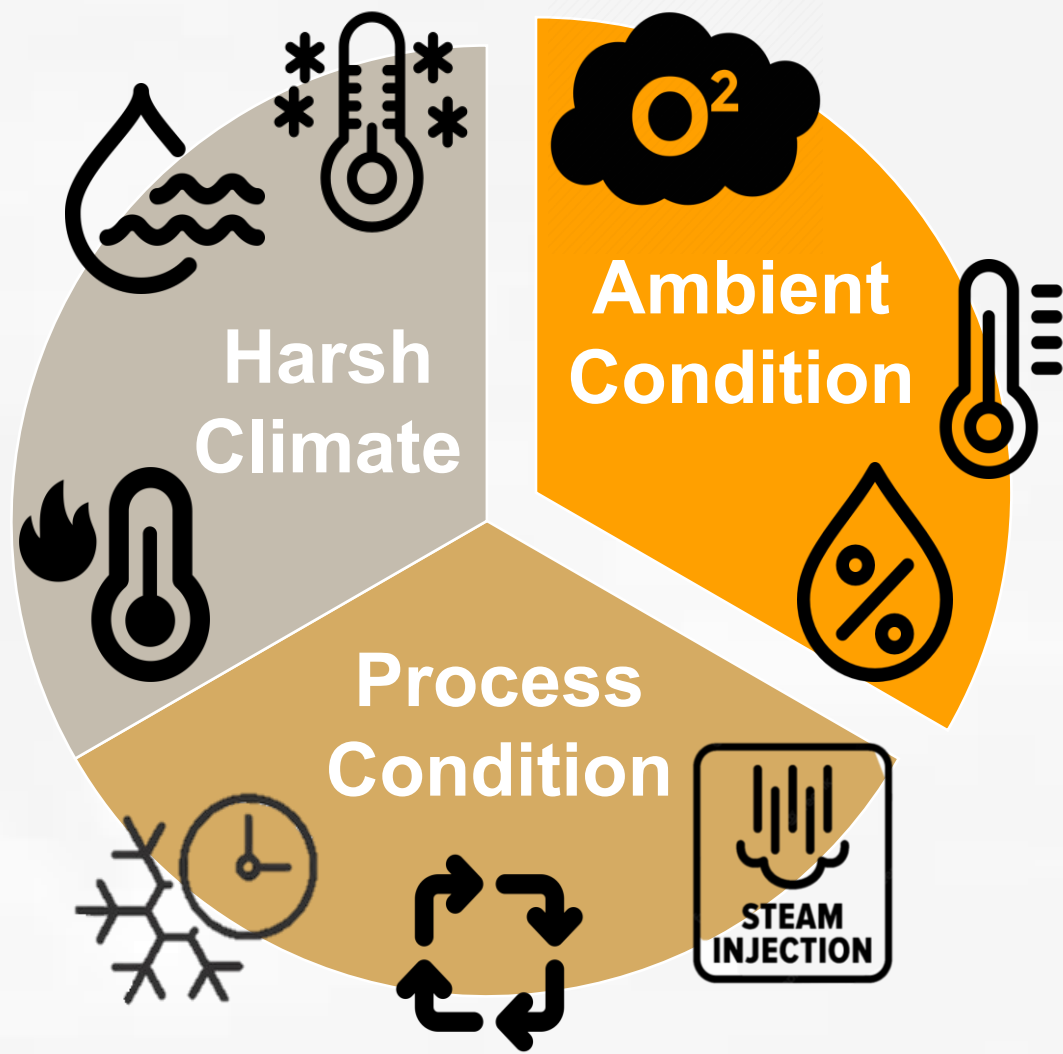


Stability of Adsorbents for Direct Air Capture: Experimental and Modeling Approach

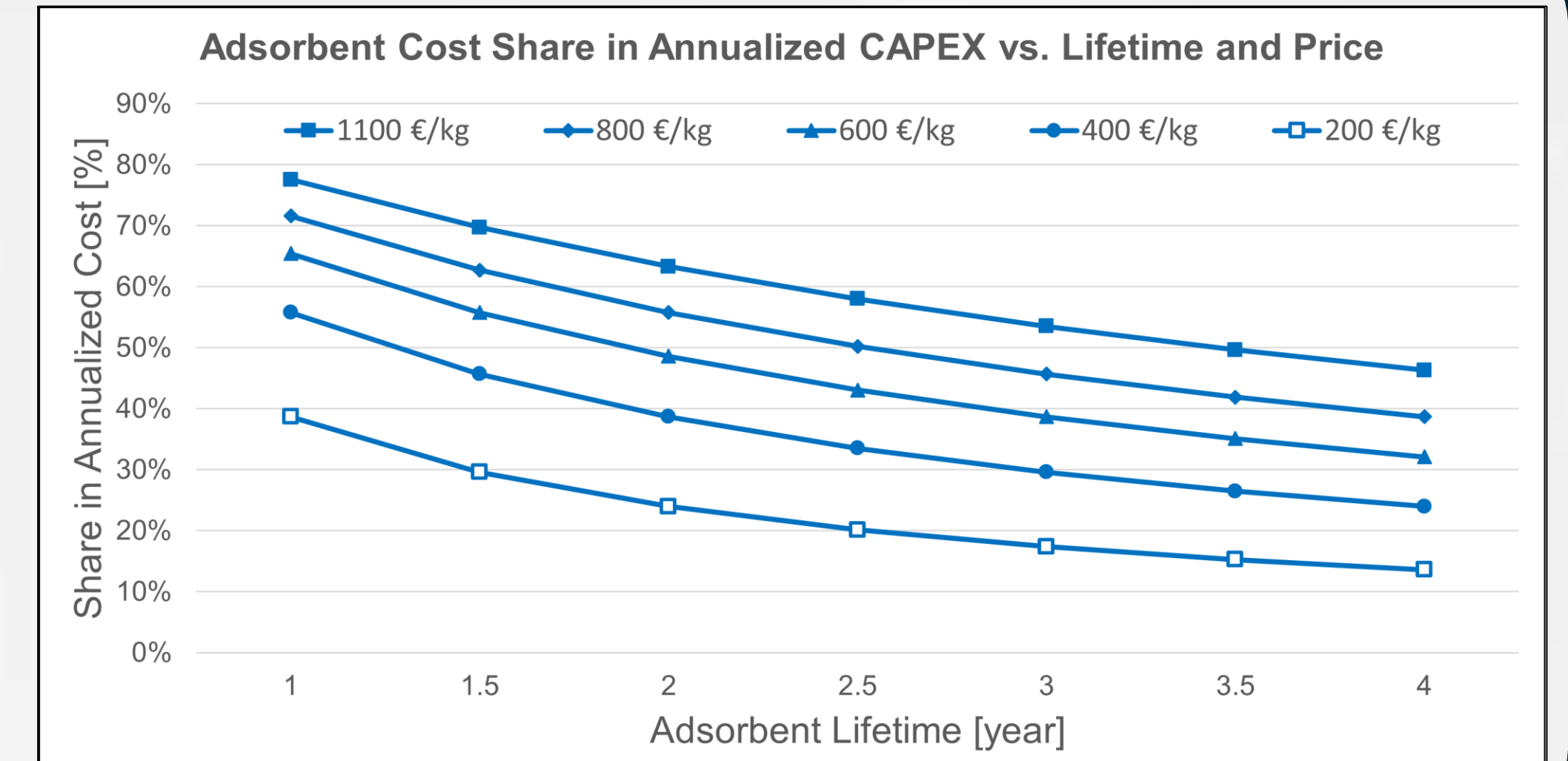
Motivation and Background



Stability of solid sorbents is a critical factor in adsorption-based separation processes, including direct air capture (DAC).

- Lifetime directly impacts **cost**, **downtime**, and **net CO₂ removal**.
- Results in **undesired emissions**, and additional material **waste**.
- While most studies focus on capacity and selectivity, **real operating conditions** (temperature, humidity, O₂) can significantly degrade sorbents.

The impact of sorbent stability on CAPEX is evaluated using the DAC model of Kim and Léonard [1] (Climeworks Hinwil, 2460 kg CO₂/day).



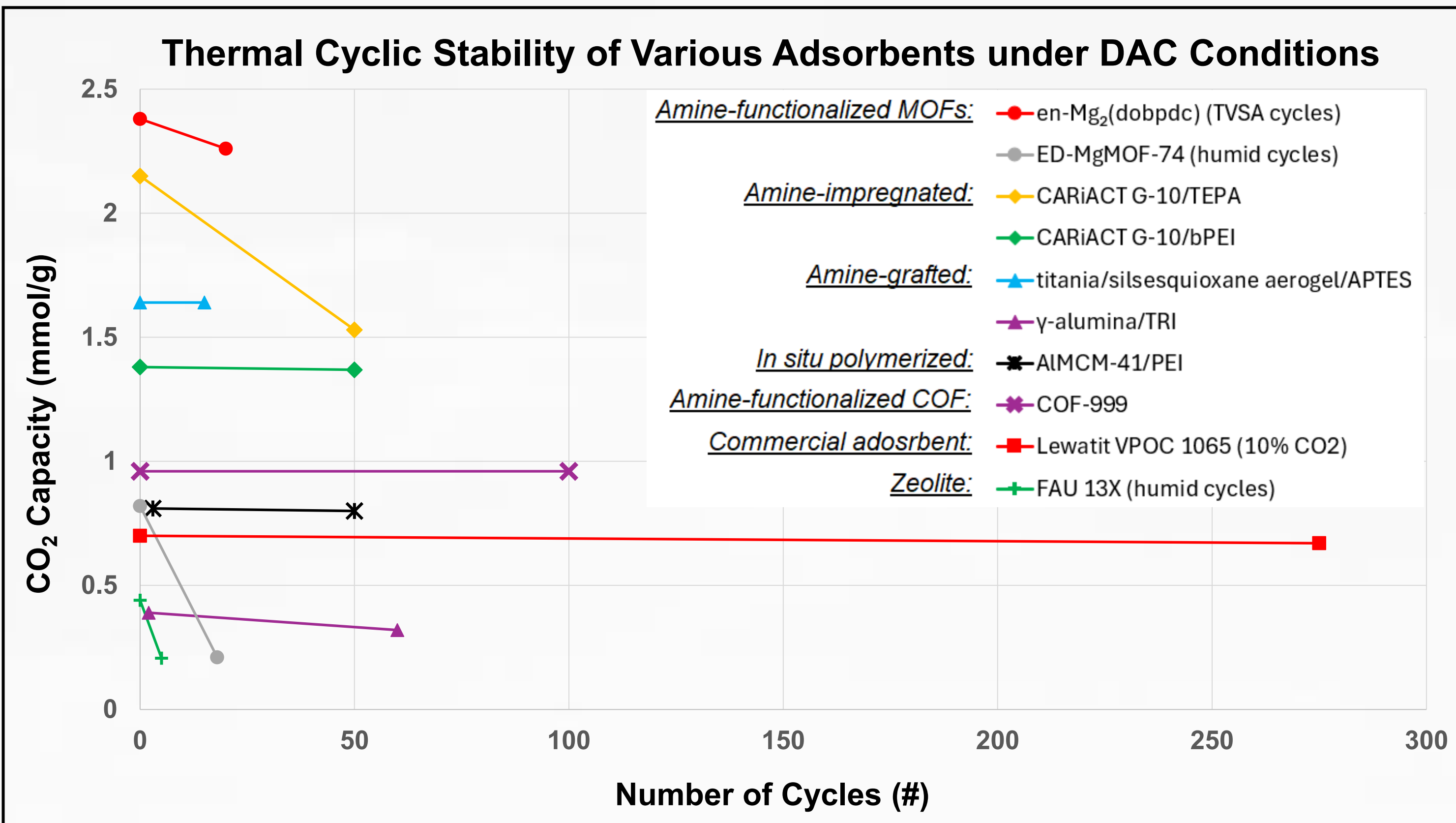
Objectives

Develop a framework integrating adsorption modeling and adsorbent degradation to predict long-term performance

1. Conduct a **literature review** to identify DAC adsorbents, major degradation mechanisms, and experimental methodologies
2. Design and build a dedicated **experimental test bench** to evaluate sorbent stability under different operating conditions
3. Develop an **adsorption process model** for DAC to predict adsorbent performance over time **considering deactivation**



Edith Flanigen (1929–2026)
"Without stability, you have nothing" ...



| | Adsorbent Types | Advantages | Disadvantages |
|---------------|--|--|---|
| Physisorbents | Zeolite | High thermal/chemical stability High CO ₂ capacity at sub-ambient conditions | Strong moisture sensitivity Low CO ₂ capacity at ambient temperature |
| | Metal-Organic Frameworks (MOFs) | High thermal stability High CO ₂ selectivity | Moisture instability for some MOFs Low capture capacity at high CO ₂ pressure |
| | Amine-impregnated | Effective under humid conditions Relatively simple synthesis | Low thermal stability (evaporation/leaching) Oxidatively sensitive |
| Chemisorbents | Aminosilane-grafted | High thermal stability High humidity tolerance | Oxidatively sensitive More complex synthesis |
| | Amine-appended MOFs | High uptake at low CO ₂ pressure High selectivity | Synthetic complexity Oxidative/moisture degradation |
| | Amine-functionalized Covalent Organic Frameworks (COFs) | Promising stability High CO ₂ selectivity | Early-stage development Limited scalability Uncertain long-term stability |
| | Carbonate-based | Low-cost materials High chemical stability | Low regeneration temperature Slow kinetics at ambient conditions |

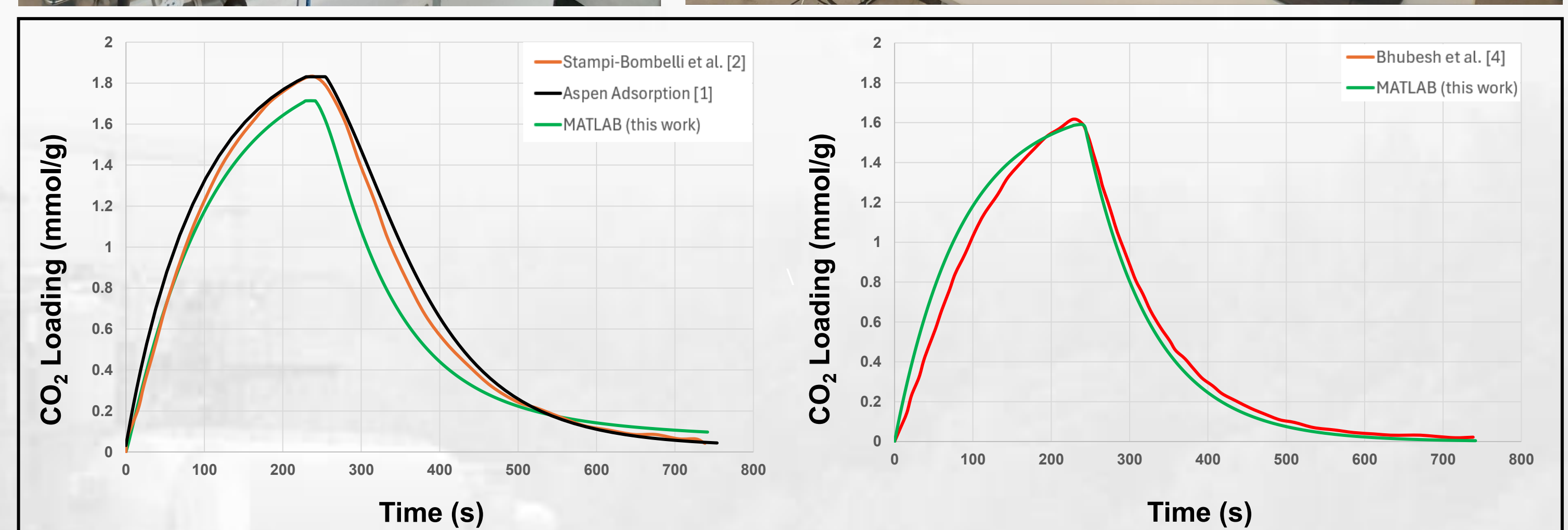
Setup for Stability Assessment

To evaluate the long-term stability of adsorbents, a dedicated automated test bench has been developed to perform repeated adsorption-desorption cycles. The setup enables stability testing of solid sorbents under controlled operating conditions:

- Temperature: -20 to 180 °C
- Gas composition: N₂/CO₂ (400–20000 ppm)/O₂
- Relative humidity: 10–90%
- Flow rate: 100–5000 Nm³/min
- Adsorbent mass: ≥ 1 g

Modeling

A 1D adsorption model for amine-functionalized Nanofibrillated cellulose (TVSA and S-TVSA) was developed in MATLAB and validated using experimental data, [2,3] with additional comparison to Aspen Adsorption simulations. [1] The model includes gas-phase component mass balances with linear driving force (LDF) mass transfer and a column energy balance. Water co-adsorption is incorporated using a modified Toth isotherm model.



Next Steps

Approaches to include sorbent degradation in adsorption models:

1. Constant capacity decay factor → large prediction errors
2. Degradation rate equations integrated into the model → complex degradation phenomena
3. Cycle-dependent isotherm parameters → selected approach

In this work, degradation will be evaluated by performing 50–200 adsorption-desorption cycles. Isotherms of aged adsorbents will then be measured, and the resulting parameters will be expressed as a function of cycle number and incorporated into the adsorption model.

Acknowledgments

The results presented were made possible thanks to infrastructure acquired under Measure BE-C[C72]-I-[I-711], Research Platform for Energy Transition of the French Community, funded through the REPowerEU chapter of the NextGenerationEU planS. We would also like to express our gratitude to Patrick Kreit and our colleagues in the group, Cédric Calberg and Sofiane Bekhti, as well as to the members of the Halle de Génie Chimique, Christian Szykula and Sergio Parra, for their valuable contributions to the construction of our experimental setup.

References

- [1] Kim, S. M., & Léonard, G. (2022). In Proceedings of the 16th Greenhouse Gas Control Technologies (GHGT-16) (pp. 23-24).
- [2] Stampi-Bombelli, V., van der Spek, M., & Mazzotti, M. (2020). *Adsorption*, 26(7), 1183-1197.
- [3] Gebald, C., Wurzbacher, J. A., Borgschulte, A., Zimmermann, T., & Steinfeld, A. (2014). *Environ. Sci. Technol.*, 48(4), 2497-2504.
- [4] Balasubramaniam, B. M., Thierry, P. T., Lethier, S., Pugno, V., Llewellyn, P., & Rajendran, A. (2024). *Chem. Eng. J.*, 485, 149568.



Review
Article



Stay in
touch

2026 French-German
Adsorption Initiative
Duisburg, Germany

