

Differential Thermal Analysis to Assist the Design of Corrosion-resistant High Entropy Alloys for Laser Powder Bed Fusion

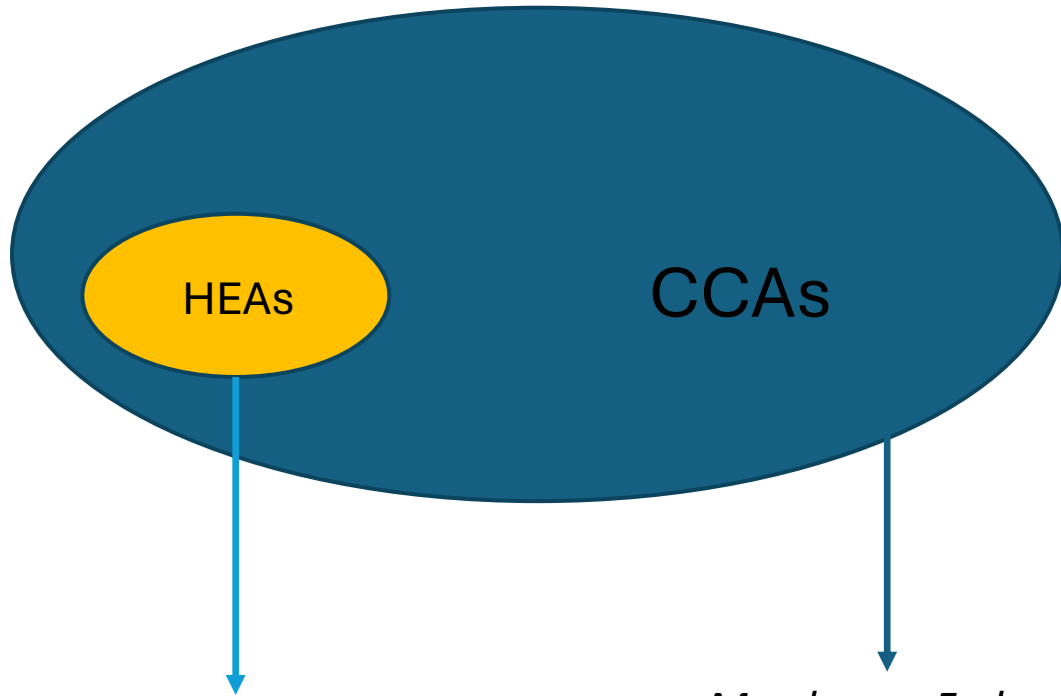
SEIDOU Herrim, BLONDIAU Catherine, DEDRY Olivier, OÑATE Angelo, TUNINETTI Victor, TCHUINDJANG Jérôme
Tchoufang and MERTENS Anne

Metallic Materials Science (MMS), Aerospace and Mechanical Engineering Dpt., University of Liège, Belgium



26/04/2024

Introduction



The common phases are FCC and BCC, while the IM are B2, Sigma and Laves.

BCC

- High Hardness
- Good electrical and thermal conductivity
- Low thermal expansion
- Good ductility

FCC

- Greater mechanical strength
- Better wear resistance
- **Better corrosion resistance**
- Better ductility

- *May have <5 elements*
- *>35% of elements*
- *Multiple phases, Intermetallics (IM) included*
- *Entropy does not matter*

The goal of this study is to design a corrosion-resistant HEA through LPBF

Crystal Structure Prediction

$$\Omega = Tm\Delta S_{mix} / \Delta H_{mix}$$

$$\Omega \geq 1.1$$

δ : atomic size difference

$$\delta \leq 6.6\%$$



Stable, Solid-solution phase

VEC: Valence electron concentration

$$VEC = \sum_{k=1}^n (c_k) VEC_k$$

$$VEC \geq 8$$



FCC

$$6.87 \leq VEC < 8$$



FCC + BCC

$$VEC < 6.87$$



BCC

TABLE I. Physiochemical properties for commonly used elements in HEAs.

Element	Atom radius (Å)	Pauling electronegativity	VEC
Al	1.432	1.61	3
B	0.820	2.04	3
C	0.773	2.55	4
Co	1.251	1.88	9
Cr	1.249	1.66	6
Fe	1.241	1.83	8
Mn	1.350	1.55	7
Mo	1.363	2.16	6
Nb	1.429	1.6	5
Ni	1.246	1.91	10
Ta	1.430	1.50	5
Ti	1.462	1.54	4
V	1.316	1.63	5
W	1.367	2.36	6

Guo S. et al., *J. Appl. Phys.* 15 May 2011; 109 (10): 103505. <https://doi.org/10.1063/1.3587228>

Materials and Methods

Selected elements: Al, Cr, Fe, Mn, and Ni

CrFeMnNi-based MEA

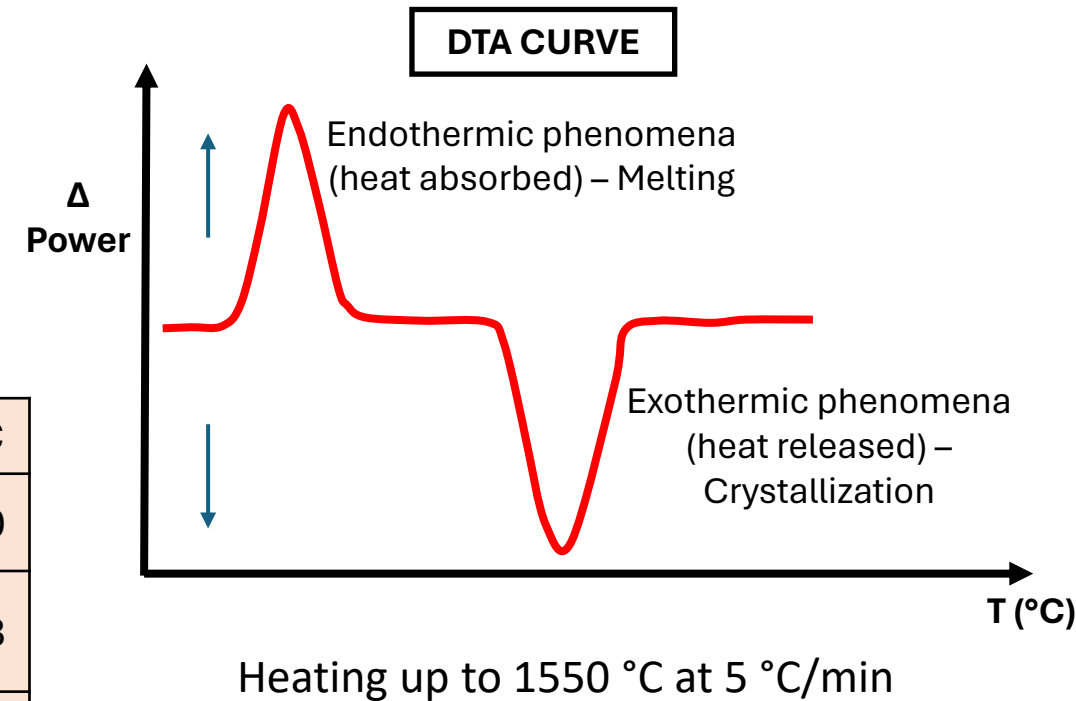
% at.	Cr	Fe	Mn	Ni	VEC
CrFeMnNi	25	25	25	25	7.75
Cr ₂ Fe ₂ MnNi	33.33	33.33	16.67	16.67	7.50
CrFe ₂ MnNi ₂	16.67	33.33	16.67	33.33	8.17

AlCrFeMnNi-based HEA

% at.	Al	Cr	Fe	Mn	Ni	VEC
AlCrFeMnNi	20	20	20	20	20	6.80
AlCrFe ₂ MnNi ₂	14.29	14.29	28.57	14.29	28.57	7.43
AlCrFe ₂ Ni ₂	16.67	16.67	33.33	-	33.33	7.50

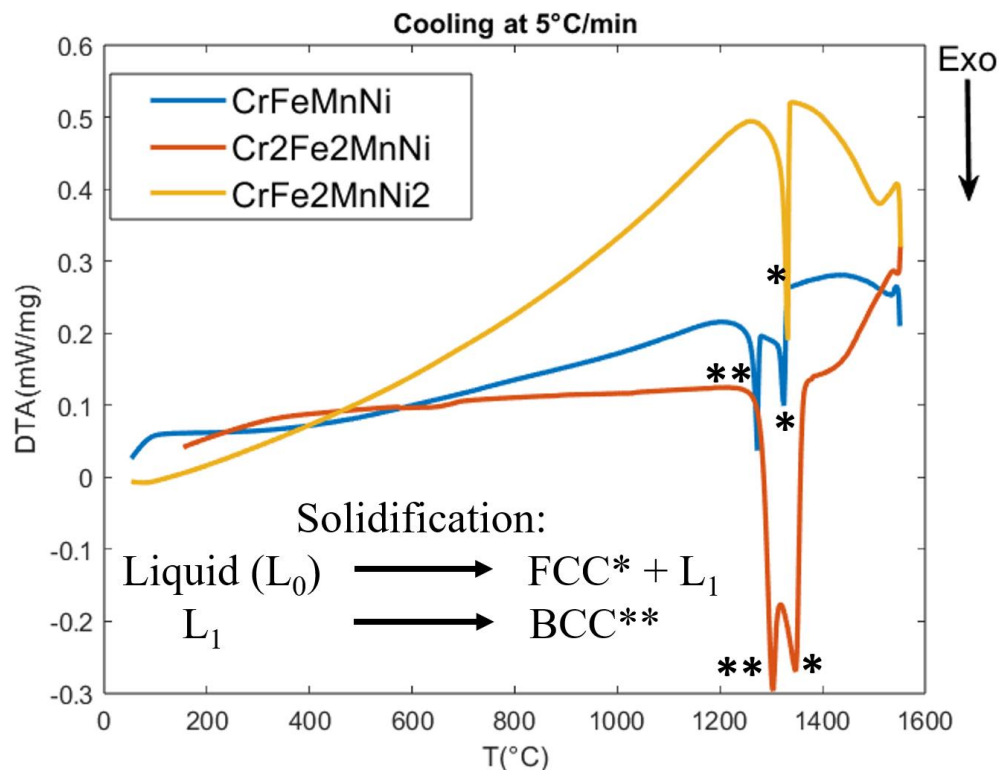
Differential Thermal Analysis (DTA)

Pre-screening of compositions

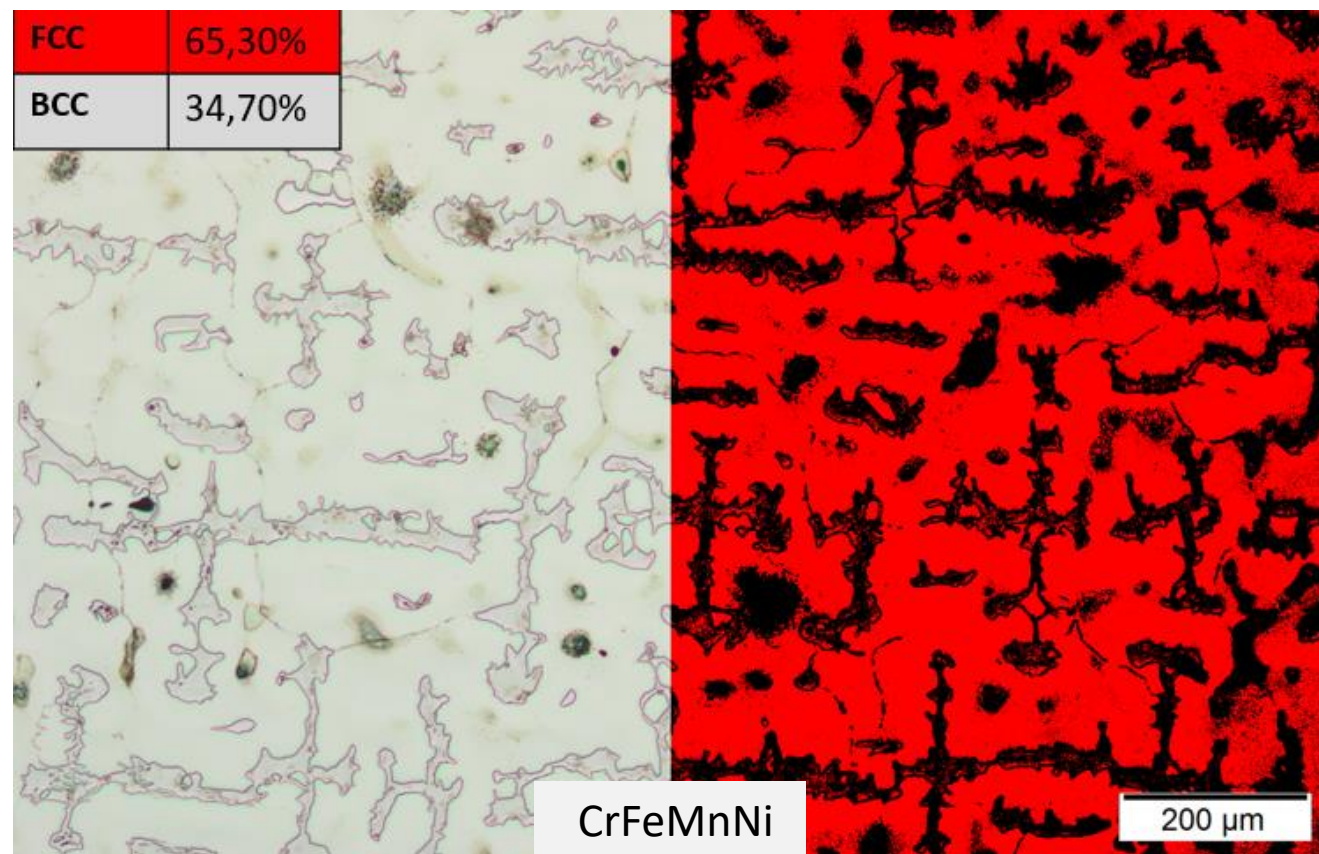
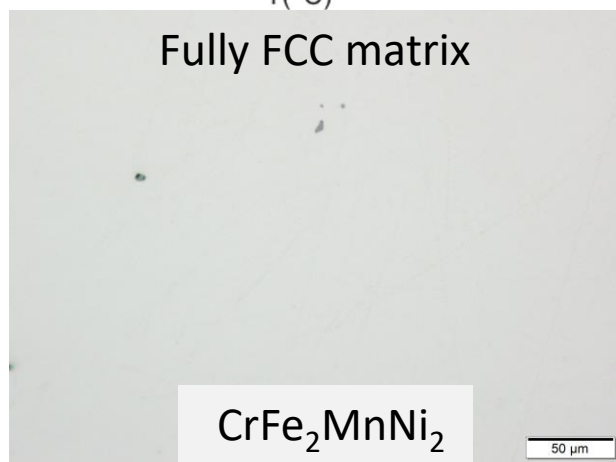


CrFeMnNi-based MEA: Solidification sequence

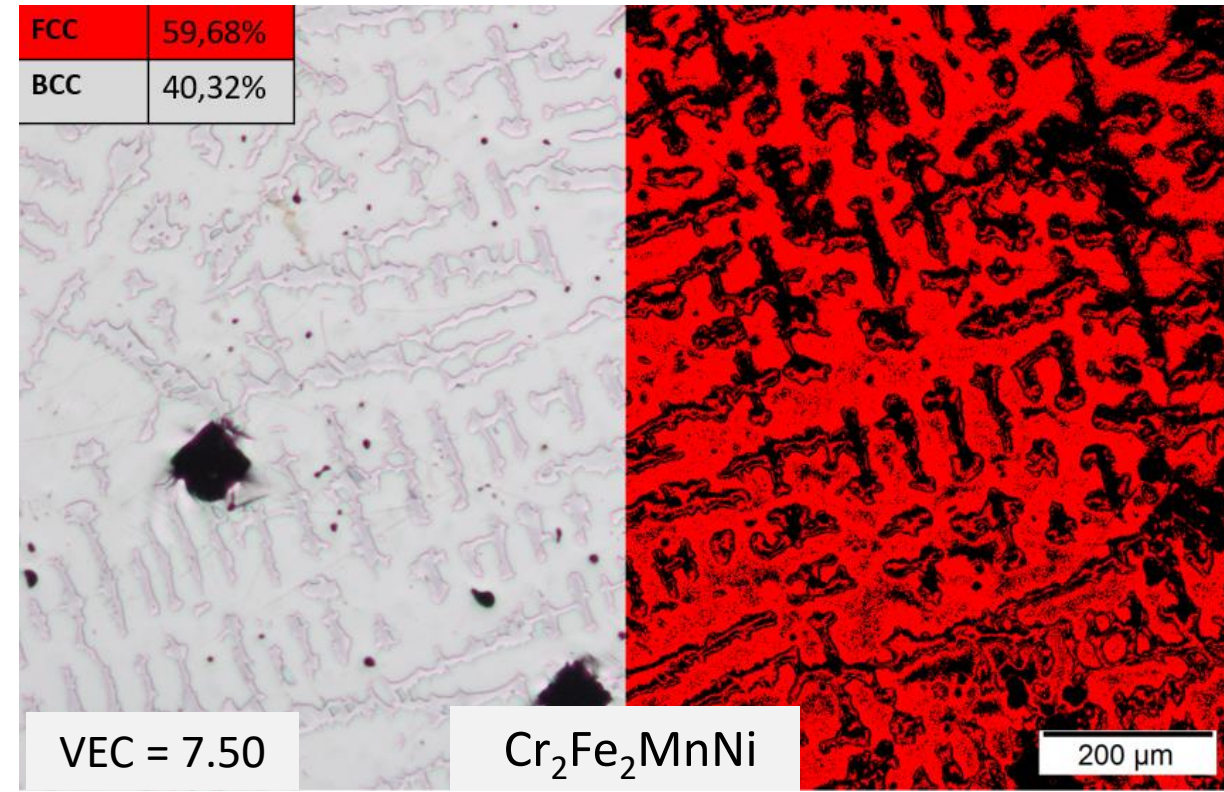
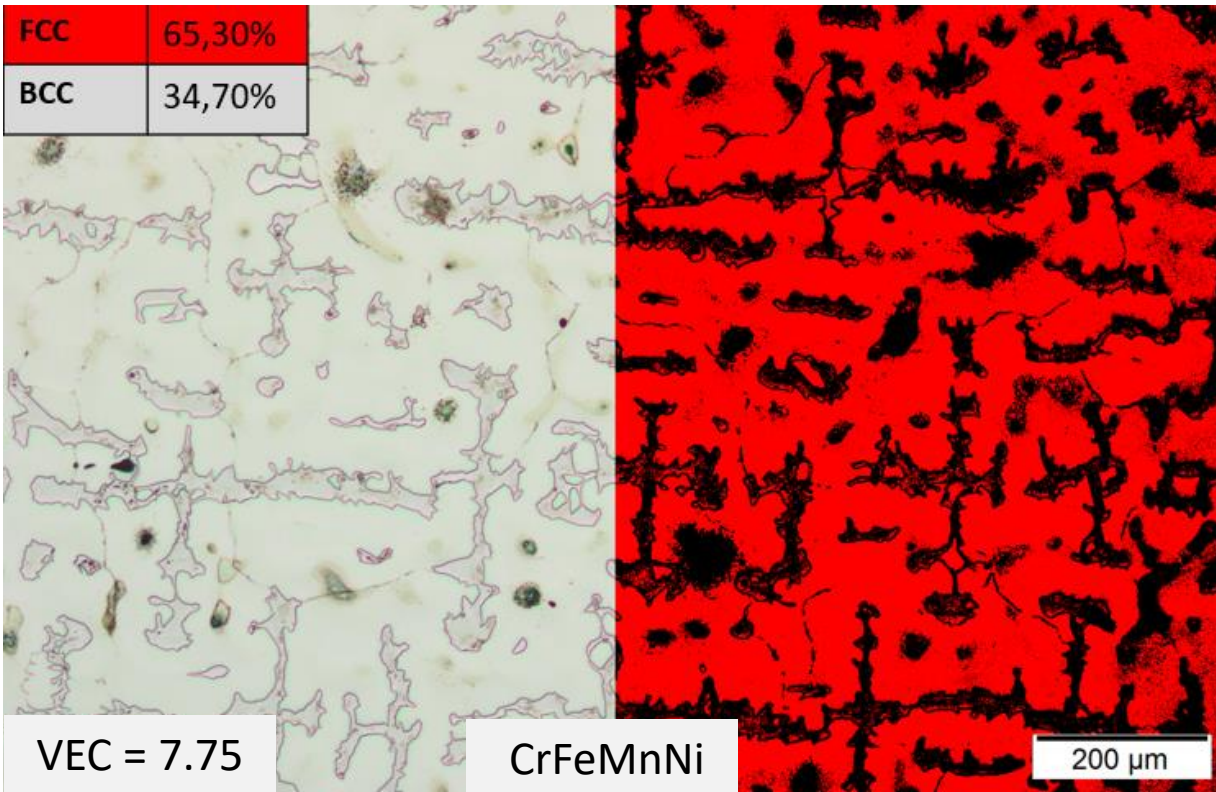
% at.	Cr	Fe	Mn	Ni	VEC
CrFeMnNi	25	25	25	25	7.75
Cr ₂ Fe ₂ MnNi	33.33	33.33	16.67	16.67	7.50
CrFe ₂ MnNi ₂	16.67	33.33	16.67	33.33	8.17



$6.87 \leq VEC < 8$
 BCC FCC



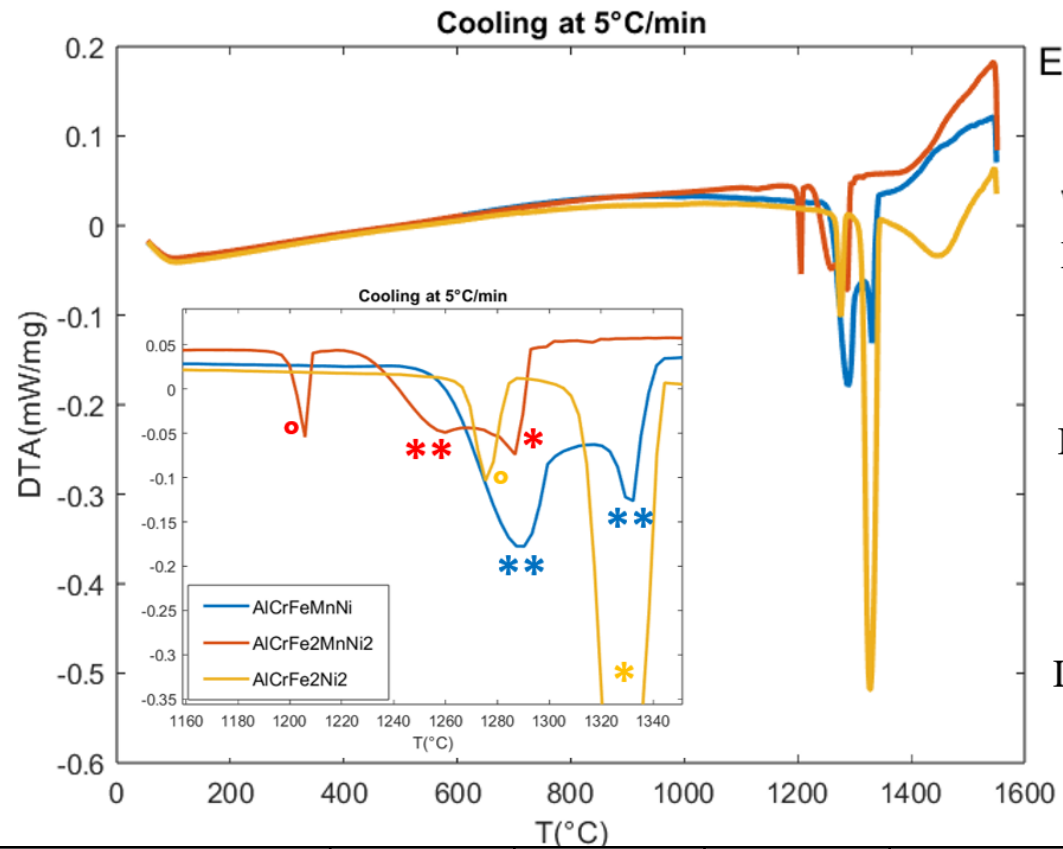
CrFeMnNi-based MEA: Composition and quantification of the phases



% at.	Cr	Fe	Mn	Ni
BCC	46.2 ± 2.0	30.3 ± 1.0	17.5 ± 0.4	6.0 ± 1.4
FCC	25.0 ± 1.4	29.5 ± 2.5	26.0 ± 2.0	19.5 ± 1.7

% at.	Cr	Fe	Mn	Ni
BCC	47.0 ± 0.7	33.2 ± 0.5	10.9 ± 0.4	8.9 ± 0.5
FCC	28.4 ± 1.0	34.8 ± 1.5	14.8 ± 1.0	22.0 ± 1.4

AlCrFeMnNi-based HEA: Solidification sequence

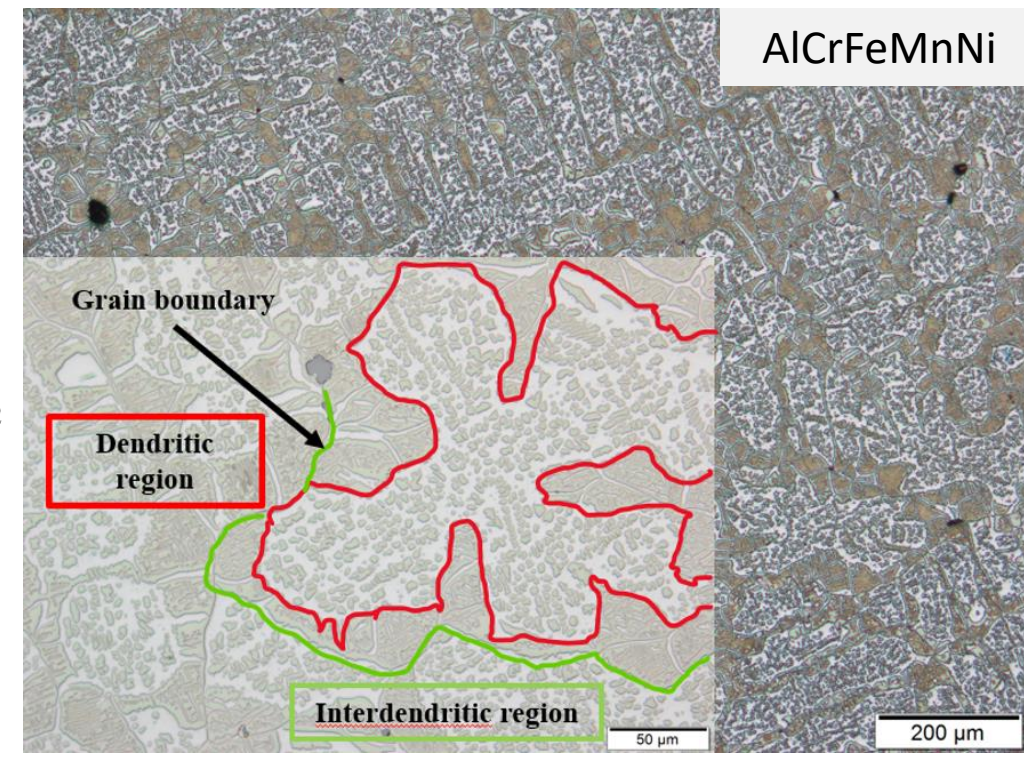


Exo ↓

AlCrFeMnNi
 Liquid (L₀) → BCC** + L₁
 L₁ → BCC**

AlCrFe₂MnNi₂
 Liquid (L₀) → FCC* + L₁
 L₁ → BCC** + L₂
 L₂ → FCC/BCC^o

AlCrFe₂Ni₂
 Liquid (L₀) → FCC* + L₁
 L₁ → FCC/BCC^o

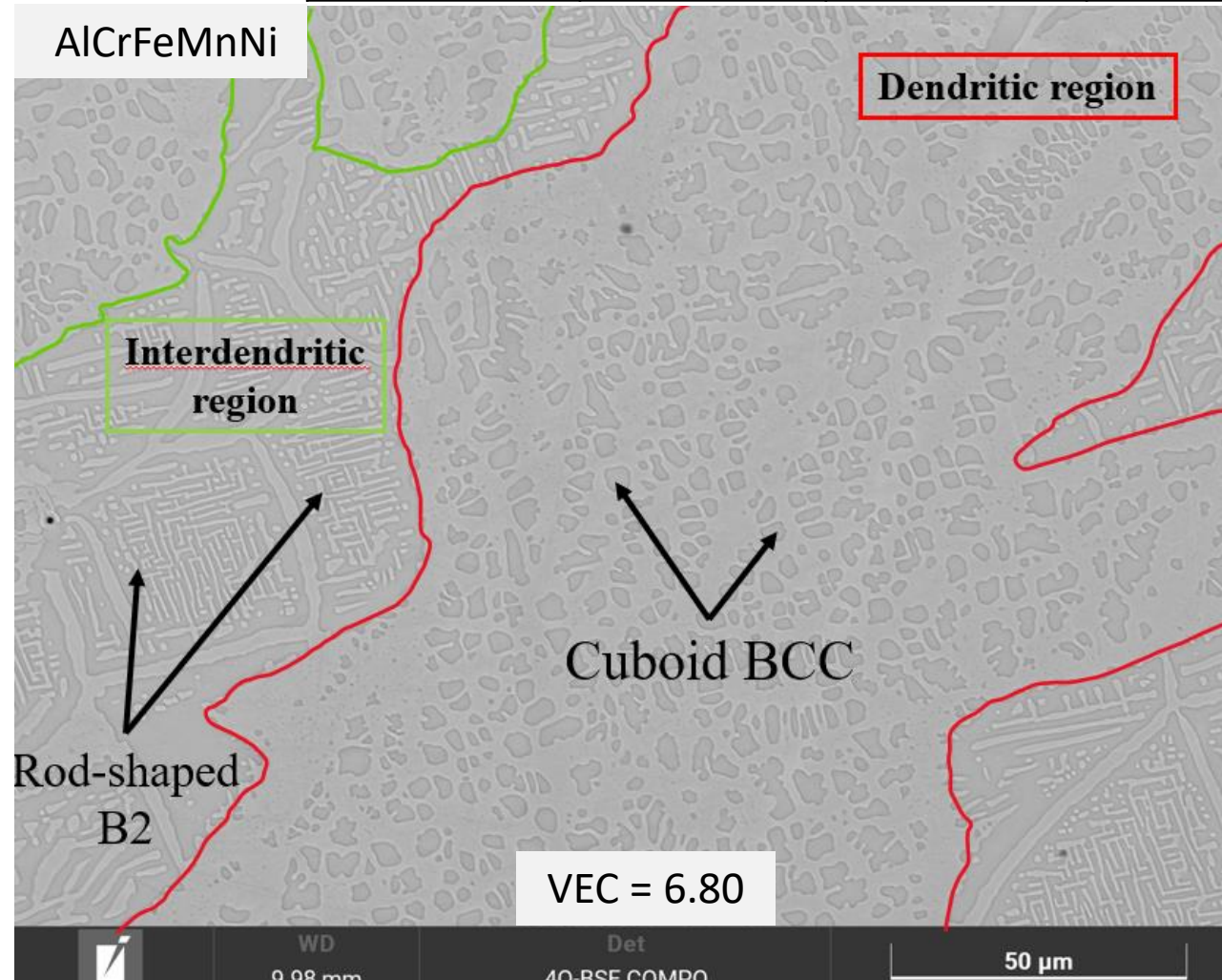


% at.	Al	Cr	Fe	Mn	Ni	VEC
AlCrFeMnNi	20	20	20	20	20	6.80
AlCrFe ₂ MnNi ₂	14.29	14.29	28.57	14.29	28.57	7.43
AlCrFe ₂ Ni ₂	16.67	16.67	33.33	-	33.33	7.50

$6.87 \leq VEC < 8$
 BCC FCC

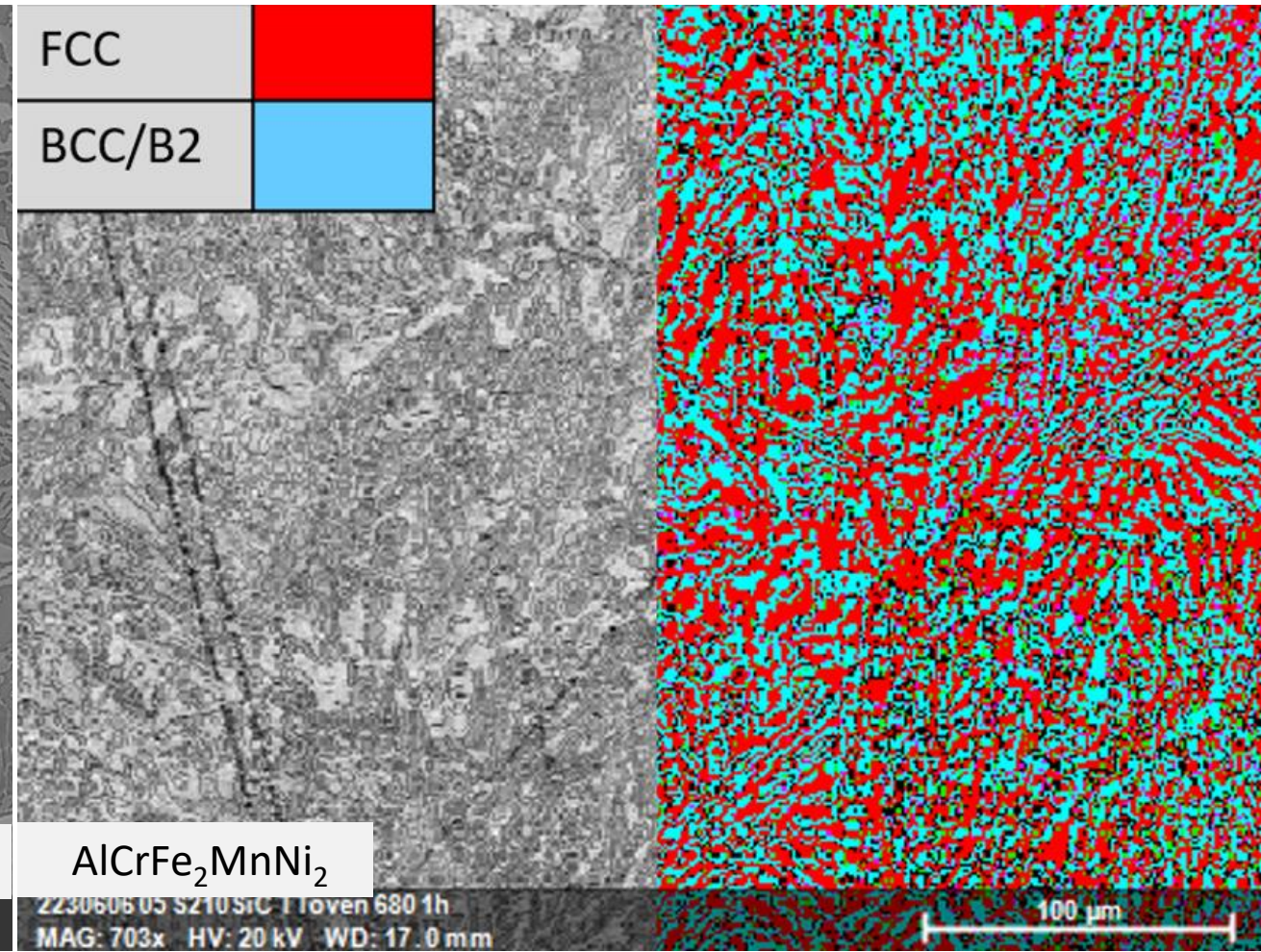
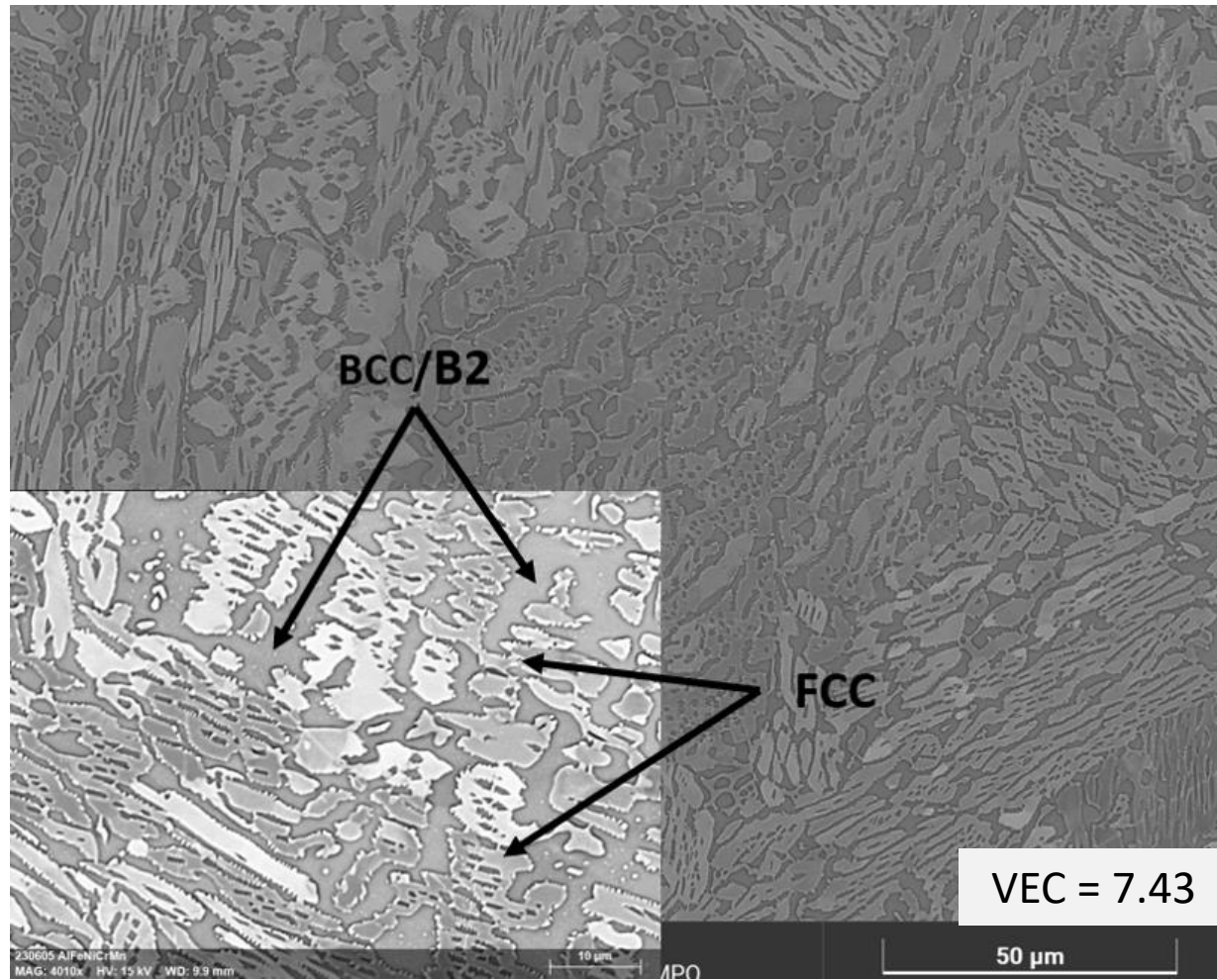
AlCrFeMnNi-based HEA: Composition of the phases

% at.	Al	Cr	Fe	Mn	Ni
Precipitates	29.4 ± 4.0	6.5 ± 3.0	9.0 ± 4.3	16.0 ± 1.5	39.1±7.9
Matrix	7.7 ± 1.1	34.6 ± 1.8	30.0 ± 0.9	24.7 ± 0.5	3.1 ± 1.6



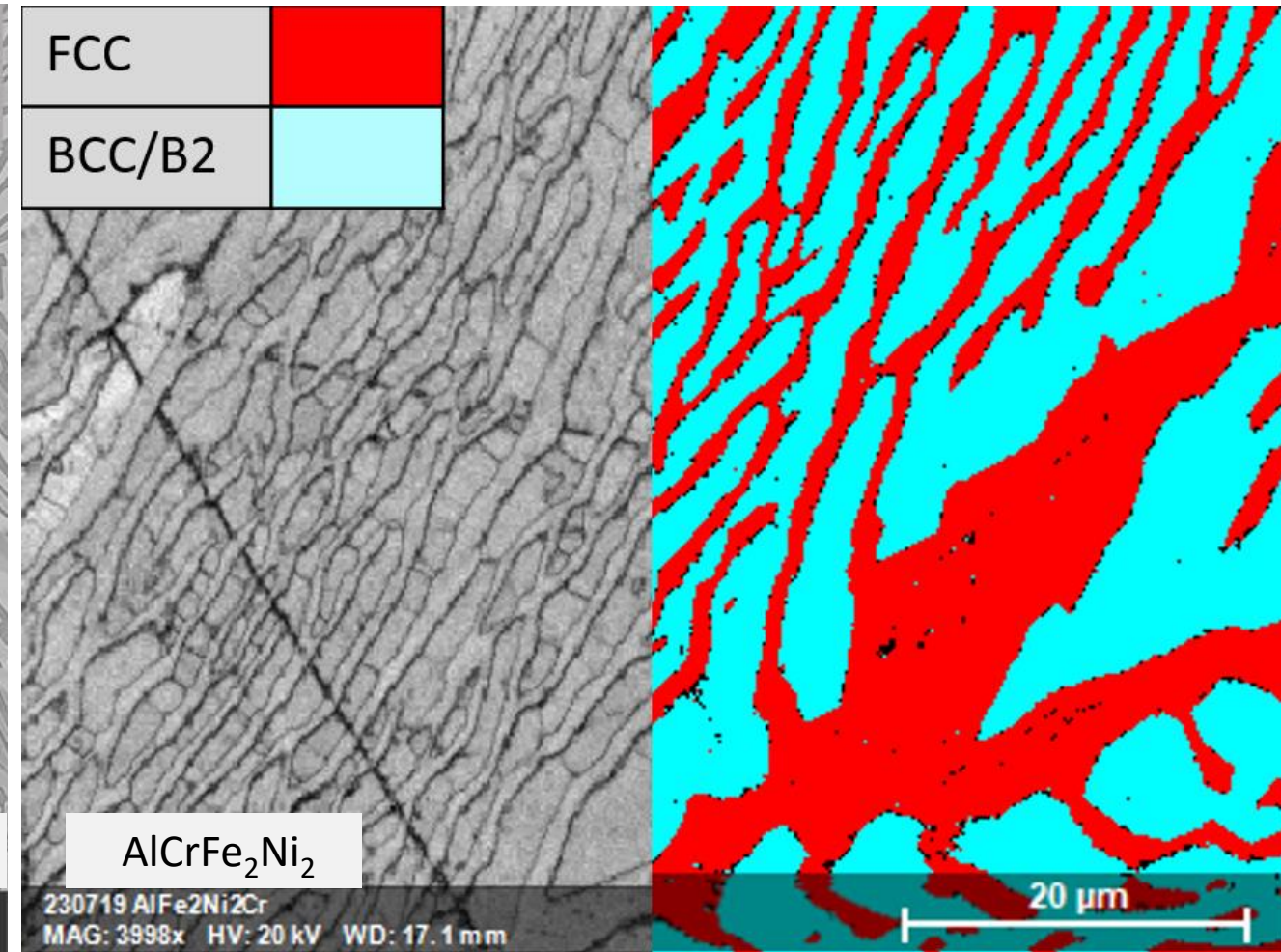
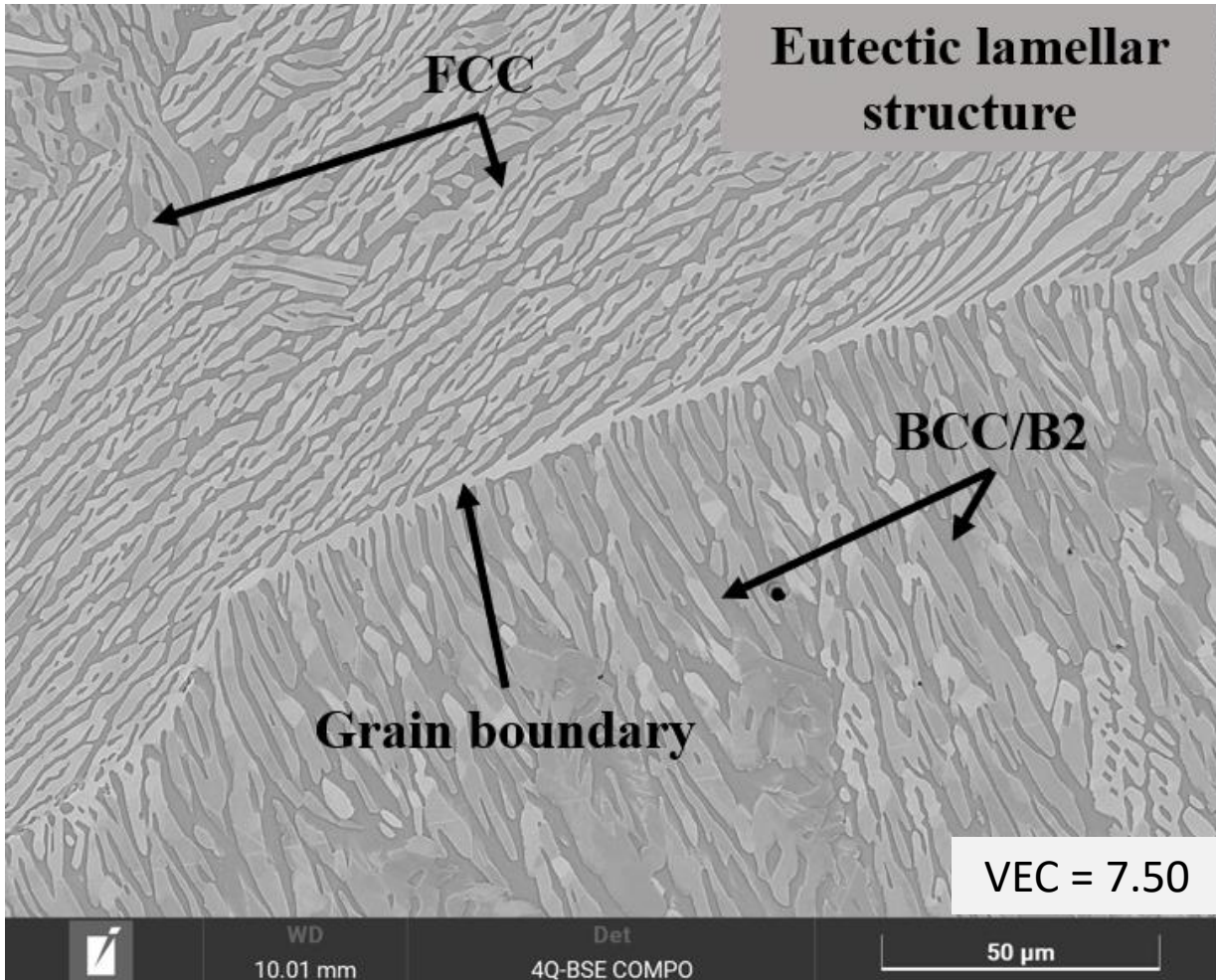
AlCrFeMnNi-based HEA: Composition and EBSD analyses

% at.	Al	Cr	Fe	Mn	Ni
FCC	4.7 ± 0.4	18.6 ± 0.9	39.6 ± 0.6	14.2 ± 0.3	22.9 ± 0.8
B2	26.0 ± 0.6	2.1 ± 0.2	7.9 ± 0.4	15.9 ± 0.5	48.1 ± 0.5
BCC	20.7 ± 4.3	9,3 ± 6.8	14.3 ± 1.4	15.4 ± 7	38.4 ± 9.8



AlCrFeMnNi-based HEA: Composition and EBSD analyses

% at.	Al	Cr	Fe	Ni
FCC	6.8 ± 0.5	22.8 ± 0.1	41.9 ± 1.1	28.4 ± 0.7
B2	29.7 ± 1.2	4.9 ± 0.4	14.2 ± 0.6	51.1 ± 0.2
BCC	27.0 ± 1.5	10.41 ± 2.9	18.6 ± 1.8	44.0 ± 3.1



Conclusions

- VEC crystal prediction theory is a useful tool to predict the first phases that solidify
- VEC theory does not consider solid-state transformations
- In Al, Cr, Fe, Mn and Ni system, FCC phase is Cr-Fe rich while BCC phase is rich in Al and Ni
- Al promotes the BCC formation, followed by spinodal decomposition

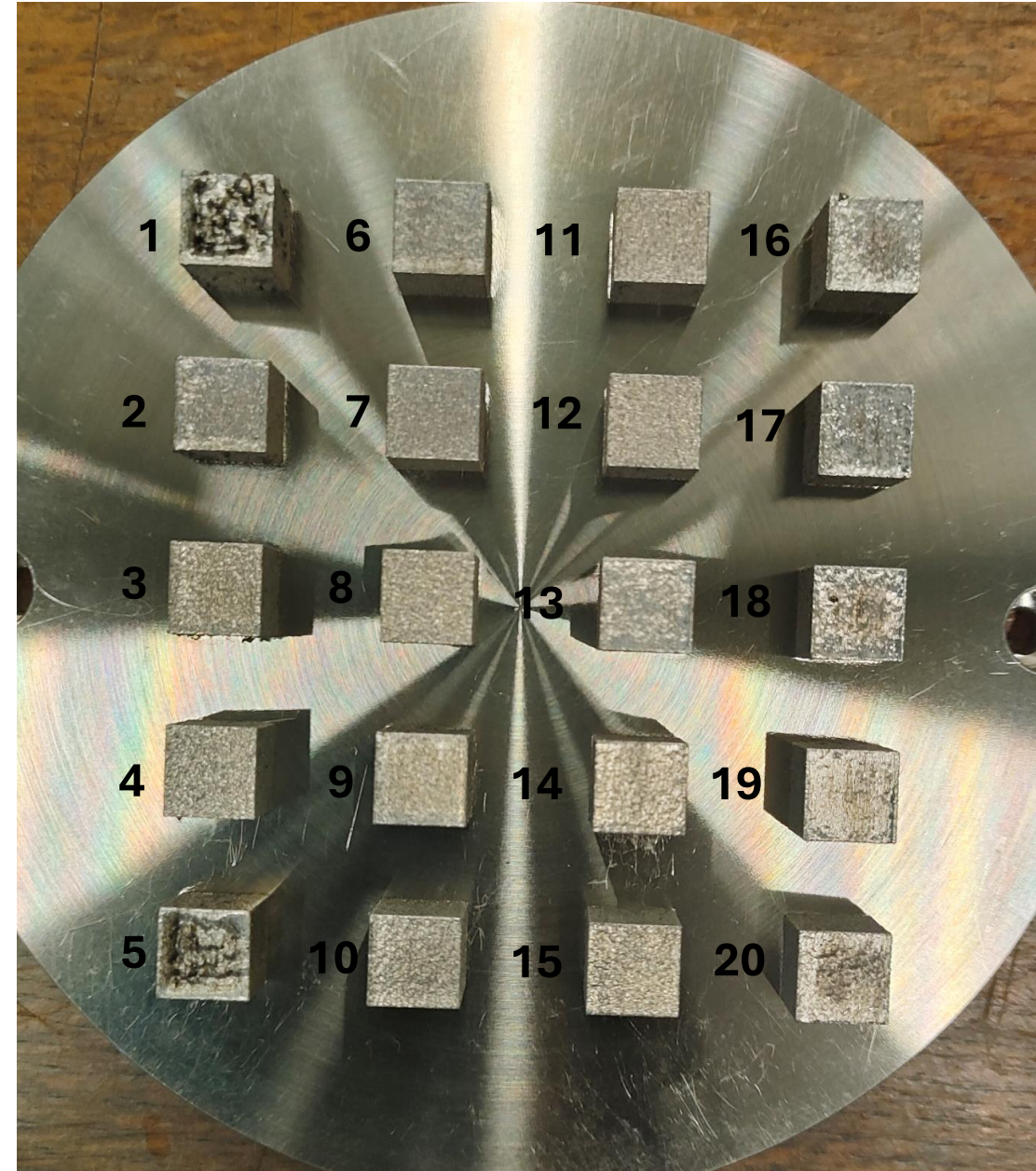
- DTA is a rapid and cost-effective test to investigate the transformations that occur at near-equilibrium conditions
- The near-equilibrium results are used to understand the out-of-equilibrium phenomena and mechanisms of LPBF
- This method guide the design of new alloys and the thermal treatments

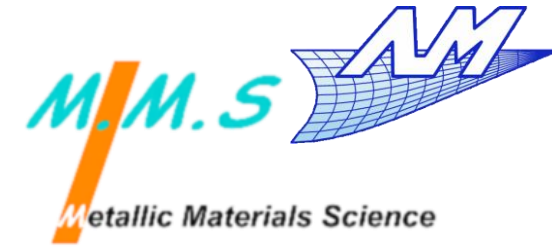
Perspectives

- A composition has been chosen for LPBF
- The parameters of the LPBF were optimized
- Cubic samples were printed



Surface and microstructural characterization





Thanks for your attention



Herrim Seidou, PhD candidate

Ahb.seidou@uliege.be

Metallic Materials Science (MMS) team
Aerospace and Mechanical Engineering Dpt.
University of Liège, Belgium

All publications of the Metallic Materials Science team are available at :

<https://orbi.uliege.be/>