





# MOSH & MOAH IN FOOD: STATE-OF-THE-ART AND RECENT ADVANCEMENTS

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2024 AOCS Annual Meeting & Expo

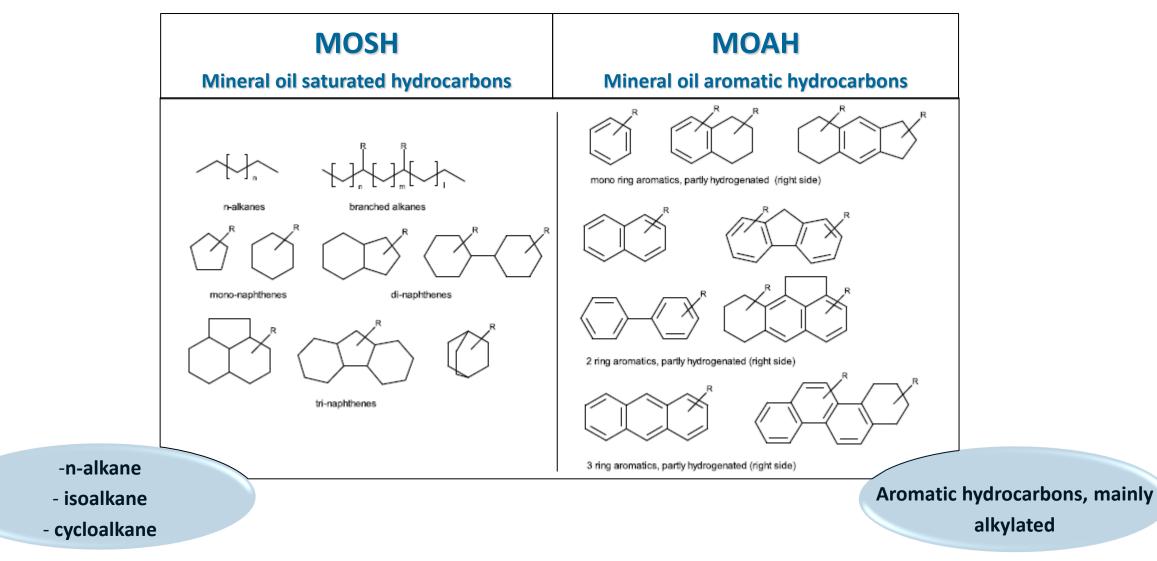
April 28-May 1, 2024, Palais des congrès de Montréal, Montréal, Québec, Canada



### **MINERAL OIL HYDROCARBONS (MOH): DEFINITION\***

a wide range of products deriving from petroleum distillation fractions







### **MOSH & MOAH: STATE-OF-THE ART**

**Vetsa** 



ef<sup>sa</sup>JOURNAL



EFSA Journal 2012;10(6):2704

#### SCIENTIFIC OPINION

2012

Scientific Opinion on Mineral Oil Hydrocarbons in Food<sup>1</sup>

EFSA Panel on Contaminants in the Food Chain (CONTAM)<sup>2,3</sup>

European Food Safety Authority (EFSA), Parma, Italy

This scientific output, published on 28 August 2013, replaces the earlier version published on 6 June 2012\*.

#### SCIENTIFIC OPINION

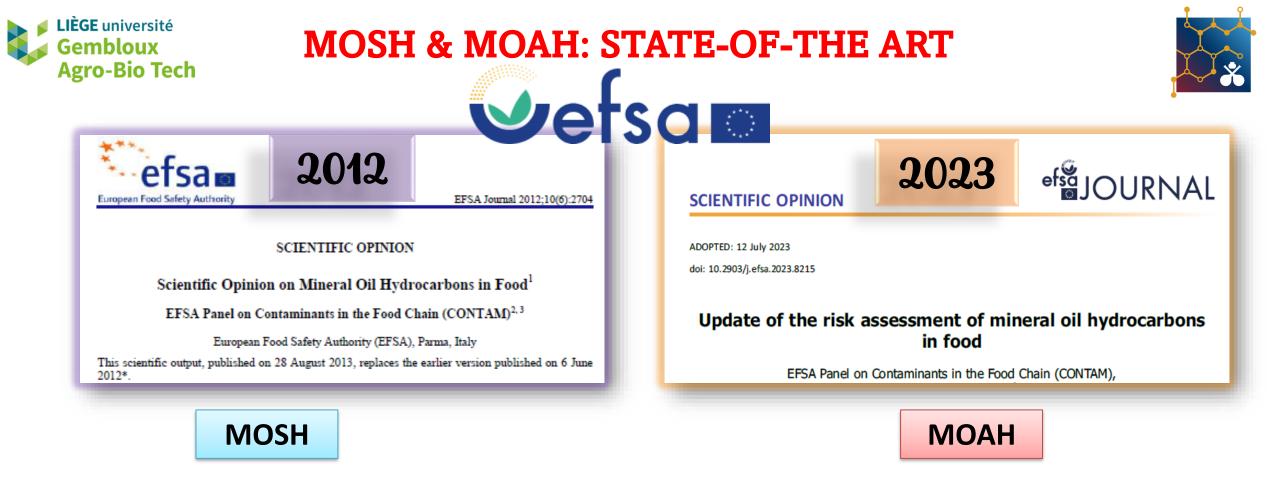
ADOPTED: 12 July 2023

doi: 10.2903/j.efsa.2023.8215

#### Update of the risk assessment of mineral oil hydrocarbons in food

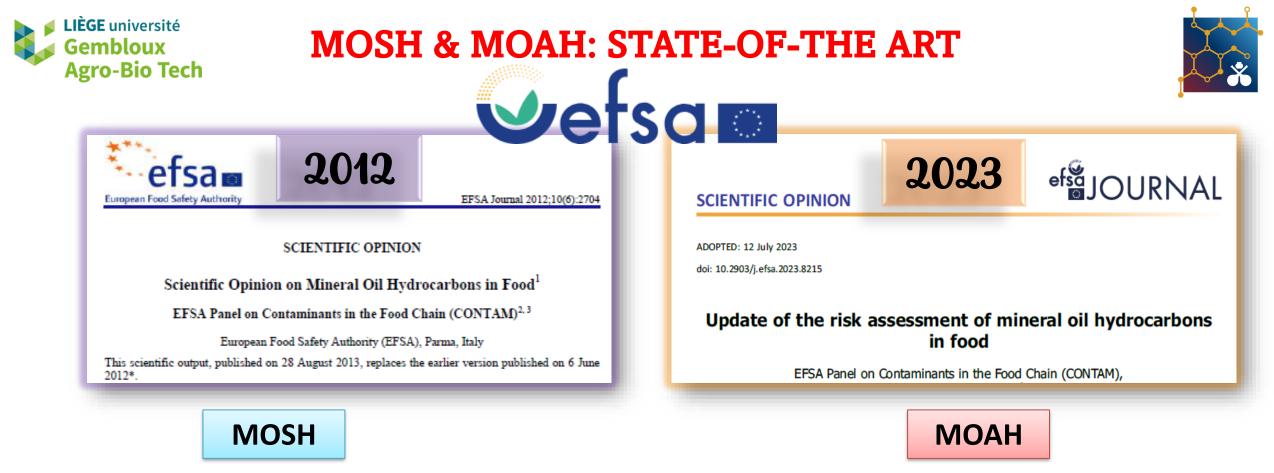
2023

EFSA Panel on Contaminants in the Food Chain (CONTAM),



- Generally considered of <u>no concern</u> at the concentration found, although accumulate in human body.
- ➢ genotoxicity of MOAH with ≥3 aromatic rings
- in the absence of reliable toxicity data, the dietary exposure to 1–2 ring MOAH might raise a concern

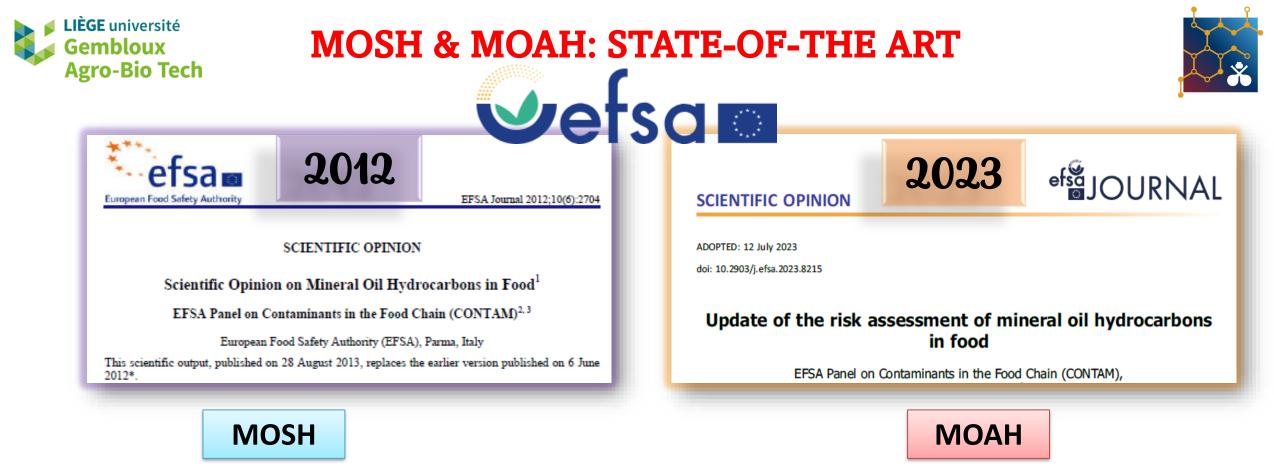
Generation of further data for the refinement of the risk assessment is needed



Generation of further data for the refinement of the risk assessment is needed

> Better investigation of the **sources** 

> Investigation on the **MOSH/MOAH structures and occurrence** 



Generation of further data for the refinement of the risk assessment is needed

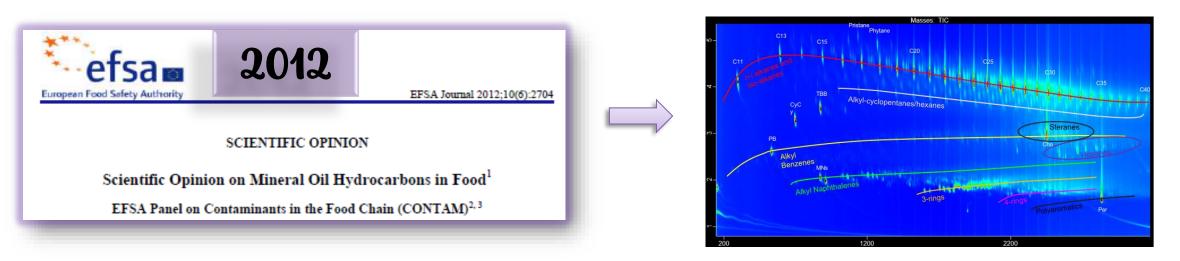
Improvement of analytical methodology

for better characterisation of MOSH&MOAH and consistency in reporting



# **MOSH & MOAH: STATE-OF-THE ART**

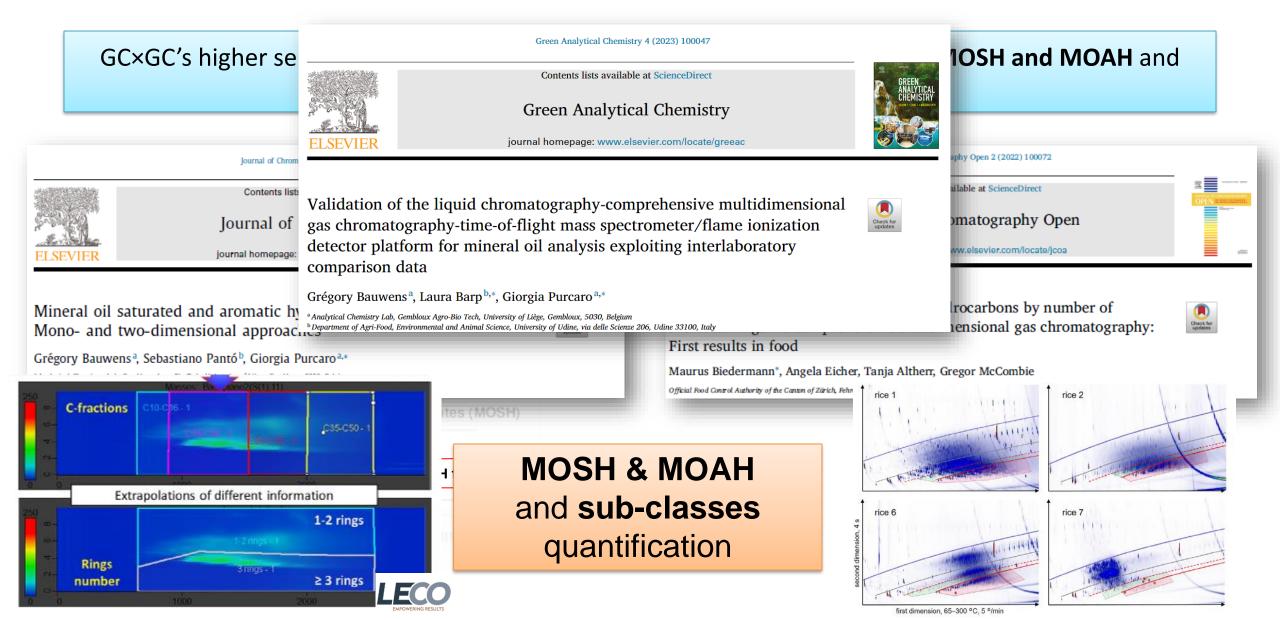




Currently, the most efficient methods for analysis of MOSH and MOAH in food and feed comprise extraction followed by pre-separation by <u>high performance liquid chromatography (HPLC) on-line coupled to GC with flame ionisation</u> <u>detection (FID)</u>. Detection limits depend on the mass distribution, the sample matrix and any prior enrichment, and can be as low as 0.1 mg/kg. <u>Comprehensive GC×GC-FID</u> enables a rough separation and quantification of paraffins and naphthenes in the MOSH fraction, but it is of limited practicality for routine analysis. Contamination with polyolefin oligomeric saturated hydrocarbons (POSH), e.g. from plastic bags, heat sealable layers or adhesives, may interfere with MOSH analysis. Analytical capacity to distinguish the different MOAH subclasses in food is limited. For this purpose, <u>GC×GC appears to be the most effective method</u>. Due to the complexity and the variable composition of MOH mixtures, it is not possible to define certified standards of general applicability.

#### Gembloux Agro-Bio Tech

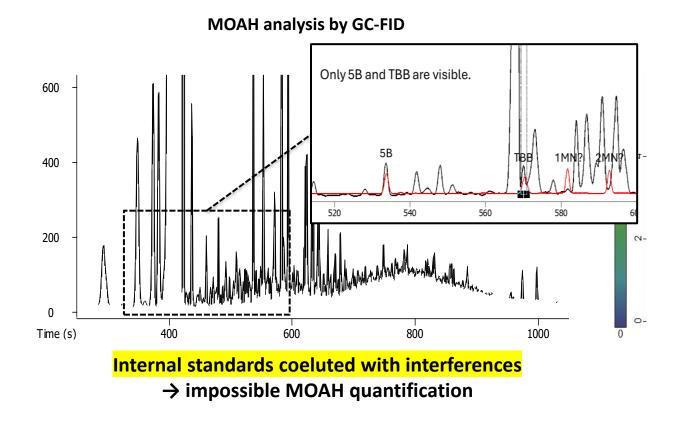








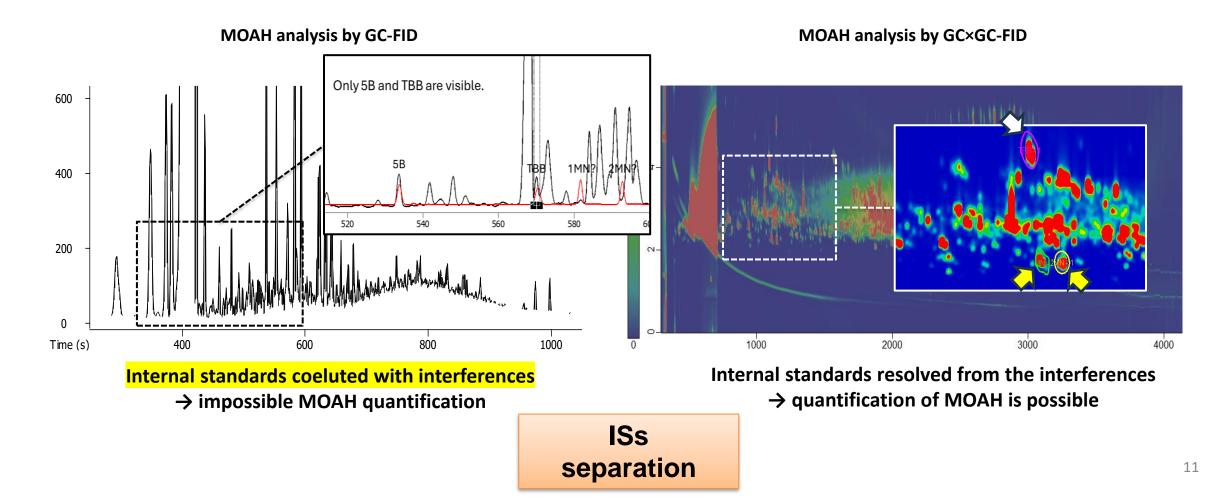
GC×GC's higher separation power is useful for **more accurately determining MOSH and MOAH** and **sub-class determination** 







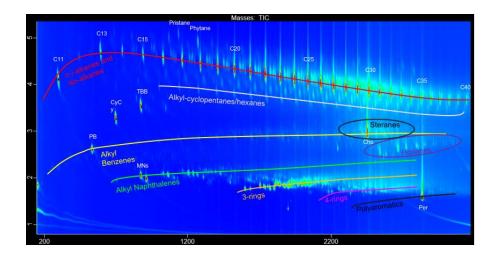
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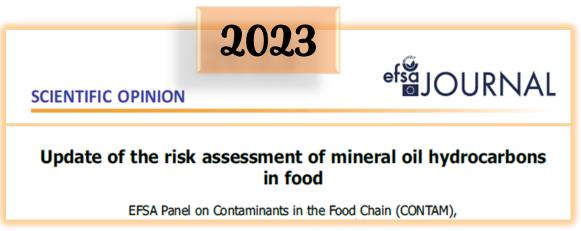




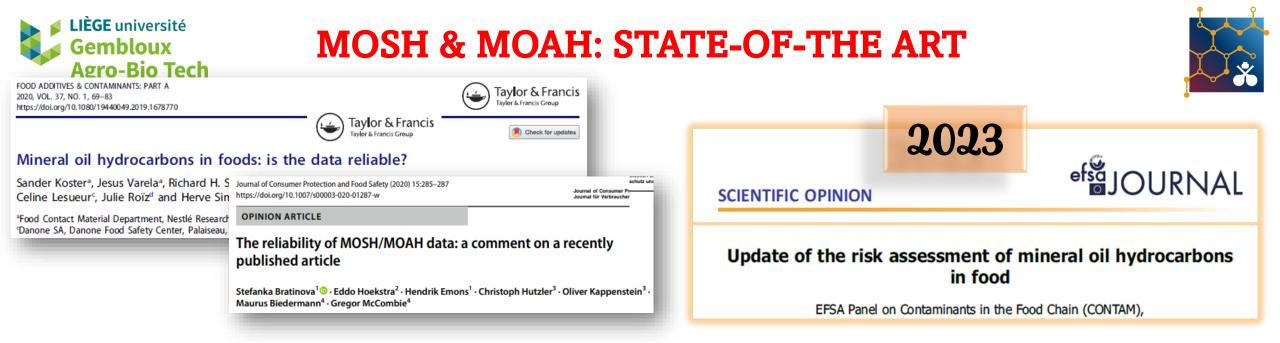
# **MOSH & MOAH: STATE-OF-THE ART**







"...Over the last 10 years, **progress in compositional analysis** has been achieved through comprehensive two-dimensional GC (GC × GC) with FID and MS. The main features of **GC** × **GC** are not only significantly **better separation** and **lower detection limits**, but also placing structurally related compounds in an order, e.g. MOAH according to the number of aromatic rings. In this way, GC × GC may <u>provide structural</u> <u>information</u> if just a single compound, or even no compound of the series could be identified owing to lacking standards or reference mass spectra. ..."



"...There have been, and still are, discussions about the reliability of the results, particularly for measurements at low concentrations. The LC-GC-FID method can be considered as standard and reliable, validated by collaborative tests. Nonetheless, sometimes there were large differences in the results from different laboratories. There were several reasons for this."

(i) Blank and cross-contamination due to the ubiquitous presence of MOH.
(ii) Interference removal is a critical step ↔ at lower limit of quantitation
(iii) Chromatogram interpretation → needs experience in the interpretation



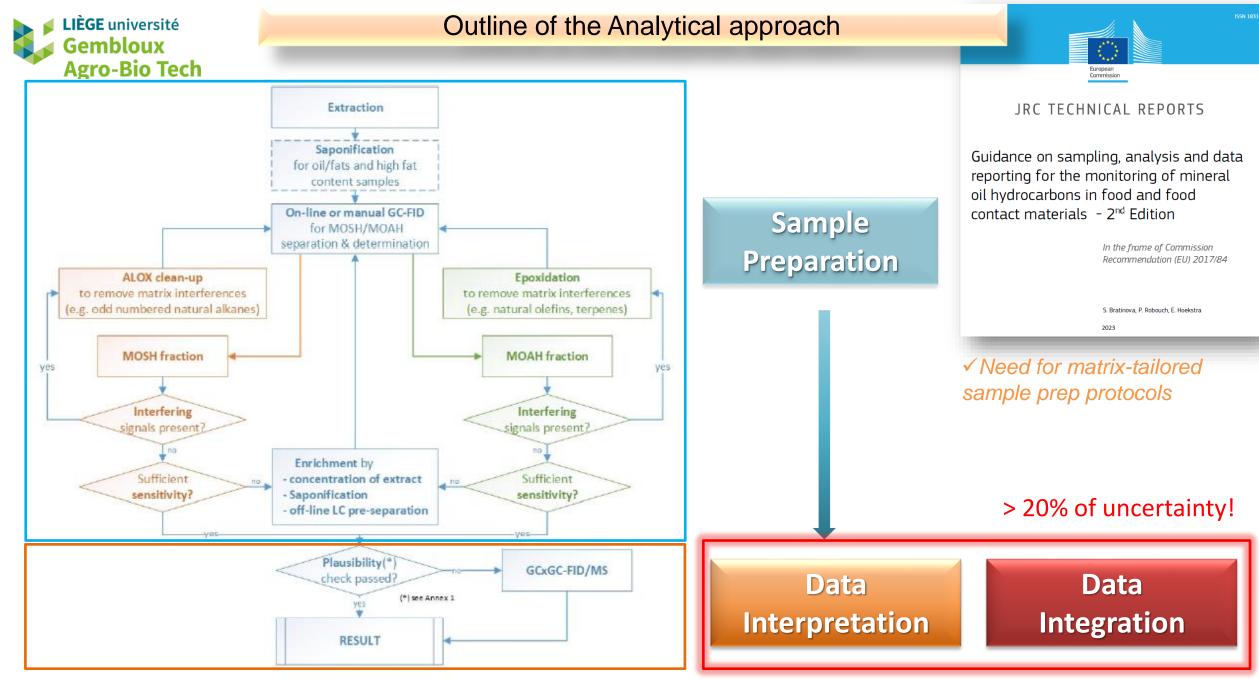
# **Consistency in reporting and data reliability**

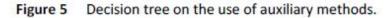


#### JRC guidance: harmonised procedures

- ✓ for sample prep (decision tree)/standardization
- ✓ C-fraction reported (extended to C50)









# **Consistency in reporting and data reliability**



#### JRC guidance: harmonised procedures

✓ for sample prep (decision tree)/standardization

 $\checkmark\,$  C-fraction reported (extended to C50)





Riding peaks subtraction

Beldi G., Senaldi C., Valzacchi S. and Hoekstra E.



Mineral oil in infant formulas - guidelines for integrating chromatograms

JRC IF 2021-04: a virtual inter-laboratory comparison

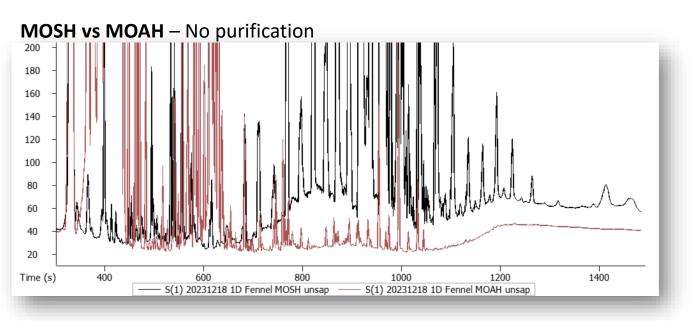
Robouch P., Bratinova S., Goncalves C., Karasek L., Beldi G., Senaldi C., Valzacchi S. and Hoekstra E.

# Interpretation Integration

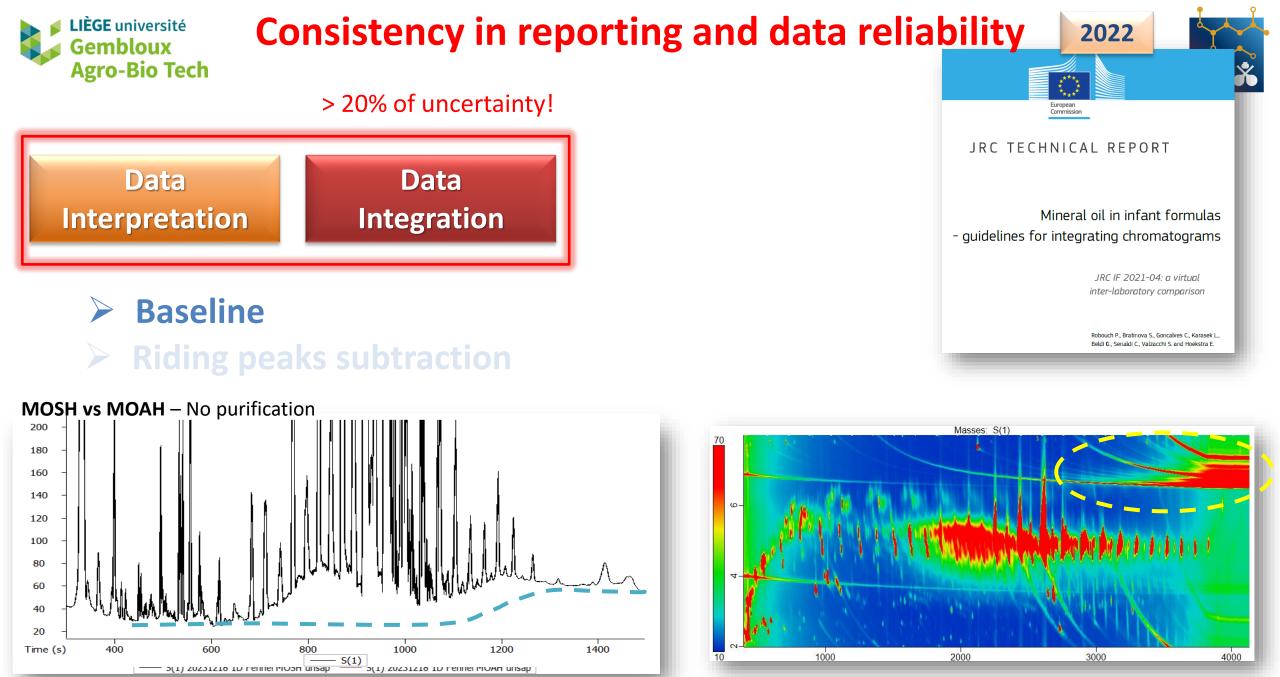
> Baseline

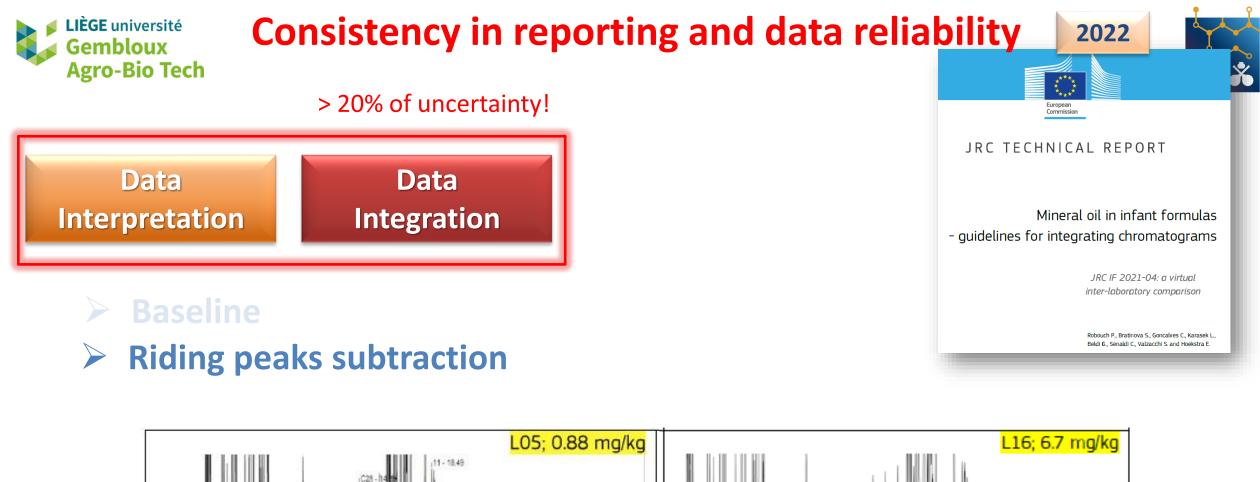
Data

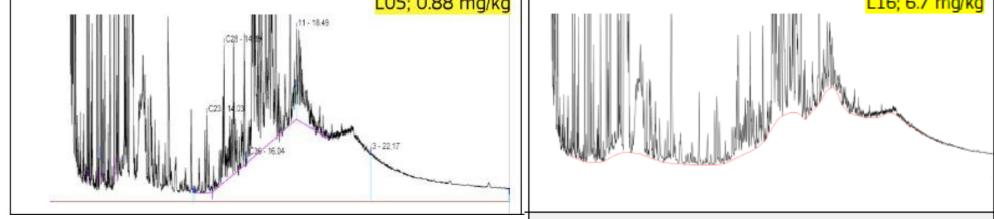
### Riding peaks subtraction



Data

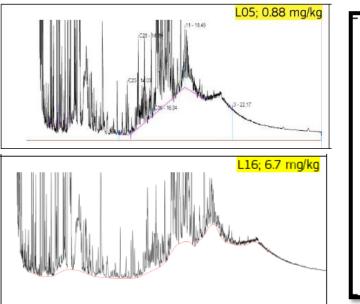








### Riding peaks subtraction



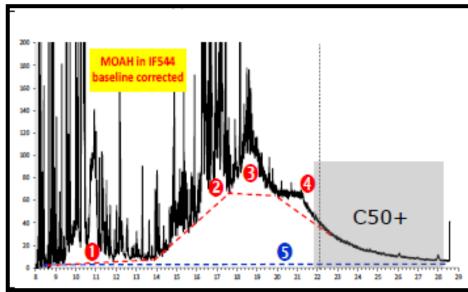
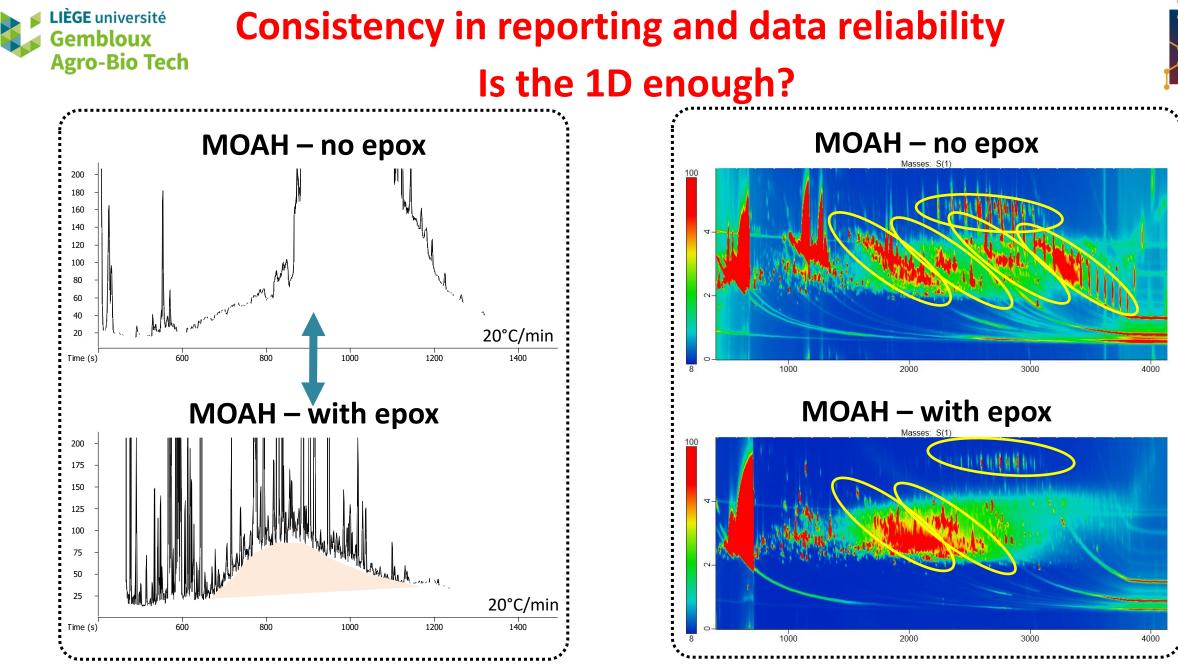
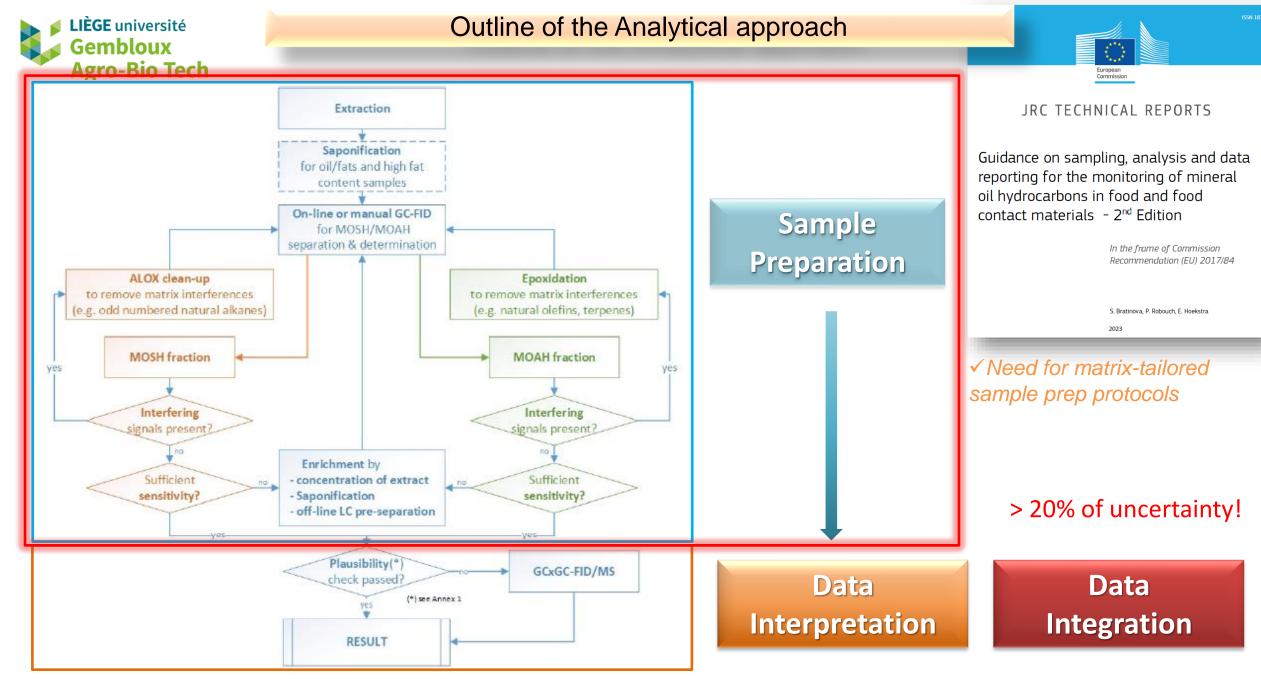


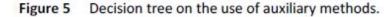
Figure 3: MOAH in IF 544 chromatograms -Comparison of different integration approaches presented by laboratories having reported total mass fractions of MOAH (C10-C50) ranging from 0.88 to 6.7 mg/kg.

← The last chromatogram (on the left) highlights various RT regions to be considered (1 to 5), while the table below summarises the riding peaks/humps included (yes, no, or partially) by the laboratories.

Reldi G. Senaldi C. Valzacchi S. and Hoekstra F









# **Consistency in reporting and data reliability**



### ISO 20122:2024



#### Vegetable oils

Determination of mineral oil saturated hydrocarbons (MOSH) and mineral oil aromatic hydrocarbons (MOAH) with online-coupled high performance liquid chromatography-gas chromatography-flame ionization detection (HPLC-GC-FID) analysis Method for low limit of quantification

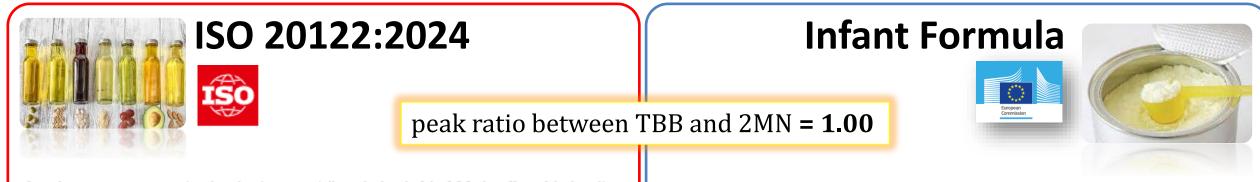
Status : Published





# **Consistency in reporting and data reliability**





In order to ensure correct signal evaluation, especially at the level of the LOQ, the offset of the baseline in the area of integration of MOSH and MOAH must not exceed one quarter of the height of the signal to be integrated (see Figure 18, Annex A). In such cases the chromatographic separation performance of the system must be improved or the LOQ must be increased.

In summary, the following conditions should be examined:

- appropriate straight baseline, blank level  $\leq 1/3$  of the LOQ,
- peaks are symmetrical and do not show a significant tailing,
- complete separation of solvent and C10,
- discrimination C10/C20 and C50/C20 between 80  $\% \leq x \leq 120$  %,
- peak ratio between TBB and 2MN ≤1,25,

- Check the LOQ of a matrix by adding a suitable mineral oil product before sample preparation. The signal of the hump at the LOQ should have a relative standard deviation  $\leq 20$  % and a signal to noise ratio  $\geq 10$ . In addition, the LOQ should be tested with a mineral oil product, whose signal width from start to end is comparable to or higher than the boiling range of the MOH in the sample to be evaluated.

#### peak ratio between TBB and 2MN $\leq$ **1.25**

Even if statistically negligible, the mass fraction of total MOAH referred to TBB was constantly 13 % lower than the one obtained using 2MN. It is therefore recommended, in order to ensure proper comparability of results, to report the MOAH in IF using TBB as the internal standard.

Still, users should be aware that this experimental protocol does not fully remove all the interfering compounds present in some challenging matrices, which may require a cautious interpretation and integration of the recorded chromatograms. Further characterisation with two-dimensional GC-GC techniques may be required.

total MOAH referred to **TBB** was constantly **13** % lower than the one obtained using **2MN**.







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TBB/2MN =1.13





TBB/2MN ≤1.25

#### JRC TECHNICAL REPORTS

Guidance on sampling, analysis and data reporting for the monitoring of mineral oil hydrocarbons in food and food contact materials - 2<sup>nd</sup> Edition



In the frame of Commission Recommendation (EU) 2017/84

S. Bratinova, P. Robouch, E. Hoekstra

2023

When complete saponification is applied and the extraction of mineral oils is performed in hexane in the absence of fat layers, the ratio TBB vs 1- or 2-MN is above 1 with a mean value around 1.15, even after a second hexane extraction of the aqueous saponification solution. This was confirmed during the two ring trial validation studies for the determination of MOSH/MOAH in edible oil and fats by CEN and in infant formula by the JRC organised in 2021/22. It might be attributed to the difference in distribution between the aqueous and hexane phases of the two IS belonging to two different classes of substances (2-ring non-branched and single ring branched aromatic hydrocarbons). In such cases, the quantification is performed vs TBB as the efficiency of the extraction of 1-MN and 2-MN in hexane is lower. Consequently, when saponification is applied, the results reported vs TBB should be reported with uncertainty reflecting the contribution of the saponification step.

### TBB/2MN **=1.15**







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TBB/2MN **≤1.25** 



TBB/2MN =1.13





#### JRC TECHNICAL REPORTS

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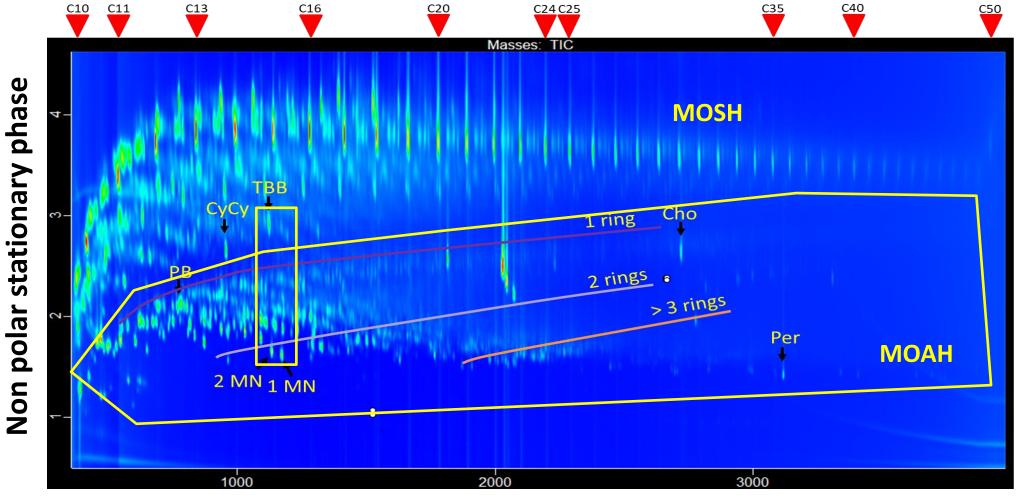




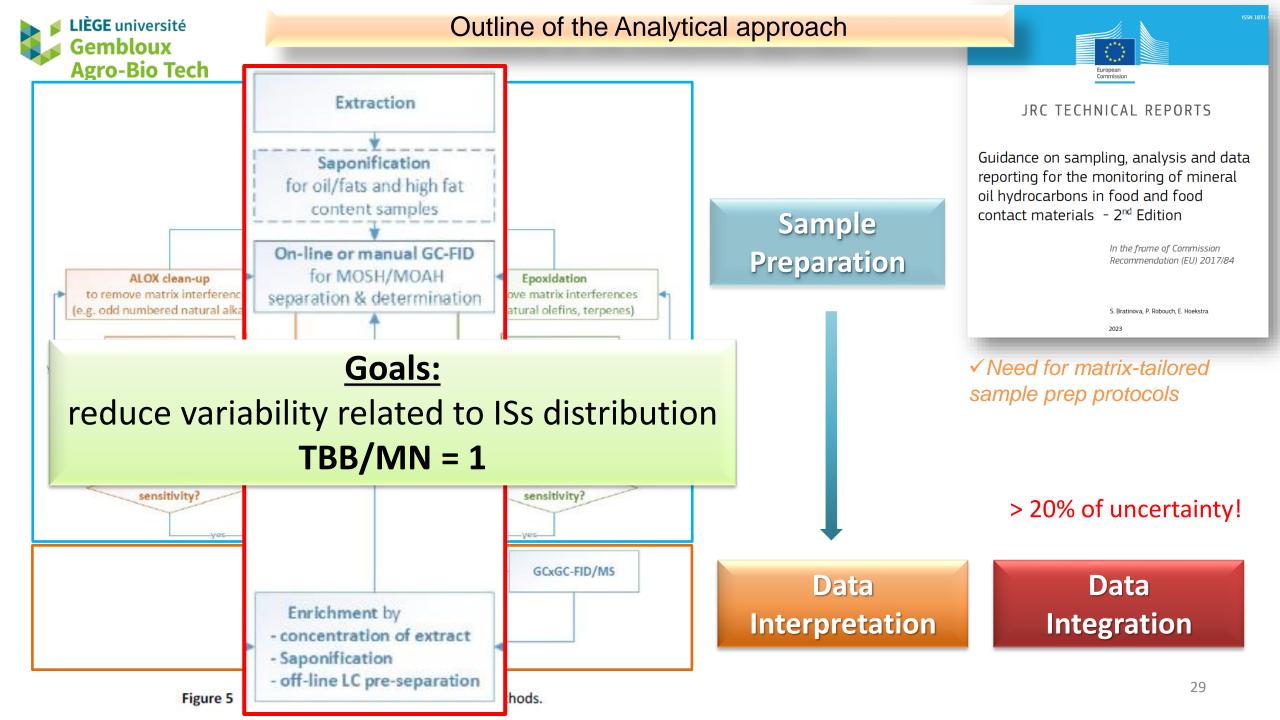
## **GC: a partition process**



### BUT an internal standard should be representative!



**Mid-polar stationary phase** 





### **Extraction:** saponification step



#### MOAH

Enrichment by concentration of extracts and/or saponification and/or off-line LC preseparation

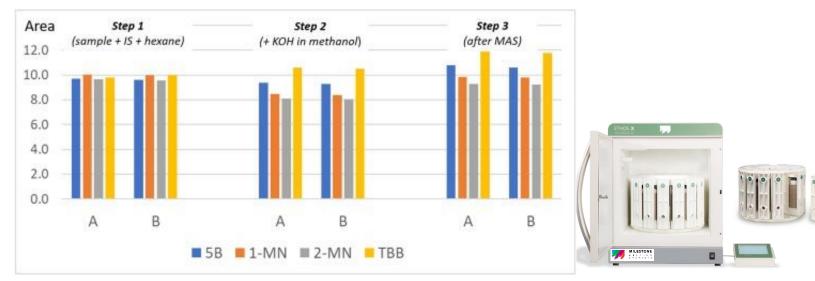
#### **Saponification**

- Traditional saponification ISO 20122:2024
- Microwave assisted saponification-MAS



TBB/MN= 1.15-1.2





Menegoz Ursol et al., Food Chemistry, 2022, 370, 13096





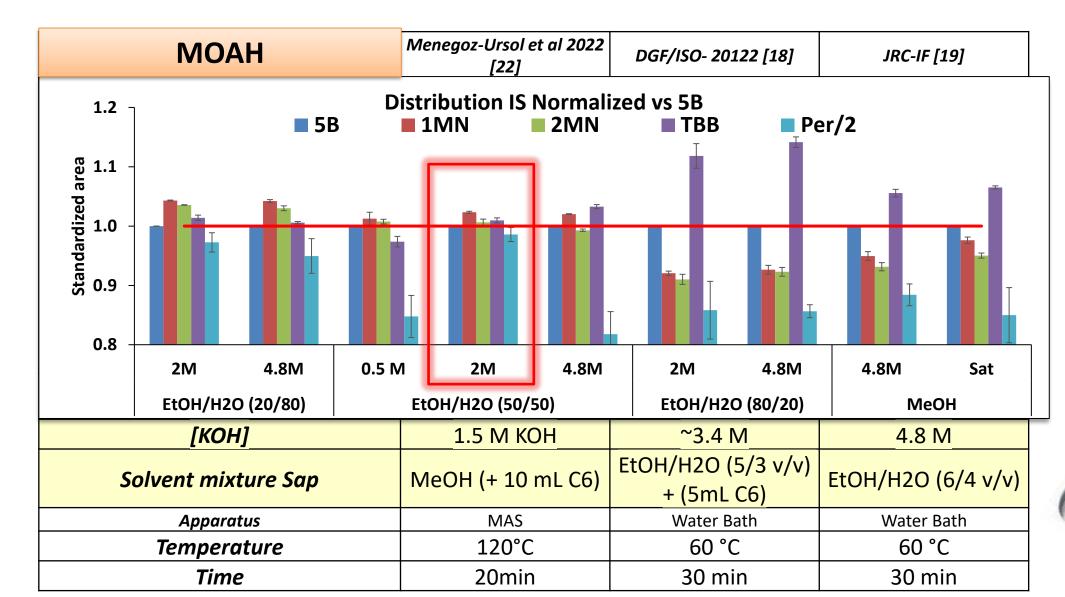
MOAH		Menegoz-Ursol et al 2022 [22]	DGF/ISO- 20122 [18]	JRC-IF [19]
Brief description	Saponification	1g oil + 10 mL 1.5 M KOH (1.5 M)/MeOH + 10 mL C6. Saponification at 120 °C for 20 min.	3 g oil in 30 mL C6/EtOH (1:1). 10 mL of this added with 3 mL KOH in H2O (0.5 g/mL). Saponification at 60 °C for 30min	5 g infant formula + 10 mL H2O. Add 10 mL of KOH (4.8 M) in EtOH/water (1:1) + 5mL EtOH. Saponification at 60°C for 30 min
	washing	40 mL of H2O + 3 mL of MeOH added in the vessel. Store at -18 °C for 30 min. Collected the C6 lyer, concentrate to 4 mL and wash with 3 mL MeOH/H2O (2/1 v/v)	Add 5mL C6 + 5 mL EtOH/H2O (1/1 v/v). Repeat twice and combine the C6 phases.	15 mL C6 + 2.5 mL EtOH twice. Wash C6 with 15 ml of EtOH/ H2O (1/1 v/v)
Mass sample		1g oil	1 g oil in 10 mL C6/EtOH (1:1)	5 g IF + 5 mL H2O
[KOH]		1.5 M KOH	~3.4 M	4.8 M
Solvent mixture Sap		MeOH (+ 10 mL C6)	EtOH/H2O (5/3 v/v) + (5mL C6)	EtOH/H2O (6/4 v/v)
Apparatus		MAS	Water Bath	Water Bath
Temperature		120°C	60 °C	_60 °C
Time		20min	30 min	30 min

















#### **Saponification**

Microwave assisted saponification-MAS

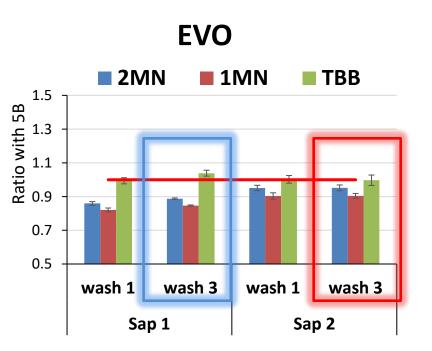
with the conditions used in ISO 20122:2024.

- 60 °C for 30 min
- Comparing **Sap: 1** 2 M KOH in EtOH/H2O (**5/3 v/v**) <mark>ISO 20122:2024</mark> **2**- 2 M KOH in EtOH/H2O (**1/1 v/v**).





 Comparing Washing : 1- Add 5 mL of C6 + 5 mL EtOH/H2O (1/1 v/v). ISO 20122:2024 3- 20 mL of H2O









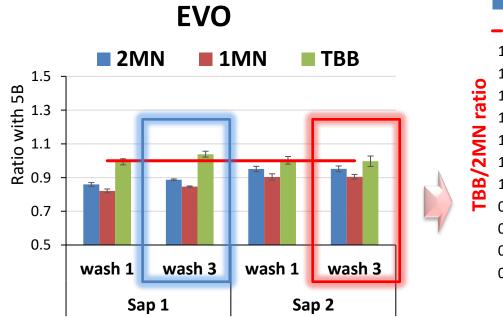
#### **Saponification**

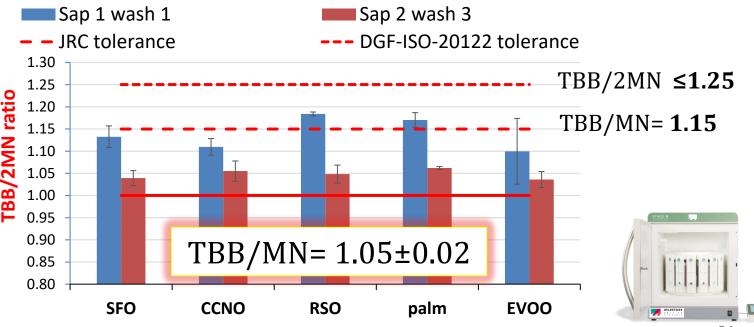
- Microwave assisted saponification-MAS with the conditions used in ISO 20122:2024.
  - 60 °C for 30 min
  - Comparing Sap: 1- 2 M KOH in EtOH/H2O (5/3 v/v) ISO 20122:2024
     2- 2 M KOH in EtOH/H2O (1/1 v/v).





Comparing Washing : 1- Add 5 mL of C6 + 5 mL EtOH/H2O (1/1 v/v). ISO 20122:2024
 3- 20 mL of H2O









2MN

wash 1

1.5

Ratio with 5B 1'1 0'0 0'0

0.7

0.5

**EVO** 

wash 3

Sap 1

1MN

wash 1

TBB

wash 3

Sap 2

## **Re-optimize the saponification step**

#### **Saponification**

Microwave assisted saponification-MAS

with the conditions used in ISO 20122:2024.

• 60 °C for 30 min

ratio

**FBB/2MN** 

Comparing **Sap: 1**- 2 M KOH in EtOH/H2O (**5/3 v/v**) <mark>ISO 20122:2024</mark> **2**- 2 M KOH in EtOH/H2O (**1/1 v/v**).



 Comparing Washing : 1- Add 5 mL of C6 + 5 mL EtOH/H2O (1/1 v/v). ISO 20122:2024 3- 20 mL of H2O

#### 2024 Posters Analytical

Session: Analytical Poster Session

#### Reducing the variability of the saponification step for MOSH&MOAH analysis using microwave-assisted extraction

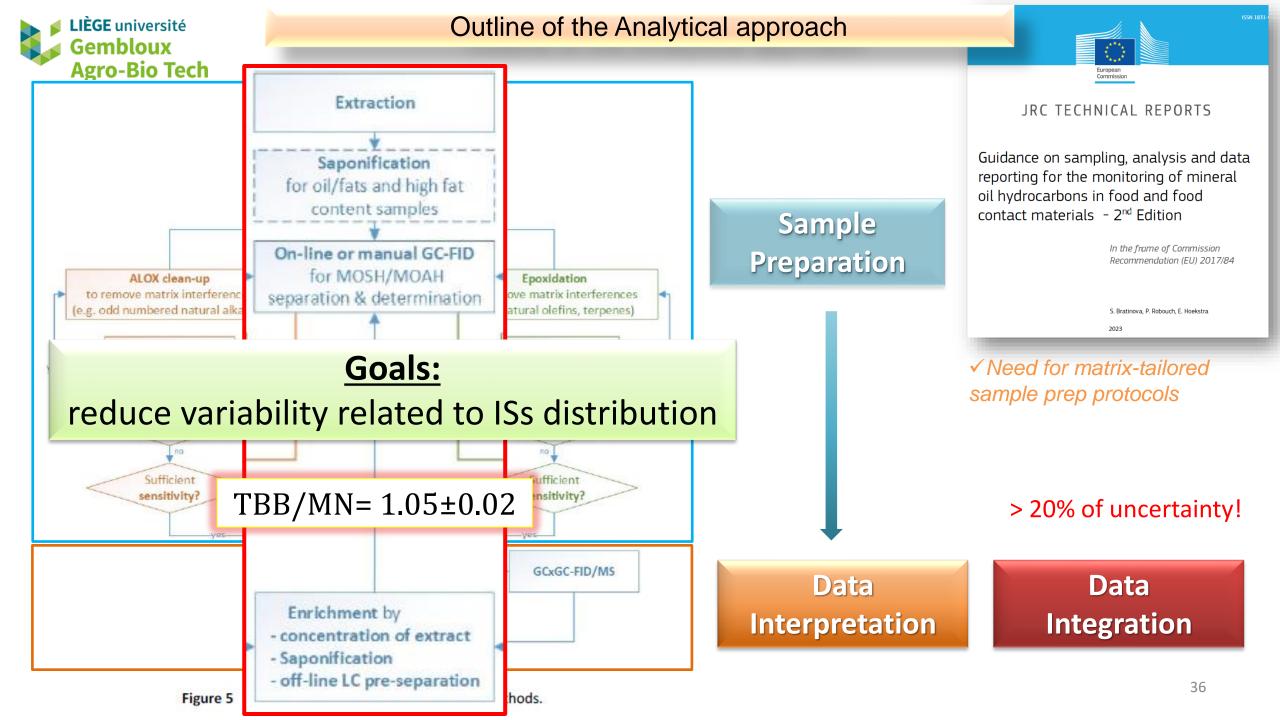
Monday, April 29, 2024 ④ 5:00 PM − 6:00 PM EDT

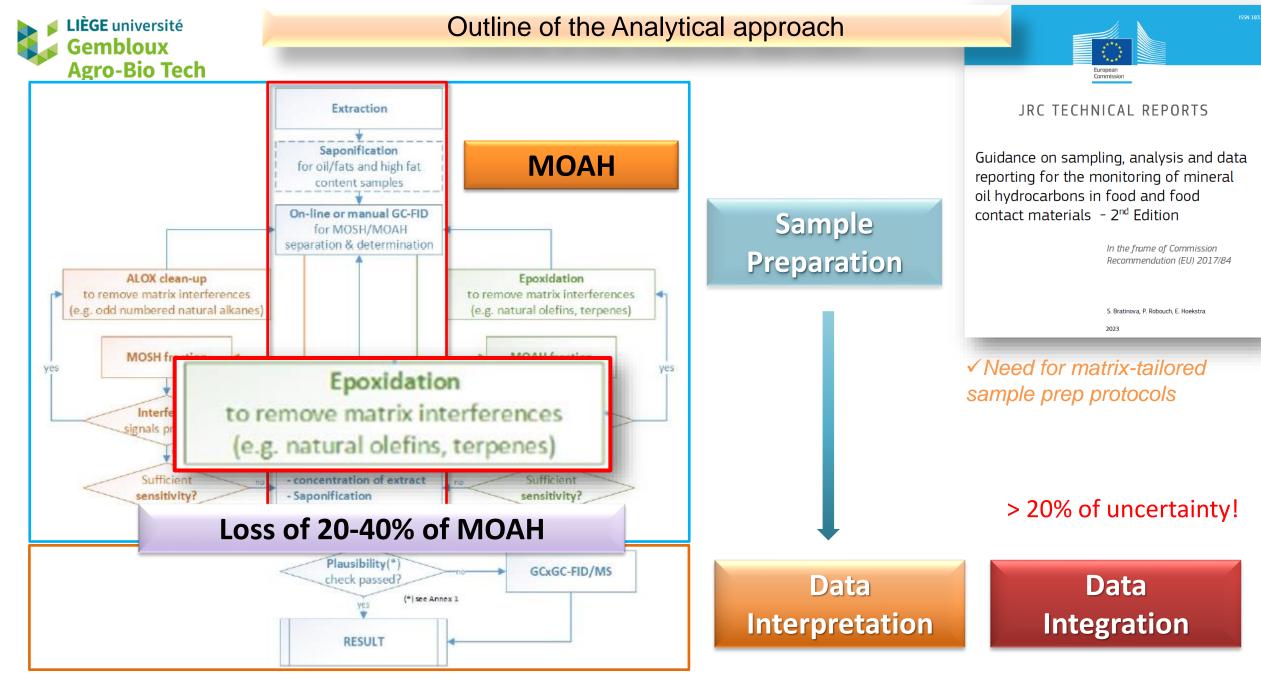
#### Presenting Author(s)

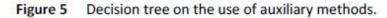


#### Aleksandra Gorska, Ir (she/her/hers)

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# **Alternative to Epoxidation**



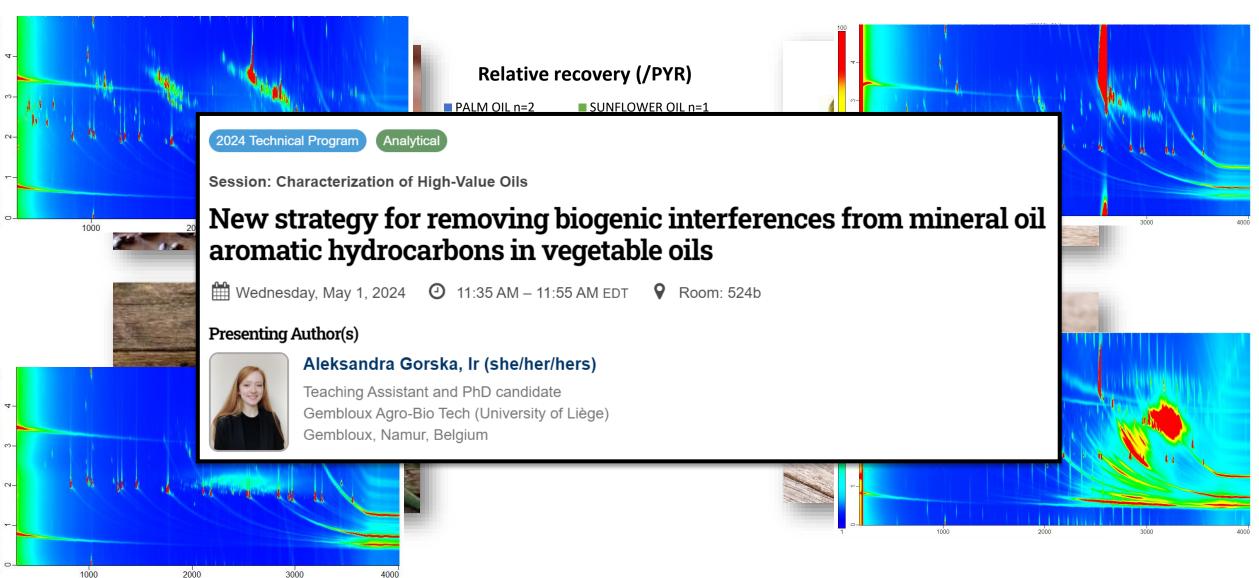






# **Alternative to Epoxidation**







#### **CONCLUSION AND FUTURE DIRECTIONS**



## **Reduction of uncertainty and analyst interpretation is fundamental**

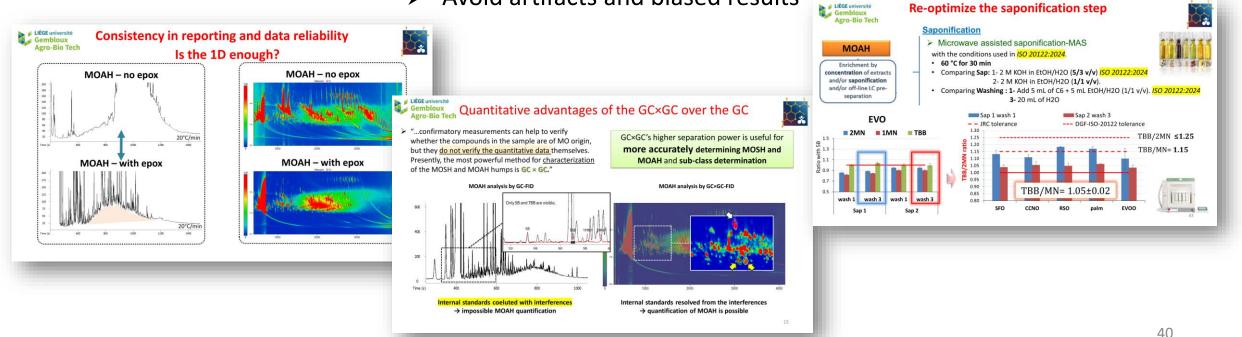
to guarantee more reliability in the results

#### **GC×GC** & sample preparation

- Reduce operator interpretation
- Handle complex matrices (e.g., essential oils)

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Avoid artifacts and biased results





#### My research group:

Sophie Vancraenenbroeck

Paula Albendea

Steven Mascrez

Damien Eggermont

Aleksandra Gorska

- Donatella Ferrara
- **Damien Pierret**
- **Grégory Bauwens**

Visiting students: Andrea Schincaglia Silvia Pranteddu

Pedro Bahia











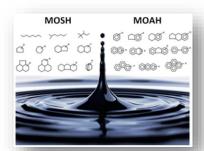
















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