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GREENHOUSE GAS (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) EMISSIONS FROM  
ECUADORIAN MOUNTAINOUS STREAMS AND HIGH  
ELEVATION LAKES



*Lake Santa Lucia in the front and in the background the Antisana Volcano - Pichincha Ecuador*

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# GREENHOUSE GAS (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) EMISSIONS FROM ECUADORIAN MOUNTAINOUS STREAMS AND HIGH ELEVATION LAKES



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# ABSTRACT

Rivers and streams contribute substantially to global emissions of carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>). Most riverine GHG emissions are from the tropics, ~60-80% for CO<sub>2</sub>, ~70% for CH<sub>4</sub>, and ~79% for nitrous oxide (N<sub>2</sub>O). Nevertheless, the total stream surface area (SSA) in the tropics (24°N-24°S) corresponds to ~44% of the total SSA globally, implying that riverine areal emissions of GHGs (per m<sup>2</sup>) are higher than in other climatic zones. Ecuador localizes on the tropics, and it is the smallest of the 17 megadiverse countries in the world harboring an astounding number of ecosystems, which offers the conditions to study in a relatively short space the GHG emissions of both, an important river tributary to the Amazon, as well as a diverse lacustrine system. The present thesis is divided into two sections. Firstly, we report for the first time the dissolved concentrations CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O in Andean headwater and piedmont streams in the Napo River basin in Ecuador, part of the Amazon River catchment. We found that concentrations increased exponentially with elevation decrease, i.e., between 3,990 and 175 m above sea level. Conversely, Andean mountainous headwater and piedmont streams are hotspots of CO<sub>2</sub> and CH<sub>4</sub> emissions; the respective areal fluxes are 1.7 and 4.5 higher in headwater streams and 1.2 and 6.6 higher in piedmont streams than in lowland streams.

Secondly, report the dissolved GHGs of 15 lakes along an elevational gradient ranging from 2,213 to 4,361 m. Lakes were grouped into three clusters as a function of their characteristics and obtained outcomes. Most lakes showed lower values of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (644-2,152 ppm) than usually attributed to tropical lakes (~1,900 ppm). Three lakes, influenced by volcanic inputs, denoted higher pCO<sub>2</sub> values (3,269-10,069 ppm), while two lakes bordered by large cities showed the lowest pCO<sub>2</sub> values (208-254 ppm), the remaining ten lakes had moderated pCO<sub>2</sub> and the highest dissolved organic carbon concentrations (8.3 to 12.8 mg L<sup>-1</sup>) that is characteristic of páramo lakes (644-2,152 ppm). Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol L<sup>-1</sup> and were negatively correlated to lake area and depth. N<sub>2</sub>O saturation levels ranged between 64 % and 101 %. The surface waters were under-saturated in N<sub>2</sub>O regarding atmospheric equilibrium, probably due to soil-water inputs with low N<sub>2</sub>O levels due to soil denitrification. By upscaling to lakes in the whole Amazon basin, we found that the average areal FCH<sub>4</sub> (per m<sup>2</sup>) was 1.3 times higher in highland lakes (0.75 gC m<sup>-2</sup> yr<sup>-1</sup>) than in lowland lakes (0.58 gC m<sup>-2</sup> yr<sup>-1</sup>) owing to the combination of higher dissolved CH<sub>4</sub> concentration (265 nmol L<sup>-1</sup> versus 179 nmol L<sup>-1</sup>) and higher gas transport velocity (k) values (3.5 cm h<sup>-1</sup> versus 2.5 cm h<sup>-1</sup>). Meanwhile, the average areal FCO<sub>2</sub> was 11 times lower in

the highlands ( $23 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) than in the lowlands ( $255 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) owing to the lower  $\text{pCO}_2$  values (530 ppm versus 2,909 ppm).



# RÉSUMÉ

Les rivières et fleuves contribuent de manière substantielle aux émissions globales de dioxyde de carbone (CO<sub>2</sub>) et de méthane (CH<sub>4</sub>). La plupart des émissions de GES par les rivières proviennent des tropiques, ~60-80% pour le CO<sub>2</sub>, ~70% pour le CH<sub>4</sub>, et ~79% pour l'oxyde nitreux (N<sub>2</sub>O). Néanmoins, la surface totale des cours d'eau situés dans les tropiques (24°N-24°S) correspond à ~44% de la surface totale des cours d'eau dans le monde, ce qui implique que les émissions fluviales de GES par unité de surface (m<sup>2</sup>) sont plus élevées que dans d'autres zones climatiques. L'Équateur se situe sous les tropiques et est le plus petit des 17 pays megadivers du monde, abritant un nombre élevé d'écosystèmes différents, ce qui permet d'étudier dans un espace relativement petit, les rejets vers l'atmosphère de GES d'un important affluent de l'Amazone, ainsi que d'un système lacustre diversifié. La recherche s'est réalisée en deux étapes. Dans un premier temps, nous avons documenté pour la première fois les concentrations de CO<sub>2</sub>, CH<sub>4</sub> et N<sub>2</sub>O dissous dans les cours d'eau andins en amont et en aval du bassin de la rivière Napo en Équateur, qui fait partie du bassin versant de l'Amazone. Nous avons observé que les concentrations augmentaient de manière exponentielle avec la diminution de l'altitude, c'est-à-dire entre 3 990 et 175 m au-dessus du niveau de la mer. Inversement, les cours d'eau en amont et en aval des montagnes andines sont des hotspots d'émission de CO<sub>2</sub> et de CH<sub>4</sub> ; les flux respectifs sont 1,7 et 4,5 fois plus élevés dans les cours d'eau en amont et 1,2 et 6,6 fois plus élevés dans les cours d'eau en aval que dans les cours d'eau en plaine.

Le second volet de notre étude a consisté à échantillonner les concentrations de GES dissous de 15 lacs le long d'un gradient d'altitude allant de 2 213 à 4 361 m. Les lacs ont été regroupés en trois groupes en fonction de leurs caractéristiques et de nos observations. La plupart des lacs présentaient des valeurs de pression partielle en CO<sub>2</sub> (pCO<sub>2</sub>) (644-2152 ppm) inférieures à celles généralement attribuées aux lacs tropicaux (~1900 ppm). Trois lacs, influencés par des apports volcaniques, présentaient des valeurs de pCO<sub>2</sub> plus élevées (3 269-10 069 ppm), tandis que deux lacs bordés par de grandes villes présentaient les valeurs de pCO<sub>2</sub> les plus basses (208-254 ppm). Les dix lacs restants présentaient des valeurs de pCO<sub>2</sub> modérées et les concentrations de carbone organique dissous les plus élevées (8.3 à 12.8 mg L<sup>-1</sup>), caractéristiques des lacs de páramo (644-2152 ppm). Les concentrations de CH<sub>4</sub> dissous variaient entre 170 et 24 908 nmol L<sup>-1</sup> et étaient négativement corrélées à la superficie et à la profondeur des lacs. Les niveaux de saturation en N<sub>2</sub>O étaient compris entre 64 % et 101 %. Les eaux de surface étaient sous-saturées en N<sub>2</sub>O par rapport à la

concentration atmosphérique, probablement en raison des apports d'eau provenant des sols avec de faibles niveaux de N<sub>2</sub>O dus à la dénitrification des sols. En comparant les résultats avec ceux du bassin de l'Amazonie, nous avons constaté que le flux moyen FCH<sub>4</sub> (par m<sup>2</sup>) était 1,3 fois plus élevé dans les lacs d'altitude (0,75 gC m<sup>-2</sup> an<sup>-1</sup>) que dans les lacs de plaine (0,58 gC m<sup>-2</sup> an<sup>-1</sup>) en raison de la combinaison d'une concentration plus élevée de CH<sub>4</sub> dissous (265 nmol L<sup>-1</sup> contre 179 nmol L<sup>-1</sup>) et de valeurs de coefficient d'échange (k) plus élevées (3.5 cm h<sup>-1</sup> contre 2.5 cm h<sup>-1</sup>). Parallèlement, le flux de CO<sub>2</sub> moyen était 11 fois plus faible dans les hautes terres (23 gC m<sup>-2</sup> an<sup>-1</sup>) que dans les basses terres (255 gC m<sup>-2</sup> an<sup>-1</sup>) en raison des valeurs plus faibles de pCO<sub>2</sub> (530 ppm contre 2 909 ppm)

# RESUMEN

Los ríos y arroyos contribuyen sustancialmente a las emisiones globales de dióxido de carbono ( $\text{CO}_2$ ) y metano ( $\text{CH}_4$ ). La mayoría de las emisiones de GEI fluviales provienen de los trópicos, ~60-80% de  $\text{CO}_2$ , ~70% de  $\text{CH}_4$  y ~79% de óxido nitroso ( $\text{N}_2\text{O}$ ). Sin embargo, el área superficial total de las corrientes fluviales (SSA) en los trópicos ( $24^\circ\text{N}$ - $24^\circ\text{S}$ ) corresponde a ~44% del SSA total a nivel mundial, lo que implica que las emisiones de GEI por área (por  $\text{m}^2$ ) son más altas que en otras zonas climáticas. Ecuador se localiza en los trópicos, y es el más pequeño de los 17 países megadiversos en el mundo, por lo que alberga una asombrosa cantidad de ecosistemas, lo que ofrece las condiciones para estudiar, en un espacio relativamente corto, las evasiones de GEI de un importante río afluente del Amazonas, así como un sistema lacustre diverso. La presente tesis se divide en dos secciones. En primer lugar, reportamos por primera vez las concentraciones disueltas de  $\text{CO}_2$ ,  $\text{CH}_4$  y  $\text{N}_2\text{O}$  en los arroyos de cabecera y piedemonte andinos en la cuenca del río Napo en Ecuador, parte de la cuenca del río Amazonas. Encontramos que las concentraciones aumentaron exponencialmente con la disminución de la elevación, es decir, entre 3990 y 175 m sobre el nivel del mar. Asimismo, las cabeceras montañosas de los Andes y los arroyos de piedemonte son puntos críticos de emisiones de  $\text{CO}_2$  y  $\text{CH}_4$ . Los flujos respectivos son 1,7 y 4,5 más altos en los arroyos de cabecera y 1,2 y 6,6 más altos en los arroyos de piedemonte que en los de tierras bajas.

En segundo lugar, reportar los GEI disueltos de 15 lagos a lo largo de un gradiente altitudinal que va de 2213 a 4361 m. Los lagos se agruparon en tres tipos; en función de sus características y resultados obtenidos. La mayoría de los lagos mostraron valores más bajos de la presión parcial de  $\text{CO}_2$  ( $p\text{CO}_2$ ) (644-2,152 ppm) de lo que se suele atribuir a los lagos tropicales (~1,900 ppm). Tres lagos, influenciados por aportes volcánicos, denotaron valores de  $p\text{CO}_2$  más altos (3,269 -10,069 ppm), mientras que dos lagos bordeados por centros urbanos mostraron los valores de  $p\text{CO}_2$  más bajos (208-254 ppm). Los diez lagos restantes tenían  $p\text{CO}_2$  moderado (644-2,152 ppm) y los valores más altos de carbón orgánico disuelto (8.3 a 12.8  $\text{mg L}^{-1}$ ), lo cual es característico de los lagos de páramo. Las concentraciones de  $\text{CH}_4$  disuelto oscilaron entre 170 y 24,908  $\text{nmol L}^{-1}$  y se correlacionaron negativamente con el área y la profundidad del lago. Los niveles de saturación de  $\text{N}_2\text{O}$  oscilaron entre el 64 % y el 101 %. Las aguas superficiales estaban subsaturadas en  $\text{N}_2\text{O}$  con respecto al equilibrio atmosférico, probablemente debido a aportes de agua al suelo con bajos niveles de  $\text{N}_2\text{O}$  debido a la desnitrificación del suelo. Al escalar a los lagos en toda la cuenca del Amazonas, encontramos que el flujo promedio de  $\text{FCH}_4$  (por  $\text{m}^2$ ) era

1,3 veces mayor en los lagos de las tierras altas ( $0,75 \text{ gC m}^{-2} \text{ año}^{-1}$ ) que en los lagos de las tierras bajas ( $0,58 \text{ gC m}^{-2} \text{ año}^{-1}$ ) debido a la combinación de una mayor concentración de  $\text{CH}_4$  disuelto ( $265 \text{ nmol L}^{-1}$  versus  $179 \text{ nmol L}^{-1}$ ) y mayores valores de velocidad de transporte de gas ( $k$ ) ( $3,5 \text{ cm h}^{-1}$  versus  $2,5 \text{ cm h}^{-1}$ ). Mientras tanto, el  $\text{FCO}_2$  promedio por área fue 11 veces menor en las tierras altas ( $23 \text{ gC m}^{-2} \text{ año}^{-1}$ ) que en las tierras bajas ( $255 \text{ gC m}^{-2} \text{ año}^{-1}$ ) debido a los valores más bajos de  $\text{pCO}_2$  ( $530 \text{ ppm}$  versus  $2,909 \text{ ppm}$ )

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# CHAPTER 1

# INTRODUCTION

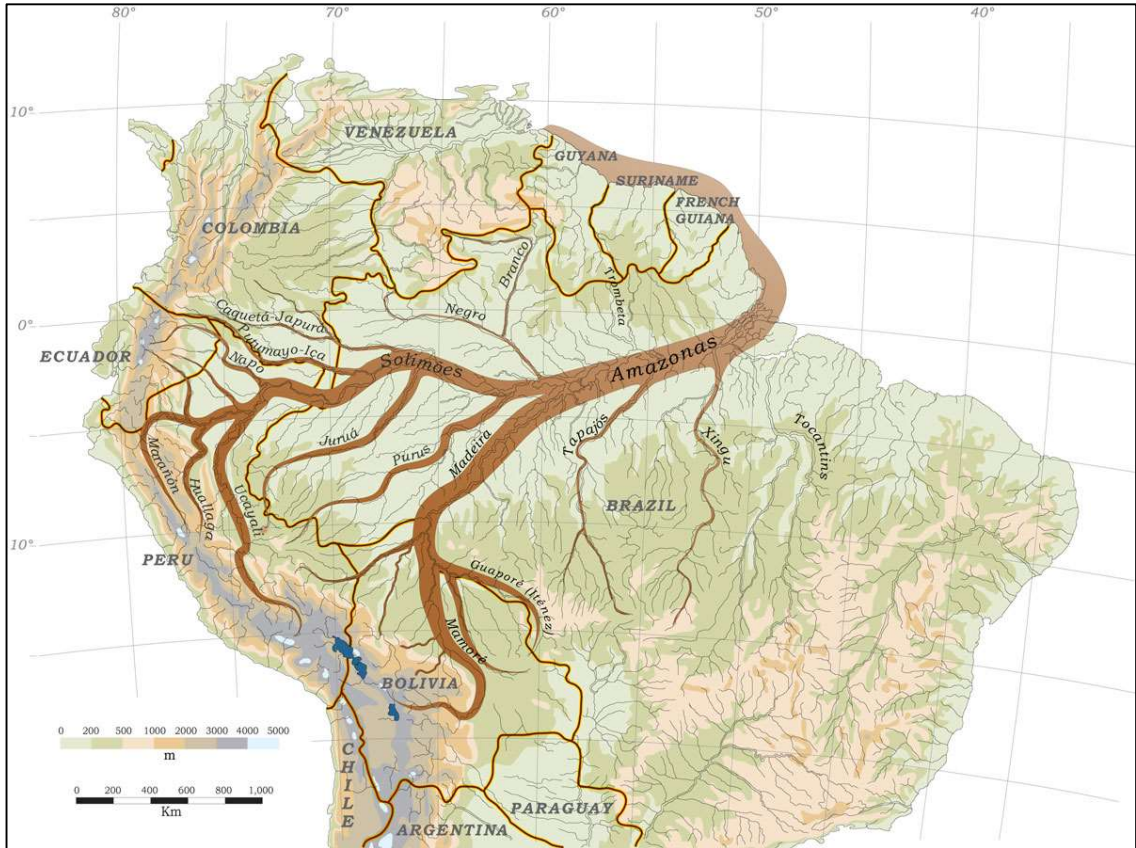


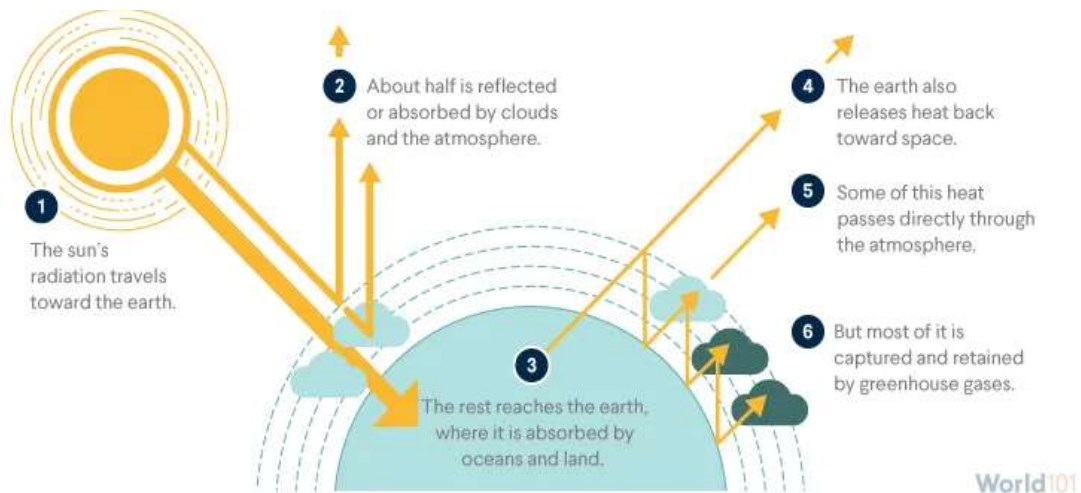
FIGURE 1.1 Amazon Basin - Fluvial Systems Source: (Amazon Waters, 2020)

## Climate is physics!

The Nobel Committee 2022 awarded the physics prize for the "Physical modeling of Earth's climate, quantifying variability and reliably predicting global warming." Two of the three laureates —Klaus Hasselmann and Syukuro Manabe— pioneered climate modeling and conceptualized elegant explanations of observed climate properties to the comprehensive analysis based on numerical fluid dynamics used today (Hegerl, 2022). The Nobel prize award highlights that Climate is physics, rendering the question "do you believe in global warming" meaningless and inferring that modeling is a crucial element of understanding and mitigating climate change.

The Climate is a forced, dissipative, chaotic system that is out of equilibrium and whose complex natural variability arises from the interplay of positive and negative feedback, instabilities, and saturation mechanisms (Ghil and Lucarini, 2020). Those factors include many chemical species and all physical phases. The system's heterogeneous phenomenology includes the microphysics of clouds, cloud-radiation interactions, atmospheric and oceanic boundary layers, land and inland waters interactions, and several scales of turbulence (Ghil, 2019). On different time scales, the climate systems' equilibrium has been strongly affected by relatively small changes in the forcing for anthropogenic and natural disturbances (Lucarini et al., 2014). On the macroscopic level, the differences in the absorption of solar radiation throughout the atmosphere drive the Climate, as well as in a thin surface layer of the ocean, continental waters, and soil. Earth's natural greenhouse effect makes life as we know it. By far, the most important GHG is water vapor. However, there is a substantial contribution from carbon dioxide (CO<sub>2</sub>), ozone (O<sub>3</sub>), methane (CH<sub>4</sub>), and nitrous oxide (N<sub>2</sub>O) in providing for the relatively mild temperature on Earth (Mitchell, 2009).

**FIGURE 1.2** Greenhouse gas effect. Sun acts as perfect blackbody that emits radiation. The balanced composition of the atmosphere sustains life and geochemical processes (World 101, 2022).



A perfect emitter or blackbody emits radiation over a range of wavelengths. The distribution of energy emitted with wavelength is a function of the temperature of this emitter: the hotter the emitter, the shorter the wavelength of peak emission (Gao et al., 2019). Thus, the Sun, which has a surface temperature of ~ 6000 K, emits most radiation in the range of 0.2 - 4  $\mu\text{m}$  (including ultraviolet, visible, and near-infrared wavelength), whereas the Earth at 255 K emits mainly in the range of 4 - 100  $\mu\text{m}$ .

Although water vapor is such a potent greenhouse molecule, human activities do not substantially influence its dynamics, and it has a short atmospheric life of some days in contrast to CO<sub>2</sub> (Inglezakis, 2016). CO<sub>2</sub> and other greenhouse gases absorb the long-wave radiation emitted from the surface and reradiate it to the surface and space, producing additional warming and maintaining the balance with incoming solar radiation Fig 1.2. As a result, the current global mean surface temperature is substantially warmer than the effective radiating temperature. This increase in surface temperature is known as the "greenhouse effect." The term applies to atmospheric gases, which are relatively transparent to solar radiation but absorb long-wave radiation like glass in a greenhouse. Thus, the term "greenhouse" is misleading, as the atmospheric greenhouse effect is solely due to radiative processes. Nevertheless, the term is universally accepted (Mitchell, 1989).

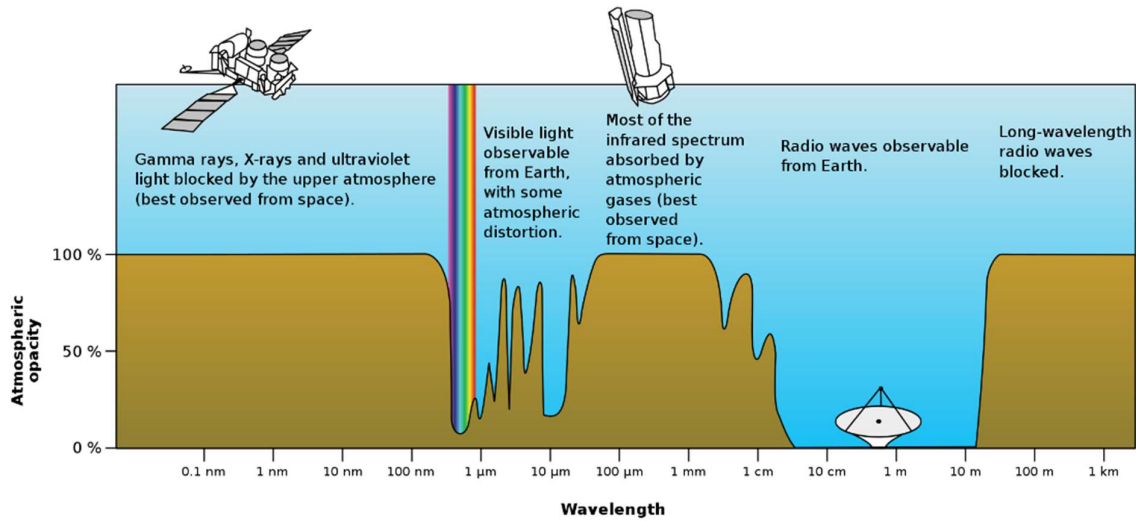
Consider the following analogy to understand the greenhouse effect. Although the surface of Venus is high enough to melt lead and the temperature on Mars is so low that almost all its relatively small amount of water is frozen, the Earth, orbiting between these two inhospitable neighbors, enjoys a climate that has evolved a vast variety of complex life (American Chemical Society, 2022). Their respective atmospheres have made Venus a furnace and have locked Mars in a deep freeze. Venus's atmosphere is composed of CO<sub>2</sub>; however, it started with roughly the same amount of CO<sub>2</sub> as the Earth. Nonetheless, the surface of Venus was so hot and dry that carbon-bearing compounds evaporated rather than remaining in the rocks accumulating a massive greenhouse gas in the atmosphere (Spencer, 2022).

Essentially 100% of the energy that fuels the Earth comes from the Sun, whose temperature in the central interior regions is around  $8 \times 10^6$  K (Duffie and Beckman, 2013). The solar constant  $S_0$  is the energy from the Sun per unit time received on a unit area of the surface perpendicular to the direction of propagation of the radiation at mean Earth-Sun distance outside the atmosphere. The World Radiation Center (WRC) has adopted a value of  $S_0 = 1,367 \text{ W m}^{-2}$ , with an uncertainty of the order of  $\pm 1\%$  (Widén and Munkhammar, 2019). This sunlight is in the visible region, where the Earth's atmosphere is almost transparent. According to the Wien Displacement Law, the wavelength of maximum emission for the Sun is  $0.5 \mu\text{m}$  (Pidwirny, 2006). Therefore, almost 99 % of this Sun's radiation is in short wavelengths from  $0.15$  to  $4.0 \mu\text{m}$ . Of this, 9 % is in the ultraviolet ( $A < 0.4 \mu\text{m}$ ), 45 % is in the visible ( $0.4 < A < 0.74 \mu\text{m}$ ), and 46 % is in the near-infrared  $A > 0.74 \mu\text{m}$ . Of the total amount of incoming radiation, the Earth's surface (land and oceans) absorbs around ~50 %; the atmosphere absorbs ~20 %; and the clouds, the Earth's surface, and different gases and particles (the Earth's albedo is 0.3 on average) reflect the remaining ~30 % to space Fig 1.3. Therefore, 70% of the Sun's energy causes Earth's surface, clouds, and atmosphere to warm (North Carolina Climate Office, 2016). Since that portion of energy makes its way back out to space the energy balance is well maintained. This cycle occurs 24 hours a day, and the energy emitted by Earth is in the longwave radiation due to the characteristic temperatures of the Planet (US National Weather Service, 2019).

The stratosphere is heated directly by O<sub>3</sub>'s absorption of solar radiation and is approximately in radiative equilibrium. All molecules with three or more atoms meet this criterion of IR absorbers. While the Earth's (dry) atmosphere is predominantly composed of non-IR absorbers, N<sub>2</sub> (78 %), O<sub>2</sub> (21 %), and Ar (~0.9 %), the 0.1 % of remaining trace gases contain CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O, and O<sub>3</sub> that absorb IR. Collisions between these energized molecules and others transfer energy through the atmosphere, increasing the average thermal energy and raising the temperature. As a result, heat distributes through the troposphere by convective (vertical) mixing (Mitchell, 2009). In the presence of an atmosphere, much of the longwave emission to space emanates from atmospheric gases rather than the surface. Thus, the level at which the mean radiating height occurs varies principally with the infrared opacity. Gases such as CH<sub>4</sub>, N<sub>2</sub>O, O<sub>3</sub>, CO<sub>2</sub>, and water vapor (IPCC, 2021) tend to be sensitive to infrared opacity (Smirnov, 2016). Regaining our planetary analogy, Mars, with a relatively thin atmosphere, has a small mean radiating height and a weak greenhouse effect. In contrast, Venus, with its dense, highly

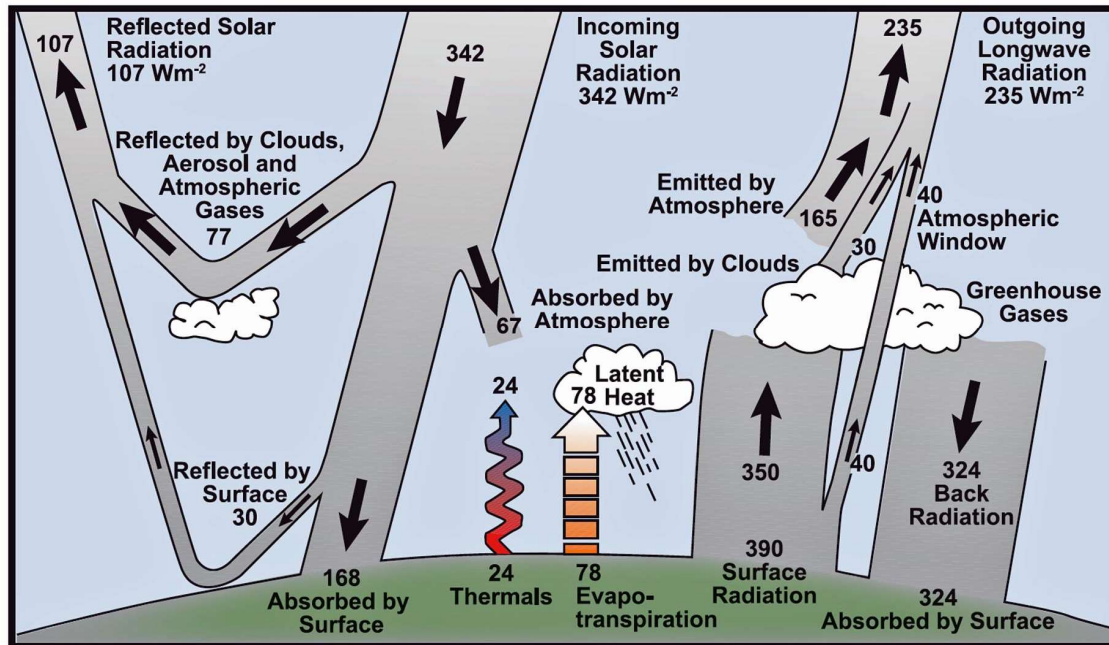
absorbing CO<sub>2</sub> atmosphere, has a considerable mean radiating height and a potent greenhouse effect (Mitchell, 2009). Manabe and Wetherald (1967) postulated that an increase of 2.3°C for a doubling of atmospheric CO<sub>2</sub>.

**FIGURE 1.3** Earth's atmospheric opacity to various wavelengths of electromagnetic radiation. The wavelength range of interest for climate science and atmospheric warming is from about 0.1 to 100 μm. (NASA, 2008).



The energy balance in the atmosphere comprehends mainly the estimates of incoming radiation 342 W m<sup>-2</sup>(I) Fig 1.4, the outgoing longwave radiation provided by satellite data (F) at the top of the atmosphere 235 W m<sup>-2</sup>, and the reflected 107 W m<sup>-2</sup> (E) solar radiation. The Earth has a mean radiating height of 5.5 km and a global mean lapse rate of 6 K km<sup>-1</sup>, giving a current greenhouse warming of 33 K (Berger and Tricot, 1992).

**FIGURE 1.4** Global energy balance based on an input of 342 Wm<sup>-2</sup> of solar radiation (NASA, 2022).



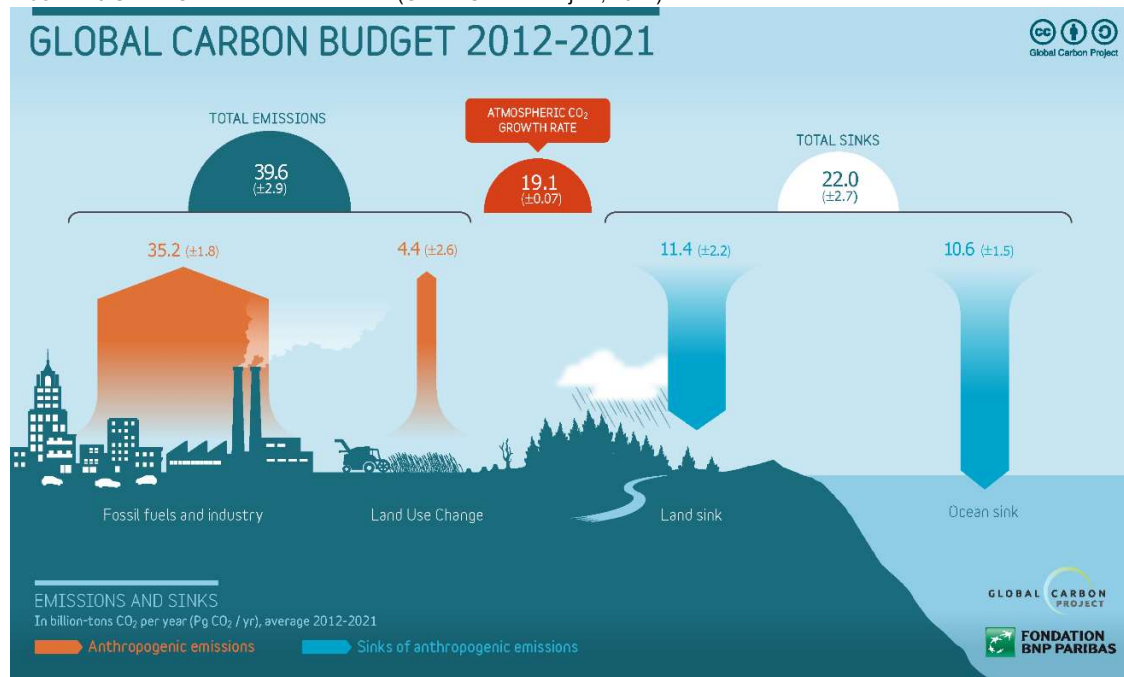


Any change in E or F, like that due to the increase in the atmospheric concentration of the GHGs, leads to a perturbation of the greenhouse effect (American Chemical Society, 2022; Pidwirny, 2006; Wei et al., 2018).

## Greenhouse recognition and role of inland waters

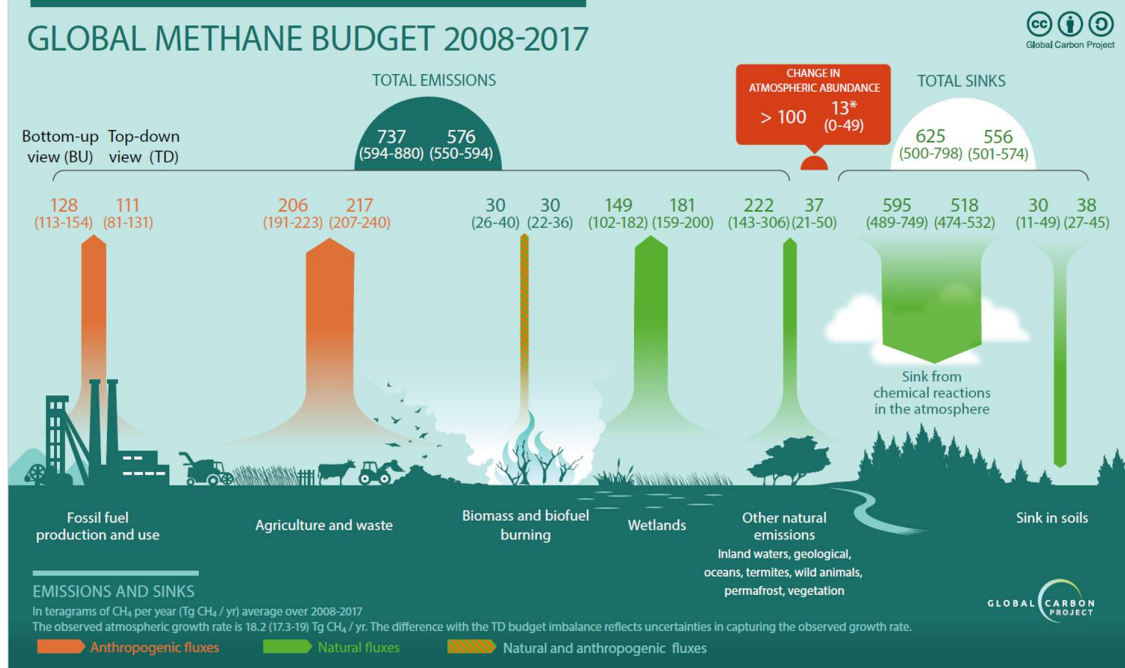
GHGs keep the Earth habitable and necessary for naturally happening atmospheric functions. In fact, without their presence, the expected effective temperature just with the Sun's radiation would be  $-19\text{ }^{\circ}\text{C}$  (Kweku et al., 2018). However, it was not until the mid-1950s that the term greenhouse effect was attached to climate alteration concerns (Mitchell, 1989). 20 years later with computing modelling came along, Manabe and his colleagues laid the foundations for predicting the climate response to GHGs emissions. At present, global average temperature has reached  $1.1\text{ }^{\circ}\text{C}$  above pre-industrial levels (European Commission, 2020) —anthropogenic activities mainly cause this. According to their contribution to the warming effect, we highlight the  $\text{CO}_2$  (74,4 %) that enters the atmosphere through the burning of fossil fuels, solid waste, wood, and other biological materials and because of specific chemical reactions such as respiration or cement production (Friedlingstein et al., 2022).  $\text{CO}_2$  is removed from the atmosphere by plants as part of the biological carbon cycle.

FIGURE 1.5 Global Carbon Dioxide Balance (Global Carbon Project, 2022).



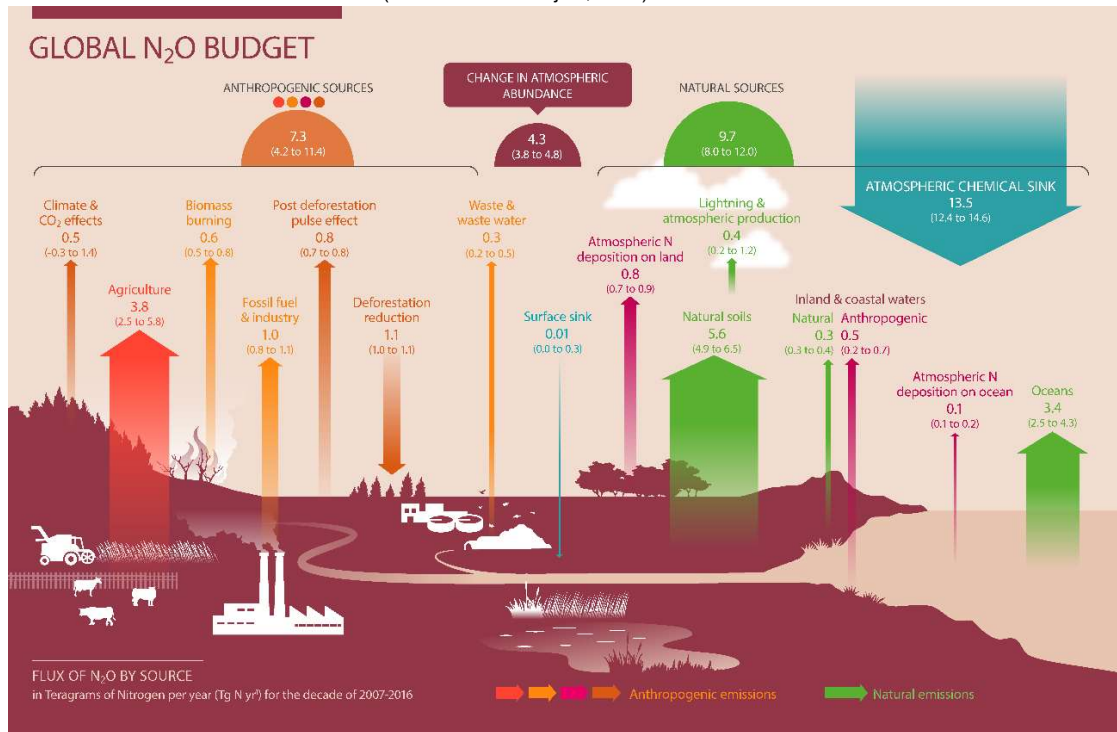
Then, we have the  $\text{CH}_4$  (17.3 %) emitted during coal, natural gas, and oil production and transport.  $\text{CH}_4$  emissions also result from livestock and other agricultural practices, land use, and the decay of organic matter in natural ways or landfills (Van Amstel, 2012; Kweku et al., 2018)

FIGURE 1. 6 Global Methane Balance (Global Carbon Project, 2018)



Finally, the N<sub>2</sub>O (6,7 %) is emitted during agricultural, land use, and industrial activities; combustion of fossil fuels and solid waste; as well as during treatment of wastewater. Together, the three account for ~98,4 % of global greenhouse gases (IPCC, 2021; EPA, 2020).

FIGURE 1. 7 Global Nitrous Oxide Balance (Global Carbon Project, 2021).



Total net anthropogenic GHG emissions have continued to rise from 2010–2019 up to  $59 \pm 6.6$  Gt CO<sub>2</sub>-eq, about 12% higher than in 2010 and 54 % higher than in 1990. Nevertheless, the average annual growth rate slowed from 2.1% yr<sup>-1</sup> between 2000 and 2009 to 1.3% yr<sup>-1</sup> between 2010 and 2019 (IPCC, 2022).

The ways how carbon (C) and nitrogen (N) interact between land, atmosphere, and oceans obey the pathways determined by their biogeochemical cycles, which correlate mutually since the productivity of plants and soil organisms strongly depends on N availability (Zaehle and Dalmonech, 2011; Schimel et al., 1997). The carbon cycle is a biogeochemical process by which C exchanges among the lithosphere, biosphere, hydrosphere, and atmosphere. The main C form exchanged with the atmosphere is CO<sub>2</sub>, whose concentration has increased from 227 ppm in 1750 to ~ 410 ppm in 2020. Initially, this increment was due to deforestation and other land-use change activities (Friedlingstein et al., 2022; House et al., 2002). Nevertheless, after the industrial revolution (Pongratz and Caldeira, 2012) fossil fuel burning became the primary anthropogenic C source.

The recognition of GHGs and climate change began in 1861 when physicist John Tyndall noted the influence of atmospheric absorption on the Climate (Tyndall, 1861). In 1896, Arrhenius highlighted that the selective absorption of the aqueous vapor and carbonic could substantially alter the surface temperature through the greenhouse effect (Arrhenius, 1896). On his part, Guy Callendar calculated that from ~1870 to ~1940, anthropogenic fuel combustion added about 150,000 million tons of CO<sub>2</sub> to the air and connected it to global warming (Callendar, 1938). In 1941, (Maslin, 2016) explained the fundamental causes of the ice ages and how humanity has disrupted the natural cycles. By 1956 Plass formulated the Carbon Dioxide Theory of Climate Change, predicting that the warming trend would continue, at least for several centuries (PLASS, 1956), and introduced the influence of temperature on the CO<sub>2</sub> equilibrium between oceans and atmosphere. By 1979, oceanographers consented that biota in oceans could be a sink (Broecker et al., 1979) and a CO<sub>2</sub> source for the atmosphere. Shortly after, limnologists came into play (Schlesinger and Melack, 1981) estimating the transport of dissolved and particulate organic carbon (DOC and POC) to the world's oceans by river flow. They emphasized the importance of these significant amounts of C stored in deep ocean sediments or released to the atmosphere for the global C cycle. Until then, studies revealed the importance of rivers in the C cycle because they enabled the transport of C (organic and inorganic) as a pipeline connection between land and ocean. This delivery of terrestrial C is partly the result of several transformations and losses in aquatic systems.

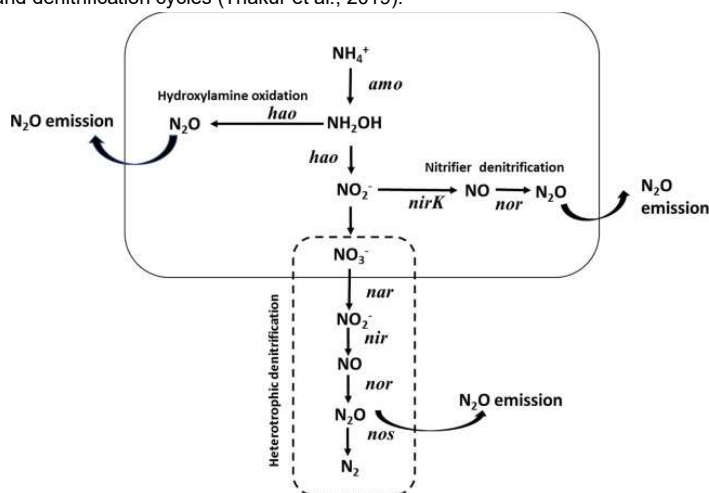
Nevertheless, they did not directly include the aquatic systems as a source of CO<sub>2</sub>. Instead, these approaches resulted in simplified depictions of the global C cycle that considered ocean and land connected through gas exchanges with the atmosphere (Siegenthaler and Sarmiento, 1993), i.e., the biota, photosynthesis, and respiration. However, they still overlooked the role of inland waters -as reactors- in addressing the global budget. By the 1990s, the modeling underwent significant improvements; for instance, (Schimel et al., 1997; Canadell et al., 2000) showed a strong correlation between C, N, and terrestrial water fluxes. A disruptive study (Cole et al., 2007) formulated a simplified mass balance to track the fate of C in an integrated freshwater and terrestrial budget. They hypothesized that the C imported annually to the global aquatic ecosystems (1.9 Pg C yr<sup>-1</sup>) could have three fates, i.e., CO<sub>2</sub> evasions to the atmosphere (0.75 Pg C yr<sup>-1</sup>), sediment storage in the river (0.23 Pg C yr<sup>-1</sup>), and transport to the oceans (0.9 Pg C yr<sup>-1</sup>). To explain this foretaste, they brought up the global role of inland aquatic systems in the C cycle pathways. They began implying that if the inland freshwaters were important CO<sub>2</sub> sources, the C should have been an active input from soil respiratory products or the oxidation of terrestrially derived organic carbon (OC) within the aquatic system (Cole et al., 2007). If this hypothesis was true, it was essential to differentiate between the inland water systems and their characteristics. For example, the type (rivers, lakes, reservoirs, and wetlands), the presence (permanent, seasonal covered), the latitude (tropical, temperate, and frozen), and the origin (raining, melting,

groundwater, headwaters), since different drivers can modify the mechanisms, velocity, and reaction yields within the processes of storage, evasion, and transport of C.

CH<sub>4</sub> is also critical for assessing realistic climate change pathways, given that it is the second most important GHG in terms of climate forcing. Its concentration is higher than ever, as observed in the ice-core record over the past 800,000 years (Van Amstel, 2012). Although global anthropogenic emissions of CH<sub>4</sub> are around 366 TgCH<sub>4</sub> yr<sup>-1</sup> (3% of the global CO<sub>2</sub> anthropogenic), it contributed ~ 25% (~0.62W m<sup>-2</sup>) to the radiative forcing accumulated in the lower atmosphere since 1750 (Friedlingstein et al., 2022). Agriculture and fossil fuels account for ~70 % of the anthropogenic CH<sub>4</sub>. Natural sources include wetlands, termites, oceans, hydrates, geological sources, wild animals, and wildfires. The first estimations of CH<sub>4</sub> emissions relied on 3D models, (Bergamaschi et al., 2007, 2005; Hein and Heimann, 1994; Houweling et al., 2000, 1999; Lelieveld et al., 1998; Tans et al., 1990) have all published analyses of the CH<sub>4</sub> budget based on measurements and model calculations. IPCC has summarized them in the Second, Third, and Fourth Assessment Reports. The bottom-up estimates issued by the Emissions Database for Global Atmospheric Research EDGAR (Wunch et al., 2009) and summarized by (Van Amstel, 2012) have contributed to the scientific literature since 1970. From 2012 onwards, bottom-up global anthropogenic inventories were available in a systematic record, which permitted (Friedlingstein et al., 2022) to elaborate on the Global Methane Budget 2000–2017, whose results are sorted by decades (2000 – 2009) 547 TgCH<sub>4</sub> yr<sup>-1</sup> and (2008 – 2017) 576 TgCH<sub>4</sub> yr<sup>-1</sup>.

Inland waters (lakes, ponds, reservoirs, streams, rivers) are natural sources of CH<sub>4</sub> (Wunch et al., 2009; Van Amstel, 2012; Bastviken et al., 2011; Borges et al., 2015c), through several pathways, such as ebullition from sediments, diffusive flux, and plant-mediated transport through emergent aquatic plants (Bastviken et al., 2011). Most fluvial systems show oversaturation of CH<sub>4</sub> with global emissions of 26.8 Tg CH<sub>4</sub> yr<sup>-1</sup>, equivalent to ~15-40% of wetland and lake effluxes, respectively (Stanley et al., 2015). Lakes and ponds prefer (3:1) ebullition rather than diffusive mechanisms to release CH<sub>4</sub> (Bastviken et al., 2011), mainly smaller and shallower lakes. Prevailing studies (Saunio et al., 2020) lead to a mean estimate of all inland freshwaters at 159 Tg CH<sub>4</sub> yr<sup>-1</sup> ranging from 117 to 212 Tg CH<sub>4</sub> yr<sup>-1</sup>, which suggests that lakes or natural ponds, flooded land–reservoirs, and streams–rivers account for 70 %, 13 %, and 17 % of the average inland water fluxes, respectively (Bastviken, 2022; Canning et al., 2021).

FIGURE 1. 8 Nitrification and denitrification cycles (Thakur et al., 2019).



N<sub>2</sub>O, like CO<sub>2</sub> and CH<sub>4</sub>, is a long-lived greenhouse gas accumulating in the atmosphere. The concentration of atmospheric N<sub>2</sub>O has increased by more than 20 % from 270 parts per billion (ppb) in 1,750 to 331 ppb in 2018 (Akshait, 2020). Two key biochemical processes—nitrification and denitrification—control N<sub>2</sub>O production in terrestrial and aquatic ecosystems and are regulated by multiple environmental and biological factors, including temperature, water and oxygen levels, acidity, and substrate availability (Butterbach-Bahl et al., 2013) Fig 1.8. Like CO<sub>2</sub> and CH<sub>4</sub>, the N<sub>2</sub>O emissions have been reported annually by IPCC. However, these inventories do not provide a complete picture of N<sub>2</sub>O emissions owing to their omission of natural sources, the limitations in methodology for attributing anthropogenic sources, and missing data on key regions like Africa or South America. Nevertheless, bottom-up and top-down approaches have improved to give consistent estimates for the decade between 2007 and 2016 ~ 17 Tg N yr<sup>-1</sup>. 23% of this value is not absorbed in natural sinks. Therefore, there is an accumulation of ~4 Tg N yr<sup>-1</sup>. Anthropogenic sources contributed, on average, 43 % of the total N<sub>2</sub>O emissions to the atmosphere (Butterbach-Bahl et al., 2013). Agricultural soil management is the largest source accounting for about 74 %.

When comparing the global annual GHG emissions from natural systems with those generated by human activity, results indicate that the global annual emissions range between 54.33 and 75.50 Gt CO<sub>2</sub>-eq, of which natural emissions account for 18.13–39.30 Gt CO<sub>2</sub>-eq, with the most likely value being approximately 29.07 Gt CO<sub>2</sub>-eq (Yue and Gao, 2018; Friedlingstein et al., 2022). The natural to anthropogenic-emissions ratio is approximately 0.8 (Yue and Gao, 2018); for both, interactions between the terrestrial C and N cycles shape the response of ecosystems to global change. Therefore, their accurate assessment of the pathways and dynamics among the atmosphere, ocean, and terrestrial biosphere is critical to developing policies and projecting future climate change (Friedlingstein et al., 2022).

## Sinks and sources

As the Earth's system has evolved, the global Climate has continuously changed. In recent decades, the issue of climate change, especially the relationship between human activities and global warming, has drawn widespread attention (Stocker et al., 2013; Yue and Gao, 2018). However, some different voices have questioned these conclusions. For example, (Barnett et al., 1999) suggested that changes in global Climate inferred from surface air temperature are likely, not due solely to anthropogenic causes. Instead, the atmospheric system must combine natural and human dynamics (sinks and sources) to explain the problem better. All these drivers were inputs to develop several modelings that simulate, every time more trustful, the evolution of the global average temperature in the 20th century (Baede et al., 2016). The natural inputs include solar variability, volcanic eruptions, and interactions within the Climate system.

Conversely, human inputs embrace the atmosphere's composition (including increases in the concentrations of greenhouse gases and aerosols) caused by industrial and social activities and changes in land use and coverage. It is worth nothing that sinks, and sources could change their storage and emission capacity according to the driving forces that control them. For instance, although emissions from fossil fuels started before the Industrial Era, they became the dominant source of anthropogenic emissions to the atmosphere from around 1950 (Friedlingstein et al., 2022; Pittock, 2013).

## **Natural sources**

### **Forest fires**

Forest fires are a significant source of GHG emissions, and their natural causes include drought, heat, and lightning. Approximately 90% of the gases emitted from forest fires are CO<sub>2</sub> (van der Werf et al., 2009). At the regional scale, boreal forest fires have received widespread attention (Ribeiro-Kumara et al., 2020). A comprehensive study based on a biogeochemical model and improved satellite-derived estimates of area burned, fire activity, and plant productivity to calculate fire emissions for 1997–2009. The study showed that most carbon emissions were from fires in grasslands and savannas (44 %), with smaller contributions from tropical deforestation and degradation fires (20 %), woodland fires (mostly confined to the tropics, 16 %), forest fires (mainly in the extratropic, 15 %), agricultural waste burning (3%), and tropical peat fires (3 %). The annual global GHG emissions from forest fires range between 1.9 to 4.3 GtC, with a most likely value of 3.0 GtC (van der Werf et al., 2010).

### **Wetlands**

Wetland soils contain some of the highest soil C stores in the biosphere. However, there needs to be more understanding of the quantity and distribution of C stored in the remaining wetlands or the potential effects of human disturbance on these stocks (Nahlik and Fennessy, 2016). Their C stocks account for approximately 15% of land surface carbon. Wetland CH<sub>4</sub> emissions are the most significant natural source in the global CH<sub>4</sub> budget, contributing to roughly one-third of natural and anthropogenic emissions (Zhang et al., 2017). Existing studies indicate that they range from 0.65 to 2.04 Gt C, with the most likely value being approximately 1.36 GtC (Friedlingstein et al., 2022; Saunio et al., 2020).

### **Permafrost**

Carbon stored in permafrost regions is one of the least understood and potentially most significant carbon-climate feedbacks because of the size of the carbon pools and the intensity of Climate forcing at high latitudes. Permafrost distributes mainly in Siberia, Canada, and Alaska, with the global permafrost area totaling approximately 18,782 x 10<sup>6</sup> km<sup>2</sup> (Tarnocai et al., 2009). Methane hydrates have been found in continental shelves and below sub-sea and land permafrost. This CH<sub>4</sub> can be either biogenic or thermogenic (Saunio et al., 2020). Research on GHG emissions from permafrost has focused on C, CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O, many of which were flux observations. For example, (Koven et al., 2011) simulated CH<sub>4</sub> emissions under different scenarios for 1860 to 2100 in high latitudes for the three major frozen regions. They found that from the end of the 19th century to the beginning of the 21st century, annual CH<sub>4</sub> emissions at high latitudes remained at 30 to 40 TgC, but this value could triplicate by 2080. The mode values for the annual GHG emissions from permafrost range between 0.6 and 2.1 GtC, with the most likely value being approximately 1.6 GtC per year (Yue and Gao, 2018).

### **Volcanoes**

Volcanic eruptions can release many volcanic gases into the atmosphere, including CH<sub>4</sub> and CO<sub>2</sub>. The CO<sub>2</sub> released directly from active volcanoes has three primary sources, dissolved in the mantle, recycled from subducted crustal material, and decarbonation of shallow crustal material (Burton et al., 2013). The level of GHG emissions from volcanic eruptions could be very high. For example, the eruption of the Pinatubo volcano in the Philippines in June 1991 released more than 20 Mt SO<sub>2</sub> and 200 Mt CO<sub>2</sub> into the atmosphere. It formed the largest volcanic cloud event of the 20th century. McCartney et al. (1990) estimated the CO<sub>2</sub> emissions from the Tegan basalt at the peak of eruption (1.36 Ma) to be 13,000 Gt. Intermittent volcanoes release gas much more slowly but last longer, so the total amount of GHGs released by intermittent volcanoes could be more significant than those from eruptions (Burton et al., 2013; Zhang et al., 2019). The GHG

emissions from global volcanic eruptions are estimated to range between 5.45 and 30 Mt per year, with an interphase of approximately 4.5. Hence, the annual total GHG emissions from volcanoes range between 187 and 277 Mt CO<sub>2</sub>-eq (Mccartney et al., 1990).

### **Mud volcanoes**

They are the eruption of mud, and the mechanisms and drivers that govern eruptions still need to be better understood. During their formation and subsequent activity, they can also produce large amounts of hydrocarbon gas (Huff and Owen, 2015). GHG mud volcanoes primarily emit CH<sub>4</sub>, which accounts for 95 % of the total hydrocarbon gas and small amounts of CO<sub>2</sub>. GHGs released during a mud volcano eruption are estimated to be approximately 60% to 80% of the total gas content. Large amounts of gas can also be released during periods of dormancy. Based on the mode values, the global annual CH<sub>4</sub> emissions from mud volcanoes are estimated to range between 0.5 and 6.15 Mt, equivalent to 0.03 to 0.43 MtC. The total GHG emission from mud volcanoes is approximately 200 MtC annually (Shi et al., 2016).

### **Earthquakes**

Earthquakes are another source of GHGs, including the immediate release of CO<sub>2</sub> from the Earth and that from the decay of plants and animals buried after earthquake events (YUE and GAO, 2018). (Saito et al., 2015) estimated the total GHGs from the Tohoku earthquake to be 5.1 Pg C. However, researchers have rarely approximated earthquake GHG.

In summary, global GHG emissions from natural systems range from 18.13 to 39.30 Gt CO<sub>2</sub>-eq per year, with a most likely value of 29.07 Gt CO<sub>2</sub>-eq. Forest fires account for the most significant component of total emissions at 37.8%, followed by oceans, permafrost, and wetlands at 21.05 %, 20.64 %, and 17.20 %, respectively. Volcanoes and mud volcanoes contribute relatively low GHG emissions, i.e., approximately 1 to 3 % of the total.

## **Anthropogenic sources**

According to the GHG inventory data for global human activity compiled by the UN, CO<sub>2</sub> emissions from global fossil fuel usage and cement production rose from 22.3 Gt in 1990 to 36.2 Gt in 2016, representing a growth rate of approximately 70%. The estimates of global and national fossil CO<sub>2</sub> emissions include the oxidation of fossil fuels through both combustion (e.g., transport, heating) and chemical oxidation (e.g., carbon anode decomposition in aluminum refining) activities, and the decomposition of carbonates in industrial processes (e.g., the production of cement) (Friedlingstein et al., 2022; Saunio et al., 2020).

### **Fossil CO<sub>2</sub> emissions**

Cumulative fossil CO<sub>2</sub> emissions for 1850–2021 were 465 GtC, including the cement carbonation sink. In this period, 46 % of fossil CO<sub>2</sub> emissions came from coal, 35 % from oil, 15 % from natural gas, 3% from the decomposition of carbonates, and 1 % from flaring. Global fossil CO<sub>2</sub> emissions were 5.1 % higher in 2021 than in 2020 because of the global rebound from the worst COVID-19 pandemic, with an increase of 0.5 GtC to 9.9 GtC (including the cement carbonation sink) in 2021 (Friedlingstein et al., 2022). Since the 1960s, there has been a noticeable increase in GHG emissions from coal, oil, and gas combustion, which have increased from 5 Gt to 14.5 Gt, from 3 Gt to 12.5 Gt, and from 0.8 Gt to 7.0 Gt, respectively. Meanwhile, GHG emissions from oil increased significantly from 1960 to 1980, and GHG emissions from coal combustion showed significant growth at the beginning of the 21st century but have decreased slightly since 2014. In 2021, the most considerable absolute contributions to global fossil CO<sub>2</sub> emissions were from China (31 %), the USA (14 %), the EU27 (8 %), and India (7 %). These four regions account for 59% of global CO<sub>2</sub> emissions, including international aviation and marine bunker fuels (2.8 % of the total) (Jiang and Green, 2017; Massetti, 2011).

## Land use changes

Cumulative CO<sub>2</sub> emissions from land-use changes (ELUC) for 1850–2021 were 205 GtC (Friedlingstein et al., 2022). The cumulative emissions from ELUC show a large spread among individual estimates of 140 GtC (updated in 2017), 280 GtC (BLUE), and 190 GtC (OSCAR) for the three bookkeeping models and a similar comprehensive estimate of 185 GtC for the Dynamic Global Vegetation Model (DGVM) (all cumulative numbers rounded the 5 GtC) (Friedlingstein et al., 2022). These estimates are broadly consistent with indirect constraints from vegetation biomass observations, giving a cumulative source of 155 GtC over the 1901–2012 period (Li et al., 2017). However, given the large spread, the best estimate is difficult to ascertain. In contrast to growing fossil emissions, CO<sub>2</sub> emissions from land use, land-use change, and forestry have remained relatively constant over the 1960–1999 period but show a slight decrease of about 0.1 GtC per decade since the 1990s, reaching 1.2 to 0.7 GtC yr<sup>-1</sup> for 2012–2021 (Hong et al., 2021).

## Total anthropogenic emissions

Cumulative anthropogenic CO<sub>2</sub> emissions for 1850–2021 totaled 670 GtC, of which 70 % (470 GtC) occurred since 1960 and 33 % (220 GtC) since 2000. Total anthropogenic emissions more than doubled over the last 60 years, from  $4.5 \pm 0.7$  GtC yr<sup>-1</sup> for the decade of the 1960s to an average of  $10.8 \pm 0.8$  GtC yr<sup>-1</sup> during 2012–2021 and reaching  $10.9 \pm 0.9$  GtC in 2021. During 1850–2021, 30 % of historical emissions were from land-use change and 70 % from fossil emissions. However, fossil emissions have grown significantly since 1960, while land-use changes have not. Consequently, the contributions of land-use change to total anthropogenic emissions were smaller during recent periods, 11 % during 2012–2021 (Hong et al., 2021).

## Natural sinks

### Atmosphere

The mass of carbon in the atmosphere increased by 48% from 1750 to 2021. Current CO<sub>2</sub> concentrations in the atmosphere are unprecedented in the last 2 million years, and the current rate of atmospheric CO<sub>2</sub> increase is at least 10 times faster than at any other time during the last 800,000 years (Van Amstel, 2012). In fact, concentrations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O increased from 1750 to 2019 by  $131.6 \pm 2.9$  ppm (47.3%),  $1137 \pm 10$  ppb (156%), and  $62 \pm 6$  ppb (23.0%) respectively (Friedlingstein et al., 2022). The growth rate in atmospheric CO<sub>2</sub> level increased from  $1.7 \pm 0.07$  GtC yr<sup>-1</sup> in the 1960s to  $5.2 \pm 0.02$  GtC yr<sup>-1</sup> during 2012–2022 with important decadal variations.

### Ocean

The carbon content of the oceans is 50 times that of the atmosphere and 20 times that of the biosphere. Global oceans serve simultaneously as a C source and sink, and their substantial thermal mass partially delays the global response to forcings (Ribeiro-Kumara et al., 2020). Since 1850, the ocean sink has added up to  $175 \pm 35$  GtC; more than two-thirds of this amount (120 GtC) has accumulated since 1960 (Kenneth L & Guy, 2016). Anthropogenic emissions represented close to 26% of this total. Besides the concentrations, the growth rate also increased; for example, by the decade 1960s, the average rate was  $1.1 \pm 0.4$  GtC yr<sup>-1</sup>. Conversely, during 2012–2021, it increased to  $2.9 \pm 0.4$  GtC yr<sup>-1</sup> with interannual variations of a few tenths of a GtC yr<sup>-1</sup>. The increased atmospheric CO<sub>2</sub> concentration primarily drives the increase in the ocean sink; nevertheless, other drivers, such as La Niña conditions, expect to weak the sink capacity (McKinley et al., 2017).



## Land

Since 1850, the terrestrial CO<sub>2</sub> sink has amounted to 31% of total anthropogenic emissions. Over the historical period, the sink increased with the exponential anthropogenic emissions increase. Over the period from 1960 to the present, the increase in the global terrestrial CO<sub>2</sub> sink is mainly due to the CO<sub>2</sub> fertilization effect. The terrestrial CO<sub>2</sub> sink from the DGVMs ensemble was  $3.5 \pm 0.9$  GtC in 2021, slightly above the decadal average of  $3.1 \pm 0.6$  GtC yr<sup>-1</sup> (Friedlingstein et al., 2022)

## Exchanges between reservoirs

Exchanges are movements called a flux or flow. Sources of carbon are releases from a particular reservoir to the atmosphere, and sinks are places where carbon is removed (taken up) from the atmosphere to another reservoir (Keller et al., 2018).

### The atmosphere and the terrestrial biosphere

When carbon that is already in the short-term carbon cycle changes form or flows to a different reservoir (for example, moving from the land to the atmosphere when organic matter is burned), the total amount of carbon in the cycle remains the same. The short-term carbon cycle is roughly balanced, and the amount of carbon-containing greenhouse gases is steady. Fossil fuels, on the other hand, are extracted from underground, so they are not a natural part of the short-term cycle. Thus, when fossil fuels are burned, carbon from long-term storage is added to the short-term cycle. When this happens, not only does the greenhouse effect increase because there are more carbon-containing greenhouse gases in the atmosphere, but the balance of the cycle is disrupted. Until about 150 years ago, fossil fuels were not an important part of the short-term cycle. Now, they must be included in any consideration of this cycle. The fossil-fuel reservoir is estimated at 5,000 Gt C—almost eight times the amount that is currently in the atmosphere. Current fossil-fuel emissions are about 6.7 Gt C per year (Keller et al., 2018).

How do we know that the increased CO<sub>2</sub> in the atmosphere comes from fossil-fuel burning and other human actions rather than from natural processes? An important indicator is provided by the isotopes (different chemical forms) of carbon. Compared to atmospheric carbon, carbon originating in the terrestrial biosphere—including fossil-fuel carbon, which started life as plant material—is also low in the isotope  $\delta^{13}\text{C}$  relative to  $\delta^{12}\text{C}$ . Measurements from ice cores show a general decrease over the last century in atmospheric levels of  $\delta^{13}\text{C}$  relative to  $\delta^{12}\text{C}$ , indicating that the increased carbon comes from either fossil-fuel burning, biomass burning, or both (Ferrio et al., 2003; Staddon, 2004).

### Oceans and the atmosphere

The role of the oceans in the global carbon cycle is twofold. First, they absorb excess atmospheric CO<sub>2</sub>, and second, changes in the physical, chemical, and biological state of the oceans may affect future atmospheric CO<sub>2</sub>. The ocean absorbs excess CO<sub>2</sub> in a three-step process: the CO<sub>2</sub> gas passes through the air-sea interface, it reacts chemically with dissolved inorganic carbon in the ocean's surface waters, and then it is transported into the deeper waters (Siegenthaler and Sarmiento, 1993; Lars J. Tranvik et al., 2009) Oceanic circulation is important in the carbon cycle because it affects carbon storage in the ocean. The ocean waters circulate mainly because of three factors: winds, which create surface currents; differences in saltness throughout the oceans (caused by differences between evaporation and precipitation as well as by freezing and thawing); and differences in water temperature. CO<sub>2</sub> is more soluble in cold water than in warmer water, so surface water from the tropics continues to absorb additional CO<sub>2</sub> as it moves toward higher latitudes and colder temperatures (Sulzman, 2000). In the rainy North Atlantic, the addition of fresh water—which weighs less than saltwater—pushes the more saline

water (and the extra CO<sub>2</sub>) toward the ocean floor. Only the movement of carbon from shallow to deeper waters in the conveyor-belt circulation allows large amounts of CO<sub>2</sub> to be stored in the oceans. The surface water has a limited capacity to absorb CO<sub>2</sub>. It will even emit CO<sub>2</sub> to the atmosphere if the local atmospheric concentration is lower than the concentration in the surface water.

Chemical processes in the ocean also affect carbon storage. When CO<sub>2</sub> enters the ocean, it reacts chemically with the water to form carbonate and bicarbonate ions. (Ions are chemical compounds of atoms that have gained or lost electrons, leaving them with a positive or negative charge.) As more of these ions are formed, more atmospheric CO<sub>2</sub> can enter the ocean. This reaction depends on temperature, the amount of dissolved inorganic carbon already in the surface water, and, to a lesser extent, the water's salinity and alkalinity (opposite of acidity) (Sulzman, 2000; Farquhar et al., 2001).

Biological processes in the oceans are driven by the marine biota (oceanic plants and animals). The marine biota is believed to play a minor role, if any at all, in the oceans' ability to absorb the CO<sub>2</sub> added to the carbon cycle by humans. In fact, marine biota contributes only about 3 Gt C to the world's total carbon balance. The marine biota plays a crucial role, however, in keeping the level of atmospheric CO<sub>2</sub> steady. Phytoplankton (microscopic ocean plants) take up dissolved inorganic carbon through photosynthesis, just as terrestrial plants take up atmospheric CO<sub>2</sub> during photosynthesis. If that dissolved inorganic carbon remained in the ocean waters, the oceans could not absorb as much CO<sub>2</sub> from the atmosphere as they do. In fact, if there were no oceanic photosynthesis, the atmospheric concentration of CO<sub>2</sub> would be approximately 520 ppm—much higher than nowadays level of 400 ppm. In the open ocean, there is more carbon available than the marine biota use for photosynthesis. Scientists think that phytoplankton populations do not use this extra carbon to grow and multiply either because they don't have enough other nutrients and/or light or because zooplankton (minute ocean animals) graze on them and keep their numbers down (Sarmiento and Gruber, 2002). If the nutrients in the ocean were evenly distributed in areas suitable for phytoplankton to live, the increase in photosynthesis would cause today's atmospheric CO<sub>2</sub> concentration to drop to only 230 ppm. Besides using dissolved inorganic carbon for photosynthesis, the marine biota uses it to manufacture both inorganic compounds (e.g., shells) and organic matter, such as cell tissue. As the organisms die and sink, the organic material falls into the deeper ocean, where it is decomposed and converted back to dissolved inorganic carbon. This mechanism of carbon transport to the deep ocean is called the biological pump. If the ocean takes up more atmospheric CO<sub>2</sub>, thus increasing the amount of dissolved inorganic (Sulzman, 2000).

### Terrestrial biosphere and oceans

Carbon moves from the terrestrial biosphere to the oceans indirectly via the global hydrologic cycle (through evaporation from land and rain over the oceans) (Sulzman, 2000), and directly via rivers. This riverine delivery is, in fact, the result of several transformations and losses in aquatic systems en route. River export of organic carbon to the sea ranges from 0.38 to 0.53 Pg C yr<sup>-1</sup>, whereas export of dissolved inorganic carbon resulting from the fixation of atmospheric carbon through rock weathering is likely to be between 0.21 and 0.3 Pg C yr<sup>-1</sup>. Collectively, using mid-range values for the river and groundwater components inland waters thus deliver about 0.9 Pg C yr<sup>-1</sup> to the oceans (Cole et al., 2007).

## **The role of inland waters in the global and regional carbon balance**

The C budget is simply one of the reallocations of C among pools. The only input to the organic pool is from Gross Primary Production (GPP); the return to CO<sub>2</sub> comes from respiration

(R) and abiotic oxidations. Storage (S) is simply the balance between GPP and oxidations. Imports (I) and exports (E) are irrelevant at this scale; the mass balance of C occurs entirely within the boundaries of the global ecosystem. Any change in S has a concomitant change in CO<sub>2</sub> (Cole, 2013).

Net ecosystem production (NEP) can be determined by:

$$NEP = GPP - R = E + \Delta s - I$$

Eq. 1

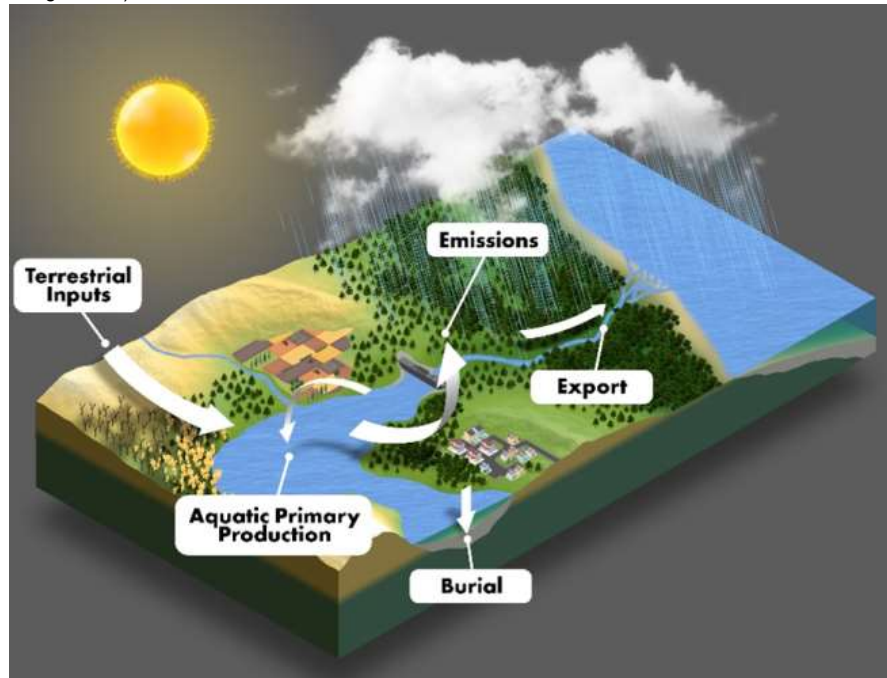
Also important is the Net primary production (NPP), equal to GPP – R<sub>a</sub> and is the fraction of primary production potentially available to consumers. Finally, R<sub>h</sub> is the respiration of heterotrophs, including all consumers and microorganisms.

$$R = R_h + R_a$$

Eq. 2

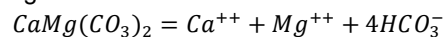
For inland waters, the interaction between CO<sub>2</sub>, CH<sub>4</sub>, and O<sub>2</sub> is essential to track, and there are five pathways shown in **Fig 1.9** with significant fluxes to consider.

**FIGURE 1.9** Carbon pathways at inland waters. About 70% of the C introduced in the water body is emitted GHG. Adapted from (Oak Ridge, 2020)



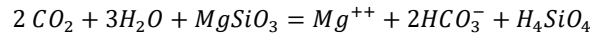
The export of dissolved bicarbonate from land to water represents 'hidden' terrestrial respiration (Cole, 2013). The HCO<sub>3</sub><sup>-</sup> ion dominates the dissolved inorganic C (DIC) that enters a water body rather than free CO<sub>2</sub>. Dissolved bicarbonates are formed principally by dissolving carbonate and aluminosilicate minerals with CO<sub>2</sub> and water. The CO<sub>2</sub> involved in these reactions comes mainly from the respiration of the terrestrial biosphere.

Carbonate weathering



Eq. 3

Silicate weathering



**Eq. 4**

Large CO<sub>2</sub> fluxes from inland waters significantly impact our understanding of overall C fluxes and cycling at the landscape or catchment scale. The respiration of upland and wetland vegetation is part of the CO<sub>2</sub> emitted to the atmosphere from inland waters. Whether it is related to lateral transport of soil/wetland CO<sub>2</sub> (that is, respiration taking place in terrestrial or wetland habitats) or lateral transport of soil or wetland DOC and POC mineralized to CO<sub>2</sub> within the aquatic domain. Hence, the lateral transport of C from the upland terrestrial biosphere and wetlands to inland waters and its subsequent emission to the atmosphere offsets the estimates of terrestrial NEP. Furthermore, the lateral transport of soil DOC leads to an underestimation of NEP based on biomass accumulation techniques. In contrast, the lateral transport of soil CO<sub>2</sub> leads to an overestimation of NEP derived from atmospheric measurements (eddy-covariance or flux towers) (Cole, 2013).

Terrestrial net ecosystem production (NEP) can enter aquatic ecosystems as DOC and POC. Both phases form a continuum along physical, chemical, and biological properties with continuous interactions. Revised carbon fluxes of the major world rivers identify the flux of terrestrial DOC (0.25 Pg C yr<sup>-1</sup>) as the most significant transfer of reduced carbon from the land to the ocean, with POC export estimated at 0.18 Pg C yr<sup>-1</sup> (J. A. Downing et al., 2008). Nevertheless, carbon fluxes into and out of inland waters have received little attention in global-scale analyses, and their presence is not totally integrated into traditional carbon budgets. Their quantification could prove critical for constraining estimates of terrestrial ecosystem fluxes, adequately integrating all vertical and lateral carbon fluxes over regional and global scales and predicting feedback to climate change (Ciais et al., 2008). For example, (Battin et al., 2009) highlighted that the 'conventional carbon cycle' pool together outgassing from inland waters with fluxes of terrestrial ecosystem respiration and underestimates the potential for lateral transport.

Nevertheless, terrestrially sourced OC can also fuel secondary production by heterotrophic biota in inland waters. Globally, that biota respire 1.2 Pg of terrestrial C each year and releases it to the atmosphere (Battin et al., 2009). In contradiction to the conventional belief that terrestrial OC is recalcitrant and contributes little to the support of aquatic metabolism, (J. A. Downing et al., 2008) suggest that the metabolism of terrestrial OC in freshwater ecosystems is responsible for a large amount of CO<sub>2</sub> outgassing to the atmosphere. Inland waters — ponds, lakes, wetlands, streams, rivers, and reservoirs — permeate terrestrial ecosystems and often shape the Earth's landscapes. Although only about 1% of the Earth's surface is assumed to be covered by inland waters, their collective contribution to global carbon fluxes is substantial compared with terrestrial and marine ecosystems (Battin et al., 2009).

Natural ecosystems must account for the Earth's energy budget because they exchange large amounts of CO<sub>2</sub> with the atmosphere and currently offset ~ 4Pg Cyr<sup>-1</sup> of anthropogenic emissions (Raymond et al., 2013). CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O transfer from inland waters to the atmosphere, known as GHG evasions, are components of the global carbon and N cycles. (Kempe, 1984) showed that CO<sub>2</sub> in inland waters calculated from alkalinity and pH were substantially higher than atmospheric values. Anthropogenic driving forces can seriously alter this natural pattern. For instance, the increase of erosion following deforestation and agriculture removes C from the biosphere into inland waters deposition centers, lakes, and reservoirs and causes the OC load of rivers to increase; organic sewage and eroded soil carbon fuel respiration in inland waters; and nutrient release provokes eutrophication mainly in lakes and coastal estuaries, resulting in enhanced primary productivity and CO<sub>2</sub> sequestration. (Raymond et al., 1997) have directly measured the partial pressure of CO<sub>2</sub> in the tidal-freshwater portion of the Hudson River Estuary over 3.5 years. At all times, the Hudson was supersaturated in CO<sub>2</sub> (1125 ± 403 SD) concerning the atmosphere averaged (416 ± 68) μatm. By 2007 this study (Cole et al., 2007) recognizes that inland waters are active components of the global C cycle that store

terrestrially derived carbon in sediments and lose CO<sub>2</sub> as emissions to the atmosphere. A conservative analysis suggests that inland waters annually receive ~ 1.9 Pg C yr<sup>-1</sup> from the terrestrial landscape, of which about 0.2 ends in aquatic sediments, 0.8 returns to the atmosphere as gas exchange, and finally, The remaining 0.9 Pg yr<sup>-1</sup> reaches the oceans. The global estimates of CO<sub>2</sub> outgassing fluxes rely on pCO<sub>2</sub> concentrations, *k* (a parameter related to the physics that determines the rate of gas exchange), and areal extent of inundation per unit area of surface water – along with a more balanced consideration of the tropics and wetlands. (Aufdenkampe et al., 2011) compiled estimates of the global flux of CO<sub>2</sub> outgassing from inland surface waters that range from 0.75 to 1.4 PtC yr<sup>-1</sup> globally. These values are essential when compared with estimates of net C accumulation on continents (2.2 Pg C y<sup>-1</sup>) or in the oceans (Cole et al., 2007; Battin et al., 2009).

By the 2010s' the CO<sub>2</sub> data distribution was skewed towards temperate and boreal systems in the Northern Hemisphere. Data in several tropical basins (including Africa and Amazon) were derived from interpolation from adjacent basins rather than actual measurements. In this context, (Borges et al., 2015) brought out the role of tropics in the global C cycle, indicating that global CO<sub>2</sub> emission from rivers and streams of 1.8 PgC y<sup>-1</sup> is related chiefly to tropical areas that account for 1.4 PgC y<sup>-1</sup> (78%). Similarly, about 49% of the CH<sub>4</sub> emission to the atmosphere from freshwater ecosystems occurs in the tropics. However, there is equally a substantial under-representation since most global syntheses resorted to extrapolating CH<sub>4</sub> fluxes from temperate rivers. Besides the under-representation of tropical waters, the prevailing significant uncertainty involved in GHG flux estimates for inland waters, essentially due to the paucity of available data, is coupled with a poor understanding of underlying processes (Borges et al., 2015), both of which preclude gauging of future fluxes in response to human pressures. There is a need to understand further the link between inland water GHG fluxes and catchment characteristics, regarding their connectivity with upland terrestrial, wetland C production and stocks (Borges et al., 2015c), gas transfer velocity (*k*), and high-quality direct CO<sub>2</sub> measurements.

## GHG drivers

Studies in rivers and streams have also progressed. Regional studies have attempted a more systematic evasion estimation thanks to developing high-resolution remote sensing, scaling computational tools, partial pressure direct measures, and water surface satellite mapping. A strong positive correlation exists between stream and river surface area and precipitation; conversely, there is a weaker negative relationship between surface area and temperature. The former is driven mainly by a strong positive correlation between stream width per Strahler stream order and precipitation. (Raymond et al., 2013) calculated the global CO<sub>2</sub> emissions from inland waters. They found that globally, a 0.07 % increase in the fraction of stream area for a 10-cm increase in precipitation and a 0.02 % decrease with an increase in temperature by 1 °C. Combining lakes and reservoirs with streams and rivers provides a total surface area of inland waters of 3 620,000 km<sup>2</sup>. The estimate of the surface area does not include wetlands.

The gas exchange velocity between water and atmosphere depends on the gradient concentration of air/water and the physical processes at the interface, usually parameterized as a gas transfer velocity (*k*) (Melack, 2017), also called a piston velocity or gas exchange coefficient. Gas transfer velocities are a function of turbulence, the water's kinematic viscosity, and the gas's molecular diffusion coefficient; the Schmidt number is the ratio of the latter two terms and is gas specific. Despite numerous studies on the subject, significant uncertainty remains regarding the parameterization of *k*<sub>600</sub> in lakes. The widely used parameterizations based on wind speed do not account for other drivers (or inhibitors) of turbulence at the interface, such as the friction of the current on the bottom evaporative heat gain and loss and the associated buoyancy fluxes, or the release of organic molecules that might behave as surfactants by the vegetation (Abril et al., 2014; MacIntyre et al., 2010). With increasing stream order, the global average gas transfer velocity of 5.7m d<sup>-1</sup> for streams and rivers increases. In a new metadata analysis of whole-stream tracer releases in streams and small rivers, the average value was 4.7md<sup>-1</sup>. The wind speed and

lake size models provided global average estimates of 0.74 and 1.33md<sup>-1</sup>, respectively. Thus, the global average gas transfer velocity for lakes and reservoirs is approximately 1.0md<sup>-1</sup>, much lower than the global average for streams and rivers but is consistent with regional studies (McDonald et al., 2013; Raymond et al., 2013).

### Global GHGs evasions from inland waters.

Climate predictions necessitate a complete and robust account of natural and anthropogenic greenhouse gas (GHG) fluxes, especially for CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O. (Borges et al., 2015c). (Raymond et al., 2013) calculated the global carbon emissions by compiling the 6 708 stream and river sampling location results. The average of these median values was 2,300 μatm, which increases to an average pCO<sub>2</sub> of 3,100 μatm after discounting for potential biases in the calculation and normalizing interpolated pCO<sub>2</sub> from each region to stream. Similarly, they assembled 20,632 pCO<sub>2</sub> observations from 7,939 lakes and reservoirs. They found that Inland waters present supersaturation of CO<sub>2</sub> concerning equilibrium with the atmosphere. At the same time, the concentration of CO<sub>2</sub> in water did not strongly relate to climatic or landscape variables for rivers and streams. Based on pCO<sub>2</sub>, three groups of lakes distinguish non-tropical freshwater, tropical, and saline lakes.

The study conducted by (Raymond et al., 2013) used two methods to estimate the gas transfer velocity for lakes and reservoirs. The first uses globally gridded wind speed and an empirical relationship between k600. The second employs new estimates of the gas transfer velocity for lakes of different sizes, which assumes a primary role of fetch, the distance traveled by the wind over the water, in regulating k in these systems. The wind speed and lake size models provided global average estimates of 0.74 and 1.33md<sup>-1</sup>, respectively.

Global CO<sub>2</sub> and CH<sub>4</sub> emissions were estimated at 2.1 PgC yr<sup>-1</sup> (Raymond et al., 2013) and 0.7 PgC yr<sup>-1</sup> (Bastviken et al., 2011) (CO<sub>2</sub>-equivalents; CO<sub>2e</sub>), respectively. The global C budget's GHG flux from inland waters is significant, considering that the oceanic and land C sinks correspond to ~1.5 and ~2.0 PgC yr<sup>-1</sup>. Over the past half-century, human activities have doubled the reactive N addition on Earth, resulting in saturation of the N assimilative capacity of some terrestrial ecosystems. This fact has led to a corresponding approximate doubling of riverine N loads. As a result, N<sub>2</sub>O emission rates from rivers have increased, with an estimated 0.3 – 2.1 Tg N<sub>2</sub>O–N yr<sup>-1</sup> or 6–30% of global anthropogenic N<sub>2</sub>O emissions. N<sub>2</sub>O emissions from rivers have been dramatically revised downwards from early estimates of 2,100 GgN<sub>2</sub>O–N yr<sup>-1</sup> to 32 GgN<sub>2</sub>O–N yr<sup>-1</sup> (Kroeze et al., 2010; Hu et al., 2016).

Approximately 70% of the stream CO<sub>2</sub> evasion originates from waters on only 20% of the total stream surface area, whereas 50% of the emissions are from the world's largest lakes, owing to their large surface area and gas transfer velocity. However, large lakes must be more adequately surveyed for concentration and k. The role of wetlands could be significant in hotspots such as Amazonia and Southeast Asia. Systematic campaigns are needed to evaluate further the relative importance of different sources (Abril et al., 2014; Raymond et al., 2013). CH<sub>4</sub>, expressed as CO<sub>2</sub> equivalents, corresponds to at least 25% of the estimated terrestrial GHG sink (2.6 ± 1.7 Pg of C yr<sup>-1</sup>). Finally, N<sub>2</sub>O, responsible for ~9% of global Climate radiative forcing, has a global warming potential of ~310 times that of CO<sub>2</sub>.

Unlike CO<sub>2</sub> and CH<sub>4</sub>, N<sub>2</sub>O has a more significant anthropogenic origin (Hu et al., 2016). Riverine N<sub>2</sub>O fluxes varied by dominant land-use type with higher fluxes (μg N<sub>2</sub>O–N m<sup>-2</sup> h<sup>-1</sup>) observed in agricultural (25.1 ± 80.6, n = 97) and residential rivers (23.3 ± 9.6, n = 31) and lower values observed in forest rivers (6.7 ± 17.9, n = 41), which is consistent with (Borges et al., 2018), suggesting that the available N level mainly determines the magnitude of N<sub>2</sub>O emission fluxes in river systems. The high dependence of N<sub>2</sub>O flux on both NH<sub>4</sub><sup>+</sup> and NO<sub>3</sub> levels further implies that N<sub>2</sub>O emission from rivers derives from nitrification and denitrification processes. Due to the role of both nitrification and denitrification, N<sub>2</sub>O emission fluxes demonstrated a higher correlation with

DIN concentrations than with either  $\text{NH}_4^+$  or  $\text{NO}_3^-$  concentrations alone. All those mechanisms intensify in small rivers due to the proximity of N sources (Hu et al., 2016; Maavara et al., 2019).

## **The Amazon's role in GHG emissions from inland waters**

Most riverine GHG emissions are from the tropics, ~60-80 % for  $\text{CO}_2$ , ~70 % for  $\text{CH}_4$ , and ~79 % for  $\text{N}_2\text{O}$ . However, the total stream surface area (SSA) in the tropics (24°N-24°S) corresponds to ~44 % of the total SSA globally, implying that riverine areal emissions of GHGs (per  $\text{m}^2$ ) are higher than in other climatic zones. However, the  $\text{CO}_2$  data distribution is skewed towards temperate and boreal systems in the Northern Hemisphere. Data in several tropical basins (including the Congo and Amazon) originated from interpolation from adjacent basins rather than actual measurements.

Although the magnitude and variability of these fluxes remain uncertain, especially in tropical regions, recent studies are improving our understanding of C dynamics in the streams, rivers, lakes, reservoirs, and wetlands of the Amazon basin (Abril et al., 2014; Borges et al., 2015a; Melack, 2017). The Amazon basin lay on the equator within the circumglobally belt of evergreen tropical rainforest. Approximately 68 % of the basin corresponds to Brazil, and the remainder splits between Bolivia, Colombia, Ecuador, French Guiana, Peru, Suriname, and Guyana (Melack, 2017) suggested that the evasion of  $\text{CO}_2$  from Amazonian rivers, lakes, and temporally inundated aquatic habitats is similar to net ecosystem exchanges in non-inundated upland forests. In addition, Amazonian aquatic ecosystems account for a significant proportion of the global methane flux from natural wetlands. The whole Amazon basin (regions <500 m) corresponds to ~ seven million square kilometers, from which about 30% comply with international criteria for wetland definition (Junk et al., 2011). To determine the role of the Amazon basin in the global scenario of the carbon budget is necessary to characterize the aquatic systems. In this context, several studies have emerged to gather valuable information to estimate the carbon dynamics, for instance, floodable area, inundated area, and areal extent of significant habitats permanently or periodically inundated. High-resolution, remotely sensed products are available for specific locations in the Amazon and provide information appropriate for the validation of basin-scale products. Geomorphological approaches are applied to determine the river areas. (Beighley and Gummadi, 2011) present a methodology for estimating full bank discharge and a station's hydraulic geometry coefficients and exponents using limited data, only stage-discharge and Landsat imagery. The approach utilizes 82 streamflow gauging locations in the Amazon Basin. They conclude that the Amazon Basin contains approximately 4.4 million kilometers of channels with a combined surface area of 59,700  $\text{km}^2$ . Other studies focus on the Modelling of Inundation (De Paiva et al., 2013) developed hydrologic/hydrodynamic modeling of the Amazon River basin. They showed that Amazon hydrological processes govern most of the Amazon terrestrial water storage changes (56 %), followed by soil water (27 %) and groundwater (8 %).

Moreover, floodplains play a significant role in stream flow routing, although backwater effects are also essential to delay and attenuate flood waves. Atmospheric stability influences gas transfer velocities (Harrison et al., 2012). Latent heat fluxes are critical in warm tropical waters like the Amazon basin, leading to convective mixing and enhanced k values. Conversely, diurnal heating under intense insolation can cause stable water column stratification that may lead to low or high k values. Hence, given the pronounced diel cycle of heating and cooling often observed in shallow tropical lakes, measuring stratification, mixing, and gas exchanges over these diel cycles are essential. However, such measurements are unusual. The lack of studies of k values and gas concentrations in vegetated habitats adds further uncertainty, mainly because of the large areas of flooded forests and floating macrophytes throughout the Amazon basin (Melack

and Hess, 2010) It is also essential to investigate the variation of  $k$  for the change in altitude above 500 m, where the mountains that feed sediments and organic matter to the Amazon basin come together. In these areas, the turbulence and speed of discharge would increase the degassing of rivers and streams. Flux and concentrations measurements intend to provide the basis for the entire basin-wide estimates. However, they need to look at the Andean component.

According to Johnson et al (2008), 77 % of carbon transported by water from the landscape was as terrestrially respired  $\text{CO}_2$  dissolved within soils, over 90% of which evaded to the atmosphere within headwater reaches of streams. Evasion of  $\text{CO}_2$  from headwater streams in the Amazon decouples from those larger rivers and wetlands. In larger lotic systems, instream processing of allochthonous organic C is the primary driver of the  $\text{CO}_2$  supersaturation typical of tropical rivers. In headwater streams, however,  $\text{CO}_2$  supersaturation results primarily from groundwater discharge of terrestrially respired  $\text{CO}_2$  dissolved within deep soils. (Neu et al., 2011) measured C losses as soil respiration, direct  $\text{CO}_2$  and  $\text{CH}_4$  fluxes from the stream surface, and fluvial export of DIC and DOC over an annual hydrologic cycle from forested Amazon perennial first-order headwater watershed at Tanguro. They reported  $\text{pCO}_2$  concentrations from 6,490 to 14,980  $\mu\text{atm}$  and evasion rates from the stream surface of  $6490 \pm 680 \text{ gCm}^{-2} \text{ yr}^{-1}$ ; methane concentrations in the stream ranged from about 290–440  $\mu\text{atm}$  and evasion averaged  $990 \pm 220 \text{ gCm}^{-2} \text{ yr}^{-1}$ . (Borges et al., 2015) sampled along the mainstem Solimoes and Amazon rivers and at the mouths of major tributaries and reported a range in  $\text{pCO}_2$  concentrations from 70 to 16,880 ppm. (Abril et al., 2014) conducted eight 800-km cruises along the main channel of the Solimoes-Amazon River and portions of its major tributaries in the central basin and measured  $\text{pCO}_2$  every minute while underway. Values of  $\text{pCO}_2$  were similar to previous studies and varied from 1,000 to 10,000 ppmv. (Raymond et al., 2013) included regional information in their global estimates of  $\text{CO}_2$  emissions from streams and rivers.  $\text{CO}_2$  concentrations in streams and rivers of the Amazon basin (including the Andes) averaged about 6,890  $\mu\text{atm}$ . The  $k_{600}$  value used is at the high end of those reported by others for the rivers of the lowland Amazon as it includes streams and rivers in highland portions of the basin. Hence, the areal efflux is probably too high as a basin-wide average. (Sawakuchi et al., 2014) assessed the spatial-temporal variability of  $\text{CH}_4$  fluxes from the Amazon and its main tributaries, the Negro, Solimoes, Madeira, Tapajos, Xingu, and Para Rivers, based on direct measurements using floating chambers. Considering only river channels, the Amazon Basin release between 0.40 and 0.58  $\text{Tg CH}_4 \text{ yr}^{-1}$ . Meanwhile, the estimates of  $\text{CH}_4$  flux from all tropical rivers and rivers globally were 19–51% to 31–84 % higher than previous estimates, with large rivers of the Amazon accounting for 22–28 % of global river  $\text{CH}_4$  emissions.

Regarding  $\text{N}_2\text{O}$ , the Bottom-up, and top-down approaches indicate that Africa was the largest source of  $\text{N}_2\text{O}$  in the last decade, followed by South America (Tian et al., 2020). Guilhen et al., (2020) quantify the average yearly denitrification and associated  $\text{N}_2\text{O}$  emissions over the entire watershed at  $17.8 \text{ kgN ha}^{-1} \text{ yr}^{-1}$ , and  $0.18 \text{ gN-N}_2\text{O m}^{-2} \text{ yr}^{-1}$  respectively for the period 2011–2015. They found that the Amazonian wetlands have similar emissions of  $\text{N}_2\text{O}$  with the Congo tropical wetlands and lower emissions than the temperate and tropical anthropogenic wetlands of the Garonne (France), the Rhine (Europe), and south-eastern Asia rice paddies. Nitrification fluxes exceed inland waters' denitrification fluxes by 5–20 %, depending on the system. (Maavara et al., 2019) estimate global inland water  $\text{N}_2\text{O}$  emissions of 0.148–0.277  $\text{Tg N yr}^{-1}$ , considerably lower than those of Tian et al., (2020), corresponding to  $\sim 17.0 \text{ Tg N yr}^{-1}$  between 2007 and 2016. Notably, most riverine GHG emissions are from the tropics, with  $\sim 79$  % for  $\text{N}_2\text{O}$  (Hu et al., 2016). Regarding to lakes, (Raymond et al., 2013) calculated  $\text{CO}_2$  concentrations in lakes of the whole Amazon basin (including the Andes) as averaging about 1,906  $\mu\text{atm}$  and an efflux, using a  $k_{600}$  of  $5.8 \text{ cm h}^{-1}$ , of  $1230 \text{ kg C km}^{-2} \text{ d}^{-1}$ . They expressed the results to the surface area of rivers and streams, not land area. Large savanna floodplains occur in the Llanos de Moxos (Bolivia) and in Roraima (Brazil) (Melack et al., 2004) approximated mean annual methane emission from these two areas as  $70 \text{ Mg C km}^{-2} \text{ yr}^{-1}$ . Jati (2013) made monthly measurements of carbon dioxide and methane flux with floating chambers in 80 wetlands near Boa Vista (Roraima); mean values from



his results were about  $9,670 \text{ kg C km}^{-2} \text{ d}^{-1}$  and  $9.6 \text{ kg C km}^{-2} \text{ d}^{-1}$  for  $\text{CO}_2$  and  $\text{CH}_4$ , respectively. (Melack et al., 2022) proposed a comprehensive study for methane emissions for the Amazon basin and summarized published fluxes and regional levels extracted from the WetCharts ensemble of wetland models corresponding to  $39.4 \pm 10.3 \text{ Tg CH}_4 \text{ y}^{-1}$ . Conversely, Gedney et al. (2019) provided  $\text{CH}_4$  flux estimates for the Amazon basin using the Joint UK Land Environment Simulator (JULES) model ranging from  $\sim 24$  to  $\sim 58 \text{ Tg CH}_4 \text{ y}^{-1}$ .

## **Andean influences on the biogeochemistry of the Amazon River**

The Amazon River exits the Andes mountains more than 4,000 kilometers (km) from its estuary. Alfred Russel Wallace (1853) was the first naturalist to write about the white-water, clear-water, and black-water river types of the Amazon basin and to relate the color of tributaries to the nature of their drainage basins. He linked the sediment load of white-water tributaries to erosion in their steep Andean headwaters and identified clear-water rivers with the crystalline mountains of Brazil. Furthermore, he linked the black-water rivers emerging from lowland sources and correctly attributed their dark coloring to the leaching of "decaying leaves, roots, and other vegetable matter" (Wallace, 1853). Similarly, Henry Bates (1963) highlighted the transport of volcanic pumice in the main-stem Amazon River and assigned its origin to volcanic ranges thousands of kilometers away in the Ecuadorian Andes. Steep terrain and young lithologies make the Andes a vital source of sediments and solutes to the lower reaches of the Amazon River. The most visible characteristics of the main-stem Amazon and its Andean tributaries are high discharge and heavy loads of suspended and bedload sediments, nutrients, and organic matter. The dominant downstream trend in biogeochemical and trophic characteristics of the main-stem Amazon and its large Andean tributaries is the progressive dilution of Andean contributions by lowland inputs of tributaries (Devol and Hedges, 2001).

The Andes mountains rise steeply along the western margin of the Amazon basin. Approximately half of the Andean Amazon lies at elevations between 500 and 2,000 m, while most of the remainder is between 2,000 and 4,000 m; about 16 % is above 4,000 m. Active volcanoes are prominent features of the Ecuadorian and Bolivian Andes. The primary vegetative cover types in the Andean Amazon are submontane (700 to 2,000 m) and montane (to 3,700 m) forests, which together constitute approximately 42 % of the region. The entire north-south length of the Andean Amazon basin is drained by eight important rivers—the Caquetá, Putumayo, Napo, Marañón, Ucayali, Madre de Dios, Beni, and Mamoré (McClain and Naiman, 2008).

The Andes cover only about 13 % of the Amazon basin. The combined flow of the Andean tributaries amounts to approximately half of the mainstem flow. Nevertheless, while Andean water contributions to the main-stem Amazon may be proportional to area, contributions of sediments and solutes are disproportionately greater (GHGs drivers). Moreover, the river's energy and nutrients from the Andes drive direct and indirect main-stem productivity through biophysical feedback with the massive lowland floodplain. Gibbs (1967) estimated that 86 percent of the total dissolved salts and 82 percent of suspended solids discharged by the Amazon were supplied from  $\sim 13$  percent of the mountainous-environment type's total area. This statement is consistent with (Meade, 1994). Most of the suspended matter sediments at the base of the Andes, measured along the entire length of the Madeira River from its Andean headwaters, show a sharp decrease in sediment load (as much as 60%) and diameter at the piedmont base of the Andes (Guyot et al., 1999). In Ecuador, the sediment discharge in the Amazon basin (Napo, Pastaza, and Santiago) corresponds to  $47 \times 10^6 \text{ t yr}^{-1}$ , whereas the suspended sediment load in the Pacific basin was  $6 \times 10^6 \text{ t yr}^{-1}$ . The difference between these contributions is likely due to the tectonic

uplift and the seismic and volcanic dynamics on the Amazon side (Armijos et al., 2013). Conclusive evidence from an Andean source depicts the mineralogical and isotopic composition of the suspended sediments (Bouchez et al., 2014) reported that 15% of fine particulate OC (FPOC) originates from rocks in the Andean part of the Madeira Basin. Andean tributaries deliver an order of magnitude more particulate nitrogen (1,170 megagrams [Mg] per year) and phosphorus (806 Mg per year) to the main stem than their lowland counterparts 119 and 43 Mg per year, respectively (McClain et al., 2008). Most particulate nitrogen is likely organic, whereas phosphorus is mainly phosphate strongly adsorbed to iron and aluminum oxide surfaces. Andes contain the only significant evaporite and carbonate deposits in the Amazon basin (Stallard and Edmond 1983). High fluxes of  $\text{Ca}^{2+}$  (calcium),  $\text{Mg}^{2+}$  (magnesium),  $\text{HCO}_3^-$  (bicarbonate), and  $\text{SO}_4^{2-}$  (sulfate) ions occur in rivers draining carbonate deposits, and high fluxes of  $\text{Na}^+$  (sodium) and  $\text{Cl}^-$  (chloride) ions occur in rivers draining evaporite deposits. Also, Among the Amazon tributaries that drain basins dominated by less-weatherable silicate rocks, Andean rivers have consistently higher total cation concentrations (Mortatti and Probst, 2003)

### Organic matter

Mountain rivers play a key role in the delivery of POC to large river systems and the ocean. Andean-derived suspended sediments carry a significant amount of organic matter, 90% of which is made up of particles less than 63 micrometers ( $\mu\text{m}$ ) in diameter FPOC (Mortatti et al., 2003; McClain et al., 2008) (. Total organic carbon is approximately 1%, by mass, of suspended sediment in the main stem, constituting a flux of 5 to 14  $\text{TgC yr}^{-1}$  to the Atlantic Ocean. Measurements show that more than 90% of particulate organic carbon (POC;  $> 0.5 \mu\text{m}$ ) in the main-stem Amazon River comes from Andean tributaries (Richey et al., 2002). Few FPOC is transported directly from the Andes; most is stored for varying periods of time in point-bar and floodplain sediments. The actual proportion of FPOC of Andean origin has been approximated using ( $\delta^{13}\text{C}$ ) stable isotopic ratios as a “fingerprint” of its origin. There is a clear separation of FPOC  $\delta^{13}\text{C}$  between Andean and lowland rivers, and this separation can be used to estimate the relative proportion of each in the main stem. For example, since FPOC in purely lowland rivers has  $\delta^{13}\text{C}$  values near to  $-28.5\text{‰}$  (McClain and Naiman, 2008) and  $-26.5\text{‰}$  corresponds to the discharge of FPOC of the exiting Andean Peruvian, a discharge in the main-stem Amazon River at Óbidos of  $-27.4\text{‰}$  would suggest a mixture of 50% Andean FPOC and 50% lowland FPOC (Quay et al., 1992). Unlike particulate organic matter, this DOM appears to derive largely from lowland sources mainly. Andes largely regulate the particulate load to the main-stem Amazon River, not simply with respect to its particulate mineral load but also with respect to associated nutrients and organic matter.

Feedback between land C pools and Climate provides one of the largest sources of uncertainty in our global climate predictions. The magnitude and timing of these fluxes depend on the hydrological regime (Borges et al., 2015) of C and have two significant pools the Above/Below-ground biomass (AGB/BGB) and the soil. The Amazon basin and Andean store a vast amount of C (Raymond et al., 2013). Overall estimations of the total AGB of intact Amazonian rainforests ( $\sim 5.76 \times 10^6 \text{ km}^2$ ) result in  $93 \pm 2 \text{ PgC}$ , considering lianas and small trees. Including dead biomass and BGB the value would increase to  $111,6 \text{ PgC}$  (Saunois et al., 2020). Riverine systems transport C (Tian et al., 2020) mainly in the form of dissolved organic DOC, POC, and dissolved inorganic (DIC); globally, rivers export approximately  $0.89 \text{ Pg C}$  to oceans every year, including  $0.24 \text{ Pg DOC}$ ,  $0.24 \text{ Pg POC}$  and  $0.41 \text{ Pg DIC}$ . DOC, POC, and DIC are associated with soil organic C amount, amount of soil erosion, and  $\text{CO}_2$  consumption by rock weathering, respectively. However, external drivers like catchment slopes, vegetation, geology, Climate, and size can also play a significant role (Tamooh et al., 2012).

Riverine POC derives from soil litterfall and autochthonous production and is a function of the total solid matter (TSM), which in turn depends on the rainfall intensity, drainage intensity, and basin slope. Usually, there is a positive correlation between elevation (basin slope) and biomass density (cultivation areas); nevertheless, the most significant driver is the TSM flux, showing that

the %POC in suspended solids decreases when increases the concentration of TSM(van der Werf et al., 2009). Regarding DOC, drainage intensity would control the DOC mainly, followed by morphology flatness and the C soil availability. DOC has two origins autochthonous (C produced by phytoplankton) and allochthonous (C produced by terrestrial biosphere) (van der Werf et al., 2009). The former is important in downstream rivers where the size, the interface area soil/water, and residence time increase, enabling leaching processes to be more intense. Conversely, steep basins result in a higher share of surface runoff concerning total runoff. This restricted contact between soil and water makes rivers less DOC concentrated (Tamooch et al., 2012).

The Amazon drainage basin comprehends the Andean Cordillera, the lowland Amazon basin, Precambrian shields, and a large floodplain (i.e., Varze) that covers - 100,000 km<sup>2</sup>. This geomorphological diversity of drainage regions results in rivers that have a large range of chemical characteristics (Quay et al., 1992). Rivers within the Amazon drainage basin have been classified previously by color. For example, whitewater rivers have high suspended and dissolved loads; clearwater rivers with depletion of suspended and dissolved material; blackwater rivers have low-suspended sediments and dissolved inorganic material but high concentrations of dissolved organic materials. The Andes are a crucial source of water and sediments to the Amazon River, the largest single source of fresh water and OC to the oceans (Gibbs, 1967). In the Amazon basin, the average export of total OC was 36.1 Tg yr<sup>-1</sup> (8.5 g m<sup>-2</sup> yr<sup>-1</sup>), of which 62% was DOC and 38% POC(Richey et al., 1990). the isotopic data (+19 Δ<sup>14</sup>C ‰) provide information that POC in the main stem have a trustworthy Andean source, much of it would have to be hundreds to thousands of years old (McClain and Naiman, 2008). Since stable isotopic δ<sup>13</sup>C ratios of plant leaves are positively correlated with elevation (-30‰ at 1000 - 2000 m to -26‰ at 4000 m)(Townsend-Small et al., 2007), they permit identifying the proportion of POC of Andean origin. Unlike POC, DOM appears to derive mainly from lowland sources.

Although scientific information asserts that Andean Cordillera significantly influences the overall biogeochemistry of the Amazon, there needs to be more research considering the integrated Andean-Amazon-fluvial system. We describe gradients of TSM, POC, DOC, δ<sup>13</sup>C-DOC, δ<sup>13</sup>C-POC in function of altitude, morphology patterns, and biomass covertures. The outcomes report a comprehensive data set of DOC, POC, and TSM of inland waters of three ecosystems in Ecuador: paramo (>3000 m), highland (500-3000 m), and lowland (<500 m). This variation in the altitude allows testing the response of the ecosystem to different drivers, for instance, at the highest altitude, significant erosion, and low organic matter inputs, except in paramo where soil characterizes for storage high levels of organic matter. In the middle altitudes, an increase in organic matter availability and moderate erosion rates. Finally, significant amounts of organic matter and minimal erosion rates at lowlands.

### Productivity

The Amazon floodplain is a highly productive regional system, with a net production of 11.9–16.8 Mg C ha<sup>-1</sup> (Clark et al., 2013) that occurs over an area of 67,900 km<sup>2</sup>. Annual floodplain deposition at any given place depends mostly on river-level fluctuation and duration of flooding. This process builds the fertile floodplain soils along Andean tributaries and the main stem. The sediments of the lateral flux between the channel floodplains in the main stem (2,100 km) range from 1,570 to 2,070 Tg yr<sup>-1</sup> and exceed the downstream flux of 1,200 Tg yr<sup>-1</sup>. Approximately 500 Tg yr<sup>-1</sup> of upstream-derived sediment and nutrients accumulate on the floodplain and the river floor (McClain and Naiman, 2008a; Amazon Waters, 2022). Most primary productivity derives from macrophyte (65%) and floodplain forest (28%) communities. Floodplains of the mainstem Solimoes emitted CH<sub>4</sub> at an annual rate of 1.3 TgC yr<sup>-1</sup>. CH<sub>4</sub> emission from a 1.77 million km<sup>2</sup> area in the central basin had a mean of 6.8 TgC yr<sup>-1</sup>, a greenhouse flux of about 0.5 PgC as CO<sub>2</sub> equivalent (Melack et al., 2004).

### Lakes

The global importance of any ecosystem type in a process or cycle is the product of the areal extent and the intensity of the process in those ecosystems. In this sense, lakes play a much

more significant role in global carbon cycling than their area would otherwise predict (Downing, 2010) While lakes make up less than 2% of Earth's surface area, they bury over three times more carbon in their sediments than all of the world's oceans combined. This statement means lakes bury over 100 times more carbon than the oceans per area (Downing et al., 2008). Lakes storage of 0.07 to 0.14 Pg of organic C  $y^{-1}$  is co-equal with long-term sediment accumulation of organic C in the ocean 0.1 Pg C  $y^{-1}$  (LaRowe et al., 2020). Although lakes bury vast amounts of carbon, they also tend to release more CO<sub>2</sub> to the atmosphere than they absorb, making them net sources of greenhouse gases. Most of the world's lakes are supersaturated in CO<sub>2</sub>, releasing some into the atmosphere. Several internal processes involve GHGs in lakes; for instance, GHGs are transported from surrounding systems, originate within sediments or in the water column, and are emitted to the atmosphere for different reasons. Any fate follows complex chemical and biological pathways that have not been fully understood. The advent in the early 1980s of techniques that could measure the rate at which bacteria synthesize new biomass revealed that pelagic bacterial production was co-equal or even more significant than primary production. Therefore, one could infer that bacterial respiration in many freshwater and marine systems is too high to explain if the only source of labile organic C is autochthonous primary production (Pradeep Ram et al., 2007; Roland et al., 1999). Actually, Cole et al., (2000) note that this pattern repeats in whole systems of fresh waters; namely, respiration (R) often exceeds GPP, and the only way to explain this excess is to have some of the external terrestrial supply of organic matter actively respired within the water body.

Most of the lakes sampled show supersaturation in CO<sub>2</sub> and CH<sub>4</sub>; therefore, they are sources of GHGs in the atmosphere (Borges et al., 2022). However, direct sampling is geographically skewed, with an overrepresentation in North America and Scandinavia global datasets. At the same time, tropical regions are much less covered, which can provoke misleading in global estimations. For example, Tropical lakes accounted for 34% of the global lake CO<sub>2</sub> emission, although they represented only 1.5% of the CO<sub>2</sub> dataset used in the global estimations (Raymond et al., 2013). The content of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O in lakes' surface waters results from the balance of C sources and sinks, which are specific to each gas (Borges et al., 2022). The main drivers are allochthonous organic matter, nutrients, internal organic matter production, and degradation. Also, the characteristics of the surrounding ecosystems play a significant role, such as catchment and land cover, size, depth, shape, and weather conditions. The interaction of these variables makes global-scale emissions predictions challenging to dimension. For example, (Borges et al., 2022) carried out several 24-lake missions that accounted for 49% of the total lacustrine surface area of the African continent to evaluate the CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O content. They found that the concentrations of dissolved CO<sub>2</sub> were much lower than values attributed in literature to tropical lakes and lower than in boreal systems because of higher productivity (triple that of boreal systems). In contrast, surface CH<sub>4</sub> concentrations were generally higher than in boreal systems. The main drivers that make the difference between the dynamics of boreal and tropical lakes were phytoplankton biomass, temperature, sunlight (restricted to the time of year), and low DOC enrichment.

Accurate estimates GHG fluxes to and from the atmosphere are essential to understanding the global carbon (C) budget and making projections of the trajectory of climate change. Several approaches have tried to calculate GHG fluxes, for example, by multiplying the average rate of CO<sub>2</sub> emission from lakes, determined from a literature review, by their total area to calculate a global emission rate of 0.81 Pg C  $yr^{-1}$  (Lars J. Tranvik et al., 2009) the same was developed by (Bastviken et al., 2011) for CH<sub>4</sub>; resulting in 69 Tg C  $y^{-1}$  or about 0.85 Pg C as CO<sub>2</sub> emission equivalents. Regarding N<sub>2</sub>O, (Soued et al., 2016) published a first-order approximation of the global emissions for lakes and reservoirs. They determined approximately 0.08 Pg of C equivalent (0.627 ± 0.296 Tg of N<sub>2</sub>O-N  $yr^{-1}$ ). (DelSontro et al., 2018) improve the estimations by applying the following considerations to the approach by linking the data on the global distribution of lake size and productivity with current empirical models of GHG emissions, including the three

GHGs in one study; and assessing the sensitivity of lentic GHG emission to expected future changes in lake and impoundment productivity. This analysis indicates that lentic systems are essential sources of GHGs driving climate change. Also, CH<sub>4</sub> emissions may be of disproportionate importance due to their link with lake productivity and the high potency of CH<sub>4</sub> as a GHG. Global emissions from lakes and impoundments are equivalent to 20% of global fossil fuel CO<sub>2</sub> emissions ( $9.5 \pm 0.5 \text{ Pg C-CO}_2 \text{ yr}^{-1}$ ) (Friedlingstein et al., 2022).

## **Project description**

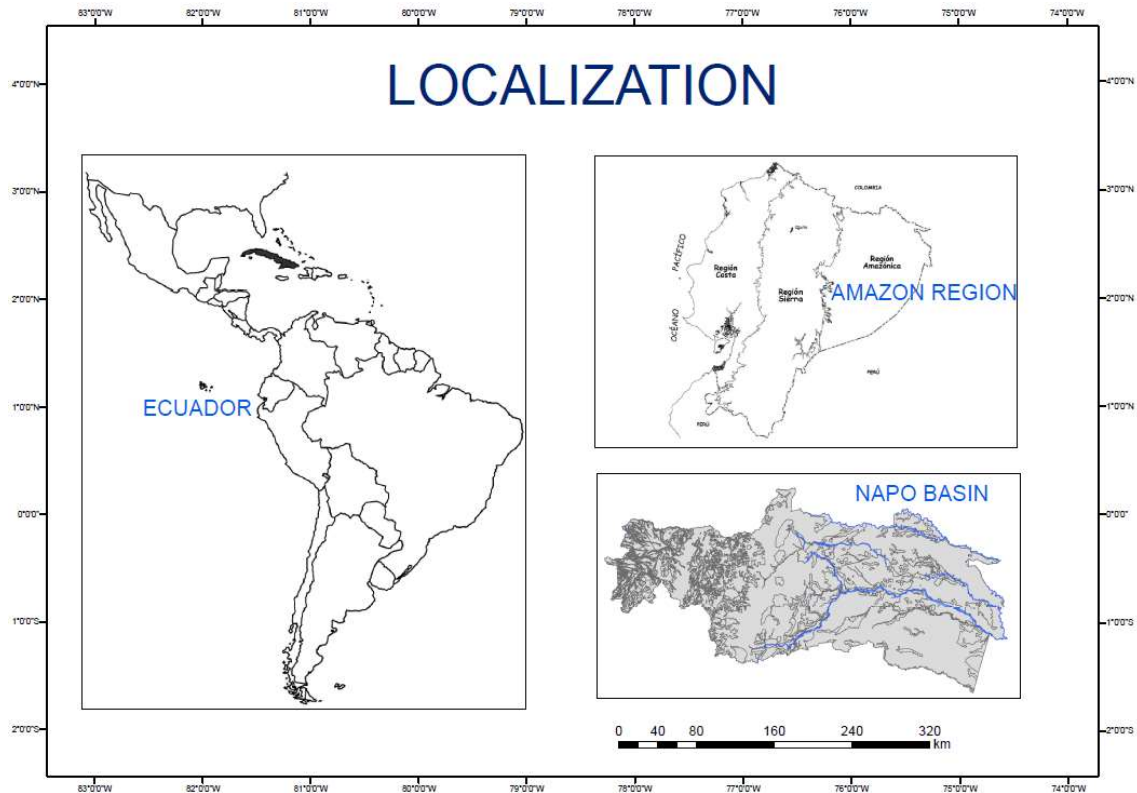
The project is carried out within the cooperation program of the Central University of Ecuador and the University of Liège under the sponsorship of ARES. Given the importance of continental waters in climate change and the understanding of geochemical processes and cycles, this Ph.D. project focuses on two aspects with current gaps in the literature. The first part corresponds to the sampling and estimation of GHGs flow from the Napo River Basin (a tributary of the Amazon). The second part focuses on studying GHGs in 15 lakes in the Andean moorlands and Ecuador's highlands. In this way, we reported data from two ecosystems and their respective continental waters.

### **Napo Basin**

The Republic of the Equator is a country in northwestern South America, bordered by Colombia on the north, Peru on the east and south, and the Pacific Ocean on the west. Four natural regions and ten hydrographic basins split according to their stream's direction. East's basins flow into the Amazon, and west's basins into the Pacific Ocean Fig 1.10. The Napo basin covers approximately 110,000 km<sup>2</sup> and represents 1,6 % of the Amazon basin. Approximately 60 % of the Napo basin is in Ecuador, while the other 40 % is in Peru.

The studies conducted to explain the GHGs in the Amazon basin focused on altitudes under 500 m. However, the degassing magnitude, lateral transportation, and the chemical dynamics occurring in low-order streams at highlands are still uncertain (Marx et al., 2017). Several aspects driving the biogeochemistry of the Amazon main stem link to the magnitude and variability of water and materials supplied from the Andes (McClain and Naiman, 2008). Here, we studied those aspects harnessing the geographical features of Ecuador, where Andean mountains and Amazon basins occur along relatively short distances.

FIGURE 1.10 Project localization. Napo Basin is in the Ecuadorian Amazon Region.



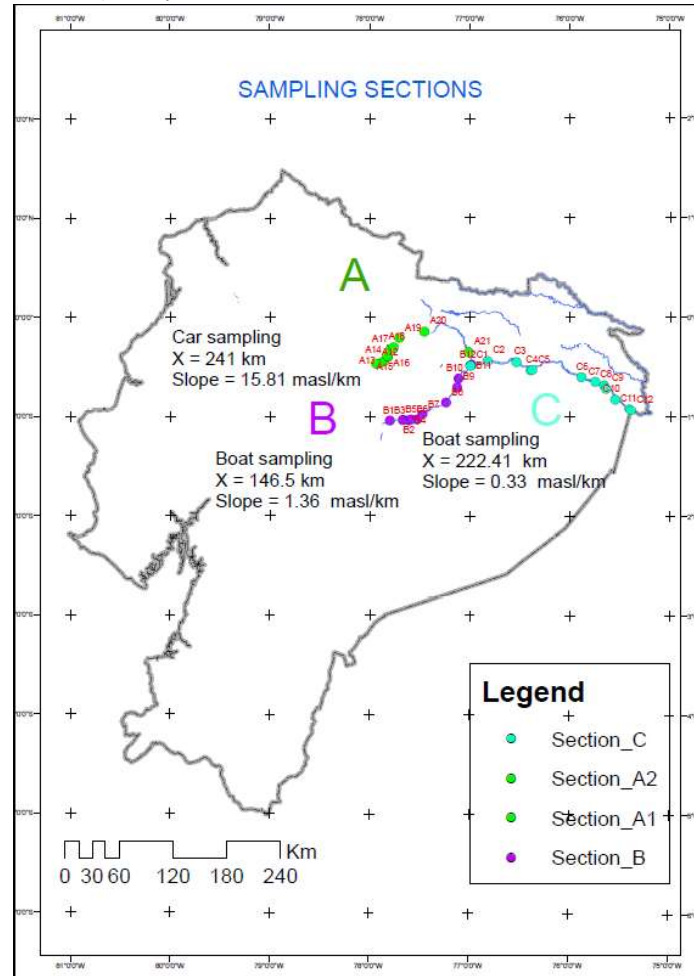
The sampling area covers the largest hydrological basin of Ecuador called Napo, which consists of about 60,000 km<sup>2</sup>, with an average discharge of 2,889 m<sup>3</sup> s<sup>-1</sup> ranging from 642 to 4,122 m<sup>3</sup> s<sup>-1</sup> for June and September, respectively (National Institute of Meteorology and Hydrology, 2021). The vegetal coverage consists mainly of forests 62 %, crops 15 %, grasslands 15 %, and others uses 7% (Technical Secretary of Planning, 2021). Therefore, we cover the basin's two longest possible routes into three sampling sections. The first one (El Coca) starts in the moorland of Salve Faccha at the Cayambe-Coca Reserve, the second starts in Puerto Napo (Napo upstream) at the piedmont of Antisana and Cotopaxi, and they both confluence at Francisco de Orellana, where starts the third section until the Yasuní National Park.

Due to the country's diversity along the sampling route, we found five climatic floors, sorted by altitude, i.e., the warm from 0 to 1,000 m, the temperate from 1,000 to 2,000 m, the cold from 2,000 to 3,000 m, and the moorland above 3,000 m (Farfan Portilla, 2018). The above-ground biomass varies along a small-scale elevation gradient from an Evergreen Amazon Forest < 500 m to the páramo grasslands > 3,000 m and their implications in terms of carbon storage (Torres et al., 2020). For instance, below 500 m values of 39.2 kgdb m<sup>-2</sup> have been reported. However, between 600 to 1,000 m, the average value decreases to 29.1 kgdb m<sup>-2</sup> (Farfan Portilla, 2018). And in the highlands above 2,700 m we find values of 19.6 kgdb m<sup>-2</sup>.

The first route is El Coca. It comprehends 241 km of sampling road and starts on the upper area of Salve Faccha páramo reserve at 4,200 m. This ecosystem is called the Andean moorland, with low temperatures throughout the year 5 ± 4 °C, precipitations 1,953 ± 134 mm yr<sup>-1</sup>, and relative humidity around 85 %. The second route is the Napo upstream, sampled entirely by boat over 146,5 km. It begins in Puerto Misahuallí, located at 408 m. The Climate is tropical, and there is rainfall throughout the year, even during the driest months. The average

annual temperature is  $22 \pm 1.5$  °C, precipitation is  $3,832 \pm 187$  mm yr<sup>-1</sup>, and humidity is about 87 %. The third route is Napo downstream, from the confluence of the Coca and Napo rivers until the Yasuní National Park at the border with Peru; the altitude corresponds to 175 m and is in a tropical forest. The average temperature is  $23 \pm 1.5$  C, and the precipitation level is around 2,832 mm yr<sup>-1</sup>.

FIGURE 1.11 Stations and river pathway.



There are two small cities in the sampling area. First, El Tena, with 23,000 inhabitants and a population density of 90 persons per km<sup>2</sup>, is located near Puerto Napo (station B1). The second is El Coca (Puerto Francisco Orellana), with 40,700 inhabitants and a density of 278 persons per km<sup>2</sup>, located at the confluence of the Coca and Napo rivers near stations B12 and C1 (Fig. 1.11). El Tena is a settlement dedicated to tourism and poorly industrialized agriculture without much use of fertilizers. In the same way, El Coca devotes to tourism development, but to a minor degree, its primary income source is the exploitation and provision of oil services. Besides these two settlements, several small towns are distributed along the river, both in the Napo and Coca. Usually, the sewage of these towns is not treated, just disposed directly into the river or indirectly through latrine systems and soil leaching.

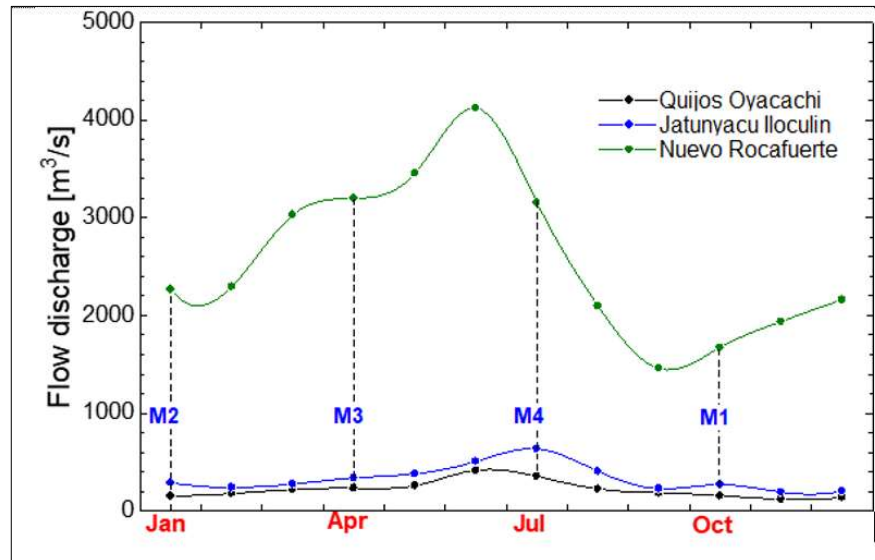
*Greenhouse gas emissions in Napo Basin and Andean Lakes*

TABLE 1.1 Sampling dates, the missions were carried out quarterly from October 2018 to March 2021.

| Month     | Missions | 2018         | 2019         | 2020          | 2021         |
|-----------|----------|--------------|--------------|---------------|--------------|
| January   |          |              |              |               |              |
| February  | M2       |              | Accomplished | Accomplished  | Accomplished |
| March     |          |              |              |               |              |
| April     |          |              |              |               |              |
| May       | M3       |              | Accomplished | Re-programmed |              |
| June      |          |              |              |               |              |
| July      |          |              |              |               |              |
| August    | M4       |              | Accomplished | Accomplished  |              |
| September |          |              |              |               |              |
| October   |          |              |              |               |              |
| November  | M1       | Accomplished | Accomplished |               |              |
| December  |          |              |              |               |              |

The second mission of 2020 was rescheduled due to mobility restrictions caused by COVID-19

FIGURE 1.12 Seasonal variation discharge and sampling planning, from April to July we have the season with the highest discharges values.





We took the series of flow discharge data from the Institute of Hydrology and Meteorology (INHAMI Table 1.2) to study the temporality's effect. We processed the available average flows from the last three years. Unfortunately, the mission planned for April 2020 was postponed due to the restrictive mobility measures (COVID 19) to next year in March 2021. Missions M1 and M2 correspond to the low-level season, typically from September to February, whereas missions M3 and M4 correspond to the high-level discharge season from March to August (Fig. 1.12). We selected one gauge station per route at 1,644, 570, and 180 m; the lower the altitude, the higher the catchment area.

TABLE 1.2 Stations, location, and length of the rivers.

| Section         | Length (km) | Altitude (m) | Estations | Main stem | Tributaries |
|-----------------|-------------|--------------|-----------|-----------|-------------|
| Coca            | 241,0       | 4,209 – 388  | 21        | 12        | 9           |
| Napo upstream   | 146,5       | 444 – 245    | 12        | 6         | 6           |
| Napo downstream | 222,4       | 244 – 175    | 12        | 6         | 6           |

The route is 610 km per campaign, from which 39 % is sampled by car and 61 % by boat. The schedule is repeated for two years, fulfilling eight quarterly campaigns. In the lab, the sampling stations were georeferenced using ArcGIS 10.8.1. The total number of stations was 45, divided between the main river and tributaries. In situ measurements of pCO<sub>2</sub> are performed with the Li-Cor 820 equipment and its software employing the headspace technique for four syringes. In the same way, we directly measured the pH, conductivity, dissolved oxygen, and temperature with the YSI ProPlus probe.

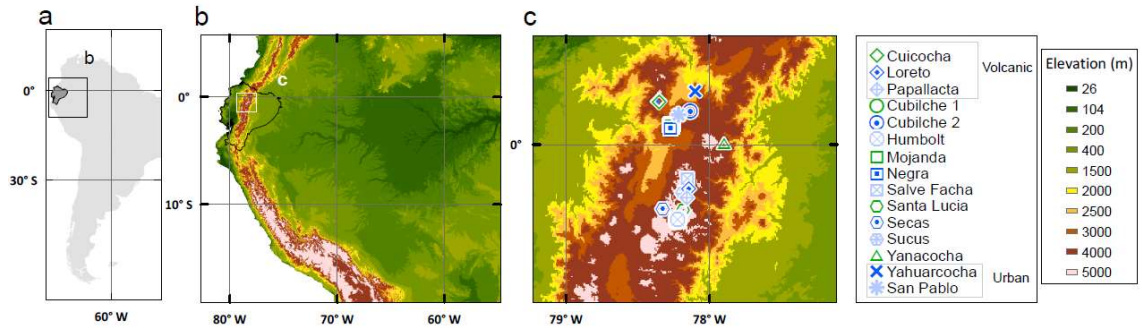
### Moorland's lakes

Paramos or moorlands are Andean Mountain ecosystems belonging to the Amazonian basin domain; they are located discontinuously in the neotropics, at altitudes between 3,000 and 4,300 m (Santander, 2015) mainly from Venezuela to northern Peru. The paramo occupies more than 30,000 km<sup>2</sup> in South America. Ecuador has the highest proportion of its territory covered by moors, with 7%. They constitute a primary source of drinking water for large populations in Ecuador, Colombia, and Venezuela. In Ecuador, the temperature is relatively constant, the average annual rainfall is 1 345 mm, and it occurs throughout most of the year (Santander, 2015). Only ~12 % of the days are dry. These conditions have favored the formation of peat and a large number of wetlands and lakes. The paramo usually presents deep organic-rich soils (30 % by volume), and grasslands dominate the vegetation (Zapata et al., 2021).

The Andean Moorland ecosystem contains the richest high-mountain flora in the world and is high in endemism. The grasslands contain many rare or endangered species, some of them restricted to just one mountain or valley. The headwaters are highly active sites for transporting, transforming, and storing considerable amounts of carbon from the terrestrial environment. Therefore, considering their spatial extent, they may have a disproportional effect in terms of GHGs emissions (Lars J Tranvik et al., 2009).

This study seeks to determine the GHG emissions, drivers, and carbon dynamics from 15 lakes in the northern Ecuadorian Andes region, which is rich in number and size. Twelve stations are in the paramos of Salve Faccha, Antisana y Mojanda above 3000 m (Fig 1.13). Three stations correspond to lower altitudes (between 2200 and 2800 m) with more anthropic influence and higher average temperatures. Given the area's diversity, we could select lakes with different characteristics such as depth, area, surrounding ecosystem, and formation. The study extends to 3 provinces Pichincha, Imbabura, and Napo.

FIGURE 1.13 Lake locations and georeferencing.



Moorland spreads out from Venezuela to Perú. In Ecuador, the soils have developed on the pyroclastic deposits resulting from volcanic eruptions. They generally have high water retention capacities (from 60 to 200%) and often significant accumulations of organic matter. The vegetation cover consists mainly of native forest, grassland, paramo, crops, humid forest, and populated areas. We sampled 54 points, performing in situ measurements of  $pCO_2$  with the Li-Cor 820 through the headspace technique for four syringes. Sampling was carried out approximately in the center of the lakes, and we directly measured pH, conductivity, dissolved oxygen, and temperature with the YSI ProPlus probe and the lake depth.

# OBJECTIVES

## **General objective**

We aim to determine the greenhouse gas emissions (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) in mountainous streams and lakes of the Andes Cordillera in Ecuador and their relative contribution at the scale of the whole Amazon basin.

## **Specific objectives**

To achieve this overarching objective, our more specific objectives were to:

Collect data along a 600 km-length river stretch in the Napo River basin, mountain streams, the foothills of the cordillera, and lowland rivers of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O concentrations, as well as ancillary variables such as organic matter concentration and inorganic nutrients.

Analyze the patterns of variations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O concentrations in streams as a function of catchment morphology (slope) and land cover (grassland, forest, and wetland coverage) basin.

Collect data of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O concentrations, as well as ancillary variables such as organic matter concentration and inorganic nutrients in 15 lakes in the Andean páramos of Salve Faccha, Mojanda, and Antisana, covering three main groups: volcanic, urban and páramo lakes.

Analyze the patterns of variations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O concentrations in lakes as a function of morphology (surface area and depth) and organic carbon content and dissolved inorganic nitrogen.

Estimate CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O emissions from rivers and lakes scale them to the entire Amazon basin and determine the relative contribution of Andean rivers and lakes in the overall basin GHG emissions.



# METHODS

## In situ sampling

As mentioned in the previous section, the study consists of two sampling areas the Napo Basin and the lakes in the Ecuadorian mountains. In the former, the total number of stations was 45, divided between the main river and tributaries and sampled quarterly. The second stage consists of 15 Andean Páramo and Ecuadorian highlands lakes. here, we sampled 54 points using an inflatable boat and procuring to reach the lake's center.

we performed in situ measurements of  $p\text{CO}_2$  with a Li-Cor Li-820 infrared gas analyser based on the headspace technique with 4 polypropylene syringes that were filled directly with river water. The  $p\text{CO}_2$  in the atmosphere was measured by injecting ambient air sampled with an additional polypropylene syringe. Water temperature, specific conductivity, pH, and %O<sub>2</sub> were measured in-situ with a YSI multi-parameter probe (ProPlus). Water for CH<sub>4</sub> and N<sub>2</sub>O samples was collected with a sampling devise consisting of a 2L polyethylene bottle with the bottom cut and fitted with a silicone tubing at the stopper. The equipment's calibration was as follows (1) YSI ProPlus Probe, oxygen in the air, and pH with the 4,0 and 7,0 buffer solutions (2) the Li-Cor 820 equipment was calibrated annually in the laboratory using standards of 388, 813, 3,788, and 8,300 ppm of CO<sub>2</sub> (Air Liquide Belgium).

Additionally, two borosilicate serum bottles (Weathon) with a volume of 40 ml were filled with the silicone tubing, poisoned with 100  $\mu\text{l}$  of a saturated solution of HgCl<sub>2</sub> and sealed with a butyl stopper and crimped with an aluminium cap. Finally, we collected a sample of 12 ml (exetainer) poisoned with 50  $\mu\text{L}$  HgCl<sub>2</sub> for  $\delta^{13}\text{C}$ -DIC.

## Laboratory analyses

The samples for CH<sub>4</sub> and N<sub>2</sub>O are processed in duplicate using the nitrogen headspace technique, then injected into a gas chromatography equipment with a flame ionization detector for CH<sub>4</sub> and an electron capture detector for N<sub>2</sub>O. For the  $\delta^{13}\text{C}$ -DIC, we used the headspace technique with helium and added 100  $\mu\text{L}$  of 99 % H<sub>3</sub>PO<sub>4</sub> to convert all the inorganic carbon into CO<sub>2</sub> (HCO<sub>3</sub><sup>1-</sup>, CO<sub>3</sub><sup>2-</sup>). The headspace formed was analyzed by an elemental IRMS analyzer.

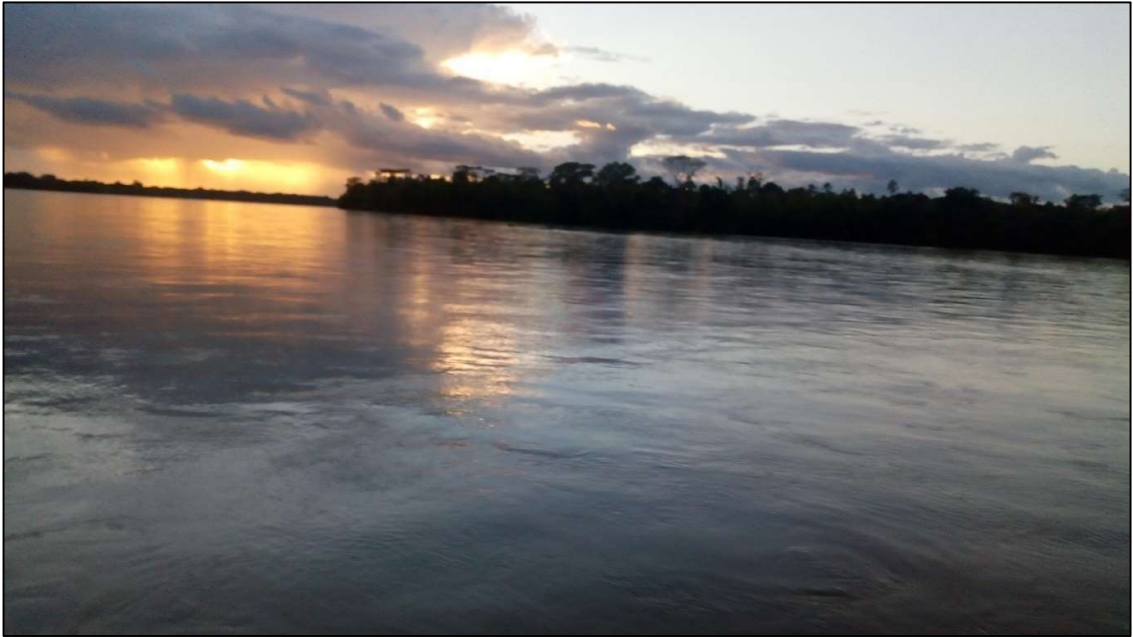
The TSM (total solid matter) is carried out by gravimetry. First, we made pass about 400 mL of water through a 47 mm filter, previously weighed. For the POC, we filter an approximate amount of 100 mL, decarbonize in an atmosphere rich in HCl so that the inorganic carbon becomes CO<sub>2</sub>, and keep only the organic one. Finally, we dried the filters and encapsulated them in metallic paper to be fed to EA-IRMS equipment to identify the concentration of POC, PN, and the composition of  $\delta^{13}\text{C}$ -POC. The POP samples are previously treated to mineralize organic phosphorus, which means orthophosphate (SRP) by hydrolysis and Plon's reagent in an autoclave. Then, in an acid medium and the presence of ammonium molybdate, the SRP forms

the phosphomolybdic complex with a blue coloration, whose concentration can be quantified in a spectrophotometer wavelength of 880 nm.

Total alkalinity is determined by automatic titration with 0.1N HCl. Measurements are verified using reference material for oceanic CO<sub>2</sub> measurements. CDOM and FDOM analysis is carried out by fluorescence and UV / VIS spectrometry, respectively, in high precision 10 mL cells. The major elements are analyzed by inductively coupled plasma MS (ICP-MS), including K, Ca, Cr, Mn, Fe, Co, Ni, Cu, Zn, As, Sr, Mo, Cd, Ba, and Ba Pb. For DOC, we use amber bottles acidified with 50 µL of H<sub>3</sub>PO<sub>4</sub> then they are analyzed through wet oxidation. In addition, we processed nutrients (NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, PO<sub>4</sub><sup>-</sup>) from a 50 mL sample by photometry. Nitrates and nitrites colored with sulfanilamide at 540 nm, Ammonium with dichloro isocyanurate-salicylate-nitroprusside at 655 nm, and Phosphates followed the technique described previously for POC.

## CHAPTER 2.

ANDEAN HEADWATER AND PIEDMONT STREAMS ARE HOT SPOTS OF CARBON DIOXIDE AND METHANE EMISSIONS IN THE AMAZON BASIN.



Sunset in the Napo River

## **Andean headwater and piedmont streams are hot spots of carbon dioxide and methane emissions in the Amazon basin.**

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### **Abstract**

Rivers substantially contribute to global greenhouse gas emissions, yet emissions from headwater streams are poorly constrained. Here, we report dissolved concentrations of carbon dioxide (CO<sub>2</sub>), methane (CH<sub>4</sub>) and nitrous oxide (N<sub>2</sub>O) in Andean headwater and piedmont streams in the Napo River basin in Ecuador, part of the Amazon River catchment. Concentrations increased exponentially with elevation decrease between 3990 and 175 m above sea level. Concentration changes scaled with catchment slope, and were attributed to variations in gas transfer velocity, forest cover, inundation extent, and water temperature. We estimate river emissions across the whole Amazon basin using existing data for the lowland Central Amazon. We find that Andean mountainous headwater and piedmont streams are hotspots of CO<sub>2</sub> and CH<sub>4</sub> emission, with respective areal fluxes being 1.7 and 4.5 higher in headwater streams, and 1.2 and 6.6 higher in piedmont streams than in lowland streams. Together, Andean mountainous headwater and piedmont streams and rivers represented 35% CO<sub>2</sub> and 72% CH<sub>4</sub> of basin scale integrated fluvial diffusive emissions. Conversely, N<sub>2</sub>O emissions from headwater and piedmont streams were low compared to lowland streams.



## Introduction

Rivers and streams contribute substantially to global natural and anthropogenic emissions of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O (Maavara et al., 2019a; Raymond et al., 2013; Stanley et al., 2015). Estimates of CO<sub>2</sub> emissions from rivers range between 0.7 and 2.9 PgC (10<sup>15</sup> gC) yr<sup>-1</sup> (Lauerwald et al., 2015; Raymond et al., 2013; Sawakuchi et al., 2014) and are comparable to the land net CO<sub>2</sub> sink of ~2 PgC yr<sup>-1</sup> (Friedlingstein et al., 2022). Estimates of CH<sub>4</sub> emissions from rivers range between 7 and 27 TgCH<sub>4</sub> (10<sup>12</sup> gCH<sub>4</sub>) yr<sup>-1</sup> (Rosentreter et al., 2021; Stanley et al., 2015), which is lower than the source of CH<sub>4</sub> from freshwater wetlands of ~150 TgCH<sub>4</sub> yr<sup>-1</sup>. Collectively, aquatic ecosystems (freshwater wetlands, rivers, streams, lakes, ponds, reservoirs, rice paddies, aquaculture ponds, coastal vegetated habitats, estuaries and ocean) contribute 41% of total global CH<sub>4</sub> emissions from natural and anthropogenic sources (Rosentreter et al., 2021). Estimates of N<sub>2</sub>O emissions from rivers have been dramatically revised downwards from early estimates of 2,100 GgN<sub>2</sub>O-N (10<sup>9</sup> gN<sub>2</sub>O-N) yr<sup>-1</sup> (Kroeze et al., 2010) to 32 GgN<sub>2</sub>O-N yr<sup>-1</sup> (Hu et al., 2016). This large difference is attributable to N<sub>2</sub>O riverine emissions being computed from nitrogen (N) deposition and emissions factors that are uncertain (Clough et al., 2006) and have been revised downwards (Webb et al., 2021).

The majority of riverine GHG emissions are from the tropics, ~60-80% for CO<sub>2</sub> (Borges, et al., 2015; Borges, Darchambeau et al., 2015a; Lauerwald et al., 2015; Raymond et al., 2013; Rosentreter et al., 2021; Sawakuchi et al., 2014), ~70% for CH<sub>4</sub> (Sawakuchi et al., 2014), and ~79% for N<sub>2</sub>O (Hu et al., 2016). Yet, the total stream surface area (SSA) in the tropics (24°N-24°S) corresponds to ~44% of the total SSA globally, implying that riverine areal emissions of GHGs (per m<sup>2</sup>) are higher than in other climatic zones. The higher areal GHG emission rates in the tropics compared to other climate zones is partly related to year-round high temperature that sustains high microbial activity leading to GHG production in soils and in-stream. Additionally, tropical forests with high organic carbon biomass (55% of total) (Pan et al., 2011) and extensive wetlands (Fluet-Chouinard et al., 2015) sustain an important supply of organic carbon to rivers that fuel in-stream microbial production of GHGs.

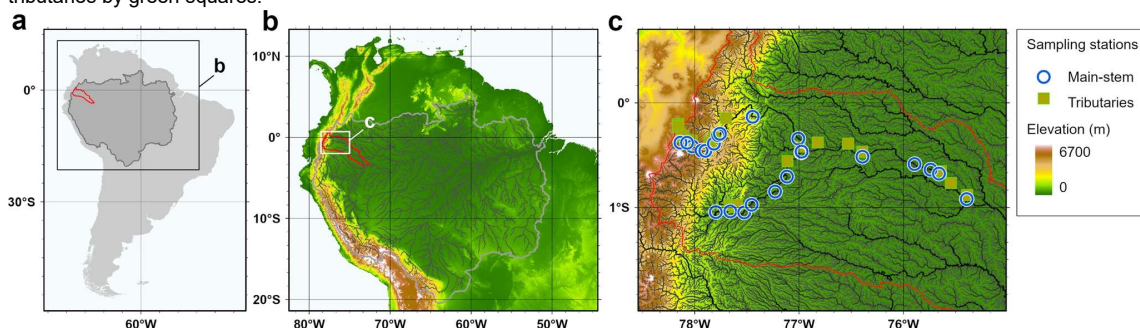
Yet, the estimates of GHG emissions from tropical rivers and streams remain highly uncertain for several reasons. One being the scarcity of data to characterize the variability within river basins, related for example to differences in river size and land cover on the catchment, but also among river basins, related for example to differences in climatic zones and concomitant differences in land cover. Additionally, emissions of GHGs are disproportionately higher in headwaters and low order streams (Marx et al., 2017), including mountainous reaches (Horgby et al., 2019). For example, in the Congo River, streams with a Strahler order 1 alone accounted for 62%, 59%, and 45% of basin wide total CO<sub>2</sub>, CH<sub>4</sub> and, N<sub>2</sub>O emissions, respectively (Borges et al., 2019). Streams with Strahler order 1 and 2 accounted, together, for even more, 79%, 77% and, 59% of basin wide total CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O emissions, respectively (Borges et al., 2019).. Higher CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions from low order streams result from a higher input of GHGs from soil-water and ground-water in low order streams, although highly variable in space and seasonally because soil-water inputs occur during short-lived high-flow events that promote shallow flow paths (Duvert et al., 2018). Additionally, low order streams and in particular mountainous ones are characterized by high values of gas transfer velocity (*k*) (Ulseth et al., 2019). The strong degassing to the atmosphere due to high *k* values results in concentrations of CO<sub>2</sub> and CH<sub>4</sub> close to equilibrium with the atmosphere in mountainous streams (Crawford et al., 2015) Conversely, in lowland river reaches, the presence of wetlands strongly enhances the fluvial riverine CO<sub>2</sub> and CH<sub>4</sub> emissions (Abril et al., 2014; Abril & Borges, 2019; Borges, Abril, et al., 2015b; Borges, Darchambeau, et al., 2015a; Borges et al., 2019). There is then a need to describe the spatial variations of GHG fluxes from headwaters to lowlands and understand the underlying drivers. This knowledge should allow choosing the appropriate landscape metrics (catchment slope, land vegetation cover, inundation cover, precipitation, among others) and/or stream attributes (river size, stream gradient, discharge, among others) to statistically model and

upscale GHG concentrations and  $k$  and compute and integrate riverine GHG fluxes, either at local scale (Borges et al., 2019) or globally (Lauerwald et al., 2015; Liu et al., 2022).

The Amazon is the largest river on the Earth in terms of freshwater discharge and watershed surface area; its basin covers a large portion of the humid Neotropics. The lowland Amazon rivers and streams (channels only at an elevation < 500 m) were reported to emit CO<sub>2</sub> at a rate ranging between 0.09 (Liu et al., 2022) and 0.14 (Melack, 2017) PgC yr<sup>-1</sup>. Additionally, headwater Amazonian streams emit CO<sub>2</sub> at a rate ranging between 0.09 (Johnson et al., 2008) and 0.17 (Liu et al., 2022) PgC yr<sup>-1</sup>. The total CO<sub>2</sub> emission from the Amazon running waters (rivers and floodplains) range between 0.5 (Richey et al., 2002) and 1.4 PgC yr<sup>-1</sup> (Sawakuchi et al., 2017). Rivers and floodplains of the Amazon River entire basin emit CH<sub>4</sub> rate of ~22 TgCH<sub>4</sub> yr<sup>-1</sup>, but rivers represented about ~0.6% of the total flux, although only integrated for the Solimões-Amazon mainstem (Melack et al., 2004). The CH<sub>4</sub> emissions from large rivers of the Amazon River have been evaluated to 0.5 TgCH<sub>4</sub> yr<sup>-1</sup> (Sawakuchi et al., 2014), and CH<sub>4</sub> emissions for the whole lowland Amazon basin (elevation < 500m) to 0.3 TgCH<sub>4</sub> yr<sup>-1</sup> (Barbosa et al., 2016). The total emission of CH<sub>4</sub> from Amazonia (including all terrestrial and aquatic compartments) converge to ~40 TgCH<sub>4</sub> yr<sup>-1</sup> and have been evaluated with a variety of methods (Basso et al., 2021; Pangala et al., 2017), either based on atmospheric observations (“top down” approach) or upscaling of flux measurements in various relevant components (“bottom up” approach).

Research on GHGs in rivers and streams has mainly focussed on the Central Amazon (Abril et al., 2014; Alin et al., 2011; Barbosa et al., 2016; Borges et al., 2015a; Call et al., 2018; de Fátima F. L. Rasera et al., 2008; Johnson et al., 2008; Melack et al., 2004; Richey et al., 2002; Sawakuchi et al., 2017, 2014; Scofield et al., 2016). Most research has focussed on high order streams and rivers, and only occasionally in headwater streams (Johnson et al., 2008a; Schneider et al., 2020). Here, we report data of the dissolved CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O concentration in the Andean headwater mountainous and piedmont rivers and streams of the Napo basin in Ecuador (Fig 2.1 and Supplementary Fig 2.1). We describe gradients of dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O with elevation and their drivers in terms of catchment attributes (slope and land cover), and we model gas emissions at the scale of whole Amazon basin using in addition for lowland rivers a compilation of published data.

FIGURE 2.1 Location of sampling stations in Andean headwater and piedmont streams of the Amazon basin a-c. The outline of the catching of the Napo River is shown in red, stations in the main-stem are shown by blue circles and in the tributaries by green squares.



## Methods

### Site description

The Napo River catchment is ranges from 100 to 6,300 m above sea level, and has a total area of 100,500 km<sup>2</sup>, distributed among Ecuador (59%), Peru (40%) and Colombia (1%). The Ecuadorian part of the Andean Cordillera is the narrowest within the Central Andes and is characterized by an important volcanic and seismic activity. Below an elevation of 500 m is located a fairly flat and homogeneous forest, part of the Amazonian plains, that covers 82% of

the total surface of the Napo basin. The rest of the basin (18%) is constituted by the Andean Cordillera with a vegetation cover composed by grasslands (puna and páramo). The drainage basin receives on average 3,200 mm of rainfall per year. At high elevations, rainfall is bimodal and is maximal in April and October and minimal in July-August and in January. In the upper basins, two discharge maxima occur during the two rainfall maxima. In the lowlands, seasonal variability of rainfall and discharge is very weak. The annual discharge of the Napo is 6,300 m<sup>3</sup> s<sup>-1</sup> corresponding to 4% of the Amazon River total discharge at Obidos. The Napo is a 'white water' river type characterized by high levels of total suspended matter, and the sediment transport of the Napo River corresponds to 6% of the sediment flow of the Amazon River at Obidos (Hall et al., 2016).

#### Sampling and chemical analysis

Sampling was carried out on eight occasions (October 2018 and 2019, January 2019 and 2020, April 2019 and 2021, July 2019 and 2020) (Supplementary Figure 2.1). Stations A1 to A21 were sampled travelling by car, and the others travelling by boat. Sampling was carried out during day-time only (early morning to late afternoon). In some streams, during day-time, oxygen concentrations can strongly increase (Gómez-Gener et al., 2021), and, conversely, CO<sub>2</sub> concentrations can decrease (Descy et al., 2017), in response to aquatic photosynthesis. Fluvial CO<sub>2</sub> emissions can, thus, be under-estimated (Descy et al., 2017) if data are only acquired during day-time, because night-time CO<sub>2</sub> concentrations and fluxes can be higher than during day-time. Note that a day-time decrease of stream CO<sub>2</sub> and increase of O<sub>2</sub> are not necessarily systematic patterns that depend on the level of in-stream productivity. In lowland streams, rich in DOC, such as in the Congo River network, day-night differences of pCO<sub>2</sub> were low (Borges et al., 2019) due to low planktonic primary production and much smaller than the differences among streams (spatial gradients).

Water temperature, specific conductivity, pH, and %O<sub>2</sub> were measured in-situ with a YSI multi-parameter probe (ProPlus). Water for CH<sub>4</sub> and N<sub>2</sub>O samples was collected with a sampling devise consisting of a 2L polyethylene bottle with the bottom cut and fitted with a silicone tubing at the stopper. Two borosilicate serum bottles (Weathon) with a volume of 40 ml were filled with the silicone tubing, poisoned with 100 µl of a saturated solution of HgCl<sub>2</sub> and sealed with a butyl stopper and crimped with an aluminium cap. Measurements were made, after over-night equilibration, on an headspace (Dickson et al., 2007) (created by injecting 15 ml of high-purity N<sub>2</sub> into the 40 ml sample bottles), with a gas chromatograph (SRI 8610C) with a flame ionisation detector for CH<sub>4</sub> and electron capture detector for N<sub>2</sub>O calibrated with CH<sub>4</sub>:N<sub>2</sub>O:N<sub>2</sub> gas mixtures (Air Liquide Belgium) with mixing ratios of 1, 10 and 30 ppm for CH<sub>4</sub>, and 0.2, 2.0 and 6.0 ppm for N<sub>2</sub>O. The precision of measurement based on duplicate samples was ±10.9% for CH<sub>4</sub> and ±5.8% for N<sub>2</sub>O based on 309 replicates.

pCO<sub>2</sub> was measured in the field with a Li-Cor Li-820 infra-red gas analyser based on the headspace technique with 4 polypropylene syringes that were filled directly with river water. The pCO<sub>2</sub> in the atmosphere was measured by injecting ambient air sampled with an additional polypropylene syringe. The Li-Cor Li-820 was calibrated with pure N<sub>2</sub> and CO<sub>2</sub>:N<sub>2</sub> gas mixtures (Air Liquide Belgium) of 388, 804, 3707 and 8146 ppm. The final pCO<sub>2</sub> value was computed taking into account the partitioning of CO<sub>2</sub> between water and the headspace, as well as equilibrium with HCO<sub>3</sub><sup>-</sup> 110 using water temperature measured in-stream and after equilibration, and total alkalinity (data not shown). Samples for total alkalinity were conditioned, stored and analysed as reported. The precision of pCO<sub>2</sub> measurement was ±5.2%.

The CO<sub>2</sub> concentration is expressed as partial pressure in parts per million (ppm) and as dissolved concentration for CH<sub>4</sub> (nmol L<sup>-1</sup>), in accordance with convention in existing topical literature. Variations of N<sub>2</sub>O were modest and concentrations fluctuated around atmospheric equilibrium, so data are presented as percent of saturation level (%N<sub>2</sub>O, where atmospheric equilibrium corresponds to 100%), computed from the global mean N<sub>2</sub>O air mixing ratios given by the Global Monitoring Division (GMD) of the Earth System Research Laboratory (ESRL) of the

National Oceanic and Atmospheric Administration (NOAA) (<https://www.esrl.noaa.gov/gmd/hats/combined/N2O.html>), using the Henry's constant (APHA, 1998).

Water was collected in the stream surface with a 2L polyethylene bottle. The water filtered through GF/F Whatman glass fiber filters was collected and further filtered through polyethersulfone syringe encapsulated filters (0.2 µm porosity) for nitrate (NO<sub>3</sub><sup>-</sup>), nitrite (NO<sub>2</sub><sup>-</sup>) and ammonium (NH<sub>4</sub><sup>+</sup>). Samples were stored frozen (-20°C) in 50 ml polypropylene vials. NO<sub>3</sub><sup>-</sup> and NO<sub>2</sub><sup>-</sup> were determined with the sulfanilamide colorimetric with the vanadium reduction method (Standing committee of Analysts., 1981), and NH<sub>4</sub><sup>+</sup> with the dichloroisocyanurate-salicylate-nitroprussiate colorimetric method (Abril et al., 2015). Detection limits were 0.3, 0.01, and 0.15 µmol L<sup>-1</sup> for NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>, respectively. Precisions were ±0.02 µmol L<sup>-1</sup>, ±0.02 µmol L<sup>-1</sup>, and ±0.1 µmol L<sup>-1</sup> for NH<sub>4</sub><sup>+</sup>, NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup>, respectively.

#### Computation and upscaling of FCO<sub>2</sub>, FCH<sub>4</sub>, FN<sub>2</sub>O

The dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O were derived from statistical relations as function of catchment slope for each of the river segments in RiverATLAS (Linke et al., 2019) (n=744,786). The statistical relations we derived from our own data in the Napo and data available from literature in the lowland rivers and streams of the Central Amazon (Figure 2.4).

The pCO<sub>2</sub> (in ppm) was computed according to:

$$\log(\text{pCO}_2) = 3.57(\pm 0.02) - 0.0043(\pm 0.0002) \times \text{CS} \quad (1)$$

$$(r^2=0.59, p<0.0001, n=473)$$

where CS is the catchment slope (x10°).

The CH<sub>4</sub> (in nmol L<sup>-1</sup>) was computed for an elevation > 180 m according to:

$$\log(\text{CH}_4) = 3.18(\pm 0.08) - 0.0081(\pm 0.0005) \times \text{CS} \quad (2)$$

$$(r^2=0.42, p<0.0001, n=305)$$

and for an elevation < 180 m according to:

$$\log(\text{CH}_4) = 2.60(\pm 0.08) - 0.0178(\pm 0.003) \times \text{CS} \quad (3)$$

$$(r^2=0.21, p<0.0001, n=174)$$

The %N<sub>2</sub>O (in %) was computed for an elevation > 200 m according to:

$$\% \text{N}_2\text{O} = 106.9(\pm 2.3) - 0.048(\pm 0.014) \times \text{CS} \quad (4)$$

$$(r^2=0.04, p=0.0011, n=277)$$

and for an elevation of < 200 m the median of values reported by Richey et al. 2016 (205.6%) was applied uniformly.

To avoid the computation of unrealistic values outside the determination bounds of the regressions, for CS > 200 (x10°), a constant value (based on the median of observations within that range of slope values) was applied: 542 ppm for pCO<sub>2</sub> and 96.3 % for %N<sub>2</sub>O. For CH<sub>4</sub>, for an elevation > 180 m, a constant value of 18.5 nmol L<sup>-1</sup> was applied for a CS > 200 (x10°); for an elevation of < 180 m, a constant value of 60.0 nmol L<sup>-1</sup> was applied for a CS > 50 (x10°).

The water temperature (WT in °C) was computed from the air temperature (AT in x10°C) from RiverATLAS (Linke et al., 2019), according to:

$$\text{WT} = 0.00054 \times \text{AT}^2 - 0.0964 \times \text{AT} + 15.2 \quad (5)$$

( $r^2=0.93$ ,  $n=371$ )

$\text{FCO}_2$ ,  $\text{FCH}_4$ ,  $\text{FN}_2\text{O}$  between surface waters and the atmosphere was computed according to:

$$F = k \times \Delta G \quad (6)$$

where  $k$  is the gas transfer velocity ( $\text{m d}^{-1}$ ) and  $\Delta G$  is the air-water gradient of a given gas.

The  $k$  normalized to a Schmidt number ( $Sc$ ) of 600 ( $k_{600}$  in  $\text{m d}^{-1}$ ) was derived from the parameterisation as a function slope and stream water velocity (Raymond et al., 2012):

$$k_{600} = 2.02(\pm 0.21) + 2841(\pm 107) \times V \times S \quad (7)$$

where  $V$  is stream velocity ( $\text{m s}^{-1}$ ) and  $S$  is stream slope (unitless).

$k_{600}$  was converted to  $k$  for the  $Sc$  at in-situ temperature according to:

$$k = k_{600} \times (Sc/600)^{-0.5} \quad (8)$$

The values of  $S$  are given by RiverATLAS (Linke et al., 2019) and  $V$  was computed from discharge ( $Q$  in  $\text{m}^3 \text{s}^{-1}$ ) also given by RiverATLAS (Linke et al., 2019) according to (Liu et al., 2022):

$$\ln(V) = 0.12 \times \ln(Q) - 1.06 \quad (9)$$

A Monte-Carlo error analysis was made by propagating errors (10,000 iterations) on the slope and constant of the equations to compute  $k_{600}$  and the gas concentrations (refer to above equations), and an error of  $\pm 1^\circ\text{C}$  for water temperature.

## **Results and discussion**

### Altitudinal patterns of stream GHG content

We sampled 25 stations in the mainstem Napo and Coca rivers and 19 stations in the tributaries along an elevation gradient from 3990 m to 175 m above sea level, on 8 occasions from 2018 to 2021 (Figure 2.1 and Supplementary Figure 2.1). In surface waters of the sampled streams, the values of the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ) ranged between 152 and 10,873 ppm, and dissolved  $\text{CH}_4$  concentrations between 4 and 38,666  $\text{nmol L}^{-1}$  (Figure 2.2a,e). Values of  $p\text{CO}_2$  below atmospheric equilibrium ( $\sim 404$  ppm, average of in-situ measurements) were only observed in 22 cases (7% of the total number of observations), while dissolved  $\text{CH}_4$  concentration values in surface waters were always above atmospheric equilibrium ( $\sim 2$   $\text{nmol L}^{-1}$ , based on the global average  $\text{CH}_4$  mixing ratio of 1.9 ppm).

## Greenhouse gas emissions in Napo Basin and Andean Lakes

FIGURE 2.2 Strong elevation gradients of  $p\text{CO}_2$ ,  $\text{CH}_4$  and  $\% \text{N}_2\text{O}$  in the Andean headwater and piedmont streams. Variations of the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ),  $\text{CH}_4$  dissolved concentration,  $\text{N}_2\text{O}$  saturation level ( $\% \text{N}_2\text{O}$ ), and oxygen saturation level ( $\% \text{O}_2$ ) in the Napo streams as a function of elevation (a, e, i, m), catchment slope (b, f, j, n), forest cover (c, g, k, o), and water temperature (d, h, l, p). Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and black lines indicate linear or exponential fits to the data (Supplementary Table 2.4).

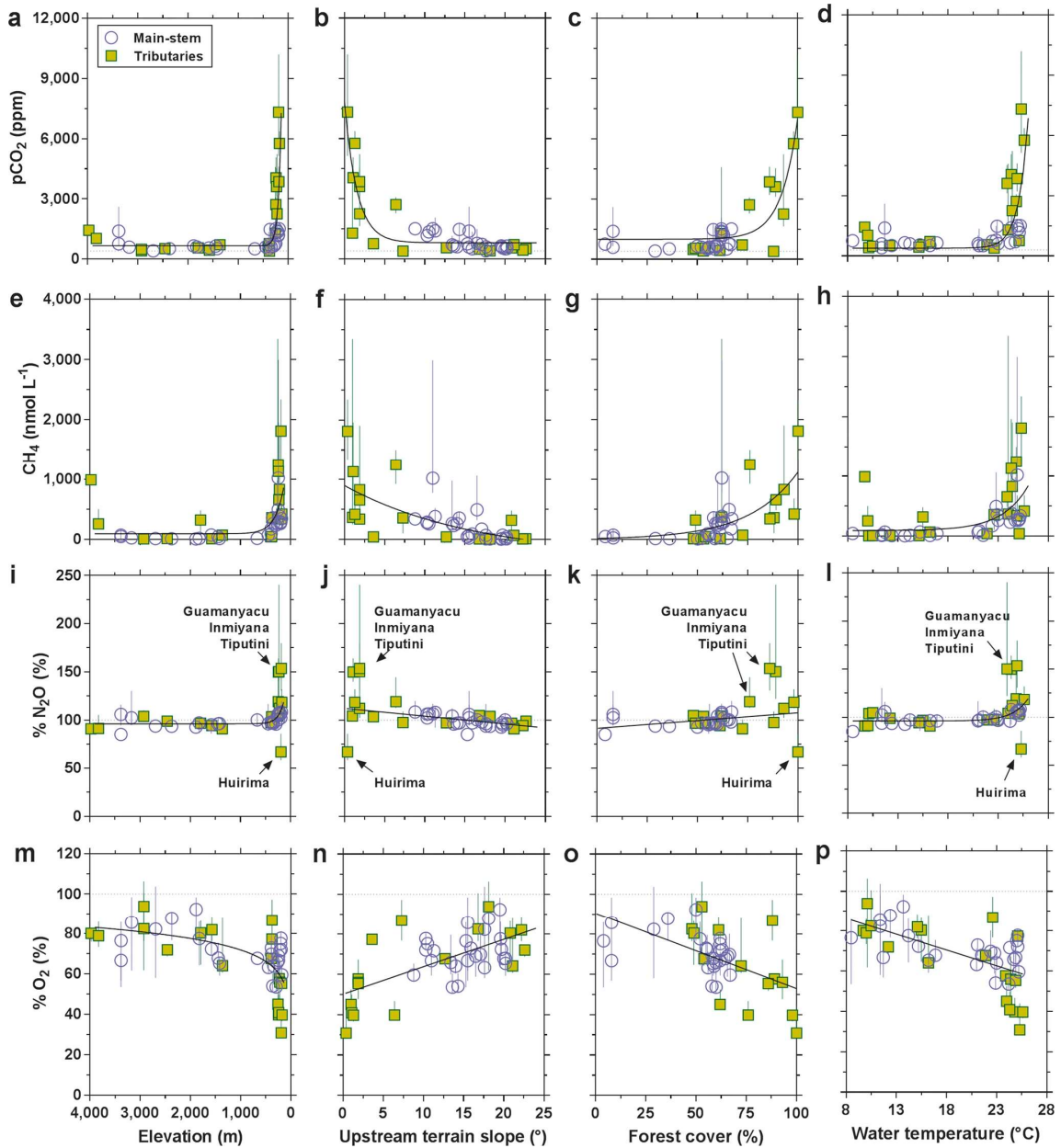
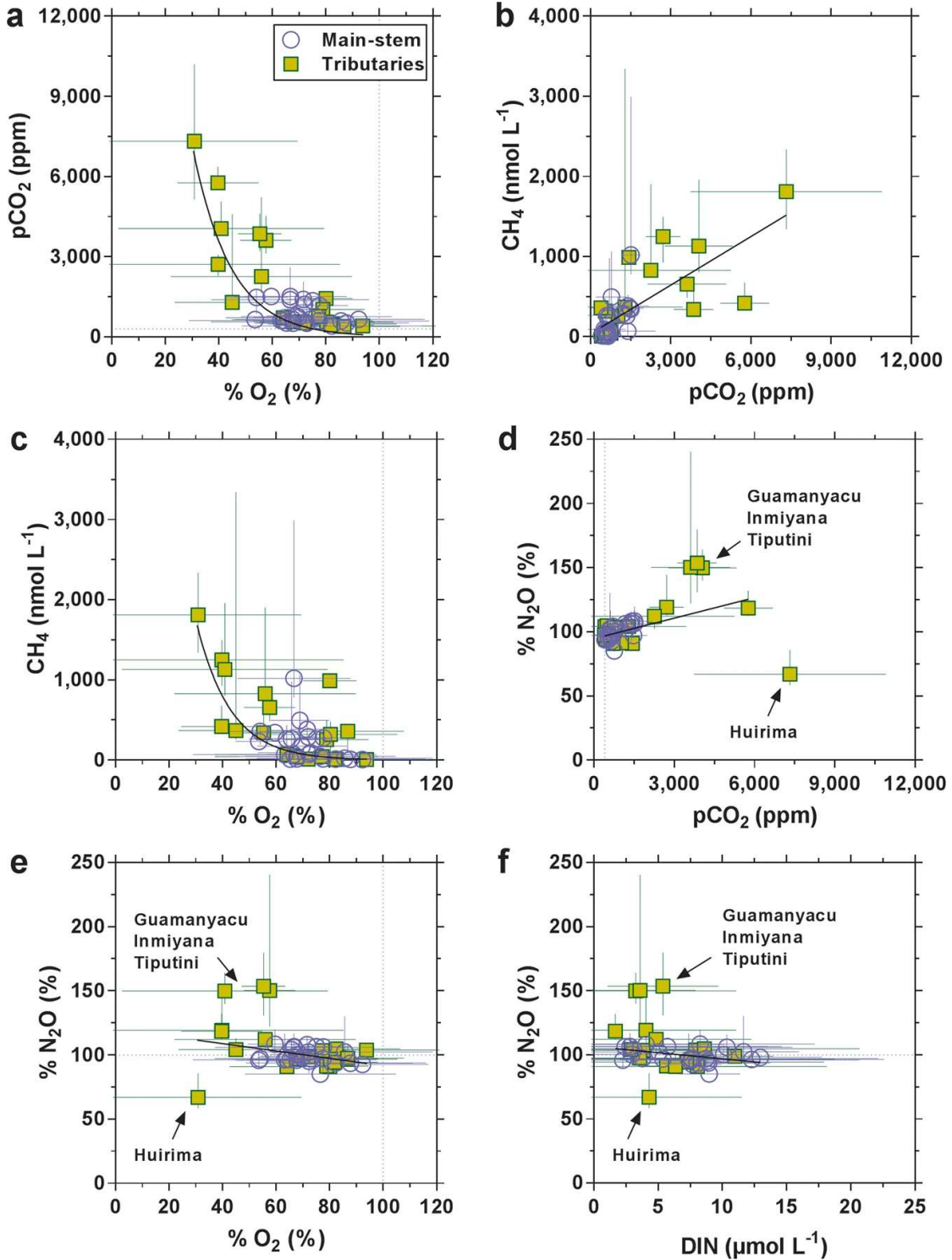


FIGURE 2.3 Strong correlations between  $p\text{CO}_2$ ,  $\text{CH}_4$ ,  $\% \text{N}_2\text{O}$  and  $\% \text{O}_2$  in the Andean headwater and piedmont streams. Variations of the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ),  $\text{CH}_4$  dissolved concentration,  $\text{N}_2\text{O}$  saturation level ( $\% \text{N}_2\text{O}$ ), as a function of oxygen saturation level ( $\% \text{O}_2$ ) (a, c, e),  $\text{CH}_4$  and  $\% \text{N}_2\text{O}$  as a function of  $p\text{CO}_2$  (b, d), and  $\% \text{N}_2\text{O}$  as a function of dissolved inorganic nitrogen (DIN) (f) in the Napo streams. Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and black lines indicate linear or exponential fits to the data. Solid lines show fits to the data (Supplementary Table 2.4).



The pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values in streams at elevations > 500 m (median 560 ppm and 17 nmol L<sup>-1</sup>, respectively) were distinctly lower than at elevations < 500 m (median 1,280 ppm and 370 nmol L<sup>-1</sup>, respectively). The pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration in streams increased exponentially from highlands to lowlands (Figure 2.2a,e). The patterns of oxygen saturation level (%O<sub>2</sub>) in streams generally mirrored those of pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration. The %O<sub>2</sub> values in streams at elevations > 500 m (median 74.4 %) were higher than at elevations < 500 m (median 63.2 %), and %O<sub>2</sub> decreased exponentially from highlands to lowlands (Figure 2.2m).

The pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values were positively correlated (Figure 2.3b), and they were negatively correlated to %O<sub>2</sub> (Figure 2.3a,c), strongly suggesting that pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration in streams were controlled by the same combination of input and output processes (that also involved opposite changes of O<sub>2</sub>). A likely important output process of CO<sub>2</sub> and CH<sub>4</sub> is the loss to the atmosphere that depends on k. The calculated k values increased with increasing elevation (Supplementary Figure 2.2b). Higher k values strongly promoted the emissions of CO<sub>2</sub> and CH<sub>4</sub> to the atmosphere, contributing to drive pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentrations to values close to saturation in the highland streams compared to lowland streams, and could explain in part the decreasing pattern of pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentrations with elevation. A likely important input process of CO<sub>2</sub> and CH<sub>4</sub> is organic matter mineralization in-stream, in fringing wetlands, and in terra firme soils of the watershed. Altitudinal changes of organic matter mineralization might have resulted from changes in temperature, soil organic carbon content and land cover. Water temperature strongly increased from highlands to lowlands, from ~13.5°C > 500 m elevation to ~24.3°C < 500 m (Figure 2.S2a). Rates of respiration (Cardoso et al., 2014; Kirschbaum, 1995) and methanogenesis (Yvon-Durocher et al., 2014) in soils and riverine sediments are positively related to temperature, suggesting the warmer conditions in the lowland rivers contributed to higher pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values. Indeed, pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values also increased exponentially with water temperature (Figure 2.2d,h). Changes of water temperature could contribute to explain the decreasing pattern of pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration with elevation.

There were also changes of land cover with elevation with a transition from a dominance of grasslands in uplands (known as páramos) to forests in lowlands (Supplementary Figure 2.3a,b) (McClain et al., 1995). Additionally, the presence of fringing wetlands (indicated by inundation extent) strongly increased in the piedmont streams (Supplementary Figure 2.3c). Stream pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values increased exponentially with forest cover (Fig 2c,g), and were generally higher where grassland cover was low to zero (Supplementary Figure 2.4a,c). Fluvial pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentrations values were positively related to inundation extent in the reach with different slopes for tributaries and main-stem (Supplementary Figure 2.4b,d). Fringing wetlands might have a local impact on pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration in smaller streams (tributaries) in the piedmont region and explain in part the increasing pattern of pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration with decreasing elevation. Lateral inputs from fringing wetlands strongly influence the fluvial content of dissolved organic carbon (DOC) (McClain et al., 1997) and CO<sub>2</sub> (Abril et al., 2014) in the Central Amazon River. Note that inundation extent was locally high in some of the lowland streams at reach level, but for the whole upstream catchment the inundation extent values were much lower (≤2%). Wetlands have been shown to strongly influence pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values in lowland African rivers where the wetland cover was higher for the whole upstream catchment (Borges et al., 2015d).

The k values strongly depend on stream gradient and water flow (Raymond et al., 2012) that both scale with catchment slope, that itself varies with elevation. Vegetation cover (forest and riparian wetlands) and temperature also vary as function of elevation, so, also with catchment slope. Consequently, catchment slope could be used as an integrative metric for processes that affect GHGs. Indeed, pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values in the sampled streams were found to be negatively related to catchment slope, while %O<sub>2</sub> was positively related to catchment slope (Figure 2.2b,f,n). The N<sub>2</sub>O saturation levels (%N<sub>2</sub>O) in streams at an altitude > 500 m



(median 95.4 %) were lower than at elevations < 500 m (median 105.0 %) (Figure 2.2i). The %N<sub>2</sub>O increased exponentially with decreasing elevation, although the exponential increase was less steep than for pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration (Figure 2.2). Lower N<sub>2</sub>O in high elevation streams was consistent with reported elevation gradients of N<sub>2</sub>O soil-air fluxes in the Ecuadorian Andes, with a soil N<sub>2</sub>O sink at high elevation (2,200-3,010 m) and a N<sub>2</sub>O source from soils to the atmosphere at lower elevation (400-1,100 m)<sup>48</sup>. The low fluvial %N<sub>2</sub>O probably resulted from inputs to river reaches of soil water with low N<sub>2</sub>O content. The removal of N<sub>2</sub>O in soils was due to complete denitrification (reduction of N<sub>2</sub>O to N<sub>2</sub>) in highlands related to low soil dissolved inorganic N (DIN) content (Pineda Lampreda et al., 2021). Conversely, the N<sub>2</sub>O emissions from soils to the atmosphere from lowland soils to the atmosphere was related to high DIN content in soils combined to higher temperature and higher soil water content (Pineda Lampreda et al., 2021). The DIN in streams (Supplementary Figure 2.5c) was higher in highland streams than in lowland streams, so did not seem to follow the pattern of soil DIN that was higher in lowlands than highlands (Pineda Lampreda et al., 2021). The same elevation pattern of stream DIN was reported in Andean streams in Venezuela (Allan et al., 2006). High DIN levels were reported in 55 streams in montane forests in Central America and the Caribbean (Brookshire et al., 2012). Large loss of inorganic nitrogen by leakage to streams from unpolluted mature tropical forests occurs ubiquitously across a range of soil types. The higher DIN in highland streams could be related to higher transfer of DIN from soils to streams related to a lower retention due to differences in vegetation types (grassland versus forest) (Templer et al., 2012) or higher catchment slopes that promote soil-river transfer of DIN (Becker et al., 2014; Fu et al., 2004). In addition, there could have been a higher DIN uptake in the water column by primary producers (phytoplankton or macrophytes) related to decrease in flow and an increase in water residence. The combination of higher stream DIN and low %N<sub>2</sub>O in highland streams, and the opposite pattern in lowland streams led to a negative relation between %N<sub>2</sub>O and DIN (Figure 2.3f). Such a negative relation is unique compared to patterns previously reported, as N<sub>2</sub>O content in rivers is in general positively related to DIN (Herreid et al., 2021; Rosamond et al., 2012; Zhang et al., 2022).

Four streams showed %N<sub>2</sub>O values diverging from general patterns, the Huirima and a cluster of three streams (Guamanyacu, Inmiyana, Tiputini). The Huirima showed distinctly lower %N<sub>2</sub>O (67%) than the other lowland streams at equivalent elevations (104-119%) but also the lowest %O<sub>2</sub> (30.9%) and highest pCO<sub>2</sub> (7,317 ppm) values (Figure 2.3d,e). The Huirima stream was also characterized by much higher DOC concentrations than all the other streams, suggesting a stronger removal of N<sub>2</sub>O by denitrification in soils and/or sediments associated to streams with high DOC and low %O<sub>2</sub>, as also shown in black-water rivers and streams of the Congo basin (Borges et al., 2019). The Huirima was characterized by the lowest catchment terrain slope (0.4°) among the sampled streams. Aerial view (Supplementary Figure 2.6) showed a dense forest cover and suggested extensive inundation of the Huirima river that probably promoted conditions favorable for complete denitrification and N<sub>2</sub>O removal (reducing conditions associated to high DOC and low %O<sub>2</sub>).

The Guamanyacu, Inmiyana, and Tiputini streams drain (in part) oil palm plantations (Supplementary Figure 2.7a-c) that could explain the higher stream %N<sub>2</sub>O levels (150-153%) compared to surrounding streams (104-119%). In the area, artificial fertilizers are not used in the oil palm plantations, and accordingly, the DIN levels in these streams were comparable to the others at same elevation (Figure 2.3f). However, the conversion of natural forest to oil palm plantations itself leads to an increase of N<sub>2</sub>O production in soils due to changes in N microbial cycling due to disturbance of the soils (Skiba et al., 2020). Overall, oil palm plantations did not seem to lose markedly DIN to rivers but there seemed to be intense N<sub>2</sub>O production in soils that was transported to streams.

Stream size has been shown to be an important predictor of CO<sub>2</sub> and CH<sub>4</sub> emissions that scale with stream Strahler order in boreal (Wallin et al., 2018), temperate (Borges et al., 2019b; Butman and Raymond, 2011) and tropical river basins. In highland streams (elevation > 500 m), there were no clear patterns with stream size for pCO<sub>2</sub>, dissolved CH<sub>4</sub> concentration, %N<sub>2</sub>O and

%O<sub>2</sub> (Supplementary Figure 2.8e-h). In the lowland streams, there was an increasing pattern with stream size of pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration (mirrored by a decreasing pattern of %O<sub>2</sub>) (Supplementary Figure 2.8i-l) as frequently observed in other river networks (Borges et al., 2019a, 2018; Butman and Raymond, 2011; Wallin et al., 2018). However, when streams were taken together irrespective of altitude, there was no clear relation with streams size for pCO<sub>2</sub>, dissolved CH<sub>4</sub> concentrations, %N<sub>2</sub>O and %O<sub>2</sub> (Supplementary Figure 2.8a-d), suggesting that the effect of altitudinal gradient overwhelmed the effect of stream size on the variability of these quantities. Exchange of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O with the atmosphere from the Napo river network

The air-water diffusive flux of CO<sub>2</sub> (FCO<sub>2</sub>) ranged between -1 and 1,074 mol m<sup>-2</sup> yr<sup>-1</sup>, and was higher in tributaries (median 128 mol m<sup>-2</sup> yr<sup>-1</sup>) than in the main-stem (median 34 mol m<sup>-2</sup> yr<sup>-1</sup>). The air-water diffusive flux of CH<sub>4</sub> (FCH<sub>4</sub>) ranged between 45 and 8,985 mmol m<sup>-2</sup> yr<sup>-1</sup>, and was generally higher in tributaries (median 999 mmol m<sup>-2</sup> yr<sup>-1</sup>) than in the main-stem (332 mmol m<sup>-2</sup> yr<sup>-1</sup>). Both FCO<sub>2</sub> and FCH<sub>4</sub> were inversely correlated to elevation and upstream terrain slope (Supplementary Figure 2.9a,b,e,f). The air-water diffusive flux of N<sub>2</sub>O (FN<sub>2</sub>O) ranged between -23 and 27 mmol m<sup>-2</sup> yr<sup>-1</sup> and did not show marked difference between tributaries and main-stem. There was a general tendency towards a sink of N<sub>2</sub>O at an elevation > 1,000m with 12 sites showing negative FN<sub>2</sub>O values and only 4 showing positive FN<sub>2</sub>O values (Supplementary Figure 2.9i,j). At an elevation < 1,000m the FN<sub>2</sub>O values were close to zero (-1.3 to 3 mmol m<sup>-2</sup> yr<sup>-1</sup>), except for the Guamanyacu, Inmiyana, and Tiputini streams that showed distinctly positive FN<sub>2</sub>O values (9.7 to 21.6 mmol m<sup>-2</sup> yr<sup>-1</sup>) and the Huirima that showed a distinctly negative FN<sub>2</sub>O value (-7.6 mmol m<sup>-2</sup> yr<sup>-1</sup>).

In river geomorphology, the concept of sediment transport capacity has been used to determine if bedload transport is primarily constrained by sediment supply (supply-limited rivers) or river flow hydraulics (transport-limited rivers) (Wainwright et al., 2015). This concept has been extended by some authors to fluvial CO<sub>2</sub> and CH<sub>4</sub> fluxes (Crawford et al., 2015; Rocher-Ros et al., 2019; Schneider et al., 2020). The underlying idea being that in highly turbulent systems (high k values) degassing brings gases close to equilibrium with atmosphere and emissions are close to zero, consequently the gaseous emission is assumed limited by gas availability. Transport- or supply-limitation of the gaseous exchange with the atmosphere can be visualized by plots of gas concentration versus k600 (Rocher-Ros et al., 2019) (Supplementary Figure 2.10). At an elevation above 1,500 m, computed k600 values were higher than 25 m d<sup>-1</sup> while pCO<sub>2</sub> (< 746 ppm) and CH<sub>4</sub> (< 72 nmol L<sup>-1</sup>) values were low indicating supply-limitation of CO<sub>2</sub> and CH<sub>4</sub> emissions (Supplementary Figure 2.10a,b). Conversely, in some of the lowland tributaries, particularly high pCO<sub>2</sub> (> 1,500 ppm) and CH<sub>4</sub> (> 500 nmol L<sup>-1</sup>) values indicate transport-limitation of CO<sub>2</sub> and CH<sub>4</sub> emissions. There was an exception for both CO<sub>2</sub> and CH<sub>4</sub>, corresponding to high concentrations values (pCO<sub>2</sub>=1,395 ppm and CH<sub>4</sub>=320 nmol L<sup>-1</sup>) in highland streams with high k600 values (74 and 88 m d<sup>-1</sup>, respectively) (indicated by asterisks in Supplementary Figs. 9 and 10). For N<sub>2</sub>O, the interpretation of concentration-k600 plot in terms of transport- or supply-limitation was not as clear as for CO<sub>2</sub> and CH<sub>4</sub> (Supplementary Figure 2.10c). The sampled streams were more or less strongly over-saturated in both CO<sub>2</sub> and CH<sub>4</sub>, consequently the flux was unidirectional from the water to the atmosphere (emission). In contrast, N<sub>2</sub>O oscillated between under- and over-saturation, corresponding to a sink or a source of N<sub>2</sub>O the atmosphere, respectively. In the case of a stream acting as a sink of a gas, the supply-limitation is from the atmosphere and not within the stream itself. In this situation, the use of concept of transport capacity as used for sediments (Wainwright et al., 2015) becomes awkward. At an elevation > 1,500 m, N<sub>2</sub>O concentrations were close to saturation in conjunction with high k600 values (Supplementary Figure 2.10c). At an elevation below 1,500 m, N<sub>2</sub>O concentrations markedly above or below saturation were observed in some streams. In this case, the direction and intensity of the N<sub>2</sub>O exchange should be more strongly driven by variations of concentrations rather than by those of k600. FCO<sub>2</sub>, FCH<sub>4</sub> and FN<sub>2</sub>O strongly correlated with pCO<sub>2</sub>, CH<sub>4</sub>, and %N<sub>2</sub>O, respectively, and did not correlate to k600 (Supplementary Figure 2.9c,g,k,d,h,l), showing that for the whole data-set the variations of gaseous exchanges with the atmosphere were strongly driven

by the variations of the concentrations (supply-limitation) rather than by  $k_{600}$  (transport-limitation). In highly turbulent systems with high  $k_{600}$  values, the gaseous exchanges with the atmosphere bring the concentrations close to equilibrium with the atmosphere, strongly decreasing the flux values that converge to zero.

#### Comparison of stream GHG content with Central Amazon basin

The  $p\text{CO}_2$  values in the rivers and streams of the Central Amazon (median 2950 ppm) were higher than piedmont (median 1280 ppm) and highland (median 555 ppm) streams of the Napo and there was a general decreasing pattern with elevation and slope (Figure 2.4a). In contrast, the  $\text{CH}_4$  concentrations in the piedmont streams (median  $341 \text{ nmol L}^{-1}$ ) were higher than in the rivers and streams of the Central Amazon (median  $120 \text{ nmol L}^{-1}$ ), that in turn were higher than in the highland streams (median  $16 \text{ nmol L}^{-1}$ ) (Figure 2.4b). There was a general decreasing tendency of  $\text{CH}_4$  as function of catchment slope, but at equivalent slope values (at the low end of the range of values),  $\text{CH}_4$  concentrations were higher in piedmont streams than lowland stream and rivers. The ratio of  $\text{CH}_4$  to  $\text{CO}_2$  dissolved concentrations ( $\text{CH}_4:\text{CO}_2$  in  $\mu\text{mol}:\mu\text{mol}$ ) followed the same patterns as dissolved  $\text{CH}_4$  concentration (Figure 2.4c), indicating that for a similar input of  $\text{CO}_2$  (lateral input from soils or in-stream mineralization) there was a larger input of  $\text{CH}_4$  in piedmont streams than in lowland rivers and streams. The forest cover and water temperatures in piedmont streams are equivalent to those in lowland rivers and streams. The higher dissolved  $\text{CH}_4$  concentration in piedmont rivers might result from the combination of relatively higher  $\text{CH}_4$  production in the piedmont streams and relatively higher losses of dissolved  $\text{CH}_4$  in the lowland rivers. The piedmont region is an area of deposition of suspended particles delivered from the highland streams originating from erosion, up to 30% of the total annual load in the Mamoré river (Vauchel et al., 2017), with an annual freshwater discharge equivalent to the one of the Napo. The deposition of sediments combined by presence of forests in the piedmont region are most probably favorable to high in-stream  $\text{CH}_4$  production. In larger lowland river channels that are relatively disconnected from floodplains, the deep water column and strong currents do not promote the accumulation of dissolved  $\text{CH}_4$  in the water column, but promote the removal of dissolved  $\text{CH}_4$  either by emission of  $\text{CH}_4$  to the atmosphere or by microbial oxidation (Borges et al., 2015).

The only  $\text{N}_2\text{O}$  data in streams and rivers of the Amazon River network were reported by Richey et al 1998. When plotted against  $p\text{CO}_2$ , the  $\%\text{N}_2\text{O}$  values were higher in the Central Amazon streams and rivers (median 205.6%) compared to piedmont (median 105.0%) and highland (median 95.7%) streams (Supplementary Figure 2.11a). We do not have a conclusive hypothesis for the higher  $\%\text{N}_2\text{O}$  in lowland streams and rivers, but the general tendency of higher  $p\text{CO}_2$  and  $\text{N}_2\text{O}$  in lowland streams and rivers than in piedmont and highland streams (Supplementary Figure 2.11b) suggests more intense mineralization in soils or in-stream leading to higher  $\text{CO}_2$  and presumably  $\text{N}_2\text{O}$  levels in rivers in the former systems (Pineda Lampreda et al., 2021). The pattern of higher  $\%\text{N}_2\text{O}$  in Central amazon streams compared to headwater and piedmont streams was also consistent with higher  $\text{N}_2\text{O}$  soil-air fluxes ( $79 \pm 7 \mu\text{gN m}^{-2} \text{ h}^{-1}$ ) in the central part of the basin from oxisols (Keller et al., 2018) that are dominant in the Amazon forests than in piedmont forests ( $8 \pm 12 \mu\text{gN m}^{-2} \text{ h}^{-1}$  at 400 m elevation) (Bernoux, 1998), Comparatively higher dissolved  $\text{CH}_4$  concentrations in piedmont streams than in lowland streams was also evident relative to  $p\text{CO}_2$  when compared with the data-set of Richey et al.1998 (Supplementary Figure 2.11a).

#### Exchange of $\text{CO}_2$ , $\text{CH}_4$ and $\text{N}_2\text{O}$ with the atmosphere from the whole Amazon basin including Andean mountainous and piedmont streams

We used catchment slope as an integrative metric to scale  $p\text{CO}_2$ ,  $\text{CH}_4$  and  $\%\text{N}_2\text{O}$  available in the Amazon River network, as detailed in the Methods section. The  $k$  values were derived from stream gradient (slope) and flow velocity (Raymond, Hartmann, 2013), itself

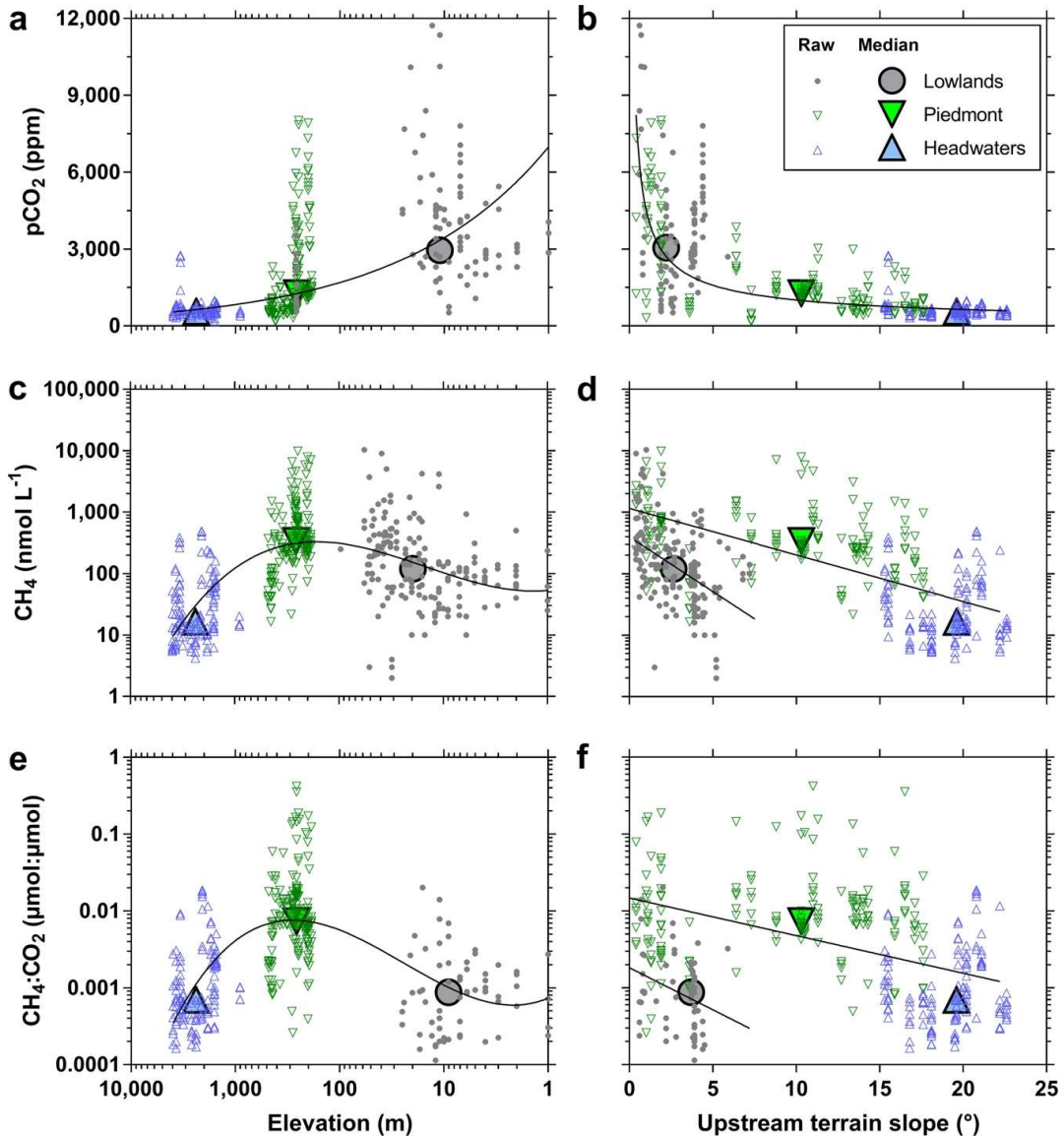
computed from discharge (Liu et al., 2022), for all the river segments given by RiverATLAS (Linke et al., 2019). The areal  $\text{FCO}_2$ ,  $\text{FCH}_4$  and  $\text{FN}_2\text{O}$  (per  $\text{m}^2$ ) were computed from  $k$  and concentrations of  $\text{CO}_2$ ,  $\text{CH}_4$ ,  $\text{N}_2\text{O}$  for each river segment of the Amazon catchment given by RiverATLAS ( $n=744,786$ ) and then integrated with the corresponding SSA.

The piedmont streams accounted for 62% of the total integrated diffusive  $\text{FCH}_4$  ( $0.4 \pm 0.1 \text{ TgCH}_4 \text{ yr}^{-1}$ ), higher than the contribution from lowland rivers (28%), while highland streams accounted for only 10% (Figure 2.5b). The diffusive  $\text{FCH}_4$  per  $\text{m}^2$  in highland streams ( $1,077 \pm 240 \text{ mmol m}^{-2} \text{ yr}^{-1}$ ) was 4.5 times higher than in lowland rivers ( $237 \pm 79 \text{ mmol m}^{-2} \text{ yr}^{-1}$ ) despite the fact that dissolved  $\text{CH}_4$  concentrations were  $\sim 2$  times lower ( $157 \pm 35$  versus  $307 \pm 100 \text{ nmol L}^{-1}$ ), due to much higher  $k$  values ( $148.8 \pm 3.2$  versus  $14.5 \pm 0.4 \text{ cm h}^{-1}$ , Fig 5e). The diffusive  $\text{FCH}_4$  per  $\text{m}^2$  in piedmont streams ( $1,573 \pm 350 \text{ mmol m}^{-2} \text{ yr}^{-1}$ ) was 6.6 times higher than in lowland rivers ( $137 \pm 79 \text{ mmol m}^{-2} \text{ yr}^{-1}$ ) because of much higher dissolved  $\text{CH}_4$  concentrations ( $797 \pm 174$  versus  $307 \pm 100 \text{ nmol L}^{-1}$ ), while  $k$  was relatively closer ( $20.0 \pm 0.5$  versus  $14.5 \pm 0.4 \text{ cm h}^{-1}$ ).

The integrated  $\text{FCO}_2$  ( $64 \pm 5 \text{ TgC yr}^{-1}$ ) was dominated by lowland rivers (65%) (Figure 2.5a). However,  $\text{FCO}_2$  per  $\text{m}^2$  in highland and piedmont rivers ( $197 \pm 35$  and  $143 \pm 3 \text{ mol m}^{-2} \text{ yr}^{-1}$ ) were 1.7 and 1.2 times higher than lowland rivers ( $115 \pm 8 \text{ mol m}^{-2} \text{ yr}^{-1}$ ), respectively (Figure 2.5d), due to higher  $k$  values, since  $\text{pCO}_2$  values were lower in highland streams ( $1,033 \pm 88 \text{ ppm}$ ), although comparatively similar in piedmont streams ( $2,599 \pm 146 \text{ ppm}$ ) and lowland rivers and streams ( $2,735 \pm 148 \text{ ppm}$ ) (Figure 2.5g).

The integrated  $\text{FN}_2\text{O}$  ( $25 \pm 4 \text{ GgN}_2\text{O-N yr}^{-1}$ ) was overwhelmingly dominated by lowland rivers and streams (99% of total), while piedmont streams were an extremely small source of  $\text{N}_2\text{O}$ , and highland streams were a very small sink of  $\text{N}_2\text{O}$  (Figure 2.5c). The computation of  $\text{FN}_2\text{O}$  is comparatively more uncertain than  $\text{FCH}_4$  and  $\text{FCO}_2$  since we extrapolated to all lowland river segments the median  $\% \text{N}_2\text{O}$  value given by Richey et al. 1988, unlike for the other two GHGs for which relations with slope were used to compute concentration values for individual river segments. The median  $\% \text{N}_2\text{O}$  value (206%) reported by Richey et al. 1988 for the Central Amazon rivers and streams was higher than the median value of  $\% \text{N}_2\text{O}$  reported in two large African rivers, the Zambezi (Teodoru et al., 2015) and the Congo (Borges et al., 2019) rivers of 116% and 131%, respectively. Another study provided a lower  $\% \text{N}_2\text{O}$  average value ( $134 \pm 25\%$ ,  $n=131$ ) in the Solimões and Negro Rivers (Oliveira, 2006) than the median  $\% \text{N}_2\text{O}$  value (206%) reported by Richey et al. 1988. Using the lower  $\% \text{N}_2\text{O}$  value of 134% for all the lowland rivers provided an integrated  $\text{FN}_2\text{O}$  value for the Amazon River basin of  $9 \pm 4 \text{ GgN}_2\text{O-N yr}^{-1}$ , which was close to the range of  $\text{N}_2\text{O}$  emissions for the Amazon River basin of 11 to 16  $\text{GgN}_2\text{O-N yr}^{-1}$  modelled from  $\text{N}$  deposition and emission factors (Maavara et al., 2019). The wide range of  $\% \text{N}_2\text{O}$  values reported in literature calls for additional data of  $\text{N}_2\text{O}$  in lowland rivers of the Amazon basin for a more accurate evaluation of  $\text{FN}_2\text{O}$ .

FIGURE 2.4 Strong elevation gradients of  $p\text{CO}_2$  and  $\text{CH}_4$  in streams and rivers across the whole Amazon basin. Variations of the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ),  $\text{CH}_4$  dissolved concentration, and the ratio of  $\text{CH}_4$  and  $\text{CO}_2$  ( $\text{CH}_4:\text{CO}_2$ ) concentration as function of elevation (a, c, e) and catchment slope (b, d, f). Data in the lowland rivers and streams was compiled from literature for  $p\text{CO}_2$  (Abril et al., 2014; Rasera et al., 2008; Rasera et al., 2013; Mitchell et al., 2018) and  $\text{CH}_4$  (Borges et al., 2015; Sawakuchi, et al 2014; Barbosa et al., 2016., Mitchell et al., 2018). For consistency, only studies based on direct measurements of  $p\text{CO}_2$  were selected, excluding data derived from the computation of  $p\text{CO}_2$  from pH and total alkalinity, as this computation can provide erroneous and unrealistically high  $p\text{CO}_2$  values (Abril et al., 2015). Data from Scofield et al.2016 was excluded because the  $p\text{CO}_2$  values reported by these authors in the Negro River (upstream of Manaus) were distinctly lower (710 ppm during low water period and 2608 ppm during higher water period) than other studies in the same system (~2000–3000 ppm during low water period and ~5000–8000 ppm during higher water period). Small symbols show individual measurements, large symbols indicate the median; the blue triangles show the data in the mountainous headwater streams, green triangles show the data in piedmont streams, and the grey circles show the data in the lowland streams and rivers, and black lines indicate linear or polynomial fits to the data. Solid lines show fits to the data (Supplementary Table 4).



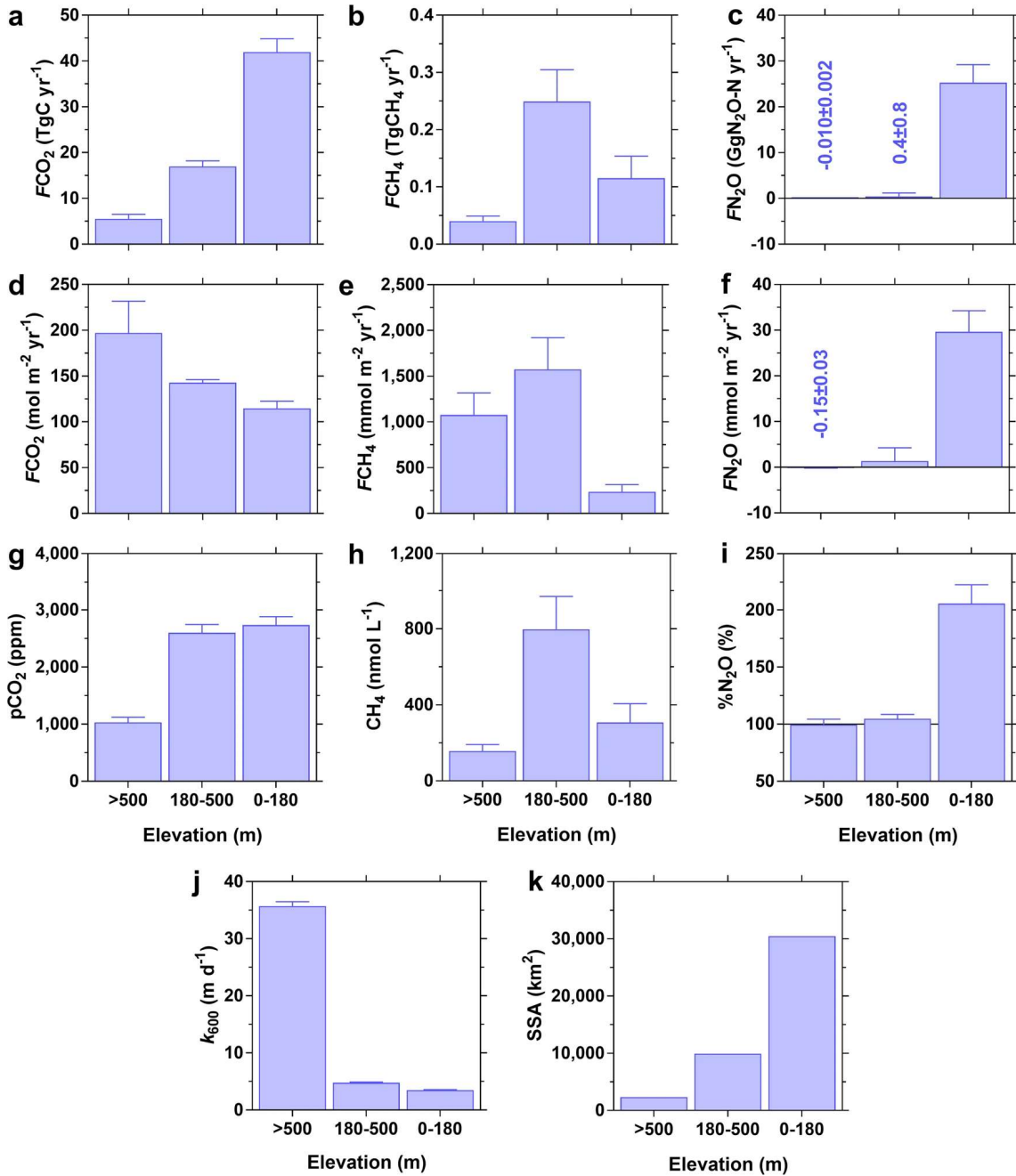
The FCO<sub>2</sub> and FCH<sub>4</sub> estimates per m<sup>2</sup> were consistent (within uncertainties) with those previously reported in the Amazon basin, for lowland basin wide FCO<sub>2</sub> 5,27 and FCH<sub>4</sub> (Barbosa et al., 2016), FCH<sub>4</sub> for large lowland rivers (Sawakuchi et al., 2014), and the Amazon/Solimões mainstem FCH<sub>4</sub> (Melack et al., 2004) (Supplementary Table 2. 1). Our lowland basin wide diffusive FCH<sub>4</sub> per m<sup>2</sup> is higher than the one reported by Barbosa et al. 2016 because our estimate includes the piedmont rivers with high diffusive FCH<sub>4</sub> values (Figure 2.5e) while the estimate of Barbosa et al. 2016 is based on data from large lowland rivers (Xingu, Tapajós, Madeira, Purus, Juruá, Japurá, Solimões, Negro).

Uncertainties in scaling the exchange of greenhouse gases with the atmosphere.

The integrated FCO<sub>2</sub>, FCH<sub>4</sub> and FN<sub>2</sub>O estimates have three sources of uncertainty from each of the terms of the flux computation: the respective modelled dissolved gas concentrations, the k values, and the SSA.

Previous studies that modelled fluvial pCO<sub>2</sub> at continental or global scales either used averaged values per stream order (Butman and Raymond, 2011) or multi-parametric approaches (Lauerwald et al., 2015; Liu et al., 2022) using several catchment properties and climate data as predictors. At the continental scale of the USA, Lauerwald et al. 2013 modelled pCO<sub>2</sub> in streams and rivers with multiple linear regressions (MLR) using catchment slope, mean air temperature, and mean precipitation as predictors. At global scale, Lauerwald et al. 2015 modelled pCO<sub>2</sub> in streams and rivers with MLR using catchment slope, mean air-temperature, land net primary production, and population density as predictors. Also at global scale, Liu et al. (2022) modelled pCO<sub>2</sub> in streams and rivers with a random forest model that used as primary inputs 15 watershed properties including catchment slope, elevation, soil pH, temperature, soil respiration, watershed area, land gross primary production, base saturation, population density, gravel, soil organic carbon, sand, precipitation, land net primary production, and wetland cover (ranked in order of decreasing importance). Note that some of these 15 watershed properties are intimately linked such as land net primary production, gross primary production and soil respiration (Lauerwald et al., 2013), or geomorphological variables such as slope and catchment area (Flint, 1974), so there could be redundancy in part of the variables used in this random forest model.

FIGURE 2.5 Andean headwater and piedmont streams are hot spots of CO<sub>2</sub> and CH<sub>4</sub> emissions within the Amazon basin. Air-water fluxes of CO<sub>2</sub> (FCO<sub>2</sub>), CH<sub>4</sub> (FCH<sub>4</sub>) and N<sub>2</sub>O (FN<sub>2</sub>O) integrated by area (a–c) and per surface area (m<sup>2</sup>) (d–f), partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (g), dissolved CH<sub>4</sub> concentration (h), and N<sub>2</sub>O saturation level (%N<sub>2</sub>O) (i), gas transfer velocity (k<sub>600</sub>) (j) and stream surface area (SSA) (k) in the Andean headwater (elevation >500 m), piedmont (elevation 180–500 m) and lowland (elevation <180 m) streams and rivers of the whole Amazon basin. Fluxes were computed from concentrations and k<sub>600</sub> modelled from attributes of RiverATLAS68 for 744,786 river segments. Error bars indicate uncertainties derived from error propagation.



There is no standardized procedure to upscale GHG emissions from inland waters which is challenging given the large inherent variability and the comparatively low data coverage. Some studies have used multi-parametric approaches, while we use a simple relation to a single predictor. Complex mathematical approaches do not necessarily provide more reliable results, and, in fact, potentially produce more uncertain estimates (Puy et al., 2022). Admittedly, it is unrealistic to model fluvial  $p\text{CO}_2$  using a single predictor at continental or global scales. Indeed, at very large spatial scales, it is required to account for strong latitudinal variations as in climate (temperature, precipitation, ice cover) and in soil organic carbon (Langeveld et al., 2020) (among others) that are expected to strongly affect fluvial  $p\text{CO}_2$ , in addition to local topographic effects that might have a more ubiquitous and uniform effect per latitude (such as the effect of stream slope on  $k$ ). Yet at the scale of a single river basin, it should be possible to reduce the number of variables to model fluvial  $p\text{CO}_2$ . A simpler model has the advantage to reduce unrealistic results computed outside the bounds of determination of available data, especially from strongly non-linear relationships. We did not find a relation between  $p\text{CO}_2$  and dissolved  $\text{CH}_4$  concentration and stream order (Supplementary Figure 2.8), as used to scale  $\text{FCO}_2$  or  $\text{FCH}_4$  in some studies at basin or continental scales (Borges et al., 2019a; Butman and Raymond, 2011; Wallin et al., 2018) 21,60,61, and we modelled fluvial  $p\text{CO}_2$  and dissolved  $\text{CH}_4$  concentration based on catchment slope. In the above mentioned three models of fluvial  $p\text{CO}_2$  at continental or global scales, catchment slope explained the largest fraction of variance in the MRL (Lauerwald et al., 2015) or random forest model (Liu et al., 2022), and as such was the most important predictor. This high predictive power shows that catchment slope strongly controls fluvial  $\text{CO}_2$  content directly or indirectly at multiple levels.

It is well established that topography of a catchment strongly influences hydrological, geomorphological and biological processes active in the landscape. Catchment slope determines to a large extent broad patterns in the variation of soil organic carbon content (de Brogniez et al., 2015; Connolly et al., 2018; Thompson and Kolka, 2005), ground water discharge (Leach et al., 2017), stream water residence time, and wetland abundance (Merot et al., 2003; Rodhe and Seibert, 1999), that, combined, have a major impact on fluvial carbon cycling, including  $\text{CO}_2$  and  $\text{CH}_4$  dynamics. These processes also determine the inputs to streams and the fluvial content of DOC. Highland steep streams with poorly developed soils and substantial exposed bedrock are characterized by lower fluvial DOC than lowland low-relief streams, and fluvial DOC content correlates to catchment slope (Harms et al., 2016; Mulholland, 2007; Ran et al., 2017). Since catchment slope strongly determines fluvial DOC (Harms et al., 2016; Mulholland, 2007; Ran et al., 2017) and wetland coverage (Merot et al., 2003; Rodhe and Seibert, 1999), it is also expected to strongly influence fluvial  $p\text{CO}_2$  that has been shown to strongly correlate with DOC (Borges et al., 2015c; Liu et al., 2021; Rasilo et al., 2017; Willgoose, 1994) and wetland coverage (Abril et al., 2014; Borges et al., 2015c, 2015) at river basin and continental scales. Wetland coverage can also explain in part the spatial variations of fluvial dissolved  $\text{CH}_4$  concentration (Horgby et al., 2019; Borges et al., 2015). Additionally, the  $k$  values strongly increase with increasing stream slope (Horgby et al., 2019; Raymond et al., 2012) that scales with catchment slope (Hua et al., 2018) and catchment area (Flint, 1974). High  $k$  values in steeper terrains will promote degassing and equilibration with the atmosphere, while lower  $k$  values will allow an accumulation of gases in the water. An additional reason to use catchment slope as a predictor is that this variable can be determined with very high confidence based on detailed and precise digital elevation models. Other variables such as land vegetation cover (Gao et al., 2020; Liu et al., 2018) or terrestrial productivity (Huotari et al., 2013; Zhang and Ye, 2022) have a relatively large uncertainty especially at finer resolution. We tested if the addition to upstream slope of other variables improved the prediction of  $p\text{CO}_2$  (Supplementary Table 2. 2). The MRL including elevation or forest cover in addition of upstream slope did not substantially improve the prediction of  $p\text{CO}_2$  compared to the linear regression using upstream slope alone (Supplementary Table 2. 2).

The  $k$  cannot be measured directly but can be calculated from simultaneous measurements of the air-water gradient of gas concentration and the gas flux. The gas flux in streams can be measured with eddy-covariance, mass balance of deliberate tracers, or floating



chambers. Eddy-covariance is the most challenging technically, has been very seldom applied in rivers, and only in large systems (Van Dam et al., 2019; Lorke et al., 2015). Eddy-covariance measurements require long deployments (weeks-months) due to large data rejection, and are not adapted for “grab-sampling” during spatial surveys of short-duration, as in the present study. The mass balance of deliberate tracers such as argon, propane, SF<sub>6</sub>, krypton-85 and tritium allow to derive *k* values at short time intervals and small scales in a non-intrusive way (Raymond et al., 2012; Ulseth et al., 2019). Floating chamber flux measurements are easy to set-up and inexpensive, however, they might not be adapted for fast flowing small streams as those sampled in the present study. In fast flowing streams, it is required to anchor the floating chamber, leading to an over-estimations of flux estimates, due to enhanced turbulence under the chamber (Belanger and Korzun, 1991; Vingiani et al., 2021). Changes in pressure and temperature inside the chamber additionally affect the flux measurements and *k* estimates (Campeau et al., 2014). Chamber measurements can advantageously be used to measure gas ebullition which is important for CH<sub>4</sub> but not CO<sub>2</sub> and N<sub>2</sub>O, in addition to gas diffusive flux. The *k* parametrization of Raymond et al. (2012) we used in the present study is based on 576 measurements of deliberate gas tracer injections, and has been used in global estimates of CO<sub>2</sub> emission from streams and rivers (Lauerwald et al., 2015; Liu et al., 2022; Raymond et al., 2013). The *k* values computed with this parameterization have been shown to satisfactorily compare to those derived from chamber measurements by several independent studies in contrasted rivers and streams (Lim et al., 2022; Rasilo et al., 2017; Schneider et al., 2020; Wallin et al., 2018; Willgoose, 1994).

To scale FCO<sub>2</sub> and FCH<sub>4</sub>, we used the spatial data, including SSA, from RiverATLAS68 that provides a standardized compilation of descriptive hydro-environmental information for all river basins of the world at high spatial resolution. The total SSA given by RiverATLAS (Linke et al., 2019) globally is 363,000 km<sup>2</sup>, which is lower than the value of 624,000 km<sup>2</sup> given by Raymond et al. (2013) and of 773,000 km<sup>2</sup> given by the Global River Widths from Landsat (GRWL) data-set (Anderson et al., 2018). At the scale of the Amazon basin, the SSA given by RiverATLAS (Linke et al., 2019) is also lower than other independent estimates, although the wide range of values reflects their uncertainty (Supplementary Table 2. 3). The SSA of the lowland Amazon basin (<500 m altitude) given by RiverATLAS (Linke et al., 2019) (37,972 km<sup>2</sup>) is between 1.6 and 2.2 times lower than the one reported by Melack (2016) (60,500-83,000 km<sup>2</sup>), but 1.4 times lower than the one reported by Liu et al. (2022) (52,640 km<sup>2</sup>) based on the GRWL data-set (Anderson et al., 2018) (Supplementary Table 2. 3). We acknowledge that the integrated fluvial emissions of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O we computed for the Amazon basin should be considered as conservative, given the lower SSA given by RiverATLAS (Linke et al., 2019) compared to other estimates (Supplementary Table 2. 3). Yet, the general spatial patterns should be correct and the under-estimation of SSA given by RiverATLAS should not invalidate our major finding regarding the relative importance of the contribution of mountainous and piedmont regions to CO<sub>2</sub> and CH<sub>4</sub> emissions from the Amazon River basin (Figure 2.5).

## **Conclusions**

We provide to the best of our knowledge the first large scale survey of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O exchange between Andean mountainous streams of the Amazon River basin and the atmosphere. We showed that these mountainous headwater and piedmont streams were hotspots of CO<sub>2</sub> and CH<sub>4</sub> emissions from the whole Amazon River basin, accounting for a very significant fraction of total emissions (35 and 72% for CO<sub>2</sub> and CH<sub>4</sub> respectively). The present study also provides a baseline evaluation of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions from headwater mountainous and piedmont streams that could change in response to human alterations and activities on the catchments. In these areas, there are high rates of deforestation and conversion to croplands (McClain and Naiman, 2008). There have been concerns about the N<sub>2</sub>O emissions from soils to the atmosphere related to oil palm plantations (Skiba et al., 2020), and here we suggest that part of the N<sub>2</sub>O emissions related to oil palm plantations indirectly occur through

streams. The oil palm plantations seemed to lead to a strong over-saturation of N<sub>2</sub>O (~150%) in streams downstream compared to streams at similar elevations that were characterized by N<sub>2</sub>O levels closer to saturation (~110%). Additionally, there has been a boom in hydropower development in the headwaters of the Amazon with potential consequences on aquatic biodiversity (Laraque et al., 2004). Hydropower development could also change sediment transport by rivers to and deposition in piedmont rivers, possibly affecting future CH<sub>4</sub> emissions that were shown here to be particularly high.

### **Data availability**

The timestamped and georeferenced data-set is available at <https://doi.org/10.5281/zenodo.6527837>

### **Acknowledgements**

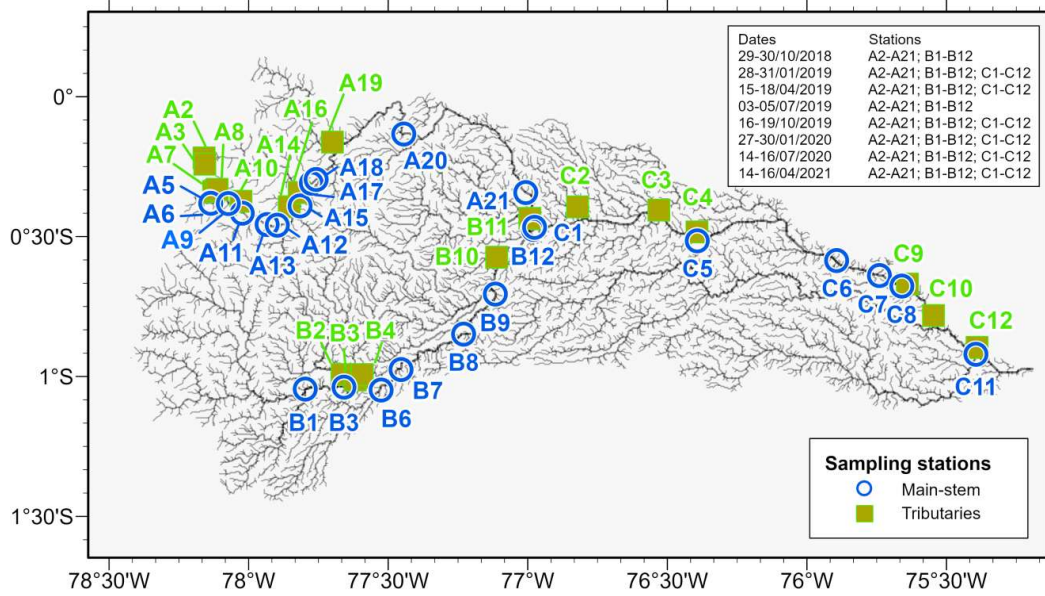
We thank three anonymous reviewers for constructive comments on the initial manuscript. This research was funded by the *Académie de recherche et d'enseignement supérieur* (ARES). AVB is a research director at the *Fonds National de la Recherche Scientifique*.

### **Author contributions**

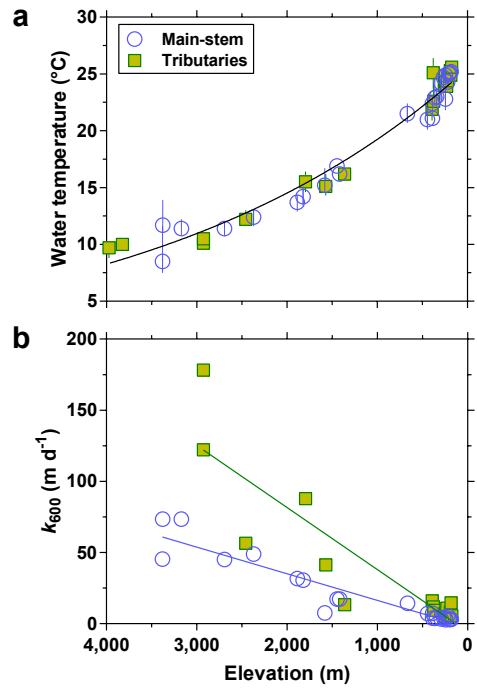
AVB designed the research and analyzed the data. GC collected and analyzed the samples. AVB and GC equally contributed to drafting the manuscript.

Supplementary Material

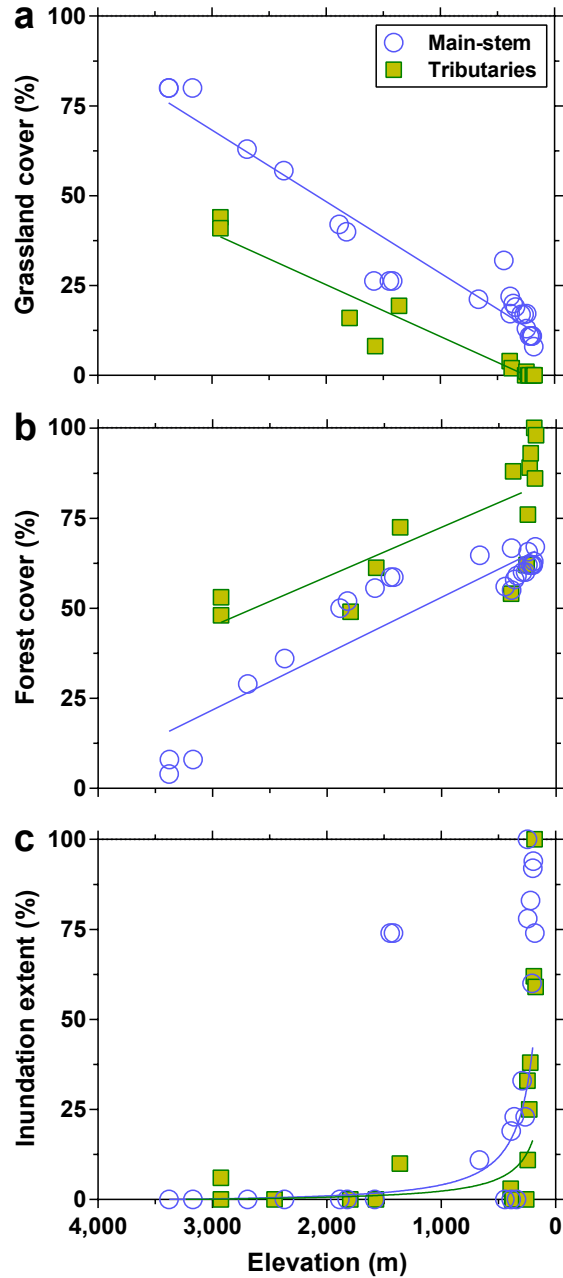
**SUPPLEMENTARY FIGURE 2.1** Location of stations in main-stem. Location of stations in main-stem (blue circles) and tributaries (green squares) and sampling dates. Stations in the main-stem are shown by blue circles and in the tributaries by green squares.



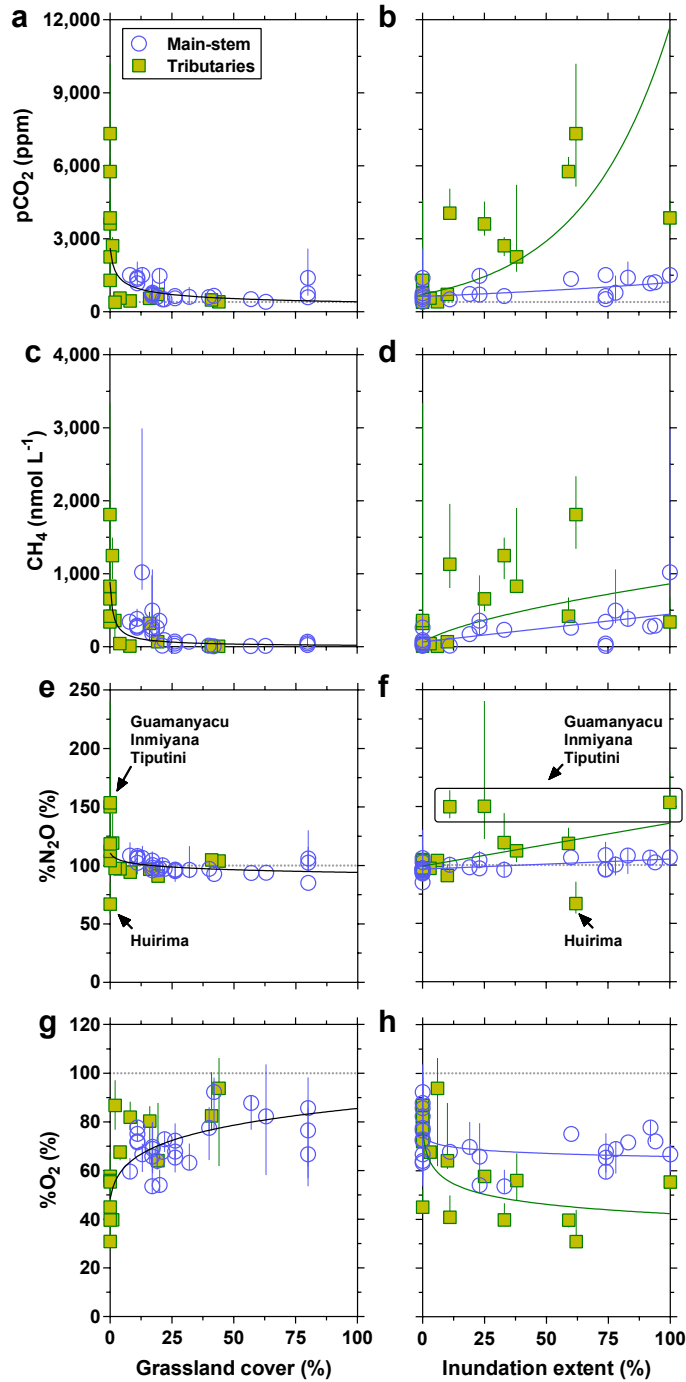
**SUPPLEMENTARY FIGURE 2.2** Strong elevation gradients of water temperature and gas transfer velocity in the Andean headwater and piedmont streams. Variations of measured water temperature (a) and the gas transfer velocity normalized to Schmidt number of 600 ( $k_{600}$ ) modelled from freshwater discharge and stream gradient from RiverATLAS (Linke et al., 2019) (b) in the Napo streams as a function of elevation. Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4).



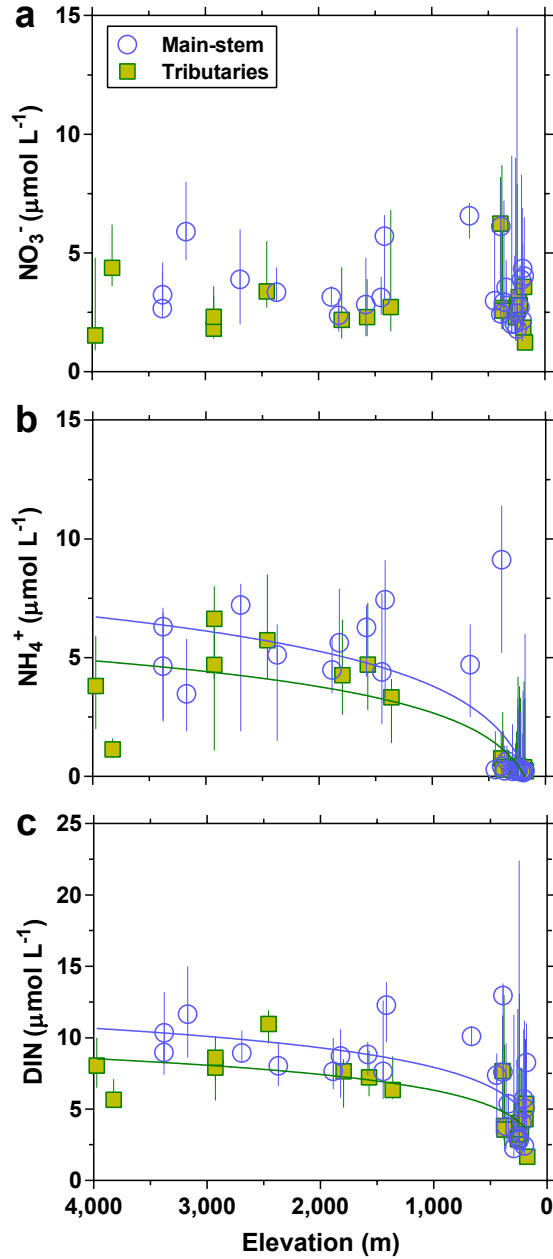
**SUPPLEMENTARY FIGURE 2.3** Strong elevation gradients of land cover in the catchment of Andean headwater and piedmont streams. Variations of grassland cover (a), forest cover (b) and inundation extent (c) in the Napo streams as a function of elevation, from Global Land Cover 2000 and Global Inundation Extent from Multi-Satellites (GIEMS-D15) extracted from RiverATLAS (linke et al., 2019). Grassland and forest cover are given over the total catchment upstream the sampled stream; inundation extent is given at the reach level. Stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4).



**SUPPLEMENTARY FIGURE 2.4** Grassland cover and inundation extent affect pCO<sub>2</sub>, CH<sub>4</sub> and %N<sub>2</sub>O in the Andean headwater and piedmont streams. Variations of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>), CH<sub>4</sub> dissolved concentration, N<sub>2</sub>O saturation level (%N<sub>2</sub>O), and oxygen saturation level (%O<sub>2</sub>) in the Napo streams as a function of grassland cover (a,c,e,g) and inundation extent (b,d,f,h) in the Napo streams, from Global Land Cover 2000<sup>115</sup> and Global Inundation Extent from Multi-Satellites (GIEMS-D15)<sup>18</sup> extracted from RiverATLAS<sup>68</sup>. Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4).



**SUPPLEMENTARY FIGURE 2.5** Elevation gradients of dissolved inorganic nitrogen in the Andean headwater and piedmont streams. Nitrate ( $\text{NO}_3^-$ ), ammonia ( $\text{NH}_4^+$ ) and dissolved inorganic nitrogen (DIN) concentration as a function of elevation in the Napo streams. Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4).

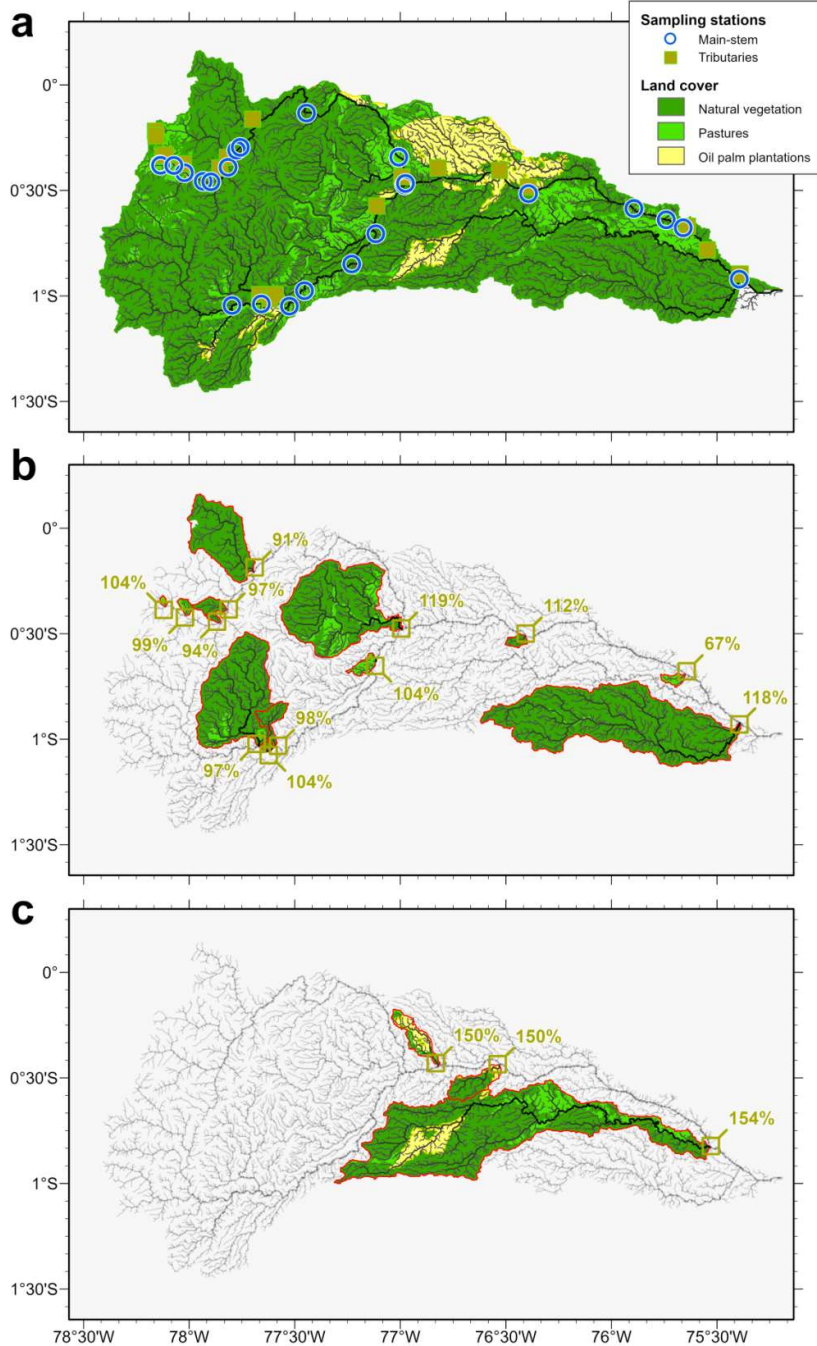


**SUPPLEMENTARY FIGURE 2.6** Huirima stream showing meandering river course and presence of small lakes indicative of lateral inundation related to flat relief (low catchment slope). Landsat image was extracted from <https://earthexplorer.usgs.gov/> and is in the U.S. Public Domain.

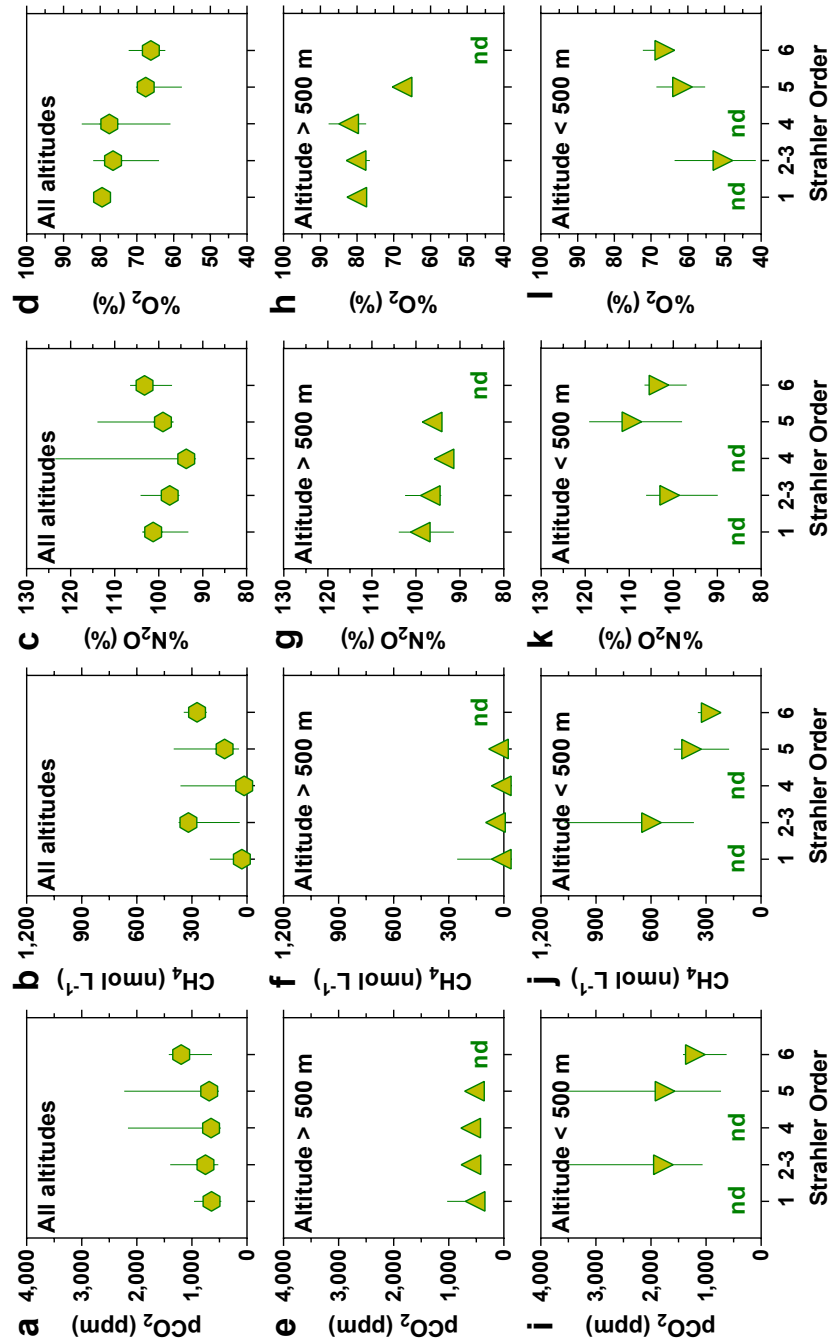




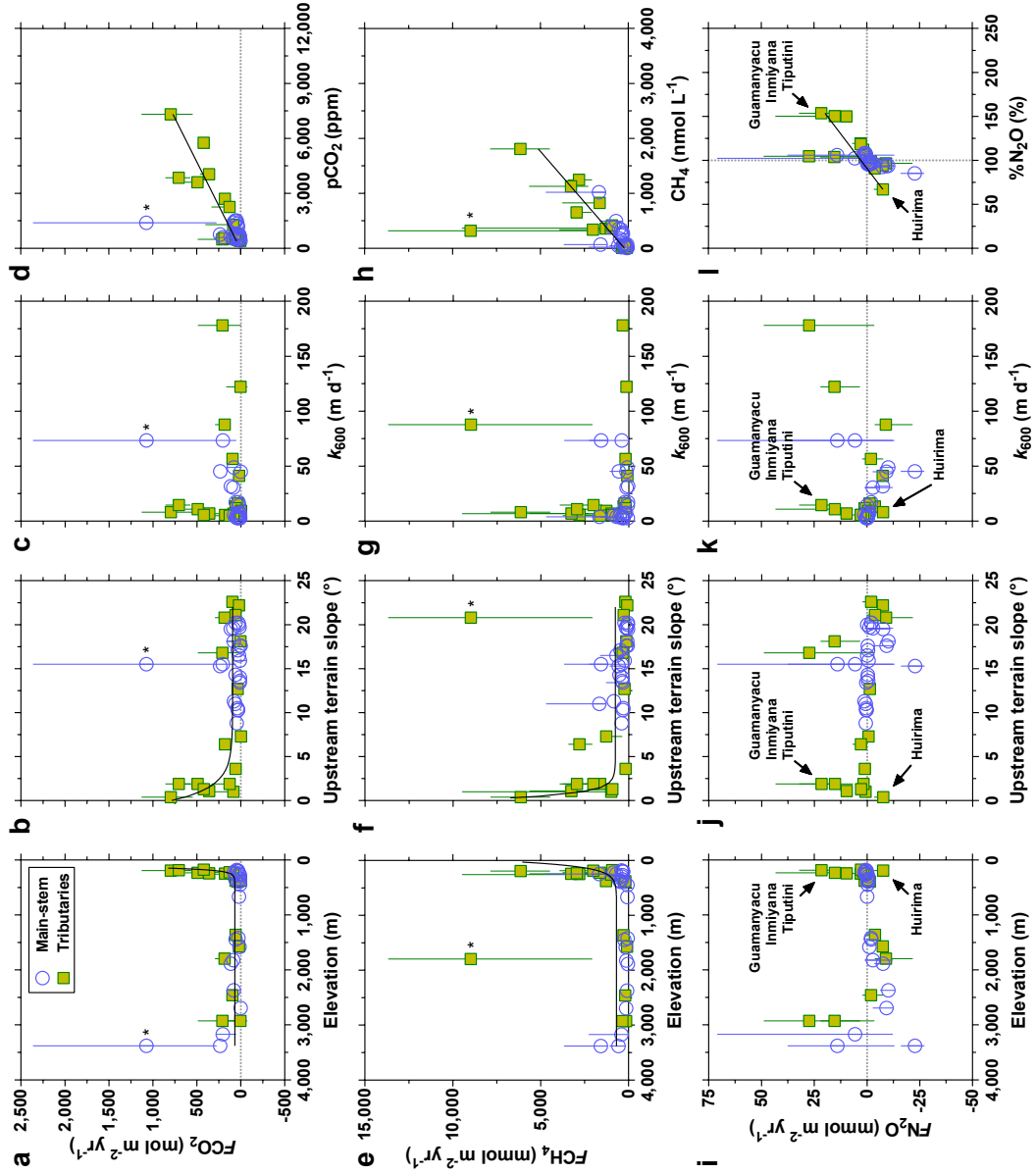
**SUPPLEMENTARY FIGURE 2.7** High N<sub>2</sub>O levels are related to the presence of oil palm plantations in the Andean piedmont streams. Distribution of “natural” vegetation (merging forest and grassland) and agriculturally impacted land (pastures and oil palm plantations) in the Napo streams, for the whole catchment upstream of the most downstream station (a), and for individual catchments for N<sub>2</sub>O saturation levels ≤ 119% (b) and ≥ 150% (c). Land covered by forest and grassland was merged into a single class (natural vegetation) by opposition to land cover strongly impacted by human activity (pastures and oil palm plantations). Stations in the main-stem are shown by blue circles and in the tributaries by green squares.



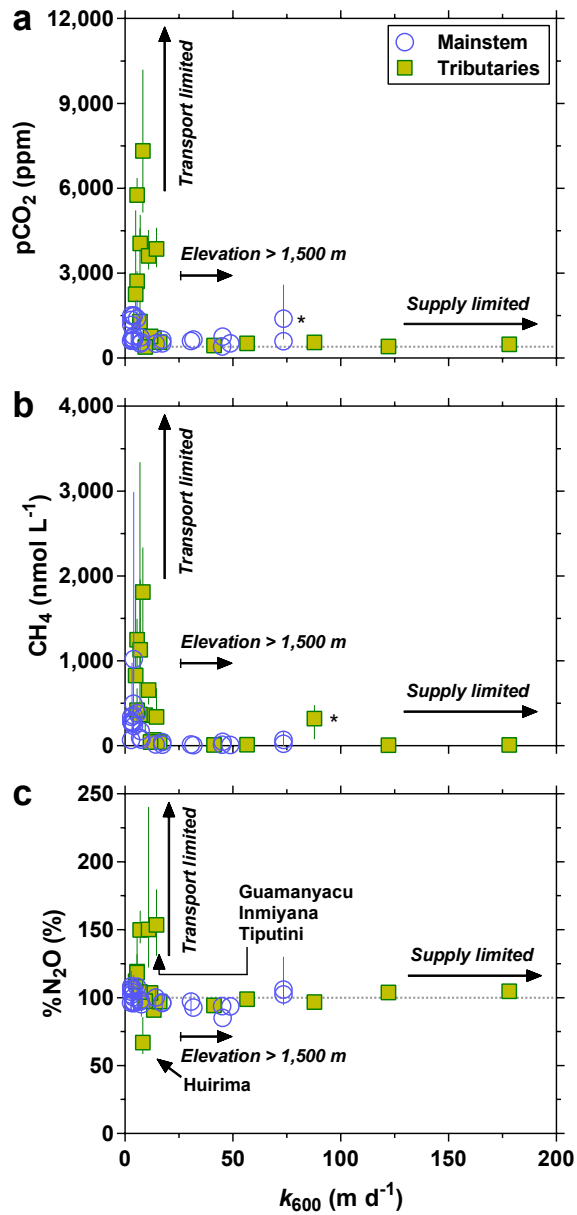
**SUPPLEMENTARY FIGURE 2.8** Low variations of pCO<sub>2</sub>, CH<sub>4</sub> and %N<sub>2</sub>O in the Andean headwater and piedmont streams as function of stream size. Variations of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>), CH<sub>4</sub> dissolved concentration, N<sub>2</sub>O saturation level (%N<sub>2</sub>O), and oxygen saturation level (%O<sub>2</sub>) in the Napo streams as a function of Strahler order irrespective of elevation (a,b,c,d), for elevation > 500 m (e,f,g,h) and < 500 m (i,j,k,l). Due to low number of samples in low order streams, orders 2 and 3 were binned. Symbols show the median and error bars the interquartile range. nd = no data.



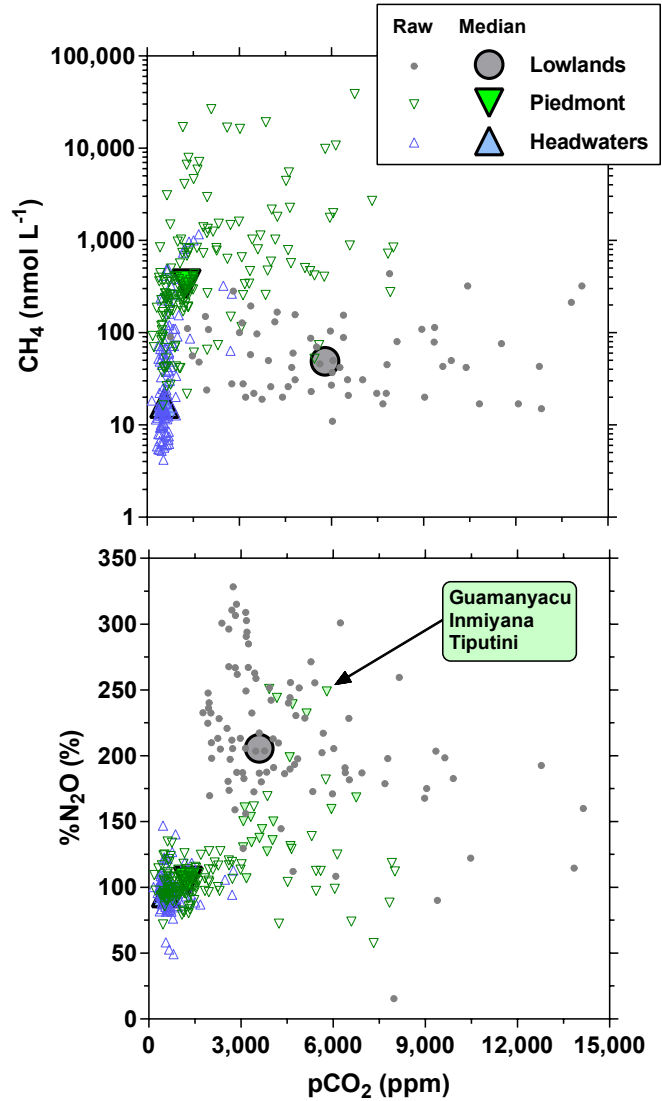
**SUPPLEMENTARY FIGURE 2.9** Strong elevation gradients of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions in the Andean headwater and piedmont streams. Air-water flux of CO<sub>2</sub> ( $F_{CO_2}$ ) (a,b,c,d), of CH<sub>4</sub> ( $F_{CH_4}$ ) (e,f,g,h) and of N<sub>2</sub>O ( $F_{N_2O}$ ) (i,j,k,l) as a function of elevation (a,e,i), upstream terrain slope (b,f,j), gas transfer velocity normalized to Schmidt number of 600 ( $k_{600}$ ) (c,g,k), and respective dissolved gas content: partial pressure of CO<sub>2</sub> ( $p_{CO_2}$ ) (d), CH<sub>4</sub> dissolved concentration (h), N<sub>2</sub>O saturation level (%N<sub>2</sub>O) (l). Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4). Asterisk indicates streams with high CO<sub>2</sub> and CH<sub>4</sub> concentrations and high  $k_{600}$  values (refer to Figure 2. S10).



**SUPPLEMENTARY FIGURE 2.10** Transport-limitation or supply-limitation of the gaseous exchange with the atmosphere in the Andean headwater and piedmont streams. Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (a), CH<sub>4</sub> dissolved concentration (b), and N<sub>2</sub>O saturation level (%N<sub>2</sub>O) (c) as a function of the gas transfer velocity normalized to Schmidt number of 600 ( $k_{600}$ ). Symbols show the median and error bars the interquartile range, stations in the main-stem are shown by blue circles and in the tributaries by green squares, and solid lines show fits to the data (Supplementary Table 2. 4). Asterisk indicates streams with high CO<sub>2</sub> and CH<sub>4</sub> concentrations and high  $k_{600}$  values leading to comparatively higher emissions (refer to Figure 2. S9).



**SUPPLEMENTARY FIGURE 2.11** Strong elevation gradients of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in streams and rivers across the whole Amazon basin. Variations of CH<sub>4</sub> dissolved concentration (a) and N<sub>2</sub>O saturation level (%N<sub>2</sub>O) (b) as a function of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>). Data in the lowland rivers and streams was compiled extracted from plots of Richey *et al.* 1988 with Plot Digitizer (<http://plotdigitizer.sourceforge.net>). Small symbols show individual measurements, large symbols indicate the median, the blue triangles show the data in the mountainous headwater streams, green triangles show the data in piedmont streams, and the grey circles show the data in the lowland streams and rivers.



*Greenhouse gas emissions in Napo Basin and Andean Lakes*

**SUPPLEMENTARY TABLE 2.1** Comparison of air-water fluxes of CO<sub>2</sub> (FCO<sub>2</sub>) and CH<sub>4</sub> (FCH<sub>4</sub>) in the Amazon rivers and streams from this study and reported in (Liu et al., 2022 ; Borges, Abril, et al., 2015; Melack J.M., 2016 ; Melack et al., 2004 ; Barbosa et al., 2016). Only the diffusive component of FCH<sub>4</sub> is reported.

|                                      |                                                                | FCO <sub>2</sub><br>mol m <sup>-2</sup><br>yr <sup>-1</sup> | FCH <sub>4</sub><br>mmol m <sup>-2</sup> yr <sup>-1</sup> |
|--------------------------------------|----------------------------------------------------------------|-------------------------------------------------------------|-----------------------------------------------------------|
| River channels < 150 m               | <b>This study</b><br>(Melack J.M., 2016)<br>(Liu et al., 2022) | <b>161±11</b><br>156<br>157                                 |                                                           |
| River channels ≥ 150 m               | <b>This study</b><br>(Melack J.M., 2016)<br>(Liu et al., 2022) | <b>91±6</b><br>71<br>135                                    |                                                           |
| Solimões/Amazon mainstem             | <b>This study</b><br>(Melack et al., 2004)                     |                                                             | <b>66±29</b><br>75                                        |
| River channels ≥ 200 m               | <b>This study</b><br>(Borges, Abril, et al., 2015)             |                                                             | <b>199±55</b><br>274±41                                   |
| River channels (whole lowland basin) | <b>This study</b><br>(Barbosa et al., 2016)                    |                                                             | <b>564±141</b><br>370                                     |

**SUPPLEMENTARY TABLE 2.2** Goodness of the linear regression of modelled partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) based on three models versus observed pCO<sub>2</sub> from the Andean headwater and piedmont streams. The models were a linear regression (LR) as function of upstream slope, a multiple linear regression (MLR) as function of upstream slope and elevation, and a MLR as function of upstream slope and forest cover.

|                                   | Modelled pCO <sub>2</sub> versus initial pCO <sub>2</sub> | r <sup>2</sup> | Sy.x  |
|-----------------------------------|-----------------------------------------------------------|----------------|-------|
| Upstream slope LR                 | y = 1.284±(0.072) + 0.594(±0.023)*x                       | 0.59           | 0.202 |
| Upstream slope + Elevation MLR    | y = 1.199(±0.071) + 0.620(±0.022)*x                       | 0.62           | 0.200 |
| Upstream slope + Forest cover MLR | y = 1.221(±0.071) + 0.613(±0.022)*x                       | 0.61           | 0.200 |

**SUPPLEMENTARY TABLE 2.3** Comparison of stream surface area in the Amazon basin used in this study (extracted from RiverATLAS68) and reported in literature (Liu et al., 2022 ; Borges, Abril, et al., 2015; Melack J.M., 2016 ; Melack et al., 2004 ; Barbosa et al., 2016). The surface area for large lowland rivers used by (Sawakuchi et al. 2014) is two to three times higher than the range of values reported by (Melack J.M., 2016) for an equivalent river size class.

|                                      |                                                    | Stream surface area<br>km <sup>2</sup> |
|--------------------------------------|----------------------------------------------------|----------------------------------------|
| River channels < 150 m               | <b>This study</b>                                  | <b>16,304 (*)</b>                      |
|                                      | (Melack J.M., 2016)                                | 31,000 (**)                            |
|                                      | (Liu et al., 2022)                                 | 27,120 (***)                           |
| River channels ≥ 150 m               | <b>This study</b>                                  | <b>21,668 (*)</b>                      |
|                                      | (Melack J.M., 2016)                                | 29,500 (**) to 52,000 (****)           |
|                                      | (Liu et al., 2022)                                 | 25,500                                 |
| Solimões/Amazon mainstem             | <b>This study</b><br>(Melack et al., 2004)         | <b>4,994 (*)</b><br>8,856              |
| River channels ≥ 200 m               | <b>This study</b><br>(Borges, Abril, et al., 2015) | <b>19,828 (*)</b><br>91,212 (*****)    |
| River channels (whole lowland basin) | <b>This study</b><br>(Barbosa et al., 2016)        | <b>40,375 (*)</b><br>52,380 (*****)    |

\* Reference (Linke et al. 2019)

\*\* Reference (Beighley et al., 2011)

\*\*\* Reference (Anderson et al., 2018)

\*\*\*\* from Shuttle Radar Topography Mission

\*\*\*\*\* Reference (Brown et al., 2003)

\*\*\*\*\* Unspecified data source

SUPPLEMENTARY TABLE 2. 4 Equations and statistics at 0.05 level of curve fits of data shown in this Chapter Figures 2...

| Figure | Function                                                                                     | Equation                                                             | Average Residual | r <sup>2</sup> | p       | n   |
|--------|----------------------------------------------------------------------------------------------|----------------------------------------------------------------------|------------------|----------------|---------|-----|
| 2 a    | pCO <sub>2</sub> versus Elevation                                                            | $y = 7.730E+04 \cdot \text{EXP}(-1.768E-02 \cdot x) + 6.623E+02$     | 6.20E+02         |                |         | 44  |
| 2 b    | pCO <sub>2</sub> versus Upstream terrain slope                                               | $y = 7.085E+03 \cdot \text{EXP}(-6.395E-01 \cdot x) + 8.108E+02$     | 5.14E+02         |                |         | 42  |
| 2 c    | pCO <sub>2</sub> versus Forest cover                                                         | $y = 2.773E-02 \cdot \text{EXP}(1.229E-01 \cdot x) + 9.977E+02$      | 5.56E+02         |                |         | 42  |
| 2 d    | pCO <sub>2</sub> versus Water temperature                                                    | $y = 6.775E-09 \cdot \text{EXP}(1.061E+00 \cdot x) + 3.882E+02$      | 7.14E+02         |                |         | 42  |
| 2 e    | CH <sub>4</sub> versus Elevation                                                             | $y = 1.695E+03 \cdot \text{EXP}(-5.725E-03 \cdot x) + 9.183E+01$     | 2.19E+02         |                |         | 44  |
| 2 f    | CH <sub>4</sub> versus Upstream terrain slope                                                | $y = 1.202E+03 \cdot \text{EXP}(-6.150E-02 \cdot x) - 3.034E+02$     | 1.71E+02         |                |         | 42  |
| 2 g    | CH <sub>4</sub> versus Forest cover                                                          | $y = 1.804E+01 \cdot \text{EXP}(4.134E-02 \cdot x) - 2.591E+00$      | 1.83E+02         |                |         | 42  |
| 2 h    | CH <sub>4</sub> versus Water temperature                                                     | $y = 3.404E-02 \cdot \text{EXP}(3.844E-01 \cdot x) + 9.299E+01$      | 2.29E+02         |                |         | 42  |
| 2 i    | %N <sub>2</sub> O versus Elevation                                                           | $y = 7.940E+01 \cdot \text{EXP}(-9.046E-03 \cdot x) + 9.615E+01$     | 4.05E+00         |                |         | 40* |
| 2 j    | %N <sub>2</sub> O versus Upstream terrain slope                                              | $y = -0.8165 \cdot x + 112.3$                                        |                  | 0.47           | <0.0001 | 38* |
| 2 k    | %N <sub>2</sub> O versus Forest cover                                                        | $y = 0.1619 \cdot x + 91.4$                                          |                  | 0.20           | 0.006   | 38* |
| 2 l    | %N <sub>2</sub> O versus Water temperature                                                   | $y = 9.648E-07 \cdot \text{EXP}(6.536E-01 \cdot x) + 9.610E+01$      | 3.62E+00         |                |         | 38* |
| 2 m    | %O <sub>2</sub> versus log(Elevation)                                                        | $y = 18.953 \cdot x + 15.0$                                          |                  | 0.40           | <0.0001 | 44  |
| 2 n    | %O <sub>2</sub> versus Upstream terrain slope                                                | $y = 1.3564 \cdot x + 50.1$                                          |                  | 0.40           | <0.0001 | 42  |
| 2 o    | %O <sub>2</sub> versus Forest cover                                                          | $y = -0.3749 \cdot x + 90.4$                                         |                  | 0.31           | 0.0002  | 42  |
| 2 p    | %O <sub>2</sub> versus Water temperature                                                     | $y = -1.5737 \cdot x + 98.8$                                         |                  | 0.40           | <0.0001 | 44  |
| 3 a    | pCO <sub>2</sub> versus %O <sub>2</sub>                                                      | $y = 5.909E+04 \cdot \text{EXP}(-7.016E-02 \cdot x) + 4.755E+02$     | 4.97E+02         |                |         | 44  |
| 3 b    | pCO <sub>2</sub> versus CH <sub>4</sub>                                                      | $y = 0.2022 \cdot x + 39$                                            |                  | 0.55           | <0.0001 | 44  |
| 3 c    | CH <sub>4</sub> versus %O <sub>2</sub>                                                       | $y = 1.863E+04 \cdot \text{EXP}(-7.885E-02 \cdot x) + 1.335E+02$     | 4.97E+02         |                |         | 44  |
| 3 d    | %N <sub>2</sub> O versus pCO <sub>2</sub>                                                    | $y = 0.0053 \cdot x + 94.8$                                          |                  | 0.46           | <0.0001 | 40* |
| 3 e    | %N <sub>2</sub> O versus %O <sub>2</sub>                                                     | $y = -0.2846 \cdot x + 120.2$                                        |                  | 0.26           | 0.0009  | 40* |
| 3 f    | %N <sub>2</sub> O versus DIN                                                                 | $y = -0.9479 \cdot x + 106.5$                                        |                  | 0.15           | 0.0126  | 40* |
| 4 a    | log(pCO <sub>2</sub> ) versus log(Elevation)                                                 | $y = -0.308 \cdot x + 3.8443$                                        |                  | 0.50           | <0.0001 | 403 |
| 4 b    | log(pCO <sub>2</sub> ) versus log(Upstream terrain slope)                                    | $y = -0.6519 \cdot x + 3.656$                                        |                  | 0.57           | <0.0001 | 403 |
| 4 c    | log(CH <sub>4</sub> ) versus log(Elevation)                                                  | $y = -0.1799 \cdot x^3 + 0.6348 \cdot x^2 - 0.1668 \cdot x + 1.7272$ |                  | 0.40           | <0.0001 | 471 |
| 4 d    | log(CH <sub>4</sub> ) versus Upstream terrain slope (for Elevation > 180 m)                  | $y = -0.0757 \cdot x + 3.0622$                                       |                  | 0.42           | <0.0001 | 297 |
| 4 d    | log(CH <sub>4</sub> ) versus Upstream terrain slope (for Elevation < 180 m)                  | $y = -0.1775 \cdot x + 2.5974$                                       |                  | 0.21           | <0.0001 | 174 |
| 4 e    | log(CH <sub>4</sub> ·CO <sub>2</sub> ) versus log(Elevation)                                 | $y = -0.2342 \cdot x^3 + 0.9761 \cdot x^2 - 0.5717 \cdot x - 3.1345$ |                  | 0.44           |         | 369 |
| 4 f    | log(CH <sub>4</sub> ·CO <sub>2</sub> ) versus Upstream terrain slope (for Elevation > 180 m) | $y = -0.0489 \cdot x - 1.8328$                                       |                  | 0.22           | <0.0001 | 305 |
| 4 f    | log(CH <sub>4</sub> ·CO <sub>2</sub> ) versus Upstream terrain slope (for Elevation < 180 m) | $y = -0.1094 \cdot x - 2.74$                                         |                  | 0.08           | 0.0239  | 64  |
| S2 a   | Water temperature versus log(Elevation)                                                      | $y = -0.0001222 \cdot x + 1.406$                                     |                  | 0.96           | <0.0001 | 44  |
| S2 b   | k <sub>600</sub> versus Elevation (for Main-stem)                                            | $y = 0.01869 \cdot x - 2.266$                                        |                  | 0.89           | <0.0001 | 25  |
| S2 b   | k <sub>600</sub> versus Elevation (for Tributaries)                                          | $y = 0.0436 \cdot x - 5.528$                                         |                  | 0.80           | <0.0001 | 17  |
| S3 a   | Grassland cover versus Elevation (for Main-stem)                                             | $y = 0.01994 \cdot x + 8.586$                                        |                  | 0.93           | <0.0001 | 25  |
| S3 a   | Grassland cover versus Elevation (for Tributaries)                                           | $y = 0.01444 \cdot x - 3.572$                                        |                  | 0.93           | <0.0001 | 17  |
| S3 b   | Forest cover versus Elevation (for Main-stem)                                                | $y = -0.01556 \cdot x + 68.6$                                        |                  | 0.85           | <0.0001 | 25  |
| S3 b   | Forest cover versus Elevation (for Tributaries)                                              | $y = -0.01371 \cdot x + 86.13$                                       |                  | 0.55           | 0.0025  | 17  |
| S3 c   | log(Inundation extent + 1) versus log(Elevation) (for Main-stem)                             | $y = -1.333 \cdot x + 4.702$                                         |                  | 0.49           | 0.0001  | 25  |
| S3 c   | log(Inundation extent + 1) versus log(Elevation) (for Tributaries)                           | $y = -1.027 \cdot x + 3.604$                                         |                  | 0.39           | 0.0078  | 17  |
| S4 a   | log(pCO <sub>2</sub> ) versus log(Grassland cover + 1)                                       | $y = -0.4003 \cdot x + 3.418$                                        |                  | 0.53           | <0.0001 | 39  |
| S4 b   | Inundation extent versus log(pCO <sub>2</sub> ) (for Main-stem)                              | $y = 0.002844 \cdot x + 2.799$                                       |                  | 0.35           | 0.0017  | 25  |
| S4 b   | Inundation extent versus log(pCO <sub>2</sub> ) (for Tributaries)                            | $y = 0.01206 \cdot x + 2.858$                                        |                  | 0.61           | 0.0002  | 17  |
| S4 c   | log(CH <sub>4</sub> ) versus log(Grassland cover + 1)                                        | $y = -0.8078 \cdot x + 2.95$                                         |                  | 0.47           | <0.0001 | 39  |
| S4 d   | log(CH <sub>4</sub> ) versus Inundation extent (for Main-stem)                               | $y = 0.009493 \cdot x + 1.635$                                       |                  | 0.34           | 0.0024  | 25  |
| S4 d   | log(CH <sub>4</sub> ) versus Inundation extent (for Tributaries)                             | $y = 0.01461 \cdot x + 1.879$                                        |                  | 0.27           | 0.0324  | 17  |
| S4 e   | log(%N <sub>2</sub> O) versus log(Grassland cover + 1)                                       | $y = -0.03509 \cdot x + 2.044$                                       |                  | 0.35           | 0.0002  | 36* |
| S4 f   | %N <sub>2</sub> O versus Inundation extent (for Main-stem)                                   | $y = 0.08749 \cdot x + 96.35$                                        |                  | 0.34           | 0.0022  | 25* |
| S4 f   | %N <sub>2</sub> O versus Inundation extent (for Tributaries)                                 | $y = 0.3674 \cdot x + 98.98$                                         |                  | 0.64           | 0.001   | 13* |
| S4 g   | log(%O <sub>2</sub> ) versus log(Grassland cover + 1)                                        | $y = 0.123 \cdot x + 1.685$                                          |                  | 0.49           | <0.0001 | 39  |
| S4 h   | log(%O <sub>2</sub> ) versus log(Inundation extent+1) (for Main-stem)                        | $y = -0.02816 \cdot x + 1.875$                                       |                  | 0.19           | 0.0309  | 25  |
| S4 h   | log(%O <sub>2</sub> ) versus log(Inundation extent+1) (for Tributaries)                      | $y = -0.1262 \cdot x + 1.88$                                         |                  | 0.46           | 0.0027  | 17  |
| S5 b   | NH <sub>4</sub> <sup>+</sup> cover versus Elevation (for Main-stem)                          | $y = 2.104 \cdot x - 10.73$                                          |                  | 0.57           | <0.0001 | 25  |
| S5 b   | NH <sub>4</sub> <sup>+</sup> cover versus Elevation (for Tributaries)                        | $y = 1.58 \cdot x - 8.237$                                           |                  | 0.71           | <0.0001 | 17  |
| S5 c   | DIN cover versus Elevation (for Main-stem)                                                   | $y = 1.989 \cdot x - 5.851$                                          |                  | 0.45           | 0.0004  | 25  |
| S5 c   | DIN cover versus Elevation (for Tributaries)                                                 | $y = 1.593 \cdot x - 4.679$                                          |                  | 0.62           | <0.0001 | 17  |
| S9 a   | F <sub>CO<sub>2</sub>+2</sub> versus Elevation                                               | $y = 3.858E+04 \cdot \text{EXP}(-2.674E-02 \cdot x) + 7.031E+01$     | 1.05E+02         |                |         | 42  |
| S9 b   | F <sub>CO<sub>2</sub>+2</sub> versus Upstream terrain slope                                  | $y = 6.936E+02 \cdot \text{EXP}(-6.181E-01 \cdot x) + 9.523E+01$     | 1.10E+02         |                |         | 42  |
| S9 d   | F <sub>CO<sub>2</sub></sub> versus pCO <sub>2</sub>                                          | $y = 0.1045 \cdot x + 6.451$                                         |                  | 0.45           | <0.0001 | 42  |
| S9 e   | F <sub>CH<sub>4</sub></sub> versus Elevation                                                 | $y = 7.324E+03 \cdot \text{EXP}(-1.036E-02 \cdot x) + 6.882E+02$     | 9.56E+02         |                |         | 42  |
| S9 f   | F <sub>CH<sub>4</sub></sub> versus Upstream terrain slope                                    | $y = 9.751E+03 \cdot \text{EXP}(-1.635E+00 \cdot x) + 7.622E+02$     | 8.02E+02         |                |         | 42  |
| S9 h   | F <sub>CH<sub>4</sub></sub> versus CH <sub>4</sub>                                           | $y = 2.711 \cdot x + 217$                                            |                  | 0.38           | <0.0001 | 42  |
| S9 i   | F <sub>N<sub>2</sub>O</sub> versus N <sub>2</sub> O                                          | $y = 0.316 \cdot x - 28.72$                                          |                  | 0.13           | 0.0206  | 42  |

\* Excluding Guamanyacu, Imiyana, Tiputini, and Huirima Rivers



# CHAPTER 3.

## DISTRIBUTION OF SUSPENDED MATTER AND ORGANIC CARBON POOLS IN THE MOUNTAINOUS AND PIEDMONT STREAMS OF THE NAPO RIVER BASIN (ECUADOR)



Rain falling in the Napo River, Community of Piñacocha (Piranha's Lake)

## **Distribution of suspended matter and organic carbon pools in the mountainous and piedmont streams of the Napo River basin (Ecuador)**

*This chapter is a draft of a manuscript in early stages of preparation, presenting a description of the patterns of organic pools in the streams of the Napo River basin that complements the analysis of greenhouse gases given in Chapter 2*

### **Introduction**

Rivers play an important role in the global carbon cycle, as conduits of organic and inorganic carbon from land to the ocean, although materials are processed during transport, leading to the production and emission to the atmosphere of greenhouse gases (GHGs) such as carbon dioxide (CO<sub>2</sub>) (Liu et al. 2022), methane (CH<sub>4</sub>) (Rosentreter et al. 2021) and nitrous oxide (N<sub>2</sub>O) (Maavara et al. 2018). Tropical rivers deliver ~60% of freshwater and an equivalent fraction of organic carbon to the oceans (Ludwig et al. 1996), and they sustain the majority of global riverine GHG emissions, ~60-80% for CO<sub>2</sub> (Raymond et al. 2013; Lauerwald et al. 2015; Sawakuchi et al. 2017; Liu et al. 2022), ~70% for CH<sub>4</sub> (Sawakuchi et al. 2014), and ~79% for N<sub>2</sub>O (Hu et al. 2016).

The Amazon River is globally the largest river in terms of drainage area (5.9 10<sup>6</sup> km<sup>2</sup>) and freshwater discharge (206,000 m<sup>3</sup> s<sup>-1</sup>) and drains the largest tropical rain forest on Earth. The Amazon River contributes 16-18% of continental freshwater discharge (Dai and Trenberth 2002), 7% of dissolved inorganic elements (Moquet et al. 2016) and ~10% of organic carbon (36 TgC yr<sup>-1</sup>; Richey et al. 1990) fluvial fluxes to the ocean globally (380 TgC yr<sup>-1</sup> Ludwig et al. 1996), of which 62% is in dissolved form. The flowing waters of the Central Amazon basin (river channels and floodplains) are important sources of CO<sub>2</sub> and CH<sub>4</sub> to the atmosphere (Richey et al. 2002; Melack et al. 2004; Abril et al. 2014; Sawakuchi et al. 2014; 2017; Barbosa et al. 2016). We recently showed that Andean mountainous headwater and piedmont streams and rivers represented 35% CO<sub>2</sub> and 72% CH<sub>4</sub> of basin scale integrated fluvial diffusive emissions from the Amazon River network (Chapter 2).

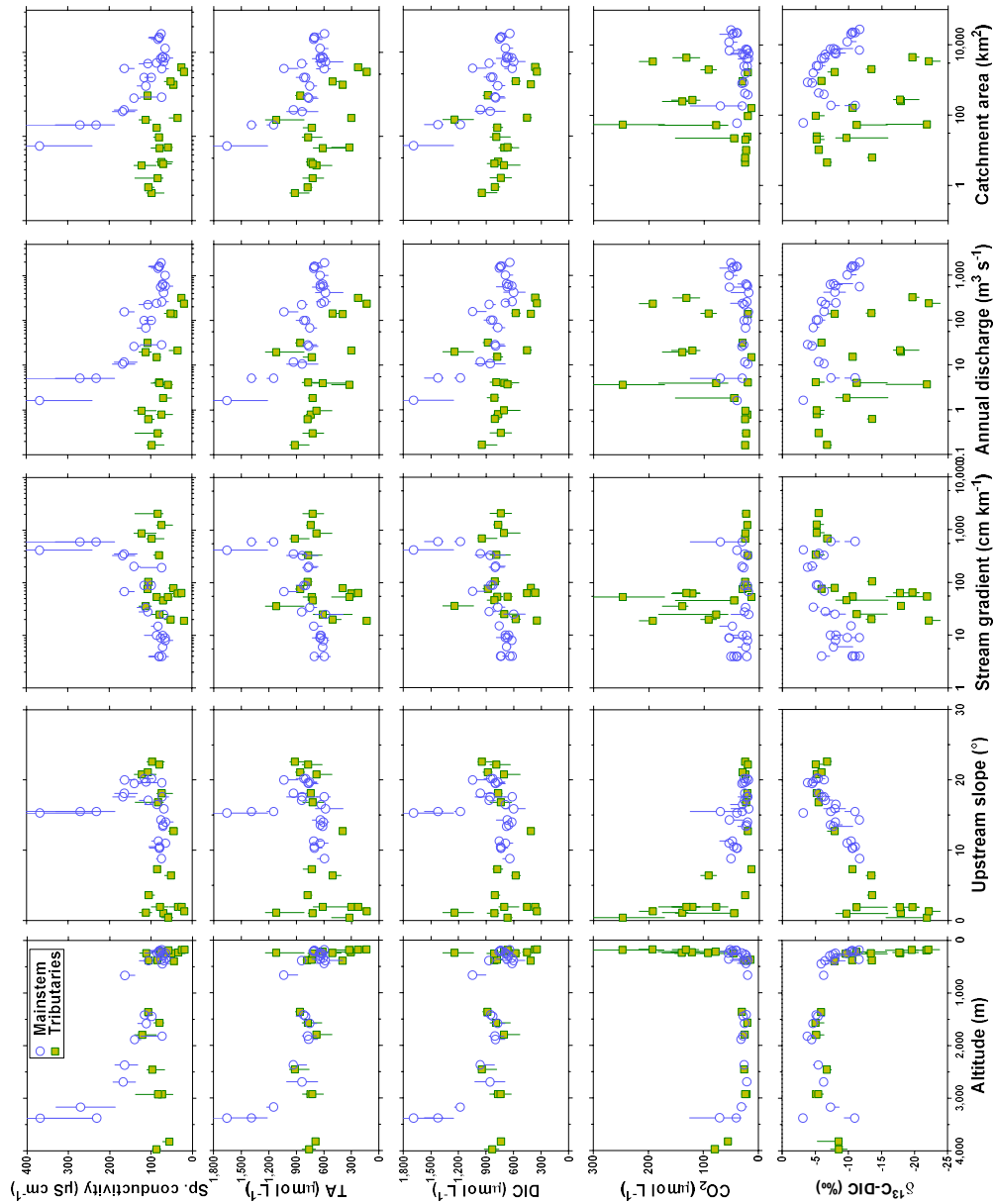
Rivers draining the Andes are part of Madeira River basin (Mamoré, Beni, Madre de Dios) and of the Solimões River (Ucayali, Huallaga, Marañón and Napo). They are characterized by high dissolved and particulate elemental concentrations and are typified as “white waters” (Sioli 1964). The rivers draining the Andes contribute to nearly 100% of the particulate (Meade et al. 1985) and 64% of inorganic solute (Gaillardet et al. 1997; Moquet et al. 2011; 2016) annual fluxes from the Amazon River to the Atlantic Ocean. Andean headwater streams largely control the distribution and composition of organic matter in the Amazon River (Richey et al. 1990; Quay et al. 1992; Hedges et al. 1994; 2000), although they remain largely unstudied (Townsend-Small et al. 2005; Aufdenkampe et al. 2007). There are differences in source, composition, and age of dissolved (DOM) and particulate (POM) organic matter in the Amazon rivers and streams, and in general DOM is richer in carbon, more degraded, and more recent than POM (Hedges et al. 1986; 1994; Richey et al. 1990; Quay et al. 1992; Mayorga et al. 2005).

Here, we report specific conductivity, total alkalinity (TA), CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O dissolved concentrations, dissolved inorganic carbon stable isotope ratios ( $\delta^{13}\text{C-DIC}$ ), TSM, POC and DOC organic carbon concentrations and stable isotope ratios ( $\delta^{13}\text{C-POC}$  and  $\delta^{13}\text{C-DOC}$ ), and three optical proxies of colored dissolved organic matter (CDOM) collected in the streams and rivers of the Napo basin along an elevational transect from 3990 and 175 m above sea level, covering the transition from headwaters in the Andes Cordillera to the western Amazonian lowland plains.

## Results and discussion

Specific conductivity showed a downstream decrease from 368 to 19  $\mu\text{S cm}^{-1}$  in the mainstem (Fig. 3.1). In the tributaries, specific conductivity varied within a narrower range from 19 to 122  $\mu\text{S cm}^{-1}$ , showing lower values at an elevation < 1000 m than > 1000 m. Specific conductivity was positively related to stream gradient, and negatively related to annual discharge and catchment area, but, was unrelated to upstream catchment slope (Fig. 3.1).

**FIGURE 3.1** Specific conductivity ( $\mu\text{S cm}^{-1}$ ), total alkalinity ( $\mu\text{mol L}^{-1}$ ), dissolved inorganic carbon (DIC in  $\mu\text{mol L}^{-1}$ ),  $\text{CO}_2$  dissolved concentration ( $\mu\text{mol L}^{-1}$ ), and carbon stable isotope ratios of DIC ( $\delta^{13}\text{C}\text{-DIC}$  in ‰) versus elevation (m), upstream catchment slope ( $^\circ$ ), stream gradient ( $\text{cm km}^{-1}$ ), annual freshwater discharge ( $\text{m}^3 \text{s}^{-1}$ ) and catchment area ( $\text{km}^2$ ) from RiverATLAS (Linke et al. 2019) in the Napo streams.



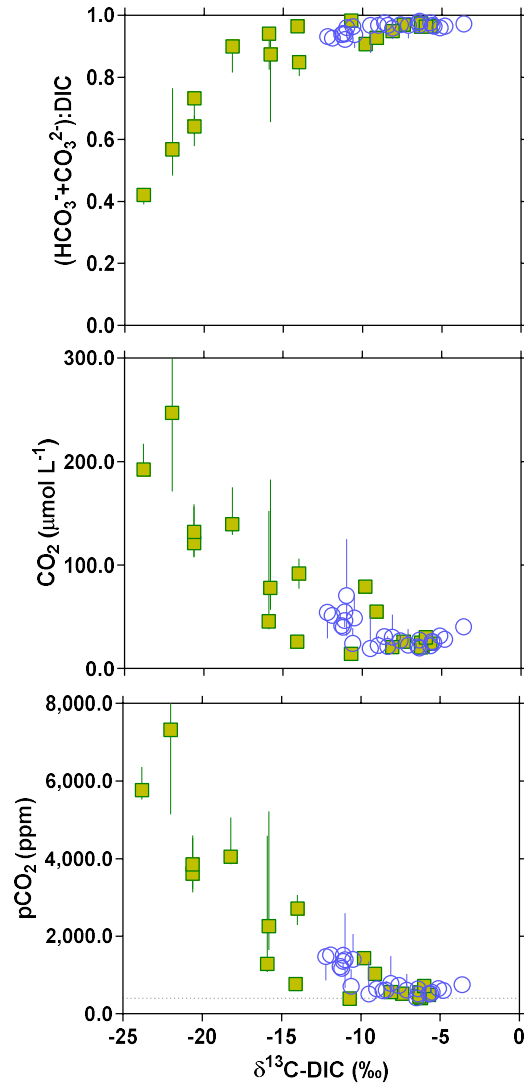
TA ranged between 135 and 1,656  $\mu\text{mol L}^{-1}$  and exhibited patterns that paralleled very closely those of specific conductivity (Fig. 3.1). The elevational patterns of specific conductivity and TA resulted from intense rock weathering in the high elevation areas of the Andes cordillera (Gaillardet et al. 1997; Moquet et al. 2011; 2016) followed by dilution in the lower reaches with the increase of stream size, as indicated by negative correlations with catchment area and discharge. DIC ranged between 348 and 1,689  $\mu\text{mol L}^{-1}$  and exhibited patterns that paralleled very closely those of TA as a result of the important contribution of  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$  to DIC and TA (Fig. 3.1). The dissolved  $\text{CO}_2$  concentrations ranged between 14 and 247  $\mu\text{mol L}^{-1}$  and showed very different patterns from those of DIC, increasing exponentially downstream and decreasing with upstream catchment slope, with no clear patterns as a function of stream gradient, freshwater discharge, and catchment area (Fig. 3.1). The drivers of the variations of dissolved  $\text{CO}_2$  concentrations were investigated in detail in Chapter 2, where the elevational gradient of the partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$ ) was attributed to variations in gas transfer velocity, forest cover, inundation extent, and water temperature, that scaled with upstream slope gradient. Upstream catchment slope determines to a large extent broad patterns in the variation of soil organic carbon content (Thompson and Kolka 2005; de Brogniez et al. 2015; Connolly et al. 2018), ground water discharge (Leach et al. 2017), stream water residence time (McGuire et al. 2005), and wetland abundance (Merot et al. 1999; Rodhe and Seibert 1999), that, combined, have a major impact on fluvial carbon cycling, including  $\text{CO}_2$  dynamics, and as such was found to be an integrative metric for processes that affect dissolved  $\text{CO}_2$  in the streams of the Napo basin, and also at larger scale for the whole Amazon basin (Chapter 2).

The  $\delta^{13}\text{C}$ -DIC values ranged between -22.1 and +3.1 ‰ and showed a downstream decrease, generally mirroring the variations of dissolved  $\text{CO}_2$  concentration (Fig. 3.1).  $\delta^{13}\text{C}$ -DIC was positively related to upstream catchment slope. When taking separately the tributaries and the main-stem, the  $\delta^{13}\text{C}$ -DIC was positively related to stream gradient and negatively related to catchment area and freshwater discharge, in line with the patterns of  $\text{HCO}_3^-$  as given by TA. The variations of  $\delta^{13}\text{C}$ -DIC can provide information on the origin of  $\text{CO}_2$ , although the signal depends on the combination of several factors such as the biological processes that remove or add  $\text{CO}_2$  to the water column (respiration and photosynthesis), rock weathering that adds  $\text{HCO}_3^-$ , and outgassing to the atmosphere which removes dissolved  $\text{CO}_2$  that is  $^{13}\text{C}$ -depleted relative to  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$ . Spatial and temporal changes of  $\delta^{13}\text{C}$  are related to the change of the relative abundance of  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$  over  $\text{CO}_2$ . In the Tana River network, the downstream increase of  $\delta^{13}\text{C}$ -DIC was attributed to a downstream accumulation of  $\text{HCO}_3^-$  due to rock weathering combined to  $\text{CO}_2$  degassing (Bouillon et al. 2009; Tamooh et al. 2003). In contrast, the downstream decrease of  $\delta^{13}\text{C}$ -DIC observed in the Napo basin (Fig. 3.1), as well as the negative relation between  $\delta^{13}\text{C}$ -DIC and  $\text{CO}_2$  concentration (and  $p\text{CO}_2$ ) (Fig. 3.2) suggest in the lowland streams the combination of addition of  $\text{CO}_2$  depleted in  $^{13}\text{C}$  from heterotrophic process and dilution of  $\text{HCO}_3^-$ . There was a general increase of  $\delta^{13}\text{C}$ -DIC from  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC} \sim 0.4$  (60% the DIC is in the form of  $\text{CO}_2$ ) towards  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC} \sim 1$  (nearly all of the DIC is in the form of  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$ ) (Fig. 3.2). This pattern reflected the mixing of two distinct types of rivers and streams: the mountainous systems draining the highland regions of the Andes with high rock weathering (high  $\text{HCO}_3^-$ ) and low  $\text{CO}_2$  due high degassing to the atmosphere owing to high gas transfer velocity values in these steep streams (Chapter 2), leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$  close to 1 with high  $\delta^{13}\text{C}$ -DIC, and the lowland streams in the Amazonian plains with a high input of  $\text{CO}_2$  from respiration in forest soils and wetlands and a dilution of  $\text{HCO}_3^-$ , leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC} \sim 0.4$  with low  $\delta^{13}\text{C}$ -DIC.

TSM, fluvial C:N, POC and DOC concentrations showed clear elevational patterns (Fig. 3.3). TSM showed a consistent downstream increase from 3 to 480  $\text{mg L}^{-1}$ . TSM values were lower than 12  $\text{mg L}^{-1}$  at elevations above 1000 m, suggesting that the higher particulate matter content transported by lowland rivers originated from low elevation areas (<1000 m) rather than from headwater regions. The stream size increased downstream, so hydraulic variables (upstream catchment slope, stream gradient, catchment area, and freshwater discharge) increased with decreasing elevation (Fig. 3.4). Consequently, TSM was negatively related to upstream

catchment slope and stream gradient, and positively related to freshwater discharge and catchment area. These patterns of TSM reflected the accumulation of particles originated from erosion with the downstream increase of stream size. The relation between TSM and upstream catchment slope was different for tributaries and main-stem, because the relation between elevation and hydraulic variables, including upstream catchment slope, was different for tributaries and main-stem (Fig. 3.4).

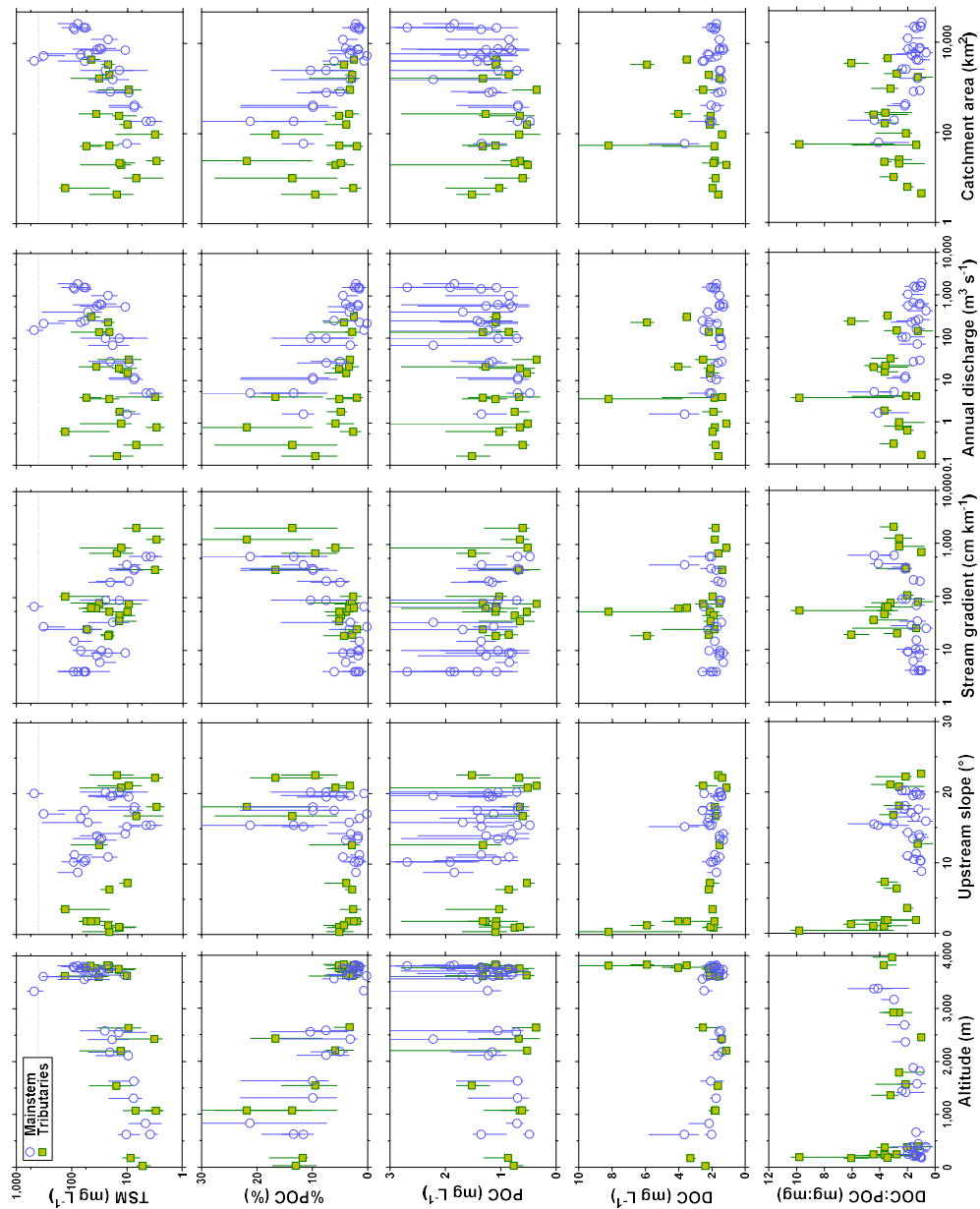
**FIGURE 3.2:** Ratio of the sum of  $\text{HCO}_3^-$  and  $\text{CO}_3^{2-}$  and dissolved inorganic carbon (DIC),  $\text{CO}_2$  dissolved concentration ( $\mu\text{mol L}^{-1}$ ), and partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$  in ppm) versus carbon stable isotope ratios of DIC ( $\delta^{13}\text{C-DIC}$  in ‰) in the Napo streams.



POC concentration ranged between 0.4 and 2.7  $\text{mg L}^{-1}$  and was positively related to TSM (Fig. 3.5). The relative contribution of POC to TSM (%POC) ranged from 0.2 and 21.8% and was negatively related to TSM (Fig. 3.5). At a comparable elevation range ( $>1000$ ;  $<4800\text{m}$ ) the range %POC (3 to 22%) we observed in the Napo streams falls within the range (2 to 60%) reported by in the Peruvian headwaters of the Amazon (Townsend-Small et al. 2005). POC followed the same patterns as TSM, increasing downstream, with freshwater discharge and catchment area, and decreasing with stream gradient and upstream catchment slope. %POC showed the opposed

patterns of TSM, decreasing downstream, with freshwater discharge and catchment area, and increasing with stream gradient and upstream catchment slope. These patterns in %POC probably reflected a better preservation of organic matter fraction in colder high elevation region, the in-stream loss by microbial respiration of organic carbon and the dilution of POC by mineral particles during downstream transport.

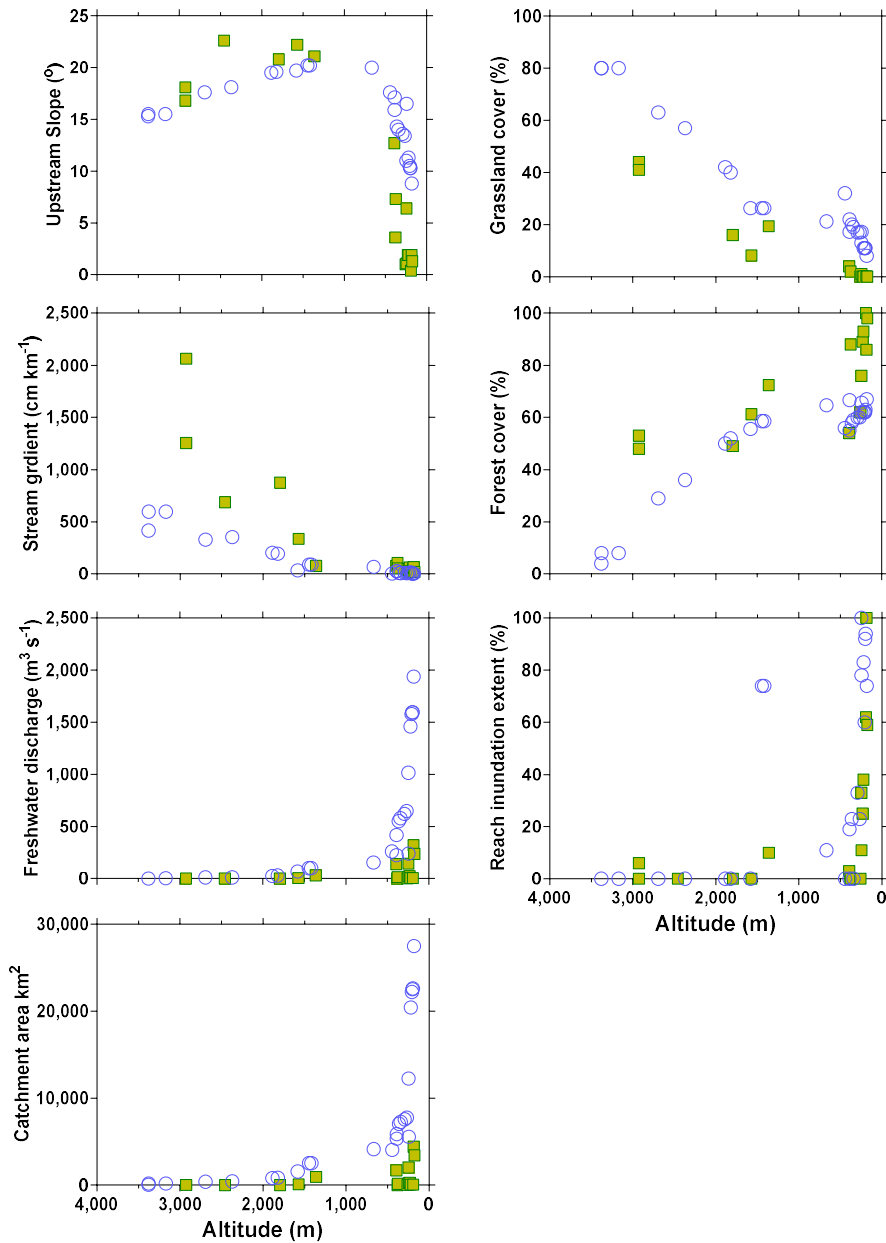
**Figure 3.3:** Total suspended matter (TSM in  $\text{mg L}^{-1}$ ), fraction of particulate organic carbon (POC) in TSM (%POC in %), POC ( $\text{mg L}^{-1}$ ), dissolved organic carbon (DOC in  $\text{mg L}^{-1}$ ), and DOC:POC ratio versus elevation (m), upstream catchment slope ( $^{\circ}$ ), stream gradient ( $\text{cm km}^{-1}$ ), annual freshwater discharge ( $\text{m}^3 \text{s}^{-1}$ ) and catchment area ( $\text{km}^2$ ) from RiverATLAS (Linke et al. 2019) in the Napo streams.



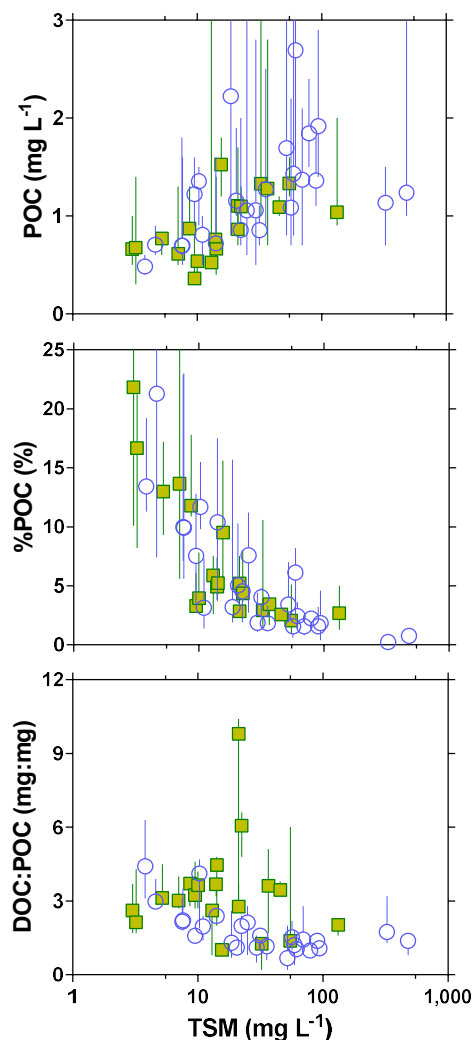
The C:N ratio of fluvial particulate organic matter (POM) ranged between 7 and 11 and decreased downstream (Fig. 3.6). The values C:N of fluvial POM we report in the Napo basin

were consistent with those reported by Townsend-Small et al. (2005) in Andean streams in Peru along an elevational gradient from 720 to 4,043 m (with the majority of C:N values ranging between 5 and 15 in the POM fraction <60  $\mu\text{m}$ ). The downstream decrease of the C:N ratio of fluvial POM is consistent with the decrease from highlands to lowlands of soil C:N ratio reported by Townsend-Small et al. (2005) in the Peruvian Andes that was attributed to a higher microbial source (with low C:N) in low-elevation soils compared to high-elevation soils with a higher contribution of plant source (with high C:N), due to elevational gradients of temperature.

**Figure 3.4:** Upstream catchment slope ( $^{\circ}$ ), stream gradient ( $\text{cm km}^{-1}$ ), annual freshwater discharge ( $\text{m}^3 \text{s}^{-1}$ ), catchment area ( $\text{km}^2$ ), grassland cover on the upstream catchment (%), forest cover on the upstream catchment (%), and inundation extent at reach level from RiverATLAS (Linke et al. 2019) versus elevation (m), in the Napo streams.



**Figure 3.5:** Particulate organic carbon (POC in  $\text{mg L}^{-1}$ ), fraction of POC in TSM (%POC in %), dissolved organic carbon (DOC in  $\text{mg L}^{-1}$ ) to POC ratio versus total suspended matter (TSM in  $\text{mg L}^{-1}$ ) in the Napo streams.



DOC concentrations ranged between 1.2 and 8.2  $\text{mg L}^{-1}$  (Fig. 3.3), and at a comparable elevation range (>1000;<4800m) the range DOC (1.2 to 3.3  $\text{mg L}^{-1}$ ) we observed in the Napo streams falls within the range (0.8 to 10.7  $\text{mg L}^{-1}$ ) reported by in the Peruvian headwaters of the Amazon (Townsend-Small et al. 2005; Audfdenkampe et al. 2007). DOC ranged within a narrow range (1.2-3.7  $\text{mg L}^{-1}$ ) with a downstream decreasing pattern in the region between 4,000 and 1,350m, reflecting the increase of the soil organic carbon content with elevation because of the decrease of temperature that leads to a decrease of organic matter mineralization in soils enhancing organic matter accumulation in soils (Poulenard et al. 2003). At elevations < 1000 m, DOC values > 3  $\text{mg L}^{-1}$  were observed in four streams. DOC decreased with upstream catchment slope but was unrelated to the other three hydraulic variables (stream gradient, freshwater discharge, and catchment area). This would suggest that elevational change of stream size was not the primary control of fluvial DOC that might be alternatively controlled by soil organic carbon inputs, as thus vary with elevational gradients in vegetation cover on the catchment. DOC was positively related to forest cover (Fig. 3.7) suggesting that leaching of DOC from forest soils strongly enhanced fluvial DOC content.

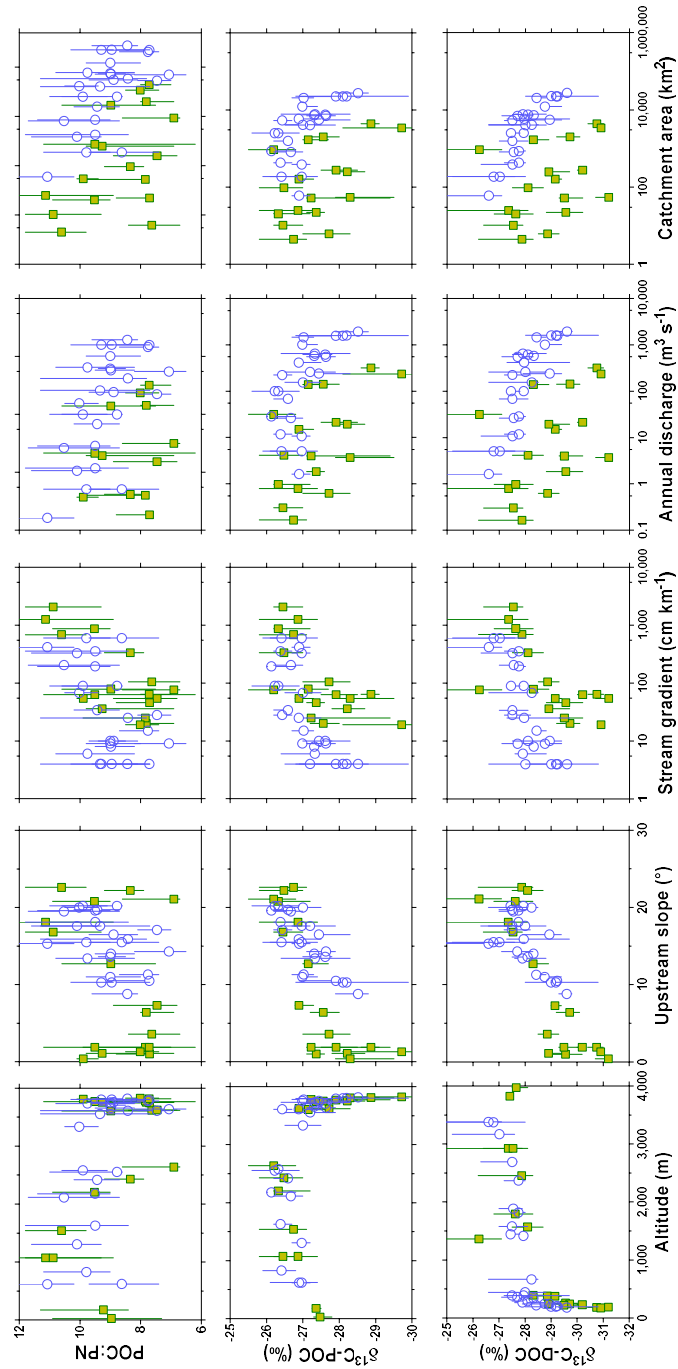


Note that the relation between DOC and upstream catchment slope (Fig. 3.3) in fact also reflected indirectly the dependency of DOC on forest cover, as forest cover was negatively related to upstream catchment slope (Fig. 3.8). Fluvial DOC did not show a clear pattern with inundation extent at reach level (Fig. 3.3). When analyzing the data separately for tributaries and main-stem, there was a weak positive relation between fluvial DOC and inundation extent for tributaries (excluding the data point with the highest freshwater discharge) but no relation for the main-stem. The DOC to POC (DOC:POC) ratio ranged between 0.7 and 9.8, and exhibited variations that generally followed those of DOC, with a general downstream decreasing weak pattern in the region between 4,000 and 1,500m, and a strong increase in the lowland forested regions (Fig. 3.3). The DOC:POC ratio decreased with upstream catchment slope but was unrelated to the other three hydraulic variables (stream gradient, catchment area and freshwater discharge), reflecting the dependency of forest cover on upstream catchment slope (Fig. 3.8), as the DOC:POC ratio was positively related to forest cover (Fig. 3.7), and negatively related to TSM (Fig. 3.5). The relations between TSM, POC, and %POC with vegetation cover (Fig. 3.7) might be spurious and reflect indirectly the relation between vegetation cover and elevation (Fig. 3.4) or upstream catchment slope (Fig. 3.8). Indeed, the patterns of TSM that imposed those of POC and %POC reflected elevational gradients of erosion and downstream accumulation of particles with stream size.

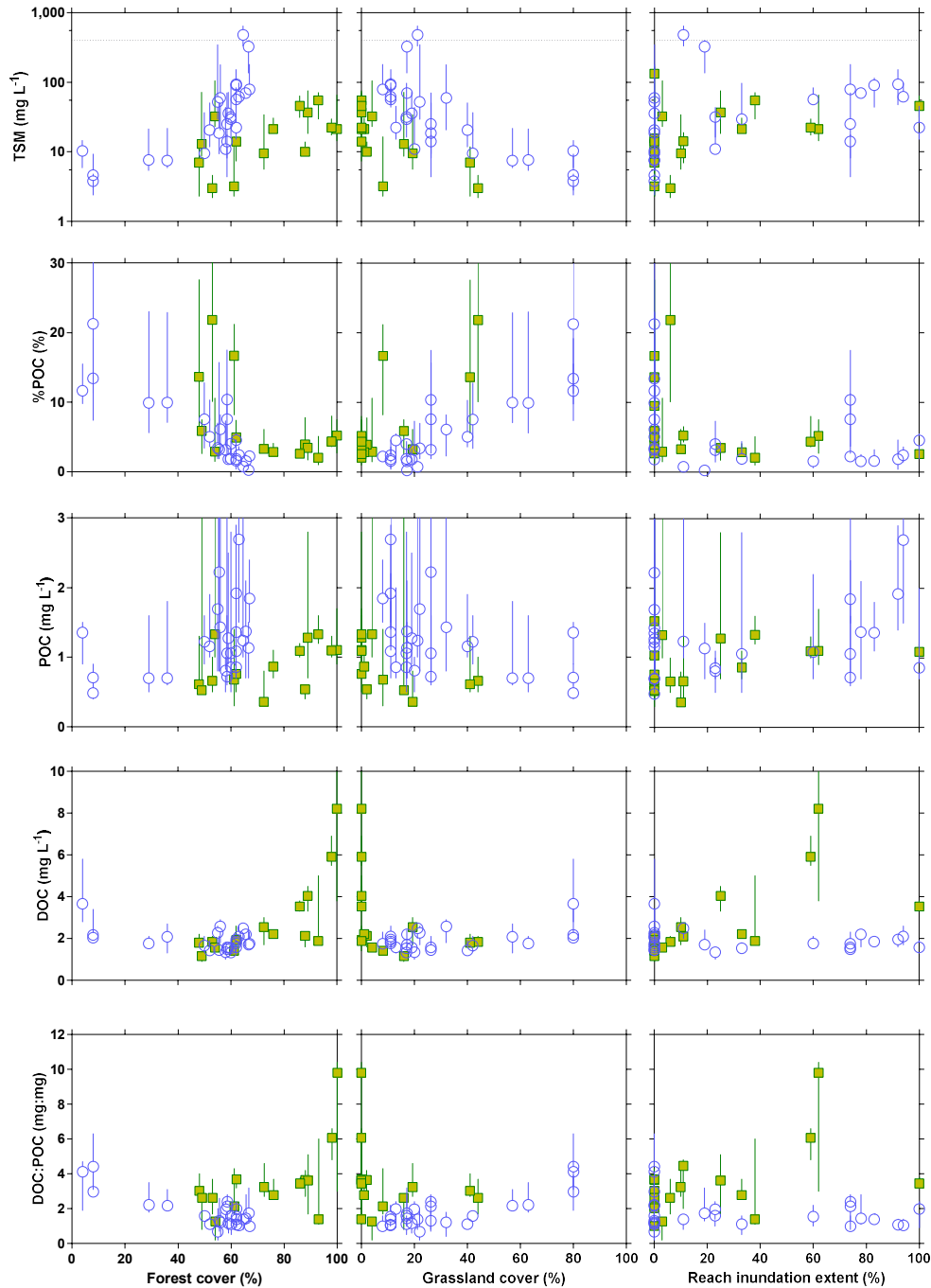
CDOM was characterized with three optical proxies: the absorption coefficient at 350 nm ( $a_{350}$ ) that quantifies the content of CDOM, the slope ratio ( $S_R$ ) that is inversely related to the molecular weight distribution of DOM, with values less than 1 indicative of enrichment in high-molecular-weight compounds (Helms et al. 2008), and the specific ultraviolet absorbance at 254 nm ( $SUVA_{254}$ ) that is positively related to the degree of DOM aromaticity, with high values ( $>3.5 \text{ L mg}^{-1} \text{ m}^{-1}$ ) indicative of the presence of more complex aromatic moieties and low values ( $<3.0 \text{ L mg}^{-1} \text{ m}^{-1}$ ) indicative of the presence of more aliphatic compounds (Weishaar et al. 2003).

The  $a_{350}$  values varied from 2.4 to 31.6  $\text{m}^{-1}$  and generally followed an exponential increase from high- to low-lands (Fig. 3.9). The  $a_{350}$  values in the two streams at the highest elevation ( $>3,500 \text{ m}$ ) were higher than the other mountainous streams at elevations  $> 1000 \text{ m}$ , reflecting the high organic carbon content in the páramos combined to small stream size. The  $SUVA_{254}$  values ranged between 2.1 and 6.1  $\text{L mg}^{-1} \text{ m}^{-1}$ , was highest in the in the two streams at the highest elevation ( $>3,500 \text{ m}$ ), then remains relatively stable (2.1-3.9  $\text{L mg}^{-1} \text{ m}^{-1}$ ) between 3,400 and 1,361 m elevation, and finally increase in the lowlands. The  $a_{350}$  and  $SUVA_{254}$  values were negatively correlated to upstream catchment slope but not  $S_R$ . Like DOC, the three DOM optical proxies were unrelated to other hydraulic variables (stream gradient, freshwater discharge, and catchment area). Like DOC, the  $a_{350}$  values were positively correlated to forest cover (Fig. 3.10), and the correlation with upstream catchment slope probably also reflects indirectly the relation with forest cover (Fig. 3.8). The  $a_{350}$  values were well correlated to DOC, the slope of the linear regression was higher for streams with  $>50\%$  forest cover in the catchment than streams with  $<50\%$  forest cover (Fig. 3.11). Lower  $a_{350}$  values at similar DOC levels in environments with a low forest cover indicates that DOM released from high-latitude grassland dominated soils is less optically active (that is, less absorptive) compared to DOM released from forest soils. Diverging patterns of  $a_{350}$  versus DOC were shown previously in Africa for streams draining catchments dominated by savannah versus forest (Lambert et al. 2006). The  $SUVA_{254}$  values were positively correlated to forest cover and negatively correlated to grassland cover on the catchment (Fig. 3.10). This is indicative that a higher forest cover in the lowlands led to a higher quantity in streams of highly aromatic DOM. The  $S_R$  values ranged between 0.6 and 1.0 and increased downstream, indicating a stronger enrichment of high-molecular-weight DOM compounds in highlands (Fig. 3.9), although, the  $S_R$  values did not show clear patterns as a function of land cover.

**Figure 3.6:** Particulate organic carbon (POC) to particulate nitrogen (N) ratio (C:N molar), carbon stable isotope ratios of POC ( $\delta^{13}\text{C-POC}$  in ‰), carbon stable isotope ratios of dissolved organic carbon ( $\delta^{13}\text{C-DOC}$  in ‰) versus elevation (m), upstream catchment slope ( $^{\circ}$ ), stream gradient ( $\text{cm km}^{-1}$ ), annual freshwater discharge ( $\text{m}^3 \text{s}^{-1}$ ) and catchment area ( $\text{km}^2$ ) from RiverATLAS (Linke et al. 2019) in the Napo streams.

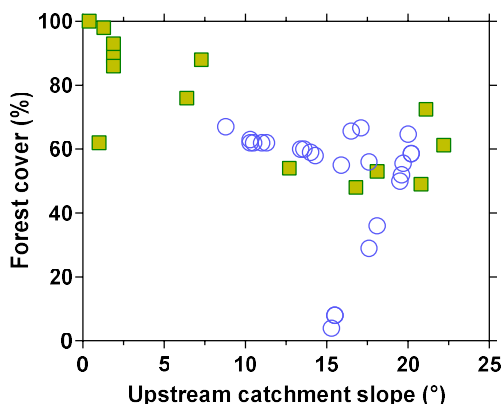


**Figure 3.7:** Total suspended matter (TSM in  $\text{mg L}^{-1}$ ), fraction of particulate organic carbon (POC) in TSM (%POC in %), POC ( $\text{mg L}^{-1}$ ), dissolved organic carbon (DOC in  $\text{mg L}^{-1}$ ), and DOC:POC ratio versus forest cover on the upstream catchment (%), grassland cover on the upstream catchment (%), and inundation extent at reach level from RiverATLAS (Linke et al. 2019) in the Napo streams.



There was a general positive relation between  $\text{pCO}_2$  and DOC (Fig. 3.12) that has been used to infer the role of terrestrial organic matter inputs on sustaining net heterotrophy and  $\text{CO}_2$  production in inland waters (Lapierre and del Giorgio 2012). However, a  $\text{pCO}_2$  and DOC relation can also be interpreted as an indication of simultaneous inputs from the catchment of both DOC and  $\text{CO}_2$  derived from soils and/or wetlands.

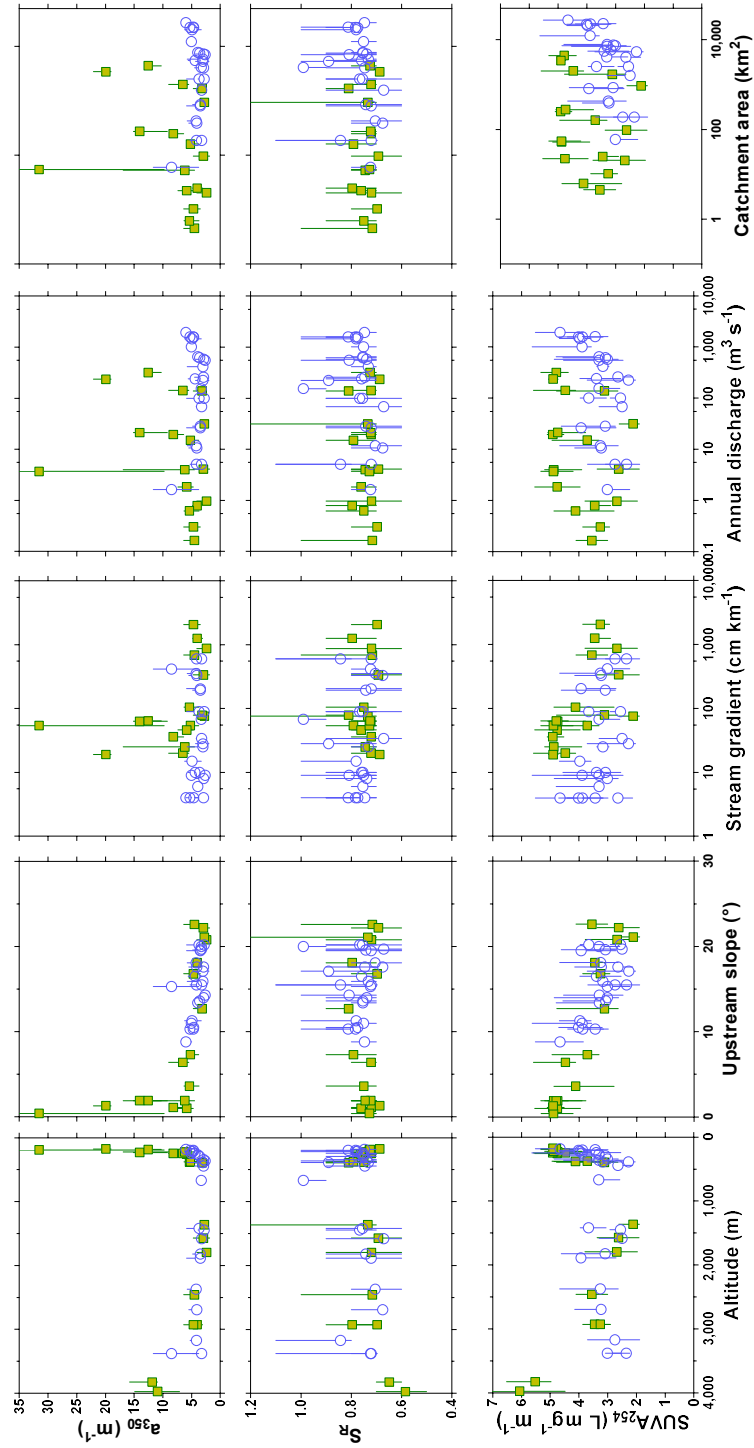
**Figure 3.8:** Forest cover on the upstream catchment (%) versus upstream catchment slope (°) from RiverATLAS (Linke et al. 2019) in the Napo streams.



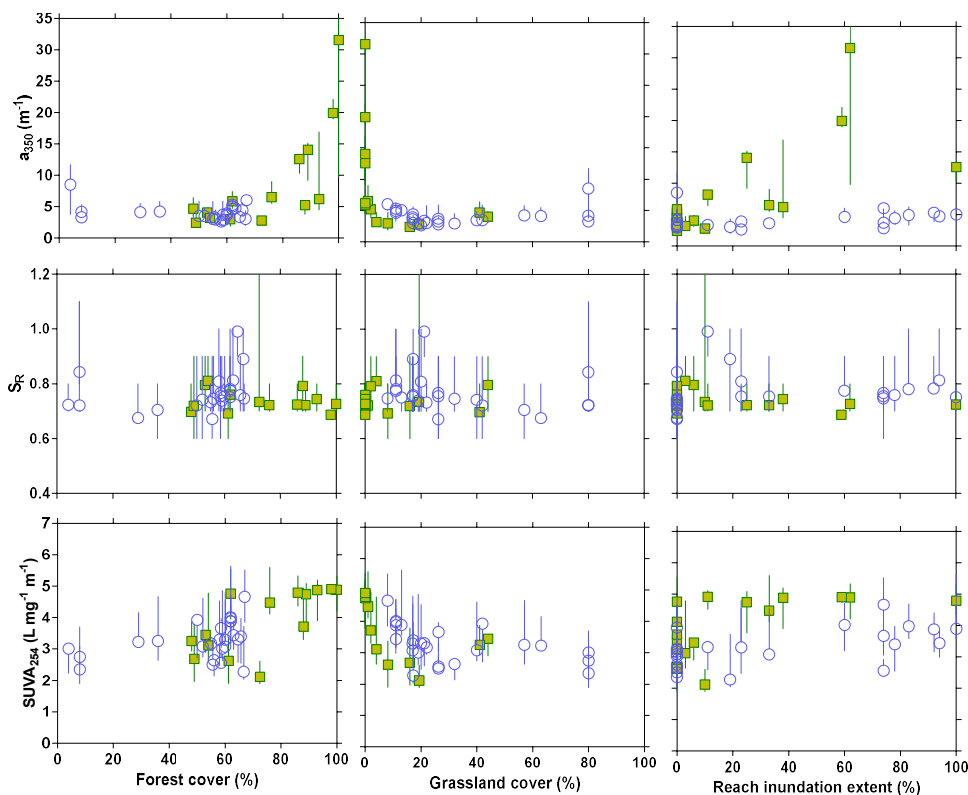
In African streams and rivers,  $p\text{CO}_2$  strongly correlated with DOC, although, pelagic respiration was insufficient to sustain  $\text{CO}_2$  emissions to the atmosphere, implying a strong lateral input of  $\text{CO}_2$  from soils and wetlands (Borges et al. 2015; 2019). There was no relation between the other two GHGs ( $\text{CH}_4$  and  $\text{N}_2\text{O}$ ) and DOC (Fig. 3.12), although a relation between  $\text{CH}_4$  and DOC emerged from a global analysis in rivers and streams (Stanley et al. 2016). In the Napo basin, the strong increase of  $\text{CH}_4$  concentrations in the piedmont streams is attributed to an increase of organic matter supply to streams from forests, but also in relation to deposition of sediments as reported in the Mamoré river (Vauchel et al. 2017), while the lower  $\text{CH}_4$  concentrations in the mountainous streams seem to be related to intense degassing to the atmosphere related high gas transfer velocities (Chapter 2). The lack of relation between  $\text{CH}_4$  and DOC could be explained by multiple drivers acting simultaneously on  $\text{CH}_4$  dynamics in the streams of the Napo in addition to organic matter delivery to streams, that might become the main explaining variable in an analysis at global scale (Stanley et al. 2016).

There was no relation between any of the three GHGs and POC (Fig. 3.12).  $p\text{CO}_2$  correlated positively with  $a_{350}$  (Fig. 3.13) in line with the correlation between  $p\text{CO}_2$  and DOC (Fig. 3.12), since  $a_{350}$  and DOC are tightly correlated. Yet, this relation is in line with the idea that inland waters with a higher CDOM content have higher  $\text{CO}_2$  levels as shown for African (Borges et al. 2022) and boreal lakes (Sobek et al. 2003). This is also in line with the idea that  $\text{CO}_2$  content increased in African rivers with presence of wetlands on the catchment (Borges et al. 2015) that in turn also increased the DOC and CDOM content (Lambert et al. 2016a;). However,  $p\text{CO}_2$  did not correlate  $S_R$  and  $\text{SUVA}_{254}$  which showed that other processes (lateral inputs and gas exchange) (Chapter 2) had an effect on  $p\text{CO}_2$  that overshadowed the effect of modest changes in the specific composition of DOM. There was no relation between  $\%\text{N}_2\text{O}$  and  $\text{CH}_4$  and CDOM optical proxies (not shown), although there was on report of a potential contribution of CDOM to  $\text{CH}_4$  production in eutrophic Lake Taihu (Zou et al. 2018).

**Figure 3.9:** Absorption coefficient at 350 nm ( $a_{350}$  in  $m^{-1}$ ), slope ratio ( $S_R$  unitless), and the specific ultraviolet absorbance at 254 nm ( $SUVA_{254}$  in  $L\ mg^{-1}\ m^{-1}$ ) of colored dissolved organic matter versus elevation (m), upstream catchment slope ( $^{\circ}$ ), stream gradient ( $cm\ km^{-1}$ ), annual freshwater discharge ( $m^3\ s^{-1}$ ) and catchment area ( $km^2$ ) from RiverATLAS (Linke et al. 2019) in the Napo streams.



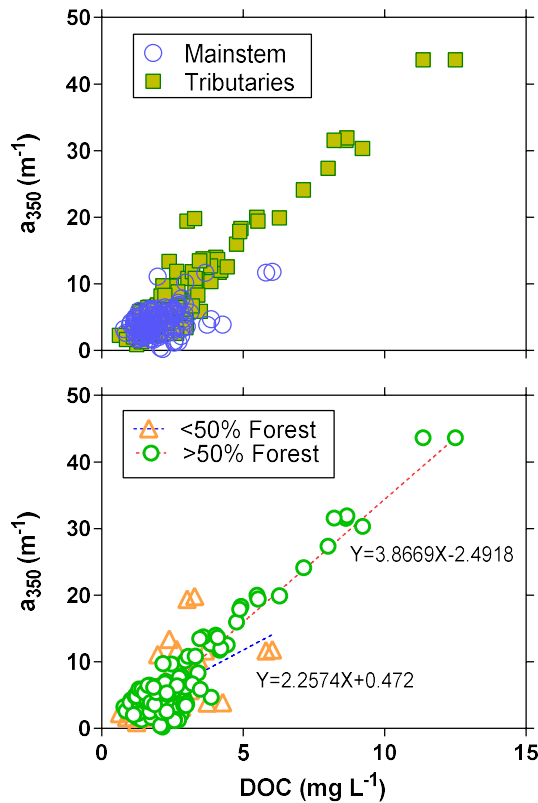
**Figure 3.10:** Absorption coefficient at 350 nm ( $a_{350}$  in  $m^{-1}$ ), slope ratio ( $S_R$  unitless), and the specific ultraviolet absorbance at 254 nm ( $SUVA_{254}$  in  $L\ mg^{-1}\ m^{-1}$ ) of colored dissolved organic matter *versus* forest cover on the upstream catchment (%), grassland cover on the upstream catchment (%), and inundation extent at reach level from RiverATLAS (Linke et al. 2019) in the Napo streams.



The fluvial  $\delta^{13}C$ -DOC values ranged between  $-31.2$  and  $-26.2$  ‰ and  $\delta^{13}C$ -POC values ranged between  $-29.7$  and  $-26.1$  ‰, and they decreased exponentially with elevation (Fig. 3.6). This range of  $\delta^{13}C$  in organic matter is much broader than the one typical reported in the lowland streams and rivers  $-29.8$  to  $-26.5$  ‰ (Quay et al. 1992; Hedges et al. 1994; Mayorga et al. 2005) and is consistent with the range reported by Audfdenkampe et al. (2007) in the Peruvian headwaters of the Amazon ( $-29.6$  to  $-21.8$  ‰). The  $\delta^{13}C$ -DOC was correlated to  $\delta^{13}C$ -POC, although the bulk concentrations of DOC was unrelated to POC (Fig. 3.14). The fluvial  $\delta^{13}C$ -DOC and  $\delta^{13}C$ -POC values followed the elevational gradient of  $\delta^{13}C$  of organic carbon from terrestrial vegetation (Fig. 3.15). The change  $\delta^{13}C$  of organic carbon from terrestrial vegetation reflected the shift from C4 dominated grasslands in highlands to C3 dominated forests in lowlands (Bird and Pousai 1997), consistent with the negative relations between fluvial  $\delta^{13}C$ -DOC and  $\delta^{13}C$ -POC with forest cover, and conversely the positive relations between fluvial  $\delta^{13}C$ -DOC and  $\delta^{13}C$ -POC with grassland cover (Fig. 3.16). Fluvial  $\delta^{13}C$ -DOC and  $\delta^{13}C$ -POC values were also negatively (albeit weakly) related to inundation extent which would suggest that riparian wetlands are dominated by flooded forest (C3 vegetation) rather than macrophytes (C4 vegetation). In addition to shifts in C3 and C4 plants, the change  $\delta^{13}C$  of organic carbon from terrestrial vegetation might also be related to a lower isotopic fractionation of carbon during photosynthesis due the decrease of atmospheric partial pressure of both  $CO_2$  and  $O_2$ , and decreasing air temperature with increasing elevation (Körner et al. 1988; 1991). The  $\delta^{13}C$ -DOC and  $\delta^{13}C$ -POC values increased with upstream catchment slope and stream gradient, and decreased with freshwater discharge and catchment area, following the gradients of these hydraulic variables with elevation, but with

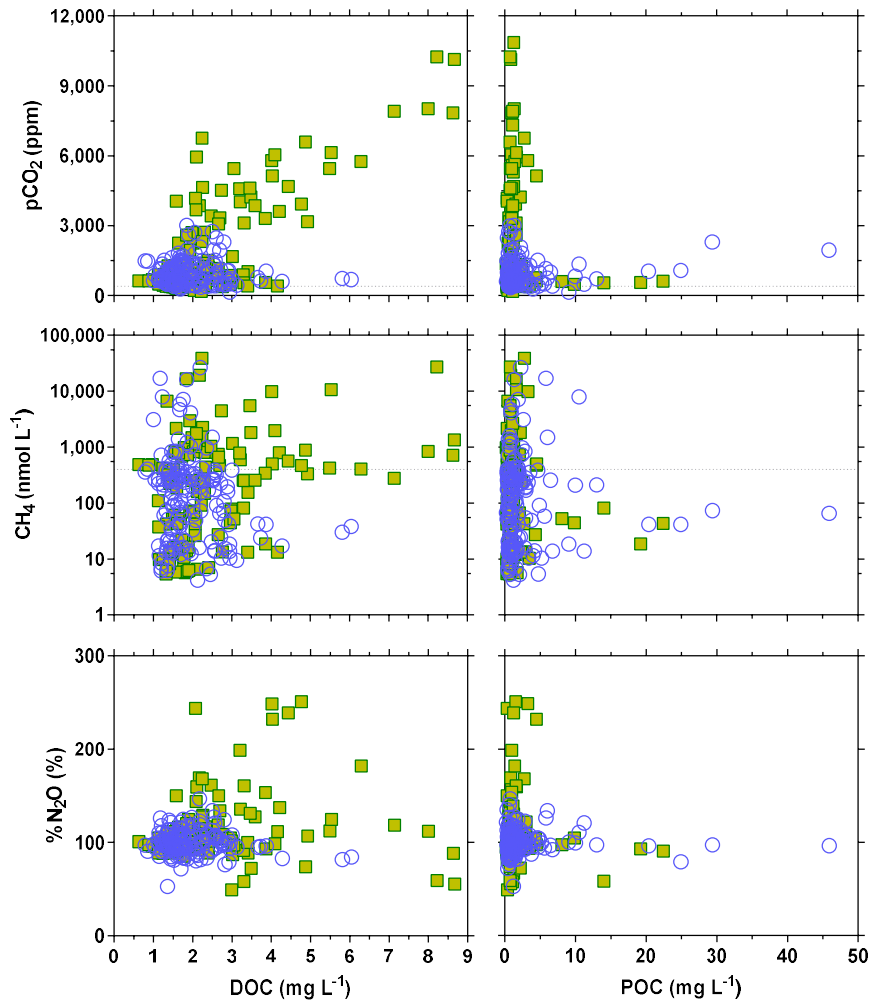
distinct relations for main-stem and tributaries. Indeed, these hydraulic variables showed distinct relations for main-stem and tributaries as a function of elevation, due to stream order effect. Additionally, the forest cover correlated with upstream catchment slope (Fig. 3.8). Although linearly correlated, the  $\delta^{13}\text{C}$  values were lower in DOC than POC by about 1‰ (Fig. 3.14), probably reflecting the more refractory nature of riverine DOC compared to POC, the latter being removed faster during downstream transport, being gradually replaced by POC a higher  $\delta^{13}\text{C}$  value from the forest vegetation.

**Figure 3.11:** Absorption coefficient at 350 nm ( $a_{350}$  in  $\text{m}^{-1}$ ) versus dissolved organic carbon (DOC in  $\text{mg L}^{-1}$ ) separated into main-stem and tributaries (top panel) and into streams with a forest cover on the upstream catchment > and < 50% (bottom panel).



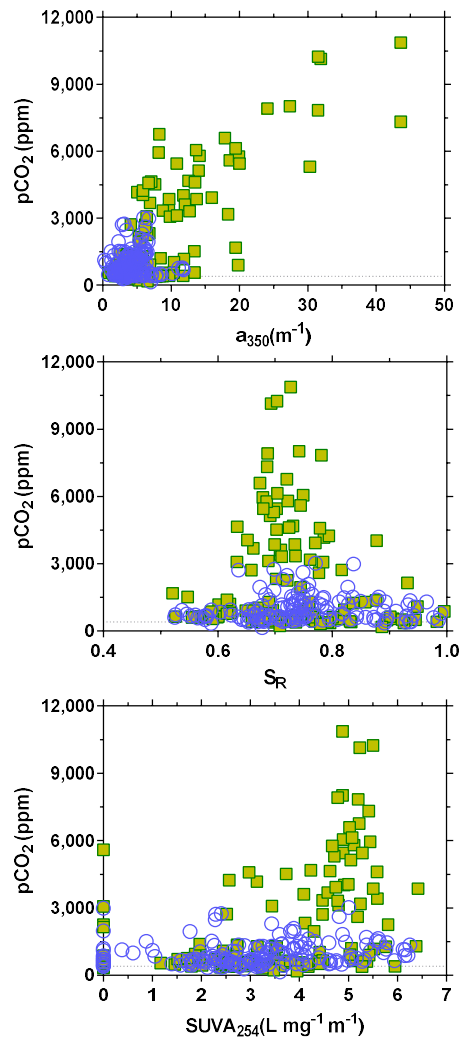
Since the transition between highland grasslands to lowland forest along the elevational gradient controlled both the carbon stable isotope ratios of DOC (Fig. 3.16) and the optical properties of DOM (Fig. 3.10),  $a_{350}$  and  $\text{SUVA}_{254}$  were negatively related to  $\delta^{13}\text{C}$ -DOC (Fig. 3.17). However, there was no relation between  $\delta^{13}\text{C}$ -DOC and  $S_R$  which might be indicative of a more uniform distribution of molecular weight of fluvial DOM despite different plant sources, compared to strong gradients in

**Figure 3.12:** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm), dissolved CH<sub>4</sub> concentration (nmol L<sup>-1</sup>), and N<sub>2</sub>O saturation level (%N<sub>2</sub>O in %) versus dissolved (DOC) and particulated (POC) organic carbon (mg L<sup>-1</sup>) in the Napo streams.

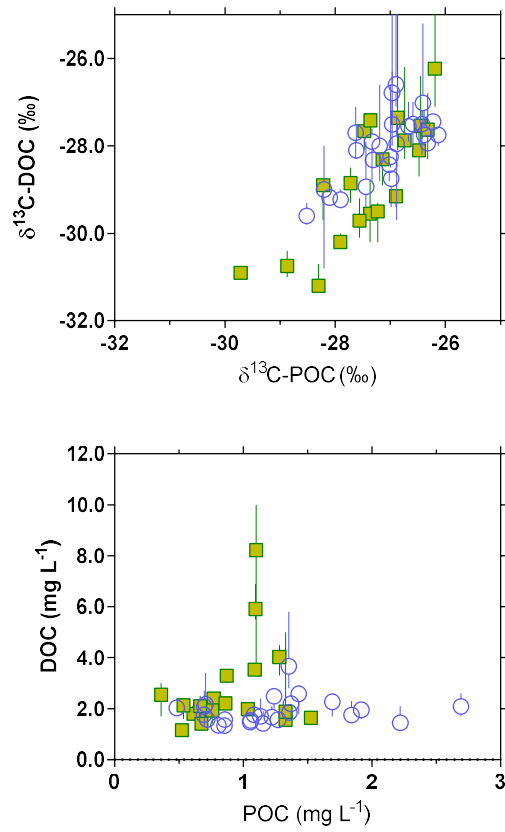




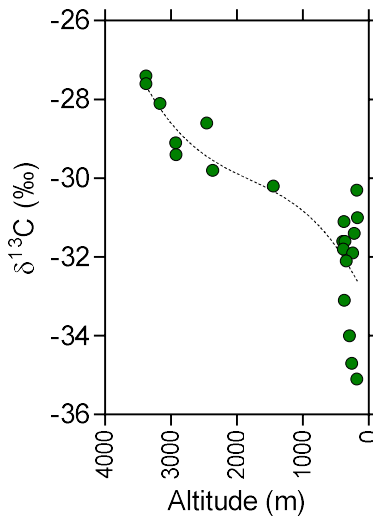
**Figure 3.13:** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm) versus absorption coefficient at 350 nm ( $a_{350}$  in m<sup>-1</sup>), slope ratio ( $S_R$  unitless), and the specific ultraviolet absorbance at 254 nm (SUVA<sub>254</sub> in L mg<sup>-1</sup> m<sup>-1</sup>) of colored dissolved organic matter in the Napo streams.



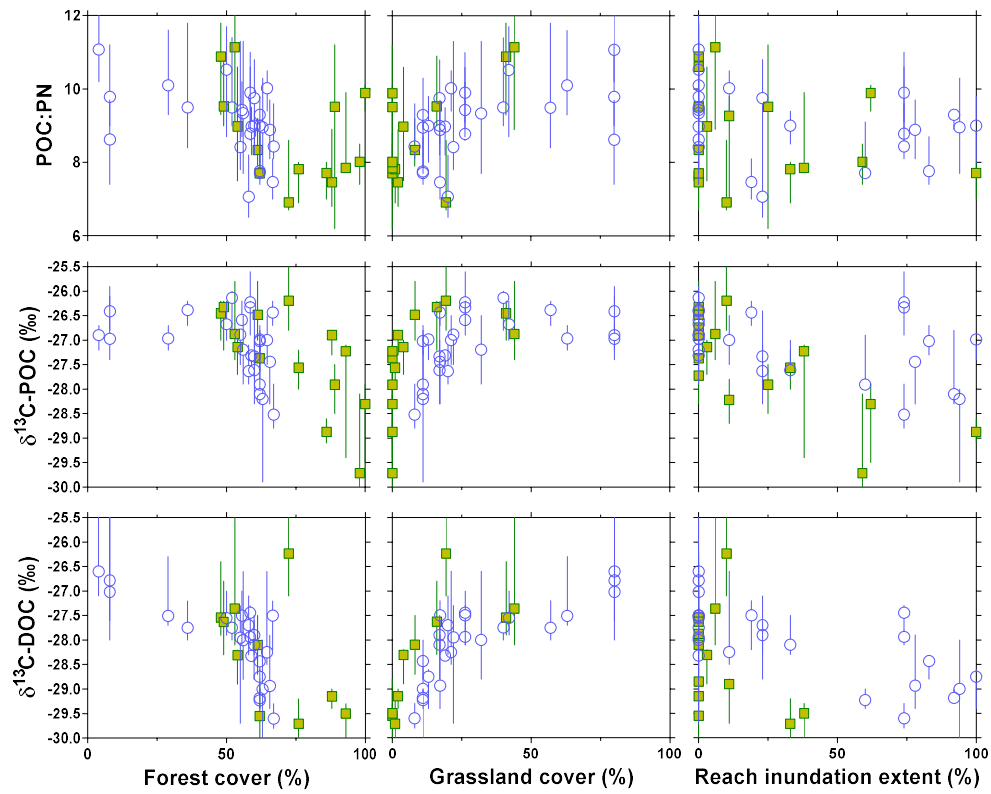
**Figure 3.14:** Carbon stable isotope ratios of dissolved organic carbon (DOC) ( $\delta^{13}\text{C}$ -DOC in ‰) versus carbon stable isotoped ratios of particulate organic carbon (POC) ( $\delta^{13}\text{C}$ -POC in ‰), and DOC versus POC in the Napo streams.



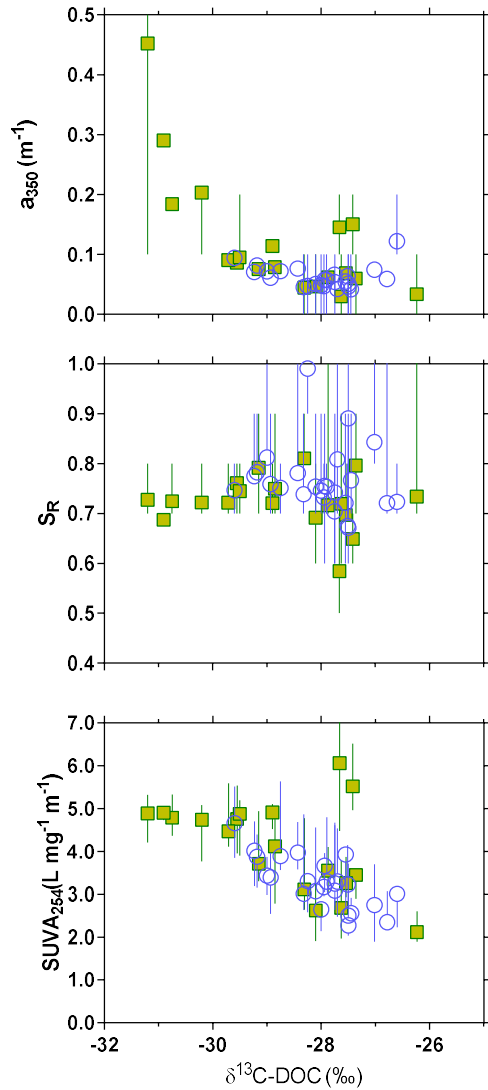
**Figure 3.15:** Carbon stable isotope ratios of plant material ( $\delta^{13}\text{C}$ -plant in ‰) versus elevation (m) from vegetation collected along the Napo streams.



**Figure 3.16:** Particulate organic carbon (POC) to particulate nitrogen (N) ratio (C:N molar), carbon stable isotope ratios of POC ( $\delta^{13}\text{C}$ -POC in ‰), carbon stable isotope ratios of dissolved organic carbon ( $\delta^{13}\text{C}$ -DOC in ‰) versus forest cover on the upstream catchment (%), grassland cover on the upstream catchment (%), and inundation extent at reach level from RiverATLAS (Linke et al. 2019) in the Napo streams.



**Figure 3.17:** Absorption coefficient at 350 nm ( $a_{350}$  in  $m^{-1}$ ), slope ratio ( $S_R$  unitless), and the specific ultraviolet absorbance at 254 nm ( $SUVA_{254}$  in  $L\ mg^{-1}\ m^{-1}$ ) of colored dissolved organic matter *versus* carbon stable isotope ratios of dissolved organic carbon ( $\delta^{13}C$ -DOC in ‰) in the Napo streams.



# CHAPTER 4

## DISSOLVED GREENHOUSE GAS (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) EMISSIONS FROM HIGHLAND LAKES OF THE ANDES CORDILLERA IN NORTHERN ECUADOR



Loreto Lake in Salve Faccha Moorland

## **Dissolved greenhouse gas (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) emissions from highland lakes of the Andes cordillera in Northern Ecuador.**

*This chapter is a submitted manuscript to the Journal Science of the Total Environment, which is in the peer-review stage, presenting a description of the emissions from highland lakes of the Andes cordillera in Northern Ecuador*

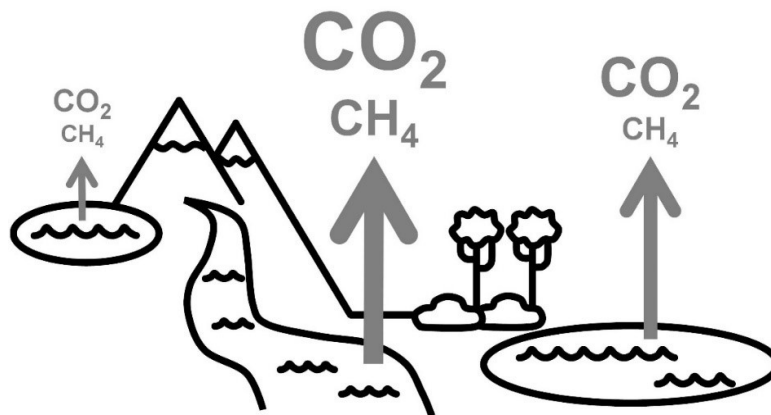
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- CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O measurements in 15 highland lakes in Andes cordillera (northern Ecuador)
- CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O correlated with lake surface area
- Three volcanic influenced lakes had much higher CO<sub>2</sub> values
- CO<sub>2</sub> in sampled highland lakes was lower than the values usually attributed to tropical lakes
- CH<sub>4</sub> in highland lakes was higher than in lowland lakes of the Amazon basin
- CO<sub>2</sub> in highland lakes was lower than in lowland lakes of the Amazon basin

## **Abstract**

We report dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in 15 lakes located in the páramos of Salve Faccha and Antisana y Mojanda (northern region of Ecuadorian Andes), along an elevational gradient from 2,213 to 4,361 m above sea level, as well as a gradient of lake surface area (0.003 to 6.1 km<sup>2</sup>) and depth (0.9 to 74 m). Most lakes were characterized by lower values of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (644-2,152 ppm) than usually attributed to tropical lakes (~1,900 ppm). Three lakes influenced by volcanic inputs were characterized by high pCO<sub>2</sub> values (3,269-10,069 ppm), while two lakes bordered by large cities were characterized by the lowest pCO<sub>2</sub> values (208-254 ppm). Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol L<sup>-1</sup> and were negatively correlated to lake area and depth. N<sub>2</sub>O saturation levels ranged between 64% and 101%. The surface waters were under-saturated in N<sub>2</sub>O with respect to atmospheric equilibrium, probably due to inputs soil-water with low N<sub>2</sub>O levels owing to denitrification in soils. The data obtained in the sampled highland lakes was combined with published data from lowland lakes to derive relations between CO<sub>2</sub> and CH<sub>4</sub> and lake surface area allowing to compute CO<sub>2</sub> and CH<sub>4</sub> lacustrine emissions at the scale of the Amazon basin using the HydroLAKES spatial data-set. The CO<sub>2</sub> and CH<sub>4</sub> emissions from highland lakes (elevation > 500 m) only represented 0.4% and 5.2% of the total lacustrine emissions at the scale of the Amazon basin, respectively. Total CO<sub>2</sub> and CH<sub>4</sub> emissions from lakes represented a small fraction (8%) of total lentic and lotic CO<sub>2</sub> and CH<sub>4</sub> emissions at the scale of the Amazon basin.

## Introduction

Inland waters (streams, rivers, lakes, and reservoirs) occupy a very small fraction of the Earth's surface but are quantitatively important global sources of CO<sub>2</sub> and CH<sub>4</sub> to atmosphere (Cole et al., 2007; Bastviken et al. 2011; Raymond et al., 2013; Rosentreter et al. 2021). Riverine CO<sub>2</sub> emissions from the tropics are disproportionately intense compared to their higher latitude counterparts (Raymond et al. 2013; Lauerwald et al. 2015; Borges et al. 2015a; 2015b; Liu et al. 2022; Battin et al. 2023). In contrast, the CO<sub>2</sub> emissions from tropical lakes might be lower than their counterparts at higher latitudes, and have been substantially over-estimated in past literature (Borges et al. 2022). The CH<sub>4</sub> emissions from tropical lakes and rivers are disproportionately intense compared to their higher latitude counterparts (Bastviken et al. 2011; Borges et al. 2015a; 2022; Rosentreter et al. 2020). Relatively higher riverine CO<sub>2</sub> and CH<sub>4</sub> emissions in tropical zones than at higher latitude are related to higher temperature and terrestrial biomass (Lauerwald et al. 2015) as well as a strong connectivity with floodplains (Abril et al. 2014; Borges et al. 2015a; 2015b; 2019). Relatively lower lacustrine CO<sub>2</sub> but higher lacustrine CH<sub>4</sub> emissions in tropical climates than at higher latitude are partly related to high phytoplankton production in non-humic lakes that leads to the uptake of dissolved CO<sub>2</sub> but at the same time high CH<sub>4</sub> production in sediments (Borges et al. 2022) due to the deposition of fresh phytoplankton detritus (Grasset et al. 2018).

Estimates of N<sub>2</sub>O emissions from inland waters have been revised downwards by a factor of 70 (Hu et al. 2016) from early estimates (Kroeze et al. 2010). Emissions of N<sub>2</sub>O from tropical lakes are lower compared to their boreal counterparts (Borges et al. 2022; 2023). Riverine N<sub>2</sub>O emissions modelled from emission factors and nitrogen deposition rates are higher in tropical rivers than rivers at higher latitudes (Hu et al. 2016; Maavara et al. 2019). However, field measurements of riverine N<sub>2</sub>O concentrations in tropical rivers are generally low and frequently below saturation (Teodoru et al. 2015; Borges et al. 2015a; 2019), except in areas impacted by urban effluents (Marwick et al. 2014). The riverine N<sub>2</sub>O concentrations are much lower in large tropical rivers such as the Congo and Zambezi than in agriculturally impacted large European rivers such as the Meuse (Borges et al. 2018). This difference between modelled and *in situ* measured N<sub>2</sub>O riverine fluxes probably reflects the uncertainty of emission factors (Clough et al. 2006) and the fact that the models do not represent the full reduction of N<sub>2</sub>O to N<sub>2</sub> by denitrification and cannot compute an uptake of N<sub>2</sub>O from inland waters (Borges et al. 2022).

Existing studies on CO<sub>2</sub> and CH<sub>4</sub> emission from tropical flowing waters have mainly focused on lowland areas, overlooking the contribution of mountainous areas. A vast amount of headwaters can be found in the Andes which contribute with the 82% of sediments and solutes to Amazon basin (McClain and Naiman 2008). Páramos are Andean mountain ecosystems which are located discontinuously in the Neotropics, at elevations between 3,000 up to 4,300 m mainly from Venezuela to northern Peru. The páramos occupy more than 30,000 km<sup>2</sup> in South America. Ecuador is the country with the highest proportion of its territory covered by páramos, with 7%. Páramo lakes constitute a primary source of drinking water for large populations in Ecuador, Colombia, and Venezuela. In Ecuador, the air temperature is relatively constant, the average annual rainfall is 1,345 mm and it occurs throughout most of the year only ~12 % of the days are dry. These conditions have favored the formation of peat and a large number of wetlands and lakes. The páramo usually presents deep and organic rich soils (30 % by volume) and its vegetation is dominated by grasslands (Zapata et al. 2021).

Here, we report the dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O in 15 lakes located in the northern region of Ecuadorian Andes along an elevational gradient from 2,213 to 4,361 m above sea level, as well as a gradient of lake surface area (0.003 to 6.1 km<sup>2</sup>) and depth (0.9 to 74 m) (Fig 4. 1). Most lakes were located in the páramos of Salve Faccha and Antisana y Mojanda.



## Material and methods

### Sampling and laboratory analysis

Sampling was carried out over a period from April 2019 to March 2022, with an inflatable boat approximately in the center of the lake, during day-time only (early morning to late afternoon). Water temperature, specific conductivity, pH, and %O<sub>2</sub> were measured in surface water with a YSI multi-parameter probe (ProPlus). Water for CH<sub>4</sub> and N<sub>2</sub>O samples was collected with a sampling devise consisting of a 2L polyethylene bottle with the bottom cut and fitted with a silicone tubing at the stopper (Abril et al. 2007). Two borosilicate serum bottles (Weathon) with a volume of 40 ml were filled with the silicone tubing, poisoned with 100 µl of a saturated solution of HgCl<sub>2</sub> and sealed with a butyl stopper and crimped with an aluminium cap. Measurements were made, after over-night equilibration, on an headspace (Weiss 1981) (created by injecting 15 ml of high-purity N<sub>2</sub> into the 40 ml sample bottles), with a gas chromatograph (SRI 8610C) with a flame ionisation detector for CH<sub>4</sub> and electron capture detector for N<sub>2</sub>O calibrated with CH<sub>4</sub>:N<sub>2</sub>O:N<sub>2</sub> gas mixtures (Air Liquide Belgium) with mixing ratios of 1, 10 and 30 ppm for CH<sub>4</sub>, and 0.2, 2.0 and 6.0 ppm for N<sub>2</sub>O. The precision of measurement based on duplicate samples was ±10.9% for CH<sub>4</sub> and ±5.8% for N<sub>2</sub>O.

The partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) was measured in the field with a Li-Cor Li-820 infra-red gas analyser based on the headspace technique with four 60 ml polypropylene syringes that were filled directly with surface water. The pCO<sub>2</sub> in the atmosphere was measured by injecting ambient air sampled with an additional polypropylene syringe. The Li-Cor Li-820 was calibrated with pure N<sub>2</sub> and CO<sub>2</sub>:N<sub>2</sub> gas mixtures (Air Liquide Belgium) of 388, 804, 3,707 and 8,146 ppm. The final pCO<sub>2</sub> value was computed taking into account the partitioning of CO<sub>2</sub> between water and the headspace, as well as equilibrium with HCO<sub>3</sub><sup>-</sup> (Dickson et al. 2007) using water temperature measured in-situ and after equilibration, and total alkalinity (TA). The precision of pCO<sub>2</sub> measurement was ±5.2%.

The CO<sub>2</sub> concentration is expressed as partial pressure in parts per million (ppm) and as dissolved concentration for CH<sub>4</sub> (nmol L<sup>-1</sup>), in accordance with convention in existing topical literature. Variations of N<sub>2</sub>O were modest and concentrations fluctuated around atmospheric equilibrium, so data are presented as percent of saturation level (%N<sub>2</sub>O, where atmospheric equilibrium corresponds to 100%), computed from the global mean N<sub>2</sub>O air mixing ratios given by the Global Monitoring Division (GMD) of the Earth System Research Laboratory (ESRL) of the National Oceanic and Atmospheric Administration (NOAA) (<https://www.esrl.noaa.gov/gmd/hats/combined/N2O.html>), using the Henry's constant (Weiss and Price 1980).

Samples for the stable isotope composition of DIC (δ<sup>13</sup>C-DIC) were collected in 12 ml Exetainer vials (Labco) and poisoned with 50 µL of a saturated solution of HgCl<sub>2</sub>. Prior to the analysis of δ<sup>13</sup>C-DIC, a 2 ml helium headspace was created and 100 µL of phosphoric acid (H<sub>3</sub>PO<sub>4</sub>, 99%) was added in the vial in order to convert CO<sub>3</sub><sup>2-</sup> and HCO<sub>3</sub><sup>-</sup> to CO<sub>2</sub>. After overnight equilibration, up to 1 mL of the headspace was injected with a gastight syringe into a coupled elemental analyser - IRMS (EA-IRMS, Thermo FlashHT or Carlo Erba EA1110 with DeltaV Advantage). The obtained data were corrected for isotopic equilibration between dissolved and gaseous CO<sub>2</sub> as described by Gillikin and Bouillon (2007). Calibration was performed with certified standards (NBS-19 or IAEA-CO<sup>-1</sup>, and LSVEC). Reproducibility of measurement based on duplicate injections of samples was typically better than ±0.2 ‰.

Water was collected in surface water with a 2L polyethylene bottle. The water filtered through 47 mm diameter GF/F Whatman glass fiber filters was collected and further filtered through polyethersulfone syringe encapsulated filters (0.2 µm porosity) for nitrate (NO<sub>3</sub><sup>-</sup>), nitrite (NO<sub>2</sub><sup>-</sup>), ammonium (NH<sub>4</sub><sup>+</sup>), TA, major elements (Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, K<sup>+</sup>), as well as dissolved silicate (DSi) and Fe, stable isotope composition of O and H of H<sub>2</sub>O (δ<sup>18</sup>O-H<sub>2</sub>O and δ<sup>2</sup>H-H<sub>2</sub>O) and dissolved organic carbon (DOC). An additional water filtration was made on 25 mm diameter GF/F Whatman glass fiber filters for particulate organic carbon (POC) analysis.

Samples for  $\text{NO}_3^-$ ,  $\text{NO}_2^-$ , and  $\text{NH}_4^+$  were stored frozen ( $-20^\circ\text{C}$ ) in 50 ml polypropylene vials.  $\text{NO}_3^-$  and  $\text{NO}_2^-$  were determined with the sulfanilamide colorimetric with the vanadium reduction method (American Public Health Association, 1998), and  $\text{NH}_4^+$  with the dichloroisocyanurate-salicylate-nitroprussiate colorimetric method (Standing committee of Analysts, 1981). Detection limits were 0.3, 0.01, and  $0.15 \mu\text{mol L}^{-1}$  for  $\text{NH}_4^+$ ,  $\text{NO}_2^-$  and  $\text{NO}_3^-$ , respectively. Precisions were  $\pm 0.02 \mu\text{mol L}^{-1}$ ,  $\pm 0.02 \mu\text{mol L}^{-1}$ , and  $\pm 0.1 \mu\text{mol L}^{-1}$  for  $\text{NH}_4^+$ ,  $\text{NO}_2^-$  and  $\text{NO}_3^-$ , respectively.

Samples for TA were stored at ambient temperature in polyethylene 55 ml vials and measurements were carried out by open-cell titration with  $\text{HCl } 0.1 \text{ mol L}^{-1}$  according to Gran (1952), and data quality checked with certified reference material obtained from Andrew Dickson (Scripps Institution of Oceanography, University of California, San Diego, USA), with a typical reproducibility better than  $\pm 3 \mu\text{mol kg}^{-1}$ . DIC was computed from TA and  $\text{pCO}_2$  measurements using the carbonic acid dissociation constants for freshwater of Millero (1979) using the  $\text{CO}_2\text{sys}$  package.

Samples for  $\delta^{18}\text{O-H}_2\text{O}$  and  $\delta^2\text{H-H}_2\text{O}$  were stored at ambient temperature in polypropylene 8 ml vials.  $\delta^2\text{H-H}_2\text{O}$  was measured on  $\text{H}_2$  gas derived from a high-temperature ( $1,030^\circ\text{C}$ ) Cr-based reactor by automated injections of water using a TriPlus autosampler on an elemental analyzer (Thermo Flash HT/EA; Thermo Finnigan) coupled to a continuous-flow isotope-ratio mass spectrometer (Delta V Advantage; Thermo Finnigan).  $\delta^{18}\text{O-H}_2\text{O}$  values were measured on a Thermo GasBench II coupled to a Thermo Delta XP IRMS after equilibration with  $\text{CO}_2$ . The long-term uncertainty for standard  $\delta^{18}\text{O}$  values was  $\pm 0.1\%$ .

Samples for major elements were stored at ambient temperature in 20 ml scintillation vials and preserved with  $50 \mu\text{l}$  of  $\text{HNO}_3$  (65%). Major elements were measured with inductively coupled plasma MS (ICP-MS; Agilent 7700x) calibrated with the following standards: SRM1640a from National Institute of Standards and Technology, TM-27.3 (lot 0412) and TMRain-04 (lot 0913) from Environment Canada, and SPS-SW2 Batch 130 from Spectrapure Standard. Limit of quantification was  $0.5 \mu\text{mol L}^{-1}$  for  $\text{Na}^+$ ,  $\text{Mg}^{2+}$  and  $\text{Ca}^{2+}$ ,  $1.0 \mu\text{mol L}^{-1}$  for  $\text{K}^+$  and  $8 \mu\text{mol L}^{-1}$  for DSi.

Samples to determine DOC were stored at ambient temperature and in the dark in 40 ml brown borosilicate vials with polytetrafluoroethylene (PTFE) coated septa and poisoned with  $50 \mu\text{l}$  of  $\text{H}_3\text{PO}_4$  (85%), and DOC concentration was determined with a wet oxidation total organic carbon analyzer (IO Analytical Aurora 1030W), with a typical reproducibility better than  $\pm 5\%$ .

Filters for POC analysis were decarbonated with  $\text{HCl}$  fumes for 4h and dried before encapsulation into silver cups; POC concentration was analysed on an EA-IRMS (Thermo FlashHT with DeltaV Advantage), with a reproducibility better than  $\pm 5\%$ . Data were calibrated with certified (IAEA-600: caffeine) and in-house standards (leucine, tuna muscle tissue) that were previously calibrated versus certified standards.

### **Flux computations and upscaling**

The dissolved concentrations of  $\text{CO}_2$  and  $\text{CH}_4$  were derived from statistical relations as function of lake surface area for each of the lakes ( $n=12,549$ ) in HydroLAKES (Messenger et al. 2016). We excluded from the analysis reservoirs that represent 0.06% of lake counts ( $n=8$ ) and 11% of total lacustrine surface area ( $2,773 \text{ km}^2$ ) in the HydroLAKES data-set. The statistical relations were derived from our own data in Ecuadorian highland lakes and data available from literature in the floodplain lakes of the Central Amazon (Abril et al. 2013; Amaral et al. 2020; Barbosa et al. 2016; Mitchell et al. 2018).

The  $\text{pCO}_2$  (in ppm) was computed from two linear regressions as a function of lake surface area (LSA in  $\text{km}^2$ ) according to:

$$\log(\text{pCO}_2) = -0.1293(\pm 0.0346) \times \text{LSA} + 3.063(\pm 0.073)$$

$$(r^2=0.58; p=0.0039; n=12)$$

for an elevation  $> 500 \text{ m}$  (highlands) excluding three lakes (Cuicocha, Loreto and Papallacta)

and

$$\log(p\text{CO}_2) = -0.0003745(\pm 0.0001642) \times \text{LSA} + 3.508(\pm 0.061)$$

( $r^2=0.26$ ;  $p=0.0376$ ;  $n=17$ )  
for an elevation < 500 m (lowlands)

The CH<sub>4</sub> (in nmol L<sup>-1</sup>) was computed from a linear regression as function of LSA (in km<sup>2</sup>) according to:

$$\log(\text{CH}_4) = -0.3753(\pm 0.0426) \times \log(\text{LSA}) + 2.545(\pm 0.066)$$

( $r^2=0.71$ ;  $p<0.0001$ ;  $n=34$ )  
for both highland and lowland lakes

The water temperature (WT in °C) was computed from the air temperature (AT in x10°C) from HydroLAKES, according to:

$$\log(\text{WT}) = 0.002192(\pm 0.000069) \times \text{AT} + 0.8905(\pm 0.0151)$$

( $r^2=0.98$ ;  $p<0.0001$ ;  $n=23$ )

The flux of CO<sub>2</sub> (FCO<sub>2</sub>) and CH<sub>4</sub> (FCH<sub>4</sub>) between surface waters and the atmosphere was computed according to:

$$F = k \times \Delta G$$

where  $k$  is the gas transfer velocity and  $\Delta G$  is the air-water gradient of a given gas.

The  $k$  normalized to a Schmidt number ( $Sc$ ) of 600 ( $k_{600}$  in cm h<sup>-1</sup>) was derived from the parameterisation as a function wind speed (Cole and Caraco 1998):

$$k_{600} = 2.07 + 0.215 \times u^{1.7}$$

where  $u$  is wind speed (m s<sup>-1</sup>) that was extracted from WorldClim (Fick and Hijmans 2017) for each lake in HydroLAKES.

$k_{600}$  was converted to  $k$  for the  $Sc$  at *in-situ* temperature (Wanninkhof 1992) according to:

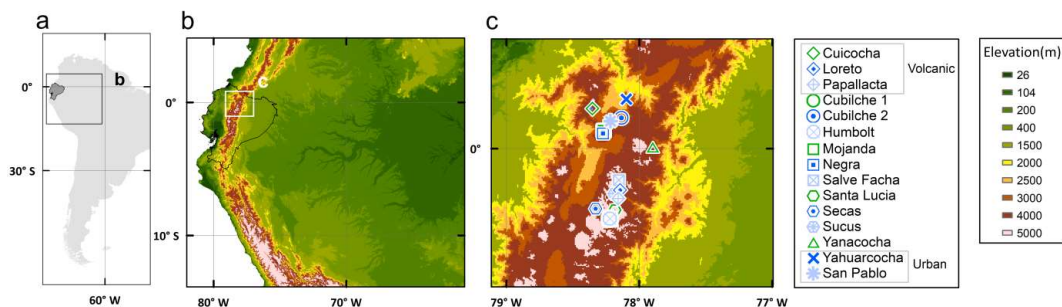
$$k = k_{600} \times (Sc/600)^{-0.5}$$

A Monte-Carlo error analysis was made by propagating errors (10,000 iterations) on the slope and constant of the equations to the gas concentrations and WT (refer to above equations) and an error of ±30% for  $k_{600}$  based on the reported variability across different lakes of  $k$  at low wind speeds (Cole and Caraco 1998).

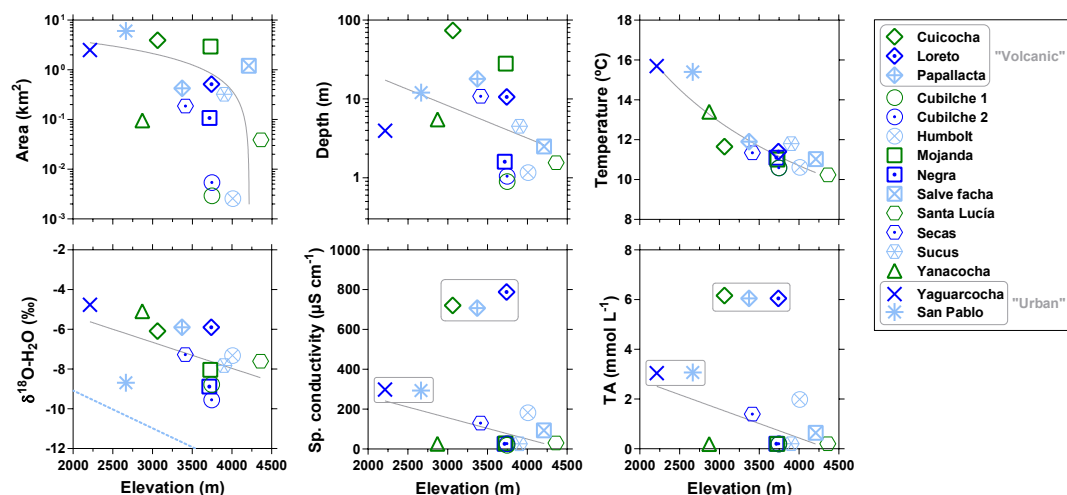
## Results and discussion

We sampled 15 lakes in Ecuador along an elevational gradient (2,213–4,361 m) (Fig 4. 1). The surface area of the sampled lakes ranged between 0.003 to 6.1 km<sup>2</sup>, and lakes at higher elevation tended to be smaller (Fig 4. 2). Lakes were generally shallow, with depths ranging between 0.9 and 72 m, with a tendency towards shallower lakes at higher elevation (Fig 4. 2). Lakes Cuicocha, Loreto and Papallacta are crater lakes (hereafter referred to as “volcanic”) and the others are of glacial origin. Among the latter, San Pablo and Yaguarcocha are surrounded by the cities of Otavalo and Ibarra, respectively (hereafter referred to as “urban”).

**Figure 4.1:** Fifteen sampled highland lakes in Northern Ecuador



**Figure 4.2:** Lake area (km<sup>2</sup>), lake depth (m), water temperature (°C), stable isotope composition of O of H<sub>2</sub>O ( $\delta^{18}\text{O}\text{-H}_2\text{O}$  in ‰), specific (Sp.) conductivity ( $\mu\text{S cm}^{-1}$ ) and total alkalinity (TA in  $\text{mmol L}^{-1}$ ) in surface waters *versus* elevation (m). Solid grey lines show fits to the data (Table 4.S2). Dotted blue line indicates the  $\delta^{18}\text{O}\text{-H}_2\text{O}$  elevation gradient for Ecuador based on the global rainwater iso-scape given by Inguaggiato et al. (2010).



Water temperature ranged between 10.2 and 15.7°C and followed a clear elevational gradient (Fig 4. 2). The specific conductivity in the sampled lakes was generally low (20-183  $\mu\text{S cm}^{-1}$ ), except for the two “urban” lakes (San Pablo and Yaguarcocha) (294-298  $\mu\text{S cm}^{-1}$ ) and the three “volcanic” lakes (Cuicocha, Loreto and Papallacta) (708-722  $\mu\text{S cm}^{-1}$ ) (Fig 4. 2). The conductivity of the three “volcanic” lakes was lower than in crater lakes in active volcanic systems in Ecuador characterized by conductivity values  $>10,000 \mu\text{S cm}^{-1}$  (Inguaggiato et al. 2010; Melián et al. 2021). Conductivity was negatively related to elevation for the non-“volcanic” lakes possibly reflecting the combination of decreasing lake size and increasing soil organic matter content with elevation. In the Andean páramo, the soil organic carbon increases with elevation because of the decrease of temperature that leads to a decrease of organic matter mineralization in soils enhancing organic matter accumulation in soils (Poulenard et al. 2003). The high organic matter content of soils and lower temperature are not favorable to rock weathering leading to soil water with low mineral content (low conductivity). Additionally, high elevation páramo areas are characterized by high precipitation leading to dilution of solutes in surface waters which should additionally contribute to decrease conductivity. TA showed very similar patterns to those of conductivity, with the three “volcanic” lakes characterized by higher TA values (6.05-6.16  $\text{mmol L}^{-1}$ ), followed by the two “urban” lakes (3.04-3.07  $\text{mmol L}^{-1}$ ), and the remaining lakes with the lowest TA values (0.19-1.99  $\text{mmol L}^{-1}$ ). The cation sequence was dominated by  $\text{Ca}^{2+}$  or  $\text{Na}^+$  followed by  $\text{Mg}^{2+}$  and  $\text{K}^+$  (Table 4.S1). The lakes with a high total cation content (21-147  $\text{mg L}^{-1}$

and 1,035-7,916  $\mu\text{eq L}^{-1}$ ) were dominated by  $\text{Na}^+$  ( $\text{Na}^+ > \text{Ca}^{2+} > \text{Mg}^{2+} > \text{K}^+$ ) characteristic of sodium bicarbonate lakes. Lakes with a lower total cation content ( $\sim 5 \text{ mg L}^{-1}$  and  $\sim 235 \mu\text{eq L}^{-1}$ ) were dominated by  $\text{Ca}^{2+}$  ( $\text{Ca}^{2+} > \text{Na}^+ > \text{Mg}^{2+} > \text{K}^+$ ) (Table 4.S1). This might indicate that the cation content of the sample lakes was related to the weathering of primarily volcanic rocks that dominate in the area (rhyolite, andesite, dacite, and diorite) (Mosquera et al. 2022). The three “volcanic” lakes (Cuicocha, Loreto and Papallacta) were characterized by higher total cation content ( $144\text{-}147 \text{ mg L}^{-1}$   $7,778\text{-}7,916 \mu\text{eq L}^{-1}$ ) than the other 12 lakes (Table 4.S1). The composition of the three “volcanic” lakes in divalent ions ( $\text{Ca}^{2+}$  and  $\text{Mg}^{2+}$ ) relative to  $\text{Na}^+$  was consistent with the other 12 lakes (Fig 4. S1). However, the three “volcanic” lakes had lower  $\text{K}^+$ , DSi and TA relative to the other 12 lakes (Fig 4. S1), which might indicate different weathering mechanism in the three “volcanic” lakes compared to the other 12 lakes: either a non-congruent dissolution of aluminosilicate minerals or alternatively the weathering of a higher portion of sodium rich volcanic rocks in the three “volcanic” lakes compared to the other 12 lakes (Mosquera et al. 2022). The Fe content was negatively related to  $\text{Na}^+$ , and the three “volcanic” lakes had lower Fe relative to  $\text{Na}^+$  compared to the other 12 lakes. This pattern might be related to the chelation of Fe to dissolved organic matter, as Fe was correlated to DOC (Fig 4. S2).

**Table 4.1:** Emissions to the atmosphere of  $\text{CO}_2$  ( $F_{\text{CO}_2}$ ) and  $\text{CH}_4$  ( $F_{\text{CH}_4}$ ) from the lakes and rivers of the Amazon basin. The  $F_{\text{CO}_2}$  and  $F_{\text{CH}_4}$  in lakes was modelled from relations of  $\text{CO}_2$  and  $\text{CH}_4$  dissolved concentration as a function of lake surface area based on our own data in Ecuadorian highland lakes and data from literature for lowland floodplain lakes (Fig 4.7), using the HydroLAKES spatial data-set (Messenger et al. 2016), and the gas transfer velocity ( $k_{600}$ ) derived from a parameterization as a function of wind speed (Cole and Caraco 1998) and wind speed from the WorldClim spatial data-set (Fick and Hijmans 2017). The  $F_{\text{CO}_2}$  and  $F_{\text{CH}_4}$  in rivers was modelled from of  $\text{CO}_2$  and  $\text{CH}_4$  dissolved concentration as a function of catchment slope,  $k_{600}$  derived from a parameterization as a function of stream slope and flow (Raymond et al. 2012) and the RiverATLAS spatial data-set (Linke et al. 2019) as detailed by Chiriboga and Borges (2023).

|                                                                     | Elevation       |                 |
|---------------------------------------------------------------------|-----------------|-----------------|
|                                                                     | <500 m          | >500 m          |
| <b>Lakes</b>                                                        |                 |                 |
| $F_{\text{CH}_4}$ (TgC yr <sup>-1</sup> )                           | 0.0130 ± 0.0036 | 0.0007 ± 0.0002 |
| $F_{\text{CH}_4}$ (gC m <sup>-2</sup> yr <sup>-1</sup> )            | 0.58 ± 0.16     | 0.75 ± 0.21     |
| Surface area weighted average $\text{CH}_4$ (nmol L <sup>-1</sup> ) | 179             | 265             |
| $F_{\text{CO}_2}$ (TgC yr <sup>-1</sup> )                           | 5.7 ± 1.2       | 0.022 ± 0.005   |
| $F_{\text{CO}_2}$ (gC m <sup>-2</sup> yr <sup>-1</sup> )            | 255 ± 55        | 23 ± 5          |
| Surface area weighted average p $\text{CO}_2$ (ppm)                 | 2,909           | 530             |
| Wind speed (m s <sup>-1</sup> )                                     | 1.4 ± 0.6       | 3.0 ± 0.6       |
| $k_{600}$ (cm h <sup>-1</sup> )                                     | 2.5 ± 0.3       | 3.5 ± 0.5       |
| Total lake surface area (km <sup>2</sup> )                          | 22,282          | 952             |
| Number of lakes                                                     | 12,549          | 1,116           |
| Average lake surface area (km <sup>2</sup> )                        | 1.78 ± 15.54    | 0.85 ± 5.54     |
| <b>Rivers</b>                                                       |                 |                 |
| $F_{\text{CH}_4}$ (TgC yr <sup>-1</sup> )                           | 0.273 ± 0.084   | 0.030 ± 0.007   |
| $F_{\text{CH}_4}$ (gC m <sup>-2</sup> yr <sup>-1</sup> )            | 6.77 ± 2.08     | 12.93 ± 2.88    |
| Average $\text{CH}_4$ (nmol L <sup>-1</sup> )                       | 427             | 157             |
| $F_{\text{CO}_2}$ (TgC yr <sup>-1</sup> )                           | 58.9 ± 4.1      | 5.5 ± 1.0       |
| $F_{\text{CO}_2}$ (gC m <sup>-2</sup> yr <sup>-1</sup> )            | 1,459 ± 103     | 2,364 ± 418     |
| Average p $\text{CO}_2$ (ppm)                                       | 2,702           | 1,033           |
| $k_{600}$ (cm h <sup>-1</sup> )                                     | 15.8 ± 0.4      | 148.8 ± 3.2     |
| Total stream surface area (km <sup>2</sup> )                        | 40,375          | 2,338           |

The  $\delta^{18}\text{O}\text{-H}_2\text{O}$  values were negatively related to elevation and paralleled elevational gradient based on the global rain water iso-scape (Inguaggiato et al. 2010) (Fig 4. 2). This indicates that the sampled lakes were mainly recharged by local precipitation (meteoric recharge). The offset between observed  $\delta^{18}\text{O}\text{-H}_2\text{O}$  values in the sampled lakes and the precipitation elevational gradient most probably resulted from fractionation during evaporation. The observed

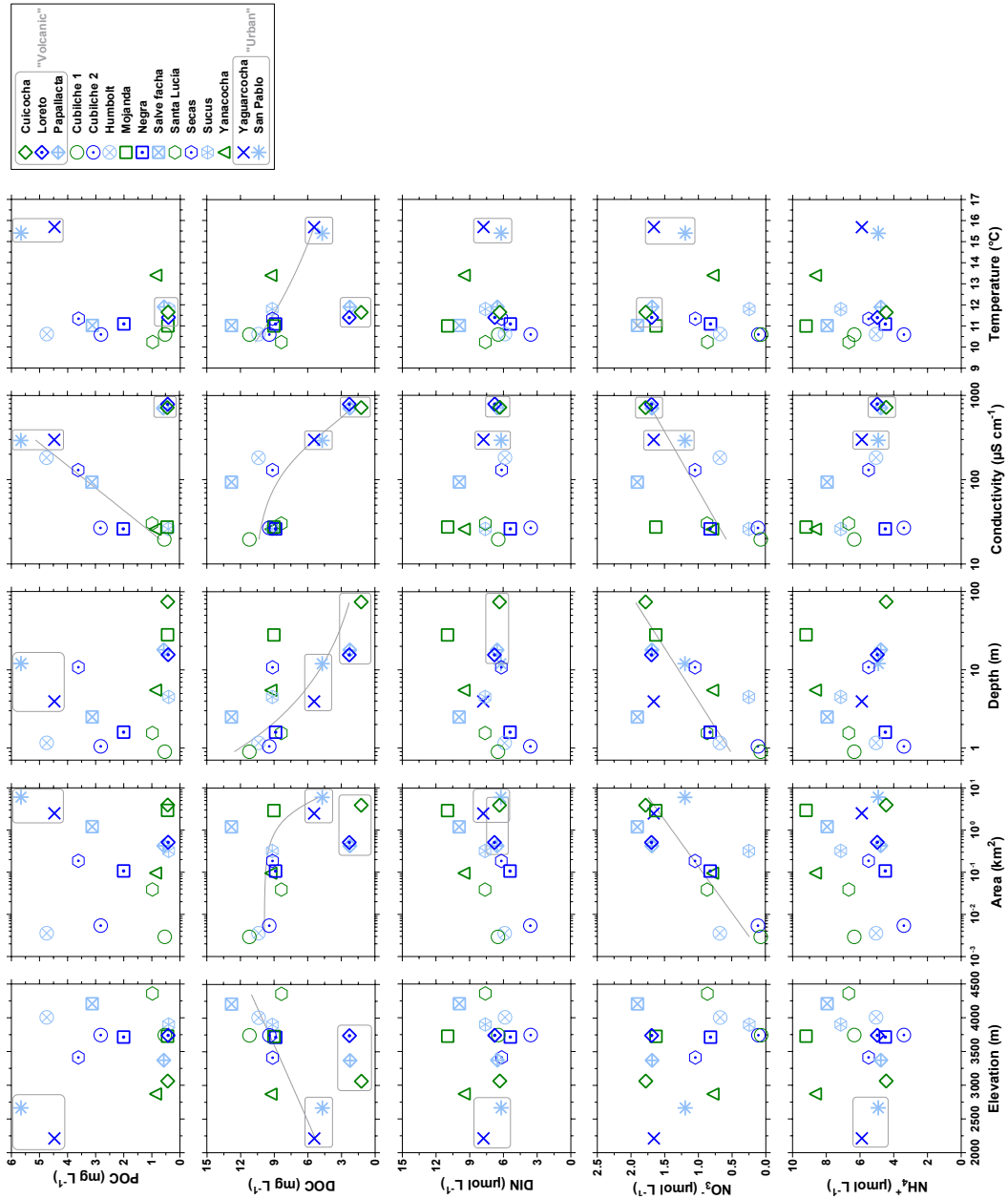
$\delta^2\text{H}$  and  $\delta^{18}\text{O}$ - $\text{H}_2\text{O}$  values in the sampled lakes deviated from local meteoric water line (LMWL) (Inguaggiato et al. 2010), confirming substantial fractionation during evaporation (Fig 4. S3).

The DOC concentrations ranged between 1.2 and 12.8  $\text{mg L}^{-1}$ , and DOC data points were grouped in clusters as a function of several variables such as elevation, lake area, lake depth, conductivity, and water temperature (Fig 4. 3). The three “volcanic” lakes (Cuicocha, Loreto and Papallacta) were characterized by the lowest DOC concentrations (1.2 to 2.3  $\text{mg L}^{-1}$ ), the two “urban” lakes (San Pablo and Yaguarcocha) were characterized by intermediate DOC concentrations (4.7 to 5.4  $\text{mg L}^{-1}$ ), and the remaining 10 lakes were characterized by the highest DOC concentrations (8.3 to 12.8  $\text{mg L}^{-1}$ ). DOC was positively correlated to elevation and negatively related to temperature (excluding the three “volcanic” lakes). This might reflect the higher inputs of soil DOC in lakes more strongly influenced by inputs from the páramos. The negative relation between DOC and conductivity is consistent with the idea that soils richer in organic matter tend to lead to less mineralized water. Low inputs of soil DOC in the three “volcanic” lakes are related to a low ratio of catchment area to lake surface area which reduces the inputs from soils to the lake. Additionally, the DOC content could be partly modulated by lake size, as indicated by the negative relation between DOC concentration and lake area (Fig 4. 3).

Small lakes have a greater connectivity and potential exchange of organic carbon with the riparian wetlands and surrounding terrestrial landscape along its periphery relative to lake surface area than larger lakes (Staehr et al. 2012). The two above explanations of DOC patterns are compatible as lake size decreases with elevation and the smaller lakes were, thus, located in the páramos with organic carbon rich soils. The POC concentrations ranged between 0.4 and 5.7  $\text{mg L}^{-1}$ , and the POC data points were less clearly clustered than DOC data points and did not correlate with elevation, lake area, lake depth, or water temperature, but did show a relationship with conductivity (Fig 4. 3). One of the two “urban” lakes (San Pablo) was characterized by the highest POC value (5.7  $\text{mg L}^{-1}$ ), while the three “volcanic” lakes were characterized by some of the lowest POC values (0.4 to 0.6  $\text{mg L}^{-1}$ ). The correlation of POC with conductivity (excluding the three “volcanic” lakes) might reflect a general dilution pattern by meteoric water of solutes and POC. Variations of DIN were modest (3.6 to 11.0  $\mu\text{mol L}^{-1}$ ) and DIN did not significantly correlate with elevation, lake area, lake depth, conductivity and water temperature. DIN was dominated by  $\text{NH}_4^+$  (70-98%) that ranged between 3.4 and 9.2  $\mu\text{mol L}^{-1}$ .  $\text{NH}_4^+$  did not significantly correlate with elevation, lake area, lake depth, conductivity, and water temperature.  $\text{NO}_3^-$  ranged between 0.1 and 1.9  $\mu\text{mol L}^{-1}$  and did not significantly correlate with elevation and temperature but correlated positively to lake area, lake depth and conductivity. These correlations might reflect a stronger impact in smaller and shallower water bodies of soil-water inputs with a low  $\text{NO}_3^-$  content related to denitrification in highland soils (Pineda et al. 2021).

The  $\text{pCO}_2$  values ranged between 208 and 10,069 ppm (Fig 4. 4). The  $\text{pCO}_2$  data points were distinctly grouped into the same three clusters observed for DOC (Fig 4. 4). The three “volcanic” lakes (Cuicocha, Loreto and Papallacta) were characterized by higher  $\text{pCO}_2$  values (3,269-10,069 ppm), the two “urban” lakes (San Pablo and Yaguarcocha) by the lowest  $\text{pCO}_2$  values (208-254 ppm), and the other páramo lakes by intermediary  $\text{pCO}_2$  values (644-2,152 ppm). All lakes had  $\text{pCO}_2$  values above atmospheric equilibrium (400 ppm) and thus acted as sources of  $\text{CO}_2$  to the atmosphere, except for the two “urban” lakes that acted as sinks of atmospheric  $\text{CO}_2$ . The highest  $\text{pCO}_2$  values in the three “volcanic” lakes seemed to be in part related to geogenic  $\text{CO}_2$  inputs, reflecting the influence of post-volcanic activities, such as emission of volcanic gases and input of hydrothermal water.

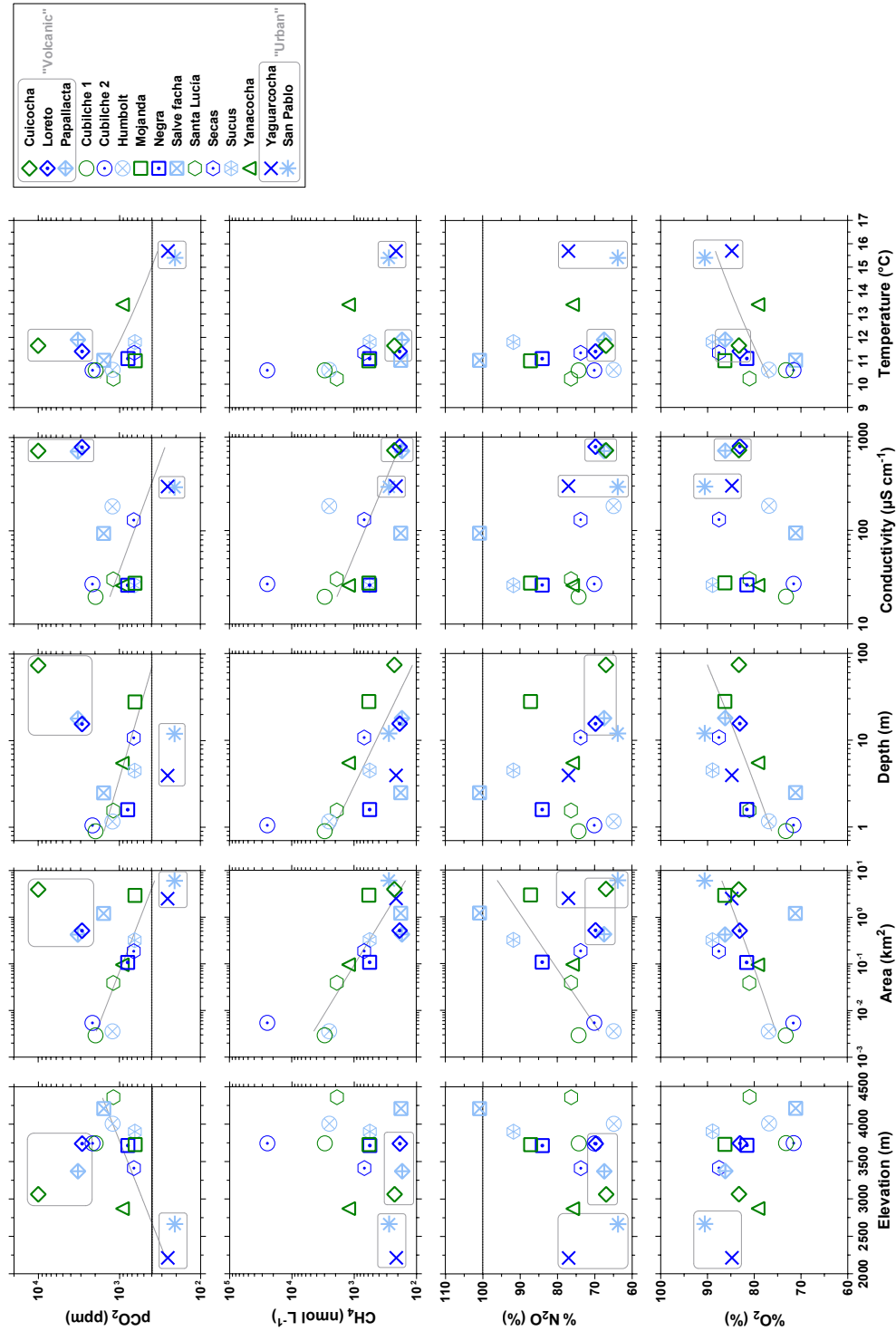
**Figure 4.3:** Particulate (POC in  $\text{mg L}^{-1}$ ) and dissolved (DOC in  $\text{mg L}^{-1}$ ) organic carbon, dissolved inorganic nitrogen (DIN in  $\mu\text{mol L}^{-1}$ ),  $\text{NO}_3^-$  ( $\mu\text{mol L}^{-1}$ ) and  $\text{NH}_4^+$  ( $\mu\text{mol L}^{-1}$ ) in surface waters versus elevation (m), lake surface area ( $\text{km}^2$ ), lake depth (m), specific conductivity ( $\mu\text{S cm}^{-1}$ ) and water temperature ( $^\circ\text{C}$ ). Solid grey lines show fits to the data (Table 4.S2).



The  $\delta^{13}\text{C}$ -DIC values in surface waters of the “volcanic” lakes were extremely high ( $\sim 3.5$  ‰) (Fig 4. S4), and magmatic inputs likely provide the dominant imprint on  $\delta^{13}\text{C}$ -DIC signatures as also reported in Lakes Kivu ( $\sim 1.0$  ‰) and Sonachi ( $\sim 9.0$  ‰) (also a crater lake) (Borges et al. 2012). The  $\delta^{13}\text{C}$  values of volcanic  $\text{CO}_2$  in gas bubbles in Ecuadorian crater lakes range between  $-2$  and  $-6$  ‰ (Inguaggiato et al. 2010; Melián et al. 2021). Consequently, the positive  $\delta^{13}\text{C}$ -DIC values observed in the three “volcanic” lakes also reflect the uptake of  $^{12}\text{C}$  by primary production and enrichment in  $^{13}\text{C}$  in the residual DIC. This effect might be enhanced by long water residence time typical of small crater lakes devoid of outlets. The two “urban” lakes were also characterized

by enriched  $\delta^{13}\text{C}$ -DIC signatures also indicative of uptake of  $^{12}\text{C}$  by primary production consistent with the low  $\text{pCO}_2$  values (and high POC values).

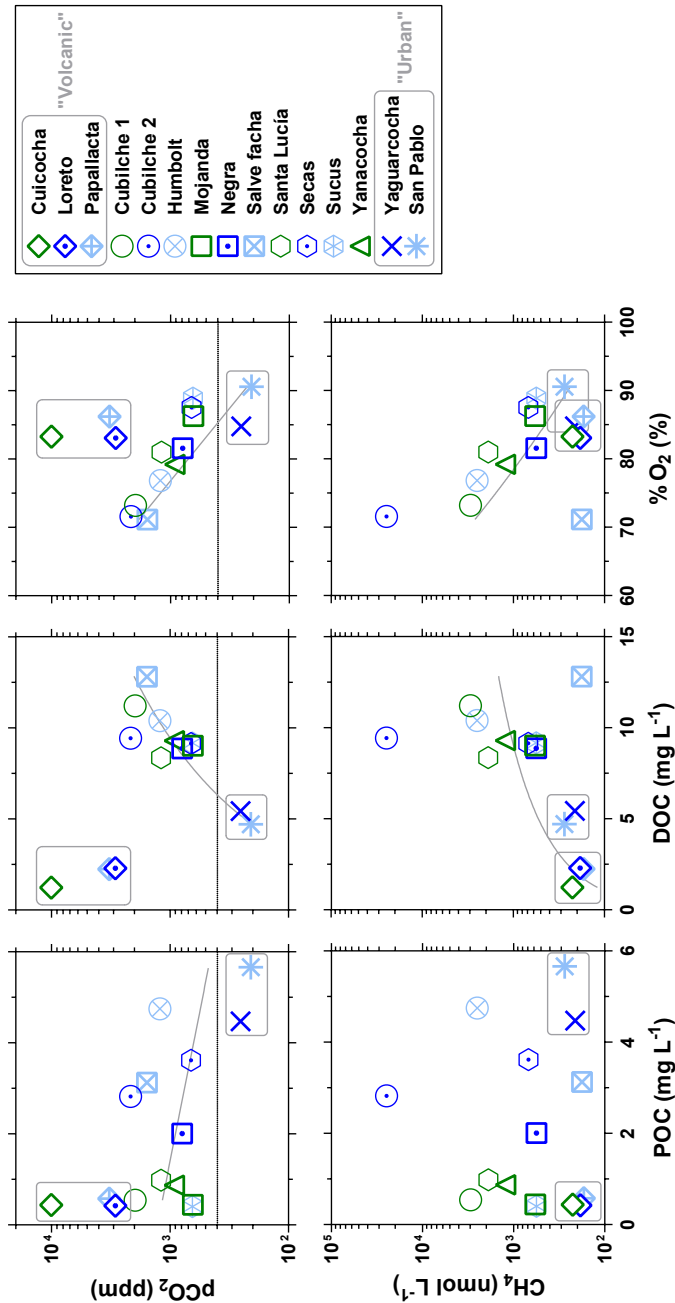
**Figure 4.4:** Partial pressure of  $\text{CO}_2$  ( $\text{pCO}_2$  in ppm), dissolved  $\text{CH}_4$  concentration ( $\text{nmol L}^{-1}$ ),  $\text{N}_2\text{O}$  saturation level ( $\%\text{N}_2\text{O}$  in %),  $\text{O}_2$  saturation level ( $\%\text{O}_2$  in %) in surface waters versus elevation (m), lake surface area ( $\text{km}^2$ ), lake depth (m), specific conductivity ( $\mu\text{S cm}^{-1}$ ) and water temperature ( $^\circ\text{C}$ ). Dotted horizontal lines indicate equilibrium of respective gases with the atmosphere and solid grey lines show fits to the data (Table 4.S2).



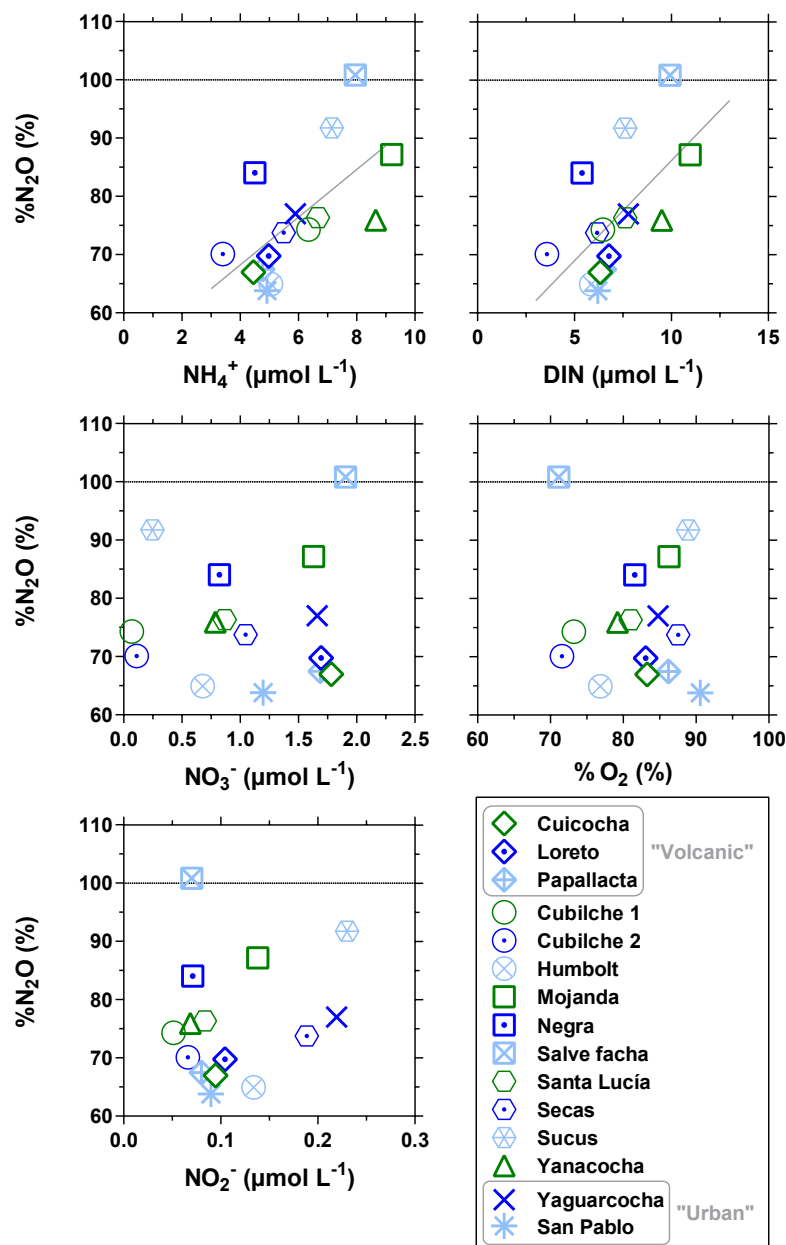


A general increase of  $\delta^{13}\text{C-DIC}$  was observed from  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}=0.5$  (half of the DIC is in the form of  $\text{CO}_2$ ) towards  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}=1$  (all of the DIC is in the form of  $\text{HCO}_3^- + \text{CO}_3^{2-}$ ). This pattern reflects a gradient of lakes from systems with low rock weathering due to dominance of deep organic soils (low  $\text{HCO}_3^-$ ) and high  $\text{CO}_2$  from respiration leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$  close to 0.5 with low  $\delta^{13}\text{C-DIC}$ , and lakes with higher rock weathering leading to  $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$  closer to 1 with high  $\delta^{13}\text{C-DIC}$ .

**Figure 4.5:** Partial pressure of  $\text{CO}_2$  ( $p\text{CO}_2$  in ppm) and dissolved  $\text{CH}_4$  concentration ( $\text{nmol L}^{-1}$ ) versus particulate (POC in  $\text{mg L}^{-1}$ ) and dissolved (DOC in  $\text{mg L}^{-1}$ ) organic carbon and  $\text{O}_2$  saturation level ( $\%\text{O}_2$  in %) in surface. Dotted horizontal lines indicate equilibrium with the atmosphere and solid grey lines show fits to the data (Table 4.S2).



**Figure 4.6:**  $\text{N}_2\text{O}$  saturation level (% $\text{N}_2\text{O}$  in %) versus  $\text{NH}_4^+$  ( $\mu\text{mol L}^{-1}$ ), dissolved inorganic nitrogen (DIN in  $\mu\text{mol L}^{-1}$ ),  $\text{NO}_3^-$  ( $\mu\text{mol L}^{-1}$ ),  $\text{NO}_2^-$  ( $\mu\text{mol L}^{-1}$ ) and  $\text{O}_2$  saturation level (% $\text{O}_2$  in %) in surface waters. Dotted horizontal lines indicate equilibrium with the atmosphere and solid grey lines show fits to the data (Table 4.S2).



The  $\text{pCO}_2$  values in the páramo lakes were positively correlated to elevation and negatively correlated to lake area, lake depth, temperature and conductivity (excluding the three "volcanic" lakes - Cuicocha, Loreto and Papallacta). This is indicative of higher lacustrine  $\text{pCO}_2$  values in páramo lakes bordered by soils with a higher organic carbon content, as shown above for DOC (Fig 4. 3). Indeed,  $\text{pCO}_2$  was positively correlated to DOC, negatively to % $\text{O}_2$  and POC (Fig 4. 5). This indicates a gradient of inputs of soil water with a low  $\text{O}_2$  and conductivity and high DOC and  $\text{pCO}_2$  among the páramo lakes that were more important at high elevations owing to

soils with a higher organic matter content. Conversely, in lakes with lower inputs of soil DOC (the “urban” lakes at lower elevation), lower DOC allowed the development of phytoplankton (lesser light limitation) contributing to the decreasing pattern of pCO<sub>2</sub> with decreasing DOC and increasing POC. Additionally, the lakes at higher elevation tended to be shallower and smaller (Fig 4. 2). As mentioned above to explain the pattern of DOC and lake size, smaller lakes have a greater connectivity and potential exchange of CO<sub>2</sub> from soils along its periphery relative to lake surface area (Staeher et al. 2012). Also, there is a stronger fetch limitation of *k* in smaller lakes (Wanninkhof 1992) that can contribute to the accumulation of CO<sub>2</sub> in lake surface water of smaller lakes, while higher *k* in larger lakes promotes gazing and brings CO<sub>2</sub> levels closer to atmospheric equilibrium. These processes combined contribute to explain the negative relation between pCO<sub>2</sub> with lake surface area as well as with lake depth (Fig 4. 4). Since the CO<sub>2</sub> dynamics were related to organic matter degradation (in soils or in-lake), %O<sub>2</sub> correlated to pCO<sub>2</sub> (Fig 4. 5) and %O<sub>2</sub> patterns mirrored those of pCO<sub>2</sub>: %O<sub>2</sub> was positively related to lake area, lake depth and water temperature but was not significantly correlated to conductivity and elevation (Fig 4. 4).

Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol L<sup>-1</sup> in the sampled lakes (Fig 4. 4). Dissolved CH<sub>4</sub> concentrations were not significantly correlated to elevation, water temperature, and POC but were negatively correlated to lake area, lake depth, conductivity, DOC and %O<sub>2</sub> (Figs. 4,5). The patterns of CH<sub>4</sub> paralleled those of pCO<sub>2</sub>. The smaller lakes bordered by soils with a higher organic carbon content were likely characterized by a higher CH<sub>4</sub> production in lake sediments or in surrounding soils.

%N<sub>2</sub>O ranged between 63.8 and 100.8%. All the sampled lakes were sinks of atmospheric N<sub>2</sub>O except Salve Fecha that was close to saturation (100.8%). This implies a removal of N<sub>2</sub>O by denitrification due to the complete reduction of N<sub>2</sub>O to N<sub>2</sub>. %N<sub>2</sub>O did not significantly correlate to elevation, lake depth, conductivity and temperature, but correlated to lake area (Fig 4. 4). NO<sub>3</sub><sup>-</sup> also correlated with lake area (Fig 4. 3). Similar patterns of %N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup> as function of lake area might reflect a stronger impact in smaller water bodies of soil-water inputs with low content in N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup> related to denitrification in highland páramo soils (Pineda et al. 2021). Yet, paradoxically, %N<sub>2</sub>O did not significantly correlate with NO<sub>3</sub><sup>-</sup> but correlated with NH<sub>4</sub><sup>+</sup> (and thus also with DIN) (Fig 4. 6), although NH<sub>4</sub><sup>+</sup> did not correlate with lake area (Fig 4. 3). %N<sub>2</sub>O did not also significantly correlate to NO<sub>2</sub><sup>-</sup> and %O<sub>2</sub>. The general lack of correlation of %N<sub>2</sub>O with other variables might reflect that there was a low internal cycling of N<sub>2</sub>O within the lakes, and that the N<sub>2</sub>O levels in lakes were primarily controlled by the inputs of soil-water in a low N<sub>2</sub>O level related to soil denitrification. Additionally, N<sub>2</sub>O was driven closer to equilibrium with the atmosphere in larger lakes where *k* increases with fetch (Wanninkhof 1992). Combined, these two processes led to a relation between %N<sub>2</sub>O and lake surface area.

#### Comparison with floodplain lakes in central Amazon

The páramo lakes were generally smaller (1.1±1.9 km<sup>2</sup>, average ± standard deviation) with higher DOC concentration (9.0±2.2 mg L<sup>-1</sup>) than the lowland floodplain lakes compiled from literature (183±311 km<sup>2</sup>; 5.3±1.1 mg L<sup>-1</sup>) (Fig 4. 7). Yet, the pCO<sub>2</sub> values in lowland floodplain lakes were higher (3,115±1,605 ppm) than in the páramo lakes (1,037±615 ppm) even for overlapping values of depth, lake surface area and DOC concentrations (Fig 4. 7). This pattern probably resulted from inputs of fluvial CO<sub>2</sub> in the floodplain lakes that led to higher pCO<sub>2</sub> values than in the páramo lakes.

The CH<sub>4</sub> values in the sampled highland lakes were higher (3,564±7,186 nmol L<sup>-1</sup>) than in the floodplain lakes in the Central Amazon basin (170±215 nmol L<sup>-1</sup>) compiled from literature (Fig 4. 7). Yet, CH<sub>4</sub> concentration was correlated with depth, lake surface area, and DOC across the full data-set including both highland and lowland lakes. A tighter relation between CH<sub>4</sub> and lake morphology (depth and surface area) would suggest a dominant effect of lake size on the internal production and external inputs from soils of CH<sub>4</sub> despite different settings across the elevational gradient in the whole Amazon basin. Such an effect seems to override the difference in water temperature between highland (12.4±1.9°C) and lowland lakes (30.7±1.4°C), despite the fact that methanogenesis shows a strong temperature dependence (Yvon-Durocher et al. 2014).

This differed from  $p\text{CO}_2$  for which there seemed to be a stronger dependency on external local drivers, in particular the connectivity with the river channel in the floodplain lakes. Consequently, a simple “unifying” correlation between  $p\text{CO}_2$  and other variables (DOC, lake size) did not emerge for both highland and lowland lakes from our data-set. Since  $\text{CH}_4$  was higher and  $\text{CO}_2$  lower in highland lakes compared to lowland lakes, the  $\text{CH}_4:\text{CO}_2$  ratio was markedly higher and also correlated with lake surface area, depth and DOC (Fig 4. 7).

The  $p\text{CO}_2$  was not correlated with DOC across the whole basin suggesting that  $p\text{CO}_2$ -DOC relationship was regionally specific, as also suggested in boreal landscapes (Roehm et al. 2009; Lapierre and del Giorgio 2012). This would suggest that the relation of  $p\text{CO}_2$  and DOC results from multiple processes and do not provide consistent and reproducible patterns across large regions as noted in boreal landscapes. This is possibly related to regional patterns in DOC composition and/or in landscape morphology and lake position in the landscape that dictate the amount and relative proportions of DOC and  $\text{CO}_2$  exports from soils to lakes (Roehm et al. 2009; Lapierre & del Giorgio 2012). This is also in line with a recent analysis of  $p\text{CO}_2$  in African lakes showing diverging  $p\text{CO}_2$ -DOC relations in different type of lakes, in particular in humic and non-humic lakes (Borges et al. 2022). In our particular case, the difference between highland and lowland  $p\text{CO}_2$ -DOC patterns is most likely related to the strong influence of riverine inputs in lowland floodplain lakes. It is unclear if the correlation between  $\text{CH}_4$  and DOC is spurious or reflects a causal relation. This relation could be spurious because shallower lakes tend to have higher  $\text{CH}_4$  concentrations due to lower effect of the removal of  $\text{CH}_4$  by MOX and stronger effect of  $\text{CH}_4$  inputs from sediments or soils (Borges et al. 2011; Holgerson and Raymond 2016). Shallower lakes tended to be smaller in our data-set, and smaller water bodies tend to have higher DOC concentrations because small lakes have a greater connectivity and potential exchange of organic carbon with the riparian wetlands and surrounding terrestrial landscape along its periphery relative to lake surface area than larger lakes (Staehr et al. 2012).

#### **Emissions of $\text{CO}_2$ and $\text{CH}_4$ from lakes and streams of the Amazon basin.**

We use relationships as a function of lake surface area (Fig 4. 7) to scale  $p\text{CO}_2$  and dissolved  $\text{CH}_4$  concentration to the all the lakes in the Amazon basin ( $n=12,549$ ) given by the HydroLAKES data-set (Messenger et al. 2016). For  $\text{CH}_4$ , there was a robust relation with lake surface area across the whole Amazon basin. For consistency, we also used a relation as function of lake surface area to scale  $p\text{CO}_2$  but two separate relations were needed for highland and lowland lakes. Such an approach is applicable because the distribution of lake numbers and surface areas given by HydroLAKES follows a bimodal distribution with the majority located in lowlands (elevation < 500 m) followed by highlands (elevation > 2,500 m) with very few lakes located at elevations between 500 and 2,500 m (Fig 4. S5).

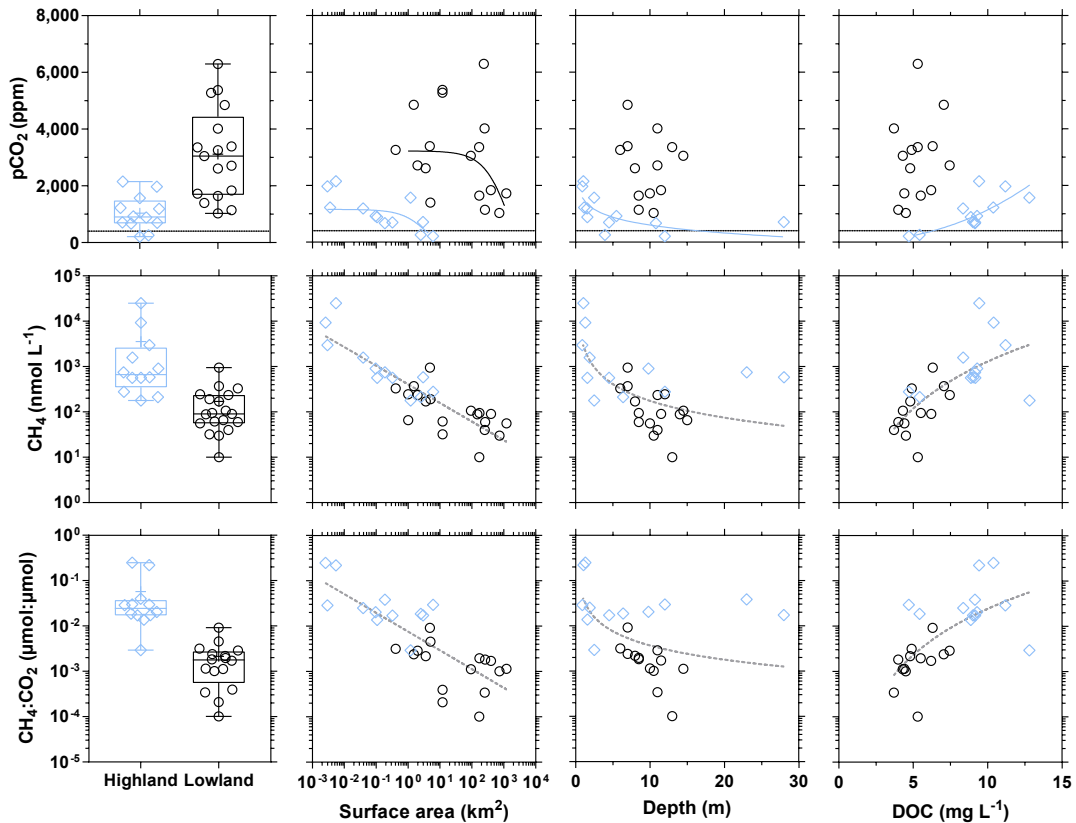
The  $\text{CH}_4$  emission ( $0.0007 \text{ TgC yr}^{-1}$ ) from highland lakes (elevation >500m) was 18 times lower than from lowland lakes (elevation <500m) ( $0.0130 \text{ TgC yr}^{-1}$ ) (Table 4.1). This was due to the fact that lakes in lowlands were 11 times more numerous (12,549 *versus* 1,116) and covered a surface area 23 times larger (22,282 *versus* 952  $\text{km}^2$ ) than in highlands (Fig 4. S5, Table 4.1). The average areal  $F\text{CH}_4$  (per  $\text{m}^2$ ) was in fact 1.3 times higher in highland lakes ( $0.75 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) than in lowland lakes ( $0.58 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) owing to the combination of higher dissolved  $\text{CH}_4$  concentration ( $265 \text{ nmol L}^{-1}$  *versus*  $179 \text{ nmol L}^{-1}$ ) and higher  $k$  values ( $3.5 \text{ cm h}^{-1}$  vs  $2.5 \text{ cm h}^{-1}$ ). The average dissolved  $\text{CH}_4$  concentration was higher in highland lakes because they were on average smaller ( $0.85 \text{ km}^2$ ) than lowland lakes ( $1.78 \text{ km}^2$ ). The  $k$  values were on average higher in highland lakes because of higher average wind speed ( $3.0 \text{ m s}^{-1}$ ) than lowland lakes ( $1.4 \text{ m s}^{-1}$ ) (Table 4.1).

The  $\text{CO}_2$  emission ( $0.022 \text{ TgC yr}^{-1}$ ) from highland lakes was 257 times lower than from lowland lakes ( $5.7 \text{ TgC yr}^{-1}$ ) (Table 4.1). The average areal  $F\text{CO}_2$  was also 11 times lower in highlands ( $23 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) than in lowlands ( $255 \text{ gC m}^{-2} \text{ yr}^{-1}$ ) owing to the lower  $p\text{CO}_2$  values (530 ppm *versus* 2,909 ppm) despite higher  $k$  values ( $3.5 \text{ cm h}^{-1}$  *versus*  $2.5 \text{ cm h}^{-1}$ ).

The total lacustrine emissions of  $\text{CH}_4$  ( $0.01 \text{ TgC yr}^{-1}$ ) and of  $\text{CO}_2$  ( $5.71 \text{ TgC yr}^{-1}$ ) were, respectively, 22 and 11 times lower than the fluvial emissions of  $\text{CH}_4$  ( $0.31 \text{ TgC yr}^{-1}$ ) and  $\text{CO}_2$

(64.42 TgC yr<sup>-1</sup>) at the scale of the Amazon basin (Table 4.1). The relative contribution of the lacustrine *FCH<sub>4</sub>* to total carbon emissions (CO<sub>2</sub>+CH<sub>4</sub>) was low, yet more important in highland lakes (3.1%) than in lowland lakes (0.2%). The relative contribution of *FCH<sub>4</sub>* to total carbon emissions (CO<sub>2</sub>+CH<sub>4</sub>) for the whole Amazon basin (both high- and low-lands) was similarly low in lakes (0.2%) and rivers (0.5%).

**Figure 4.7:** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm), dissolved CH<sub>4</sub> concentration (nmol L<sup>-1</sup>) and CH<sub>4</sub>:CO<sub>2</sub> ratio (μmol:μmol) versus lake surface area (km<sup>2</sup>), lake depth (m) and dissolved organic carbon (DOC in mg L<sup>-1</sup>). Samples in highland lakes in Northern Ecuador (excluding lakes Cuicocha, Loreto and Papallacta influenced by volcanic inputs) and in lowland floodplain lakes of the Amazon basin gathered from literature (Abril et al. 2013; Albéric et al. 2018; Amaral et al. 2020; Barbosa et al. 2016; Mitchell et al. 2018). For pCO<sub>2</sub> only data directly measured by equilibration were compiled, data computed from pH and total alkalinity were excluded as they can be prone to large errors (Abril et al. 2015). Dotted horizontal lines indicate equilibrium with the atmosphere and solid lines show fits to the data (Table 4.S2) (blue solid lines = highland lakes; black solid lines = lowland lakes; grey dotted lines = high- and low-land lakes).



## Conclusions

Global estimates of CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O emissions from lakes and rivers are uncertain because available data-sets do not capture the inherent large variability, and are insufficient to derive robust statistical functions to scale fluxes with confidence, despite the recent emergence of detailed spatial data-sets of location and surface area for rivers (Allen and Pavelsky 2018; Linke et al. 2019) and lakes (Messenger et al. 2016) that can be used to scale fluxes. Tropical lakes are under-sampled with regards to GHG emissions compared to higher latitude systems, and this is particularly true for mountainous lakes. The surface-area averaged pCO<sub>2</sub> of 465 ppm we reported for 12 Andean mountainous lakes at elevations between ~2,200 and ~4,350 m (excluding three lakes influenced by volcanic activity) was lower than the values usually attributed

to tropical lakes of ~1,900 ppm (Aufdenkampe et al. 2011; Raymond et al. 2013; Marotta et al. 2009). The average pCO<sub>2</sub> in the sampled 12 Andean mountainous lakes was also lower than in the floodplain lakes of the Central Amazon with a surface averaged pCO<sub>2</sub> of 2,020 ppm. The surface-area averaged CH<sub>4</sub> in the 12 Andean mountainous lakes of 322 nmol L<sup>-1</sup> was higher than in the floodplain lakes of the Central Amazon with a surface-area averaged CH<sub>4</sub> of 58 nmol L<sup>-1</sup>. The CO<sub>2</sub> and CH<sub>4</sub> emissions from highland lakes only represented 0.4% and 5.2% of the total lacustrine emissions at the scale of the Amazon basin. Total CO<sub>2</sub> and CH<sub>4</sub> emissions from lakes (5.7 TgC yr<sup>-1</sup>) represented a small fraction (8%) of total lentic and lotic CO<sub>2</sub> and CH<sub>4</sub> emissions at the scale of the Amazon basin (70.5 TgC yr<sup>-1</sup>).

## **Acknowledgments**

We thank Yannick Strobandt for analytical assistance with stable isotope and POC and DOC concentration measurements, and Gwenaël Abril for providing CH<sub>4</sub> data in Central Amazonian lakes. This research was funded by the *Académie de recherche et d'enseignement supérieur* (ARES). AVB is a research director at the *Fonds National de la Recherche Scientifique*.

## Supplementary material

TABLE 4S.1: Major cations (Na<sup>+</sup>, K<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>) (in mg L<sup>-1</sup> and µequivalents L<sup>-1</sup>) in surface waters.

|              | Na <sup>+</sup><br>mg L <sup>-1</sup> | K <sup>+</sup><br>mg L <sup>-1</sup> | Mg <sup>2+</sup><br>mg L <sup>-1</sup> | Ca <sup>2+</sup><br>mg L <sup>-1</sup> | Na <sup>+</sup><br>µeq L <sup>-1</sup> | K <sup>+</sup><br>µeq L <sup>-1</sup> | Mg <sup>2+</sup><br>µeq L <sup>-1</sup> | Ca <sup>2+</sup><br>µeq L <sup>-1</sup> | Cation sequence                                                     |
|--------------|---------------------------------------|--------------------------------------|----------------------------------------|----------------------------------------|----------------------------------------|---------------------------------------|-----------------------------------------|-----------------------------------------|---------------------------------------------------------------------|
| Cubilche 1   | 1.6                                   | 1.2                                  | 0.6                                    | 1.7                                    | 69                                     | 30                                    | 48                                      | 84                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Cubilche 2   | 1.6                                   | 1.4                                  | 0.6                                    | 1.8                                    | 70                                     | 35                                    | 51                                      | 88                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Cuicocha     | 62.3                                  | 5.7                                  | 34.6                                   | 44.4                                   | 2,709                                  | 145                                   | 2,845                                   | 2,216                                   | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Humbolt      | 16.4                                  | 5.0                                  | 9.2                                    | 9.7                                    | 711                                    | 128                                   | 760                                     | 485                                     | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Loreto       | 60.6                                  | 5.8                                  | 34.2                                   | 43.7                                   | 2,637                                  | 149                                   | 2,814                                   | 2,179                                   | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Mojanda      | 1.7                                   | 1.2                                  | 0.6                                    | 1.6                                    | 73                                     | 30                                    | 48                                      | 78                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Negra        | 1.8                                   | 1.3                                  | 0.6                                    | 1.6                                    | 78                                     | 33                                    | 50                                      | 81                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Papallacta   | 60.6                                  | 5.7                                  | 34.4                                   | 43.7                                   | 2,637                                  | 145                                   | 2,828                                   | 2,182                                   | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Salve Faccha | 10.2                                  | 1.5                                  | 2.2                                    | 7.4                                    | 445                                    | 38                                    | 184                                     | 369                                     | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| San Pablo    | 23.7                                  | 4.0                                  | 12.7                                   | 16.9                                   | 1,032                                  | 104                                   | 1,046                                   | 845                                     | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Santa lucía  | 1.6                                   | 1.2                                  | 0.6                                    | 1.8                                    | 71                                     | 31                                    | 52                                      | 88                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Secas        | 10.6                                  | 3.3                                  | 5.8                                    | 6.9                                    | 459                                    | 83                                    | 475                                     | 343                                     | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Sucus        | 1.64                                  | 1.18                                 | 0.58                                   | 1.60                                   | 71                                     | 30                                    | 48                                      | 80                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Yaguarcocha  | 24.0                                  | 4.2                                  | 12.7                                   | 17.0                                   | 1,045                                  | 107                                   | 1,046                                   | 848                                     | Na <sup>+</sup> >Ca <sup>2+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |
| Yanacocha    | 1.57                                  | 1.19                                 | 0.58                                   | 1.58                                   | 68                                     | 30                                    | 48                                      | 79                                      | Ca <sup>2+</sup> >Na <sup>+</sup> >Mg <sup>2+</sup> >K <sup>+</sup> |

**TABLE 4S.2:** Equations and statistics at 0.05 level of curve fits of data shown in Figures 4...

| Fig. | Function                                                                                   | Equation                         | $r^2$ | $p$     | $n$   |
|------|--------------------------------------------------------------------------------------------|----------------------------------|-------|---------|-------|
| 2    | surface area <i>versus</i> elevation                                                       | $Y = -0.001756 \cdot X + 7.4$    | 0.32  | 0.0272  | 15    |
| 2    | log(depth) <i>versus</i> elevation                                                         | $Y = -0.0004121 \cdot X + 2.151$ | 0.18  | 0.1150  | 15    |
| 2    | log(water temperature) <i>versus</i> log(elevation)                                        | $Y = -0.6515 \cdot X + 3.376$    | 0.84  | <0.0001 | 15    |
| 2    | $\delta^{18}\text{O-H}_2\text{O}$ <i>versus</i> elevation                                  | $Y = -0.001312 \cdot X - 2.713$  | 0.26  | 0.0636  | 15    |
| 2    | conductivity <i>versus</i> elevation                                                       | $Y = -0.1072 \cdot X + 479$      | 0.43  | 0.0204  | 12 *  |
| 2    | TA <i>versus</i> elevation                                                                 | $Y = -0.001162 \cdot X + 5.084$  | 0.44  | 0.0189  | 12 *  |
| 3    | POC <i>versus</i> log(conductivity)                                                        | $Y = 3.767 \cdot X - 4.175$      | 0.85  | <0.0001 | 12 *  |
| 3    | log(DOC) <i>versus</i> log(elevation)                                                      | $Y = 1.074 \cdot X - 2.865$      | 0.61  | 0.0029  | 12 *  |
| 3    | log(DOC) <i>versus</i> log(surface area)                                                   | $Y = -0.04877 \cdot X + 0.9942$  | 0.56  | 0.0052  | 12 *  |
| 3    | log(DOC) <i>versus</i> log(depth)                                                          | $Y = -0.3862 \cdot X + 1.081$    | 0.55  | 0.0016  | 15    |
| 3    | log(DOC) <i>versus</i> conductivity                                                        | $Y = -0.001015 \cdot X + 1.035$  | 0.89  | <0.0001 | 15    |
| 3    | log(DOC) <i>versus</i> temperature                                                         | $Y = -0.05368 \cdot X + 1.578$   | 0.69  | 0.0008  | 12 *  |
| 3    | $\text{NO}_3^-$ <i>versus</i> log(surface area)                                            | $Y = 0.4539 \cdot X + 1.388$     | 0.62  | 0.0005  | 15    |
| 3    | $\text{NO}_3^-$ <i>versus</i> log(depth)                                                   | $Y = 0.7412 \cdot X + 0.5505$    | 0.47  | 0.0049  | 15    |
| 3    | $\text{NO}_3^-$ <i>versus</i> log(conductivity)                                            | $Y = 0.7178 \cdot X - 0.3462$    | 0.48  | 0.0041  | 15    |
| 4    | log(pCO <sub>2</sub> ) <i>versus</i> elevation                                             | $Y = 0.0003642 \cdot X + 1.626$  | 0.56  | 0.0054  | 12 *  |
| 4    | log(pCO <sub>2</sub> ) <i>versus</i> log(surface area)                                     | $Y = -0.2228 \cdot X + 2.74$     | 0.60  | 0.0033  | 12 *  |
| 4    | log(pCO <sub>2</sub> ) <i>versus</i> log(depth)                                            | $Y = -0.3118 \cdot X + 3.176$    | 0.59  | 0.0092  | 10 ** |
| 4    | log(pCO <sub>2</sub> ) <i>versus</i> log(conductivity)                                     | $Y = -0.4246 \cdot X + 3.668$    | 0.38  | 0.0338  | 12 *  |
| 4    | log(pCO <sub>2</sub> ) <i>versus</i> log(temperature)                                      | $Y = -3.824 \cdot X + 7.105$     | 0.72  | 0.0005  | 12 *  |
| 4    | log(CH <sub>4</sub> ) <i>versus</i> log(surface area)                                      | $Y = -0.4573 \cdot X + 2.53$     | 0.69  | 0.0001  | 15    |
| 4    | log(CH <sub>4</sub> ) <i>versus</i> log(depth)                                             | $Y = -0.6757 \cdot X + 3.323$    | 0.42  | 0.0085  | 15    |
| 4    | log(CH <sub>4</sub> ) <i>versus</i> log(conductivity)                                      | $Y = -0.6283 \cdot X + 4.089$    | 0.40  | 0.0111  | 15    |
| 4    | %N <sub>2</sub> O <i>versus</i> log(surface area)                                          | $Y = 8.459 \cdot X + 89.49$      | 0.65  | 0.0048  | 10 ** |
| 4    | log(%O <sub>2</sub> ) <i>versus</i> log(surface area)                                      | $Y = 0.01921 \cdot X + 1.924$    | 0.39  | 0.0122  | 15    |
| 4    | log(%O <sub>2</sub> ) <i>versus</i> log(depth)                                             | $Y = 0.03767 \cdot X + 1.884$    | 0.43  | 0.0082  | 15    |
| 4    | log(%O <sub>2</sub> ) <i>versus</i> log(temperature)                                       | $Y = 0.3271 \cdot X + 1.555$     | 0.38  | 0.0139  | 15    |
| 5    | log(pCO <sub>2</sub> ) <i>versus</i> POC                                                   | $Y = -0.07622 \cdot X + 3.107$   | 0.20  | 0.1421  | 12 *  |
| 5    | log(pCO <sub>2</sub> ) <i>versus</i> log(DOC)                                              | $Y = 2.287 \cdot X + 0.7695$     | 0.77  | 0.0002  | 12 *  |
| 5    | pCO <sub>2</sub> <i>versus</i> %O <sub>2</sub>                                             | $Y = -83.21 \cdot X + 7766$      | 0.82  | <0.0001 | 12 *  |
| 5    | log(CH <sub>4</sub> ) <i>versus</i> log(DOC)                                               | $Y = 1.068 \cdot X + 1.981$      | 0.29  | 0.0393  | 15    |
| 5    | log(CH <sub>4</sub> ) <i>versus</i> log(%O <sub>2</sub> )                                  | $Y = -9.975 \cdot X + 21.9$      | 0.31  | 0.0320  | 15    |
| 6    | %N <sub>2</sub> O <i>versus</i> NH <sub>4</sub> <sup>+</sup>                               | $Y = 4.101 \cdot X + 51.83$      | 0.42  | 0.0088  | 15    |
| 6    | %N <sub>2</sub> O <i>versus</i> DIN                                                        | $Y = 3.441 \cdot X + 51.79$      | 0.38  | 0.0150  | 15    |
| 7    | log(pCO <sub>2</sub> ) <i>versus</i> surface area (highland)                               | $Y = -0.1293 \cdot X + 3.063$    | 0.58  | 0.0039  | 12 *  |
| 7    | log(pCO <sub>2</sub> ) <i>versus</i> surface area (lowland)                                | $Y = -0.0003745 \cdot X + 3.508$ | 0.26  | 0.0376  | 17    |
| 7    | pCO <sub>2</sub> <i>versus</i> log(depth) (highland)                                       | $Y = -927.4 \cdot X + 1531$      | 0.52  | 0.0083  | 12 *  |
| 7    | log(pCO <sub>2</sub> ) <i>versus</i> log(DOC) (highland)                                   | $Y = 2.287 \cdot X + 0.7695$     | 0.77  | 0.0002  | 12 *  |
| 7    | log(CH <sub>4</sub> ) <i>versus</i> log (surface area) (highland+lowland)                  | $Y = -0.3753 \cdot X + 2.545$    | 0.71  | <0.0001 | 34 *  |
| 7    | log(CH <sub>4</sub> ) <i>versus</i> log (depth) (highland+lowland)                         | $Y = -1.232 \cdot X + 3.477$     | 0.44  | 0.0001  | 28 *  |
| 7    | log(CH <sub>4</sub> ) <i>versus</i> log (DOC) (highland+lowland)                           | $Y = 3.442 \cdot X - 0.33$       | 0.50  | <0.0001 | 25 *  |
| 7    | log(CH <sub>4</sub> :CO <sub>2</sub> ) <i>versus</i> log (surface area) (highland+lowland) | $Y = -0.4136 \cdot X - 2.118$    | 0.63  | <0.0001 | 28 *  |
| 7    | log(CH <sub>4</sub> :CO <sub>2</sub> ) <i>versus</i> log (DOC) (highland+lowland)          | $Y = 3.391 \cdot X - 5.006$      | 0.43  | 0.0004  | 25 *  |
| 7    | log(CH <sub>4</sub> :CO <sub>2</sub> ) <i>versus</i> log (depth) (highland+lowland)        | $Y = -1.042 \cdot X - 1.39$      | 0.27  | 0.0074  | 25 *  |
| S1   | Ca <sup>2+</sup> <i>versus</i> Na <sup>+</sup>                                             | $Y = 0.6659 \cdot X + 0.4364$    | 0.99  | <0.0001 | 12 *  |
| S1   | Mg <sup>2+</sup> <i>versus</i> Na <sup>+</sup>                                             | $Y = 0.5443 \cdot X - 0.4691$    | 0.96  | <0.0001 | 12 *  |
| S1   | TA <i>versus</i> Na <sup>+</sup>                                                           | $Y = 0.1261 \cdot X - 0.05166$   | 0.97  | <0.0001 | 12 *  |
| S1   | K <sup>+</sup> <i>versus</i> Na <sup>+</sup>                                               | $Y = 0.1487 \cdot X + 1.022$     | 0.81  | <0.0001 | 12 *  |
| S1   | DSi <sup>+</sup> <i>versus</i> Na <sup>+</sup>                                             | $Y = 0.8402 \cdot X - 0.7169$    | 0.94  | <0.0001 | 12 *  |
| S1   | Fe <i>versus</i> Na <sup>+</sup>                                                           | $Y = -2.512 \cdot X + 97.74$     | 0.89  | <0.0001 | 12 *  |
| S2   | log(Fe) <i>versus</i> DOC                                                                  | $Y = 0.06804 \cdot X + 1.284$    | 0.87  | <0.0001 | 15    |
| S3   | $\delta\text{D}$ <i>versus</i> $\delta^{18}\text{O-H}_2\text{O}$                           | $Y = 5.131 \cdot X - 19.57$      | 0.91  | <0.0001 | 15    |
| S4   | log(pCO <sub>2</sub> ) <i>versus</i> $\delta^{13}\text{C-DIC}$                             | $Y = -0.06483 \cdot X + 2.258$   | 0.44  | 0.0196  | 12 *  |
| S4   | $\delta^{13}\text{C-DIC}$ <i>versus</i> $(\text{HCO}_3^- + \text{CO}_3^{2-})\text{:DIC}$   | $Y = 0.02482 \cdot X + 1.092$    | 0.53  | 0.0069  | 12 *  |

\* Excluding "volcanic" lakes (Cuicocha, Loreto and Papallacta)

\*\* Excluding "volcanic" lakes (Cuicocha, Loreto and Papallacta) and "urban" lakes (Yaguarcocha and San Pablo)



FIGURE 4S.1: Major cations ( $K^+$ ,  $Mg^{2+}$ ,  $Ca^{2+}$  in  $mg\ L^{-1}$ ), dissolved silica (DSi) ( $mg\ L^{-1}$ ), total alkalinity (TA in  $mmol\ L^{-1}$ ) and Fe ( $\mu g\ L^{-1}$ ) versus  $Na^+$  in surface waters of fifteen highland lakes in Northern Ecuador (Fig 4. 1). Solid grey lines show fits to the data (Table 4.S2).

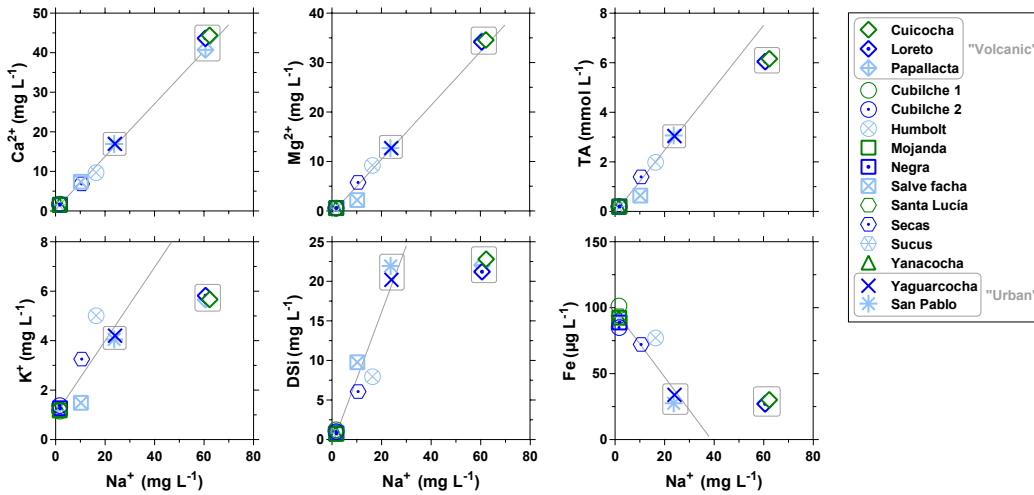
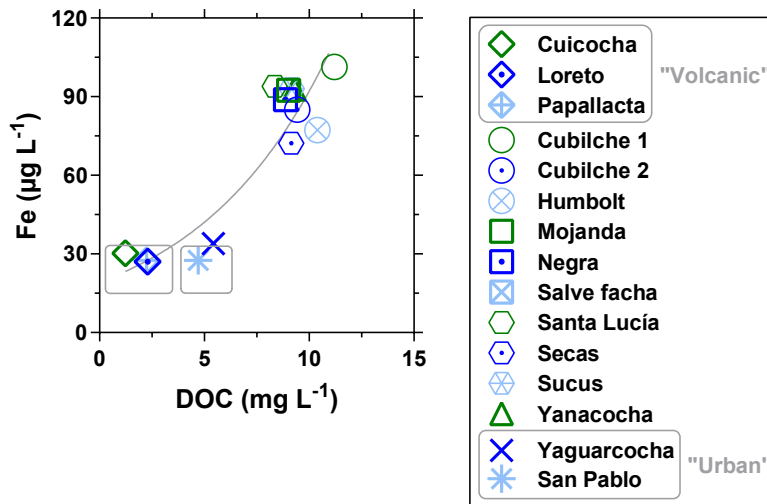
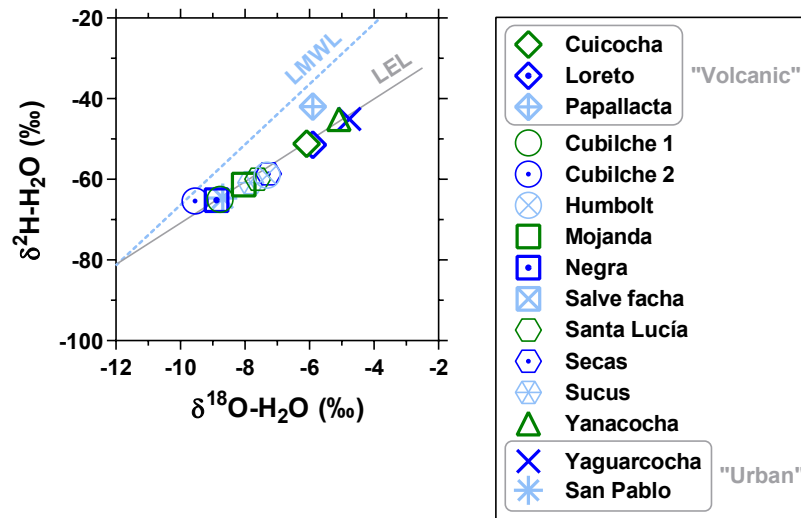


FIGURE 4S.2: Fe ( $\mu g\ L^{-1}$ ) versus dissolved organic carbon (DOC in  $mg\ L^{-1}$ ) in surface waters of fifteen highland lakes in Northern Ecuador (Fig 4. 1). Solid grey line shows fit to the data (Table 4.S2).



**FIGURE 4S.3:** Stable.isotope composition of H and O of H<sub>2</sub>O ( $\delta^{18}\text{O}\text{-H}_2\text{O}$  and  $\delta\text{D}\text{-H}_2\text{O}$  in ‰) in surface waters of fifteen highland lakes in Northern Ecuador (Fig 4. 1). Solid grey line shows fit to the data (Table 4.S2) and gives the local evaporative line (LEL). Dotted line is the local meteoric water line (LMWL) (Inguaggiato et al. 2010).



**FIGURE 4S.4:** Partial pressure of CO<sub>2</sub> (pCO<sub>2</sub> in ppm) and the ratio of HCO<sub>3</sub><sup>-</sup>+CO<sub>3</sub><sup>2-</sup> to dissolved inorganic carbon (DIC) (mol:mol) versus the stable.isotope composition of DIC ( $\delta^{13}\text{C}\text{-DIC}$  in ‰) in surface waters of fifteen highland lakes in Northern Ecuador (Fig 4. 1). Solid grey lines show fits to the data (Table 4.S2).

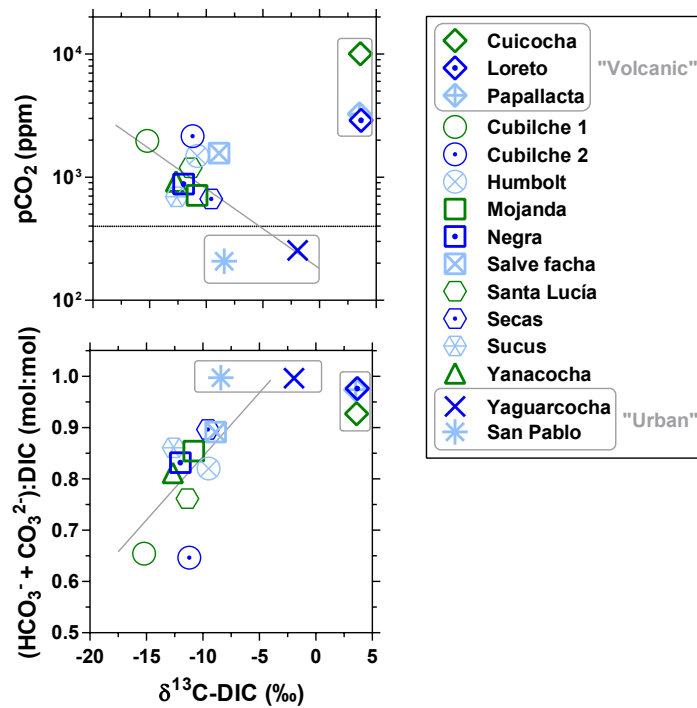
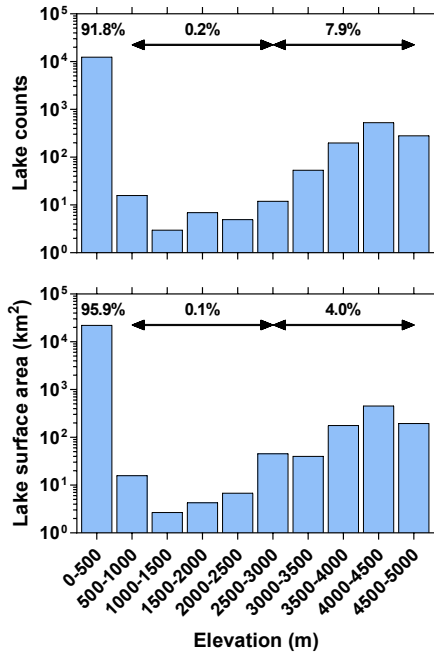


FIGURE 4S.5: Lake counts and cumulated lake surface area (km<sup>2</sup>) as function of elevation (in m by bins of 500 m) in the Amazon basin from the HydroLAKES spatial data-set (Messenger et al. 2016).





# CONCLUSIONS

This study provides 1) the first large-scale survey across an altitudinal gradient of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O exchange between Andean mountainous streams of the Amazon River basin and the atmosphere 2) the C dynamics within the Basin based on properties such as TSM, DOC, POC, specific conductivity, total alkalinity, stable isotope ratios, and three optical proxies of CDOM and 3) the dissolved concentrations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O in 15 lakes located in the northern region of Ecuadorian Andes along an elevational gradient from 2,213 to 4,361 m above sea level.

## Reveling the significance of the project.

Understanding the functioning of the natural drivers of the C and N cycles is crucial to face climate change. Inland waters are the primary natural source of greenhouse gases for CO<sub>2</sub> (~21%) and CH<sub>4</sub> (~41%), and the third one for N<sub>2</sub>O (~2%). In this sense, estimates of CO<sub>2</sub> emissions from inland waters could range from 0.7 to 2.9 PgC yr<sup>-1</sup>, being the most probable value of 2.1 PgC yr<sup>-1</sup> (Borges et al., 2015c; Raymond et al., 2013). This number is significant considering global anthropogenic CO<sub>2</sub> emissions total 10.0 PgC yr<sup>-1</sup>. Compared with the primary natural C sinks, these emissions are similar to the uptake capacity of all oceans (2.0 PgC yr<sup>-1</sup>) and slightly lower than the total land sink (3.1 PgC yr<sup>-1</sup>). Collectively, aquatic ecosystems contribute 41% of total global CH<sub>4</sub> emissions from anthropogenic sources (359 TgCH<sub>4</sub> yr<sup>-1</sup>) (Saunois et al., 2020), which is equivalent to the enteric fermentation and manure management (111 TgCH<sub>4</sub> yr<sup>-1</sup>) and rice paddies (30 TgCH<sub>4</sub> yr<sup>-1</sup>) together. Estimates of N<sub>2</sub>O emissions from rivers have been dramatically revised downwards to 0.4 Tg N yr<sup>-1</sup>, which represents ~1.7% of the natural and anthropogenic sources (Tian et al., 2020); nevertheless, N<sub>2</sub>O has a GWP 273 times that of CO<sub>2</sub> for a 100-year timescale. Despite these data, GHGs from inland waters have not been fully incorporated into national GHG inventories due to the uncertainties in estimations, the paucity of in situ direct measures, the spatial skewing in sampling, and the immense heterogeneity between ecosystems and driving factors. With about 44% of the global inland waters surface (Raymond et al., 2013), the tropics can produce ~60-80% of CO<sub>2</sub>, ~70% of CH<sub>4</sub>, and ~79% of N<sub>2</sub>O of the global inland water emissions. Favorable conditions in the tropics are adequate for GHG formation within the water body and surrounding ecosystems. For instance, temperature, light availability, hydrological inputs, plenty of labile organic C, bacterial degradation, biogeochemical processing, and connectivity between riparian zones and the watershed (Borges et al., 2015a; Stanley et al., 2015). Therefore, several studies encompassing the most significant number of drivers and territories in the tropics are required to comprehensively understand biogeochemical cycles.

The interest in CO<sub>2</sub> evasion fluxes has centered mainly on rivers draining low-altitude catchments, mostly in boreal regions. Nevertheless, mountains account for 25% of the Earth's land surface, and their streams contribute more than a third to the global runoff (Meybeck et al., 2001). Higher CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O emissions from low-order streams in the mountains are characterized by high gas transfer velocity (*k*) (Ulseth et al., 2019). This massive degassing results in concentrations of CO<sub>2</sub> and CH<sub>4</sub> close to equilibrium with the atmosphere (Crawford et al., 2015). This gap has increased uncertainty in global GHG estimates since models employ spatial extrapolations without considering particular evasion rates from rivers of Strahler order 1 or 2 (width < 200m). Consequently, the studies should incorporate specific information on ecosystems scarcely sampled, such as the Andean zone of South America.

As important as measuring GHGs is understanding the C dynamics that produce them. For instance, it is known that Andes are a crucial source of sediments and dissolved matter to the Amazon River, the largest single source of fresh water and OC to the oceans (Gibbs, 1967). In

the Amazon basin, the average export of DOC and POC are 21.6 and 14.4 Tg yr<sup>-1</sup> (Richey et al., 1990). Of these, between 30 and 50% of the POC could have Andean origin in large tributaries such as the Óbidos River (Hedges et al., 2000; Quay et al., 1992). Unlike POC, DOC appears to derive mainly from lowland sources. So, the study includes a report of the C dynamics within the Basin. Like rivers, lacustrine systems have also been under-sampled in the tropics; nonetheless, the CH<sub>4</sub> emissions from lakes are disproportionately more intense than their higher latitude counterparts (Bastviken et al., 2011; Rosentreter et al., 2021), possibly, due to the higher phytoplankton production in non-humic lakes and the high CH<sub>4</sub> methanogenesis rates, driven by light availability and temperatures in tropics, respectively. A particular ecosystem of the tropics is the páramo located between 3,000 to 4,300 m. with intensive peat formation and many wetlands and lakes. However, the low temperature significantly limits the development of biochemical processes.

Ecuador is the smallest of the 17 megadiverse countries in the world and harbors an astounding number of ecosystems (Cuesta et al., 2017), which offered the conditions to study in a relatively small space, the GHGs evasions of the Napo, as well as a diverse lacustrine system. First, the fluvial system includes 44 stations along the Napo and Coca Rivers basin (25 main rivers and 19 tributary rivers). The campaigns included the Andean páramos (>3,000 m), the highlands (3,000 – 1,500 m), the foothills (1,500 – 500), and the lowlands of the Napo basin (<500 m). Second, a lacustrine system composed of 15 lakes located in the Andean páramos and the highlands (4,361 -2,213 m) with different characteristics of size (0.003 to 6.1 km<sup>2</sup>), depth (0.9 to 74 m) and type (moorland, urban, and volcanic lakes). Together, the two stages focus on areas whose ecosystems are analogous to more than 13% of the Amazon basin and for which there are few emissions records.

## **GHGs dynamics in the Napo River Basin**

We sampled 25 stations in the mainstem Napo and Coca rivers and 19 stations in the tributaries along an elevation gradient from 3990 m to 175 m on eight occasions from 2018 to 2021. The sampled streams were over-saturated in CO<sub>2</sub> and CH<sub>4</sub>; consequently, the flux was unidirectional from the water to the atmosphere (emission). In contrast, N<sub>2</sub>O oscillated between under and over-saturation, corresponding to a sink or source of N<sub>2</sub>O in the atmosphere.

The pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values in streams at elevations > 500 m (median 560 ppm and 17 nmol L<sup>-1</sup>, respectively) were distinctly lower than at elevations < 500 m (median 1,280 ppm and 370 nmol L<sup>-1</sup>) and increased exponentially from highlands to lowlands. By including O<sub>2</sub> in the analysis, we observed that the %O<sub>2</sub> decreased exponentially from highlands to lowlands. For instance, values in streams at elevations > 500 m (median 74.4 %) were higher than at elevations < 500 m (median 63.2 %). This trend suggests that the same input and output processes controlled pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration. For instance, at highlands, CO<sub>2</sub> and CH<sub>4</sub> degasification (high k values), primary production (clear waters that enables photosynthesis), and water aeration (turbulence and velocity). Conversely, lowlands intensive respiration (O<sub>2</sub> consumption), organic matter mineralization (wetlands presence), low light penetration (black and white waters), low oxygen availability (methanogenesis), and low degassing rates (low k values).

Temperature varies markedly with altitude and controls the rates of respiration and methanogenesis in soils and riverine sediments. The outcomes suggest the warmer conditions in the lowland rivers contributed to higher pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values since pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values increased exponentially with water temperature. The variability of land cover changed considerably with elevation. A clear transition from the dominance of grasslands in uplands (páramos) to evergreen forests in lowlands was observed.

Stream pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values increased exponentially with forest cover and were generally higher where grassland cover was low to zero. Also, pCO<sub>2</sub> and dissolved CH<sub>4</sub> concentration values were found to be negatively related to the catchment slope due to massive evasions but directly related with %O<sub>2</sub>. Both phenomena imply the same result of oxygenation of the water.

The N<sub>2</sub>O saturation levels (%N<sub>2</sub>O) in streams at an altitude > 500 m (median 95.4 %) were lower than at elevations < 500 m (median 105.0 %). The %N<sub>2</sub>O increased exponentially with decreasing elevation, although lower than for the other GHGs. Lower N<sub>2</sub>O in high-elevation streams was consistent with reported elevation gradients of N<sub>2</sub>O soil-air fluxes in the Ecuadorian Andes, with a soil N<sub>2</sub>O sink at high elevation (2,200-3,010 m) and an N<sub>2</sub>O source from soils to the atmosphere at lower elevation (400-1,100 m). The removal of N<sub>2</sub>O in soils was due to complete denitrification (reduction of N<sub>2</sub>O to N<sub>2</sub>) in highlands related to low soil dissolved inorganic N (DIN) content. The combination of higher stream DIN and low %N<sub>2</sub>O in highland streams and the opposite pattern in lowland streams led to a negative relation between %N<sub>2</sub>O and DIN. Such a negative relation is unique compared to previously reported patterns, as N<sub>2</sub>O content in rivers is generally positively related to DIN.

The air-water diffusive flux of CO<sub>2</sub> (FCO<sub>2</sub>) ranged between -1 and 1,074 mol m<sup>-2</sup> yr<sup>-1</sup> and was higher in tributaries (median 128 mol m<sup>-2</sup> yr<sup>-1</sup>) than in the main stem (median 34 mol m<sup>-2</sup> yr<sup>-1</sup>). The air-water diffusive flux of CH<sub>4</sub> (FCH<sub>4</sub>) ranged between 45 and 8,985 mmol m<sup>-2</sup> yr<sup>-1</sup> and was also higher in tributaries (median 999 mmol m<sup>-2</sup> yr<sup>-1</sup>) than in the main stem (332 mmol m<sup>-2</sup> yr<sup>-1</sup>). Both FCO<sub>2</sub> and FCH<sub>4</sub> were inversely correlated to elevation and upstream terrain slope. The air-water diffusive flux of N<sub>2</sub>O (FN<sub>2</sub>O) ranged between -23 and 27 mmol m<sup>-2</sup> yr<sup>-1</sup> and did not show a marked difference between tributaries and the main stem. There was a general tendency towards a sink of N<sub>2</sub>O at an elevation > 1,000m, with 12 sites showing negative FN<sub>2</sub>O values and only 4 showing positive FN<sub>2</sub>O values. At an elevation < 1,000m, the FN<sub>2</sub>O values were close to zero (-1.3 to 3 mmol m<sup>-2</sup> yr<sup>-1</sup>), except for the Guamanyacu, Inmiyana, and Tiputini streams that showed distinctly positive FN<sub>2</sub>O values (9.7 to 21.6 mmol m<sup>-2</sup> yr<sup>-1</sup>) and the Huirima that showed a distinctly negative FN<sub>2</sub>O value (-7.6 mmol m<sup>-2</sup> yr<sup>-1</sup>). The underlying idea is that in highly turbulent systems (high k values), degassing brings gases to equilibrium with the atmosphere, and emissions are near zero. Consequently, the gaseous emission is assumed to be limited by gas availability.

Lowland rivers dominate the integrated FCO<sub>2</sub> (64±5 TgC yr<sup>-1</sup>) (65%). However, FCO<sub>2</sub> per m<sup>2</sup> in highland and piedmont rivers (197±35 and 143±3 mol m<sup>-2</sup> yr<sup>-1</sup>) were 1.7 and 1.2 times higher than in lowland rivers (115±8 mol m<sup>-2</sup> yr<sup>-1</sup>), respectively, due to higher k values. The deposition of sediments and the presence of forests in the piedmont region are favorable to high in-stream CH<sub>4</sub> production. In larger lowland river channels that are relatively disconnected from floodplains, the deep water column and strong currents do not promote the accumulation of dissolved CH<sub>4</sub> in the water column but promote the removal of dissolved CH<sub>4</sub> either by emission of CH<sub>4</sub> to the atmosphere or by microbial oxidation (Borges et al., 2015). The piedmont streams accounted for 62% of the total integrated diffusive FCH<sub>4</sub> (0.4±0.1 TgCH<sub>4</sub> yr<sup>-1</sup>), higher than the contribution from lowland rivers (28%). In comparison, highland streams accounted for only 10% due to much higher k values (148.8±3.2 versus 14.5±0.4 cm h<sup>-1</sup>). Interestingly, the diffusive FCH<sub>4</sub> per m<sup>2</sup> in piedmont streams (1,573±350 mmol m<sup>-2</sup> yr<sup>-1</sup>) was 6.6 times higher than in lowland rivers (137±79 mmol m<sup>-2</sup> yr<sup>-1</sup>) because of much higher dissolved CH<sub>4</sub> concentrations (797±174 versus 307±100 nmol L<sup>-1</sup>), while k was relatively closer (20.0±0.5 versus 14.5±0.4 cm h<sup>-1</sup>). The integrated FN<sub>2</sub>O (25±4 GgN<sub>2</sub>O-N yr<sup>-1</sup>) was dominated by lowland rivers and streams (99% of total), while piedmont streams were a minimal source of N<sub>2</sub>O, and highland streams were a tiny sink of N<sub>2</sub>O. The computation of integrated FN<sub>2</sub>O value for the Amazon River basin resulted in 9±4 GgN<sub>2</sub>O-N yr<sup>-1</sup>, close to the range of N<sub>2</sub>O emissions for the Amazon River basin of 11 to 16 GgN<sub>2</sub>O-N yr<sup>-1</sup> modeled from N deposition and emission factors (Maavara et al., 2019).

We concluded that mountainous headwater and piedmont streams were hotspots of CO<sub>2</sub> and CH<sub>4</sub> emissions from the Amazon River basin, accounting for a significant fraction of total emissions (35 and 72% for CO<sub>2</sub> and CH<sub>4</sub>, respectively). Although the present study constitutes a baseline for the integration of the systems that make up the Amazon Basin, such as the Andes, Piedmont, and Amazonia, it is necessary to expand the sampling area and analyze the factors that drive the degassing rate (k values) a significant parameter assumed to the obtained results.

## **Suspended matter and organic carbon pools.**

We report specific conductivity, TA, stable isotope ratios ( $\delta^{13}\text{C-DIC}$ ,  $\delta^{13}\text{C-POC}$ ,  $\delta^{13}\text{C-DOC}$ ), TSM, total organic matter (POC and DOC) and three optical proxies of CDOM. Specific conductivity showed a downstream decrease from 368 to 19  $\mu\text{S cm}^{-1}$  in the main-stem. Conversely, in the tributaries, specific conductivity varied within a narrower range from 19 to 122  $\mu\text{S cm}^{-1}$ . Specific conductivity was positively related to stream gradient. TA ranged between 135 and 1,656  $\mu\text{mol L}^{-1}$  and mimicked of specific conductivity. The elevational patterns of specific conductivity and TA resulted from intense rock weathering in the high elevation areas followed by dilution in the lower reaches with the increase of stream size, as indicated by negative correlations with catchment area and discharge. The  $\delta^{13}\text{C-DIC}$  values ranged between -22.1 and +3.1 ‰, and by differentiating between the tributaries and the main-stem,  $\delta^{13}\text{C-DIC}$  is in line with the patterns of HCO<sub>3</sub><sup>-</sup>. A negative relation between  $\delta^{13}\text{C-DIC}$  and CO<sub>2</sub> concentration suggests the addition of CO<sub>2</sub> depleted in <sup>13</sup>C from heterotrophic process and the gradual dilution of HCO<sub>3</sub><sup>-</sup> in lowlands waters.

TSM showed a consistent downstream increase from 3 to 480 mg L<sup>-1</sup>. Nevertheless, the indicates that the higher particulate matter content transported by lowland rivers originated from moderate elevation areas (<1000 m) rather than from headwater regions. TSM was negatively related to upstream catchment slope and stream gradient, and positively related to freshwater discharge and catchment area, implying an accumulation of particles when the river size increases and reaches lowlands. TSM and POC are positively correlated, Conversely, TSM and %POC, ranged from 0.2 and 21.8%, were negatively correlated, this trend probably reflected a better preservation of organic matter fraction in colder high elevation region, the in-stream loss by microbial respiration of organic carbon and the dilution of POC by mineral particles during downstream transport. The downstream decrease of the C:N ratio of fluvial POM is consistent with the decrease from highlands to lowlands of soil C:N ratio reported by Townsend-Small et al. (2005) in the Peruvian Andes that was attributed to a higher microbial source (with low C:N) in low-elevation soils compared to high-elevation soils with a higher contribution of plant source (with high C:N), due to elevational gradients of temperature.

DOC ranged from 1.2 to 8.2 mg L<sup>-1</sup>, which is consistent with (Townsend-Small et al. 2005; Aufdenkampe et al. 2007) for comparable elevations (>1000;<4800m). Between 4,000 and 1,350m, DOC showed a decreasing pattern from 13.7 to 1.2 mg L<sup>-1</sup>; the decrease of temperature at paramos leads to a decrease of organic matter mineralization and enhancing organic matter accumulation in soils (Poulenard et al. 2003). Hydraulic variables were not the primary driver controlling DOC, at contrary it was controlled by soil organic carbon inputs. For example, leaching forest soils strongly enhanced fluvial DOC content.

CDOM was characterized with three optical proxies. The  $a_{350}$  and the SUVA<sub>254</sub> values in the two streams at the highest elevation (>3,500 m) were higher than the other mountainous streams at elevations > 1000 m, reflecting the high organic carbon content in the páramos combined to small stream size. The  $a_{350}$  and SUVA<sub>254</sub> values were negatively correlated to upstream catchment slope but not S<sub>R</sub>. Like DOC, the three DOM optical proxies were unrelated to other hydraulic



variables, but related with forest cover. Lower  $a_{350}$  values at similar DOC levels in environments with a low forest cover indicates that DOM released from high-latitude grassland dominated soils is less optically active compared to DOM released from forest soils. High values of  $SUVA_{254}$  at lowlands suggest that a higher forest cover in the lowlands led to a higher quantity in streams of highly aromatic DOM. Conversely, The  $S_R$  values ranged between 0.6 and 1.0 and increased downstream, indicating a stronger enrichment of high-molecular-weight DOM compounds in highlands.

Positive relation between  $pCO_2$  and DOC was observed, which suggests heterotrophy  $CO_2$  production and simultaneous inputs from the catchment of both DOC and  $CO_2$  derived from soils. Conversely, the lack of relation between  $CH_4$  and DOC could be explained by multiple drivers acting simultaneously on  $CH_4$ , in addition to organic matter delivery to streams, that might become the main explaining variable in an analysis at global scale. The fluvial  $\delta^{13}C$ -DOC values ranged between -31.2 and -26.2 ‰ and  $\delta^{13}C$ -POC values ranged between -29.7 and -26.1 ‰ and followed the elevational gradient of  $\delta^{13}C$  of organic carbon from terrestrial vegetation, which reflected the shift from C4 dominated grasslands in highlands to C3 dominated forests in lowlands.

We collected 21 representative plant samples from each area to complement the analysis, maintaining an altitudinal variation between 181 and 3,300 m. A proportional relationship between the  $\delta^{13}C$  content and the altitude was observed. This trend follows the pattern between the  $\delta^{13}C$  POC and the altitude presented previously. Significant differences between average values of  $\delta^{13}C$  in lowland (-28.78 ‰) and Andean-draining rivers (-32.28 ‰) confirm the influence of sources of plant material on the  $\delta^{13}C$  POC signature. The trend could also suggest high dilution of the  $\delta^{13}C$  POC downstream in the main river and depleted POC input to replace the POC that respired in the river. C dynamics reinforced the conclusions reached in Chapter 2 regarding GHG emissions. Clearly, DOC supports  $CO_2$  generation. Although, more research is needed to interpret the relationships with the other GHGs.

## **GHGs dynamics in the lacustrine system**

Lakes at high altitudes have a particular dynamic concerning GHGs due to the influence of organic matter stored in the soil. However, low temperatures could counteract this effect. In the frame of the present thesis, we reported the dissolved concentrations of  $CO_2$ ,  $CH_4$ , and  $N_2O$  in 15 lakes located in the northern region of the Ecuadorian Andes along an elevational gradient from 2,213 to 4,361 m of altitude, as well as a gradient of lake surface area (0.003 to 6.1 km<sup>2</sup>) and depth (0.9 to 74 m). Lakes were grouped in clusters as a function of several variables, such as elevation, the lake area, lake depth, conductivity, and water temperature. Lakes Cuicocha, Loreto, and Papallacta are "volcanic lakes" with high conductivity and TA. Yaguarcocha and San Pablo "urban lakes" are surrounded by the cities of Otavalo and Ibarra and present medium values of TA and the highest surface area. The rest, "páramo" lakes, are of glacial origin, shallows, and located in the Andean paramo above 3000 m altitude.

Water temperature ranged between 10.2 and 15.7°C and followed the elevational pattern. Conductivity was negatively related to elevation for the non-"volcanic" lakes, possibly reflecting the combination of decreasing lake size and increasing soil organic matter content with elevation. In the Andean páramo, the soil organic carbon increases with elevation because of the decrease in temperature that leads to a decrease of organic matter mineralization in soils, enhancing organic matter accumulation in soils.

The DOC ranged between 1.2 and 12.8 mg L<sup>-1</sup>. The three "volcanic" lakes showed the lowest DOC concentrations (1.2 to 2.3 mg L<sup>-1</sup>), the two "urban" lakes (San Pablo and Yaguarcocha) showed intermediate DOC concentrations (4.7 to 5.4 mg L<sup>-1</sup>). The remaining ten lakes were

characterized by the highest DOC content (8.3 to 12.8 mg L<sup>-1</sup>). DOC was positively correlated to elevation and negatively related to temperature (excluding the three “volcanic” lakes). This might reflect the higher inputs of soil DOC in lakes from the páramos. The negative relation between DOC and conductivity is consistent with the idea that soils richer in organic matter tend to lead to less mineralized water. Low inputs of soil DOC in the three “volcanic” lakes are related to a low ratio of catchment area to lake surface area, which reduces the inputs from soils to the lake.

The POC values ranged between 0.4 and 5.7 mg L<sup>-1</sup>, and the POC data points were less clearly clustered than DOC. POC only correlated positively with conductivity, possibly due to a general dilution pattern by meteoric water of solutes and POC. One of the two “urban” lakes (San Pablo) was characterized by the highest POC value (5.7 mg L<sup>-1</sup>), while the three “volcanic” lakes were characterized by among the lowest POC values (0.4 to 0.6 mg L<sup>-1</sup>).

The pCO<sub>2</sub> values ranged between 208 and 10,069 ppm. The three “volcanic” lakes were characterized by higher pCO<sub>2</sub> values (3,269-10,069 ppm), the two “urban” lakes (San Pablo and Yaguarcocha) by the lowest pCO<sub>2</sub> values (208-254 ppm) and the other páramo lakes by intermediary pCO<sub>2</sub> values (644-2,152 ppm). Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol L<sup>-1</sup>. Higher methane production is characterized by the smaller lakes bordered by soils with higher organic carbon content in lake sediments or surrounding soils CH<sub>4</sub> follows the patterns of pCO<sub>2</sub>, and was not significantly correlated to elevation, water temperature, and POC but was negatively correlated to the lake area, lake depth, conductivity, DOC, and %O<sub>2</sub>.

Variations of DIN ranged from 3.6 to 11.0 μmol L<sup>-1</sup>. DIN was dominated by NH<sub>4</sub><sup>+</sup> (70-98%) that ranged between 3.4 and 9.2 μmol L<sup>-1</sup>. NH<sub>4</sub><sup>+</sup> did not significantly correlate with elevation, the lake area, depth, conductivity, and water temperature. NO<sub>3</sub><sup>-</sup> ranged between 0.1 and 1.9 μmol L<sup>-1</sup> and did not significantly correlate with elevation and temperature but positively correlated to the lake area, depth, and conductivity. %N<sub>2</sub>O ranged between 63.8 and 100.8%. All the sampled lakes were sinks of atmospheric N<sub>2</sub>O except Salve Faccha, which was close to saturation (100.8%). This implies the removal of N<sub>2</sub>O by denitrification due to the complete reduction of N<sub>2</sub>O to N<sub>2</sub>. %N<sub>2</sub>O did not significantly correlate to elevation, lake depth, conductivity, and temperature but correlated to the lake area.

Considering the global estimations of the whole basin, the CH<sub>4</sub> emission (0.0007 TgC yr<sup>-1</sup>) from highland lakes (elevation >500m) was 18 times lower than from lowland lakes (elevation <500m) (0.0130 TgC yr<sup>-1</sup>). The average areal FCH<sub>4</sub> (per m<sup>2</sup>) was, in fact, 1.3 times higher in highland lakes (0.75 gC m<sup>-2</sup> yr<sup>-1</sup>) than in lowland lakes (0.58 gC m<sup>-2</sup> yr<sup>-1</sup>) owing to the combination of higher dissolved CH<sub>4</sub> concentration (265 nmol L<sup>-1</sup> versus 179 nmol L<sup>-1</sup>) and higher k values (3.5 cm h<sup>-1</sup> vs. 2.5 cm h<sup>-1</sup>). As a result, the CO<sub>2</sub> emission (0.022 TgC yr<sup>-1</sup>) from highland lakes was 257 times lower than from lowland lakes (5.7 TgC yr<sup>-1</sup>). The average areal FCO<sub>2</sub> was also 11 times lower in highlands (23 gC m<sup>-2</sup> yr<sup>-1</sup>) than in lowlands (255 gC m<sup>-2</sup> yr<sup>-1</sup>) owing to the lower pCO<sub>2</sub> values (530 ppm versus 2,909 ppm) despite higher k values (3.5 cm h<sup>-1</sup> versus 2.5 cm h<sup>-1</sup>).

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# ANNEXES

## Highlights on the field.

**FIGURE AN 1. SHOT TAKEN FROM THE PÁRAMO SALVE FACCHA DAM. THE SALVE FACCHA RESERVOIR IS A VITAL CATCHMENT POINT FOR THE DRINKING WATER SUPPLY OF THE BIGGEST CITY IN ECUADOR. IT IS AT 3,900 M ALTITUDE IN AN ANCIENT VOLCANIC CALDERA. IT PRESENTS HYDROTHERMAL SPRINGS THAT MAY INFLUENCE ITS CHEMISTRY. FIDEICOMISO FONDO PARA LA PROTECCIÓN DEL AGUA (FONAG) FACILITATED ADMISSION TO THE ZONE FOR SAMPLING.**



**FIGURE AN 2 PHOTOGRAPH TAKEN FROM THE BASE OF THE SALVE FACCHA RESERVOIR, THE TEMPERATURE IS 8°C AND RELATIVE HUMIDITY OF 90%. THE COVERTURE IS DOMINATED BY PAJONALES, THESE ECOSYSTEMS LOSE AROUND 51% LESS WATER TO THE ATMOSPHERE THAN OTHER VEGETATION COVER TYPES, WHICH IS CRUCIAL FOR CAPTURING AND CONSERVING WATER.**



**FIGURE AN 3** PAPALLACTA IS A STREAM BORN IN THE ANDEAN PÁRAMOS AND ENDS IN THE PROVINCE OF NAPO. IT IS LOCATED AT AN ALTITUDE OF 2,300 M ALTITUDE. ITS MOST IMPORTANT TRIBUTARIES ARE THE CHALPI AND THE CUYUJA. DOWNSTREAM IT TAKES THE NAME OF QUIJOS.



**FIGURE AN 4.** PHOTOS OF THE SAME POINT, ON THE RIGHT, THE VEGETATION OF QUIJOS AT 2000 M, HIGHER HEIGHT OF THE PLANTS IS OBSERVED DUE TO THE WARMER AVERAGE TEMPERATURE. SAMPLING IS FROM A BRIDGE ON THE LEFT DUE TO DIFFICULTY ACCESSING THE POINT.



**FIGURE AN 5 CAPTURE OF SATELLITE IMAGERY. THE STATION EL SALADO, SAMPLED FROM THE BRIDGE AND THE STATION COCA CODO, SAMPLED IN THE RIVER AFTER THE HYDROPOWER DAM. MOREOVER, THE PICTURE SHOWS THE UNION OF EL SALADO AND COCA RIVER.**



**FIGURE AN 6 SAMPLING POINT IN COCA RIVER AT 600 M ALTITUDE.**



**FIGURE AN 7** AFTER BREAKING THE MOUNTAIN RANGE OF THE ECUADORIAN ANDES, IT ENTERS THE ENORMOUS AMAZONIAN PLAIN, PASSING SOMETHING TO THE SOUTH OF THE CITY OF TENA, THE CAPITAL OF THE NAPO PROVINCE, AT 500 M ALTITUDE. AT THE BACK OF THE PICTURE, THE PÁRAMOS AND EASTERN FOOTHILLS OF THE LLANGANATES NATIONAL PARK.



**FIGURE AN 8** NAPO UPSTREAM OF THE CONFLUENCE WITH COCA. IN PUERTO MISAHUALLI, BOATS ARE THE PRIMARY MEANS OF TRANSPORTATION FOR THE COMMUNITIES SCATTERED ON THE BANKS OF THE RIVER.



**FIGURE AN 9 SUNSET IN THE NAPO RIVER TAKEN FROM NUEVO ROCAFUERTE THE EASTERNMOST TOWN OF ECUADOR A FEW METERS FROM THE BORDER WITH PERU.**



**FIGURE AN 10 LAKE SAN PABLO, SURROUNDED BY THE COMMUNITY OF OTAVALO. AT THE BACK OF THE PICTURE, IT IS THE IMBABURA VOLCANO. IT PRESENTS AN IRREGULAR FIGURE OF 3.5 X 2.2 KM WITH A WATER MIRROR OF APPROXIMATELY 583 HECTARES WITH AN APPROXIMATE DEPTH OF 83 M.**



**FIGURE AN 11 WETLAND AT ANTISANA NATURAL RESERVE. THE WETLANDS AT PÁRAMOS ACT AS A SPONGE TO "CAPTURE" WATER. THEY ARE CRUCIAL FOR THE EXISTENCE OF ECOSYSTEMS SUCH AS FAUNA AND FLORA AND EVEN FOR HUMAN WELL-BEING SINCE CARBON CAPTURE IS LINKED TO THE CONSERVATION OF PÁRAMOS AND WETLANDS.**



**FIGURE AN 12. SECAS LAKE IN THE ANTISANA PARAMO, IT WAS FORMED WHEN THE ANTISANA VOLCANO ERUPTED MORE THAN 400 YEARS AGO.**



FIGURE AN 13 THE TEAM, SEBASTIÁN PINTADO, BRYAN MOROCHO, JOHNNY CHÁVEZ AND GONZALO CHIRIBOGA SAILING FROM MISAHUALLI.



## Greenhouse gas (CO<sub>2</sub>, CH<sub>4</sub>, N<sub>2</sub>O) emissions from Ecuadorian mountainous streams and high elevation lakes

Rivers and streams contribute substantially to global emissions of carbon dioxide (CO<sub>2</sub>) and methane (CH<sub>4</sub>). Most riverine GHG emissions are from the tropics, ~60-80% for CO<sub>2</sub>, ~70% for CH<sub>4</sub>, and ~79% for nitrous oxide (N<sub>2</sub>O). Nevertheless, the total stream surface area (SSA) in the tropics (24°N-24°S) corresponds to ~44% of the total SSA globally, implying that riverine areal emissions of GHGs (per m<sup>2</sup>) are higher than in other climatic zones. Ecuador localizes on the tropics, and it is the smallest of the 17 megadiverse countries in the world harboring an astounding number of ecosystems, which offers the conditions to study in a relatively short space the GHG emissions of both, an important river tributary to the Amazon, as well as a diverse lacustrine system. The present thesis is divided into two sections. Firstly, we report for the first time the dissolved concentrations CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O in Andean headwater and piedmont streams in the Napo River basin in Ecuador, part of the Amazon River catchment. We found that concentrations increased exponentially with elevation decrease, i.e., between 3,990 and 175 m above sea level. Conversely, Andean mountainous headwater and piedmont streams are hotspots of CO<sub>2</sub> and CH<sub>4</sub> emissions; the respective areal fluxes are 1.7 and 4.5 higher in headwater streams and 1.2 and 6.6 higher in piedmont streams than in lowland streams.

Secondly, report the dissolved GHGs of 15 lakes along an elevational gradient ranging from 2,213 to 4,361 m. Lakes were grouped into three clusters as a function of their characteristics and obtained outcomes. Most lakes showed lower values of the partial pressure of CO<sub>2</sub> (pCO<sub>2</sub>) (644-2,152 ppm) than usually attributed to tropical lakes (~1,900 ppm). Three lakes, influenced by volcanic inputs, denoted higher pCO<sub>2</sub> values (3,269-10,069 ppm), while two lakes bordered by large cities showed the lowest pCO<sub>2</sub> values (208-254 ppm), the remaining ten lakes had moderated pCO<sub>2</sub> and the highest dissolved organic carbon concentrations (8.3 to 12.8 mg L<sup>-1</sup>) that is characteristic of páramo lakes (644-2,152 ppm). Dissolved CH<sub>4</sub> concentrations ranged between 170 and 24,908 nmol L<sup>-1</sup> and were negatively correlated to lake area and depth. N<sub>2</sub>O saturation levels ranged between 64 % and 101 %. The surface waters were under-saturated in N<sub>2</sub>O regarding atmospheric equilibrium, probably due to soil-water inputs with low N<sub>2</sub>O levels due to soil denitrification. By upscaling to lakes in the whole Amazon basin, we found that the average areal FCH<sub>4</sub> (per m<sup>2</sup>) was 1.3 times higher in highland lakes (0.75 gC m<sup>-2</sup> yr<sup>-1</sup>) than in lowland lakes (0.58 gC m<sup>-2</sup> yr<sup>-1</sup>) owing to the combination of higher dissolved CH<sub>4</sub> concentration (265 nmol L<sup>-1</sup> versus 179 nmol L<sup>-1</sup>) and higher gas transport velocity (k) values (3.5 cm h<sup>-1</sup> versus 2.5 cm h<sup>-1</sup>). Meanwhile, the average areal FCO<sub>2</sub> was 11 times lower in the highlands (23 gC m<sup>-2</sup> yr<sup>-1</sup>) than in the lowlands (255 gC m<sup>-2</sup> yr<sup>-1</sup>) owing to the lower pCO<sub>2</sub> values (530 ppm versus 2,909 ppm).