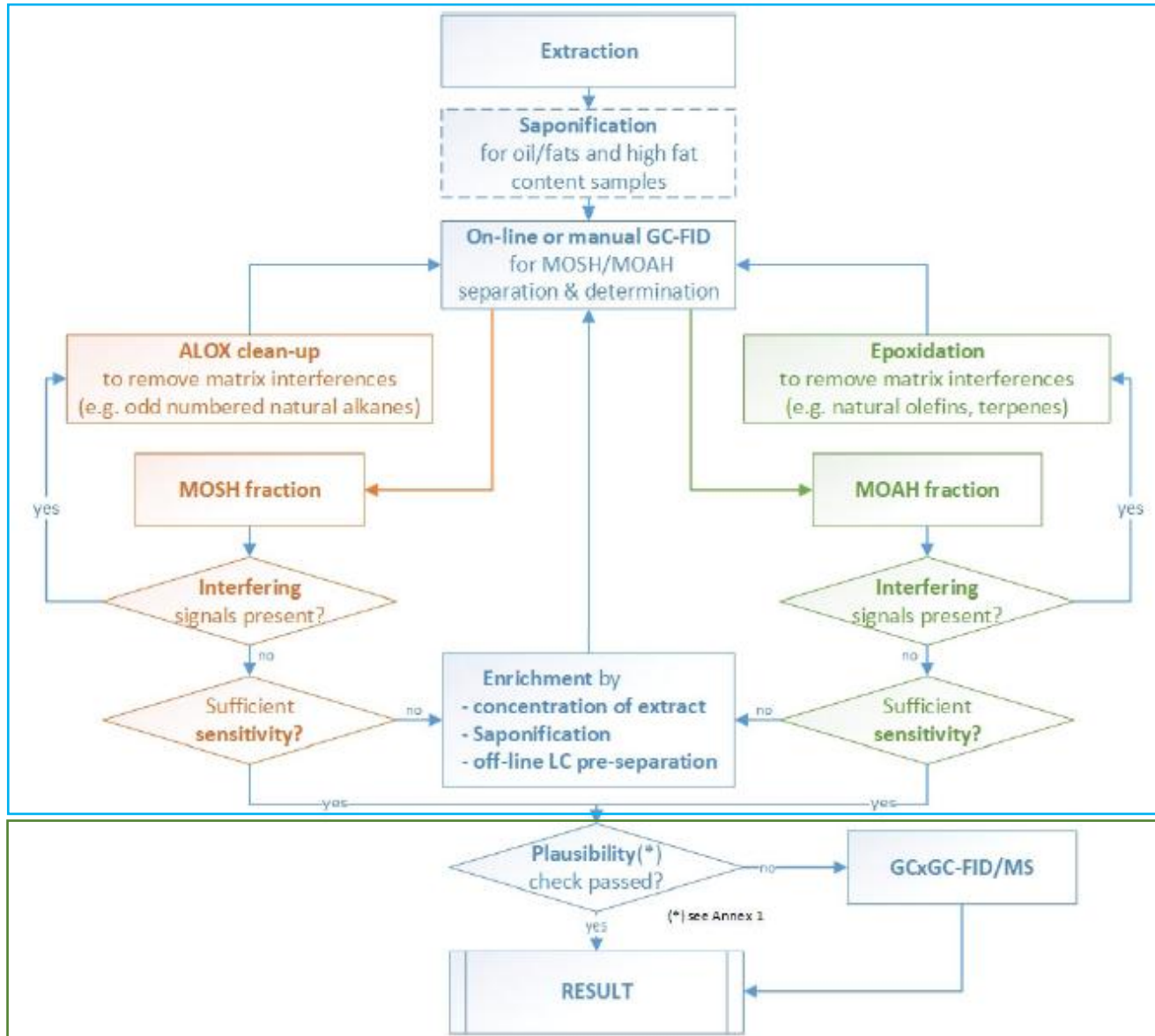
Sunflower





Microwave-assisted saponification and extraction (MASE) method





Sample Preparation

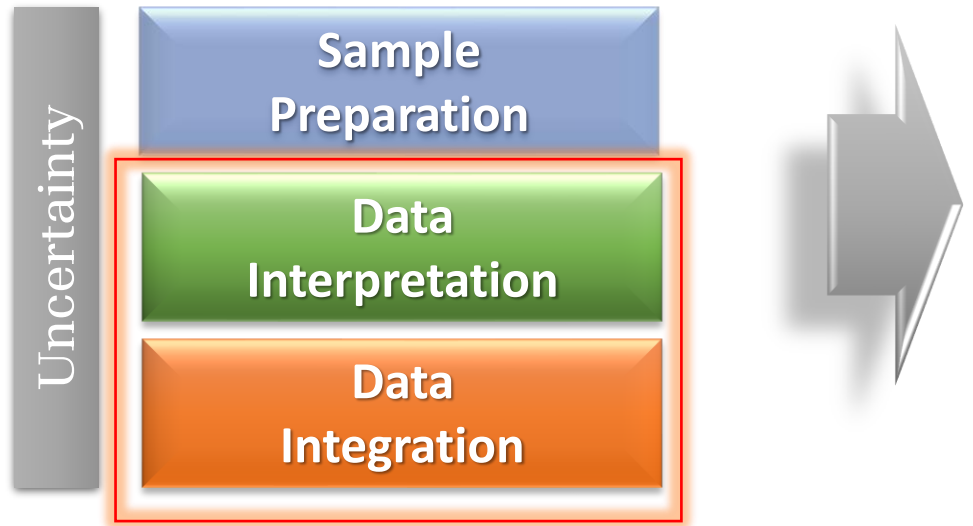


Data Interpretation

Data Integration

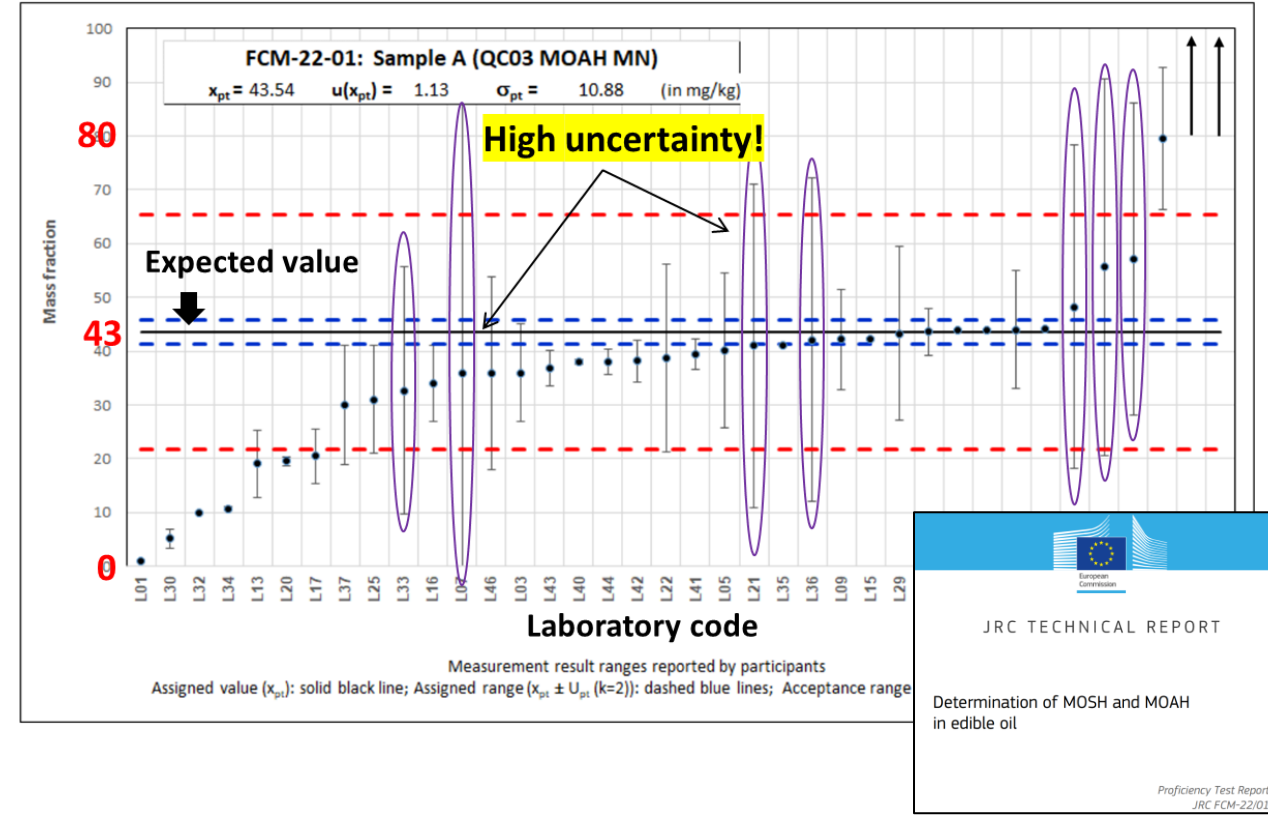
✓ Need for matrix-tailored sample prep protocols

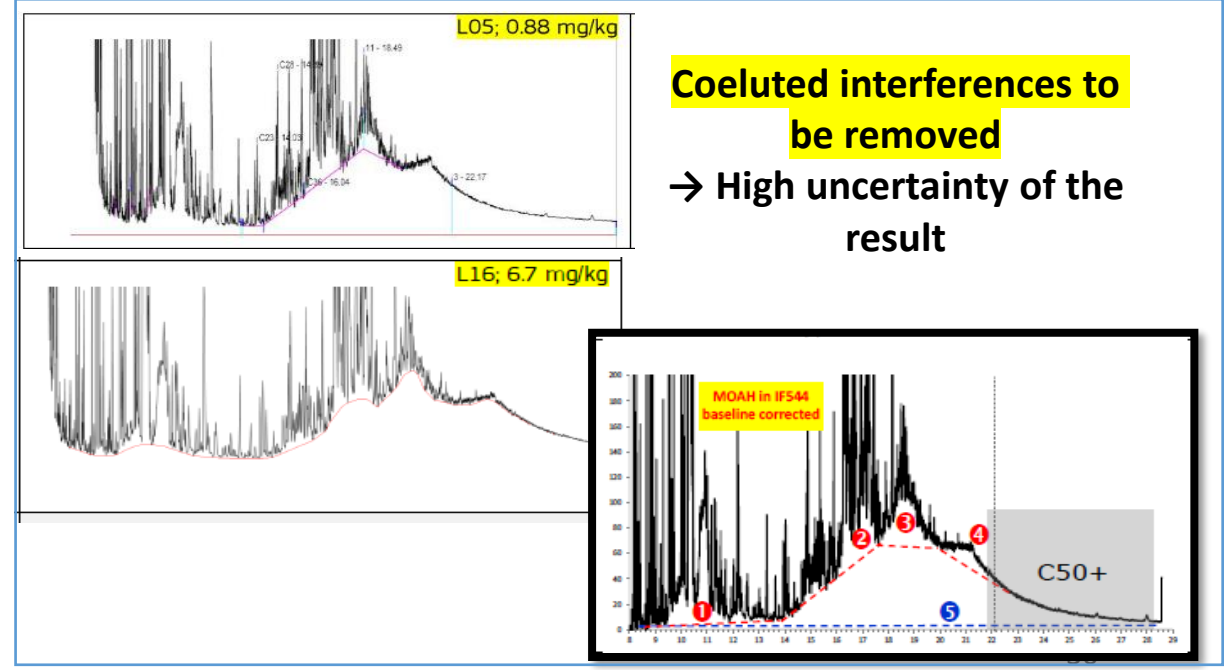
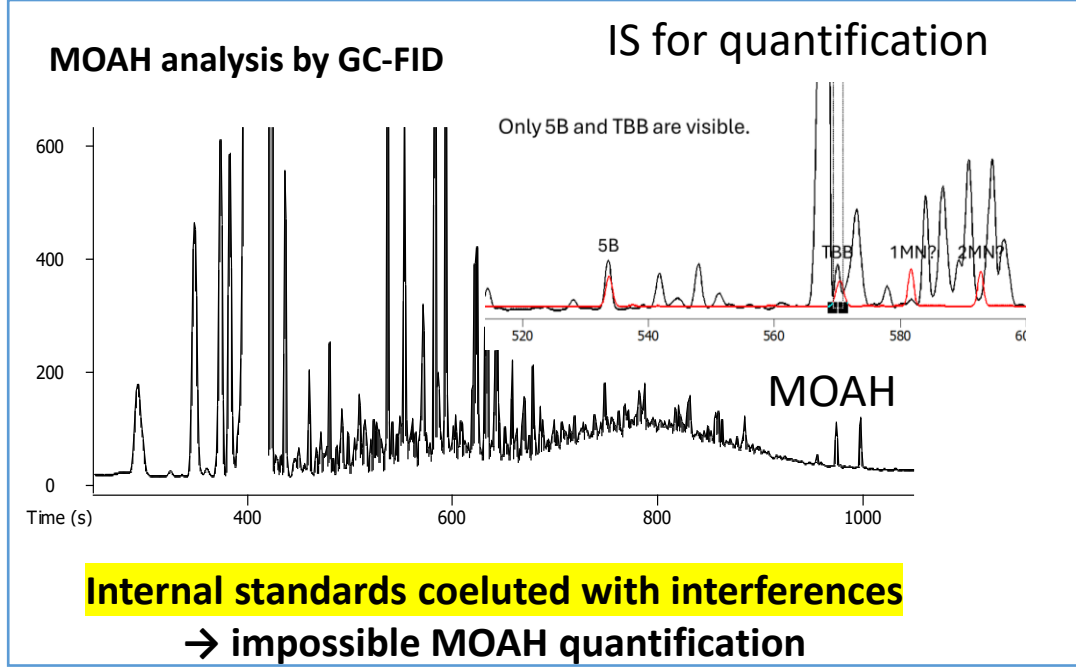
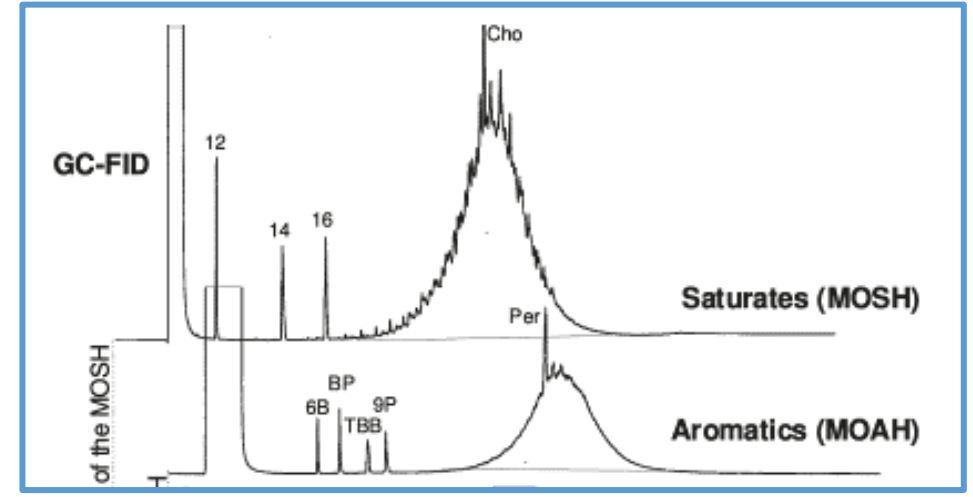
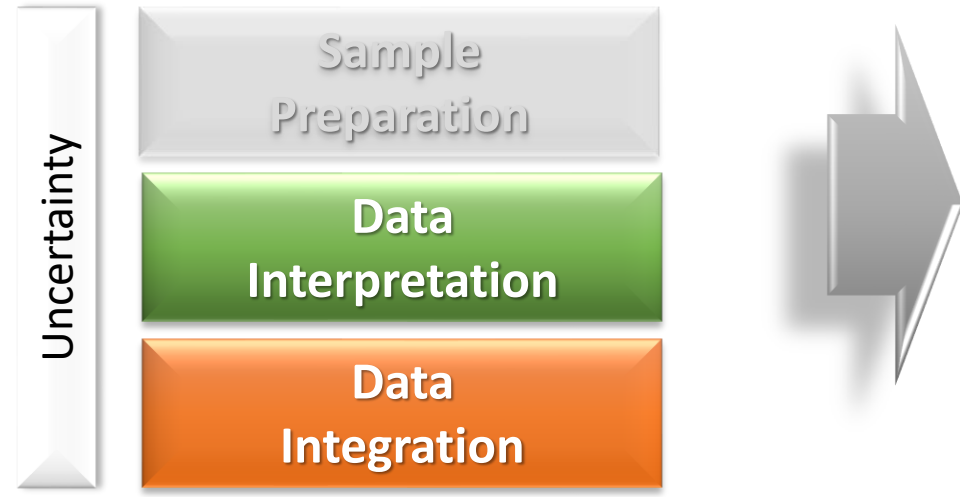
Figure 5 Decision tree on the use of auxiliary methods.



> 20% of uncertainty!

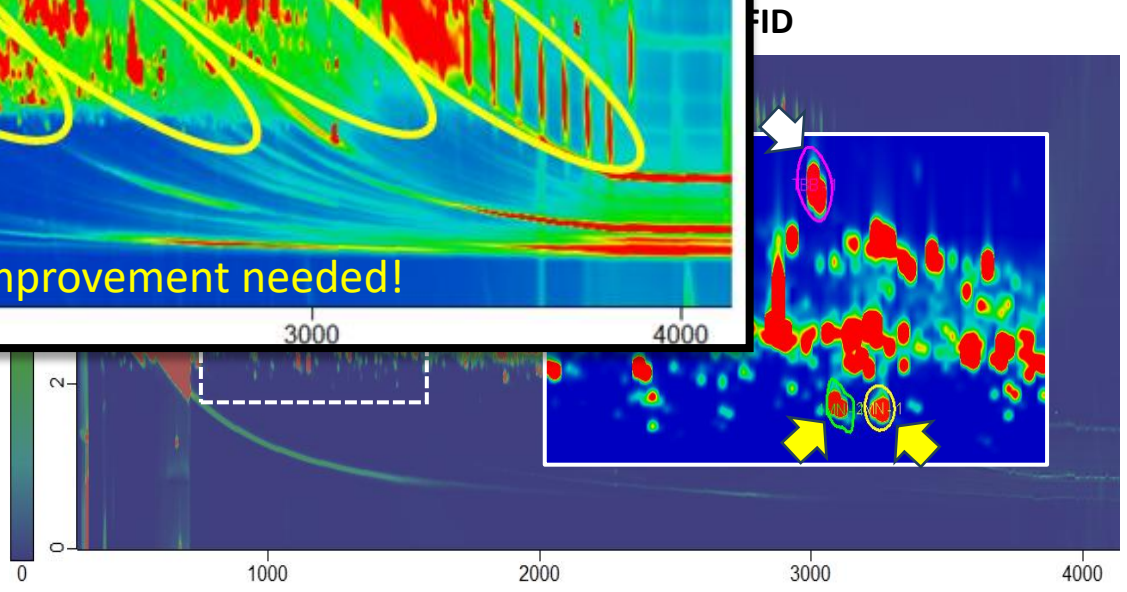
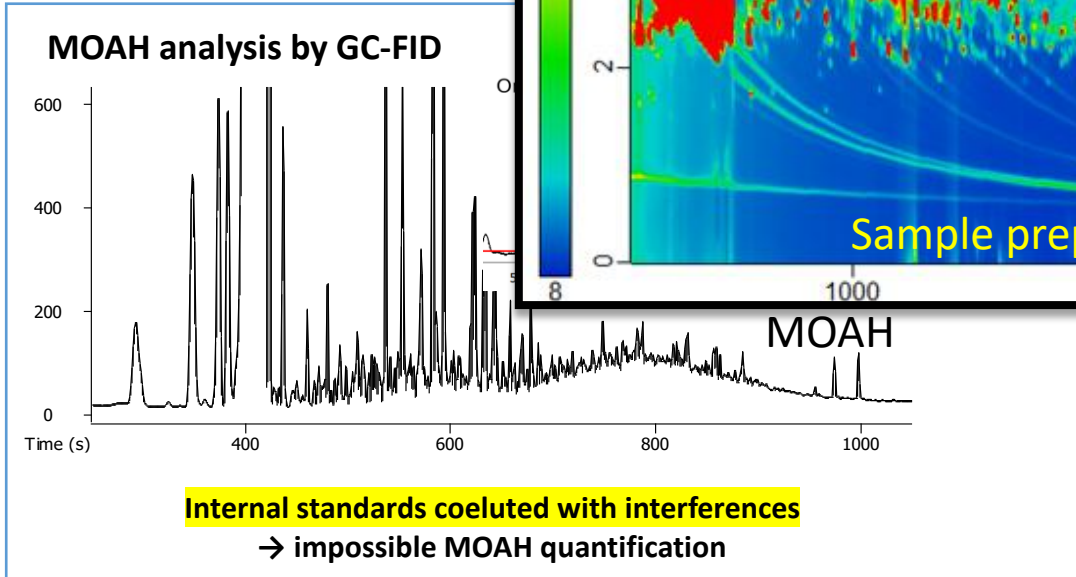
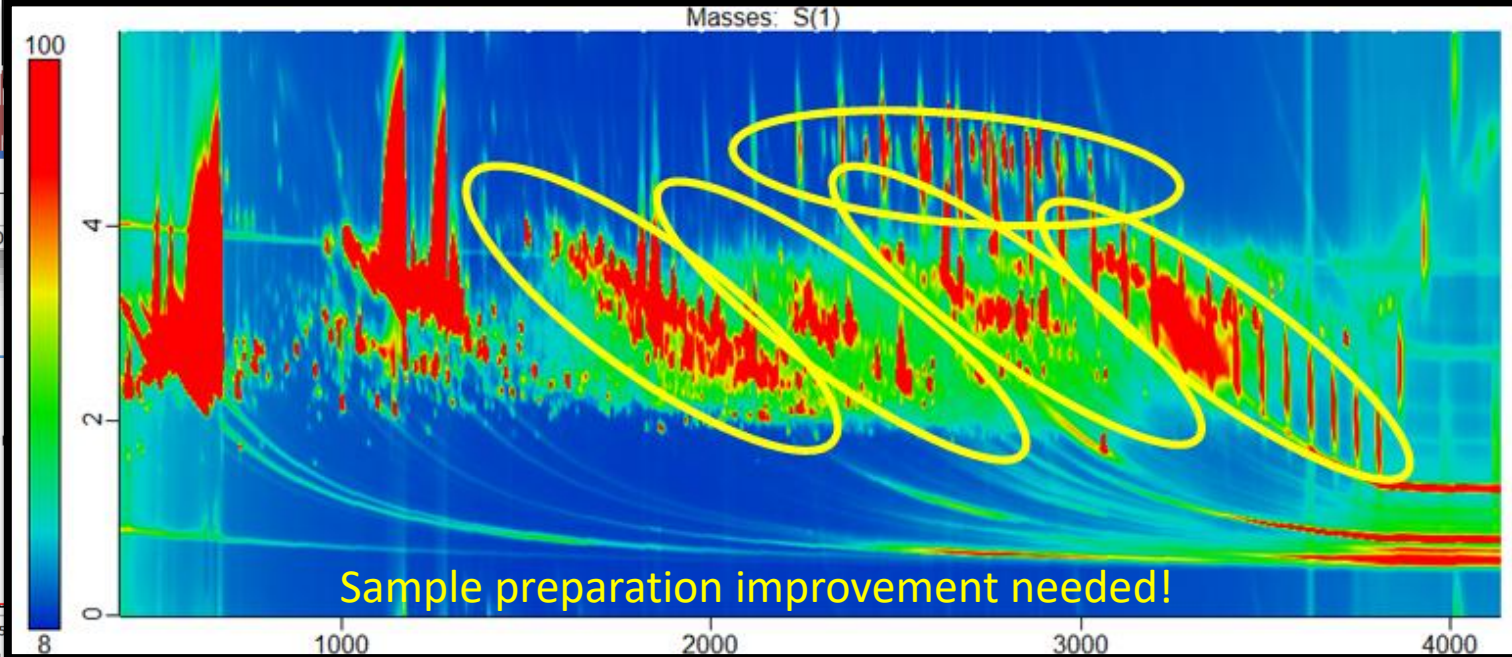
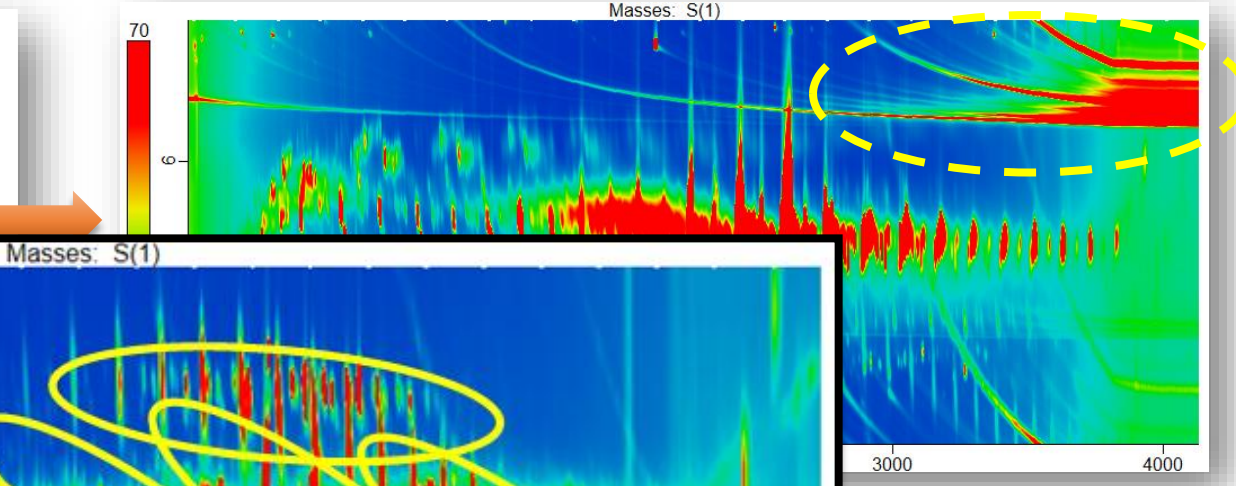
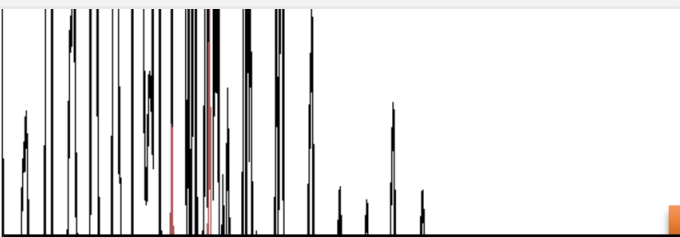
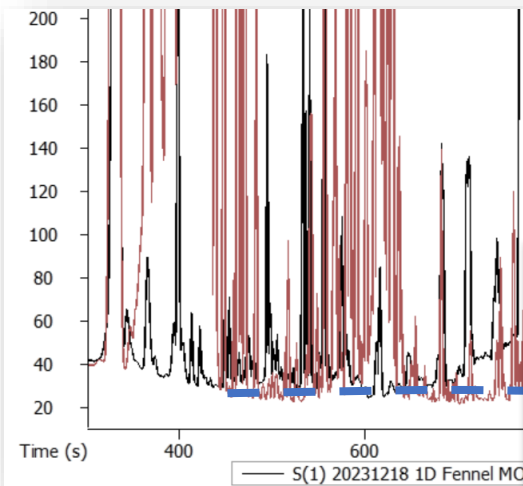
JRC proficiency test 2023 on olive oil





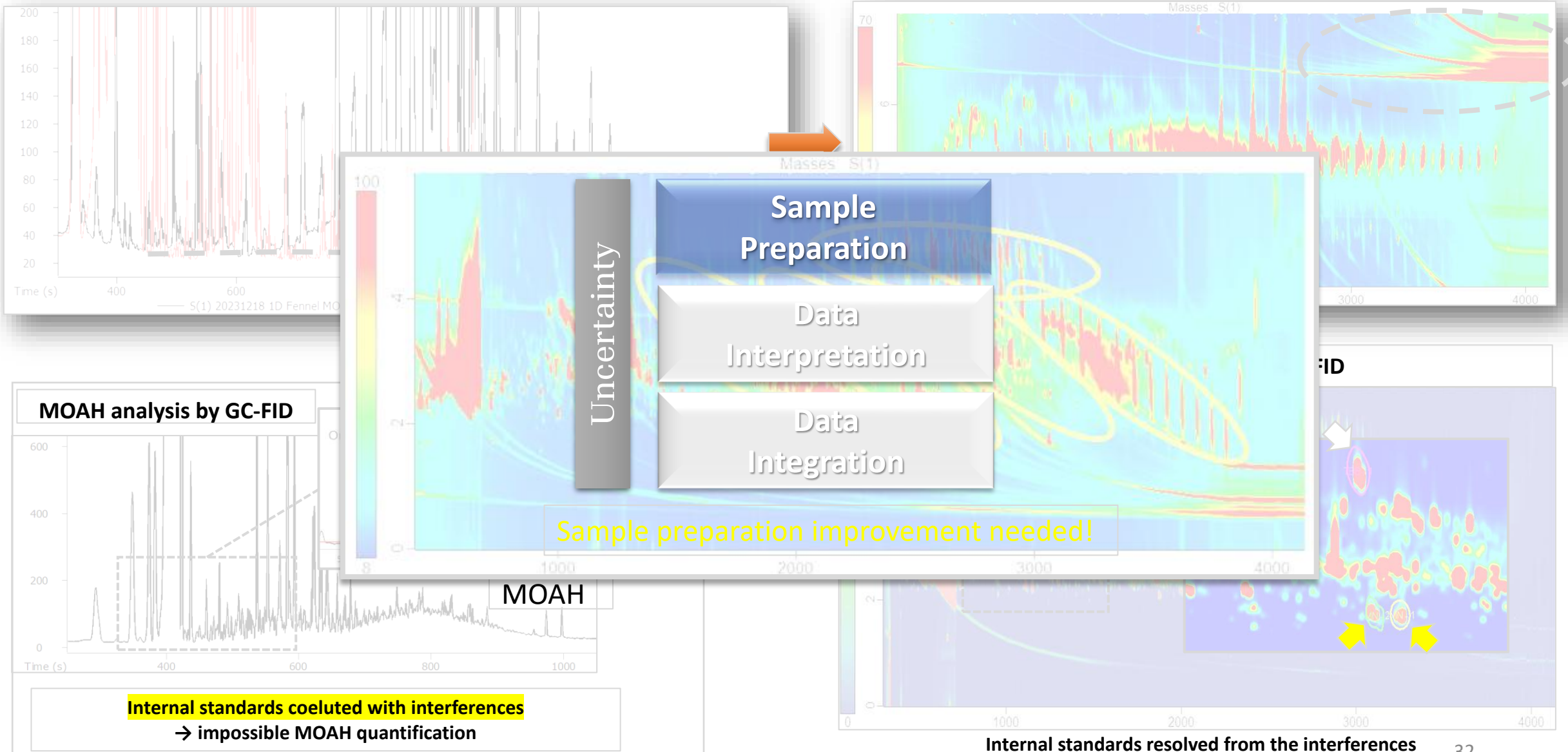
MOSH vs MOAH – No purification

More accurate quantification!

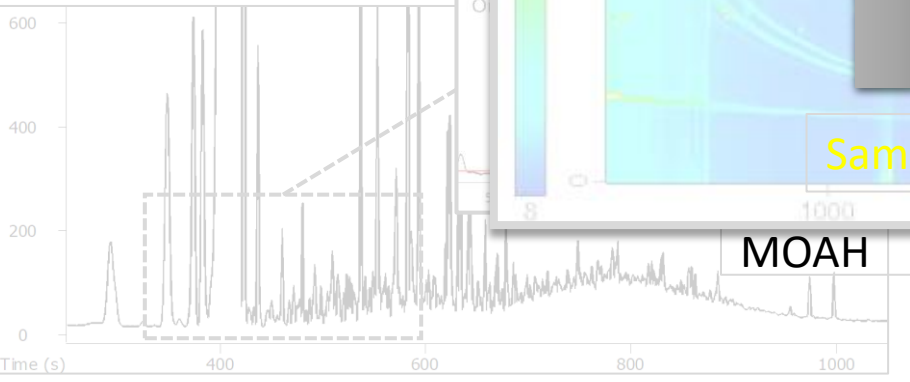


MOSH vs MOAH – No purification

More accurate quantification!

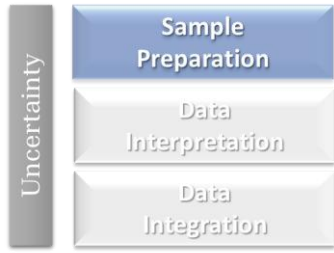


MOAH analysis by GC-FID

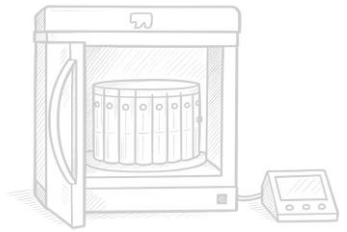


Internal standards coeluted with interferences
→ impossible MOAH quantification

Internal standards resolved from the interferences
→ quantification of MOAH is possible



SAPONIFICATION

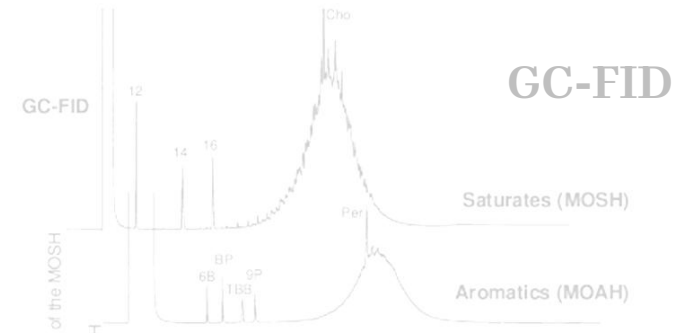
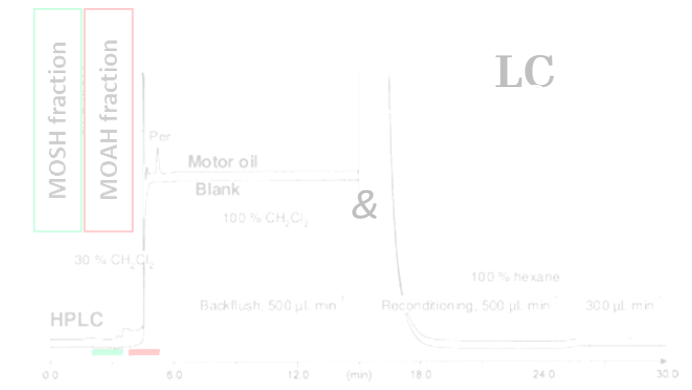
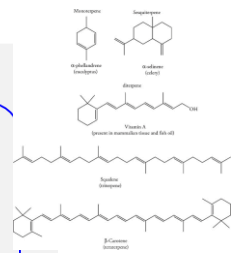
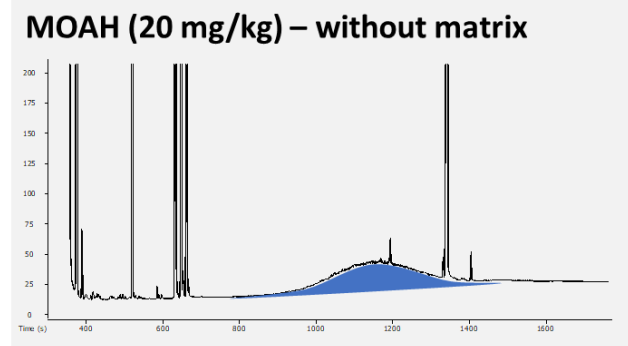
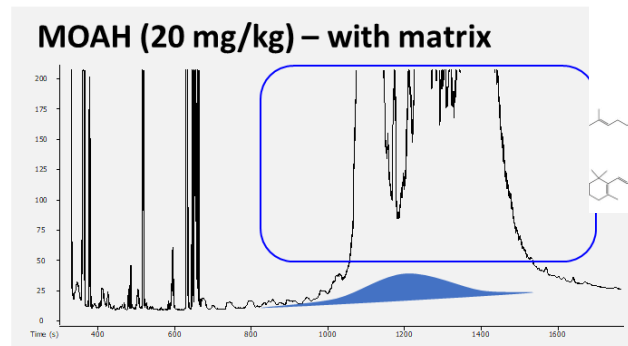


Microwave-assisted saponification and extraction (MASE) method

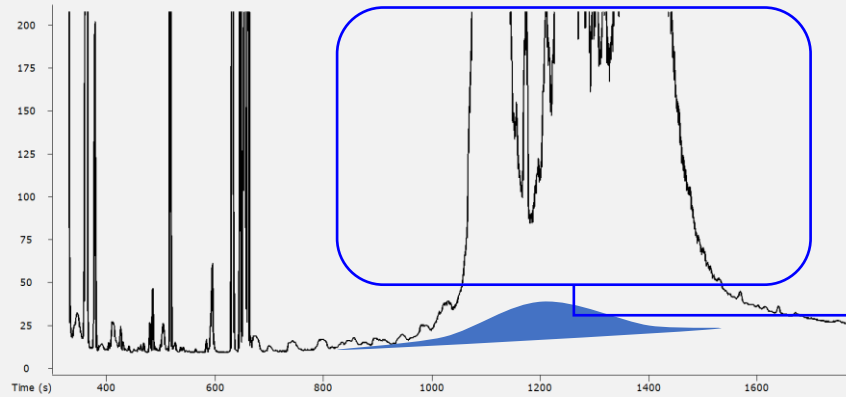
$$TBB/MN = 1.05 \pm 0.02$$

PURIFICATION

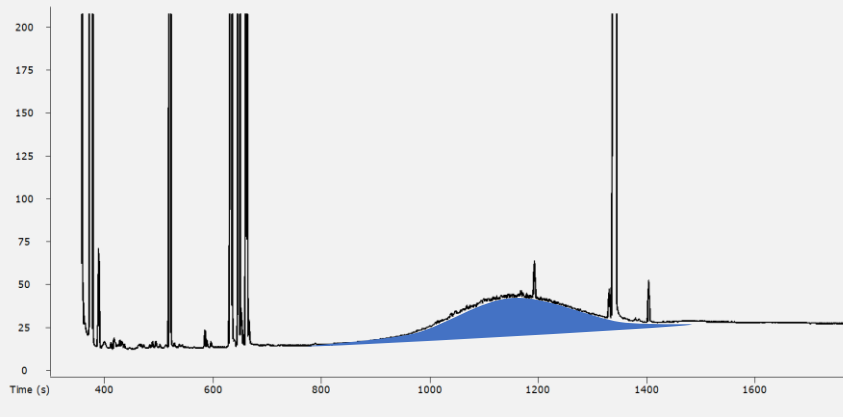
LC-GCxGC-FID



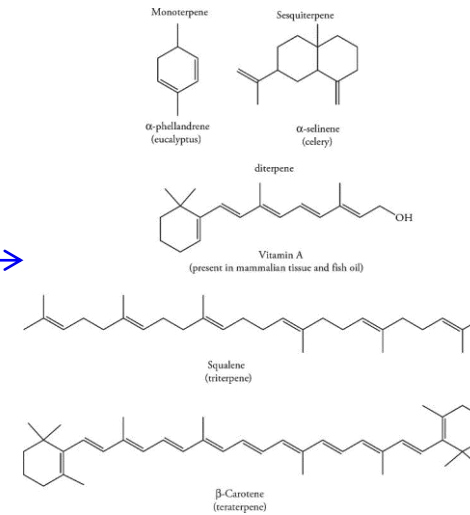
MOAH (20 mg/kg) – with matrix



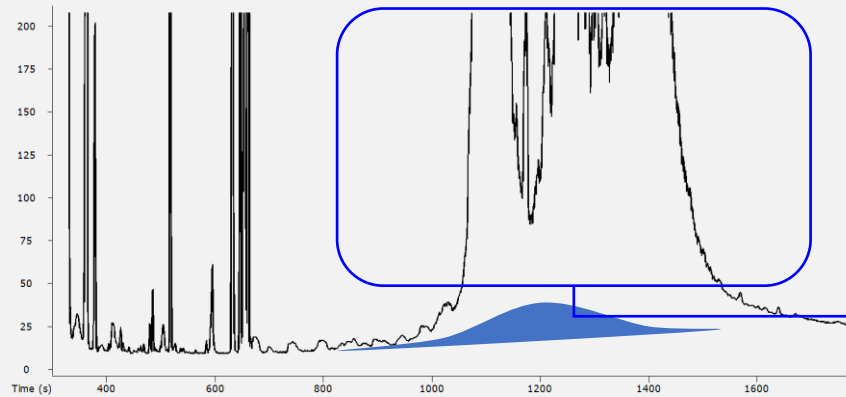
MOAH (20 mg/kg) – without matrix



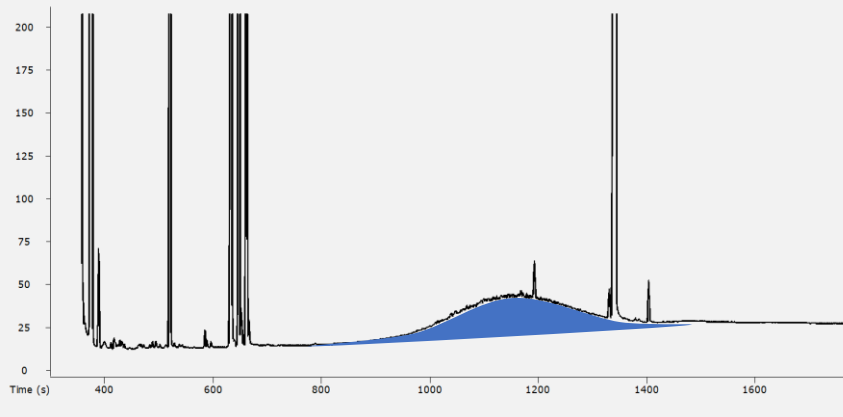
Food matrices can contain **terpenes** at concentrations that overload MOAH chromatograms



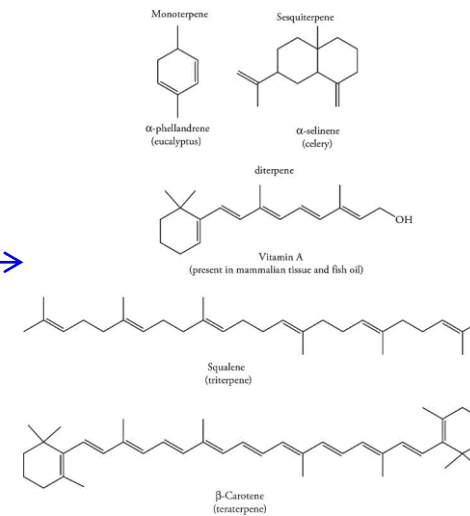
MOAH (20 mg/kg) – with matrix



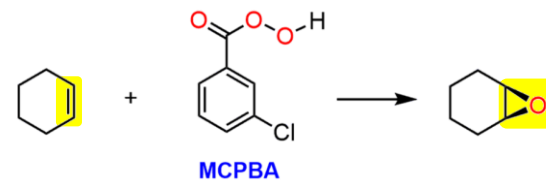
MOAH (20 mg/kg) – without matrix



Food matrices can contain **terpenes** at concentrations that overload MOAH chromatograms

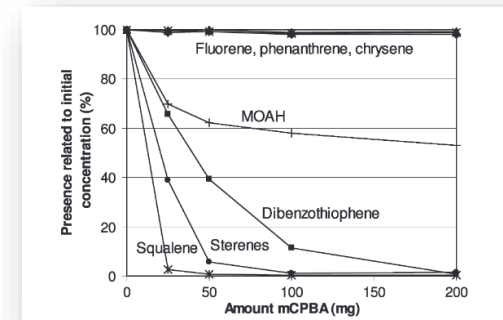


Epoxidation is used to remove these terpenes

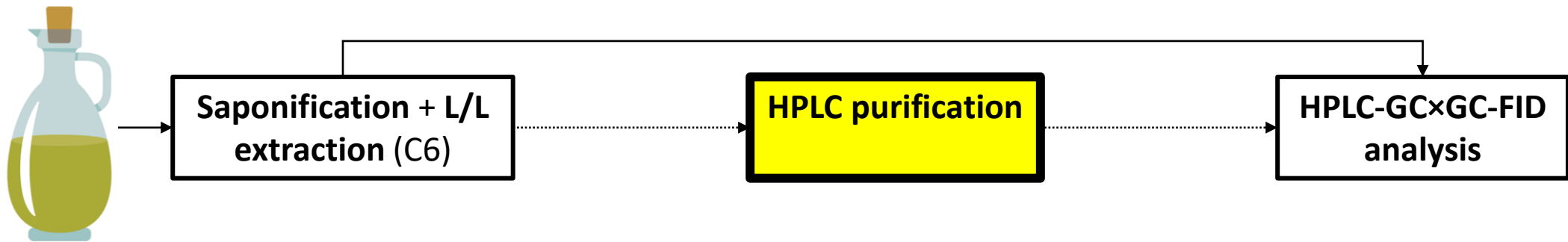
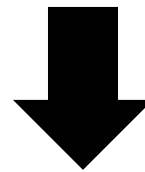
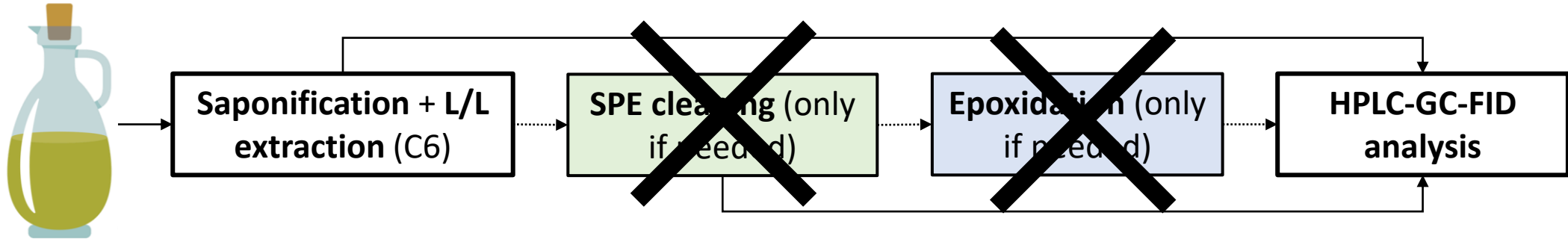


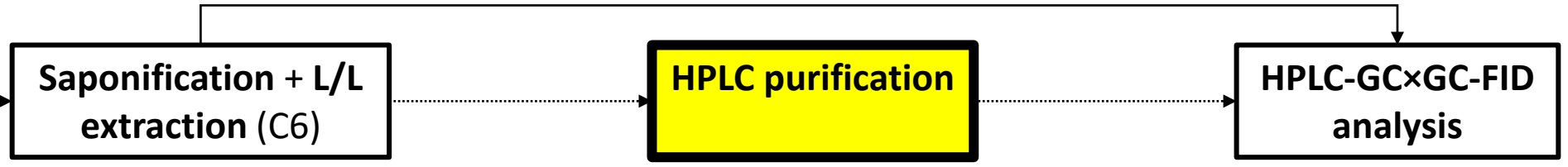
20-40% of MOAH can be also be lost (particularly those having many d.b.)

Another cause of uncertainty!



ISO 20122:2024 **Vegetable oils** — Determination of mineral oil saturated hydrocarbons (MOSH) and aromatic hydrocarbons (MOAH) with online coupled HPLC-GC-FID analysis — Method for low limit of quantification



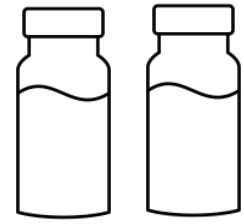
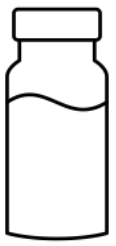


Allure Silica (250 mm × 2.1 mm i.d. × 5 μm d_p, 60 Å (Restek))

HPLC

C6/DCM gradient 1

Non-purified oil extract (after saponification)

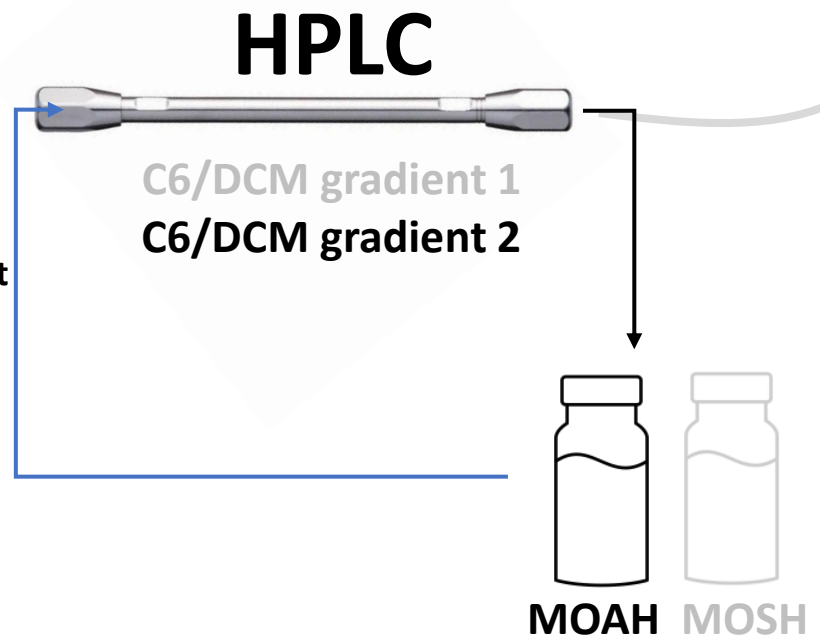
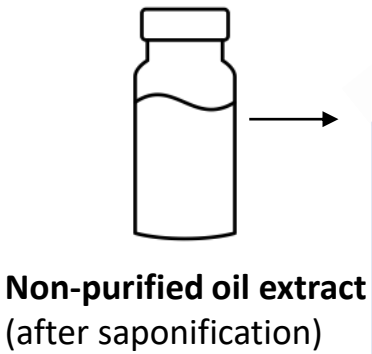


MOAH MOSH

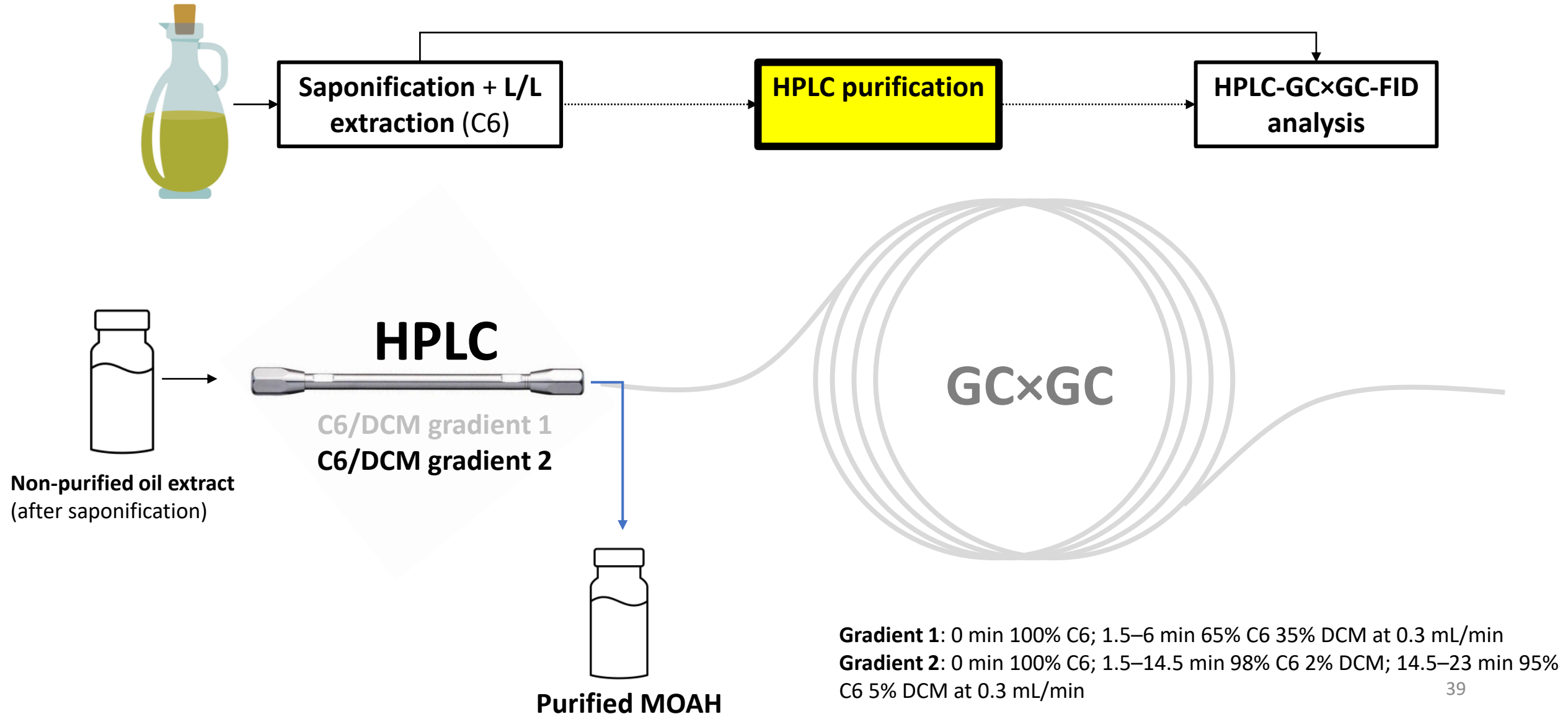


GC×GC

Gradient 1: 0 min 100% C6; 1.5–6 min 65% C6 35% DCM at 0.3 mL/min



Gradient 1: 0 min 100% C6; 1.5–6 min 65% C6 35% DCM at 0.3 mL/min
Gradient 2: 0 min 100% C6; 1.5–14.5 min 98% C6 2% DCM; 14.5–23 min 95% C6 5% DCM at 0.3 mL/min



2025

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Purification of mineral oil aromatic hydrocarbons and separation based on the number of aromatic rings using a liquid chromatography silica column. An alternative to epoxidation

Aleksandra Gorska ^a, Grégory Bauwens ^a, Marco Beccaria ^b, Giorgia Purcaro ^{a,*}



LC-GC×GC-FID/MS

Very good removal of carotenoids and squalene

Other terpenoids are less well removed



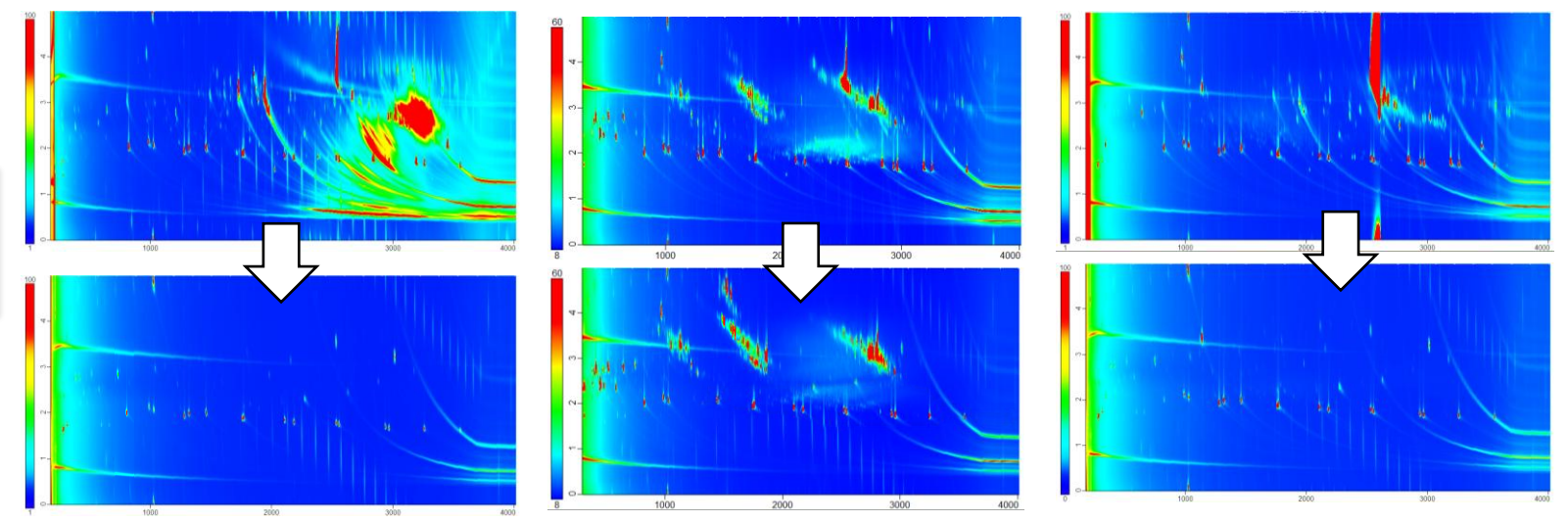
Epoxidation

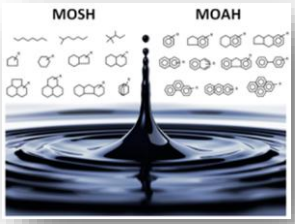
mCPBA
82% ± 10%

performic acid
71 ± 16%



MOAH Recovery LC Purification:
94% ± 2%





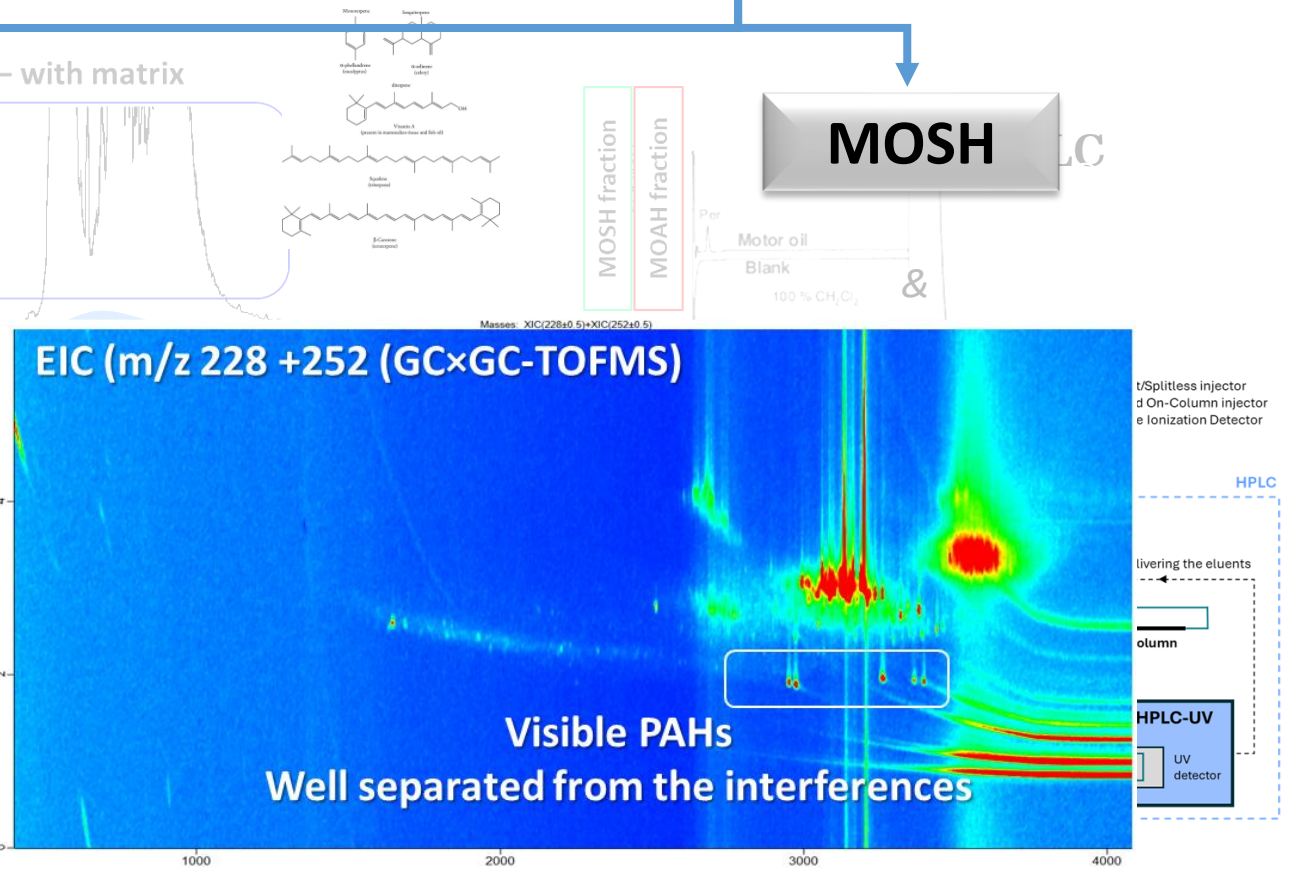
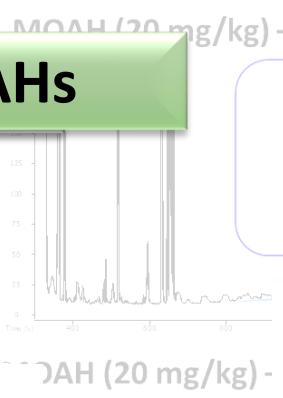
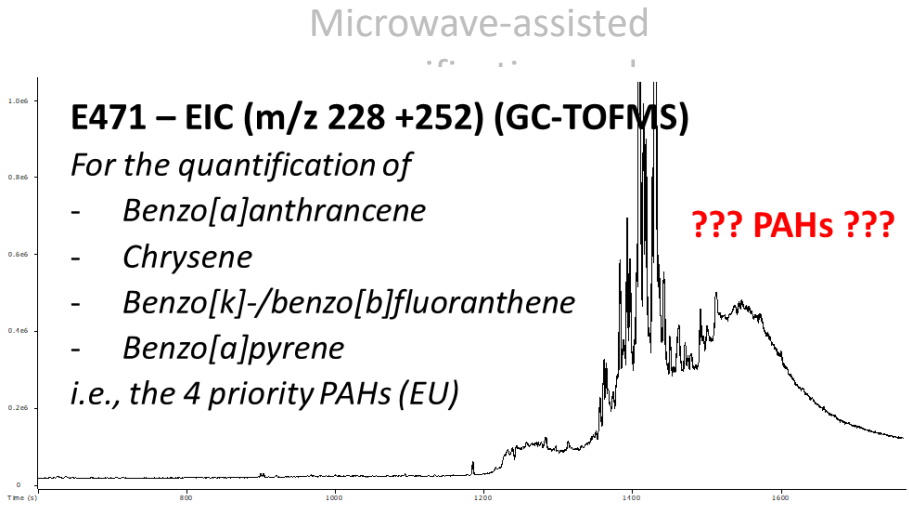
SAPONIFICATION

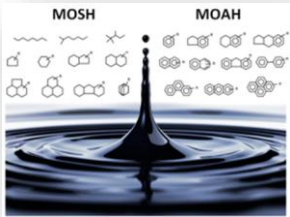
PURIFICATION

LC-GC×GC-FID/MS

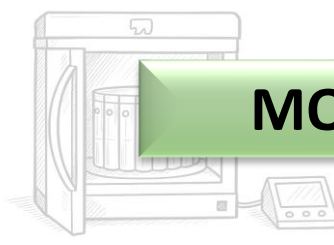
MOAH & PAHs

MOSH



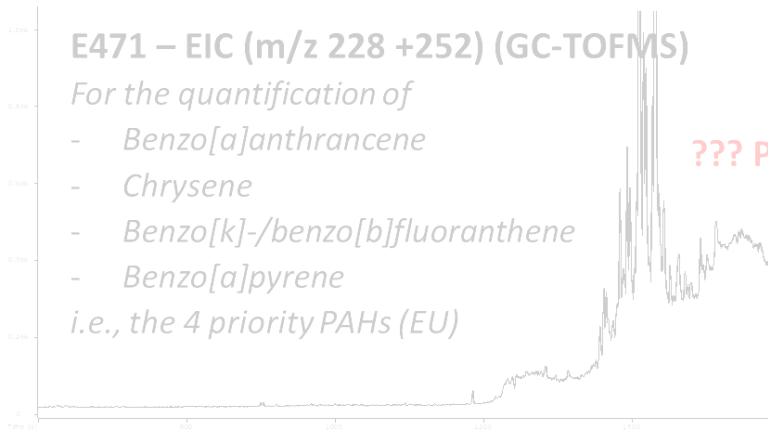


SAPONIFICATION



MOAH

Microwave-assisted



LIÈGE université Gemboux Agro-Bio Tech

Università degli Studi di Ferrara

ACCESS

ExTech 2025 Mülheim an der Ruhr

Simultaneous analysis of MOSH/MOAH and PAHs in fat oils derived food additives using an adapted sample preparation workflow and HPLC/GC×GC-

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INTRODUCTION

Mineral oil hydrocarbons (MOHs) represent a broad class of compounds derived from petroleum distillation and refining processes. Based on their chemical structure, they are classified into mineral oil saturated hydrocarbons (MOSH) and mineral oil aromatic hydrocarbons (MOAH). The potential human health impact of MOH varies widely. MOSH, can accumulate in the human tissues like liver and lymphoid system, while MOAH (3-7 ring) may contain genotoxic substances [1].

In addition to MOHs, polycyclic aromatic hydrocarbons (PAHs) are contaminants that result from incomplete combustion of organic matter, and they have carcinogenic, genotoxic and mutagenic effects [2].

MOSH & MOAH

PAHs

MATERIALS & METHODS

Analytical workflow

Instrumental system

HPLC	GC×GC-FID/TOFMS
<p>250 mm × 4.6 mm i.d. × 5 µm, 60 Å (Restek)</p>	<p>Guard Column 10 m × 0.53 mm i.d. (1) Rui 1758 MS 15m × 0.25 mm × 0.25 µm; (2) Rui Tms HT 0.8 m × 0.15 mm × 0.15 µm. TOFMS line (PMI) - (1) Rui 1758 MS 15m × 0.25 mm × 0.25 µm; (2) Rui Tms HT 1.3 m × 0.15 mm × 0.15 µm. TOFMS acquisition range: 50-700 m/z</p>

RESULTS & DISCUSSION

Evaluation of the sample preparation

- All food additives tested were saponified efficiently
- TBB/2MNH (i.e., two internal standards that can be used for MOAH quantification) ratios were below the tolerance value of 1.15 defined in the JRC guidance for MOSH/MOAH[4]
- The applied workflow yielded repeatable peak areas for MOSH and MOAH standards, with coefficients of variation between 12% and 19% across all standards in the ten E471 samples (data not shown).

HPLC/GC×GC-FID MOSH/MOAH profiles

- Most of the tested food additives showed very high levels of interferences.
- Epoxidation was helpful, but not 100% efficient for purification of MOAH.
- In general, MOSH and MOAH levels were low. MOAH were all below 2 mg/kg. Other tested additives were compliant, too, although it was not possible to determine the MOSH/MOAH of some specific samples, such as the E570 shown below.

Analysis of PAHs by GC×GC-TOFMS

- Analysis of PAHs is performed on non-epoxidized MOAH fractions.
- PAHs are determined using extracted ion chromatograms (EIC). Quantification is performed using deuterated PAHs (e.g., BaP-d12) used as internal standards.
- As visible, many interferences persist when working in EIC with monodimensional GC. GC×GC helps overcoming this limitation.

CONCLUSION

The following analytical workflow was successfully applied for the determination of MOSH/MOAH and PAHs in food additives as E471, E322, E472, E473, E475 and 570. However, food additives proved to be a complex matrix due to a high abundance of coeluting interferences, particularly affecting MOAH identification. For this reason, epoxidation step facilitated MOAH analysis. In conclusion, GC×GC significantly improved the reliability of both MOSH/MOAH and PAH determination and this workflow could be adapted for the analyses of similar contaminants in other complex food matrices.

ACKNOWLEDGMENTS

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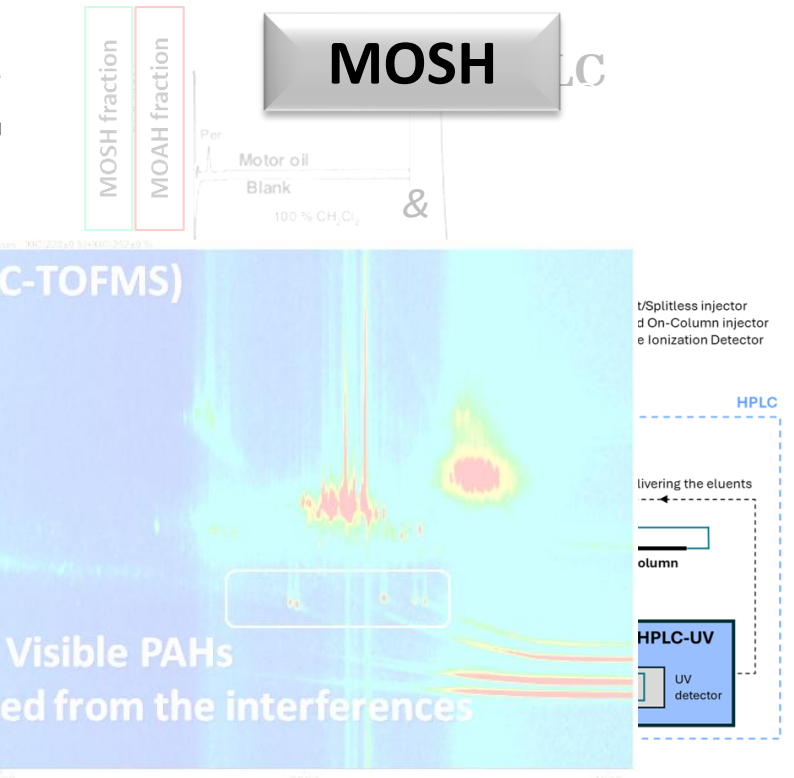
[2] Kawanishi, S. et al. (2006). Polycyclic Aromatic Hydrocarbons in Food: Scientific Opinion of the Panel on Contaminants in the Food Chain. EFSA Journal, 4(1), 124. <https://doi.org/10.2903/efsa.2006.124>

[3] Beccaria, M., & Purcaro, G. (2016). Improved microwave assisted saponification to reduce the variability of MOAH determination in edible oils. Analytical Chemistry, 88(12), 4976-4982. <https://doi.org/10.1021/acs.analychem.5b01789>

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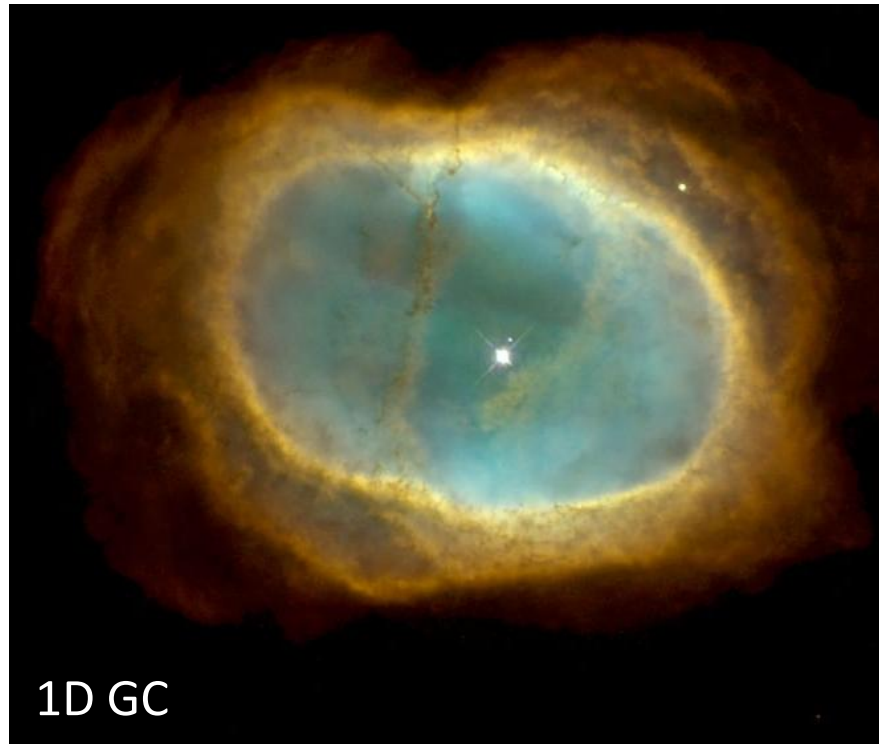
Poster Food 10

LC-GC×GC-FID



Is GC×GC worth the effort?

Southern Ring Nebula



1998 Hubble Telescope



Webb Space Telescope's view today

GC×GC an opportunity for simplifying sample preparation or an extra difficulty in food analysis?

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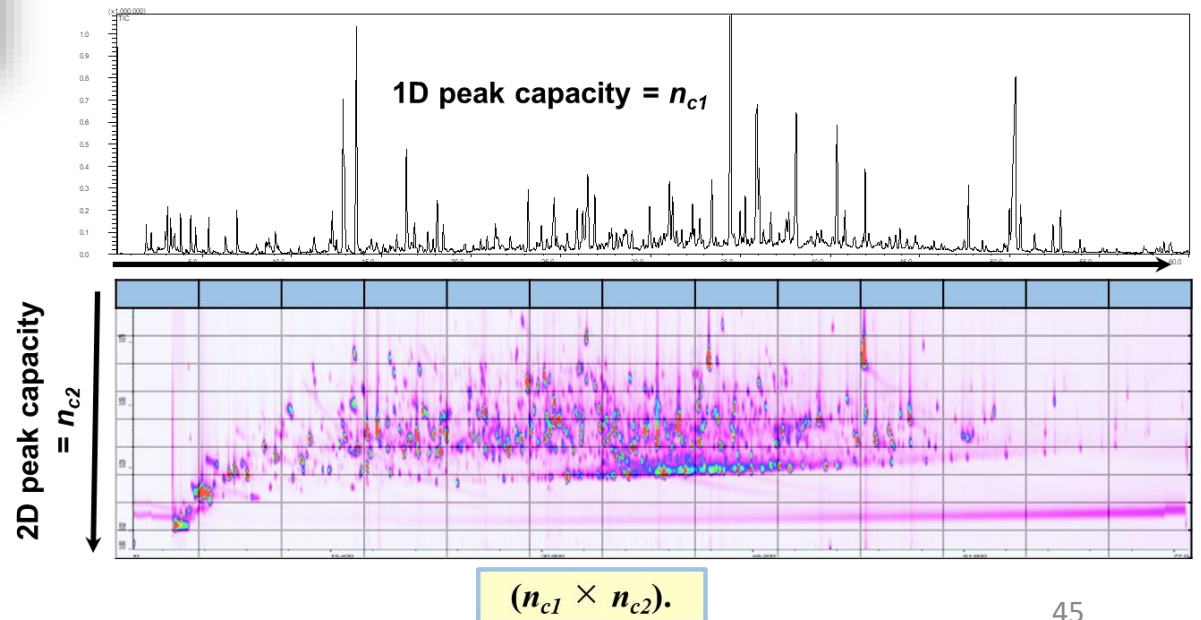


GC×GC doesn't replace **sample preparation** —
They provide *reciprocal reinforcement!*

GC×GC

- ✓ Inform optimization
- ✓ maximizes the information
- ✓ Simplify sample preparation
- ✓ Support data interpretation

Comprehensive: CC × CC



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