Abstract Submission

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CRYSTAL CHEMISTRY OF THE LONDONITE-RHODIZITE SOLID SOLUTION Kim Totaro^{* 1}, Frédéric Hatert¹, Zelemir Gabelica² ¹Minéralogie, Université de Liège, Liège, Belgium, ²Minéralogie, Université de Haute Alsace, Mulhouse Cedex, France

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Londonite and rhodizite are rare borates which mainly contain beryllium, aluminium, as well as large alkaline cations K and Cs. These large cations substitute to each other to form a complete solid solution between the two end-members: londonite, $CsBe_4AI_4(B_{11}Be)O_{28}$, and rhodizite, $KBe_4AI_4(B_{11}Be)O_{28}$. The 32 samples investigated herein were collected in the LTC-type granitic pegmatites located in central Madagascar; they form rhombododecahedral crystals that can reach 3 cm in diameter. The samples were investigated by single-crystal X-ray diffraction techniques (Rigaku Xcalibur diffractometer, EOS detector, MoK α radiation), electron microprobe (Cameca SX-100, 15 kV, 15 nA) and LA-ICP-TOF-MS. Structure refinements were performed in space group *P*-43*m*, to final R₁ values between 2.64 and 6.64 %. Unit-cell parameters show significant variations: *a* = 7.291-7.326 Å, and V = 387.91-393.13 Å³ (with *Z*=1). The structure of the londonite-rhodizite series is composed of five types of cations distributed over four different crystallographic sites. Aluminium is located at the centre of octahedral sites, whereas Be and B are located in tetrahedral coordination polyhedra. Cs and K are coordinated by 12 oxygen atoms and form polyhedra resembling large truncated tetrahedra. Within a single grain, significant compositional variations in Cs and K were observed. A positive correlation exists between Cs and the number of vacancies *pfu*, and the maximal number of vacancies *pfu* reaches 0.40, thus approaching to the new species "kenolondonite" not yet described in the literature.

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