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Phytochemical characterisation and aromatic potential for brewing of wild hops (*Humulus lupulus* L.) from Northern France: towards a lead for local hop varieties

Anne-Sophie Paguet¹, Ali Siah¹, Gabriel Lefèvre¹, Mathilde Vandenberghe¹, David Lutun², Norman Degardin², Jennifer Samaillie¹, David Mathiron³, Charles Dermont¹, Franck Michels¹, Marie-Laure Fauconnier¹, Sylvie Chollet¹, Roland Molinié¹, Jean-Xavier Fontaine¹, Sevser Sahpaz¹, Céline Rivière^{1*}

¹ Joint Research Unit 1158 BioEcoAgro, University of Lille, Junia-ISA, University of Liège, University of Picardy Jules Verne, University of Artois, ULCO, INRAE, Villeneuve d'Ascq, France

² High School Biotech Douai - Campus Wagnonville, 458 Rue de la Motte Julien, 59500 Douai, France

³ University of Picardie Jules Verne (UPJV), Analytical Platform, Rue Dallery – Passage du sourire d'Avril, 80039 Amiens cedex, France

* Corresponding author: celine.riviere@univ-lille.fr

Abstract

In the current context of developing aromatic beers, our study aims at deciphering the chemical characterisation of cones from 39 wild hop genotypes collected in the North of France and replanted in an experimental hop farm, as well as 10 commercial and 3 heirloom varieties, using HS-SPME/GC-MS for the volatile compounds, UHPLC-UV for phenolic compound quantification, and UHPLC-IMS-HRMS for untargeted metabolomics. These analyses revealed a strong opposition between wild accessions and reference varieties, and an original chemical composition of some genotypes. 27 beers were produced with the same recipe, analysed by SBSE-GC-MS and evaluated by panellists. The unique difference relates to the hops to be assessed in order to determine their sensory profile. The different datasets were compared by OPLS-DA analysis in order to identify chemical markers which may influence the hop aromatic potential. Our results highlight the aromatic potential of some wild accessions, close to the commercial variety Cascade.

Keywords: *Humulus lupulus* L.; wild hops; chemodiversity; sensory analysis; mass spectrometry; multivariate analysis

30 1. Introduction

31 Beer is a fermented drink, which has traditionally been flavoured with hop (*Humulus lupulus* L.,
32 Cannabaceae) since the Middle Ages. Female inflorescences, commonly referred as hops or hop cones,
33 are added to the wort for their bitterness and their aromatic originality as well as for their antiseptic
34 properties, linked to their original chemical composition (De Keukeleire et al., 1992). Original phenolic
35 compounds, including prenylated chalcones as xanthohumol and desmethylxanthohumol and
36 acylphloroglucinol derivatives (α -acids or humulone derivatives and β -acids or lupulone derivatives),
37 are produced in lupulin glands at the base of the bracts. Hop essential oil is also rich in non-oxygenated
38 monoterpenes and sesquiterpenes, among them β -myrcene, α -humulene and β -caryophyllene
39 (Bocquet et al., 2018). When bittering hops are added to the wort during the boiling, α -acids are
40 isomerised into iso- α -acids, in *cis* or *trans* position (Schönberger & Kostelecky, 2011) and provide up
41 to 80% of a beer's bitterness (De Keukeleire et al., 1992). By contrast, aromatic hops are rather added
42 to the wort at the end or after its boiling, to minimise the evaporation of their volatile compounds
43 (Sharpe & Laws, 1981). Monoterpenes and sesquiterpenes are considered to be responsible for the
44 "hoppy" aroma of beer (Van Opstaele et al., 2010). α -humulene and β -caryophyllene can impart spicy
45 and woody notes to beer in particular due to the formation of oxygenated derivatives during wort
46 boiling. β -myrcene, an important contributor to fresh hop aroma, is largely lost during beer processing
47 except in dry hopping, but oxygenated derivatives formed can contribute to the fragrance. Among
48 other terpenoids, alcohol monoterpenes (linalool, geraniol, β -citronellol, α -terpineol) and aldehyde
49 monoterpenes (citral) provide floral and citrus notes to beer, respectively (Rettberg et al., 2018). By
50 contrast, other fruity aroma can be attributed to the presence of some esters, such as red berries-like
51 for ethyl-3-methylbutanoate, whereas aldehydes, such as hexanal, confer grassy fragrances to the beer
52 (Machado et al., 2021). Perception of hop volatile compounds after brewing remains complex.
53 Interactions between hop volatile compounds and other beer components (ethanol, carbohydrates,
54 yeast, hop bitter acids...), combined to oxidation phenomena, require a thorough investigation to
55 understand their resulting effects on the final sensory profile of beer (Dietz et al., 2020). Hops also play
56 a role in the characteristics and stabilisation of the foam (De Keukeleire et al., 1992). The hydroxyl
57 group of the isohumulone acts as a surfactant, which helps to strengthen the electrostatic bonds
58 between the bubbles and thus stabilises the foam (Asano & Hashimoto, 1980).

59 Northern France is an historical region for beer craft and hop production. In the beginning of the
60 XXth century, 1220 ha of hops and nearly 2,000 breweries were inventoried in the region (Ducloux et
61 al., 1910). If beer production remains important (around 6 Bn hL in 2020), hop production for its part
62 clearly decreased during the XXth century, to reach only 35 ha today (Bart-Haas Group, 2021). This
63 decline can be explained by the economic context of the last century, when French brewers started to
64 import hops from the USA or Germany. However, in the context of the "beer craft movement", current
65 brewers have been looking for local and aromatic hops from sustainable or organic agriculture for a
66 decade (Paguet et al., 2022). In this context, the objective of our study is to investigate the diversity of
67 wild hops from Northern France, that can be used further to support a future varietal development
68 (Paguet et al., 2022). In a previous study, we investigated the genetic and the chemical diversity of 50
69 wild hop accessions collected *in-situ* from eleven natural sites of the North of France, in the Hauts-de-
70 France region (Paguet et al., 2023). These wild accessions were compared to ten commercial varieties
71 and three old varieties coming from the same region for their genetic and chemical characteristics.
72 This study underlined a high genetic diversity and chemical variability among wild accessions (Paguet
73 et al., 2023). During the collection of samples, rhizomes were also collected and transplanted into our
74 experimental field to get a germplasm resource. The goal of the current study was to assess these
75 accessions in *ex-situ* conditions and to analyse their chemical diversity under more standardised
76 conditions. The more productive accessions were used for the production of beers using the same beer
77 style (lager style). The only difference in the recipe was the aromatic hops, with wild or commercial
78 origin. The beers were subjected to a sensory study in order to highlight the aromatic traits of these
79 hops after brewing. Recently, Machado et al., (2021) have studied the aromatic qualities of wild hops

80 on dry-hopped beers. Hong et al. (2022) evaluated, using partial least-squares discriminant analysis,
81 the link between chemical characteristics of cones and their influence in hop-tea or after brewing.

82 While our previous article Paguet et al. (2023) concerned wild hop accessions collected under *in-situ*
83 conditions, the present study deals with these same accessions replanted in our experimental hop field,
84 and therefore constitute a new dataset. The chemical composition of wild hop accessions, grown under
85 *ex-situ* conditions, and their aromatic potential were determined using combined analytical methods,
86 including HS-SPME GC-MS, UHPLC-UV and UHPLC-IMS-HRMS. The physico-chemical characterisation of
87 the beers was carried out using analytical methods such as SBSE GC-MS, while a panel of experts was
88 used to characterise the beers according to a certain range of descriptors. Moreover, sensory evaluation
89 and multiple statistical methods (heatmap, principal component analysis (PCA), orthogonal partial least
90 squares discriminant analysis OPLS-DA) were carried out to explore the relationship between wild hops'
91 chemical composition and beer aroma.

92 2. Material and methods

93 2.1. Samples

94 2.1.1. Hop collection

95 50 wild hops coming from 11 different locations, with ecological or ethnobotanical interests, were
96 harvested in September 2019, in accordance with the rules of the Nagoya Protocol and the French
97 biodiversity law of 2017 (decision of June 9, 2020 issued by the Ministry of Ecological and Inclusive
98 Transition; NOR: TREL2002508 S/284) (Paguet et al., 2023). The rhizomes of these accessions had also
99 been collected and transplanted in November 2019 into our experimental hop field (Ferme brasserie
100 Wagnonville, Douai, France). The present study focuses on the characterisation of the 32 most
101 productive accessions in cones to enable their analysis among the 50 that had been replanted. Cones
102 were collected at their maturity (80% humidity) in September 2021, oven-dried at a temperature below
103 40°C until they reached a moisture content of 10% and then stored under vacuum at -20°C before being
104 used for analysis. These wild accessions cultivated in field conditions were compared to 10 commercial
105 cultivars grown in the North of France region. They were selected among the ten most relevant varieties,
106 including so-called top hops used by craft brewers, such as Brewers Gold, Cascade, Challenger, Fuggles,
107 Goldings, Magnum, Northern Brewer, Nugget, Strisselspalt and Target. Cones of commercial hops,
108 collected in September 2021, were provided by the Northern France Hop Cooperative, Coopounord
109 (Bailleul, France). They were also compared to three heirloom varieties collected the same year. Three
110 genbank clones representing former Flander cultivars were included: Groene Bel, Coigneau, and Record,
111 coming from Belgium (Hoppecruyt, Poperinge, Belgium). Dried hop cones, kept vacuum-packed in cold
112 storage, were grounded in liquid nitrogen using a blender IKA A11 (Staufen, Germany) before being used
113 for extraction.

114 2.1.2. Beer samples: Brewing processes

115 To investigate the aromatic potential of these wild hop accessions after brewing, the 17 more
116 productive hops of our collection were brewed, according to a recipe of blond Pilsner, for chemical
117 characterisation and sensory evaluation. These beers were compared to beers elaborated with 2 old
118 varieties (Groene Bel and Record) and seven commercial varieties (Nugget, Strisselspalt, Cascade,
119 Magnum, Brewers Gold, Target and Northern Brewers). The mash was brewed in 60 L in mash-in with
120 13 kg of Pilsen malt (La Malterie du Château, Beloeil, Belgium) at 67°C for 1 hour and 15 minutes and
121 filtered. Afterwards, the wort was boiled and 30 grams of bittering hops cv. Magnum in pellets
122 (Coopounord, Bailleul, France) were added 10 minutes after the beginning of boiling. After boiling,
123 the wort was divided into 6 brew tests of 10 L each for aroma hopping. 50 grams of dried hops to
124 evaluate, the only ingredient differentiating the brews, were added. A beer control without aromatic

125 hop was also performed. Aromatic hops were added after the boiling step in infusion during the
126 intonation for 20 min (steeping/whirlpool hops mode). After cooling, fermentation took place at
127 atmospheric pressure at 10°C (low fermentation) for one week, with *Saccharomyces cerevisiae* LalBrew
128 Nottingham yeasts (Lallemand Brewing, Felixstowe, United Kingdom). Lagering was carried out at 4°C
129 for 10 days. The beers were then racked in bottles, without filtration or pasteurisation.

130 2.2. Chemical analysis

131 2.2.1. Volatile compound analysis

132 2.2.1.1. Volatile compound analysis in hops by HS-SPME GC-MS

133 Extraction of volatile compounds by HS-SPME was reached using a triple-phase fiber, 50/30µm
134 DVB/CAR/PDMS (Supelco, Bellefonte, PA, USA), that was preconditioned according to the instructions
135 of the manufacturer. Extraction was done on 2 grams of hop cone powder placed in a 20 mL vial
136 incubated at 45°C for 5 min. We used the same chromatographic conditions as those detailed in Paguet
137 et al., 2023, on an Agilent 7890 A gas chromatograph coupled to an Agilent 5975C mass spectrometer
138 (Agilent Technologies, Santa Clara, CA, USA) equipped with an MPS auto-sampler and an HP-5 MS
139 capillary column (30 m × 250 µm × 0.25 µm, Agilent Technologies, Santa Clara, CA, USA). Injections
140 were performed in splitless mode at 280 °C and helium was employed as the carrier gas at a flow rate
141 of 1.2 mL.min⁻¹. The separation conditions were as follows: initial column temperature of 40°C for 2
142 min; then the temperature was successively increased by 4°C/min up to 200°C then by 20 °C/min up
143 to 300°C, where it was maintained for 5 min. The mass spectrometer was set to have a temperature
144 of the ion source at 230°C and was programmed with SCAN acquisition mode.

145 2.2.1.2. Beer analysis by SBSE GC-MS.

146 Ten milliliters of beer samples were taken and placed in a vial with 2 grams of NaCl (VWR,
147 France) adding a stir bar (10mm x 0.5mm) with PDMS coating (SBSE Gerstel-Twister, USA) and stirring
148 for 120 min at 1000 rpm. Some samples were analysed twice on two different bottles, to ensure the
149 inter-bottle repeatability. After extraction, the stir bar was rinsed with some distilled water, dipped on
150 a filter paper and introduced in a glass thermal desorption tube (4 mm i.d. x 178 mm L). The stir bar
151 was then placed in the thermal desorption unit of an Agilent 7890 GC coupled to an Agilent 5975 C
152 mass spectrometer (Agilent Technologies, Santa Clara, CA, USA) equipped with a MPS auto-sampler
153 and a HP-5 MS capillary column (30 m x 0.25 mm x 0.25 µm, Agilent Technologies, Santa Clara, CA,
154 USA). Helium was used as the carrier gas at a flow rate of 1.2 mL/min. The oven was programmed as
155 follows: initial temperature of column of 40°C with a hold of 2 minutes; then increased by 6°C/min to
156 300°C, where it was maintained for 5 min. Stir bars were thermally desorbed by programming the
157 system from 40°C to 260°C with a rate of 6°C/min and held at this temperature for 5 min. The desorbed
158 analytes were cryofocused at -10°C. Injection was performed in the splitless mode. These analyses are
159 intended to verify intra-batch homogeneity and inter-batch variations.

160 2.2.1.3. Volatile compound identification

161 MassHunter Version B.06.00 (Agilent Technologies, Santa Clara, CA, USA) was used for data
162 acquisition and processing. General volatile compound profiles were established through a
163 chromatographic deconvolution process (Agilent MassHunter Unknowns Analysis) and
164 chromatographic areas were obtained for each volatile compound. Identification of the individual
165 components was based on: (a) comparison of the mass spectrum (MS) outcomes to those of
166 commercial databases: National Institute of Standards and Technology (NIST17) and Wiley7 (match
167 factor threshold > 700); (b) comparison of the retention index (RI) of each peak with literature RI data
168 (± 20) from the NIST WebBook. Experimental retention index (RI) of the compounds were calculated
169 following the injection of a mixture of n-alkanes C8-C20 (Sigma Aldrich, Darmstadt, Germany). They

170 were first reported as a percentage of the total chromatographic area on Excel (Microsoft Excel 2016)
171 to allow a general analysis.

172 2.2.2. Analysis of non-volatile compounds

173 2.2.2.1. Sample preparation

174 For each hop studied in our study, hydro-ethanolic extracts were performed on 50 mg of dried
175 cone powder with 1 mL of ethanol-water mixture (9:1, v/v). A one-hour maceration session was carried
176 out in an ultrasonic bath. Afterwards, tubes were centrifuged at 4000 rpm at 20°C for 5 min.
177 Supernatant was then transferred in a tared tube. The exhausted matrix was then re-extracted two
178 times more following the same protocol and each time the supernatant was pooled with the first
179 extract. At the end of the three maceration sessions, extracts were then evaporated in a centrifugal
180 concentrator Genevac™. Extracts were diluted at 1 mg/mL in methanol (Carlo Erba), filtered through
181 0.22 µm PTFE membrane and placed in glass vials for further LC-MS analysis. Quality control (QC)
182 samples were prepared by pooling 10 µL from each sample of the same round preparation and
183 thoroughly mixed.

184 2.2.2.2. Setting up and validation of the quantification method of main phenolic compounds

185 Quantification of xanthohumol, *co*-, *n*- and *ad*-humulone, *co*-, *n*- and *ad*-lupulone in each
186 hydro-ethanolic extract was performed on an Acquity UPLC® H-Class Waters® system (Waters,
187 Guyancourt, France) coupled with a Diode Array Detector (DAD) and a QDa ESI-Quadrupole Mass
188 Spectrometer. Separation was achieved using a Waters® Acquity BEH C18 column (pore size 300 Å,
189 particle size 1.7 µm, 2.1 x 50 mm, Waters, Milford MA) connected to a 0.2 µm in-line filter. Solvent A
190 (water with 0.1% formic acid, v/v) and solvent B (acetonitrile with 0.1% formic acid, v/v) were used as
191 mobile phases. Compounds were eluted using the following chromatographic conditions: the flow rate
192 was 0.3 mL/min; the column temperature was set at 30°C; the injection volume was 2 µL. The gradient
193 elution was performed using eluent A and eluent B: initial condition at 50% B, 0-1 min isocratic step,
194 1-3 min linear gradient to 75% B, 3-5 min isocratic step at 75% B, 5-7 min linear gradient to 100% B, 7-
195 9 min isocratic step at 100% B, 9-9.5 min linear gradient to 50% B, 9.5–13 min isocratic step at 50% B
196 (total analysis time: 13 min). The ionisation was performed in negative mode. Cone voltage was set at
197 10 V. Probe temperature was 600 °C. Capillary voltage was 0.8 kV. The MS-Scan mode was used from
198 100 to 1000 Da. Xanthohumol and acylphloroglucinol derivatives were quantified according to the
199 International Conference for Harmonisation (ICH) of Technical Requirements for Pharmaceuticals for
200 Human Use guideline Q2-R1 (ICH, 2005). Quantification was performed in UV at 370 nm for
201 xanthohumol and 330 nm for acylphloroglucinols. The quantitation method was set up using standards
202 purified in the laboratory according to the protocol detailed by Bocquet et al. (2019) and following the
203 methodology detailed in Paguet et al. (2023). *Ad*-humulone and *ad*-lupulone were also quantified by
204 the establishment of calibration range using standards purified in the same way as the one detailed in
205 Bocquet et al. (2019) for *n*- and *ad*- humulone and lupulone. The quantitation of each compound was
206 performed in five technical replicates. QC samples were analysed regularly along with unknown
207 samples. To reduce the effects of systematic errors, all samples were assigned to a random LC–MS run
208 order and interspersed after every 20 injections with QC sample injections. Xanthohumol and α - and
209 β -acids were identified based on the retention time of purified standards and their mass spectra and
210 quantified using the quantification methods of previously set up on the Empower 3 software. Data
211 were exported from Empower 3 to Excel (Microsoft Excel 2016).

212 2.2.2.3. UHPLC-IMS-HRMS analysis

213 UHPLC-IMS-HRMS analysis were performed using a Waters ACQUITY UPLC I-Class system
214 interfaced with a Vion IMS Q-TOF (Ion Mobility Quadrupole Time-of-flight) hybrid mass spectrometer,

215 equipped with an electrospray ionisation (ESI) source (Waters, Manchester, UK). The autosampler was
216 programmed to inject 2 μ L of each sample. Chromatographic separation of the analytes was carried
217 out using the same column, a Waters® Acquity BEH C18 column (2.1 \times 50 mm, 1.7 μ m, Waters, Milford
218 MA) connected to a 0.2 μ m in-line filter maintained at 40°C. The UHPLC method was carried out
219 following the same protocol and procedure described previously for the quantification: the flow rate
220 was 0.3 mL/min; the column temperature was set at 30°C; solvent A (water with 0.1% formic acid, v/v)
221 and solvent B (acetonitrile with 0.1% formic acid, v/v); the gradient was the following: initial conditions
222 at 50% B, 0-1 min isocratic mode, 1-3 min linear gradient to 75% B, 3-5 min isocratic mode at 75% B,
223 5-7 min linear gradient to 100% B, 7-9 min isocratic mode at 100% B, before re-equilibration time of 4
224 min. For HRMS, the ESI parameters were set as follows: capillary voltage, 2.4 kV in negative mode;
225 source temperature, 120°C; desolvation temperature of 450°C. Time-of-flight (TOF) MS was operated
226 in sensitive mode. The data were acquired in high-definition MS^E (HDMS^E) over a mass range of m/z
227 50-1200 at a mass resolving power of 50,000 FWHM and a scan time of 0.2 s. Spectra were acquired
228 and processed with the UNIFI software (version 1.9.4, Waters), allowing to generate the data matrix
229 with default parameters comprising the retention time, mass-to-charge ratio (m/z) values, and peak
230 intensity. The untargeted metabolomic data matrix was cleaned by removing the variables with a
231 significant variance (>35%) in the quality control (QC) and the variables present in the blank. A
232 minimum intensity threshold was chosen at 500 to keep the variable. Metabolites were principally
233 identified by matching the accurate masses, retention times and fragmentation patterns with those of
234 the reference standards and literature references. Supplementary analyses in DDA mode were done,
235 with a collision energy of 20 eV on QC to perform dereplication analysis on UNIFI and MassLynx
236 (version 4.1, Waters) softwares.

237 2.3. Beer characterisation

238 2.3.1. Physico-chemical characterisation of beers.

239 Beers were characterised according to different physico-chemical parameters following the
240 American Society of Brewing Chemists recommendations (1992). Samples were preliminarily degassed
241 by agitation in the open air.

- 242 (i) *Beer colour (EBC)*. Beer colour was determined by an official method from ASBC, in which the
243 absorbance of beer samples was measured at 430 nm using a spectrophotometer UV-1280
244 (Shimadzu, France). EBC was achieved following equations: $EBC = 50 * D * Abs_{430nm}$, where D is the
245 dilution factor, and Abs_{430nm} the absorbance at 430 nm.
- 246 (ii) *Bitterness dosage (IBU)*. Bitterness was determined following Kawa-Rygielska et al. (2019)
247 method: 10 mL of beer sample was added to 1 mL of HCl and 20 mL of isooctane. Tubes were
248 stirred for 15 min and centrifuged at 1880 G for 3 min. The supernatant was measured at 275
249 nm against a blank control (pure isooctane) with a spectrophotometer UV-1280 (Shimadzu,
250 France). IBU values were achieved as follows: $IBU = Abs_{275nm} * 50$ where IBU is the International
251 Bitterness Unit, and Abs_{275nm} is the absorbance at 275 nm.
- 252 (iii) *Determination of total acidity*. Acidity of beer is considered as mainly due to the presence of
253 malic acid, measured by acid-base titration with soda NaCl 0.1 M, using phenolphthalein as
254 indicator of equivalence.
- 255 (iv) *Wort density and alcohol content*. Wort density (expressed in °Plato) and alcohol content (% v/v)
256 were measured with a densimeter Alex 500 (Anton Paar, France) on 50 mL of beer sample
257 vacuum filtered through Kieselguhr. Repeatability of bottles was tested using a Student test.

258 2.3.2. Sensory characterisation of beers

259 2.3.2.1. Panel training

260 Twenty-one trained subjects (13 men and 8 women) were enrolled in a training program
261 designed to produce beer experts. They were trained 1hour a week to describe beers on seventeen
262 descriptors (intensity of the smell, hoppy smell, malty smell, fruity smell, intensity of taste, acid, bitter,
263 floral, hops, citrus, malt, sweet, spicy, red fruits, yellow fruits, astringent and bitter persistence) and to
264 evaluate the intensity of these descriptors on a non-structured linear scale. These descriptors have
265 been chosen in the literature and were recalled by the panelists during vocabulary generation sessions.
266 At the time of the experiment, the panelists had already received 12 hours of training. At this point,
267 they should have developed a consensual vocabulary, along with the ability to detect the descriptors
268 on which they have been trained.

269 2.3.2.2. Procedure

270 The assessors evaluated the 26 beer samples (17 with wild hops, 7 with commercial varieties,
271 2 with old varieties) in duplicate during 12 sessions under standard sensory conditions (ISO, 2016).
272 Samples were presented in transparent glasses and 20 mL were served between 8 and 10°C, in a
273 sequential monadic way. Their presentation order was different for each assessor and based on a
274 Williams' Latin-square arrangement. Subjects had to evaluate beers on the 17 descriptors according to
275 a non-structured linear scale from 0 to 10 using FIZZ software (Biosystemes, Dijon, France). The
276 significant descriptors were identified with three factors ANOVA (product, repetition and panelist),
277 measuring the p -value on the factor product.

278 2.4. Data analysis

279 For each simple dataset collected, data were compiled into an Excel sheet. Statistical analysis as
280 Principal Component Analysis (PCA) for quantitation of prenylated phenolic compounds, untargeted
281 metabolomics and sensory analysis, Agglomerative Hierarchical Clustering (AHC) for composition of
282 volatile compounds and sensory analysis, as well as heatmaps for composition of volatile compounds
283 and quantitation of phenolic compounds were achieved using XLStat (Addinsoft, 2022). On the
284 quantitation triplicate, one-way analysis of variance (ANOVA) followed by Tukey's post-hoc test with p
285 < 0.05 was conducted using XLStat (Addinsoft, 2022). Student tests for the verification of the inter-
286 bottles repeatability and the choice of the selection of the most significant descriptors were achieved
287 using the statistical tests on XLStat (Addinsoft, 2022).

288 To link analytical measurement and sensory analysis, OPLS-DA model was performed using MATLAB
289 R2014a (MathWorks, Natick, MA) software. OPLS-DA model provide insights into separations between
290 experimental groups, here based on sensorial analysis discrimination results, based on analytical
291 measurements (here the phytochemical characterisation of cones). Hence, the OPLS model comprises
292 two blocks of model variations: 1) the Y-predictive block, which represents the variation between the
293 classes, and 2) the Y-orthogonal block also referred to the uncorrelated variation, which constitutes
294 the within class variation (Bylesjö et al., 2007). As we preferred to focus on the link between cone
295 phytochemistry and panel-assessed beer aromas, the SBSE-GC-MS analysis dataset was not included
296 in our orthogonal analysis.

297

298 3. Results

299 3.1. Chemical characterisation of cones

300 3.1.1. Volatile compound analysis by HS-SPME GC-MS

301 The putative identification of volatile compounds by HS-SPME GC-MS revealed the presence
302 of 65 different compounds among hops studied in the study. Thirty-two compounds were detected in

303 fixing a threshold at 1%. This volatile compound analysis was represented by a heatmap and a
304 dendrogram (**Figure 1**). The majority of these compounds were monoterpenes or sesquiterpenes
305 (**Table S1**). In particular, we identified β -caryophyllene, β -myrcene and α -humulene, known as the
306 main volatile compounds of hops, in all accessions characterised (**Figure 1A**). The dendrogram
307 associated with the AHC (**Figure 1B**) distinguished 4 different classes of accessions according to their
308 content of volatile compounds. The blue cluster was mainly composed of commercial and old cultivars,
309 as well as wild accessions I3 and G2. These accessions were characterised by a high content of the main
310 hop volatile compounds, as α -humulene, β -myrcene and β -caryophyllene. In **Figure 1B**, we then noted
311 that the commercial accessions Cascade, Fuggle and Strisselspalt, known for their aromatic
312 characteristics, were distributed in different clusters from other commercial accessions known for
313 their bitter potential. The purple cluster, including Strisselspalt, showed a high level of other volatile
314 compounds, such as limonene, germacrene and β -elemene. The green and purple classes, represented
315 by different accessions, both wild, old and commercial, as well as the red class, only composed of a
316 few wild accessions, showed a lower content of the majority non-oxygenated terpenes (α -humulene,
317 β -myrcene and β -caryophyllene). By contrast, they showed higher contents of minor volatile
318 compounds that were not taken into account in this qualitative classification as γ -muurolene (for B4
319 and H4 in the purple cluster) or α -cedrol (for I8 in the green one), known for their woody aromas (**Table**
320 **S1**). We also identified the presence of some particular sesquiterpenes in both wild and commercial
321 accessions, such as α -bergamotene (in Cascade, Fuggle, H2, H3 and I8) or β -farnesene, α - and β -
322 selinene, or even β - and γ - elemene mostly present in wild accessions (**Figure 1A**).

323 3.1.2. Quantification of main phenolic compounds by UHPLC-UV

324 UHPLC-UV analysis revealed the presence of seven main peaks at the retention times 2.76,
325 4.35, 4.73, 4.83, 6.26, 6.87 and 7.00 minutes attributed to xanthohumol (XN), α -acids (*co*-, *n*-, *ad*-
326 humulone) and β -acids (*co*-, *n*-, *ad*-lupulone) respectively. This attribution was based both on their
327 retention times compared with those of the standards purified in the laboratory and on their mass
328 spectra. Xanthohumol, *co*-humulone, *n*-humulone, *ad*-humulone, *co*-lupulone, *n*-lupulone and *ad*-
329 lupulone were then quantified in 52 crude extracts of cone samples using the quantitation method
330 setting up by UHPLC-UV. Acceptable linearity was observed for each compound over the concentration
331 range used for calibration (**Table S2**). Evaluation of the recovery data of the quantification method
332 showed acceptable intra and inter-day precisions for xanthohumol (RSD % = 18.64, 13.69), *co*-
333 humulone (RSD % = 13.13, 6.40), *n*-humulone (RSD % = 4.01, 2.55), *ad*-humulone (RSD % = 15.55,
334 16.58), *co*-lupulone (RSD % = 10.19, 11.17), *n*-lupulone (RSD % = 16.12, 9.7) and *ad*-lupulone (RSD % =
335 15.15, 16.68). Results were expressed in $\mu\text{g}/\text{mL}$ and converted in percentages of dry matter. Statistical
336 analysis with a Tukey's test revealed significant differences among hop samples in terms of
337 composition in phenolic compounds (**Table S3**). Overall, commercial varieties showed a higher content
338 of *n*-humulone, *ad*-humulone and xanthohumol, which may assert their bitter potential, unlike in wild
339 accessions which produced more β -acids. Nevertheless, the cultivars Strisselspalt and Fuggle were
340 distinguished by low contents of phenolic compounds than other commercial varieties. On the other
341 side, some wild hops showed higher levels than commercial varieties. In the detail, higher contents in
342 xanthohumol, total α -acids and total β -acids were observed for the accessions Target (0.045 %),
343 Magnum (2.748 %) and Groene Bel (3.138 %) respectively. Lower contents were observed for C3 (0.005
344 %), D1 (0.100%) and I5 (0.353 %) respectively (**Figure 2A and Table S3**). **Figure 2A** thus revealed low
345 contents of humulone for wild hops compared to commercial varieties. Some exceptions were noted
346 for the wild hops I3, I1, G2, B3, D3 and J3 that showed contents of humulone similar to commercial or
347 old cultivars (from 1.274 % to 0.621 %). Wild hops also displayed low contents of *ad*-humulone and
348 xanthohumol. By contrast, most wild accessions had levels of *co*-humulone and β -acids which were
349 not negligible. These phenolic compounds will have a different impact on bitterness. The output
350 process of the quantification represented in the form of PCA (**Figures 2B and 2C**) revealed a good
351 explanation of the distribution of variables, equal to 92.32 %, with a contribution of PC1 and PC2 equal
352 to 72.48 % and 19.84 % respectively. Moreover, the biplot clearly revealed a good correlation between

353 α -acid and xanthohumol contents on one side and β -acid contents on the other side (**Figure 2C**).
354 According to this figure, most of the commercial and heirloom varieties had higher contents in *n*-
355 humulone, *ad*-humulone and xanthohumol. On the contrary, wild hops (Locations A to K) and the old
356 variety Groene Bel showed higher β -acid contents (*co*-, *n*- and *ad*-lupulone). The strong separation
357 between commercial cultivars and wild hops was also illustrated by the dendrogram (**Figure S1**). The
358 colouration of the accessions, on the biplot and on the dendrogram, based on their clustering
359 previously achieved with their composition in volatile compounds, allowed us to evaluate the
360 composition both in main phenolic compounds and in volatile compounds. Except for the old variety
361 Groene Bel, this new biplot showed that the accessions were globally grouped according to their
362 colour, and therefore according to their composition in volatile compounds (**Figure 2B**). This study also
363 underlined that the accession I3 had a chemical composition very close to those of commercial
364 varieties because it belonged to the same cluster both for GC-MS analysis and quantification by UHPLC-
365 UV (**Figure 1B and Figure S1**).

366 3.2.2.3. Dereplication and untargeted metabolomics by UHPLC-IMS-HRMS

367 From 164 601 variables acquired, we finally kept 262 variables (160 variables from 1 to 7 min)
368 after cleaning the matrix. From these 160 variables, 15 compounds were identified based on previous
369 works including Farag et al. (2012) and Nicácio et al. (2022) (**Table S4**). These partially annotated
370 metabolites were classified as putative MS/MS annotations and one compound was labeled as an
371 unknown compound (**Figure 3A**). Among them, the 7 compounds previously quantified (xanthohumol,
372 *co*-, *n*- and *ad*-humulone and *co*-, *n*- and *ad*-lupulone) were also identified using PDA spectrum and
373 retention times. As presented in the **Table S5** and on the **Figure 3C**, different classes of metabolites
374 were represented, in particular chalcones and flavanones [desmethylxanthohumol (1), xanthohumol
375 (2), isoxanthohumol (3)], as well as acylphloroglucinol derivatives including α -acids and oxidised
376 derivatives [cohumulone (5), 4-deoxycohumulone (6), humulone (7), humulinone (8), adhumulone (9),
377 deoxyhumulone (10), préhumulone (11)] and β -acids [postlupulone (12), colupulone (13), lupulone
378 (14), adlupulone (15), prelupulone (16)]. **Figures 3B and C** showed the PCA score plots for data acquired
379 in HRMS in negative ion mode corresponding to the UHPLC retention times ranging from 1 to 7 min.
380 Considering the two principal components, we explained 44.78% of the variance distributed as follows:
381 28.31% for the PC1 and 16.47% for the PC2. On **Figure 3B**, individuals were again coloured according
382 to their composition in volatile compounds after the AHC shown on the **Figure 1B**. Thus, we noted that
383 we globally had the same repartition of blue varieties after PCA treatment on the UHPLC-HRMS data;
384 while accessions from other groups were more mixed. The identification of some compounds on **Figure**
385 **3C** can provide some explanations to this distribution. Individuals belonging to the first cluster coloured
386 in blue, which gathered most of the commercial and old varieties (in square), globally showed higher
387 contents in α - and β -acids. **Figure 3C** also highlighted that chemical families, including chalcones, α -
388 and β -acids, were close on the loading plot. These results therefore allowed the identification of some
389 accessions with high contents of α -acids, not only in *co*-, *n*- and *ad*-humulone previously quantified.
390 We have previously identified two accessions, I3 and G2, with a composition in volatile compounds
391 and α -acids close to those of commercial varieties. **Figures 3D and 3E**, only focused on wild accessions,
392 confirmed the high content of α -acid derivatives in these two accessions. These figures also allowed
393 us to better compare wild accessions with each other. It hence appeared that wild accessions, with a
394 composition in volatile compounds close to those of commercial varieties (in red and in blue), had a
395 lower content of acylphloroglucinol derivatives, while some wild accessions, as the accessions A5, B2,
396 B3, I1 or H1, had a higher content of these compounds. However, it would be relevant to continue the
397 identification of the chemical markers located at the top of PC2 and which differentiate several wild
398 individuals. The distribution of individuals on the loading plot was strongly different from the biplot of
399 the quantification (**Figure 2B**), and the explanation of the variance was relatively weak (less than 50%),
400 thereby reflecting the high chemical diversity of the accessions studied.

401 3.2. Chemical characterisation after brewing: beer analysis and sensory analysis

402 Once the hop accessions have been chemically characterised, we got interested in their aromatic
403 and gustative potential after brewing by sensory analysis. The beers elaborated according to a same
404 base beer style (lager) and with the different hops to evaluate were first characterised by physico-
405 chemical analysis before being tasted by a panel. The main objective of these analyses was to
406 characterise the matrix and to ensure the homogeneity of the batches tasted by the panelists.

407 3.2.1. Physico-chemical analysis

408 Physico-chemical characterisations were based on a measurement of IBU, EBC, alcohol content,
409 residual sugars and total acidity. These analyses revealed IBUs ranging from 13.1 (Cascade) to 38.2
410 (Magnum); alcohol content from 4.69 (F1) to 6.6 (Groene Bel) % v/v; EBC from 8.2 (H2) to 20.8 (H1);
411 total acidity from 4.1 (G1) to 8.1 (Northern Brewers); residual sugars from 1.190 (Target) to 2.3 °P (B3)
412 (**Table S5**). Higher IBUs were measured for beers brewed with commercial bittering hops (Brewers
413 Gold, Magnum), while beers brewed with wild hops had IBUs comparable to those of beers brewed
414 with commercial aromatic hops (Striesselspalt and Target). EBCs measured were those of a Lager beer,
415 according to the Pilsen malt used. Alcohol content, total acid concentration and density were not
416 statistically influenced by the hop used for the brewing. Furthermore, student test ($p = 0.95$) revealed
417 no significant differences among bottles of a same batch for alcohol, colour and residual sugars (with
418 p -value respectively equal to 0.282, 0.618 and 0.223), while for total acidity, p -value was less than
419 0.0001, which indicated a high level of variability among bottles from a same batch (**Table S6**).

420 3.2.2. Analysis of volatile compounds by SBSE GC-MS

421 In total, 180 volatile compounds were identified by SBSE GC-MS in beer samples. These compounds
422 were mainly represented by the ester chemical class (**Figure S2**). We also detected some
423 sesquiterpenes, as α -humulene, β -caryophyllene, α -, β -, γ - and δ -selinene and alloaromadendrene
424 (**Table S7**) previously identified in hop cones. Some monoterpenes, such as citronellol, were detected
425 in beers but were not found in cones. The analysis of the beer brewed without aromatic hop revealed
426 the quasi absence of terpenoids in this sample, and thus confirmed that terpenes and their derivatives
427 present in other beer samples could come from hop tested. Lastly, analysis on several different bottles
428 from the same batch revealed a similar composition in volatile compounds and thus confirmed the
429 homogeneity of the flavours of the beers tasted by the panelists, even if they come from different
430 bottles.

431 3.2.3. Sensory results

432 On the 17 descriptors on which the panel was trained (**Table S8**), we finally retained nine
433 descriptors which were significant (p -value <0.05%) (**Table S9**). These descriptors were: hoppy smell,
434 fruity smell, intensity of the taste, malt taste, hoppy taste, citrus taste, yellow fruits, bitter taste and
435 bitter persistence. To process data by PCA, we kept the first two principal components explaining 75.06
436 % of the variance, and we considered the beer brewed with no aromatic hop as a supplementary
437 observation (**Figure 4A**). PC1 (56.27% of the variance explained) translated the intensity of the taste,
438 bitter taste, bitter persistence, fruity smell, yellow fruits, and hoppy taste and smell; while PC2
439 (18.79%) translated the citrus and malt tastes (**Figure 4B**). **Figure 4A** and **Figure 4B** highlighted the
440 bitter taste of beers brewed with so-called bitter commercial varieties (such as Northern Brewers,
441 Magnum, Brewers Gold or Target) which had previously revealed high α -acid levels. The HAC (**Figure**
442 **4C**) clearly opposed the cluster made of beers brewed with commercial varieties on one side (in blue)
443 and another cluster composed of beers brewed with wild accessions on the other side (in red). The
444 clear separation already observed for the chemical analysis of the 52 hop samples was thus confirmed
445 after brewing (**Figure S3**). Nevertheless, the accessions I3 and G2 whose chemical composition was
446 fairly close to the marketed varieties no longer belonged this time to the same cluster concerning the
447 taste felt by the panelists after brewing. By contrast, in this sensory analysis, 5 wild accessions (B3, H1,

448 H2, I9, K5) of our collection belonged to the same cluster as commercial and old varieties (**Figure 4C**).
449 The chemical composition of these accessions was far from the reference varieties in regards to non-
450 targeted metabolomic analysis. However, accessions H1, H2, K5 and I9 revealed the presence of some
451 particular volatile compounds, while the accession B3 had high levels of some phenolic compounds.
452 The proximity of the wild accessions H2 with the aromatic cultivar Cascade, and I9 with K5, observed
453 on **Figure 4A**, had already been observed on the dendrogram **1B** for the analysis of volatile compounds.
454 α -bergamotene was identified in accession H2, γ -elemene in accession K5 as well as β -elemene, γ -
455 elemene, α -selinene, β -selinene and alloaromadendrene in accession I9. This composition was close
456 to those of the accession Strisselspalt for example, also containing alloaromadendrene, α -selinene, β -
457 selinene and γ -elemene and that could explain the fruity, hoppy and citrus perception of these
458 accessions according to panellists. Concerning the chemical composition of beers analysed by SBSE GC-
459 MS, some accessions (e.g. Brewers Gold, Magnum, B3, G1, I3 and I5) had a high level of monoterpenes
460 and sesquiterpenes. Nevertheless, among these beers, those brewed with wild accessions did not
461 show similar aroma profiles according to the panel. For the beer brewed with no aromatic hop (in
462 yellow on **Figure 4A**), analysed in the PCA as a supplementary observation, and therefore not included
463 in the AHC, panelists did not discriminate it from the other samples tasted, and seemed to be fairly
464 close to the accession G2 and the old variety Record, which may not have significant aromatic impact
465 on the beer.

466 3.3. Discriminant analysis: OPLS-DA

467 To objectively assess the influence of the chemical composition of hop cones used for brewing in
468 the final beer, a supervised OPLS-DA approach based on the chemical profiling and gustative evaluation
469 was performed. Two classes were selected in this study, according to the results of the AHC of the
470 sensory analysis: group 1 ($n = 13$ in blue) and group 2 ($n = 13$ in red) (**Figure 4C**). In total, 225 variables
471 identified in the two datasets previously collected to characterise the chemical diversity of hops
472 (volatile compound analysis by HS-SPME GC-MS: 65 variables, untargeted metabolomic analysis by
473 UHPLC-IMS-HRMS: 160 variables) were used for the OPLS-DA model to explain the aromatic diversity
474 of the 26 accessions used in brewing tests. The combination of these data was expected to provide a
475 global profiling of the gustative qualities of hops in an integrative brewing perspective. The consensus
476 OPLS-DA strategy was applied for the differential analysis of the two aromatic groups and the
477 simultaneous analysis of the two blocks of data.

478 The block contributions of the predictive latent variable indicated an equivalent importance of the
479 volatile cone analysis by HS-SPME GC-MS (p : 51.91%, o : 52.05%) or of the untargeted metabolomics
480 by UHPLC-IMS-HRMS (p : 48.08 %, o : 47.94 %) (**Figure S4**). A model with one predictive and one
481 orthogonal latent variable was evaluated as the best model based on the DQ2 value computed during
482 leave one-out cross validation, i.e., $DQ2 = 0.393$. Hence, in view of this relative low value, the proposed
483 model is an explanatory model and not a predictive model. Score plots and the contributions of the
484 model were presented in **Figure 5**. This model confirmed the clear opposition between the two beer
485 groups defined by the panelists, separated by the predictive component (**Figure 5A**) and used for the
486 model construction. As discussed before, **Figure 5B** highlighted the influence of some volatile
487 compounds on the aromatic profile of beers. Beers brewed with commercial varieties were influenced
488 by their content of main hop volatile compounds (as α -humulene, β -myrcene, β -caryophyllene, linalool
489 or β -pinene), while beers brewed with wild accessions were more distinguished by the composition of
490 their cones in α - and β -selinene, alloaromadendrene, β - and γ -elemene or α -bergamotene. This model
491 also revealed that the bitterness of beers was imparted by acylphloroglucinols derivatives (**Figure 5C**),
492 and then corroborated the bitter potential of commercial varieties compared to wild hops, which were
493 not distributed along the axes of the bitterness (**Figure 4A**).

494 4. Discussion

495 This study was focused on the chemical composition of wild hops and their aromatic potential in
496 beers (**Figure 6**). As regards the chemistry of cones, we identified the main volatile compounds of hops,
497 α -humulene, β -myrcene and β -caryophyllene, in all accessions of our collection. In particular, α -
498 humulene and β -caryophyllene are known for their spicy and woody smell (Rettberg et al., 2018). We
499 also found original volatile compounds, present in a few wild accessions, as α - and β -selinene,
500 alloaromadendrene or β - and γ -elemene. Patzak et al. (2010) reported that selinene was particularly
501 present in wild hops compared with North American hops. This work also highlighted the strong
502 chemical opposition between commercial and old varieties compared with wild accessions on their
503 phytochemical composition (volatile compounds and phenolic compounds) and gustatory quality. A
504 certain form of opposition between varieties and experimental germplasms had already been reported
505 by Yan et al. (2019) especially on volatile compound composition, using GC–acTOFMS, on 30
506 Australian hop genotypes. Furthermore, recent studies such as Morcol et al., (2020) or Van Holle et al.
507 (2021) suggested a terroir effect on hop chemical composition. The results of this study also served
508 this hypothesis as chemical diversity underlined by non-targeted analyses was much reduced on the
509 accessions of this study, all coming from the same experimental hop field, compared to the accessions
510 collected *in-situ* in our previous study Paguet et al. (2023). Nevertheless, the *ex-situ* collection also
511 allowed a monitoring of hop maturation and thus collection of the cones at their optimum maturity,
512 which avoided the appearance of degradation compounds. For instance, contrary to our previous
513 study, we noticed high levels of β -myrcene in all accessions without exception. Myrcene usually does
514 not make a contribution to hop aroma in beer, because its concentration is often far below the sensory
515 threshold level, due to its evaporation during wort boiling Kishimoto et al. (2005). However, it may be
516 used as a marker for cones ripening because monoterpenes are the last produced metabolites (Briggs
517 et al., 2004). Some specific hop volatile components, such as geraniol for Cascade or β -pinene for
518 Centennial, were identified as statistically relevant for forecasting dry-hop aroma quality (Lafontaine
519 et al., 2018). Phytochemical profiling of our collection led to the identification of some wild accessions
520 with a profile close to those of commercial cultivars (e.g. accessions I3 and G2). Previous genotyping
521 in Paguet et al. (2023) revealed that wild accessions of our collection were significantly different from
522 commercial and old cultivars grown in the north of France and were therefore not from an existing
523 commercial variety (**Figure S5**). Hence, the chemical diversity of wild accessions, even under *ex-situ*
524 conditions, was probably due to the wide genetic diversity previously observed and is also reflected in
525 significant morphological differences (**Figure S6**). Among this chemical diversity, we have identified
526 some derivatives of bitter acids. Wild accessions revealed high levels in β -acids. β -acids do not directly
527 impact the taste of beer, but their oxidation in hulupones provides a very bitter taste to the beer (Van
528 Cleemput et al., 2009).

529 Hence, after highlighting the chemical diversity of these hops grown under the same conditions,
530 we evaluated their aromatic potential after brewing. An intra-batch physico-chemical characterisation
531 made it possible to check that all the panellists tasted the same product. Nevertheless, inter-batch
532 variations were observed, particularly in terms of the product's acidity and its residual sugar content.
533 These differences may result from slightly different fermentation conditions, and did not really impact
534 the descriptors appreciated by the panellists and therefore should not have significantly affected the
535 taste characterisation of the products. After evaluation by a panel, we underlined that some wild
536 accessions had a gustative potential fairly close to commercial varieties, as the accession H1, with a
537 bitter perception. Some other wild accessions, H2, I9 and K5, belonging to the same cluster on the
538 dendrogram, rather showed hoppy and fruity taste as the commercial variety Cascade or the old
539 variety Groene Bel. This fruity and hoppy taste may be due to the presence of some volatile
540 compounds, as γ -elemene or α - or β -selinene, which were also detected after brewing by SBSE and
541 were known for their herbal and fresh flavors (**Table S1**). These statements have been confirmed by
542 the OPLS model, which underlined the influence of acylphloroglucinol derivatives and main hop volatile
543 compounds on the bitterness and aromatic properties of beers brewed with commercial varieties on
544 one hand, and the influence of less common volatile compounds and typical of our wild hop collection,
545 on the other. However, the aromatic impact of hops in beer remains very complex to study because

546 beer is a complex matrix due to the interaction with the other ingredients of the beer and the effects
547 of fermentation etc. (Dietz et al., 2020), as evidenced by the relative low validation score of our OPLS
548 model. For example, the presence of citronellol, found in some beer composition but in no single hop,
549 can be derived from a *de novo* synthesis, induced by yeast during the fermentation (Dietz et al., 2020).
550 Hence, Van Holle et al. (2017) chose to carry out a sensory analysis on single hop beers in order to
551 enhance the perception of the aromatic contribution of tested hop on beer. Machado et al., (2021)
552 studied wild accessions coming from Portugal and assessed their gustative potential on dry-hopped
553 beers. Dry-hopping allowed highlighting of the aromatic variations between different accessions
554 (Podeszwa & Harasym, 2016). In our study, we also greatly reduced bias because we tested our hops
555 in a same lager-style recipe, with a unique magnum bittering hop to provide bitterness, and in steeping
556 (whirlpool) hop mode. The only ingredient modifying the recipe was the aromatic hops, wild or
557 commercial, to be tested and it was added after boiling in infusion to preserve the volatile compounds.
558 However, it could also be interesting to test in parallel the organoleptic profile of hops, without
559 brewing, to avoid the interaction with malt and fermentation metabolites, as Martins et al. (2020)
560 evaluated the aromatic potential of hops by check-all-that-apply analysis.

561 **5. Conclusion**

562 By means of a targeted and untargeted chemical fingerprinting and chemometrics approach, this
563 study was able to identify compounds in our wild hops collection from Northern France that make
564 them distinct from commercial varieties. Volatile compound analysis revealed a strong opposition
565 between commercial and old cultivars and wild accessions of our collection; this opposition was also
566 noticeable on the quantification of phenolic compounds and sensory evaluation, but less strong as
567 regards untargeted metabolomics.

568 Some hops of wild accessions, such as I3 and G2, showed a chemical composition fairly close to
569 commercial varieties. However, other wild accessions, such as H2, K5 and I9, showed an aromatic
570 potential more pronounced during sensory analysis and close to the commercial variety Cascade or
571 the old variety Groene Bel. OPLS-DA allowed the identification of chemical markers of cones which
572 lead to a beer with sensory characteristics close to those brewed with the cultivars currently used by
573 brewers. This study underlined the potential for further investigating genetic markers to understand
574 the differences in the volatile compound concentrations as the phenotypic expression for each hop
575 cultivar. These results illustrated the difference between commercially germplasm resulting from
576 robust selection programs and the wild germplasms. This analytical approach thereby confirmed the
577 potential of wild hops for varietal development and gave some information to implement a systematic
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592 **Author's contribution**

593 **ASP:** Investigation; Methodology; Formal analysis; Writing – original draft; **AS:** Project
594 administration; Supervision; Writing-review & editing; **GL:** Conceptualization; Investigation in
595 particular of the harvesting and monitoring of hop field; Writing-review & editing; **MV, SC:** Supervision
596 of the sensory analysis; Methodology; Formal analysis; Writing-review & editing; **DL, ND:** Investigation
597 in particular of brewing; **JS, FM, CD:** Investigation in particular of the chemistry part; **DM:** Investigation
598 and Methodology in particular of High resolution mass spectrometry part; **MLF:** Supervision of GC-MS
599 analysis; Methodology; Writing-review & editing; **RM, JXF:** Supervision of untargeted metabolomic
600 analysis; Methodology; Formal analysis; Writing-review & editing; **SS:** Project administration; Writing-
601 review & editing; **CR:** Conceptualization; Investigation and Supervision in particular of the harvesting
602 and chemistry part, as well as the overall approach of the project; Methodology; Validation; Project
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604

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702 **Declaration of interests**

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704 The authors declare that they have no known competing financial interests or personal
705 relationships that could have appeared to influence the work reported in this paper.

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707 The authors declare the following financial interests/personal relationships which may be
708 considered as potential competing interests:

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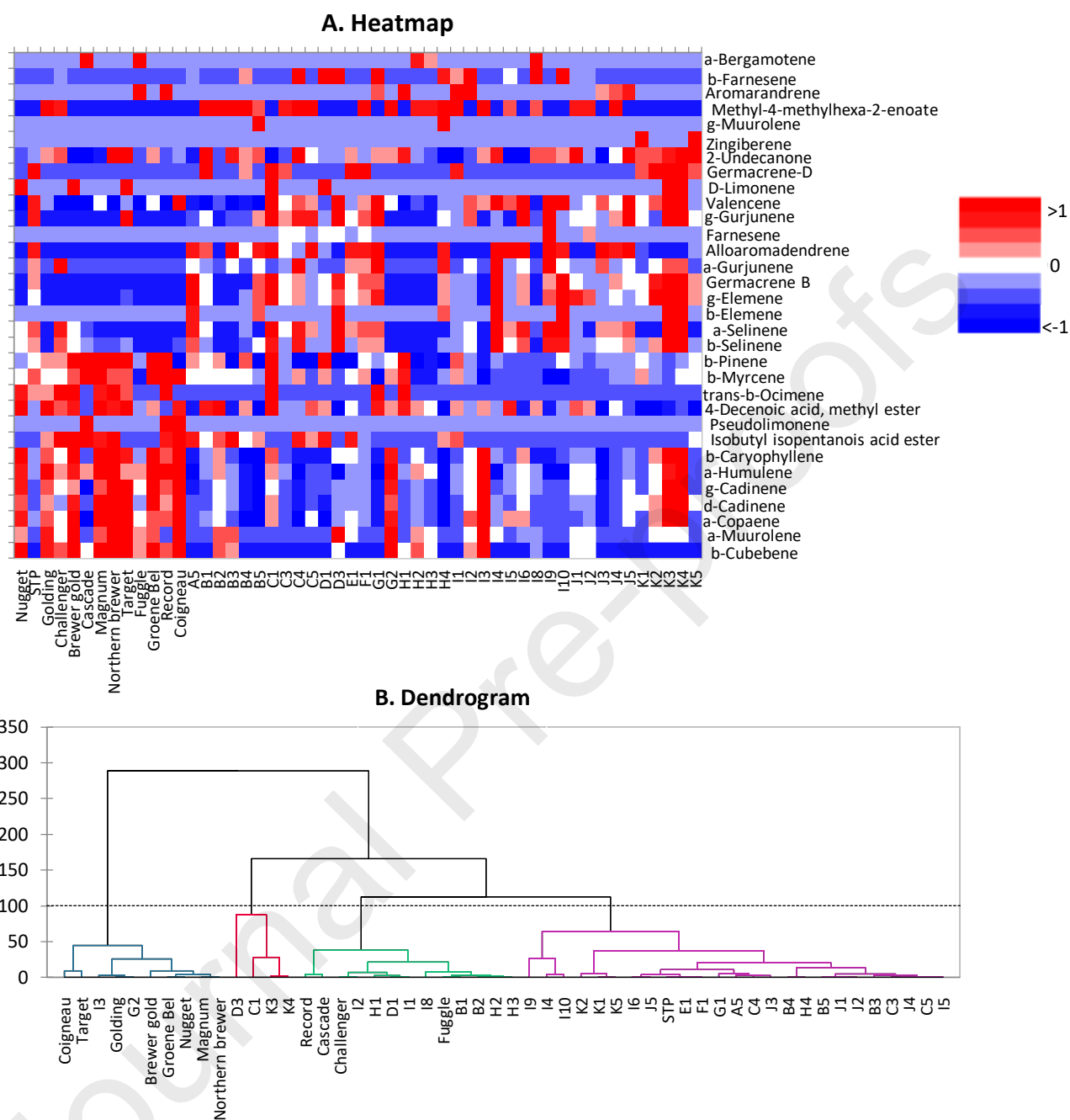
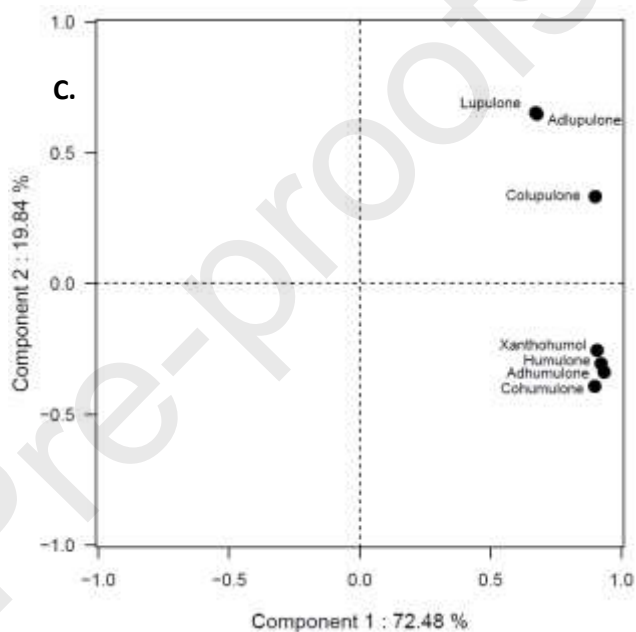
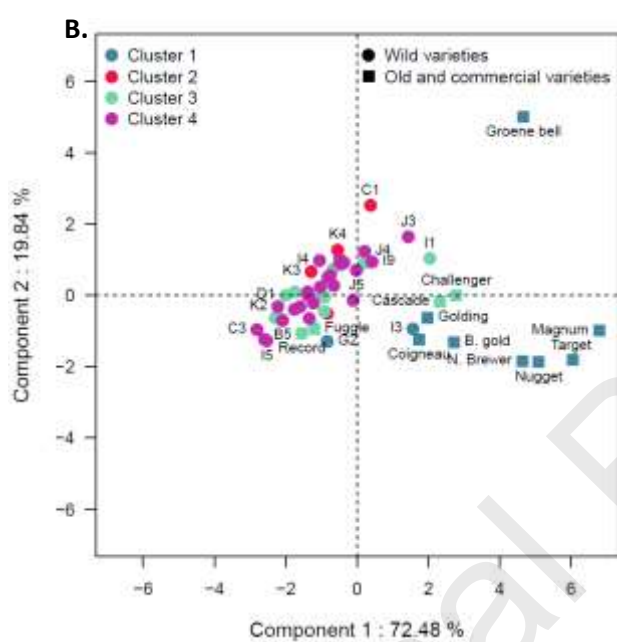
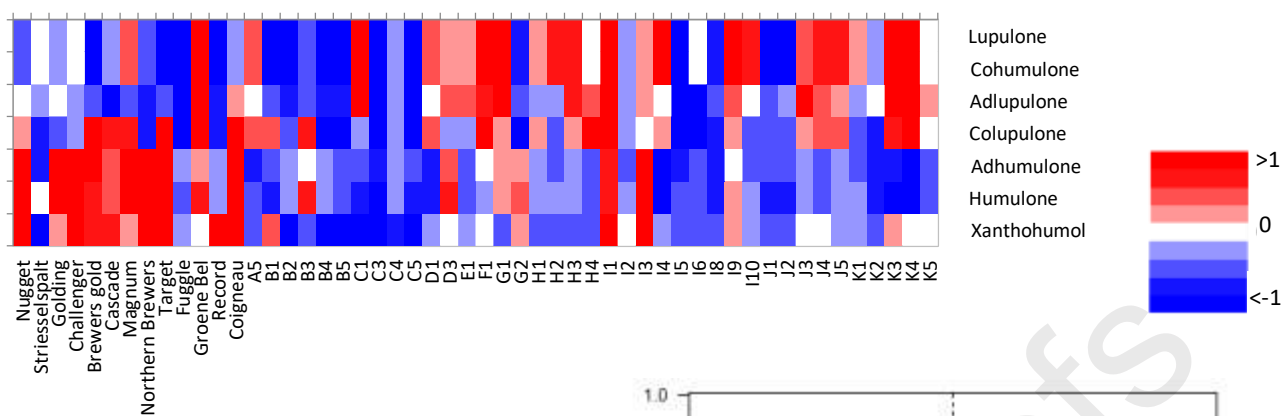


Figure 1. Volatile compound analysis by HS-SPME GC-MS based on the 32 compounds detected at least 1% in one accession. A. Heatmap generated with normalized data for the top 32 compounds responsible for the differences between the profiles. B. Dendrogram associated to the AHC (Ward's method, $n=4$). STP = Strisselspalt.

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A. Heatmap



nds by UHPLC-UV. A: Heatmap; B: Loading plot of the 52 previous clusterisation regarding their composition in volatile compounds, old and commercial cultivars represented by a square; C: Loading plot of the seven compounds quantified.

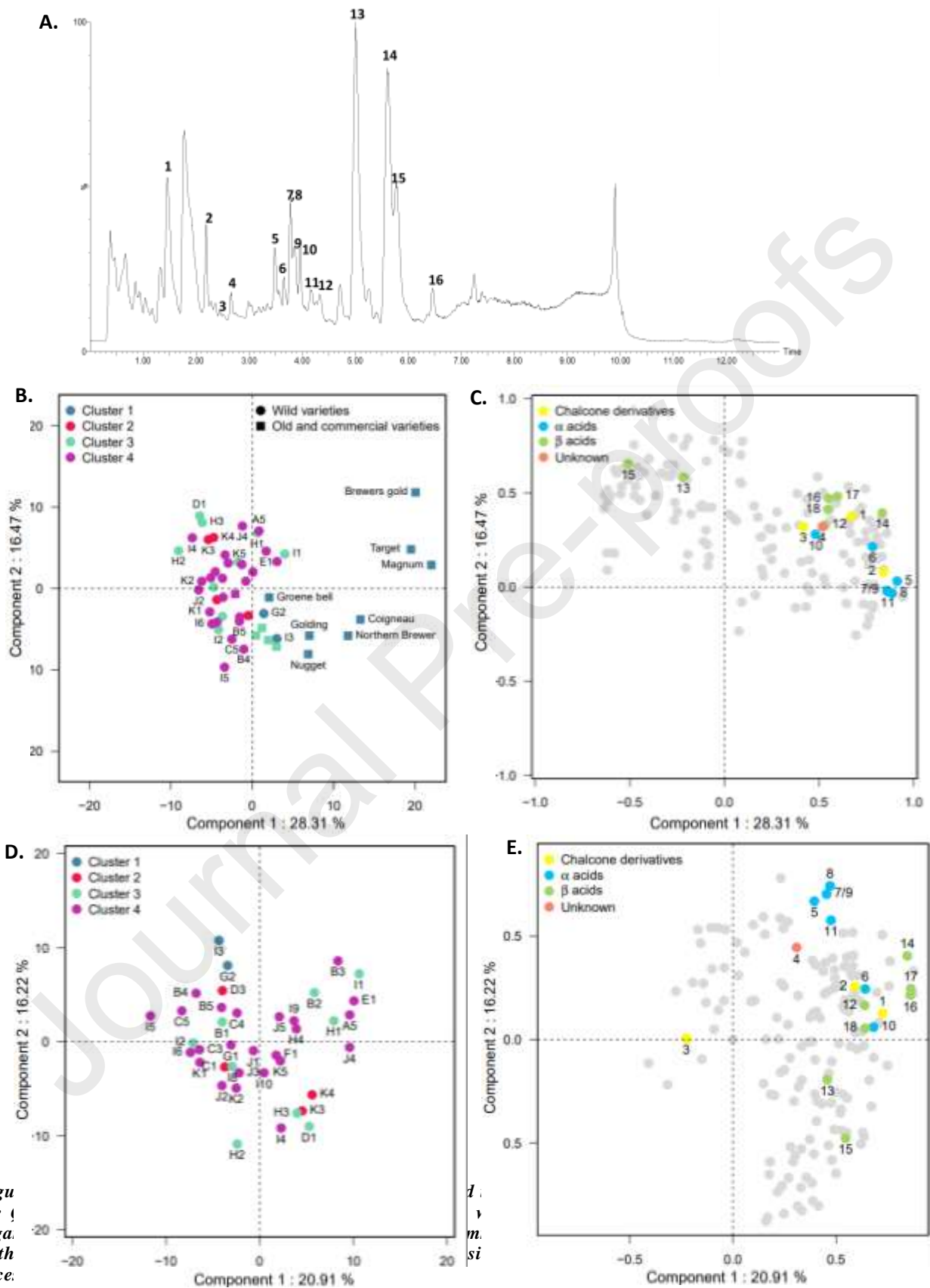


Fig. 1
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Isoxanthohumol, 4 = Unknown, 5 = Cohumulone, 6 = 4-Deoxycohumulone or adhumulone, 7 = Humulone, 8 = Humulinone, 9 = Adhumulone, 10 = Deoxyhumulone, 11 = Prehumulone, 12 = Postlupulone, 13 = Colupulone, 14 = Lupulone, 15 = Adlupulone, 16 = Prelupulone.

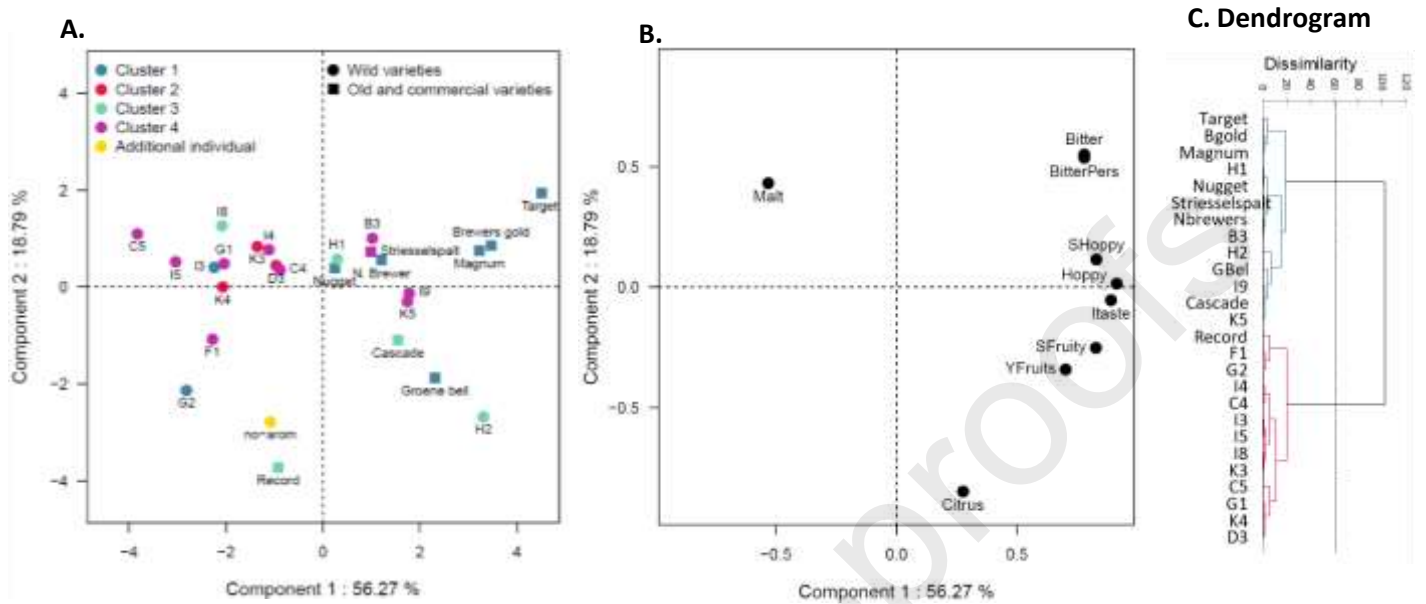


Figure 4. Analysis of the data panel. **A:** Loading plot of the 27 beers on the dimensions 1 and 2 on the data of the sensory characterization by a panel. Individuals were coloured according to their composition in volatile compounds. The observation “no-arom” is brewed with no aromatic hop; **B:** Loading plot of the sensorial descriptors used for the sensory analysis. “SHoopy”: hoppy smell; “Hoppy”: hoppy taste; “SFruity”: fruity smell “YFruits”: yellow fruits flavour; “Itaste”: intensity of the taste; “Citrus”: citrus flavour; “Malt”: malt flavour; “Bitter”: bitter flavour; “BitterPers”: bitter persistence; **C:** Dendrogram obtained by AHC plotting the data of 27 beers for the 9 descriptors (Ward’s method, $n=2$).

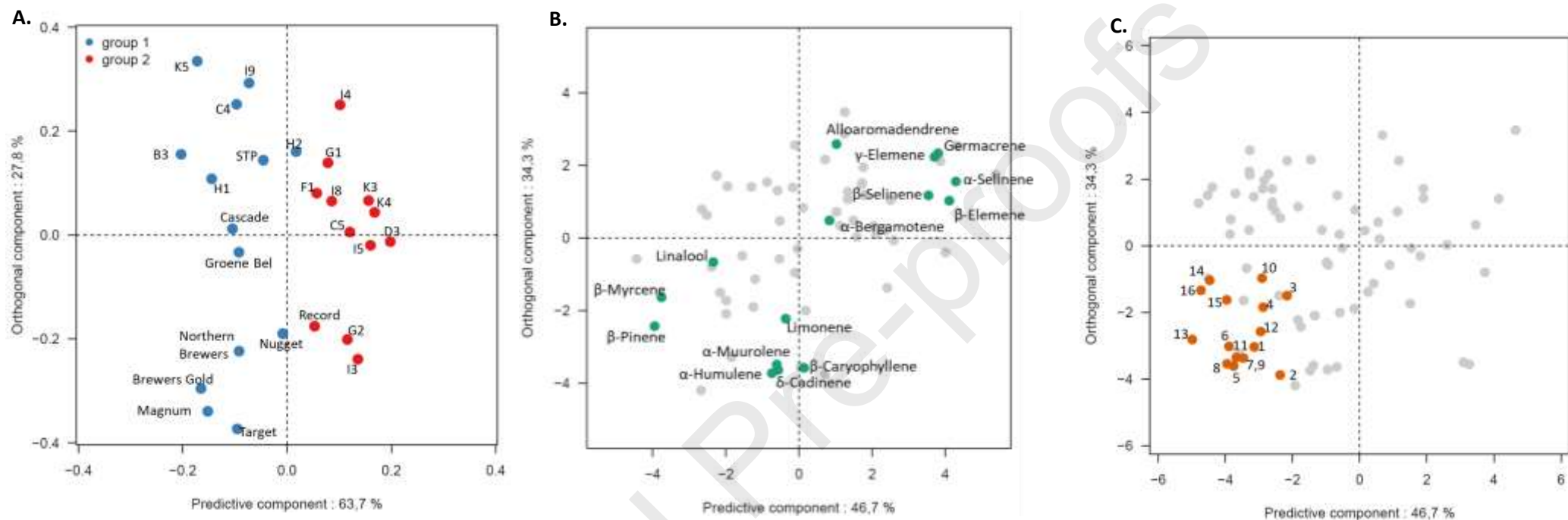
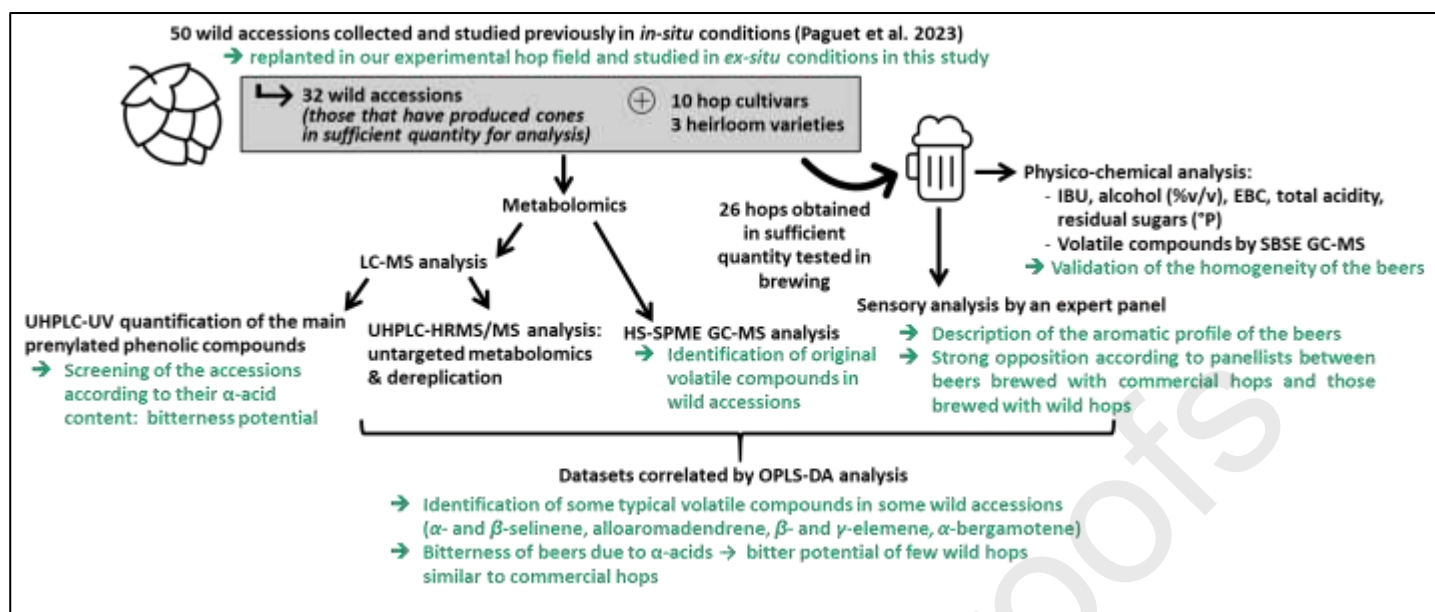


Figure 5. Multivariate association of beer metabolites with panel sensory traits. The association between chemical characterization by HS-SPME GC-MS (68 variables) and the UHPLC-ESI-Qtof (160 variables) was evaluated by OPLSA-DA using HCA clusters from sensory analysis. A: score plot of the aromatic qualities of beers according the panel test; B: loading plot hop volatile compounds analyzed by HS-SPME GC-MS; C: loading plot untargeted metabolomics by UHPLC-HRMS-ESI-Qtof. 1 = Desmethylxanthohumol, 2 = Xanthohumol, 3 = Isoxanthohumol, 4 = Unkown, 5 = Cohumulone, 6 = 4-Deoxycohumulone or adhulupone, 7 = Humulone, 8 = Humulinone, 9 = Adhumulone, 10 = Deoxyhumulone, 11 = Prehumulone, 12 = Postlupulone, 13 = Colupulone, 14 = Lupulone, 15 = Adlupulone, 16 = Prelupulone.

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3 **Figure 6.** Summary diagram of the analyses carried out in this research and the main findings obtained.

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5 **Highlights**

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- Few wild hops, such as I3, have a similar chemical composition to commercial hops.
 - Wild hops produce original volatile compounds compared with commercial hops.
 - Some wild hops show an aromatic potential after brewing.
 - Discriminant analysis opposes beers brewed with wild hops to those with commercial hops.
 - This observation is consistent with their chemical composition.

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