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Title: Characterization and differentiation of boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrac  $\times$  Yorkshire) pigs by volatiles profiling and chemometrics analysis

Article Type: Research Articles

Keywords: GC-MS/O; E-nose; pork breeds; odour-active compounds; potential flavour markers

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Abstract: To characterize and differentiate boiled pork from three different breeds of pig (Tibetan, Sanmenxia and Duroc × (Landrace × Yorkshire)), the volatile compounds in each were analysed by gas chromatography-olfactometry-mass spectrometry (GC-MS/O) and electronic nose (E-nose) combined with chemometrics analysis. In total, 61 volatile compounds were identified, among which 25 compounds were selected as odour-active compounds in boiled pork. Moreover, seven odour-active compounds (hexanal, nonanal, 1-octen-3-ol, dimethyl disulphide, heptanal, 2-pentylfuran and 2-ethylfuran) were the main contributors to the integral flavour of boiled pork due to their higher odour activity values (OAVs) ranging from 17.3-524.2. The odour-active compounds were examined by principal component analysis (PCA), agglomerative hierarchical clustering (AHC) and partial least squares-discriminant analysis (PLS-DA). The results showed that boiled pork from the three pig breeds could be clearly distinguished, and twelve odour-active compounds, including (E,E)-2,4-decadienal, ethyl hexanoate, dimethyl disulphide, hexanal, 2acetylthiazole, (E)-2-nonenal, 1-octen-3-ol, (E,E)-2,4-nonadienal, heptanal, (E)-2-octen-1-ol, styrene and (E)-2-octenal, were determined as potential flavour markers for discrimination. This study indicated that GC-MS/O and E-nose with chemometrics analysis are feasible methods to characterize and discriminate boiled pork from three pig breeds.

Aug. 19th 2019

### Dr. A. Sant'Ana, Editor-in-Chief

#### Food Research International

#### Dear Dr. A. Sant'Ana,

Please find submitted our research article entitled "Characterization and differentiation of boiled pork from different breeds of pigs by volatiles profiling and chemometrics analysis" by Dong Han, Chunhui Zhang, Marie-Laure Fauconnier and Si Mi for publication in *Food Research International*.

The aim of this study was to characterize the volatiles profile of boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrace  $\times$  Yorkshire) pigs by application of GC-MS/O, followed by confirmation of the potential flavour compounds by chemometrics analysis. Moreover, multivariate statistical methods were used to explore the feasibility of differentiating boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrace  $\times$  Yorkshire) pigs using volatile compounds. The results of this study will provide a better understanding of the odour-active compounds in boiled pork from different breeds of pigs and can be expected to establish a new method to discriminate pork from different pig breeds. This article contains original data that is not being considered for publication, in whole or in part, elsewhere. All of the authors have reviewed and approved the manuscript and there are no conflicts of interest to report. We look forward to the review and sincerely hope that our paper will be of sufficient merit to be published in *Food Research International*.

Thank you in advance for your consideration.

Sincerely,

Chun-Hui Zhang, PhD

# Response to Editor's Comments

## **Editor's comments:**

**Point:** The list of references must be updated. There was a lot done in this field in the last 3 years (2017-2019). Making reference to recent work in the field is particularly key to highlight the current context of the present manuscript and to make it more comprehensive and to highlight the novelty to the readers as well as its assessment and contribution to the field. Please, address this request by adding new data, critical analysis and not by simply citing papers published in this subject in the mentioned years.

**Response:** Many thanks for your valuable comments. According to your suggestion, the introduction has been completely revised and as follows (**Line 44-137**). Fonts marked in red are added, updated and modified sections. Additionally, the paper has been edited by **Elsevier Language Editing Services** to read smoothly.

### 1. Introduction

According to a United States Department of Agriculture (USDA) report, global pork production is expected to rise to 113.0 million tons. China has the largest pork share in the global market, accounting for 48.7% (55.0 million tons) of total production in the whole world. Moreover, pork is popular with consumers due to its sensory attributes, such as tender texture, rich nutritional composition (Purriños, Franco, Carballo, & Lorenzo, 2012; Sivakumar 2016) and unique flavour (Straadt, Aaslyng, & Bertram, 2013). Flavour is one of the most important sensory attributes for consumers to judge the quality of pork (Wang, Song, Zhang, Tang, & Yu, 2016) and mainly associated with the generation of volatile compounds (Zhao et al., 2017). Previous studies have indicated that over 1000 volatile compounds have been identified in meat and meat products, including aldehydes, ketones, alcohols, acids, esters, hydrocarbons, ethers, heterocyclic compounds and sulphur compounds (Shahidi 1998). These compounds are mainly derived from a complex series of chemical reactions (e.g., lipid oxidation, the Maillard reaction and lipid-Maillard interactions) between precursors, intermediate reaction products and degradation products (Jayasena, Ahn, Nam, & Jo, 2013).

To explore the composition, origin and formation of volatile compounds in different pork products, many studies have been performed in recent years. A total of 149 volatile compounds (25 aldehydes, 18 phenols, 12 alcohols, 16 terpenes, 27 aromatic hydrocarbons, 18 aliphatic hydrocarbons, 17 ketones, 9 esters and 7 acids) were identified from dry-cured hams using four different processing methods, among which aldehydes and phenols were the more abundant volatiles (Petričević, Radovčić, Lukić, Listeš, & Medić, 2018). The volatile compounds in six dry-cured meat products were detected using a GC/MS technique; these compounds were then used to identify the possible source of the typical volatiles (Domínguez et al., 2019). Due to lipid oxidation, brine permeation and carbohydrate fermentation, the levels of volatile compounds under high pressure treatment contributed were more than 70% of the typical aroma, except for acetic acid (Yang, Sun, Pan, Wang, & Cao, 2018). Although a large number of volatile compounds associated with different processing technologies for specific types of pork have been fully analysed, information on volatile profiles in different varieties of processed pork products are still lacking.

As reported, the intramuscular fat, colour and flavour of pork from different pig breeds have been studied by many researchers (Lee et al. 2012; Lu, Li, Yin, Zhang, & Wang, 2008; Meinert, Christiansen, Kristensen, Bjergegaard, & Aaslyng, 2008), and the results show that the breed greatly impacted pork flavour quality. In China, Tibetan and Sanmenxia pigs, as the local pig breeds, are well known for their favourable organoleptic properties and rich nutritional composition (Mi et al., 2019; Shen et al., 2014). As a typical hybrid pig, Duroc × (Landrace × Yorkshire) is now widely used for commercial production and the texture and flavour of this pork is notably different from that of Chinese local pig breeds (Lu, Li, Yin, Zhang, & Wang, 2008). The above studies were mainly focused on the physic-chemical and sensory characteristics in the different pork breeds, and it is unclear which flavour compounds are important for sensory attributes. Furthermore, due to the higher nutrition and flavour quality, Chinese local pork is usually at a higher price compared the typical hybrid pork. To obtain the higher profits, some people pass off the typical hybrid pork as Chinese local pork. This behaviour results in economic loss to the meat industry and has negative effects on the reputation of Chinese local pork. Therefore, it is necessary to find a new method to identify Chinese local pork.

At present, a variety of analytical techniques have been employed for differentiation of meat in the scientific literature. Grunert, Stephan, Ehling-Schulz, and Johler (2016) provide a promising method to differentiate fresh and frozen/thawed chicken using Fourier transform infrared spectroscopy. Visible and near infrared spectroscopy technology can rapidly identify enhanced quality pork (Prieto, Juárez, Zijlstra, López-Campos, & Aalhus, 2015). Lopez-Oceja, Nuñez, Baeta, Gamarra, and Pancorbo (2017) reported that eight common meat species were identified using a high-resolution melt screening method. Additionally, our research team have reported an ICP-MS-based element profile (Mi, Shang, Jia, Zhang, & Fan, 2019) for the authentication of Taihe black-boned silky fowl. To the best of our knowledge, most of the above analytical methods are widely used to distinguish different types of meat; however, there are only a few reports describing the use of chemometrics analysis of volatile flavour compounds to discriminate different breeds. A recent study has found that beef, pork and mixed (70% beef and 30% pork) minced meat could be easily discriminated and classified by a volatilomic approach based on volatile fingerprints (Pavlidis, Mallouchos, Ercolini, Panagou, & Nychas, 2019). Hence, volatiles analysis together with multivarise statistics is a promising approach for the differentiatation of different varieties of pork.

The aim of this study was to characterize the volatile profile in boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrace  $\times$  Yorkshire), and then to confirm the key odor-active compounds and potential flavour markers. Multivariate statistical methods for volatile compounds were used to explore the feasibility to differentiate boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrace  $\times$  Yorkshire) pigs. The results of this study should provide a better understanding of the aroma characteristics of boiled pork and provide a novel strategy for the authentication of boiled pork from Tibetan, Sanmenxia and Duroc  $\times$  (Landrace  $\times$  Yorkshire) pigs.

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## Highlight:

- A total of 61 volatile compounds were identified and quantified.
- 25 volatile compounds were considered as odour-active compounds.
- Boiled pork was clearly separated into three groups by chemometric analysis.
- The potential flavour markers were found in boiled pork of different breeds of pigs.
- GC-MS/O coupled with E-nose method is feasible to distinguish the different boiled pork.

- 2 1 3	Characterization and differentiation of boiled pork from Tibetan, Sanmenxia
2	and Duroc $\times$ (Landrac $\times$ Yorkshire) pigs by volatiles profiling and chemometrics
2 7 3	analysis
) 4	Dong Han <sup>a,b</sup> , Chun-Hui Zhang <sup>a,*</sup> , Marie-Laure Fauconnier <sup>b</sup> , Si Mi <sup>a</sup>
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) 8	<sup>b</sup> Laboratory of Chemistry of Natural Molecules, Gembloux Agro-bio Technology,
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2 7 11	*Chun-Hui Zhang, E-mail: dr_zch@163.com, Tel: 86-10-62819469
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2	13	Abstract: To characterize and differentiate boiled pork from three different
4	14	nig (Tibetan Sanmenxia and Duroc $\times$ (Landrace $\times$ Yorkshire)) the volatile
5	14	pig (Trocuit, Sumienxia and Baroe × (Eandrace × Torksmie)), the volutio
7 8	15	compounds in each were analysed by gas chromatography-olfactometry-ma
9 10	16	spectrometry (GC-MS/O) and electronic nose (E-nose) combined with cher
11 12	17	analysis. In total, 61 volatile compounds were identified, among which 25 of
13 14 15	18	were selected as odour-active compounds in boiled pork. Moreover, seven
16 17	19	odour-active compounds (hexanal, nonanal, 1-octen-3-ol, dimethyl disulphi
18 19 20	20	heptanal, 2-pentylfuran and 2-ethylfuran) were the main contributors to the
21 22	21	flavour of boiled pork due to their higher odour activity values (OAVs) range
23 24	22	17.3-524.2. The odour-active compounds were examined by principal comp
25 26 27	23	analysis (PCA), agglomerative hierarchical clustering (AHC) and partial lea
28 29	24	squares-discriminant analysis (PLS-DA). The results showed that boiled po
30 31 32	25	the three pig breeds could be clearly distinguished, and twelve odour-active
33 34	26	compounds, including $(E,E)$ -2,4-decadienal, ethyl hexanoate, dimethyl disu
35 36 27	27	hexanal, 2-acetylthiazole, (E)-2-nonenal, 1-octen-3-ol, (E,E)-2,4-nonadiena
38 39	28	(E)-2-octen-1-ol, styrene and $(E)$ -2-octenal, were determined as potential fl
40 41	29	markers for discrimination. This study indicated that GC-MS/O and E-nose
42 43 44	30	chemometrics analysis are feasible methods to characterize and discriminat
45 46	31	pork from three pig breeds.
47 48 49	32	Keywords: GC-MS/O; E-nose; pork breeds; odour-active compounds; pote
50 51	33	flavour markers
52 53	34	Chemical compounds studied in this article:
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om three	differe	nt breed	ls of

y (GC-MS/O) and electronic nose (E-nose) combined with chemometrics

total, 61 volatile compounds were identified, among which 25 compounds

pentylfuran and 2-ethylfuran) were the main contributors to the integral

biled pork due to their higher odour activity values (OAVs) ranging from

The odour-active compounds were examined by principal component

riminant analysis (PLS-DA). The results showed that boiled pork from

including (E,E)-2,4-decadienal, ethyl hexanoate, dimethyl disulphide,

cetylthiazole, (E)-2-nonenal, 1-octen-3-ol, (E,E)-2,4-nonadienal, heptanal,

1-ol, styrene and (E)-2-octenal, were determined as potential flavour

discrimination. This study indicated that GC-MS/O and E-nose with

cs analysis are feasible methods to characterize and discriminate boiled

GC-MS/O; E-nose; pork breeds; odour-active compounds; potential

35 Hexanal (PubChem CID: 6184)

36 Nonanal (PubChem CID: 31289)

37 1-Octen-3-ol (PubChem CID: 18827)

38 Dimethyl disulphide (PubChem CID: 12232)

- 39 Heptanal (PubChem CID: 8130)
- 40 2-Pentylfuran (PubChem CID: 19602)
- 41 2-Ethylfuran (PubChem CID: 18554)
- (E,E)-2,4-Decadienal (PubChem CID: 5283349)
- 43 2-Acetylthiazole (PubChem CID: 520108)

#### **1. Introduction**

According to a United States Department of Agriculture (USDA) report, global pork production is expected to rise to 113.0 million tons. China has the largest pork share in the global market, accounting for 48.7% (55.0 million tons) of total production in the whole world. Moreover, pork is popular with consumers due to its sensory attributes, such as tender texture, rich nutritional composition (Purriños, Franco, Carballo, & Lorenzo, 2012; Sivakumar 2016) and unique flavour (Straadt, Aaslyng, & Bertram, 2013). Flavour is one of the most important sensory attributes for consumers' choice of to judge the quality of pork (Wang, Song, Zhang, Tang, & Yu, 2016) and mainly associated with the generation of volatile compounds (Zhao et al., 2017).- Previous studies have indicated that So far, over 1000 volatile compounds have been identified in meat and meat products, including aldehydes, ketones, alcohols, acids, esters, hydrocarbons, ethers, heterocyclic compounds and sulphur 

57	compounds (Shahidi 1998). These compounds are mainly derived from a complex	Formatted: Font color: Text 1
58	series of chemical reactions (e.g., lipid oxidation, the Maillard reaction and	
59	lipid-Maillard interactions) between precursors, intermediate reaction products and	
60	degradation products (Jayasena, Ahn, Nam, & Jo, 2013).	Formatted: Font color: Text 1
61	To explore the composition, origin and formation of volatile compounds in	
62	different pork products, many studies have been performed in recent years. A total of	Formatted: Font color: Text 1 Formatted: Font color: Text 1
63	149 volatile compounds (25 aldehydes, 18 phenols, 12 alcohols, 16 terpenes, 27	
64	aromatic hydrocarbons, 18 aliphatic hydrocarbons, 17 ketones, 9 esters and 7 acids)	
65	were identified from dry-cured hams using four different processing methods, among	
66	which aldehydes and phenols were the more abundant volatiles (Petričević, Radovčić,	
67	Lukić, Listeš, & Medić, 2018). The volatile compounds in six dry-cured meat	Formatted: Font color: Text 1 Formatted: Font color: Text 1
68	products were detected using a GC/MS technique; these compounds were then used to	Formatted: Font color: Text 1
69	identify the possible source of the typical volatiles (Domínguez et al., 2019). Due to	Formatted: Font color: Text 1 Formatted: Font color: Text 1
70	lipid oxidation, brine permeation and carbohydrate fermentation, the levels of volatile	Formatted: Font color: Text 1
71	compounds under high pressure treatment contributed were more than 70% of the	
72	typical aroma, except for acetic acid (Yang, Sun, Pan, Wang, & Cao, 2018), Although	Formatted: Font color: Text 1
73	a large number of volatile compounds associated with different processing	Formatted: Font color: Text 1 Formatted: Font color: Text 1
74	technologies for specific types of pork have been fully analysed, information on	Formatted: Font color: Text 1 Formatted: Font color: Text 1
75	volatile profiles in different varieties of processed pork products are still lacking.	Formatted: Font color: Text 1
76	Aroma compounds such as pyrazines, pyridines and furans in fried bacon and fried	Formatted: Font color: Text 1
77	pork loin are thought to be responsible for the meaty aromas (Cross & Ziegle 2006;	Formatted: Font color: Text 1 Formatted: Font color: Text 1
78	Timón, Carrapiso, Jurado, & van, 2004). A total of 38 volatiles were determined in-	Formatted: Font color: Text 1

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79	raw ham using the HS-Trap GC-MS method (Bosse, Wirth, Konstanz, Becker, Weiss,
80	& Gibis, 2017). A total of 149 volatile compounds of dry cured hams from four-
81	different processing methods were identified and 15 of them were quantified
82	(Petricevic, Marusic, Luki, Listes, & Medic, 2018). The above studies have mainly-
83	focus on the qualitative and quantitative analysis of volatile profiles of the pork from
84	different processing technologies. However, there is a small number of studies-
85	regarding the volatile profiles of boiled pork of different breeds of pigs.
86	As reported, the intramuscular fat, colour and flavour of pork from different pig
87	breeds have been studied by many researchers (Lee et al. 2012; Lu, Li, Yin, Zhang, &
88	Wang, 2008; Meinert, Christiansen, Kristensen, Bjergegaard, & Aaslyng, 2008), and
89	the results show that the breed greatly impacted pork flavour quality. In China,
90	Tibetan and Sanmenxia pigs, as the local pig breeds, are well known for their
91	favourable organoleptic properties and rich nutritional composition (Mi et al., 2019;
92	Shen et al., 2014). As a typical hybrid pig, Duroc × (Landrace × Yorkshire) is now
93	widely used for commercial production and the texture and flavour of this pork is
94	notably different from that of Chinese local pig breeds (Lu, Li, Yin, Zhang, & Wang,
95	2008). The above studies were mainly focused on the physic-chemical and sensory
96	characteristics in the different pork breeds, and it is unclear which flavour compounds
97	are important for sensory attributes. The cooked pork from Chinese indigenous breed-
98	pigs have higher flavor intensity than hybrid pigs (Lu, Li, Yin, Zhang, & Wang, 2008).
99	In this study, both Tibetan pigs and Sanmenxia pigs belong to Chinese native breeds,
100	Duroc × (Landrace × Yorkshire) pigs belong to hybrid pigs. Furthermore, Due-due to

-	
101	the higher nutrition and flavour quality, (Yang et al. 2014; Zhang et al. 2014) Chinese
102	local pork, the pork of Tibetan pigs was _ is usually at a higher price compared the
103	typical hybrid pork. To obtain the higher profits, some people pass off the typical
104	hybrid pork as Chinese local pork Tibetan pork. This behaviour results in economic
105	loss to the meat industry and has negative effects on the reputation of Chinese local
106	pork Tibetan pork. Therefore, it is necessary to find a new method to identify Chinese
107	local porkthe pork of Tibetan pigs.
108	At present, a variety of analytical techniques have been employed for
109	differentiation of meat in the scientific literature. Grunert , Stephan, Ehling-Schulz,
110	and Johler (2016) provide a promising method to differentiate fresh and
111	frozen/thawed chicken using Fourier transform infrared (FTIR) spectroscopy. Visible
112	and near infrared spectroscopy (Vis-NIRS) technology can rapidly identify enhanced
113	quality pork (Prieto, Juárez, Zijlstra, López-Campos, & Aalhus, 2015). Lopez-Oceja,
114	Nuñez, Baeta, Gamarra, and Pancorbo (2017) reported that eight common meat
115	species were identified using a high-resolution melt (HRM)-screening method.
116	Additionally, our research team have reported ICP-MS-based element profile (Mi,
117	Shang, Jia, Zhang, & Fan, 2019) for the authentication of Taihe black-boned silky
118	fowl. To the best of our knowledge, most of the above analytical methods are widely
119	used to distinguish different types of meat; however, there are only a few reports
120	describing the use of chemometrics analysis of volatile flavour compounds to
121	discriminate different breeds. A recent study has found that beef, pork and mixed (70%
122	beef and 30% pork) minced meat could be easily discriminated and classified by a

1		
2 3	123	volatilomic approach based on volatile fingerprints (Pavlidis, Mallouchos, Ercolini,
4 5	124	Panagou, & Nychas, 2019). Hence, volatiles analysis together with multivarise
6 7 8	125	statistics is a promising approach for the differentiatation of different varieties of
9 10	126	pork.However, there is few researches regarding the discrimination method of
11 12 13	127	different breeds of pigs though chemometrics analysis of volatile flavor compounds.
14 15	128	The aim of this study was to characterize the volatile profile in boiled pork from
16 17	129	Tibetan, Sanmenxia and Duroc × (Landrace × Yorkshire), different breeds of pigs and
19 20	130	then to confirm the key odor-active compounds and potential flavour markers.
21 22	131	Multivariate statistical methods for volatile compounds were used to explore the
23 24 25	132	feasibility to differentiate boiled pork from Tibetan, Sanmenxia and Duroc $\times$
26 27	133	(Landrace × Yorkshire) pigsdifferent breeds of pigs. The results of this study should
28 29 20	134	provide a better understanding of the aroma characteristics of boiled pork and provide
31 32	135	a novel strategy for the authentication of <b>boiled pork from Tibetan</b> , Sanmenxia and
33 34	136	<u>Duroc <math>\times</math> (Landrace <math>\times</math> Yorkshire) pigsthe different breeds of pork</u> .
35 36 37	137	2. Materials and methods
38 39 40	138	2.1. Materials and chemicals
41 42 43	139	A total of 18 pigs from three breeds, including Tibetan pigs ( $n = 6$ , aged 5-6
44 45	140	months), Sanmenxia pigs (n = 6, aged 5-6 months) and Duroc $\times$ (Landrac $\times$ Yorkshire)
46 47 48	141	(n = 6, aged 5-6 months) were studied. Tibetan pigs (TB) was provided by Tibet Woye
49 50	142	Tibetan Pig Development Co. Ltd. (Nyingchi, Tibet Autonomous Region, China).
51 52 53	143	Sanmenxia pigs (SMX) and Duroc $\times$ (Landrace $\times$ Yorkshire) pigs (DLY) were
53 54 55 56 57 58 59 60	144	obtained from Chuying Agro-Pastoral Group Co. Ltd. (Zhengzhou, Henan Province,
51 52		
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145	China). All the pigs were reared under the same conditions and provided with the
146	same feed. They were slaughtered following the same commercial procedures in the
147	nearby abattoir. After cooling at 0-4°C for 24 h, Triceps brachii and Biceps femoris
148	muscles of all of the pigs were dissected from the carcasses. Tibetan, SMX and DLY
149	pork samples ( $n = 6$ for each breed) from two different muscle were collected, and the
150	same muscle gathered from two individual pigs of the same breed was combined as
151	one sample ( $n = 3$ for each muscle) for volatiles analysis. All pork samples were
152	placed into ice-boxes and sent to the laboratory of Chinese Academy of Agricultural
153	Sciences, Beijing. The study procedures were approved by the Animal Care and Use
154	Committee of the Institute of Food Science and Technology, Chinese Academy of
155	Agricultural Sciences, and performed in accordance with animal welfare and ethics.
156	$C_7$ - $C_{30}$ saturated alkanes (1000 µg/mL for each component in hexane) and
157	2-methyl-3-heptanone (99%) were purchased from Sigma-Aldrich (Shanghai, China).
158	2.2. Boiled pork muscles pretreatment
159	The skin, visible fat and connective tissues were removed from the pork of TB,
160	SMX and DLY. Approximately 200 g of meat supplemented with 1.0% sodium
161	chloride (based on the raw meat weight) and 150% ( $w/w$ ) tap water were boiled in a
162	low-density polyethylene bag. The pork samples were first heated from room
163	temperature (22.3 $\pm$ 0.5 °C) to a core temperature (80.0 $\pm$ 0.5 °C), then held for 30 min.
164	The boiled pork was cut into $1.0 \times 1.0 \times 1.0$ cm <sup>3</sup> cubes, ground with a pulveriser in
165	liquid nitrogen and stored in a frozen state (-18°C) until use.

#### 2.3. Solid-phase micro-extraction (SPME) of volatile compounds

The extraction method was modified from a previous study (Wang, Song, Zhang, Tang, & Yu, 2016). The volatile compounds of boiled pork were extracted using a manual SPME equipped with a  $50/30 \ \mu m$ divinylbenzene/carboxen/polydimethylsiloxane (DVB/CAR/PDMS) fibre (Supelco, Inc., Bellefonte, PA, USA). The pork sample was precisely weighed-5.00 g-and transferred to a 40 ml vial. Thereafter, 2-methyl-3-heptanone was dissolved in hexane as an internal standard solution to a final concentration of 0.41 mg/ml, 1µl of this solution was added and the vial was tightly capped with Teflon/silicon septum. The vial was equilibrated at 60°C for 20 min in a water bath. The selected fibre was exposed to the headspace of the samples to absorb the volatile compounds for 40 min at 60°C. Upon completion, the fibre was inserted into the injection port (250°C) of the GC instrument to desorb the analyses for 5 min. All samples were extracted in triplicate. 2.4. GC-MS/O analysis The method was performed according to the method of Liu, He, and Song (2018) with minor modifications. Analyses of volatile compounds were performed out on an Agilent 7890A gas chromatograph coupled with an Agilent Model 7000B series mass spectrometer (Agilent Technologies, Inc., Santa Clara, CA, USA). The GC-MS system was equipped with an olfactory detector port (Sniffer 9000; Brechbuhler, Schlieren, Switzerland). The volatiles were separated on polar DB-wax and non-polar DB-5 capillary column (30 m  $\times$  0.32 mm i.d., 0.25  $\mu$ m film thickness; J & W

188	Scientific, Inc., Folsom, CA, USA). Ultra-high purity helium (≥99.999%) was used as
189	the carrier gas and the constant flow rate was 1.2 ml/min. Temperature programme
190	began with isothermal heating at 40°C for 3 min, then rising to 200°C at a rate of
191	5°C/min, followed by another increase to 230°C (DB-wax) and 250°C (DB-5) at
192	10°C/min. Final temperature was held for 3 min. The transfer line temperatures were
193	maintained at 240°C (DB-wax) and 270°C (DB-5). The effluent from the capillary
194	column was split 5:1 ( $\nu/\nu$ ) between the mass spectrometry detector and the olfactory
195	detector port. Electro-impact mass spectra were generated at 70 eV with an $m/z$ scan
196	range from 50 to 400 amu. The ion source temperature was 230°C. A panel that
197	contains eight trained staff was utilized for the sniffing test on the GC-O. Humidified
198	air was supplied to the sniff port with a flow of 30 ml/min to avoid dryness of the
199	nasal mucosa.
199 200	nasal mucosa. 2.5. Identification and quantification of volatile compounds
199 200 201	<ul><li>nasal mucosa.</li><li>2.5. Identification and quantification of volatile compounds</li><li>The volatile components were identified by comparing their electron ionization</li></ul>
199 200 201 202	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds</li> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> </ul>
199 200 201 202 203	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds</li> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> <li>and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices</li> </ul>
199 200 201 202 203 203	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds</li> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> <li>and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices</li> <li>(RIs) and odour descriptions described in the literature and in online databases</li> </ul>
199 200 201 202 203 204 205	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds</li> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> <li>and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices</li> <li>(RIs) and odour descriptions described in the literature and in online databases</li> <li>(http://www.flavornet.org; http://www.odour.org.uk).</li> </ul>
199 200 201 202 203 204 205 206	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds The volatile components were identified by comparing their electron ionization (EI) spectra with the database records provided by the National Institute of Standards and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices (RIs) and odour descriptions described in the literature and in online databases (http://www.flavornet.org; http://www.odour.org.uk). Quantitative analysis of the volatile compounds was performed using a</li></ul>
199 200 201 202 203 204 205 206 207	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds</li> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> <li>and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices</li> <li>(RIs) and odour descriptions described in the literature and in online databases</li> <li>(http://www.flavornet.org; http://www.odour.org.uk).</li> <li>Quantitative analysis of the volatile compounds was performed using a</li> <li>calibration method with an internal standard (Zhou, Chong, Ding, Gu, &amp; Liu, 2016).</li> </ul>
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199 200 201 202 203 204 205 206 207 208 209	<ul> <li>nasal mucosa.</li> <li>2.5. Identification and quantification of volatile compounds <ul> <li>The volatile components were identified by comparing their electron ionization</li> <li>(EI) spectra with the database records provided by the National Institute of Standards</li> <li>and Technology (NIST) Mass Spectral Library (Version 2.0), GC retention indices</li> <li>(RIs) and odour descriptions described in the literature and in online databases</li> <li>(http://www.flavornet.org; http://www.odour.org.uk).</li> <li>Quantitative analysis of the volatile compounds was performed using a</li> <li>calibration method with an internal standard (Zhou, Chong, Ding, Gu, &amp; Liu, 2016).</li> <li>The concentrations of the volatile constituents were measured by the calibration</li> <li>curves of the GC-peak area and the amount ratios for the target analyte relative to</li> </ul> </li> </ul>

2-methyl-3-heptanone. The final results were expressed as µg volatile compounds/kg of the boiled pork. Each value represented the average of triplicate determinations. 2-methyl-3-heptanone was used as the internal standard without considering the calibration factors, that is, all calibration factors were considered to be 1.00. The involved equation can be written as follows:  $\operatorname{Conc}\left(\frac{\mu g}{kg}\right) = \frac{\operatorname{Peak area ratio}\left(\frac{\operatorname{volatile}}{IS}\right) \times 0.41 \mu g(IS)}{5 \text{ g(boiled pork)}} \times 1000$ OAVs were calculated according to the method of Liu, He, and Song (2018) using the following addition equation:  $OAV_i = \frac{C_i}{OT_i}$ where  $C_i$  is the concentration of the compound in the boiled pork and  $OT_i$  is the odour threshold in water. OT<sub>i</sub> was obtained from the online database (http://www.odour.org.uk) and some references related to flavour. Compounds with and  $OAV \ge 1$  were considered to be the main contributors to total flavour. 2.6. E-nose analysis In this study, the odour profile of different boiled pork samples was discriminated using a portable electronic nose (PEN3) that operates with an enrichment and desorption unit (EDU) from Win Muster Airsense Analytics, Inc. (Airsense, Germany). This instrument consisted of a sampling apparatus, a detector unit that contains ten metal oxide sensors (Gao, Liu, An, Zhang, Ma, & Cui, 2017), and pattern identification software for data recording and elaboration (Wang, Wang, Liu, & Liu, 2012). Table 1 lists all sensors and their major applicants.

Approximately 1.00 g of a boiled pork sample was added to a 10 ml glass vial. A filtered and dried air flow (99%, 300 ml/min) was used as a carrier gas for E-nose detection. The data acquisition period lasted for 60 s, and an additional 180 s was required for system rebalance. For each sample, the E-nose analysis was repeated three times for the same conditions. 2.7. Statistical analysis The contents of all volatile compounds and OAVs of the odour-active compounds were performed using one-way analysis of variance (ANOVA) and 

Duncan's multiple range tests in the SPSS software (v. 19.0, SPSS, Inc., Chicago, IL,

USA). The significance level was set at P < 0.05. Principal component analysis 

(PCA), agglomerative hierarchical clustering (AHC) and partial least 

squares-discriminant analysis (PLS-DA) were performed based on the odour-active

compounds (OAV > 1) using the software XLSTAT (2016) from Addinsoft 

(Barcelona, Spain). The odour-active compounds with variable importance in the 

projection (VIP) score > 1 in the PLS-DA analysis and a p-value < 0.05 in the 

ANOVA were considered significantly different among all boiled pork samples.

E-nose linear discriminate analysis (LDA) was conducted using the WinMuster 

software (version 1.6, Airsense Analytics, Schwerin, Germany) to differentiate the 

boiled pork samples according to the overall flavour. 

#### 3. Results and discussion

3.1. Volatile profiling of boiled pork by GC-MS/O 

#### 3.1.1. Volatile composition of boiled pork

253	A total of 61 volatile components were identified in boiled pork from the Triceps
254	brachii and Biceps femoris muscles from different pig breeds by SPME-GC-MS/O, as
255	is shown in Table 2. These compounds can be classified into nine chemical families,
256	including aldehydes (50.5%-65.7%, 25/61), alcohols (4.8%-10.3%, 7/61), ketones
257	(0.4%-0.9%, 1/61), esters (2.8%-12.3%, 3/61), aromatics (0.6%-8.4%, 8/61),
258	hydrocarbons (2.1%-5.7%, 8/61), furans (5.3%-7.7%, 3/61), N-containing compounds
259	(3.9%-6.5%, 2/61) and S-containing compounds (4.0%-9.5%, 4/61). The results
260	showed that most of these compounds have been reported in the three different pig
261	breeds (Pan, Yang, Zhu, and Wu, 2014). Among them, the largest number of
262	aldehydes were found in boiled pork, followed by hydrocarbons, and aromatic
263	compounds. Moreover, aldehyde compounds, which accounted for greater than 50.0%
264	of the total volatile compounds, were the most abundant in boiled pork.
265	The concentration ratios and quantities of each group of volatile compounds in
266	boiled pork is presented in Table 2. For the pig breeds yielding the boiled pork, there
267	were 52, 44 and 54 volatile compounds in DLY, SMX and TB, respectively. The
268	proportions of aldehydes and ketones (64.2%-65.7% and 0.5%-0.9%, respectively)
269	were highest in TB, and the ratios of ethers, furans and S-containing compounds
270	(9.0%-12.3%, 6.7%-7.7% and 8.6%-9.5%, respectively) were the highest in SMX. In

271	contrast, aromatic compounds (7.7%-8.4%) were the most abundant in DLY. Owing
272	to the main flavour of pork from aldehydes, furans and S-containing compounds and
273	their presence in TB and SMX, TB and SMX had significant contributions to overall
274	flavour. This result is in accordance with the study of Zhao et al., (2017). With respect
275	to the parts for boiled pork, the major volatile components in the Triceps brachii and
276	Biceps femoris muscles of DLY were aldehydes, alcohol and aromatics, which
277	accounted for total concentrations of 55.1%-56.5%, 8.6%-10.3% and 6.6%-8.8%,
278	respectively. The abundant volatiles in Triceps brachii and Biceps femoris muscles
279	from SMX were aldehydes, ethers and S-containing compounds, which maintained
280	the relationship of aldehydes $>$ ethers $>$ S-containing compounds. The main volatile
281	compounds in Triceps Brachii and Biceps Femoris from TB were aldehydes
282	(64.2%-65.7%) and S-containing compounds (6.2%-8.1%). These analyses concluded
283	that aldehydes and S-containing compounds had a dominant role in cooked pork
284	(Aaslyng & Meinert 2017).
285	Qualitative and quantitative analyses of the volatile components in boiled pork
286	from different pig breeds are listed in Table 3. Aldehyde compounds, similar to the
287	important volatile compounds in all types of meat products, were produced primarily
288	by lipid oxidation and degradation reactions. Strecker degradation products of amino
289	acids (Zhao et al., 2017; Li, Li, Zhang, Wang, Tang, and Chen, 2016) are also known
290	to be major contributors to the unique flavour of cooked pork due to their low odour
291	threshold (Lorenzo & Fonseca 2014). In this study, aldehydes were the most abundant
292	groups and had the highest number of compounds in boiled pork samples. Eight of

293	these compounds were simultaneously detected in all the boiled pork samples,
294	including four alkenals (hexanal, heptanal, nonanal and hexadecanal), two alkadienals
295	((E)-2-octenal and $(E)$ -2-nonenal) and two phenyl-containing aldehydes
296	(benzaldehyde and 4-ethylbenzaldehyde). The four alkenals and two alkadienals are
297	unsaturated fatty acid degradation products (Karahadian & Lindsay 1989). Meanwhile,
298	the two phenyl-containing aldehydes are usually derived from the Strecker reaction
299	(MacLeod, Ames, & Betz, 1988). Hexanal was the most abundant aldehyde and
300	presented grassy notes, while ( <i>E</i> )-2-octenal had a low odour threshold (3 $\mu$ g·kg <sup>-1</sup> ) and
301	was described as having fatty notes (Gu, Wang, Tao, & Wu, 2013; Wang et al., 2018).
302	Moreover, the hexanal and $(E)$ -2-octenal contents in DLY were significantly higher
303	(P < 0.01) than in SMX and TB, indicating that the extent of lipid oxidation in DLY
304	was greater. The nonanal and benzaldehyde contents in TB were significantly higher
305	(P < 0.001) than in DLY and SMX. This showed that TB had an advantage in the
306	contribution to fruity and floral notes. Additionally, $(E,E)$ -2,4-heptadienal and
307	9,12,15-octadecatrienal were exclusively found in boiled pork from SMX and TB and
308	promoted a sweeter and fruit aroma (Allen & Hamilton 1989).
309	Alcohols are mainly generated by the oxidative decomposition of lipids (Zou,
310	Kang, Liu, Qi, Zhou, & Zhang, 2018). Compared with short straight chain alcohols,
311	long chain alcohols are considered to have more contributions to the aroma of meat
312	products due to their lower odour thresholds (Li, Li, Zhang, Wang, Tang, & Chen,
313	2016). Seven alcohols were detected in this study, including three straight chain
314	alcohols (1-pentanol, 1-hexanol and 1-octanol) and four branched chain alcohols

315	(1-octen-3-ol, 2-hexyldecanol, $(E)$ -2-octen-1-ol and anethole). Among these volatile
316	compounds, 1-octen-3-ol, 1-octanol and $(E)$ -2-octen-1-ol were found in all three
317	varieties of boiled pork. The average contents of 1-octen-3-ol, with mushroom notes,
318	and $(E)$ -2-octen-1-ol, with green apple notes, in boiled pork from DLY were
319	significantly ( $P < 0.01$ ) higher than those from TB and SMX. Moreover,
320	2-hexyldecanol was only present in DLY, which indicated that it contributes more
321	pleasant fruity and floral aromas (Petričević, Radovčić, Lukić, Listeš, & Medić, 2018)
322	to overall flavour.
323	Furan, nitrogen and sulphur-containing compounds are well known as important
324	heterocyclic compounds in meat products (Wang et al., 2018). Among the three furan
325	compounds, 2-pentylfuran, with a fruity and buttery odour, had the highest contents
326	$(108.5-244.0 \ \mu g \cdot k g^{-1})$ in all the boiled pork, which could be due to linoleic acid
327	oxidization (Aparicio, Morales, & Alonso, 1996). 2-ethylfuran and 2-furanmethanol
328	usually have pungent and caramel odours and have been reported in cooked meat (Gu,
329	Wang, Tao, & Wu, 2013; Yang, Pan, Zhu, & Zou, 2014). For two nitrogen-containing
330	compounds, the contents of pyridine and 2-acetylpyrazine were significantly higher
331	(P < 0.01) in boiled meat from pig <i>Triceps brachii</i> muscle than in the boiled meat
332	from pig Biceps Femoris muscle. Furthermore, 3-methylthiophene and benzothiazole
333	were very abundant in the boiled meat from pig Biceps Femoris muscle, while the
334	amounts of dimethyl disulphide and 2-acetylthiazole identified in TB were greater
335	than in DLY and SMX. The comparative analysis indicated that these four
336	sulphur-containing compounds in boiled pork of Biceps Femoris muscle in TB are

regarded as the major contributor to the cooked cabbage and roasted flavours (Zhou, Chong, Ding, Gu, & Liu, 2016). Previous reported noted that these compounds might originate from sulphur amino acids (free, peptidic and proteinic amino acids), thiamine or glutathione (Girard & Durance, 2010). Eight aromatic hydrocarbons and eight aliphatic hydrocarbons were identified in all the boiled pork samples. All of these compounds are usually formed by lipid oxidation (Kang, Gao, Ge, Zhou, & Zhang, 2017). Hydrocarbons had few effects on the aromatic profiles of meat products due to their high odour thresholds (Qi, Liu, Zhou, & Xu, 2017). Compared to boiled pork from SMX and TB, there were more and greater amounts of hydrocarbons in boiled pork from DLY, which may be due to the higher levels of lipid oxidation. Ester compounds can be formed by the esterification of acids and alcohols. A previous study showed that short-chain esters have fruity notes and long-chain esters have fatty notes (Wang et al., 2018). Terpinyl acetate and ethyl hexanoate were only found in SMX and may be used to distinguish boiled pork from different pig breeds. Vinyl hexanoate was present in all the investigated boiled pork samples. 3.1.2. Odour-active compounds in boiled pork The odour-active compounds in this study were defined as the compounds with OAVs > 1. The OAVs of odour-active compounds in boiled pork are presented in Table 4. Statistical analysis showed that the OAVs of 25 volatile compounds showed 

significant differences (P < 0.05) in boiled pork from the three pig breeds. Among the 

three varieties of boiled pork, TB contained the largest number of the odour-active

359	constituents, including fourteen aldehydes, three alcohols, one hydrocarbon, two
360	furans, one N-containing compound and three S-containing compounds (total of
361	twenty-five), which indicated that TB displayed the most overall flavour among the
362	meat samples. Compared with Triceps Brachii muscle from TB, there was a greater
363	variety of odour-active compounds in Biceps Femoris muscle from TB. SMX
364	contained 22 aroma-active constituents, which was the fewest odour-active
365	compounds among the boiled pork from the three different pig breeds. For boiled pork
366	from both the Triceps Brachii and Biceps Femoris muscles, the OAVs of half of the
367	odour-active compounds did not show significant differences ( $P > 0.05$ ), indicating
368	that the muscles (Triceps Brachii and Biceps Femoris) presented similar flavour
369	characteristics. In a word, the breed was considered as the main influencing factor for
370	the overall flavour of boiled meat.
371	As shown in Table 4, the following seven odour-active constituents with
372	relatively high OAVs were detected in all samples: hexanal (OAV at 213.9-524.2),
373	nonanal (OAV at 248.7-454.6), 1-octen-3-ol (OAV at 56.9-194.3), dimethyl
374	disulphide (OAV at 76.8-141.3), heptanal (OAV at 19.0-41.7), 2-pentylfuran (OAV at
375	18.1-40.7) and 2-ethylfuran (OAV at 17.3-22.9). These constituents were regarded as
376	key odour-active compounds due to their significant contributions to the integral
377	flavour. Linear aldehydes such as hexanal, nonanal and heptanal, with grass and fatty
378	notes (Petričević, Radovčić, Lukić, Listeš, & Medić, 2018), come from lipid oxidation
379	and may contribute to the overall flavour. 1-Octen-3-ol was the only alcohol among
380	the odour-active compounds and has been reported to be generated by $\beta$ -oxidation

(Girard & Durance, 2010). Two furans, namely, 2-pentylfuran and 2-ethylfuran, might impart rubber and sweet flavours to the boiled pork, respectively. Dimethyl disulphide, with cooked cabbage notes, is an important fraction of aroma in fish paste products (Giri, Osako, & Ohshima, 2010). 3.2. Discrimination of boiled pork by PCA, PLS-DA and AHC To better visualize the data and reduce the dimensions of the original variables, PCA was performed to discriminate the boiled pork from the Triceps Brachii and Biceps Femoris muscles in the three pig breeds. Twenty-five odour-active compounds (OAVs > 1) were analysed by PCA. PCA scoring and a loading plot are presented in Fig. 1. The first two principal components account for 34.94% and 27.81% of the variance, respectively, (62.76% in total). The six sample groups are clearly were well discriminated from one another. A clear separation between SMX and DLY can be observed for PC1, while TB was significantly different from SMX and DLY with respect to PC2. As shown in Fig. 1 and Table 5, nine aldehydes (hexanal, r=0.727; heptanal, r=0.740; (*E*)-2-octenal, r=0.945; (*E*)-2-nonenal, r=0.617; (*E*,*E*)-2,4-nonadienal, r=0.846; (*E*,*E*)-2,4-decadienal, r=0.790; 1-octen-3-ol, r=0.747 and (E)-2-octen-1-ol, r=0.708) and one hydrocarbon (styrene, r=0.892) had high correlation coefficients with the positive side of PC1, which were present in DLY with high OAVs (Table 3). In contrast, only 2-acetylthiazole (r=-0.909), 2-ethlfuran (r=-0.677) and dimethyl disulphide (r=-0.671) showed high correlation coefficients with the negative side of PC1. Moreover, PC2 on the positive axis was highly influenced by benzaldehyde (r=0.988), nonanal (r=0.923), octanal (r=0.791), anethole

403	(r=0.863) and ( $E$ )-2-nonenal (r=0.671), indicating that these compounds were the
104	important odour-active compounds in TB (Fig. 1), while PC2 on the negative axis was
405	highly influenced by ethyl hexanoate and ( $E$ )-2-octen-1-ol (r=-0.682). Hence,
406	( $E$ )-2-nonenal was highly associated with the PC1 and PC2 positive axis and
407	( $E$ )-2-octen-1-ol was highly associated with the PC1 positive and PC2 negative axes.
108	However, 2-acetylpyrazine and decanal were lowly relevant to the plane (correlation
109	coefficients $< 0.400$ ). This suggested that these two compounds could not be
<b>1</b> 10	described by PC1 and PC2. Additionally, according to the VIP scores in the PLS-DA
111	analysis and p-values in the ANOVA of the studied odour-active compounds (Table
112	5), it could be concluded that a total of twelve volatile compounds with a VIP score $>$
413	1 and p-value $< 0.05$ were considered as potential flavour markers for the
414	differentiation of boiled pork. These odour-active compounds were
415	(E,E)-2,4-decadienal, ethyl hexanoate, dimethyl disulphide, hexanal, 2-acetylthiazole,
116	(E)-2-nonenal, 1-octen-3-ol, (E,E)-2,4-nonadienal, heptanal, (E)-2-octen-1-ol, styrene
117	and (E)-2-octenal.
118	AHC can be used to depict the similarities and differences among different
119	boiled pork. Ward's method with a metric of Euclidean distance was applied in this
120	study. The results as a dendrogram are presented in Fig. 2. The boiled meat samples
121	were divided into three clusters. The third cluster included Triceps Brachii and Biceps
122	Femoris muscles from SMX, with the lowest dissimilarity index, indicating that the
123	Triceps Brachill and Biceps Femoris muscles from SMX have the most similar
124	volatile profiles. Similarly, the second cluster, with Triceps Brachii and Biceps

Femoris muscles from DLY, possessed similar volatile profiles. The first cluster consisted of Triceps Brachii and Biceps Femoris muscles from TB and had the highest dissimilarity index; this illustrated that the overall flavour of Triceps Brachii and Biceps Femoris muscles from TB was greatly different from that of the boiled pork from the other two pig varieties. 3.3. Volatile profiling of boiled pork using E-nose The signal from 10 sensors in response to volatile compounds from different boiled pork are presented in Fig. 3. Fig. 3a-c show the that responses of all ten sensors to boiled pork from Triceps Brachii and Biceps Femoris muscles from the three pig breeds had no significant differences, which explained their similar flavour. This also showed that W5S (broad-range nitrous oxides), W1W (terpenes and sulphur-containing organic compound), and W2W (aromatics and organic sulphides) had higher responses than other sensors, which suggested that they may contain more heterocyclic compounds, such as furans and N- and S-containing compounds. The response values of W1C and W3C were less than one. Both sensors were mainly sensitive to aroma components and ammonia. As shown in Fig. 3d and e, the signals from the W5S and W1W sensors to the boiled pork from the three pig breeds obviously varied. Sensor W5S showed stronger responses to SMX, and while sensor W1W showed weaker responses to that breed. In a word, this result indicated that the influence of different pig breeds on flavour is greater than from different pig parts for boiled pork. 

*3.4. Discrimination of boiled pork by PCA and LDA* 

447	E-nose analysis was performed to obtain a description of the odour profiles of
448	boiled pork from different pig parts and breeds, and the PCA analysis results are
449	shown in Fig. 4a. The plot consists of two axes showing PC1 and PC2, which could
450	explain 99.46% of the total variance. PC1 accounted for 94.83% and PC2 accounted
451	for 4.63%. The contribution variance of PC1 and PC2 is over 90%, indicating that the
452	first two PCs were sufficient to explain the maximum variation in the original types of
453	boiled pork. According to Fig. 4a, the dots corresponding to the Triceps Brachii and
454	Biceps Femoris muscles from SMX and DLY had some overlap on PC1, and the
455	sample points for the Triceps Brachii and Biceps Femoris muscles for TB were close
456	to one another. Thus, the boiled pork samples can be divided into three groups (SMX,
457	DLY and TB). This result illustrated that the boiled pork from different pig breeds
458	had significantly different flavours, and that boiled pork from the Triceps Brachii and
459	Biceps Femoris muscles of pigs had similar aroma compositions.
460	An LDA was also performed to investigate the similarities and differences
461	among the six sample groups. As shown in Fig. 4b, the first two PCs account for
462	88.58% of the variation in the data. Comparative analysis of all samples including
463	each boiled pork sample could be clearly discriminated using LDA. The sample
464	points for the Triceps Brachii and Biceps Femoris muscles from DLY and TB were
465	close on PC1 (72.58%) and PC2 (16.00%), respectively. Moreover, the SMX Triceps
466	Brachii and Biceps Femoris muscles clustered together on PC1 and PC2, which
467	demonstrated that the odour profiles of the boiled pork from the Triceps Brachii and
468	Biceps Femoris muscles of pigs appeared similar. The boiled pork from different pig

breeds were far from one another. This demonstrated that there were considerable differences in their flavours. Both multivariate analyses (PCA and LDA) performed on E-nose data were suitable to distinguish the boiled pork based on its flavour profile.

4. Conclusions 

In this study, a total of 61 volatile compounds were identified and quantified in boiled pork from different pig breeds by SPME-GC-MS/O. These compounds can be divided into nine categories: aldehydes, alcohols, ketones, esters, aromatics, hydrocarbons, furans, N- and S-containing compounds. The key odour-active volatiles from the evaluated samples were hexanal (OAV at 213.9-524.2), nonanal (OAV at 248.7-454.6), 1-octen-3-ol (OAV at 56.9-194.3), dimethyl disulphide (OAV at 76.8-141.3), heptanal (OAV at 19.0-41.7), 2-pentylfuran (OAV at 18.1-40.7) and 2-ethylfuran (OAV at 17.3-22.9), which significantly contribute to the overall flavour. Moreover, according to multicomponent statistics analyses, including PCA, PLS-DA, AHC and LDA, the boiled pork from different pig breeds could be classified into three separate groups. Twelve odour-active compounds were confirmed as potential flavour makers for the differentiation of boiled pork among the three pig breeds. Overall, it can be concluded that the characterization and differentiation of boiled pork from TB, SMX and DLY pigs by volatiles profiling and chemometrics analysis has the potential to be a feasible method to evaluate pork from different breeds. Moreover, further studies should include more pork samples to build a more reliable data model and validate key flavour compounds by aroma recombination and

491 omission analysis.

#### 492 Compliance with Ethical Standards

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497 Conflicts of Interest: Dong Han declares that he has no conflict of interest. Chun-Hui

498 Zhang declares that he has no conflict of interest. Marie-Laure Fauconnier declares

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500 Ethical Approval: All applicable international, national, and/or institutional

501 guidelines for the care and use of animals were followed.

502 Informed Consent: Informed consent was obtained from all individual participants503 included in the study.

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Biplot (axes PC1 and PC2: 62.76 %)

**Fig. 1** PCA for odour-active compounds of the different boiled pork (TB1 = *Triceps Brachii* muscle of Tibetan pigs, TB2 = *Biceps Femoris* muscle of Tibetan pigs, DLY1 = *Triceps Brachii* muscle of Duroc × (Landrace × Yorkshire), DLY2 = *Biceps Femoris* muscle of Duroc × (Landrace × Yorkshire), SMX1 = *Triceps Brachii* muscle of Sanmenxia pigs, SMX2 = *Biceps Femoris* muscle of Sanmenxia pigs). The blue dots represent the samples from the different boiled pork, and the red dots respsent odour-active compounds.



**Fig. 2** AHC results of the different boiled pork (TB1 = *Triceps Brachii* muscle of Tibetan pigs, TB2 = *Biceps Femoris* muscle of Tibetan pigs, DLY1 = *Triceps Brachii* muscle of Duroc × (Landrace × Yorkshire), DLY2 = *Biceps Femoris* muscle of Duroc × (Landrace × Yorkshire), SMX1 = *Triceps Brachii* muscle of Sanmenxia pigs, SMX2 = *Biceps Femoris* muscle of Sanmenxia pigs).











**Fig. 3** Radar charts of E-nose data from DLY1 and DLY2 (a), SMX1 and SMX2 (b), TB1 and TB2 (c), DLY1, SMX1 and TB1 (d), DLY2, SMX2 and TB2 (e) (TB1 = *Triceps Brachii* muscle of Tibetan pigs, TB2 = *Biceps Femoris* muscle of Tibetan pigs, DLY1 = *Triceps Brachii* muscle of Duroc × (Landrace × Yorkshire), DLY2 = *Biceps Femoris* muscle of Duroc × (Landrace × Yorkshire), SMX1 = *Triceps Brachii* muscle of Sanmenxia pigs, SMX2 = *Biceps Femoris* muscle of Sanmenxia pigs).



**Fig. 4** PCA (a) and LDA (b) plot of e-nose response from different boiled pork (TB1 = *Triceps Brachii* muscle of Tibetan pigs, TB2 = *Biceps Femoris* muscle of Tibetan pigs, DLY1 = *Triceps Brachii* muscle of Duroc  $\times$  (Landrace  $\times$  Yorkshire), DLY2 = *Biceps Femoris* muscle of Duroc  $\times$  (Landrace  $\times$  Yorkshire), SMX1 = *Triceps Brachii* muscle of Sanmenxia pigs, SMX2 = *Biceps Femoris* muscle of Sanmenxia pigs).

Iubi		description and sensitivity of metal oxide sensors	for TER(5 electronic hose
No.	Sensor name	Performance description	Reference
1	W1C	Benzene and aromatic compounds	Methylbenzene, 10 ppm
2	W5S	Broad range sensitivity, very sensitive to	Nitrogen dioxide, 1 ppm
		nitrogen oxides	
3	W3C	Ammonia, sensitive to aromatic compounds	Benzene, 10 ppm
4	W6S	Mainly hydrogen, selectively	Hydrogen, 100 ppm
5	W5C	Alkane, aromatic compounds	Propane, 1 ppm
6	W1S	Sensitive to methane, broad range.	Methane, 100 ppm
7	W1W	Sensitive to many sulfur organic compounds	Hydrogen sulfide, 1 ppm
		and terpenes.	
8	W2S	Alcohol, sensitive to aromatic compounds	Carbon monoxide, 100
		with broad range, similar to No. 6.	ppm
9	W2W	Aromatic compounds and sulfur organic	Hydrogen sulfide, 1 ppm
		compounds	
10	W3S	Reacts on high concentrations, very sensitive	Methane, 100 ppm
		to several compounds	

 Table 1. Performance description and sensitivity of metal oxide sensors for PEN3 electronic nose

	Ratio% (qu	antities)								
Classes of components	Triceps Bra	<i>chii</i> muscle		Biceps Fem	oris muscle		Triceps Brachii	and Biceps Fem	oris muscle	Breeds and parts
	DLY	SMX	TB	DLY SMX		TB	DLY	SMX	ТВ	DLY-SMX-TB
Aldehydes	56.5 (16)	50.0 (18)	65.7 (12)	55.1 (20)	56.2 (17)	64.2 (23)	55.1-56.5 (21)	50.0-56.2 (20)	64.2-65.7 (24)	50.5-65.7 (25)
Alcohols	8.6 (5)	9.2 (4)	4.8 (6)	10.3 (6)	8.5 (3)	6.1 (4)	8.6-10.3 (7)	8.5-9.2 (4)	4.8-6.1 (6)	4.8-10.3 (7)
Ketones	0.6 (1)	0.4 (1)	0.5 (1)	0.4 (1)	0.5 (1)	0.9 (1)	0.4-0.6 (1)	0.4-0.5 (1)	0.5-0.9 (1)	0.4-0.9 (1)
Ethers	8.0 (1)	12.3 (3)	3.6 (1)	6.6 (1)	9.0 (2)	2.8 (1)	6.6-8.0 (1)	9.0-12.3 (3)	2.8-3.6 (1)	2.8-12.3 (3)
Aromatics	8.4 (7)	0.6 (2)	3.4 (4)	7.3 (7)	0.9 (3)	4.4 (8)	7.3-8.4 (7)	0.6-0.9 (3)	3.4-4.4 (8)	0.6-8.4 (8)
Hydrocarbons	3.4 (4)	2.4 (4)	2.1 (3)	3.1 (6)	3.8 (3)	5.7 (5)	3.1-3.4 (6)	2.4-3.8 (4)	2.1-5.7 (5)	2.1-5.7 (8)
Furans	5.4 (3)	7.7 (3)	5.3 (3)	4.9 (3)	6.7 (3)	5.8 (3)	4.9-5.4 (3)	6.7-7.7 (3)	5.3-5.8 (3)	5.3-7.7 (3)
N-containing compounds	5.0 (2)	8.0 (2)	6.5 (2)	6.0 (2)	5.8 (2)	3.9 (2)	5.0-6.0 (2)	5.8-8.0 (2)	3.9-6.5 (2)	3.9-6.5 (2)
S-containing compounds	4.0 (4)	9.5 (4)	8.1 (4)	6.2 (4)	8.6 (4)	6.2 (4)	4.0-6.2 (4)	8.6-9.5 (4)	6.2-8.1 (4)	4.0-9.5 (4)
Total	100.0 (43)	100.0 (41)	100.0 (36)	100.0 (50)	100.0 (38)	100.0 (51)	100.0 (52)	100.0 (44)	100.0 (54)	100.0 (61)

Table 2. Concentration ratios and quantities of volatile composition of boiled pork from different breeds of pigs.

*Note:* DLY, Duroc × (Landrace × Yorkshire); SMX, Sanmenxia pig; TB, Tibetan pig.

		<sup>2</sup> DD 5	<sup>3</sup> Identification $-\frac{7}{2}$	Triceps B	<i>rachii</i> mus	cle			Biceps Fe	<i>emoris</i> muse	ele			Sign. parts		
Compounds	DB-wax	-DB-2	Identification	DLY	SMX	ТВ	SEM	Sign.	DLY	SMX	TB	SEM	Sign.	DLY	SMX	TB
Aldehydes (25)				3506.0 <sub>a</sub> <sup>x</sup>	1626.5 <sup>y</sup>	2832.6 <sub>b</sub> <sup>y</sup>	275.4	***	2158.4 <sup>y</sup>	$2023.7^{x}_{b}$	3511.9 <sup>x</sup>	238.7	***	***	**	***
2,3-Dimethylpentanal	<900	-	MS,RI	$0.0_{b}^{y}$	$0.0_{b}^{y}$	13.8 <sup>x</sup>	2.3	***	2.8 <sup>x</sup>	$4.7_{b}^{x}$	$13.0_{a}^{x}$	1.6	***	***	***	NS
Pentanal	927	702	MS,RI,O	$41.7_{b}^{y}$	$11.7_{c}^{y}$	$145.3_{a}^{x}$	20.2	***	$60.9_{a}^{x}$	$26.0_{b}^{x}$	0.0c <sup>y</sup>	8.8	***	**	***	***
Hexanal	1074	797	MS,RI,O	2620.9 <sup>x</sup>	1128.3 <sup>v</sup>	1509.5 <sub>b</sub> <sup>y</sup>	224.5	***	1373.3 <sup>a</sup>	1341.1 <sup>x</sup>	1069.6 <sub>b</sub> <sup>x</sup>	51.1	**	***	*	**
Heptanal	1175	900	MS,RI,O	$125.2_{a}^{x}$	56.9 <sup>x</sup>	96.7 <sup>x</sup>	10.2	***	85.3 <sub>b</sub> <sup>y</sup>	$62.0^{x}_{c}$	$91.6_{a}^{x}$	4.6	***	**	NS	NS
Octanal	1281	1002	MS,RI,O	82.2 <sub>b</sub> <sup>x</sup>	0.0 <sup>y</sup>	87.0 <sub>a</sub> <sup>x</sup>	14.1	***	57.0 <sup>y</sup>	$62.2_{b}^{x}$	85.1 <sup>y</sup>	4.4	***	***	***	*
Nonanal	1388	1112	MS,RI,O	333.8 <sub>b</sub> <sup>x</sup>	255.7 <sup>x</sup>	$454.6_{a}^{x}$	29.4	***	$267.4_{b}^{y}$	$248.7_{b}^{x}$	362.6 <sup>y</sup>	18.1	***	**	NS	**
(E)-2-Octenal	1424	1062	MS,RI,O	$47.2_{a}^{x}$	$21.4_{b}^{x}$	23.0 <sub>b</sub> <sup>x</sup>	4.2	***	31.3 <sup>y</sup>	23.3 <sup>x</sup>	27.0 <sub>b</sub> <sup>x</sup>	1.3	**	**	NS	NS
(E,E)-2,4-Heptadienal	1486	1015	MS,RI,O	$0.0_{b}$	$0.0_{b}$	$9.2_{a}^{y}$	1.5	***	$0.0_{b}$	$0.0_{b}^{y}$	$10.8_{a}^{x}$	1.8	***	NS	NS	**
Decanal	1494	1210	MS,RI,O	$0.0_{b}^{y}$	$11.0_{a}^{x}$	$0.0_{b}^{y}$	1.8	***	$21.7_{a}^{x}$	0.0c <sup>y</sup>	$13.8_{b}^{x}$	3.2	***	***	***	***
Benzaldehyde	1514	926	MS,RI,O	135.9 <sub>b</sub> <sup>x</sup>	$60.2^{y}_{c}$	$294.5_{a}^{x}$	34.5	***	$107.2_{b}^{y}$	113.0 <sub>b</sub> <sup>x</sup>	$276.3_{a}^{y}$	27.8	***	**	***	*
(E)-2-Nonenal	1531	1165	MS,RI,O	$16.6_{a}^{x}$	8.7 <sup>y</sup>	$13.9_{b}^{y}$	1.2	***	$14.2_{b}^{y}$	$13.0_{b}^{x}$	$18.6_{a}^{x}$	0.9	***	***	**	***
cis-4-Decenal	1534	1202	MS,RI	$22.4_{a}^{x}$	$19.2_{b}^{x}$	$0.0_{c}^{y}$	3.5	***	$24.0_{a}^{x}$	$13.5_{b}^{y}$	$15.2_{b}^{x}$	1.7	***	NS	***	***
(E)-2-Decenal	1600	1265	MS,RI,O	$0.0_{b}$	13.6 <sup>y</sup>	$0.0_{b}^{y}$	2.3	***	0.0c	$18.1_{a}^{x}$	$14.2_{b}^{x}$	2.8	***	NS	**	***
2-Butyl-2-octenal	1665	1392	MS,RI	13.3 <sub>a</sub>	$7.0_{b}^{x}$	0.0 <sub>c</sub>	1.9	***	20.7 <sub>b</sub>	$7.2_{c}^{x}$	21.3 <sub>a</sub>	2.3	***	***	NS	***
9,12,15-Octadecatrienal	1675	-	MS,RI	$0.0_{b}$	5.3 <sup>x</sup>	$0.0_{b}$	0.9	***	0.0	$0.0^{\mathrm{y}}$	0.0	0.0	NS	NS	***	NS
(E,E)-2,4-Nonadienal	1695	1219	MS,RI,O	$8.9_{a}^{x}$	$4.1_{b}^{x}$	$0.0_{c}^{y}$	1.3	***	$7.6_{a}^{x}$	$4.1_{b}^{x}$	$7.4_{a}^{x}$	0.6	**	NS	NS	**
4-Ethylbenzaldehyde	1701	1173	MS,RI	$10.8_{b}^{x}$	5.5 <sup>y</sup>	33.5 <sup>y</sup>	4.3	***	$9.1_{b}^{x}$	$9.4_{b}^{x}$	$69.5_{a}^{x}$	10.0	***	NS	*	***
Dodecanal	1706	1418	MS,RI,O	$4.5_{a}^{y}$	$2.9_{b}^{y}$	$0.0_{c}^{y}$	0.7	***	$7.2_{b}^{x}$	$6.0_{b}^{x}$	$23.6_{a}^{x}$	2.8	***	**	**	***
2-Undecenal	1747	1153	MS,RI,O	$7.3_{a}^{x}$	$0.0_{b}$	$0.0_{b}^{y}$	1.2	***	$0.0_{b}^{y}$	$0.0_{b}$	$19.9_{a}^{x}$	3.5	**	***	NS	**
(E,E)-2,4-Decadienal	1758	1324	MS,RI,O	$11.4_{a}^{x}$	$5.8_{b}^{x}$	$0.0_{c}^{y}$	1.7	***	$4.0_{b}^{y}$	0.0 <sup>y</sup>	$7.5_{a}^{x}$	1.1	***	**	**	***
Tridecanal	1812	1525	MS,RI	$0.0_{b}^{y}$	$3.9_{a}^{y}$	$0.0_{b}^{y}$	0.7	***	$10.8^{x}_{b}$	$12.1_{b}^{x}$	$67.0_{a}^{x}$	9.3	***	***	***	***
Tetradecanal	1918	1626	MS,RI	$0.0^{\mathrm{y}}$	0.0	$0.0^{\mathrm{y}}$	0.0	NS	$9.4_{b}^{x}$	0.0 <sub>c</sub>	85.1 <sup>x</sup>	13.5	***	***	NS	***
4-Pentylbenzaldehyde	1999	1472	MS,RI	$0.0^{\mathrm{y}}$	0.0	$0.0^{\mathrm{y}}$	0.0	NS	$2.0_{b}^{x}$	0.0 <sub>c</sub>	$11.3_{a}^{x}$	1.8	***	***	NS	***
Pentadecanal	2025	1712	MS,RI	0.0	0.0	0.0 <sup>y</sup>	0.0	NS	$0.0_{b}$	$0.0_{b}$	$211.8_{a}^{x}$	35.3	***	NS	NS	***

**Table 3.** Identification and quantification of volatile compounds in boiled pork from different breeds of pigs by GC-MS/O ( $\mu g \cdot kg^{-1}$ ).

Hexadecanal	2132	1832	MS,RI	$24.0_{\rm b}{}^{\rm y}$	$5.3_{b}^{y}$	151.5 <sup>y</sup>	23.2	***	42.4 <sup>x</sup>	$59.2_{\rm b}^{\rm x}$	989.6 <sup>x</sup>	156.5	***	**	***	***
Alcohols (7)				534.2 <sup>x</sup>	298.6 <sup>x</sup>	208.9 <sup>°y</sup>	48.7	***	405.1 <sup>y</sup>	$305.5_{b}^{x}$	331.8 <sup>°</sup> <sub>b</sub> x	15.9	**	**	NS	**
1-Pentanol	1245	756	MS,RI,O	$72.7_{a}^{x}$	$32.0_{b}^{x}$	$31.4_{b}^{x}$	6.9	***	37.3 <sup>y</sup>	$0.0_{b}^{y}$	$0.0_{b}^{y}$	6.2	***	***	***	***
1-Hexanol	1347	865	MS,RI,O	$20.7_{a}^{x}$	0.0 <sub>c</sub>	$16.2_{b}^{y}$	3.1	***	$0.0_{b}^{y}$	$0.0_{b}$	$116.1_{a}^{x}$	19.4	***	***	NS	***
1-Octen-3-ol	1443	981	MS,RI,O	388.6 <sup>x</sup>	$228.9_{b}^{y}$	113.7 <sup>°</sup>	40.0	***	$306.4_{a}^{y}$	257.8 <sub>b</sub> <sup>x</sup>	155.0 <sup>x</sup>	22.9	***	**	*	*
2-Hexyldecanol	1546	-	MS,RI	$0.0^{y}$	0.0	0.0	0.0	NS	$5.3_{a}^{x}$	$0.0_{b}$	$0.0_{b}$	0.9	***	***	NS	NS
1-Octanol	1552	1072	MS,RI,O	19.6 <sup>x</sup>	13.3 <sup>y</sup>	$19.3_{a}^{x}$	1.1	**	$20.3_{b}^{x}$	$22.5_{a}^{x}$	$19.7_{b}^{x}$	0.5	*	NS	***	NS
( <i>E</i> )-2-Octen-1-ol	1608	1079	MS,RI,O	$32.7_{a}^{x}$	$24.4_{b}^{x}$	13.6 <sup>v</sup>	2.8	***	$30.6_{a}^{x}$	$25.2_{b}^{x}$	19.0 <sup>x</sup>	1.8	**	NS	NS	**
Anethole	1818	-	MS,RI,O	$0.0_{b}^{y}$	$0.0_{b}$	$14.7_{a}^{y}$	2.5	***	$5.2_{b}^{x}$	0.0 <sub>c</sub>	22.0 <sub>a</sub> <sup>x</sup>	3.3	***	***	NS	***
Ketones (1)				35.0 <sub>a</sub> <sup>x</sup>	13.6 <sup>x</sup>	23.6 <sub>b</sub> <sup>y</sup>	3.1	***	$16.8_{b}^{y}$	$16.3_{b}^{x}$	$47.3_{a}^{x}$	5.1	***	***	NS	***
(E,Z)-3,5-Octadien-2-one	1564	1094	MS,RI,O	35.0 <sub>a</sub> <sup>x</sup>	13.6 <sup>x</sup>	23.6 <sub>b</sub> <sup>y</sup>	3.1	***	$16.8_{b}^{y}$	$16.3_{b}^{x}$	$47.3_{a}^{x}$	5.1	***	***	NS	***
Esters (3)				$498.7_{a}^{x}$	398.8 <sub>b</sub> <sup>x</sup>	156.2 <sup>x</sup>	51.3	***	$259.2_{b}^{y}$	$322.7_{a}^{y}$	155.1 <sup>x</sup>	25.0	***	***	**	NS
Terpinyl acetate	1186	1361	MS,RI,O	$0.0_{b}$	$18.1_{a}^{x}$	$0.0_{b}$	3.0	***	0.0	$0.0^{\mathrm{y}}$	0.0	0.0	NS	NS	***	NS
Ethyl hexanoate	1225	1006	MS,RI,O	$0.0_{b}$	$7.8_{a}^{y}$	$0.0_{b}$	1.3	***	$0.0_{b}$	$8.9_{a}^{x}$	$0.0_{b}$	1.5	***	NS	**	NS
Vinyl hexanoate	1313	-	MS,RI	$498.7_{a}^{x}$	372.9 <sup>x</sup>	156.2 <sup>x</sup>	50.3	***	$259.2_{b}^{y}$	313.8 <sup>y</sup>	155.1 <sup>x</sup>	23.8	***	***	**	NS
Aromatics (8)				523.0 <sub>a</sub> <sup>x</sup>	$19.2^{y}_{c}$	$148.0_{b}^{y}$	75.6	***	$285.5_{a}^{y}$	34.0 <sup>x</sup>	239.5 <sup>x</sup>	38.7	***	***	***	***
Ethylbenzene	1112	872	MS,RI,O	$47.8_{a}^{x}$	$0.0_{b}$	$0.0_{b}^{y}$	8.0	***	$24.9_{a}^{y}$	0.0 <sub>c</sub>	$8.4_{b}^{x}$	3.7	***	***	NS	***
p-Xylene	1119	868	MS,RI	$71.8_{a}^{x}$	0.0 <sub>c</sub>	$34.7_{b}^{x}$	10.4	***	$40.5_{a}^{y}$	0.0 <sub>c</sub>	$35.8_{b}^{x}$	6.4	***	***	NS	NS
o-Xylene	1126	888	MS,RI	$96.9_{a}^{x}$	$0.0_{b}$	$0.0_{b}^{y}$	16.2	***	$56.5_{a}^{y}$	0.0 <sub>c</sub>	$34.1_{b}^{x}$	8.2	***	***	NS	
Styrene	1246	895	MS,RI,O	$244.1_{a}^{x}$	0.0 <sub>c</sub>	91.3 <sup>y</sup>	35.6	***	131.3 <sup>y</sup>	0.0 <sub>c</sub>	$111.7_{b}^{x}$	20.5	***	***	NS	**
1,2,4-Trimethylbenzene	1271	945	MS,RI	$15.9_{a}^{x}$	$0.0_{b}^{y}$	$0.0_{b}^{y}$	2.7	***	$4.5_{b}^{y}$	$8.0_{a}^{x}$	$4.5_{b}^{x}$	0.6	**	***	***	**
1,2,4,5-Tetramethylbenzene	1418	1218	MS,RI	37.2 <sup>x</sup> <sub>a</sub>	15.0 <sub>b</sub> <sup>y</sup>	0.0c <sup>y</sup>	5.4	***	$21.9_{a}^{y}$	$20.1_{a}^{x}$	13.2 <sup>x</sup>	1.4	**	**	**	***
Naphthalene	1733	-	MS,RI,O	$9.2_{b}^{x}$	4.2 <sup>v</sup>	13.2 <sup>y</sup>	1.3	***	$5.9_b^{y}$	$5.9_{b}^{x}$	$18.6_{a}^{x}$	2.1	***	**	*	**
2-Methylnaphthalene	1845	-	MS,RI	$0.0_{b}$	$0.0_{b}$	$8.7_{a}^{y}$	1.5	***	$0.0_{b}$	$0.0_{b}$	$13.4_{a}^{x}$	2.2	***	NS	NS	***
Hydrocarbons (8)				212.2 <sup>x</sup> <sub>a</sub>	$76.8_{c}^{y}$	$91.5_{b}^{y}$	21.5	***	$122.6_{b}^{y}$	135.6 <sub>b</sub> <sup>x</sup>	$314.2_{a}^{x}$	31.1	***	***	***	***
Limonene	1190	1032	MS,RI,O	$0.0^{y}$	0.0	0.0	0.0	NS	$14.4_{a}^{x}$	$0.0_{b}$	$0.0_{b}$	2.4	***	***	NS	NS
Dodecane	1195	1104	MS,RI,O	$39.7_{a}^{x}$	$15.7_{b}^{x}$	0.0 <sub>c</sub>	5.8	***	$10.4_{a}^{y}$	$0.0_{b}^{y}$	$0.0_{b}$	1.7	***	***	***	NS
Tridecane	1296	1301	MS,RI	$42.7_{a}^{x}$	$20.9^{y}_{c}$	$24.6_b^{y}$	3.4	***	$25.3_{b}^{y}$	$52.2_{a}^{x}$	$43.1_{a}^{x}$	4.2	**	**	**	**
3-Methyltridecane	1368	-	MS,RI	$0.0^{\mathrm{y}}$	0.0	0.0	0.0	NS	$6.4_{a}^{x}$	0.0 <sub>b</sub>	$0.0_{b}$	1.1	***	***	NS	NS

Tetradecane	1400	1401	MS,RI	103.3 <sup>x</sup>	29.0 <sup>y</sup>	32.5 <sub>b</sub> <sup>y</sup>	12.1	***	44.2 <sub>b</sub> <sup>y</sup>	46.8 <sup>x</sup>	$94.8_{a}^{x}$	8.2	***	***	**	***
Pentadecane	1501	1500	MS,RI	26.5 <sup>x</sup>	11.2 <sup>y</sup>	$34.4_{a}^{y}$	3.4	***	21.8 <sup>y</sup>	36.6 <sup>x</sup>	$138.2_{a}^{x}$	18.4	***	**	***	***
Longifolene	1574	1403	MS,RI	0.0	0.0	0.0 <sup>y</sup>	0.0	NS	$0.0_{b}$	$0.0_{b}$	$16.4_{a}^{x}$	2.7	***	NS	NS	***
Hexadecane	1600	1600	MS,RI	0.0	0.0	0.0 <sup>y</sup>	0.0	NS	$0.0_{b}$	$0.0_{b}$	$21.6_{a}^{x}$	3.6	***	NS	NS	***
Furans (3)				$335.7_{a}^{x}$	251.5 <sub>b</sub> <sup>x</sup>	226.6 <sup>y</sup>	16.9	***	192.0 <sup>v</sup>	$240.0_{b}^{y}$	316.3 <sup>x</sup>	18.2	***	***	NS	***
2-Ethylfuran	952	708	MS,RI,O	39.8 <sup>x</sup>	$42.5_{b}^{y}$	52.2 <sup>x</sup>	2.0	***	$41.1_{c}^{x}$	50.6 <sup>x</sup>	$47.7_{b}^{x}$	1.4	***	NS	**	NS
2-Pentylfuran	1218	988	MS,RI,O	$244.0_{a}^{x}$	162.6 <sup>x</sup>	132.6 <sup>y</sup>	17.0	***	$108.5_{c}^{y}$	155.5 <sup>x</sup>	$231.4_{a}^{x}$	18.0	***	***	NS	***
2-Furanmethanol	1625	851	MS,RI,O	$51.9_{a}^{x}$	$46.5_{b}^{x}$	41.8 <sup>x</sup>	1.5	***	$42.4_{a}^{y}$	33.9 <sup>v</sup>	37.3 <sup>y</sup>	1.3	**	**	**	*
N-containing compounds (2)				$311.0_{a}^{x}$	259.4 <sup>x</sup>	278.3 <sub>b</sub> <sup>x</sup>	7.6	***	$233.5_{a}^{y}$	209.7 <sup>y</sup>	214.6 <sub>b</sub> <sup>y</sup>	3.7	***	***	***	***
Pyridine	1156	751	MS,RI,O	$223.0_{a}^{x}$	175.2 <sup>x</sup>	183.6 <sup>x</sup>	7.4	***	$182.7_{a}^{y}$	$157.5_{b}^{y}$	152.8 <sup>y</sup>	4.7	***	***	**	***
2-Acetylpyrazine	1978	1095	MS,RI,O	88.0 <sub>b</sub> <sup>x</sup>	84.3 <sup>x</sup>	$94.7_{a}^{x}$	1.5	***	$50.8_{b}^{y}$	52.2 <sub>b</sub> <sup>y</sup>	$61.8_{a}^{y}$	1.8	***	***	***	***
S-containing compounds (4)				246.4 <sup>x</sup>	309.1 <sub>b</sub>	347.5 <sub>a</sub>	14.8	***	243.7 <sup>x</sup>	310.4 <sub>b</sub>	341.8 <sub>a</sub>	14.5	***	NS	NS	NS
Dimethyl disulphide	1109	785	MS,RI,O	84.5 <sup>x</sup>	128.2 <sub>b</sub> <sup>x</sup>	$155.5_{a}^{x}$	10.4	***	54.0 <sup>v</sup>	116.5 <sup>y</sup>	$140.2_{a}^{y}$	12.9	***	**	**	***
3-Methylthiophene	1185	773	MS,RI,O	$20.3_{a}^{y}$	17.2 <sub>b</sub> <sup>y</sup>	18.2 <sub>ab</sub> <sup>y</sup>	0.6	***	$25.7_{a}^{x}$	$27.4_{a}^{x}$	$25.4_{a}^{x}$	0.5	NS	**	**	**
2-Acetylthiazole	1632	1016	MS,RI,O	32.8 <sup>y</sup>	$68.3_{a}^{x}$	$69.5_{a}^{x}$	6.0	***	$42.6_{b}^{x}$	$49.9_{a}^{y}$	$51.3_{a}^{y}$	1.4	***	**	***	***
Benzothiazole	1934	1225	MS,RI,O	$108.7_{a}^{y}$	$95.4_{c}^{y}$	104.3 <sup>y</sup>	2.0	***	$121.5_{ab}^{x}$	116.7 <sup>x</sup>	$124.9_{a}^{x}$	1.5	*	***	***	**
Total				$6202.1_{a}^{x}$	3253.6 <sup>v</sup>	4313.2 <sub>b</sub> <sup>y</sup>	431.5	***	3916.6 <sub>b</sub> <sup>y</sup>	3597.8 <sup>x</sup>	$5472.4_{a}^{x}$	291.0	***	***	*	***

*Note:* DLY, Duroc × (Landrace × Yorkshire); SMX, Sanmenxia pig; TB, Tibetan pig.<sup>a-b</sup>means in the same row not followed by a common subscript letter differ significantly

(P < 0.05, Duncan test) (differences among the breeds of pigs).<sup>x-y</sup> means in the same row not followed by a common superscript letter differ significantly (P < 0.05, Duncan

test) (differences between Triceps Brachii and Biceps Femoris muscle of pigs). All experiments done in n=3 independent boiled pork samples. Significance: \*\*\*P < 0.001,

\*\* P < 0.01, \*P < 0.05; NS, not significant; SEM, standard error of the mean.

<sup>1</sup> Linear retention index calculated on DB-Wax capillary column.

<sup>2</sup> Linear retention index calculated on DB-5 capillary column.

<sup>3</sup>Means of identification: MS, mass spectrum comparison using NIST libraries; RI, retention index compared with literature value; O, aroma description (odor).

	<sup>1</sup> Odor thresholds	201 1	Triceps B	<i>rachii</i> mus	cle			Biceps Fe	emoris musc	le			Sign. parts			
Compounds	$(\mu g \cdot k g^{-1})$	Odor descriptions	DLY	SMX	TB	SEM	Sign.	DLY	SMX	TB	SEM	Sign.	DLY	SMX	TB	
Pentanal	9	Fruity	$4.6_{b}^{y}$	1.3 <sup>y</sup>	$16.1_{a}^{x}$	2.2	***	$6.8_{a}^{x}$	$2.9_{b}^{x}$	0.0 <sup>y</sup>	1.0	***	**	***	***	
Hexanal	5	Green, grass	$524.2_{a}^{x}$	225.7 <sup>°</sup>	301.9 <sub>b</sub> <sup>x</sup>	44.9	***	$274.7_{a}^{y}$	$268.2_{a}^{x}$	213.9 <sub>b</sub> <sup>y</sup>	10.2	**	***	*	**	
Heptanal	3	Fatty, putty	$41.7_{a}^{x}$	19.0 <sup>x</sup>	32.2 <sub>b</sub> <sup>x</sup>	3.4	***	$28.4_{b}^{y}$	20.7 <sup>x</sup>	$30.5_{a}^{x}$	1.5	***	**	NS	NS	
Octanal	0.578	Fatty, pungent	142.2 <sub>b</sub> <sup>x</sup>	$0.0_{c}^{y}$	$150.5_{a}^{x}$	24.4	***	98.5 <sup>y</sup>	107.6 <sup>x</sup>	$147.3_{a}^{y}$	7.5	***	***	***	*	
Nonanal	1	Fatty, floral, wax	333.8 <sub>b</sub> <sup>x</sup>	255.7 <sup>x</sup>	$454.6_{a}^{x}$	29.4	***	$267.4_{b}^{y}$	$248.7_{b}^{x}$	$362.6_{a}^{y}$	18.1	***	**	NS	**	
(E)-2-Octenal	3	Burdock, fatty	$15.7_{a}^{x}$	$7.1_{b}^{x}$	$7.7_{b}^{x}$	1.4	***	$10.4_{a}^{y}$	$7.8_{c}^{x}$	$9.0_{b}^{x}$	0.4	**	**	NS	NS	
Decanal	2	Orange peel, soapy	$0.0_{b}^{y}$	$5.5_{a}^{x}$	$0.0_{b}^{y}$	0.9	***	$10.9_{a}^{x}$	$0.0_{c}^{y}$	$6.9_{b}^{x}$	1.6	***	***	***	***	
Benzaldehyde	41.7	Bitter, almond	$3.3_{b}^{x}$	$1.4_{c}^{y}$	$7.1_{a}^{x}$	0.8	***	2.6 <sub>b</sub> <sup>y</sup>	$2.7_{b}^{x}$	$6.6_{a}^{y}$	0.7	***	**	***	*	
(E)-2-Nonenal	1	Cardboard, cucumber	$16.6_{a}^{x}$	8.7 <sup>y</sup>	13.9 <sub>b</sub> <sup>y</sup>	1.2	***	14.2 <sub>b</sub> <sup>y</sup>	13.0 <sub>b</sub> <sup>x</sup>	$18.6_{a}^{x}$	0.9	***	***	**	***	
(E)-2-Decenal	0.4	Fatty, green	$0.0_{b}$	$34.0_a^{y}$	$0.0_{b}^{y}$	5.7	***	0.0 <sub>c</sub>	$45.2_{a}^{x}$	35.5 <sup>x</sup>	6.9	***	NS	**	***	
(E,E)-2,4-Nonadienal	0.16	Fatty, green	$55.7_{a}^{x}$	$25.7_{b}^{x}$	$0.0^{y}_{c}$	8.2	***	$47.5_{a}^{x}$	$25.3_{b}^{x}$	$45.9_{a}^{x}$	4.0	**	NS	NS	**	
Dodecanal	2	Herbaceous, fatty	$2.2_{a}^{y}$	$1.5_{b}^{y}$	$0.0^{y}_{c}$	0.3	***	$3.6_{b}^{x}$	$3.0_{b}^{x}$	$11.8_{a}^{x}$	1.4	***	**	**	***	
2-Undecenal	0.78	Wax, fatty	$9.3_{a}^{x}$	$0.0_{b}$	$0.0_{b}^{y}$	1.6	***	$0.0_{b}^{y}$	$0.0_{b}$	$25.6_{a}^{x}$	4.5	**	***	NS	**	
(E,E)-2,4-Decadienal	0.07	Fatty, deep-fried	$163.4_{a}^{x}$	83.1 <sup>x</sup>	$0.0^{y}_{c}$	24.0	***	57.8 <sup>y</sup>	$0.0_{c}^{y}$	$107.0_{a}^{x}$	15.5	***	**	**	***	
1-Octen-3-ol	2	Mushroom	$194.3_{a}^{x}$	$114.5_{b}^{x}$	56.9 <sup>v</sup>	20.0	***	$153.2_{a}^{y}$	128.9 <sup>x</sup>	77.5 <sup>°</sup> x	11.5	***	**	NS	*	
(E)-2-Octen-1-ol	3	Fruity, green apple	$10.9_{a}^{x}$	$8.1_{b}^{x}$	4.5 <sup>y</sup>	0.9	***	$10.2_{a}^{x}$	$8.4_{b}^{x}$	$6.3_{c}^{x}$	0.6	**	NS	NS	**	
Anethole	15	Rubber, paint	$0.0_{b}^{y}$	$0.0_{b}$	$1.0_{a}^{y}$	0.2	***	$0.3_{b}^{x}$	0.0 <sub>c</sub>	$1.5_{a}^{x}$	0.2	***	***	NS	***	
Ethyl hexanoate	1	Fatty, green	$0.0_{b}$	$7.8_{a}^{y}$	$0.0_{b}$	1.3	***	$0.0_{b}$	$8.9_{a}^{x}$	$0.0_{b}$	1.5	***	NS	**	NS	
Styrene	65	Herbaceous, fatty	$3.8_{a}^{x}$	0.0 <sub>c</sub>	$1.4_{b}^{y}$	0.5	***	$2.0_{a}^{y}$	0.0 <sub>c</sub>	$1.7_{b}^{x}$	0.3	***	***	NS	**	
2-Ethylfuran	2.3	Rubber, pungent	17.3 <sup>x</sup>	$18.5_{b}^{y}$	$22.7_{a}^{x}$	0.8	***	17.9 <sup>x</sup>	$22.0_{a}^{x}$	$20.7_{b}^{x}$	0.6	***	NS	**	NS	
2-Pentylfuran	6	Pungent, sweet	$40.7_{a}^{x}$	$27.1_{b}^{x}$	22.1 <sup>y</sup>	2.8	***	18.1 <sup>°</sup>	25.9 <sub>b</sub> <sup>x</sup>	$38.6_{a}^{x}$	3.0	***	***	NS	***	
Dimethyl disulphide	1.1	Cooked cabbage	76.8 <sup>x</sup>	116.5 <sup>x</sup>	$141.3_{a}^{x}$	9.4	***	49.1 <sup>v</sup>	105.9 <sub>b</sub> <sup>y</sup>	$127.4_{a}^{y}$	11.7	***	**	**	***	
2-Acetylpyrazine	62	Nutty, popcorn	$1.4_{b}^{x}$	$1.4_{b}^{x}$	$1.5_{a}^{x}$	0.02	***	$0.8_{b}^{y}$	$0.8_{b}^{y}$	$1.0_{a}^{y}$	0.03	***	***	***	***	

**Table 4.** Odor-active compounds (OACs, OAVs > 1) in boiled pork from different breeds of pigs.

2-Acetylthiazole	10	Caramel, sweaty	3.3 <sub>b</sub> <sup>y</sup>	6.8 <sup>x</sup>	$7.0_{a}^{x}$	0.6	***	$4.3_{b}^{x}$	5.0 <sub>a</sub> <sup>y</sup>	5.1 <sup>y</sup> <sub>a</sub>	0.1	***	**	***	***
Benzothiazole	80	Caramel, cheese	$1.4_{a}^{y}$	1.2 <sup>y</sup>	$1.3b^{y}$	0.03	***	$1.5_{b}^{x}$	$1.5_{b}^{x}$	$1.6_{a}^{x}$	0.02	*	***	***	**
Total			1662.6 <sub>a</sub> <sup>x</sup>	970.6 <sup>x</sup>	1243.7 <sub>b</sub> <sup>x</sup>	100.9	***	1080.0 <sub>b</sub> <sup>y</sup>	1052.6 <sub>b</sub> <sup>x</sup>	1302.5 <sup>x</sup>	40.8	***	***	NS	NS

*Note:* DLY, Duroc × (Landrace × Yorkshire); SMX, Sanmenxia pig; TB, Tibetan pig. <sup>a-b</sup>means in the same row not followed by a common subscript letter differ significantly (P < 0.05, Duncan test) (differences among the breeds of pigs). <sup>x-y</sup>means in the same row not followed by a common superscript letter differ significantly (P < 0.05, Duncan test) (differences between *Triceps Brachii* and *Biceps Femoris* muscle of pigs). All experiments done in n=3 independent boiled pork samples. Significance: \*\*\*P < 0.001, \*\* P < 0.01, \*P < 0.05; NS, not significant; SEM, standard error of the mean.

<sup>1</sup>Odor thresholds were mainly obtained from the literature and an online database, with water applied as the matrix: (Gu, S. Q., Wang, X. C., Tao, N. P., & Wu, N, 2013; Sansone-Land, A., Takeoka, G.R., & Shoemaker, C. F, 2014; Czerny et al., 2008; Mayuoni-Kirshinbaum, L., Daus, A., & Porat, R, 2013; Liu, Y., He, C., & Song, H, 2018), (http://www.flavornet.org, http://www.odour.org.uk).

<sup>2</sup> Odor descriptions were mainly gathered from the following literature and online database: (Gu, S. Q., Wang, X. C., Tao, N. P., & Wu, N, 2013; Czerny et al. 2008; Mayuoni-Kirshinbaum, L., Daus, A., & Porat, R, 2013), (http://www.flavornet.org).

Odor active compounds	PCA		PLS-DA	One way analysis of variance
	PC1	PC2	VIP score	<i>p</i> value
Pentanal	-0.236	0.468	0.894	< 0.01
Hexanal	0.727	0.041	1.026	< 0.01
Heptanal	0.740	0.541	1.028	< 0.01
Octanal	0.386	0.791	0.947	< 0.01
Nonanal	-0.061	0.923	0.852	< 0.01
(E)-2-Octenal	0.945	0.006	1.069	< 0.01
Decanal	0.119	-0.225	0.750	< 0.01
Benzaldehyde	-0.113	0.988	0.883	< 0.01
(E)-2-Nonenal	0.617	0.671	1.011	< 0.01
(E)-2-Decenal	-0.452	-0.273	0.945	< 0.01
(E,E)-2,4-Nonadienal	0.864	-0.247	1.044	< 0.01
Dodecanal	0.249	0.337	0.871	< 0.01
2-Undecenal	0.387	0.517	0.910	< 0.01
(E,E)-2,4-Decadienal	0.790	-0.046	1.033	< 0.01
1-Octen-3-ol	0.747	-0.598	1.039	< 0.01
(E)-2-Octen-1-ol	0.708	-0.682	1.023	< 0.01
Anethole	-0.129	0.863	0.882	< 0.01
Ethyl hexanoate	-0.555	-0.676	1.002	< 0.01
Styrene	0.892	0.345	1.079	< 0.01
2-Ethylfuran	-0.677	0.527	0.981	< 0.01
2-Pentylfuran,	0.562	0.258	0.973	< 0.01
Dimethyl disulphide	-0.671	0.543	1.040	< 0.01
2-Acetylpyrazine	-0.067	0.309	0.973	< 0.01
2-Acetylthiazole	-0.909	0.158	1.067	< 0.01
Benzothiazole	0.362	0.250	0.897	< 0.01

Table 5. Information of PCA, PLS-DA and one way analysis of variance.

*Note:* VIP, variable importance in projection.



### **Conflicts of Interest:**

Dong Han declares that he has no conflict of interest. Chun-Hui Zhang declares that he has no conflict of interest. Marie-Laure Fauconnier declares that she has no conflict of interest. Si Mi declares that she has no conflict of interest.

## **Author Contributions Section:**

D. Han performed the experiments and analysed the results. C-H Zhang contributed to the conception of the research and invaluable experimental design. M-L. Fauconnier and S. Mi critically revised the manuscript.