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Methane, carbon dioxide, and nitrous oxide emissions from two clear-water and two turbid-water urban ponds in Brussels (Belgium)

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Abstract. Shallow ponds can occur either in a clear-water state dominated by macrophytes or a turbid-water state dominated by phytoplankton, but it is unclear if and how these two alternative states affect the emission of greenhouse gases (GHGs) such as carbon dioxide (CO₂), methane (CH₄), and nitrous oxide (N2O) to the atmosphere. We measured the dissolved concentration of CO₂, CH₄, and N₂O from which the diffusive air-water fluxes were computed, in four urban ponds in the city of Brussels (Belgium): two clearwater macrophyte-dominated ponds (Silex and Tenreuken), and two turbid-water phytoplankton-dominated ponds (Leybeek and Pêcheries) on 46 occasions over 2.5 years (between June 2021 and December 2023). Ebullitive CH₄ fluxes were measured with bubble traps in the four ponds during deployments in spring, summer, and autumn, totalling 48 d of measurements. Measured ancillary variables included water temperature, oxygen saturation level (%O₂), concentrations of chlorophyll-a (Chl-a), total suspended matter (TSM), soluble reactive phosphorus (SRP), nitrite (NO_2^-) , nitrate (NO_3^-) , and ammonium (NH_4^+) . The turbid-water and clear-water ponds did not differ significantly in terms of diffusive emissions of CO₂ and N₂O. Clear-water ponds exhibited higher values of ebullitive CH₄ emissions compared to turbid-water ponds, most probably in relation to the delivery of organic matter from macrophytes to sediments, but the diffusive CH₄ emissions were not significantly different between clearand turbid-water ponds. Across seasons, CH₄ emissions increased with water temperature in all four ponds, with ebullitive CH₄ fluxes having a stronger dependence on water temperature (Q_{10}) than diffusive CH₄ fluxes. The temperature sensitivity of ebullitive CH4 fluxes decreased with increasing water depth, implying that shallow sediments would respond more strongly to warming (e.g. heat waves). Total annual CH₄ emissions (diffusive + ebullitive) in CO₂ equivalents equalled those of CO2 in turbid-water ponds and exceeded those of CO₂ in clear-water ponds, while N₂O emissions were negligible compared to the other two GHGs. Total annual GHG emissions in CO2 equivalents from all four ponds increased from 2022 to 2023 due to higher CO₂ diffusive fluxes, likely driven by higher annual precipitation in 2023 compared to 2022 (leading putatively to higher inputs for organic or inorganic carbon from run-off), possibly in response to the intense El Niño event of 2023. The findings of this work suggest that it might be necessary to account for the presence of submerged macrophytes when extrapolating ebullitive CH₄ fluxes in ponds at a larger scale (regional or global) (particularly if Chl-a is used as a descriptor), although it might be less critical for the extrapolation of diffusive CH₄, CO₂, and N₂O fluxes.

1 Introduction

Greenhouse gas (GHG) emissions from inland water (rivers, lakes, and reservoirs) to the atmosphere such as carbon dioxide (CO₂), methane (CH₄), and nitrous oxide (N₂O) are quantitatively important for global budgets (Lauerwald et al., 2023). Global GHG emissions from lakes are lower than from rivers for CO₂ (Raymond et al., 2013) and for N₂O (Lauerwald et al., 2019; Maavara et al., 2019). However, reported global emissions of CH₄ from lakes (Rosentreter et al., 2021; Johnson et al., 2022) are equivalent or even higher compared to rivers (Stanley et al., 2016; Rocher-Ros et al., 2023). Global emissions of CO₂ and CH₄ from lakes to the

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atmosphere represent 1.25 to 2.30 Pg CO₂ equivalents (CO₂eq) annually, with a significant proportion from CH₄ emissions, and represent nearly 20 % of global CO₂ emissions from fossil fuels (DelSontro et al., 2018). The contribution of CO₂ and CH₄ emissions from small lentic water bodies (small lakes and ponds) can be disproportionately high compared to large systems (Holgerson and Raymond, 2016) as small lakes and ponds are the most abundant of all water body types in number (Verpoorter et al., 2014; Cael et al., 2017), and fluxes per square metre are usually higher from smaller water bodies. The emissions of GHGs from artificial water bodies such as agricultural reservoirs, urban ponds, and storm-water retention basins could be higher than those from natural systems (Martinez-Cruz et al., 2017; Grinham et al., 2018; Herrero Ortega et al., 2019; Gorsky et al., 2019; Ollivier et al., 2019; Peacock et al., 2019, 2021; Webb et al., 2019; Bauduin et al., 2024a). These higher GHG emissions seem to result from higher external inputs of anthropogenic organic carbon and dissolved inorganic nitrogen (DIN) into artificial systems but might also reflect other differences compared to natural systems such as in hydrology (Clifford and Heffernan, 2018). Among artificial systems, urban ponds are the subject of a growing body of literature on GHG emissions (Singh et al., 2000; Natchimuthu et al., 2014; van Bergen et al., 2019; Audet et al., 2020; Peacock et al., 2021; Goeckner et al., 2022; Ray and Holgerson, 2023; Ray et al. 2023; Bauduin et al., 2024a). Urban areas can have numerous small artificial water bodies mostly associated with green spaces such as public parks, and their number is increasing due to rapid urbanization worldwide (Brans et al., 2018; Audet et al., 2020; Gorsky et al., 2024; Rabey and Cotner, 2024). Urban ponds are generally small and shallow, and usually their catchment consists of a majority of impervious surfaces with a smaller contribution from soils (Davidson et al., 2015; Peacock et al., 2021). In general, the main function of urban ponds is for storm-water management, but they provide additional benefits including aesthetic/recreational amenities and habitats for wildlife (e.g. Tixier et al., 2011; Hassall, 2014).

Shallow ponds and lakes occur in two alternative states corresponding to systems with either clear waters (macrophyte-dominated) or turbid waters (phytoplanktondominated) during the productive period of the year (spring and summer in mid-latitudes) (Scheffer et al., 1993). Submerged macrophytes and phytoplankton regulate CO2 dynamics directly through photosynthesis that can be more or less balanced by community respiration in the water column (e.g. Sand-Jensen and Staehr, 2007). However, it is not clear whether the presence of macrophytes increases or decreases the net CO₂ emissions from ponds and lakes. Some studies have shown a decrease in CO₂ emissions with increasing macrophyte density (Kosten et al., 2010; Ojala et al., 2011; Davidson et al., 2015), but other studies showed the opposite pattern (Theus et al., 2023). In phytoplankton-dominated lakes, CO₂ concentrations depend in part on the developmental stage of phytoplankton, with the growth and peak phases generally coinciding with lower CO₂ concentrations due to photosynthesis (Grasset et al., 2020; Vachon et al., 2020).

CH₄ emissions have been reported to increase with the concentration of chlorophyll-a (Chl-a) in phytoplanktondominated lakes (DelSontro et al., 2018; Borges et al., 2022). The presence of macrophytes strongly affects the production of CH₄ in freshwaters (Bastviken et al., 2023), and vegetated littoral zones of lakes exhibit higher CH₄ emissions than non-vegetated zones (Hyvönen et al., 1998; Huttunen et al., 2003; Juutinen et al., 2003; Desrosiers et al., 2022; Theus et al., 2023). Macrophytes influence organic matter decomposition processes in sediments depending on the quality and quantity of plant matter they release into their environment (Reitsema et al., 2018; Grasset et al., 2019; Harpenslager et al., 2022; Theus et al., 2023). Yet, few studies have consistently compared CH₄ emissions in clear-water and turbidwater ponds (Hilt et al., 2017). A study in Argentina reported higher dissolved CH₄ concentrations in clear-water ponds with submerged macrophytes compared to turbidwater phytoplankton-dominated ponds but no differences in measured CH₄ emissions (Baliña et al., 2023).

The production of N₂O predominantly occurs through microbial nitrification and denitrification that depend on DIN, O₂ levels, and temperature (Codispoti and Christensen, 1985; Mengis et al., 1997; Velthuis and Veraart, 2022). Competition for DIN between primary producers and N2O-producing microorganisms can impact N₂O production. Additionally, the transfer of labile phytoplankton organic matter to sediments fuels benthic denitrification and impacts N2O fluxes. Eutrophication is assumed to drive high N₂O emissions from lakes and ponds (Audet et al., 2020; Webb et al., 2021; Wang et al., 2021; Xie et al., 2024), but some lakes with elevated Chl-a concentrations can act as sinks of N₂O due to removal of N₂O by denitrification (Webb et al., 2019; Borges et al., 2022, 2023). The presence of macrophytes also strongly influences nitrogen cycling in sediments of lakes and ponds (Barko et al., 1991; Choudhury et al., 2018; Deng et al., 2020; Dan et al., 2021) and should in theory also affect N₂O emissions, although this is seldom investigated, and available studies provide contradictory conclusions. N₂O emissions have been shown to follow diurnal cycles of O2 concentrations in areas dominated by submerged macrophytes in Lake Wuliangsuhai (China) (Ni et al., 2022) and the seasonal cycle of aboveground biomass of emerged macrophytes (Phragmites) in Baiyangdian Lake (China) (Yang et al., 2012). On the contrary, a study showed there was no significant difference of N2O production in sediments of macrophyte-rich (n = 10) and macrophyte-free (n = 12) lakes in subtropical China (Liu et al., 2018). There has been a very limited number of studies systematically investigating how emissions differ between ponds dominated by phytoplankton and those dominated by macrophytes (Baliña et al., 2023) and none simultaneously investigating CO₂, CH₄, and N₂O emissions including both diffusive and ebullitive components.

The emissions of CO₂ and N₂O from aquatic systems are exclusively through diffusion across the air-water interface (diffusive flux), while CH₄ can be additionally emitted as bubbles are released from sediments to the atmosphere (ebullitive flux). At annual scale, ebullitive CH₄ flux usually represents more than half of total (diffusive + ebullitive) CH₄ emissions from shallow lakes (Wik et al., 2013; Deemer and Holgerson, 2021), although the relative contribution of ebullitive and diffusive CH₄ emissions is highly variable seasonally (e.g. Wik et al., 2013; Ray and Holgerson, 2023; Rabaey and Cotner 2024). Ebullitive CH₄ fluxes are particularly high in the littoral zone of lakes at depths <5 m (Wik et al., 2013; DelSontro et al., 2016; Borges et al., 2022) and strongly increase in response to temperature (DelSontro et al., 2016; Aben et al., 2017; Rabaey and Cotner, 2024), as well as organic matter availability (DelSontro et al., 2016, 2018). Ebullitive CH₄ fluxes tend to be higher in small and shallow water bodies (Deemer and Holgerson, 2021) but are notoriously variable in time and space and are difficult to estimate reliably (DelSontro et al., 2011).

Here, we report a data set of CO₂, CH₄, and N₂O dissolved concentrations in four shallow and small urban ponds (Leybeek, Pêcheries, Silex, and Tenreuken) in the city of Brussels (Belgium) (Fig. 1), with data collected 46 times at regular intervals (between June 2021 and December 2023) on each pond. The air-water diffusive fluxes of CO₂, CH₄, and N₂O were calculated from dissolved concentrations and the gas transfer velocity, while the ebullitive CH₄ fluxes were measured with inverted funnels during eight deployments (totalling 48 d) in the four ponds. The four ponds have similar depth, surface area, and catchment urban coverage and mainly differ by the phytoplankton-macrophyte dominance, a clear-water state dominated by macrophytes and a turbidwater state dominated by phytoplankton (alternative states) (Fig. 1). We assess whether the differences in terms of (i) CO₂, CH₄, and N₂O dissolved concentration and diffusive emissions; (ii) ebullitive CH₄ emissions; and (iii) the relative contribution of CO₂, CH₄, and N₂O to the total GHG emissions in CO₂-eq between the four ponds are explained by the two alternative states.

2 Material and methods

2.1 Field sampling and meteorological data

Sampling was carried out at a single fixed station (pontoon) in each of the four ponds, on the same day between 09:00 am and 11:00 am CET, 46 times on each pond between June 2021 and December 2023 at a frequency ranging from one (winter) to three (summer) times per month. Water was sampled 5 cm below the surface with 60 mL polypropylene syringes for analysis of dissolved concentrations of CO₂, CH₄, and N₂O. Samples for CH₄ and N₂O were transferred from the syringes with a silicone tube into 60 mL borosil-

icate serum bottles (Wheaton), preserved with 200 µL of a saturated solution of HgCl2, sealed with a butyl stopper and crimped with an aluminium cap, without a headspace, and stored at ambient temperature in the dark prior to analysis in the laboratory. The partial pressure of CO₂ (pCO₂) was measured directly in the field, within 5 min of sample collection, with a Li-Cor Li-840 infrared gas analyser (IRGA) based on the headspace technique with four polypropylene syringes (Borges et al., 2019). A volume of 30 mL of sample water was equilibrated with 30 mL of atmospheric air within the syringe by shaking vigorously for 5 min. The headspace of each syringe was then sequentially injected into the IRGA, and a fifth syringe was used to measure atmospheric CO_2 . The final pCO₂ value was computed taking into account the partitioning of CO₂ between water and the headspace, as well as equilibrium with HCO₃⁻ (Dickson et al., 2007) using water temperature measured in situ and after equilibration and total alkalinity (data not shown). Samples for total alkalinity were conditioned, stored, and analysed as described by Borges et al. (2019). The IRGA was calibrated in the laboratory with ultrapure N₂ and a suite of gas standards (Air Liquide Belgium) with CO₂ mixing ratios of 388, 813, 3788, and 8300 ppm. The precision of pCO₂ measurements was $\pm 2.0\%$. Water temperature, specific conductivity, and oxygen saturation level ($\%O_2$) were measured in situ with VWR MU 6100H probe 5 cm below the surface. A 2 L polyethylene water container was filled with surface water for conditioning the samples for other variables at the laboratory in Université Libre de Bruxelles.

Three bubble traps were deployed 50 cm apart for measuring ebullitive CH₄ flux. The bubble traps consisted of inverted polypropylene funnels (diameter 23.5 cm) mounted with 60 mL polypropylene syringes, with three-way stop valves allowing the gas to be collected without contamination from ambient air. The polypropylene funnel was attached with steel rods to a polystyrene float. The volume of gas collected in the funnels was sampled with graduated polypropylene 60 mL syringes every 24 h. The value of the collected volume of gas was logged, and the gas was transferred immediately after collection to pre-evacuated 12 mL vials (Exetainers, Labco, UK) that were stored at ambient temperature in the dark prior to the analysis of CH₄ concentration in the laboratory. The time series of measurements was longer at the Silex pond than the other three ponds.

Surveys to identify and quantify visually the relative coverage of emerged and submerged macrophytes were conducted in summer 2023 (Table S1). The resulting list of macrophyte species agreed with past studies in Brussels' ponds (Peretyatko et al., 2007). Air temperature, precipitation, wind speed, and atmospheric pressure were retrieved from https://wow.meteo.be/en (last access: 18 January 2024) for the meteorological station of the Royal Meteorological Institute of St-Lambert (50.8408° N, 4.4234° E) in Brussels, located between 2.5 and 5.0 km from the surveyed ponds. Air temperature, wind speed, and atmospheric pressure were av-

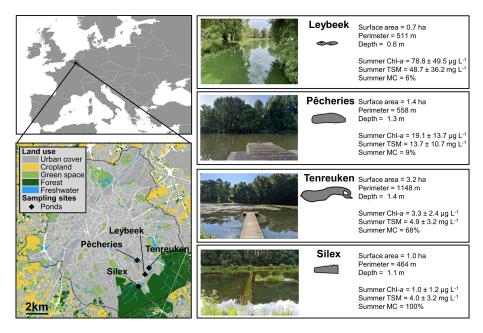


Figure 1. Location of the four sampled urban ponds (black diamonds) in the city of Brussels (Belgium) delineated by the black line. Right panels indicate, for each pond, the shape of the ponds, surface area (ha), perimeter (m), average depth (m), mean \pm standard deviation of chlorophyll-a (Chl-a, in μ g L⁻¹), and total suspended matter (TSM, in mg L⁻¹) in summer (21 June to 21 September 2021, 2022, 2023) and summer total macrophyte cover (MC; in %) (Table S1).

eraged over 24 h to obtain a daily mean value. Precipitation was integrated each day to obtain cumulated daily value.

2.2 Laboratory analysis

2.2.1 Chlorophyll-*a*, total suspended matter, and dissolved inorganic nutrients

Water was filtered through Whatman GF/F glass microfiber filters (porosity 0.7 µm) with a diameter of 47 mm for total suspended matter (TSM) and Chl-a. Filters for TSM were dried in an oven at 50 °C, and filters for Chl-a were kept frozen (-20 °C). The weight of each filter was determined before and after filtration of a known volume of water using an ExplorerTM Pro EP214C analytical microbalance (accuracy ± 0.1 mg) for determination of TSM concentration. Chla concentration was measured on extracts with 90% acetone by fluorimetry (Kontron model SFM 25) (Yentsch and Menzel, 1963) with a limit of detection of $0.01 \,\mu g \, L^{-1}$. Filtered water was stored frozen (-20 °C) in 50 mL polypropylene bottles for analysis of dissolved nutrients. Soluble reactive phosphorus (SRP) was determined by the ammonium molybdate, ascorbic acid, and potassium antimony tartrate staining method (Koroleff, 1983), with a limit of detection of $0.1 \,\mu\text{mol}\,\text{L}^{-1}$. Ammonium (NH_4^+) was determined by the nitroprusside-hypochlorite-phenol staining method (Grasshoff and Johannsen, 1972), with a limit of detection of $0.05 \,\mu\text{mol}\,\text{L}^{-1}$. Nitrite (NO_2^-) and nitrate (NO_3^-) were determined before and after reduction of NO₃ to NO₂ by a cadmium-copper column, using the Griess acid reagent staining method (Grasshoff et al., 2009), with a detection limit of 0.01 and 0.1 μ mol L⁻¹, respectively. Concentration of dissolved inorganic nitrogen (DIN) was calculated as the sum of NH₄⁺, NO₂⁻, and NO₃⁻ concentrations (in μ mol L⁻¹).

2.2.2 CH₄ and N₂O measurements by gas chromatography

Measurements of N_2O and CH_4 concentrations dissolved in water were made with the headspace technique (Weiss, 1981) with a headspace volume of 20 mL of ultra-pure N_2 (Air Liquid Belgium) and a gas chromatograph (GC) (SRI 8610C) with a flame ionization detector for CH_4 and an electron capture detector for N_2O calibrated with $CH_4:N_2O:N_2$ gas mixtures (Air Liquide Belgium) with mixing ratios of 1, 10, and 30 ppm for CH_4 and 0.2, 2.0, and 6.0 ppm for N_2O . The precision of measurement based on duplicate samples was $\pm 3.9\,\%$ for CH_4 and $\pm 3.2\,\%$ for N_2O . Measurements of CH_4 concentration in the gas samples from bubble traps were also made by GC with the same set-up and calibration as for the determination of the dissolved concentrations in water samples.

The CO_2 concentration is expressed as partial pressure (pCO₂) in parts per million (ppm) and CH_4 as dissolved concentration (nmol L^{-1}), as frequently used in the topical literature. CH_4 concentrations were systematically and distinctly above saturation level (2–3 nmol L^{-1}), and pCO₂ values were below saturation only five times out of the 187 measurements. The N_2O concentrations fluctuated around atmo-

spheric equilibrium, so data are presented as a percentage of saturation level (% N_2O , where atmospheric equilibrium corresponds to 100%). The equilibrium with atmosphere for N_2O was calculated from the average air mixing ratios of N_2O provided by the Global Monitoring Division (GMD) of the National Oceanic and Atmospheric Administration (NOAA) Earth System Research Laboratory (ESRL) (Dutton et al., 2023) and using Henry's constant given by Weiss and Price (1980).

2.3 Calculations

2.3.1 Diffusive GHG emissions

The diffusive air–water CO₂, CH₄, or N₂O fluxes (F_G) were computed according to

$$F_{G} = k \times \Delta[G], \tag{1}$$

where k is the gas transfer velocity, and $\Delta[G]$ is the air–water gas concentration gradient.

The atmospheric pCO₂ was measured in the field with the Li-Cor Li-840. For CH₄, the global average present-day atmospheric mixing ratio of 1.9 ppm was used (Lan et al., 2024). k was computed from a value normalized to a Schmidt number of 600 (k_{600}) and from the Schmidt number of CO₂, CH₄, and N₂O in freshwater according to the algorithms as a function of water temperature given by Wanninkhof (1992). k_{600} was calculated from the parameterization as a function of wind speed of Cole and Caraco (1998). CH₄ and N₂O emissions were converted into CO₂ equivalents (CO₂-eq) considering a 100-year time frame, using global warming potentials of 32 and 298 for CH₄ and N₂O, respectively (Myrhe et al., 2013).

2.3.2 Ebullitive flux

Bubble flux (mL m⁻² d⁻¹) was calculated according to

$$F_{\text{bubble}} = \frac{V_{\text{g}}}{A \times \Delta t},\tag{2}$$

where V_g is the volume of gas collected in the inverted funnel (mL), A is the cross-sectional surface area of the funnel (m²), and Δt is the collection time (d).

A multiple linear regression model of F_{bubble} dependent on water temperature (T_{w} in °C) and drops of atmospheric pressure (Δp in atm) was fitted to the data according to

$$\log_{10}(F_{\text{bubble}}) = \alpha \times T_{\text{w}} + \beta \times \Delta p + \gamma, \tag{3}$$

where γ is the y intercept, and α and β are the slope coefficients of the multiple linear regression model.

 Δp was calculated according to Zhao et al. (2021):

$$\Delta p = -\frac{1}{\Delta t} \int_0^t p - p_0; \forall p < p_0, \tag{4}$$

where p is the atmospheric pressure (atm), p_0 a threshold value fixed at 1 atm, and Δt the time interval between two measurements (d) (Fig. S1).

A linear regression model of F_{bubble} dependent on T_{w} alone was fitted to the data according to

$$\log_{10}(F_{\text{bubble}}) = \alpha' \times T_{\text{w}} + \gamma', \tag{5}$$

where α' is the slope coefficient, and γ' is the y intercept.

To evaluate the relative importance of $T_{\rm w}$ and Δp combined and $T_{\rm w}$ alone in driving $F_{\rm bubble}$, the values modelled based on Eqs. (3) and (5) were compared to the observations in the Silex pond alone or in all four ponds together, for three $T_{\rm w}$ ranges ($T_{\rm w}$ <15 °C, $T_{\rm w}$ >15 °C, and the full $T_{\rm w}$ range).

Ebullitive CH₄ fluxes (E_{CH_4} in mmol m⁻² d⁻¹) were calculated according to

$$E_{\text{CH}_4} = [\text{CH}_4] \times F_{\text{bubble}},\tag{6}$$

where $[CH_4]$ is the measured CH_4 concentration in bubbles (mmol mL⁻¹).

The CH₄ content in bubbles expressed as a percentage of total gas (%CH₄) was fitted with a linear regression model dependent on $T_{\rm w}$ in the Silex pond. The correlation of %CH₄ with $F_{\rm bubble}$ was tested on the merged data of all the four ponds.

A linear regression model of $E_{\mathrm{CH_4}}$ dependent on T_{w} was fitted to the data according to

$$\log_{10}(E_{\text{CH}_4}) = b \times T_{\text{W}} + c, \tag{7}$$

where b is the slope, and c is the y intercept of the linear regression.

Equation (7) is used to predict E_{CH_4} in each pond from measured time series of T_{w} , allowing matching to each diffusive CH₄ flux estimate derived from Eq. (1).

The ratio of ebullitive CH₄ flux to total (diffusive + ebullitive) CH₄ flux $\left(\frac{\text{Ebul}}{\text{Tot}}\right)$ was fitted as a function of T_{w} according to DelSontro et al. (2016):

$$\frac{\text{Ebul}}{\text{Tot}} = \frac{1}{1 + f \times e^{g \cdot T_{\text{w}}}}.$$
 (8)

The methane ebullition Q_{10} represents the change in E_{CH_4} per 10 °C change in T_{w} and was computed according to DelSontro et al. (2016):

$$Q_{10} = 10^{10b}, (9)$$

where b is given by Eq. (7).

The Q_{10} of diffusive CH₄ fluxes was also computed from Eq. (9) but using the *b* value derived by replacing E_{CH_4} by the diffusive CH₄ flux in Eq. (7).

2.3.3 Statistical analysis

Generalized linear mixed models (GLMMs) were used (1) to find differences in variables among ponds and (2) to relate

GHG variables to their putative controls across all ponds and within individual ponds. GLMMs were computed with the *lme4* package (Bates et al., 2015) in R version 4.4.1 (R Core Team, 2021).

GLMMs allowed us to compare pCO₂, dissolved CH₄ concentration, $\%N_2O$, F_{bubble} , $\%CH_4$ in bubbles, E_{CH_4} , and diffusive CH₄ fluxes among the four ponds, using the sampling date as a random effect, and post hoc tests were performed using estimated marginal means (emmeans package in R) to assess pairwise differences between ponds. This analysis aimed at investigating if patterns in GHG concentrations and emissions differed among ponds, in particular with regards to clear-water and turbid-water states. Data were also compared among the four ponds separated by seasons, but GLMMs did not converge due to an insufficient number of data points. Comparisons on log₁₀-transformed data were then made using repeated measures analysis of variance (ANOVA) (ezANOVA package in R) with Tukey's honestly significant difference (HSD) post hoc tests. This analysis aimed at investigating if patterns in data shown by the data analysis with the full data set (four seasons merged) were also observed when analysing the data separated by seasons.

For the data sets covering the whole sampling period (four seasons merged), GLMMs were constructed for pCO₂, dissolved CH₄ concentration, %N₂O, F_{bubble}, %CH₄ in bubbles, E_{CH_4} , and diffusive CH₄ fluxes that included T_w , precipitation, %O2, Chl-a, TSM, DIN, and SRP as fixed effects and "pond" and "sampling date" as a random effect to account for repeated measurements with the lme4 package in R. This analysis aimed at investigating (1) the impact of photosynthesis–respiration on CO₂ concentrations and emissions based on the relationships with Chl-a, DIN, and SRP; (2) the impact of the response of methanogenesis to warming on CH₄ concentrations and diffusive and ebullitive emissions based on the relationships to T_w ; and (3) the impact on DIN availability and Tw on N2O concentrations and diffusive emissions. A linear regression model was used to assess the relationship between Chl-a, TSM, %O₂, SRP, and DIN versus $T_{\rm w}$. This analysis aimed at investigating if some of the patterns between GHG variables versus Chl-a, TSM, %O₂, SRP, and DIN might in fact have reflected a relation between these variables and $T_{\rm w}$ indirectly. The relationships between the annual means of CH₄, CO₂ and N₂O fluxes and the annual means of a subset of variables (Chl-a, macrophyte cover, pond surface area, and depth) were assessed with linear or quadratic regressions. The modelled F_{bubble} values in the Silex pond and in the four ponds were compared to measured F_{bubble} values with a linear regression in order to evaluate the model performance. A correlation analysis was used to assess the relationship between %CH₄ and F_{bubble} , as well as between anomalies in annual air temperature and annual precipitation.

The linear regression model and the Pearson correlation coefficient (r) were computed in R using the functions lm and cor(method = "pearson"), respectively. Statistical sig-

nificance was set at p<0.05 for all analyses. Significant differences between groups presented in box plots are indicated by lower-case letters in the figures.

3 Results

3.1 Seasonal variations of meteorological conditions and GHG concentrations

The city of Brussels experiences a temperate climate with mild weather year-round and evenly distributed abundant precipitation, totalling 837 mm on average annually for the reference period 1991-2020. The average annual air temperature was 11 °C, with a summer average of 17.9 °C and a winter average of 4.1 °C for the reference period 1991-2020. During the sampling period, from June 2021 to December 2023, $T_{\rm w}$ in the surface of the four sampled ponds (Leybeek, Pêcheries, Silex, and Tenreuken; Fig. 1) tracked the air temperature closely, which ranged between -1.5 and 30.0 °C following the typical seasonal cycle at mid-latitudes in the Northern Hemisphere (Fig. S2). The years 2022 and 2023 were about 1 °C warmer than the average for the period 1991-2020 (11 °C), while the year 2021 was closer to the long-term average (Fig. 2). The year 2022 was warmer and drier than 2021 and 2023 (Fig. 2), with positive air temperature anomalies observed evenly throughout the year (9 months out of 12) and negative precipitation anomalies in summer, autumn, and early winter (Fig. S2). The year 2021 had warmer and drier months in June and September and colder and wetter months in July and August and was wetter and colder than 2022 overall (Fig. 2). The year 2023 was marked by both positive air temperature and precipitation anomalies (Fig. S2), resulting in a wetter and warmer year than normal and compared to 2021 and 2022 (Fig. 2). Daily wind speed was generally low ($<1 \text{ m s}^{-1}$) except for a windier period in spring 2022 (up to 5.8 m s⁻¹, corresponding to the Eunice storm) and in autumn 2023 (up to $9.7 \,\mathrm{m \, s^{-1}}$, corresponding to the Ciarán storm) (Fig. S2).

The four sampled ponds are situated in the periphery of the city of Brussels, with the Silex pond being bordered by the Sonian Forest (Fig. 1). The four ponds are relatively small (0.7-3.2 ha) and shallow (0.6-1.4 m) and have not been drained or dredged since at least 2018 (Table S2). The four studied ponds had significantly different Chl-a concentration values during summer, with the Leybeek pond having higher Chl-a $(78.8 \pm 49.5 \,\mu \,\mathrm{g} \,\mathrm{L}^{-1})$, followed by the Pêcheries pond $(19.1 \pm 13.7 \,\mu\text{g}\,\text{L}^{-1})$, the Tenreuken pond $(3.3 \pm 2.4 \,\mu\text{g}\,\text{L}^{-1})$, and the Silex pond $(1.0 \pm 1.2 \,\mathrm{\mu g}\,\mathrm{L}^{-1})$ (Figs. 1, 3, Table S3). The Leybeek and Pêcheries ponds with higher summer Chl-a concentration had turbid water (summer TSM = 48.7 ± 36.2 and $13.7 \pm 10.7 \,\mathrm{mg}\,\mathrm{L}^{-1}$, respectively) and undetectable submerged macrophyte cover in summer (Fig. 1, Table S1). The Tenreuken and Silex ponds with lower summer Chl-a concentrations had clear water (summer TSM = 4.9 ± 3.2 and

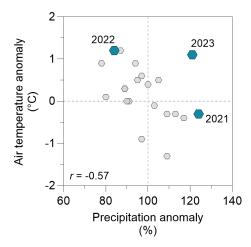


Figure 2. Anomaly of annual air temperature (°C) as a function of anomaly of annual precipitation (%) from 2003 to 2023 with respect to the average of the 1991–2020 period (11 °C and 837 mm, respectively). Small grey hexagons represent anomaly values for the years 2003–2020 and larger blue hexagons represent anomaly values for the years 2021–2023. The air temperature anomaly was correlated to the precipitation anomaly for the years 2003–2020 (Pearson r = -0.57, p = 0.0147, n = 18). Note the anomalous rainy year in 2023 relative to the pattern of precipitation as a function of temperature anomalies for the other years, possibly in response to the strong El Niño event of 2023 (Chen et al., 2024).

 $4.0 \pm 3.2 \,\mathrm{mg} \,\mathrm{L}^{-1}$, respectively) and a high total macrophyte cover during summer (68 % and 100 %, respectively, Fig. 1, Table S1). Seasonally, the highest values of Chl-a were observed in summer in the turbid-water Leybeek and Pêcheries ponds, related to algal blooms. Conversely, lowest values of Chl-a were observed in summer in the clear-water Tenreuken and Silex ponds (Figs. 1, 3), probably related to competition for dissolved inorganic nutrients with macrophytes.

The $\%O_2$ values ranged from 11 % to 191 % (Fig. 3). The highest $\%O_2$ values in the four ponds were observed in spring and summer compared to autumn and winter owing to aquatic primary production. In summer, $\%O_2$ was significantly higher in the Leybeek pond (109 \pm 46 %), characterized by higher Chl-a concentration compared to the Pêcheries pond (75 \pm 23 %, p = 0.0212, Table S3). The lowest average $\%O_2$ was observed in autumn in the Pêcheries pond (46 \pm 22 %) and was significantly lower than in the Leybeek (85 \pm 34 %, p = 0.0146, Table S3) and Silex ponds (81 \pm 19 %, p = 0.0130, Table S3).

The pCO₂ values ranged from 40 to 13 804 ppm (Fig. 3). Undersaturation of CO_2 with respect to atmospheric equilibrium (\sim 410 ppm) was only observed on five occasions out of the 187 measurements, three times in the turbidwater Leybeek pond in summer (40 ppm on 13 August 2021, 220 ppm on 27 June 2022 and 149 ppm on 13 June 2023) and twice in the clear-water Tenreuken pond in spring and summer (383 ppm on 13 August 2021 and 55 ppm on 2

May 2022). Low values of pCO₂ were generally observed in spring and summer, and high values of pCO₂ were observed in autumn in the four ponds (Fig. 3). In summer, pCO₂ was lower in the Leybeek pond (2187 \pm 2012 ppm) than in the Pêcheries (3427 \pm 1672 ppm, p=0.0015, Table S3) and the Silex (3222 \pm 1175 ppm, p=0.0002, Table S3) ponds. When data were pooled, pCO₂ was negatively influenced by %O₂ and positively by DIN, SRP, and precipitation (Table S4). In individual ponds, pCO₂ was negatively influenced by %O₂ and positively by precipitation in the four ponds, positively by DIN in the Leybeek pond, positively by DIN and SRP in the Tenreuken pond, and negatively by Chl-a in the Silex pond (Table S5).

The CH₄ dissolved concentrations ranged from 194 to $48\,380\,\mathrm{nmol}\,\mathrm{L}^{-1}$ (Fig. 3) and were always above saturation ($\sim 2 \,\mathrm{nmol}\,\mathrm{L}^{-1}$). High values of CH₄ dissolved concentrations were generally observed in spring and summer, and low values of CH₄ dissolved concentrations were generally observed in winter in the four ponds (Fig. 3). In summer, CH₄ dissolved concentration was higher in the Silex pond $(4898 \pm 3384 \, \text{nmol L}^{-1})$ than in the Pêcheries $(2518 \pm 2105 \text{ nmol L}^{-1}, p = 0.0385, \text{ Table S3})$ and the Tenreuken (2189 \pm 1365 nmol L⁻¹, p = 0.0055, Table S3) ponds. When data were pooled, dissolved CH₄ concentration was influenced positively by $T_{\rm w}$ (Table S4). In individual ponds, CH₄ dissolved concentration was also influenced positively by $T_{\rm w}$ in each of the four ponds (Table S5). Additionally, CH₄ dissolved concentration was positively influenced by precipitation in the Leybeek pond and by SRP in the Silex pond and negatively by DIN in the Pêcheries pond and by Chl-a in the Tenreuken and the Silex ponds. (Table S5). These relationships between CH₄ and other variables (SRP, DIN, Chl-a) probably indirectly reflected the seasonal variations of these other variables that were also influenced by $T_{\rm w}$. Indeed, DIN was negatively influenced by $T_{\rm w}$ in the Pêcheries pond, Chl-a was negatively influenced by $T_{\rm w}$ in the Tenreuken and the Silex ponds, and SRP was positively influenced by $T_{\rm w}$ in the Silex pond (Table S6).

The $\%N_2O$ values ranged from 32 % to 826 % (Fig. 3). Undersaturation of N₂O with respect to atmospheric equilibrium was observed 66 times out of the 187 measurements. Low values of %N₂O were generally observed in spring and summer, and high values of %N2O were generally observed in autumn and winter in the four ponds (Fig. 3). During spring, $\%N_2O$ was lower in the Pêcheries pond (90 ± 11 %) than the Leybeek (138 \pm 30 %, p = 0.0043, Table S3) and the Tenreuken (138 \pm 41, p = 0.0057, Table S3) ponds. During summer, %N2O was lower in the Pêcheries pond $(78 \pm 17 \%)$ than the Leybeek $(191 \pm 104 \%, p < 0.0001, Ta$ ble S3) and the Silex $(126 \pm 49 \%, p = 0.001, Table S3)$ pond and lower in the Tenreuken pond $(133 \pm 106 \%)$ than the Leybeek pond (p = 0.0219, Table S3). During autumn, % N_2O was lower in the Pêcheries pond (103 ± 33 %) than the Leybeek pond (190 \pm 70 %, p = 0.0174, Table S3). For the whole sampling period, %N₂O was lower in the Pêcheries

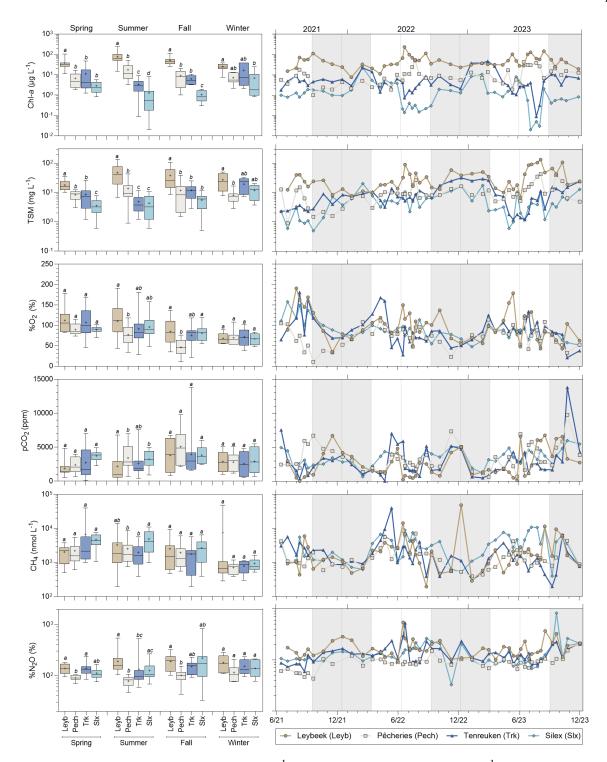


Figure 3. Seasonal variations of Chlorophyll-a (Chl-a, in μ g L⁻¹), total suspended matter (TSM, in μ g L⁻¹), oxygen saturation (μ O₂, in %), partial pressure of CO₂ (μ CO₂ in ppm), dissolved CH₄ concentration (CH₄, in nmol L⁻¹), and N₂O saturation level (μ N₂O, in %) in four urban ponds (Leybeek (Leyb), Pêcheries (Pech), Tenreuken (Trk), and Silex (Slx)) in the city of Brussels (Belgium) from June 2021 to December 2023. Box plots show median (horizontal line), mean (cross), and 25 %–75 % percentiles (box limits). Whiskers extend from minimum to maximum values. Grey and white bands in the plots on the right correspond to the autumn/winter and spring/summer periods, respectively, and dotted vertical bars indicate the first day of each season. Lower-case letters indicate significant differences between ponds (Tables S3 and S4).

pond $(94 \pm 28 \%)$ than the Leybeek $(178 \pm 82 \%, p < 0.0001,$ Table S7), the Tenreuken ($140 \pm 77 \%$, p < 0.0001, Table S7), and the Silex (144 \pm 113 %, p<0.0001, Table S7) ponds and was lower in the Tenreuken pond than the Leybeek pond (p = 0.0038, Table S7). When data were pooled, %N₂O was influenced negatively by $T_{\rm w}$ and positively by DIN and NH₄⁺ (Table S4). In individual ponds, %N₂O was influenced negatively by $T_{\rm w}$ in the Leybeek, the Pêcheries, and the Tenreuken ponds (Table S5). %N₂O was influenced positively by NO₂ in the Leybeek pond and by NH₄⁺ in the Pêcheries and Tenreuken ponds (Table S8). %N2O was influenced positively by Chl-a and TSM in the Tenreuken pond and negatively by Chl-a in the Leybeek pond (Table S5), probably reflecting the negative influence on Chl-a and TSM by $T_{\rm w}$ in the Tenreuken pond and the positive influence on Chl-a by $T_{\rm w}$ in the Leybeek pond (Table S6).

3.2 Drivers of bubble flux

The F_{bubble} measured with inverted funnels in the four sampled ponds in the city of Brussels ranged between 0 and 2078 mL m⁻² d⁻¹ and was influenced positively by T_{w} in all four systems (Fig. 4). The mean %CH₄ of the bubbles in the four sampled ponds in the city of Brussels was 31 ± 21 %, and values were influenced positively by T_{w} in the Silex pond (Fig. 4). The %CH₄ of the bubbles was correlated with F_{bubble} (Fig. S3) as both variables were positively influenced by T_{w} (Fig. 4).

The time series at the Silex pond allowed the effects of $T_{\rm w}$ and atmospheric pressure variations on F_{bubble} to be investigated in more detail (Fig. 5). In spring 2022, the F_{bubble} at the Silex pond increased during drops in atmospheric pressure (depressions) (Fig. 5). There was no relation between wind speed and peaks of F_{bubble} ($r^2 = 0.01$, p = 0.463), suggesting a more important role of changes in atmospheric pressure than of wind speed in triggering bubble fluxes in the Silex pond in spring 2022. The F_{bubble} values at the Silex pond were higher in summer $(1152 \pm 433 \text{ mL m}^{-2} \text{ d}^{-1})$ than during spring $(198 \pm 170 \,\mathrm{mL \, m^{-2} \, d^{-1}})$, and the temporal changes of F_{bubble} tracked those of T_{w} (Fig. 5). In order to evaluate the relative importance of changes of atmospheric pressure and water temperature in triggering bubble fluxes, the F_{bubble} was modelled as a function of T_{w} alone or as a function of both $T_{\rm w}$ and Δp (Figs. 5, S4). For periods of low $T_{\rm w}$ (<15 °C), the inclusion of the term of Δp in the model improved the performance of the model by comparison to the measurements (Figs. 5, S4). But for warmer periods (>15 °C), when bubbling fluxes were quantitatively more important, the inclusion in the model of the term of Δp did not improve the performance of the model (Figs. 5, S4). For the full T_w range (<15 and >15 °C), the inclusion of the term of Δp only improved the performance of the model very marginally (Fig. S4).

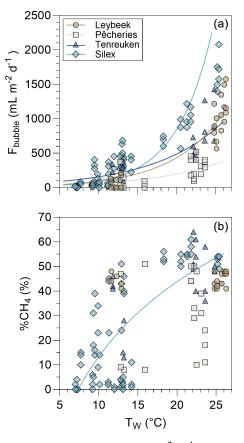


Figure 4. Bubble flux (F_{bubble} in mL m⁻² d⁻¹) and the relative CH₄ content in bubbles (%CH₄, in %) as a function of surface water temperature (Tw in °C) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium) from June 2021 to December 2023. Bubbles fluxes were measured with three bubble traps in spring, summer, and autumn of 2022 and 2023, totalling 8 d in the Leybeek, Pêcheries, and Tenreuken ponds and 24 d in the Silex pond. Given the shallowness of the sampled systems (<1.5 m, Fig. 1), we assumed that sediments experience the same temperature as surface waters. In panel (a), solid lines represent linear regression of $log_{10}(F_{bubble})$ as a function of T_{w} for the Leybeek $(\log_{10}(F_{\text{bubble}}) = 0.0664 \times T_{\text{w}} + 1.3095, n = 22, p < 0.0001),$ the Pêcheries $(\log_{10}(F_{\text{bubble}}) = 0.0486 \times T_{\text{w}} + 1.3257, n = 22, p =$ 0.0146), the Tenreuken $(\log_{10}(F_{\text{bubble}}) = 0.0492 \times T_{\text{w}} + 1.7039,$ n = 19, p < 0.0001), and the Silex $(\log_{10}(F_{\text{bubble}}) = 0.0945 \times T_{\text{W}} +$ 1.0373, n = 72, p < 0.0001) ponds. In panel (**b**), the solid line represents the linear regression of %CH₄ as a function of $log_{10}(T_w)$ for the Silex pond (%CH₄ = $101.11 \times \log_{10}(T_w) - 87.8$, n = 72, p < 0.0001).

3.3 Drivers of methane ebullitive fluxes

The $E_{\rm CH_4}$ values in the four ponds ranged between 0 and 59 mmol m⁻² d⁻¹ and were positively related to $T_{\rm w}$ (Fig. 6). The fitted relations between $E_{\rm CH_4}$ and $T_{\rm w}$ were specific to each pond and encompassed the fitted relations established in similar systems: four small ponds in Quebec (DelSontro et al., 2016) and a small urban pond in the Netherlands (Aben et

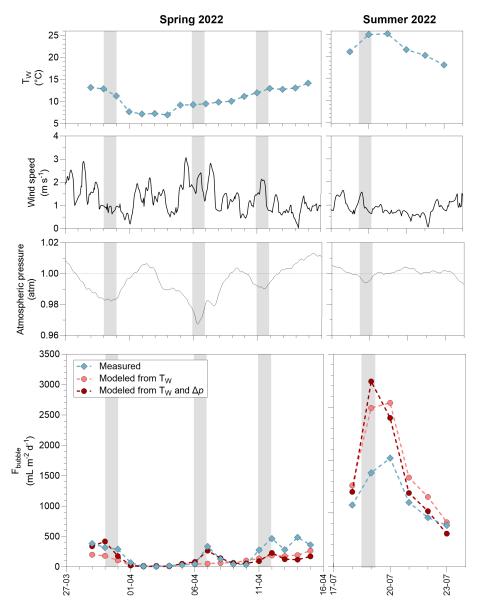


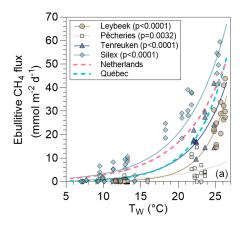
Figure 5. Time series of surface water temperature $(T_{\rm w}, {^{\circ}\rm C})$, wind speed $({\rm m\,s^{-1}})$, atmospheric pressure (atm), and measured and modelled bubble flux $(F_{\rm bubble} \ {\rm in\,mL\,m^{-2}\,d^{-1}})$ in the Silex pond from 29 March 2022 to 15 April 2022 and from 18 July 2022 to 23 July 2022. The bubble flux was modelled from a fit to data based on $T_{\rm w}$ alone $(\log_{10}(F_{\rm bubble}) = 3973 \times \log_{10}(T_{\rm w}) - 215, \, p < 0.0001, \, n = 72)$ and based on both $T_{\rm w}$ and drops in atmospheric pressure (Δp) $(\log_{10}(F_{\rm bubble}) = 4.551 \times \log_{10}(T_{\rm w}) + 1.962 \times \Delta p - 3.006, \, p < 0.0001, \, n = 72)$.

al., 2017). The Q_{10} of CH₄ ebullition values ranged between 4.4 in the deeper Pêcheries pond and 26.9 in the shallower Leybeek pond (Table S9). The Q_{10} values of CH₄ ebullition in the four studied ponds of the city of Brussels, in Quebec (DelSontro et al., 2016), and in the Netherlands (Aben et al., 2017) were negatively related to water depth (Fig. 6).

3.4 Relative contribution of methane ebullitive and diffusive fluxes

Diffusive CH₄ fluxes computed from dissolved CH₄ concentration and *k* derived from wind speed ranged between 0.1

and 19.7 mmol m⁻² d⁻¹ (Fig. 7). The diffusive CH₄ fluxes tended to be higher in summer and spring than in autumn and winter, owing to the strong positive influence on CH₄ dissolved concentration by $T_{\rm w}$ (Fig. 3; Tables S4, S5). In addition, wind speed only showed small seasonal variations during sampling $(0.6\pm0.6\,{\rm m\,s^{-1}}$ in spring, $0.3\pm0.2\,{\rm m\,s^{-1}}$ in summer, $0.7\pm0.7\,{\rm m\,s^{-1}}$ in autumn, and $0.6\pm0.2\,{\rm m\,s^{-1}}$ in winter) (Fig. S2). $E_{\rm CH_4}$ values were calculated from the relations with $T_{\rm w}$ for each pond given in Figure 6 from the $T_{\rm w}$ data coincident with the diffusive CH₄ fluxes (Fig. 7). The resulting calculated $E_{\rm CH_4}$ values allowed us both to compare and integrate components of CH₄ emissions to the at-



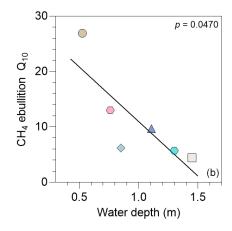


Figure 6. (a) Measured ebullitive CH₄ fluxes (mmol m⁻² d⁻¹) as a function of surface water temperature (°C) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium), in spring, summer, and autumn of 2022 and 2023, totalling 8 d in the Leybeek, Pêcheries, and Tenreuken ponds and 24 d in the Silex pond, with three bubble traps. Solid lines represent exponential fit for the Leybeek ($Y = 0.01 \cdot e^{0.32 \cdot X}$, n = 22), Pêcheries ($Y = 0.16 \cdot e^{0.15 \cdot X}$, n = 22), Tenreuken ($Y = 0.10 \cdot e^{0.23 \cdot X}$, n = 19), and Silex ($Y = 0.54 \cdot e^{0.18 \cdot X}$, n = 72) ponds (Table S7). Dashed lines represent the published exponential fit established in similar systems: four small ponds in Quebec ($Y = 0.06 \cdot e^{0.25 \cdot X}$) (DelSontro et al., 2016) and a small urban pond in the Netherlands ($Y = 0.51 \cdot e^{0.17 \cdot X}$) (Aben et al., 2017). (b) Each exponential curve allows us to determine a Q_{10} of CH₄ ebullition, plotted against water depth; the solid line represents linear regression ($Y = 30.64 - 19.67 \cdot X$, N = 6).

mosphere seasonally and to calculate the relative contribution of $E_{\rm CH_4}$ to total (diffusive + ebullitive) CH₄ emissions. The relative contribution of $E_{\rm CH_4}$ to total CH₄ emissions ranged between 1 % and 99 % in the four sampled ponds in the city of Brussels (Fig. 7) and was influenced positively by $T_{\rm w}$ (Fig. S5). The values of Q_{10} of diffusive CH₄ fluxes were lower than those for $E_{\rm CH_4}$ in each pond and less variable seasonally (1.2 in the Pêcheries pond to 2.9 in the Silex pond) (Table S9).

The annually averaged diffusive and ebullitive fluxes of CH₄ in the four ponds in the city of Brussels were plotted against annually averaged Chl-a concentration, total macrophyte cover in summer, water depth, and lake surface area (Fig. 8) that are frequent predictors of variations of CH₄ fluxes among lakes and ponds (Holgerson and Raymond, 2016; DelSontro et al., 2018, Deemer and Holgerson, 2021; Casas-Ruiz et al., 2021; Borges et al., 2022). The annually averaged E_{CH_4} values were significantly higher in the two clear-water ponds (7.3 ± 2.9) and $13.4 \pm 3.7 \,\mathrm{mmol}\,\mathrm{m}^{-2}\,\mathrm{d}^{-1}$ in the Tenreuken and the Silex ponds, respectively) than the two turbid-water ponds $(3.8 \pm 3.2 \text{ and } 2.5 \pm 1.4 \, \text{mmol m}^{-2} \, \text{d}^{-1} \text{ in the Leybeek and}$ the Pêcheries ponds, respectively) (Table S7). The annually averaged E_{CH_4} values were significantly higher in the Silex pond, which showed a higher macrophyte cover during summer (100% in the Silex pond and 68% in the Tenreuken pond) than the Tenreuken pond (p < 0.0001, Table S7), and were not significantly different in the two turbidwater Leybeek and Pêcheries ponds (p = 0.0617, Table S7) that showed similar low macrophyte cover during summer (6% and 9% in the Leybeek and Pêcheries ponds, respectively) (Fig. 8). The annually averaged E_{CH_4} values were overall influenced positively by macrophyte cover and negatively by Chl-a (Fig. 8).

In the four sampled urban ponds, annually averaged CH₄ diffusive fluxes were higher in the pond with the highest total macrophyte cover in the clear-water ponds and higher in the pond with the highest Chl-a concentration in the turbid-water ponds (Fig. 8). The annually averaged relative contribution of $E_{\rm CH_4}$ to total CH₄ emissions was higher in the two clearwater ponds than the two turbid-water ponds (Table S7). The relative contribution of $E_{\rm CH_4}$ to the total CH₄ flux was influenced positively by macrophyte cover and negatively by Chl-a (Fig. 8).

The annually averaged diffusive fluxes of CO_2 (F_{CO_2}) and N_2O (F_{N_2O}) in the four ponds in the city of Brussels were also plotted against annually averaged Chl-a concentration, total macrophyte cover in summer, water depth, and lake surface area (Fig. S6). Annually averaged F_{CO_2} values were lower in the Leybeek pond than the Pêcheries and the Silex ponds (Table S7). F_{CO_2} was not significantly influenced by the other variables (Chl-a concentration, total macrophyte cover, water depth, and lake surface area) (Fig. S6). Annually averaged F_{N_2O} was not significantly different between clear-water and turbid-water ponds. F_{N_2O} was significantly lower in the deeper Pêcheries pond than the two shallower Leybeek and Silex ponds (Table S7), and F_{N_2O} showed a significant negative relationship with water depth (Fig. S6).

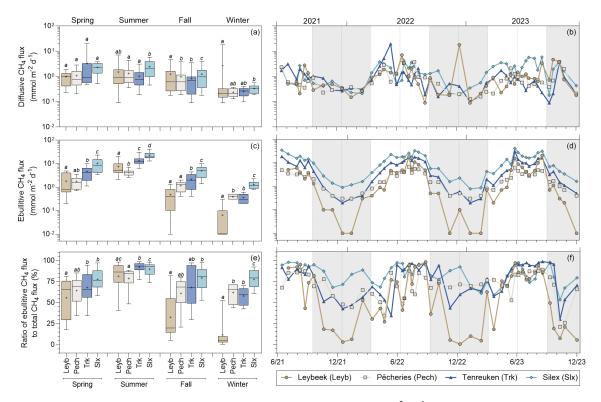


Figure 7. Seasonal variations of (**a, b**) diffusive and (**c, d**) ebullitive CH₄ fluxes (mmol m⁻² d⁻¹) and (**e, f**) the ratio of ebullitive CH₄ flux to total (ebullitive + diffusive) CH₄ flux (%) in four urban ponds (Leybeek (Leyb), Pêcheries (Pech), Tenreuken (Trk), and Silex (Slx)) in the city of Brussels (Belgium) from June 2021 to December 2023. Diffusive fluxes were calculated from CH₄ concentration and gas transfer velocity derived from wind speed. Ebullitive CH₄ fluxes were calculated from the relations with water temperature for each pond (Fig. 6; Table S7) from the water temperature data coincident with the diffusive CH₄ fluxes. Note that the relations of ebullitive CH₄ fluxes as a function of water temperature were established over a temperature range (7.0 to 26.3 °C) that is consistent with the range of water temperature values (2.0–25.9 °C) over which the ebullitive CH₄ fluxes were modelled. Box plots show median (horizontal line), mean (cross), and 25 %–75 % percentiles (box limits). Whiskers extend from minimum to maximum values. Grey and white bands in the plots on the right correspond to the autumn/winter and spring/summer periods, respectively, and dotted vertical bars indicate the first day of each season. Lower-case letters indicate significant differences between ponds (Tables S3 and S4).

3.5 Relative contribution of CO₂, CH₄, and N₂O emissions

The emissions in CO₂-eq for the three GHGs averaged per season for both 2022 and 2023 peaked seasonally in summer in the Silex $(2.9 \,\mathrm{mg}\,\mathrm{CO}_2\text{-eq}\,\mathrm{m}^{-2}\,\mathrm{d}^{-1})$, the Tenreuken $(1.7 \text{ mg CO}_2\text{-eq m}^{-2} \text{ d}^{-1})$, and the Leybeek $(1.1 \text{ mg CO}_2\text{-eq})$ $m^{-2} d^{-1}$) pond (Fig. 9) but peaked in autumn in the Pêcheries pond (1.3 mg CO_2 -eq m⁻² d⁻¹). The higher value of the total GHG emissions in autumn compared to other seasons in the Pêcheries pond was due to an increase of CO2 emissions in autumn that surpassed the peak of CH₄ emissions in summer. The GHG fluxes were lowest in winter in the Silex (1.3 mg CO_2 -eq m⁻² d⁻¹), the Tenreuken (0.9 mg CO_2 -eq m⁻² d⁻¹), the Pêcheries (0.8 mg CO_2 -eq m⁻² d⁻¹), and the Leybeek $(0.6 \text{ mg CO}_2\text{-eq m}^{-2} \text{ d}^{-1})$ ponds. The relative contribution of E_{CH_4} peaked in summer in the Silex (73.8 %), the Tenreuken (70.9%), the Pêcheries (23.6%), and the Leybeek (58.3 %) ponds. The relative contribution of E_{CH_4} was lowest in winter in the Silex (22.1%), the Tenreuken (10.0%), the Pêcheries (6.7%), and the Leybeek (1.0%) ponds.

The annual emissions in CO₂-eq of the three GHGs (CO₂, CH₄, and N₂O) in 2022 and 2023 were higher in the two clear-water ponds $(1.3 \pm 0.5 \text{ and } 1.8 \pm 0.9 \text{ mg CO}_2\text{-eq})$ $m^{-2} d^{-1}$ in the Tenreuken and Silex ponds, respectively) than in the two turbid-water ponds $(1.0 \pm 0.2 \text{ and } 0.9 \pm 0.5 \text{ mg})$ CO₂-eq m⁻² d⁻¹ in the Leybeek and Pêcheries ponds, respectively) (Fig. 9) due to higher total CH₄ emissions (diffusive + ebullitive) in clear-water ponds (0.7 ± 0.4) and $1.2 \pm 0.5 \,\mathrm{mg} \,\mathrm{CO}_2$ -eq m⁻² d⁻¹ in the Tenreuken and Silex ponds, respectively) than in turbid-water ponds (0.2 ± 0.2) and $0.4 \pm 0.3 \,\mathrm{mg}$ CO₂-eq m⁻² d⁻¹ in the Leybeek and Pêcheries ponds, respectively). The contribution of N₂O to the total GHG emissions was marginal and did not affect the differences in total GHG fluxes between ponds, with the highest contribution observed in the Leybeek pond, with a contribution of 1.7 %.

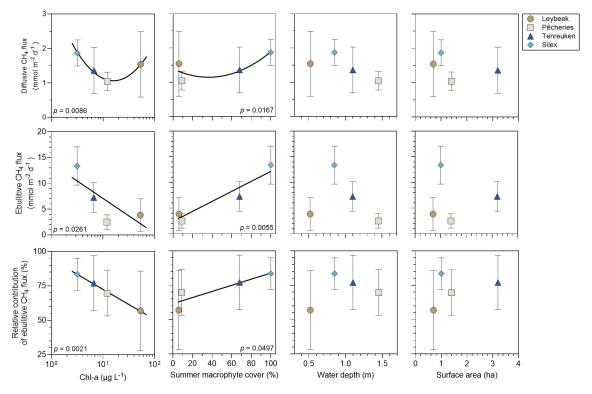


Figure 8. Mean diffusive and ebullitive CH_4 fluxes (mmol m⁻² d⁻¹) and mean ratio of ebullitive CH_4 flux to total (diffusive + ebullitive) CH_4 flux (%) versus chlorophyll-a (Chl-a, in μ g L⁻¹), total macrophyte cover in summer (%), water depth (m), and lake surface area (ha) in four ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium) from June 2021 to December 2023. Error bars indicate the standard deviation. Solid lines indicate either linear or polynomial fits. Statistical comparisons between the four ponds are summarized in Table S3.

The majority of GHG emissions in CO_2 -eq were related to CO_2 and CH_4 (diffusive + ebullitive) in the four ponds. In turbid-water ponds CO_2 represented the largest fraction of GHG emissions (68.5 % (2022) and 79.3 % (2023) in the Pêcheries pond, and 49.0 % (2022) and 58.3 % (2023) in the Leybeek pond). In clear-water ponds CH_4 represented the largest fraction of GHG emissions (66.5 % (2022) and 63.3 % (2023) in the Silex pond and 60.8 % (2022) and 50.0 % (2023) in the Tenreuken pond). The higher annual GHG emissions in CO_2 -eq from the two clear-water ponds than from the turbid-water ponds were related to the higher contribution of E_{CH_4} .

The annual GHG fluxes increased from 2022 to 2023 due to an increase in the relative contribution of CO_2 diffusive emissions in all four ponds. Diffusive CO_2 emissions averaged 0.5 mg CO_2 m⁻² d⁻¹ annually in 2022 and 0.7 mg CO_2 m⁻² d⁻¹ annually in 2023 in all four ponds. Diffusive CO_2 emissions were 2.1 times higher in summer 2023 than in summer 2022 and 2.5 times higher in autumn 2023 than in autumn 2022 and showed similar values between 2023 and 2022 in spring and winter (1.1 higher and 1.1 lower, respectively).

4 Discussion

The Leybeek and Pêcheries ponds are turbid-water systems (high Chl-a and TSM values, low submerged macrophyte cover), and the Tenreuken and Silex ponds are clear-water systems (low Chl-a and TSM values, high submerged macrophyte cover) (Figs. 1, 3). All four ponds have a relatively similar size (0.7 to 3.2 ha) and depth (0.5 to 1.4 m) and are uniformly located in an urban landscape in the city of Brussels. It can be assumed that, among the four systems, the major difference that is expected to affect GHG emissions is the dominance of aquatic primary producer, either phytoplankton or macrophytes, corresponding to two alternative states sensu Scheffer et al. (1993). Our data set provides the opportunity to investigate the effect of the two alternative states on GHG emissions from small lentic systems.

The reported pCO₂ values (40 to 13 804 ppm) (Fig. 3) in the four ponds in the city of Brussels were within the range of values typically observed in ponds (Holgerson and Raymond, 2016; Peacock et al., 2019; Audet et al., 2020) (Fig. 3). The pCO₂ values were influenced negatively by %O₂ and positively by DIN and SRP across seasons (Tables S4, S5), showing that their seasonal variability was driven by aquatic primary production and degradation of organic matter (e.g.

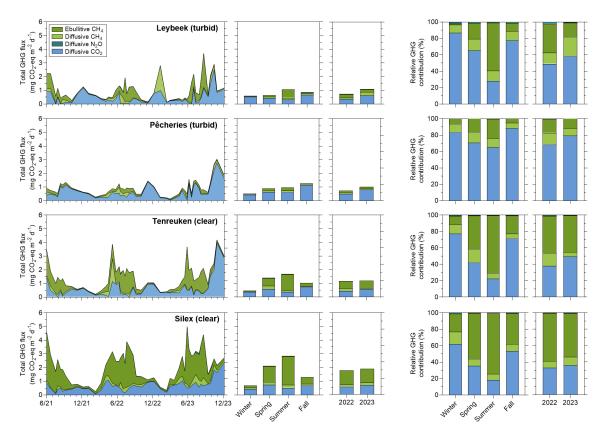


Figure 9. Seasonal and year-to-year variations of the emissions to the atmosphere of CO_2 (diffusive), CH_4 (diffusive and ebullitive), and N_2O (diffusive) expressed in CO_2 equivalents (in mg CO_2 -eq m⁻² d⁻¹) and their relative contribution (in %), in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium) from June 2021 to December 2023. Seasonal averages include data from 2021, 2022, and 2023. The annual precipitation was higher in 2023 (1011 mm) than in 2022 (701 mm).

Holgerson, 2015). Accordingly, low values of pCO₂ were generally observed in spring and summer, probably due to uptake of CO₂ by primary production from either phytoplankton or submerged macrophytes. High values of pCO₂ were observed in autumn in the four ponds and probably reflect the release of CO₂ from degradation of organic matter due to the senescence of phytoplankton or macrophytes (Fig. 3). In all four ponds, pCO₂ values were influenced positively by precipitation (Tables S4, S5), suggesting an additional control of external inputs of carbon either as organic carbon sustaining internal degradation of organic matter or as soil CO₂ (e.g. Marotta et al., 2010; Ojala et al., 2011; Rasilo et al., 2012; Vachon and del Giorgio, 2014; Holgerson, 2015). The %N₂O values (32 % to 826 %) (Fig. 3) in the four ponds were within the range of values typically observed in ponds (Audet et al., 2020; Rabaey and Cotner, 2022). When all the data were pooled, the %N2O was influenced positively by DIN (Table S4), as also frequently reported by other studies in ponds and interpreted as a control of nitrification and/or denitrification (hence N2O production) by DIN levels (Audet et al., 2020; Webb et al., 2021; Wang et al., 2021; Xie et al., 2024). The negative influence on %N2O by Tw (Table S4) might reflect the effect of the inhibition at low temperatures of the final step of denitrification leading to an accumulation of N_2O (Velthuis and Veraart, 2022) but could also indirectly result from the higher DIN concentrations observed at low T_w values (Table S6). The CH₄ dissolved concentrations (194 to 48 380 nmol L⁻¹) (Fig. 3) in the four ponds were within the range of values typically observed in ponds (Natchimuthu et al., 2014; Holgerson and Raymond, 2016; Peacock et al., 2019; Audet et al., 2020; Rabaey and Cotner, 2022; Ray et al., 2023) and were influenced positively by T_w in all four ponds individually and, when pooled (Tables S4, S5), most probably reflecting the increase in sedimentary methanogenesis with temperature (Schulz and Conrad, 1996).

Temperature also exerted a strong control on bubble flux from sediments and ebullitive CH₄ emissions. The F_{bubble} values (0 and 2078 mL m⁻² d⁻¹) in the four sampled ponds (Fig. 4) were within the range of values reported in lentic systems of equivalent size by Wik et al. (2013) (0 to 2772 mL m⁻² d⁻¹), DelSontro et al. (2016) (11 to 748 mL m⁻² d⁻¹), and Ray and Holgerson (2023) (0 to 2079 mL m⁻² d⁻¹). The F_{bubble} was influenced positively by T_{w} (Fig. 4), in agreement with previous studies (e.g. Wik et al., 2013; DelSontro et al., 2016; Aben et al., 2017; Ray and

Holgerson, 2023). Bubbling events from lake sediments are known to also be triggered by a decrease in hydrostatic pressure on the sediments due to water level fluctuations or drops in atmospheric pressure (Tokida et al., 2007; Scandella et al., 2011; Varadharajan and Hemond, 2012; Wik et al., 2013; Taoka et al., 2020; Zhao et al., 2021). In the Silex pond, in spring 2022, some peaks in F_{bubble} were related to drops in atmospheric pressure (Fig. 5) but unrelated to wind speed as shown in Gatun Lake (Keller and Stallard, 1994). A statistical model of F_{bubble} that included the contributions of T_{w} and Δp was used to quantify the relative importance of each of these two drivers (Fig. S4) and showed that an air pressure drop only seemed quantitatively important at low $T_{\rm w}$ and that the intensity of bubble flux was mainly driven by temperature change at yearly scales, in agreement with previous studies (e.g. Wik et al., 2013; DelSontro et al., 2016; Aben et al., 2017; Ray and Holgerson, 2023).

The mean %CH₄ of the bubbles $(31 \pm 21 \%)$ in the four sampled ponds in the city of Brussels was comparable to the values obtained by Wik et al. (2013) (35 \pm 25 %), DelSontro et al. (2016) (58 \pm 25 %), and Ray and Holgerson (2023) $(25 \pm 13 \%)$ in lentic systems of similar size. The increasing pattern of %CH₄ of the bubbles with $T_{\rm w}$ (Fig. 4) was most probably related to the strong dependence of methanogenesis on temperature (Schulz and Conrad, 1996). The increase in methanogenesis with temperature leads to the build-up of gas bubbles in sediments that are richer in CH₄, and consequently to higher bubble fluxes with a higher CH₄ content at higher temperatures (Figs. 4, S3). Since both F_{bubble} and the %CH₄ of the bubbles increased with $T_{\rm w}$ (Fig. 4), $E_{\rm CH_4}$ values in the four ponds were also positively related to $T_{\rm w}$ (Fig. 6), as shown previously in other small lentic systems (e.g. Wik et al., 2013; DelSontro et al., 2016; Natchimuthu et al., 2016; Aben et al., 2017; Ray and Holgerson, 2023; Rabaey and Cotner, 2024). Yet, the dependency of CH₄ ebullition on temperature (Q_{10}) was different among the four ponds and was negatively related to depth including data from systems in Quebec (DelSontro et al., 2016) and the Netherlands (Aben et al., 2017) (Fig. 6). This implies that an increase in $T_{\rm w}$ leads to a smaller increase in CH₄ ebullitive fluxes (lower Q_{10}) in deeper ponds as the impact of hydrostatic pressure on sediments is higher in deeper ponds compared to shallow ponds, restricting bubble formation and release (e.g. DelSontro et al., 2016). This dependence of Q_{10} of CH₄ ebullition on depth suggests that the response of CH₄ ebullition to heat waves (or longer-term warming) might be more intense the shallower the pond, in addition to other effects from heat waves on GHG emissions (e.g. Audet et al., 2017).

The values of Q_{10} for diffusive CH₄ fluxes in the four ponds were lower than those for ebullitive CH₄ fluxes (Table S9), as reported by other studies in lentic systems (DelSontro et al., 2016; Xun et al., 2024). The lower dependence on $T_{\rm w}$ of diffusive CH₄ fluxes compared to ebullitive CH₄ fluxes might be related to a lower relative change of CH₄ concentrations and k with the variation of $T_{\rm w}$. Dis-

solved CH₄ concentrations in surface waters of lentic systems are strongly affected by microbial methane oxidation (e.g. Bastviken et al., 2002). A relative increase in methanogenesis in sediments might lead to a stronger increase in CH₄ emission by ebullition than by diffusion because of a reduction of CH₄ diffusive emissions resulting from microbial methane oxidation. Additionally, k depends on wind speed, but in the four ponds, the warmer periods of the year (summer) tended to be less windy ($\sim 0.3 \text{ m s}^{-1}$) than the other seasons ($>0.6 \text{ m s}^{-1}$), also contributing to a lower dependence on $T_{\rm w}$ of CH₄ diffusive fluxes compared to ebullitive fluxes (lower Q_{10} values).

The difference in the Q_{10} of diffusive and ebullitive CH₄ fluxes was consistent with a seasonally variable contribution of the diffusive and ebullitive CH4 fluxes as a function of $T_{\rm w}$. The contribution of $E_{\rm CH_4}$ to total (diffusive + ebullitive) CH_4 emissions strongly increased with T_w in the four ponds (Fig. S5). At an annual scale, E_{CH_4} represented between 55 % and 83 % of the total CH₄ emissions in the Leybeek and Silex ponds, respectively. This finding is consistent with other studies showing that ebullitive CH₄ fluxes can account for more than half of total CH₄ emissions in small and shallow lentic systems (e.g. Wik et al., 2013; Deemer and Holgerson, 2021; Ray and Holgerson, 2023; Rabaey and Cotner, 2024). The averaged $E_{\mathrm{CH_4}}$ values were higher in the two clear-water ponds (10.4 mmol $m^{-2} d^{-1}$) than the two turbid-water ponds (3.2 mmol $m^{-2} d^{-1}$) (Fig. 7). The averaged E_{CH_4} values in the four ponds were influenced positively by macrophyte cover and negatively by Chl-a (Fig. 8). The higher E_{CH_4} values from the two clear-water ponds would suggest that the delivery of organic matter to sediments from macrophytes sustained a quantitatively larger methane production than from phytoplankton. This finding is consistent with the notion that vegetated littoral zones of lakes are hotspots of CH₄ production and emission (e.g. Hyvönen et al., 1998; Huttunen et al., 2003; Juutinen et al., 2003; Desrosiers et al., 2022). CH₄ fluxes in lentic systems have been extrapolated at a global scale assuming a dependency on aquatic productivity using Chl-a as a predictor (e.g. DelSontro et al., 2018). The negative relation between E_{CH_4} values with Chl-a (Fig. 8) shows that Chl-a concentration alone fails to predict ebullitive fluxes in macrophytedominated clear-water ponds.

The annually averaged diffusive CH₄ emissions in the four ponds seemed to respond positively to both increasing phytoplankton and macrophyte biomass, resulting in a U-shaped relation between diffusive CH₄ emissions and Chl-*a*, as well as macrophyte cover (Fig. 8). Higher values of annually averaged CH₄ diffusive fluxes occurred at the extreme values of Chl-*a* or of macrophyte cover (minimum or maximum), and lower values occurred at the intermediate values of Chl-*a* or macrophyte cover. Such a U-shape relation resulted from the inverse relationship between macrophyte cover and Chl-*a* (alternative states) and is consistent with the reported positive relation between diffusive CH₄ fluxes with both macrophyte

cover (e.g. Ray et al., 2023; Theus et al., 2023) and phytoplankton biomass (e.g. DelSontro et al., 2018; Yan et al., 2019; Bartosiewicz et al., 2021). The relative contribution of E_{CH_4} to the total annual CH₄ flux increased with the macrophyte cover (Fig. 8), in agreement with the idea of an increase in CH₄ ebullition relative to diffusive CH₄ emissions in vegetated sediments compared to unvegetated sediments (e.g. Desrosiers et al., 2022; Ray et al., 2023; Theus et al., 2023).

Fluxes of CH₄ and CO₂ have been reported to be negatively related to surface area and depth by numerous studies in ponds (e.g. Holgerson, 2015; Holgerson and Raymond, 2016; Ray et al., 2023; Theus et al., 2023) and lakes (e.g. Kankaala et al., 2013; DelSontro et al., 2018, Deemer and Holgerson, 2021; Casas-Ruiz et al., 2021; Borges et al., 2022). Annual diffusive F_{CH_4} and F_{CO_2} were both unrelated to surface area and depth in the four studied ponds (Figs. 8, S6) resulting from the narrow range of variation of water depth (0.6 to 1.4 m) and surface area (0.7 to 3.2 ha). The lack of relationship between annual F_{CO_2} and both Chl-a and macrophyte cover in the four ponds (Fig. S6) might be surprising since other studies have reported lower CO₂ fluxes in more productive lentic systems (e.g. Sand-Jensen and Staehr, 2007; Borges et al., 2022). We hypothesize that given that the four systems were either phytoplankton-dominated or macrophyte-dominated (alternative states), the ponds had an important submerged productivity, in both cases, resulting in a relatively invariant F_{CO_2} as a function of either Chl-a or macrophyte cover.

Global average emissions of GHGs in CO₂-eq from inland waters are dominated by CO₂ followed by CH₄ with a small contribution from N₂O according to Lauerwald et al. (2023). However, in small lentic systems such as ponds, the CO₂-eq emissions from CH₄ can match or dominate those of CO₂ (e.g. Webb et al., 2023; Ray and Holgerson, 2023; Rabaey and Cotner, 2024). The meta-analysis of Holgerson and Raymond (2016) suggested that the CO₂ and CH₄ emissions in CO₂-eq are numerically close in small lentic systems such as ponds but become increasingly dominated by CO₂ emissions in larger lentic systems. In the four studied ponds, the GHG emissions in CO₂-eq were dominated by CO₂ and CH₄ with a marginal contribution (<2%) from N₂O (Fig. 9). Annually, CO₂ represented the largest fraction of GHG emissions in CO₂-eq (\sim 60 %) in turbid-water ponds (Leybeek and Pêcheries), while CH₄ represented the largest fraction of GHG emissions in CO₂-eq (\sim 60 %) in clear-water ponds (Silex and Tenreuken) as a result of higher E_{CH_4} values in the clear-water ponds (Fig. 7).

The annual GHG emissions in CO_2 -eq increased from 2022 to 2023 due to an increase in the relative contribution of CO_2 diffusive emissions in all four ponds (Fig. 9) as a result of higher precipitation in 2023 (Fig. 2). Air temperatures were similar in both years (annual average of 12.2 °C in 2022 and 12.1 °C in 2023), and precipitation was 1.5 times higher in 2023 than in 2022. Higher precipitation is likely

to increase the inputs of organic and inorganic carbon from soils to ponds, as previously shown in other lentic systems (e.g. Marotta et al., 2010; Ojala et al., 2011; Rasilo et al., 2012; Vachon and del Giorgio, 2014; Holgerson, 2015). This hypothesis is only based on the comparison of 2 years, but the increase in the relative contribution of CO₂ diffusive emissions in 2023 was observed in all four ponds. The synchronicity of the increase in CO₂ diffusive emissions in 2023 compared to 2022 suggests a common uniform driver in all four ponds that would be consistent with a large variation in weather between the 2 years, such as annual precipitation. The strong El Niño event in 2023 induced low-level cyclonic wind anomalies and higher precipitation over western Europe, including Belgium (Chen et al., 2024).

5 Conclusions

Ebullitive CH₄ emissions in 2022–2023 were higher in the two clear-water, macrophyte-dominated ponds (Tenreuken and Silex) than in the two turbid-water, phytoplanktondominated ponds (Pêcheries and Leybeek) of the city of Brussels, although the diffusive CH₄ fluxes were not significantly different between the clear-water ponds and the turbidwater ponds. The annually averaged diffusive N₂O and CO₂ fluxes were not significantly different in the two clear-water ponds from those in the two turbid-water ponds. Other studies have found no difference in N₂O sedimentary production in lakes with a high and low density of submerged macrophytes. We hypothesize that CO₂ fluxes were relatively invariant among the four sampled ponds because of their similar size, depth, and putatively productivity (either from phytoplankton or submerged macrophytes). The total (diffusive and ebullitive) CH₄ emissions represented 58 % of total annual GHG emissions in CO₂-eq in the two clear-water ponds compared to 41 % in the two turbid-water ponds. CO₂ represented nearly all the remainder of total annual GHG emissions in CO₂-eq, and N₂O represented a very marginal fraction (<2%).

The seasonal variations in GHG emissions were mainly driven by CH₄ ebullitive emissions that peaked in summer (both quantitatively and relatively), as CH₄ ebullition was related positively to $T_{\rm w}$, resulting from an increase in both the flux of bubbles and CH₄ content of bubbles with warming. The pCO₂ values in the four sampled ponds increased with precipitation at a seasonal scale, probably in relation to higher inputs of organic and inorganic carbon from soils. The years 2022 and 2023 were abnormally dry and wet, respectively. The GHG emissions were higher in 2023 than 2022, mainly due to an increase in the relative contribution of CO₂ emissions, probably due to higher precipitation in response to a strong El Niño event of 2023. This would suggest that variations in precipitation also affected year-to-year variations in CO₂ emissions in addition to partly regulating seasonal variations in CO₂ emissions from the four studied ponds.

Data availability. The full data set is available at https://doi.org/10.5281/zenodo.11103556 (Bauduin et al., 2024b).

Supplement. The supplement related to this article is available online at https://doi.org/10.5194/bg-22-3785-2025-supplement.

Author contributions. AVB and NG conceived the study; TB collected field samples; TB and AVB conducted the laboratory analysis; and TB and AVB jointly interpreted data and drafted the manuscript with substantial inputs from NG.

Competing interests. The contact author has declared that none of the authors has any competing interests.

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Supplement of

Methane, carbon dioxide, and nitrous oxide emissions from two clearwater and two turbid-water urban ponds in Brussels (Belgium)

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Table S1: Emerged, submerged, and total macrophyte cover and main species in summer 2023 in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium).

Pond	Emergent macrophyte (EM) cover (%)	Submerged macrophyte (SM) cover (%)	Total macrophyte cover (%)	Main species
Leybeek	6	0 (*)	6	Caltha palustris (EM), Lythrium salicaria (EM), Phragmites australis (EM), Veronica beccabunga (EM)
Pêcheries	9	0	9	Phragmites australis (EM)
Tenreuken	11	57	68	Certophyllum demersum (SM), Iris pseudocorus (EM), Lythrum salicaria (EM), Mentha aquatica (EM), Phragmites australis (EM), Potamogeton pectinatus (SM)
Silex	13	87	100	Lemna trisulca (SM), Phragmites australis (EM).

^(*) Due to high turbidity, the bottom sediment of the Leybeek pond was not visible and the submerged macrophytes were assumed absent. This assumption is based on the fact that in the Pêcheries pond where the bottom sediment was visible owing to lower turbidity, the presence of submerged macrophytes was not observed.

 $Table \ S2: \ Recent \ operations \ in \ four \ urban \ ponds \ (Leybeek, \ P\hat{e}cheries, \ Tenreuken, \ and \ Silex) \ of \ the \ city \ of \ Brussels \ (Belgium) \ (provided \ by \ Bruxelles \ Environnement).$

Pond	Operation	Period	
Leybeek	Riverbank redevelopment	Winter 2017-2018	
Pêcheries	Dredging	Winter 2017-2018	
Tenreuken	Draining	Winter 2014-2015	
Silex	Partial draining	Winter 2015-2016	

Table S3: Outcomes of post-hoc tests of the repeated measures analysis of variance (ANOVA) for data grouped by season on log₁₀ transformed data, for chlorophyll-*a* (Chl-*a*, μg L⁻¹), total suspended matter (TSM, mg L⁻¹), oxygen saturation level (%O₂, %), partial pressure of CO₂ (pCO₂, ppm), dissolved CH₄ concentration (CH₄, nmol L⁻¹), N₂O saturation level (%N₂O, %), diffusive and ebullitive CH₄ fluxes (mmol m⁻² d⁻¹), mean ratio of ebullitive CH₄ flux to total (diffusive + ebullitive) CH₄ flux (%)) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium) from June 2021 to December 2023. Each matrix shows means of the log₁₀ transformed data along the diagonal, the *p*-values in the upper triangle, and the differences in the lower triangle.

		Spring					Summer					Fall					Winter		
									Chl-a	(μg L ⁻¹)									
	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[3.51]	< 0.0001	< 0.0001	0.0001	Leybeek	[4.143]	< 0.0001	< 0.0001	< 0.0001	Leybeek	[3.843]	< 0.0001	< 0.0001	< 0.0001	Leybeek	[3.18]	0.0565	0.0072	0.3504
Pêcheries	1.68	[1.83]	0.295	0.9739	Pêcheries	1.512	[2.631]	< 0.0001	< 0.0001	Pêcheries	1.797	[2.046]	< 0.0001	0.9356	Pêcheries	1.197	[1.98]	0.8024	0.7448
Silex	2.247	0.567	[1.26]	0.1388	Silex	3.521	2.009	[0.622]	0.0052	Silex	3.17	1.373	[0.673]	<0.0001	Silex	1.598	0.401	[1.58]	0.2483
Tenreuken	1.545	-0.135	-0.702	[1.96]	Tenreuken	2.828	1.316	-0.693	[1.315]	Tenreuken	1.933	0.136	-1.237	[1.910]	Tenreuken	0.749	-0.448	-0.849	[2.43]
	Levbeek	Pêcheries	Silex	Tenreuken	ı	Leybeek	Pêcheries	Silex	Tenreuken	(mg L ⁻¹)	Leybeek	Pêcheries	Silex	Tenreuken	ı	Levbeek	Pêcheries	Silex	Tenreuken
Leybeek	[2.87]	0.0291	<0.0001	0.0045	Leybeek	[3,62]	<0.0001	<0.0001	<0.0001	Leybeek	[3.39]	0.0046	0.0001	0.0316	Leybeek	[3,13]	0.0180	0.1334	0.8349
Pêcheries	0.696	[2.17]	0.0128	0.8902	Pêcheries	1.168	[2.45]	<0.0001	0.0007	Pêcheries	1.225	[2.16]	0.5603	0.8778	Pêcheries	1.036	[2.10]	0.7800	0.1088
Silex	1.469	0.773	[1.40]	0.0717	Silex	2.148	0.98	[1.47]	0.8801	Silex	1.667	0.443	[1.72]	0.1857	Silex	0.731	-0.305	[2.40]	0.4936
Tenreuken	0.866	0.17	-0.603	[2.00]	Tenreuken	1.996	0.828	-0.152	[1.62]	Tenreuken	0.973	-0.251	-0.694	[2.41]	Tenreuken	0.27	-0.765	-0.46	[2.86]
				[====]), (%)				[[2.00]
	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[4.70]	0.2813	0.3066	0.8806	Leybeek	[4.64]	0.0212	0.8238	0.3765	Leybeek	[4.37]	0.0146	1	0.9451	Leybeek	[4.23]	1	0.9968	0.9998
Pêcheries	0.20486	[4.49]	0.9999	0.7039	Pêcheries	0.3311	[4.31]	0.1574	0.5317	Pêcheries	0.64372	[3.73]	0.0130	0.0559	Pêcheries	-0.00327	[4.23]	0.9955	0.9999
Silex	0.19876	-0.00611	[4.50]	0.7358	Silex	0.0968	-0.2343	[4.54]	0.8718	Silex	-0.00935	-0.65307	[4.38]	0.9315	Silex	0.02749	0.03076	[4.20]	0.9911
Tenreuken	0.08326	-0.12161	-0.1155	[4.62]	Tenreuken	0.1811	-0.15	0.0843	[4.46]	Tenreuken	0.11136	-0.53236	0.12071	[4.26]	Tenreuken	-0.01136	-0.00809	-0.03885	[4.24]
										(ppm)									
	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[7.39]	0.8774	0.1255	1	Leybeek	[7.03]	0.0015	0.0020	0.0879	Leybeek	[8.02]	0.4447	0.8843	0.9995	Leybeek	[7.83]	0.9999	0.9985	0.8730
Pêcheries	-0.2679	[7.66]	0.4377	0.8923	Pêcheries	-0.9821	[8.02]	0.9998	0.4744	Pêcheries	-0.4009	[8.42]	0.8650	0.5111	Pêcheries	0.0205	[7.81]	0.9996	0.8947
Silex	-0.8124	-0.5444	[8.20]	0.1349	Silex	-0.9607	0.0214	[8.00]	0.5264	Silex	-0.1944	0.2065	[8.21]	0.9255	Silex	0.0614	0.0409	[7.77]	0.9317
Tenreuken	-0.0131	0.2548	0.7992	[7.40]	Tenreuken	-0.6143	0.3678	0.3464	[7.65]	Tenreuken mol L ⁻¹)	-0.0297	0.3712	0.1647	[8.05]	Tenreuken	0.293	0.2725	0.2317	[7.54]
	Leybeek	Pêcheries	Silex	Tenreuken	ı	Levbeek	Pêcheries	Silex	Tenreuken	imoi L)	Leybeek	Pêcheries	Silex	Tenreuken	·	Levbeek	Pêcheries	Silex	Tenreuken
Leybeek	[7,47]	1	0.0940	0.3894	Leybeek	[7.59]	0.9975	0.0673	0.8204	Leybeek	[7,35]	0.9956	0.9483	0.9094	Leybeek	[7.02]	0.7150	0.9563	0.8229
Pêcheries	-0.0085	[7.47]	0.0988	0.4026	Pêcheries	0.05	[7.54]	0.0385	0.9022	Pêcheries	0.0959	[7.26]	0.8669	0.9720	Pêcheries	0.5289	[6.49]	0.9445	0.9972
Silex	-0.8562	-0.8477	[8.32]	0.8527	Silex	-0.65	-0.7	[8.24]	0.0055	Silex	-0.2256	-0.3214	[7.58]	0.6268	Silex	0.253	-0.2759	[6,77]	0.9841
Tenreuken	-0.5706	-0.5621	0.2856	[8.04]	Tenreuken	0.226	0.176	0.876	[7.36]	Tenreuken	0.2773	0.1815	0.5029	[7.08]	Tenreuken	0.4304	-0.0985	0.1774	[6.59]
										O (%)									
	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[2.68]	0.0043	0.3503	0.9996	Leybeek	[2.73]	< 0.0001	0.3954	0.0219	Leybeek	[3.01]	0.0174	0.9676	0.7439	Leybeek	[3.20]	0.4370	0.8081	0.9700
Pêcheries	0.3847	[2.29]	0.2136	0.0057	Pêcheries	0.739	[1.99]	0.0010	0.0597	Pêcheries	0.6261	[2.38]	0.0525	0.1667	Pêcheries	0.3187	[2.88]	0.9198	0.7025
Silex	0.1763	-0.2084	[2.50]	0.4070	Silex	0.214	-0.525	[2.51]	0.5175	Silex	0.0916	-0.5344	[2.92]	0.9451	Silex	0.1864	-0.1323	[3.01]	0.9691
Tenreuken	0.0111	-0.3736	-0.1652	[2.67]	Tenreuken	0.398	-0.341	0.184	[2.33]	Tenreuken	0.2023	-0.4237	0.1107	[2.81]	Tenreuken	0.0927	-0.226	-0.0937	[3.10]
				T = .						ux (mmol m ⁻² d ⁻¹)	T =		T = .			1		
.	Leybeek	Pêcheries	Silex	Tenreuken	L	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken	L	Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[0.731]	0.9848	0.5416	0.2461	Leybeek	[0.763]	0.9816	0.0860	0.7022	Leybeek	[0.755]	0.9211	0.9995	0.8276	Leybeek	[0.799]	0.5670	0.5960	0.5253
Pêcheries	-0.0153	[0.746]	0.7547	0.4205	Pêcheries	0.00703	[0.756]	0.0312	0.8919	Pêcheries	0.01584	[0.739]	0.9536	0.9960	Pêcheries	0.08433	[0.715]	1 10 710	0.9999
Silex Tenreuken	-0.0586 -0.0828	-0.0432 -0.0674	[0.790] -0.0242	0.9444	Silex Tenreuken	-0.04465 0.02014	-0.05168 0.01311	[0.808]	0.0039 [0.743]	Silex Tenreuken	0.00279	-0.01305 0.0056	[0.752] 0.01865	0.8786	Silex Tenreuken	0.08126 0.08877	-0.00307 0.00444	[0.718] 0.00751	0.9994
тешенкеп	-0.0626	-0.0074	-0.0242	[0.814]	тешенкеп	0.02014	0.01311	0.00479	[01.10]	lux (mmol m ⁻² d ⁻¹	0.02144	0.0036	0.01803	[0./34]	тешенкеп	0.06677	0.00444	0.00731	[0.710]
	Leybeek	Pêcheries	Silex	Tenreuken	1	Leybeek	Pêcheries	Silex	Tenreuken	iux (iiiiii) kui	Leybeek	Pêcheries	Silex	Tenreuken	ı	Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[0.735]	0.9924	<0.0001	0.0291	Leybeek	[0,966]	0.0020	<0.0001	<0.0001	Leybeek	[0,756]	0.8897	0.0003	0.0448	Leybeek	[0.695]	<0.0001	<0.0001	<0.0001
Pêcheries	-0.0151	[0.750]	<0.0001	0.0557	Pêcheries	0.16	[0.806]	<0.0001	<0.0001	Pêcheries	0.0238	[0.732]	< 0.0003	0.0265	Pêcheries	-0.014288	[0.710]	<0.0001	0.9996
Silex	-0.354	-0.3389	[1.089]	0.0050	Silex	-0.478	-0.638	[1.444]	<0.0001	Silex	-0.1525	-0.1763	[0.909]	0.0074	Silex	-0.061422	-0.047134	[0.757]	<0.0001
Tenreuken	-0.1586	-0.1435	0.1954	[0.894]	Tenreuken	-0.233	-0.393	0.245	[1.199]	Tenreuken	-0.0375	-0.0613	0.115	[0.794]	Tenreuken	-0.015041	-0.000753	0.04638	[0.710]
									Ebullitive C	CH ₄ ratio (%)									
	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken	, , ,	Leybeek	Pêcheries	Silex	Tenreuken		Leybeek	Pêcheries	Silex	Tenreuken
Leybeek	[3.68]	0.2437	0.0005	0.0130	Leybeek	[4.36]	0.0819	0.0576	0.0154	Leybeek	[3.67]	0.6749	0.0207	0.1442	Leybeek	[2.14]	< 0.0001	< 0.0001	< 0.0001
Pêcheries	-0.318	[3.99]	0.0775	0.5463	Pêcheries	0.1376	[4.22]	< 0.0001	< 0.0001	Pêcheries	-0.26	[3.93]	0.2325	0.7120	Pêcheries	-1.85036	[3.99]	0.0429	1
Silex	-0.735	-0.417	[4.41]	0.6554	Silex	-0.1461	-0.2838	[4.51]	0.9578	Silex	-0.705	-0.445	[4.37]	0.8217	Silex	-2.15178	-0.30142	[4.29]	0.0397
Tenreuken	-0.541	-0.223	0.194	[4.22]	Tenreuken	-0.1745	-0.3121	-0.0284	[4.54]	Tenreuken	-0.505	-0.246	0.2	[4.17]	Tenreuken	-1.85239	-0.00203	0.29939	[3.99]

Table S4: Results of generalized linear mixed models (GLMM) assessing the relationships between environmental variables and partial pressure of CO_2 (p CO_2 , ppm), dissolved CH_4 concentration (CH_4 , nmol L^{-1}), and N_2O saturation level (% N_2O , %) for merged data collected in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium) from June 2021 to December 2023. Fixed effects include water temperature (Tw, °C), daily precipitation (mm d⁻¹), concentration of chlorophyll-a (Chl-a, μ g L^{-1}), concentration of total suspended matter (TSM, mg L^{-1}), oxygen saturation level (% O_2 , %), concentration of soluble reactive phosphorus (SRP, μ mol L^{-1}), concentration of dissolved inorganic nitrogen (DIN, μ mol L^{-1}), and concentrations of nitrite (NO_2 , μ mol L^{-1}), nitrate (NO_3 , μ mol L^{-1}), and ammonium (NH_4 , μ mol L^{-1}). Regression coefficients (μ), standard errors (SE) and p-values are presented for each relationship tested on \log_{10} transformed data. The models include "pond" and "sampling date" as a random effect to account for repeated measurements.

	pCO ₂ (ppm)	CH ₄ (nmol L ⁻¹)	%N ₂ O (%)
T (0C)	β = -0.05, SE = 0.16, p = 0.774	$\beta = 0.71$, SE = 0.15, p < 0.001	β = -0.44, SE = 0.08, p < 0.001
T _w (°C)	Intercept = 7.91 , SE = 0.46 , p < 0.001	Intercept = 5.57 , SE = 0.42 , p < 0.001	Intercept = 3.80 , SE = 0.25 , p < 0.001
Precipitation (mm d ⁻¹)	$\beta = 0.23$, SE = 0.06, p < 0.001	β = -0.02, SE = 0.07, p = 0.812	β = -0.02, SE = 0.04, p = 0.599
r recipitation (mm u)	Intercept = 7.54 , SE = 0.17 , p < 0.001	Intercept = 7.51 , SE = 0.18 , p < 0.001	Intercept = 2.62 , SE = 0.14 , p < 0.00
Chl-a (µg L ⁻¹)	β = -0.16, SE = 0.05, p = 0.008	β = -0.09, SE = 0.07, p = 0.186	β = -0.00, SE = 0.03, p = 0.941
Cm-a (µg L)	Intercept = 8.13 , SE = 0.16 , p < 0.001	Intercept = 7.70 , SE = 0.20 , p < 0.001	Intercept = 2.60 , SE = 0.15 , p < 0.001
TSM (mg L ⁻¹)	β = -0.09, SE = 0.08, p = 0.228	β = -0.04, SE = 0.09, p = 0.679	$\beta = 0.01$, SE = 0.04, p = 0.761
15W (ling L)	Intercept = 8.00 , SE = 0.22 , p < 0.001	Intercept = 7.58 , SE = 0.26 , p < 0.001	Intercept = 2.56 , SE = 0.16 , p < 0.001
%O ₂ (%)	β = -1.16, SE = 0.12, p < 0.001	β = -0.12, SE = 0.17, p = 0.495	$\beta = 0.01$, SE = 0.08, p = 0.942
/002 (/0)	Intercept = 12.88 , SE = 0.53 , p < 0.001	Intercept = 8.02 , SE = 0.78 , p < 0.001	Intercept = 2.57 , SE = 0.37 , p < 0.001
SRP (µmol L ⁻¹)	$\beta = 0.45$, SE = 0.09, p < 0.001	$\beta = 0.08$, SE = 0.12, p = 0.468	β = -0.06, SE = 0.05, p = 0.254
SKI (µmoi L)	Intercept = 7.22 , SE = 0.18 , p < 0.001	Intercept = 7.39 , SE = 0.21 , p < 0.001	Intercept = 2.67 , SE = 0.14 , p < 0.001
DIN (µmol L ⁻¹)	$\beta = 0.30$, SE = 0.07, p < 0.001	β = -0.09, SE = 0.09, p = 0.309	$\beta = 0.09$, SE = 0.04, p = 0.029
DIN (µmor L)	Intercept = 6.91 , SE = 0.26 , p < 0.001	Intercept = 7.76 , SE = 0.30 , p < 0.001	Intercept = 2.32 , SE = 0.18 , p < 0.001
NO ₂ (μmol L ⁻¹)			$\beta = 0.12$, SE = 0.08, p = 0.145
14O ₂ (μιιίοι Ε΄)			Intercept = 2.54 , SE = 0.13 , p < 0.001
NO3 (µmol L-1)			$\beta = 0.06$, SE = 0.04, p = 0.152
11Ο3 (μποι Ε΄)			Intercept = 2.45 , SE = 0.16 , p < 0.001
NH₄⁺(μmol L⁻¹)			$\beta = 0.09$, SE = 0.03, p = 0.007
14114 (µ11101 L)			Intercept = 2.45 , SE = 0.14 , p < 0.001

Table S5: Results of linear regression models between partial pressure of CO_2 (p CO_2 , ppm), dissolved CH_4 concentration (CH_4 , nmol L^{-1}), and N_2O saturation level (% N_2O , %) and water temperature (T_W , °C), daily precipitation (mm d⁻¹), concentration of chlorophyll-a (Chl-a, µg L^{-1}), concentration of total suspended matter (TSM, mg L^{-1}), oxygen saturation level (% O_2 , %), concentration of soluble reactive phosphorus (SRP, µmol L^{-1}), concentration of dissolved inorganic nitrogen (DIN, µmol L^{-1}) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium) from June 2021 to December 2023.. Regression coefficients (β), standard errors (SE) and significance values (β) are presented for each relationship tested on \log_{10} transformed data.

		Leybeek						
	pCO ₂ (ppm)	CH ₄ (nmol L ⁻¹)	%N ₂ O (%)					
	$\beta = -0.58$, SE = 0.31, p = 0.068	$\beta = 0.63$, SE = 0.31, p = 0.048	$\beta = -0.41$, SE = 0.11, p < 0.001					
Tw (°C)	Intercept = 9.02, SE = 0.86, p < 0.001	Intercept = 5.72 , SE = 0.85 , p < 0.001	Intercept = 3.97, SE = 0.29 , p < 0.001					
1	$\beta = 0.42$, SE = 0.11, p < 0.001	$\beta = 0.25$, SE = 0.12, p = 0.043	β = -0.05, SE = 0.05, p = 0.315					
Precipitation (mm d ⁻¹)	Intercept = 6.96, SE = 0.18, $p < 0.001$	Intercept = 7.14, SE = 0.20 , p < 0.001	Intercept = 2.90, SE = 0.08, p < 0.001					
	$\beta = -0.39$, SE = 0.24, p = 0.104	$\beta = 0.22$, SE = 0.24, p = 0.376	$\beta = -0.13$, SE = 0.09, p = 0.150					
Chl-a (µg L ⁻¹)	Intercept = 8.92, SE = 0.91 , p < 0.001	Intercept = 6.61, SE = 0.93 , p < 0.001	Intercept = 3.35, SE = 0.35 , p < 0.001					
	$\beta = 0.05$, SE = 0.22, p = 0.810	$\beta = 0.23$, SE = 0.21, p = 0.281	$\beta = -0.11$, SE = 0.08, p = 0.169					
TSM (mg L ⁻¹)	Intercept = 7.26, SE = 0.74, p < 0.001	Intercept = 6.65, SE = 0.73, p < 0.001	Intercept = 3.22, SE = 0.28, p < 0.001					
	$\beta = -1.66$, SE = 0.31, p < 0.001	β = -0.24, SE = 0.40, p = 0.539	$\beta = -0.22$, SE = 0.15, p = 0.150					
%O ₂ (%)	Intercept = 14.96, SE = 1.41, p < 0.001	Intercept = 8.54 , SE = 1.80 , p < 0.001	Intercept = 3.83, SE = 0.67, p < 0.001					
		$\beta = -0.11$, SE = 0.25, p = 0.650	<u> </u>					
SRP (µmol L ⁻¹)	$\beta = 0.44$, SE = 0.24, p = 0.078	1 1	β = -0.03, SE = 0.10, p = 0.728					
	Intercept = 6.97, SE = 0.30, p < 0.001	Intercept = 7.55, SE = 0.31, p < 0.001	Intercept = 2.88, SE = 0.12, p < 0.001					
DIN (µmol L-1)	$\beta = 0.49$, SE = 0.16, p = 0.004	$\beta = 0.08$, SE = 0.17, p = 0.658	$\beta = 0.14$, SE = 0.06, p = 0.032					
	Intercept = 5.97 , SE = 0.50 , p < 0.001	Intercept = 7.19 , SE = 0.55 , p < 0.001	Intercept = 2.42 , SE = 0.20 , p < 0.001					
		Pêcheries						
	pCO ₂ (ppm)	CH ₄ (nmol L ⁻¹)	%N ₂ O (%)					
T _W (°C)	β = -0.01, SE = 0.18, p = 0.954	$\beta = 0.83$, SE = 0.20, p < 0.001	β = -0.71, SE = 0.08, p < 0.001					
IW (C)	Intercept = 8.02 , SE = 0.50 , p < 0.001	Intercept = 5.04, SE = 0.57, p < 0.001	Intercept = 4.22 , SE = 0.22 , p < 0.001					
Precinitation (mm d-1)	β = 0.20, SE = 0.06, p = 0.003	β = -0.16, SE = 0.09, p = 0.077	$\beta = 0.01$, SE = 0.05, p = 0.770					
Precipitation (mm d ⁻¹)	Intercept = 7.78 , SE = 0.10 , p < 0.001	Intercept = 7.49 , SE = 0.15 , p < 0.001	Intercept = 2.25 , SE = 0.08 , p < 0.001					
Chl., (. T.b.	β = -0.02, SE = 0.11, p = 0.870	$\beta = 0.07$, SE = 0.15, p = 0.650	β = -0.10, SE = 0.08, p = 0.206					
Chl-a (μg L ⁻¹)	Intercept = 8.04 , SE = 0.26 , p < 0.001	Intercept = 7.16 , SE = 0.35 , p < 0.001	Intercept = 2.49 , SE = 0.19 , p < 0.001					
mone (rely	β = -0.06, SE = 0.13, p = 0.647	$\beta = 0.21$, SE = 0.16, p = 0.212	β = -0.09, SE = 0.09, p = 0.307					
TSM (mg L ⁻¹)	Intercept = 8.13 , SE = 0.30 , p < 0.001	Intercept = 6.84 , SE = 0.39 , p < 0.001	Intercept = 2.48 , SE = 0.22 , p < 0.001					
2/ 5 /2/	β = -0.88, SE = 0.15, p < 0.001	$\beta = 0.21$, SE = 0.26, p = 0.416	$\beta = -0.07$, SE = 0.14, p = 0.648					
%O ₂ (%)	Intercept = 11.71, SE = 0.62, p < 0.001	Intercept = 6.43 , SE = 1.09 , p < 0.001	Intercept = 2.54 , SE = 0.60 , p < 0.001					
	$\beta = 0.21$, SE = 0.12, p = 0.099	$\beta = 0.16$, SE = 0.17, p = 0.332	β = -0.12, SE = 0.09, p = 0.211					
SRP (µmol L-1)	Intercept = 7.71, SE = 0.19, p < 0.001	Intercept = 7.09 , SE = 0.26 , p < 0.001	Intercept = 2.42 , SE = 0.14 , p < 0.001					
	$\beta = 0.17$, SE = 0.09, p = 0.059	$\beta = -0.41$, SE = 0.10, p < 0.001	$\beta = 0.09$, SE = 0.06, p = 0.156					
DIN (µmol L ⁻¹)	Intercept = 7.51, SE = 0.27, p < 0.001	Intercept = 8.53 , SE = 0.31 , p < 0.001	Intercept = 2.00, SE = 0.20, p < 0.001					
	intercept = 7.51, SE = 0.27, p < 0.001	Tenreuken	Intercept = 2.00, SE = 0.20, p < 0.001					
pCO ₂ (ppm) CH ₄ (nmol L ⁻¹) %N ₂ O (%)								
	$\beta = 0.03$, SE = 0.26, p = 0.921	$\beta = 0.58$, SE = 0.26, p = 0.031	$\beta = -0.50$, SE = 0.11, p < 0.001					
T _w (°C)	Intercept = 7.59, SE = 0.72, p < 0.001	Intercept = 5.75, SE = 0.72, p < 0.001	Intercept = 3.96, SE = 0.31, p < 0.001					
	$\beta = 0.23$, SE = 0.10, p = 0.024	β = -0.06, SE = 0.11, p = 0.576	β = -0.03, SE = 0.05, p = 0.631					
Precipitation (mm d ⁻¹)								
	Intercept = 7.41, SE = 0.17, p < 0.001	Intercept = 7.40, SE = 0.19, p < 0.001	Intercept = 2.64, SE = 0.09, p < 0.001					
Chl-a (μg L ⁻¹)	β = -0.24, SE = 0.16, p = 0.136	$\beta = -0.37$, SE = 0.16, p = 0.024	$\beta = 0.16$, SE = 0.08, p = 0.044					
	Intercept = 8.07, SE = 0.30 , p < 0.001	Intercept = 7.97, SE = 0.31, p < 0.001	Intercept = 2.33 , SE = 0.15 , p < 0.001					
TSM (mg L-1)	$\beta = 0.03$, SE = 0.18, p = 0.862	β = -0.29, SE = 0.18, p = 0.122	$\beta = 0.30$, SE = 0.08, p < 0.001					
	Intercept = 7.60 , SE = 0.39 , p < 0.001	Intercept = 7.92 , SE = 0.40 , p < 0.001	Intercept = 1.99, SE = 0.17, p < 0.001					
%O ₂ (%)	$\beta = -1.46$, SE = 0.24, p < 0.001							
4 (/ •)		$\beta = 0.10$, SE = 0.34, p = 0.764	β = -0.27, SE = 0.16, p = 0.091					
	Intercept = 14.13, SE = 1.05, p < 0.001	Intercept = 6.88 , SE = 1.51 , p < 0.001	Intercept = 3.82 , SE = 0.70 , p < 0.001					
SRP (umol L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007	,	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768					
SRP (µmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 14.13, \text{SE} = 1.05, p < 0.001 \\ & \beta = 0.57, \text{SE} = 0.20, p = 0.007 \\ & \text{Intercept} = 6.90, \text{SE} = 0.30, p < 0.001 \end{aligned}$	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001	Intercept = 3.82 , SE = 0.70 , p < 0.001					
	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000	$\begin{aligned} & \text{Intercept} = 3.82, \text{SE} = 0.70, p < 0.001 \\ & \beta = 0.03, \text{SE} = 0.11, p = 0.768 \\ & \text{Intercept} = 2.57, \text{SE} = 0.17, p < 0.001 \\ & \beta = 0.11, \text{SE} = 0.07, p = 0.163 \end{aligned}$					
SRP (μmol L ⁻¹) DIN (μmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 14.13, \text{SE} = 1.05, p < 0.001 \\ & \beta = 0.57, \text{SE} = 0.20, p = 0.007 \\ & \text{Intercept} = 6.90, \text{SE} = 0.30, p < 0.001 \end{aligned}$	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001					
	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147	$\begin{aligned} & \text{Intercept} = 3.82, \text{SE} = 0.70, p < 0.001 \\ & \beta = 0.03, \text{SE} = 0.11, p = 0.768 \\ & \text{Intercept} = 2.57, \text{SE} = 0.17, p < 0.001 \\ & \beta = 0.11, \text{SE} = 0.07, p = 0.163 \end{aligned}$					
	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001	$\begin{aligned} & \text{Intercept} = 3.82, \text{SE} = 0.70, p < 0.001 \\ & \beta = 0.03, \text{SE} = 0.11, p = 0.768 \\ & \text{Intercept} = 2.57, \text{SE} = 0.17, p < 0.001 \\ & \beta = 0.11, \text{SE} = 0.07, p = 0.163 \end{aligned}$					
DIN (μmol L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex	Intercept = 3.82 , SE = 0.70 , p < 0.001 $\beta = 0.03$, SE = 0.11 , p = 0.768 Intercept = 2.57 , SE = 0.17 , p < 0.001 $\beta = 0.11$, SE = 0.07 , p = 0.163 Intercept = 2.30 , SE = 0.23 , p < 0.001					
	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001 pCO_2 (ppm)	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH_4 (nmol L ⁻¹)	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%)					
DIN (μmol L ⁻¹) T _W (°C)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex $CH_4 (nmol L^{-1})$ β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001 $ pCO_2 (ppm) $ β = 0.16, SE = 0.12, p = 0.190 Intercept = 7.62, SE = 0.33, p < 0.001 β = 0.11, SE = 0.05, p = 0.021 Intercept = 7.98, SE = 0.09, p < 0.001 β = -0.26, SE = 0.09, p = 0.004	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001 $ pCO_2 (ppm) $ $ β = 0.16, SE = 0.12, p = 0.190 $ Intercept = 7.62, SE = 0.33, p < 0.001 β = 0.11, SE = 0.05, p = 0.021 Intercept = 7.98, SE = 0.09, p < 0.001 β = -0.26, SE = 0.09, p = 0.004 Intercept = 8.29, SE = 0.10, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001 $\frac{pCO_2(ppm)}{\beta} = 0.16, SE = 0.12, p = 0.190$ Intercept = 7.62, SE = 0.33, p < 0.001 β = 0.11, SE = 0.05, p = 0.021 Intercept = 7.98, SE = 0.09, p < 0.001 β = -0.26, SE = 0.09, p = 0.004 Intercept = 8.29, SE = 0.10, p < 0.001 β = -0.06, SE = 0.10, p = 0.511 Intercept = 8.16, SE = 0.17, p < 0.001 β = -0.41, SE = 0.26, p = 0.012	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180 Intercept = 8.32, SE = 0.33, p < 0.001 β = -0.02, SE = 0.52, p = 0.973	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001 β = 0.06, SE = 0.12, p = 0.626 Intercept = 2.57, SE = 0.21, p < 0.001 β = -0.14, SE = 0.31, p = 0.650					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹) TSM (mg L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex CH ₄ (nmol L ⁻¹) β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180 Intercept = 8.32, SE = 0.33, p < 0.001 β = -0.02, SE = 0.52, p = 0.973 Intercept = 7.98, SE = 2.31, p = 0.001	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001 β = 0.06, SE = 0.12, p = 0.626 Intercept = 2.57, SE = 0.21, p < 0.001 β = -0.14, SE = 0.31, p = 0.650 Intercept = 3.30, SE = 1.39, p = 0.022					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹) TSM (mg L ⁻¹) %O ₂ (%)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001 $\frac{pCO_2(ppm)}{\beta} = 0.16, SE = 0.12, p = 0.190$ Intercept = 7.62, SE = 0.33, p < 0.001 β = 0.11, SE = 0.05, p = 0.021 Intercept = 7.98, SE = 0.09, p < 0.001 β = -0.26, SE = 0.09, p = 0.004 Intercept = 8.29, SE = 0.10, p < 0.001 β = -0.06, SE = 0.10, p = 0.511 Intercept = 8.16, SE = 0.17, p < 0.001 β = -0.41, SE = 0.26, p = 0.012 Intercept = 9.86, SE = 1.15, p < 0.001 β = 0.24, SE = 0.12, p = 0.059	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex $CH_4 (nmol L^{-1})$ β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180 Intercept = 8.32, SE = 0.33, p < 0.001 β = -0.02, SE = 0.52, p = 0.973 Intercept = 7.98, SE = 2.31, p = 0.001 β = 0.74, SE = 0.23, p = 0.002	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N ₂ O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001 β = 0.06, SE = 0.12, p = 0.626 Intercept = 2.57, SE = 0.21, p < 0.001 β = -0.14, SE = 0.31, p = 0.650 Intercept = 3.30, SE = 1.39, p = 0.022 β = -0.21, SE = 0.15, p = 0.173					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹) TSM (mg L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex $CH_4 (nmol L^{-1})$ β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180 Intercept = 8.32, SE = 0.33, p < 0.001 β = -0.02, SE = 0.52, p = 0.973 Intercept = 7.98, SE = 2.31, p = 0.001 β = 0.74, SE = 0.23, p = 0.002 Intercept = 6.90, SE = 0.33, p < 0.001	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N2O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001 β = 0.06, SE = 0.12, p = 0.605 Intercept = 2.57, SE = 0.21, p < 0.001 β = -0.14, SE = 0.31, p = 0.650 Intercept = 3.30, SE = 1.39, p = 0.022 β = -0.21, SE = 0.15, p = 0.173 Intercept = 2.95, SE = 0.22, p < 0.001					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹) TSM (mg L ⁻¹) %O ₂ (%) SRP (μmol L ⁻¹)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex	Intercept = 3.82, SE = 0.70, p < 0.001 β = 0.03, SE = 0.11, p = 0.768 Intercept = 2.57, SE = 0.17, p < 0.001 β = 0.11, SE = 0.07, p = 0.163 Intercept = 2.30, SE = 0.23, p < 0.001 %N2O (%) β = -0.25, SE = 0.14, p = 0.080 Intercept = 3.34, SE = 0.38, p < 0.001 β = -0.04, SE = 0.06, p = 0.556 Intercept = 2.71, SE = 0.11, p < 0.001 β = -0.11, SE = 0.11, p = 0.337 Intercept = 2.77, SE = 0.13, p < 0.001 β = 0.06, SE = 0.12, p = 0.626 Intercept = 2.57, SE = 0.21, p < 0.001 β = -0.14, SE = 0.31, p = 0.650 Intercept = 3.30, SE = 1.39, p = 0.022 β = -0.21, SE = 0.15, p = 0.173 Intercept = 2.95, SE = 0.22, p < 0.001 β = 0.10, SE = 0.11, p = 0.361					
DIN (μmol L ⁻¹) T _W (°C) Precipitation (mm d ⁻¹) Chl-a (μg L ⁻¹) TSM (mg L ⁻¹) %O ₂ (%)	Intercept = 14.13, SE = 1.05, p < 0.001 β = 0.57, SE = 0.20, p = 0.007 Intercept = 6.90, SE = 0.30, p < 0.001 β = 0.43, SE = 0.13, p = 0.003 Intercept = 6.38, SE = 0.42, p < 0.001	Intercept = 6.88, SE = 1.51, p < 0.001 β = 0.00, SE = 0.23, p = 1.000 Intercept = 7.33, SE = 0.34, p < 0.001 β = -0.23, SE = 0.15, p = 0.147 Intercept = 8.01, SE = 0.48, p < 0.001 Silex $CH_4 (nmol L^{-1})$ β = 0.83, SE = 0.21, p < 0.001 Intercept = 5.68, SE = 0.57, p < 0.001 β = -0.09, SE = 0.10, p = 0.395 Intercept = 8.00, SE = 0.17, p < 0.001 β = -0.34, SE = 0.18, p = 0.064 Intercept = 8.21, SE = 0.21, p < 0.001 β = -0.26, SE = 0.19, p = 0.180 Intercept = 8.32, SE = 0.33, p < 0.001 β = -0.02, SE = 0.52, p = 0.973 Intercept = 7.98, SE = 2.31, p = 0.001 β = 0.74, SE = 0.23, p = 0.002 Intercept = 6.90, SE = 0.33, p < 0.001	Intercept = 3.82 , SE = 0.70 , p < 0.001 β = 0.03 , SE = 0.11 , p = 0.768 Intercept = 2.57 , SE = 0.17 , p < 0.001 β = 0.11 , SE = 0.07 , p = 0.163 Intercept = 2.30 , SE = 0.23 , p < 0.001 %N ₂ O (%) β = -0.25 , SE = 0.14 , p = 0.080 Intercept = 3.34 , SE = 0.38 , p < 0.001 β = -0.04 , SE = 0.06 , p = 0.556 Intercept = 2.71 , SE = 0.11 , p < 0.001 β = -0.11 , SE = 0.11 , p = 0.337 Intercept = 2.77 , SE = 0.13 , p < 0.001 β = 0.06 , SE = 0.12 , p = 0.626 Intercept = 2.57 , SE = 0.21 , p < 0.001 β = -0.14 , SE = 0.31 , p = 0.650 Intercept = 3.30 , SE = 1.39 , p = 0.022 β = -0.21 , SE = 0.15 , p = 0.173 Intercept = 2.95 , SE = 0.22 , p < 0.001					

Table S6: Results of linear regressions in individual ponds and generalized linear mixed models (GLMM) merging all four ponds assessing the relationships with water temperature (T_W in $^{\circ}$ C) of concentration of chlorophyll-a (Chl-a, μg L^{-1}), concentration of total suspended matter (TSM, mg L^{-1}), oxygen saturation level ($^{\circ}$ O₂, $^{\circ}$), concentration of soluble reactive phosphorus (SRP, μ mol L^{-1}), and concentration of dissolved inorganic nitrogen (DIN, μ mol L^{-1}). Regression coefficients (β), standard errors (SE) and significance values (ρ) are presented for each relationship tested on \log_{10} transformed data.

Leybeek (l	Linear regression)							
Chl-a (μg L ⁻¹) versus T _w (°C)	$\beta = 0.62$, SE = 0.17, p < 0.001							
	Intercept = 2.10, SE = 0.48, p < 0.001 β = 0.24, SE = 0.22, p = 0.269							
TSM (mg L ⁻¹) versus T _W (°C)	Intercept = 2.67 , SE = 0.60 , p < 0.001							
%O ₂ (%) versus T _w (°C)	$\beta = 0.32$, SE = 0.11, p = 0.006							
	Intercept = 3.66, SE = 0.31, p < 0.001 β = 0.03, SE = 0.19, p = 0.889							
SRP (µmol L ⁻¹) versus T _W (°C)	Intercept = 1.00 , SE = 0.53 , p = 0.065							
DIN (µmol L-1) versus Tw (°C)	β = -0.51, SE = 0.26, p = 0.060 Intercept = 4.41, SE = 0.73, p < 0.001							
Pêcheries (Linear regression)							
-								
Chl - (I ·) T (0C)	$\beta = 0.46$, SE = 0.23, p = 0.051							
Chl-a (μg L ⁻¹) versus T _w (°C)	Intercept = 0.97 , SE = 0.64 , p = 0.138							
TSM (mg L-1) versus Tw (°C)	β = 0.39, SE = 0.20, p = 0.061 Intercept = 1.22, SE = 0.56, p = 0.035							
%O ₂ (%) versus T _w (°C)	$\beta = 0.23$, SE = 0.13, p = 0.083							
	Intercept = 3.57, SE = 0.37, p < 0.001 β = 0.38, SE = 0.20, p = 0.066							
SRP (µmol L ⁻¹) versus Tw (°C)	Intercept = 0.34 , SE = 0.56 , p = 0.543							
DIN (μmol L ⁻¹) versus T _W (°C)	β = -0.71, SE = 0.28, p = 0.014							
	Intercept = 4.91, SE = 0.78, p < 0.001 (Linear regression)							
Telli curcii	(Linear regression)							
	β = -0.86, SE = 0.20, p < 0.001							
Chl-a (μg L ⁻¹) versus T _W (°C)	Intercept = 4.07, SE = 0.57, p < 0.001							
TSM (mg L ⁻¹) versus T _W (°C)	β = -0.83, SE = 0.18, p < 0.001							
	Intercept = 4.30, SE = 0.49, p < 0.001 β = 0.17, SE = 0.12, p = 0.155							
%O ₂ (%) versus T _w (°C)	Intercept = 3.96 , SE = 0.32 , p < 0.001							
SRP (µmol L-1) versus Tw (°C)	β = 0.42, SE = 0.16, p = 0.012 Intercept = 0.20, SE = 0.45, p = 0.666							
DIN (umal I :1) uargus T (9C)	β = -0.86, SE = 0.22, p < 0.001							
DIN (μmol L ⁻¹) versus T _W (°C)	Intercept = 5.33 , SE = 0.62 , p < 0.001							
Silex (Li	Silex (Linear regression)							
	0 - 0.47 SE - 0.10 0.010							
Chl-a (µg L ⁻¹) versus Tw (°C)	β = -0.47, SE = 0.18, p = 0.010 Intercept = 2.17, SE = 0.48, p < 0.001							
TSM (mg L ⁻¹) versus T _W (°C)	β = -0.52, SE = 0.17, p = 0.004							
	Intercept = 3.03, SE = 0.46, p < 0.001 β = 0.19, SE = 0.06, p = 0.003							
%O ₂ (%) versus T _W (°C)	Intercept = 3.93, SE = 0.17, p < 0.001							
SRP (μmol L ⁻¹) versus T _W (°C)	$\beta = 0.47$, SE = 0.12, p < 0.001							
	Intercept = 0.10, SE = 0.34, p = 0.766 β = -0.34, SE = 0.19, p = 0.079							
DIN (μmol L ⁻¹) versus T _W (°C)	Intercept = 3.86 , SE = 0.51 , p < 0.001							
All data for	ur ponds (GLMM)							
Chl-a (µg L ⁻¹) versus T _w (°C)	β = -0.11, SE = 0.11, p = 0.320 Intercept = 2.46, SE = 0.68, p = 0.018							
TSM (mg L ⁻¹) warrang T (0C)	β = -0.27, SE = 0.14, p = 0.064							
TSM (mg L ⁻¹) versus T _W (°C)	Intercept = 3.06 , SE = 0.53 , p < 0.001							
%O ₂ (%) versus T _W (°C)	β = 0.22, SE = 0.06, p = 0.001 Intercept = 3.80, SE = 0.19, p < 0.001							
SRP (µmol L ⁻¹) versus T _W (°C)	$\beta = 0.32$, SE = 0.12, p = 0.011							
	Intercept = 0.43 , SE = 0.34 , p = 0.214							
DIN (μmol L ⁻¹) versus T _W (°C)	β = -0.46, SE = 0.19, p = 0.017 Intercept = 4.24, SE = 0.52, p < 0.001							

Table S7: Outcomes of post-hoc tests of the generalized linear mixed models (GLMMs) for chlorophyll-a (Chl-a, μg L⁻¹), total suspended matter (TSM, mg L⁻¹), oxygen saturation level (%O₂, %), partial pressure of CO₂ (pCO₂, ppm), dissolved CH₄ concentration (CH₄, nmol L⁻¹), N₂O saturation level (%N₂O, %), diffusive and ebullitive CH₄ fluxes (mmol m⁻² d⁻¹), mean ratio of ebullitive CH₄ flux to total (diffusive + ebullitive) CH₄ flux (%)) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium) from June 2021 to December 2023. Each matrix shows adjusted means of each pond based on GLMM along the diagonal, the p-values in the upper triangle, and the differences in the lower triangle.

		Chl-a (μg L ⁻¹)								
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[3.9809]	< 0.0001	< 0.0001	<0.0001						
Pêcheries	1.553	[2.4279]	<0.0001	<0.0001						
Silex	3.3036	1.7505	[0.6774]	<0.0001						
Tenreuken	2.1335	0.5805 TSM (mg L ⁻¹)	-1.17	[1.8474]						
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[3.42]	<0.0001	<0.0001	<0.0001						
Pêcheries	1.157	[2.26]	<0.0001	0.2334						
Silex	1.908	0.751	[1.51]	0.0002						
Tenreuken	1.393	0.236	-0.515	[2.03]						
	%O ₂ (%)									
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[4.56]	< 0.0001	0.262	0.2038						
Pêcheries	0.3086	[4.25]	0.0052	0.0085						
Silex	0.1101	-0.1985	[4.45]	0.999						
Tenreuken	0.1185	-0.1901	0.0084	[4.44]						
	T	pCO ₂ (ppm)								
.	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[7.68]	0.004	0.0011	0.6695						
Pêcheries	-0.3748	[8.05]	0.9824	0.1062 0.0419						
Silex Tenreuken	-0.4153 -0.1269	-0.0405 0.2479	[8.09] 0.2883	[7.80]						
Tenreuken	-0.1209	CH ₄ (nmol L ⁻¹)	0.2883	[7.80]						
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[7.68]	0.4687	0.0152	0.8924						
Pêcheries	0.213	[7.47]	<0.0001	0.8845						
Silex	-0.443	-0.656	[8.12]	0.0009						
Tenreuken	0.107	-0.105	0.55	[7.57]						
		%N ₂ O (%)		[,,,,]						
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[2.78]	< 0.0001	0.1275	0.0038						
Pêcheries	0.612	[2.17]	< 0.0001	<0.0001						
Silex	0.15	-0.462	[2.63]	0.6371						
Tenreuken	0.231	-0.381	0.081	[2.55]						
		ive CH4 flux (mmol 1								
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[0.132]	0.5752	0.5256	0.8845						
Pêcheries	0.032	[0.100]	0.039	0.9458						
Silex Tenreuken	-0.0339 0.0183	-0.0659 -0.0137	[0.166] 0.0522	0.1513 [0.114]						
Tellieukeli		tive CH ₄ flux (mmol		[0.114]						
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[0.262]	0.0617	<0.0001	<0.0001						
Pêcheries	0.116	[0.146]	<0.0001	<0.0001						
Silex	-0.5	-0.616	[0.761]	<0.0001						
Tenreuken	-0.221	-0.337	0.279	[0.483]						
	Relative contr	ribution of ebullitive								
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[3.89]	0.0818	< 0.0001	< 0.0001						
Pêcheries	-0.195	[4.09]	0.0002	0.0304						
Silex	-0.524	-0.33	[4.42]	0.5134						
Tenreuken	-0.414	-0.22	0.11	[4.31]						
		ive N ₂ O flux (mmol 1		T- 1						
T avils a -1-	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek Pêcheries	[0.011667] 0.011149	0.0001 [0.000518]	0.9971 0.0030	0.2096 0.0924						
Silex	0.000528	-0.010621	[0.011139]	0.0924						
Tenreuken	0.005095	-0.010621	0.004568	[0.006571]						
Temedicii		ive CO ₂ flux (mmol 1		[0.0003/1]						
	Leybeek	Pêcheries	Silex	Tenreuken						
Leybeek	[0.344]	0.0045	0.0030	0.0876						
Pêcheries	-0.125	[0.469]	0.8933	0.7459						
Silex	-0.1512	-0.0263	[0.495]	0.3152						
Tenreuken	-0.0877	0.0372	0.0635	[0.432]						

Table S8: Results of linear regression models between N_2O saturation level (% N_2O , %) and concentrations of nitrite (NO_2 , μ mol L^{-1}), nitrate (NO_3 , μ mol L^{-1}) and ammonium (NH_4 , μ mol L^{-1}). Regression coefficients (β), standard errors (SE) and p-values are presented for each relationship tested on log_{10} transformed data.

Leybeek							
work and	$\beta = 0.19$, SE = 0.13, p = 0.167						
NO2 (µmol L-1)	Intercept = 2.77, SE = 0.08, p < 0.001						
NO : (11 · l)	$\beta = 0.13$, SE = 0.06, p = 0.029						
NO ₃ (µmol L ⁻¹)	Intercept = 2.51, SE = 0.16, p < 0.001						
NH ₄ ⁺ (μmol L ⁻¹)	$\beta = 0.06$, SE = 0.06, p = 0.325						
NH4 (μιιιοι L)	Intercept = 2.76 , SE = 0.11 , p < 0.001						
Pêcheries							
NO2 (µmol L-1)	$\beta = 0.26$, SE = 0.15, p = 0.092						
- NO ₂ (μιιιοι L)	Intercept = 2.16 , SE = 0.09 , p < 0.001						
NO3 (µmol L-1)	$\beta = 0.01$, SE = 0.06, p = 0.930						
- 14O3 (μποι Ε΄)	Intercept = 2.25 , SE = 0.15 , p < 0.001						
NH4+ (µmol L-1)	$\beta = 0.17$, SE = 0.05, p = 0.001						
	Intercept = 1.95 , SE = 0.11 , p < 0.001						
Tenreuken							
NOs (umol L-1)	$\beta = 0.18$, SE = 0.13, p = 0.176						
NO ₂ - (μmol L ⁻¹)	Intercept = 2.52 , SE = 0.09 , p < 0.001						
	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528						
NO ₂ (µmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 2.52, \text{SE} = 0.09, p < 0.001 \\ & \beta = 0.05, \text{SE} = 0.08, p = 0.528 \\ & \text{Intercept} = 2.47, \text{SE} = 0.23, p < 0.001 \end{aligned}$						
NO ₃ (μmol L ⁻¹)	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528 Intercept = 2.47, SE = 0.23, p < 0.001 β = 0.17, SE = 0.06, p = 0.004						
	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528 Intercept = 2.47, SE = 0.23, p < 0.001 β = 0.17, SE = 0.06, p = 0.004 Intercept = 2.36, SE = 0.10, p < 0.001						
NO ₃ (μmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 2.52, \text{SE} = 0.09, p < 0.001 \\ & \beta = 0.05, \text{SE} = 0.08, p = 0.528 \\ & \text{Intercept} = 2.47, \text{SE} = 0.23, p < 0.001 \\ & \beta = 0.17, \text{SE} = 0.06, p = 0.004 \\ & \text{Intercept} = 2.36, \text{SE} = 0.10, p < 0.001 \\ & \text{Silex} \end{aligned}$						
NO ₃ · (μmol L ⁻¹) NH ₄ ⁺ (μmol L ⁻¹)	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528 Intercept = 2.47, SE = 0.23, p < 0.001 β = 0.17, SE = 0.06, p = 0.004 Intercept = 2.36, SE = 0.10, p < 0.001 Silex β = -0.12, SE = 0.15, p = 0.449						
NO ₃ (μmol L ⁻¹)	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528 Intercept = 2.47, SE = 0.23, p < 0.001 β = 0.17, SE = 0.06, p = 0.004 Intercept = 2.36, SE = 0.10, p < 0.001 Silex β = -0.12, SE = 0.15, p = 0.449 Intercept = 2.72, SE = 0.10, p < 0.001						
NO ₃ · (μmol L ⁻¹) NH ₄ ⁺ (μmol L ⁻¹) NO ₂ · (μmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 2.52, \text{SE} = 0.09, p < 0.001 \\ & \beta = 0.05, \text{SE} = 0.08, p = 0.528 \\ & \text{Intercept} = 2.47, \text{SE} = 0.23, p < 0.001 \\ & \beta = \textbf{0.17}, \text{SE} = \textbf{0.06}, p = \textbf{0.004} \\ & \textbf{Intercept} = \textbf{2.36}, \text{SE} = \textbf{0.10}, p < \textbf{0.001} \\ & \textbf{Silex} \\ & \beta = -0.12, \text{SE} = 0.15, p = 0.449 \\ & \textbf{Intercept} = 2.72, \text{SE} = 0.10, p < 0.001 \\ & \beta = 0.11, \text{SE} = 0.10, p = 0.278 \end{aligned}$						
NO ₃ · (μmol L ⁻¹) NH ₄ ⁺ (μmol L ⁻¹)	Intercept = 2.52, SE = 0.09, p < 0.001 β = 0.05, SE = 0.08, p = 0.528 Intercept = 2.47, SE = 0.23, p < 0.001 β = 0.17, SE = 0.06, p = 0.004 Intercept = 2.36, SE = 0.10, p < 0.001 Silex β = -0.12, SE = 0.15, p = 0.449 Intercept = 2.72, SE = 0.10, p < 0.001 β = 0.11, SE = 0.10, p = 0.278 Intercept = 2.38, SE = 0.27, p < 0.001						
NO ₃ · (μmol L ⁻¹) NH ₄ ⁺ (μmol L ⁻¹) NO ₂ · (μmol L ⁻¹)	$\begin{aligned} & \text{Intercept} = 2.52, \text{SE} = 0.09, p < 0.001 \\ & \beta = 0.05, \text{SE} = 0.08, p = 0.528 \\ & \text{Intercept} = 2.47, \text{SE} = 0.23, p < 0.001 \\ & \beta = \textbf{0.17}, \text{SE} = \textbf{0.06}, p = \textbf{0.004} \\ & \textbf{Intercept} = \textbf{2.36}, \text{SE} = \textbf{0.10}, p < \textbf{0.001} \\ & \textbf{Silex} \\ & \beta = -0.12, \text{SE} = 0.15, p = 0.449 \\ & \textbf{Intercept} = 2.72, \text{SE} = 0.10, p < 0.001 \\ & \beta = 0.11, \text{SE} = 0.10, p = 0.278 \end{aligned}$						

Table S9: Exponential fit of ebullitive and diffusive CH_4 fluxes (mmol m⁻² d⁻¹) as function of water temperature (Tw, in $^{\circ}C$) and corresponding Q_{10} in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium). Ebullitive CH_4 fluxes were measured in spring, summer, and fall in 2022 and 2023, totaling 8 days in Leybeek, Pêcheries and Tenreuken ponds and 24 days in Silex pond, with three bubble traps. Diffusive CH_4 fluxes were computed from CH_4 concentration measurements collected 46 times on each pond from June 2021 to December 2023. Dfd = degrees of freedom. Original data and fits are shown in Figure S9.

	Statistics	Leybeek	Pêcheries	Tenreuken	Silex
	Function (Tw) (mmol m ⁻² d ⁻¹)	$0.01 \cdot e^{0.32 \cdot Tw}$	$0.16 \cdot e^{0.15 \cdot Tw}$	$0.10 \cdot e^{0.23 \cdot Tw}$	$0.54 \cdot e^{0.18 \cdot Tw}$
	Q10	26.9	4.4	9.7	6.2
Ebullitive CH ₄ fluxes	r^2	0.89	0.23	0.72	0.92
	<i>p</i> -value	< 0.0001	0.0032	< 0.0001	< 0.0001
	Dfd (N-2)	22	22	19	70
	Function (Tw) (mmol m ⁻² d ⁻¹)	$0.22 \cdot e^{0.07 \cdot Tw}$	$0.39 \cdot e^{0.02 \cdot Tw}$	$0.26 \cdot e^{0.06 \cdot TW}$	$0.22 \cdot e^{0.11 \cdot Tw}$
	Q10	2.1	1.2	1.9	2.9
Diffusive CH4 fluxes	r^2	0.11	0.13	0.10	0.18
	<i>p</i> -value	0.0012	0.0009	0.0032	0.0004
	Dfd (N-2)	45	45	46	46
	Function (Tw) (%)	$\frac{1}{1 + 22 \times e^{-0.25 \cdot Tw}}$	$\frac{1}{1 + 2.43 \times e^{-0.13 \cdot T_W}}$	$\frac{1}{1 + 2.60 \times e^{-0.17 \cdot Tw}}$	$\frac{1}{1+0.40\times e^{-0.07\cdot Tw}}$
Ratio of ebullitive CH4 flux to total (diffusive + ebullitive) CH4 flux	r^2	0.71	0.17	0.45	0.30
(unituality) Cliq ilux	<i>p</i> -value	< 0.0001	0.0043	< 0.0001	< 0.0001
	Dfd (N-2)	45	45	46	46

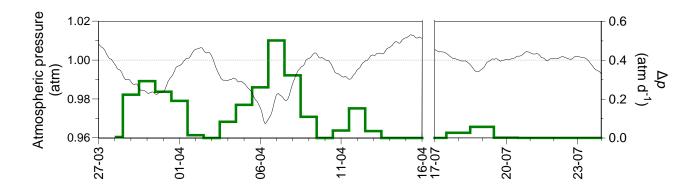


Figure S1: Atmospheric pressure (atm) and atmospheric pressure drop factor (Δp in atm d⁻¹) computed from Equation (4) according to Zhao et al. (2017) from 27 March 2022 to 15 April 2022 and from 18 July 2022 to 23 July 2022 corresponding to the period of bubble flux measurements in the Silex pond in the city of Brussels (Belgium) (Fig. 5).

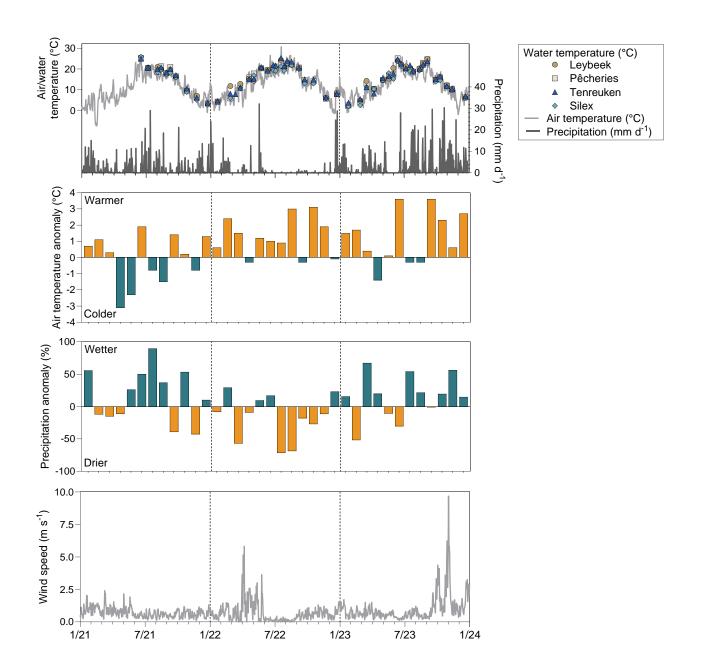


Figure S2: Daily mean air temperature (light grey line) and water temperature during sampling four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium), daily precipitation (dark grey line), monthly temperature anomaly ($^{\circ}$ C) relative to the 1991-2020 period, and daily wind speed (m s $^{-1}$) from January 2021 to December 2023.

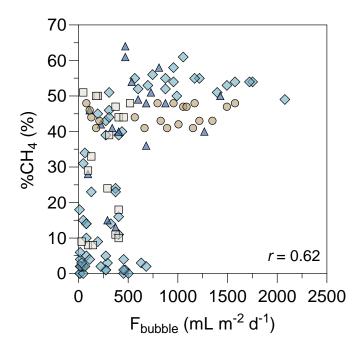


Figure S3: The relative CH_4 content in bubbles (% CH_4 , in %) as a function of bubble flux (F_{bubble} in ml m⁻² d⁻¹) in four ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium). Bubbles fluxes were measured with three bubble traps in spring, summer, and fall of 2022 and 2023, totalling 8 days in the Leybeek, Pêcheries, and Tenreuken ponds and 24 days in the Silex pond. % CH_4 correlated positively to $log_{10}(F_{bubble})$ (Pearson r=0.62, p<0.0001, n=122).

Modelled from T_W only Modelled from T_W and Δp

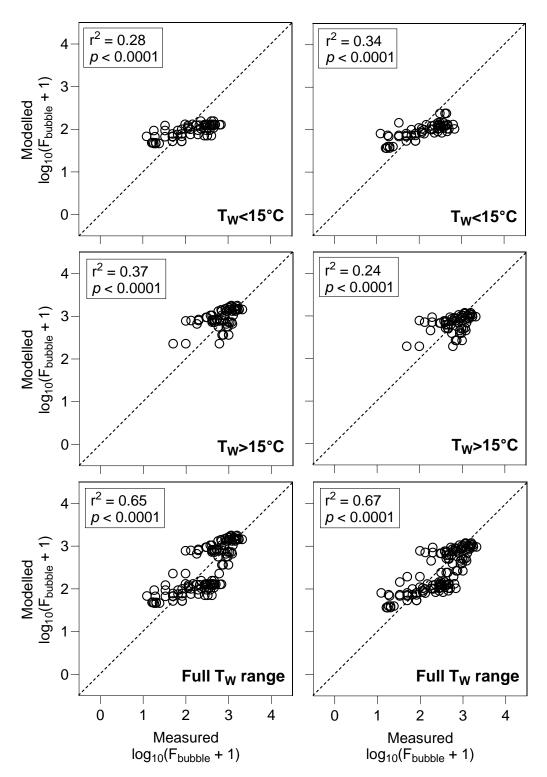


Figure S4: Comparison between measured bubble flux (F_{bubble}) and modelled F_{bubble} at low water temperature $(T_w < 15^{\circ}C)$, high temperature $(T_w > 15^{\circ}C)$, and full T_w range from multiple linear regression models with T_w alone or both T_W and atmospheric pressure drops (Δp) as drivers in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium). Bubble fluxes were measured at different seasons in 2022 and 2023, totaling 8 days in Leybeek, Pêcheries and Tenreuken ponds, and 24 days in Silex pond, with three bubble traps. To account for zero values, \log_{10} was computed from F_{bubble} plus 1. The model based on Tw alone corresponds to $\log_{10}(F_{bubble} + 1) = 0.096 \times T_w + 0.821$. The model based on both TW and Δp corresponds to $\log_{10}(F_{bubble} + 1) = 0.078 \times T_w - 0.731 \times \Delta p + 0.991$

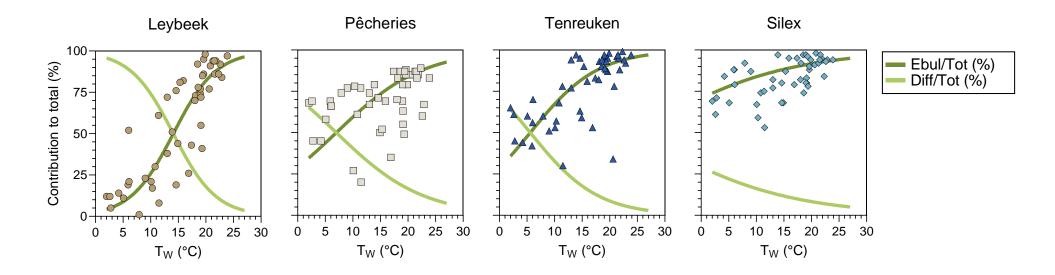


Figure S5: Relative contribution of ebullitive and diffusive CH_4 fluxes to total (ebullitive + diffusive) CH_4 flux as function of water temperature (T_W in $^{\circ}C$) in four urban ponds (Leybeek, Pêcheries, Tenreuken, and Silex) of the city of Brussels (Belgium). Ebullitive CH_4 fluxes were measured at different seasons in 2022 and 2023, totaling 8 days in Leybeek, Pêcheries and Tenreuken ponds, and 24 days in Silex pond, with three bubble traps. Data were collected 46 times on each pond from June 2021 to December 2023 for diffusive CH_4 fluxes. Regression lines of fitted data of the relative contribution of ebullition to total (Ebul/Tot) are shown in light green and the relative contribution of diffusion to total (Diff/Tot) in dark green (Table S9). These patterns of temperature dependency are consistent with those previously shown in ponds in Québec (DelSontro et al., 2016).

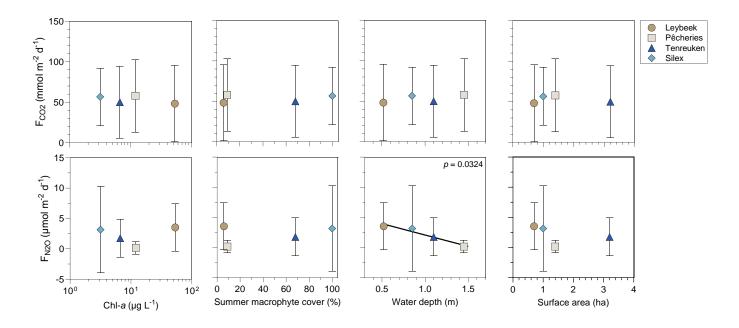


Figure S6: Mean annual diffusive CO_2 flux (F_{CO2} in mmol m^{-2} d^{-1}) and mean annual N_2O flux (F_{N2O} , μ mol m^{-2} d^{-1}) versus chlorophyll-a (Chl-a, μ g L^{-1}), total macrophyte cover in summer (%), water depth (m), and lake surface area (ha) in four ponds (Leybeek, Pêcheries, Tenreuken, and Silex) in the city of Brussels (Belgium) from June 2021 to December 2023. Error bars indicate the standard deviation.