

Synthesis, characterization, biological and radical scavenging activities of new palladium complexes

Université de Liège

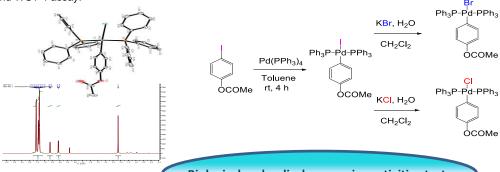
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Introduction

Synthesis and characterization

In recent years, there has been considerable interest in the chemistry of transition metal complexes containing novel ligands. In this contribution, we will present the synthesis of new palladium complexes and evaluate their radical scavenging activities in ABTS and DPPH tests. The cells toxicity of complexes was evaluated using trypan blue test and WST-1 assay.

The complex with iodide anion was obtained by oxidative addition of 4-OAcC₆H₄I on Pd(PPh₃)₄. While bromo- and chloro-containing palladium complexes were prepared by anion metathesis from the iodide. All compounds have been characterized by ¹H, ¹³C and ³¹P NMR and by X-ray diffraction.

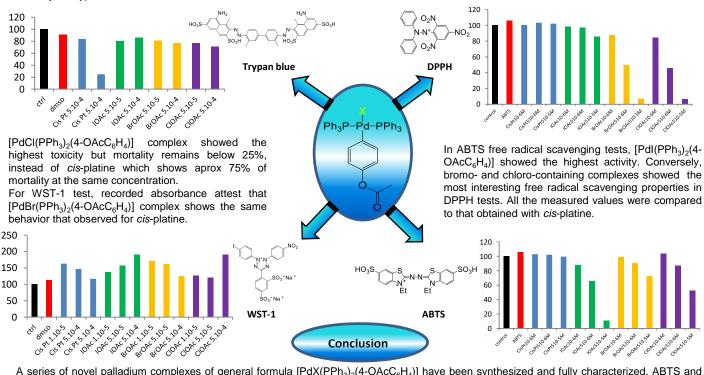




Biological and radical scavenging activities tests

Cells toxicity of the 3 complexes was evaluated using trypan blue and WST-1 tests. Cells were incubated with solutions of complexes during one day for trypan blue test and 2 hours for WST-1 test.

Some antioxydant properties were studied.1 After 30 min. of incubation at room temperature, absorbances were measured at 734 nm for ABTS and at 517 nm for DPPH.2



A series of novel palladium complexes of general formula [PdX(PPh₃)₂(4-OAcC₆H₄)] have been synthesized and fully characterized. ABTS and DPPH free radical scavenging tests were performed, showing the highest activity for the iodo-containing complex in the ABTS tests, whereas the chloro complex was the most efficient for the DPPH test. In the WST-1 versus Trypan blue test, there is no correlation for the cellular toxicity in spite of the low toxicity of compounds. Action mode of those complexes in WST-1 test is currently investigated.

Conclusion

References