

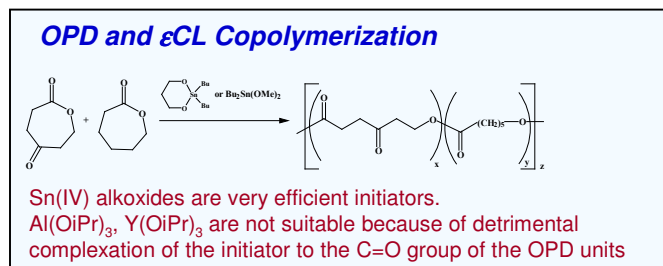
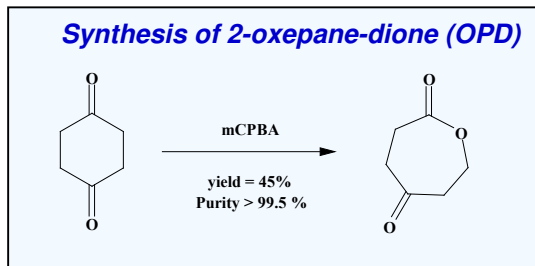


2-Oxepane-1,5-dione: A Precursor of a Novel Class of Versatile Semicrystalline Biodegradable (Co)polyesters.

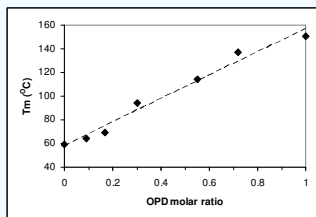
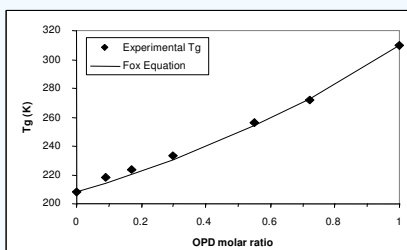
Jean-Pierre Latere Dwan'Isa, **Philippe Lecomte**, Philippe Dubois¹, Robert Jérôme
Center for Education and Research on Macromolecules (CERM), University of Liège, B6
Sart-Tilman, B 4000 Liège, Belgium: <http://www.ulg.ac.be/cerm>



¹Current address: University of Mons-Hainaut, Place du Parc, 20, B-7000 Mons, Belgium



Thermal properties : P(OPD-co-εCL), a semicrystalline copolymer

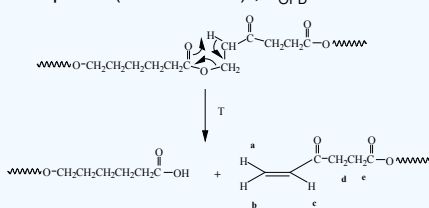
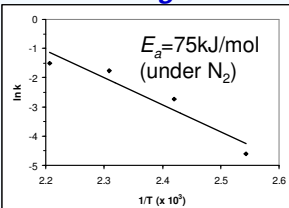


Cell parameters (Å)	PCL	P(OPD)
A	7.49	6.80
B	4.98	5.18
C	17.05	17.17
ε	6.2	6.2
δ	2.98	2.82

Regular and continuous increasing T_m PCL and P(OPD) units are isomorphous

P(OPD-co-εCL) containing 30 mol% OPD exhibits a $T_m = 90^\circ\text{C}$, which opens up new opportunities for applications that require a higher service temperature than PCL, i. e., packaging applications

Thermal degradation : sample : P(OPD-co-εCL) ; $F_{\text{OPD}}=0.3$



k (%/min) determined by Isothermal TGA

Degradation of P(OPD-co-εCL) is faster than PCL ($E_a=92\text{kJ/mol}$) because of the ketone group of the OPD units

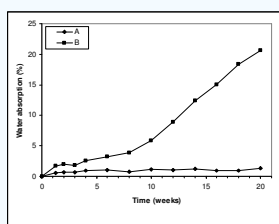
P(OPD-co-εCL) containing 30 mol% of OPD is stable up to 140°C under nitrogen for one hour, such that processing by extrusion is possible.

Mechanical properties

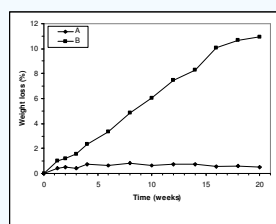
Copolymer	Mn (K)	ϵ_b (%)	σ_b (MPa)	σ_y (MPa)
P(OPD-co-OPD), $F_{\text{OPD}}=0.33$	42	800	26	15
CAPA 650	50	1000	36	13
Impact energy (kJ/m ²)				
P(OPD-co-OPD), $F_{\text{OPD}}=0.34$	33K	18-20		
P(OPD-co-OPD), $F_{\text{OPD}}=0.33$	42K	No breaking		
P(OPD-co-OPD), $F_{\text{OPD}}=0.33$	50K	No breaking		

Above a critical molecular weight, the tensile and impact properties are comparable to those ones of commercially available PCL (CAPA, Solvay)

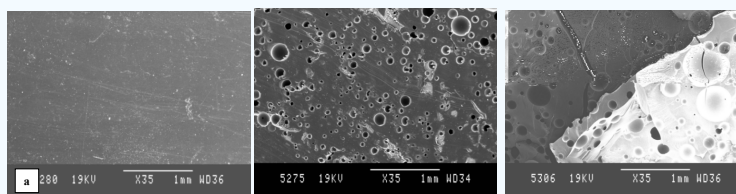
Hydrolytic degradation Conditions : pH=7.4 at 37°C ; P(OPD-co-εCL) (Mn=35000 ; 30 mol% of OPD)



A = PCL ; B = P(OPD-co-εCL)



SEM image after immersion



After 0 days

After 9 days

After 20 weeks

The C=O group imparts higher hydrophilicity, which has a key role in the increase of the hydrolytic degradation rate compared to PCL

Crosslinking under UV Exposure of P(OPD-co-εCL) ($F_{\text{OPD}}=0.3$) to UV induces loss of solubility. Frequency sweep experiments show higher elasticity above T_m . An increased brittleness was shown by tensile properties. No significant effect on T_m and ΔH_m was observed. The C=O group of the OPD units is at the origin of this behavior.