



### Recycling brick fines for new alcaliactivated binders

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#### Context





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#### Context

Flow of brick waste: 1-2% of C&DW in BE/North of France

Valorization

Reuse of bricks Aggregates: landfilling/recycling for backfilling Brick fine particles









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# **Objectives**

#### Brick fine particles treatment

Increase specific surface

Activate amorphous characteristics

#### Two ways of valorisation

Supplementary cementitious material

Alcali-activated material

Investigations on paste



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### **Objectives**

# Brick fine particles treatment grinding crushing Jaw crusher Impact crusher Bloc 238x138x138 mm

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Brick fine particles/GGBFS granulometry



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#### Brick fine particles

3 types of granulometry

#### B1: $d_{50} = 3.3 \mu m$ (with supplementary cyclogrinding)

B2:  $d_{50} = 20 \ \mu m$ Specific surface, BET (m²/kg)B3:  $d_{50} = 190 \ \mu m$ Specific surface, BET (m²/kg)Ground Granulated Blast Furnace SlagWater absorption (%)Granulometry (µm)d10

	Brick fines	GGBFS
	B2	
Specific surface, BET (m <sup>2</sup> /kg)	833	1
Water absorption (%)	1.1	-
Granulometry (µm)		
d10	1.95	1
d50	19.1	8.5
d90	56.6	30
Ca(OH) <sub>2</sub> quantity fixed (mg/g brick fines)	394	-



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#### Brick fine particles/GGBFS mineralogy

Oxides (%)	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	Fe <sub>2</sub> O <sub>3</sub>	K <sub>2</sub> O	Na <sub>2</sub> O	MgO	TiO <sub>2</sub>	Total
Brick fine	1.7	62.8	10.4	16.3	2.1	0.6	2.2	2.4	99.3
GGBFS	42.9	38	10.8	0.5	0.3	-	6.5	0.7	99.5

Mineral (%)	Brick fine		
Quartz SiO <sub>2</sub>	58.6		
Hematite Fe <sub>2</sub> O <sub>3</sub>	12.8		
Albite NaAlSi <sub>3</sub> O <sub>8</sub>	3.9		
Microline KAlSi <sub>3</sub> O <sub>8</sub>	6.0		
Cristobalite SiO <sub>2</sub>	2.8		
Amorphicity	15.9		

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#### Two hypothesis

- BL : brick fines = mineral addition
  - Concentration of the alcali-activating solution calculated versus GGBFS mass

#### • BLM : brick fines = precursor like GGBFS

• Concentration of the alcali-activating solution calculated versus GGBFS+brick fines mass



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#### Two hypothesis

Control 100% GGBFS

GGBFS
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BL	BLM
Concentration 🖌	Constant Concentration
BL 10%	BML 10%
BL 20%	BML 20%
BL 30%	BML 30%
BL 50%	BML 50%



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Evolution of spread with time

Short time of maniability with AAM

Slowing down of "stiffening" with continuous mixing on BL 30% mix

### Results





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#### Evolution of setting time

Setting time usually lower with AAM

BL: hardening time *¬* from 30 % substitution

BLM: hardening time *¬* from 50 % substitution

### Results





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#### Development of hydrates (TGA) – loss of mass

Impact on phase precipitation from 50% substitution

BL more impacted than BLM

Samples	C-A-S-H (%)		Hydrotalcite (%)		Total mass loss (%)	
	7 days	90 days	7 days	90 days	7 days	90 days
GGBFS	4,9	6,4	3,9	4,2	8,7	10,6
BL 10%	4,4	5,7	3,7	3,5	8,1	9,1
BLM 10%	4,7	6,1	3,4	4,6	8	10,6
BL 50%	2,8	3,5	4,5	2,8	7,3	6,3
BLM 50%	4,1	4,4	3,5	3,1	7,6	7,6

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#### Development of hydrates (TGA) – loss of mass

BLM: brick fines react with AA solution  $\rightarrow$  compensation of GGBFS

Samples	С-А-Ѕ-Н (%)		Hydrotalcite (%)		Total mass loss (%)	
	7 days	90 days	7 days	90 days	7 days	90 days
GGBFS	4,9	6,4	3,9	4,2	8,7	10,6
BL 10%	4,4	5,7	3,7	3,5	8,1	9,1
BLM 10%	4,7	6,1	3,4	4,6	8	10,6
BL 50%	2,8	3,5	4,5	2,8	7,3	6,3
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#### Poral distribution

Finer porosity with time for all the mixes

Finer porosity with BLM 50% than BL 50

### Results





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Mechanical strength BL: slower kinetics – Rc ↓ when [brick fines] ↑ BLM: quicklier kinetics -Rc ≥ GGBFS from 90 days



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BL : brick fines = mineral addition

- ↗ Substitution 30 % ↗ Workability if continous mixing ↘ stiffening time
- ↗ [Brick fines] ↗ time for casting
- ↘ compressive strength but not proportional to substitution rate

→ Economy on activator

BLM : brick fines = precursor = GGBFS

- $\nearrow$  [Brick fines]  $\rightarrow$  Workability and consistency constant
- ↗ alcali-activation kinetics and hydrates production
- compressive strength at 90 days ≥ GGBFS

→ Economy on precursor





# Acknowledgements

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