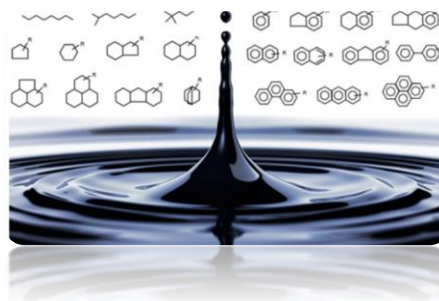




MOSH & MOAH : remediation strategies and analytical challenges



Giorgia Purcaro, Aleksandra Gorska, Sabine Danthine, Nicolas Jacquet

Gembloux Agro Bio-Tech, University of Liège, Belgium

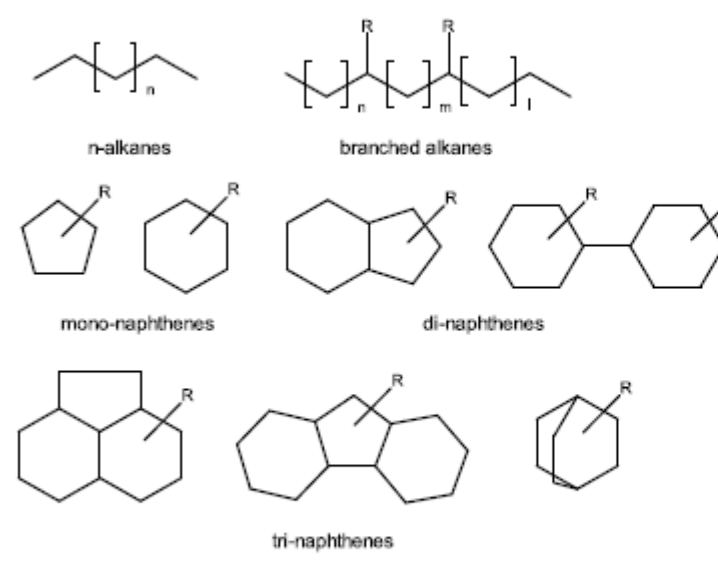
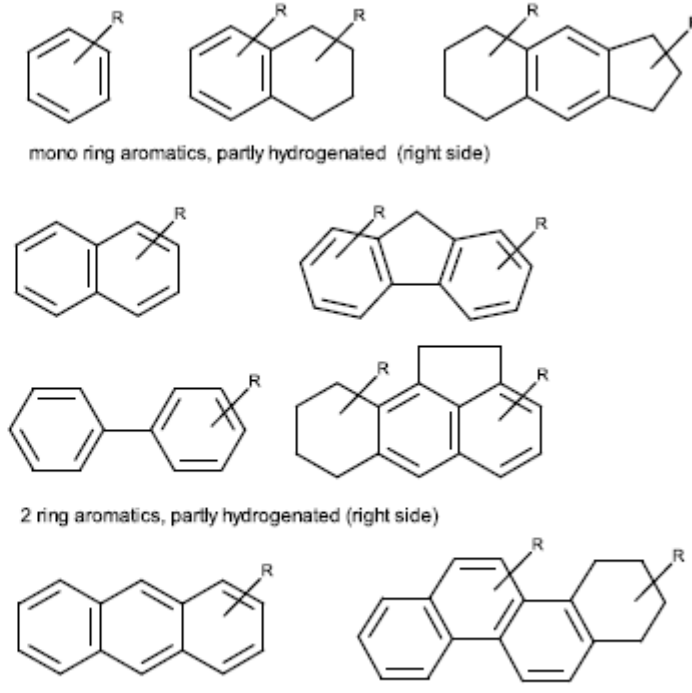
gpurcaro@uliege.be

2025 AOCS annual Meeting & Expo

April 27–30, 2025 Oregon Convention Center, Portland, Oregon, USA

MINERAL OIL HYDROCARBONS (MOH): DEFINITION*

a wide range of products deriving from petroleum distillation fractions

| MOSH Mineral oil saturated hydrocarbons | MOAH Mineral oil aromatic hydrocarbons |
|--|---|
|  <p>n-alkanes</p> <p>branched alkanes</p> <p>mono-naphthenes</p> <p>di-naphthenes</p> <p>tri-naphthenes</p> |  <p>mono ring aromatics, partly hydrogenated (right side)</p> <p>2 ring aromatics, partly hydrogenated (right side)</p> <p>3 ring aromatics, partly hydrogenated (right side)</p> |

-n-alkane
- isoalkane
- cycloalkane

Aromatic hydrocarbons, mainly
alkylated

1989

Original Research Papers

Partially Concurrent Eluent Evaporation with an Early Vapor Exit; Detection of Food Irradiation through Coupled LC-GC Analysis of the Fat

Maurus Biedermann, Konrad Grob*, and Werner Meier
Kantonales Labo

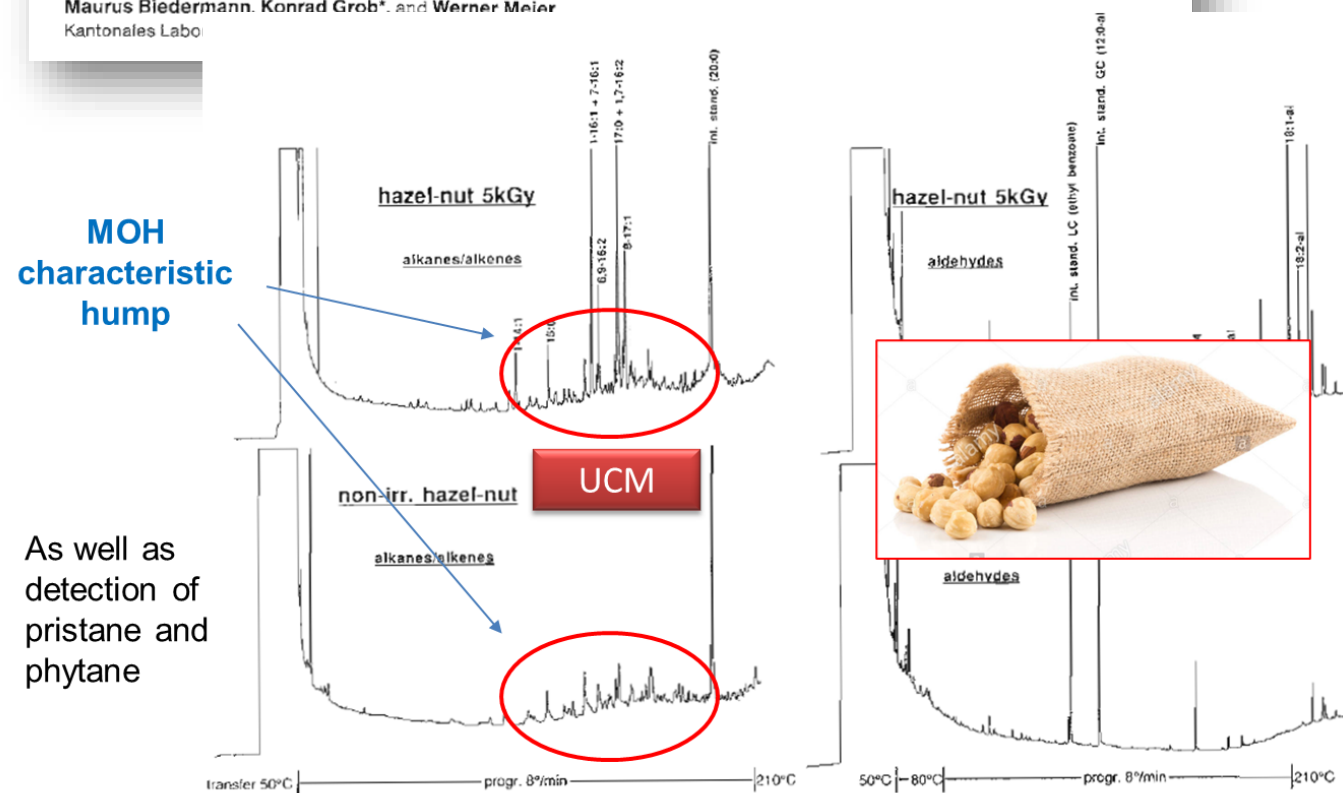


Figure 6

Alkane/alkene and aldehyde fractions of irradiated and non-irradiated hazelnuts. The poorly resolved alkanes in the matrix of the alkane/alkene fraction probably indicate a contamination of the nuts with mineral oil.



MINERAL OIL HYDROCARBONS (MOH)

1989 →
2001 →
2008/2009 →

Original Research Papers

Mineral Paraffins in Vegetable Oils and Feeds



efsa
European Food Safety Authority

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Sunflower oil: contamination with mineral oil from Ukraine - Update

Published: 28 April 2008



Eur. J. Lipid Sci. Technol. 2009, 111, 313–319

Highlight Article

How “white” was the mineral oil in the contaminated Ukrainian sunflower oils?

Maurus Biedermann and Koni Grob

Official Food Control Authority of the Canton of Zurich, Zurich, Switzerland

Up to **1800 mg/kg**
of **MOAH**

1989 →
2001 →
2008/2009 →

Original Research Papers

Mineral Paraffins in Vegetable Oils

and
Feed



ABOUT ▾ **NEWSROOM** ▾ TOPICS ▾ RESOURCES ▾ PUBLICATIONS APPLICATIONS ▾ ENGAGE ▾ CALENDAR

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JOURNAL OF AGRICULTURAL AND FOOD CHEMISTRY ARTICLE

J. Agric. Food Chem. 2009, 57, 8711–8721
DOI:10.1021/jf901375e

Aromatic Hydrocarbons of Mineral Oil Origin in Foods: Method for Determining the Total Concentration and First Results

MAURUS BIEDERMANN, KATELL FISELIER, AND KONI GROB*

Kantonales Labor (Official Food Control Authority of the Canton of Zurich), Fehrenstrasse 15,
CH-8032 Zurich, Switzerland



MINERAL OIL HYDROCARBONS (MOH)



1989

2001

2008/2009

JOURNAL OF AGRICULTURAL AND FOOD CHEMISTRY ARTICLE

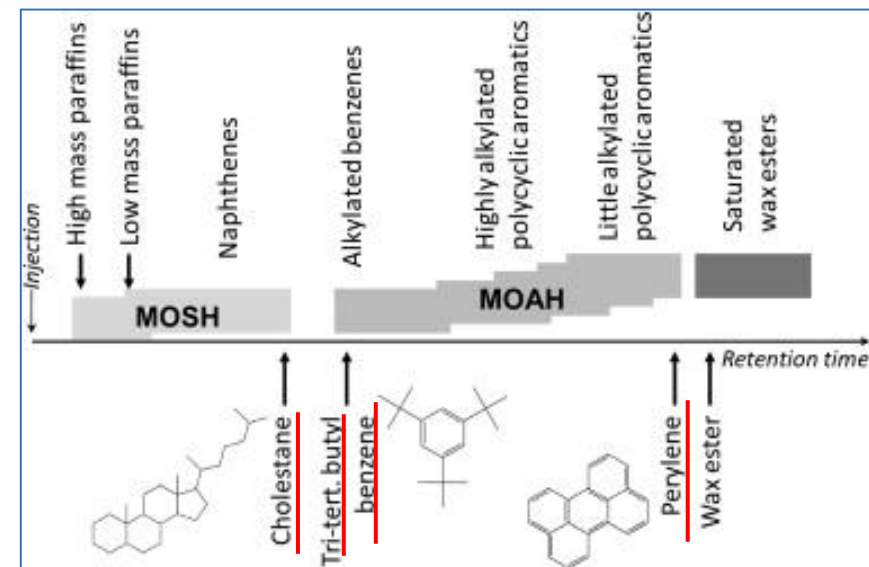
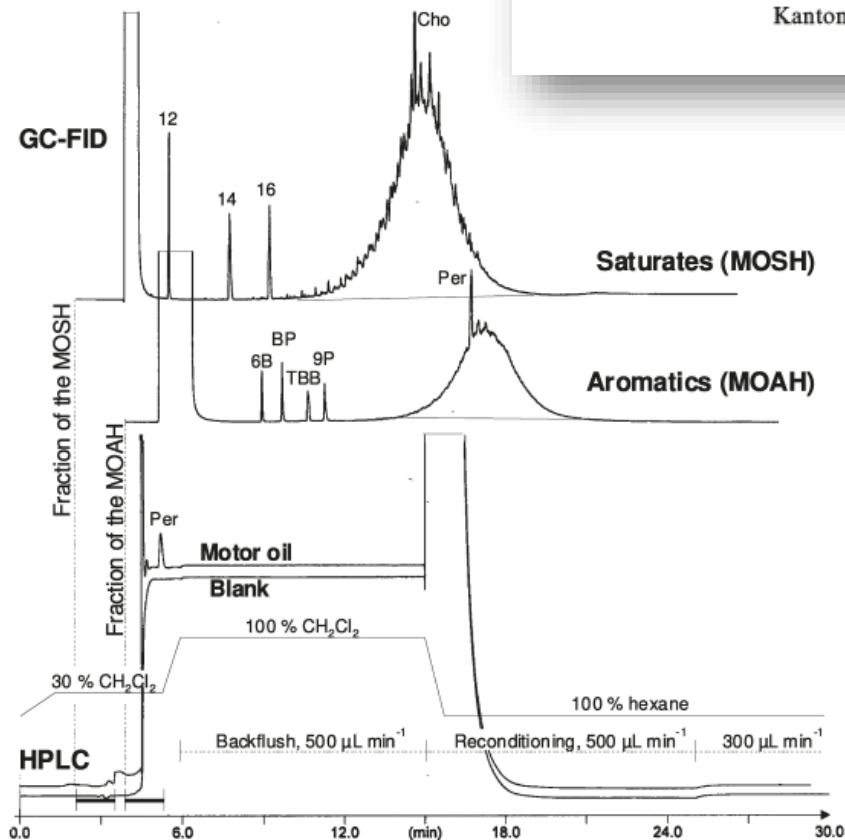
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CH-8032 Zurich, Switzerland

8714 J. Agric. Food Chem., Vol. 57, No. 19, 2009





J. Sep.

3726

Maurus Biedermann
Koni Grob

Official Food Control Authority of
the Canton of Zurich, Zurich,
Switzerland

Received May 23, 2009
Revised August 3, 2009
Accepted August 5, 2009

Research Article

Comprehensive two-dimensional GC after HPLC preseparation for the characterization of aromatic hydrocarbons of mineral oil origin in contaminated sunflower oil

J. Agric. Food Chem. 2009, 57, 8711–8721
DOI:10.1021/jf901375e

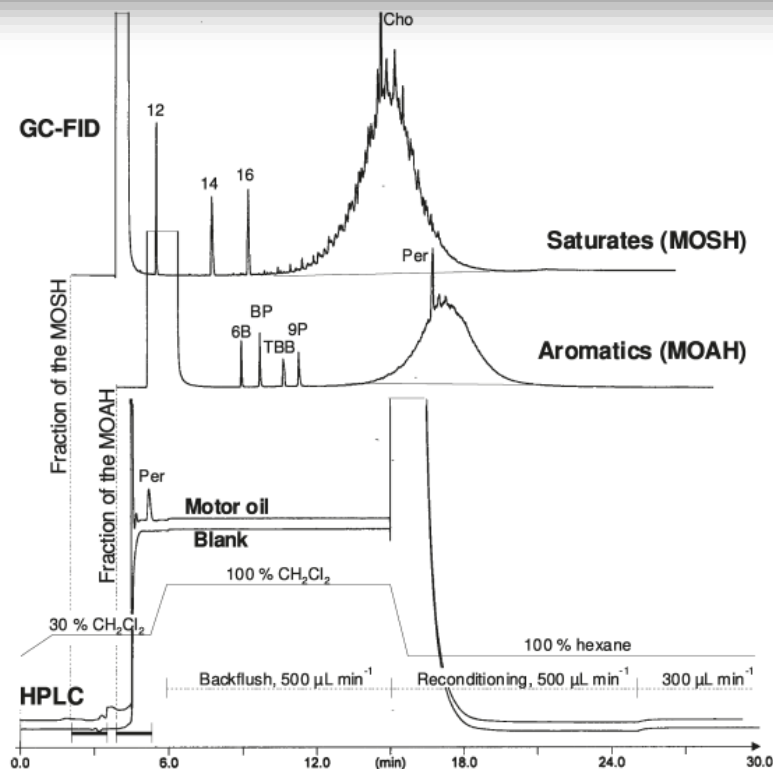
JOURNAL OF
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ARTICLE

Aromatic Hydrocarbons of Mineral Oil Origin in Foods: Method for Determining the Total Concentration and First Results

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CH-8032 Zurich, Switzerland

8714 J. Agric. Food Chem., Vol. 57, No. 19, 2009 Biedermann and Grob



“For **risk evaluation**, the concentration of the sum of all MOAH might not be satisfactory. The method of Moret et al. used HPLC to group the MOAH by ring number. In the meantime, **comprehensive two-dimensional GC (GC×GC)** was established [12, 13], which is an attractive alternative because of its simplicity, the enhanced sensitivity resulting from focusing by modulation and the additional information provided on the extent of alkylation.

Mineral oil analysis is the most widely described application of GC×GC, though mainly on products like gasoline, heavy naphtha or kerosene with a lower molecular mass than the food contaminants most frequently found.”

MINERAL OIL HYDROCARBONS (MOH)



J. Sep.

3726

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Official Food Control Authority of
the Canton of Zurich, Zurich,
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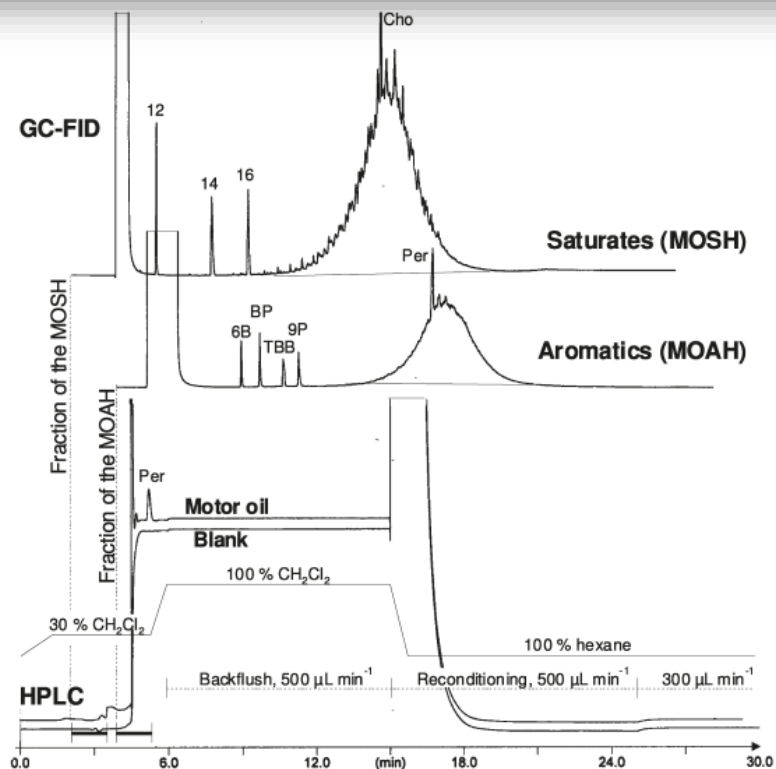
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8711 J. Agric. Food Chem., Vol. 57, No. 19, 2009 Biedermann and Grob



Up to 1800 mg/kg
of **MOAH**

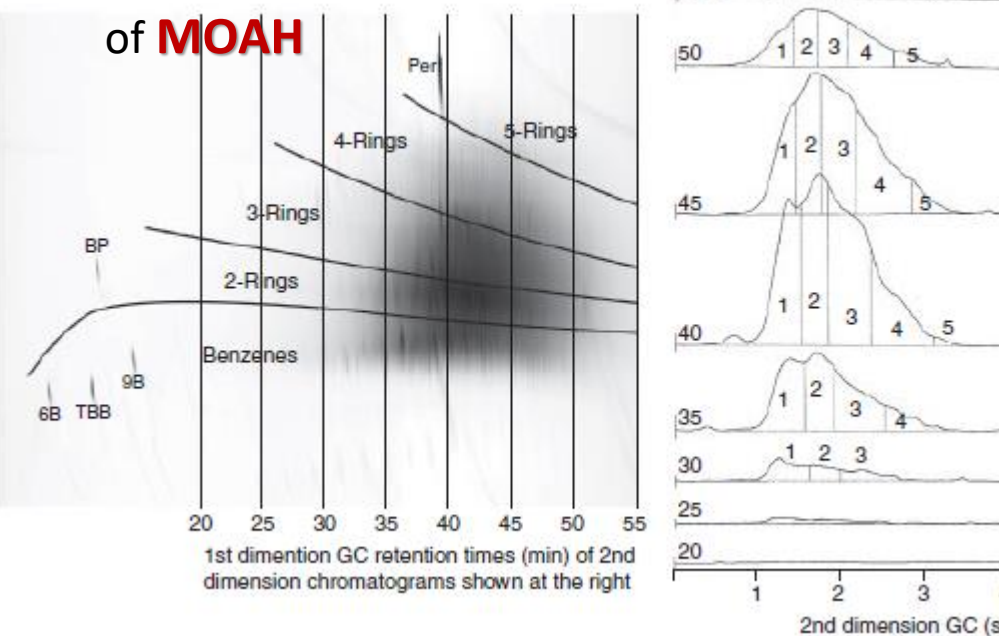
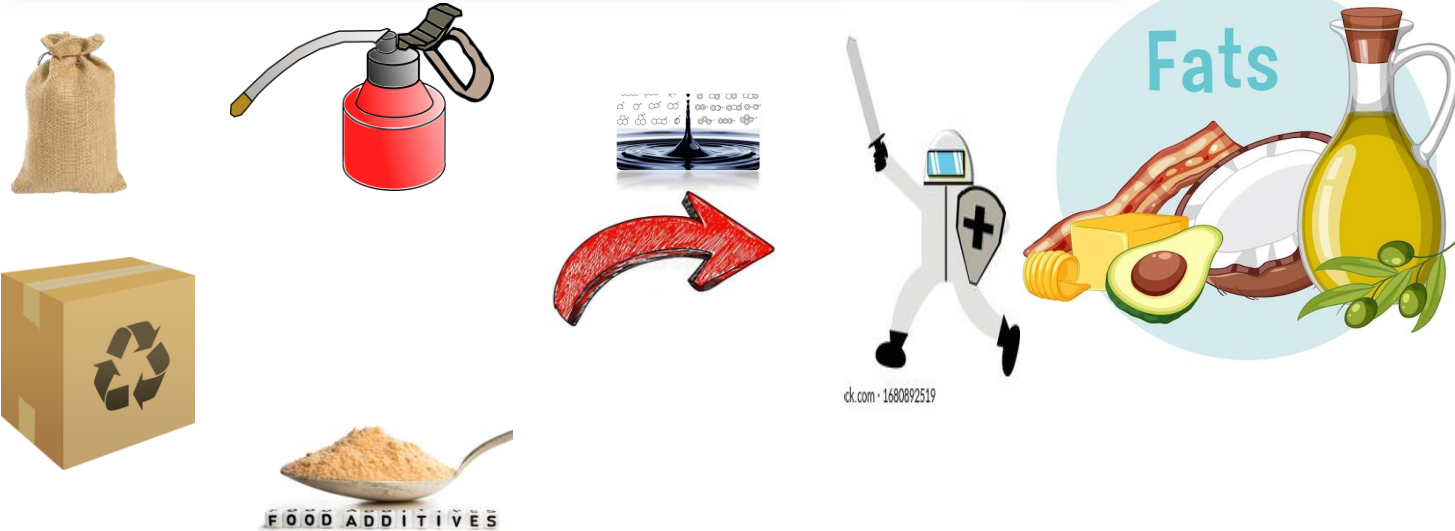
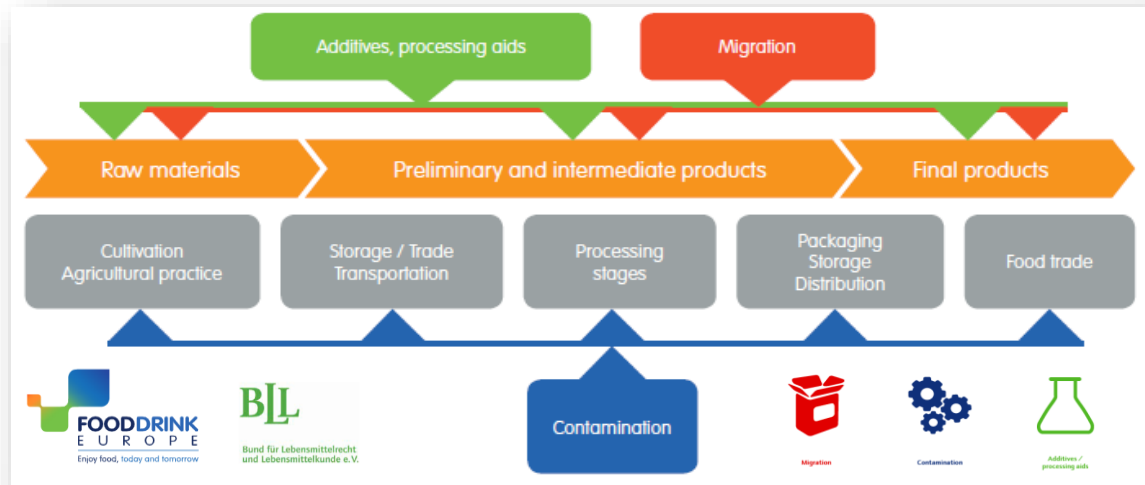
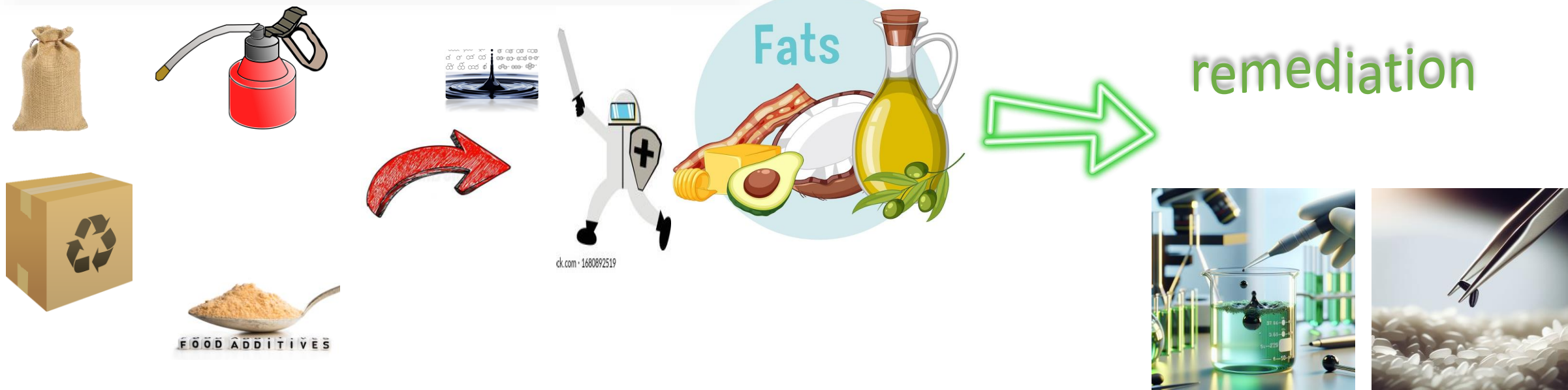
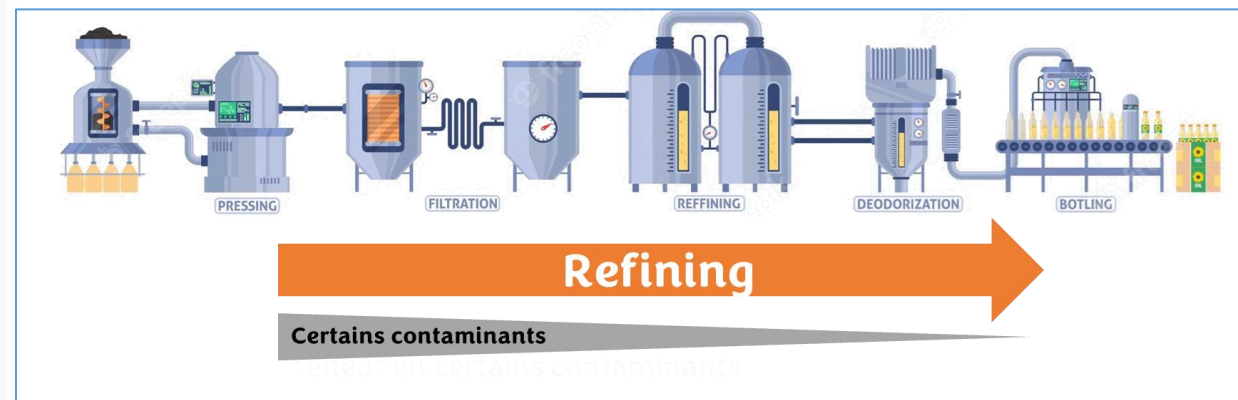
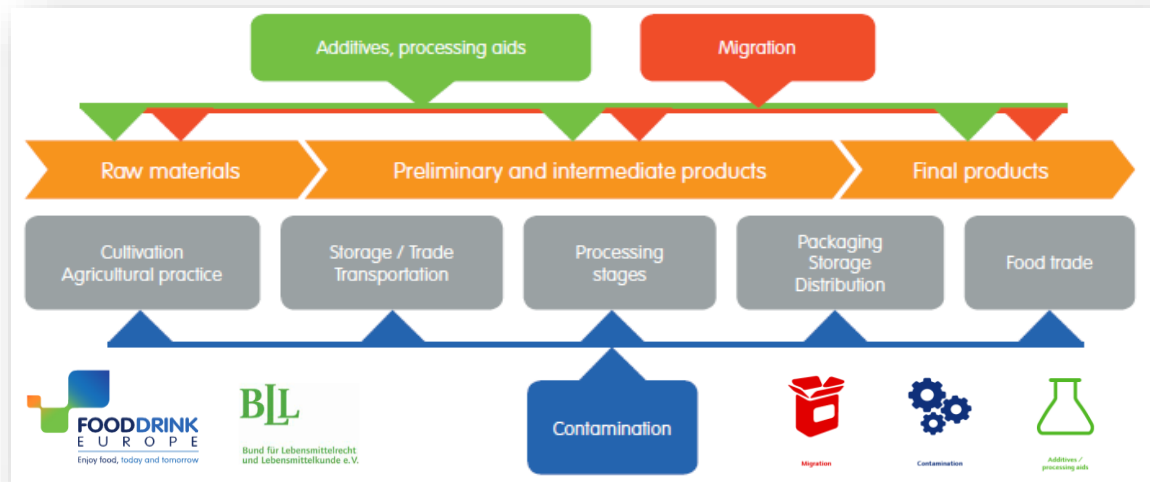


Figure 7. MOAH in the crude mineral oil fraction by GC×GC

MINERAL OIL HYDROCARBONS (MOH)



MINERAL OIL HYDROCARBONS (MOH)



Steps of edible oil refining (simplified)



Refining:

- Industrial process applied to edible oils to **remove undesirable compounds** that are either **naturally present** (e.g., free fatty acids, waxes, phospholipids) or **introduced during cultivation or processing** (e.g., pesticides, metals, MOH, PAH).
- The objective is to obtain a standardized product with low odour, neutral taste, and clear appearance, which is stable over time, and **safe for consumption**.



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- The objective is to obtain a standardized product with low odour, neutral taste, and clear appearance, which is stable over time, and **safe for consumption**.

Effect of refining on MOH content

- Limited literature, mostly published between 2017 and 2025



1989
2001

Original Papers

Mitt. Lebensm. Hyg. 92, 499–514 (2001)

Mineral Paraffins in Vegetable Oils and Refinery By-Products for Animal Feeds

Christoph Wagner, Hans-Peter Neukom and Koni Grob, Official Food Control Authority of the Canton of Zurich (Kantonales Labor), Zurich, Switzerland

Sabrina Moret, Tiziana Populin and Lanfranco S. Conte, Department of Food Science, University of Udine, Udine, Italy

Received 2 August 2001, accepted 23 August 2001

Acids for animal feed

The by-products of edible oil refining, consisting of the free fatty acids and the condensate from the deodorization process, contained between 120 and 6800 mg/kg of mineral paraffins. This by far exceeds the Swiss limit of 30 mg/kg for fats used in animal feeds (22).

Edible oils

Probably more than half of all raw vegetable oils and fats produced worldwide contain more than 10 mg/kg mineral paraffins and still a substantial proportion exceeds 50 mg/kg. Deodorization removes the hydrocarbons up to C25–C30, depending on the conditions. Usually this corresponds to about two thirds of the contamination (depending on the composition of the paraffins and the conditions of deodorization). However, many refined oils on the market still contained 20–80 mg/kg mineral paraffins. Some exceptional samples reached concentrations up to 3000 mg/kg.

Total MOH
(no MOSH+MOAH distinction)



Summary of literature (2017-2025)

| Authors | Key Process | Main Findings |
|-------------------------------|--|---|
| Stauff et al. 2020 | Deodorization (140-240°C, 18-27 mbar, 2% water) | <ul style="list-style-type: none"> Min. temperature for noteworthy MOH reduction: 210 °C 10-75% of removal of MOH \leq C24 |
| Gelmez et al. 2017 | Molecular distillation | <ul style="list-style-type: none"> \sim85% MOH \leq C40 removed |
| Zhang et al. 2022 | (220°C, 10^{-3} mbar, 1 kg/h feedstock) | |
| Bauwens et al. 2023 | Bleaching + Deodorization (230°C, 3h, 1 mbar) | <ul style="list-style-type: none"> 66% C16-C25 MOH removed No modifications in C25-C50 MOH >98.9% spiked alkylated PAHs \geq 3 rings removed |
| Gorska et al. 2024 | Deodorization (150–240°C, 3 mbar, 1% water) | <ul style="list-style-type: none"> At 200 °C, < LOQ MOAH < C24 (incl. weakly alkylated triaromatics) At 230 °C, >60% reduction of C24–C35 MOAH (incl. low-alkylated pentaromatics) |
| Ursol et al. 2025 | Bleaching + Deodorization (180-227 °C, 2.5-5h, 0.8-1 mbar) | <ul style="list-style-type: none"> \sim10–30% total MOH reduction, \sim90% C10-25 MOH reduction, \sim40% C10–35 MOH reduction Deodorization was the only effective MOH-reducing step Bleaching reduced weakly alkylated/non-alkylated 2–4 ring PAHs |

RESEARCH ARTICLE

European Journal of
Lipid Science and Technology
www.ejlst.com

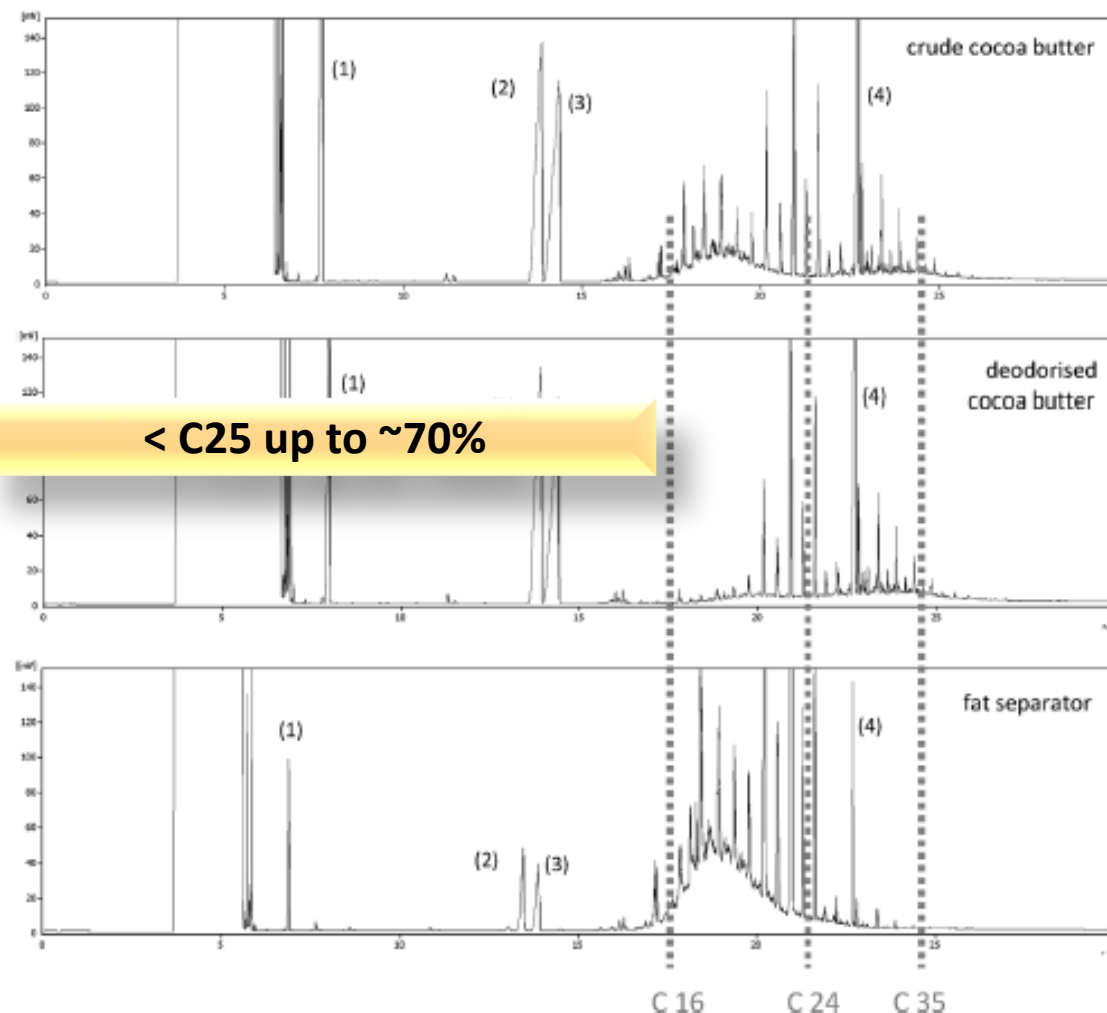
Mineral Oil Hydrocarbons (MOSH/MOAH) in Edible Oils and Possible Minimization by Deodorization Through the Example of Cocoa Butter

Eur. J. Lipid Sci. Technol. 2020, 122, 1900383

Anna Stauff,* Julia Schnapka, Frank Heckel, and Reinhard Matissek

Table 2. Reduction of MOSH/MOAH subfractions $\leq C_{24}$ by industrial deodorization of cocoa butter (LOD: 1.5 mg kg^{-1} , LOQ: 2.5 mg kg^{-1} ; $n = 9$).

| | MOSH | | | MOAH | | |
|-------|--|------------------------|--|--|------------------------|--|
| | MOSH $\leq C_{24} [\text{mg kg}^{-1}]$ | | | MOAH $\leq C_{24} [\text{mg kg}^{-1}]$ | | |
| | Crude | Deodorized (reduction) | | Crude | Deodorized (reduction) | |
| Set 1 | 12 | 5.8 (53%) | | <LOQ | <LOD (40%) | |
| Set 2 | 4.9 | 2.5 (49%) | | 3.4 | <LOD (56%) | |
| Set 3 | 8.9 | 5.0 (44%) | | <LOD | <LOD — | |
| Set 4 | 27 | 12 (54%) | | 8.8 | 3.3 (62%) | |
| Set 5 | 21 | 6.3 (70%) | | 16 | 4.0 (75%) | |
| Set 6 | 17 | 13 (25%) | | 9.0 | 7.1 (21%) | |
| Set 7 | 5.8 | 5.2 (10%) | | 2.8 | <LOQ (21%) | |
| Set 8 | 5.3 | 4.2 (21%) | | <LOD | <LOD — | |
| Set 9 | 4.3 | 3.3 (23%) | | <LOQ | <LOD — | |



Min 210°C for removal

MOH – Remediation Molecular distillation

Eur. J. Lipid Sci. Technol. 2017, 119, 1600001

2017

Research Article

Removal of di-2-ethylhexyl phthalate (DEHP) and mineral oil from crude hazelnut skin oil using molecular distillation–multiobjective optimization for DEHP and tocopherol

Beyza Gelmez¹, Onur Ketenoglu², Huseyin Yavuz³ and Aziz Tekin²

Only MOSH

Table 1. Chain length distribution of MOSH in the samples

| | | MOSH fractions, mg/kg | | | |
|-------------------------|--------------------|-----------------------|-----------------|---------|-------|
| | | C(0)– C(15) | C(16)– C(24) | > C(25) | Total |
| Crude hazelnut skin oil | | 4.1 | 13.95 | 45.95 | 64 |
| Temp. (°C) | Pressure (mbar) | | | | |
| 200 | 1 | 0.86 | 3.09 | 7.24 | 11.19 |
| | 2 | 1.53 | 3.83 | 3.49 | 8.85 |
| | 3 | 0.66 | 3.11 | 3.16 | 6.93 |
| 210 | 1 | 0.72 | 4.15 | 7.41 | 12.28 |
| | 2 | 1.35 | 5.36 | 6.67 | 13.38 |
| | 3 | 1.16 | 3.93 | 3.18 | 8.27 |
| 220 | 1 | 0.34 | 1.45 | 3.79 | 5.58 |
| | 2 | 0.97 | 5.83 | 5.81 | 12.61 |
| | 3 | 0.41 | 3.65 | 2.24 | 6.30 |
| 230 | 1 | 0.20 | 4.28 | 4.95 | 9.43 |
| | 2 | 0.72 | 4.80 | 5.60 | 11.12 |
| | 3 | 1.10 | 2.96 | 3.61 | 7.67 |

ORIGINAL ARTICLE

2022

AACS WILEY

Mineral saturated hydrocarbons and mineral aromatic hydrocarbons in tropical plant oils and their removal by molecular distillation

Mingming Zhang¹ | Hai Zhang | Tosin Michael Olajide | Wenming Cao | Yan Wang² | Hong Zhang | Yuanrong Jiang

MOSH+MOAH

TABLE 3 Reduction of MOSH/MOAH sub-fractions (C10–C50) by molecular distillation

| Palm olein | | MOSH sub-fractions (mg kg ⁻¹) | | | | | | | MOAH sub-fractions (mg kg ⁻¹) | | | | |
|------------|---------------------------------|---|---------|---------|-----------------------|----------|---------|---------------|---|---------|----------|---------|---------------|
| Temp. (°C) | Feed rate (kg h ⁻¹) | C10–C16 | C16–C20 | C20–C25 | C25–C35 | C35–C40 | C40–C50 | Sum (C10–C50) | C10–C16 | C16–C25 | C25–C35 | C35–C50 | Sum (C10–C50) |
| - | - | <LOQ | <LOQ | 4.5 | 36.8 | 8.1 | 8.7 | 58.1 | <LOQ | 3.0 | 6.1 | 1.2 | 10.3 |
| 200 | 1 | <LOQ | <LOQ | <LOQ | 6.7 (82) ^a | 4.8 (40) | 8.5 (2) | 20.0 (65) | <LOQ | <LOQ | 1.9 (69) | 1.2 (–) | 3.1 (70) |
| | 2 | <LOQ | <LOQ | <LOQ | 17.1 (54) | 6.1 (24) | 8.5 (2) | 31.7 (45) | <LOQ | <LOQ | 3.0 (51) | 1.2 (–) | 4.2 (59) |
| | 3 | <LOQ | <LOQ | <LOQ | 21.1 (43) | 6.9 (14) | 8.6 (1) | 36.6 (37) | <LOQ | <LOQ | 4.1 (33) | 1.2 (–) | 5.3 (49) |
| 210 | 1 | <LOQ | <LOQ | <LOQ | 1.7 (95) | 3.1 (61) | 8.4 (3) | 13.2 (77) | <LOQ | <LOQ | <LOQ | 1.2 (–) | 1.2 (88) |
| | 2 | <LOQ | <LOQ | <LOQ | 9.8 (73) | 5.1 (36) | 8.5 (2) | 23.4 (60) | <LOQ | <LOQ | 1.8 (71) | 1.2 (–) | 3.0 (71) |
| | 3 | <LOQ | <LOQ | <LOQ | 10.3 (72) | 5.8 (28) | 8.5 (2) | 24.6 (58) | <LOQ | <LOQ | 2.2 (64) | 1.3 | 3.5 (67) |
| 220 | 1 | <LOQ | <LOQ | <LOQ | <LOQ | 1.6 (80) | 8.2 (5) | 9.8 (83) | <LOQ | <LOQ | <LOQ | 1.2 (–) | 1.2 (89) |
| | 2 | <LOQ | <LOQ | <LOQ | 2.1 (94) | 3.4 (58) | 8.4 (3) | 13.9 (76) | <LOQ | <LOQ | <LOQ | 1.2 (–) | 1.2 (89) |
| | 3 | <LOQ | <LOQ | <LOQ | 7.0 (81) | 4.0 (50) | 8.5 (2) | 19.5 (66) | <LOQ | <LOQ | 1.5 (75) | 1.2 (–) | 2.7 (74) |

Abbreviations: LOQ, limit of quantification; MOAH, mineral aromatic hydrocarbons; MOSH, mineral saturated hydrocarbons.

^aRemoval rate (%) is given in brackets.

< C40 up to ~80%

MOH – Remediation

Analytical strategy

FOOD ADDITIVES & CONTAMINANTS: PART A
<https://doi.org/10.1080/19440049.2022.2164621>

2022



Check for updates

Investigation of the effect of refining on the presence of targeted mineral oil aromatic hydrocarbons in coconut oil

Grégory Bauwens^a, Alexandre Cavaco Soares^b, Florence Lacoste^b, Daniel Ribera^c, Coen Blomsma^d, Iekje Berg^e, Fernando Campos^f, Alwin Coenradie^g, Adina Creanga^h, Ralph Zwagermanⁱ and Giorgia Purcaro^a

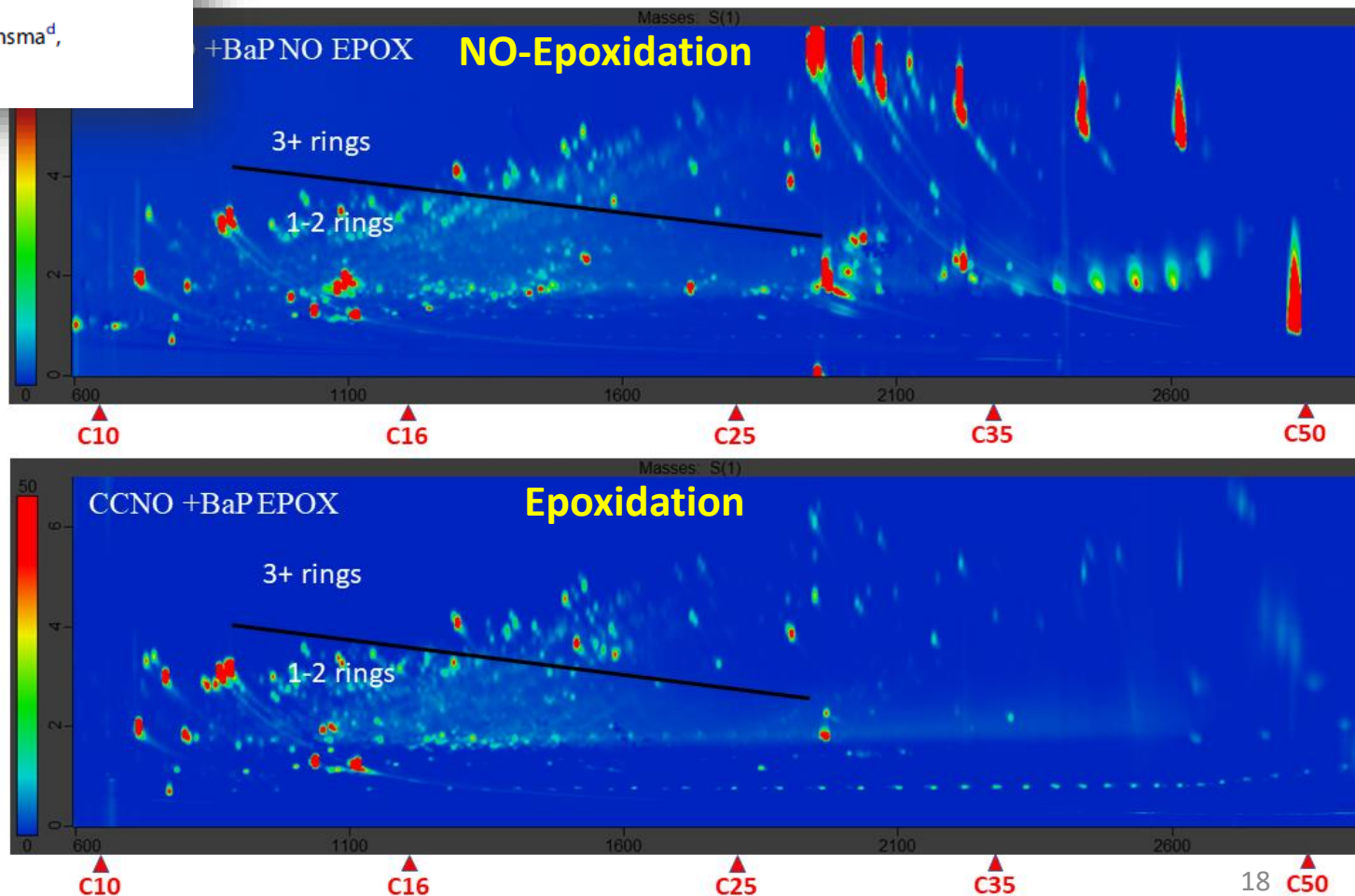
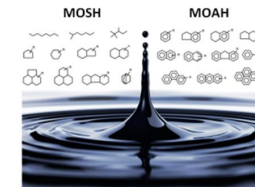
Deodorization + bleaching

(230°C, 3h, 1 mbar + 1.7% mass of bleaching earth (Pureflow B80) 95°C x 30 min)

Epoxidation cannot be applied for this kind of studies!

VERNOF

Crude coconut oil



MOH – Remediation Analytical strategy

FOOD ADDITIVES & CONTAMINANTS: PART A
<https://doi.org/10.1080/19440049.2022.2164621>

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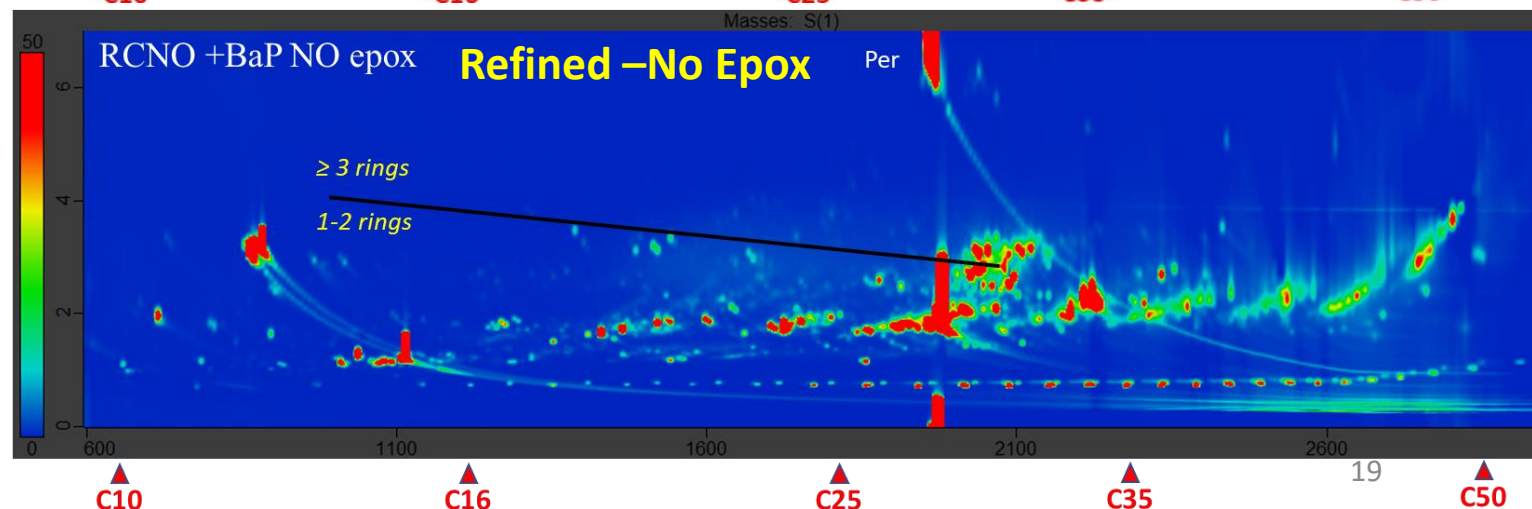
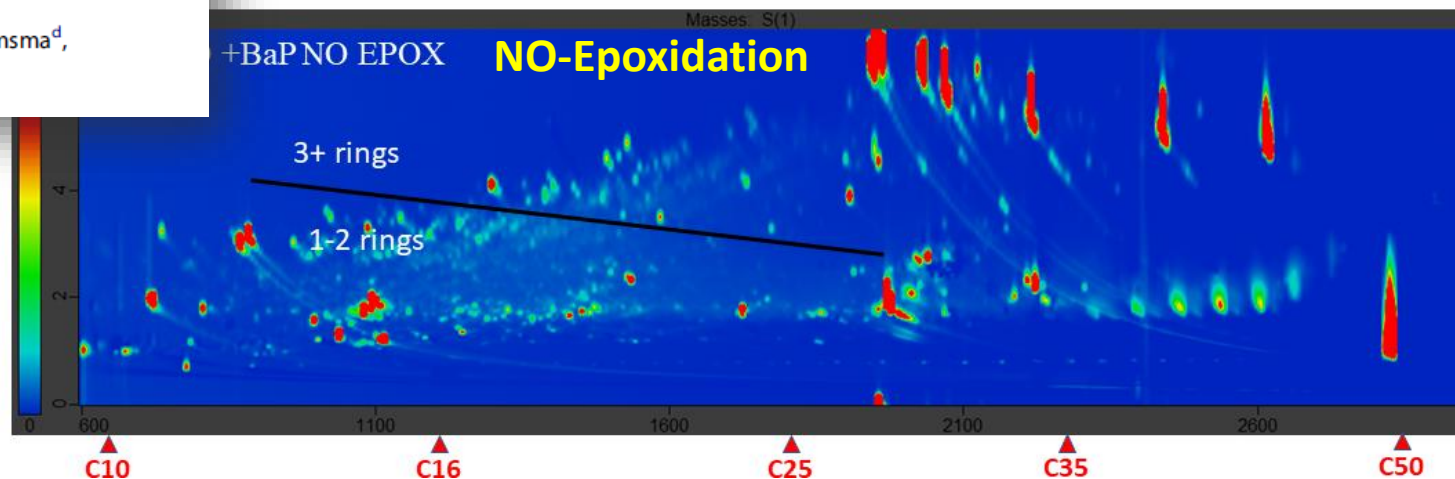
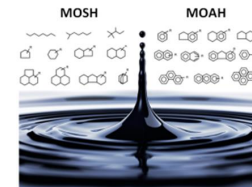
Deodorization + bleaching

Epoxidation cannot be applied for this kind of studies!

- 66% C16-C25 MOH removed
- No modifications in C25-C50 MOH (but not clear contamination there)
- **>98.9% spiked alkylated PAHs ≥ 3 rings removed**

VERNOF

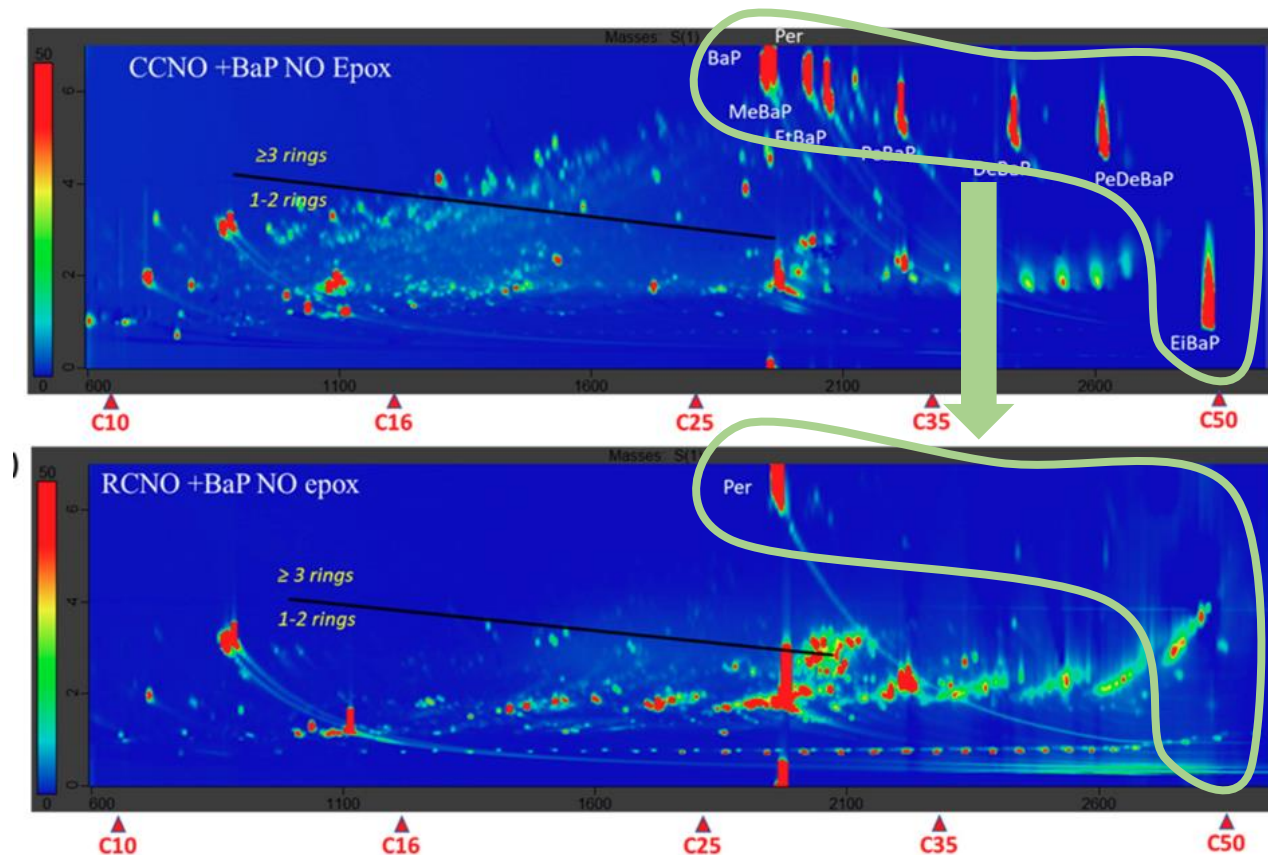
Crude coconut oil





Removal of alkylated PAHs by bleaching

Bauwens et al. (2023) – bleaching + deodorization



Which is the contribution of **deodorization** and **bleaching** separately?

>98.9% removal of the spiked alkylated PAHs

MOH – Remediation Deodorization

FOOD ADDITIVES & CONTAMINANTS: PART A
<https://doi.org/10.1080/19440049.2024.2371925>

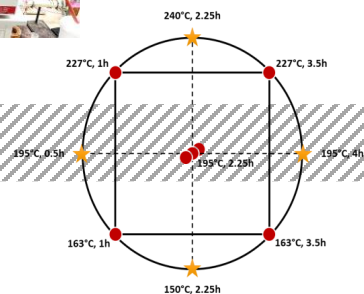


Check for updates

Impact of deodorisation time and temperature on the removal of different MOAH structures: a lab-scale study on spiked coconut oil

Aleksandra Gorska^a, Sabine Danthine^b, Nicolas Jacquet^b and Giorgia Purcaro^a

^aAnalytical Chemistry, Gembloux Agro-Bio Tech, University of Liège, Gembloux, Belgium; ^bFood Technology, Gembloux Agro-Bio Tech, University of Liège, Gembloux, Belgium

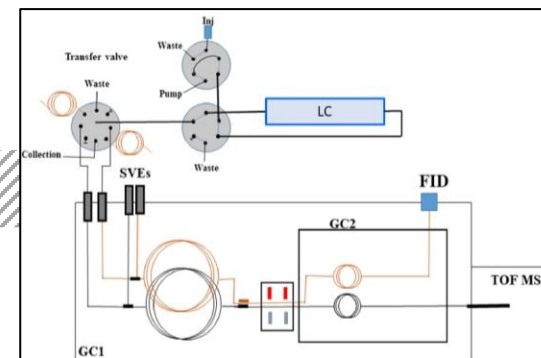


Deodorization at lab-scale
(11 experiments - CCD)

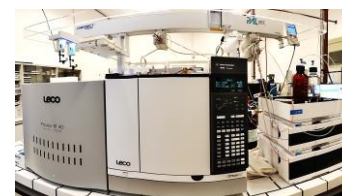
Variables:
time (0.5-4h), T° (150-240°C)

Constants:
pressure (3mbar), steam flow (1%_w/h)

**Coconut oil spiked with
MOAH**
(9 mg/kg)

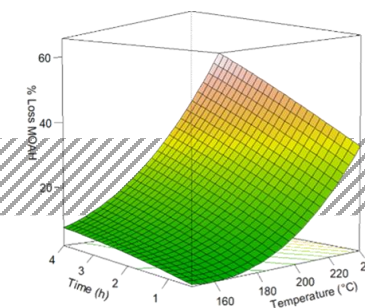


MOAH analysis before/after
deodorization (by LC-GC×GC-FID)



DEODORIZATION

The most delicate step of this project
was the MOAH analysis → required an
adapted **strategy**

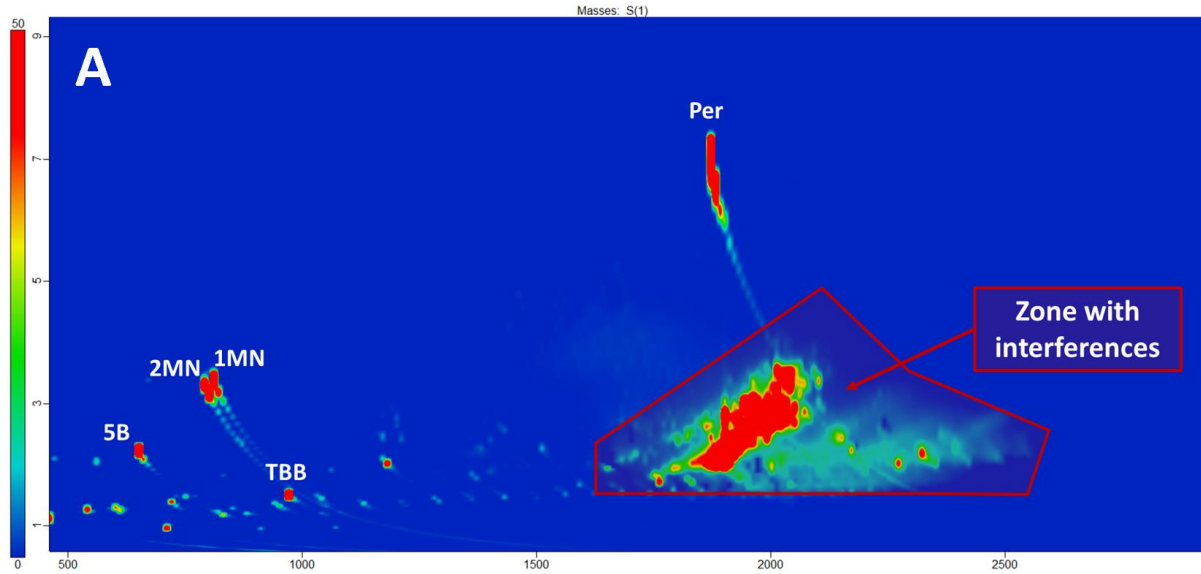


Response surfaces
MOAH loss according to
time and temperature

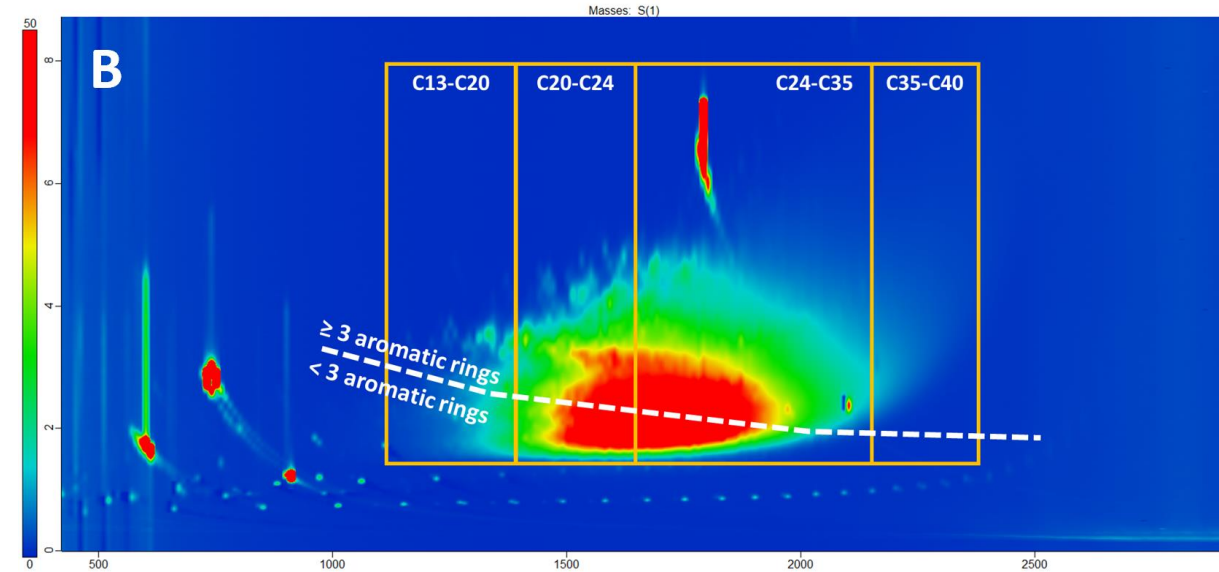
MOH – Remediation

Deodorization time-temperature

Coconut oil

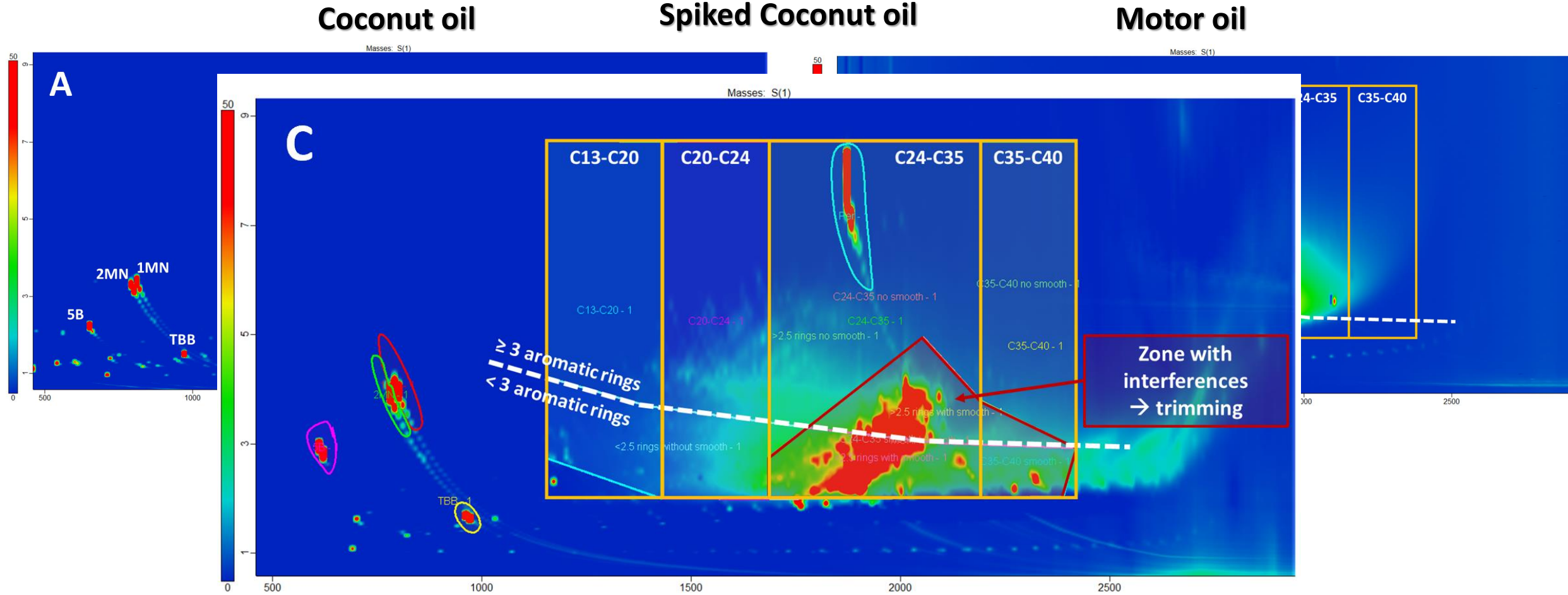


Motor oil



MOH – Remediation

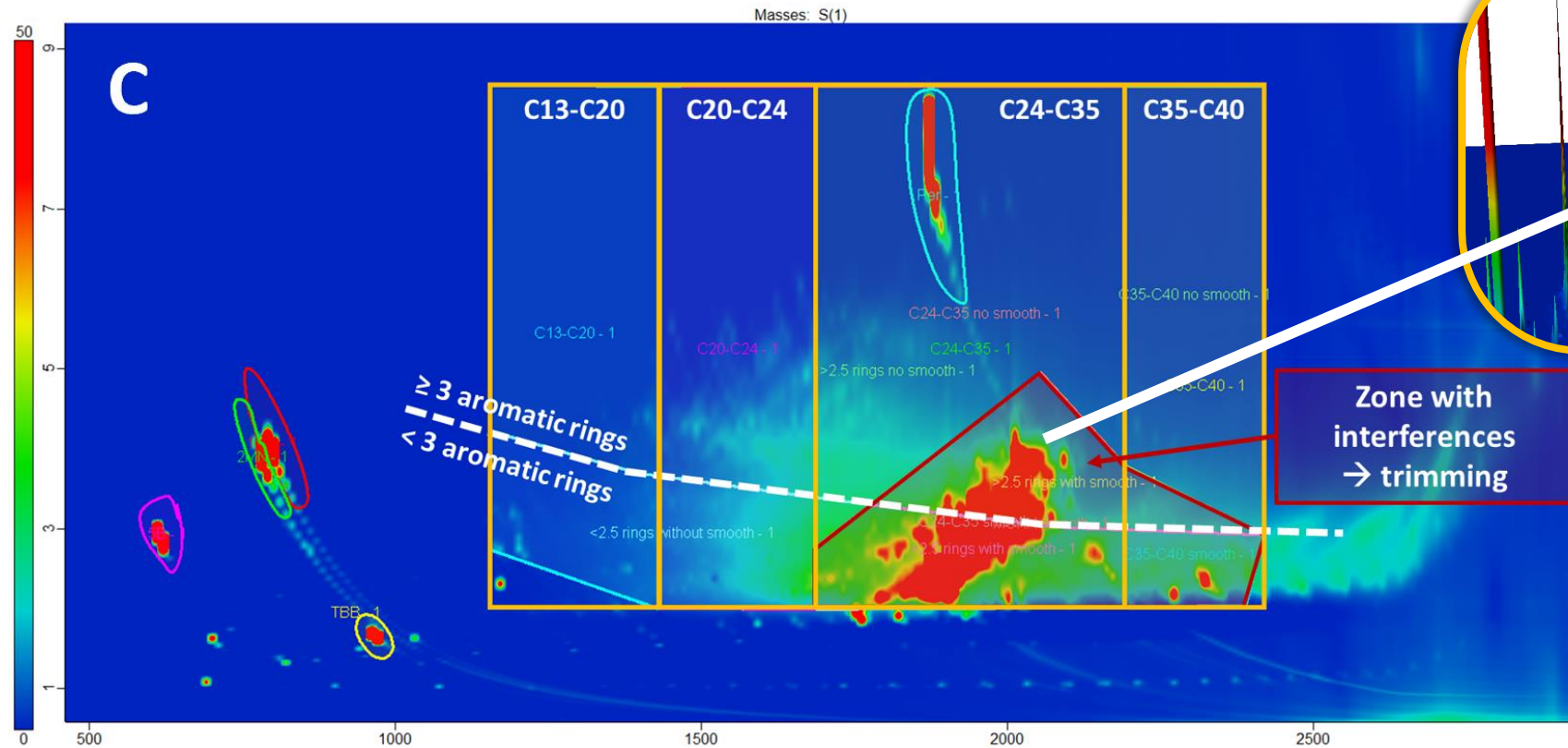
Deodorization time-temperature



MOH – Remediation

Deodorization time-temperature

Spiked Coconut oil



Interferences hinder accurate determination of MOAH → **purification method needed!**

MOH – Remediation Analytical strategy

Journal of Chromatography A 1743 (2025) 465684

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journal homepage: www.elsevier.com/locate/chroma

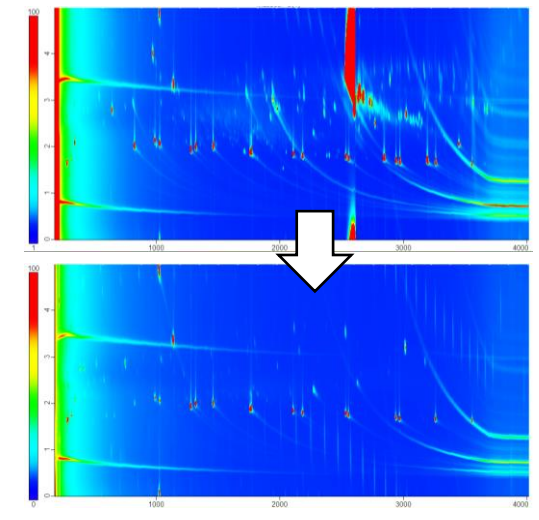
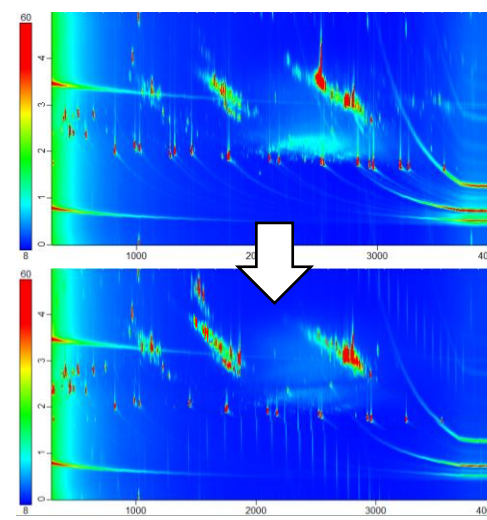
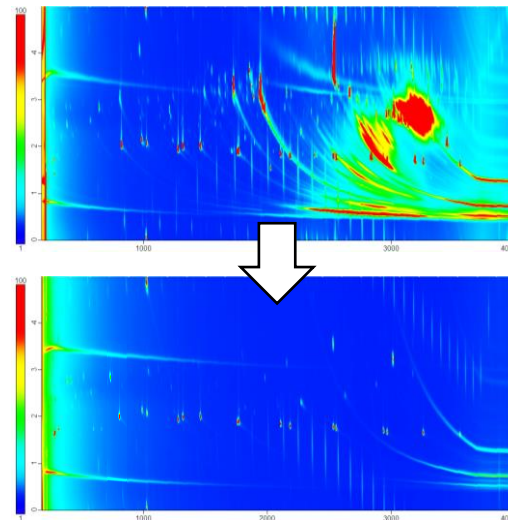


Purification of mineral oil aromatic hydrocarbons and separation based on the number of aromatic rings using a liquid chromatography silica column. An alternative to epoxidation

Aleksandra Gorska^{a,*}, Grégory Bauwens^a, Marco Beccaria^b, Giorgia Purcaro^{a,*}

Very good removal of carotenoids and squalene

Other terpenoids are less well removed



MOH – Remediation

Analytical strategy

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Purification of mineral oil aromatic hydrocarbons and separation based on the number of aromatic rings using a liquid chromatography silica column. An alternative to epoxidation

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Very good removal of carotenoids and squalene

Other terpenoids are less well removed



MOAH Recovery LC Purification:

94% ± 2%

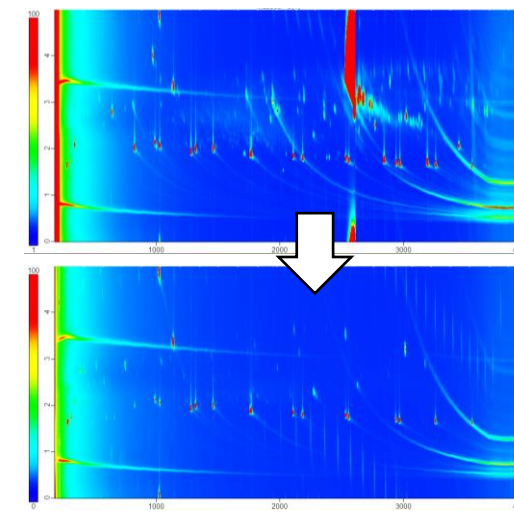
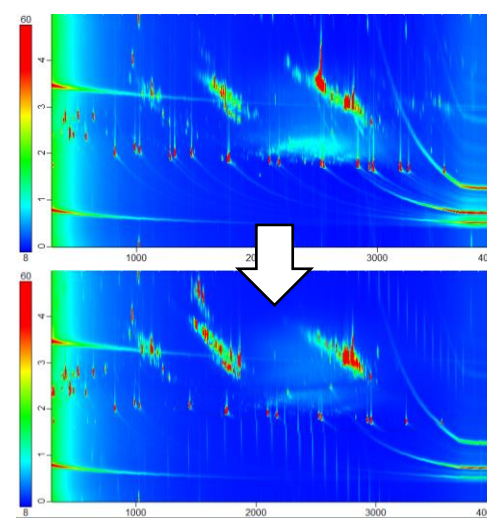
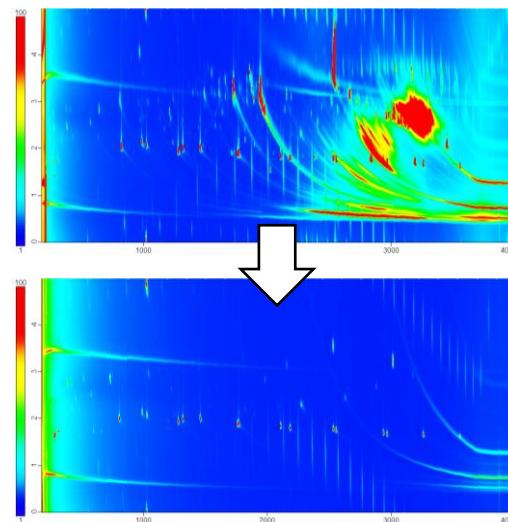
Epoxidation

mCPBA

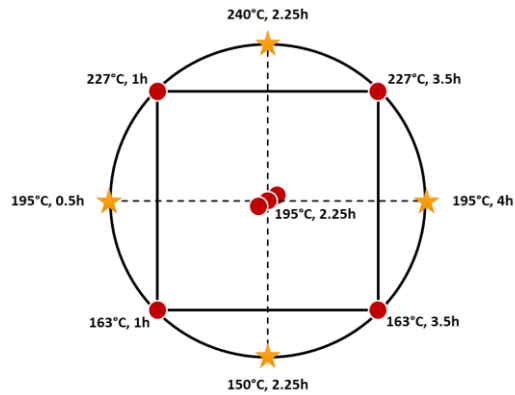
82% ± 10%

performic acid

71 ± 16%



MOH – Remediation Analytical strategy



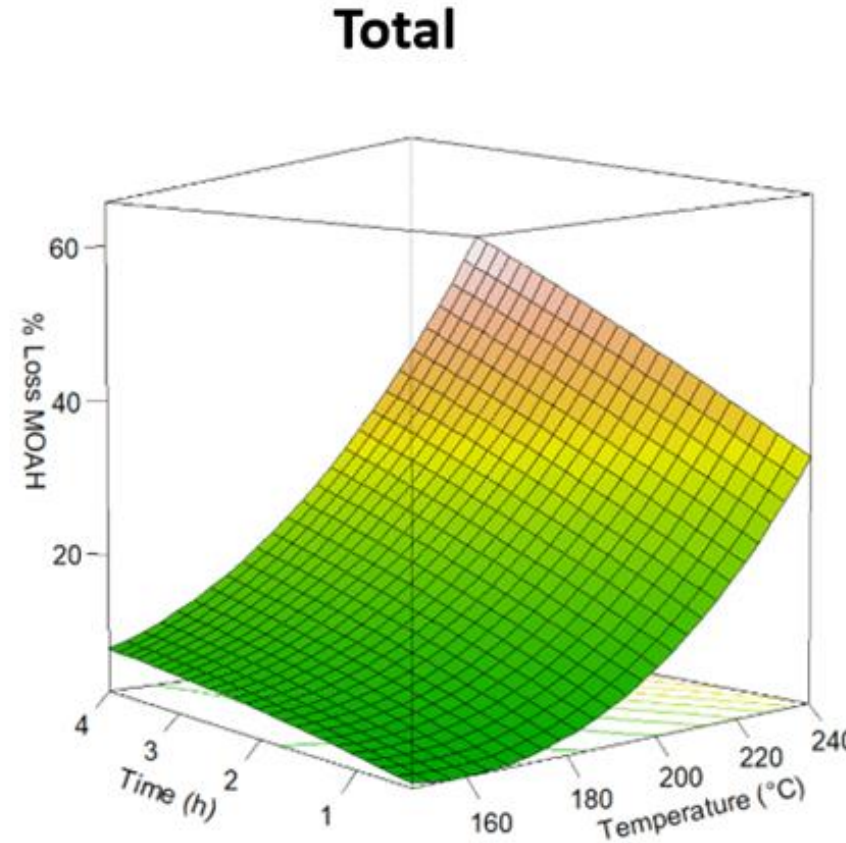
Deodorization at lab-scale

Variables:

time (0.5-4h), T° (150-240°C)

Constants:

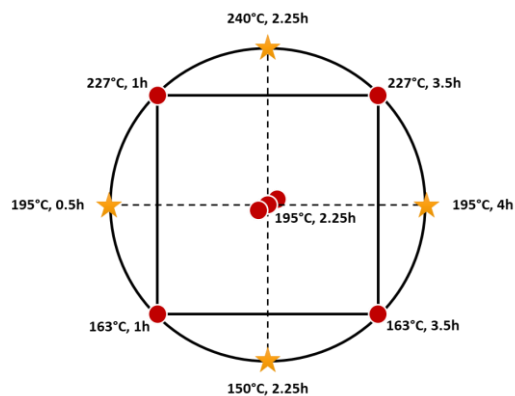
pressure (3mbar),
steam flow (1%_w/h)



**Max removal:
60%**

MOH – Remediation

Analytical strategy

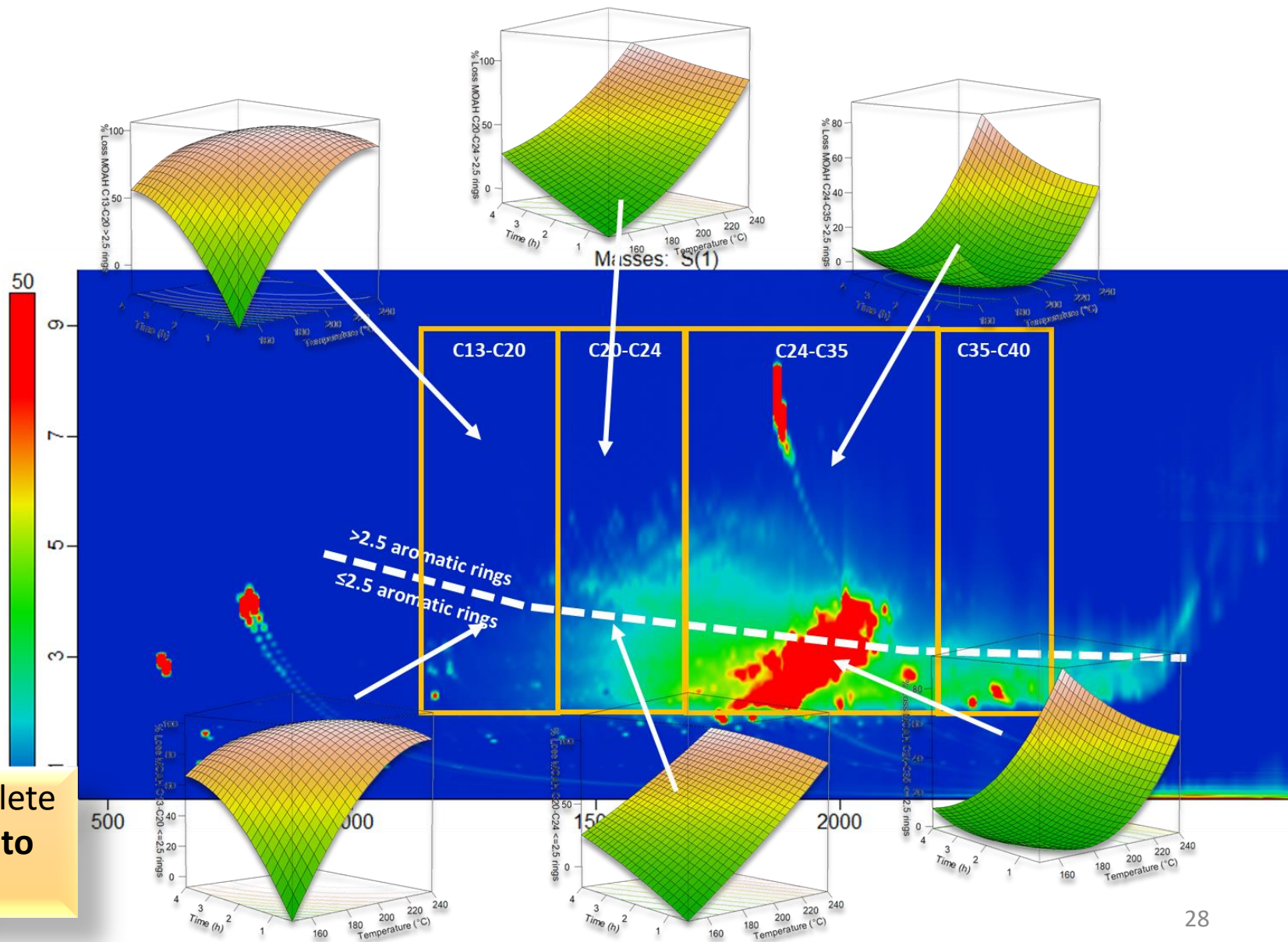


Deodorization at lab-scale

Variables:
time (0.5-4h), T° (150-240°C)

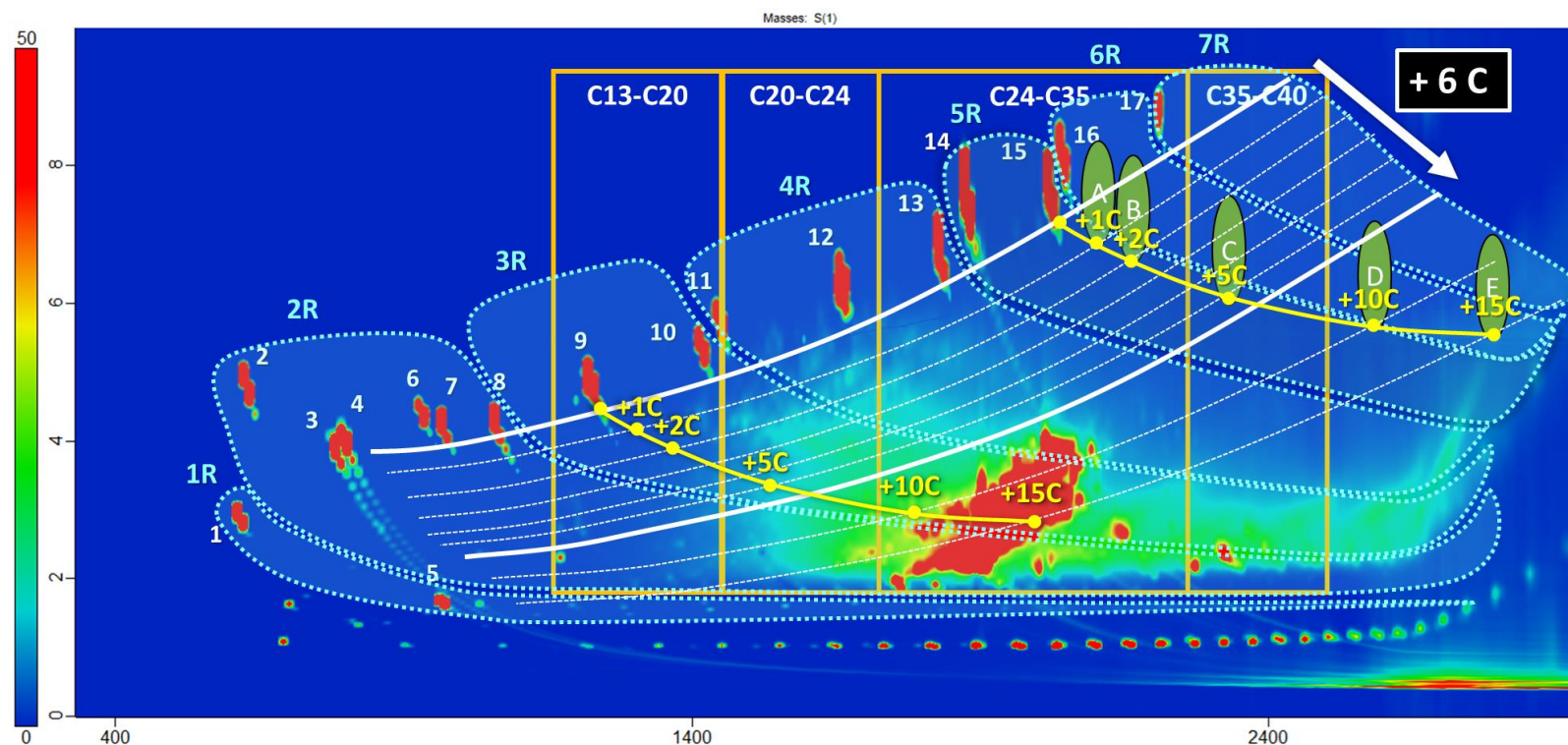
Constants:
pressure (3mbar),
steam flow (1%_w/h)

Almost complete
removal up to
C30!



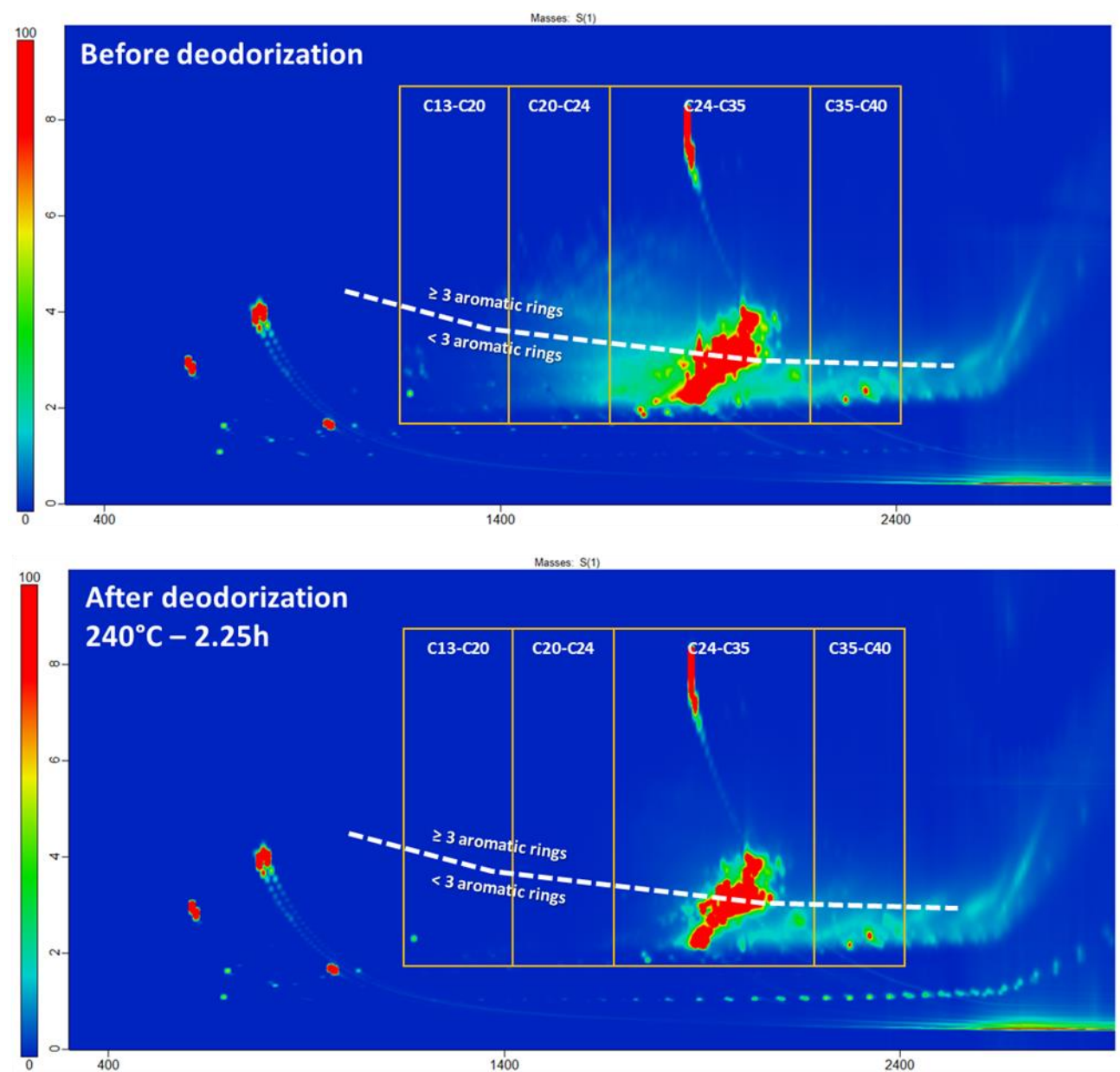
What chemical structures are removed?

Possible to study the **MOAH chemical structures** that are removed thanks to an **LC-GC×GC-FID/TOFMS** analysis



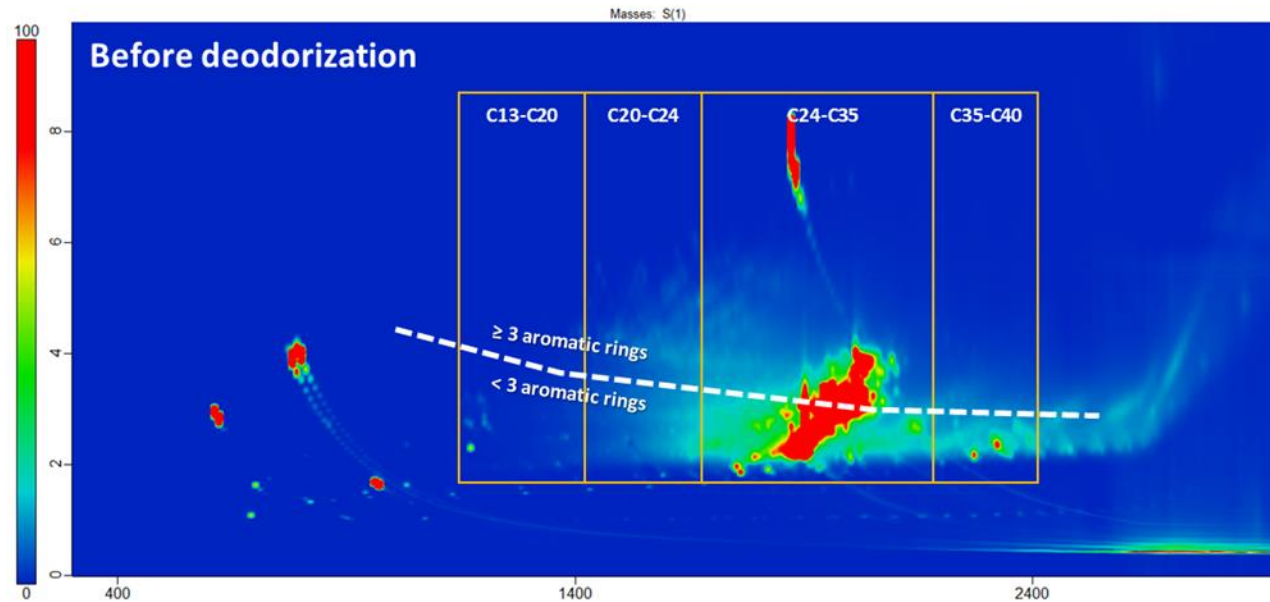
Possible estimation of the
number of aromatic rings and
alkylation degree of present
compounds

MOH – Remediation Analytical strategy



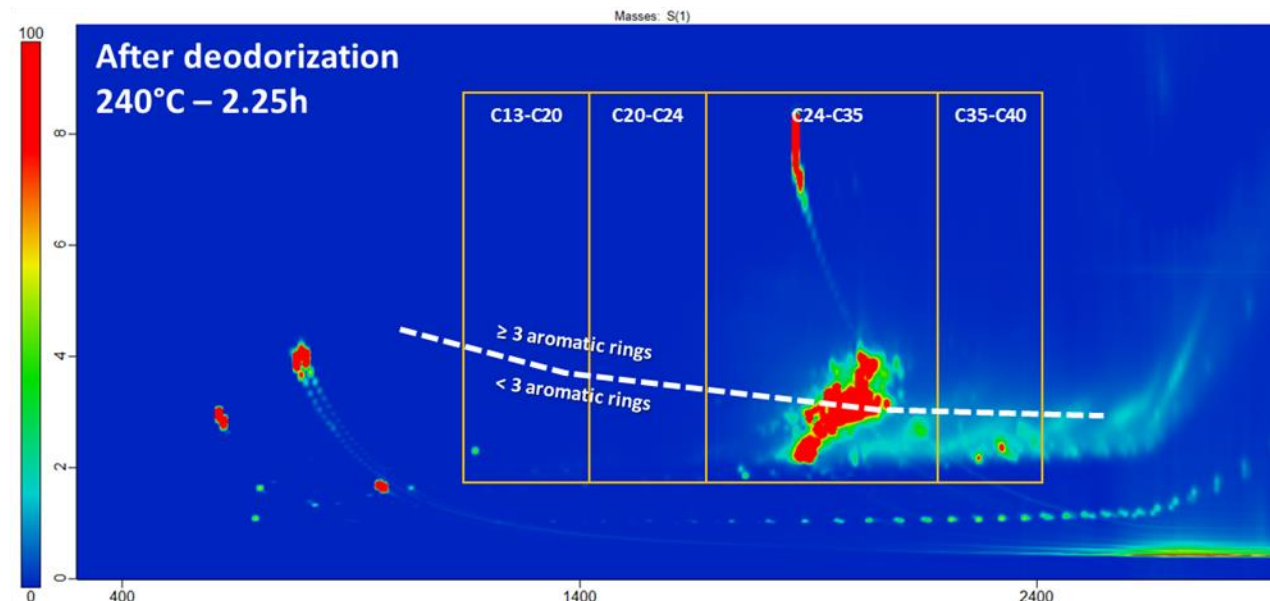
| | | C-fraction | | | |
|---|----|------------|---------|---------|---------|
| | | C13-C20 | C20-C24 | C24-C35 | C35-C40 |
| 1 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 2 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 3 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 4 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 5 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 6 | - | | | | |
| | + | | | | |
| | ++ | | | | |
| 7 | - | | | | |
| | + | | | | |
| | ++ | | | | |

MOH – Remediation Analytical strategy



After deodorization (T° /time variable, 3 mbar, 1% steam)

- C13-C20: reduction <LOQ at $\sim 200^\circ\text{C}$ ✓
- C20-C24: reduction <LOQ $\sim 220^\circ\text{C}$ ✓
- C24-C35: reduction of $\sim 60\%$ at $>230^\circ\text{C}$ ~
- C35-C40: no visible reduction at 240°C ✗

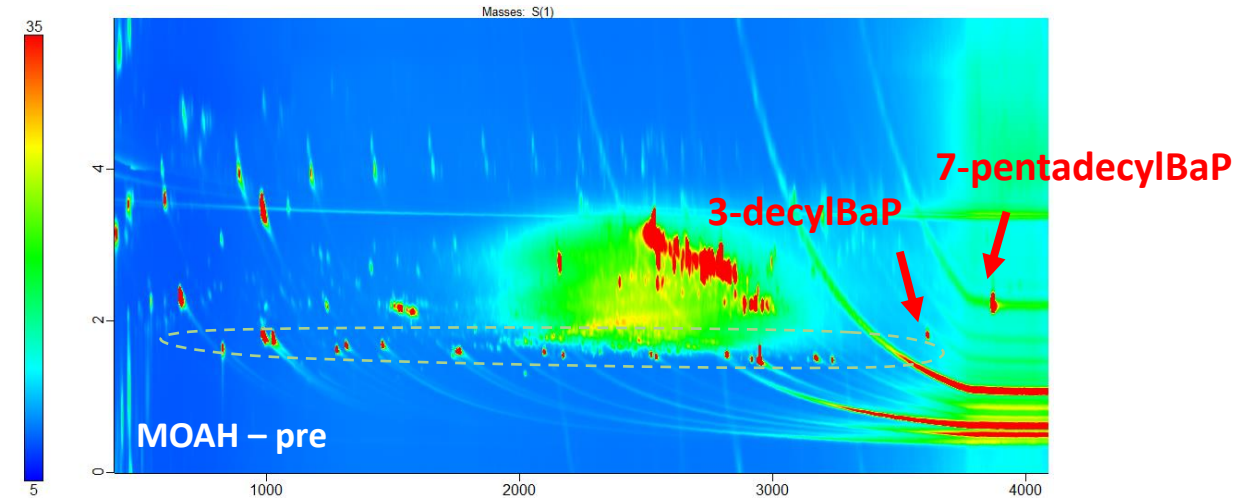
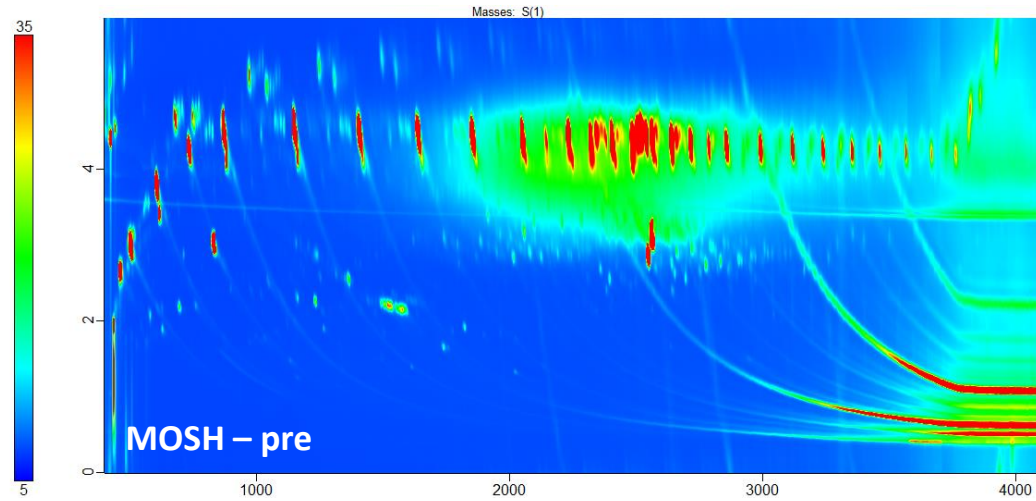


Open question at the end of the study

*Is it possible to reduce **C35-C40 MOH** applying lower (but technically feasible) pressures?*

New deodorization trials with lower pressure

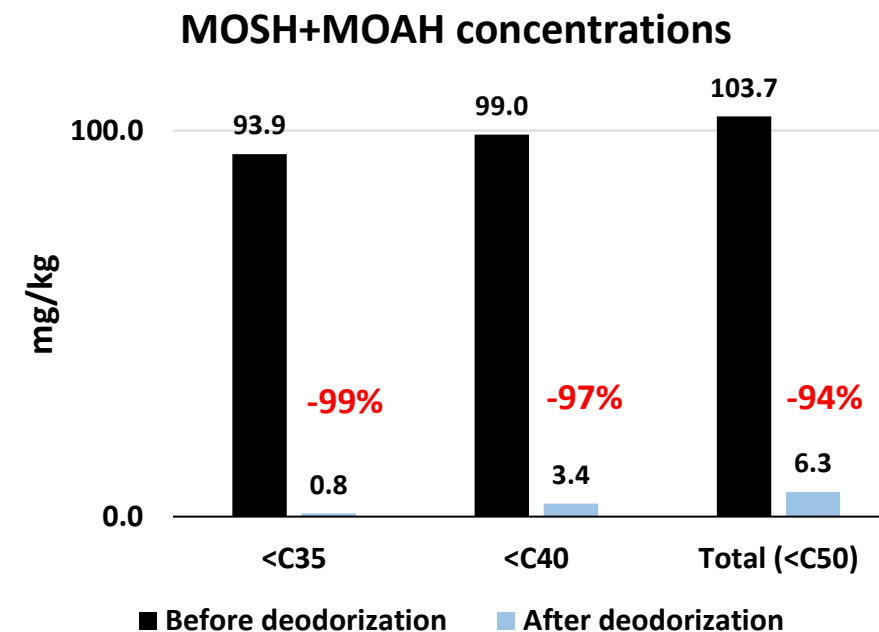
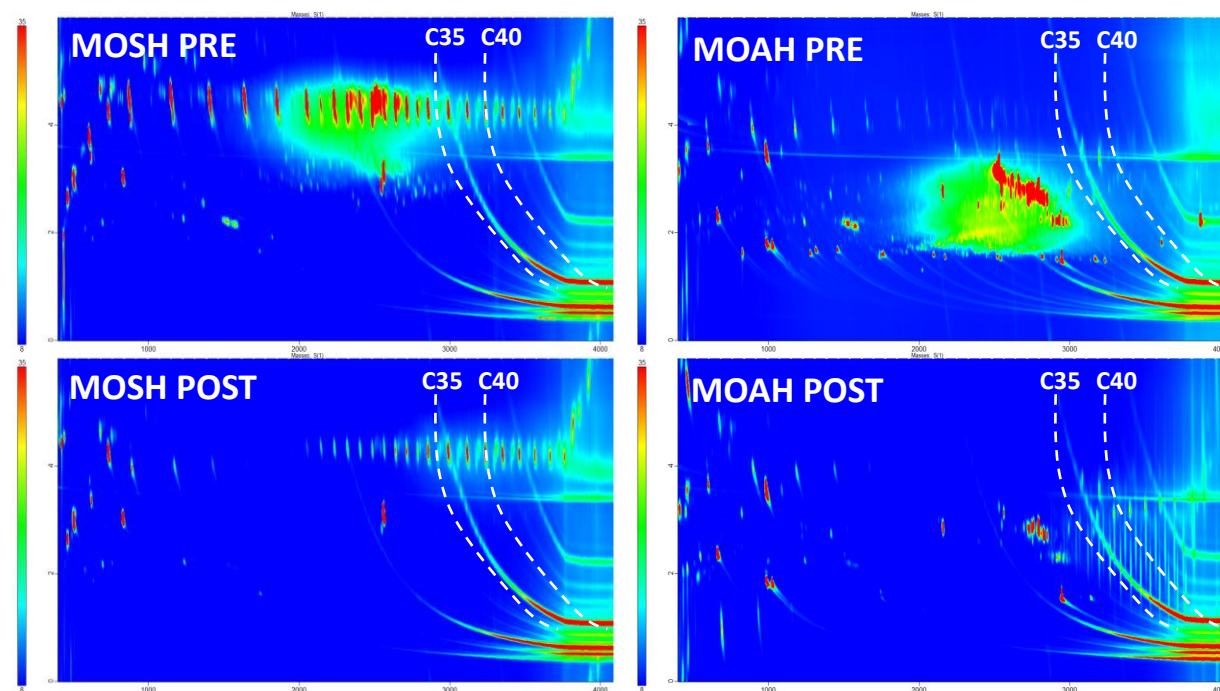
Coconut oil spiked with lubricant oil and PAHs (included two alkylated PAHs)



New deodorization trials with lower pressure

Deodorization conditions: 240°C, 4h, **1 mbar** (instead of 3 mbar as in *Gorska et al.*) , 1%/h steam

Matrix: spiked coconut oil



Reduction of 99% (from 66.9 to <1 mg/kg) of the C25-C35 fraction (compared to ~60% in *Gorska et al.* (2024))

Weak reduction of ~50% of the C35-C40 (5.5 to 2.7 mg/kg MOH) and C40-C50 (4.8 to 2.9 mg/kg MOH)

What about bleaching?

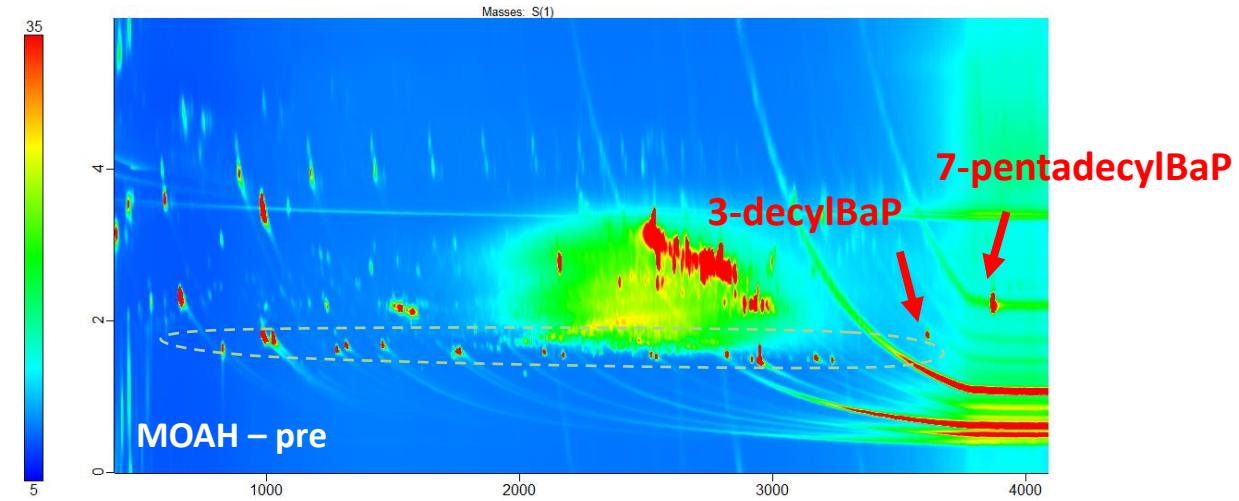
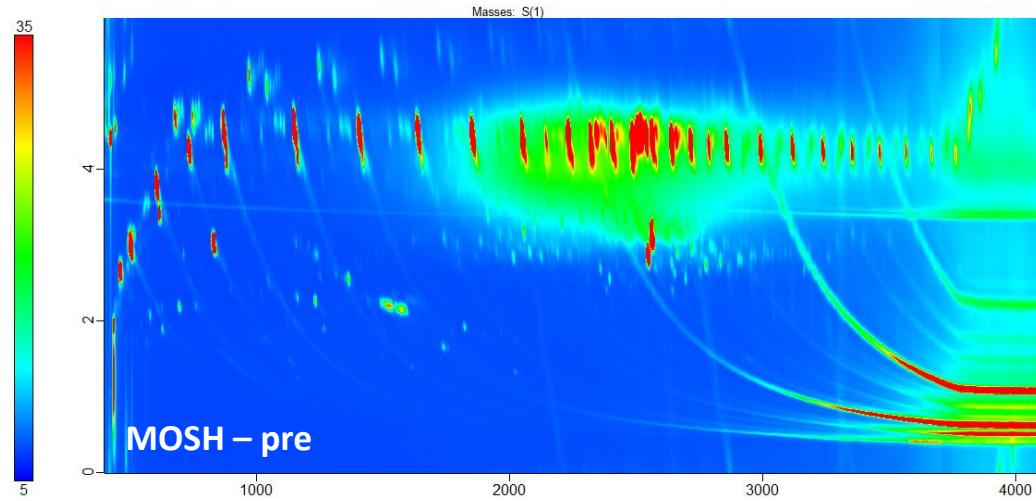
Bleaching: Adsorptive treatment (e.g. bleaching earth) under vacuum and heat to remove pigments (carotenoids, chlorophyll), peroxides, soaps, and residual salts.

Limited available literature for MOH

- **Heavy PAHs** are known to be able to adsorb on **activated carbon**
- Little information on the **effect of alkylation** of the parent PAH
- Significant (> 98.5%) reduction of **alkylated BaP** was observed by Bauwens et al. (2023) after **combined bleaching and deodorization**, but the effect of each individual step was not investigated
- Ursol et al. (2025) observed a **reduction of low alkylated PAHs** in bleached samples, but **epoxidation** was applied



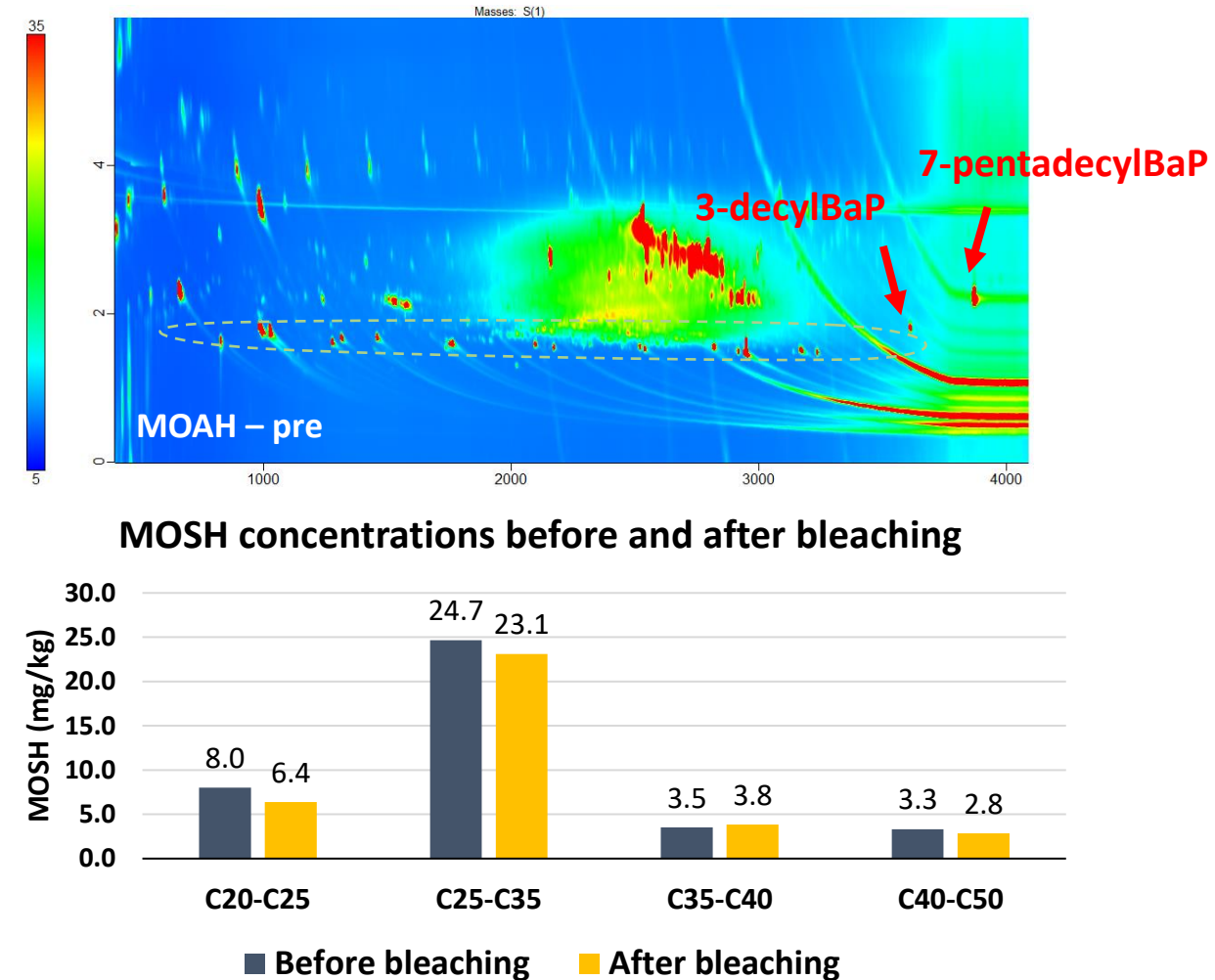
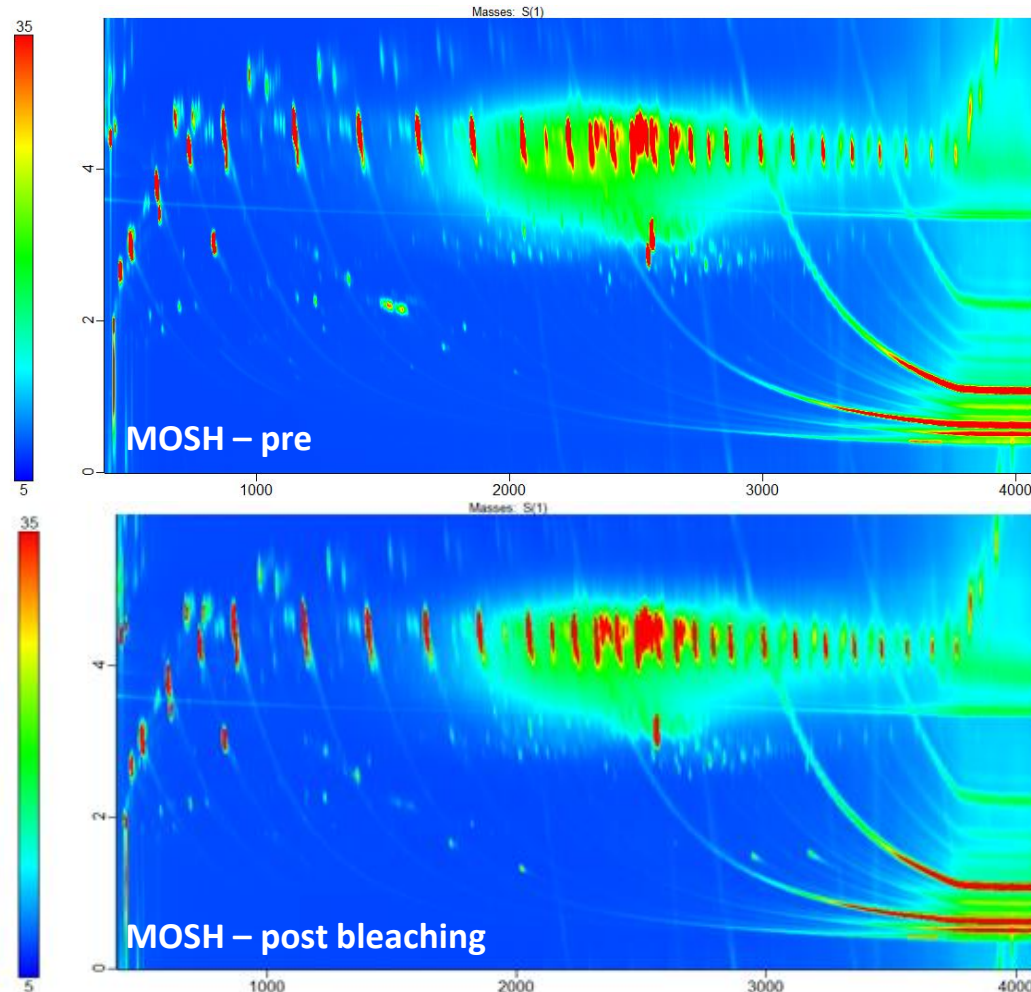
Coconut oil spiked with lubricant oil and PAHs (included two alkylated PAHs)



Bleaching conditions: activated carbon (Jacobi ColorSorb™ XFP21), 240°C, 1 mbar, 4 h

What about bleaching?

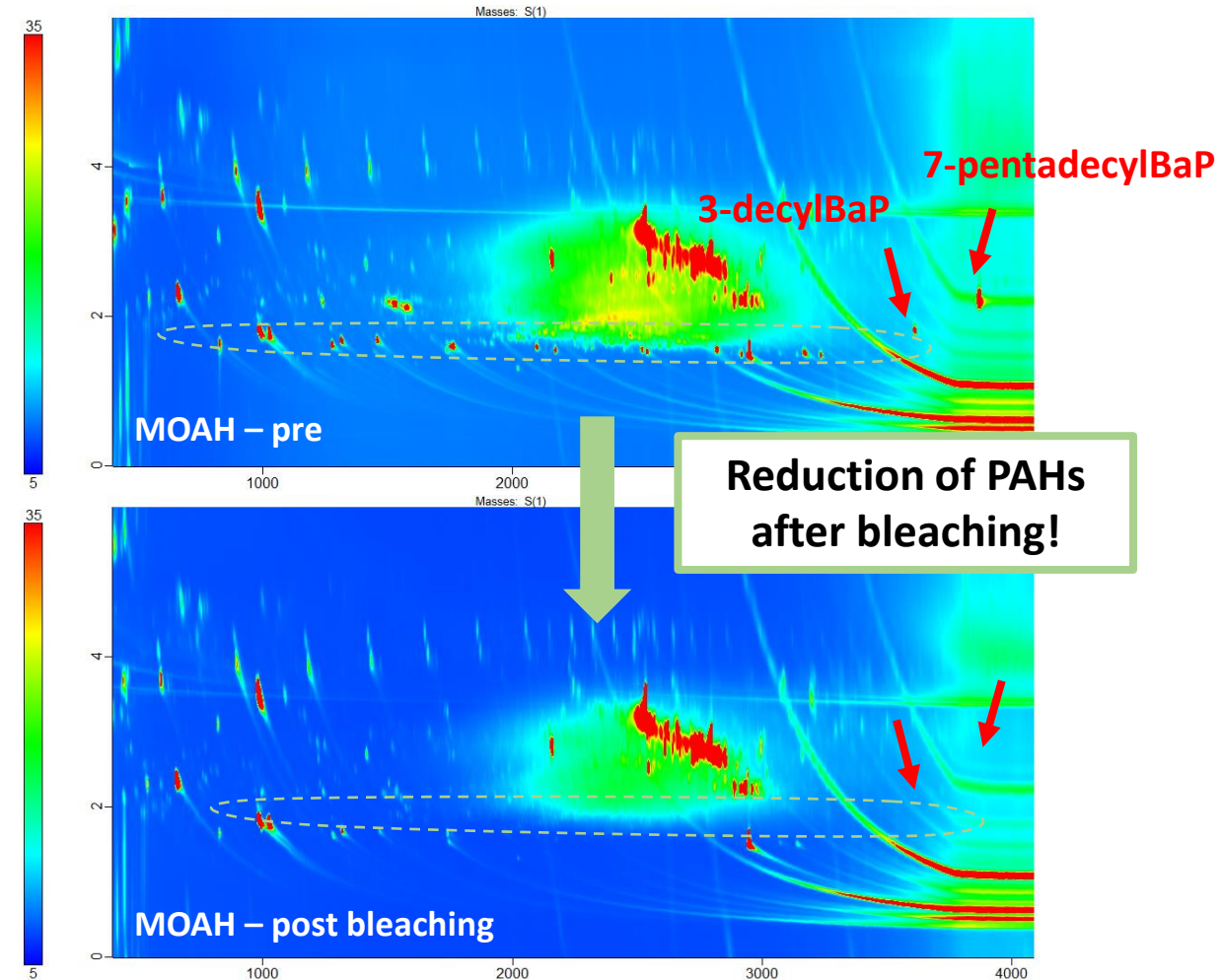
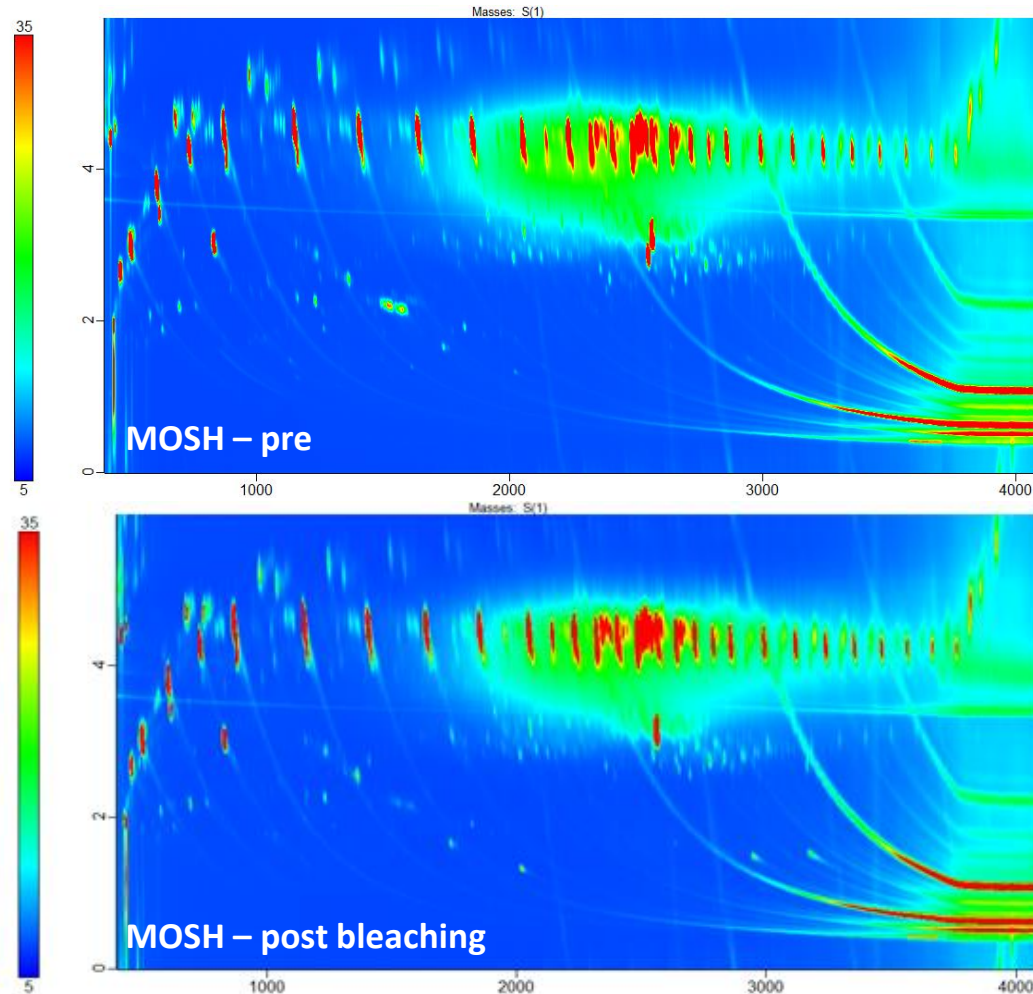
Coconut oil spiked with lubricant oil and PAHs (included two alkylated PAHs)



Bleaching conditions: activated carbon (Jacobi ColorSorb™ XFP21), 240°C, 1 mbar, 4 h

What about bleaching?

Coconut oil spiked with lubricant oil and PAHs (included two alkylated PAHs)

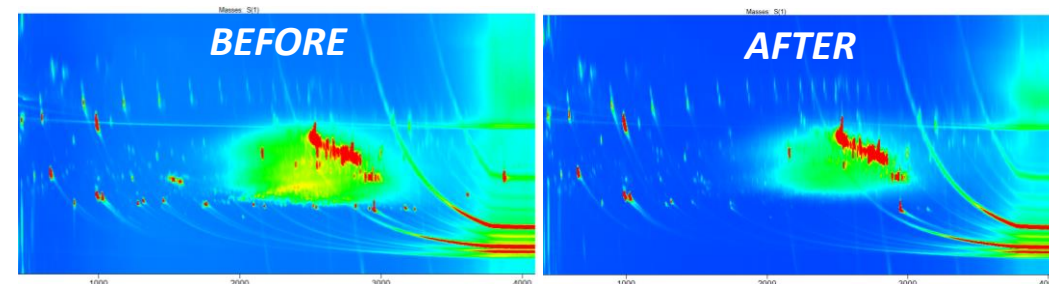
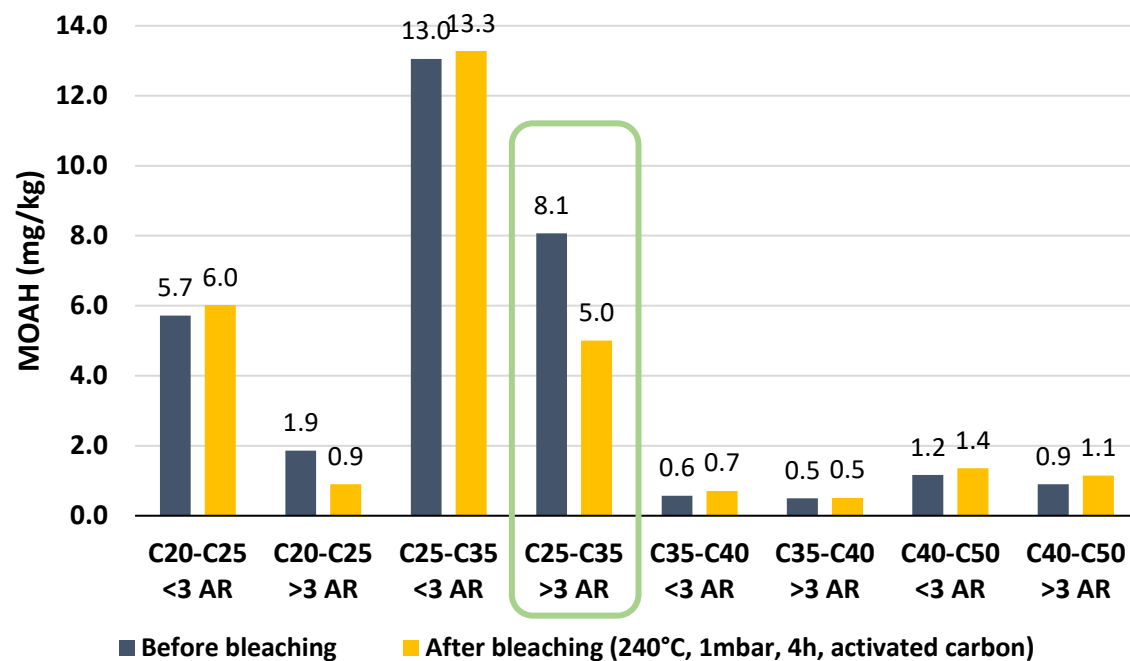


Bleaching conditions: activated carbon (Jacobi ColorSorb™ XFP21), 240°C, 1 mbar, 4 h

What about bleaching?

Effect of bleaching on MOAH

MOAH concentrations before and after bleaching

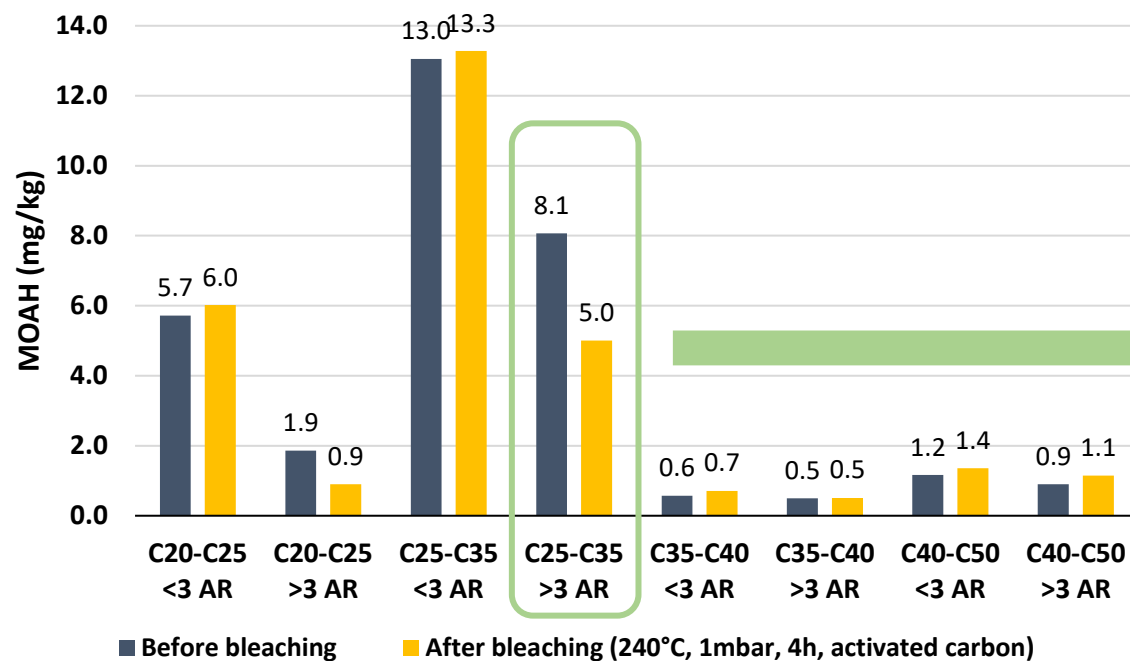


Noticeable reduction of MOAH >3 aromatic rings between C25 and C35

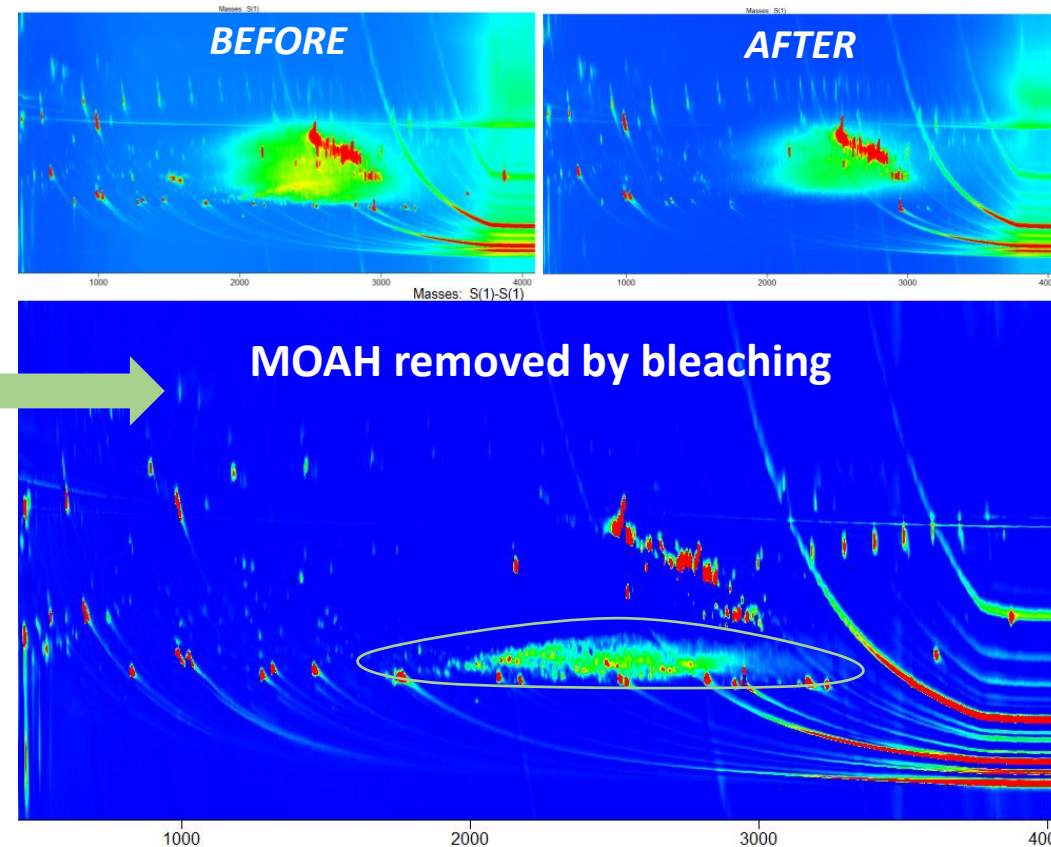
What about bleaching?

Effect of bleaching on MOAH

MOAH concentrations before and after bleaching

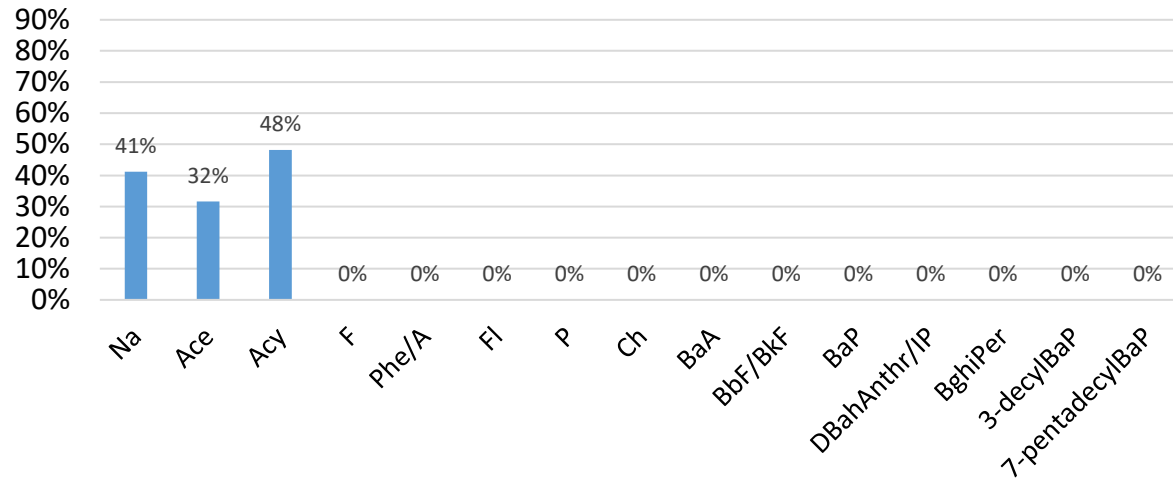


Noticeable reduction of MOAH >3 aromatic rings between C25 and C35

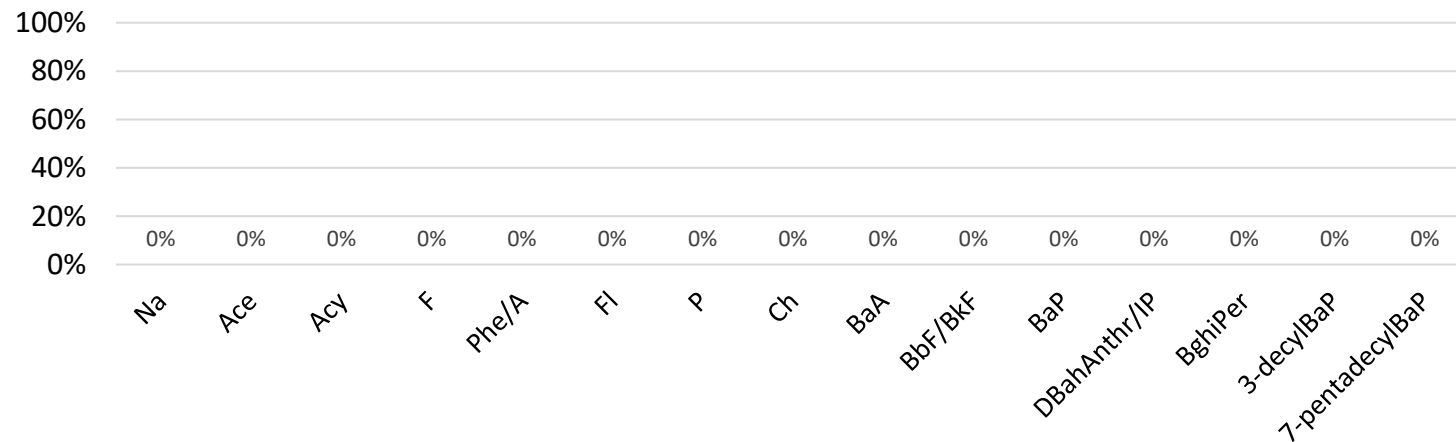


Remaining PAHs and PAC

■ After bleaching

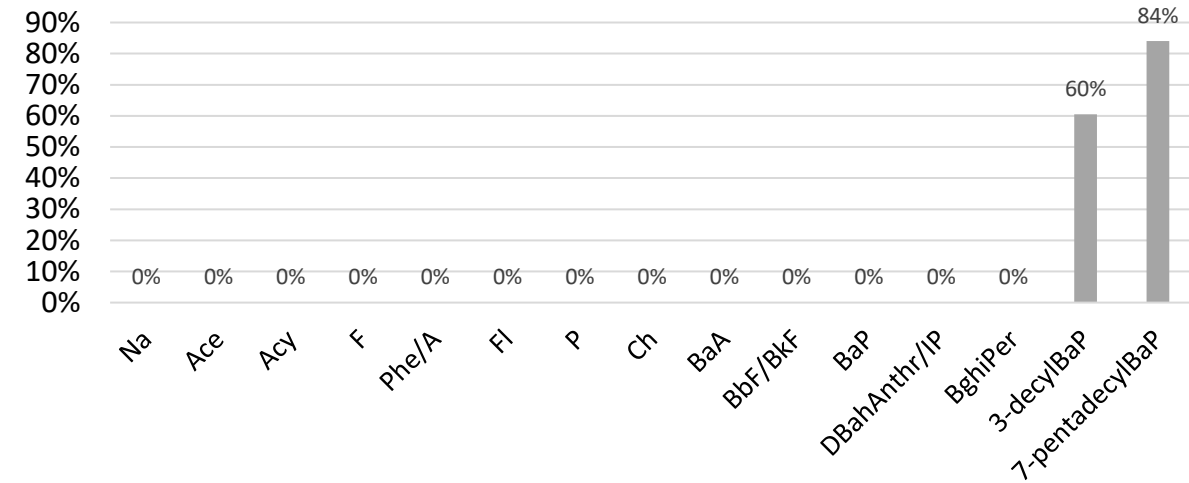


■ After bleaching and deodorization



Remaining PAHs and PAC

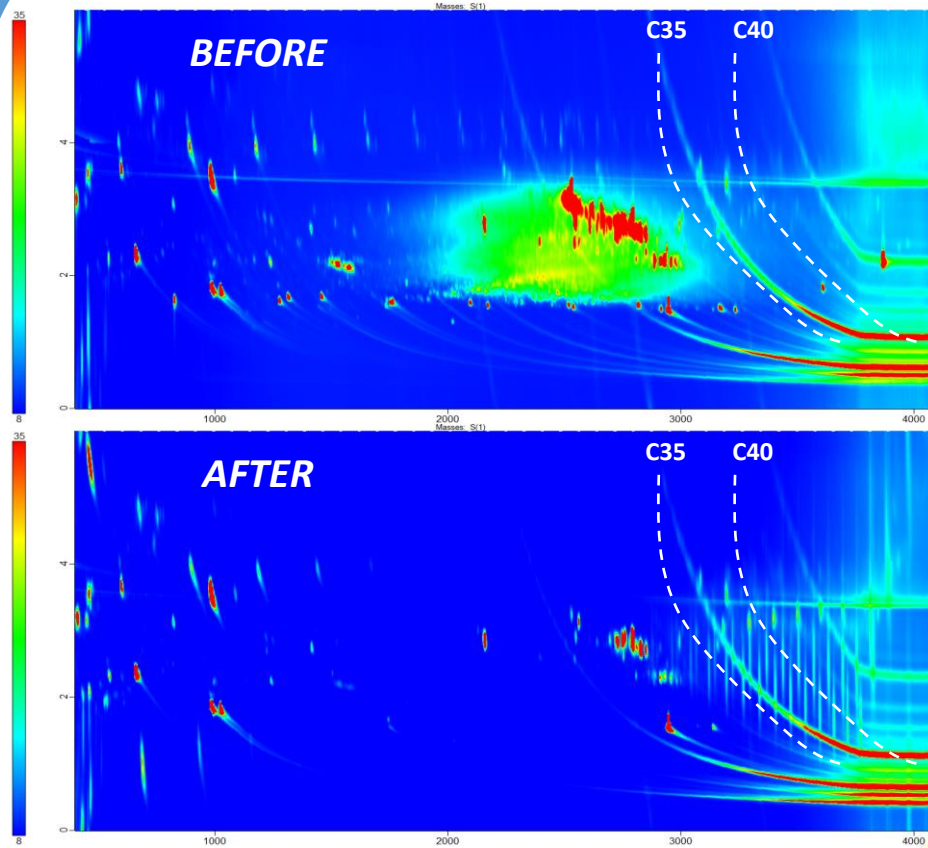
■ After deodorization



Bleaching removed all PAHs apart from Na, Ac, Ap, which could then anyway be removed during the deodorization step

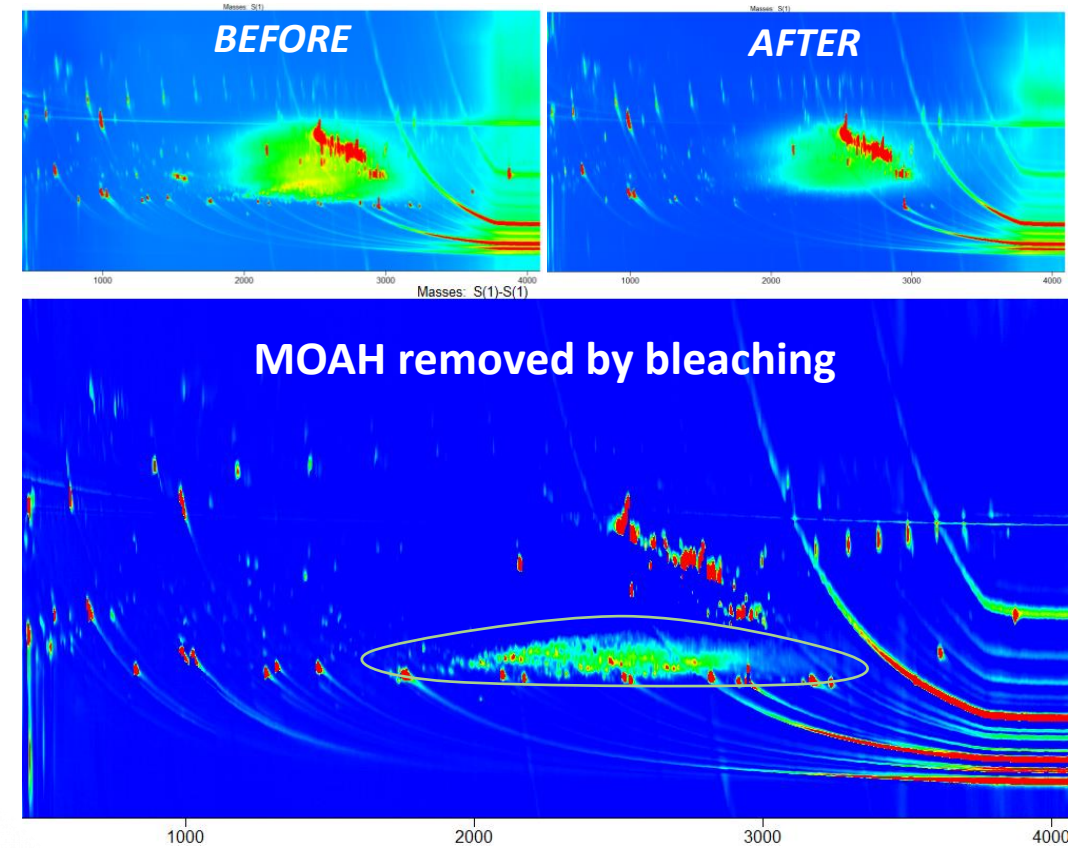
Effect of deodorization and/or bleaching on MOAH

Deodorization



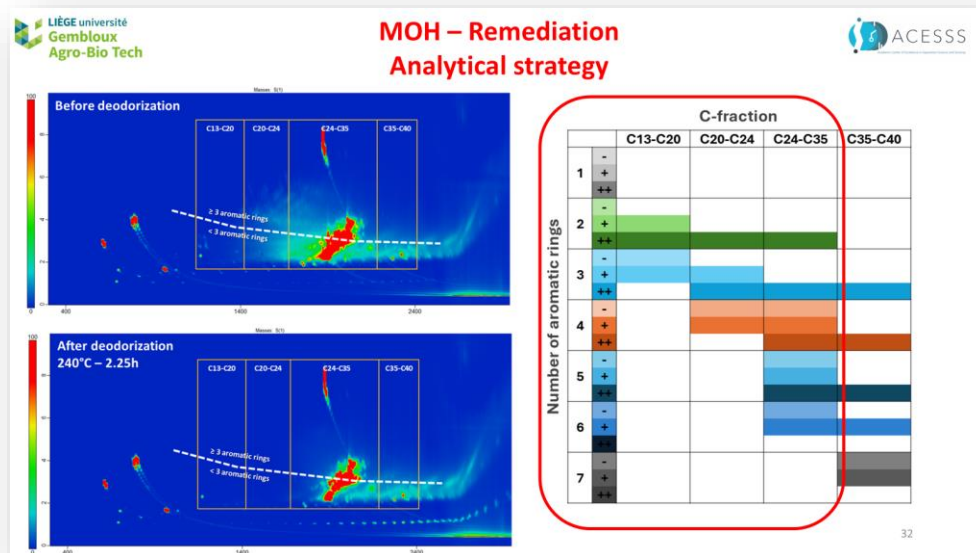
240°C – 4h; 1mbar

Bleaching

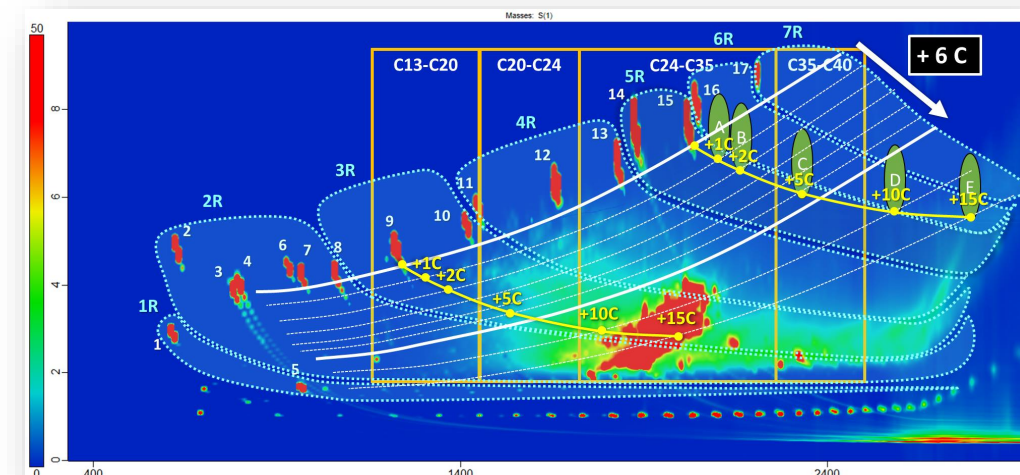


**WORK IN
 PROGRESS**
 CHECK BACK SOON!

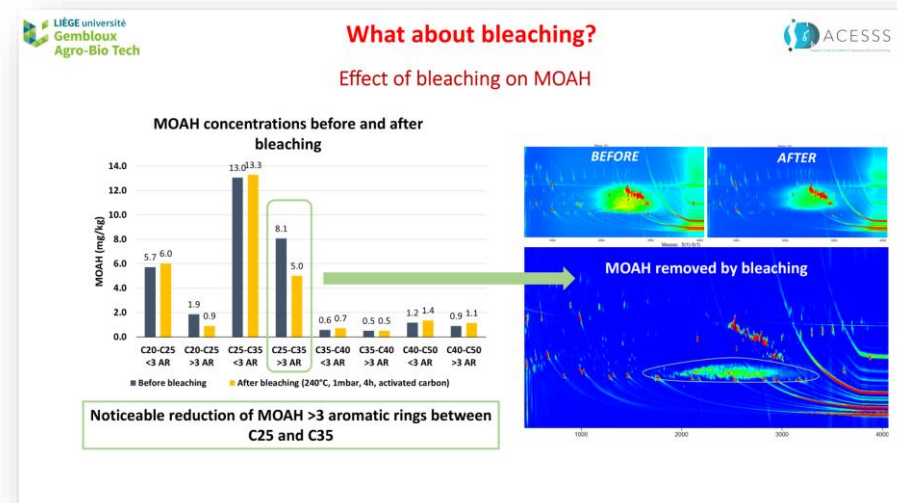
- **GC×GC** is a **strategic tool**, not only for getting more insight on the profile of the MOSH & MOAH contamination but also to elucidate the role of **remediation strategy**.



- **Bleaching** significantly reduce **MOAH >3 aromatic rings between C25 and C35**



- **Deodorization** significantly remove MOH contamination depending on the **T/t** and **pressure** applied



June, 12-13

3rd ADVANCES IN SEPARATION SCIENCE WORKSHOP

Gembloux, Belgium



J. Pawliszyn



C. Cordero



G. Hopfgartner



H. Mol



B. Bojko



E. Leitner



P. Tranchida



E. Rosenberg



K. Schug



H.-G. Janssen



E. Gionfriddo

June, 11: 1-day course in

- **SPME**
- **GC**



My research group:

Sophie Vancraenenbroeck

Paula Albendea

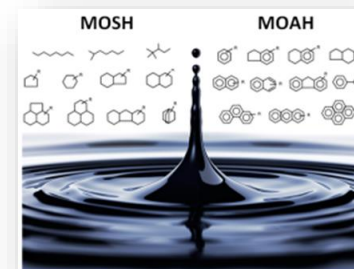
Damien Eggermont

Aleksandra Gorska

Donatella Ferrara

Damien Pierret

Carlo Bellinghieri



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