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Qingheiite-(Fe²⁺), a new member of the wyllieite group

F. Hatert, M. Baijot & S. Philippo

IMA 2010

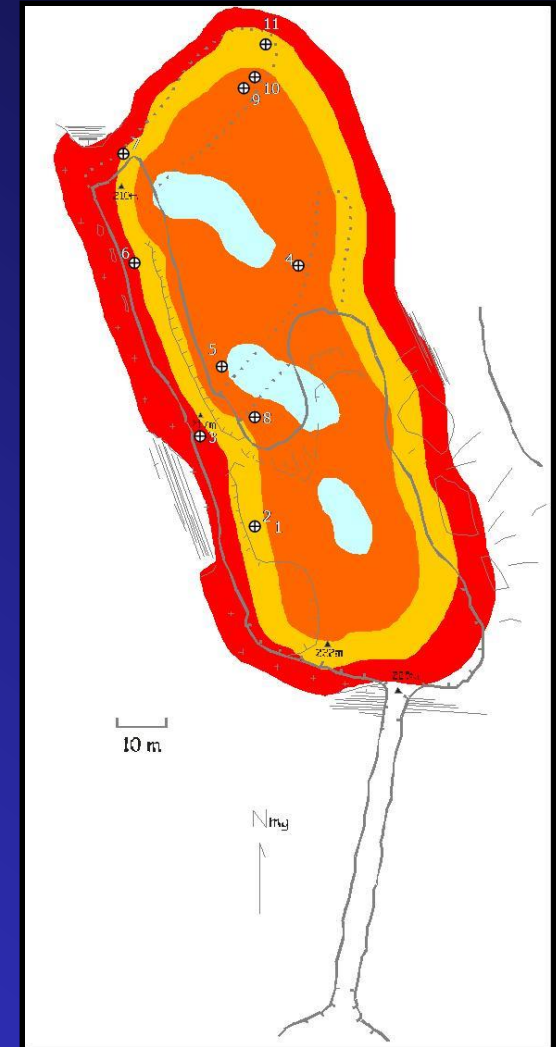
Budapest, August 27th, 2010



M. Baijot, F. Hatert, S. Philippo & J. Cassedanne (September 2008)

The Conselheiro Pena district

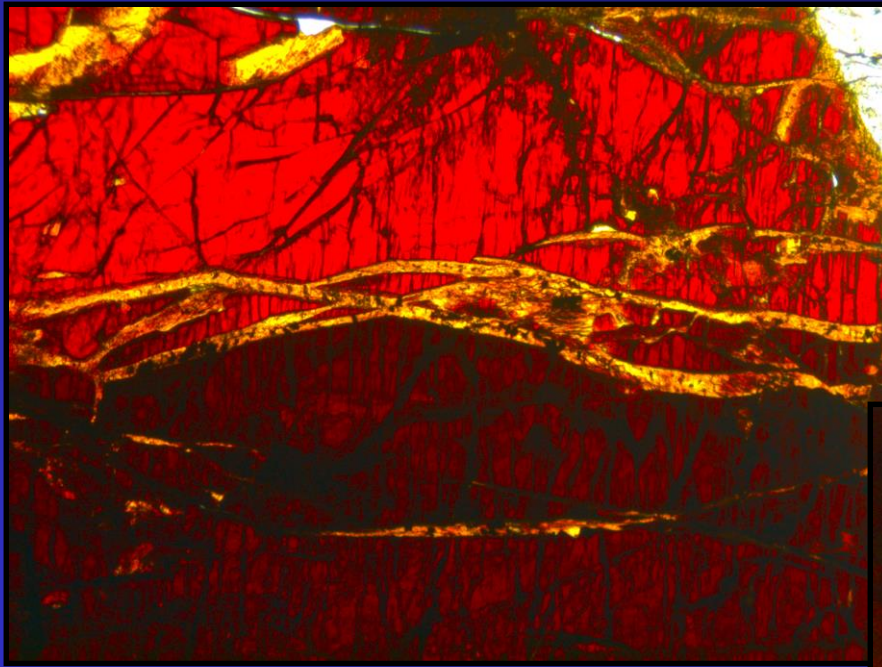
- EBPP, Minas Gerais, Brazil
- 15 new mineral species (Atencio, 2000, 2008)
- Sapucaia pegmatite





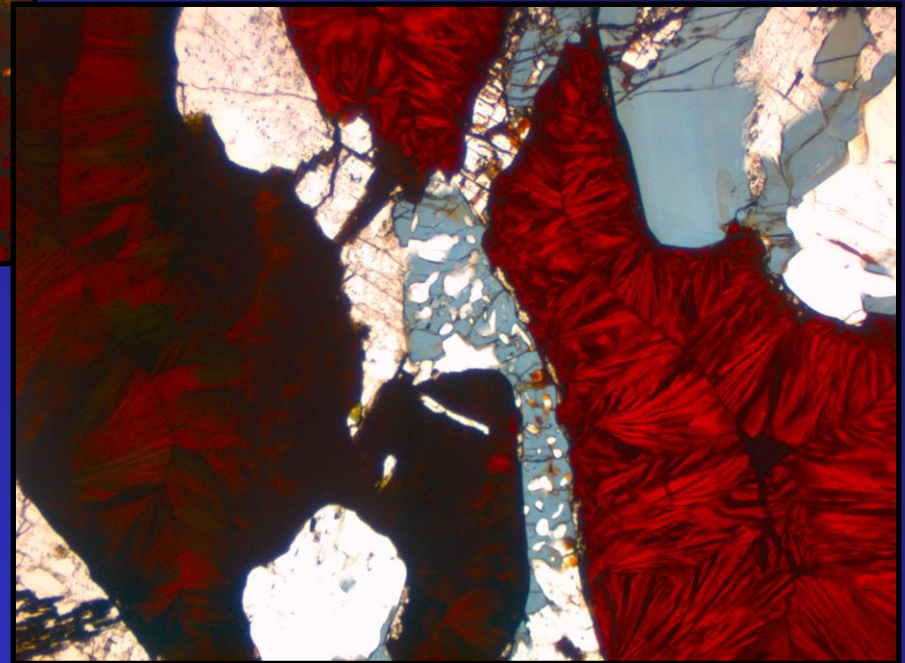
Dendritic phosphates, Sapucaia pegmatite, Minas Gerais, Brazil

Dendritic phosphates



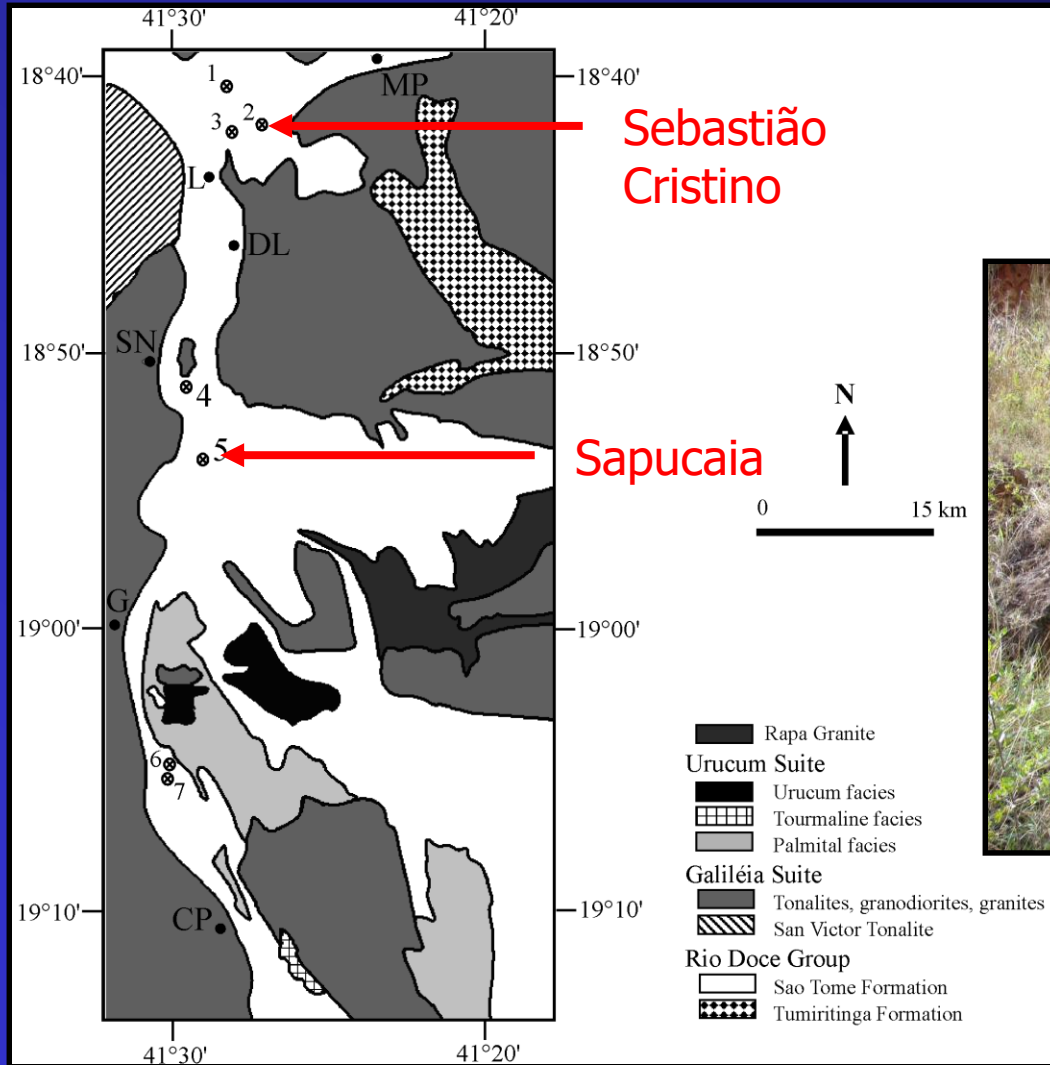
- Frondelite: $\text{Mn}^{2+}\text{Fe}^{3+}_4(\text{PO}_4)_3(\text{OH})_5$
- Schorl
- Quartz and albite

- Ferrisicklerite, $\text{Li}_{1-x}(\text{Fe}^{3+}, \text{Mn}^{2+})\text{PO}_4$
- Heterosite, $(\text{Fe}^{3+}, \text{Mn}^{3+})\text{PO}_4$
- Jahnsite-(CaMnMg):
 $\text{CaMn}^{2+}\text{Mg}_2\text{Fe}^{3+}_2(\text{PO}_4)_4(\text{OH})_2 \cdot 8\text{H}_2\text{O}$

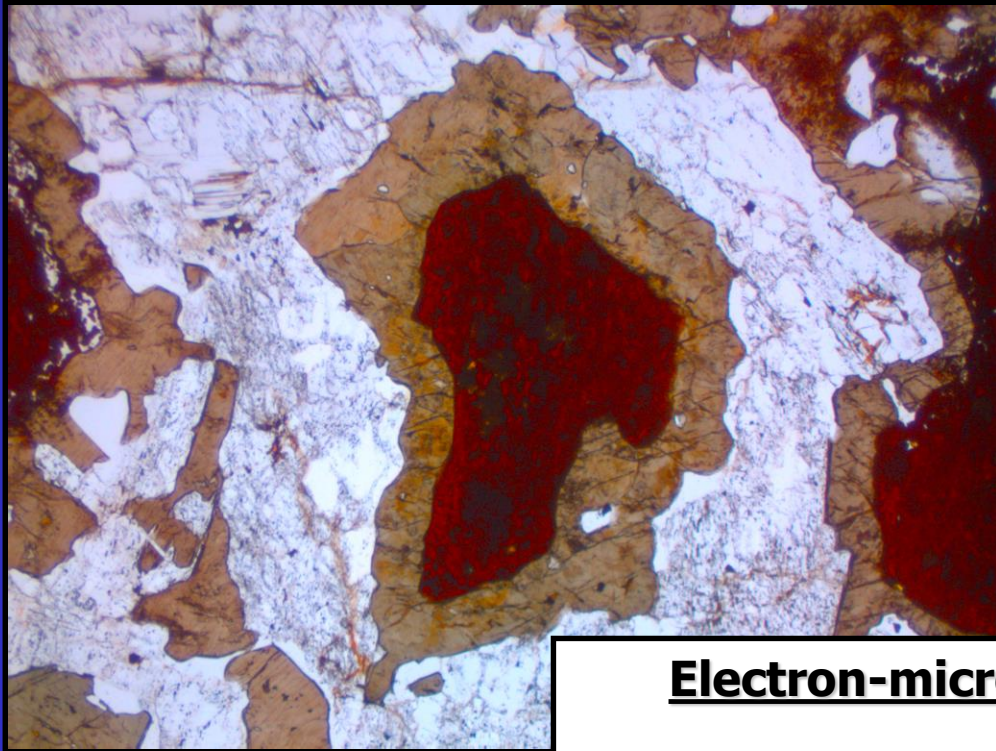


Dendritic phosphates, Sapucaia pegmatite, Minas Gerais, Brazil

The Sebastião Cristino pegmatite



A new mineral species?



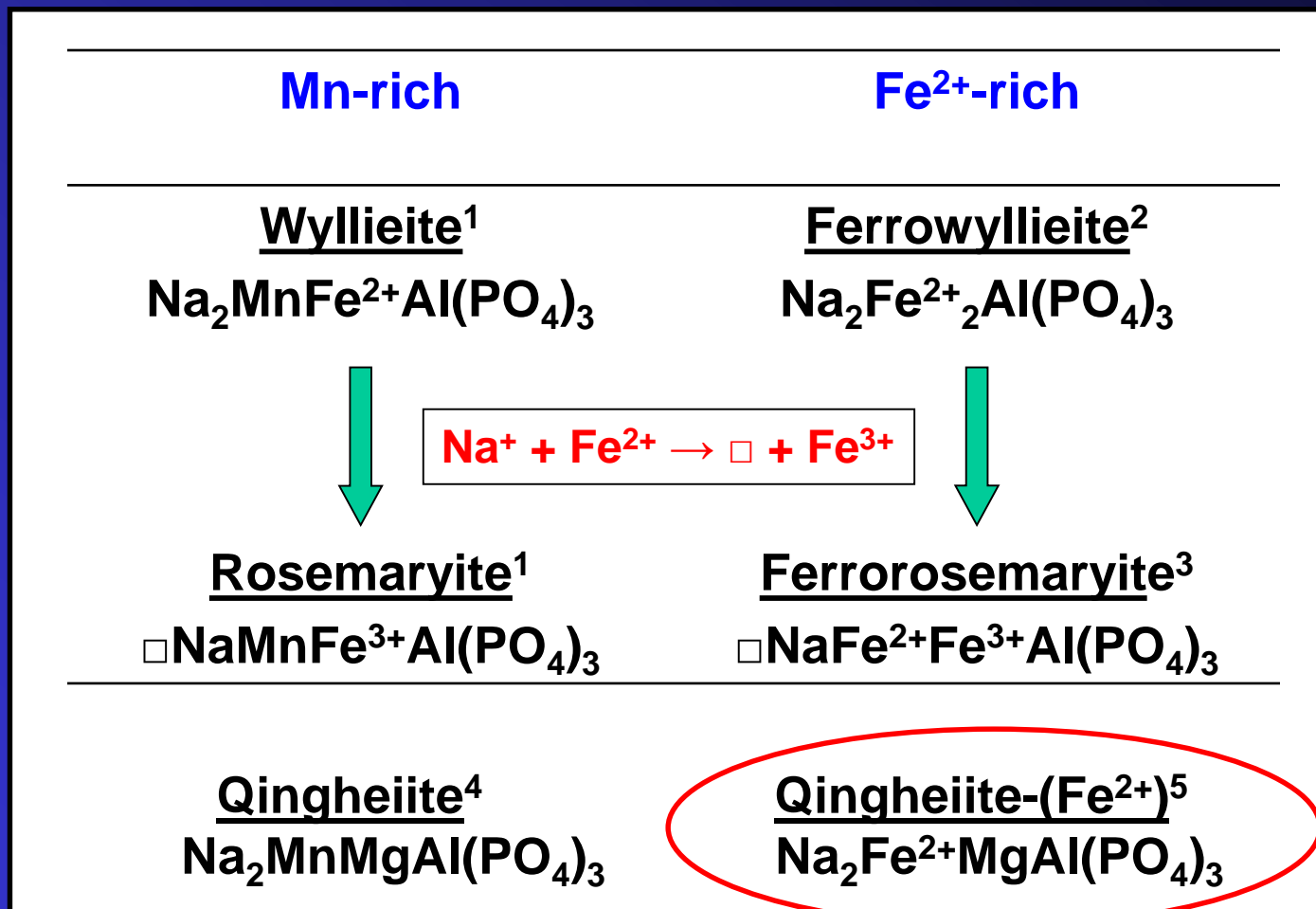
- $\text{Na}_2\text{Fe}^{2+}\text{MgAl}(\text{PO}_4)_3$
- New member of the
wylleite group
(IMA 2009-076)

- Green-brown mineral
forming rims around
frondelite grains
- X-ray powder diffraction
pattern similar to those of
alluaudites and wylleites

Electron-microprobe analyses

- X sites: 0.93 Na + 0.40 Mn + 0.02 Ca + 0.65 □
- M(1): 0.68 Fe^{2+} + 0.32 Mn
- M(2a): 0.72 Mg + 0.23 Fe^{3+} + 0.05 Fe^{2+}
- M(2b): 0.62 Al + 0.38 Fe^{3+}

Phosphates of the wyllieite group

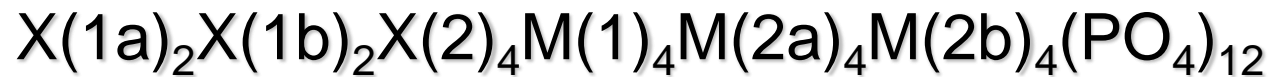


1. Moore & Ito, 1979; 2. Moore & Molin-Case, 1974; 3. Hatert *et al.*, 2005;
4. Zhesheng *et al.*, 1983; 5. Hatert *et al.*, 2010

Crystal structure

Qingheiite-(Fe²⁺) shows the wyllieite structure, which is topologically identical to the alluaudite structure.

Moore & Molin-Case (1974)



Alluaudite \Rightarrow Wyllieite

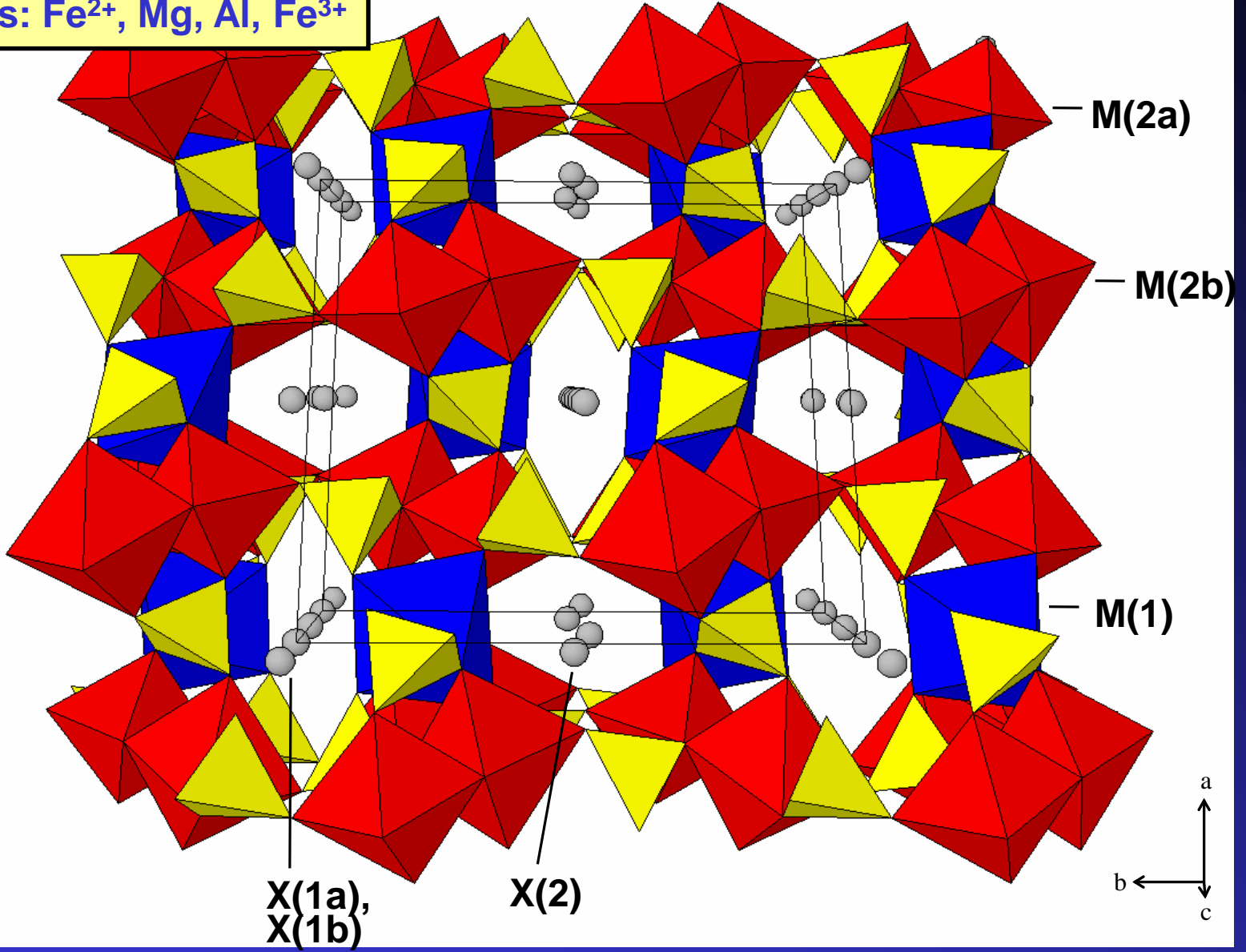
M(1) \Rightarrow M(1)

M(2) \Rightarrow M(2a), M(2b)

X(1) \Rightarrow X(1a), X(1b)

X(2) \Rightarrow X(2)

X sites: Na, Mn, []
M(1): Mn, Fe²⁺
M(2) sites: Fe²⁺, Mg, Al, Fe³⁺



Previous single-crystal structure refinements

	Ferrowyllieite	Qingheite	Synthetic compound
References	[1]	[2]	[3]
Ideal formula	$\text{Na}_2\text{Fe}^{2+}_2\text{Al}(\text{PO}_4)_3$	$\text{Na}_2\text{MnMgAl}(\text{PO}_4)_3$	$\text{Na}_2\text{Mn}^{2+}_2\text{Mn}^{3+}(\text{PO}_4)_3$
Space group	$P2_1/n$	$P2_1/n$	$P2_1/c$
<i>a</i> (Å)	11.868(15)	11.856(3)	6.5291(6)
<i>b</i> (Å)	12.382(12)	12.411(3)	12.653(1)
<i>c</i> (Å)	6.354(9)	6.421(1)	10.952(1)
β (°)	114.52(8)	114.45(2)	97.18(1)
M(2a)	1.00Fe²⁺	0.42Al³⁺ + 0.26Fe ²⁺ + 0.18Mg ²⁺ + 0.13Fe ³⁺ + 0.01Zn ²⁺	1.00Mn ²⁺
M(2b)	0.75Al³⁺ + 0.25Fe ²⁺	0.95Mg²⁺ + 0.05Mn ²⁺	0.79Mn ³⁺ + 0.21Mn ²⁺
M(1)	0.75Fe ²⁺ + 0.25Mg ²⁺	1.00Mn ²⁺	1.00Mn ²⁺
X(1a)	0.46Na ⁺ + 0.04[]	0.50Mn ²⁺	0.41Na ⁺ + 0.09[]
X(1b)	0.25Ca ²⁺ + 0.25Mn ²⁺	0.50Na ⁺	0.48Mn ²⁺ + 0.02[]
X(2)	0.70Na ⁺ + 0.30[]	0.81Na ⁺ + 0.08Ca ²⁺ + 0.11[]	0.85Na ⁺ + 0.15[]

New single-crystal structure refinements

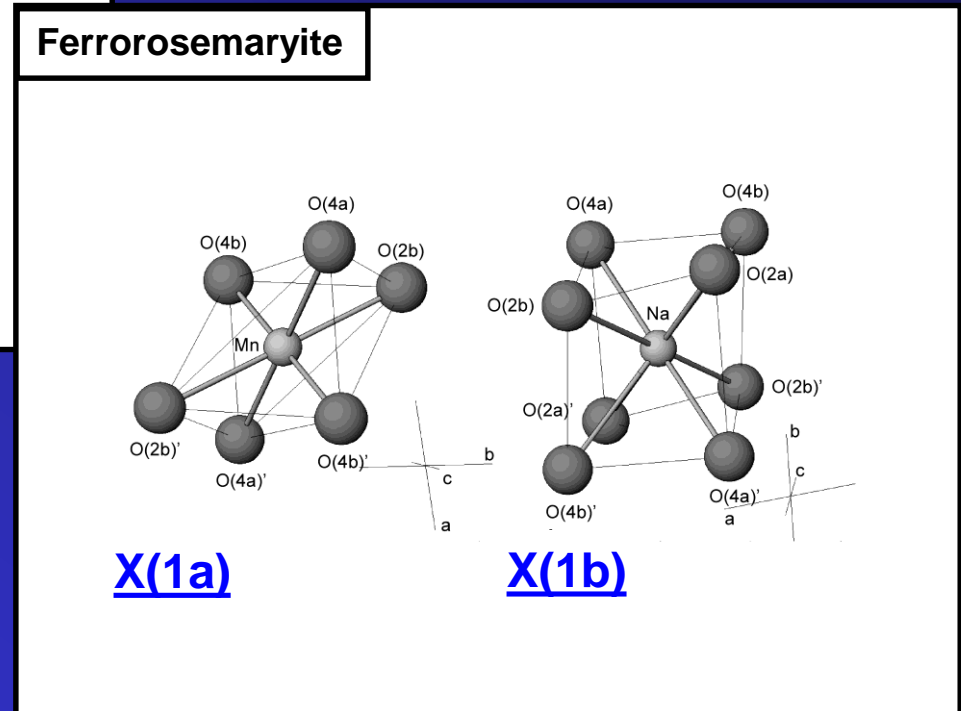
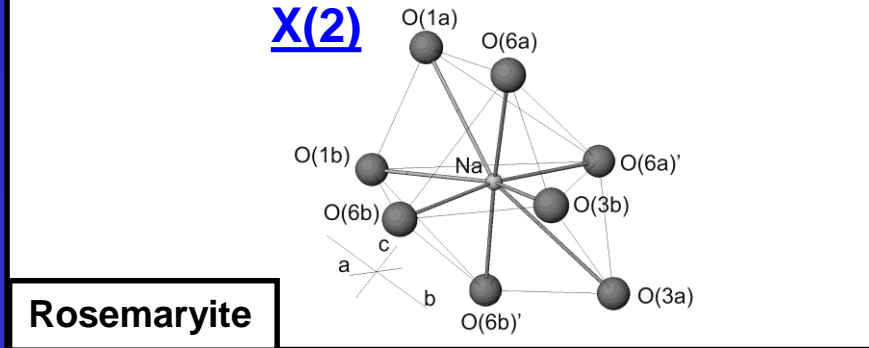
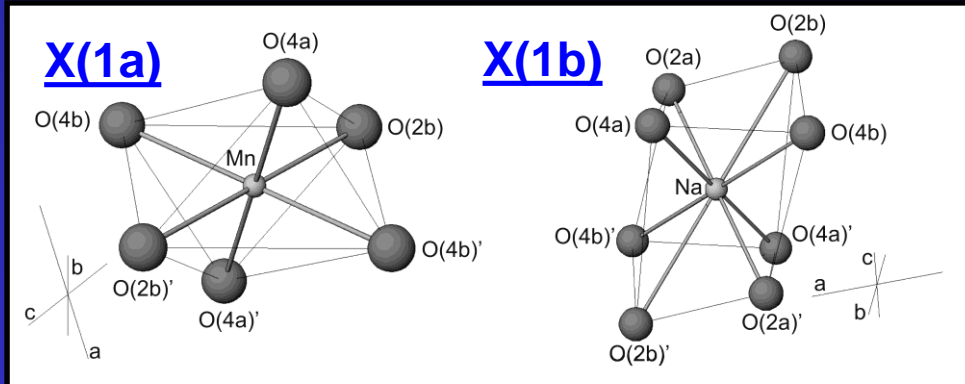
	Ferrosermaryite	Rosemaryite	Qingheite-(Fe²⁺)	Wyllieite	Qingheite
References	[1]	[2]	[3]	This work	This work
Locality	Rubindi, Rwanda	Buranga, Rwanda	Sebastião Cristino, Brazil	Buranga, Rwanda	Santa Ana, Argentina
Ideal formula	NaFe ²⁺ Fe ³⁺ Al(PO ₄) ₃	NaMn ²⁺ Fe ³⁺ Al(PO ₄) ₃	Na ₂ Fe ²⁺ MgAl(PO ₄) ₃	Na ₂ MnFe ²⁺ Al(PO ₄) ₃	Na ₂ MnMgAl(PO ₄) ₃
Space group	<i>P2₁/n</i>	<i>P2₁/n</i>	<i>P2₁/n</i>	<i>P2₁/n</i>	<i>P2₁/n</i>
a (Å)	11.838(1)	12.001(2)	11.910(2)	11.954(2)	11.878(3)
b (Å)	12.347(1)	12.396(1)	12.383(3)	12.439(2)	12.448(2)
c (Å)	6.2973(6)	6.329(1)	6.372(1)	6.406(1)	6.438(2)
β (°)	114.353(6)	114.48(1)	114.43(3)	114.54(1)	114.49(1)
M(2a)	0.72Al³⁺ + 0.14Fe ²⁺ + 0.14Fe ³⁺	0.69Al³⁺ + 0.24Fe ³⁺ + 0.08Fe ²⁺	0.62Al³⁺ + 0.28Mg ²⁺ + 0.10Fe ³⁺	0.70Al³⁺ + 0.25Fe ²⁺ + 0.05Mn ²⁺	0.47Al³⁺ + 0.34Mg ²⁺ + 0.19Fe ³⁺
M(2b)	0.88Fe³⁺ + 0.10Al ³⁺ + 0.02Mg	0.80Fe³⁺ + 0.10Al ³⁺ + 0.10Mn ²⁺	0.44Mg²⁺ + 0.35 Fe ³⁺ + 0.21Fe ²⁺	0.55Fe²⁺ + 0.20Fe ³⁺ + 0.20Mg ²⁺ + 0.05Al ³⁺	0.60Mg²⁺ + 0.40Fe ²⁺
M(1)	0.57Fe ²⁺ + 0.20Fe ³⁺ + 0.18Na ⁺ + 0.05Mn ²⁺	0.75Mn ²⁺ + 0.10Ca ²⁺ + 0.08Mg ²⁺ + 0.07Fe ²⁺	0.52 Fe ²⁺ + 0.25 Na ⁺ + 0.07Mn ²⁺ + 0.16Fe ³⁺	0.75Mn ²⁺ + 0.20Na ⁺ + 0.05Ca ²⁺	0.93Mn ²⁺ + 0.07Fe ³⁺
X(1a)	0.43Mn ²⁺ + 0.07Na ⁺	0.45Mn ²⁺ + 0.05Na ⁺	0.50Mn ²⁺	0.50Mn ²⁺	0.50Mn ²⁺
X(1b)	0.17Na ⁺ + 0.04Mn ²⁺ + 0.04Ca ²⁺ + 0.25[]	0.10Na ⁺ + 0.07Mn ²⁺ + 0.33[]	0.50Na ⁺	0.50Na ⁺	0.50Na ⁺
X(2)	1.0[]	0.44Na ⁺ + 0.56[]	0.18Na ⁺ + 0.15 Mn ²⁺ + 0.02Ca ²⁺ + 0.65 []	1.00Na ⁺	0.95Na ⁺ + 0.05Ca ²⁺

1. Hatert *et al.*, 2005; 2. Hatert *et al.*, 2006; 3. Hatert *et al.*, 2010



Variation in the distribution of Al and Fe on the M(2a) and M(2b) sites

Morphology of the X sites



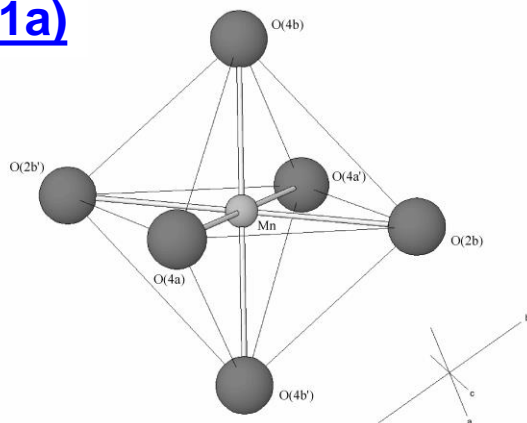
X(1a): Distorted octahedron [6]

X(1b): Distorted cube [4+4]

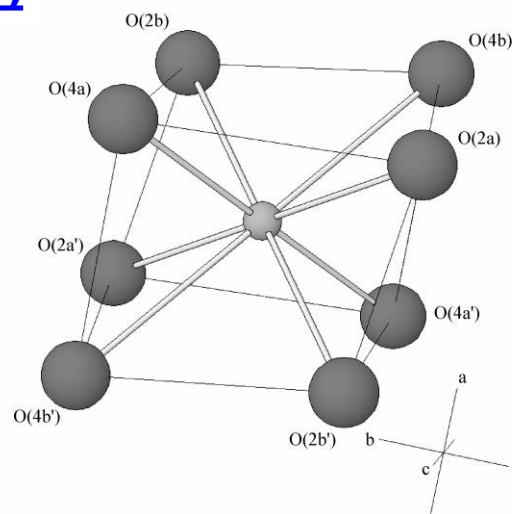
X(2): Distorted gable disphenoid [7+1]

The X sites in qingheiite-(Fe²⁺)

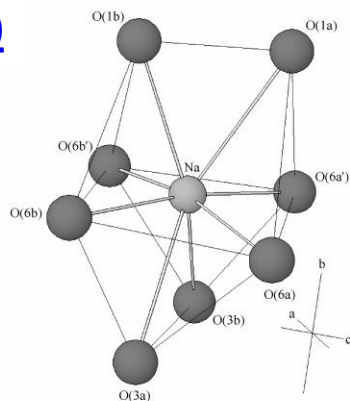
X(1a)



X(1b)



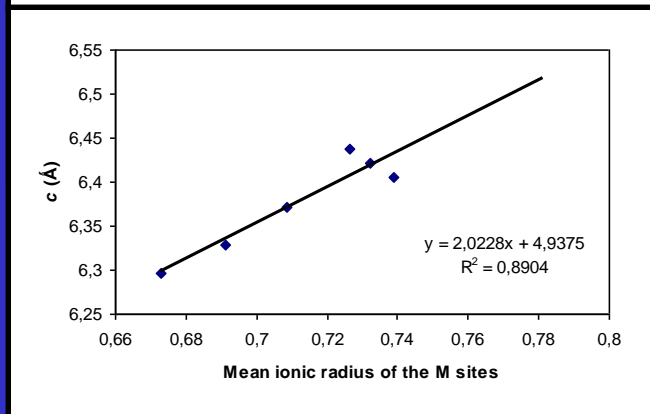
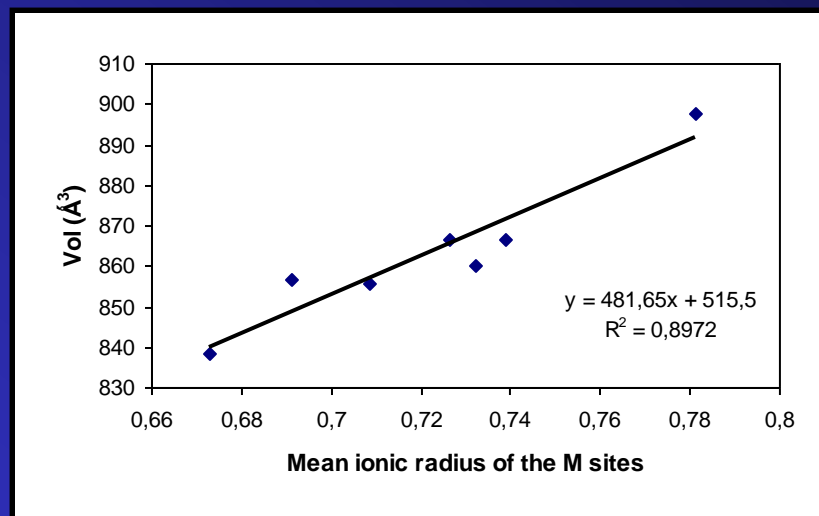
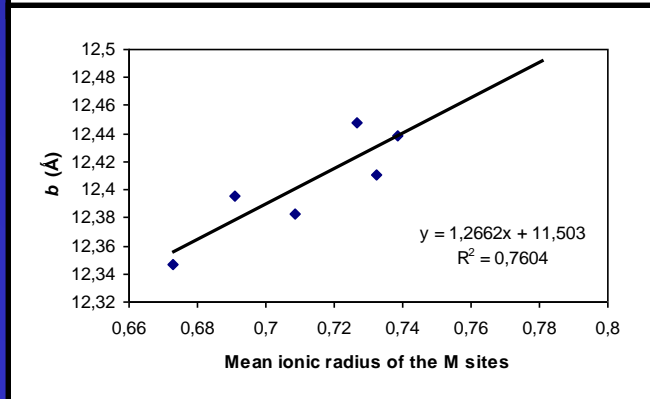
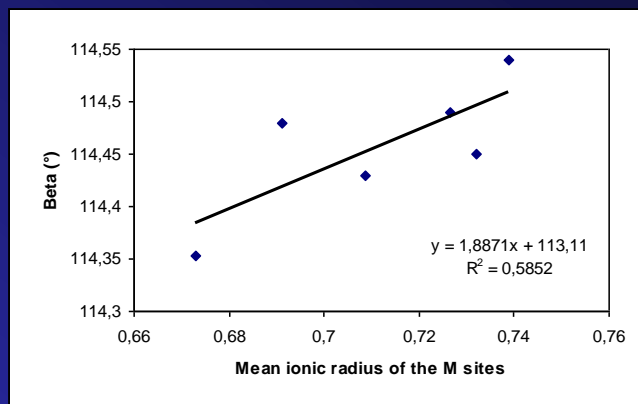
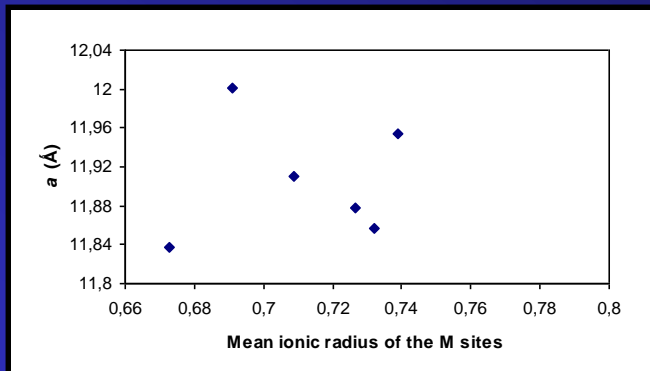
X(2)



BLD coefficients:

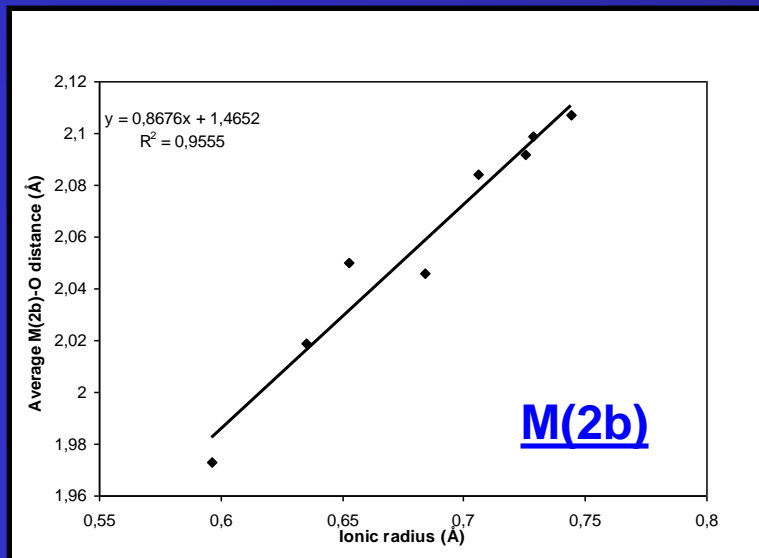
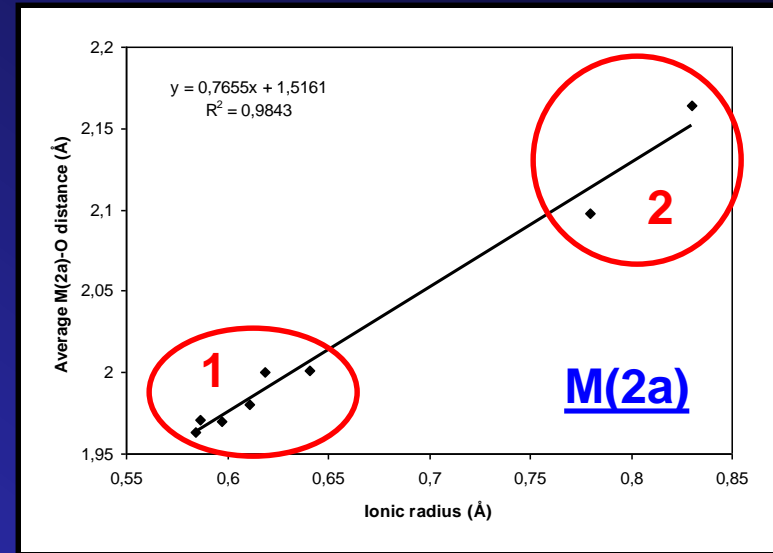
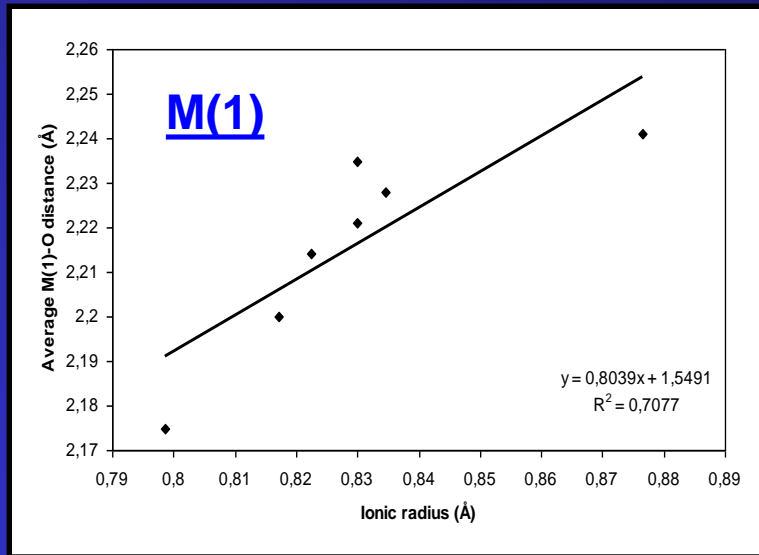
- Rosemaryite: 8.07 %
- Ferrorosemaryite: 7.76 %
- Qingheiite-(Fe²⁺): 6.72 %

Variations of the unit-cell parameters

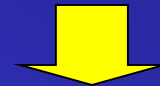


Good correlations between b , c , $Vol.$, and the mean ionic radius of the cations occurring on the M sites

Variations of the M-O bond lengths



- When the M(2) sites contain Al and a cation with an *e.i.r.* smaller than 0,75 Å (Fe³⁺, Mg), then Al occupies the M(2a) site (zone 1).
- When the M(2) sites contain a cation with an *e.i.r.* larger than 0,75 Å (Fe²⁺, Mn²⁺), this cation occupies the M(2a) site (zone 2) and the small cations (Al, Mn³⁺) occupy the M(2b) site.



M(2a) can accommodate larger cations than M(2b)

Conclusions

- Qingheiite-(Fe²⁺) is a new phosphate mineral of the wyllieite group, which occurs in the Sebastião Cristino pegmatite, Minas Gerais, Brazil
- A comparison of the structural data of qingheiite-(Fe²⁺), with those of other minerals of the wyllieite group, indicates: (I) that the distortion coefficients calculated for the X(1b) sites show significant variations. (II) that the M(2a) site can accommodate larger cations than M(2b).
- Investigations are in progress in order to better understand the genesis of qingheiite-(Fe²⁺), and of other phosphates from pegmatites of the Conselheiro Pena district.