



# Dissolved greenhouse gas (CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O) emissions from highland lakes of the Andes cordillera in Northern Ecuador

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## Abstract

We report, in 15 Ecuadorian mountainous lakes, dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O and a suite of ancillary biogeochemical variables (dissolved inorganic nutrients, oxygen, major cations, dissolved and particulate organic carbon, and the stable isotope composition of dissolved inorganic carbon and water). The sampled lakes were located in the páramos of Salve Facha and Antisana y Mojanda (northern region of Ecuadorian Andes), along an elevational gradient from 2213 to 4361 m above sea level, as well as a gradient of lake surface area (0.003–6.1 km<sup>2</sup>) and depth (1–74 m). Most lakes were characterized by lower values of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (644–2152 ppm) than usually attributed to tropical lakes (~1900 ppm). Three lakes influenced by volcanic inputs were characterized by the highest pCO<sub>2</sub> values (3269–10,069 ppm), while two lakes bordered by large cities were characterized by the lowest pCO<sub>2</sub> values (208–254 ppm). Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol/L and were negatively correlated to lake area and depth. N<sub>2</sub>O saturation levels ranged between 64% and 101%. The surface waters were undersaturated in N<sub>2</sub>O with respect to atmospheric equilibrium, probably due to inputs of soil–water with low N<sub>2</sub>O levels owing to denitrification in soils. The data obtained in the sampled highland lakes was combined with published data from lowland lakes (elevation < 500 m, *n* = 17 for CO<sub>2</sub> and *n* = 16 for CH<sub>4</sub>) to derive relations between CO<sub>2</sub> and CH<sub>4</sub> and lake surface area, allowing to compute CO<sub>2</sub> and CH<sub>4</sub> lacustrine emissions at the scale of the Amazon basin using the HydroLAKES spatial data set. The CO<sub>2</sub> and CH<sub>4</sub> emissions from highland lakes (elevation > 500 m) only represented 0.4% and 2% of the total lacustrine emissions at the scale of the Amazon basin, respectively. Total CO<sub>2</sub> and CH<sub>4</sub> emissions from lakes represented a small fraction (8.6%) of total lentic and lotic CO<sub>2</sub> and CH<sub>4</sub> emissions at the scale of the Amazon basin. The lake and river emissions of CH<sub>4</sub> represented ~3% of total CH<sub>4</sub> emissions from all compartments of the Amazon basin, mainly attributable to wetlands.

**Keywords** Amazon River · Ecuador · Mountainous lakes · CO<sub>2</sub> · CH<sub>4</sub> · N<sub>2</sub>O

## Introduction

Lakes and rivers are important components of the global carbon cycle in terms of transport and storage of organic carbon, as well as emissions of CO<sub>2</sub> and CH<sub>4</sub> to the atmosphere. Rivers transport from land to the ocean ~500 TgC/yr of organic carbon (Stallard 1998). The organic carbon

buried in sediments of lakes (70–140 TgC/yr) is coequal to the rate in the oceans (120 TgC/yr), and comparable to the accumulation of organic carbon on land (soils and vegetation) during the last interglacial period (230 TgC/yr) (Cole 2013). This is striking because the total area of lakes is <4% of the Earth's nonglaciaded land area and ~1% of that of the oceans. Lakes and rivers also emit to the atmosphere CO<sub>2</sub> and CH<sub>4</sub>, two potent greenhouse gases (GHGs). The CO<sub>2</sub> emission from lakes globally was estimated to ~300 TgC/yr (Raymond et al. 2013) and would be six times lower than that from rivers of ~1800 TgC/yr (Raymond et al. 2013; Liu et al. 2022). On the contrary, the emission of CH<sub>4</sub> from lakes (42 TgC/yr) could be approximately ten times higher than that from rivers (4.5 TgC/yr), accounting for 21% of total aquatic emissions, according to Rosentreter et al. (2021). The global CH<sub>4</sub> emission

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from lakes has been recently revised downwards (31 TgC/yr) (Johnson et al. 2022) and that from rivers upwards (21 TgC/yr) (Rocher-Ros et al. 2023). The relative importance of CH<sub>4</sub> emissions from lakes compared with those from rivers at global scale is variable from one study to another, while at regional scale, some studies suggest that integrated CH<sub>4</sub> emissions from rivers might be higher than those from lakes, such as at the African continental scale (Borges et al. 2022). Yet, it is clear that the exchange of CO<sub>2</sub> and CH<sub>4</sub> between inland waters and the atmosphere is large enough to command further interest and research effort at the global scale. Further, a large source of uncertainty in the global CH<sub>4</sub> budget is attributed to emissions from inland waters (Saunois et al. 2020), thus, reducing the uncertainty of CH<sub>4</sub> emissions from lakes and rivers is clearly a priority (Johnson et al. 2022; Melack et al. 2022).

The uncertainty on the estimates of CO<sub>2</sub> and CH<sub>4</sub> emissions from lakes and rivers cannot be reduced by measurements alone, because it will be impossible to measure in situ GHG emissions in all rivers and lakes globally, given the global extent of river networks (~2 × 10<sup>6</sup> km by length and ~770,000 km<sup>2</sup> by area) (Allen and Pavelsky 2018) and of lakes (~1.4 × 10<sup>6</sup> lakes > 0.1 km<sup>2</sup> with a combined surface area of 2.7 × 10<sup>6</sup> km<sup>2</sup>) (Messenger et al. 2016). It is then required to model GHG emissions in rivers and lakes using either mechanistic models (e.g. Guo et al. 2020) or statistical relations with relevant predictors represented by spatial data sets. The choice of predictors needs to be anchored on a sound conceptual understanding of drivers of GHG emissions to minimize the application of spurious correlations and avoid redundancy in predictors of multivariate models. Numerous relevant potentially applicable spatial data sets are already available such as RiverATLAS (Linke et al. 2019) and HydroLAKES (Messenger et al. 2016) and are expanding rapidly (Toming et al. 2020; Altenau et al. 2021; Bonnema et al. 2022; Pi et al. 2022; Sikder et al. 2023). The most readily available predictors for lakes are morphometric, so regional or global upscaling attempts of CO<sub>2</sub> and CH<sub>4</sub> emissions from lakes have relied on lake surface area (Holgerson and Raymond 2016; Rosentreter et al. 2021) or depth (Borges et al. 2022). Others have integrated lake productivity based on statistical distributions of chlorophyll-*a* by lake size (DelSontro et al. 2018), although not accounting for strong latitudinal variations of productivity (Lewis 1987), or by integrating ecoclimatic zones (Johnson et al. 2022). So, there is an increasing need to expand and possibly standardize upscaling methods (Lauerwald et al. 2023). Yet, the largest limitation to reduce the uncertainty of the estimates of GHG emissions from rivers and lakes is the scarcity of field data of GHG emissions to adequately capture the very large spatial and temporal variability, allowing developing robust and reliable statistical models (e.g., Deemer and Holgerson 2021).

There are fewer studies of CO<sub>2</sub> and CH<sub>4</sub> variations in mountainous lakes than in lowland lakes. Available data suggest that there is no systematic difference in CH<sub>4</sub> emissions between mountainous and lowland lakes, while CO<sub>2</sub> emissions seem to be lower in mountainous lakes that occasionally act as sinks for atmospheric CO<sub>2</sub>, based on data mainly collected in the Alps (Pighini et al. 2018), Rocky Mountains (McCrackin and Elser 2011; Crawford et al. 2015), Sierra Nevada (Cohen and Melack 2020), and the Tibetan Plateau (Yan et al. 2018, 2023; Wang et al. 2022). Available data sets of GHG emissions from lakes and rivers are also spatially skewed toward higher latitudes, mainly in North America and Scandinavia. These data sets cannot capture the features of GHG emissions from lakes and rivers in other parts of the globe with different climates and land cover. Yet, the majority of riverine GHG emissions are from the tropics (20 °N to 20 °S), ~60–80% for CO<sub>2</sub> (Raymond et al. 2013; Borges et al. 2015; Lauerwald et al. 2015; Sawakuchi et al. 2017; Liu et al. 2022) and ~50–70% for CH<sub>4</sub> (Sawakuchi et al. 2014; Rocher-Ros et al. 2023). Emissions from tropical lakes represent 37–50% of the total for CH<sub>4</sub> (Bastviken et al. 2011; Johnson et al. 2022) and 34% for CO<sub>2</sub> (Raymond et al. 2013). The estimate of emissions of CO<sub>2</sub> from tropical lakes as given in existing literature has been recently questioned and seems to have been overestimated (Borges et al. 2022). The CO<sub>2</sub> emissions from tropical lakes might, in fact, be lower than their counterparts at higher latitudes (Borges et al. 2022), while the CH<sub>4</sub> emissions from tropical lakes are disproportionately intense compared with their higher-latitude counterparts (Bastviken et al. 2011; Johnson et al. 2022; Borges et al. 2022). Relatively lower lacustrine CO<sub>2</sub> but higher lacustrine CH<sub>4</sub> emissions in tropical climates than at higher latitudes are partly related to high phytoplankton production in nonhumic lakes (Morana et al. 2023) that leads to the uptake of dissolved CO<sub>2</sub> but at the same time high CH<sub>4</sub> production in sediments (Borges et al. 2022) due to the deposition of fresh phytoplankton detritus (Grasset et al. 2018). Similarly, prior estimates of emissions of N<sub>2</sub>O (Kroeze et al. 2010) from both rivers and lakes globally have been recently questioned and revised downwards (Hu et al. 2016; Maavara et al. 2019; Lauerwald et al. 2019). Even so, the N<sub>2</sub>O emissions modeled from nitrogen deposition rates and emission factors (Lauerwald et al. 2019) seem to provide higher estimates in tropical lakes compared with those derived from field measurements (Borges et al. 2022). Additionally, emissions of N<sub>2</sub>O from tropical lakes are lower compared with their boreal counterparts (Borges et al. 2022; 2023). This calls for a careful evaluation and a reduction of uncertainty of the GHG emissions from tropical lakes and rivers.

Until recently, the majority of research on GHG emissions from tropical inland waters was carried out in South America and in particular on the Amazon basin, as reviewed by Melack (2016) for CO<sub>2</sub>, and Melack et al. (2022) for

CH<sub>4</sub>. Melack et al. (2022) summarized published CH<sub>4</sub> fluxes and regional estimates in several types of aquatic systems (including rivers, lakes, reservoirs, and several types of wetlands), evaluated the adequacy of methods, and listed the limitations in current estimates. Melack et al. (2022) also identified regions or habitats of the Amazon basin with few or no data that include systems at altitudes above 500 m, mainly in the Andes cordillera. These systems are expected to be sensitive to climate change as increased air temperature accelerates the thawing of glaciers, leading to an increase in surface runoff (Bliss et al. 2014; LópezMoreno et al. 2014), that should affect GHG emissions from existing aquatic environments, as well as their coverage by the emergence of new lakes and ponds (Bosson et al. 2023).

Here, we report the dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in 15 lakes located in the northern region of Ecuadorian Andes along an elevational gradient from 2213 to 4361 m above sea level, as well as a gradient of lake surface area (0.003–6.1 km<sup>2</sup>) and depth (0.9–74 m) (Fig. 1). Most lakes were located in the páramos of Salve Facha and Antisana y Mojanda. We also report a large suite of biogeochemical variables that allow us to characterize the lakes and provide some insights into the biogeochemical and physiographic drivers of the variability of GHGs in the sampled lakes, such as water temperature, specific conductivity, oxygen saturation level (%O<sub>2</sub>), stable isotope composition of carbon (δ<sup>13</sup>C-DIC) in dissolved inorganic carbon (DIC), nitrate (NO<sub>3</sub><sup>-</sup>), nitrite (NO<sub>2</sub><sup>-</sup>), ammonium (NH<sub>4</sub><sup>+</sup>), total alkalinity (TA), major cations (Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, K<sup>+</sup>), dissolved silicate (DSi), Fe, stable isotope composition of O and H of H<sub>2</sub>O (δ<sup>18</sup>O-H<sub>2</sub>O and δ<sup>2</sup>H-H<sub>2</sub>O), and dissolved organic carbon (DOC) and particulate organic carbon (POC) concentrations. We explore modelling of CO<sub>2</sub> and CH<sub>4</sub> emissions to the atmosphere from the lakes using our data in highland lakes and literature data for lowland lakes (totaling about 30 lakes for each gas), statistical relations with relevant predictors from HydroLAKES (Messenger et al. 2016), and recently published estimates of ebullitive CH<sub>4</sub> emissions

(Borges et al. 2022). Finally, the upscaled lacustrine CO<sub>2</sub> and CH<sub>4</sub> emissions are compared with the emissions from rivers based on a recent estimate at the scale of the Amazon basin that included data from the mountainous Andean streams (Chiriboga and Borges 2023).

## Material and methods

### Site description

A vast amount of headwaters can be found in the Andes, which contribute 82% of sediments and solutes to the Amazon basin (McClain and Naiman 2008). Páramos are Andean mountain ecosystems that are located discontinuously in the Neotropics at elevations between 3000 m and up to 4300 m, mainly from Venezuela to northern Peru. The páramos occupy more than 30,000 km<sup>2</sup> in South America. Ecuador is the country with the highest proportion of its territory covered by páramos (with 7%) Páramo lakes constitute a primary source of drinking water for large populations in Ecuador, Colombia, and Venezuela. In Ecuador, the air temperature is relatively constant, the average annual rainfall is 1345 mm and occurs throughout most of the year; only ~ 12% of the days are dry. These conditions have favored the formation of peat and a large number of wetlands and lakes. The páramo usually presents deep and organic rich soils (30% by volume) and its vegetation is dominated by grasslands (Zapata et al. 2021).

### Lake sampling

The lakes were sampled at different periods from April 2019 to March 2022, with an inflatable boat, during daytime (early morning to late afternoon), and using three strategies: (1) Lake Salve Fecha was sampled at a single location (close to the edge of the lake) during four occasions in 2019 and 2020 (1 April 2019, 1 September 2019, 1 October 2019, and 1

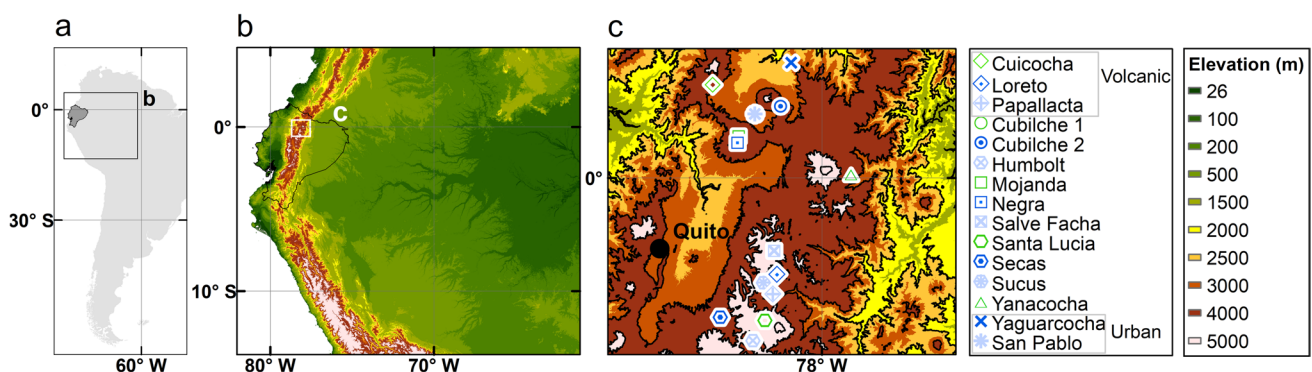


Fig. 1 Fifteen sampled highland lakes in Northern Ecuador

January 2019) as part of the sampling of the Napo river network (Chiriboga and Borges 2023); (2) Lakes Santa Lucía, Secas, and Humbolt were sampled on two occasions (24–25 September 2021 and 17–21 December 2021), along a transect across the longest length with equidistant points ( $n = 9, 6, 3$ , respectively); (3) the remaining 11 lakes (Cubilche 1 and 2, Cuicocha, Loreto, Mojanda, Negra, Papallacta, San Pablo, Sucus, Yaguarcocha, and Yanacocha) were sampled at only one station in the center of the lake on one occasion, from 22 March 2022 to 28 March 2022. Admittedly, CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in surface waters of lakes show variations in time (from subdaily to seasonal) and horizontally (from shore to the deepest part frequently in the center) (Fernandez et al. 2020; Ngochera and Bootsma 2020; Mandryk et al. 2021; Borges et al. 2022; Peacock et al. 2023), but a “one station in the center” snapshot survey of lakes has frequently been used to investigate broad geographical gradients as a function, for instance, of climate or catchment land cover (Juutinen et al. 2009; Finlay et al. 2010; Lapiere and del Giorgio 2012; Kortelainen et al. 2020; Casas-Ruiz et al. 2021; Rodríguez-Cardona et al. 2023). Albeit not optimal, the “one station in the center” snapshot survey approach is a tradeoff between the detail of the description of within lake horizontal/temporal variability and a broad description of geographical gradients, with typical resources allocated to sampling effort. A statistical summary [average  $\pm$  standard deviation (st. dev.), coefficient of variation (CV), median, first quartile (Q1), second quartile (Q2), minimum–maximum range] of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) and

the dissolved CH<sub>4</sub> and N<sub>2</sub>O concentrations in Lakes Salve Fecha, Santa Lucía, Secas, and Humbolt is given in Table 1. The data collected in Lake Salve Facha provided an indication of the temporal variability (during four different sampling occasions) at a single site, while data collected in Lakes Santa Lucía, Secas, and Humbolt provided an indication of the spatial variability (multiple sampling points), although also integrating temporal variability (during two different sampling occasions). Despite a relatively important variability indicated by the minimum–maximum and Q1–Q3 range, the median and average values of the three gases tended to converge, except for Lake Humbolt, possibly because of a lower number of samples ( $n = 6$ ) (Lewis 1993). The CV indicated that CH<sub>4</sub> was more variable (35–89%) than pCO<sub>2</sub> (29–66%), while N<sub>2</sub>O was the least variable (6–16%). The variability of the three GHGs, as given by the CV, was highest in the smallest lake (Humbolt).

### Sample collection and analysis

Water temperature, specific conductivity, pH, and %O<sub>2</sub> were measured in surface water with a YSI multiparameter probe (ProPlus). Water samples for CH<sub>4</sub> and N<sub>2</sub>O analysis were collected in duplicate borosilicate serum bottles (Weathon; volume 40 mL) with a 2L polyethylene bottle with the bottom cut and fitted with a silicone tubing at the stopper (Abril et al. 2007), poisoned with 100  $\mu$ L of a saturated solution of HgCl<sub>2</sub> and sealed with a butyl stopper and crimped with an aluminum cap. After overnight equilibration, measurements

**Table 1** Statistical summary (average  $\pm$  st. dev., coefficient of variation (CV), median, first quartile (Q1), second quartile (Q2), minimum–maximum) of CH<sub>4</sub> concentration (nmol/L; relative analytical precision  $\pm$  10.9%), partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>; ppm, relative

analytical precision  $\pm$  5.2%), and N<sub>2</sub>O concentration (nmol/L, relative analytical precision  $\pm$  5.8%) in lakes where multiple samplings were carried out (Salve Fecha, Santa Lucía, Secas, and Humbolt). SA, surface area

		<i>n</i>	Average $\pm$ st. dev	CV (%)	Median	Q1	Q3	Minimum	Maximum
Santa Lucía *	CH <sub>4</sub> (nmol/L)	18	1577 $\pm$ 961	61	1467	709	2081	49	3347
	pCO <sub>2</sub> (ppm)	18	1193 $\pm$ 344	29	1121	1012	1385	672	1882
	N <sub>2</sub> O (nmol/L)	18	6.3 $\pm$ 0.8	13	6.1	5.6	6.9	5.0	8.0
Secas **	CH <sub>4</sub> (nmol/L)	12	743 $\pm$ 562	76	711	270	1173	46	1842
	pCO <sub>2</sub> (ppm)	12	667 $\pm$ 229	34	760	701	793	129	818
	N <sub>2</sub> O (nmol/L)	12	6.0 $\pm$ 0.9	15	6.2	5.6	6.6	4.4	7.2
Humbolt ***	CH <sub>4</sub> (nmol/L)	6	2513 $\pm$ 2243	89	1388	1267	4634	110	5165
	pCO <sub>2</sub> (ppm)	6	1344 $\pm$ 888	66	704	692	2297	691	2336
	N <sub>2</sub> O (nmol/L)	6	5.2 $\pm$ 0.8	16	5.7	4.4	5.8	4.2	5.9
Salve Facha ****	CH <sub>4</sub> (nmol/L)	4	178 $\pm$ 62	35	199	167	209	87	228
	pCO <sub>2</sub> (ppm)	4	1570 $\pm$ 618	39	1862	1541	1891	644	1912
	N <sub>2</sub> O (nmol/L)	4	7.5 $\pm$ 0.4	6	7.5	7.2	7.8	7.0	7.9

\*24 September 2021 ( $n = 9$ ) 17 December 2021 ( $n = 9$ )

\*\*25 September 2021 ( $n = 6$ ) 21 December 2021 ( $n = 6$ )

\*\*\*25 September 2021 ( $n = 3$ ) 21 December 2021 ( $n = 3$ )

\*\*\*\*1 April 2019 ( $n = 1$ ) 1 July 2019 ( $n = 1$ ) 1 October 2019 ( $n = 1$ ) 1 January 2020 ( $n = 1$ )

were made, on a headspace (Weiss 1981) (15 mL of high-purity N<sub>2</sub> into the 40 mL sample bottles), with a gas chromatograph (SRI 8610C) with a flame ionization detector for CH<sub>4</sub> and electron capture detector for N<sub>2</sub>O calibrated with CH<sub>4</sub>:N<sub>2</sub>O:N<sub>2</sub> gas mixtures (Air Liquide Belgium) with mixing ratios of 1, 10, and 30 ppm for CH<sub>4</sub>, and 0.2, 2.0, and 6.0 ppm for N<sub>2</sub>O, with a precision of  $\pm 10.9\%$  for CH<sub>4</sub> and  $\pm 5.8\%$  for N<sub>2</sub>O (based on duplicate samples).

The pCO<sub>2</sub> in the water was measured in the field with a Li-Cor Li-820 infrared gas analyzer based on the headspace technique from four 60-mL polypropylene syringes filled directly with surface water. A headspace of 30 mL was created with ambient air that was equilibrated in the syringe with 30 mL of water by vigorous shaking during 5 min. The water temperature after equilibration was measured immediately after injection of the headspace with a portable thermometer (Testo 925). The pCO<sub>2</sub> in the atmosphere was measured by injecting ambient air sampled with an additional polypropylene syringe. The Li-Cor Li-820 was calibrated with pure N<sub>2</sub> and CO<sub>2</sub>:N<sub>2</sub> gas mixtures (Air Liquide Belgium) of 388, 804, 3707, and 8146 ppm. The final pCO<sub>2</sub> value was computed taking into account the partitioning of CO<sub>2</sub> between water and the headspace, as well as equilibrium with HCO<sub>3</sub><sup>-</sup> (Dickson et al. 2007) using water temperature measured in situ, after equilibration, and TA. The precision of pCO<sub>2</sub> measurement was  $\pm 5.2\%$ .

The CO<sub>2</sub> concentration is expressed as partial pressure in parts per million (ppm) and as dissolved concentration for CH<sub>4</sub> (nmol/L). N<sub>2</sub>O data are presented as percent of saturation level (%N<sub>2</sub>O, where atmospheric equilibrium corresponds to 100%), computed from the global mean N<sub>2</sub>O air mixing ratios (<https://www.esrl.noaa.gov/gmd/hats/combind/N2O.html>), using the Henry's constant (Weiss and Price 1980).

Samples for  $\delta^{13}\text{C}$ -DIC were collected in 12-mL Exetainer vials (Labco) and poisoned with 50  $\mu\text{L}$  of a saturated solution of HgCl<sub>2</sub>. Prior to the analysis of  $\delta^{13}\text{C}$ -DIC, a 2-mL helium headspace was created and 100  $\mu\text{L}$  of phosphoric acid (H<sub>3</sub>PO<sub>4</sub>, 99%) was added in the vial to convert carbonate (CO<sub>3</sub><sup>2-</sup>) and bicarbonate (HCO<sub>3</sub><sup>-</sup>) ions to CO<sub>2</sub>. After overnight equilibration, up to 1 mL of the headspace was injected with a gastight syringe into a coupled elemental analyser—IRMS (EA-IRMS, Thermo FlashHT or Carlo Erba EA1110 with DeltaV Advantage). Calibration was performed with certified standards (NBS-19 or IAEA-CO-1 and LSVEC). The obtained data were corrected for isotopic equilibration between dissolved and gaseous CO<sub>2</sub> (Gillikin and Bouillon 2007). Reproducibility of measurement based on duplicate injections was typically better than  $\pm 0.2\%$ .

Water was collected in surface water with a 2-L polyethylene bottle, and a known volume was filtered through 47 mm diameter GF/F Whatman glass fiber filters. The filtrated water was collected and further filtered through

polyethersulfone syringe encapsulated filters (0.2  $\mu\text{m}$  porosity) for NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, TA, major elements (Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, and K<sup>+</sup>), as well as DSi and Fe,  $\delta^{18}\text{O}$ -H<sub>2</sub>O,  $\delta^2\text{H}$ -H<sub>2</sub>O and DOC. An additional water filtration was made on 25 mm diameter GF/F Whatman glass fiber filters for POC analysis.

Water samples for nutrient analysis (NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, and NH<sub>4</sub><sup>+</sup>) were stored frozen ( $-20\text{ }^\circ\text{C}$ ) in 50-mL polypropylene vials. NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> were determined with the sulfanilamide colorimetric with the vanadium reduction method (American Public Health Association 1998), and NH<sub>4</sub><sup>+</sup> with the dichloroisocyanurate–salicylate–nitroprussiate colorimetric method (Standing committee of Analysts 1981). Detection limits were 0.3, 0.01, and 0.15  $\mu\text{mol/L}$  for NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>, respectively. Precisions were  $\pm 0.02\text{ } \mu\text{mol/L}$ ,  $\pm 0.02\text{ } \mu\text{mol/L}$ , and  $\pm 0.1\text{ } \mu\text{mol/L}$  for NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>, respectively.

Samples for TA analysis were stored at ambient temperature in polyethylene 55-mL vials. TA was measured by open-cell titration with HCl 0.1 mol/L according to Gran (1952), and data quality was checked with certified reference material obtained from Andrew Dickson (Scripps Institution of Oceanography, University of California, San Diego, CA), with a typical reproducibility better than  $\pm 3\text{ } \mu\text{mol/kg}$ . DIC was computed from TA and pCO<sub>2</sub> measurements using the carbonic acid dissociation constants for freshwater of Millero (1979) using the CO2sys package.

Samples for  $\delta^{18}\text{O}$ -H<sub>2</sub>O and  $\delta^2\text{H}$ -H<sub>2</sub>O were stored at ambient temperature in polypropylene 8-mL vials.  $\delta^2\text{H}$ -H<sub>2</sub>O was measured on H<sub>2</sub> gas derived from a high-temperature (1030  $^\circ\text{C}$ ) Cr-based reactor by automated injections of water using a TriPlus autosampler on an elemental analyzer (Thermo Flash HT/EA; Thermo Finnigan) coupled to a continuous-flow isotope-ratio mass spectrometer (Delta V Advantage; Thermo Finnigan).  $\delta^{18}\text{O}$ -H<sub>2</sub>O values were measured on a Thermo GasBench II coupled to a Thermo Delta XP IRMS after equilibration with CO<sub>2</sub>. The long-term uncertainty for standards was  $\pm 0.1\text{ } \text{‰}$  for  $\delta^{18}\text{O}$  and  $\pm 2.0\text{ } \text{‰}$  for  $\delta^2\text{H}$ -H<sub>2</sub>O.

Samples for major elements were stored at ambient temperature in 20-mL scintillation vials and preserved with 50  $\mu\text{L}$  of HNO<sub>3</sub> (65%). Major elements were measured with inductively coupled plasma MS (ICP-MS; Agilent 7700x) calibrated with SRM1640a from National Institute of Standards and Technology, TM-27.3 (lot 0412) and TMRain-04 (lot 0913) from Environment Canada, and SPS-SW2 Batch 130 from Spectrapure Standard. The limit of quantification that in this case also corresponds to the background equivalent concentration was 0.038 mg/L for Na<sup>+</sup>, 0.005 mg/L for Mg<sup>2+</sup>, 0.077 mg/L for K<sup>+</sup>, 0.029 mg/L for Ca<sup>2+</sup>, 0.185 mg/L for DSi, and 0.37  $\mu\text{g/L}$  for Fe. The precision was  $\pm 4.1\%$  for Na<sup>+</sup>,  $\pm 3.8\%$  for Mg<sup>2+</sup>,  $\pm 4.9\%$  for K<sup>+</sup>,  $\pm 2.4\%$  for Ca<sup>2+</sup>,  $\pm 5.8\%$  for DSi, and  $\pm 4.6\%$  for Fe.

Samples for DOC analysis were stored at ambient temperature and in the dark in 40-mL brown borosilicate vials with polytetrafluoroethylene (PTFE) coated septa and poisoned with 50  $\mu\text{L}$  of  $\text{H}_3\text{PO}_4$  (85%). DOC concentration was measured with a wet oxidation total organic carbon analyzer (IO Analytical Aurora 1030W), with a typical reproducibility better than  $\pm 5\%$ .

Filters for POC analysis were decarbonated with HCl fumes for 4 h and dried before encapsulation into silver cups; POC was measured on an EA-IRMS (Thermo FlashHT with DeltaV Advantage), with a reproducibility better than  $\pm 5\%$ . Data were calibrated with certified (IAEA-600: caffeine) and in-house standards (leucine, tuna muscle tissue) that were previously calibrated versus certified standards.

### Flux computations and upscaling

The dissolved concentrations of  $\text{CO}_2$  and  $\text{CH}_4$  were estimated for each of 12,549 lakes in HydroLAKES (Messenger et al. 2016) from statistical relations as function of lake surface area (LSA), derived from our own data in Ecuadorian highland lakes and data available from literature in the floodplain lakes of the Central Amazon (Abril et al. 2013; Amaral et al. 2020; Barbosa et al. 2016; Mitchell et al. 2018). We excluded from the analysis reservoirs that represent 0.06% of lake counts ( $n=8$ ) and 11% of total lacustrine surface area (2773  $\text{km}^2$ ) in the HydroLAKES data set.

The  $\text{pCO}_2$  (in ppm) was computed from two linear regressions as a function of LSA (in  $\text{km}^2$ ) according to:

$$\log(\text{pCO}_2) = -0.1293(\pm 0.0346) \times \text{LSA} + 3.063(\pm 0.073) \\ (r^2 = 0.58; p = 0.0039; n = 12)$$

for an elevation  $> 500$  m (highlands), excluding three lakes (Cuicocha, Loreto, and Papallacta) and

$$\log(\text{pCO}_2) = -0.0003745(\pm 0.0001642) \times \text{LSA} + 3.508(\pm 0.061) \\ (r^2 = 0.26; p = 0.0376; n = 17)$$

for an elevation  $< 500$  m (lowlands).

The  $\text{CH}_4$  (in  $\text{nmol/L}$ ) was computed from a linear regression as function of LSA (in  $\text{km}^2$ ) according to:

$$\log(\text{CH}_4) = -0.3753(\pm 0.0426) \times \log(\text{LSA}) + 2.545(\pm 0.066) \\ (r^2 = 0.71; p < 0.0001; n = 34)$$

for both highland and lowland lakes.

The water temperature (WT in  $^\circ\text{C}$ ) was computed from the air temperature (AT in  $\times 10$   $^\circ\text{C}$ ) from HydroLAKES, according to:

$$\log(\text{WT}) = 0.002192(\pm 0.000069) \times \text{AT} + 0.8905(\pm 0.0151) \\ (r^2 = 0.98; p < 0.0001; n = 23)$$

The flux of  $\text{CO}_2$  ( $F_{\text{CO}_2}$ ) and  $\text{CH}_4$  ( $F_{\text{CH}_4}$ ) between surface waters and the atmosphere was computed according to:

$$F = k \times \Delta G,$$

where  $k$  is the gas transfer velocity and  $\Delta G$  is the air–water gradient of a given gas.

The  $k$  normalized to a Schmidt number ( $Sc$ ) of 600 ( $k_{600}$  in  $\text{cm/h}$ ) was derived from the parameterization as a function wind speed (Cole and Caraco 2018):

$$k_{600} = 2.07 + 0.215 \times u^{1.7},$$

where  $u$  is wind speed ( $\text{m/s}$ ) that was extracted from WorldClim (Fick and Hijmans 2017) for each lake in HydroLAKES.

$k_{600}$  was converted to  $k$  for the  $Sc$  at in situ temperature (Wanninkhof 1992) according to:

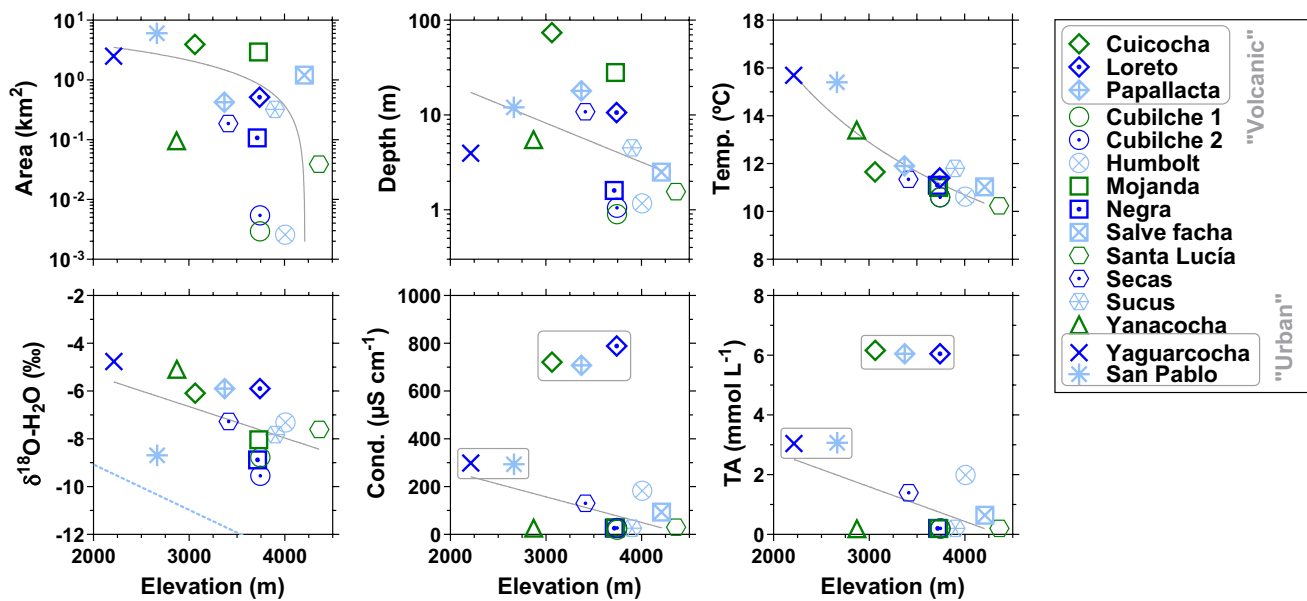
$$k = k_{600} \times (Sc/600)^{-0.5}.$$

A Monte Carlo error analysis (10,000 iterations) was conducted by propagating errors in the slope and constant of the equations of the gas concentrations and WT (refer to above equations), an uncertainty of  $\pm 65\%$  and  $\pm 40\%$  on dissolved  $\text{CH}_4$  concentration and  $\text{pCO}_2$ , respectively, to account for within-lake variability (based on Table 1), and an error of  $\pm 30\%$  for  $k_{600}$  based on the reported variability across different lakes of  $k$  at low wind speeds (Cole and Caraco 2018).

## Results and discussion

We sampled 15 lakes in Ecuador along an elevational gradient (2213–4361 m) (Fig. 1). The surface area of the sampled lakes ranged between 0.003 and 6.1  $\text{km}^2$ , and lakes at higher elevation tended to be smaller (Fig. 2). Lakes were generally shallow, with depths ranging between 0.9 and 72 m, with a tendency toward shallower lakes at higher elevation (Fig. 2). Lakes Cuicocha, Loreto, and Papallacta are crater lakes (hereafter referred to as “volcanic”), and the others are of glacial origin. Among the latter, San Pablo and Yaguarcocha are surrounded by the cities of Otavalo and Ibarra, respectively (hereafter referred to as “urban”).

Water temperature ranged between 10.2 and 15.7  $^\circ\text{C}$  and followed a clear elevational gradient (Fig. 2). The specific conductivity in the sampled lakes was generally low (20–183  $\mu\text{S/cm}$ ), except for the two “urban” lakes (San Pablo and Yaguarcocha) (294–298  $\mu\text{S/cm}$ ) and the three “volcanic” lakes (Cuicocha, Loreto and Papallacta) (708–722  $\mu\text{S/cm}$ ) (Fig. 2). The conductivity of the three “volcanic” lakes was lower than in crater lakes in active volcanic systems in Ecuador characterized by conductivity values  $> 10,000$   $\mu\text{S/cm}$  (Inguaggiato et al. 2010; Melián et al. 2021). Conductivity was negatively related to elevation for the non-“volcanic”



**Fig. 2** Lake area (km<sup>2</sup>), lake depth (m), water temperature (Temp. °C, analytical precision  $\pm 0.1$  °C), stable isotope composition of O of H<sub>2</sub>O ( $\delta^{18}\text{O}\text{-H}_2\text{O}$  in ‰, analytical precision  $\pm 0.1$ ‰), specific conductivity (cond.  $\mu\text{S}/\text{cm}$ , analytical precision  $\pm 0.1$   $\mu\text{S}/\text{cm}$ ), and total alkalinity (TA in mmol/L, analytical precision  $\pm 0.003$  mmol/L) in surface

waters of 15 highland lakes in Northern Ecuador (Fig. 1) versus elevation (m). Solid grey lines show fits to the data (Table S2). Dotted blue line indicates the  $\delta^{18}\text{O}\text{-H}_2\text{O}$  elevation gradient for Ecuador based on the global rain water isoscape given by Inguaggiato et al. (2010)

lakes, possibly reflecting the combination of decreasing lake size and temperature, and increasing soil organic matter content and precipitation with increasing elevation. In the Andean páramo, the soil organic carbon increases with elevation because of the decrease in temperature that leads to a decrease of organic matter mineralization in soils enhancing organic matter accumulation in soils (Poulenard et al. 2003). The high organic matter content of soils and lower temperature are not favorable to rock weathering leading to soil water with low mineral content (low conductivity). Additionally, high elevation páramo areas are characterized by high precipitation leading to dilution of solutes in surface waters, which should additionally contribute to decrease conductivity. TA showed very similar patterns to those of conductivity, with the three “volcanic” lakes characterized by higher TA values (6.05–6.16 mmol/L), followed by the two “urban” lakes (3.04–3.07 mmol/L), and the remaining lakes with the lowest TA values (0.19–1.99 mmol/L).

The cation sequence was dominated by Ca<sup>2+</sup> or Na<sup>+</sup> followed by Mg<sup>2+</sup> and K<sup>+</sup> (Table S1). The lakes with a high total cation content (21–147 mg/L and 1035–7916  $\mu\text{eq}/\text{L}$ ) were dominated by Na<sup>+</sup> (Na<sup>+</sup> > Ca<sup>2+</sup> > Mg<sup>2+</sup> > K<sup>+</sup>), characteristic of sodium bicarbonate lakes. Lakes with a lower total cation content ( $\sim 5$  mg/L and  $\sim 235$   $\mu\text{eq}/\text{L}$ ) were dominated by Ca<sup>2+</sup> (Ca<sup>2+</sup> > Na<sup>+</sup> > Mg<sup>2+</sup> > K<sup>+</sup>) (Table S1). This might indicate that the cation content of the sample lakes was related to the weathering of primarily volcanic rocks that dominate in the area (rhyolite, andesite, dacite, and diorite)

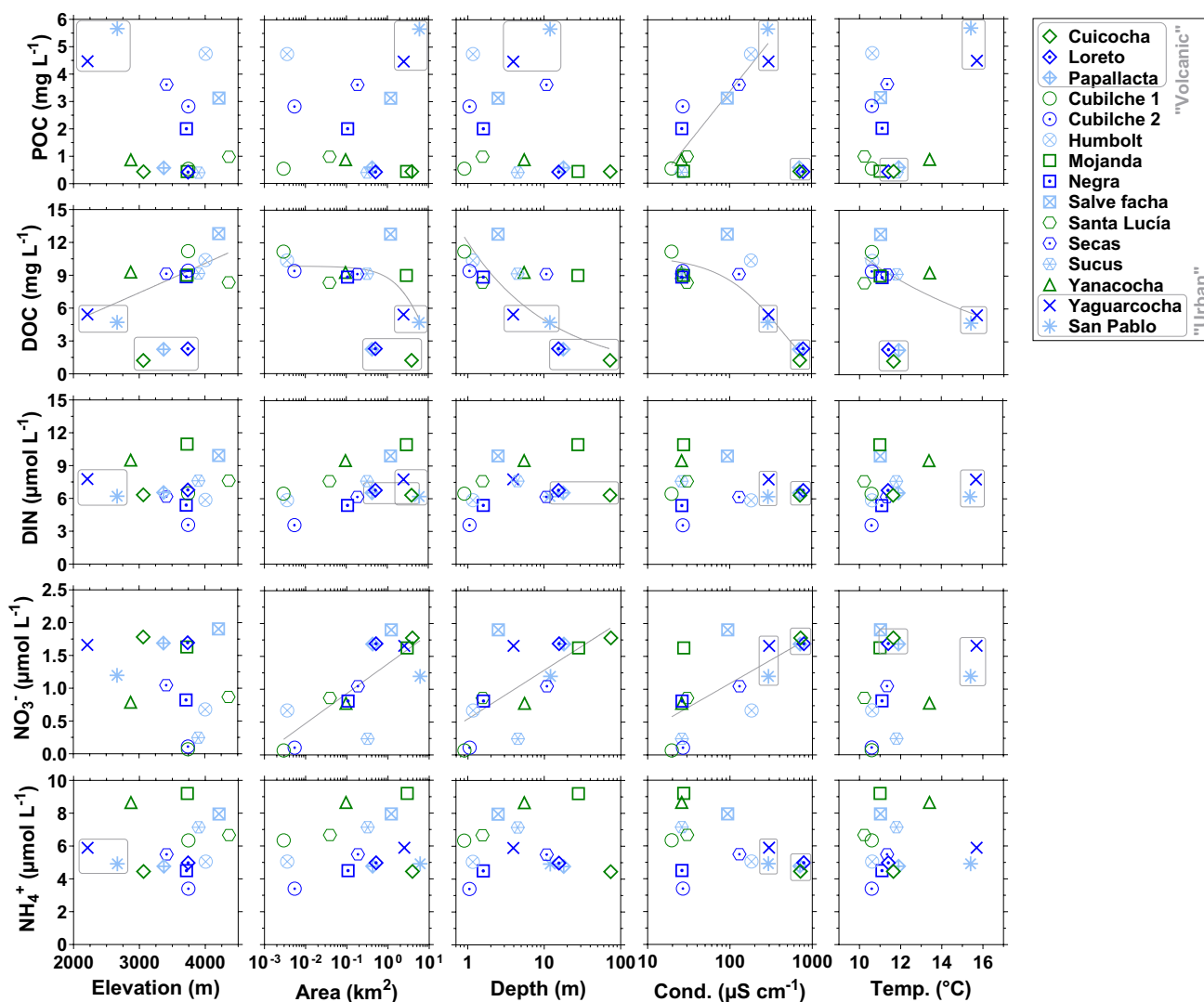
(Mosquera et al. 2022). The three “volcanic” lakes (Cuicocha, Loreto, and Papallacta) were characterized by higher total cation content (144–147 mg/L 7778–7916  $\mu\text{eq}/\text{L}$ ) than the other 12 lakes (Table S1). The composition of the three “volcanic” lakes in divalent ions (Ca<sup>2+</sup> and Mg<sup>2+</sup>) relative to Na<sup>+</sup> was consistent with the other 12 lakes (Fig. S1). However, the three “volcanic” lakes had lower K<sup>+</sup>, DSi and TA relative to the other 12 lakes (Fig. S1), which might possibly indicate different weathering mechanism in the three “volcanic” lakes compared with the other 12 lakes: either a noncongruent dissolution of aluminosilicate minerals or alternatively the weathering of a higher portion of sodium rich volcanic rocks in the three “volcanic” lakes compared with the other 12 lakes (Mosquera et al. 2022). The Fe content was negatively related to Na<sup>+</sup>, and the three “volcanic” lakes had lower Fe relative to Na<sup>+</sup> compared with the other 12 lakes. This pattern might be possibly related to the chelation of Fe to dissolved organic matter, as Fe was correlated to DOC (Fig. S2).

The  $\delta^{18}\text{O}\text{-H}_2\text{O}$  values were negatively related to elevation and paralleled elevational gradient based on the global rain water isoscape (Inguaggiato et al. 2010) (Fig. 2). This might possibly indicate that the sampled lakes were mainly recharged by local precipitation (meteoric recharge). The offset between observed  $\delta^{18}\text{O}\text{-H}_2\text{O}$  values in the sampled lakes and the precipitation elevational gradient most probably resulted from fractionation during evaporation. The observed  $\delta^2\text{H}$  and  $\delta^{18}\text{O}\text{-H}_2\text{O}$  values in the sampled lakes

deviated from the local meteoric water line (LMWL) (Inguaggiato et al. 2010), confirming substantial fractionation during evaporation (Fig. S3).

The DOC concentrations ranged between 1.2 and 12.8 mg/L, and DOC data points appeared to be grouped in clusters as a function of several variables such as elevation, lake area, lake depth, conductivity, and water temperature (Fig. 3). The three “volcanic” lakes (Cuicocha, Loreto, and Papallacta) were characterized by the lowest DOC concentrations (1.2–2.3 mg/L), the two “urban” lakes (San Pablo and Yaguarcocha) were characterized by intermediate DOC concentrations (4.7–5.4 mg/L), and the remaining 10 lakes were characterized by the highest DOC concentrations (8.3–12.8 mg/L). DOC was positively correlated to elevation

and negatively related to temperature (excluding the three “volcanic” lakes). This might reflect the higher inputs of soil DOC in lakes more strongly influenced by inputs from the páramos. The negative relation between DOC and conductivity is consistent with the idea that soils richer in organic matter tend to lead to less mineralized water. Low inputs of soil DOC in the three “volcanic” lakes are possibly related to a low ratio of catchment area to lake surface area, reducing the inputs from soils to the lake. Additionally, the DOC content could be partly modulated by lake size, as indicated by the negative relation between DOC concentration and lake area (Fig. 3). Small lakes might have a greater connectivity and potential exchange of organic carbon with the riparian wetlands and surrounding terrestrial landscape along



**Fig. 3** Particulate (POC in mg/L, relative analytical precision  $\pm 5\%$ ) and dissolved (DOC in mg/L, relative analytical precision  $\pm 5\%$ ) organic carbon, dissolved inorganic nitrogen (DIN in  $\mu\text{mol/L}$ ),  $\text{NO}_3^-$  ( $\mu\text{mol/L}$ , analytical precision  $0.1 \mu\text{mol/L}$ ) and  $\text{NH}_4^+$  ( $\mu\text{mol/L}$ , analytical precision  $0.02 \mu\text{mol/L}$ ) in surface waters of 15 highland lakes

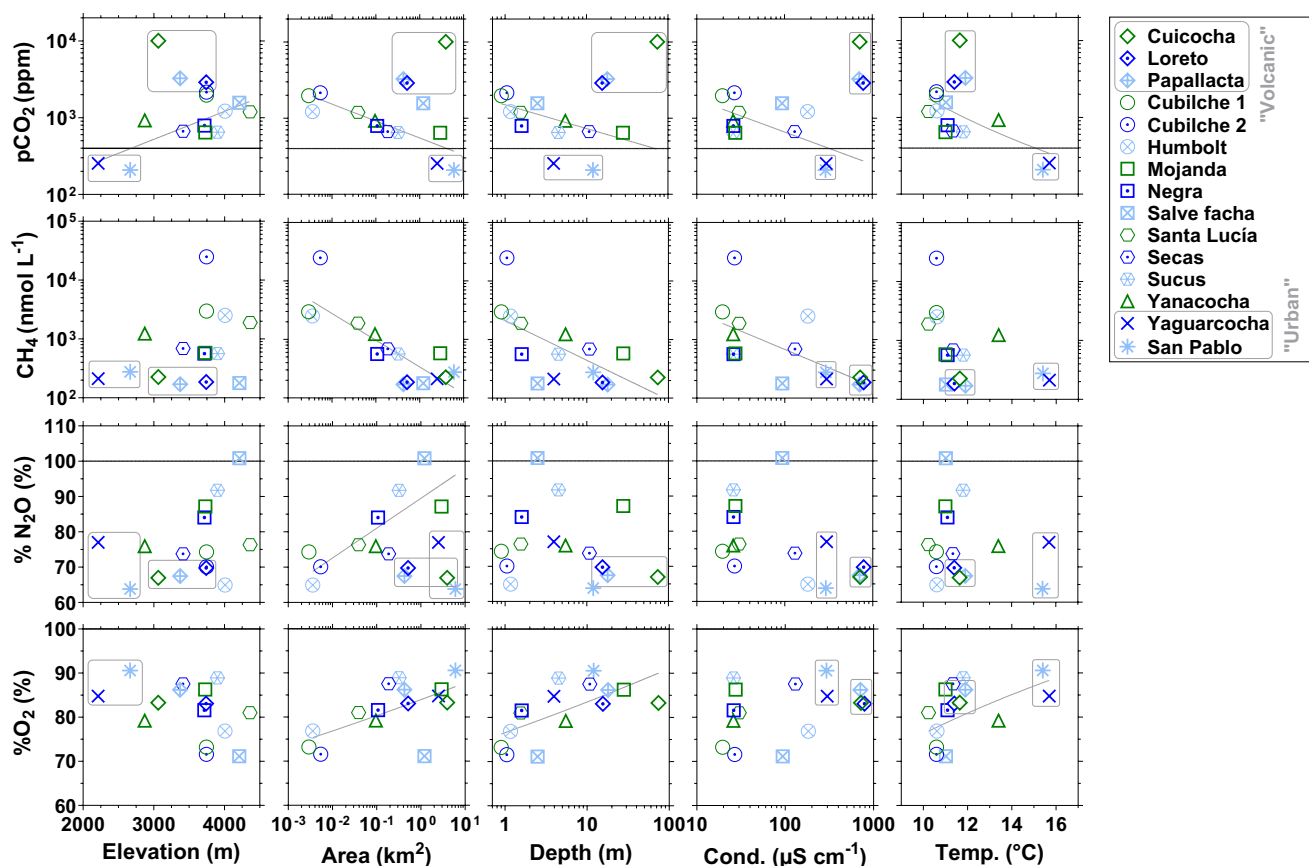
in Northern Ecuador (Fig. 1) versus elevation (m), lake surface area ( $\text{km}^2$ ), lake depth (m), specific conductivity (cond.  $\mu\text{S/cm}$ , analytical precision  $\pm 0.1 \mu\text{S/cm}$ ), and water temperature (temp.  $^\circ\text{C}$ , analytical precision  $\pm 0.1 \text{ }^\circ\text{C}$ ). Solid grey lines show fits to the data (Table S2)



its periphery relative to lake surface area than larger lakes (Staehr et al. 2012). The two above explanations of DOC patterns are compatible as lake size decreases with elevation and the smaller lakes were, thus, located in the páramos with organic carbon rich soils. The POC concentrations ranged between 0.4 and 5.7 mg/L, and the POC data points were less clearly clustered than DOC data points and did not correlate with elevation, lake area, lake depth, or water temperature but did show a relationship with conductivity (Fig. 3). One of the two “urban” lakes (San Pablo) was characterized by the highest POC value (5.7 mg/L), while the three “volcanic” lakes were characterized by some the lowest POC values (0.4–0.6 mg/L). The correlation of POC with conductivity (excluding the three “volcanic” lakes) might possibly reflect a general dilution pattern by meteoric water of solutes and POC. Variations of DIN were modest (3.6–11.0 μmol/L) and DIN did not significantly correlate with elevation, lake area, lake depth, conductivity, and water temperature. DIN was dominated by NH<sub>4</sub><sup>+</sup> (70–98%) that ranged between 3.4

and 9.2 μmol/L. NH<sub>4</sub><sup>+</sup> did not significantly correlate with elevation, lake area, lake depth, conductivity, and water temperature. NO<sub>3</sub><sup>-</sup> ranged between 0.1 and 1.9 μmol/L and did not significantly correlate with elevation and temperature but correlated positively to lake area, lake depth, and conductivity. These correlations might possibly reflect a stronger impact in smaller and shallower water bodies of soil–water inputs with a low NO<sub>3</sub><sup>-</sup> content related to denitrification in highland soils (Pineda et al. 2021). The NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub><sup>-</sup> values we report in Ecuadorian lakes were consistent with those in 51 páramo lakes in the Colombian Andes investigated by Zapata et al. (2021) who reported an average NH<sub>4</sub><sup>+</sup> concentration of 36 μmol/L (range 20–60 μmol/L), and an average NO<sub>3</sub><sup>-</sup> concentration of 1.1 μmol/L (range 0.1–12.8 μmol/L).

The pCO<sub>2</sub> values ranged between 208 and 10,069 ppm (Fig. 4). The pCO<sub>2</sub> data points appeared to be grouped into the same three clusters observed for DOC (Fig. 4). The three “volcanic” lakes (Cuicocha, Loreto, and Papallacta) were characterized by higher pCO<sub>2</sub> values (3269–10,069 ppm),



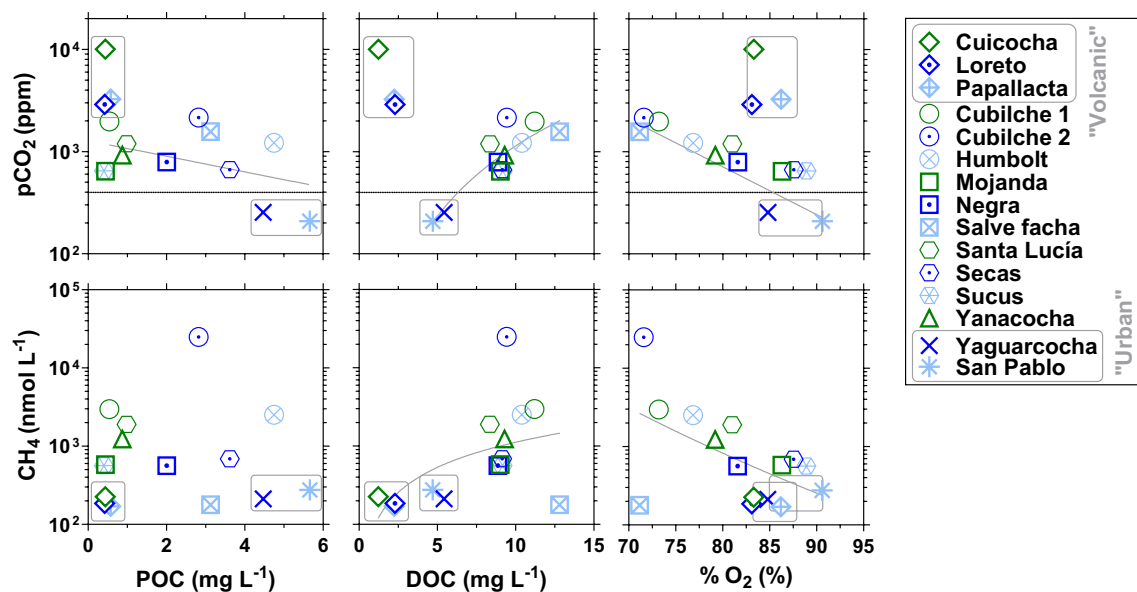
**Fig. 4** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm, relative analytical precision ± 5.2%), dissolved CH<sub>4</sub> concentration (nmol/L; relative analytical precision ± 10.9%), N<sub>2</sub>O saturation level (%N<sub>2</sub>O in %, relative analytical precision ± 5.8%), and O<sub>2</sub> saturation level (%O<sub>2</sub> in %, analytical precision ± 0.1%) in surface waters of 15 highland lakes in Northern Ecuador (Fig. 1) versus elevation (m), lake surface area

(km<sup>2</sup>), lake depth (m), specific conductivity (cond. μS/cm, analytical precision ± 0.1 μS/cm), and water temperature (Temp. °C analytical precision ± 0.1 °C). Dotted horizontal lines indicate equilibrium with the atmosphere and solid grey lines show fits to the data (Table S2)

the two “urban” lakes (San Pablo and Yaguarcocha) by the lowest  $p\text{CO}_2$  values (208–254 ppm), and the other páramo lakes by intermediary  $p\text{CO}_2$  values (644–2152 ppm). All lakes had  $p\text{CO}_2$  values above atmospheric equilibrium (400 ppm) and, thus, acted as sources of  $\text{CO}_2$  to the atmosphere, except for the two “urban” lakes that acted as sinks of atmospheric  $\text{CO}_2$ . The highest  $p\text{CO}_2$  values in the three “volcanic” lakes seemed to be in part related to geogenic  $\text{CO}_2$  inputs, reflecting the influence of postvolcanic activities, such as emission of volcanic gases and input of hydrothermal water. The  $\delta^{13}\text{C}$ -DIC values in surface waters of the “volcanic” lakes were extremely high ( $\sim 3.5\text{‰}$ ) (Fig. S4), and magmatic inputs likely provide the dominant imprint on  $\delta^{13}\text{C}$ -DIC signatures as also reported in Lakes Kivu ( $\sim 1.0\text{‰}$ ) and Sonachi ( $\sim 9.0\text{‰}$ ) (also a crater lake) (Borges et al. 2012). The  $\delta^{13}\text{C}$  values of volcanic  $\text{CO}_2$  in gas bubbles in Ecuadorian crater lakes range between  $-2\text{‰}$  and  $-6\text{‰}$  (Inguaggiato et al. 2010; Melián et al. 2021). Consequently, the positive  $\delta^{13}\text{C}$ -DIC values observed in the three “volcanic” lakes also probably reflect the uptake of  $^{12}\text{C}$  by primary production and enrichment in  $^{13}\text{C}$  in the residual DIC. This effect might be possibly enhanced by long water residence time typical of small crater lakes devoid of outlets. The two “urban” lakes were also characterized by enriched  $\delta^{13}\text{C}$ -DIC signatures, also possibly indicative of uptake of  $^{12}\text{C}$  by primary production consistent with the low  $p\text{CO}_2$  values (and high POC values). A general increase of  $\delta^{13}\text{C}$ -DIC was observed from  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC} = 0.5$  (half of the

DIC is in the form of  $\text{CO}_2$ ) toward  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC} = 1$  (all of the DIC is in the form of  $\text{HCO}_3^- + \text{CO}_3^{2-}$ ). This pattern possibly reflected a gradient of lakes from systems with low rock weathering due to dominance of deep organic soils (low  $\text{HCO}_3^-$ ) and high  $\text{CO}_2$  from respiration leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$  close to 0.5 with low  $\delta^{13}\text{C}$ -DIC and lakes with higher rock weathering leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$  closer to 1 with high  $\delta^{13}\text{C}$ -DIC.

The  $p\text{CO}_2$  values in the páramo lakes were positively correlated to elevation and negatively correlated to lake area, lake depth, temperature, and conductivity (excluding the three “volcanic” lakes—Cuicocha, Loreto and Papallacta). This was indicative of higher lacustrine  $p\text{CO}_2$  values in páramo lakes bordered by soils with a higher organic carbon content, as shown above for DOC (Fig. 3). Indeed,  $p\text{CO}_2$  was positively correlated to DOC, negatively to  $\% \text{O}_2$  and POC (Fig. 5). This possibly indicated a gradient of inputs of soil water with a low  $\text{O}_2$  and conductivity and high DOC and  $p\text{CO}_2$  among the páramo lakes that were more important at high elevations owing to soils with a higher organic matter content. Alternatively, the negative relation between  $p\text{CO}_2$  and  $\% \text{O}_2$  could also reflect an internal production of  $\text{CO}_2$  within the lakes (and concurrent consumption of  $\text{O}_2$ ) by respiration of allochthonous organic matter, mostly in dissolved form, indirectly leading to a positive relation between  $p\text{CO}_2$  and DOC. The implications of the relation between  $p\text{CO}_2$  and DOC are discussed in more detail hereafter. Conversely, in lakes with lower inputs of soil DOC (the “urban” lakes



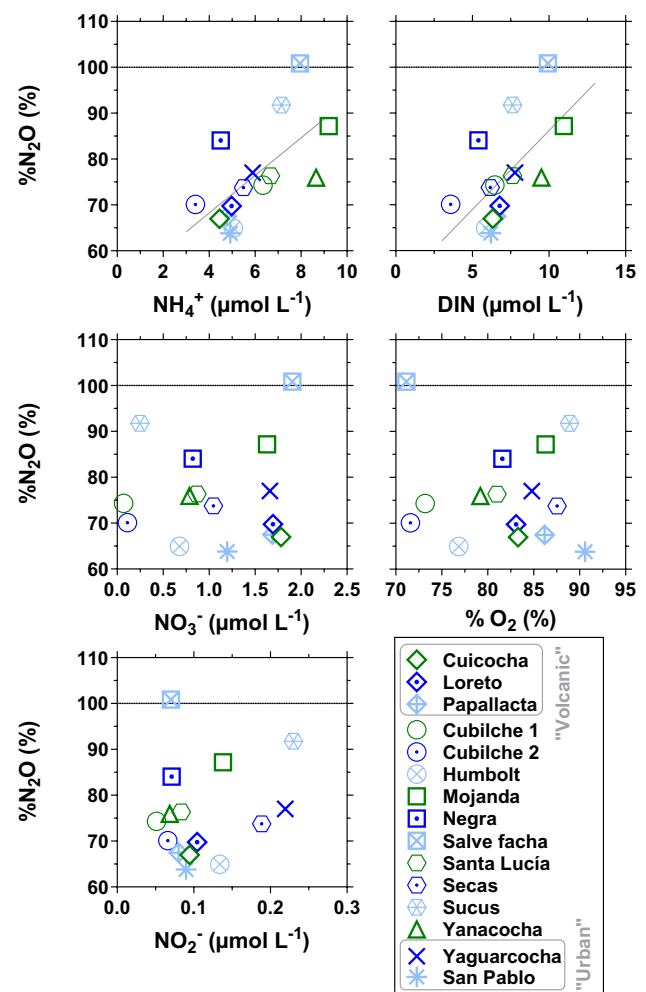
**Fig. 5** Partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$  in ppm, relative analytical precision  $\pm 5.2\%$ ) and dissolved  $\text{CH}_4$  concentration ( $\text{nmol/L}$ , relative analytical precision  $\pm 10.9\%$ ) versus particulate (POC in  $\text{mg/L}$  relative analytical precision  $\pm 5\%$ ) and dissolved (DOC in  $\text{mg/L}$  relative analytical precision  $\pm 5\%$ ) organic carbon and  $\text{O}_2$  saturation level ( $\% \text{O}_2$  in

$\%$ , analytical precision  $\pm 0.1\%$ ) in surface waters of 15 highland lakes in Northern Ecuador (Fig. 1). Dotted horizontal lines indicate equilibrium with the atmosphere and solid grey lines show fits to the data (Table S2)

at lower elevation), the lower DOC concentrations might have allowed the development of phytoplankton (lesser light limitation, e.g. Karlsson et al. 2009; Seekell et al. 2015) contributing to the decreasing pattern of pCO<sub>2</sub> with decreasing DOC and increasing POC. Additionally, the lakes at higher elevation tended to be shallower and smaller (Fig. 2). As mentioned above to explain the pattern of DOC and lake size, smaller lakes should have a greater connectivity and potential exchange of CO<sub>2</sub> from soils along their periphery relative to lake surface area (Staehr et al. 2012). Also, there is a stronger fetch limitation of *k* in smaller lakes (Wanninkhof 1992) that can possibly contribute to the accumulation of CO<sub>2</sub> in lake surface water of smaller lakes, while higher *k* in larger lakes can promote degassing and bring CO<sub>2</sub> levels closer to atmospheric equilibrium. These processes combined possibly contributed to the explanation of the negative relation between pCO<sub>2</sub> with lake surface area as well as with lake depth (Fig. 4). Since the CO<sub>2</sub> dynamics were related to organic matter degradation (in soils or in lake), %O<sub>2</sub> correlated to pCO<sub>2</sub> (Fig. 5) and %O<sub>2</sub> patterns mirrored those of pCO<sub>2</sub>: %O<sub>2</sub> was positively related to lake area, lake depth, and water temperature but was not significantly correlated to conductivity and elevation (Fig. 4).

Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol/L in the sampled lakes (Fig. 4). Dissolved CH<sub>4</sub> concentrations were not significantly correlated to elevation, water temperature, and POC but were negatively correlated to lake area, lake depth, conductivity, DOC and %O<sub>2</sub> (Figs. 4, 5). The patterns of CH<sub>4</sub> paralleled those of pCO<sub>2</sub>. The smaller lakes bordered by soils with a higher organic carbon content were likely characterized by a higher CH<sub>4</sub> production in lake sediments or in surrounding soils. Additionally, smaller lakes were generally shallower, also potentially affecting the relative removal of CH<sub>4</sub> by bacterial oxidation compared with the sedimentary input of CH<sub>4</sub>, probably resulting in higher CH<sub>4</sub> dissolved concentrations in shallower lakes (Borges et al. 2011; 2022).

%N<sub>2</sub>O ranged between 63.8% and 100.8%. All the sampled lakes were sinks of atmospheric N<sub>2</sub>O, except Salve Fecha, which was close to saturation (100.8%). This possibly implies a removal of N<sub>2</sub>O by denitrification due to the complete reduction of N<sub>2</sub>O to N<sub>2</sub>. %N<sub>2</sub>O did not significantly correlate to elevation, lake depth, conductivity, and temperature but correlated to lake area (Fig. 4). NO<sub>3</sub><sup>-</sup> also correlated with lake area (Fig. 3). Similar patterns of %N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup> as a function of lake area might possibly reflect a stronger impact in smaller water bodies of soil–water inputs with low content in N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup> related to denitrification in highland páramo soils (Pineda et al. 2021). Yet, paradoxically, %N<sub>2</sub>O did not significantly correlate with NO<sub>3</sub><sup>-</sup> but correlated with NH<sub>4</sub><sup>+</sup> (and thus also with DIN) (Fig. 6), although NH<sub>4</sub><sup>+</sup>



**Fig. 6** N<sub>2</sub>O saturation level (%N<sub>2</sub>O in %, relative analytical precision  $\pm 5.8\%$ ) versus NH<sub>4</sub><sup>+</sup> ( $\mu\text{mol/L}$ , analytical precision  $\pm 0.02 \mu\text{mol/L}$ ), dissolved inorganic nitrogen (DIN in  $\mu\text{mol/L}$ ), NO<sub>3</sub><sup>-</sup> ( $\mu\text{mol/L}$ , analytical precision  $\pm 0.1 \mu\text{mol/L}$ ), NO<sub>2</sub><sup>-</sup> ( $\mu\text{mol/L}$ , analytical precision  $\pm 0.02 \mu\text{mol/L}$ ) and O<sub>2</sub> saturation level (%O<sub>2</sub> in %, analytical precision  $\pm 0.1\%$ ) in surface waters of 15 highland lakes in Northern Ecuador (Fig. 1). Dotted horizontal lines indicate equilibrium with the atmosphere and solid grey lines show fits to the data (Table S2)

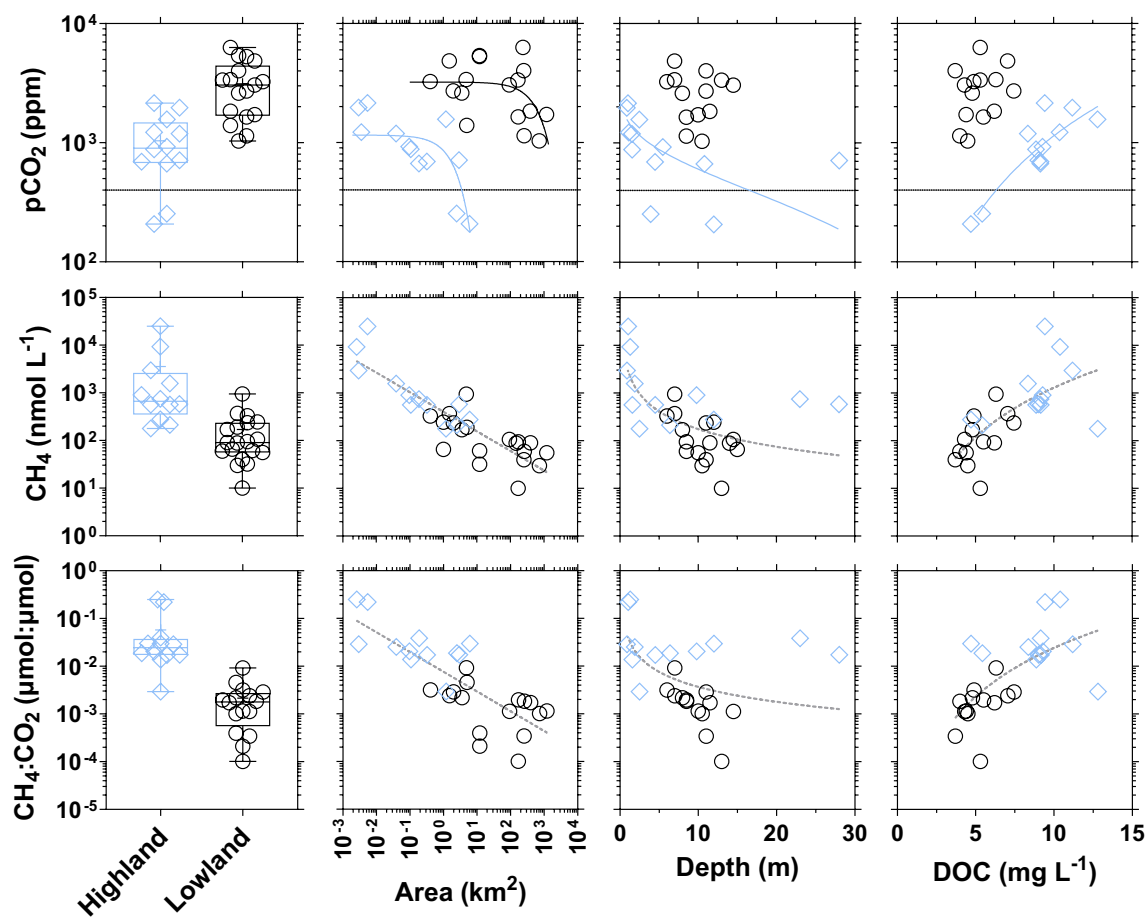
did not correlate with lake area (Fig. 3). %N<sub>2</sub>O did not also significantly correlate to NO<sub>2</sub><sup>-</sup> and %O<sub>2</sub>. The general lack of correlation of %N<sub>2</sub>O with other variables might possibly reflect that there was a low contribution of internal cycling of N<sub>2</sub>O within the lakes and that the N<sub>2</sub>O levels in lakes were primarily controlled by the inputs of soil–water in a low N<sub>2</sub>O level related to soil denitrification. Additionally, N<sub>2</sub>O was driven closer to equilibrium with the atmosphere in larger lakes where *k* increases with fetch (Wanninkhof 1992). Combined, these two processes might possibly explain the relation between %N<sub>2</sub>O and lake surface area.

## Comparison with floodplain lakes in central Amazon

The sampled páramo lakes were generally smaller ( $1.1 \pm 1.9$  km<sup>2</sup>, average  $\pm$  standard deviation) with higher DOC concentration ( $9.0 \pm 2.2$  mg/L) than the lowland floodplain lakes compiled from literature ( $183 \pm 311$  km<sup>2</sup>;  $5.3 \pm 1.1$  mg/L) (Fig. 7). Yet, the pCO<sub>2</sub> values in lowland floodplain lakes were higher ( $3115 \pm 1605$  ppm) than in the páramo lakes ( $1037 \pm 615$  ppm), even for overlapping values of depth, lake surface area, and DOC concentrations (Fig. 7). This pattern probably resulted from inputs of fluvial CO<sub>2</sub> in the floodplain lakes that led to higher pCO<sub>2</sub> values than in the páramo lakes. Bidirectional flows between rivers and floodplains determine the ecological functioning and biogeochemical fluxes of Amazonian floodplain lakes (Junk et al. 1989), including CO<sub>2</sub> dynamics (Abril et al. 2014;

Amaral et al. 2018). The pattern of lower pCO<sub>2</sub> values in mountainous lakes compared with lowland lakes generally agrees with other studies of CO<sub>2</sub> dynamics in mountainous lakes at higher latitude, such as in the Alps (Pighini et al. 2018), Rocky Mountains (McCrackin and Elser 2011; Crawford et al. 2015), Sierra Nevada (Cohen and Melack 2020), and the Tibetan Plateau (Yan et al. 2018, 2023; Wang et al. 2022).

The pCO<sub>2</sub> values were positively correlated to water temperature across the data set merging highland and lowland lakes (Fig. S5). This correlation could be spurious, reflecting the altitudinal gradient in pCO<sub>2</sub> and does not necessarily reflect a temperature dependence of inputs of CO<sub>2</sub> to the lake, such as aquatic respiration within the lake or inputs of CO<sub>2</sub> from the soils of the watershed related to soil respiration. Regarding soil respiration, it is difficult to disentangle



**Fig. 7** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm, relative analytical precision  $\pm 5.2\%$ ), dissolved CH<sub>4</sub> concentration (nmol/L, relative analytical precision  $\pm 10.9\%$ ), and CH<sub>4</sub>:CO<sub>2</sub> ratio ( $\mu\text{mol}:\mu\text{mol}$ ) versus lake surface area (km<sup>2</sup>), lake depth (m), and dissolved organic carbon (DOC in mg/L, relative analytical precision  $\pm 5\%$ ) in the sampled highland lakes in Northern Ecuador (excluding lakes Cuicocha, Loreto, and Papallacta influenced by volcanic inputs) and in lowland floodplain lakes of the Amazon basin gathered from literature (Abril

et al. 2013; Albéric et al. 2018; Amaral et al. 2020; Barbosa et al. 2016; Mitchell et al. 2018). For pCO<sub>2</sub>, only data directly measured by equilibration were compiled, data computed from pH and total alkalinity were excluded as they can be prone to large errors (Abril et al. 2015). Dotted horizontal lines indicate equilibrium with the atmosphere and solid lines show fits to the data (Table S2) (blue solid lines are highland lakes; black solid lines are lowland lakes; grey dotted lines are high- and lowland lakes)

the effects of variations of soil properties and of temperature across altitudinal gradients, since both vary along the altitudinal gradient (Badraghi et al. 2021). Pineda et al. (2021) reported a data set of soil–air emissions of CO<sub>2</sub> in Ecuadorian Andes along an altitudinal gradient (3000–400 m). Soil–air emissions of CO<sub>2</sub> were higher at 3000 m due to higher organic carbon content than at 400 m despite lower temperatures (~ 12 °C versus ~ 22 °C, respectively). Since soil–air CO<sub>2</sub> emissions are a function of soil respiration, the potential CO<sub>2</sub> inputs from soils to lakes could in fact be stronger at higher altitudes than lower altitudes; however, lower CO<sub>2</sub> concentrations were observed in the higher altitude lakes compared with the lakes in the Central Amazon. This would suggest an additional external CO<sub>2</sub> input in lowland lakes compared with highland lakes.

Respiration in lake sediments scales with temperature (Cardoso et al. 2014); however, respiration in the water column of lakes seems more strongly dependent on organic carbon availability than on temperature (Pace and Prairie 2005). This is in line with a lack of relation between aquatic respiration and temperature in rivers in tropical settings (for a temperature gradient between ~ 12 °C and ~ 27 °C) (Marzolf and Ardón 2021), as well as across climate zones (Bernhardt et al. 2022). So, other factors than temperature, such as organic matter availability, seem to govern the variability of aquatic respiration in lakes and rivers.

We hypothesize that the positive relation between pCO<sub>2</sub> and temperature did not result primarily from the temperature dependence of aquatic respiration within the lakes. If the CO<sub>2</sub> was mainly coming from soils, higher CO<sub>2</sub> would have been expected in higher altitude lakes due to the higher soil respiration (Pineda et al. 2021) and, generally, lower lake size of sampled lakes that should have led to a stronger influence from soil inputs (Staehr et al. 2012). Hence, the hypothesis of riverine inputs of CO<sub>2</sub> from the Amazon River in driving higher pCO<sub>2</sub> values in lowland lakes might possibly be more appropriate and in line with the importance of bidirectional flows between rivers and floodplains for the carbon fluxes of Amazonian floodplain lakes (Junk et al. 1989; Abril et al. 2014; Amaral et al. 2018).

The CH<sub>4</sub> values in the sampled highland lakes were higher ( $3564 \pm 7186$  nmol/L) than in the floodplain lakes in the Central Amazon basin ( $170 \pm 215$  nmol/L) compiled from literature (Fig. 7). Yet, CH<sub>4</sub> concentration was correlated with depth, lake surface area, and DOC across the full data set including both highland and lowland lakes. A tighter relation between CH<sub>4</sub> and lake morphology (depth and surface area) would suggest a dominant effect of lake size on the internal production and external inputs from soils of CH<sub>4</sub> despite different settings across the elevational gradient in the whole Amazon basin. Such an effect seems to override the difference in water temperature between highland ( $12.4 \pm 1.9$  °C) and lowland

lakes ( $30.7 \pm 1.4$  °C), despite the fact that methanogenesis shows a strong temperature dependence (Yvon-Durocher et al. 2014). The CH<sub>4</sub> values were negatively correlated to water temperature across the data set merging highland and lowland lakes (Fig. S5). Such a correlation seemed spurious since CH<sub>4</sub> fluxes would be expected to be positively related to temperature (Yvon-Durocher et al. 2014). In a recent meta-analysis at global scale, no relation was found between temperature and CH<sub>4</sub> emissions from lakes (Deemer and Holgerson 2021) and rivers (Rocher-Ros et al. 2023).

The consistency across the data set of the relation between CH<sub>4</sub> and surface area differed from pCO<sub>2</sub>, for which there were distinct relationships with surface area for highlands and lowlands, possibly reflecting local drivers specific to lowlands compared with highlands, in particular the connectivity with the Amazon river channel in the floodplain lakes. Consequently, unlike CH<sub>4</sub>, a simple “unifying” correlation between pCO<sub>2</sub> and other variables (DOC, lake size) did not emerge for both highland and lowland lakes from our data set. Since CH<sub>4</sub> was higher and CO<sub>2</sub> lower in highland lakes compared with lowland lakes, the CH<sub>4</sub>:CO<sub>2</sub> ratio was markedly higher and also correlated with lake surface area, depth and DOC (Fig. 7).

The pCO<sub>2</sub> was not correlated with DOC across the whole basin suggesting that pCO<sub>2</sub>-DOC relationship was regionally specific, as also suggested in boreal landscapes (Roehm et al. 2009; Lapierre and del Giorgio 2012). This would suggest that the relation of pCO<sub>2</sub> and DOC results from multiple processes and do not provide consistent and reproducible patterns across large regions as noted in boreal landscapes. This is possibly related to regional patterns in DOC composition and/or in landscape morphology and lake position in the landscape that dictate the amount and relative proportions of DOC and CO<sub>2</sub> exports from soils to lakes (Roehm et al. 2009; Lapierre and del Giorgio 2012). This is also in line with a recent analysis of pCO<sub>2</sub> in African lakes showing diverging pCO<sub>2</sub>-DOC relations in different type of lakes, in particular in humic and nonhumic lakes (Borges et al. 2022). In our particular case, the difference between highland and lowland pCO<sub>2</sub>-DOC patterns is most likely related to the strong influence of riverine inputs in lowland floodplain lakes. It is unclear if the correlation between CH<sub>4</sub> and DOC is spurious or reflects a causal relation. This relation could be spurious because shallower lakes tend to have higher CH<sub>4</sub> concentrations due to the lower effect of the removal of CH<sub>4</sub> by methane oxidation and stronger effect of CH<sub>4</sub> inputs from sediments or soils (Borges et al. 2011; Holgerson and Raymond 2016). Shallower lakes tended to be smaller in our data set, and smaller water bodies tend to have higher DOC concentrations because small lakes have a greater connectivity and potential exchange of organic carbon with the riparian wetlands and surrounding terrestrial

landscape along its periphery relative to lake surface area than larger lakes (Staehr et al. 2012).

### Emissions of CO<sub>2</sub> and CH<sub>4</sub> from lakes and streams of the Amazon basin

We use relationships as a function of lake surface area (Fig. 7) to scale pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration to all the lakes in the Amazon basin ( $n = 12,549$ ) given by the HydroLAKES data set (Messenger et al. 2016). The relationships combined our data in Ecuadorian highland lakes and data compiled from the literature for lowland lakes in the Central Amazon, totaling 34 lakes for CH<sub>4</sub> and 29 lakes for CO<sub>2</sub>. For CH<sub>4</sub>, there was a robust relation with lake surface area across the whole Amazon basin (Fig. 7). For consistency, we also used a relation as a function of lake surface area to scale pCO<sub>2</sub>, but two separate relations were needed for highland and lowland lakes. Such an approach is applicable because the distribution of lake numbers and surface areas given by HydroLAKES follows a bimodal distribution, with the majority located in lowlands (elevation < 500 m) followed by highlands (elevation > 2500 m) with very few lakes located at elevations between 500 and 2500 m (Fig. S6). We upscaled CH<sub>4</sub> diffusive fluxes using surface area as a predictor (Fig. 7), because the range of depth variations was restricted in the highland (average  $8 \pm 9$  m range 1–28 m) and in the lowland (average  $10 \pm 3$  m range 6–15 m) lakes, in line with previous studies in small and shallow water bodies (Holgerson and Raymond 2016). The ebullitive  $F_{CH_4}$  was not measured in the sampled Ecuadorian lakes, and we extracted for lakes of the Amazon basin the values modeled by Borges et al. (2022) in tropical lakes globally. This model relies on a compilation of ebullitive  $F_{CH_4}$  data in tropical lakes extrapolated to the littoral zone (depth < 10 m), for which the surface area was derived from morphometric relations based on maximum depth (Håkan et al. 2007) applied to HydroLAKES (Messenger et al. 2016). The CO<sub>2</sub> and CH<sub>4</sub> diffusive fluxes in the streams and rivers of the Amazon basin were reported by Chiriboga and Borges (2023) and relied on statistical modelling of CO<sub>2</sub> and CH<sub>4</sub> concentrations applied to RiverAtlas (Linke et al. 2019). Data of ebullitive  $F_{CH_4}$  have only been reported in nine large rivers of the Amazon basin (Sawakuchi et al. 2014). We estimated ebullitive  $F_{CH_4}$  for the rivers and streams assuming they equaled the values of diffusive  $F_{CH_4}$ , as reported by a global compilation ( $n = 93$ ) by Rocher-Ros et al. (2023).

The diffusive CH<sub>4</sub> emission ( $0.0007 \pm 0.0003$  TgC/yr) from highland lakes (elevation > 500 m) was 18 times lower than from lowland lakes (elevation < 500 m) ( $0.0130 \pm 0.006$  TgC/yr) (Table 2). This was due to lakes in the lowlands being 11 times more numerous (12,549 versus 1116) and covering a surface area 23 times larger (22,282 versus 952 km<sup>2</sup>) than in the highlands (Fig. S6,

Table 2). The average areal diffusive  $F_{CH_4}$  (per m<sup>2</sup>) in highland lakes ( $0.75 \pm 0.36$  gC/m<sup>2</sup>/yr) was close to the  $F_{CH_4}$  in lowland lakes ( $0.58 \pm 0.28$  gC/m<sup>2</sup>/yr) owing to similar dissolved CH<sub>4</sub> concentrations (265 nmol/L versus 179 nmol/L) but marginally higher  $k$  values ( $3.5 \pm 0.5$  cm/h versus  $2.5 \pm 0.3$  cm/h). The average dissolved CH<sub>4</sub> concentration was similar in highland lakes because of equivalent average surface areas ( $0.85 \pm 5.54$  km<sup>2</sup>) in highland and lowland ( $1.78 \pm 15.54$  km<sup>2</sup>) lakes. The  $k$  values were on average marginally higher in highland lakes because of higher average wind speed ( $3.0 \pm 0.6$  m/s) than lowland lakes ( $1.4 \pm 0.6$  m/s) (Table 2). The ebullitive  $F_{CH_4}$  in lakes ( $0.41 \pm 0.21$  TgC/yr) modeled by Borges et al. (2022) was 31 times higher than the diffusive  $F_{CH_4}$  ( $0.014 \pm 0.006$  TgC/yr), as generally reported in lakes (Deemer and Holgerson 2021; Borges et al. 2022; Johnson et al. 2022). The total  $F_{CH_4}$  (ebullitive and diffusive,  $0.42 \pm 0.21$  TgC/yr) from lakes in the Amazon basin we computed was within the range (0.10–0.53 TgC/yr) modeled with one-dimensional process-based lake biogeochemical model (Guo et al. 2020) calibrated with data from lakes Calado and Janauacá (respectively) (Melack et al. 2022), and similar to the estimate of ~0.5 TgC/yr reported by Malhi et al. (2021) based on the extrapolation of the average fluxes from a single lake (Lake Janauacá, Barbosa et al. 2020).

The areal diffusive  $F_{CH_4}$  was higher in rivers than lakes in both lowlands ( $6.8 \pm 2.1$  versus  $0.6 \pm 0.3$  gC/m<sup>2</sup>/yr) and highlands ( $12.9 \pm 2.9$  versus  $0.8 \pm 0.4$  gC/m<sup>2</sup>/yr) due to much higher  $k$  values (16–149 versus 3–4 cm/h), while average CH<sub>4</sub> was higher in rivers (427 nmol/L) than lakes (179 nmol/L) in lowlands, but the opposite was observed in highlands (157 versus 265 nmol/L). Integrated diffusive  $F_{CH_4}$  was higher in rivers than lakes in both lowlands ( $0.27 \pm 0.08$  versus  $0.01 \pm 0.01$  TgC/yr) and highlands ( $0.03 \pm 0.01$  versus  $0.001 \pm 0.001$  TgC/yr) because the surface area covered by rivers was higher than lakes in lowlands (40,375 versus 22,282 km<sup>2</sup>) and highlands (2338 versus 952 km<sup>2</sup>), in addition to higher areal diffusive  $F_{CH_4}$  values. Ebullitive  $F_{CH_4}$  was equivalent in rivers and lakes in lowlands ( $0.27 \pm 0.08$  versus  $0.41 \pm 0.21$  TgC/yr) and highlands ( $0.03 \pm 0.01$  versus  $0.01 \pm 0.01$  TgC/yr). The total (diffusive and ebullitive)  $F_{CH_4}$  from rivers ( $0.61 \pm 0.12$  TgC/yr) was close to the total (diffusive and ebullitive)  $F_{CH_4}$  from lakes ( $0.43 \pm 0.21$  TgC/yr), at the scale of the whole Amazon basin. The total (diffusive and ebullitive)  $F_{CH_4}$  from lakes alone ( $0.43 \pm 0.21$  TgC/yr) and from lakes plus rivers ( $1.04 \pm 0.24$  TgC/yr) only represented ~1.4% and ~3.3%, respectively, of total CH<sub>4</sub> emissions from all compartments of the Amazon basin ranging between 30 and 34 TgC/yr derived from top-down estimates based on aircraft or satellite data (Wilson et al. 2016, 2021; Pangala et al. 2017), which are mainly attributable to wetlands (Basso et al. 2021).

**Table 2** Emissions to the atmosphere of CO<sub>2</sub> (*F*CO<sub>2</sub>) and CH<sub>4</sub> (*F*CH<sub>4</sub>) from the lakes and rivers of the Amazon basin

	Elevation	
	< 500 m	> 500 m
<i>Lakes</i>		
Diffusive <i>F</i> CH <sub>4</sub> (TgC/yr)	0.0130 ± 0.006	0.0007 ± 0.0003
Diffusive <i>F</i> CH <sub>4</sub> (gC/m <sup>2</sup> /yr)	0.58 ± 0.28	0.75 ± 0.36
Ebullitive <i>F</i> CH <sub>4</sub> (TgC/yr)	0.41 ± 0.21	0.009 ± 0.005
Ebullitive <i>F</i> CH <sub>4</sub> (gC/m <sup>2</sup> /yr) *	18.5 ± 9.2*	9.5 ± 4.7*
Surface area weighted average CH <sub>4</sub> (nmol/L)	179	265
Diffusive <i>F</i> CO <sub>2</sub> (TgC/yr)	5.7 ± 1.9	0.022 ± 0.015
Diffusive <i>F</i> CO <sub>2</sub> (gC/m <sup>2</sup> /yr)	255 ± 86	23 ± 16
Surface area weighted average pCO <sub>2</sub> (ppm)	2909	530
Wind speed (m/s)	1.4 ± 0.6	3.0 ± 0.6
<i>k</i> <sub>600</sub> (cm/h)	2.5 ± 0.3	3.5 ± 0.5
Total lake surface area (km <sup>2</sup> )	22,282	952
Number of lakes	12,549	1116
Average lake surface area (km <sup>2</sup> )	1.78 ± 15.54	0.85 ± 5.54
<i>Rivers</i>		
Diffusive <i>F</i> CH <sub>4</sub> (TgC/yr)	0.273 ± 0.084	0.030 ± 0.007
Diffusive <i>F</i> CH <sub>4</sub> (gC/m <sup>2</sup> /yr)	6.77 ± 2.08	12.93 ± 2.88
Ebullitive <i>F</i> CH <sub>4</sub> (TgC/yr) **	0.273**	0.030**
Ebullitive <i>F</i> CH <sub>4</sub> (gC/m <sup>2</sup> /yr) **	6.77**	12.93**
Average CH <sub>4</sub> (nmol/L)	427	157
Diffusive <i>F</i> CO <sub>2</sub> (TgC/yr)	58.9 ± 4.1	5.5 ± 1.0
Diffusive <i>F</i> CO <sub>2</sub> (gC/m <sup>2</sup> /yr)	1459 ± 103	2364 ± 418
Average pCO <sub>2</sub> (ppm)	2702	1033
<i>k</i> <sub>600</sub> (cm/h)	15.8 ± 0.4	148.8 ± 3.2
Total stream surface area (km <sup>2</sup> )	40,375	2338

The diffusive *F*CO<sub>2</sub> and *F*CH<sub>4</sub> in lakes was modeled from relations of CO<sub>2</sub> and CH<sub>4</sub> dissolved concentration as a function of lake surface area based on our data in Ecuadorian highland lakes and data from literature for lowland floodplain lakes (Fig. 7), using the HydroLAKES spatial data set (Messenger et al. 2016) and the gas transfer velocity (*k*<sub>600</sub>) derived from a parameterization as a function of wind speed (Cole and Caraco 2018) and wind speed from the WorldClim spatial data set (Fick and Hijmans 2017). The ebullitive *F*CH<sub>4</sub> in lakes was modeled by Borges et al. (2022) from a compilation of data in tropical lakes extrapolated to the littoral zone (depth < 10 m) for which the surface area was derived from morphometric relations based on maximum depth (Håkan et al. 2007) applied to HydroLAKES (Messenger et al. 2016). The diffusive *F*CO<sub>2</sub> and *F*CH<sub>4</sub> in rivers was modeled by Chiriboga and Borges (2023) from CO<sub>2</sub> and CH<sub>4</sub> dissolved concentration as a function of catchment slope, *k*<sub>600</sub> derived from a parameterization as a function of stream slope and flow (Raymond et al. 2012) and the RiverATLAS spatial data set (Linke et al. 2019). The ebullitive *F*CH<sub>4</sub> in rivers was assumed identical to the diffusive *F*CH<sub>4</sub> as reported by a global compilation (*n* = 93) by Rocher-Ros et al. (2023)

\*The areal flux was rescaled for the total lake surface rather than the surface area of the littoral zone (depth < 10 m)

\*\*For rivers, ebullitive *F*CH<sub>4</sub> was assumed equal to diffusive *F*CH<sub>4</sub> based on Rocher-Ros et al. (2022)

The CO<sub>2</sub> emission (0.022 ± 0.015 TgC/yr) from highland lakes was 257 times lower than from lowland lakes (5.7 ± 1.9 TgC/yr) (Table 2). The average areal *F*CO<sub>2</sub> from lakes was also 11 times lower in highlands (23 ± 16 gC/m<sup>2</sup>/yr) than in lowlands (255 ± 86 gC/m<sup>2</sup>/yr) owing to the lower pCO<sub>2</sub> values (530 ppm versus 2909 ppm) despite higher *k* values (3.5 ± 0.5 cm/h versus 2.5 ± 0.3 cm/h). The CO<sub>2</sub> emission from lakes from the Amazon basin we computed (5.7 ± 1.9 TgC/yr) was 4.1 times lower than the one reported

by Raymond et al. (2013) (23.3 TgC/yr). This difference is related to a lower lake surface area (23,234 versus 51,629 km<sup>2</sup>) and a lower areal *F*CO<sub>2</sub> (245 versus 451 gC/m<sup>2</sup>/yr) that is due to lower *k* values (2.5 versus 5.7 cm/h). Our averaged pCO<sub>2</sub> value was in fact marginally higher than the one used by Raymond et al. (2013) (2811 versus 1906 ppm). The lake surface area used by Raymond et al. (2013) was derived from Downing et al. (2006), which includes lakes with a surface area down to 0.001 km<sup>2</sup>, while HydroLAKES only

reports lakes with a surface area  $\geq 0.1 \text{ km}^2$ . The number of lakes with a surface area  $< 0.1 \text{ km}^2$  reported by Downing et al. (2006) was computed assuming a Pareto distribution which has been questioned on empirical (McDonald et al. 2012) and theoretical (Seekell and Pace 2011) grounds. The HydroLAKES data set has recently been recommended and applied in global scaling of lacustrine GHG emissions (Johnson et al. 2022; Lauerwald et al. 2023). Raymond et al. (2013) used a constant  $k$  value for several lake size classes calculated from the compilation of  $k$  values derived from tracer experiments in temperate or boreal systems. We computed the  $k$  values for each lake from local wind speed derived from WorldClim (Fick and Hijmans 2017) using the  $k$ -wind parameterization of Cole and Caraco (2018). Wind speed is zonally variable as a function of latitude, and tropical landmasses show distinctly lower values than higher latitude areas, which explained the difference in average  $k$  for lakes in the Amazon basin we computed and those used by Raymond et al. (2013). Raymond et al. (2013) used an average  $p\text{CO}_2$  of 1906 ppm for all tropical lakes based on data set of Sobek et al. (2005) that compiles mostly  $p\text{CO}_2$  data computed from pH and TA that are usually overestimated (Abril et al. 2015). The average  $p\text{CO}_2$  value proposed by Raymond et al. (2013) for tropical lakes (1906 ppm) was higher than the value we scaled for highland lakes (on average 530 ppm) but lower for lowland lakes (on average 2909 ppm).

The relative contribution of ebullitive and diffusive  $F\text{CH}_4$  to total carbon emissions ( $\text{CO}_2 + \text{CH}_4$ ) was 31% in highland lakes and 7% in lowland lakes (Table 2). The relative contribution of  $F\text{CH}_4$  to total carbon emissions ( $\text{CO}_2 + \text{CH}_4$ ) for the whole Amazon basin (both high- and lowlands) was low in lakes (7%) and rivers (1%).

## Conclusions

Global estimates of  $\text{CO}_2$ ,  $\text{CH}_4$ , and  $\text{N}_2\text{O}$  emissions from lakes and rivers are uncertain because available data sets do not capture the inherent large variability and are insufficient to derive robust statistical functions to scale fluxes with confidence, despite the recent emergence of detailed spatial data sets of location and surface area for rivers (Allen and Pavelsky 2018; Linke et al. 2019) and lakes (Messenger et al. 2016) that can be used to scale fluxes. Tropical lakes are undersampled with regards to GHG emissions compared with higher latitude systems, and this is particularly true for mountainous lakes. The  $\text{CH}_4$  and  $p\text{CO}_2$  values we report in mountainous lakes of the Ecuadorian Andes were very different from those reported in literature for the floodplain lakes of Central Amazon. This seemed to have resulted from smaller size of highland lakes that were surrounded by the páramo organic matter rich soils, which might have possibly led to higher  $\text{CH}_4$  levels in highland lakes. The higher

$p\text{CO}_2$  values in lowland floodplain lakes probably resulted from inputs of  $\text{CO}_2$  from the Amazon River, although we do not exclude other processes such as a higher respiration, in particular in lake sediments, related to higher temperatures.

We modeled the  $\text{CH}_4$  and  $\text{CO}_2$  emissions from lakes at the scale of the whole basin, using simple statistical relations based on lake surface area. Such a relation with surface area is not fortuitous but is based on well-established links between lake size and external subsidies of carbon from the surrounding landscape (Staehr et al. 2012). Given the restricted range of depth variations in the data set, surface area seemed to be a better predictor than depth to scale the  $\text{CO}_2$  and  $\text{CH}_4$  data. The resulting estimated  $\text{CH}_4$  and  $\text{CO}_2$  emissions from lakes had an inherent uncertainty, which was quantified, but should represent an improvement compared with the previous estimate of  $\text{CH}_4$  and  $\text{CO}_2$  emissions from lakes in the Amazon basin that relied on the extrapolation of the value from a single lake for  $\text{CH}_4$  (Malhi et al. 2021) or on the extrapolation of an average value of  $p\text{CO}_2$  from tropical lakes in general (Raymond et al. 2012). The uncertainty on the estimated  $\text{CH}_4$  and  $\text{CO}_2$  emissions from lakes resulted in part from the “one station in the center” snapshot survey approach, although frequently used to investigate broad geographical gradients as a function, for instance, of climate or catchment land cover (Juutinen et al. 2009; Finlay et al. 2010; Lapierre and del Giorgio 2012; Kortelainen et al. 2020; Casas-Ruiz et al. 2021; Rodríguez-Cardona et al. 2023). Additionally, the ebullitive  $\text{CH}_4$  emissions were not measured in the Ecuadorian sampled lakes but derived from a statistical model for tropical lakes (Borges et al. 2022). The uncertainty on the estimation of  $\text{CH}_4$  and  $\text{CO}_2$  emissions from lakes of the Amazon basin could be improved in future by increasing the number of sampled lakes and better accounting for within lake spatial and temporal variability by increasing the number of samples horizontally and in time. The uncertainty on the estimation of  $\text{CH}_4$  emissions from lakes of the Amazon basin could be further improved by additional direct measurements of ebullitive  $\text{CH}_4$  emissions.

The integrated basin-wide diffusive  $\text{CO}_2$  emissions from lakes were much lower than those from the river network. The integrated basin-wide diffusive and ebullitive lake  $\text{CH}_4$  emissions were roughly equivalent to those from the river network. The total diffusive and ebullitive  $\text{CH}_4$  emissions from lakes and rivers corresponded to a very small fraction ( $< 2\%$ ) of the total  $\text{CH}_4$  emissions from all compartments of the Amazon basin, derived from top-down estimates, which are mainly attributable to wetlands.

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**Author contributions** A.V.B. designed the research and analyzed the data. G.C. collected the samples. G.C. measured the GHGs, TA, and nutrients, and S.B. measured DOC and POC concentrations, H<sub>2</sub>O, DIC isotopes, and major elements. A.V.B. drafted the manuscript, with contributions from G.C. and S.B.

**Data availability** The dataset generated during the current study is publicly available from <https://doi.org/https://doi.org/10.5281/zenodo.7867344>.

## Declarations

**Conflict of interest** The authors declare that they have no conflict of interest.

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