

Contents lists available at [ScienceDirect](www.sciencedirect.com/science/journal/03014797)

## Journal of Environmental Management



journal homepage: [www.elsevier.com/locate/jenvman](https://www.elsevier.com/locate/jenvman)

## Research article

# Sub-daily variability of carbon dioxide, methane, and nitrous oxide emissions from two urban ponds in Brussels (Belgium)

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#### ARTICLE INFO

*Keywords:* Carbon dioxide Methane Nitrous oxide  $\delta^{13}$ C-CH<sub>4</sub> Brussels Urban ponds Urban ecology

Handling Editor: Jason Michael Evans

#### ABSTRACT

Sub-daily variations might significantly impact the estimates of GHG emissions from lakes and ponds. The objectives of this study are (i) to quantify sub-daily variations of the emissions of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O from two urban ponds (Silex and Pêcheries) in the city of Brussels, (ii) to quantify if the sub-daily variations of GHG emissions were significant compared to their seasonal variations and to inter-pond variations among 20 other ponds in the city of Brussels. The partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>), CH<sub>4</sub> concentration, and N<sub>2</sub>O saturation level (% N<sub>2</sub>O) were measured hourly from dawn to dusk in the Pêcheries turbid-water pond and in the Silex clear-water pond during the four seasons in 2023–2024. pCO2 followed the day-night cycle of photosynthesis in spring and summer but was more erratic in winter and fall. The variations of CH<sub>4</sub> concentration and %N<sub>2</sub>O were on most occasions erratic and difficult to attribute systematically to specific biogeochemical processes. The sub-daily variations of computed GHG emissions were mostly driven by variability in wind speed that usually peaked around mid-day. The comparison with previously acquired seasonal and inter-pond data ( $n = 22$ ) showed that sub-daily variations of GHG fluxes were lower than seasonal variations, which were in turn lower than inter-pond variations. Consequently, to design sampling strategies to reduce the uncertainty on the estimate of  $CO<sub>2</sub>$ ,  $CH<sub>4</sub>$ , and  $N_2O$  emissions a priority should be given to describe inter-system variability, followed by seasonal variability, and lastly sub-daily variability, in the context of the environmental management of inland waters, including urban ponds.

#### **1. Introduction**

Emissions to the atmosphere from inland waters (rivers, lakes, and reservoirs) of greenhouse gases (GHGs) such as carbon dioxide  $(CO<sub>2</sub>)$ , methane (CH<sub>4</sub>), and nitrous oxide (N<sub>2</sub>O) are quantitatively important for global budgets (Lauerwald et al.,  $2023$ ). The contribution of  $CO<sub>2</sub>$  and CH4 emissions from small water bodies (ponds) could be disproportionately high compared to larger lakes owing to higher values of areal emission density ([Holgerson and Raymond, 2016\)](#page-12-0). Yet, GHG emissions from lakes and ponds remain poorly constrained, due to the paucity of data coverage that is inadequate to correctly capture the enormous spatial heterogeneity and temporal variability ([Deemer and Holgerson,](#page-12-0)  [2021;](#page-12-0) [Borges et al., 2022](#page-12-0); [Ray et al., 2023\)](#page-13-0). This might be even more critical for small water bodies such as ponds that are more abundant (numerous) and potentially more diverse than larger lentic water bodies ([Verpoorter et al., 2014;](#page-13-0) [Cael et al., 2017\)](#page-12-0). Emissions of CO<sub>2</sub> and CH<sub>4</sub> span over very wide ranges of values across different lakes and ponds depending on a combination of lake and pond morphology (depth and surface area), catchment topography (slope), climate, and land cover on the catchment ([Staehr et al., 2012](#page-13-0); [Maberly et al., 2013](#page-13-0); [Casas-Ruiz](#page-12-0)  [et al., 2021a](#page-12-0), [2021b;](#page-12-0) [Borges et al., 2022](#page-12-0)). Combined, these factors determine the amount and fate of allochthonous organic carbon inputs to lakes and ponds as well as the level of autochthonous organic carbon production. Both largely determine the production and emission of  $CO<sub>2</sub>$ and CH4 from lakes and ponds to the atmosphere [\(Staehr et al., 2012](#page-13-0); [Maberly et al., 2013](#page-13-0); [Borges et al., 2022](#page-12-0)). Additional processes such as lateral inputs from  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  from ground-water or soil-water will also depend on a combination of the same drivers ([Weyhenmeyer et al.,](#page-13-0)  [2015\)](#page-13-0).

Additionally, the exchanges of  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  in lakes and ponds with the atmosphere also vary across temporal scales from sub-daily to seasonal and to inter-annual. While seasonal (e.g. [Borges et al., 2023;](#page-12-0) [Ray](#page-13-0) 

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<https://doi.org/10.1016/j.jenvman.2024.123627>

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Received 8 July 2024; Received in revised form 8 November 2024; Accepted 2 December 2024

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<span id="page-1-0"></span>[and Holgerson, 2023\)](#page-13-0) and even decadal (e.g. [Xiao et al., 2024](#page-14-0)) variations of  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  emissions have been frequently described, their diel variations are not well documented in lakes and ponds. Yet, diel variations of  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  emissions are potentially important and could lead to a substantial revision of estimates since most data are usually acquired during day-time only, and day-time  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  emissions are not necessarily equivalent to those during night-time ([Sieczko et al.,](#page-13-0)  [2020; Rudberg et al., 2021;](#page-13-0) [Golub et al., 2023](#page-12-0)). Nevertheless, no studies have so far compared consistently and quantitatively the variability of GHG emissions from lakes and ponds at diel scale with variability at larger temporal scales (seasonal) or with the variability across different lakes and ponds driven by differences in productivity, depth, and surface

area ([Holgerson and Raymond, 2016;](#page-12-0) [DelSontro et al., 2018](#page-12-0); [Deemer](#page-12-0)  [and Holgerson, 2021](#page-12-0); [Casas-Ruiz et al., 2021a; Borges et al., 2022](#page-12-0)).

Diel variations of GHG emissions to the atmosphere from inland waters can result from night-day variations of dissolved concentration of the GHGs and/or from the night-day variations of the gas transfer velocity (*k*). To investigate the diel variations of GHG emissions to the atmosphere from inland waters it is possible to directly measure the GHG fluxes with either eddy-covariance or floating chambers ([Podgrajsek et al., 2014;](#page-13-0) Erkkilä [et al., 2018](#page-12-0)) or to measure the dissolved GHG concentrations and compute the fluxes with modelled *k* values, usually from parameterizations as function of wind speed (e.g. Klaus and [Vachon, 2020](#page-12-0); [MacIntyre et al., 2021a,b\)](#page-13-0). Each of these three



 $A) CO<sub>2</sub>$ 



**Fig. 1.** Diagram summarizing the processes that can increase CH<sub>4</sub> (A) and CO<sub>2</sub> (B) emissions from lakes to the atmosphere during daytime (right) and night-time (left). Available evidence from automated floating domes suggests that the effect of higher daytime wind speed leads to higher CH4 ([Sieczko et al., 2020](#page-13-0)) and CO2 [\(Rudberg et al., 2021](#page-13-0)) emissions in Swedish lakes. (1) [MacIntyre et al. \(2010\);](#page-13-0) (2) [Sieczko et al. \(2020\);](#page-13-0) (3) [Martinez-Cruz et al. \(2020\);](#page-13-0) (4) [Podgrajsek et al.](#page-13-0)  [\(2014\)](#page-13-0); (5) [McGinnis et al. \(2017\)](#page-13-0); (6) [Wanninkhof \(1992\);](#page-13-0) (7) [Hofmann et al. \(2010\)](#page-12-0); (8) [Lorke and Peeters \(2006\)](#page-13-0); (9) [MacIntyre et al. \(2006\)](#page-13-0); (10) [Murase and](#page-13-0)  [Sugimoto \(2005\)](#page-13-0); (11) [Morana et al. \(2020\);](#page-13-0) (12) Bižić et al. (2020); (13) [Oswald et al. \(2015\);](#page-13-0) (14) [Rudberg et al. \(2021\);](#page-13-0) (15) [Rocher-Ros et al. \(2021\)](#page-13-0) (16) Borges [et al. \(2022\);](#page-12-0) MOX = methane oxidation; P = gross primary production; R = community respiration.

approaches to derive GHG fluxes has advantages but also limitations that have been documented and several inter-comparisons have evaluated the consistency of the three approaches (e.g. Guérin et al., 2007; [Podgrajsek et al., 2014](#page-13-0); [Holgerson et al., 2017;](#page-12-0) Erkkilä [et al., 2018](#page-12-0); [MacIntyre et al., 2021a,b\)](#page-13-0). Several processes generate differences in the concentrations of CO2 and CH4 and in the *k* values between day-time and night-time resulting in differences in emissions of  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  as summarized in [Fig. 1.](#page-1-0) Available evidence from eddy covariance in 13 lakes and reservoirs in the Northern Hemisphere indicated that day-time and night-time  $CO<sub>2</sub>$  fluxes were close numerically in 8 systems, but night-time CO<sub>2</sub> fluxes were higher in 5 systems ([Golub et al., 2023](#page-12-0)). Most of the lakes and reservoirs studied by [Golub et al. \(2023\)](#page-12-0) were located *>*60◦N so the lower average fluxes during day-time could be related to a very long photo-period at higher latitudes during summer when aquatic primary production lowers day-time  $CO<sub>2</sub>$  fluxes. Several studies have reported CH4 flux measurements in lakes with eddy-covariance [\(Podgrajsek et al., 2014;](#page-13-0) [Franz et al., 2016;](#page-12-0) [Xiao et al.,](#page-14-0)  [2017;](#page-14-0) [Jammet et al., 2017;](#page-12-0) [Sollberger et al., 2017](#page-13-0); Erkkilä [et al., 2018](#page-12-0)), and approximately half of them reported higher day-time fluxes, and the other half showed higher night-time emissions, as summarized by [Sieczko et al. \(2020\)](#page-13-0). Available evidence from floating chamber measurements in four Swedish lakes [\(Sieczko et al., 2020](#page-13-0); [Rudberg et al.,](#page-13-0)   $2021$ ) indicates that both  $CO<sub>2</sub>$  and CH<sub>4</sub> emissions tended to be higher during day-time because of higher wind speed leading to higher *k* values ([Fig. 1](#page-1-0)). Under light wind conditions, turbulent kinetic energy can increase during periods of heating (buoyancy flux) related to the establishment of a diurnal thermocline leading to an increase of *k* during the morning till mid-day [\(MacIntyre et al., 2021a,b\)](#page-13-0). Additionally, higher wind speed during day-time can increase the vertical transfer to surface waters of  $CO<sub>2</sub>$  and  $CH<sub>4</sub>$  from deep waters rich in these gases, as well as lead to waves that induce pressure variations on sediments and pump pore-waters rich in  $CO<sub>2</sub>$  and CH<sub>4</sub> and also induce CH<sub>4</sub> ebullition ([Hofmann et al., 2010\)](#page-12-0). During day-time, photosynthesis leads to a decrease of  $CO<sub>2</sub>$ , although changes of  $CO<sub>2</sub>$  during day-time and night-time are generally symmetrical, so that average day-time  $CO<sub>2</sub>$ concentration should be nearly identical to average night-time  $CO<sub>2</sub>$ concentration if the night-time and day-time periods are similar (~12h:12h). Photo-oxidation of dissolved organic matter can lead to a production of CO<sub>2</sub> during day-time, although this process seems quantitatively modest compared to other processes driving  $CO<sub>2</sub>$  variability in lakes ([Rocher-Ros et al., 2021](#page-13-0)). Aerobic  $CH<sub>4</sub>$  production seems to be linked directly or indirectly to photosynthesis so it should mainly occur during day-time (Bižić et al., 2020; [Morana et al., 2020\)](#page-13-0). Light might additionally inhibit methane oxidation (MOX) ([Dumestre et al., 1999](#page-12-0); [Murase and Sugimoto, 2005](#page-13-0); [Morana et al., 2020\)](#page-13-0). However, the increase of photosynthetically produced oxygen  $(O_2)$  during day-time might favour aerobic MOX at the oxycline ([Oswald et al., 2015\)](#page-13-0) as well as superficial sediments in shallow systems. MOX modifies the  ${}^{13}C/{}^{12}C$  ratio ( $\delta {}^{13}C$ -CH<sub>4</sub>) of residual CH<sub>4</sub> in water, as methanotrophs preferentially consume CH<sub>4</sub> with the lighter carbon stable isotope  $(^{12}C)$ over the heavier one  $(^{13}C)$ . Altogether these processes might result in CH4 dissolved concentrations distributed asymmetrically during the diel cycle, possibly higher during day-time, unlike  $CO<sub>2</sub>$  concentrations that should be nearly symmetric.

The diel variability of  $N_2O$  fluxes in lakes and ponds is less documented than for  $CO<sub>2</sub>$  and CH<sub>4</sub>. Measurements of  $CO<sub>2</sub>$  and CH<sub>4</sub> exchange with the atmosphere can be routinely automated with either eddycovariance or floating chamber methods coupled to low-cost instrumentation for the latter. This is not the case for  $N_2O$  because of lower fluxes (lower signal), inadequate sensitivity or response time of available instrumentation, or unavailability of low cost instrumentation. So, studies of  $N_2O$  in lakes and ponds have mostly relied so far on gas chromatography (GC) limiting the possibility of automation.  $N_2O$ sources and sinks in lakes depend on denitrification and nitrification that in turn depend on dissolved inorganic nitrogen (DIN) availability and O<sub>2</sub> levels [\(Mengis et al., 1997;](#page-13-0) [Borges et al., 2023\)](#page-12-0). O<sub>2</sub> levels change

in surface waters at daily scale, increasing during day-time with photosynthesis and decreasing during night-time. In shallow systems, these oscillations of  $O_2$  in the water column might modify oxygenation of the top layer of sediments and affect sedimentary processes including nitrification and denitrification. In rivers, a night-time increase of  $N_2O$ has been attributed to the lowering of  $O<sub>2</sub>$  in the water column (Harrison [et al., 2005;](#page-12-0) [Clough et al., 2007;](#page-12-0) [Baulch et al., 2012;](#page-12-0) [Xia et al., 2013](#page-14-0); [Chen et al., 2021](#page-12-0)). In Lake Wuliangsuhai (China), [Ni et al. \(2022\)](#page-13-0) showed that  $N_2O$  emissions followed diurnal cycles, peaking in the middle of the day when  $O_2$  concentrations were maximal in areas dominated by submerged macrophytes. Other studies in Lake Taihu (China) [\(Xiao et al., 2019](#page-14-0)) and in Ashumet pond (Massachusetts, United states) ([Smith et al., 2024](#page-13-0)) did not report marked and systematic sub-daily changes in  $N_2O$  concentrations.

The present paper reports a dataset of  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O dynamics in two contrasted shallow small urban ponds (Pêcheries and Silex) in the city of Brussels (Belgium) [\(Fig. 2](#page-3-0)). The Pêcheries pond is a turbid-water phytoplankton dominated system and the Silex pond is a clear-water macrophyte dominated system. The urban ponds in the city of Brussels fall into either of these two categories [\(Scheffer et al., 1993\)](#page-13-0) that have distinct GHG emission rates [\(Bauduin et al., 2024a,](#page-12-0) [2024b\)](#page-12-0). Data were collected every hour starting before dawn and ending after dusk in spring (3rd and  $8<sup>th</sup>$  April 2023), summer (7th and  $13<sup>th</sup>$  September 2023), fall (15th and  $16^{\text{th}}$  November 2023) and winter (7th and  $8^{\text{th}}$  January 2024) at the Pêcheries and the Silex ponds. The objectives of this study are to investigate if diurnal variations in the concentrations and fluxes of  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O (i) are detectable and systematic in the two ponds; (ii) change with season in occurrence or amplitude; (iii) can be explained by biogeochemical processes and/or meteorological conditions such as wind speed or light intensity; (iv) are quantitatively significant compared to seasonal variations in the two ponds and compared to inter-system variation among other ponds of the city of Brussels.

#### **2. Material and methods**

#### *2.1. Site description*

Belgium is characterized by a temperate climate, with year-round mild temperatures and uniform high precipitation. Over the 1991–2020 period, annual precipitation averaged 837 mm, annual temperature averaged 11 ◦C, with summer and winter averages of 17.9 ◦C and 4.1 ◦C, respectively. The ponds in the city of Brussels are artificial, small, shallow, and eutrophicated [\(De Backer et al., 2010;](#page-12-0) [Van](#page-13-0)  [Onsem et al., 2010](#page-13-0); [Peretyatko et al., 2012\)](#page-13-0). The ponds in the periphery of the city are located in the Sonian Forest, the ponds closer to the city center are situated primarily within recreational parks. Several ponds of the city of Brussels are located in protected Natura 2000 areas and are valuable for biodiversity conservation, as well as for recreational purposes such as boating and fishing. The ponds in the city of Brussels are eutrophic, and occur in two states *sensu* [Scheffer et al. \(1993\)](#page-13-0): clear-water ponds with dominance of a few macrophyte species with a variable abundance ([Peretyatko et al., 2007](#page-13-0)); turbid-water ponds with dominance of phytoplankton composed mainly of cyanobacteria with occasional harmful algal blooms [\(De Backer et al., 2010](#page-12-0); [Van Onsem](#page-13-0)  [et al., 2010](#page-13-0); [Peretyatko et al., 2012\)](#page-13-0). The accumulation of high cyanobacteria biomass can lead to anoxic/hypoxic conditions and/or accumulation of toxins causing occasional fish kills, with a negative effect on the aquatic ecological functioning and leading to public health concerns ([De Backer et al., 2010](#page-12-0); [Van Onsem et al., 2010; Peretyatko et al., 2012](#page-13-0)).

The two sampled ponds (Pêcheries and Silex) are located in the city of Brussels, in the Woluwe river basin, at the periphery of the Sonian forest ([Fig. 2\)](#page-3-0). The Pêcheries and the Silex ponds are both relatively small (1.0–1.4 ha) and shallow (1.1–1.3 m) ([Fig. 2](#page-3-0)), and mainly differ in the type of dominant aquatic primary producer during spring-summer. The Pêcheries pond is a turbid-water pond with higher chlorophyll-*a* (Chl-*a)* and total suspended matter (TSM) values during summer [\(Fig. 2\)](#page-3-0)

<span id="page-3-0"></span>

**Fig. 2.** Location of the two sampled ponds in Brussels (Belgium, Europe). Bottom left map shows the metropolitan area of the region of Brussels delineated by the black line and surrounding region of Flanders in Belgium, showing land cover and sampled urban ponds (black diamonds). The star corresponds to the center of the city (50.8504◦N, 4.3487◦E). Additional information for each pond is indicated on right panels: shapes of the ponds, surface area (ha), perimeter (m), average depth (m) (obtained thanks to Bruxelles Environnement, the regional water management agency), mean ± standard deviation of summer chlorophyll-*a* (Chl−*a*, μg L<sup>−1</sup>) and summer total suspended matter (TSM, mg L<sup>-1</sup>) of periods from 21 June to 21 September in 2021, 2022, 2023, and summer total macrophyte cover (MC, %) from [Bauduin et al. \(2024a\)](#page-12-0).

due to a dominance of aquatic primary production by phytoplankton (mainly cyanobacteria). The Silex pond is a clear-water pond with lower Chl-*a* and TSM values due to a dominance of aquatic primary production by submerged macrophytes (mainly *Lemna trisulca*). The seasonal and inter-annual variations of  $CO<sub>2</sub>$ ,  $CH<sub>4</sub>$ , and  $N<sub>2</sub>O$  dissolved concentrations and emissions were reported in two companion papers ([Bauduin et al.,](#page-12-0)  [2024a,](#page-12-0) [2024b](#page-12-0)), but information on variations at sub-daily scale were lacking, so their relative importance is unknown.

#### *2.2. Meteorological data*

Hourly data acquired by the Royal Meteorological Institute of Belgium of air temperature, rainfall, wind speed, and atmospheric pressure referenced at 10m height were retrieved from [https://wow.me](https://wow.meteo.be/en)  [teo.be/en](https://wow.meteo.be/en) for the meteorological station closest to the two ponds (Institute of St-Lambert; 50.8408◦N 4.4234◦E) located 2.5 km away from the Pêcheries pond and 5 km way from the Silex pond. Hourly data of solar irradiance referenced at 37m height and corrected for cloud cover and aerosols were retrieved from the Prediction Of Worldwide Energy Resources Data Archive [\(https://power.larc.nasa.gov/](https://power.larc.nasa.gov/)) for the same location. The hourly data of solar irradiance were integrated at daily scale using the trapezoidal method. The meteorological data are used to contextualize the differences of the patterns of sub-daily variations of  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O concentrations and fluxes between the two ponds and between seasons. Wind speed is also used to compute the *k*  values from a *k*-wind parameterization (section [2.5\)](#page-4-0).

#### *2.3. Sampling strategy and field sampling*

Sampling was designed to investigate sub-daily variability of GHG concentrations and was not designed to describe spatial variability within the ponds (assumed low since the ponds are small and uniformly shallow), so sampling was carried out at a single location. Two ponds were investigated to test if sub-daily variability changed among clearwater and turbid-water ponds. Four sampling periods were chosen (03–08/04/2023, 07–13/09/2023, 15–16/11/2023, 07–08/01/2024) to test if the patterns and the amplitude of sub-daily variations changed with season (spring, summer, fall, and winter, respectively). Sampling was carried out at a single location, from a pontoon, every hour, starting 1–2h hour before dawn and ending 2–3h after dusk.Water was collected manually 5 cm below the surface with seven polypropylene syringes (60 ml) for gases ( $CO<sub>2</sub>$ ,  $CH<sub>4</sub>$ , N<sub>2</sub>O). Prior to sampling, the syringes were rinsed three times with surface water. Water was sampled taking care to avoid the inclusion of air bubbles, to prevent the alteration of the content of dissolved gases. After sample collection, the syringes were processed as quickly as possible to minimize biological alteration of the content of dissolved gases. Three syringes were used to preserve water samples for subsequent analysis of CH<sub>4</sub> and N<sub>2</sub>O, and  $\delta^{13}$ C-CH<sub>4</sub> in the laboratory and four syringes were used to measure the partial pressure of  $CO<sub>2</sub>$  (pCO<sub>2</sub>) directly in the field. The water was transferred, taking care to avoid the inclusion of air bubbles, from the syringes with a polyvinyl chloride flexible tube (Tubclair®) into three 60 ml borosilicate serum bottles (Wheaton™) for CH<sub>4</sub> and N<sub>2</sub>O, and  $\delta^{13}$ C-CH<sub>4</sub>, preserved with 200 μl of a saturated solution of HgCl<sub>2</sub>, sealed with a butyl stopper

<span id="page-4-0"></span>without a headspace and crimped with an aluminium cap, for subsequent analysis at the laboratory. The  $pCO<sub>2</sub>$  measurements were carried out directly in the field with a Li-Cor® Li-840  $CO<sub>2</sub>/H<sub>2</sub>O$  gas analyser using the headspace technique in four syringes by equilibrating 30 ml of sample water with 30 ml of atmospheric air by vigorous shaking during 5 min ([Borges et al., 2019\)](#page-12-0). The Li-Cor® Li-840 was calibrated 1–2 days before each sampling with ultrapure  $N_2$  and a suite of gas standards (Air Liquide<sup>™</sup> Belgium) with  $CO<sub>2</sub>$  mixing ratios of 388, 813, 3,788, and 8300 ppm manufactured and certified at  $\pm 2\%$  by Air Liquide™ Belgium. The 388 ppm and 813 ppm  $CO<sub>2</sub>$  gas standards from Air Liquide™ Belgium were recalibrated against  $CO<sub>2</sub>$  standard reference materials (SRM) traceable to United States of America National Institute of Standards and Technology (NIST) reference materials provided by the National Oceanic and Atmospheric Administration (NOAA) with CO<sub>2</sub> mixing ratios of 360.5 and 773.8 ppm. Drift in the calibration of the Li-Cor® Li-840 was checked by measuring the standards again after each sampling within 1–2 days, and was systematically undetectable (within the precision of instrument). The overall precision of  $pCO<sub>2</sub>$  measurements was  $\pm 2.0\%$  with a detection limit of 5 ppm.

Water temperature (precision  $\pm 0.1$  °C), specific conductivity (precision  $\pm 0.1$  μS cm $^{-1}$ ), and O<sub>2</sub> saturation level (%O<sub>2</sub>) (precision  $\pm 0.1$ %) were also measured in-situ simultaneously every hour with a VWR® MU 6100H probe. The  $O_2$  probe was calibrated in humidity saturated ambient air at the start of the experiment following the manufacturer recommendation. The conductivity probe was calibrated with a commercial standard (Merck®) of 1000  $\upmu\text{S cm}^{-1}$  (United Nations Standard Products and Services (UNSPSC) Code 41,116,107). Polyethylene containers (2l) were filled with water three to four times a day, stored in cool boxes with frozen packs and processed at the laboratory for soluble reactive phosphorus (SRP), ammonium (NH $_4^+$ ), nitrite (NO $_2^-$ ), nitrate (NO<sub>3</sub>), Chl-a, and TSM concentrations. Prior to sampling, the polyethylene container was rinsed three times with surface water.

#### *2.4. Laboratory analysis*

### 2.4.1. *CH<sub>4</sub>* and  $N_2O$  measurements by gas chromatography and  $\delta^{13}C$ -CH<sub>4</sub> *by cavity ring-down spectrometry*

Measurements of  $N_2O$  and CH<sub>4</sub> dissolved concentrations were made with the headspace technique [\(Weiss, 1981](#page-13-0)) and a GC, following standard operating procedures to quantify GHG emissions from inland waters ([UNESCO/IHA, 2010](#page-13-0)). The used instrumental set-up and analytical protocol compared satisfactorily with 14 other laboratories during an international inter-comparison exercise of  $CH_4$  and  $N_2O$  measurements ([Wilson et al., 2018](#page-14-0)). The headspace (20 ml) was made with ultra-pure N2 (Air Liquide™ Belgium) in two 60 ml serum bottles for a replicate analysis of both CH<sub>4</sub> and N<sub>2</sub>O.The GC (SRI<sup>TM</sup> 8610C) was equipped with a flame ionisation detector for CH4 and an electron capture detector for N<sub>2</sub>O, and was calibrated with CH<sub>4</sub>:N<sub>2</sub>O:N<sub>2</sub> gas mixtures with mixing ratios of 1, 10 and 30 ppm for CH<sub>4</sub>, and 0.2, 2.0 and 6.0 ppm for  $N_2O$ , manufactured and certified at  $\pm 2\%$  for CH<sub>4</sub> and  $\pm 10\%$  for N<sub>2</sub>O by Air Liquide™ Belgium. The precision of measurements based on duplicate samples was  $\pm 3.9\%$  for CH<sub>4</sub> and  $\pm 3.2\%$  for N<sub>2</sub>O. The detection limit was 0.5 nmol  $L^{-1}$  for both CH<sub>4</sub> and N<sub>2</sub>O.

The CO2 concentration is expressed as partial pressure in parts per million (ppm) and CH<sub>4</sub> as dissolved concentration (nmol  $L^{-1}$ ), in accordance with convention in existing topical literature, and because both quantities were systematically and distinctly above saturation level ( $\sim$ 400 ppm and 2–3 nmol L<sup>-1</sup>, respectively). Variations of N<sub>2</sub>O fluctuated around atmospheric equilibrium, so data are presented as percent of saturation level (%N2O, where atmospheric equilibrium corresponds to 100%).

The  $\delta^{13}$ C-CH<sub>4</sub> was measured in the headspace gas (20 ml of synthetic air, Air Liquide™ Belgium) equilibrated with the water sample (total volume 60 ml) on a single serum bottle. All of the samples had a headspace with a partial pressure of  $CH_4$  (pCH<sub>4</sub>) higher than 10 ppm, so the gas samples were diluted to achieve a final  $pCH<sub>4</sub>$  around 6 ppm to

fall within instrumental operational concentration range (2–10 ppm) recommended by the manufacturer. The diluted gas was then injected into a cavity ring-down spectrometer (Picarro™ G2201-i, isotopic analyzer) equipped with a Small Sample Introduction Module 2 (Picarro<sup>™</sup>). The data were corrected using calibration curves of  $\delta^{13}$ C-CH<sub>4</sub> as a function of concentration, based on two gas standards from Airgas™ Specialty Gases with certified  $\delta^{13}$ C-CH<sub>4</sub> values of  $-23.9 \pm 0.3$  ‰ and  $-69.0 \pm 0.3$  ‰ referenced against Vienna Pee Dee Belemnite SRM. The precision of measurement based on repeated analysis of the two standards was better than  $\pm 1.0$  ‰.

#### *2.4.2. Chlorophyll-a, total suspended matter, and dissolved inorganic nutrients*

Chl-a, TSM, and dissolved inorganic nutrients (SRP, NH<sub>4</sub>, NO<sub>2</sub>, NO<sub>3</sub>) were measured using standard operating procedures for the analysis of water quality [\(APHA, 1998\)](#page-12-0) by fluorimetry, differential weight, and colorimetric-spectrophotometric measurements, respectively. A known volume of water was filtered through Whatman™ GF/F glass microfiber filters (porosity 0.7 μm) with a diameter of 47 mm for Chl-*a* and TSM determination. Filters were kept frozen (− 20 ◦C) until extraction of Chl-*a* in 90% acetone and the measurement of the concentration by fluorimetry with a Kontron™ model SFM 25 according to [Yentsch and Menzel \(1963\),](#page-14-0) with a precision of  $\pm 0.1$  µg L<sup>-1</sup> and detection limit of 0.05 μg L<sup>-1</sup>. A calibration curve was made from sequential dilutions with 90% acetone of a commercial (Merck®) Chl-*a*  standard (UNSPSC Code 41,116,107). Filters used for the determination of TSM were weighed with a Explorer™ Pro EP214C analytical microbalance with  $\pm 0.1$  mg precision scale after oven drying at 50  $\degree$ C during 12h before and after filtration of a known volume of water, with a precision of  $\pm 10$ % and detection limit of 0.05 mg L<sup>-1</sup>. Filtered water for the determination of dissolved nutrients was stored frozen (− 20 ◦C) in 50 ml polypropylene vial. Absorption measurements for the colorimetric determination of dissolved nutrients were made with a Perkin Elmer™ Lambda 650S spectrophotometer. SRP was measured by the ammonium molybdate, ascorbic acid, and potassium antimony tartrate staining method [\(Koroleff, 1983\)](#page-13-0). NH4 was measured by the nitroprusside-hypochlorite-phenol staining method ([Grasshoff and](#page-12-0)  [Johannsen, 1972\)](#page-12-0).  $NO<sub>2</sub><sup>-</sup>$  and  $NO<sub>3</sub><sup>-</sup>$  were measured before and after reduction of NO<sub>3</sub> to NO<sub>2</sub> by a cadmium-copper column, using the Griess acid reagent staining method ([Grasshoff et al., 2009\)](#page-12-0). Calibration curves were made from sequential dilutions with pure water of commercial standards (Merck®) traceable to NIST SRMs (84L for  $PO_4^{3-}$ , 999b for NH<sub>4</sub>, 8040 for NO<sub>2</sub>, 723 for NO<sub>3</sub>). Precisions were  $\pm 0.05$ ,  $\pm 0.02$ ,  $\pm 0.02$ , and  $\pm 0.1$ , µmol L<sup>-1</sup> for SRP, NH<sub>4</sub>, NO<sub>2</sub>, and NO<sub>3</sub>, respectively. Detection limits were 0.01, 0.3, 0.01, and 0.15 µmol  $L^{-1}$  for SRP, NH<sub>4</sub>,  $NO<sub>2</sub>$ , and  $NO<sub>3</sub>$ , respectively.

#### *2.5. Calculation of diffusive GHG emissions*

The diffusive air-water  $CO_2$ ,  $CH_4$ , or  $N_2O$  fluxes ( $F_G$ ) were calculated according to eq. (1):

$$
F_G = k \Delta[G] \tag{1}
$$

where  $k$  is the gas transfer velocity and  $\Delta[G]$  is the air-water gas concentration gradient. Δ[G] was computed from the measured dissolved concentrations of  $CO_2$ ,  $CH_4$ , and  $N_2O$  and from dissolved concentrations at equilibrium with the partial pressure in air of the respective gases using the Henry constants computed from water temperature using the algorithms of [Weiss \(1974\)](#page-13-0) for  $CO_2$ , [Yamamoto et al. \(1976\)](#page-14-0) for CH<sub>4</sub>, and [Weiss and Price \(1980\)](#page-13-0) for N<sub>2</sub>O. The atmospheric  $pCO<sub>2</sub>$  was measured on the field with the Li-Cor Li-840. For CH4, the global average present day atmospheric mixing ratio of 1.9 ppm was used ([Lan](#page-13-0)  [et al., 2024\)](#page-13-0). The monthly global average air mixing ratios of  $N_2O$ provided by the Global Monitoring Division of the NOAA Earth System Research Laboratory [\(Dutton et al., 2017](#page-12-0)) were used. *k* was computed

<span id="page-5-0"></span>from a value normalized to a Schmidt number of 600  $(k_{600})$  and from the Schmidt number of  $CO<sub>2</sub>$ , CH<sub>4</sub> and N<sub>2</sub>O in freshwater computed from water temperature with the algorithms given by [Wanninkhof \(1992\)](#page-13-0).  $k_{600}$  was calculated from the parameterization as a function of wind speed of [Cole and Caraco \(1998\)](#page-12-0) that was chosen for consistency and comparability with two companion papers on GHG emissions from urban ponds in the city of Brussels [\(Bauduin et al., 2024a, 2024b\)](#page-12-0). The *k*600-wind parametrization of [Cole and Caraco \(1998\)](#page-12-0) intrinsically integrates other sources of turbulence such as night-time convection due to cooling and the effect of fetch limitation, because it is derived from a large compilation of repeated tracer measurements in 11 lakes with a wide range of surface area  $(0.2-487 \text{ km}^2)$  and maximum depth  $(1-109m)$ .

#### *2.6. Statistical analysis*

Statistical analyses were carried out with R package version 4.4.1 [\(R](#page-13-0)  [Core Team, 2021\)](#page-13-0). Relationships between variables were evaluated by linear regression, using the coefficient of determination  $(r^2)$  to measure correlation strength and Fisher's F test and the associated *p*-value to



**Fig. 3.** Hourly measurements of water temperature (℃), oxygen saturation level (%O<sub>2</sub>, in %), partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>, in ppm), dissolved CH<sub>4</sub> concentration (CH<sub>4</sub>, in nmol L<sup>−1</sup>), <sup>13</sup>C/<sup>12</sup>C ratio of CH<sub>4</sub> in surface waters (δ<sup>13</sup>C-CH<sub>4</sub>, in ‰) and N<sub>2</sub>O saturation level (%N<sub>2</sub>O, in %) in spring (3rd and 8<sup>th</sup> April 2023), summer (7th and  $13<sup>th</sup>$  September 2023), fall (15th and 16<sup>th</sup> November 2023) and winter (7th and 8<sup>th</sup> January 2024) at the Pêcheries and Silex ponds in Brussels, respectively. The vertical grey bands represent times of day before sunrise and after sunset.

<span id="page-6-0"></span>measure statistical significance. Differences in GHG fluxes between three periods of the day (late night, mid-day, and early night) were tested with Tukey's honestly significant difference post-hoc test following one-way repeated measures analysis of variance (ANOVA).

The abbreviations used in this work are listed and defined in Table S1.

The data supporting this study are available at [https://zenodo.](https://zenodo.org/records/11395489)  [org/records/11395489](https://zenodo.org/records/11395489)

#### **3. Results and discussion**

*3.1. Variations of meteorological conditions and GHG concentrations at sub-daily scale*

The seasonal range of recorded air temperature for all sub-daily samplings at the two urban ponds in the city of Brussels was from −2.8 °C in winter to 30.3 °C in summer (Fig. S1). The amplitude between the minimum and the maximum values of air temperature variations during individual sub-daily samplings was minimal in winter (1.4 ◦C at the Silex pond) and maximal in summer (13.6  $°C$  at the Pêcheries pond). During sub-daily samplings in both ponds, precipitation did not occur in spring and summer, low precipitation occurred in winter, and higher precipitation occurred in fall in particular during the sampling of the Silex pond (Fig. S1). In both ponds, wind speed was generally low ( $\leq$ 4 m  $\rm s^{-1})$  in spring and summer, and higher in fall and winter with maximum values  $> 5$  m s<sup> $-1$ </sup> (except at the Pêcheries pond in winter with wind speed  $<$ 2 m s<sup> $-1$ </sup>). In spring, summer, and fall, wind speed was usually higher at the middle of the day than at dawn and dusk. In winter, wind speed increased regularly from dawn to dusk at the Pêcheries pond and was relatively constant at the Silex pond (Fig. S1).

The seasonal range of water temperature for all sub-daily samplings in both ponds was from 2.5  $\degree$ C in winter to 23.4  $\degree$ C in summer [\(Fig. 3](#page-5-0)). The amplitude of water temperature sub-daily variations was minimal in winter and fall ( $\leq$ 0.6 °C) and maximal in spring (4.4 °C at the Pêcheries pond and 3.1 ℃ at the Silex pond). The turbid-water Pêcheries pond was characterized by high Chl-*a* (19.1  $\pm$  13.7 μg L<sup>−1</sup>) and TSM (13.7  $\pm$  10.7 mg  $L^{-1}$ ) values while the clear-water Silex pond was characterized by low Chl-*a* (1.0  $\pm$  1.2 μg L $^{-1}$ ) and TSM (4.0  $\pm$  3.2 mg L $^{-1}$ ) values ([Fig. 2](#page-3-0) and Table 1). The comparison of average Chl-*a* during the sub-daily samplings with the seasonally resolved higher data-density time series reported by [Bauduin et al. \(2024a\)](#page-12-0) showed that the sampling in September at the Pêcheries pond occurred after the peak of

phytoplankton in August, but that the Chl-*a* values were consistent on the other three occasions, as well on all four occasions at the Silex pond where Chl-*a* values were uniformly low during the year (Fig. S2). The seasonal range of SRP concentrations for all sub-daily samplings in both ponds was from 0.5 to 5.5 µmol L<sup>-1</sup> and concentrations peaked in fall in both ponds (Table 1). The measured SRP concentrations were higher during each season at the Silex pond. The seasonal range of NH $_4^+$  concentrations for all sub-daily samplings in both ponds was from 0.9 to 11.0 μmol  $L^{-1}$ , with maximal values in fall and minimal values in summer in both ponds (Table 1). The seasonal range of  $NO<sub>2</sub>$  concentrations for all sub-daily samplings in both ponds was from 0.1 to 3.3  $\mu$ mol L<sup>-1</sup>, with maximal values in fall in both ponds (Table 1). The seasonal range of  $\mathrm{NO_3^-}$  concentrations for all sub-daily samplings in both ponds was from 6.1 to 28.8  $\mu$ mol L<sup>-1</sup>, with maximal values in fall at the Silex pond and in winter at the Pêcheries pond (Table 1). The TSM, Chl-a, SRP, NH<sub>4</sub>, NO<sub>2</sub>, and NO<sub>3</sub> data collected at sub-daily scale from this study were within the range and the means were comparable to the respective seasonal data from [Bauduin et al. \(2024a\)](#page-12-0) in the same ponds based on a dense seasonal sampling from June 2021 to December 2023 (once a day from one (winter) to three (summer) times per month) (Table S2).

The seasonal range of  $\%O<sub>2</sub>$  for all sub-daily samplings was from 33% in fall to 106% in summer, at the Pêcheries pond  $(Fig. 3)$ . The amplitude of %O<sub>2</sub> sub-daily variations was minimal in winter ( $\leq$ 7% at both ponds) and highest in spring (45% at the Pêcheries pond and 32% at the Silex pond). In spring and summer,  $\%O_{2}$  followed a temporal pattern consistent with diel variations of photosynthesis with a gradual increase from dawn to mid-afternoon followed by a gradual decrease until dusk ([Fig. 3\)](#page-5-0). In fall and winter,  $\%O<sub>2</sub>$  decreased during the course of the sampling at both ponds, expect in fall at the Pêcheries pond when  $\%O<sub>2</sub>$ variations were erratic.

The seasonal range of  $pCO<sub>2</sub>$  for all sub-daily samplings was from 1026 ppm in winter to 8417 ppm in fall at the Pêcheries pond in both cases [\(Fig. 3](#page-5-0)). The  $pCO<sub>2</sub>$  values were systematically above atmospheric equilibrium  $(\sim 400 \text{ ppm})$  showing that both ponds acted as sources of  $CO<sub>2</sub>$  to the atmosphere whatever the time of the day and whatever the season. The amplitude of  $pCO<sub>2</sub>$  sub-daily variations was minimal in winter (≤120 ppm at both ponds) and highest in spring (2373 ppm at the Pêcheries pond and 440 ppm at the Silex pond). The amplitude of subdaily variations of  $pCO<sub>2</sub>$  was highest in spring probably due to more favourable light conditions for photosynthesis that led to the onset of the spring phytoplankton bloom in the Pêcheries pond and the growth of

#### **Table 1**

Daily mean  $\pm$  standard deviation of soluble reactive phosphorus concentration (SRP, in µmol L $^{-1}$ ), ammonium concentration (NH $^+_4$ , in µmol L $^{-1}$ ), nitrite concentration (NO<sub>2</sub>, in µmol L<sup>-1</sup>), nitrate concentration (NO<sub>3</sub>, in µmol L<sup>-1</sup>), chlorophyll-a concentration (Chl-a, in µg L<sup>-1</sup>), and total suspended matter concentration (TSM, in mg  $L^{-1}$ ) measured at the Pêcheries and Silex ponds in the city of Brussels for the four seasons, from samples collected three times a day (in the morning, noon, and end of the day), and daily integrated solar irradiance (in W m<sup>-2</sup>). Table S2 provides a comparison of the means and minimum and maximum range of sub-daily measurements from this study and seasonal data from [Bauduin et al. \(2024a\)](#page-12-0) based on sampling once a day from one (winter) to three (summer) times per month from June 2021 to December 2023.

Season	Date	Pond	$SRP$ ( $\mu$ mol $L^{-1}$ )	$NH4+$ (µmol $L^{-1}$ )	NO <sub>2</sub> (µmol) $L^{-1}$ )	$NO_3^-$ (µmol $L^{-1}$ )	Chl- $a$ ( $\mu$ g $L^{-1}$ )	TSM (mg) $L^{-1}$ )	Daily solar irradiance (W $m^{-2}$
Spring 2023	$03-04-$ 23	Pêcheries	$0.5 \pm 0.1$	$4.9 \pm 0.6$	$0.2 \pm 0.0$	$10.4 \pm 3.0$	$5.5 \pm 0.3$	$4.3 \pm 0.9$	2471
	$08-04-$ 23	Silex	$1.0\pm0.2$	$2.7 \pm 0.5$	$0.1\pm0.0$	$7.3 \pm 0.8$	$0.9 \pm 0.1$	$1.6 \pm 0.3$	2344
Summer 2023	$07-09-$ 23	Pêcheries	$1.6 \pm 0.3$	$0.9 \pm 0.5$	$0.3\pm0.0$	$6.1 \pm 0.7$	$14.5 \pm 1.1$	$27.1 \pm 2.1$	3403
	13-09- 23	Silex	$3.8 \pm 0.5$	$2.7 \pm 0.2$	$0.8 \pm 0.3$	$10.7 \pm 0.2$	$1.2 \pm 0.1$	$2.9 \pm 0.5$	2651
Fall 2023	$15-11-$ 23	Pêcheries	$1.6 \pm 0.2$	$2.7 \pm 0.2$	$0.7 \pm 0.1$	$7.6 \pm 0.8$	$14.8 \pm 1.5$	$19.0 \pm 1.5$	482
	$16-11-$ 23	Silex	$5.5 \pm 0.2$	$11.0 \pm 0.7$	$3.3 \pm 0.5$	$28.8 \pm 2.5$	$6.9 \pm 0.9$	$9.2 \pm 1.9$	509
Winter 2024	$07-01-$ 24	Pêcheries	$0.7 \pm 0.2$	$5.9 \pm 0.3$	$0.3\pm0.0$	$15.9 \pm 0.8$	$2.7 \pm 0.7$	$13.4 \pm 2.4$	403
	$08-01-$ 24	Silex	$0.8\pm0.2$	$5.6 \pm 0.3$	$1.6 \pm 0.2$	$19.4 \pm 0.9$	$0.6\pm0.1$	$8.6 \pm 1.4$	456

<span id="page-7-0"></span>macrophytes in the Silex pond. Indeed, daily solar irradiance values in spring were higher than in fall and winter but lower than in summer ([Table 1\)](#page-6-0). The sub-daily sampling in summer in the Pêcheries pond was carried out after the phytoplankton peak in August (Fig. S2). Consequently, despite higher daily solar irradiance values in summer than spring ([Table 1](#page-6-0)), the amplitude of sub-daily variations of  $pCO<sub>2</sub>$  (351) ppm) in the Pêcheries pond was lower in summer during senescence of the phytoplankton bloom than in spring (2373 ppm) during presumably the onset of the spring phytoplankton bloom. In spring and summer, pCO2 followed a temporal pattern consistent with diel variations of photosynthesis as function of light with a gradual decrease from dawn to mid-afternoon followed by a gradual increase until dusk. This was consistent with  $%$ O<sub>2</sub> variations mirroring the pCO<sub>2</sub> variations in spring and summer ([Fig. 3\)](#page-5-0). The daily photosynthesis cycle is presumably due to phytoplankton in the Pêcheries pond and macrophytes in the Silex pond. In fall at the Pêcheries pond, pCO<sub>2</sub> also followed the cycle expected from daily variations of photosynthesis, but not at the Silex pond where  $pCO_2$  showed erratic variations. In winter,  $pCO_2$  increased and decreased during day-time at the Pêcheries pond and at the Silex pond, respectively ([Fig. 3](#page-5-0)), but these changes were marginal compared to other seasons, with amplitudes of  $pCO<sub>2</sub>$  sub-daily variations of 121 ppm at the Silex pond and 93 ppm at the Pêcheries pond.

The seasonal range of CH<sub>4</sub> dissolved concentration for all sub-daily samplings in both ponds was from 391 nmol  $L^{-1}$  in winter at the Pêcheries pond to 16,736 nmol L<sup>-1</sup> in summer at the Silex pond ([Fig. 3](#page-5-0)). The CH4 dissolved concentration values were systematically above atmospheric equilibrium (2–3 nmol  $\text{L}^{-1}$ ) showing that both ponds acted as sources of CH4 to the atmosphere whatever the time of the day and whatever the season. During all seasons, the CH<sub>4</sub> dissolved concentrations were higher at the Silex pond than the Pêcheries pond (Table S2) in agreement with the previously reported pattern across 22 urban ponds in the city of Brussels showing higher  $CH<sub>4</sub>$  emissions in clear-water ponds than turbid-water ponds ([Bauduin et al. 2024a,](#page-12-0) [2024b](#page-12-0)). The amplitude of sub-daily variations of CH<sub>4</sub> dissolved concentration was minimal in winter ( $\leq$ 103 nmol L<sup>-1</sup> in both ponds) and highest in summer (7186 nmol L<sup>-1</sup> at the Pêcheries pond and 10,774 nmol L<sup>-1</sup> at the Silex pond). Sub-daily variations in CH4 dissolved concentrations did not correlate with the sub-daily variations in water temperature. The amplitude of water temperature variations at sub-daily scale (0.5 ◦C in winter to 2.6 ℃ in summer) were small compared to the seasonal scale (23.5  $\degree$ C) for which a dependence between CH<sub>4</sub> concentrations and temperature was reported in these two ponds [\(Bauduin et al., 2024a](#page-12-0)). For most samplings, the  $CH<sub>4</sub>$  dissolved concentrations did not show a discernible temporal pattern during day-time and variations were erratic (with a few exceptions, see hereafter). These erratic temporal patterns of CH4 dissolved concentrations were mirrored by variations of



**Fig. 4.** Hourly measurements of <sup>13</sup>C/<sup>12</sup>C ratio of CH<sub>4</sub> in surface waters ( $\delta$ <sup>13</sup>C-CH<sub>4</sub>, in ‰) versus dissolved CH<sub>4</sub> concentration (CH<sub>4</sub>, in nmol  $L^{-1}$ ) in spring (3rd and 8<sup>th</sup> April 2023), summer (7th and 13<sup>th</sup> September 2023), fall (15th and  $16<sup>th</sup>$  November 2023) and winter (7th and  $8<sup>th</sup>$  January 2024) at the Pêcheries (Pech) and Silex (Slx) ponds in Brussels, respectively.

 $\delta^{13}$ C-CH<sub>4</sub> [\(Figs. 3 and 4\)](#page-5-0). This suggests that CH<sub>4</sub> dissolved concentration and  $\delta^{13}$ C-CH<sub>4</sub> resulted from spurious changes in balance of the removal of CH4 in the water column by MOX and degassing to the atmosphere and inputs of  $CH_4$  fluxing out of the sediment. The  $CH_4$  fluxing out of the sediment is itself also partly function of MOX at the sediment-water interface and in the superficial oxic sediment layer, and of methanogenesis deeper in the sediment.

At the Pêcheries pond, in spring, CH<sub>4</sub> dissolved concentrations showed a regular pattern, with an increase of values during day-time that could be related to an increase of light limitation of MOX ([Dumestre et al., 1999;](#page-12-0) [Murase and Sugimoto, 2005;](#page-13-0) [Morana et al.,](#page-13-0)  [2020\)](#page-13-0). The intensity and amplitude of variations of light during the day were highest in spring and summer, which also corresponded to the largest variations in  $\delta^{13}$ C-CH<sub>4</sub> observed in both ponds (4.8 ‰ and 1.4 ‰ in spring, and 4.3 ‰ and 6.8 ‰ in summer, compared to 0.3 ‰ and 0.8 ‰ in fall, and 0.6 ‰ and 0.2 ‰ in winter at the Pêcheries pond and the Silex pond, respectively).

The relatively strong decrease of CH<sub>4</sub> dissolved concentration observed between morning and afternoon at the Silex pond in spring (decrease of 2169 nmol L<sup>-1</sup>) and summer (decrease of 10,774 nmol L<sup>-1</sup>) occurred as  $\%O_2$  increased in the water column. Higher  $O_2$  in the water column should lead to a higher diffusion of  $O<sub>2</sub>$  into the top layer of the sediment enhancing MOX and decreasing the CH4 fluxing from the sediment, resulting in a lowering of CH4 dissolved concentration in surface waters.

In summer at the Pêcheries pond, CH<sub>4</sub> dissolved concentrations were relatively uniform from dawn to 17:30 (950–1798 nmol L-1) and at 18:30 suddenly increased by 1555 nmol  $L^{-1}$ , and increased again at 22:30 by 7000 nmol  $L^{-1}$ . These two sharp increases of CH<sub>4</sub> dissolved concentrations coincided with decreases of  $\delta^{13}$ C-CH<sub>4</sub> and, more surprisingly, with equally sharp increase in  $\%N_2O$  ([Fig. 3\)](#page-5-0). Water temperature was comparatively high during this sampling (up to 23.5 ◦C), favourable to ebullition from the sediments that was shown to be positively related to water temperature in shallow ponds including urban ponds of the city of Brussels ([Bauduin et al., 2024a](#page-12-0)). The strong temperature increase during day-time, as well as a decrease of atmospheric pressure (Fig. S1), were probably very favourable to ebullition at the end of the day during the sampling in summer at the Pêcheries pond. The high temperature was probably more important in triggering ebullition than change in atmospheric pressure that seems to be important in driving ebullition only at low temperatures in these ponds [\(Bauduin](#page-12-0)  [et al., 2024a\)](#page-12-0). Additionally, the drop in atmospheric pressure was modest compared to other sampling periods when more important changes in atmospheric pressure were observed but without a sharp increase of CH<sub>4</sub> dissolved concentrations (Fig.  $S1$ ). Ebullition could have directly increased CH4 dissolved concentrations in the water column through dissolution of rising bubbles (McGinnis et al., 2016). Additionally, it is hypothesized that the physical disruption of the sediments by bubbling also contributed to mix pore water rich in  $CH<sub>4</sub>$  and  $N<sub>2</sub>O$ contributing to the observed peaks of these two gases (e.g. [Liikanen](#page-13-0)  [et al., 2002\)](#page-13-0).

In both ponds, the variations of CH<sub>4</sub> dissolved concentration in surface waters during the course of the day in fall and winter did not follow the pattern observed in summer and spring, possibly due to lower methanogenesis related to lower temperatures and lower fluxing of CH4 from sediments, as well as a lower increase of  $O<sub>2</sub>$  in the water column during day-time, and lower light intensities susceptible of inducing light limitation of MOX. The decrease of CH4 dissolved concentration observed between morning and afternoon at the Pêcheries pond in fall (decrease of 520 nmol  $L^{-1}$ ) and at the Silex pond in winter (decrease of 103 nmol L<sup>-1</sup>) occurred as %O<sub>2</sub> decreased as well in the water column. These variations of CH4 dissolved concentration could have been erratic as it was the case in other occasions (the Silex pond in fall and the Pêcheries pond in winter).

The seasonal range of  $\%N_2O$  variations for all sub-daily samplings was from 73.6% in spring at the Pêcheries pond to 182.0% in summer at the Silex pond ([Fig. 3\)](#page-5-0). %N<sub>2</sub>O oscillated around saturation (100%) showing these ponds acted alternatively as sinks or sources of  $N_2O$ . The amplitude of %N<sub>2</sub>O sub-daily variations was minimal in winter ( $\leq$ 8.0 %) in both ponds) and maximal in summer (81.2% at the Pêcheries pond and 34.9% at the Silex pond). The very high daily amplitude in summer at the Pêcheries pond (81.2%) could be related to high inputs of  $N_2O$ from sediments putatively due to an ebullition event (see above) that caused physical disruption of the sediments, leading to the mixing of pore-waters with bottom pond waters, promoting the transport of DIN and  $N_2O$  to the upper part of the water column (e.g. Liikanen et al.,  $2002$ ). The %N<sub>2</sub>O levels were generally close to saturation, with notable deviations during summer and fall at the Silex pond, when higher %N<sub>2</sub>O values were observed. These higher %N<sub>2</sub>O levels coincided with periods of high NH $_4^+$  concentrations at the Silex pond (Table S2) and could have resulted from enhanced nitrification as suggested by equally high  $NO<sub>2</sub>$ values. In spring at both ponds and in summer at the Pêcheries pond (excluding the two extreme peaks of %N2O), %N2O increased during the day in parallel to %O<sub>2</sub>, and %N<sub>2</sub>O was positively correlated to %O<sub>2</sub> (Fig. S3). This pattern is unexpected as  $N_2O$  is usually negatively related to  $O_2$  in aquatic environments rich in DIN (e.g. [Rosamond et al., 2012](#page-13-0)), although in DIN poor environments a positive relation can emerge, most probably related to  $N_2O$  removal by denitrification (e.g. Borges et al., [2018,](#page-12-0) [2019,](#page-12-0) [2022,](#page-12-0) [2023\)](#page-12-0), which is unlikely in the DIN rich eutrophic sampled ponds. A possible explanation of the positive relationship between %N<sub>2</sub>O and %O<sub>2</sub> is that an increase of O<sub>2</sub> in the water column led to a transfer of  $O_2$  to sediments and enhanced nitrification leading to an increase of  $N_2O$  production. Alternatively,  $N_2O$  could have been produced by microalgae themselves ([Weathers, 1984](#page-13-0); [Guieysse et al., 2013](#page-12-0); [Burlacot et al., 2020;](#page-12-0) [Plouviez et al., 2017,](#page-13-0) [2019,](#page-13-0) [2020;](#page-13-0) [Fabisik et al.,](#page-12-0)  [2023\)](#page-12-0). This could explain the day-time increase of  $\%$ N<sub>2</sub>O in summer and spring at the Pêcheries pond where Chl-*a* was more abundant [\(Table 1\)](#page-6-0) than at the Silex pond. At the Silex pond, phytoplankton was less abundant ([Table 1\)](#page-6-0) and the macrophyte *Lemna trisulca* was dominant. Yet, higher plants can also produce  $N_2O$  [\(Lenhart et al., 2019](#page-13-0)), so the daytime increase of %N2O in Silex in spring could in theory be also linked to primary production leading to a parallel increase with  $\%O_{2}$ .

In summer and fall at the Silex pond, a negative relationship between  $%N<sub>2</sub>O$  and  $%O<sub>2</sub>$  was observed (Fig. S3), a common feature in aquatic environments rich in DIN (e.g. [Rosamond et al., 2012\)](#page-13-0) including rivers in Belgium ([Borges et al., 2018\)](#page-12-0). The negative relationship between %  $N_2O$  and % $O_2$  probably reflected an increase in the yield of  $N_2O$  production by nitrification at lower  $O_2$  levels ([Goreau et al., 1980;](#page-12-0) Ni et al.,  $2011$ ). Additionally, low %O<sub>2</sub> conditions reflect higher organic matter degradation at community level, hence, also  $\mathrm{NH}_4^+$  production by ammonification, leading to higher rates of nitrification and  $N_2O$  production. Indeed, the highest daily %N<sub>2</sub>O values observed in summer and fall at the Silex pond coincided with higher NH<sup> $+$ </sup> and NO<sub>2</sub> concentrations [\(Table 1\)](#page-6-0).

In fall at the Silex pond and in winter at both the Pêcheries and Silex ponds, there were clear sub-daily variations in  $\%$ N<sub>2</sub>O that were uncorrelated to %O<sub>2</sub> variations. %O<sub>2</sub> decreased during day-time, but %N<sub>2</sub>O either showed erratic changes (the Silex pond in winter), a decreasing tendency (the Pêcheries pond in winter), or an increasing tendency (the Silex pond in fall). There was no clear explanation for these divergent patterns and why they differed from those described above for spring and summer.

The %O<sub>2</sub>,  $pCO_2$ , CH<sub>4</sub>, and %N<sub>2</sub>O data collected at sub-daily scale from this study were within the range and the means were comparable to the seasonal data from [Bauduin et al. \(2024a\)](#page-12-0) in the same ponds based on a dense seasonal sampling from June 2021 to December 2023 (once a day from one (winter) to three (summer) times per month) (Table S2).

#### *3.2. Variations of CO2, CH4 and N2O fluxes at sub-daily scale*

The air-water diffusive  $CO_2$  fluxes ( $F_{CO2}$ ) at the Pêcheries pond for all sub-daily samplings ranged from 10 to 671 mmol  $m^{-2} d^{-1}$  in winter and in fall, respectively (Fig. S4). The amplitude of sub-daily variations of  $F_{CO2}$  at the Pêcheries pond was minimal in winter (14 mmol m<sup>-2</sup> d<sup>-1</sup>) and maximal in fall (496 mmol m<sup>-2</sup> d<sup>-1</sup>). The  $F_{CO2}$  fluxes at the Silex pond for all sub-daily samplings ranged from 50 to 467 mmol  $m^{-2} d^{-1}$  in summer and fall, respectively. The amplitude of sub-daily variation of  $F_{\text{CO2}}$  at the Silex pond was minimal in summer (39 mmol m<sup>-2</sup> d<sup>-1</sup>) and maximal in fall (274 mmol m<sup>-2</sup> d<sup>-1</sup>). The air-water diffusive CH<sub>4</sub> fluxes ( $F<sub>CH4</sub>$ ) at the Pêcheries pond for all sub-daily samplings ranged from 130 to 4847 µmol m<sup> $-2$ </sup> d<sup>-1</sup> in winter and summer, respectively. The amplitude of sub-daily variation of  $F_{\text{CH4}}$  at the Pêcheries pond was minimal in winter (59 µmol m<sup>-2</sup> d<sup>-1</sup>) and maximal in fall (3076 µmol m<sup>-2</sup> d<sup>-1</sup>) (Fig. S4). The  $F_{CH4}$  at the Silex pond for all sub-daily samplings ranged from 348 to 9334 µmol m<sup>-2</sup> d<sup>-1</sup> in winter and summer, respectively. The amplitude of sub-daily variation of  $F_{CH4}$  at the Silex pond was minimal in winter (497 μmol m<sup>-2</sup> d<sup>-1</sup>) and maximal in summer (6291 μmol m<sup>-2</sup> d<sup>-1</sup>). The air-water diffusive N<sub>2</sub>O fluxes (*F*<sub>N2O</sub>) in the two ponds were sometimes positive (source of  $N_2O$  to the atmosphere) and sometimes negative (sink of atmospheric N<sub>2</sub>O), given that  $%N<sub>2</sub>O$  oscillated around saturation (100%) [\(Fig. 3,](#page-5-0) S4). The  $F_{N2O}$  at the Pêcheries pond for all sub-daily samplings ranged from  $-1.3$  to 0.5 µmol m<sup>-2</sup> d<sup>-1</sup> in spring and summer, respectively. The amplitude of variation of  $F_{N2O}$  at the Pêcheries pond was minimal in winter (0.2 µmol  $m^{-2}$  d<sup>-1</sup>) and maximal in spring (1.3 µmol m<sup>-2</sup> d<sup>-1</sup>) (Fig. S4). The  $F_{\rm N2O}$  at the Silex pond for all sub-daily samplings ranged from  $-1.8$  to 8.1 µmol  $m^{-2} d^{-1}$  in spring and fall, respectively. The amplitude of variation of  $F_{N2O}$  at the Silex pond was minimal in winter (0.8 μmol m<sup>-2</sup> d<sup>-1</sup>) and maximal in fall (3.9 μmol  $m^{-2} d^{-1}$ ).

The  $F_{\text{CO2}}$ ,  $F_{\text{CH4}}$ , and  $F_{\text{N2O}}$  were calculated from *k* with a parameterization as a function of wind speed ([Cole and Caraco, 1998\)](#page-12-0) and from the air-water gradient of the respective GHG concentration. Both wind speed and GHG concentrations can vary at daily scale leading potentially to differences in the intensity of flux during day-time and during night-time [\(Fig. 1](#page-1-0)). To investigate sources of sub-daily variability of  $F_{\text{CO2}}$ ,  $F_{\text{CH4}}$ , and  $F_{\text{N2O}}$ , the data were averaged over 3 periods each day: end of the night and early morning; mid-morning to mid-afternoon, late after-noon and early night (Figs. S5 and S6).

At the Pêcheries pond, the wind speed was higher in the middle of the day in spring, summer and fall, and increased from late night to early night in winter (Fig.  $S_5$ ). The  $F_{CO2}$  at the Pêcheries pond were higher in the middle of the day than during the rest of the day in summer and fall, following the pattern of the wind speed. In spring, the  $F_{C_0/2}$  at the Pêcheries pond were higher at the start of the day and decreased over the course of the day, as  $pCO_2$ . In winter, the  $F_{CO2}$  at the Pêcheries pond increased throughout the day, as wind speed and  $pCO<sub>2</sub>$  (Fig. S5). The  $F<sub>CH4</sub>$  at the Pêcheries pond followed concentration changes in spring, and wind speed changes in summer, fall and winter (Fig. S5).

At the Silex pond, wind speed was higher in the middle of the day in spring and fall (Fig. S5). In summer, the wind was higher at the beginning of the day than at the end, and in winter, the wind was nearly constant throughout the day, at the Silex pond ( $Fig. S5$ ). The  $F_{CO2}$  at the Silex pond followed wind speed variations in each season. The  $F_{\text{CH4}}$  at the Silex pond followed wind speed variations in spring, fall and winter. In summer, the  $F_{CH4}$  at the Silex pond were highest in the middle of the day and followed the  $CH_4$  concentrations (Fig. S5).

In both ponds, N2O fluxes sometimes followed wind speed (winter at the Pêcheries pond), sometimes changes in concentrations (summer at the Pêcheries pond), sometimes wind speed and concentrations (fall at the Silex pond). In most cases, the sub-daily variations of  $F_{\text{N2O}}$  did not show clear patterns for the three periods of the day (Fig. S5).

#### *3.3. Comparison of sub-daily variability to seasonal and spatial variability of GHG concentrations and fluxes*

The coefficient of variation (CV, in %) was used to compare the subdaily variability of the GHG concentrations and fluxes with the variability at seasonal scale within the same pond (hereafter "intra-pond" CV) and the variability among several ponds in the city of Brussels (hereafter "inter-pond" CV). The intra-pond seasonal CV was estimated from a data-set acquired in the Pêcheries and Silex ponds with a sampling from one (winter) to three (summer) times per month in 2021, 2022 and 2023 [\(Bauduin et al., 2024a](#page-12-0)). The inter-pond CV was estimated from a data-set acquired in 22 ponds sampled only once per season, during the four seasons, in 2021–2022 [\(Bauduin et al., 2024b](#page-12-0)).

The sub-daily CV of  $pCO<sub>2</sub>$  ranged between 2 and 32% in the Pêcheries pond and between 0.9 and 4.5% in the Silex pond (Fig. 5). The highest sub-daily CV of  $pCO<sub>2</sub>$  (32%) was observed in the Pêcheries pond in spring presumably due to intense phytoplankton photosynthesis during the spring bloom. In both ponds, the sub-daily variability of  $pCO<sub>2</sub>$ was lower than the intra-pond seasonal CV (45–67% in the Pêcheries pond and 22–47% in the Silex pond) and the inter-pond CV (55–90%).

The sub-daily CV of CH4 dissolved concentration ranged between 2 and 101% in the Pêcheries pond and between 4 and 33% in the Silex pond (Fig. 5). The highest sub-daily CV of  $CH_4$  dissolved concentration (101%) was observed in the Pêcheries pond in summer presumably due to an ebullition event (see above). In both ponds, the sub-daily variability of CH4 dissolved concentration was lower than the intra-pond seasonal CV (45–100% in the Pêcheries pond and 46–75% in the Silex pond) and the inter-pond CV (67–156%).

The sub-daily CV of  $\%N_2O$  ranged between 1 and 13% in the Pêcheries pond and between 2 and 7% in the Silex pond (Fig. 5). In both ponds, the sub-daily variability of %N2O was lower than the intra-pond

seasonal CV (22–35% in the Pêcheries pond and 19–74% in the Silex pond) and the inter-pond CV (52–332%).

The sub-daily, intra-pond seasonal, and inter-pond variability (CV) of diffusive  $F_{\text{CO2}}$  and  $F_{\text{CH4}}$  showed the same patterns and were numerically close to those of  $pCO<sub>2</sub>$  and CH<sub>4</sub> dissolved concentration (Figs. 5 and 6). The sub-daily CV of  $F_{\text{N2O}}$  ranged between 28 and 189% in the Pêcheries pond and between 21 and 58% in the Silex pond [\(Fig. 6](#page-10-0)), and was substantially higher than the respective CV of  $\%$ N<sub>2</sub>O ([Fig. 6](#page-10-0)). In both ponds, the sub-daily variability of  $F_{N2O}$  was lower than the intra-pond seasonal CV (74–1492% in the Pêcheries pond and 150–252% in the Silex pond) and the inter-pond CV (115–389%). The studied ponds oscillated seasonally and among different ponds between a sink and source of N<sub>2</sub>O [\(Bauduin et al. 2024a](#page-12-0), [2024b\)](#page-12-0), so the average was close to zero, leading to a numerical increase of CV of  $F_{\text{N2O}}$  compared to the respective CV of %N<sub>2</sub>O. The CV of  $F_{CO2}$ ,  $F_{CH4}$ , and  $F_{N2O}$  computed with a *k* derived from a constant average wind speed  $(1 \text{ m s}^{-1})$  (Fig. S7) exhibited similar patterns and were numerically comparable to those calculated using actual wind speed data ([Fig. 6\)](#page-10-0). This showed that the variability (CV) of  $F_{CO2}$ ,  $F_{CH4}$ , and  $F_{N2O}$  was mainly driven by changes in  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O concentrations rather than changes in computed *k* from a parameterization function of wind speed. Indeed, in the studied ponds, variations of wind speed are modest at sub-daily (Fig. S1) and seasonal ([Bauduin et al. 2024a](#page-12-0), [2024b](#page-12-0)) scales.

For the three GHGs, whether concentrations or fluxes, the variability (CV) tended to be higher in spring-summer than fall-winter at all scales



Fig. 5. Coefficients of variation of data collected from June 2021 to December 2023 at Pêcheries (Pech) and Silex (Slx) ponds (intra-pond seasonal CV, in %) from [Bauduin et al. \(2024a\)](#page-12-0) and coefficients of variation of data collected on 22 ponds in Brussels at the four seasons in 2021 and 2022 (inter-pond seasonal CV, in %) from [Bauduin et al. \(2024b\)](#page-12-0) versus coefficients of variation of sub-daily data at the Pêcheries and Silex ponds from this study (sub-daily CV, in %) for partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>, in ppm), dissolved CH<sub>4</sub> concentration (CH<sub>4</sub>, in nmol L<sup>-1</sup>), and N<sub>2</sub>O saturation level (%N<sub>2</sub>O, in %). Dotted line indicates the 1:1 line.

<span id="page-10-0"></span>

Fig. 6. Coefficients of variation of data collected from June 2021 to December 2023 at Pêcheries (Pech) and Silex (Slx) ponds (intra-pond seasonal CV, in %) from [Bauduin et al. \(2024a\)](#page-12-0) and coefficient of variation of data collected on 22 ponds in Brussels at the four seasons in 2021 and 2022 (inter-pond seasonal CV, in %) from [Bauduin et al. \(2024b\)](#page-12-0) versus coefficients of variation of sub-daily data at the Pêcheries and Silex ponds from this study (sub-daily CV, in %) for diffusive fluxes of  $CO_2$  ( $F_{CO2}$ , in mmol m<sup>-2</sup> d<sup>-1</sup>), CH<sub>4</sub> ( $F_{CH4}$ , in µmol m<sup>-2</sup> d<sup>-1</sup>) and N<sub>2</sub>O ( $F_{N2O}$ , in µmol m<sup>-2</sup> d<sup>-1</sup>), computed from hourly wind speed measurements; Fig. S7 reports  $F_{CO2}$ ,  $F_{\text{CH4}}$ , and  $F_{\text{N2O}}$  computed with a constant wind speed of 1 m s<sup>-1</sup>. Dotted line indicates the 1:1 line.

(sub-daily, seasonal intra-pond, and inter-pond). Indeed, the CV of the GHG concentrations was in most cases positively related to water temperature (Fig. S8), although less clear for fluxes due to the additional variability from wind speed (Fig. S9). In spring and summer, higher temperatures increase all aquatic metabolic processes, and, in addition, higher primary production in response to better light conditions directly affects  $CO<sub>2</sub>$  levels and cascades to enhance microbial processes that also affect  $CO_2$  (respiration), as well as CH<sub>4</sub> (methanogenesis), and N<sub>2</sub>O (nitrification and denitrification).

#### *3.4. Comparison of GHG sub-daily variations with other lakes and ponds*

There is only a limited number of equivalent studies of the sub-daily variability of  $CO_2$ , CH<sub>4</sub>, and N<sub>2</sub>O emissions from lakes and ponds (Zhang [et al., 2018](#page-14-0), [2023; Xiao et al., 2019;](#page-14-0) [Martinez-Cruz et al., 2020; Sieczko](#page-13-0)  [et al., 2020](#page-13-0); [Rudberg et al., 2021;](#page-13-0) [Borges et al., 2022](#page-12-0); [Ni et al., 2022](#page-13-0); S[ø](#page-13-0)  [et al., 2023](#page-13-0), [2024](#page-13-0); [Shen et al., 2024](#page-13-0); [Smith et al., 2024;](#page-13-0) [Wang et al.,](#page-13-0)  [2024a,](#page-13-0) [2024b\)](#page-12-0), and, none reporting simultaneous measurements of the three GHGs. In general, previous studies showed that during periods of active aquatic primary production, the  $CO<sub>2</sub>$  concentration followed the

light cycle with decreasing values during day-time and increasing values during night-time, in response to cycles of photosynthesis and respiration ([Borges et al., 2022](#page-12-0); [Wang et al., 2024a](#page-13-0), [2024b;](#page-12-0) [Shen et al., 2024](#page-13-0)). The actual net  $F_{CO2}$  also depended on wind speed that was generally higher during day-time ([Martinez-Cruz et al., 2020](#page-13-0); [Rudberg et al.,](#page-13-0)  [2021;](#page-13-0) [Borges et al., 2022](#page-12-0); Sø [et al., 2023](#page-13-0), [2024\)](#page-13-0). These patterns of CO<sub>2</sub> dissolved concentrations and computed  $F_{CO2}$  were also observed in spring and summer in the Silex and the Pêcheries ponds (Fig.  $3$ , S4). During fall and winter, when aquatic primary production was putatively lower owing to lower temperature and light availability ([Table 1\)](#page-6-0), the patterns of CO<sub>2</sub> dissolved concentrations and computed  $F_{CO2}$  were more erratic and with lower amplitude in the Silex and the Pêcheries ponds ([Fig. 3,](#page-5-0) S4).

In general, previous studies showed that the higher wind speeds during day-time seemed to systematically lead to higher  $F_{CH4}$  from lakes and ponds during day-time [\(Zhang et al., 2018](#page-14-0), [2023;](#page-14-0) [Martinez-Cruz](#page-13-0)  [et al., 2020](#page-13-0); [Sieczko et al., 2020](#page-13-0); Sø [et al., 2023](#page-13-0), [2024\)](#page-13-0). In the Pêcheries and the Silex ponds, in nearly all the occasions, the sub-daily variations of  $F_{\text{CH4}}$  followed those of wind speed that frequently peaked in the middle of the day (Figs. S5 and S6). However, nearly all of the reported

studies on diel variations of  $F_{\text{CH4}}$  from lakes and ponds were based on floating chambers ([Zhang et al., 2018,](#page-14-0) [2023](#page-14-0); [Sieczko et al., 2020](#page-13-0); S[ø](#page-13-0)  [et al., 2023](#page-13-0), [2024\)](#page-13-0). Floating chambers provide the net  $F_{\text{CH4}}$  but do not allow discriminating the individual contribution to the observed variability of *F*<sub>CH4</sub> from changes in CH<sub>4</sub> dissolved concentration and from those of *k*, with the additional complication that floating chambers can potentially also capture the ebullitive component of  $F_{\text{CH4}}$ . There are theoretical grounds for hypothesizing variations of CH4 dissolved concentrations at sub-daily scale in surface waters of ponds and lakes summarized in [Fig. 1](#page-1-0), but they do not seem discernible in observational studies available so far. [Zhang et al. \(2018\)](#page-14-0) observed erratic sub-daily variations of CH4 dissolved concentration in an urban sub-tropical pond in the city of Yichang (Central China). In the Pêcheries and the Silex ponds, the variations of CH4 dissolved concentrations were also mostly erratic in fall and winter or inconsistent among ponds in spring and summer [\(Fig. 4](#page-7-0)). If such a lack of consistent night-day variations in CH4 dissolved concentrations is ubiquitous, it would suggest the prevalence of the variability of wind speed in driving sub-daily variations of  $F<sub>CH4</sub>$  from lakes and ponds.

In nearly all occasions, N<sub>2</sub>O dissolved concentrations showed erratic variations at sub-daily scale in the Pêcheries and the Silex ponds [\(Fig. 4\)](#page-7-0) as also observed in Lake Taihu (China) [\(Xiao et al., 2019](#page-14-0)) and in Ashumet pond (U.S.A.) [\(Smith et al., 2024](#page-13-0)). On a limited number of occasions in the Pêcheries pond (spring and summer) and the Silex pond (spring), the N<sub>2</sub>O dissolved levels peaked with the  $O<sub>2</sub>$  dissolved concentration, as also reported by [Ni et al. \(2022\)](#page-13-0) in Lake Wuliangsuhai (China). However, an unequivocal explanation could not be provided for this pattern. On another limited number of occasions in the Silex pond (summer and fall), the  $N_2O$  levels were negatively correlated to  $O_2$ dissolved concentration. Although such a pattern has not been reported at sub-daily scales in lakes and ponds, a night-time increase of  $N_2O$  in parallel to the lowering of  $O_2$  in the water column has been frequently reported in rivers [\(Harrison et al., 2005](#page-12-0); [Clough et al., 2007](#page-12-0); [Baulch](#page-12-0)  [et al., 2012;](#page-12-0) [Xia et al., 2013;](#page-14-0) [Chen et al., 2021\)](#page-12-0).

#### **4. Conclusions**

Data of  $pCO_2$ , CH<sub>4</sub> concentration, and %N<sub>2</sub>O were collected hourly in two equally shallow (depth  $\sim$ 1m) and small (surface area  $\sim$ 1ha) urban ponds in the city of Brussels, the Pêcheries pond dominated by phytoplankton and the Silex pond dominated by macrophytes, during each of the four seasons in 2023–2024, with the aim to investigate if diurnal variations in the concentrations and fluxes of  $CO_2$ ,  $CH_4$ , and  $N_2O$  in the two ponds were detectable, systematic, and could be explained mechanistically by biogeochemical drivers or meteorological conditions.

pCO2 and %O2 followed expected temporal patterns from dawn to dusk in response to photosynthetic activity in the Pêcheries and the Silex ponds in spring and summer, as well as in fall at the Pêcheries pond. In winter,  $pCO<sub>2</sub>$  sub-daily variations were erratic in both ponds. The patterns of temporal changes from dawn to dusk of  $CH<sub>4</sub>$  dissolved concentration and  $\%$ N<sub>2</sub>O were more variable, were not systematically consistent during the different sampling periods and among the two ponds, and were more difficult to interpret than those of CO<sub>2</sub>. Some of the temporal patterns of CH<sub>4</sub> dissolved concentration and %N<sub>2</sub>O could be interpreted as resulting from the effect of an increase during day-time of O2 diffusion from the water column to the sediments, and the resulting effect on sediment biogeochemical processes (MOX and nitrification, respectively). Some temporal patterns fitted with alternate explanations such as light inhibition of MOX leading to an increase of  $CH<sub>4</sub>$  during day-time or production of N<sub>2</sub>O by microalgae themselves leading to an increase of %N2O during day-time.

The sub-daily variations of *F*<sub>CO2</sub> and *F*<sub>CH4</sub> were in most cases dominated by the variations of wind speed that was frequently but not systematically higher during the middle of the day compared to early morning or late afternoon. In other cases, the sub-daily variations of *F*<sub>CO2</sub> and *F*<sub>CH4</sub> were dominated by the variations of GHG concentrations when their amplitude was more important and/or when the wind speed was relatively invariant. The sub-daily variations of  $F_{N2O}$  were mostly erratic whatever the pond and season.

The major limitation of the present work arose from the manual sampling leading to limited time-series with a high demand in humanpower. The automation of measurements of dissolved concentrations with moored detectors of GHG dissolved concentrations (e.g. [Hunt et al.,](#page-12-0)  [2017;](#page-12-0) [Wang et al., 2024a,b\)](#page-13-0) or of air-water GHG fluxes with automated floating chambers (e.g. [Sieczko et al., 2020;](#page-13-0) [Rudberg et al., 2021](#page-13-0); S[ø](#page-13-0)  [et al., 2023,](#page-13-0) [2024\)](#page-13-0) should allow to provide more extensive time series. More extensive time series of GHG dissolved concentrations and/or fluxes should allow to better understand the drivers of daily variations of CO2 and CH4 emissions from ponds and lakes, however, technology to make equivalent instrumentation is at the moment lacking for  $N_2O$ .

The sub-daily variability, quantified by the CV, in the two ponds during the four seasons was lower for  $pCO<sub>2</sub>$  (CV $<$ 10% except one occasion), highest for CH4 dissolved concentration (CV*<*50% except one occasion), and intermediary for %N2O (CV*<*20%). The sub-daily variability of the three GHG concentrations tended to be more important in spring-summer than fall-winter. In both ponds, the sub-daily variability of the concentrations of the three GHGs was lower than the seasonal variability within individual ponds (CV*>*20%). In both ponds, the subdaily variability of the concentrations of the three GHGs was also lower than the variability among 22 ponds in the city of Brussels (CV $>$ 50%). The same patterns of CV were observed for  $F_{CO2}$ ,  $F_{CH4}$ , and *F*<sub>N2O</sub> than for their respective concentrations, although the CV of the fluxes of  $N_2O$  was higher than for % $N_2O$ . The studied ponds oscillated seasonally and among different ponds between a sink and source of  $N_2O$ , so the average was close to zero, leading to a numerical increase of CV, unlike  $CO<sub>2</sub>$  and CH<sub>4</sub> that were systematically distinctly above saturation.

It is concluded that the variability of  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O concentrations and fluxes either seasonally within a given system or among different ponds seemed more important than the variability at sub-daily scales. This implies that to reduce the uncertainty on the estimate of  $CO<sub>2</sub>$ , CH<sub>4</sub>, and N<sub>2</sub>O emissions from urban ponds a priority should be given to describe inter-system variability, followed by seasonal variability, and then sub-daily variability. Furthermore, the very strong dominance of sub-daily variations of wind speed in driving the diel variability of GHG emissions rather than the sub-daily variations of the dissolved concentrations of GHGs, allows concluding that the timing of the day for a spot concentration measurement when continuous measurements are not possible is of marginal importance. The bias in the computation of the flux can be reduced by correctly accounting for subdaily variability in wind speed, especially for  $CH<sub>4</sub>$  and  $N<sub>2</sub>O$  for which sub-daily variability of dissolved concentrations seemed mostly erratic. Both these conclusions can be used to inform when designing sampling strategies and optimizing the allocation of available resources for a systematic quantification of GHG emissions from water bodies in the context of the environmental management of inland waters, including urban ponds.

#### **CRediT authorship contribution statement**

**Thomas Bauduin:** Writing – original draft, Visualization, Investigation, Formal analysis, Data curation, Conceptualization. **Nathalie Gypens:** Writing – original draft, Validation, Supervision, Resources, Project administration, Methodology, Funding acquisition, Formal analysis. **Alberto V. Borges:** Writing – original draft, Visualization, Supervision, Resources, Project administration, Methodology, Funding acquisition, Formal analysis, Data curation, Conceptualization.

#### **Funding**

TB received funding from Institute for the Encouragement of Scientific Research and Innovation (Innoviris) of the Brussels-Capital Region as part of the Smartwater project (RBC/2020-EPF-6 h) and from the <span id="page-12-0"></span>"Fonds pour la formation à la Recherche dans l'Industrie et dans l'Agriculture" (FRIA, Belgium). The Picarro G2201-i isotopic analyzer was funded by FRS-FNRS (U.N005.21).

#### **Declaration of competing interest**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

#### **Acknowledgements**

Ozan Efe (University of Liège) analysed  $CH_4/N_2O$  concentrations and  $\delta^{13}$ C-CH<sub>4</sub>, Cédric Morana (University of Liège) helped setting up the Picarro G2201-*i* isotopic analyzer, Brussels Environment provided data on pond morphology, five anonymous reviewers provided comments that allowed to improve the original manuscript. AVB is a Research Director at the FRS-FNRS.

#### **Appendix A. Supplementary data**

Supplementary data to this article can be found online at [https://doi.](https://doi.org/10.1016/j.jenvman.2024.123627)  [org/10.1016/j.jenvman.2024.123627](https://doi.org/10.1016/j.jenvman.2024.123627).

#### **Data availability**

The data supporting this study are available at [https://zenodo.](https://zenodo.org/records/11395489)  [org/records/11395489.](https://zenodo.org/records/11395489)

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