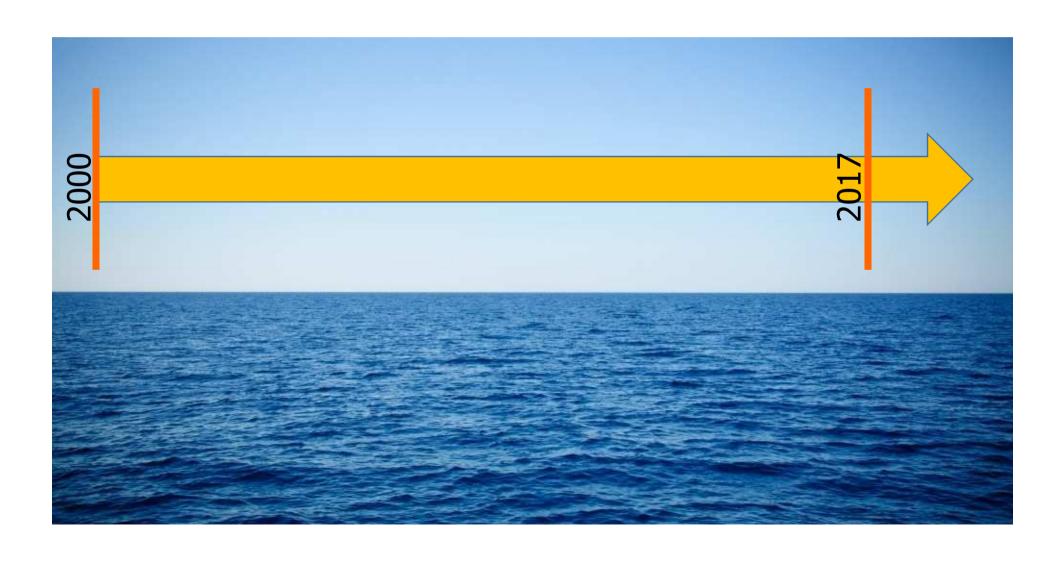
From master...s to post-doc...s



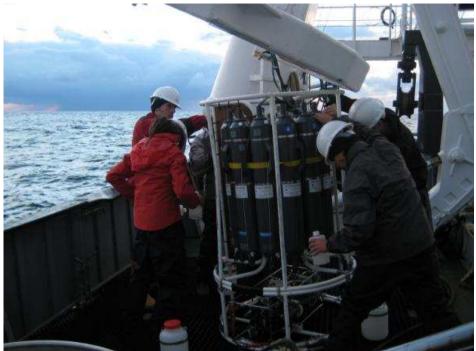






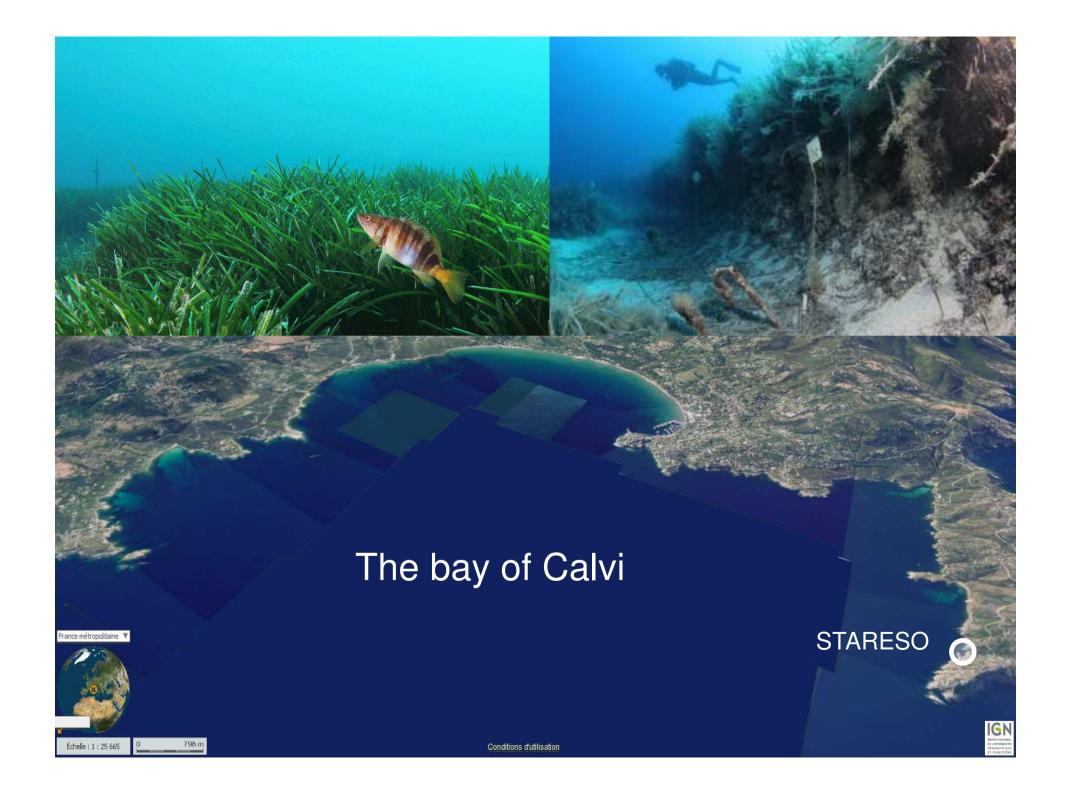


































DMITRI MENDELEYEV [1834 - 1907]

The Russian chemist, Dmitri Mendeleyev, was the first to observe that if elements were listed in order of atomic mass, they showed regular (periodical) repeating properties. He formulated his discovery in a periodic table of elements, now regarded as the backbone of modern chemistry.

The crowning achievement of Mendeleyev's periodic table lay in his prophecy of then, undiscovere elements. In 1869, the year he published his periodic classification, the elements gallium, germanium and scandium were unknown. Mendeleyev left spaces for them in his table and even predicted their atomic masses and other chemical properties. Six years later, gallium was discovered and his predictions were found to be accurate. Other discoveries followed and their chemical behaviour matched that predicted by Mendeleyev.

This remarkable man, the youngest in a family of 17 children, has left the scientific community with a classification system so powerful that it became the cornerstone in chemistry teaching and the prediction of new elements ever since. In 1955, element 101 was named after him: Md, Mendelevium.



























































TABLE



























































Ta



















































































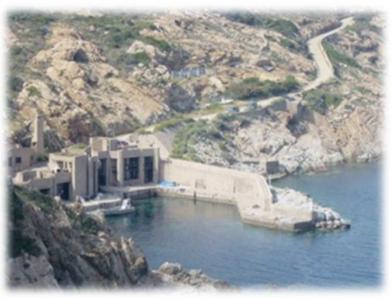




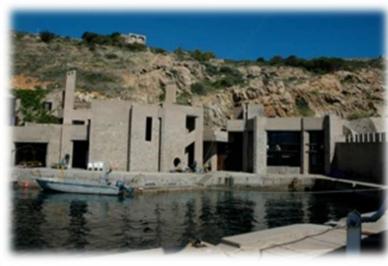




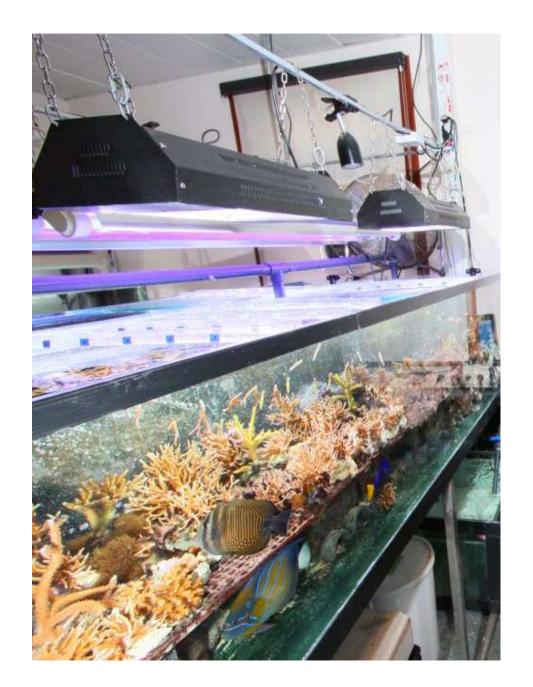




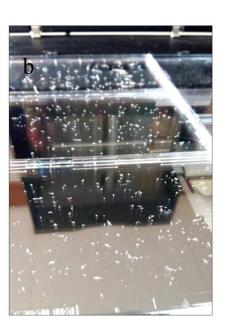


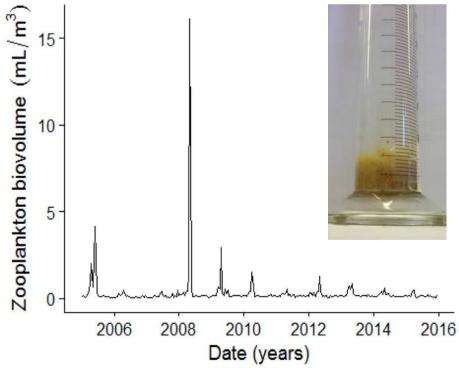




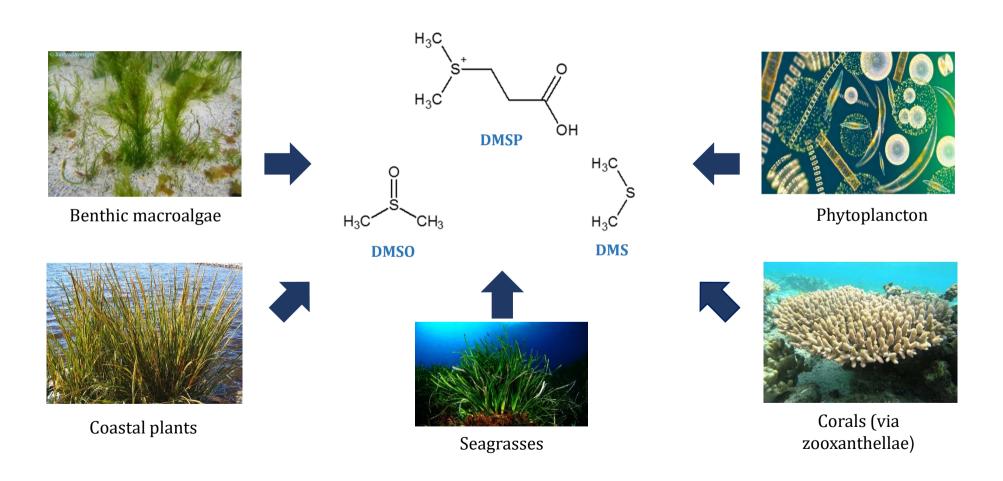


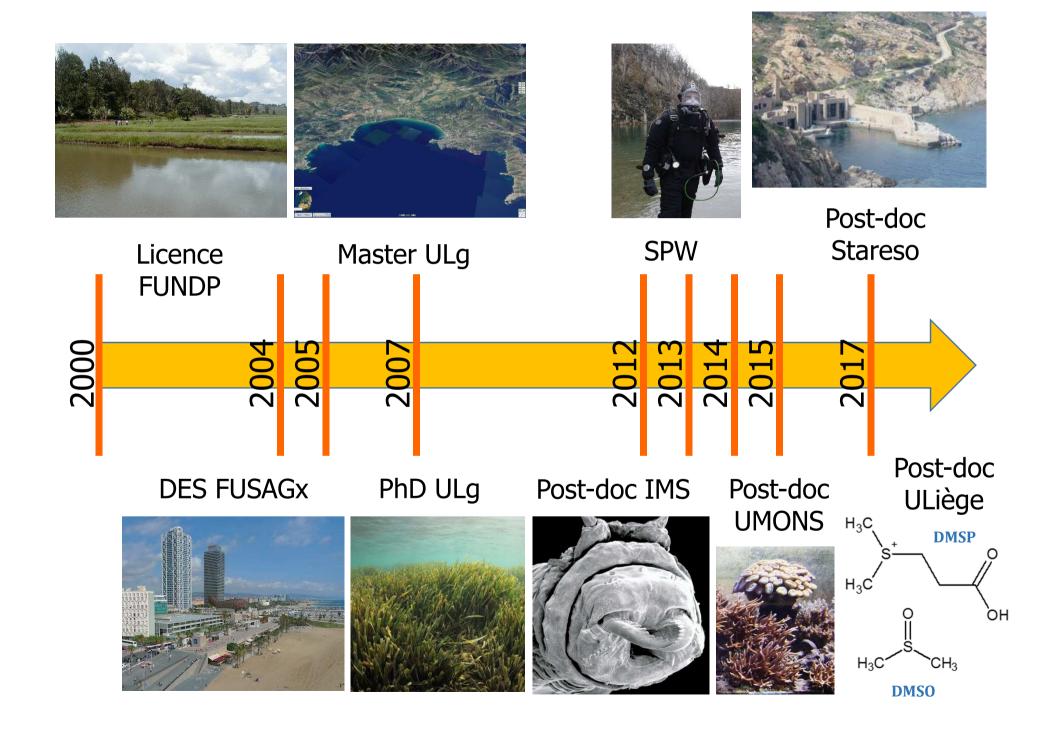






Producers of DMSP and its derivatives



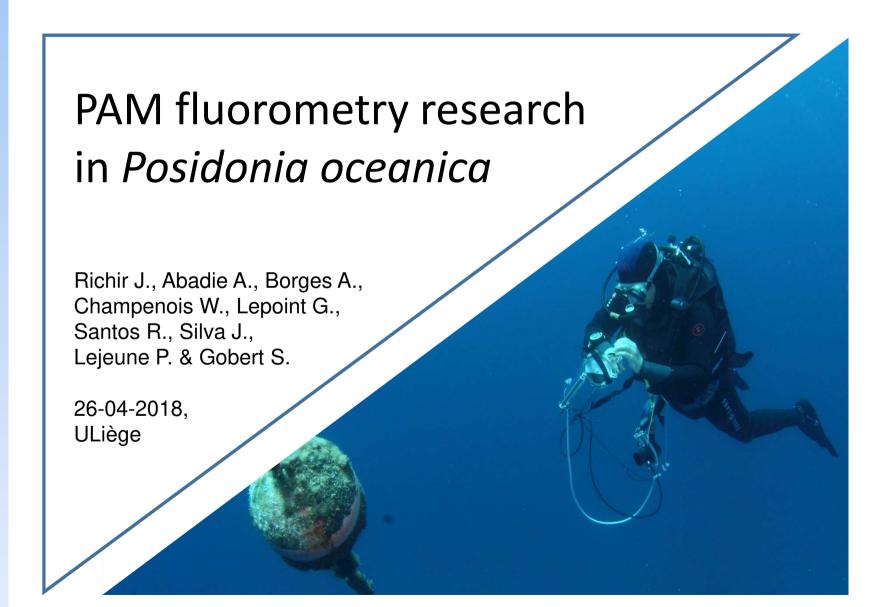






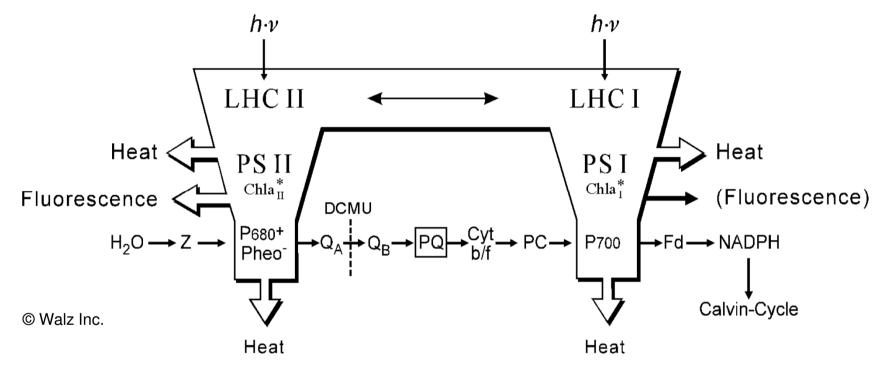








Fluorescence emission



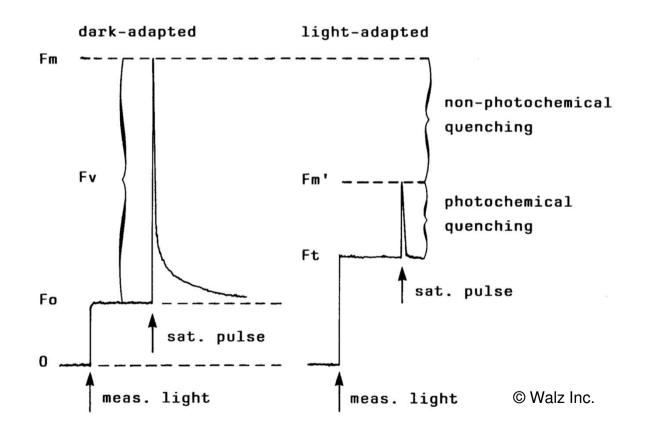
Fluorescence emission is complementary to the alternative pathways of de-excitation, which are photochemistry and heat dissipation.





Fluorescence measurement

- ❖ Dark adapted YIELD = (Fm-F0)/Fm = Fv/Fm
 - maximum photochemical efficiency
- Light adapted YIELD = (Fm'-Fm)/Fm' = ΔF/Fm'
 - effective photochemical efficiency





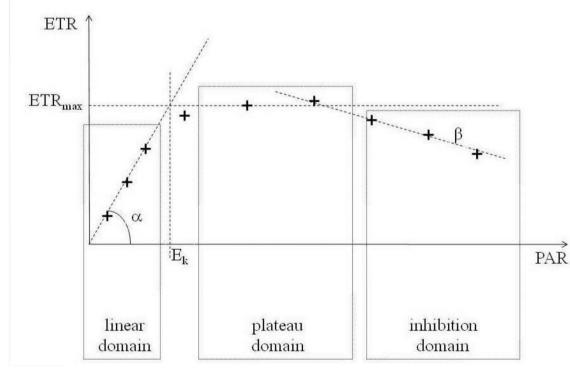


ETR - RLC

- Electron Transfert Rate: ETR = YIELD x PAR x 0.5 x ETR-factor
- Rapid Light Curve: photosynthetic flexibility with which a sample can adapt its photosynthetic aparatus to rapid changes of ligh intensity







ETR_{max} = maxixmum electron tranport rate

 E_k = compensation irradiance

(Lassauque, 2008)



Diving-PAM

- Underwater study of in situ photosynthesis
- Optimized to determine the effective quantum yield of photosynthetic energy conversion, ΔF/Fm'



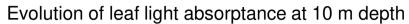


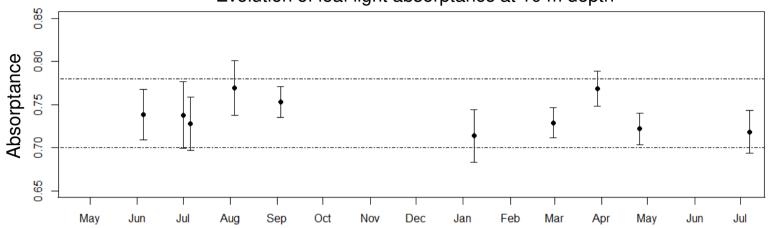




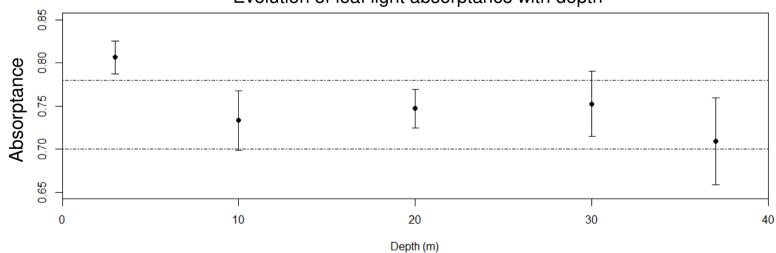


ETR = YIELD x PAR x 0.5 x ETR-factor



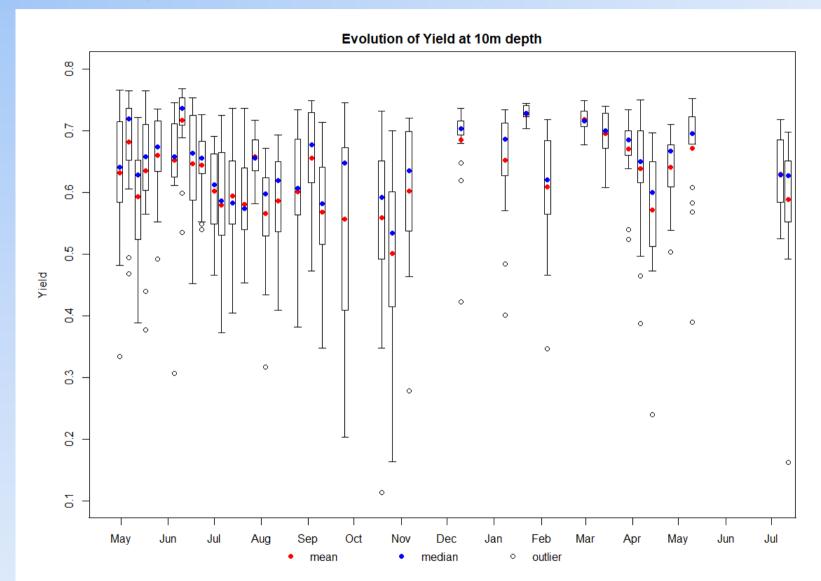


Evolution of leaf light absorptance with depth



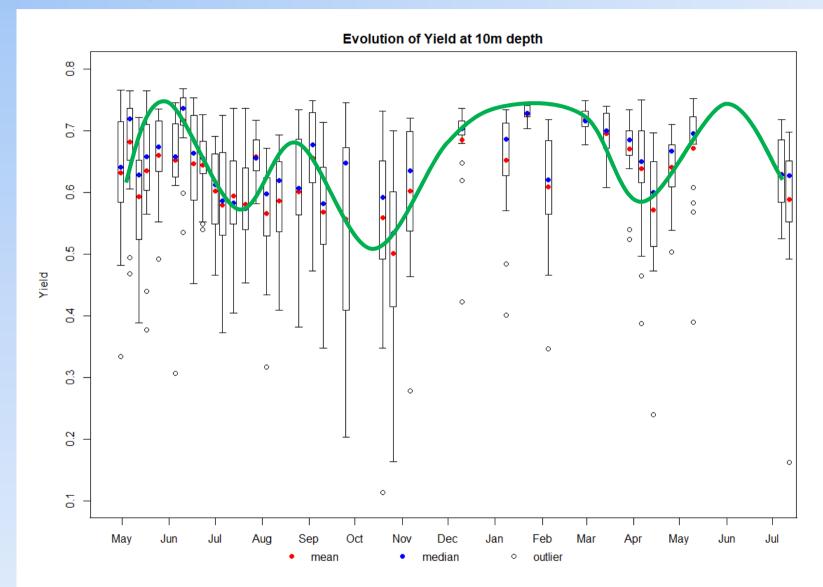


Effective photochemical efficiency



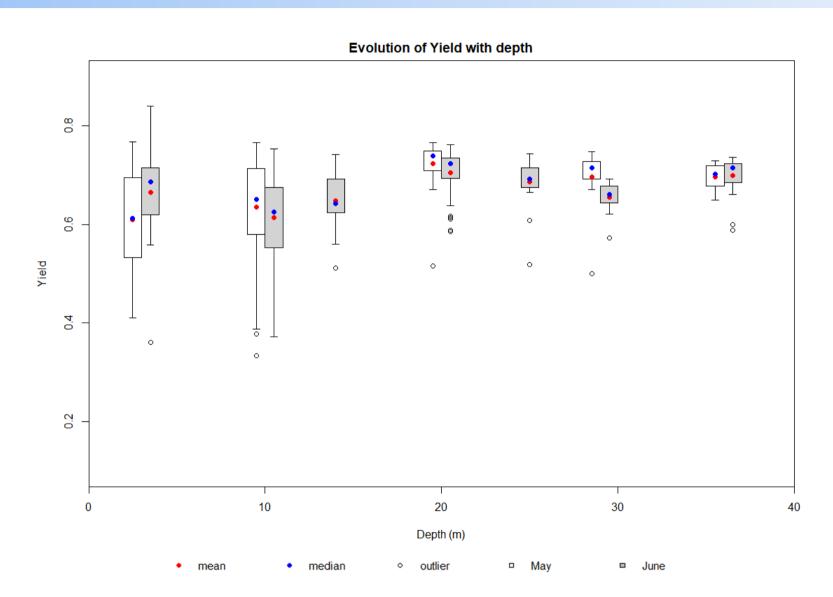


Effective photochemical efficiency



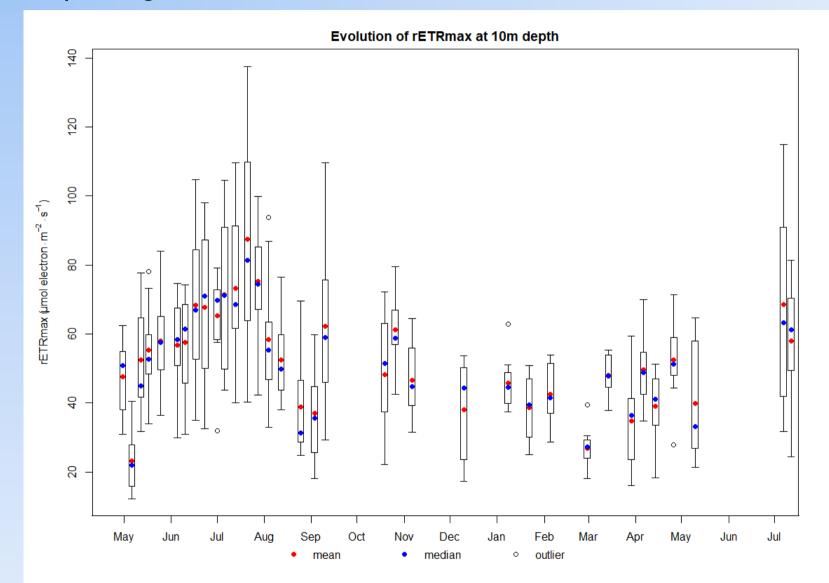


Effective photochemical efficiency



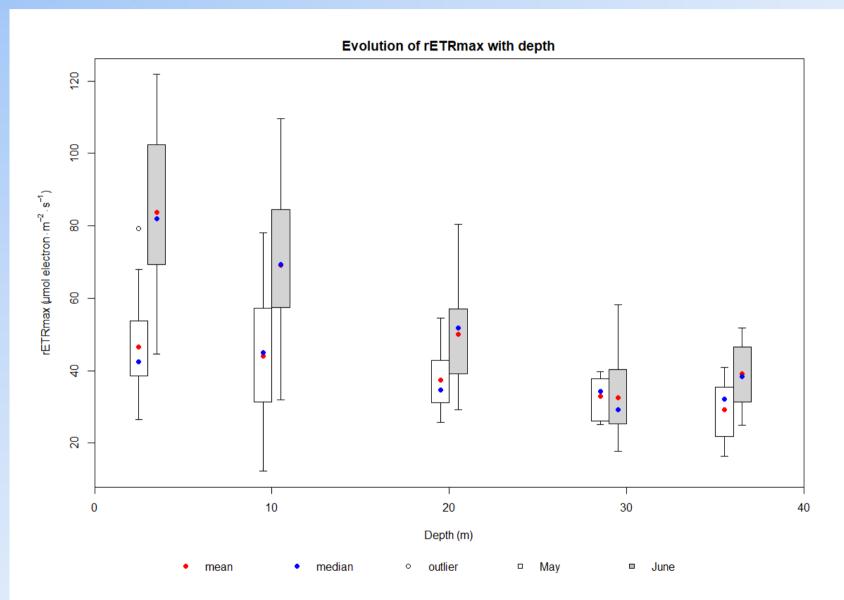


Rapid Light Curve



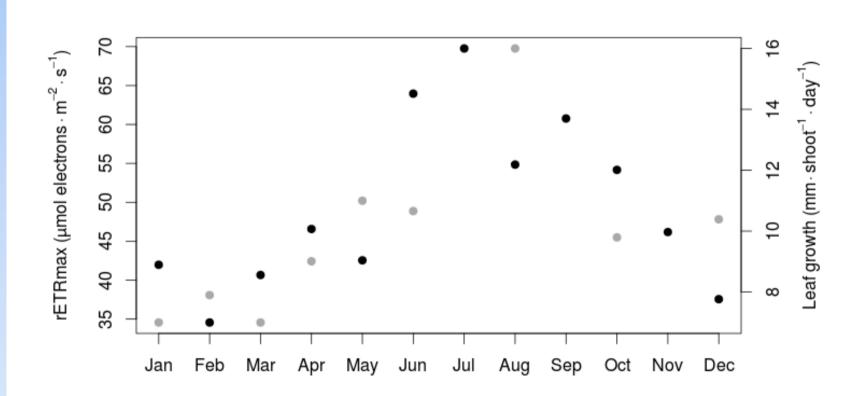


Rapid Light Curve





rETRmax = leaf growth proxy?





Conclusion et perspectives

Study of *Posidonia oceanica* photosynthesis by pulse amplitude modulated (PAM) fluorometry ...

- rapid and cost-effective method for:
 - ecophysiological studies,
 - ecotoxicological studies.

Next:

- photosynthesis vs foliar growth relationship,
- photosynthesis vs PP (O₂) relationship,
- field biomarker.

