

Université de Liège  
Faculté des Sciences  
Département de Géologie  
Laboratoire de Minéralogie



# The structural complexity of phosphates in the $\text{Na-Fe}^{2+}\text{-Fe}^{3+}$ (+ $\text{PO}_4$ ) system

Frédéric Hatert  
ECM-24

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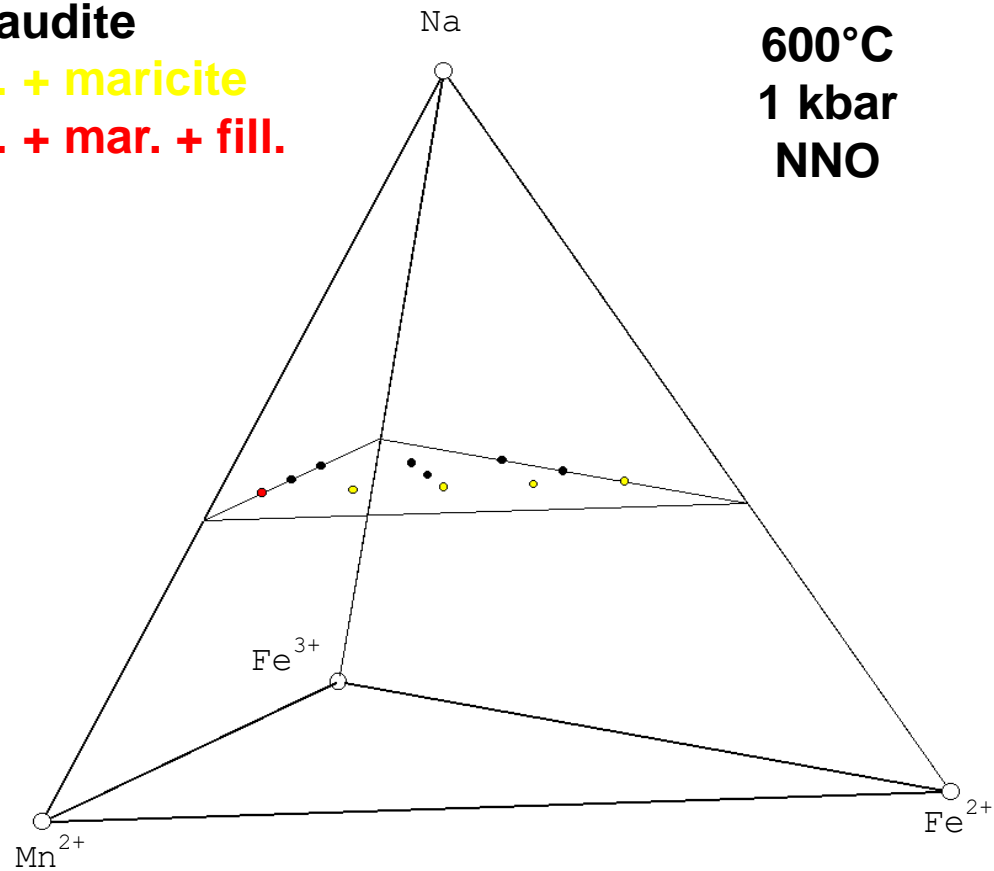


**Alluaudite, Buranga pegmatite, Rwanda**

# The Na-Mn-Fe<sup>2+</sup>-Fe<sup>3+</sup>-P-O system

- Alluaudite
- Allu. + maricite
- Allu. + mar. + fill.

600°C  
1 kbar  
NNO



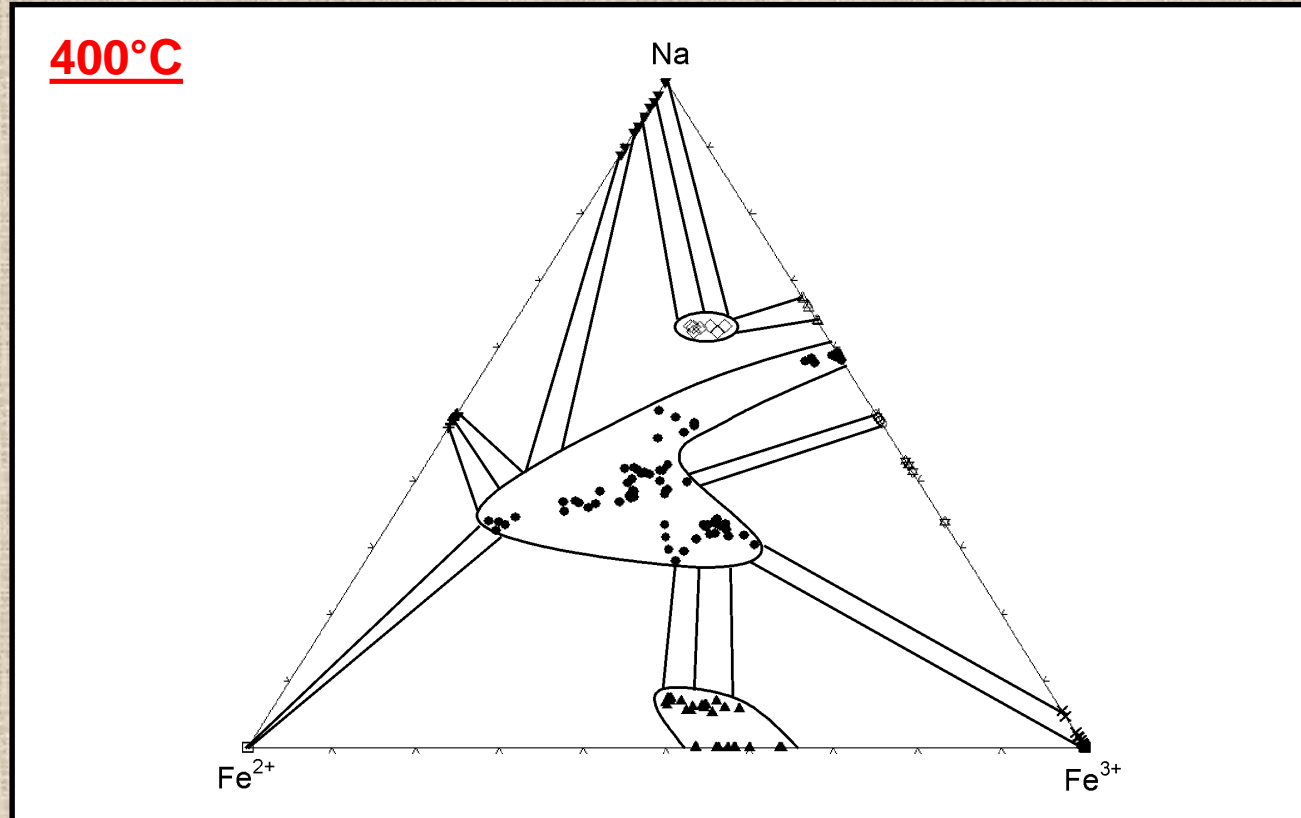
# Experimental



## Na-Fe<sup>2+</sup>-Fe<sup>3+</sup> (+ PO<sub>4</sub>) system

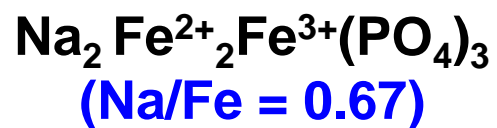
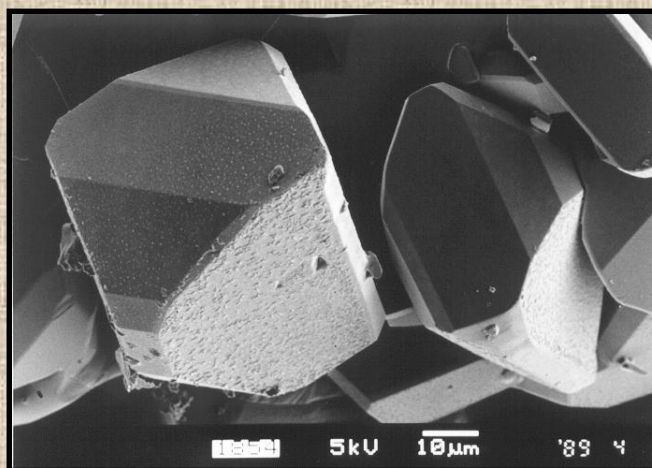
- Hydrothermal synthesis
- Tuttle-type cold-seal bombs
- T = 400-700 °C
- P = 1 kbar
- Oxygen fugacity: close to Ni/NiO (NNO)

# Phase diagram at 400°C / 1 kbar



- **Center**  $\Rightarrow$  alluaudite
- **Fe<sup>3+</sup> part**  $\Rightarrow$   $\text{Fe}^{3+}_4(\text{PO}_4)_3(\text{OH})_3$
- **Fe<sup>2+</sup> part**  $\Rightarrow$   $\text{Fe}^{2+}_3(\text{PO}_4)_2$  (sarcopside)
- **Na-rich part**  $\Rightarrow$   $\text{Na}_2\text{HPO}_4 \cdot n\text{H}_2\text{O}$
- $\blacktriangle$   $\Rightarrow$   $\text{Fe}^{3+}_4\text{Fe}^{2+}_3(\text{PO}_4)_6$
- $\Delta$   $\Rightarrow$   $\text{Na}_2\text{Fe}^{3+}(\text{HPO}_4)_2(\text{OH})$  (Phase A)
- $\diamond$   $\Rightarrow$   $\text{Na}_7\text{Fe}^{3+}_3\text{Fe}^{2+}(\text{PO}_4)_6$
- $+$   $\Rightarrow$   $\text{NaFe}^{2+}(\text{PO}_4)$  (maricite)

# Alluaudite-type phosphates



***C2/c***

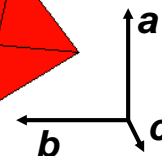
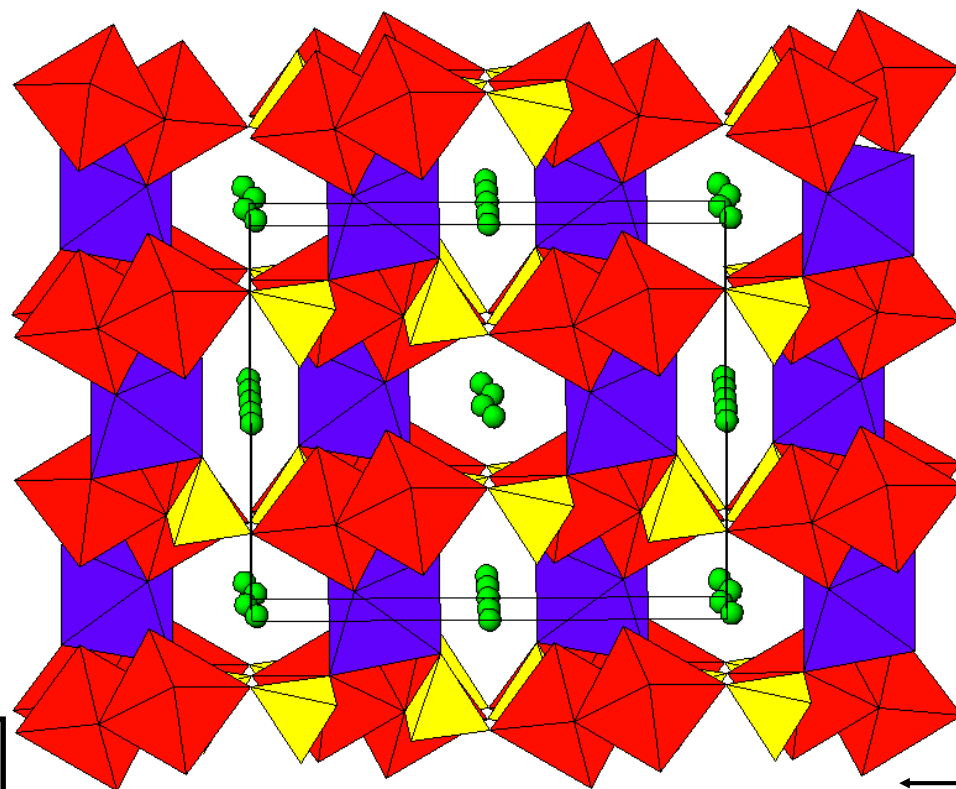
$$a = 12.001(2) \text{ \AA}$$

$$b = 12.538(2) \text{ \AA}$$

$$c = 6.405(1) \text{ \AA}$$

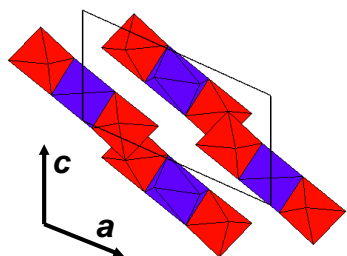
$$\beta = 114.45(1)^\circ$$

**Chains of edge-sharing  
M octahedra**



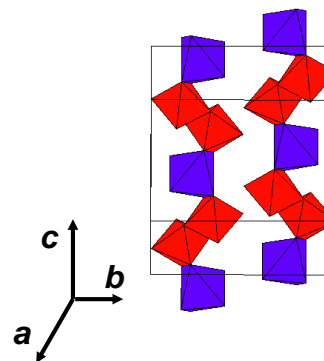
# Alluaudite-type phosphates

## M(1)-M(2) chains

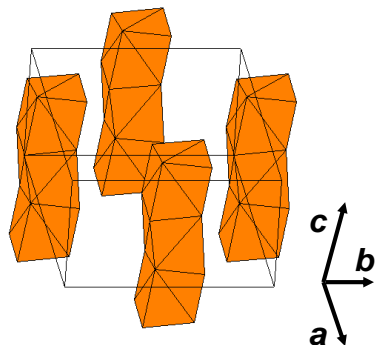


M(1):  $\text{Fe}^{2+}$   
M(2):  $\text{Fe}^{2+}, \text{Fe}^{3+}$

## M(1)-M(2) chains

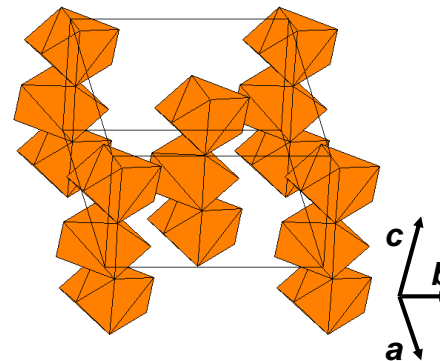


## A(1) chains



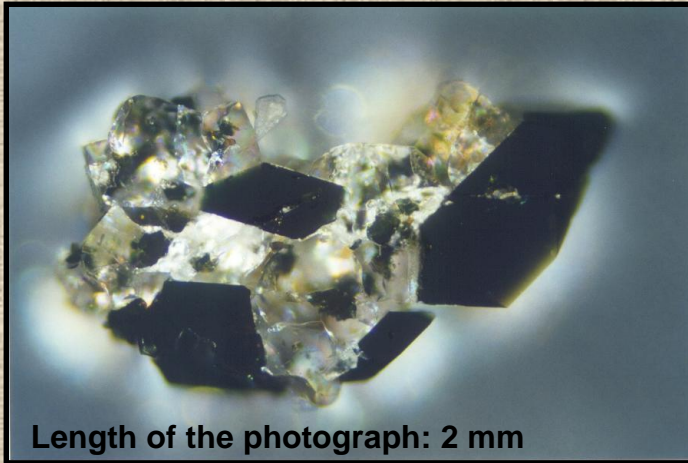
A(1): Na

## A(2)' chains



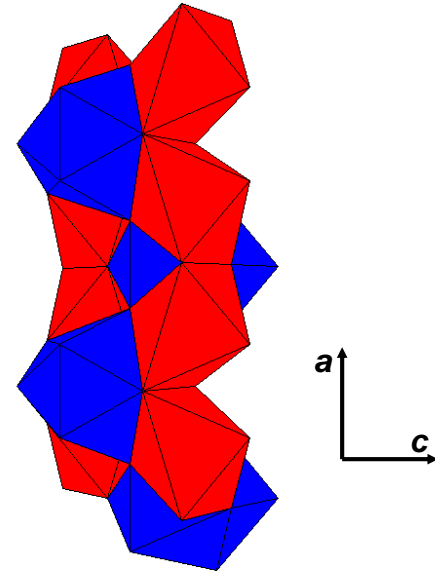
A(2)': Na

# Maricite-type phosphates



**Chains of edge-sharing  
M(1) octahedra**

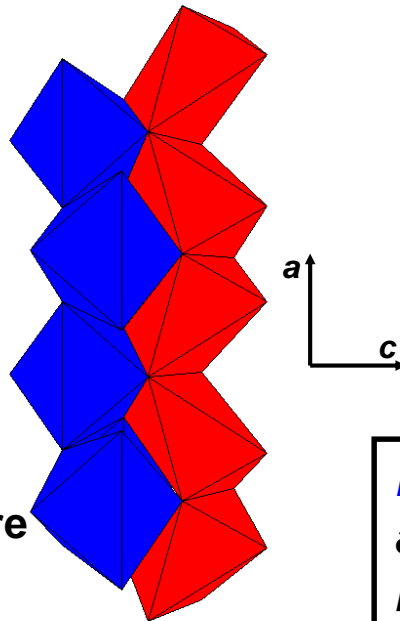
Maricite  
 $\text{NaFe}^{2+}(\text{PO}_4)$   
(Na/Fe = 1)



M(1):  $[\text{Fe}^{2+}]_6$   
M(2):  $[\text{Na}]_{10}$

***Pmnb***  
 $a = 6.864(4) \text{ \AA}$   
 $b = 8.986(6) \text{ \AA}$   
 $c = 5.040(2) \text{ \AA}$

Natrophilite  
 $\text{NaMn}(\text{PO}_4)$

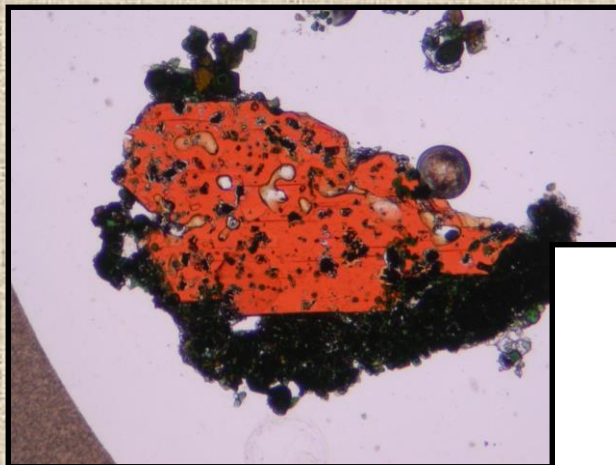
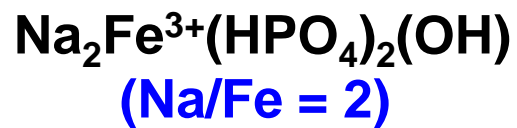


***Pmnb***  
 $a = 6.312(3) \text{ \AA}$   
 $b = 10.523(5) \text{ \AA}$   
 $c = 4.987(2) \text{ \AA}$

Olivine structure  
M(1):  $[\text{Fe}^{2+}]_6$   
M(2):  $[\text{Na}]_6$



# Phase A



Length of the photograph: 3 mm

$I2/m$

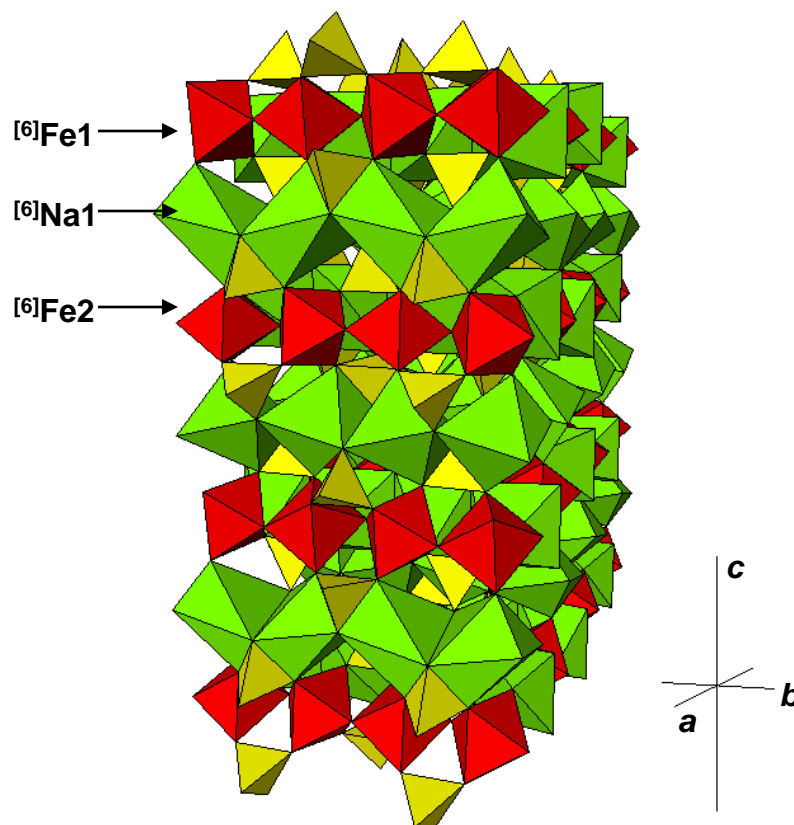
$$a = 14.605(4) \text{ \AA}$$

$$b = 7.144(2) \text{ \AA}$$

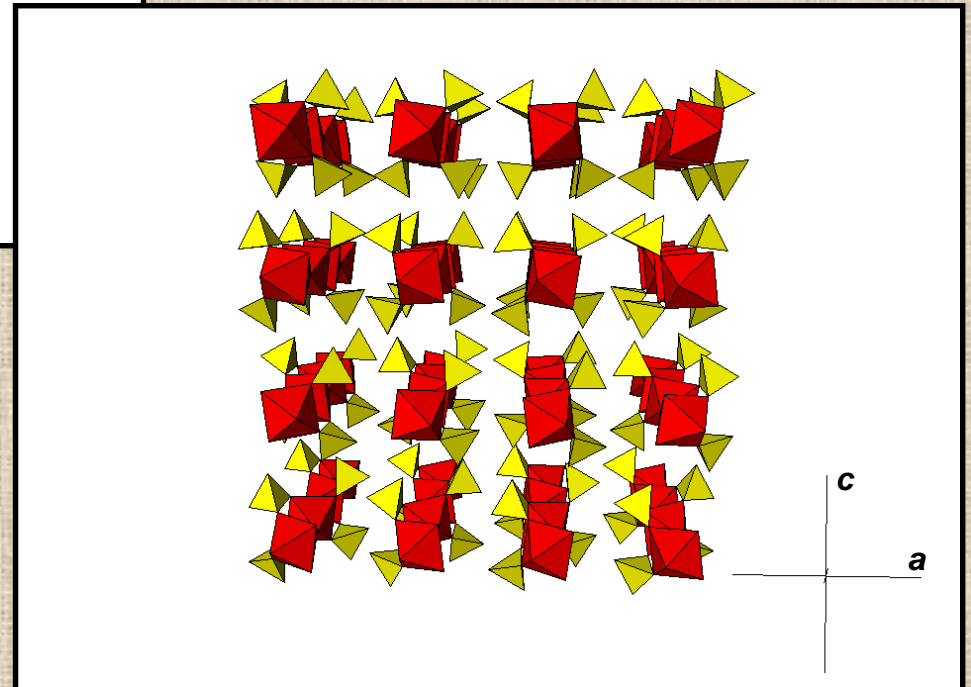
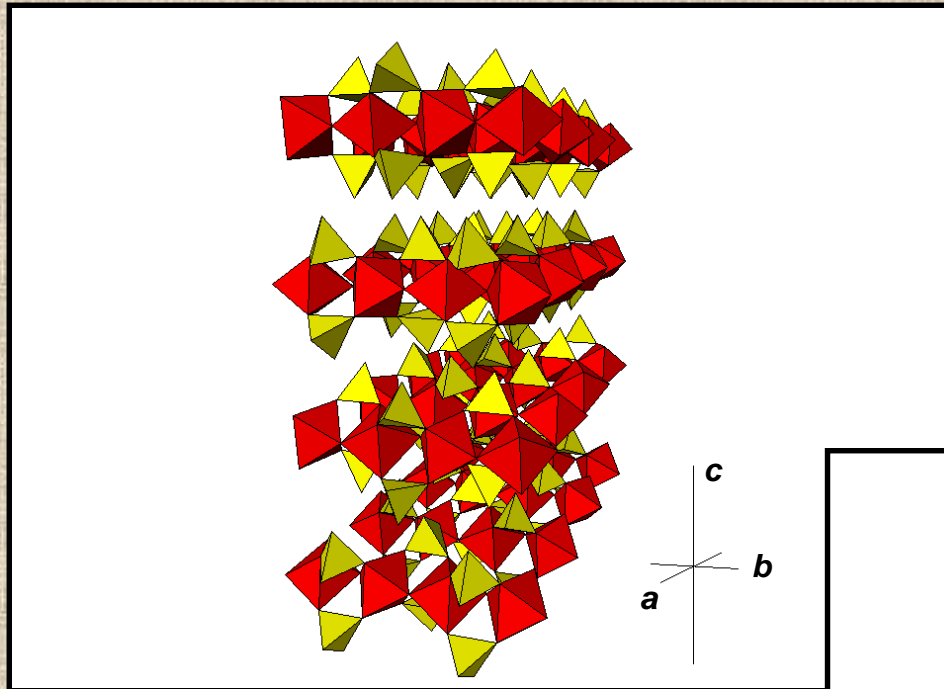
$$c = 15.490(4) \text{ \AA}$$

$$\beta = 90.06(2)^\circ$$

$$R_1 = 5.34 \%$$



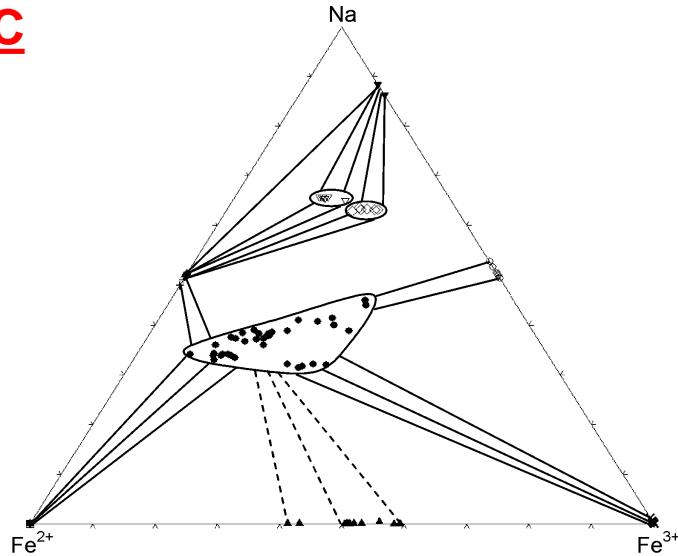
# Phase A



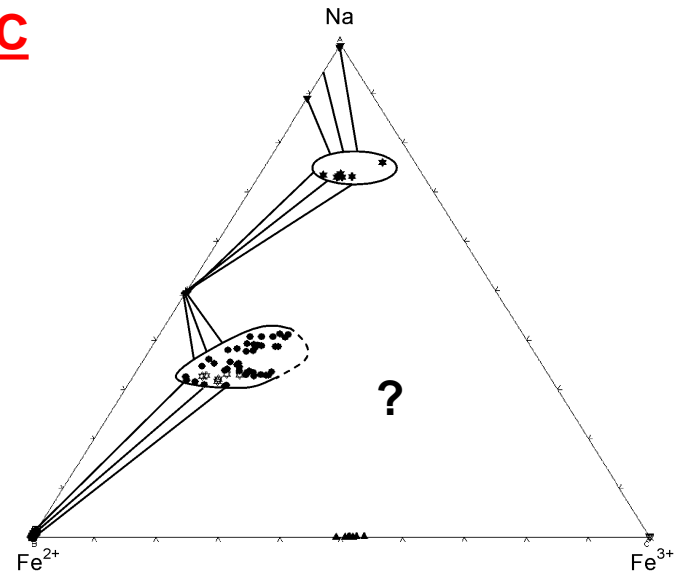
**Chains of corner-sharing  
octahedra**

# Phase diagrams at 500-700°C / 1 kbar

**500°C**

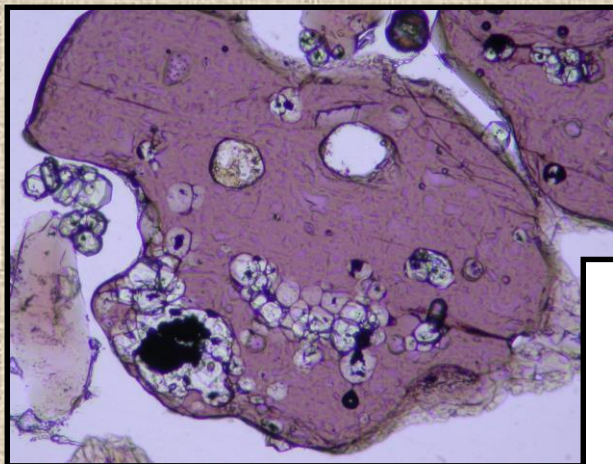
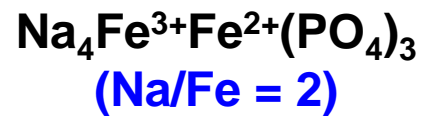


**700°C**



- $\nabla \Rightarrow \text{Na}_4\text{Fe}^{3+}\text{Fe}^{2+}(\text{PO}_4)_3$  (Phase B)
- $\diamond \Rightarrow \text{Na}_7\text{Fe}^{3+}_3\text{Fe}^{2+}(\text{PO}_4)_6$
- $\star \Rightarrow \text{Na}_3\text{Fe}^{3+}(\text{PO}_4)_2$  (Phase C)

# Phase B



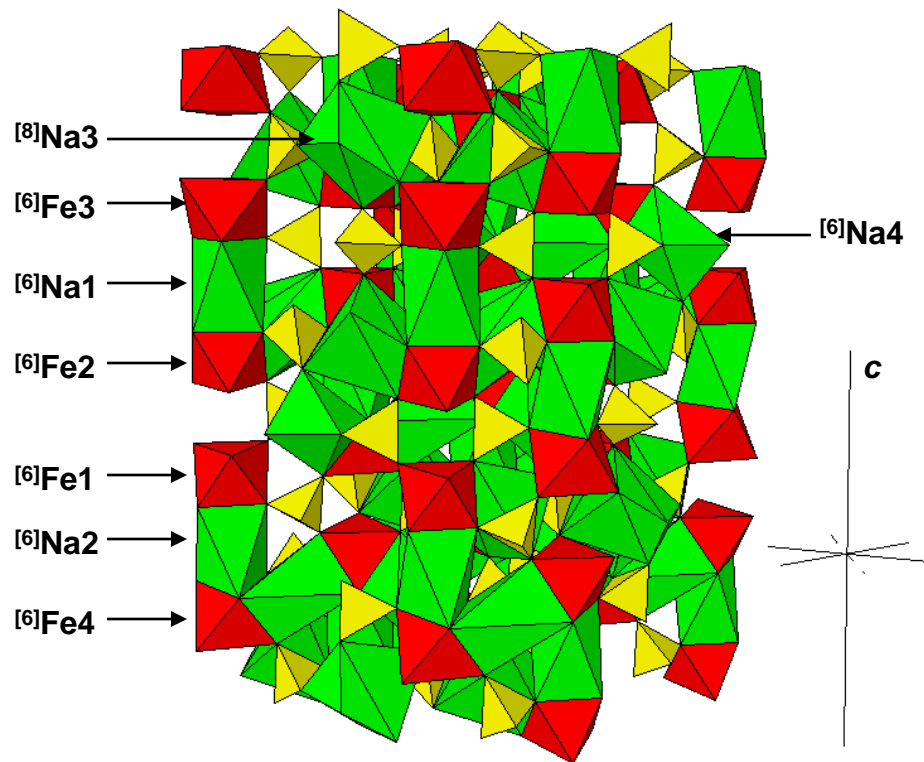
Length of the photograph: 2 mm

**R3**

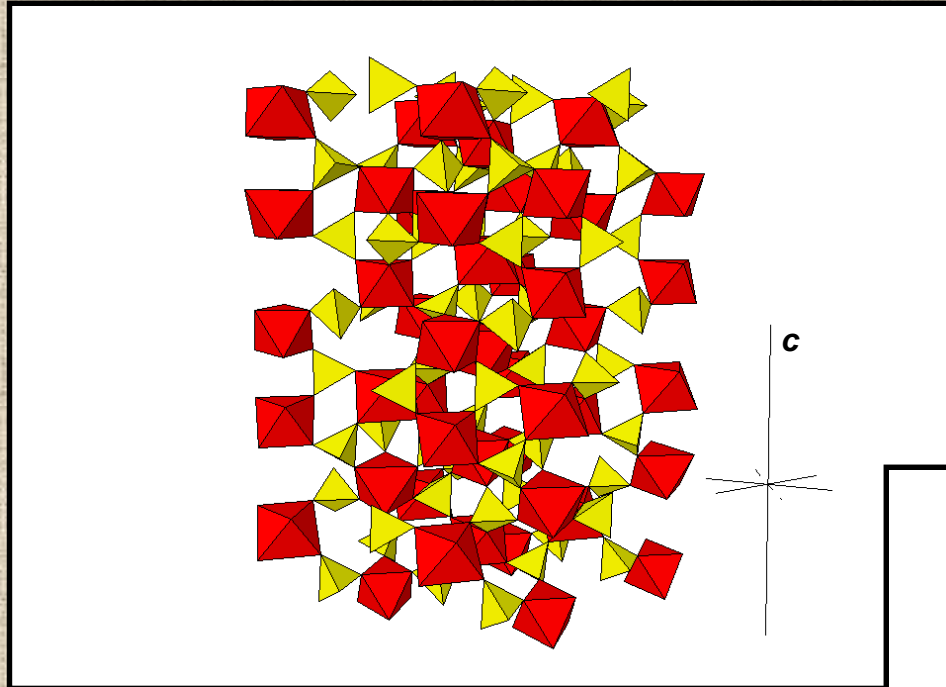
$$a = 8.954(1) \text{ \AA}$$

$$c = 21.280(4) \text{ \AA}$$

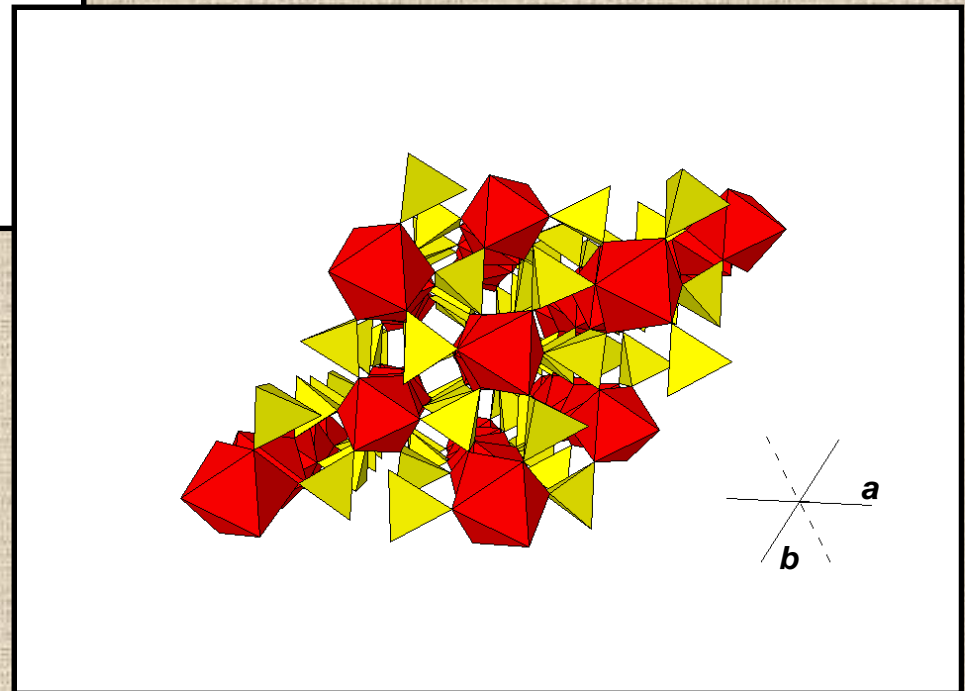
$$R_1 = 3.28 \%$$



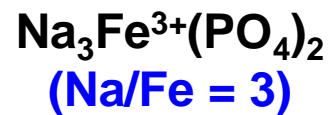
# Phase B



**Heteropolyhedral  
framework**



# Phase C



Length of the photograph: 2 mm

**P-1**

$$a = 5.3141(6) \text{ \AA}$$

$$b = 8.5853(9) \text{ \AA}$$

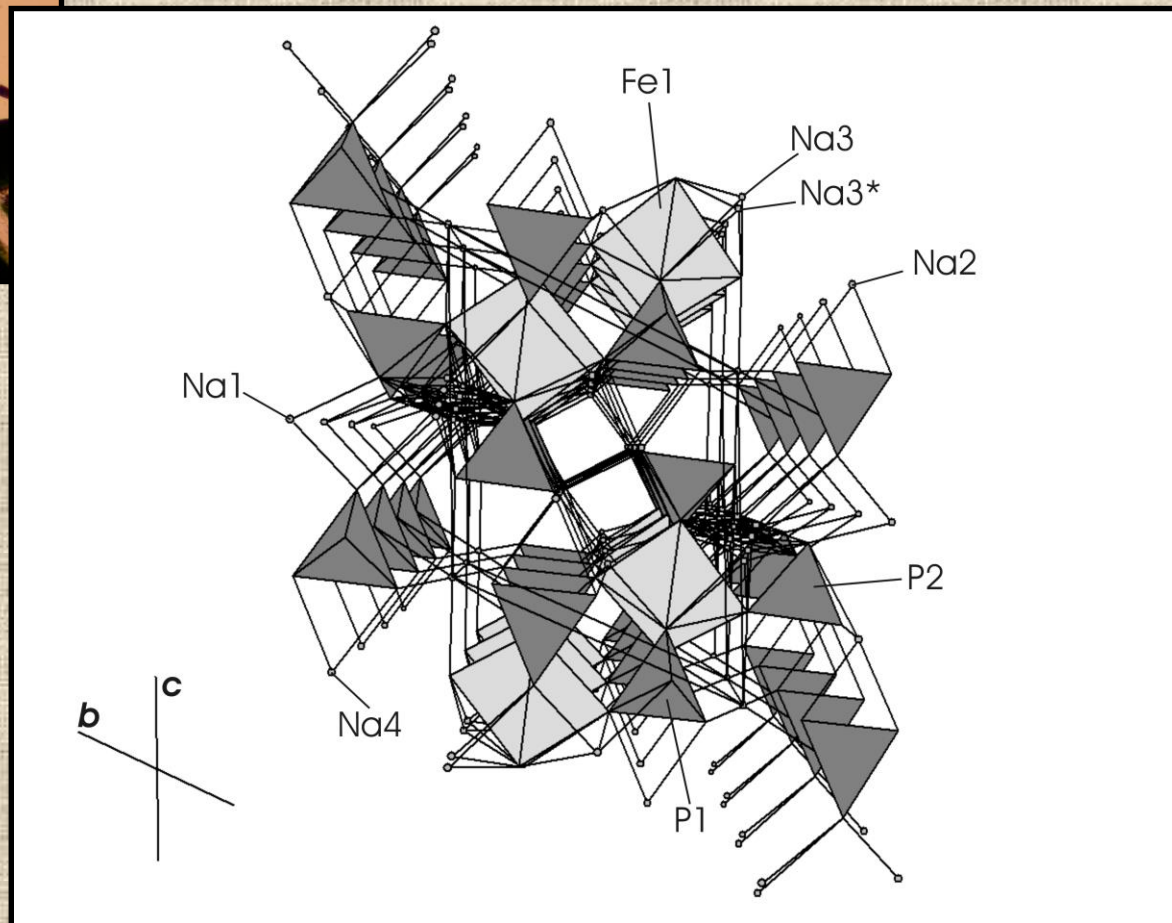
$$c = 8.7859(8) \text{ \AA}$$

$$\alpha = 114.429(9)^\circ$$

$$\beta = 92.327(9)^\circ$$

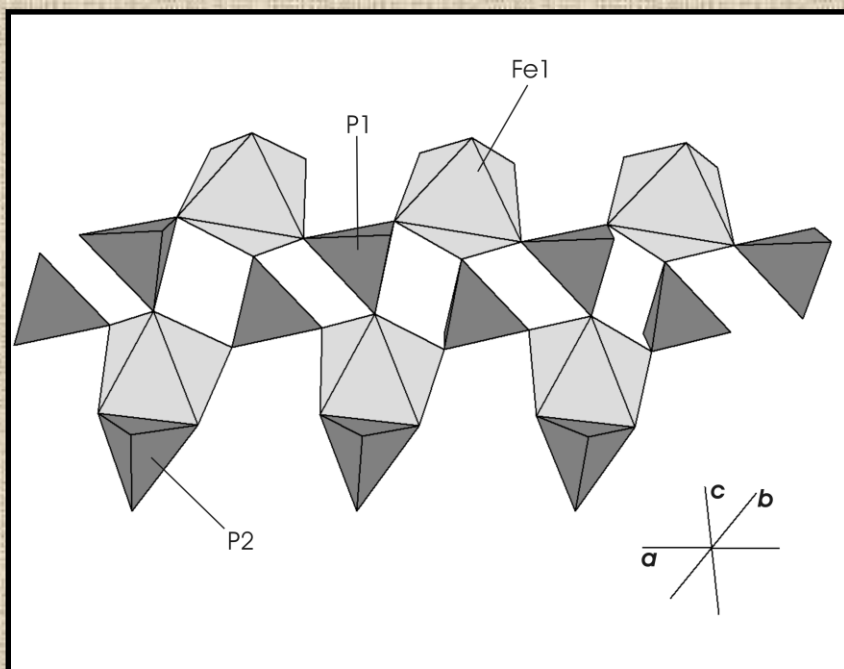
$$\gamma = 106.08(1)^\circ$$

$$R_1 = 2.77 \%$$

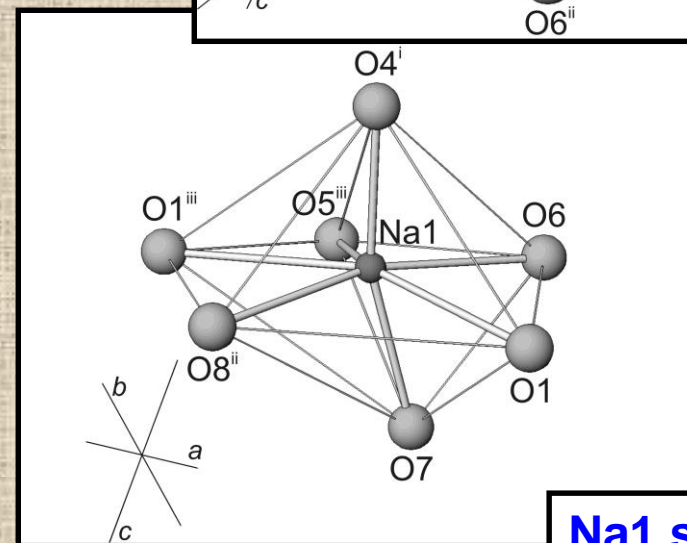
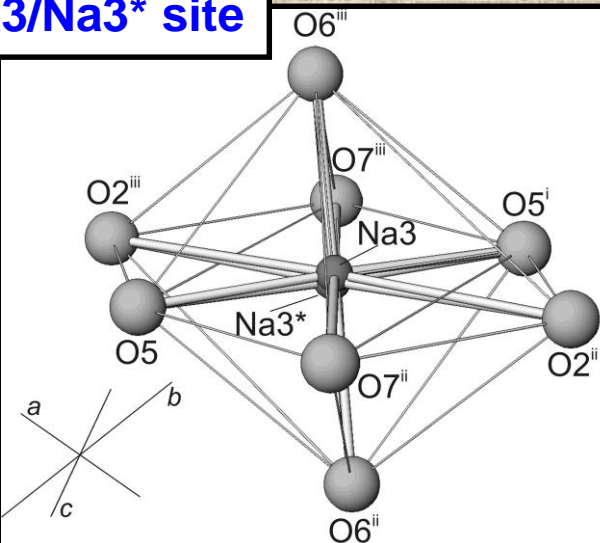


# Phase C

## Hannayite-type heteropolyhedral chains



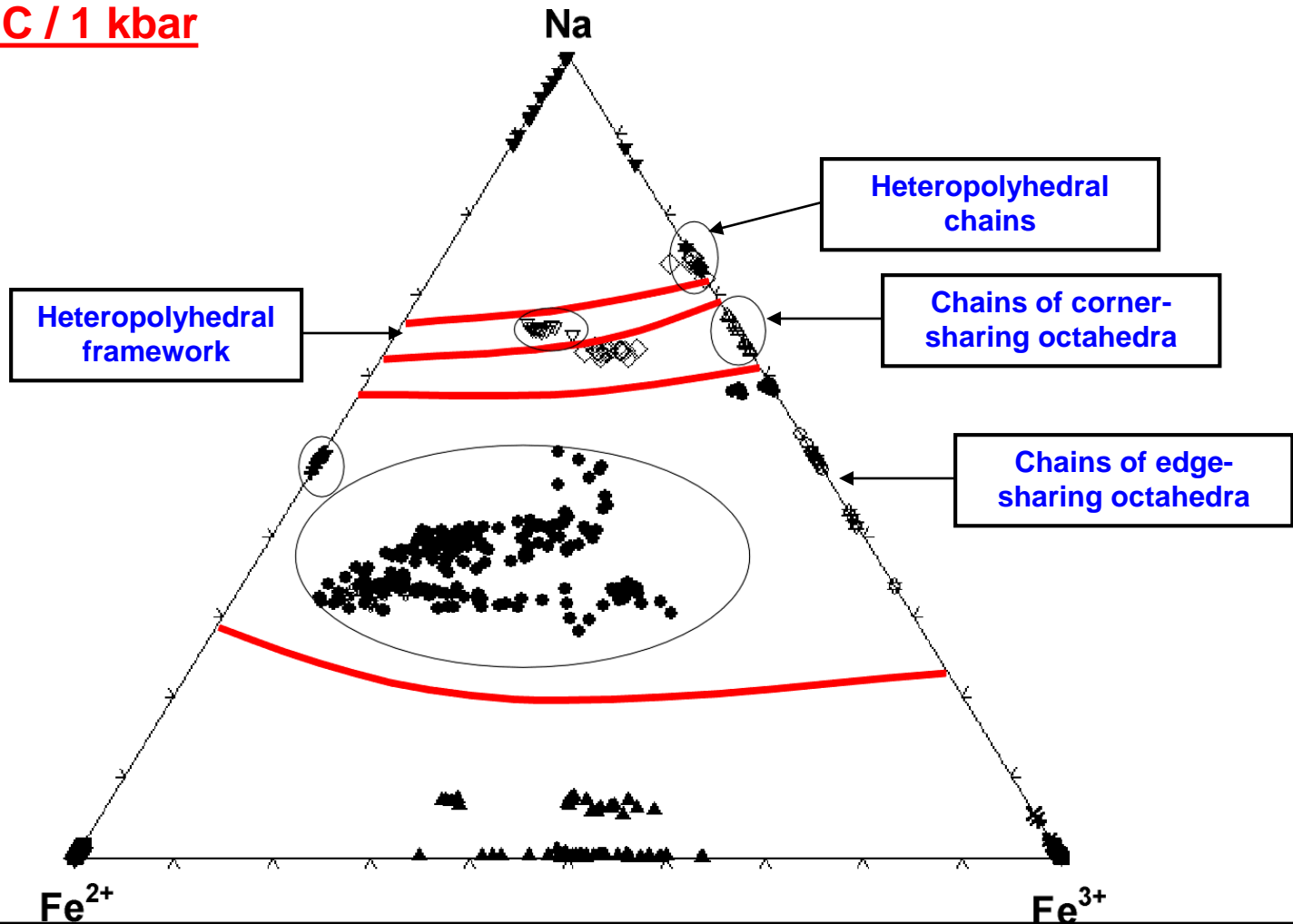
### Na3/Na3\* site



### Na1 site

# Structural domains

400-700°C / 1 kbar





# Conclusions



- Alluaudite-type phosphates cover a wide compositional field in the centre Na-Fe<sup>2+</sup>-Fe<sup>3+</sup> (+PO<sub>4</sub>) diagram
- In the Na-rich portion of the diagram, three phosphates with new crystal structures were synthesized
- With increasing Na content, the FeO<sub>6</sub> octahedra become progressively diluted in the framework
- Progressive transition from crystal structures based on octahedral chains, to structures based on heteropolyhedral units.

# Acknowledgements



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