



Variability of edaphic conditions in metalcontaminated sites at multiple scales. A temperate and a tropical situations

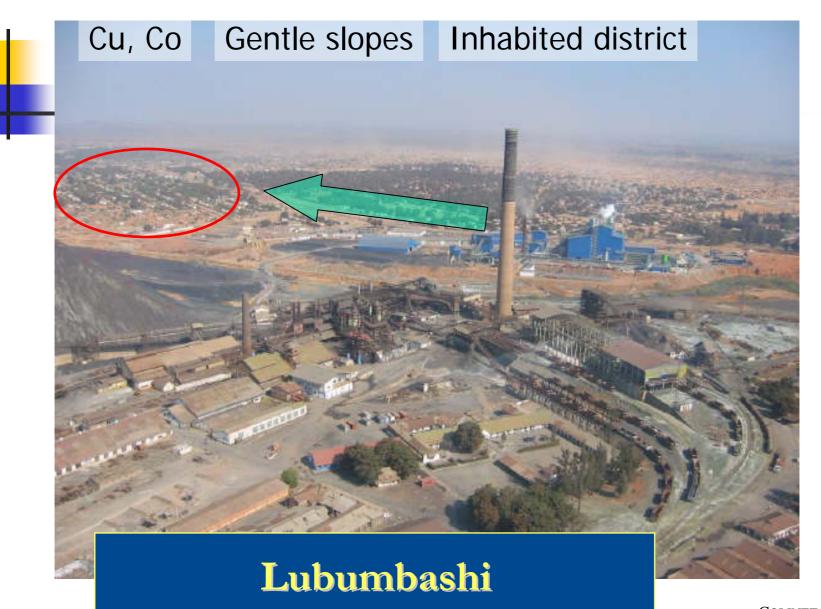
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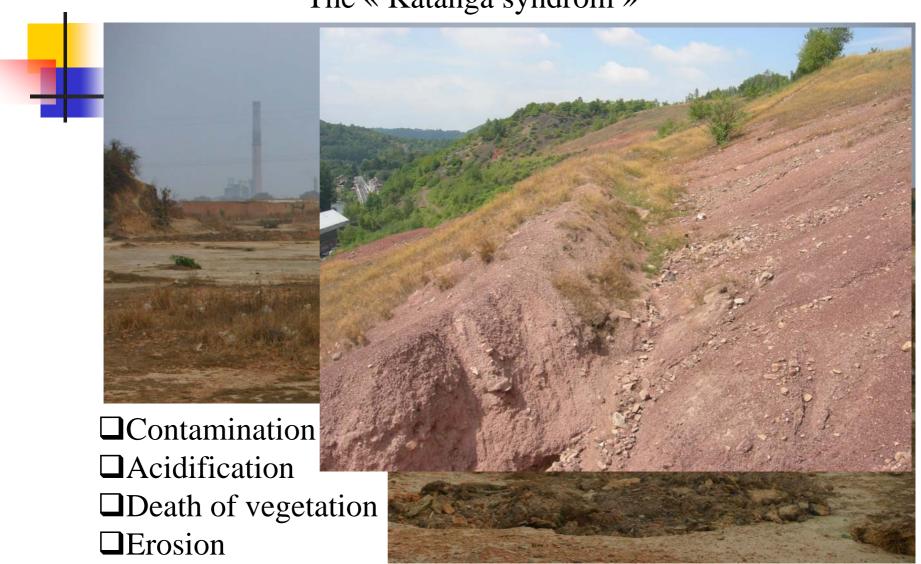




Gembloux agricultural University
Soil Science Department
Edaphic conditions in contaminated sites - 4



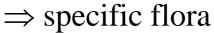
The « Katanga syndrom »





Theme

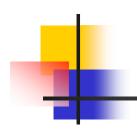
Soil TE content generates ecological pressure











Questions

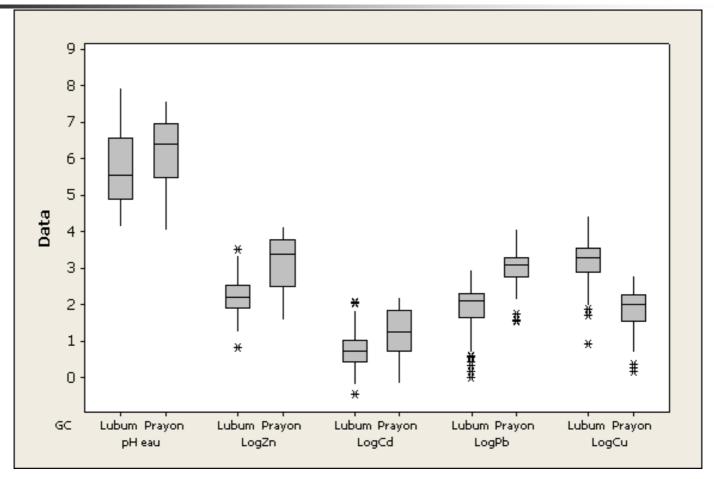
What range of ecological conditions?

What consequences for site characterization? remediation?



Main results







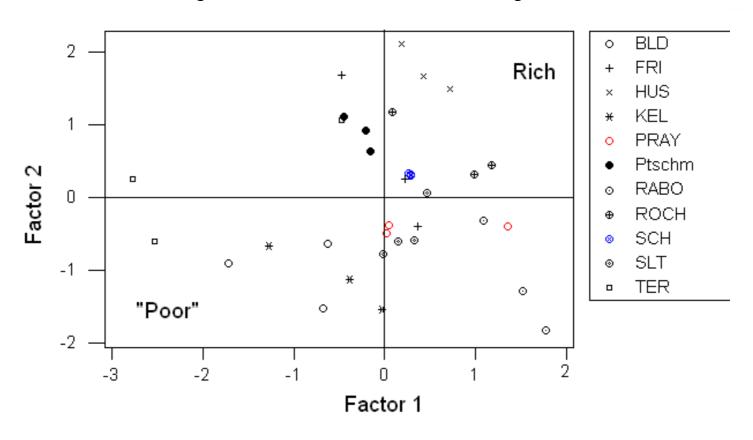


Site specific relationships





2. Variability within the sites: Prayon

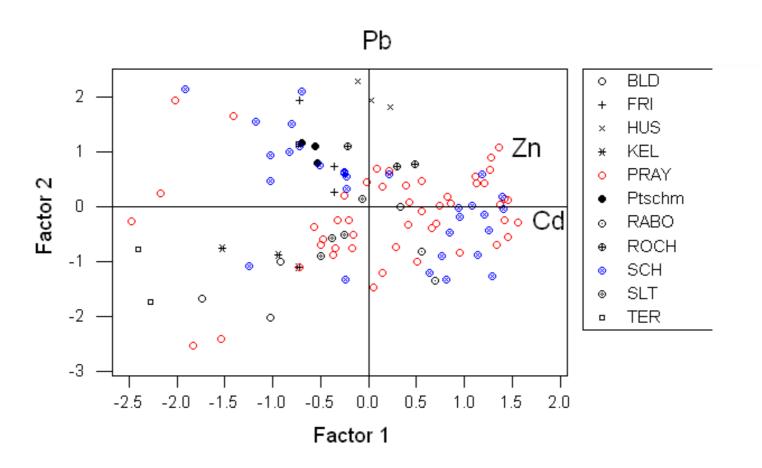


Differences between sites appear important because triplets (n=3) seem closer to each other than to other groups

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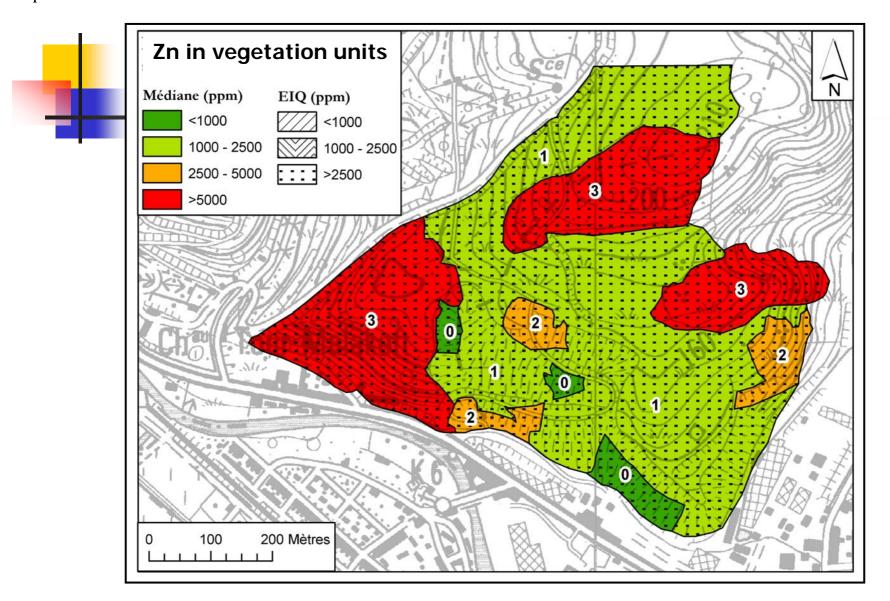






When sampling density rises, the within-site diversity appears much more important.









2. Variability within the sites: Lubumbashi

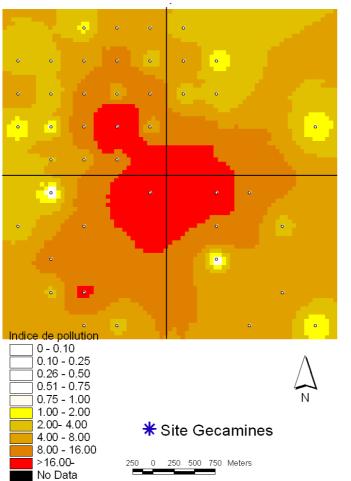


No significant difference of TE content between districts

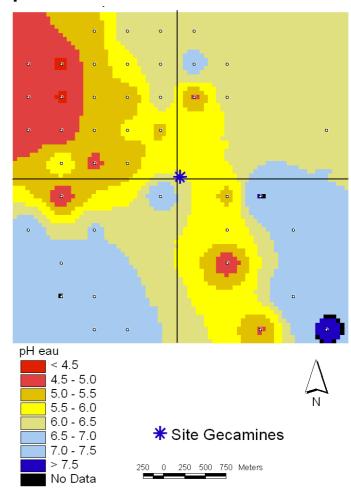




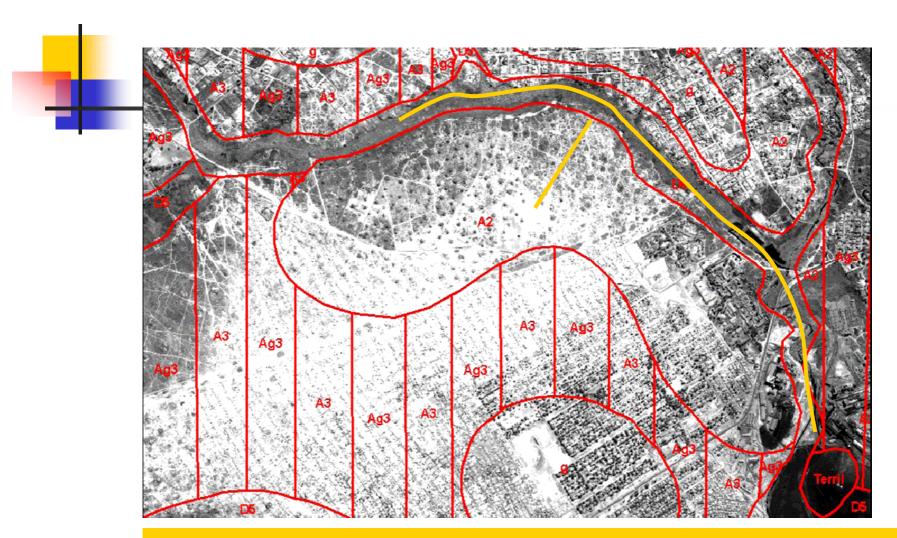
Synthetic pollution index



pHwater in Lubumbashi

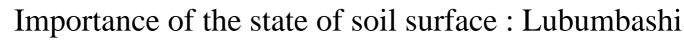






The Gecamines district







0,50

0,00

 α

SR

SR

က

SR

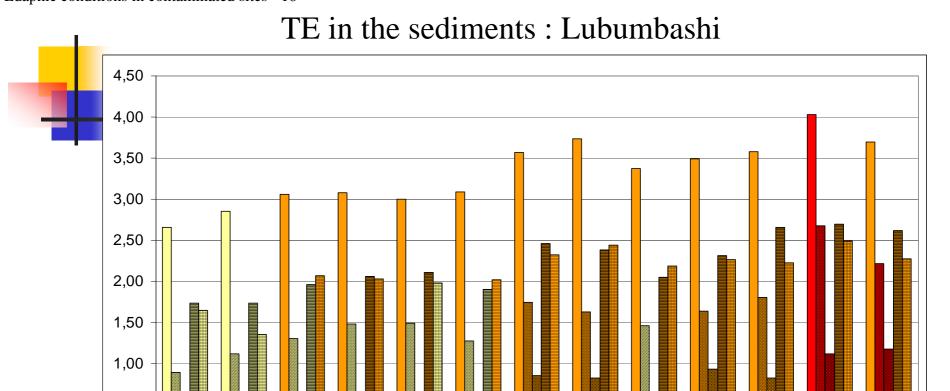
4

SR

2

SR





9

SR

/

SR

□ Cu
□ Co
□ Cd
□ Pb
□ Zn

 ∞

SR

0

SR

10

SR

SR

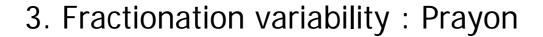
12

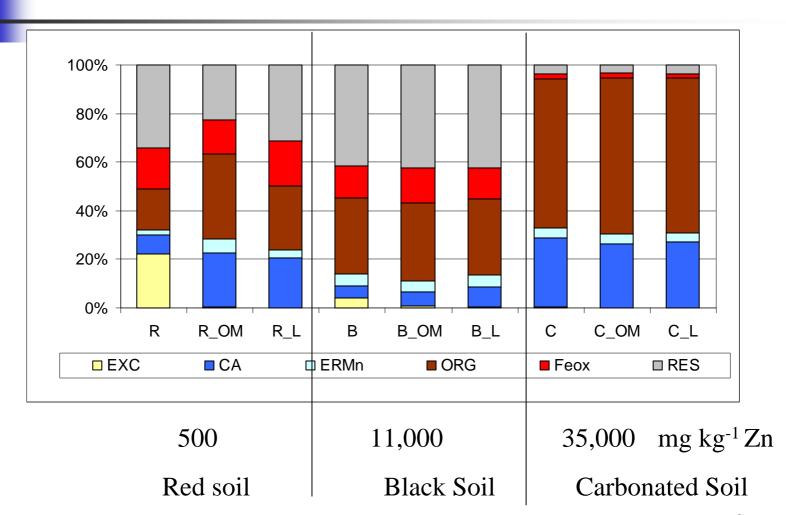
SR

13

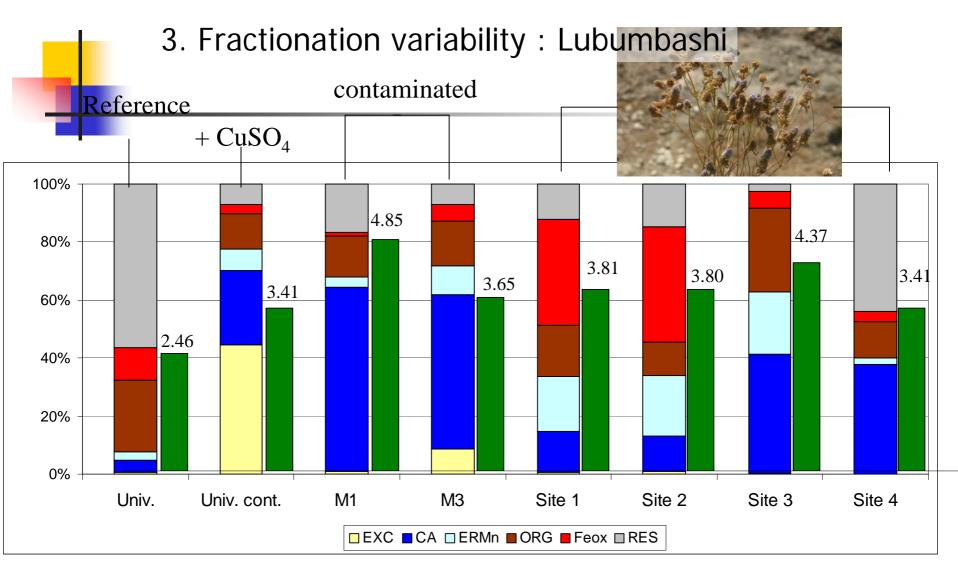
SR











Need for a typology of the contaminations...





Conclusions

High variability whatever the scale considered Importance of redistribution processes



Definition of differentiated management areas



More research is conducted about...





- (i) contaminations and physical-chemical conditions in the rhizosphere,
- (ii) plant tolerance for growing in multi-metallic contaminations,
- (iii) the ways to reduce the limitations for installation of vegetation



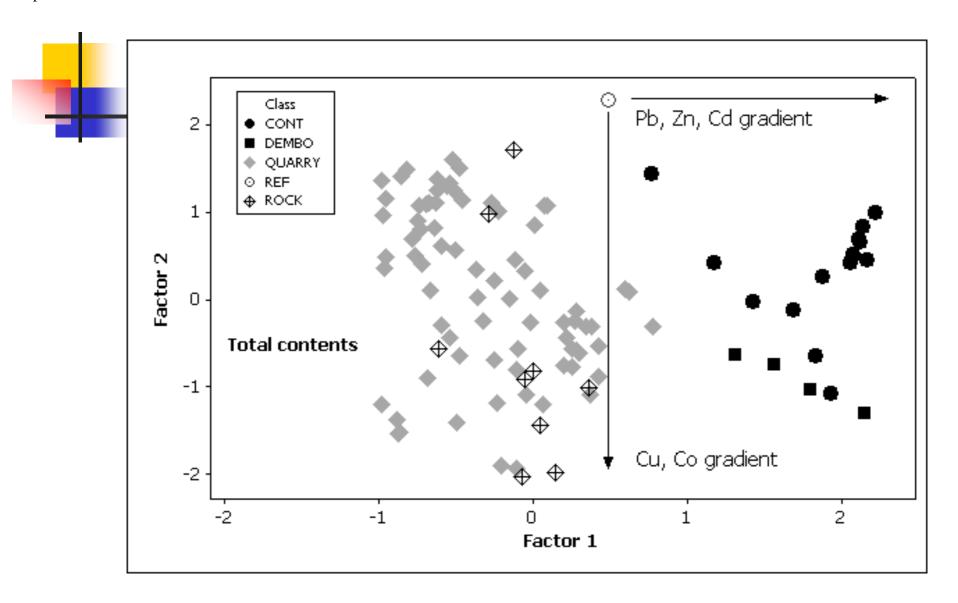


Thank You!

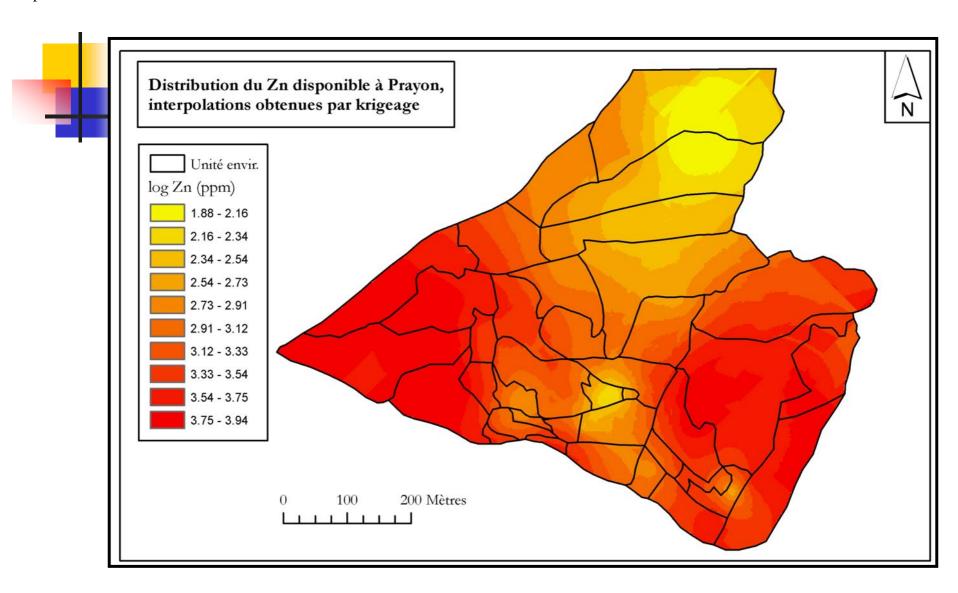


Form	Abbr.	Method
Exchangeable	EXC	25 ml 1 M MgCl ₂ , pH 7, shaking 1h
Acido-soluble	CAR	25 ml 1 M CH ₃ COO-Na + CH ₃ COOH, pH 5, shaking 3h
Easily reducible	ERM	25 ml 0.1 N NH ₂ OH.HCl + 0.01 N HNO ₃ , pH 2, shaking 30'
Organic matter	ORG	5 ml 35% (v:v) $H_2O_2 + 3$ ml 0.02 N HNO ₃ , 2 h at 85°C
		3 ml 35% (v:v) H ₂ O ₂ , 2 h at 85°C
		10 ml 2 N NH ₄ NO ₃ in 20% (v:v) HNO ₃ , 30'
Fe and Al oxides	FOX	20 ml 1 N NH ₂ OH.HCl + 25% (v:v) CH ₃ COOH, 4h at 90°C
Residual forms	RES	2x (5 ml conc. HF + 1,5 ml conc. HClO ₄) + 10 ml 10% (v:v) HCl





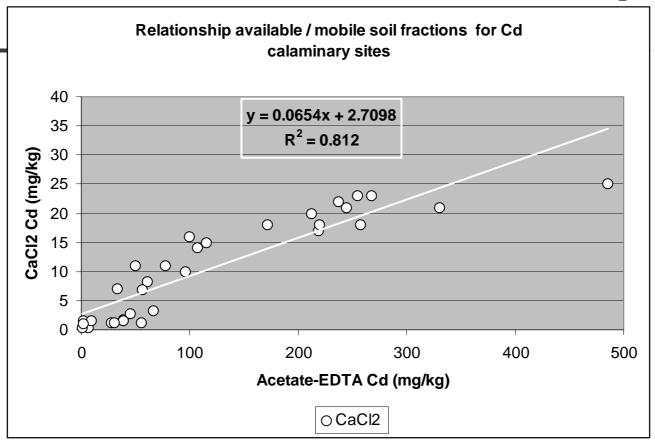








Identification of mobilizable-mobile relationships



 r^2 Zn: 0.444; Pb: 0.265

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