Original network of zig-zag chains in the β polymorph of Fe₂WO₆: crystal structure and magnetic ordering

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The structural and physical properties of the β polymorph of iron tungstate Fe₂WO₆ have been investigated by synchrotron and neutron diffraction *vs* temperature, combined with magnetization and dielectric properties measurements. The monoclinic *P*2₁/*a* crystal structure of β -Fe₂WO₆ has been determined and consists in an original network of zig-zag chains of FeO₆ and WO₆ octahedra sharing trans- and skew-edges, connected through corners into a 3D structure. Magnetization measurements indicate an antiferromagnetic transition at T_N = 264 K, which corresponds to a $\uparrow \uparrow \downarrow \downarrow \downarrow$ nearly collinear ordering of iron moments inside sequences of four edge-sharing FeO₆ octahedra as determined by neutron diffraction. A canting of the moments out of the *ac* plane is observed below 150 K, leading to a non-collinear antiferromagnetic structure, the *P*2₁/*a*' magnetic space group remaining unchanged. These results are discussed in comparison with the crystal and magnetic structures of γ -Fe₂WO₆ and with the magnetic couplings in other iron tungstates and trirutile Fe₂TeO₆.

Introduction

The iron-tungsten-oxygen system has recently been the subject of considerable interest, particularly in the search for iron-based oxide semiconductors for photoelectrochemical applications.^{1–3} Within this family, Fe₂WO₆ (in which iron is trivalent) was for instance investigated as a photocatalyst to achieve the degradation of organic pollutants in wastewater or the selective reduction of NO_x, or as a photoanode for water oxidation.^{4–6} Another interesting feature of Fe₂WO₆ is the existence of three polymorphs. Most of the literature concerns the so-called γ -Fe₂WO₆, synthesized by Kozmanov⁷ in 1957. Senegas and Galy⁸ described the orthorhombic structure of this γ polymorph as a tri- α -PbO₂ cell, which is typically obtained by

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: https://dx.doi.org/10.1021/acs.inorgchem.0c01024 solid-state reaction at 900°C or above. Its magnetic ordering is characterized by the existence of a weak ferromagnetic component^{9,10} and magnetodielectric properties were recently investigated.¹¹ Parant et al.¹² showed that a second polymorph, called α -Fe₂WO₆, exists at temperatures as low as 800°C. This low-temperature phase was described as a columbite structure, another superlattice variant of the α -PbO₂ cell. Literature on the physical properties of this polymorph is scarce.^{13,14} In 1992, Walczak et al.¹⁵ obtained a third polymorph, called β -Fe₂WO₆, through long heat treatment between 750 and 840 °C. Its monoclinic structure was not solved at the time. The only published investigation of the physical properties of β -Fe₂WO₆, conducted by Guskos et al¹³ using EPR and bulk magnetic measurements, reported a magnetic transition slightly below room temperature, around 260 K, similar to the α - and γ - polymorphs. To this day, experimental data dealing with the crystal structure and the macroscopic properties of β -Fe₂WO₆ have remained largely unavailable, most likely because the synthesis of β -Fe₂WO₆ requires very long heat treatment and singlephased material cannot be obtained easily.¹⁵

In this article, we report on the synthesis of single-phase β -Fe₂WO₆ and its characterization by high angular-resolution synchrotron X-ray and neutron powder diffraction and measurement of bulk magnetic and dielectric properties. An original crystal structure is evidenced, in which the alternation of trans- and skew-edge sharing octahedra creates sequences of four edge-sharing FeO₆ octahedra, with a nearly collinear up-up-down-down magnetic configuration below 264 K. Below 150 K, a reorientation of the moments outside the *ac* plane leads to non-collinear antiferromagnetic ordering with a small component along the *b* axis. The discussion involves a comparison between the crystal and magnetic structures of the different Fe₂WO₆ polymorphs and considers the magnetic couplings in other motifs of edge-sharing FeO₆ octahedra such as the dimers in tri-rutile Fe₂TeO₆.

Experimental Section

Synthesis and preliminary characterizations

Polycrystalline samples of β -Fe₂WO₆ were prepared by solid-state reaction between α -Fe₂O₃ and WO₃. Commercial iron oxide (α -Fe₂O₃, 99.9%) and tungsten trioxide (WO₃, 99.8%) were both purchased from Alfa-Aesar. The precursor powders were weighed in a Fe/W = 2 ratio and mixed in an agate mortar. The powder mixture was then pressed with a uniaxial press in the shape of bars (2 x 2 x 12 mm). During preliminary experiments, trial preparation conditions taken from Walczak et al.¹⁵ were adjusted by monitoring the evolution of the synthesis with ex-situ X-ray diffraction. Brownish-black single-phase β -Fe₂WO₆ was obtained by sintering at 850 °C for 21 days in air, with intermediate grinding and pelletizing steps after 7 and 14 days.

Sample quality was checked by room temperature (RT) X-ray powder diffraction (XRPD), using a Panalytical Xpert Pro diffractometer (CoK α radiation). The Fe/W stoichiometry was determined by induced coupled plasma – optical emission spectroscopy (ICP-OES) using an Agilent ICP-OES spectrometer. β -Fe₂WO₆ was dissolved in fuming hydrochloric acid (37wt.%, Acros, for analysis) overnight at 60 °C. Standard solutions of iron and tungsten were prepared by dilution of the respective standard solutions (Merck, traceable to SRM from NIST). The calibration curves were in the 10 – 100 mg L⁻¹ range. The Fe/W ratio of 2.03 ± 0.03 (confidence interval of ± 2 standard deviations) is in good agreement with the Rietveld refinements of synchrotron data at RT indicating a content in hematite of about 0.4 % wt% (1 mol%).

Synchrotron X-ray and neutron powder diffraction

Synchrotron X-ray powder diffraction (SXRPD) experiments were performed on the BL04-MSPD beamline of the ALBA synchrotron.¹⁶ Data were collected using a high intensity mode detection setup (position sensitive detector MYTHEN), at wavelength $\lambda = 0.4427$ Å, with the sample enclosed in a spinning glass capillary (0.3 mm inner diameter) at 295 K and in the low temperature He flow Dynaflow cryostat at 10 K.¹⁷

High-resolution neutron powder diffraction (HRNPD) data were collected at 295 K on the WISH time-of-flight (TOF) diffractometer at the ISIS neutron source.¹⁸ The sample was loaded into a 6 mm diameter cylindrical vanadium can and mounted in a cryostat.

Neutron powder diffraction (NPD) data were collected in the temperature range 1.5 - 300 K at LLB-Orphée on the G4.1 diffractometer ($\lambda = 2.427$ Å). The sample was loaded into a 10 mm diameter cylindrical vanadium can.

Structure refinements

Possible sets of crystal systems and cell parameters were determined using both TREOR¹⁹ and Dicvol²⁰ programs from SXRPD data collected at 295 K. Atomic positions were determined by the charge flipping method using Superflip²¹ with the JANA2006 program.²² Refinements of the crystal and magnetic structures, and symmetry analysis (magnetic irreducible representations calculations) were performed using the FullProf Suite.²³ HRNPD data collected on banks 2–9 were included in the Rietveld calculations. A convolution pseudo-Voigt with back-to-back exponential functions was employed to model the peak shapes. The scattering angles for the different detector banks are as follows: banks 2/9, 58°; banks 3/8, 90°; banks 4/7, 122° and banks 5/6, 153°.

Magnetic and electrical properties

Magnetic characterizations were performed using a SQUID magnetometer (Quantum Design) for zero-field-cooled warming (*zfcw*) and field-cooled warming (*fcw*) measurements from 5 to 300 K in a magnetic field of 100 Oe and for isothermal magnetic field dependent measurements up to 50000 Oe at 5, 25, 50 and 100 K.

For dielectric measurements, a 3.52 x 2.62 x 0.74 mm sample was prepared and silver paste was deposited on the two larger surfaces to attach electric copper wires. The sample was soldered to a home-made sample probe which was inserted in a 14 T-PPMS (Quantum Design). The dielectric permittivity was obtained using an Agilent 4284A LCR meter.

Results

Room temperature structural investigation of β -Fe₂WO₆

The SXRPD pattern of β -Fe₂WO₆ is indexed with a monoclinic unit cell whose space group is $P2_1/a$ (n° 14) with $a \approx 15.092$ Å, $b \approx 4.594$ Å, $c \approx 5.596$ Å and $\beta \approx 98.54^\circ$. The crystallographic parameters obtained by combined refinements using SXRPD and HRNPD data and the corresponding interatomic distances are presented in Tables 1 and 2, respectively. Figure 1a shows RT HRPND experimental and calculated patterns, and the corresponding crystal structure is drawn in Figure 2.

RT				10 K					
	a (Å)		15.09245 (8	3)		a (Å)		15.06343 (4	1)
	b (Å)		4.59451 (2)		b (Å)		4.58614 (1)
	<i>c</i> (Å)		5.59628 (3)		<i>c</i> (Å)		5.58757 (1)
	β (°) 98.5360 (5)			β (°) 98.5791 (2))		
	V (Å ³)		383.761 (1)		V (Å ³)		381.687 (1)
Atom	x	у	z	Biso (Å ²)	Atom	x	У	z	Biso (Å ²)
W	0.5876(1)	-0.004(1)	0.3613(3)	0.22(1)	W	0.5877(1)	0.0058(4)	0.3617(1)	0.20(1)
Fe1	0.0853(2)	0.008(1)	0.8549(6)	0.63(6)	Fe1	0.0857(2)	0.000(2)	0.8551(4)	0.15(3)
Fe2	0.2551(2)	-0.004(2)	0.2292(6)	0.70(7)	Fe2	0.2548(2)	-0.004(2)	0.2284(4)	0.17(3)
01	0.9686(6)	-0.243(2)	0.868(1)	1.05(8)	01	0.970(1)	-0.234(3)	0.862(2)	0.41(8)
O2	0.6890(5)	0.231(2)	0.464(1)	1.05(8)	02	0.6899(9)	0.227(3)	0.447(2)	0.41(8)
O3	0.6328(5)	-0.226(1)	0.665(1)	0.8(2)	O3	0.637(1)	-0.219(3)	0.670(2)	0.41(8)
O4	0.5240(6)	0.219(1)	-0.371(1)	0.9(2)	04	0.526(1)	0.217(3)	-0.383(2)	0.41(8)
05	-0.1365(6)	0.233(2)	0.821(1)	1.05(8)	05	-0.131(1)	0.236(3)	0.841(3)	0.41(8)
O6	0.6945(5)	0.300(1)	0.948(1)	1.05(8)	O6	0.6966(9)	0.295(3)	0.953(3)	0.41(8)

($R_{Bragg} = 3.55$ %) and 10 K ($R_{Bragg} = 4.43$ %), obtained by using simultaneously SXRPD and HRNPD data at RT; and SXRPD data only at 10 K. All atoms stand on general Wyckoff positions (4*e*) of the *P*2₁/*a* space group.

Table 1. Cell parameters and crystallographic parameters of β -Fe₂WO₆ at room temperature

Cation	0	RT	10K	Cation	0	RT	10K
W	01	1.866(8)	1.869(13)	Fe1	01	2.143(9)	2.162(14)
	O2	1.890(8)	1.875(13)		01	2.118(10)	2.049(15)
	O3	2.012(8)	2.057(12)		03	1.887(9)	1.887(14)
	O4	2.152(8)	2.046(14)		O4	1.918(8)	1.967(13)
	O4	1.962(9)	2.029(15)		05	2.168(8)	2.000(13)
	O5	1.810(9)	1.883(13)		O6	1.872(8)	1.933(14)
WO	<d></d>	1.949(3)	1.960(5)	E-10	<d></d>	2.018(4)	2.000(6)
W 06	$\Delta d \ 10^{-4}$	32.9(1)	18.6(1)	FeIO ₆	∆d 10 ⁻⁴	39.6(1)	19.6(1)
Fe1-Fe1 (intra chain)		3.243(7)	3.254(7)	Fe2	O2	2.077(8)	2.136(12)
Fe1-Fe2 (intra chain)		3.060(5)	3.040(4)		O2	2.165(9)	2.089(14)
Fe2-W (intra chain)		3.045(4)	3.044(3)		03	2.008(8)	1.943(14)
W-W (intra chain)		3.259(4)	3.251(2)		05	2.059(10)	2.151(14)
Fe1-Fe2 (inter chains)		3.451(8)	3.429(9)		06	1.907(9)	1.872(16)
Fe2-Fe2 (inter chains)		3.430(5)	3.419(4)		06	1.940(8)	1.910(16)
Fe1-Fe2 (inter chains)		3.372(7)	3.378(9)	E-20	<d></d>	2.026(4)	2.017(6)
Fe1-W (inter chains)		3.516(6)	3.561(6)	ге206	∆d 10 ⁻⁴	18.2(1)	30.8(1)

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> **Table 2.** Selected interatomic distances (in Å) in β -Fe₂WO₆ corresponding to crystallographic parameters of Table 1. <d> is the average bond length (in Å) in MO₆ octahedra (M = W, Fe1, or Fe2) and Δd is the distortion parameter calculated for each MO₆ octahedron as $\Delta d = (\frac{1}{6}) \sum_{n=1,6} \{(d_n - \langle d \rangle)/\langle d \rangle\}^2$, with $\langle d \rangle$ being the average M–O distance.



Figure 1. Rietveld refinements of β -Fe₂WO₆: (a) HRNPD data at 295 K and (b) SXRPD data at 10 K. Experimental, black dots; calculated, red continuous line; Bragg reflections, green vertical marks. The difference between the experimental and calculated profiles is displayed at the bottom of the graph as a blue continuous line.



Figure 2. β -Fe₂WO₆ crystal structure: (a,b) Projections along [010] showing the zigzag patterns in the two types of layers (y \approx 0 and \approx ½). (c,d,e) Corner-sharing octahedra in the three types of *bc*-planes. WO₆ octahedra, blue; Fe1O₆ octahedra, red; Fe2O₆ octahedra, orange.

The three cations (Fe1, Fe2 and W) are in octahedral oxygen environments. The structure can be described as the stacking, along the *b*-axis, of layers made of zig-zag chains of edge-sharing octahedra (Figure 2 a,b). The connectivity between edge-sharing octahedra is conveniently described using the skew/trans nomenclature²⁴ (Figure S1). In each zig-zag chain, the sequence of cations is Fe1-Fe2-W-W-Fe2-Fe1. Fe2O₆ octahedra are sharing trans edges with Fe1O₆ and WO₆ octahedra, thus making a linear fragment of three octahedra. Linear fragments are connected to each other into zig-zag chains by sharing skew edges: WO₆ octahedra share skew edges with WO₆ and Fe2O₆ octahedra while Fe1O₆ octahedra share skew edges with Fe1O₆ and Fe2O₆. The Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> sequence of cations is identical in the first ($y \approx 0$) and second ($y \approx \frac{1}{2}$) layers, the zig-zag chains being translated by a vector ($\frac{1}{2}$, $\frac{1}{2}$, $\frac{1}{2}$) from one layer to the other.

Layers are connected to each other through corner-sharing, as illustrated in Figure 2(c,d,e), leading to three kinds of *bc*-planes. Two of them consist in alternating Fe1O₆ and WO₆ (planes labelled 1 and 3 in Figure 2) and the other one is made of Fe2O₆ only (plane 2 in Figure 2).

All octahedra are distorted (Table 2 and Figure S2), with a noticeable off-centering of Fe and W cations (Table S1); there are three long and three short distances, in the range 2.118-2.168 Å and 1.872-1.918 Å respectively, in Fe1O₆, and in the range 2.059-2.165 Å and 1.907-2.008 Å respectively, in Fe2O₆. The average Fe-O or W-O distances in each octahedron are in agreement with the Shannon ionic radii²⁵ and very similar to those reported for polymorph γ .⁸

Low-temperature structural investigation of β -Fe₂WO₆

SXRPD and NPD data show that β -Fe₂WO₆ does not undergo a structural transition between 350 and 1.5 K, as illustrated in Figure 1b by the 10 K SXRPD pattern. The crystallographic parameters and corresponding interatomic distances at 10 K are presented in Tables 1 and 2, respectively, and display only slight variations compared to RT. With decreasing temperature, the unit cell parameters *a*, *b* and *c* decrease by 0.10 to 0.15 %, while the monoclinic angle β becomes slightly larger (0.03 %), causing an overall reduction in the unit cell volume by 0.4 %, as shown in Figure 3. In a zig-zag chain, W atoms in edge-sharing octahedra units get slightly closer to each other at low temperature (d_{W-W} \approx 3.251 Å at 10 K against 3.259 Å at RT). In contrast, Fe1 atoms in edge-sharing octahedra move away from each other, with Fe1-Fe1 interatomic distance increasing from 3.243 Å at room temperature to 3.254 Å at 10 K.



Figure 3. Temperature dependence of the lattice parameters and cell volume of β -Fe₂WO₆ between 1.5 and 300 K, from NPD data (G4.1). Dots are refined values; error bars are taken as \pm two standard deviations; continuous lines are guides to the eye.

Investigation of the magnetic properties of β -Fe₂WO₆

The magnetic properties of β -Fe₂WO₆ were investigated by a combination of magnetic measurements (Figure 4) and neutron diffraction (Figures 5 and 6). The results are presented and discussed in an integrated manner to analyze the rather complex antiferromagnetic ground state already hinted at in previous reports^{13,14}. The transition temperatures T_N and T* identified in the magnetic susceptibility curves are in good agreement with the ones taken from neutron diffraction data, taking into account that, over the first few degrees below the magnetic transition, the intensity of magnetic origin is too weak for a reliable refinement.

The magnetic susceptibility curves plotted in Figure 4a evidence a weak variation over the temperature range from RT to 5K and display a somewhat unusual behavior. Decreasing temperature from RT, the *zfcw* and *fcw* susceptibility curves are superimposed and increase slowly down to $T_N \approx 264$ K, below which the signal first decreases slightly and then (from 250 K) increases again with a marked split between the *zfcw* and *fcw* curves. The curves feature a small cusp in the *zfcw* data (T* \approx 160 K) with a corresponding change of slope in the *fcw* mode. In

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Figure 4. (a) Magnetic susceptibility curves of β-Fe₂WO₆ in 100 Oe as a function of temperature.
Black lines are guides to the eye to highlight the change of slope. (b) Field-dependent magnetization curves at 5, 25, 50 and 100 K. (c) Magnetic susceptibility taken at 2500, 10000 and 20000 Oe in the virgin M(H) curves shown in (b).

The magnetic transition is also detected in the NPD patterns; an increased intensity is observed below T_N on top of existing Bragg peaks, as illustrated in Figure 5a, around Q = 0.85, 1.3, 1.4 and 1.8 Å⁻¹. The magnetic structure can therefore be described with the propagation vector $\mathbf{k} = (0\ 0\ 0)$. A symmetry analysis performed for the $P2_1/a$ space group with propagation vector $\mathbf{k} = (0\ 0\ 0)$ and Wyckoff site 4e yields four possible magnetic irreducible representations (Irrep) of dimension one, each containing 3 basis vectors: $\Gamma_{mag} = 3\Gamma_1 \oplus 3\Gamma_2 \oplus 3\Gamma_3 \oplus 3\Gamma_4$. A satisfactory refinement (Figure 5c) in the whole temperature range below T_N is achieved using Γ_2 Irrep, whose basis vectors are listed in Table 3.



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Figure 5. (a) Temperature evolution of NPD (G4.1) patterns of β -Fe₂WO₆. (b) Temperature evolution of the components of the Fe magnetic moments. (c) NPD patterns at 300, 150 and 1.5 K. Experimental data, red dots; calculated profile, black continuous line; Bragg reflections, green vertical marks. The difference between the experimental and calculated profiles is displayed at the bottom of the graph as a blue continuous line. The insets at 150 K and 1.5 K illustrate the increasing magnetic intensity on the (301) Bragg position, indicating a non-zero m_y component of the magnetic moment.

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Γ_2	(x, y, z)	(-x+1/2, y+1/2, -z+1/2)	(y+1/2,- x+1/2, z+1/2)	(y, x,-z+1)
ψ_1	100	-1 0 0	-1 0 0	100
ψ_2	010	010	0 -1 0	0 -1 0
ψ_3	001	0 0 -1	0 0 -1	001

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Table 3. Basis functions for axial vectors bound to the Wyckoff site 4*e* of the $P2_1/a$ space group, within the irreducible representation Γ_2 (propagation vector $\mathbf{k} = 0.00$).

Below T_N and down to T*, the best magnetic structure modelling corresponds to magnetic moments on the Fe1 and Fe2 sites confined in the *ac*-plane. In each zig-zag chain, the Fe spins in the *ac*-plane follow a $\uparrow \uparrow \downarrow \downarrow$ sequence, each tetramer being separated from the following one by two diamagnetic hexavalent tungsten cations (Figure 6a). The coupling between Fe spins is ferromagnetic for the short Fe1-Fe2 distance (3.040 Å) and antiferromagnetic for the long Fe1-Fe1 distance (3.254 Å). Fe1 and Fe2 are not actually constrained by symmetry to have collinear moments, refining them independently lead to the m_x and m_z values given in Figure 5b and Table 4. The observed small misalignment of the spins (a few degrees) is questionable since the refinement is not significantly improved compared to the one including constraints between Fe1 and Fe2 moments. However, the evolution of the magnetic susceptibility below T_N (see Figure 4) supports the canting of the AFM structure.

Below T*, the appearance of a peak at the (301) Bragg position (Figure 5c) indicates the onset of an out-of-plane magnetic component m_y (Figures 5b and Table 4). In contrast with the in-*ac*plane magnetic order observed at and above T*, refining the components m_y independently on the two Fe sites led to a much better agreement factor; m_x and m_z were also refined independently for coherence (Figure 5b). The resulting magnetic order is therefore non collinear with an antiferromagnetic m_y component highlighted in Figure 6b. This reorientation of the spins below T* is accompanied by an increase in the calculated moment value which reaches about 4 μ_B for Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> both sites at 1.5 K, noticeably lower than the 5 μ_B value expected for Fe(III), but much higher than the moments of 2.55 and 2.15 μ_B reported in *γ*-Fe₂WO₆.¹⁰

In summary, the picture of magnetism emerging from the neutron diffraction study and the magnetic curves is as follows: (i) below T_N , there appears an antiferromagnetic structure which is nearly collinear with spins in the ac-plane; (ii) the small difference in the spin orientation of iron in the two sites (i.e. canting) and their different and low values compared to the theoretical one are in agreement with the splitting of the *zfcw* & *fcw* curves; (iii) the cusp observed on the *zfcw* susceptibility curve around T* corresponds to the spin reorientation inducing a new component along the *b*-axis. Since there is no clear change in the NPD data around 60K (such as an evolution of the background at low angle or in the width of the Bragg peaks in their lower part), the downturn in the susceptibility could be related to a small ferromagnetic-like component related to the evolution of the slightly different temperature dependences of the magnetic components of the two iron cations (Figure 5b).



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Figure 6. (a) Magnetic structure of β-Fe₂WO₆ at 150 K: the Fe1 and Fe2 moments are confined in the ac planes from T_N to T*. Moments shown in octahedra belong to the y ≈ 0 layer while moments shown as isolated arrows belong to the y ≈ ½ layer. (b) Arrangement of the m_y component below T* (positive (+) and negative (-) signs indicate a positive or a negative m_y component, respectively); this small m_y component is in addition to the magnetic structure given in (a). W, blue; Fe1, red; Fe2, orange.

	150) K	1.5 K		
	Fe1 Fe2		Fe1	Fe2	
mx	2.4(2)	2.7(2)	3.2(2)	3.2(2)	
my	0	0	-0.8(2)	0.4(3)	
mz	2.1(1)	2.0(3)	2.9(1)	2.4(2)	
m tot	2.9(1)	3.1(1)	4.1(2)	3.7(3)	

Table 4. Magnitude (in μ_B) of the components of the magnetic moment on Fe1 and Fe2 at 150 K and 1.5 K.

Irrep Γ_2 corresponds to the Shubnikov group $P2_1/a'$ (BNS #14.78), which allows a magnetoelectric (direct effect) tensor. Coherently with the rather low synthesis temperature, leading to a sintering of low quality, the measurements of the dielectric permittivity as a function of temperature (Figure 7) reveal a leaky behavior, characterized by losses (tan δ) which increase rapidly as the temperature increases above ≈ 120 K and shift towards higher temperatures as the frequency increases. This feature unfortunately hinders the electrical polarization measurements. The same problem of sinterability also precluded measurement of the specific heat capacity, which might have provided valuable insight into the nature of the transitions. However, in the case of a

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> sample of γ polymorph sintered at 950°C, the C_p(T) curve reported by Panja et al.¹¹ between 150

K and RT revealed only a very weak signature in the vicinity of the high temperature transition.



Figure 7. Dielectric permittivity (ϵ ') of β -Fe₂WO₆ as a function of temperature. Inset: Dielectric loss (tan δ) as a function of temperature.

Discussion

Structural relationships in iron tungstates

As illustrated in Figure 8, the newly-solved structure of β -Fe₂WO₆ joins γ -Fe₂WO₆ and Fe(II) tungstate FeWO₄ in a group of compounds that can be described as stackings of zig-zag chains made of edge-sharing MO₆ octahedra. In polymorph β (Figure 8a), these chains are built from trans- and skew-edge-sharing octahedra; all chains follow the same Fe-Fe-Fe-Fe-W-W sequence and are connected through corners from one layer to the next. Polymorph γ (Figure 8b) crystallizes in the tri- α -PbO₂ structure (*Pbcn*), with zig-zag chains of octahedra sharing skew edges⁸. Assuming ideal ordering of Fe and W cations, 1/3 of the zig-zag chains in a layer contain Fe1O₆ octahedra only while 2/3 of the chains consist in alternating Fe2O₆ and WO₆ octahedra. The layers of zig-zag chains are stacked along the *a*-axis with a displacement ($\frac{1}{2}$, $\frac{1}{2}$, 0). From one layer to the next, octahedra share corners. FeWO₄ ^{26,27} (Figure 8c) presents a monoclinic structure *P2/c* with a

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slight monoclinic distortion β = 90.17°. This structure displays the same zig-zag chains as in γFe₂WO₆ but there is no mixing of Fe and W in the chains, and layers of FeO₆ zig-zags alternate with layers of WO₆ zig-zags.



Figure 8. Representations of the structures of (a) β -Fe₂WO₆; (b) γ -Fe₂WO₆ and (c) FeWO₄ at room temperature. WO₆ octahedra, blue; Fe1O₆ octahedra, red; Fe2O₆ octahedra, orange. The off-centering of the cations is evidenced by green spheres indicating the positions of the barycenters.

In all three structures, Fe and W cations are clearly off-centered in their MO₆ octahedra as shown in Figure 8 and Tables S1-S3. Cations in edge-sharing octahedra shift away from each other in a

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798-9806 Link to paper on ACS editor's website: https://dx.doi.org/10.1021/acs.inorgchem.0c01024 direction roughly perpendicular to the axis of the zig-zag chains, except for the Fe2 cations in polymorph β , which by shifting away from their W neighbor move closer to their Fe1 neighbor, due to the trans configuration. The amplitude of off-centering for this Fe2 cation in polymorph β (0.1296 Å along the *c*-axis) is smaller than for Fe1 and W (0.2711 Å and 0.2531 Å, respectively). Such off-centering seems to be specific to Fe(II) and Fe(III) tungstates, since it is not observed in Fe₂TeO₆²⁸ but is seen in ordered aeschynite RFeWO₆.²⁹ Indeed a majority of the compounds with d⁰ W⁶⁺ in octahedral coordination display an out-of-center distortion driven by the second-order Jahn-Teller (SOJT) effect.^{30–33} In the SOJT effect, the near degeneracy which is removed by the distortion is that of states created by the mixing of the empty d levels of tungsten with the full 2p levels of the oxygen neighbors. As argued by Kunz and Brown³¹, such a distortion driven by electronic effects is usually supported by an adaptation of the whole structural network, which stabilizes the structure. As a complement to the interatomic distances W-O, Fe1-O and Fe2-O reported in Table 2 and Figure S2, Figure 9 plots the distance of each oxygen to its three cationic neighbors. In the case of oxygen atoms connected to 2 Fe and 1 W, the W-O distance is usually the shortest of the three (except for O3). Oxygen O4, which connects the two tungstens in the W₂O₁₀ unit, exhibits two long W-O bond lengths probably due to the repulsion between cations of high oxidation state. In the case of O6 which bonds to 3 Fe, the Fe^{3+} -O distances are similar to the typical W⁶⁺-O distances. This complex pattern of off-centered cations and distorted octahedra results in an adapted structure where the environment of all cations corresponds to a stable situation according to Pauling's second rule³⁴, as shown by the calculation of charge distribution summarized in Table S4.



Figure 9. Distribution of the interatomic Fe1 \blacktriangle /Fe2 \checkmark /W \bigcirc -O distances (y-axis) for each of the six oxygen atoms in the structure (x-axis). The error bars correspond to three times the standard deviations calculated using FullProf.

Magnetic networks of Fe(III) and magnetic exchange paths in Fe₂WO₆ polymorphs

The structural differences between the β and γ polymorphs of Fe₂WO₆ create different networks of magnetic Fe(III) cations, which can be described as follows. In polymorph β , the edge-sharing octahedra in the zig-zag chains form (Fe2-Fe1-Fe1-Fe2) zig-zag tetramers; the Fe2O₆ octahedra at each end of the tetramers share corners with Fe2O₆ octahedra of four adjacent tetramers, resulting in planes of corner-sharing Fe2O₆ octahedra (plane 2 in Figure 2d). In polymorph γ , the zig-zag chains of edge-sharing Fe1O₆ are connected through corners to Fe2O₆ octahedra; the Fe2O₆ octahedra share edges with non-magnetic WO₆ and are therefore involved in magnetic interactions only through corner-sharing with Fe2O₆ and Fe1O₆ in adjacent chains.

Common to both polymorphs is the presence of Fe(III) cations (3d⁵): according to the semiempirical Goodenough-Kanamori (GK) rules, $3d^5-3d^5$ couplings are expected to be antiferromagnetic in the direct exchange ($\approx 90^\circ$ orbital configuration between edge-sharing octahedra for instance) and super-exchange ($\approx 180^\circ$ orbital configuration, e.g., between cornersharing octahedra) cases. Table 5 summarizes the magnetic couplings (established from the magnetic ground states determined in the present work for β -Fe₂WO₆ and in the study by Pinto et al.¹⁰ for γ -Fe₂WO₆) inside and between the zig-zag chains and provides the corresponding values for the Fe-Fe distances and Fe-O-Fe angles. An illustrated version of the same information is available in Figure S3.

	Polymorph	β(10 K)	Polymorph γ (4.2 K)		
	AFM coupling	FM coupling	AFM coupling	FM coupling	
Intra chain (edge-sharing octahedra)	Fe1-Fe1: 3.254 Å 101.0°/101.0°	Fe1-Fe2: 3.040 Å 94.1°/104.6°	/	Fe1-Fe1: 3.122 Å 97.1°/97.1°	
Inter chains (corner-sharing octahedra)	Fe1-Fe2: 3.378 Å Fe1-Fe2: 3.429 Å Fe2-Fe2: 3.419 Å Fe2-Fe2: 3.822 Å from 125.1° to 129.5°	/	Fe2-Fe2: 3.486 Å Fe1-Fe2: 3.626 Å 125.8°-126.1°	/	
Around Fe1	3	1	2	2	
Around Fe2	6	1	6	0	

Table 5. Magnetic couplings, interatomic Fe-Fe distances and Fe-O-Fe angles in polymorphs β -Fe₂WO₆ (this work) and γ -Fe₂WO₆.¹⁰

In both polymorphs, the magnetic interactions between chains are super-exchange mediated (Fe-Fe distances between 3.3 Å and 3.8 Å, with Fe-O-Fe angles between 125° and 130°) and always antiferromagnetic, in good agreement with the GK rules.

The description of magnetic couplings within zig-zag chains is more complex. In both compounds, the Fe-O-Fe angles in chains are smaller (between 94° and 105°) than between chains. In polymorph β , coupling within a tetramer in the zig-zag chain is ferromagnetic between Fe1 and

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: https://dx.doi.org/10.1021/acs.inorgchem.0c01024 Fe2 and antiferromagnetic between Fe1 and Fe1, leading to a $\uparrow \uparrow \Psi \Psi$ magnetic configuration. In polymorph γ , only Fe1 are magnetically coupled through edge-sharing in zig-zag chains and the coupling is also ferromagnetic. In both cases, the FM coupling takes place for short Fe-Fe distances (3.04 Å in polymorph β , 3.12 Å in polymorph γ), suggesting therefore a ferromagnetic direct exchange contribution, in contrast to what is expected from the GK rules. The AFM coupling between Fe1 and Fe1 within the tetramers of polymorph β is observed for a longer Fe-Fe distance (3.25 Å). As a comparison, the magnetic network in trirutile Fe₂TeO₆²⁸ is made of dimer units of two edge-sharing FeO₆ octahedra, connected through corners, with antiferromagnetic coupling even for the shortest Fe-Fe distance (3.03 Å in the dimer, i.e. shorter than the shortest distance in the Fe₂WO₆ polymorphs), in good agreement with the GK rules. Likewise to results on the Cr₂Te₁. xWxO₆ system³⁵, this could suggest that another key parameter controlling the sign of direct exchange magnetic interactions is the orbital hybridization between Fe 3*d* and O 2*p* orbitals through the 5*d* W states, therefore beyond the GK formalism.

The spin orientation in an individual FeO₆ octahedron is another difference between the Fe₂WO₆ polymorphs (Figure 6 and S4). Single-ion magnetic anisotropy, that is, the term in the magnetic energy dependent on the relative direction of the spin with respect to the local octahedral environment, is not easily assessed from structural considerations, since the Fe(*4e*) site symmetry is 1. In both polymorphs, there are two independent Fe sites in the crystal structure, so that each site could have its own local anisotropy, owing to the strong impact of local environment on single-ion magnetic anisotropy, even for a d^5 element (L = 0).³⁶ Figure S4 shows that, in β -Fe₂WO₆, the canting away from the order in the *ac*-plane below 150 K corresponds to a realignment of the spin towards the basal plane of the octahedron, together with an increase in $|m_{TOT}|$ (Table 4). However,

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> at this stage, it is not possible to conclude as to whether the observed spin reorientation is a result of competing magnetic exchanges owing to the complex topology of the magnetic network, if single-ion anisotropy is involved or not, or if another scenario altogether is at play. If single crystals could be obtained, inelastic neutron scattering experiments would allow to probe further into the details of the relevant parameters (hierarchy of magnetic exchanges, role of single-ion anisotropy in particular) stabilizing the magnetic ground state of β-Fe₂WO₆.

Conclusion

The monoclinic polymorph β -Fe₂WO₆ exhibits an original crystal structure, based on the alternation of FeO₆ and WO₆ octahedra sharing trans- or skew- edges. The corresponding magnetic network is built on Fe(III) tetramers, which order below 264 K in a nearly collinear up-up-down-down configuration in the *ac*-plane. These tetramers are coupled antiferromagnetically, the magnetic ordering having the $P2_1/a'$ symmetry. Below 150 K, a reorientation of the Fe(III) moments is observed, which leads to a non-collinear antiferromagnetic order with a small component of the Fe(III) moments along the *b*-axis.

This determination of the β polymorph structure unlocks the possibility to use the Fe₂WO₆ $\alpha/\beta/\gamma$ system as a platform to study the mechanisms governing the phase transitions and functional properties in such iron oxides. Further work is underway to build a clearer view of the formation conditions of each polymorph, including through solution-based procedures.

In addition, it is hoped that such synthesis routes would make it possible to prepare the different polymorphs with similar characteristics of particle size and surface area, in order to study whether and how the structural differences affect the (photo)catalytic/photoelectrochemical properties. In 2017, Abdi et al.⁵ used spray pyrolysis of an alcoholic solution to prepare a 400-nm-thick film whose diffractogram displayed broad diffraction peaks at some positions corresponding to a

Author's postprint version of paper published in Inorg. Chem. 2020, 59, 9798–9806 Link to paper on ACS editor's website: <u>https://dx.doi.org/10.1021/acs.inorgchem.0c01024</u> partially-ordered γ phase. They assessed the suitability of this film as a candidate photo-anode for the oxidation of water and concluded that it was better suited to other photocatalytic applications, especially in environmental remediation. Obviously, if the so-far-untested β polymorph turns out to display promising properties, practical prospects would only be realistic if the heat treatment time is drastically reduced thanks to the solution synthesis route.

ASSOCIATED CONTENT

Supporting Information.

Representation of "skew" and "trans" edges forming a zig-zag chain of octahedra.

Projections along [010], interatomic distances and angles in β -Fe₂WO₆ at room temperature and 10K.

Representations of the environments and magnetic interactions of Fe1 and Fe2 in β -Fe₂WO₆ and γ -Fe₂WO₆ and orientations of the Fe³⁺ moments in the FeO₆ octahedra in β -Fe₂WO₆ and γ -Fe₂WO₆.

Tables of atomic positions, barycenter positions and off-centering in the MO₆ octahedra in polymorph β , polymorph γ and FeWO₄.

Table with the formal charge, q, of the cations, and the charge distribution Q calculated for each octahedron with and without off-centering of the cations. (PDF)

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The structural and physical properties of the β polymorph of iron tungstate Fe₂WO₆ have been investigated. The crystal structure of β -Fe₂WO₆ has been solved and consists in an original network of zig-zag chains of MO₆ octahedra. Below 264 K, a $\uparrow \uparrow \downarrow \downarrow$ nearly collinear order of the Fe(III) moments was determined, with a further canting of the moments below 150 K. These results are compared with other iron tungstates and relevant compounds.

