

Pharmacomodulation of ellagic acid, a promising antiplasmodial agent, to improve its bioavailability.

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Context

Vector-borne diseases represent more than 17 percent of all infectious sicknesses and affect millions of people each year.¹ Among them, malaria is the most important causing more than 400 000 deaths and 216 million of cases. This parasitosis is caused by a protozoan belonging to the *Plasmodium* genus. More than 90% of cases and deaths following malaria are linked to *P. falciparum* which is transmitted thanks to mosquitoes from *Anopheles* family.²

While many efforts are consent to develop a vaccine, chemotherapy remains the most valid therapeutic strategy, nowadays. Combination of artemisinin derivatives (artemether, artesunate) and a long acting agent such as lumefantrine, amodiaquine, mefloquine... are particularly recommended. Unfortunately, as for antibiotics, resistance to antimalarials emerge from their use. Considering this resistance and the prevalence of malaria, development of new antiplasmodials represents an urgent need.³

Objectives

Nature remains an important source of medicinal compounds⁴. Screening of natural molecules identified from plants used in traditional medicine leads to identification of metabolites which are active against *Plasmodium*'s strains, including the one which are resistant to modern antiplasmodials.

Among the candidates, ellagic acid (EA, **1**) seems to be one of the most promising compound thanks to its proven *in vitro* efficacy (105-330 nM) and *in vivo* activity after intraperitoneal injection. Interestingly, no toxic effect could be demonstrated at >5 g/kg (*per os*) or 1g/kg/d (IP). This polyphenolic compound seems thus very attractive for malaria treatment.⁵ However, EA suffers from a poor oral bioavailability, partly explained by a reduced water solubility (9.7 µg/mL) itself linked to strong intermolecular bounds⁶.

The aim of our work is the design and the preparation of crowded analogues and prodrugs with enhanced solubility and by extension, improved bioavailability which would permit its oral use.

Pharmacomodulation

As described in Fig.1, taking advantage of phenolic functions, we plan to insert bulky/polar chains on selected positions. This should lead to crowded analogues of EA. The planned pharmacomodulations should also give insights about the pharmacophore of this polyphenol.

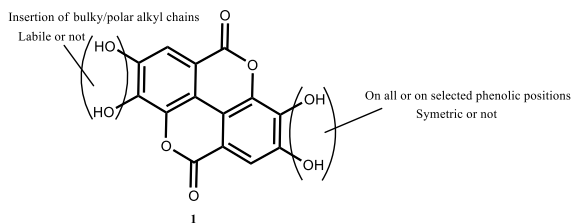
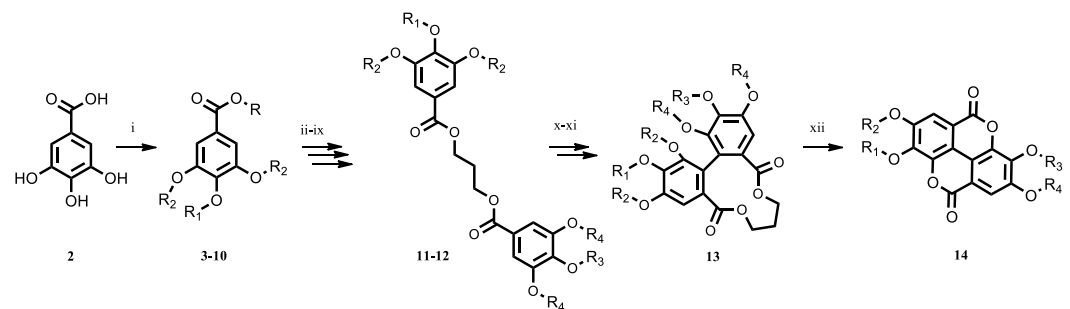


Fig. 1 Pharmacomodulation opportunities

Synthesis

Considering issues encountered following a direct synthesis approach, a total synthesis strategy is preferred in order to obtain the desired molecules.^{7,8}



i) CH₃OH, H₂SO₄ ii) Ac₂O, H₂SO₄ iii) BnBr, KI, K₂CO₃, acetone iv) K₂CO₃, CH₃OH, H₂O v) NaH, MOMCl, DMF vi) LiOH, CH₃OH, THF, H₂O vii) 1,3-propanediol-PMB, DMAP, EDCI-HCl, CH₂Cl₂ viii) DDQ, Sorenson's buffer, CH₂Cl₂ ix) DMAP, EDCI-HCl, CH₂Cl₂ x) THF, IPA/HCl xi) CuCl₂, *n*-BuNH₂, CH₃OH xii) LiOH, CH₃OH, THF, H₂O

Antiplasmodial effect

In order to explore the effect of substituted derivatives on malaria parasite, final compounds and all the intermediates are submitted in an *in vitro* assay to determine their inhibitory effect, expressed as IC₅₀ (µM), on 3D7 strain of *Plasmodium falciparum* (chloroquin-sensitive). This evaluation follows the procedure established by Frederich *et al.* (2002)⁸ and is based on colorimetric measure of *Plasmodium* lactate dehydrogenase activity⁹ realized at least in triplicates.

Conclusion & Prospects

As observed during the pharmacological assay, some intermediates showed a higher effect (< 125 µM) than gallic acid (**2**). Interestingly, this inhibitory activity clearly increases when there is a linker between 2 gallate moieties. This observation suggests the importance to look like as a dimer to observe the same rank of potency of EA. We can also observe that substituents on the phenolic functions seem to negatively impact this effect (**11** vs **12**). The next step will be testing against 3D7 to confirm pharmacological results and the final substituted compound. Moreover, other mono- and disubstituted derivatives of EA's scaffold will be synthesized. Another part of the future work will consist to determine the aqueous solubility of the new compounds vs EA.

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Product (n=)	R	R ₁ /R ₃	R ₂ /R ₄	IC ₅₀ (µM)
1	/	H	H	IP
2 (3)	H	H	H	125,19±5,76
3 (3)	CH ₃	H	H	118,14±16,62
4 (2)	CH ₃	CH ₃ CO	CH ₃ CO	47,29±17,98
5 (3)	CH ₃	BnCH ₂	CH ₃ CO	33,27±12,86
6 (3)	CH ₃	BnCH ₂	H	54,11±13,02
7 (1)	CH ₃	BnCH ₂	MOM	43,83
8 (3)	H	BnCH ₂	MOM	58,74±28,25
9 (4)	CH ₂ CH ₃ CH ₂ OPMB	BnCH ₂	MOM	19,89±4,16
10 (2)	CH ₂ CH ₃ CH ₂ OH	BnCH ₂	MOM	82,02±15,64
11 (2)	CH ₂ CH ₃ CH ₂	BnCH ₂	MOM	17,19±5,23
12 (2)	CH ₂ CH ₃ CH ₂	BnCH ₂	H	3,67±0,19
13	CH ₂ CH ₃ CH ₂	BnCH ₂	H	IP
14	/	BnCH ₂	H	IP

IP = In Progress