1 Preparation of active films with enhanced antioxidant and antibacterial properties

2	by	incorpora	ating	ginger	essential	oil	nanoemulsions	with	xylan	and	PVA	١
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31 Abstract

In this study, active films were successfully prepared using xylan/polyvinyl alcohol (PVA) as the film-forming matrix, combined with ginger essential oil nanoemulsions (GEO-NEs) at varying concentrations (2.0%, 4.0%, 6.0%, and 8.0% w/w). The GEO-NEs, produced via ultrasound, had an average particle size of 176.4±1.2 nm. FTIR and XRD analyses revealed that interactions between GEO-NEs and the film matrix occurred through hydrogen bonding, indicating good compatibility between the components. Incorporating GEO-NEs significantly enhanced the UV shielding performance and mechanical properties of the composite films, achieving mechanical properties comparable to those of commercial packaging materials such as high-density polyethylene (HDPE). The GEO-NEs also boosted the antioxidant and antimicrobial activities of the films, producing inhibition zones against Staphylococcus aureus and Escherichia coli. These results suggest that the composite films have excellent UV shielding, mechanical properties, as well as antioxidant and antibacterial activities, indicating their potential application as active food packaging materials.

47 Keywords

48 Ginger essential oil, Nanoemulsions, Xylan, Polyvinyl alcohol, Active films

61 **1. Introduction**

As conventional plastic production continues to deplete petroleum resources, significant attention has shifted towards novel packaging materials derived from biopolymers such as polysaccharides, proteins, and lipids. These biopolymers offer advantages such as non-toxicity, biodegradability, renewability, and edibility, thereby presenting broad prospects for various applications (Chen et al., 2024; Jurić, Maslov Bandić, Carullo, & Jurić, 2024).

Xylan is a primary component of plant hemicellulose, characterized by a backbone 68 composed of β -(1,4)-linked D-xylopyranosyl residues. It is widely distributed in nature 69 and plentiful in dicots, grasses, and gymnosperms, making it the second most abundant 70 renewable plant polysaccharide on Earth after cellulose (Curry, Peña, & Urbanowicz, 71 72 2023; Ye & Zhong, 2022). However, xylan main chains contain numerous hydroxyl groups, leading to the formation of dense hydrogen bonding structures during the film 73 formation process. This results in defects such as brittleness and hygroscopicity in pure 74 xylan films, which are unsuitable for food packaging (Höije, Gröndahl, Tømmeraas, & 75 76 Gatenholm, 2005; Sárossy et al., 2013). To address these issues, researchers have modified the structure of xylan (Mikkonen et al., 2015; Rao et al., 2021; Šimkovic et 77 al., 2014; Zhang, Li, Qi, & Xiang, 2024) or blended xylan/modified xylan with 78 plasticizers and other polymers to enhance the properties of pure xylan films (Bao et 79 al., 2018; Liu et al., 2019b; Sousa, Ramos, Evtuguin, & Gamelas, 2016; Wang et al., 80 2022; Wang et al., 2021). Polyvinyl alcohol (PVA) is a widely used, non-toxic, water-81 soluble polymer known for its excellent chemical stability, biocompatibility, and 82 biodegradability, making it highly promising for food packaging applications (Oun, 83 84 Shin, Rhim, & Kim, 2022). Previous studies have shown that composite films made by 85 blending xylan with PVA exhibit good compatibility and mechanical properties (Liu et al., 2019b; Wang et al., 2014). However, the antioxidant and antibacterial activities of 86 xylan/PVA films are relatively weak or absent, posing a challenge for extending the 87 shelf life of food products. 88

Ginger essential oil (GEO), derived from the rhizomes of ginger (*Zingiber officinale* 90 *Roscoe*), contains an array of compounds including α-zingiberene, geranial, β-

sesquiphyllandrene, α -curcumene, and β -bisabolene. These compounds impart GEO 91 with its distinctive aroma and confer antimicrobial and antioxidant activities (He et al., 92 2023). Essential oils (EOs) have been classified as Generally Recognized as Safe 93 (GRAS) by the U.S. Food and Drug Administration as natural preservatives (Pandey, 94 Islam, Shams, & Dar, 2022). However, EOs typically exhibit strong taste, high volatility, 95 and low water solubility, which limits their effective application in food packaging 96 (Lakshmayya et al., 2023). An effective strategy to overcome these limitations is to 97 98 encapsulate EOs into oil-in-water nanoemulsions, creating essential oil-based nanoemulsions (EO-NEs). This approach helps mask the strong aroma of EOs and 99 enhances their compatibility with various biopolymer matrixes (Mirsharifi, Sami, 100 Jazaeri, & Rezaei, 2023). EO-NEs can be prepared using high-energy emulsification 101 techniques such high-pressure homogenization, ultrasonication, 102 as and microfluidization, yielding droplets with an average size ranging from 20 to 200 nm 103 (Singh & Pulikkal, 2022). Research indicates that EO-NEs not only effectively preserve 104 the bioactive constituents of EOs but also exhibit superior antibacterial activity (Shi, 105 106 Zhang, Chen, & Wang, 2022). In recent years, researchers have attempted to incorporate EO-NEs into various biopolymer-based films such as gelatin (Li et al., 2020; 107 Sun et al., 2021), chitosan (Rui et al., 2024), sodium alginate (Mutlu, 2023), pectin 108 (Norcino et al., 2020), starch (Fan et al., 2024; Kong et al., 2020; Sanchez, Pinzon, & 109 Villa, 2022), and zein (Li et al., 2022). Furthermore, the results indicate good 110 compatibility between these biopolymers and EO-NEs. 111

To our knowledge, no studies have been reported on xylan/PVA films containing GEO-NEs. The primary objective of this study was to develop active films based on xylan/PVA with enhanced antioxidative and antibacterial properties (Fig. 1). Furthermore, the influence of different concentrations of GEO-NEs on the physicochemical, antioxidative, and antibacterial properties of the films was examined to evaluate the potential application of xylan/PVA-based active films in food preservation.

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121 **2. Materials and methods**

122 *2.1. Materials*

Xylan derived from corn cobs was purchased from Aladdin Reagent Co., Ltd. 123 (Shanghai, China). Polyvinyl alcohol, glycerol, and Tween 80 were obtained from 124 Sinopharm Chemical Reagent Co., Ltd. (Shanghai, China). Ginger essential oil was 125 supplied by Huashuo Spice Oil Co., Ltd. (Jiangxi, China). Mueller Hinton agar was 126 purchased from Weiju Biotechnology Co., Ltd. (Nanjing, China). Escherichia coli 127 ATCC 25922 and Staphylococcus aureus ATCC 6538 were deposited in the College of 128 Food Science and Engineering, Shandong Agricultural University. All other reagents 129 were of analytical grade. 130

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132 2.2. Preparation and characterization of nano-emulsions

According to the method of Li et al. (2023), with slight modifications, the crude emulsion was obtained by mixing a 1% Tween-80 solution with 2 g of ginger essential oil and shearing at $20,000 \times$ g for 5 min using a high-speed shear (T18 digital Ultra-Turrax, IKA, Germany). The nanoemulsion was then produced using an ultrasonic cell crusher (VCX 130, SONICS, USA) with a 10-min ultrasonic crushing time, a 3-sec ultrasonic crushing interval, and an ultrasonic crushing power of 130 W.

The particle size, polydispersity index (PDI), and zeta potential of GEO-NEs were 139 determined at 25°C using a nanoparticle size and potential analyzer (NS-90Z, OMEC, 140 China). To reduce the multiple light scattering effect, the GEO-NEs were diluted 100 141 142 times with deionized water and left for 1 minute before testing (Shi et al., 2022). Additionally, the average particle size, PDI, and zeta potential of GEO-NEs were 143 144 recorded at 7, 14, 21, and 28 d to assess the stability of GEO-NEs. The GEO-NEs were also observed at 40× magnification using an inverted fluorescence microscope (MF53-145 N, MSHOT, China) 146

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148 2.3. Preparation of films

149 The composite films were prepared using the solution casting method, based on the 150 procedure outlined by Wang et al. (2014), with slight modifications.

Briefly, 1.5 g of PVA particles were dissolved in 90 mL of deionized water and stirred 151 at 95°C until the PVA formed a homogeneous solution. Once cooled to 80°C, 0.5 g of 152 xylan was added to the PVA solution and stirred for 30 min. After further cooling to 153 40°C, 20% glycerol (based on the total weight of PVA and xylan) was added and stirred 154 for an additional 30 min to obtain a film-forming solution (FFS). GEO-NEs (at 155 concentrations of 2%, 4%, 6%, and 8% of the total film volume) were added to the FFS 156 and stirred for 20 min at room temperature to obtain active FFS, labeled as GEO-NEs 157 2%, 4%, 6%, and 8%, respectively. After stationary defoaming, 25 g of the resulting 158 active FFS was cast onto polystyrene molds ($10 \text{ cm} \times 10 \text{ cm}$) and dried in an oven at 159 40°C for 6 to 8 h. All films were conditioned at 23°C and 50% RH for 48 h before 160 analysis. 161

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163 2.4. Characterization of active films

164 2.4.1. Attenuated total reflectance–Fourier transform infrared (ATR–FTIR)

165 The ATR-FTIR spectra of the films were recorded using a Nicolet iS10 FTIR 166 spectrometer (Thermo Fisher Scientific, USA). The absorption spectra were collected 167 in the wavenumber range of 4000 to 400 cm⁻¹ with a scan rate of 32 scans and a 168 resolution of 4 cm⁻¹.

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170 2.4.2. Scanning electron microscopy (SEM) observation

The morphology characteristics of the film surface and cross-section were observed using a field emission scanning electron microscope (SEM, JSM-7800F, JEOL Ltd., Tokyo, Japan). To observe the cross-section, the film was immersed in liquid nitrogen for 5 min and then fractured. All samples were sputter-coated with gold for 50 sec at a current of 15 mA to make them conductive, followed by observation at magnifications of 500× and 1000×.

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178 2.4.3. X-ray diffraction test (XRD)

The crystallinity of the films was evaluated using an X-ray diffractometer (D8
ADVANCE, Bruker, Germany). Samples were scanned at diffraction angles (20)

181 between 5° and 80° at a scanning rate of $5^{\circ}/\text{min}$.

182

183 *2.4.4. Optical properties*

The white standard plate (L*=92.7, a*=1.01, b*=0.64) was used as the background, and the L (lightness), a (redness), and b (yellowness) values were determined by using a colorimeter (CR-400, Konica Minolta Co., Ltd., Japan) to take samples from six random positions of each film. The total color difference (ΔE) was calculated by the following equation:

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$$\Delta E = \sqrt{(L^* - L)^2 + (a^* - a)^2 + (b^* - b)^2}$$

where L*, a*, b* are the color values of the white standard plate, L, a, b were the color
values of film samples.

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193 The UV-visible transmittance of the films was recorded using a UV-visible 194 spectrophotometer (UV-2600i, Shimadzu, Japan) in the wavelength range of 200-800 195 nm. The opacity of the films was determined according to the following equation:

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$$Opacity = \frac{A_{600}}{X}$$

197 where A_{600} is the absorbance at 600 nm and X is the film thickness (mm).

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199 2.4.5. Thickness and mechanical properties

Film thickness was measured using a digital micrometer caliper (Sanfeng Precision Meter Co., Ltd., Shanghai, China) at twelve random positions on the film. The tensile strength (TS) and elongation at break (EB) of the films were determined using an auto tensile tester (Labthink Instruments Co. Ltd., Jinan, China) based on a previous method by Qin, Yang, Zhu, and Wei (2022). Before analysis, films were cut into strips measuring 150 mm × 15 mm. The initial clamping distance and the velocity were set at 100 mm and 100 mm/min, respectively.

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208 2.4.6. Water contact angle (WCA)

A contact angle goniometer (JC2000C1, Shanghai Chen Digital Technology Instrument Co., Ltd., China) was utilized to measure the spread of water droplets on the film surface using the sessile droplet method. The contour data of water droplets were fit using the Laplace-Young equation (Zhang et al., 2023c).

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215 2.4.7. Water vapor permeability (WVP)

The WVP of the films was determined using a WVP tester (W3/031, Labthink Instruments Co., Ltd., Jinan, China). Round films with a diameter of 80 mm were tested at 38.0°C and 90% relative humidity (RH) with a preheating time of 4 h and a weighing

219 interval of 2 h. The WVP value was calculated from triplicate measurements.

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221 2.4.8. Water solubility (WS)

WS was determined according to the method of Gao et al. (2022) with some modifications. The films were cut into samples measuring 40 mm \times 40 mm and dried at 40°C until a constant weight (m₁) was achieved over 24 h. Subsequently, the dried film samples were immersed in 50 mL of distilled water for 24 h, then removed, dried again at 60°C for 24 h, and weighed (m₂). The calculation method for WS (%) was as follows:

 $WS = \frac{m_1 - m_2}{m_2}$

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230 *2.4.9. Thermogravimetric (TG) analysis*

Thermal degradation behavior was measured using a TG analyzer (TGDTA7300, Hitachi, Ltd., Japan). The TG test involved placing 10 mg of film sample in an alumina crucible and heating it in the range of 40-600°C with a heating rate of 10°C/min and an N_2 flow rate of 50 mL/min. Derivative TG (DTG) curves were derived from the differential of TG values.

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237 2.4.10. Release of essential oils in different food simulants

Two food simulants were selected: 10% ethanol (for aqueous-based foods) and 90%

ethanol (for fatty foods). Film samples measuring 20×40 mm were immersed in 10 mL of food simulants, and 2 mL of the solutions were removed at specific time intervals at room temperature to determine the released GEO. Subsequently, 2 mL of the simulant solutions were added to maintain the original conditions. The release of GEO from the films was measured using a UV-visible spectrophotometer at 278 nm (Zhang et al., 2021).

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246 2.4.11. Antioxidant properties

The DPPH and ABTS radical scavenging rate (%) of the films were determined using
a previous method with some modifications (Shen et al., 2021).

A 50 mg film was immersed in 10 mL of anhydrous ethanol and shaken for 4 h at 25°C. Subsequently, 1 mL of the film extract solution was mixed with 3 mL of 0.1 mM DPPH anhydrous ethanol solution and reacted in the dark for 30 min. Afterward, the absorbance at 517 nm was measured.

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Equal amounts of ABTS (7 mM) solution and $K_2S_2O_8$ solution (2.45 mM) were mixed and placed in the dark for 12 h. The absorbance of the mixed solution was then adjusted to be 0.70 ± 0.02 (at 734 nm) to prepare the ABTS working solution. Subsequently, 200 μ L of film extract solution was mixed with 3 mL of ABTS working solution and reacted in the dark for 30 min. Afterward, the absorbance at 734 nm was measured. The DPPH radical scavenging activity (%) and ABTS radical scavenging activity (%) were calculated as follows:

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$$scavenging rate(\%) = 1 - \frac{A_1 - A_2}{A_0} \times 100$$

where A_1 is the absorbance of the sample, A_0 is the absorbance measured under the same conditions with anhydrous ethanol instead of the film extract solution, and A_2 is the absorbance measured with anhydrous ethanol instead of DPPH or ABTS.

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266 2.4.12. Antibacterial properties

267 The antibacterial activity of the films against *S. aureus* and *E. coli* was analyzed using

the disc diffusion method (Hasheminya et al., 2019). Pathogens were inoculated and 268 streaked on Mueller-Hinton agar medium, which was then incubated at 37°C for 12 h. 269 The bacterial cultures were subsequently transferred to 0.85% (w/v) NaCl solution to 270 adjust the turbidity of the bacterial suspension to 0.5 McFarland (1.5×10^{8} CFU/mL). 271 The bacterial suspension was then evenly spread onto Mueller-Hinton agar plates using 272 a cotton swab. Sterile filter paper discs (6 mm) were immersed in the film extraction 273 solution and placed on the agar surface using forceps. The plates were then incubated 274 275 at 37°C for 24 h. After incubation, the antibacterial activity was assessed by measuring 276 the size of the zone of inhibition surrounding the discs.

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278 2.5. Statistical analysis

279 SPSS statistical software version 22.0 (SPSS Inc., Chicago, USA) was applied to 280 analyze the Data, and the difference between the mean values was analyzed by Waller-281 Duncan's multiple range test at the significance level of p < 0.05.

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283 3. Results and discussion

284 3.1 Characterization of GEO-NEs

Oil-in-water (O/W) type GEO-NEs were prepared using the ultrasonic emulsification 285 method. As depicted in Fig. 2A, the average particle size of the GEO-NEs was 286 176.4±1.2 nm, with a PDI of 0.276±0.012, indicating the homogeneous dispersion of 287 the GEO-NEs droplets. This observation was further confirmed by microscopy images 288 (Fig. 2B). GEO-NEs were formulated using Tween 80, a non-ionic surfactant, with 289 steric repulsion as the primary mechanism for stabilizing the nanoemulsion (Campolo 290 291 et al., 2020). Consequently, the zeta potential of the nanoemulsion is expected to approximate zero, whereas nanoemulsions stabilized by electrostatic repulsion require 292 293 a zeta potential of at least ± 30 mV (Zhang et al., 2022). The zeta potential of GEO-NEs was measured to be -12.37 ± 0.05 mV, indicating a negative surface charge, which may 294 be attributed to the dissociation of ionizable compounds present in GEO (Campolo et 295 296 al., 2020; Hasheminya & Dehghannya, 2021). Additionally, no phase separation was observed for GEO-NEs after storage at 4°C and 25°C for 28 days (Fig. 2C), and there 297

were no significant changes in the average particle size and PDI (Fig. 2D), indicating
the excellent stability of GEO-NEs.

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301 *3.2. Characterizations of composite films*

302 *3.2.1 Fourier transform infrared (FT-IR) analysis*

The intermolecular interactions within the film matrix were analyzed using FTIR 303 spectroscopy. As illustrated in Fig. 3A, a broad and strong band was observed in the 304 range of 3500-3300 cm⁻¹, attributed to the stretching vibrations of hydroxyl groups (-305 OH) present in PVA, xylan, and glycerol (Wang et al., 2022; Yang et al., 2024). A 306 gradual shift of the peak from 3290 cm⁻¹ to 3280 cm⁻¹ was noted in the wavenumber of 307 films containing GEO-NEs compared to the control group, implying the formation of 308 309 hydrogen bonds between GEO-NEs and the film matrix (Wu et al., 2023). Their potential hydrogen-bonding interactions are depicted in Fig. 3C. The weak band around 310 2927 cm⁻¹ was related to the stretching vibrations of C-H in the CH₂ and CH₃ groups. 311 Additionally, films containing GEO-NEs all exhibited a characteristic peak at 1740 cm⁻ 312 ¹, with the peak amplitude progressively strengthening with increasing concentrations 313 of GEO-NEs. This peak was associated with the C=O stretching vibration of the 314 aldehyde or ester carbonyl groups of GEO. Similar findings were reported by Cai and 315 Wang (2021). The characteristic peaks at 1417 and 1034 cm⁻¹ corresponded to the 316 stretching vibrations of -C-H and C-O-C in the pyranose ring of xylan, respectively, 317 and the β -pyranose configuration of xylan was confirmed by the characteristic peak at 318 846 cm⁻¹ (Liu et al., 2019b; Shahrampour & Razavi, 2023). FTIR results indicated that 319 hydrogen bonding dominated the interaction between the GEOs-NEs and the 320 xylan/PVA base matrix, and GEOs-NEs did not change the chemical structure of the 321 films. 322

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324 *3.2.2 X-ray diffraction (XRD) analysis*

The crystal structure of the films was investigated using XRD. As depicted in Fig. 3B, a prominent diffraction peak was observed at $2\theta = 19.3^{\circ}$ for all films. Importantly, the incorporation of GEO-NEs did not result in any new diffraction peaks, suggesting that

GEO-NEs did not induce alterations in the crystal structure of the xylan/PVA base 328 matrix. This observation validates the compatibility between GEO-NEs and the film 329 matrix (Fan et al., 2023). However, the incorporation of GEO-NEs resulted in a 330 reduction in the crystallinity of the films. This effect was particularly noticeable at 331 GEO-NEs concentrations of 2% and 4%, where a weakening in the intensity of the film 332 diffraction peaks was observed. On the one hand, FTIR analysis confirmed the 333 formation of new hydrogen bonds among GEO-NEs and the film matrix, thereby 334 335 weakening the original ordered structure formed by xylan and PVA, consequently reducing the crystallinity (Liu et al., 2019b). On the other hand, reduced crystallinity 336 implies increased mobility among xylan and PVA molecules, which may affect the 337 mechanical properties of the films, such as elongation at break (Tavassoli et al., 2021). 338

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340 *3.2.3 SEM analysis*

The surface and cross-section SEM micrographs of films are presented in Fig. 4. The 341 control film appeared smooth and continuous without any pores, confirming the good 342 343 compatibility between xylan and PVA. However, as the concentration of GEO-NEs increased, the film surfaces became rougher, and pores began to appear in the cross-344 sections. This phenomenon was particularly noticeable at GEO-NEs concentrations of 345 6% and 8%. The presence of pores and roughness can be attributed to the high 346 concentration of GEO-NEs hindering the ability of the film matrix to capture more oil 347 droplets through hydrogen bonding. This leads to increased flocculation and 348 coalescence of oil droplets during the film formation process, causing them to migrate 349 towards the film surface and resulting in a non-homogeneous morphology. Additionally, 350 351 the evaporation of the oil droplets during this process contributes to the formation of pores (Acevedo-Fani, Salvia-Trujillo, Rojas-Graü, & Martín-Belloso, 2015; 352 Hasheminya et al., 2021; Zhang et al., 2021). Similar observations have been reported 353 in other studies where biopolymer-based films became rough and porous after mixing 354 with EO-NEs (Cai et al., 2021; Chu et al., 2020; Fan et al., 2024; Norcino et al., 2020). 355 These changes in microstructure may affect the barrier properties of the film, such as 356 water barrier performance (Hasheminya et al., 2021), as confirmed in section 3.2.7. 357

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359 *3.2.4 Optical properties*

As shown in Table S1, with the increasing concentration of GEO-NEs, the L* and a* 360 values of the composite film decreased while b* increased, indicating that the film 361 became darker and more yellowish. Moreover, the opacity gradually increased, which 362 is attributed to light scattering caused by the surface roughness and internal porous 363 structures of the film (Lin et al., 2020). Transparency is particularly important for food 364 365 packaging materials, as foods retaining their original colors and appearances are more likely to be accepted by consumers (Liu et al., 2022). From a visual perspective (Fig. 366 5A), the composite film containing 8% GEO-NEs still exhibited relatively high 367 368 transparency.

369 The UV transmittance of the films is illustrated in Fig. 5B. It can be observed that the incorporation of GEO-NEs effectively shielded the spectra of UVC (275-200 nm), 370 UVB (320-275 nm), and UVA (400-320 nm), which significantly enhanced the UV 371 shielding performance of the films. The light scattering by the small lipid droplets of 372 373 the nanoemulsion and the absorption of UV and visible light by the phenolic compounds in the GEO may be responsible for the enhanced UV shielding performance 374 of the films (Chen et al., 2016; Fu et al., 2023), which is positive for inhibiting the 375 oxidative deterioration of foodstuffs (Zhang et al., 2023c). 376

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378 *3.2.5 Mechanical properties*

Mechanical properties are crucial attributes of films, ensuring their ability to withstand 379 external stresses during transportation, handling, and storage. Tensile strength (TS) and 380 381 elongation at break (EAB) represent the strength and flexibility of the films, respectively (Erfanifar, Majdinasab, & Shaghaghian, 2023). As illustrated in Fig. 5C 382 and D, the xylan/PVA composite films exhibited the capability to withstand various 383 deformations, with their TS and EAB comparable to the mechanical properties of the 384 prevalent commercial packaging material, high-density polyethylene (HDPE) (TS, 385 approximately 22-23 MPa; EAB, approximately 150%) (Lee, Garcia, Shin, & Kim, 386 2019). As the concentration of GEO-NEs increased, the TS gradually decreased from 387

27.38±3.25 MPa in the control film to 20.27±2.55 MPa in the 8% GEO-NEs film. 388 Conversely, the EAB of the films significantly increased (p<0.05) and reached its 389 maximum at a moderate concentration (GEO-NEs 4%), measuring 258.4±38.6%. Due 390 to the weakened hydrogen bonding between xylan and PVA caused by GEO-NEs, the 391 cohesive forces of the film matrix were reduced, resulting in a lower TS (Almasi, Azizi, 392 & Amjadi, 2020; Zhang et al., 2021). At the same time, the chain entanglement of xylan 393 and PVA was weakened, leading to an increase in molecular mobility and consequently 394 395 enhancing the EAB (Chen et al., 2016). These results were also verified in the XRD analysis. It is worth noting that the continued addition of GEO-NEs resulted in a 396 reduction in EAB. This may be attributed to the uneven dispersion of high 397 concentrations of GEO-NEs within the film matrix, leading to flocculation and the 398 399 formation of localized hardened regions during the film formation process. Ultimately, this phenomenon affected the overall flexibility of the thin film. 400

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402 *3.2.6 Thermostability*

403 The thermostability of the films was investigated using thermogravimetric analysis (TGA) and the first derivative of TGA curves (DTG). As shown in Fig. 5E, the films 404 exhibited multiple stages of weight loss within the temperature range of 40-600°C. The 405 initial stage of weight loss (40-120°C) was attributed to the loss of moisture from the 406 film matrix. Subsequently, a second stage of weight loss occurred in the temperature 407 range of 150-250°C, attributed to the degradation of glycerol and PVA chains (Chen, 408 Ren, & Meng, 2015). A third stage of weight loss was observed between 250-330°C, 409 involving further degradation of PVA chains and degradation of xylan chains (Liu et al., 410 411 2019b; Wang et al., 2014). In the final stage, from 350-600°C, polymer matrix carbonization occurred, leading to a stabilization of film mass loss (Zhang et al., 2023b). 412 In Fig. 5F, it was observed that the maximum thermal degradation rate of the film was 413 achieved at around 320°C. The maximum thermal degradation temperature (T_{max}) 414 reflects the thermostability of the film, with a higher T_{max} indicating better thermal 415 stability (Qin et al., 2022). Compared to the control group with a T_{max} of 322.9°C, the 416 T_{max} for films containing 2%, 4%, 6%, and 8% of GEO-NEs were 326.6°C, 322.3°C, 417

325.4°C, and 325.9°C, respectively, showing a slight improvement in thermostability.
In summary, the inclusion of GEO-NEs exerted negligible influence on the
thermostability of the xylan/PVA films.

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422 *3.2.7 WCA, WS and WVP analysis*

Film surface hydrophobicity was assessed by WCA, with $\theta > 65^{\circ}$ generally indicating 423 a hydrophobic surface (Zhang et al., 2023c). As shown in Fig. 6A, all films exhibited 424 425 hydrophilic surfaces, which can be attributed to the inherent hydrophilic properties of xylan and PVA in the film-forming matrix. The control film had the highest WCA of 426 $50.1 \pm 3.9^{\circ}$. However, with the increasing concentration of GEO-NEs, there was a 427 significant decrease in the WCA of films (P < 0.05). This decrease can be associated 428 with the strong surface hydrophilicity of GEO-NEs droplets, as Tween80 and GEO can 429 form emulsion droplets with hydrophilic tails pointing to the aqueous solution (Liu et 430 al., 2022). WCA was also influenced by the surface roughness and porosity of the film 431 (Liu et al., 2019a; Qin et al., 2022). Higher surface roughness and internal porous 432 433 structures increased the surface area of the film in contact with water molecules, leading to increased hydrophilicity or even water solubility of the film (Fig. 6B). 434

The efficiency of moisture penetration through the film was assessed by determining 435 the WVP. A higher WVP indicates insufficient water barrier properties of the film, 436 which would create favorable conditions for microbial growth and lead to food spoilage. 437 As illustrated in Fig. 6C, there was a significant decrease followed by a remarkably 438 increased trend in WVP with increasing concentration of GEO-NEs (P < 0.05). 439 Compared to the WVP of the control film, which was $10.45 \pm 0.72 \times 10^{-13} \text{ g} \cdot \text{cm}^{-1} \cdot \text{s}^{-1} \cdot \text{Pa}^{-1}$ 440 ¹, the film containing 2% GEO-NEs exhibited the lowest WVP of $7.45 \pm 0.71 \times 10^{-13}$ 441 g·cm⁻¹·s⁻¹·Pa⁻¹. The decrease in WVP is associated with an increase in the tortuosity of 442 the path of water molecules through the film. On the one hand, GEO-NEs droplets at 443 low concentrations were able to disperse uniformly in the film matrix, thereby 444 increasing the length of the path for water molecules through the film (Hasheminya et 445 al., 2021). On the other hand, XRD results indicated that GEO-NEs caused 446 discontinuities in the film matrix, which could potentially further increase the tortuosity 447

path. However, films containing high concentrations of GEO-NEs (6% and 8%) 448 exhibited a large number of pores within their structure, which provided additional 449 space and channels for water molecules to traverse through the film, thus causing an 450 increase in WVP (Mirsharifi et al., 2023; Rui et al., 2024). Despite the significant 451 improvement in water barrier performance observed in composite films containing 2% 452 GEO-NEs, their WVP remains higher than that of commercial low-density 453 polyethylene (LDPE) films, which typically exhibit a WVP of approximately 7.00×10⁻ 454 ¹⁴ g·m⁻¹·s⁻¹·Pa⁻¹ (Silva et al., 2024). Therefore, further research is required to improve 455 the water barrier performance, it is also the direction of future efforts for bio-based 456 packaging materials. 457

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459 *3.2.8 Release of GEO in different food simulants*

The release curves of GEO in two different food simulants are shown in Fig. 7. Initially, 460 GEO exhibited a rapid release, which gradually decelerated until reaching a constant 461 level. As described by Dong et al. (2024), water molecules initially diffused from the 462 463 simulated solution into the film matrix, resulting in swelling and dissolution of the film, which led to the detachment of GEO from the film matrix and its subsequent release 464 into the simulated solution until achieving thermodynamic equilibrium. Due to the 465 higher hydrophilicity of the xylan/PVA composite film, its swelling phenomenon was 466 more obvious in a 10% ethanol solution with lower polarity, which led to a faster release 467 of GEO and reached equilibrium in a shorter time. Fig. 7 also demonstrates that the 468 release rate of GEO in both simulated systems decreased with increasing concentration 469 of GEO-NEs. This trend can be explained by the reduction in the mass transfer 470 471 concentration gradient, as the increasing accumulation of GEO in the simulant reduced the concentration gradient between the film matrix and the simulant, making the release 472 of GEO to the simulant slower (Xu et al., 2019). 473

474

475 *3.2.9 Antioxidant and antimicrobial activities*

476 As shown in Fig. 8A, xylan/PVA films without GEO-NEs also exhibited certain 477 antioxidant activity, with DPPH and ABTS radical scavenging rates of 5.53±1.20% and

 $11.06\pm1.22\%$, respectively, which was related to the fact that xylan (derived from corn 478 cobs) as a reducing polysaccharide branched by arabinose and glucuronic acid (Bao et 479 480 al., 2018). With the increase in GEO-NEs concentration, the antioxidant activity of the composite films significantly increased (P < 0.05). Films containing 8% GEO-NEs 481 showed a 45.90% and 59.87% increase in DPPH and ABTS radical scavenging rates, 482 respectively, compared to the control films. A study by Badrunanto et al. (2024), 483 indicated that terpene compounds such as β -myrcene, D-limonene, α -sabinene, geranyl 484 485 acetate, α -curcumene. α -zingiberene, α -farnesene. β-bisabolene, and βsesquiphellandrene were closely associated with the antioxidant activity of GEO. 486

The antimicrobial activity of the films was assessed using the disc diffusion method, as 487 depicted in Fig. 8B. It was noted that inhibition zones against both Gram-positive 488 bacteria (S. aureus) and Gram-negative bacteria (E. coli) were observed when the 489 concentration of GEO-NEs in the film reached 6%. Similar findings were reported by 490 Zhang et al. (2021) and Mutlu (2023), suggesting that the films exhibited antimicrobial 491 activity only upon the addition of high concentrations of EO-NEs. Generally, Gram-492 493 negative bacteria tend to exhibit greater tolerance to EOs when compared to Grampositive bacteria. Apart from the peptidoglycan layer, the outer membrane of Gram-494 negative bacteria contains lipopolysaccharides, rendering the diffusion of hydrophobic 495 EOs into the cells more challenging (Zhang et al., 2023a). However, in this study, it was 496 observed that composite films strongly inhibited E. coli compared to S. aureus. This 497 could be attributed to the accumulation of certain specific hydrophobic compounds 498 499 from the essential oils on the outer membrane of E. coli, disrupting its permeability and 500 inhibiting bacterial growth (Ran et al., 2023). Additionally, studies have also reported 501 controversies regarding the effectiveness of black pepper essential oil (Acharya et al., 2024; Amalraj, Haponiuk, Thomas, & Gopi, 2020; Saranti et al., 2021), oregano 502 essential oil (Hosseini, Rezaei, Zandi, & Farahmandghavi, 2015; Lee et al., 2019), and 503 cinnamon essential oil (Ran et al., 2023; Wu et al., 2023) in various film matrixes 504 against both Gram-positive and Gram-negative bacteria. In conclusion, numerous 505 factors, including the type of film matrix, the type and concentration of EOs, as well as 506 the initial number of bacteria and growth conditions, collectively influenced the 507

antibacterial activity of the films (Hasheminya et al., 2021).

509

510 **4. Conclusion**

In this study, active films with UV resistance, antioxidant, and antimicrobial properties 511 were prepared using xylan and PVA as the film-forming matrix and blended with GEO-512 513 NEs. The results indicated good compatibility among xylan, PVA, and GEO-NEs. The incorporation of GEO-NEs improved the UV shielding properties, with films 514 515 containing 8% GEO-NEs maintaining high transparency. Despite a decrease in tensile strength and an increase in elongation at break, the mechanical properties of all films 516 met the standards of commercial packaging material HDPE. Furthermore, the 517 antioxidant activities of xylan/PVA films were significantly enhanced upon the addition 518 519 of GEO-NEs, exhibiting inhibitory activities against both S. aureus and E. coli at a 6% concentration. Therefore, xylan/PVA films containing GEO-NEs demonstrate potential 520 applications as active packaging materials. However, further research is needed to 521 improve their hydrophobicity and water barrier properties. 522

523

524 **Declaration of competing interest**

- 525 The authors declare no conflict of interest.
- 526

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Fig. 1. Schematic illustration of the preparation of xylan/PVA-based active films.



Fig. 2. Particle size distribution, zeta-potential and PDI of GEO-NEs (A), microscopy images of GEO-NEs (B), and changes in appearance, average particle size, and PDI of GEO-NEs during 0~28 d.



Fig. 3. FT-IR spectra (A), X-ray diffraction spectra (B) of xylan/PVA films containing different concentrations of GEO-NEs and schematic illustration of hydrogen bonding interactions in the film matrix (C).



Fig. 4. SEM micrographs of the surfaces (left column) and the cross-sections (right column) of xylan/PVA films containing different concentrations of GEO-NEs.



Fig. 5. Digital photographs (A), UV transmittance (B), various deformations (C), tensile strength and elongation at break (D), TGA (A) and DTG (B) thermograms of xylan/PVA films containing different concentrations of GEO-NE.



Fig. 6. WCA (A), WS (B) and WVP (C) of xylan/PVA films containing different concentrations of GEO-NEs.



Fig. 7. Release of GEO from films in 10 % ethanol (A) and 90 % ethanol (B).



Fig. 8. Antioxidant activity (A), inhibition zones (B) and antibacterial pictures (C, D) of xylan/PVA films containing different concentrations of GEO-NEs.

Supplementary material

		-	-	-	
	L*	a*	b*	$\triangle E$	Opacity
GEO-NEs 0%	92.72±0.03ª	1.08±0.03ª	0.76 ± 0.08^{e}	$0.15{\pm}0.06^{e}$	$0.84{\pm}0.07^{\circ}$
GEO-NEs 2%	92.17±0.11 ^b	$0.89{\pm}0.06^{b}$	$1.65{\pm}0.18^d$	$1.16{\pm}0.19^{d}$	$3.05{\pm}0.14^{\text{b}}$
GEO-NEs 4%	$92.01{\pm}0.29^{\text{b}}$	$0.83{\pm}0.05^{bc}$	$2.00{\pm}0.05^{\circ}$	1.55±0.19°	$3.89{\pm}0.11^{ab}$
GEO-NEs 6%	$91.94{\pm}0.10^{b}$	$0.78{\pm}0.08^{\circ}$	$2.38{\pm}0.16^{b}$	$1.93{\pm}0.17^{b}$	5.07±0.64ª
GEO-NEs 8%	91.53±0.25°	$0.65{\pm}0.05^{d}$	2.95±0.31ª	2.63±0.39ª	5.47±1.45ª

Table S1 The color parameters and opacity of films.

Note: Different letters in the same column indicate significant differences (P < 0.05).