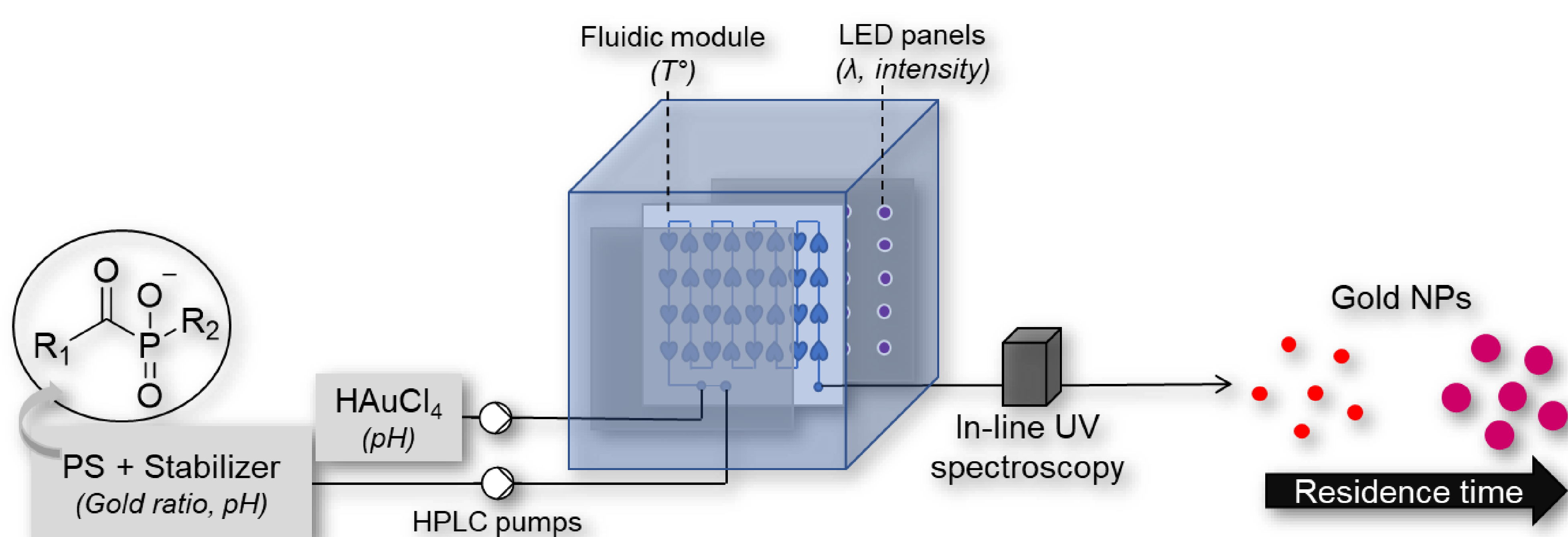
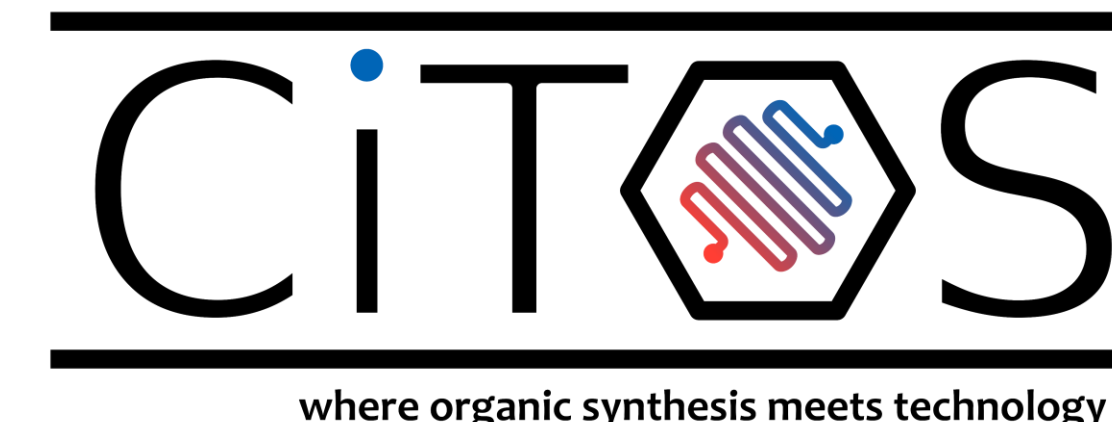


# CONTINUOUS FLOW TECHNOLOGY FOR THE PHOTOCHEMICAL SYNTHESIS OF SPHERICAL GOLD NANOPARTICLES



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Gold nanoparticles (Au NPs) are used in a wide range of applications because of their localized surface plasmon resonance (LSPR), leading to shape- and size-dependent optical properties. The **photochemical generation of gold NPs** has emerged as a potential alternative to the well-known Turkevitch method since it offers a **spatial and temporal control**.

We present herein a **continuous flow process** for the controlled photochemical synthesis of spherical gold NPs. The implementation of a continuous flow setup **overcomes issues related to batch synthesis**, namely poor mixing efficiency, poor light penetration and inefficient control of the local parameters. After optimization, the NPs size is only dictated by the irradiation time.

## STEP 1 : Identification of the best photosensitizer (PS)

Tests with different PS:

### Type I PS:

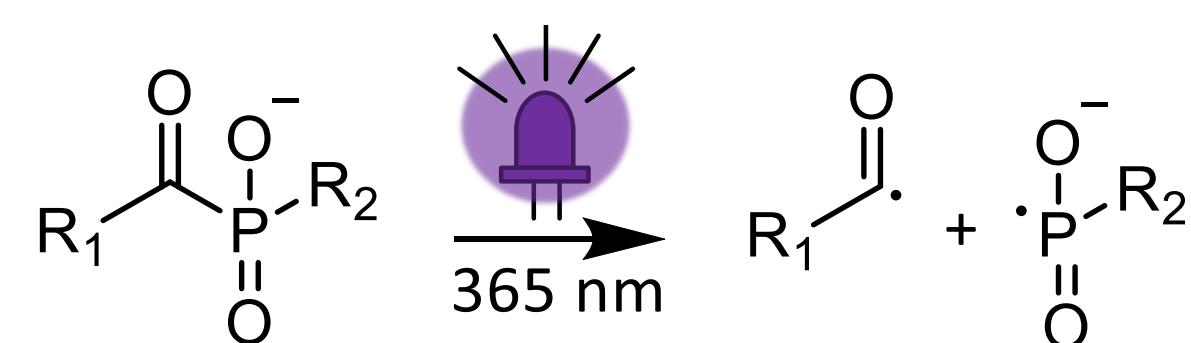
Unimolecular reaction  
 May be water soluble  
 Better stabilization

### Type II PS:

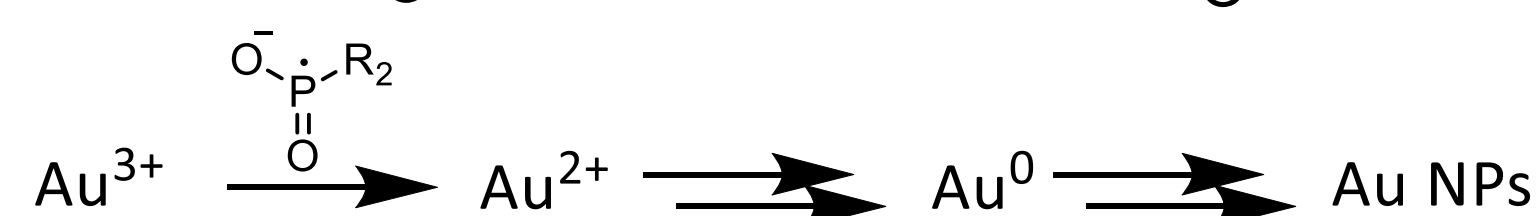
Bimolecular reaction  
 Need of a co-initiator  
 Water solubility issues  
 Bad stabilization

Identification of a new, water-soluble acylphosphinate PS : generation of free radicals acting as internal reductants of chloroauric acid (HAuCl<sub>4</sub>) into Au(0)

Photocleavage



Gold reduction

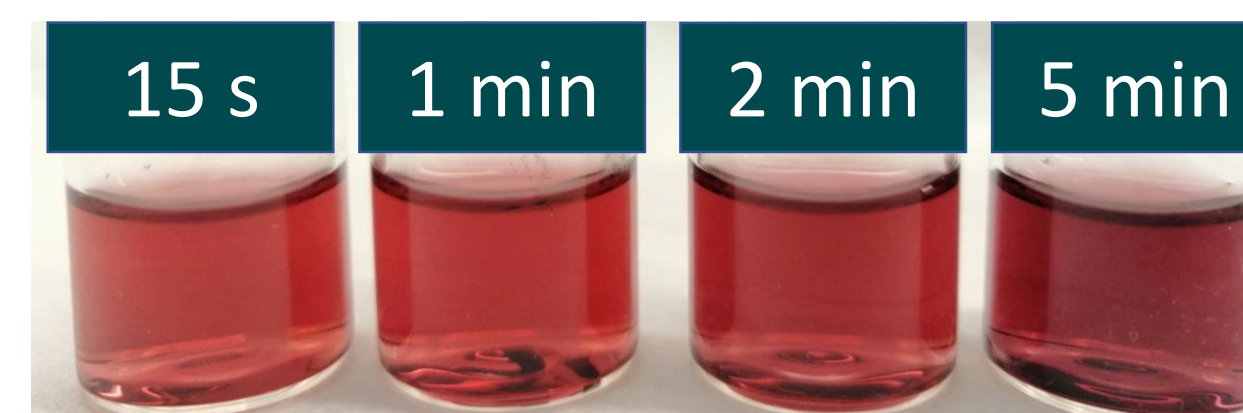


## STEP 3 : Optimization

Tuning experimental parameters : Monitoring with in-line UV spectrometry

Increase of	Light intensity	T°	pH	C <sub>stab</sub>
Maximum intensity	↗	↗	↘	↗
Maximum wavelength	↘	↗	↗	↘

Residence time : bigger NPs obtained when the irradiation time increases



Smaller NPs

Larger NPs

Stability : The addition of an external stabilizer and the optimization of some parameters enable the production of **stable** NPs up to several weeks

## STEP 2 : Avoiding gold coating

The formation of a gold coating on the internal surface of the reactor decreases the light intensity and leads to a less efficient photoredox process

Solutions:



Coating the glass with Sigmacote® before the synthesis

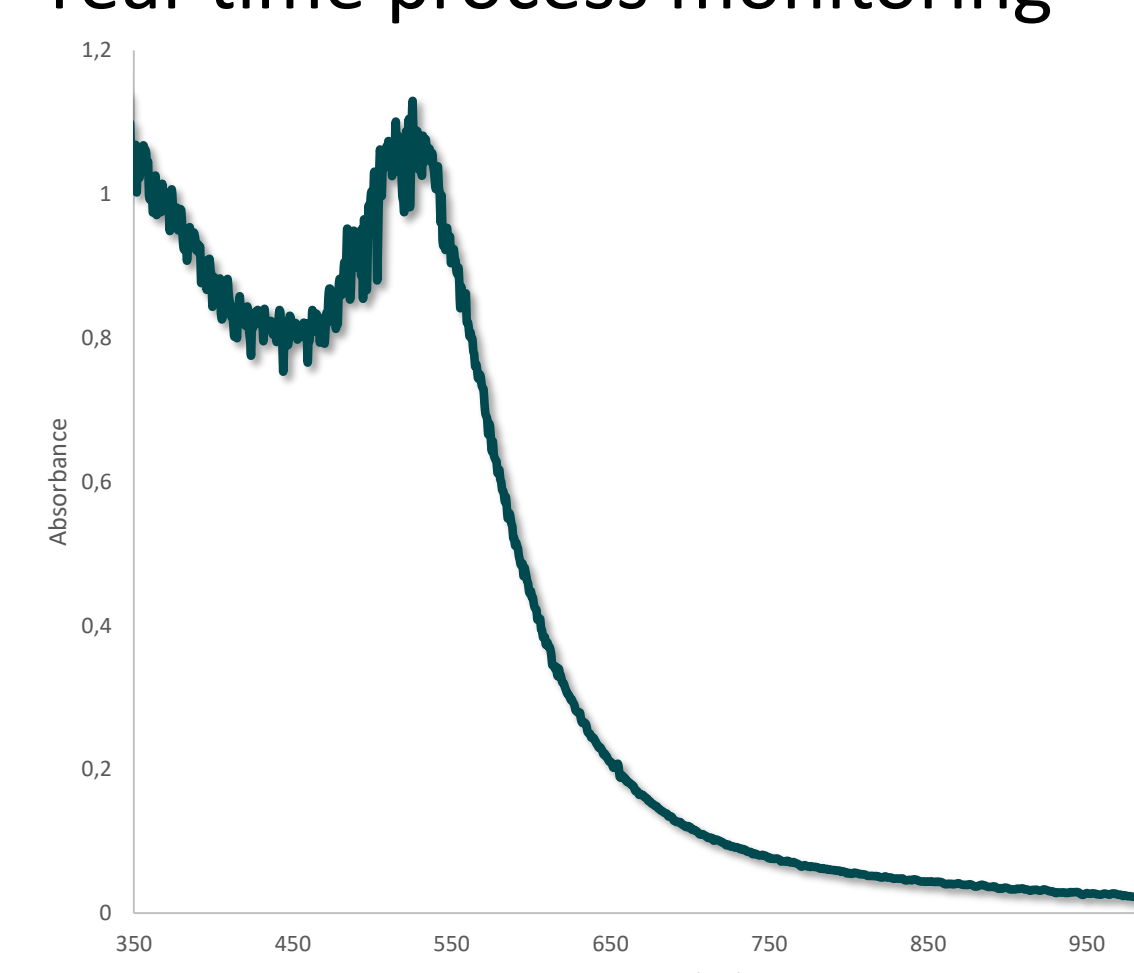
Working under biphasic conditions

Adding some **electrostatic repulsions** by working at slightly basic pH and with a deprotonated stabilizer

➔ The selection of the most appropriate stabilizer is critical

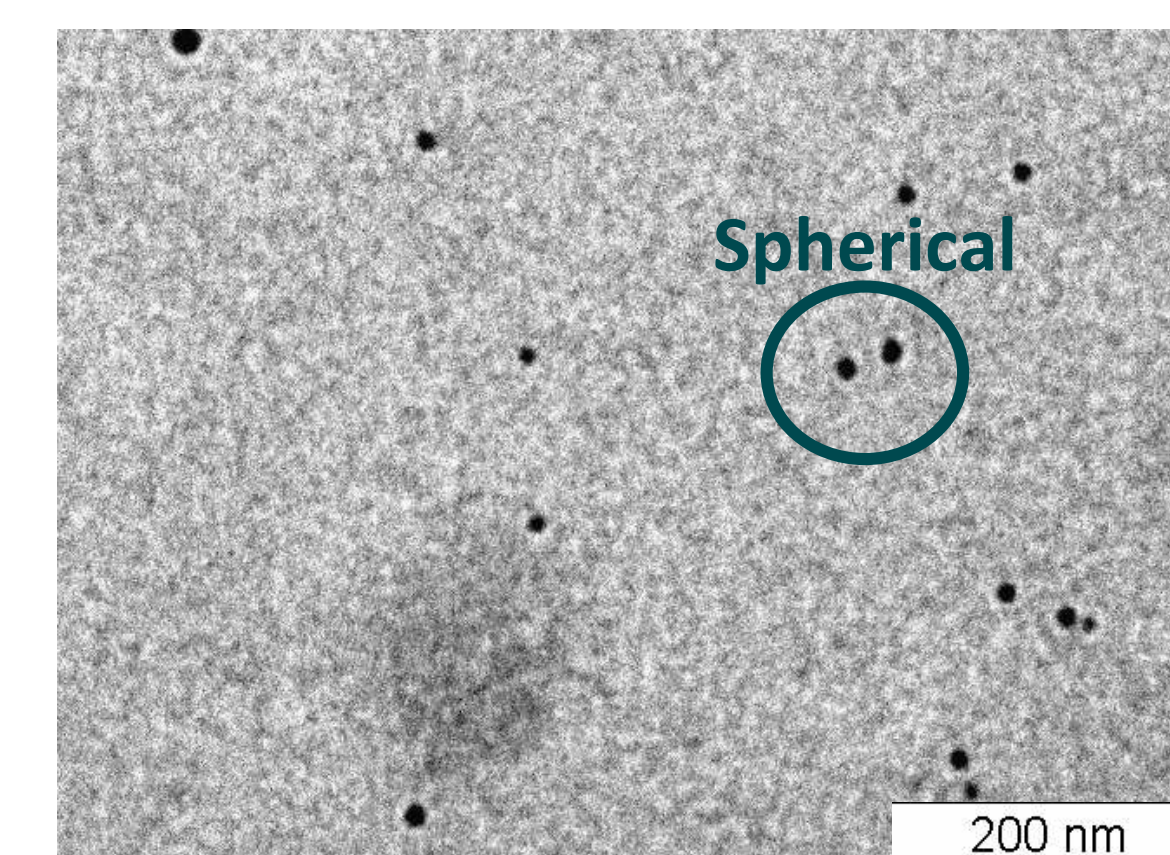
## NPs characterization : size and shape

In-line UV spectroscopy : real-time process monitoring



Particle or crystallite size

TEM : information about polydispersity and shape



Crystallite size radius

## CONCLUSION AND PERSPECTIVES

- ✓ Fast photogeneration of **spherical gold NPs** : control of the size only by tuning the irradiation time
- ✓ Use of the **flow technology** to overcome issues related to batch synthesis
- ✓ Identification of the parameters **avoiding gold coating**
- ✓ Identification of the best conditions to obtain **stable** NPs
- ✓ **Real-time monitoring** with in-line UV spectroscopy

Perspectives : scale-up

1. Lab photo reactor : Optimization (2 to 10 mL/min)
2. G1 : Scalability test (10 to 200 mL/min)
3. G3 : Industrial production (400 to 2000 mL/min)