

CO<sub>2</sub>-rich mineral groundwaters from East Belgium

> Geological origin and where can we find them ?

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Inspiring Groundwater







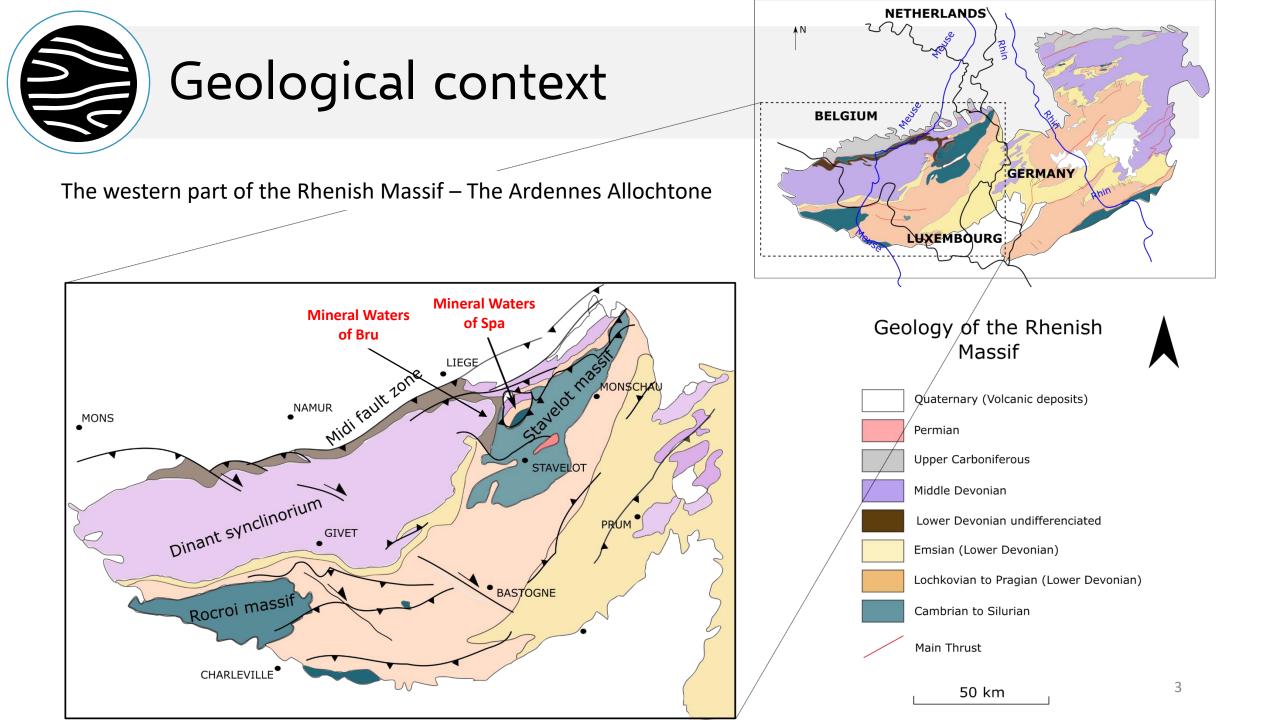
- Naturally enriched in dissolved CO<sub>2</sub> (2-5 g/l)
- Low mineralization (TDS 80 mg/l to 160 mg/l)
- Rich in iron (1-2 mg/l)

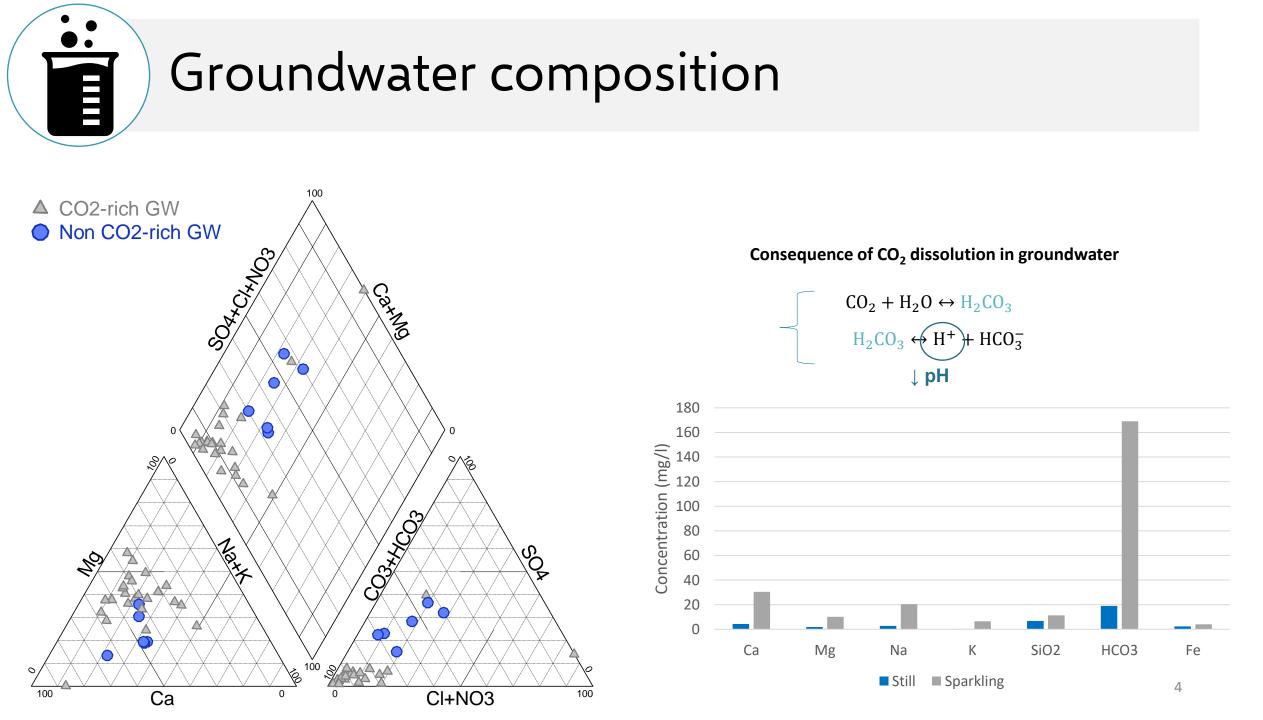


Where does the dissolved gaz come from ?

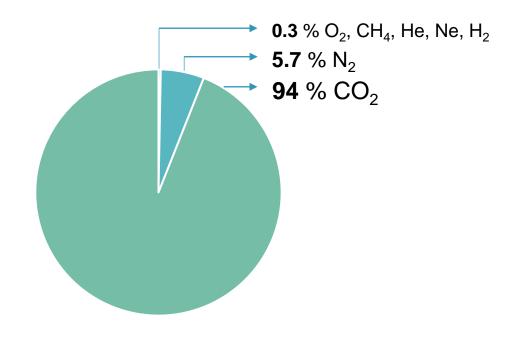


What can be a suitable method to efficiently prospect thoses waters ?





## Question 1. Where does the gas come from ?



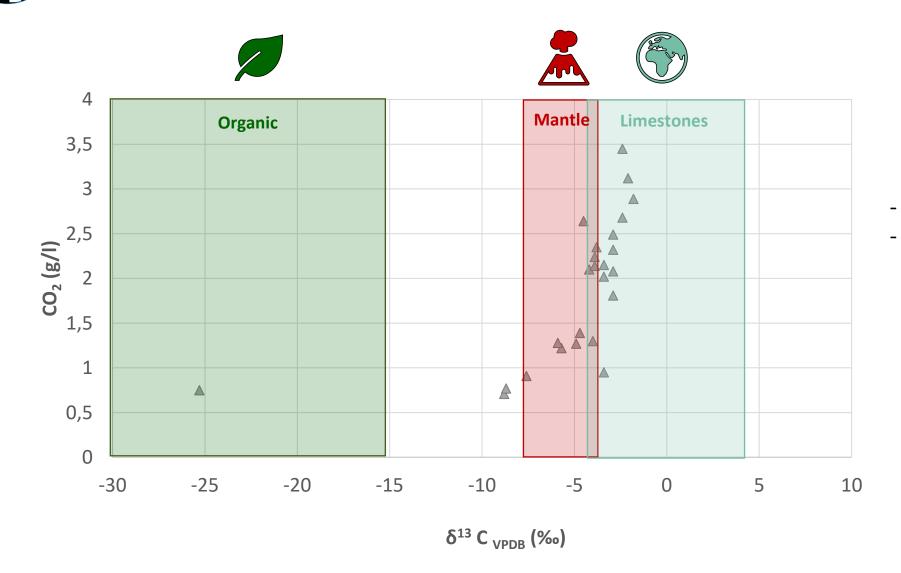
Usually, 3 main origins are considered







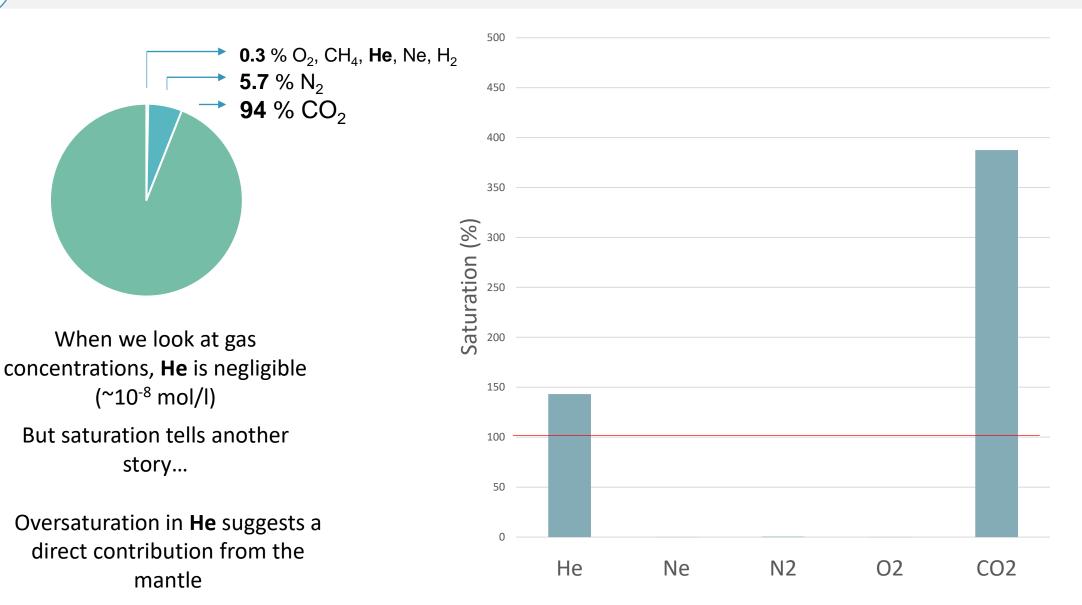
## Isotopic analysis of the dissolved gases



- Clearly not an organic origin
- But the distinction between mantle and limestones is tricky, due to degassing and potential isotopic fractionation

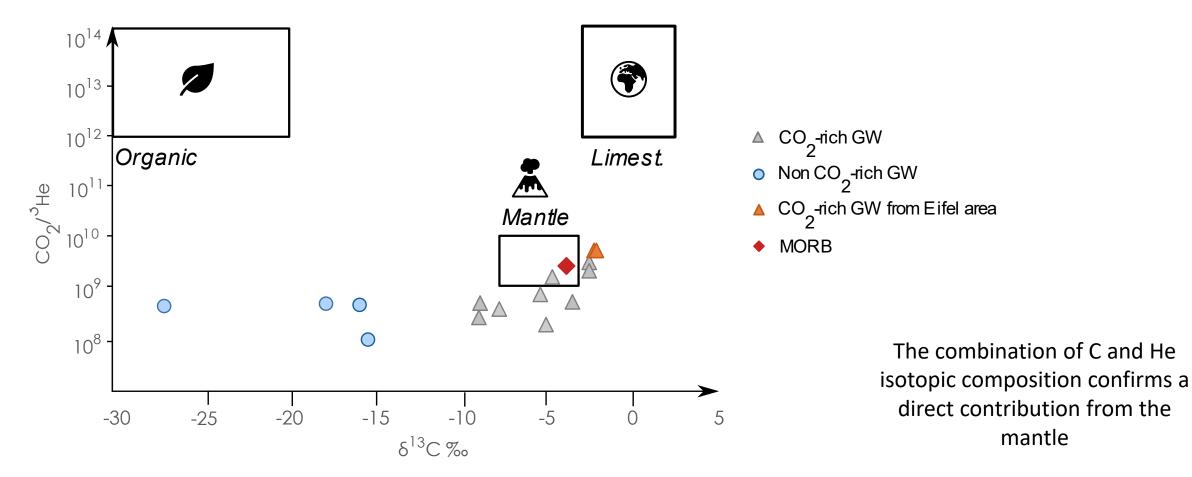
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## Isotopic analysis of the dissolved gases



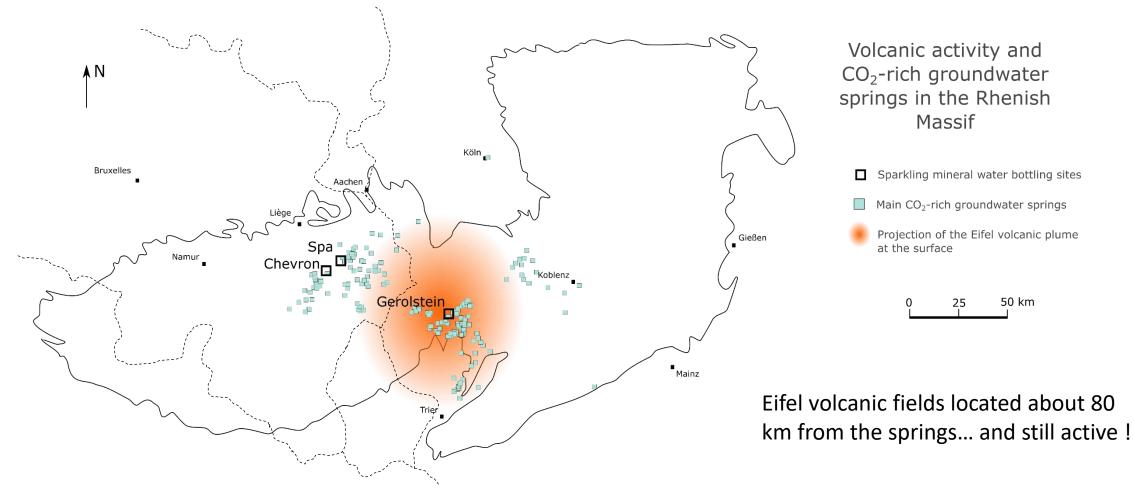


## Isotopic analysis of the dissolved gases





## From the mantle ?



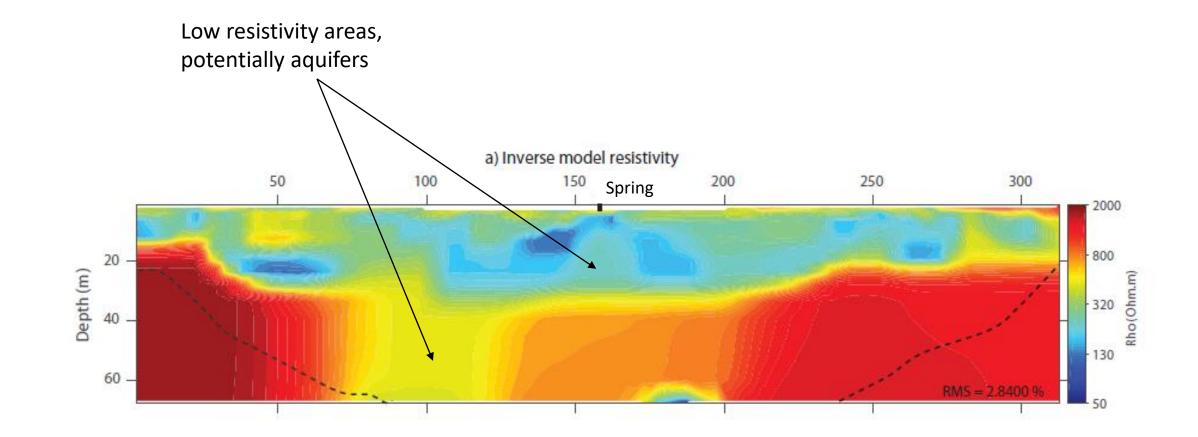
Bräur et al. 2013 - Indications for the existence of different magmatic reservoirs beneath the Eifel area (Germany): A multi-isotope (C,N,He,Ne,Ar) approach Kreemer et al. 2020 – Geodetic evidence for a buoyant mantle plume beneath the Eifel volcanic area, NW Europe

## Question 2. Where can we drill to find CO<sub>2</sub>-rich groundwater?

From hydrochemistry to geophysics...



ERT measurements



This might be water-saturated zones. But is it sparkling water ?

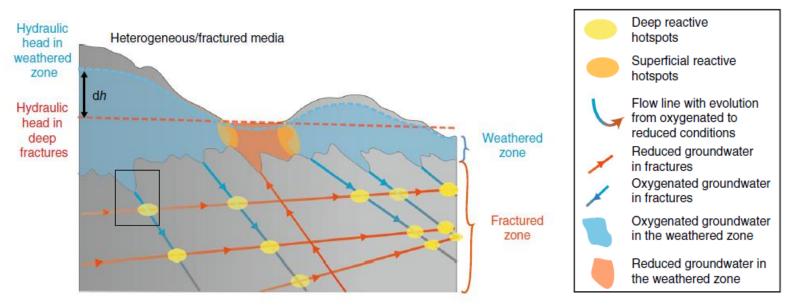
### **IP** measurements



Iron hydroxydes

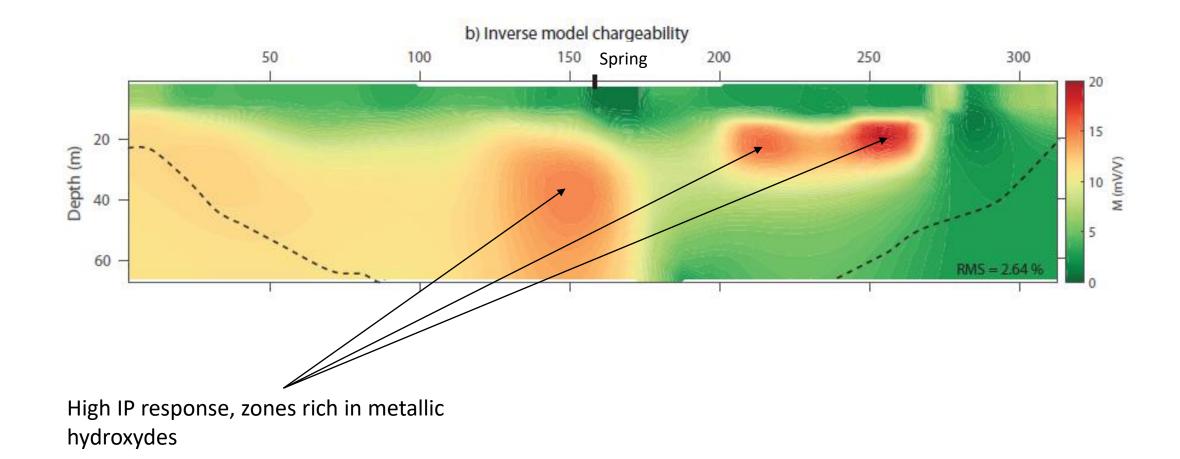
CO<sub>2</sub>-rich groundwaters characterized by the presence of **iron hydroxydes hot spots** formed by intermittent oxic–anoxic fluid mixing in fractured rocks.

Goal : highlight their presence thanks to the **polarizable** character of theses elements.

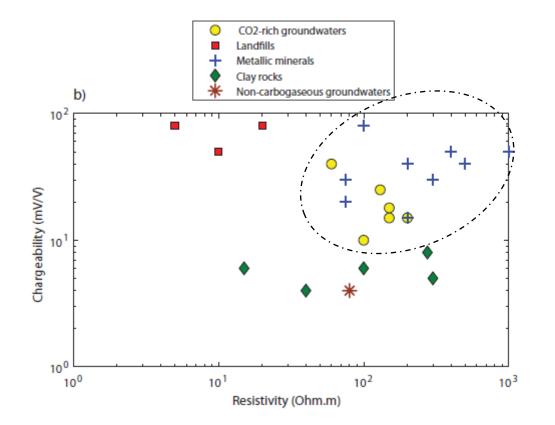


Bochet et al. (2021) – Iron-oxidizers hotspots formed by the intermittent oxic-anoxic fluid mixing in fractured rocks. *Nature Geoscience* 

IP measurements



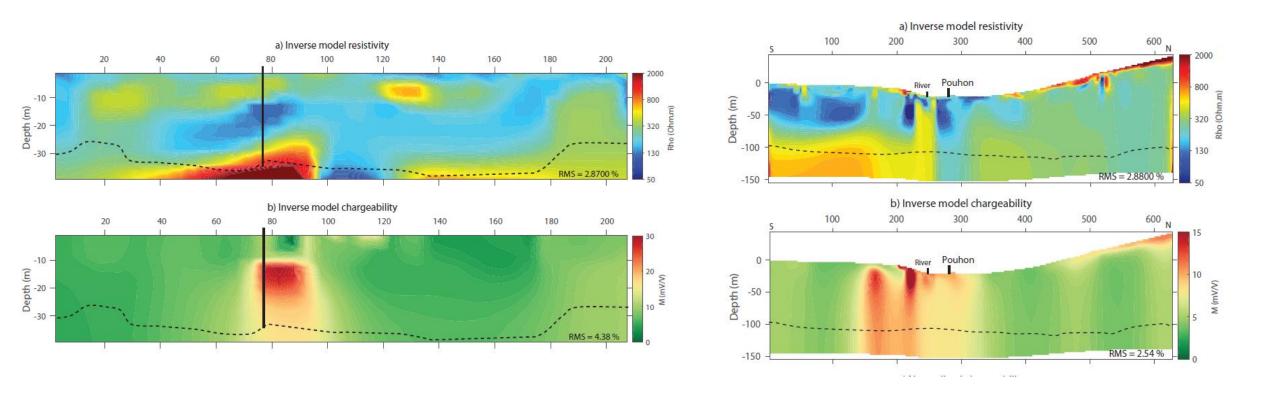
### Comparison with the litterature



Resistivity and chargeability values measured at the springs location are similar to the one commonly accepted for metallic minerals in the literature...

Which tends to confirm the hypothesis of iron hydroxides precipitation at shallow depths as a signature of the presence of CO2-rich groundwater.

More details in : Defourny, A.; Nguyen, F.; Collignon, A.; Jobé, P.; Dassargues, A.; Kremer, T. Induced Polarization as a Proxy for CO<sub>2</sub>-Rich Groundwater Detection—Evidences from the Ardennes, South-East of Belgium. *Water* 2020, *12*, 1394. https://doi.org/10.3390/w12051394 Other results



#### Question is:

Can we find a similar signature at other locations where there is no spring? Or does this signature exist at that precise location because of a contact with oxygen ?

Defourny et al. (2020). Induced Polarization as a Proxy for CO2-Rich Groundwater Detection—Evidences from the Ardennes, South-East of Belgium. Water



# Conclusions and further research

#### Natural mineral waters rich in dissolved CO<sub>2</sub>

Where does the CO<sub>2</sub> comes from ?



Combination of  $\delta^{13}$ C and <sup>3</sup>He isotopic values



Where can we find them ?



Combination of ERT and IP measurements

## Thank you for your kind attention...

