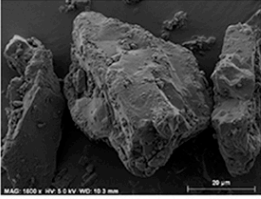


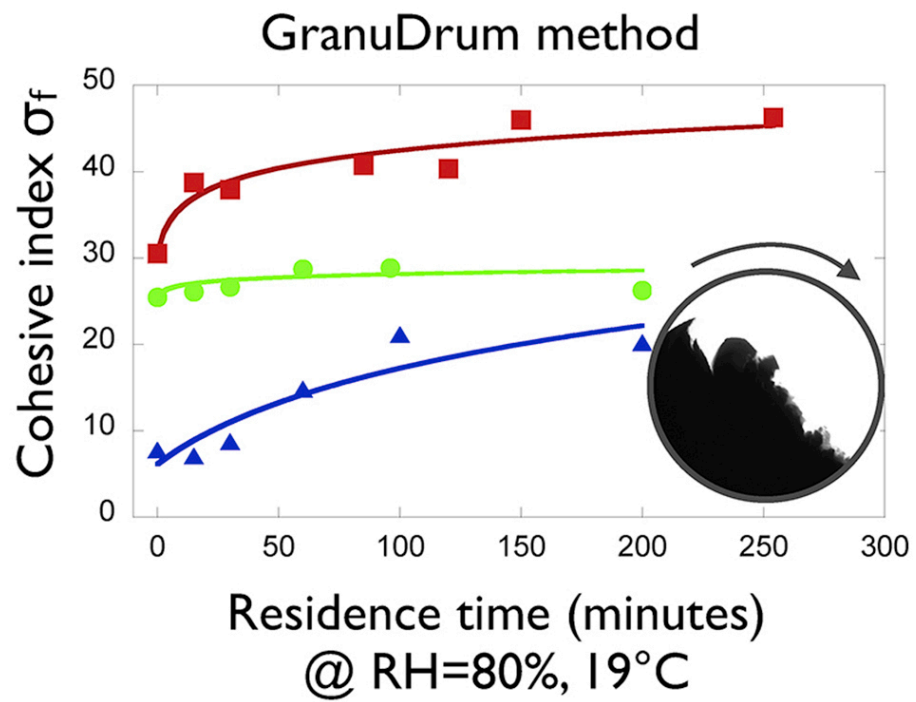
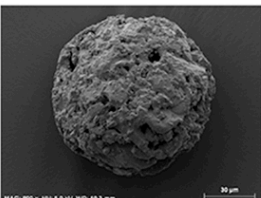
■ GranuLac 140



● InhaLac 230



▲ FlowLac 100



1 Effect of relative air humidity on the flowability of
2 lactose powders

3 G. Lumay^{a,b,*}, K. Traina^d, F. Boschini^{a,c}, V. Delaval^a, A. Rescaglio^a, R.
4 Cloots^{a,c}, N. Vandewalle^{a,b}

5 ^a*APTIS, University of Liege, Sart-Tilman, B-4000 Liège, Belgium*

6 ^b*GRASP, Institute of Physics, University of Liege, B-4000 Liège, Belgium.*

7 ^c*GreenMat, Institute of Chemistry, University of Liege, B-4000 Liège, Belgium*

8 ^d*Galephar MF, R&D center, B-6900 Marche en Famenne, Belgium*

9 **Abstract**

10 Moisture is known to affect the flowing properties of powders. However,
11 the quantification and the understanding of the observed effects are far to
12 be obvious. To study air moisture influence on powder flowability in a labo-
13 ratory, a conditioning system and a precise flowmeter are necessary. Simple
14 flow testers, which are commonly used R&D laboratories, are not able to
15 quantify precisely the effect of moisture on powders. The use of advanced
16 techniques is necessary. In this paper, we present how a precise and user-
17 friendly flowability test associated with a dynamic conditioning system can
18 be used to quantify the influence of moisture on the flowability properties of
19 lactose powders. The effect of residence time T in high humidity conditions
20 is analyzed. Afterward, we show that the relative humidity range that opti-
21 mizes the flowability of lactose powders is between 30% and 50%. For lower
22 values of the relative humidity, the apparition of electric charges inside the
23 bulk induces cohesive forces. For higher relative humidity, the condensation
24 of the air humidity at the contact between the grains forms capillary bridges

*Corresponding author, Email: geoffroy.lumay@ulg.ac.be, Tel: 0032 4 366 44 21

25 which favors also the cohesive interactions. A model taking both triboelectric
26 and capillary effects into account is proposed to fit the experimental data.

27 *Keywords:* Cohesion, Flowability, Humidity, moisture, Electric charges,
28 Lactose

29 **1. Introduction**

30 Granular materials, fine powders and nanopowders are widely used in
31 industrial applications [1, 2]. In particular, lactose powders are commonly
32 used as an excipient for dry powder pharmaceutical formulations (dry powder
33 inhaler, tabs, capsules). Indeed, lactose is inert, non-toxic and cheap. There-
34 fore, any progress in the understanding of lactose powders behaviors could
35 have huge consequences for pharmaceutical industries for the optimization
36 of industrial processes or to avoid technical issues (caking, clogging, non-
37 compliance or unconformity of the by-product). In order to control and to
38 optimize processing methods, these materials have to be precisely character-
39 ized. This characterization methods is related either to the properties of the
40 grains or to the behavior of the assembly of grains. The relation between
41 physico-chemical (size, shape, crystallinity, *etc.*) grain properties and the
42 macroscopic behaviors (flowability, density, cohesiveness, *etc.*) of the powder
43 is far to be obvious. Therefore, both measurement types have to be per-
44 formed. Many advanced methods are available to measure physico-chemical
45 grain characteristics: laser diffraction to obtain the grain size distribution,
46 morphometer to measure grains shape, X-ray diffractometer to characterize
47 crystallinity, ... However, concerning the physical behaviors of an assembly
48 of grains, most of the techniques used in R&D or control quality laborato-

49 ries are based on old measurement techniques [3]. During the last decades,
50 some interesting techniques have been developed like shear cells and pow-
51 der rheometers. However, the evolution of this field is still at his beginning.
52 Indeed, even at the fundamental point of view, the determination of the
53 physical laws that govern the behavior of a granular material is still a matter
54 of intense debates in the physics community [4, 5].

55 Recently, the rotating drum measurement method has been revisited ac-
56 cording to the present fundamental knowledge [6]. The existence of the
57 optimized equipment called GranuDrum opens new perspectives due to both
58 its precision and conceptual simplicity. This equipment measures the flowing
59 angle inside a rotating drum as a function of the rotating speed. The cohe-
60 sion between the grains is also estimated through the fluctuations of the flow
61 free surface. Figure 1 shows two typical flows in the rotating drum: (left) a
62 non cohesive flow and (right) a cohesive flow. In contrary to the shear tester
63 where the powder is loaded, the rotating drum is a free flow test. The free
64 flowing behavior inside the rotating drum is comparable with powder dy-
65 namics inside many industrial processes such as conveyor, mixer, granulator,
66 *etc.*

67 Granular materials behavior is influenced by (i) steric repulsions, (ii) fric-
68 tion forces (iii) cohesive forces and (iv) interaction with the surrounding
69 fluid (gaz or liquid). The steric repulsion is related to the grains geometry.
70 Friction forces are influenced by both surface state (rough or smooth sur-
71 face) and chemical nature of the grains. Cohesive forces may be induced by
72 the presence of liquid bridges [7, 8, 9], electric charges [10], van der Waals
73 interactions [11] or magnetic dipole-dipole interactions [12, 13, 14, 15, 16].

74 The predominance of one of these forces depends on both the environmental
75 conditions and the physicol-chemical properties of the grains.

76 Moisture is known to affect both static and dynamic behaviors of a gran-
77 ular materials [8, 9, 18, 19]. However, the effect of moisture is far to be
78 obvious. Indeed, moisture influences surface grains conductivity, capillary
79 bridges formation and relative air permittivity [20]. For low relative air hu-
80 midity, the electrical conductivity necessary for charge dissipation is reduced
81 and the relative permittivity comes close to vacuum permittivity. Therefore,
82 the electric charges created by triboelectric effects lead to uncontrolled elec-
83 tric field, electrostatic forces between the grains and/or between the grains
84 and the container. For high relative air humidity, the electrical conductivity
85 increases, liquid bridges may be formed at the contacts between the grains
86 and the relative air permittivity increases. Therefore, the electrical charges
87 are dissipated more easily. However, the apparition of liquid bridges induces
88 cohesive forces inside the packing. At intermediate relative humidity values
89 the cohesion is expected to decrease. In addition to these effects, the humid-
90 ity could also modify the chemical characteristics of the grain surface [21],
91 the strength of the Van des Waals interactions [22] and the friction due to
92 lubrication effects [23].

93 To measure the effect of relative air humidity on the flowability of a
94 powder sample in a laboratory, a conditioning system is necessary before
95 the flowability measurement. Powders are usually conditioned with a static
96 method. The sample is generally poured in a container, which is put in
97 a climate chamber. The exchanges between the air and the powder bulk
98 is limited. Indeed, only the upper part of the powder bed can have ex-

99 change with the controlled atmosphere. In the worse case, a crust forming
100 phenomenon prevents access of the humidity to the powder situated in the
101 bottom of the container. Moreover, the static approach may generate physi-
102 cal and/or chemical heterogeneity in the sample, leading to artefacts during
103 the following measurement. In the present study, the powder is conditioned
104 at controlled relative air humidity in a slowly rotating reactor to optimize
105 the exchanges between the surface grain and the moisture.

106 Watling *et al* [21] have found a significative aging of lactose powders after
107 an exposure at 75% RH, 40°C during six months. The observed effect was
108 mainly related to a dissolution of the smallest particles covering the big ones
109 leading to a modification of the surface state of the coarse lactose. This effect,
110 which needs long time exposure, is not expected in the present study. The
111 influence of shorter residence times of pharmaceutical powders in controlled
112 humidity conditions has been investigated Crouter *et al* [24] (48h) and by
113 Emery *et al* [9] (24h). In particular, Emery *et al* have shown that simple
114 flow testers (tapped density test, angle of repose measurement, flow through
115 an aperture, ...) are not able to quantify the effect of moisture on lactose
116 powders. Only the Jenike shear test was found to give acceptable results.
117 Unfortunately, this test is difficult to operate and time consuming. Therefore,
118 to quantify efficiently the effect of moisture on powders, a reliable and user-
119 friendly flow tester is necessary in addition to the simple tests and to the
120 shear cell measurements.

121 In the present paper, we show how relative air humidity influences lactose
122 powder flowability after a short residence time (only a few hours). The
123 powder samples are first conditioned at controlled relative air humidity in

124 a rotating reactor to optimize the exchanges between the surface grain and
125 the moisture. Afterward, the flowability is measured with the GranuDrum
126 equipment [6].

127 **2. Materials and methods**

128 *2.1. Lactose powders*

129 Three lactose powders produced by the company Meggle and widely used
130 in pharmaceutical industries have been used: GranuLac 140 (G140), InhaLac
131 230 (I230) and FlowLac 100 (F100). All the samples are α -lactose monohy-
132 drate powders. Granulac powders come from milling process. Inhalac pow-
133 ders are milled and sieved lactose powders. Finally, FlowLac powders are
134 obtained from spray-dried lactose suspensions. Powders have been selected
135 in order to have similar grain size distributions but resulting from different
136 process.

137 *2.1.1. Particle sizing*

138 The grain size distributions have been measured with the dry method
139 with a laser diffraction particle size analyzer (Malvern, Mastersizer Sirocco
140 2000). The measurements have been made with an air injection pressures
141 fixed at 1.2 bar. The refractive index of particles was fixed to 1.5 and the
142 default model was selected.

143 *2.1.2. Scanning electron microscopy*

144 Scanning Electronic Microscopy (Philips ESEM XL30 FEG, Eindhoven,
145 NL) with an acceleration voltage of 5kV was used to analyse qualitatively

146 the grains morphology. Samples were deposited on carbon tapes. Sputter-
147 ing deposition was done with gold target under argon atmosphere (Balzers,
148 SCD004, Sputter coater).

149 *2.2. Conditioning*

150 Before the flowability measurement, the powder is conditioned during a
151 time T in a slowly rotating reactor. The reactor is a 316L stainless steel tube
152 of internal diameter $D = 48\text{mm}$ and length $L = 320\text{mm}$. The tube axis is
153 placed horizontally. The tube is rotating around its axis at 0.66 RPM. A
154 flux $F = 0.3\text{ l/min}$ of air with controlled relative humidity flows through the
155 reactor. The humidity of the air passing through the reactor is adjusted by
156 varying the flow ratio of dry air and humidified air. The humidified air is ob-
157 tain by bubbling through water in a column. The slow motion of the reactor
158 allows the humid air to penetrate inside the bulk. Moreover, the low value
159 of the rotating speed avoid granulation. The accessible values of the relative
160 humidity range from $\text{RH}=0\%$ to $\text{RH}=95\%$. Directly after this preparation,
161 the drum cell is half filled by the powder and is closed hermetically. All the
162 manipulations have been performed at room temperature ($19^\circ\text{C} \pm 1^\circ\text{C}$).

163 *2.3. Flow measurement method*

164 Experimentally, the most practical geometry to study a free flow of gran-
165 ular materials is the rotating drum. This flow geometry has been extensively
166 studied with non cohesive granular materials [25, 26, 27, 28] (see Figure
167 1(left)) and with cohesive powders [29, 30, 31] (see Figure 1(right)). In the
168 case of cohesive grains, the flow is intermittent in the whole range of rotating
169 speed.

170 The experimental setup used to perform the flowability measurement is
171 the GranuDrum from the company GranuTools. This measurement methods
172 has been presented in detail in a previous paper [6]. The measurement cell
173 is a horizontal aluminum cylinder of diameter $D = 84\text{mm}$ and length $L = 20$
174 mm with glass side walls. The drum is half-filled with the powder. The
175 cylinder rotates around its axis at an angular velocity Ω producing the flow
176 of the grains. For each angular velocity, 50 images of the rotating powder
177 bed separated by 0.5 s are recorded. The position of the air/powder interface
178 is measured by image processing and analysis. The average interface position
179 and the fluctuations around this average position are computed. From the
180 fluctuations of the interface, the standard deviation σ_f is calculated [6]. This
181 parameter, also called cohesive index, is directly related to the cohesion inside
182 the drum. Other parameters like flowing angle, aeration, thixotropy, first
183 avalanche angle can be measured [6, 34]. In the present paper, we are focused
184 on the analysis of powder cohesiveness through the cohesive index σ_f .

185 3. Results

186 Typical scanning electron micrographs of lactose grains are presented in
187 Figure 2 (top). Granulac and Inhalac grains show angular edges due to
188 the milling process. Flowlac grains have roughly spherical shapes related to
189 the spray drying production process from lactose suspension. The grain size
190 distributions obtained by laser diffraction are presented in Figure 2 (bottom)
191 and the main descriptors extracted from these distributions are summarized
192 in Table 1. The presented distributions are corresponding to an air pressure
193 of 1.2 bars. The main parameter selected to characterize the average grain

194 size in the present study is the volume mean diameter (de Brouckere Mean
195 Diameter) $D[4/3]$. Granulac 140 powder has a significative fraction of fine
196 grains and the lowest average size. Inhalac 230 powder size distribution is
197 mainly a sharp peak around $100 \mu\text{m}$. Finally, the spherical Flowlac 100
198 grains have the highest average size.

199 The cohesive index σ_f measured with GranuDrum instrument for differ-
200 ent rotating speeds with the powder extracted directly after the first opening
201 from a new container are presented in Figure 3. These results show that the
202 powders considered in the present study have different flow behaviors and
203 are covering a wide range of cohesiveness. Flowlac 100 is characterized by
204 low cohesive indices σ_f whatever the range of rotating speed. At the oppo-
205 site, Granulac 140 powder has the higher cohesive indices σ_f at slow rotating
206 speeds but the cohesive index decreases for higher rotating speeds. Results
207 obtained at slow rotating speeds regarding powder cohesiveness can be di-
208 rectly related to grain sizes. Indeed, when the average grain size decreases
209 ($D[4/3]_{F100} > D[4/3]_{I230} > D[4/3]_{G140}$), the influence of the cohesive forces
210 (van der Waals, electrostatic and capillary forces) on the powder bed behav-
211 ior increases. A deeper analysis of the curves obtained with the GranuDrum
212 instrument allows to obtain additional information about powder rheology.
213 For exemple, a decrease of the cohesive index σ_f with the rotating speed is
214 observed for Granulac 140. In others words, the flowability of the Granulac
215 140 is improved as the shear rate increases, corresponding to a shear-thinning
216 behavior. We have previously demonstrated that this effect is due to pow-
217 der aeration during high speed flows [34]. This observation evidences the
218 necessity of deep analysis when powder flowability is considered.

219 In order to study the temporal evolution of powder cohesiveness when
220 submitted to high humidity conditions, a series of GranuDrum measurement
221 have been performed with different residence times T in the conditioning
222 system. For each measurement, a sample of 80ml coming directly from the
223 box was conditioned in the rotating reactor with a relative humidity fixed at
224 80%RH and 19°C. After a residence time T , the sample is extracted directly
225 from the reactor and the GranuDrum measurement cell is half filled with
226 55ml of powder. Afterward, a new sample is taken from the original container
227 to prepare the next measurement. Figure 4 shows the temporal evolution of
228 powder cohesiveness when submitted to high humidity conditions (80% RH).
229 The intermediate drum rotating speed of 8RPM has been selected to plot the
230 results. The relative air humidity does not affect equally the different lactose
231 powders. GranuLac and FlowLac cohesiveness increases significantly with
232 the residence time while InhaLac is less sensitive. After an increase, the
233 cohesive index reaches a plateau. The modification of powders cohesiveness
234 is quite fast. Indeed, the cohesive index is already modified after less than
235 one hour in the reactor.

236 To measure the influence of the relative humidity RH on powder cohesiveness,
237 a residence time of $T = 150$ minutes has been selected. This residence
238 time corresponds roughly to the beginning of the plateau observed in Figure 4.
239 Figures 5 shows the influence of relative humidity RH on powder
240 cohesiveness for the three selected lactose powders. The cohesion is found to
241 decrease slightly in the range of relative humidity from 0% to 50%. Therefore,
242 the cohesive forces inside the bulk decrease and the flowability is enhanced.
243 Afterward, the cohesion increase importantly from $RH=50\%$ to $RH=100\%$.

244 Then, the relative humidity that optimizes the flowability of lactose grains is
245 situated between 30% and 50%. For low humidity conditions, the cohesion is
246 induced by the apparition of electric charges on the surface of powder grain.
247 For the extremal case $RH=0\%$ this effect was particularly important. Indeed,
248 the powder was sticking on all surfaces. During GranuDrum measurement,
249 the analysis of the flow was impossible for rotating speed higher than 8RPM
250 due to this sticking effect. For this reason, the rotating drum speed of 8RPM
251 has been selected to perform the comparisons in the whole present paper.
252 For high humidity conditions, the condensation of water at the surface of the
253 grains leads presumably to the formation of capillary bridges at the contact
254 between the grains, increasing powder cohesiveness. For the short residence
255 times considered in the present study, we do not expect a water absorption
256 inside the grains.

257 These experimental results show the importance of moisture and its con-
258 trol on powder flowing properties. For example, the cohesive index of FlowLac
259 100 samples directly from a new container is $\sigma_f^{8RPM} = 7.7$. Therefore, the
260 flowability of this powder can be considered as good (almost excellent) [34].
261 After a residence time $T = 200$ minutes at high humidity conditions, the
262 cohesive index becomes $\sigma_f^{8RPM} = 20.2$. This cohesive index corresponds
263 to a powder with passable flowing properties. Therefore, moisture modifies
264 drastically the flowing properties of this powder even during a short-term
265 exposition.

266 3.1. Measurement uncertainties

267 In order to estimate the uncertainties on the measured cohesive index, the
268 measurements have been repeated six times with a residence time $T = 150$

269 minutes at a relative humidity of $RH = 95\%$. For each measurement, we
270 started with a new sample coming from the box, conditioned at controlled
271 humidity and analyzed with the rotating drum method. The error bars pre-
272 sented in Figure 5 correspond to the standard deviation calculated from the
273 six measurements. Typically, the uncertainty on the cohesive index corre-
274 sponds to about 10% of its average value.

275 4. Discussion

276 The modification of powder cohesiveness is related to a combined effect
277 of capillary bridges induced by air moisture and triboelectric charges. In
278 the present discussion, a model combining (i) adhesion forces F_{adh} induced
279 by capillary bridges and (ii) electrostatic forces F_e between the grains is
280 proposed to fit the experimental results. We assume that the powder cohe-
281 siveness measured by GranuDrum instrument is directly related to the sum
282 of the cohesive forces acting between the grains. This model has been used
283 previously to fit experimental data related to taped density measurement
284 [18].

285 The effect of moisture on the first avalanche angle in a rotating drum
286 has been investigated by Crassous *et al.* [35]. A model based on capillary
287 condensation of water vapor between the grains is proposed to explain the
288 modification of the avalanche angle due to moisture. By assuming that the
289 kinetics of the process is governed by thermally activated nucleation of cap-
290 illary bridges, Crassous *et al.* were able to reproduce both time and relative
291 humidity dependence of the process. In this model, the adhesion force F_{adh}
292 between two contacting grains due to capillary bridges formation during a

293 period T at a relative humidity RH can be expressed as

$$F_{adh}(T, RH) \approx \frac{1}{\ln\left(\frac{100}{RH}\right)} \ln\left(1 + \frac{T}{\tau}\right), \quad (1)$$

294 where τ is a characteristic time.

295 Concerning the electrostatic charges, it has been found empirically that
 296 the electrical charge q on beads decreases exponentially as a function of the
 297 RH with a characteristic rate 30%, i.e., $q = q_0 \exp(-RH/30)$ [36]. This
 298 particular decrease is associated with microscopic capillary condensation on
 299 the beads which neutralizes a fraction of the charges. The electric force
 300 associated with this charge is proportional to q^2 and therefore decays as

$$F_e(RH) \approx \exp(-RH/15). \quad (2)$$

301 Associating both triboelectric and capillary effects, we obtain following
 302 relation for powder cohesiveness (assuming that the powder cohesiveness σ_f
 303 measured with the GranuDrum instrument is directly proportional to cohe-
 304 sive forces):

$$\begin{aligned} \sigma(T, RH) = & \sigma_0 + \sigma_e \exp(-RH/15) \\ & + \sigma_c \frac{1}{\ln\left(\frac{100}{RH}\right)} \ln\left(1 + \frac{T}{\tau}\right), \end{aligned} \quad (3)$$

305 where σ_0 is the intrinsic cohesion inside the powder without any charges and
 306 capillary bridges, σ_e is the electrostatic contribution to cohesion and σ_c is the
 307 capillary contribution to cohesion. This relation gives the temporal evolution
 308 of powder cohesiveness and also the effect of relative humidity RH . The pow-
 309 der cohesiveness temporal evolution in high humidity conditions presented in

310 Figure 4 is well fitted by this model with a logarithmic saturation. Moreover,
311 the decrease of powder cohesiveness followed by an increase as a function of
312 relative humidity (see Figure 5) is also fitted by the model. Despite the sim-
313 plicity of this model regarding real powder complexity, this model fits quite
314 well the set of measurements.

315 5. Conclusion

316 We have shown that the relative air humidity affects strongly the free
317 flow of lactose powders. Moreover, the dynamics is quite fast. Indeed, an
318 increase of cohesiveness is observed after less than one hour in high humidity
319 conditions if powder is conditioned in dynamic approach. However, a precise
320 flowability characterization method is necessary to measure this influence in
321 a laboratory. The GranuDrum instrument is found to be a good method for
322 this kind of study.

323 Cohesion in powders appears for both dry and wet conditions. When
324 the air relative humidity is low, grain-grain friction inside the flow induces
325 the appearance of electrical charges in the packing. The electrical charges
326 induce cohesive forces between the grains. The condensation in high humidity
327 conditions implies the formation of liquid bridges between contacting grains
328 which induce also cohesive forces. Taking both effects (triboelectricity and
329 capillary) into account, the cohesion is minimized in intermediate values of
330 the relative humidity, i.e. between $RH=30\%$ and $RH=50\%$.

331 Therefore, it is important to take these effects into account in industrial
332 processes and laboratory where surrounding is not always under control (sea-
333 sonal effect) or for long process where initial powders have not the same time

334 residence, and, so, the same physical and/or chemical properties as the final
335 powders being on the production line.

336 **6. Acknowledgements**

337 We Thanks Walloon region (Fonds de maturation - convention 1318086)
338 for the financial support and the Cat μ (microscopy center of the University
339 of Liège) for the SEM micrographs. Thanks to Meggle company for sending
340 lactose powder samples.

341 **References**

342 **References**

- 343 [1] M.J. Donovan, H.D.C. Smyth, *Influence of size and surface roughness*
344 *of large lactose carrier particles in dry powder inhaler formulations*, Int.
345 J. Pharm. **402** (2010) 1
- 346 [2] P. Ruenraroengsak, J.M. Cook, and A.T. Florence, *Nanosystem drug*
347 *targeting: Facing up to complex realities*, J. of Controlled Release **141**
348 (2010) 265
- 349 [3] European pharmacopoeia 7.0, Chapter 2.9.36. : Powder flow, p. 308
- 350 [4] M. Z. Miskin, and M. H. Jaeger, *Adapting granular materials through*
351 *artificial evolution*, Nature Materials **12**, 326 (2013)
- 352 [5] M. Bouzid, M. Trulsson, P. Claudin, E. Clément, and B. Andreotti,
353 *Nonlocal rheology of granular flows across yield conditions*, Phys. Rev.
354 Lett. **111**, 238301 (2013)

- 355 [6] G. Lumay, F. Boschini, K. Traina, S. Bontempi, J.-C. Remy, R. Cloots,
356 and N. Vandewalle, *Measuring the flowing properties of powders and*
357 *grains*, Powder Technology **224** (2012) 19
- 358 [7] A. Kudrolli, *Granular Matter - Sticky Sand*, Nature Materials **7**, 174
359 (2008).
- 360 [8] J. E. Fiscina, G. Lumay, F. Ludewig, and N. Vandewalle, *Compaction*
361 *Dynamics of Wet Granular Assemblies*, Phys. Rev. Lett. **105** (2010)
362 048001
- 363 [9] E. Emery, J. Oliver, T. Pugsley, J. Sharma, and J. Zhou, *Flowability of*
364 *moist pharmaceutical powders*, Powder Technology **189** (2009) 409
- 365 [10] E. Mersch, G. Lumay, F. Boschini, and N. Vandewalle, *Effect of an*
366 *electric field on an intermittent granular flow*, Phys. Rev. E **81** (2010)
367 041309
- 368 [11] J. M. Valverde, and A. Castellanos, *Random loose packing of cohesive*
369 *granular materials*, Europhysics Letters **75** (2006) 985
- 370 [12] A. J. Forsyth, S. R. Hutton, M. J. Rhodes, and C. F. Osborne, *Effect*
371 *of applied interparticle force on the static and dynamic angles of repose*
372 *of spherical granular material*, Phys. Rev. E **63** (2001) 031302
- 373 [13] J. M. Valverde, M. J. Espin, M. A. S. Quintanilla, and A. Castellanos,
374 *Magnetofluidization of fine magnetite powder*, Phys. Rev. E **79** (2009)
375 031306

- 376 [14] G. Lumay, and N. Vandewalle, *Controlled flow of smart powders*, Phys.
377 Rev. E **78** (2008) 061302
- 378 [15] G. Lumay, S. Dorbolo, and N. Vandewalle, *Compaction dynamics of a*
379 *magnetized powder*, Phys. Rev. E **80** (2009) 041302
- 380 [16] G. Lumay, and N. Vandewalle, *Flow of magnetized grains in a rotating*
381 *drum*, Phys. Rev. E **82** (2010) 040301
- 382 [17] B. Chaudhuri, A. W. A. Alexandar, A. Faqih, F. J. Muzzio, C. Davis, M.
383 S. Tomassone , *Avalanching flow of cohesive powders*, Powder Techno.
384 **164** (2006) 13-21
- 385 [18] N. Vandewalle, G. Lumay, F. Ludewig, and J. E. Fiscina, *How relative*
386 *humidity affects random packing experiments*, Phys. Rev. E **85** (2012)
387 031309
- 388 [19] F. Podczek, J. M. Newton, M. B. James, *The influence of constant and*
389 *changing relative humidity of the air on the autoadhesion force between*
390 *pharmaceutical powder particles*, Int. J. Pharm. **145** (1996) 221
- 391 [20] M. Santo Zarnik, D. Belavic, *An Experimental and Numerical Study of*
392 *the Humidity Effect on the Stability of a Capacitive Ceramic Pressure*
393 *Sensor*, Radioengineering **21** (2012) 201
- 394 [21] C.P. Watling, J.A. Elliott, C. Scruton, R.E. Cameron, *Surface modifica-*
395 *tion of lactose inhalation blends by moisture*, Int. J. Pharm. **391** (2010)
396 29

- 397 [22] M.C. Coelho, N. Harnby, *The effect of humidity on the form of water*
398 *retention in a powder*, Powder Technology **20** (1978) 209
- 399 [23] Q. Xu, A. V. Orpe, and A. Kudrolli, *Lubrication effects on the flow of*
400 *wet granular materials*, Phys. Rev. E **76** (2007) 031302
- 401 [24] A. Crouter, and L. Briens, *The effect of moisture on the flowability of*
402 *pharmaceutical excipients*, AAPS PharmSciTech. **15** (2014) 65-74
- 403 [25] J. Rajchenbach, *Flow in powders: From discrete avalanches to continu-*
404 *ous regime*, Phys. Rev. Lett. **65** (1990) 2221
- 405 [26] R. Fischer, P. Gondret, and M. Rabaud, *Transition by Intermittency in*
406 *Granular Matter: From Discontinuous Avalanches to Continuous Flow*,
407 Phys. Rev. Lett. **103** (2009) 128002
- 408 [27] N. Taberlet, P. Richard, and E. J. Hinch, *S shape of a granular pile in*
409 *a rotating drum*, Phys. Rev. E **73** (2006) 050301
- 410 [28] Xiao Yan Liua, E. Spechta, and J. Mellmann, *Experimental study of the*
411 *lower and upper angles of repose of granular materials in rotating drums*
412 Powder Technology **154** (2005) 125 ? 131
- 413 [29] A. W. Alexander, B. Chaudhuri, A. Faqih, F. J. Muzzio, C. Davies and
414 M. S. Tomasson, *Avalanching flow of cohesive powders*, Powder Technol.
415 **164** (2006) 13
- 416 [30] A. Castellanos, J.M. V., A.T. Perez, A. Ramos and P.K. Watson, *Flow*
417 *regimes in fine cohesive powders*, Phys. Rev. Lett. **82** (1999) 1156

- 418 [31] M. A. S. Quintanilla, J. M. Valverde and A. Castellanos, *The transitional*
419 *behaviour of avalanches in cohesive granular materials*, J. Stat. Mech.
420 (2006) P07015
- 421 [32] R. P. Hegde , J. L. Rheingold, S. Welch, and C. T. Rhodes, *Studies of*
422 *powder flow using a recording powder flowmeter and measurement of the*
423 *dynamic angle of repose*, J Pharm Sci. (1985) **74** 11
- 424 [33] V. R. Nalluri, and M. Kuentz, *Flowability characterisation of drug-*
425 *excipient blends using a novel powder avalanching method*, Eur. J. of
426 Pharmaceutics and Biopharmaceutics **74** (2010) 388
- 427 [34] F. Boschini, V. Delaval, K. Traina, N. Vandewalle, and G. Lumay, *Link-*
428 *ing flowability and granulometry of lactose powders*, International Jour-
429 nal of Pharmaceutics **494** (2015) 312
- 430 [35] L. Bocquet, E. Charlaix, S. Ciliberto, and J. Crassous, *Moisture-induced*
431 *ageing in granular media and the kinetics of capillary condensation*, Na-
432 ture **396** 735 (1998)
- 433 [36] M. Rhodes, S. Takeuchi, K. Liffman, K. Muniandy, *The role of intersti-*
434 *tial gas in the Brazil Nut effect*, Granular Matter **5**, 107 (2003)

435 **List of Figures**

- 436 1 Two typical flows obtained with GranuDrum Instrument. (left)
 437 Regular flow obtained with the non cohesive powder FlowLac
 438 100. (right) Irregular flow obtained with the cohesive powder
 439 GranuLac 140. The powder cohesiveness measured with the
 440 GranuDrum instrument corresponds to the fluctuations of the
 441 air-interface position. 23
- 442 2 (Top) Typical SEM micrographs of the lactose grains consid-
 443 ered in the present study. (Bottom) Size distributions of the
 444 three powder samples obtained with a dry laser diffraction
 445 method for air dispersive pressure fixed at 1.2 bars. 24
- 446 3 Powder cohesiveness σ_f measured with the GranuDrum instru-
 447 ment as a function of the rotating speed. These measurements
 448 have been performed with powder samples taken in the box
 449 directly after the first opening. The results corresponding to
 450 GranuLac 140 (G140), InhaLac 230 (I230) and FlowLac 100
 451 (F100) lactose powders are presented. Each point correspond
 452 to an average over three measurements. Error bars are indi-
 453 cated but are sometimes smaller than the symbol size. 25
- 454 4 Powder cohesiveness σ_f measured with the GranuDrum in-
 455 strument at 8 RPM as a function of the residence time T in
 456 the reactor at a relative humidity of 80%. The results cor-
 457 responding to GranuLac 140 (G140), InhaLac 230 (I230) and
 458 FlowLac 100 (F100) lactose powders are presented. The solid
 459 lines are corresponding to a fit with Eq(3). 26

460 5 Powder cohesiveness σ_f measured with the GranuDrum in-
461 strument at 8 RPM as a function of the relative humidity RH
462 for a residence time in the reactor fixed at $T = 150$ minutes.
463 The results corresponding to GranuLac 140 (G140), InhaLac
464 230 (I230) and FlowLac 100 (F100) lactose powders are pre-
465 sented. The solid lines are corresponding to a fit with Eq(3).
466 Error bars corresponding to the standard deviation of six mea-
467 surements are presented for $RH=95\%$ 27

Table 1: Main granulometric descriptors extracted from grain size distributions obtained with the laser diffraction method in dry method for air dispersive pressure at 1.2 bars. The sizes are expressed in μm . Std is the standard deviation corresponding to the volume mean diameter (de Brouckere Mean Diameter) $D[4/3]$.

Powder	d(0.1)	d(0.5)	d(0.9)	$D[4/3]$	Std
G140	10,3	64,8	152,4	74,0	54,5
I230	58,6	97,5	152,2	101,0	38,9
F100	41,6	129,6	247,7	138,8	76,9

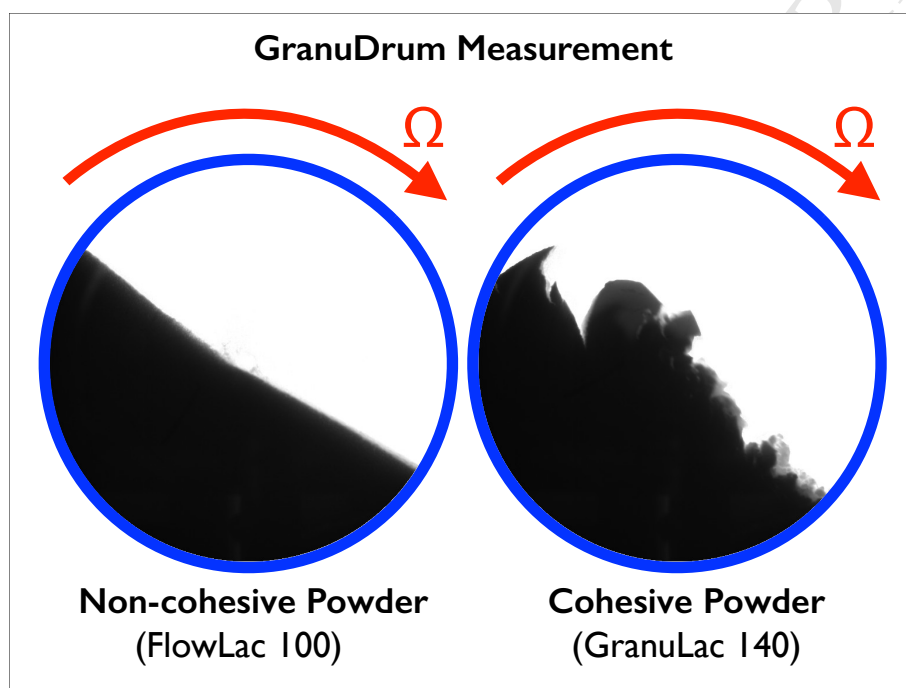


Figure 1: Two typical flows obtained with GranuDrum Instrument. (left) Regular flow obtained with the non-cohesive powder FlowLac 100. (right) Irregular flow obtained with the cohesive powder GranuLac 140. The powder cohesiveness measured with the GranuDrum instrument corresponds to the fluctuations of the air-interface position.

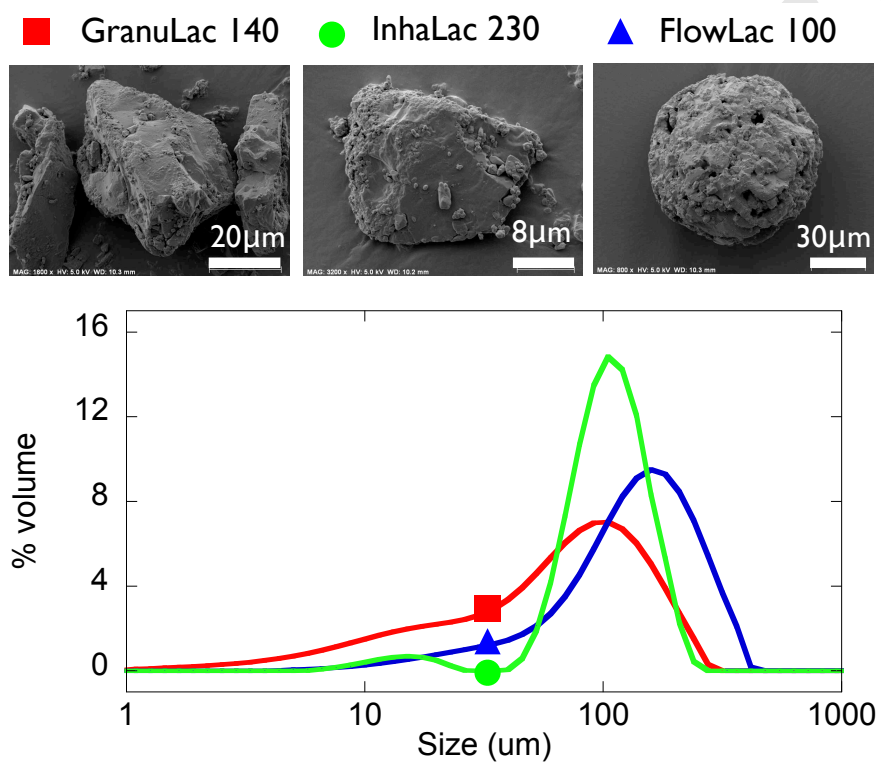


Figure 2: (Top) Typical SEM micrographs of the lactose grains considered in the present study. (Bottom) Size distributions of the three powder samples obtained with a dry laser diffraction method for air dispersive pressure fixed at 1.2 bars.

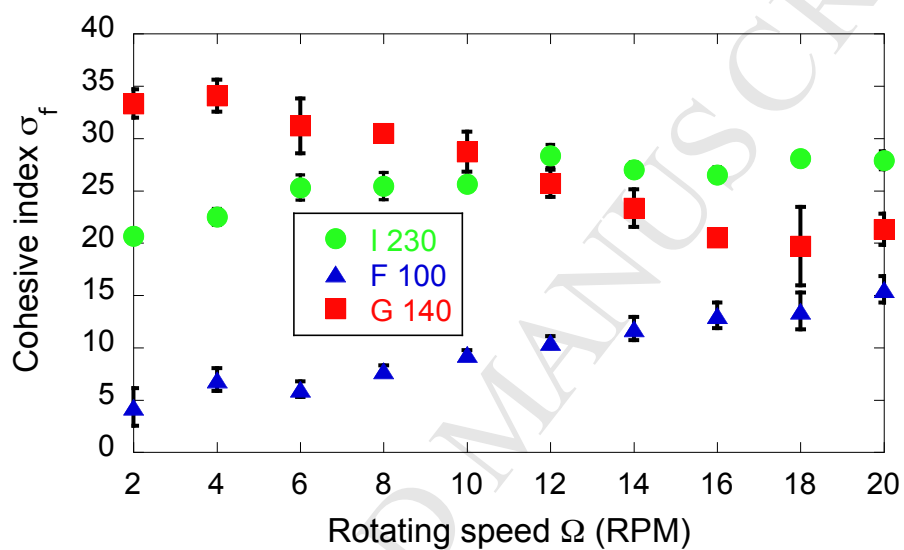


Figure 3: Powder cohesiveness σ_f measured with the GranuDrum instrument as a function of the rotating speed. These measurements have been performed with powder samples taken in the box directly after the first opening. The results corresponding to GranuLac 140 (G140), InhaLac 230 (I230) and FlowLac 100 (F100) lactose powders are presented. Each point correspond to an average over three measurements. Error bars are indicated but are sometimes smaller than the symbol size.

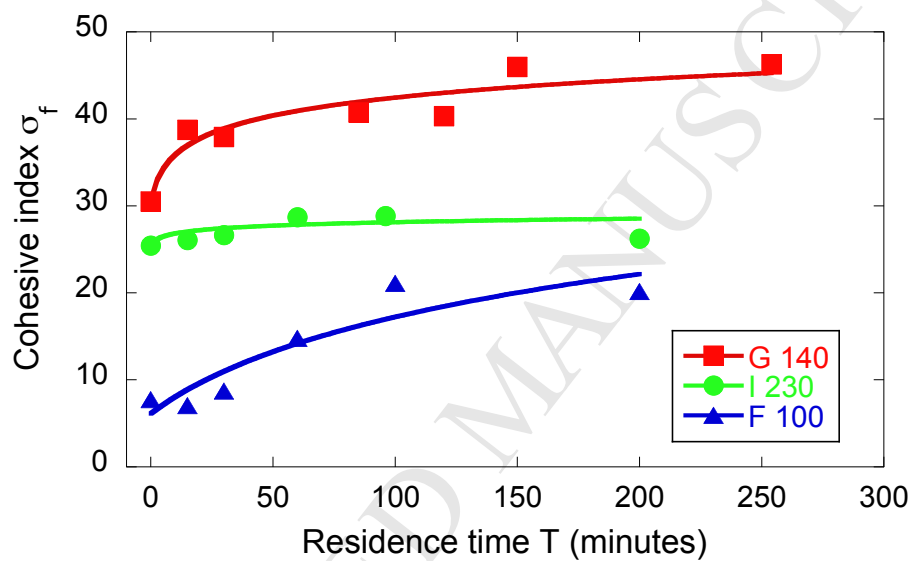


Figure 4: Powder cohesiveness σ_f measured with the GranuDrum instrument at 8 RPM as a function of the residence time T in the reactor at a relative humidity of 80%. The results corresponding to GranuLac 140 (G140), InhaLac 230 (I230) and FlowLac 100 (F100) lactose powders are presented. The solid lines are corresponding to a fit with Eq(3).

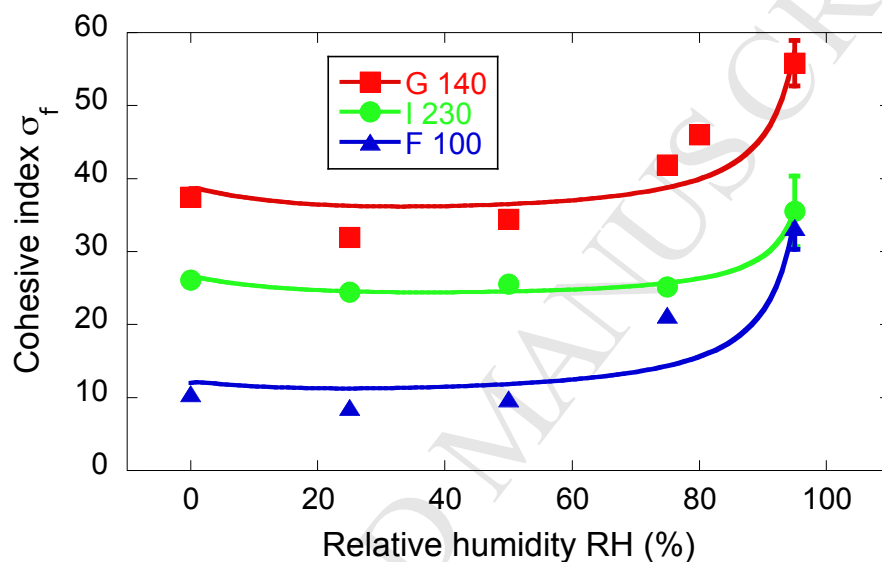


Figure 5: Powder cohesiveness σ_f measured with the GranuDrum instrument at 8 RPM as a function of the relative humidity RH for a residence time in the reactor fixed at $T = 150$ minutes. The results corresponding to GranuLac 140 (G140), InhaLac 230 (I230) and FlowLac 100 (F100) lactose powders are presented. The solid lines are corresponding to a fit with Eq(3). Error bars corresponding to the standard deviation of six measurements are presented for $RH=95\%$.