

MD-simulated ices

High-level method design

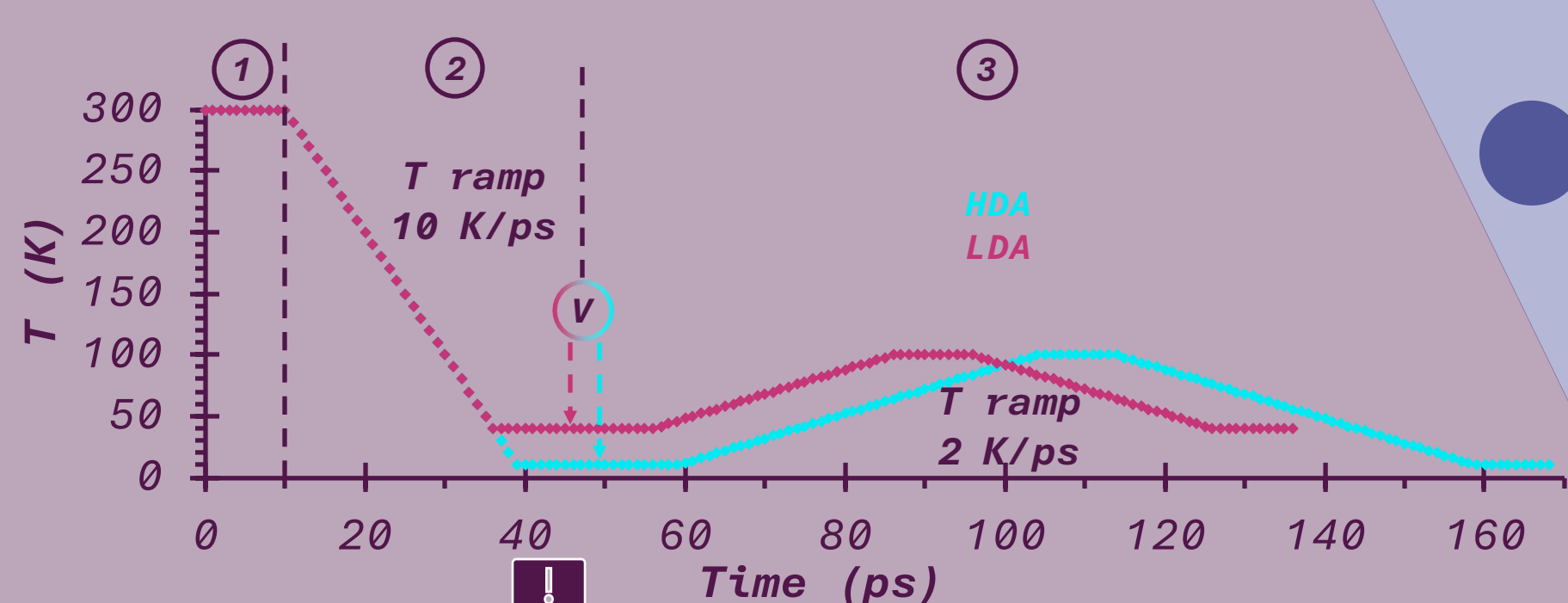
A DFT-functional benchmark on quadrimeric structures (1 adsorbate + 3 H₂O) has been performed, with **6-311+G(d,p)** as basis set.

	Mean absolute relative difference (%) w.r.t. CCSD(T) over the three adsorbate test cases		Mean absolute relative difference (%) w.r.t. CCSD(T) over the three adsorbate test cases
B3LYP-D3BJ	14.3	B3PW91-D3BJ	26.0
PBE0-D3BJ	51.3	B3PW91	84.9
PBE0	22.8	CAMP-B3LYP	18.6
BMK-D3BJ	145.0	M06-2X	22.7
BMK	89.0	ωB97XD	17.8

Two interstellar ice analogues have been built, with

- $\rho \sim 0.94$ g/cm³ for Low density Amorphous Solid Water (LDA)
- $\rho \sim 1.13$ g/cm³ for High density Amorphous Solid Water (HDA)

Heat & quench protocol

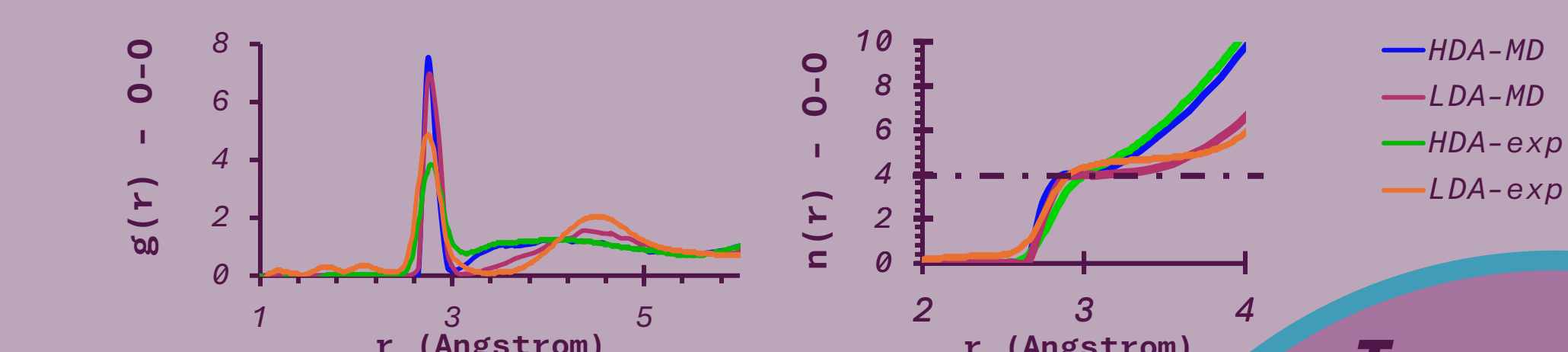


- Bulk Modelling**
- Equilibration to 300 K
 - Quenching
- Surface Modelling**
- Addition of a slab of void in the z-direction - "2D" PBC
 - Re-heating to 100 K and new quenching
- Periodic Boundary Conditions (PBC)

The bulk structure reliability has been checked through

- Comparison to experimental O-O, O-H and H-H Radial Distribution Functions ($g(r)$)/Running Coordination Number ($n(r)$)

E.g. O-O distances analysis

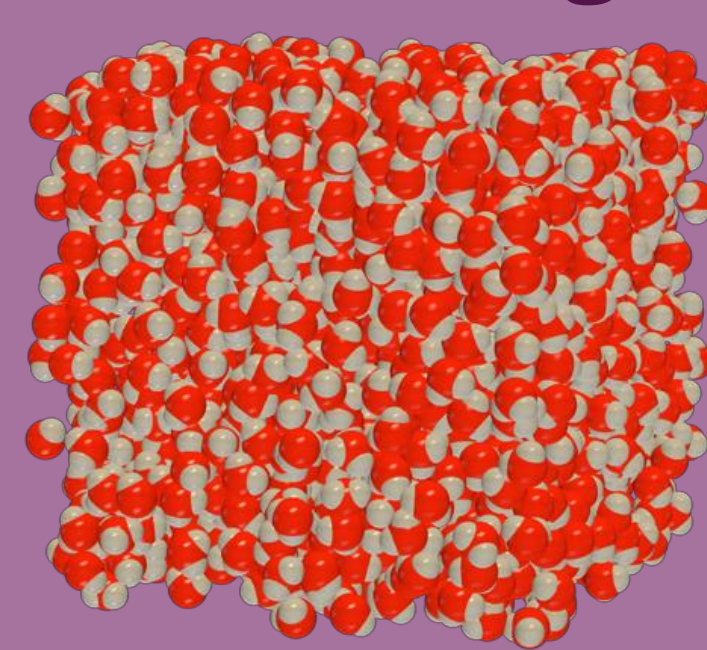


H-bonds analysis

Both check-up gave consistent results

- 2000 H₂O
- TIP4P/2005
- Heat & quench
- NVT

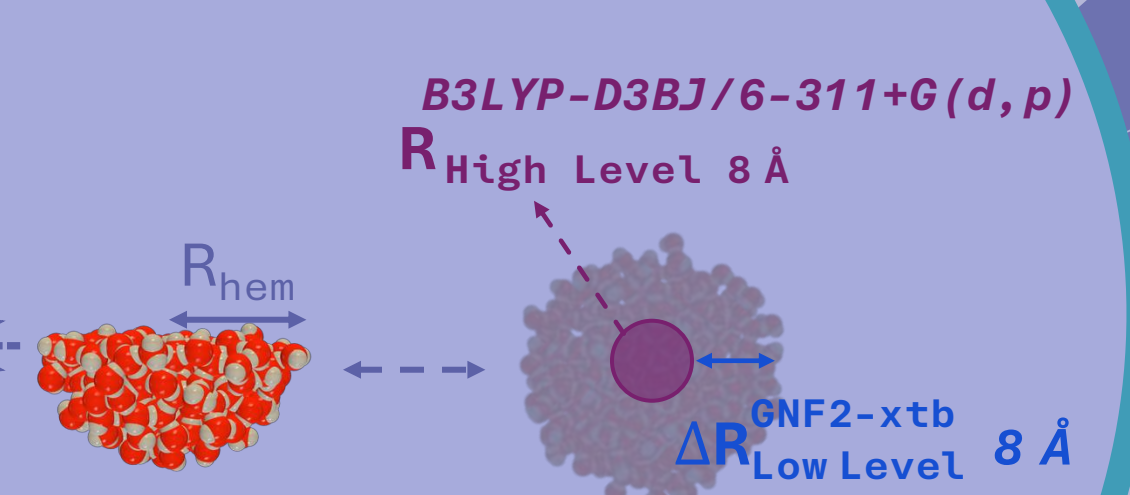
Ice model building



Molecular Dynamics
NAMD 2.14^[2]

NH₃, CO & CH₄

Binding energy inference



ONIOM-2 framework design
Gaussian 16^[3]

- High-Level Basis Set & Functional check
- High-Level Size 8 Å
- Low-Level Size benchmark
- Retro-check

Retro-checking on the high and low level sizes

BE values have been re-computed with R_{High-L} of 12 Å for a ΔR_{Low-L} of 13 Å.

Very consistent with the converged values from the low-level size benchmark with relative discrepancies < 6%



What are the main sources of BE diversity?

- Complexity of the substrate typology (amorphicity)
- Adsorbate orientation & local roughness of the potential energy landscape at a given binding site (in-site reorientation, metastable states with kinetic locking under a given T_{cutoff})

Automated Two-fold sampling

In practise - automated workflow

- Replication of the box in x-y plane (PBC)
- Grid of 100 equally spaced hemispheric centers
- First starting guess for the adsorbate barycenter position at 2 Å in z from each grid point
- Sampling of locally distinct binding sites
- Definition of the substrate-to-adsorbate starting orientation(s) based on the adsorbate symmetry
- 3 starting orientations for NH₃ & CO
- Unique starting orientation for CH₄ (very symmetric)
- Sampling of the local PES within each locally distinct site
- Atoms grouped by ONIOM layers
- Input written + submitted

Influence of the ice model density

The same sampling has been applied on the HDA model. Global statistics are conserved, even for the NH₃ GMM profile.

Results are well-reproducible, with high stability w.r.t. the ice density.

Perspectives

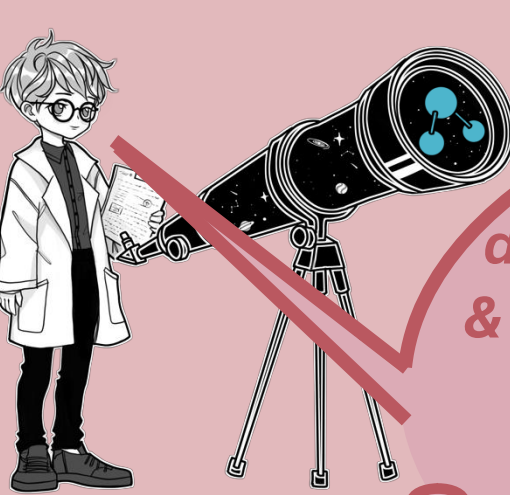
- From vibrational frequency results, exact the desorption pre-exponential factors ($k_{des} = A \cdot e^{-BE/RT}$)
- What about such distributions on CO ices?
- Inclusion in an astrochemical model

Scientific context

Dense molecular clouds:

- Birthplace of stars and planets
- Astrochemically very efficient

Great interest in the investigation of the exogenous origin of prebiotic molecules



With T of 10-20 K, densities of 10^3 - 10^6 part./cm³, & almost complete UV extinction within the cloud interior... How can they be efficient chemical factories under such conditions?

Slow gas phase kinetics BUT presence of dust grains (1 %w/w w.r.t gas) surrounded by an icy mantle (molecular clouds & outer regions of protoplanetary disks)

Heterogeneous catalyst; third body (energy excess) & hub to put in close proximity potential reaction partners

Icy mantles have been proven to be water-rich & amorphous in molecular clouds

Distribution of Binding Energies (BE) per species required

Unique value used in most current codes

No consensus on the structure of the ice model nor the BE inference scheme



Binding Energy Distributions Inference on M. Groyne^{1,2}, B. Champagne², C. Baijot¹, M. De Becker¹

[1] ESO/APEX (MPIFR/ESO/OSO)/Hacar, A. et al./Digitized Sky Survey 2.

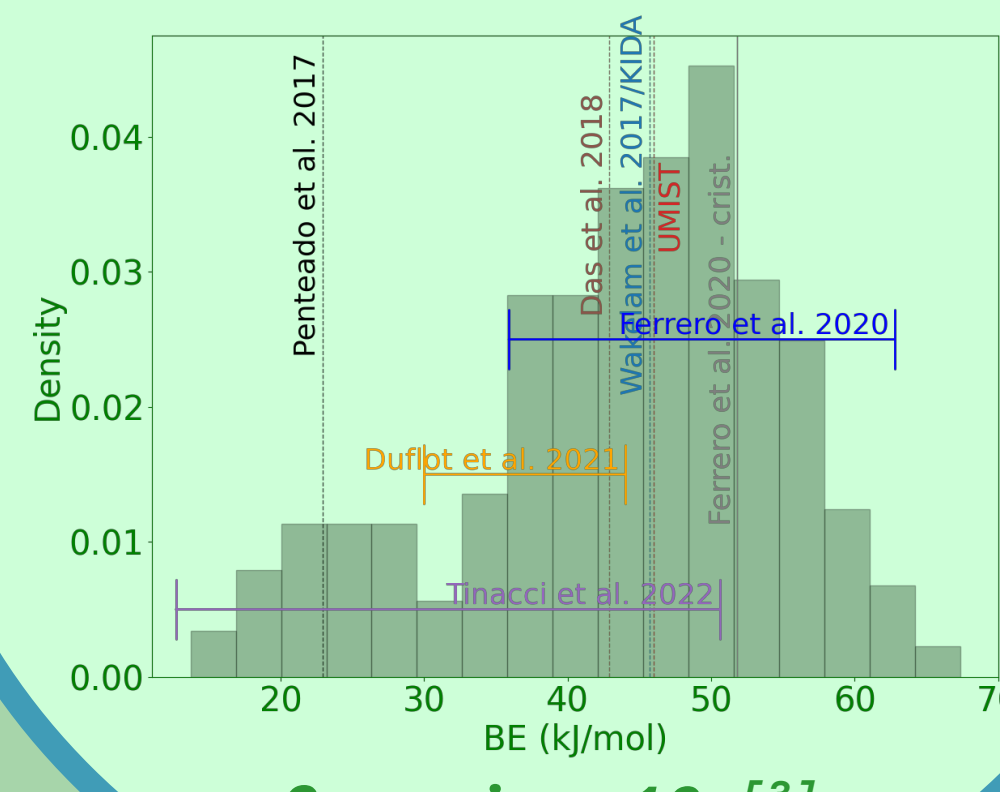
[2] Phillips, J.C. et al., 2020, J. Chem. Phys., 153

[3] Frisch, M.J. et al., 2016, Gaussian 16

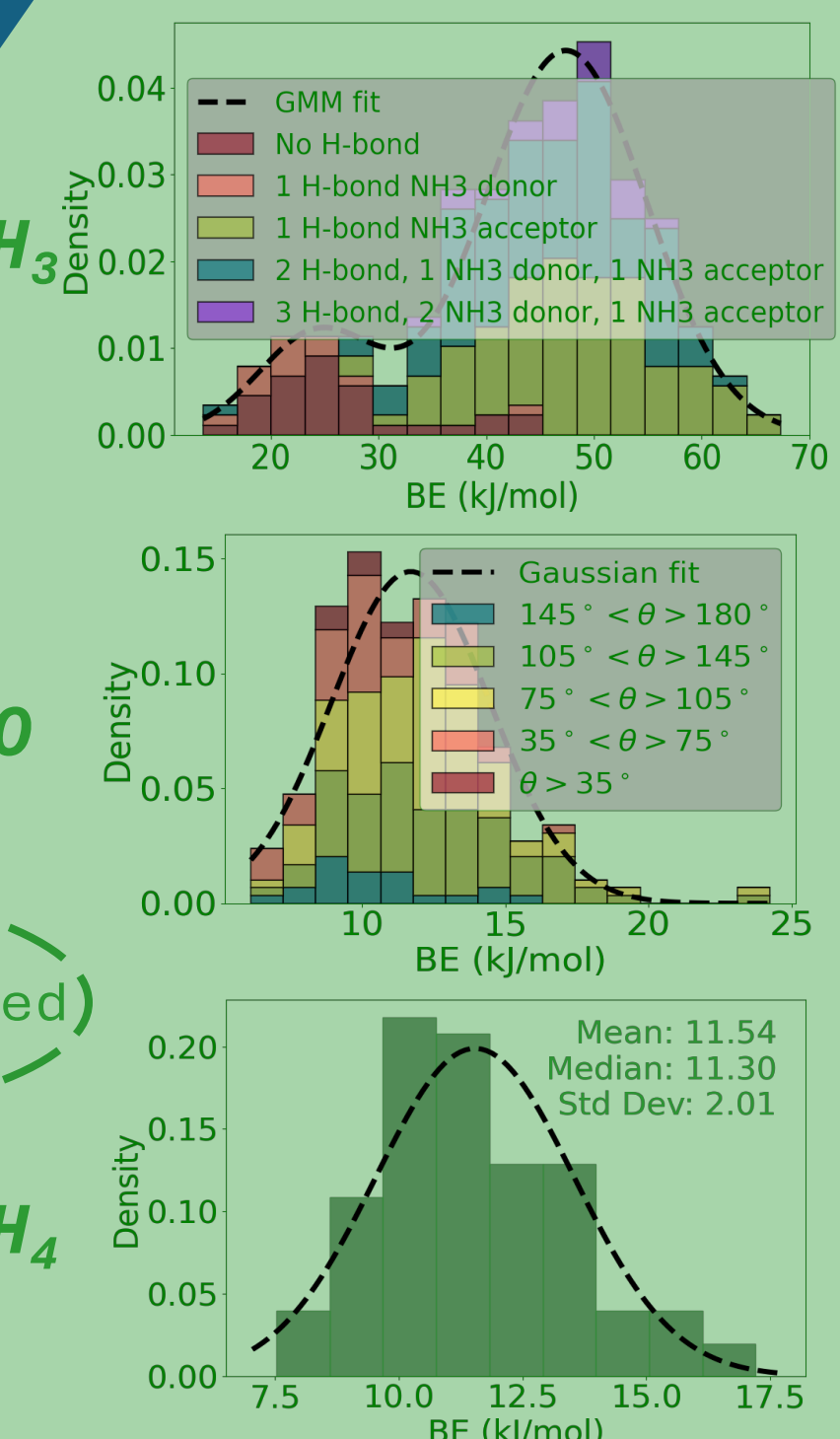
[4] Tinacci, L. et al., 2022, ACS Earth Space Chem, 6

[5] Bannwarth, C. et al., 2019, WIREs Comput. Mol. Sci., 11

Results & perspectives



PDFs on LDA ice

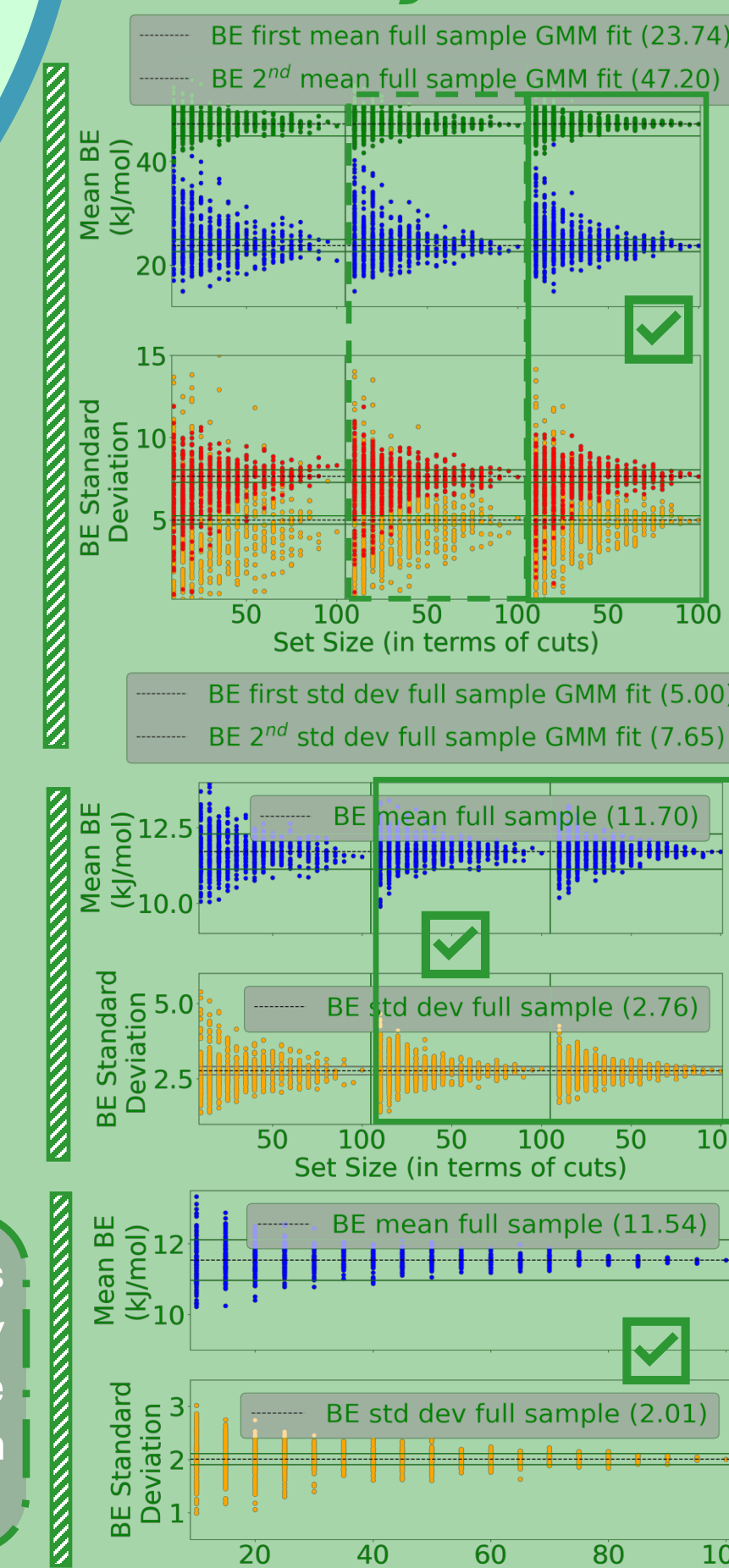


- Gaussian Mixture Model (GMM)
- 1st peak - config. without H-bond
- 2nd peak - config. with NH₃ at least H-bond acceptor

- $\theta = \arccos$ of the angle between the C-O bond axis and the z-reference vector
- θ [75-145]° favored

General note - These results are encompassing previously reported results, which were disparate or presenting non entirely overlapping ranges.

Convergence Analysis



Acknowledgments

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