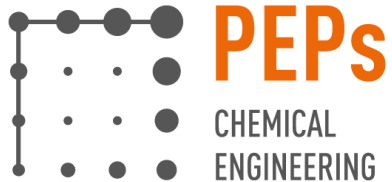


---

# CO2 FUTURE at the University of Liège

---

Prof. Grégoire LEONARD



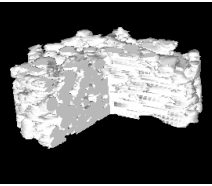
# University of Liège - ULiège

- Liège: 3<sup>rd</sup> urban area in Belgium
  - ~750 000 inh.
- ULiège = a pluralist university
  - 11 faculties, 23 000+ students, 122 Nationalities
  - 38 bachelor study lines, more than 200 master study lines

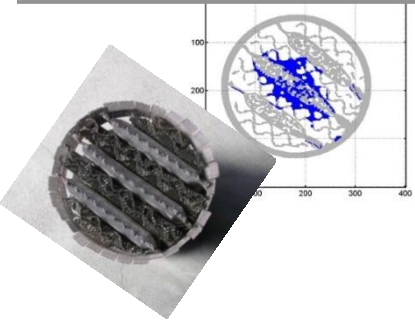


# PEPs - Products, Environment and processes

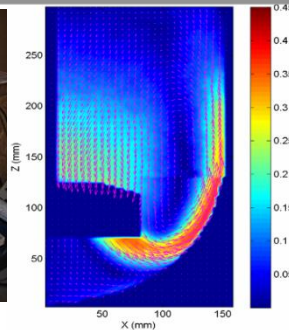
Solid waste and flue gas treatment



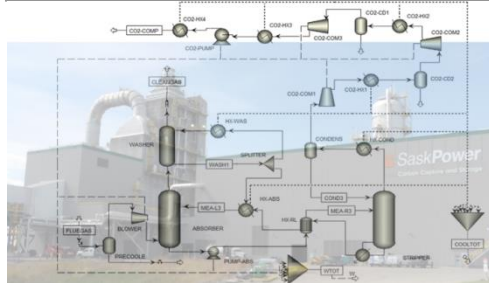
Hydrodynamics in multiphase systems



Mixing in (bio)reactors



CO<sub>2</sub> capture and reuse



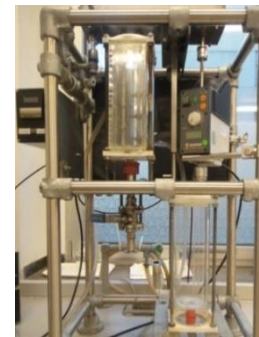
Life Cycle Assessment



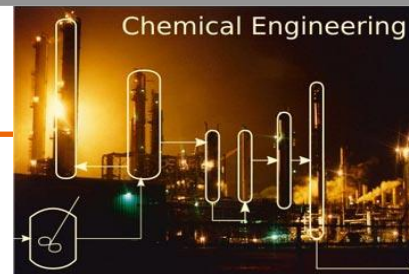
## Department of Chemical Engineering

<http://chemeng.ulg.ac.be>

Solvent and reactive extraction



Computer-Aided Process Engineering (CAPE)



# PEPs

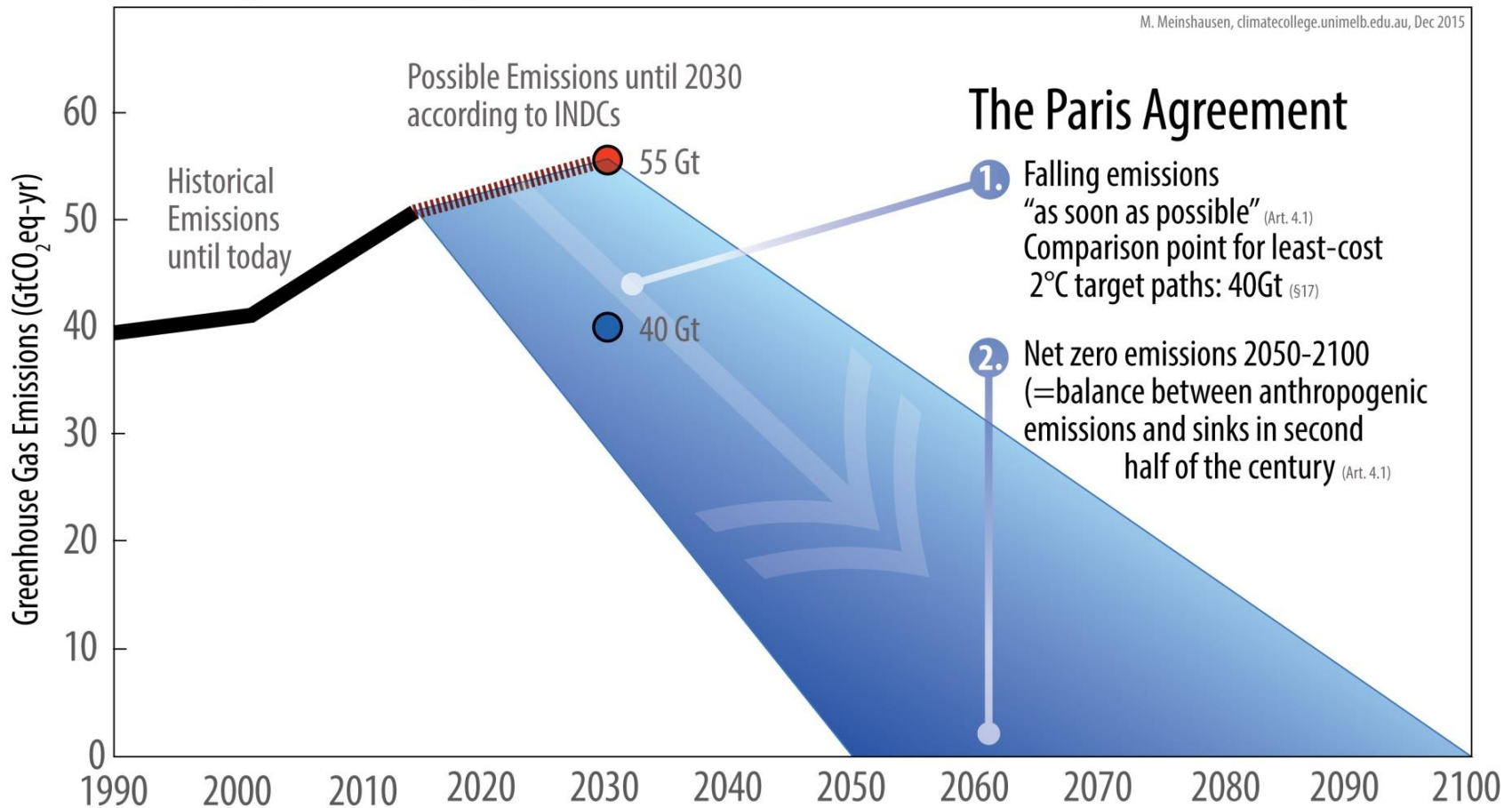
CHEMICAL  
ENGINEERING

# Content

- Context
- CO<sub>2</sub> capture
- CO<sub>2</sub> re-use and power-to-fuel
- CO<sub>2</sub>FUTURE platform

# The Energy Transition has started... ... but is far from being over!

## Global greenhouse gas emissions



# European objectives

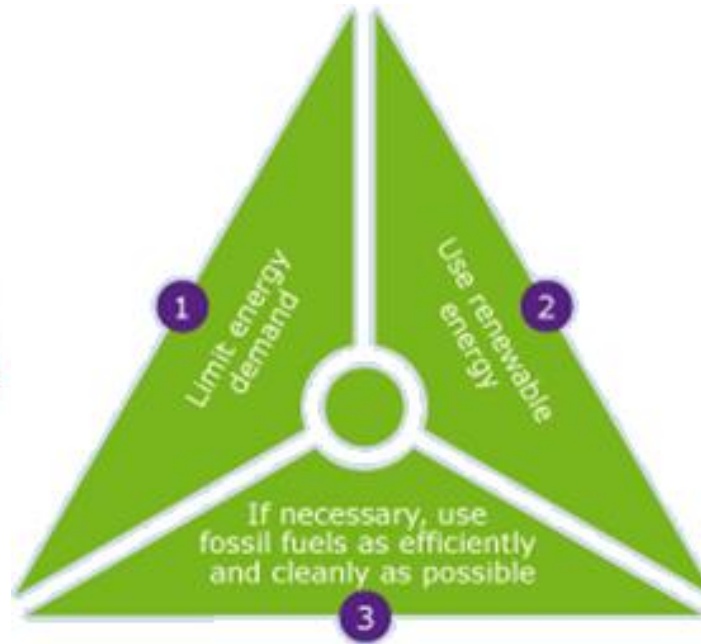
## ■ Roadmap for low carbon economy in 2050

**Table 1: Sectoral reductions**

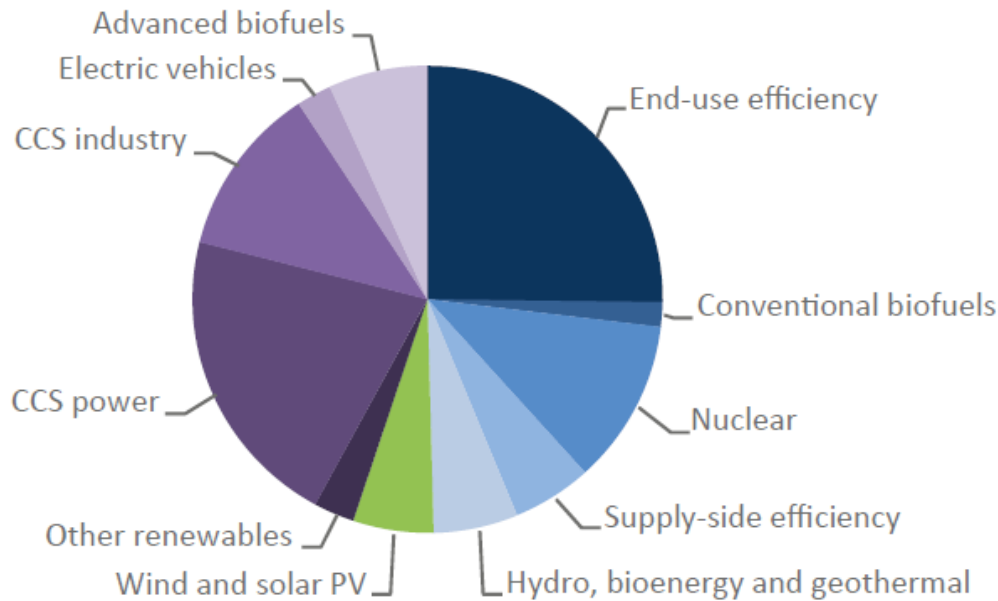
GHG reductions compared to 1990	2005	2030	2050
Total	-7%	-40 to -44%	-79 to -82%
Sectors			
Power (CO <sub>2</sub> )	-7%	-54 to -68%	-93 to -99%
Industry (CO <sub>2</sub> )	-20%	-34 to -40%	-83 to -87%
Transport (incl. CO <sub>2</sub> aviation, excl. maritime)	+30%	+20 to -9%	-54 to -67%
Residential and services (CO <sub>2</sub> )	-12%	-37 to -53%	-88 to -91%
Agriculture (non-CO <sub>2</sub> )	-20%	-36 to -37%	-42 to -49%
Other non-CO <sub>2</sub> emissions	-30%	-72 to -73%	-70 to -78%

European Commission communication, 2011, COM(2011) 112 final

# Possible answers: Trias Energetica



# What efforts are needed?



World CO<sub>2</sub> emissions abatement in the 450 Scenario (Bridge Scenario 2015-2040), IEA 2015, WEO special report, Energy & Climate Change

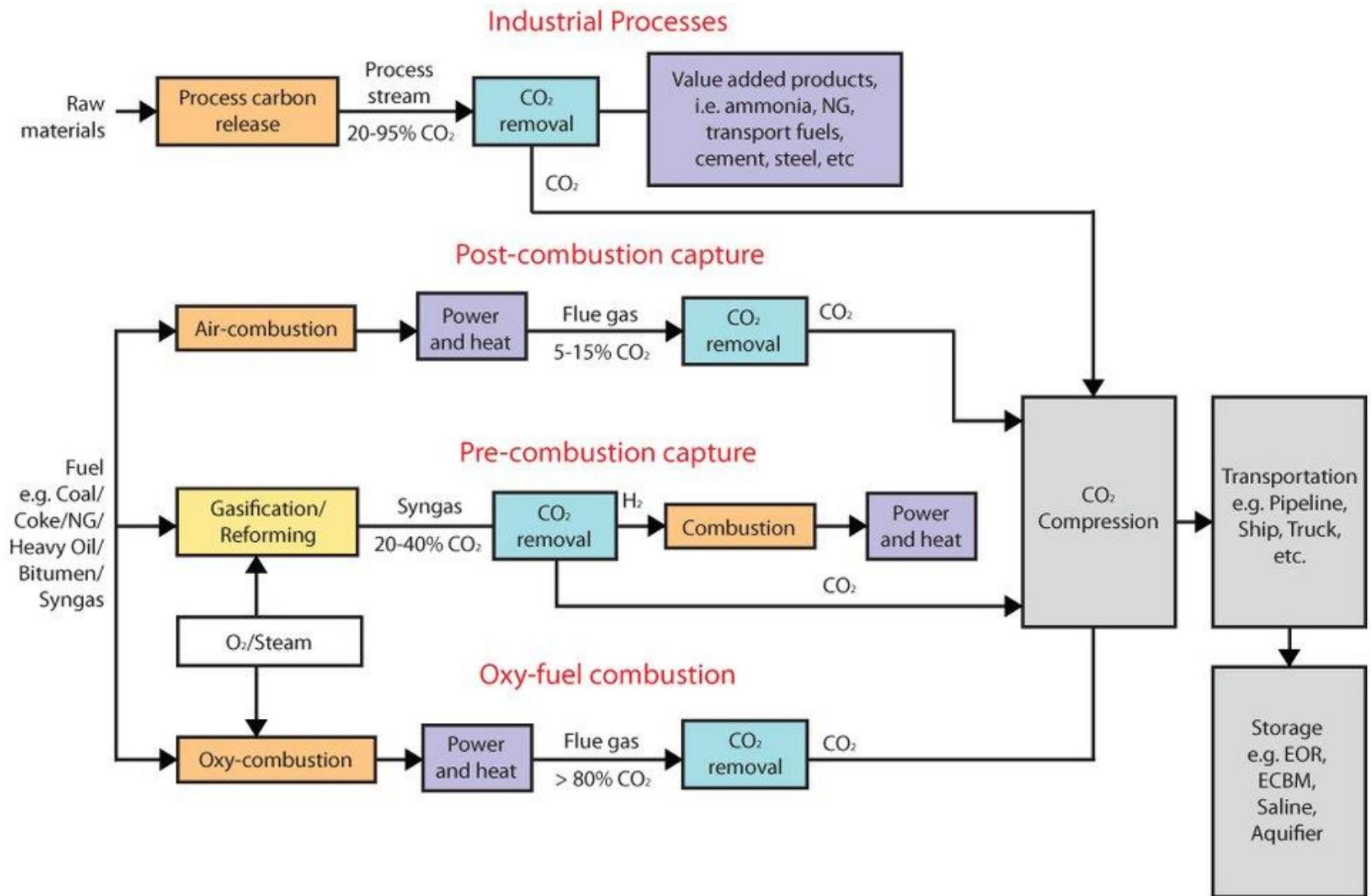
- **CCS** mature technology, but cost only!
- **CCU**: different maturity levels, depending on product

---

## 2. CO<sub>2</sub> Capture technologies & configurations

---

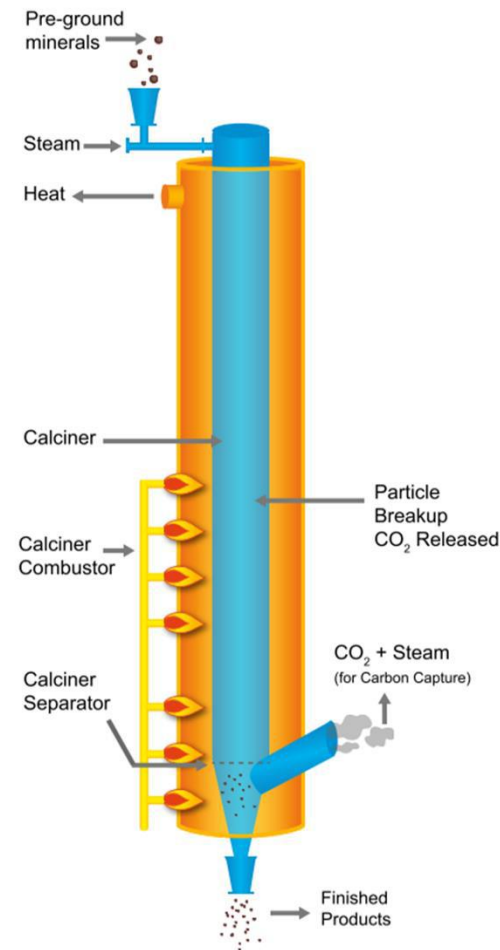
# CO<sub>2</sub> capture configurations



# Industrial processes

## 1. CO<sub>2</sub> not resulting from combustion

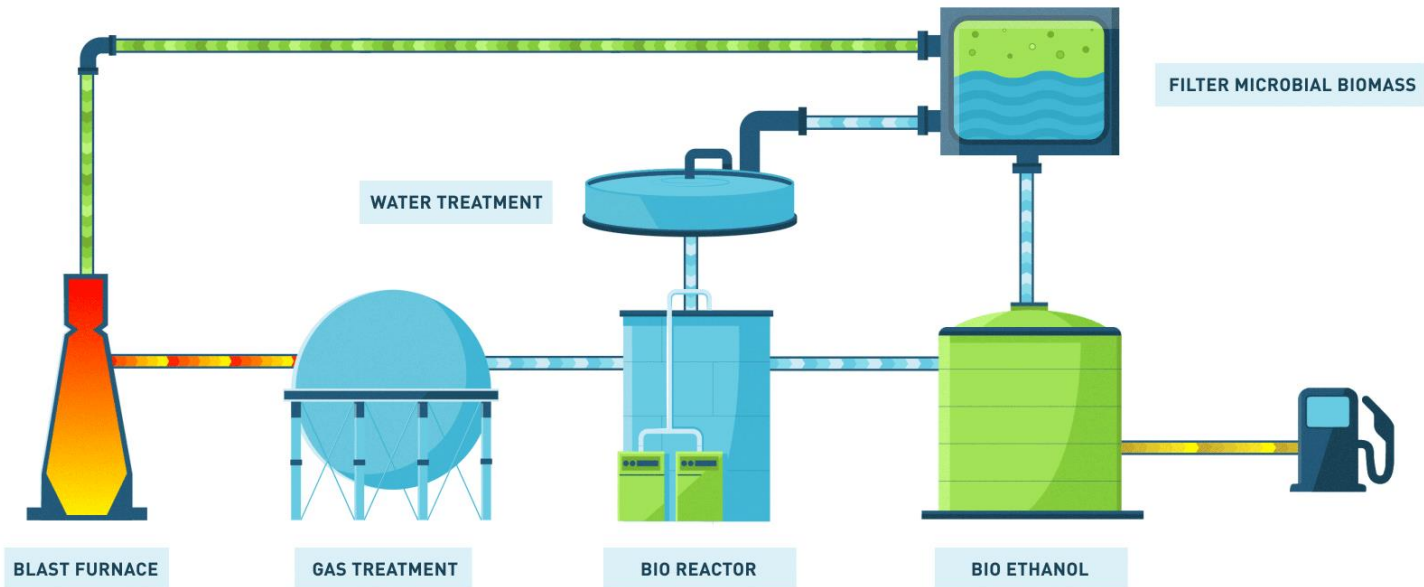
- Cement plants
  - $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$
  - High Temperature needed
  - Leilac: 21 M€, -60% CO<sub>2</sub>



# Industrial processes

## 1. CO<sub>2</sub> not resulting from combustion

- ❑ Steel mills
  - Fermentation of CO into ethanol
  - Steelanol: 87 M€, -70% CO<sub>2</sub>



# Post-combustion capture

## 2. Capture CO<sub>2</sub> from combustion gases

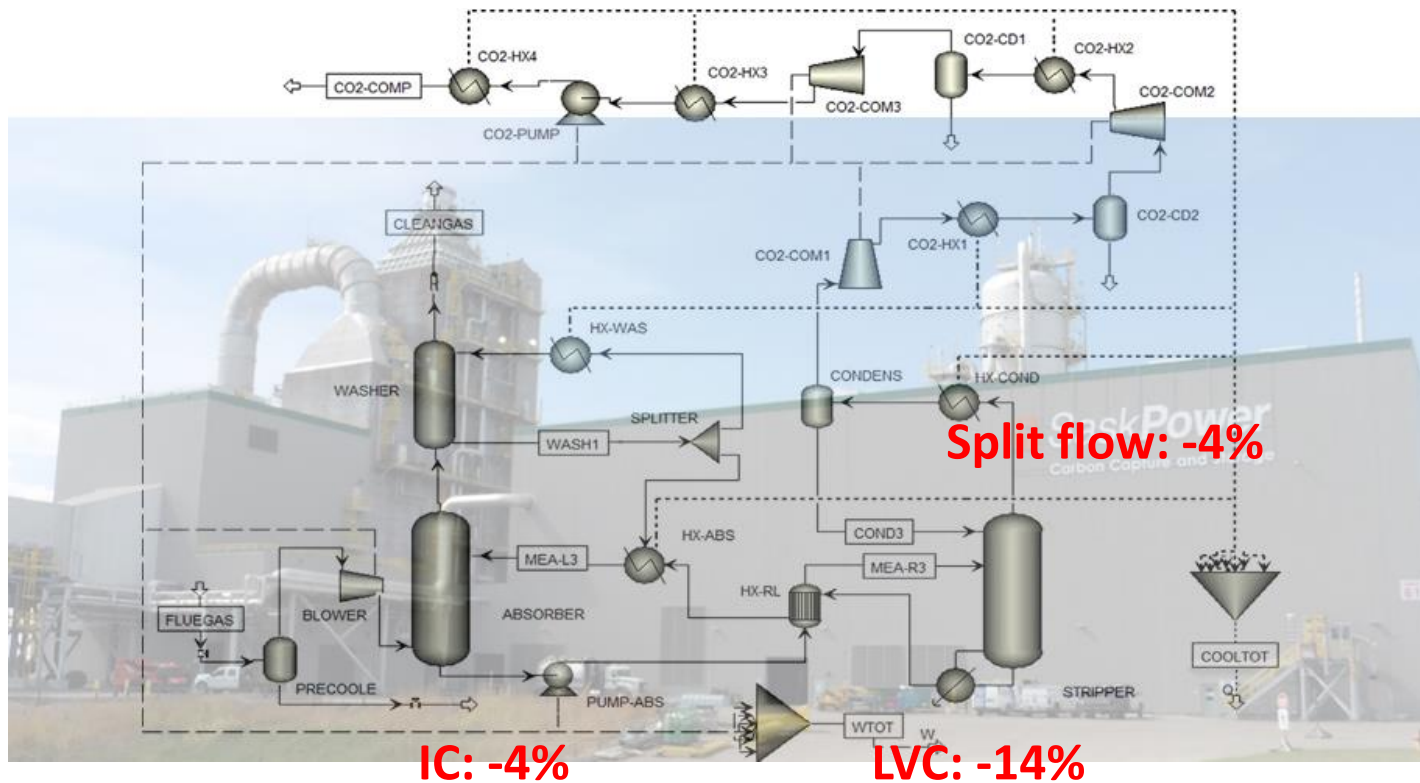
- Absorption – Regeneration with chemical solvents
- Boundary Dam (Ca), 2700 tCO<sub>2</sub>/day from Coal PP
  - Flue gas: 180 Nm<sup>3</sup>/s ; Solvent: 550 L/s



# Post-combustion capture

## 2. Capture CO<sub>2</sub> from combustion gases

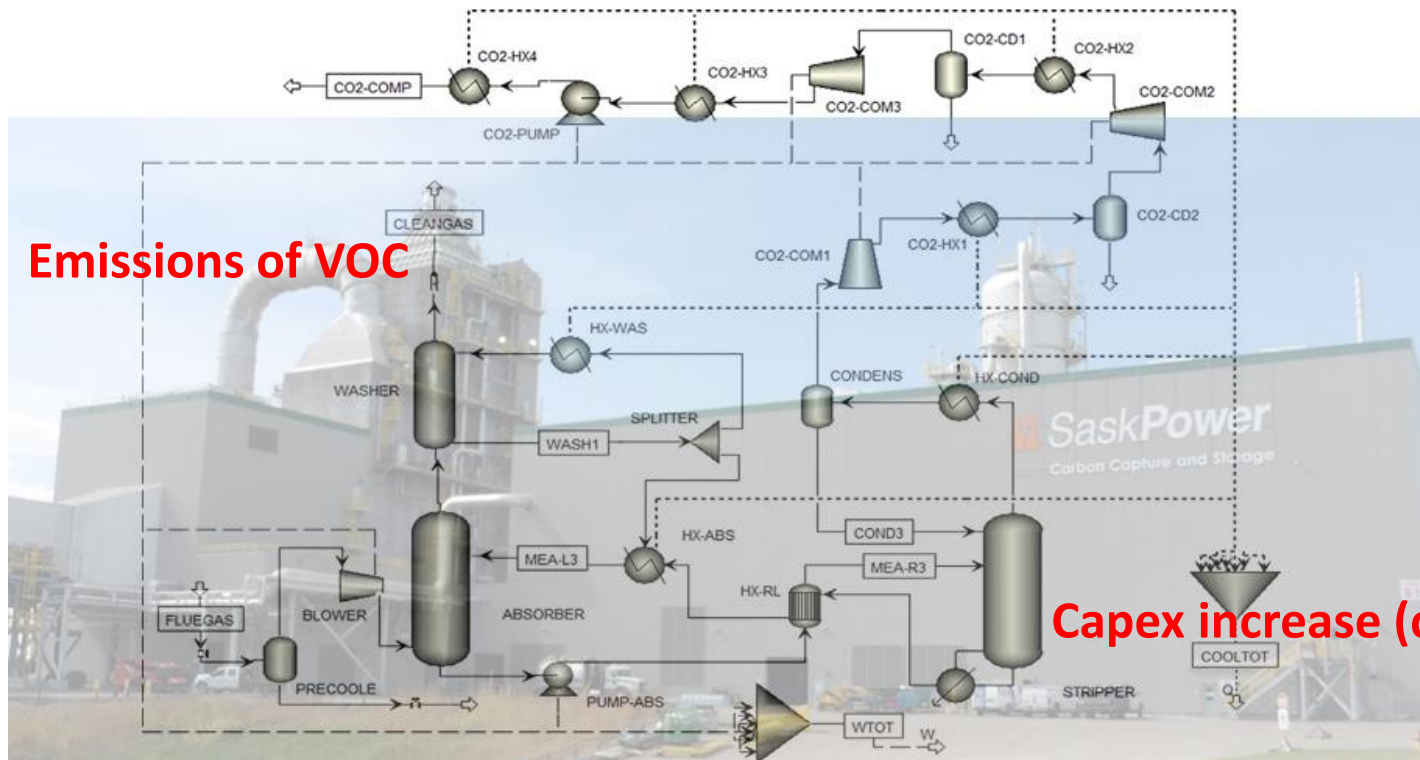
- 2 main focus at ULiège: Process modeling



# Post-combustion capture

## 2. Capture CO<sub>2</sub> from combustion gases

- 2 main focus at ULiège: Solvent stability



Emissions of VOC

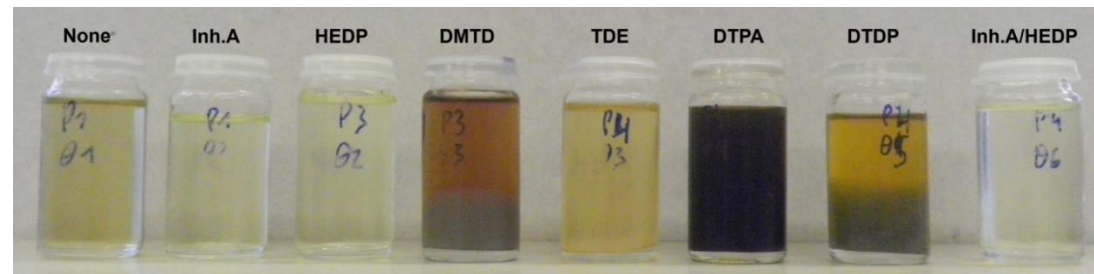
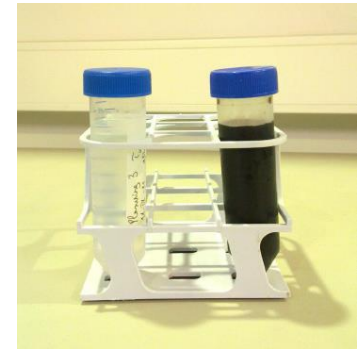
Capex increase (corrosion)

OPEX increase: viscosity, solvent properties...

# Post-combustion capture

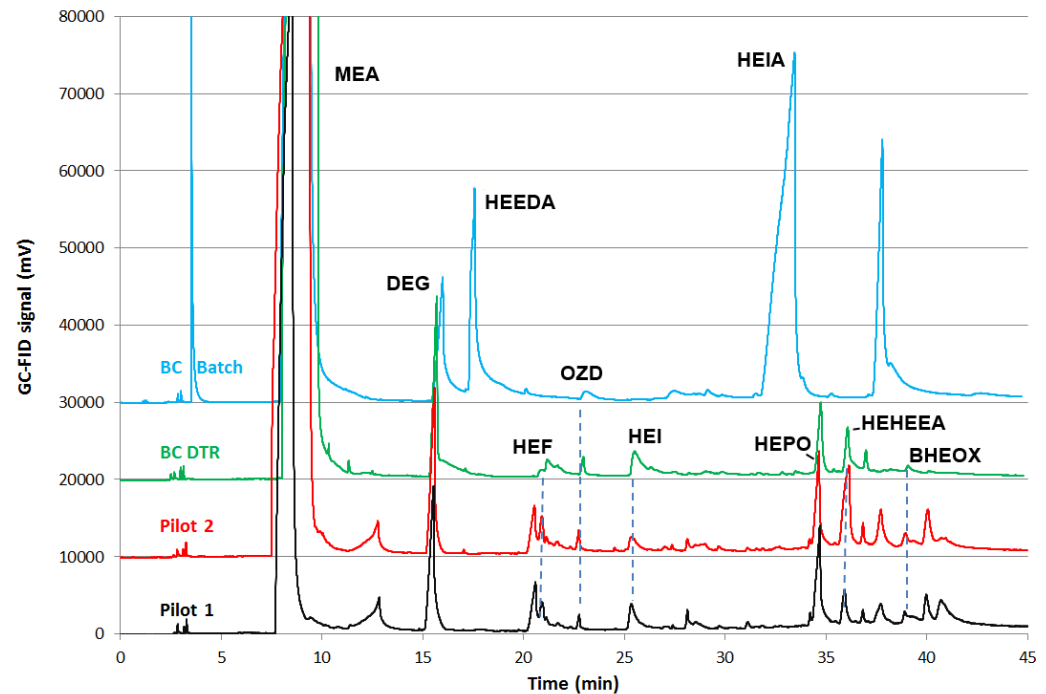
4 main types of solvent degradation

- Thermal decomposition  $> 200^{\circ}\text{C}$
- Thermal degradation ( $120 < T < 140^{\circ}\text{C}$ )
  - Irreversible reactions with  $\text{CO}_2$
- Oxidative degradation
  - Oxidation of amines with  $\text{O}_2$  present in flue gas
- Degradation with other flue gas contaminants
  - $\text{SO}_x$ ,  $\text{NO}_x$  ...



# Post-combustion capture

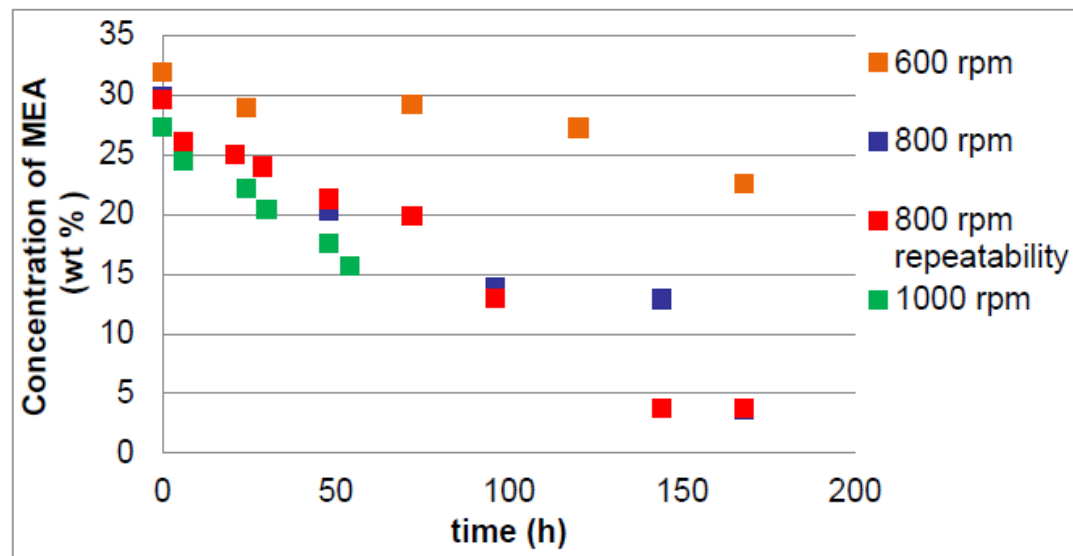
1st Objective: Representativity of accelerated degradation versus industrial degradation



=> 21% MEA loss after 7 days vs. 4% loss in 45 days

# Post-combustion capture

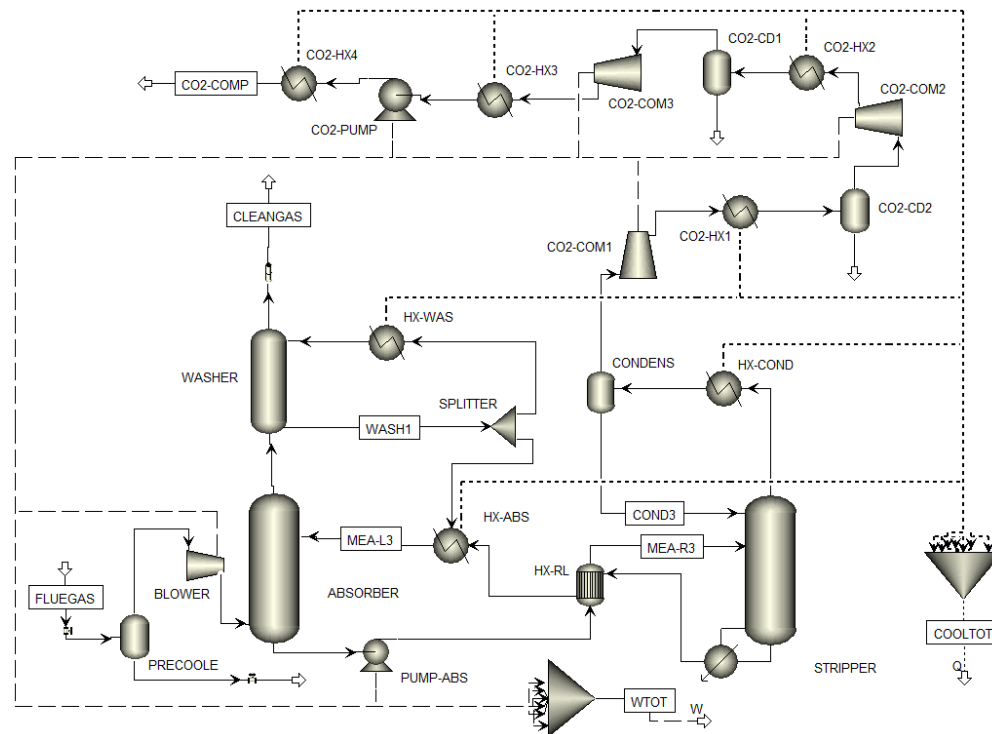
- Influence of operating conditions on solvent degradation
  - O<sub>2</sub>, CO<sub>2</sub> content in flue gas
  - Temperature & Mass transfer
  - Additives (dissolved metals, degradation inhibitors, SO<sub>x</sub>...)



=> kinetic model for oxidative and thermal degradation

# Post-combustion capture

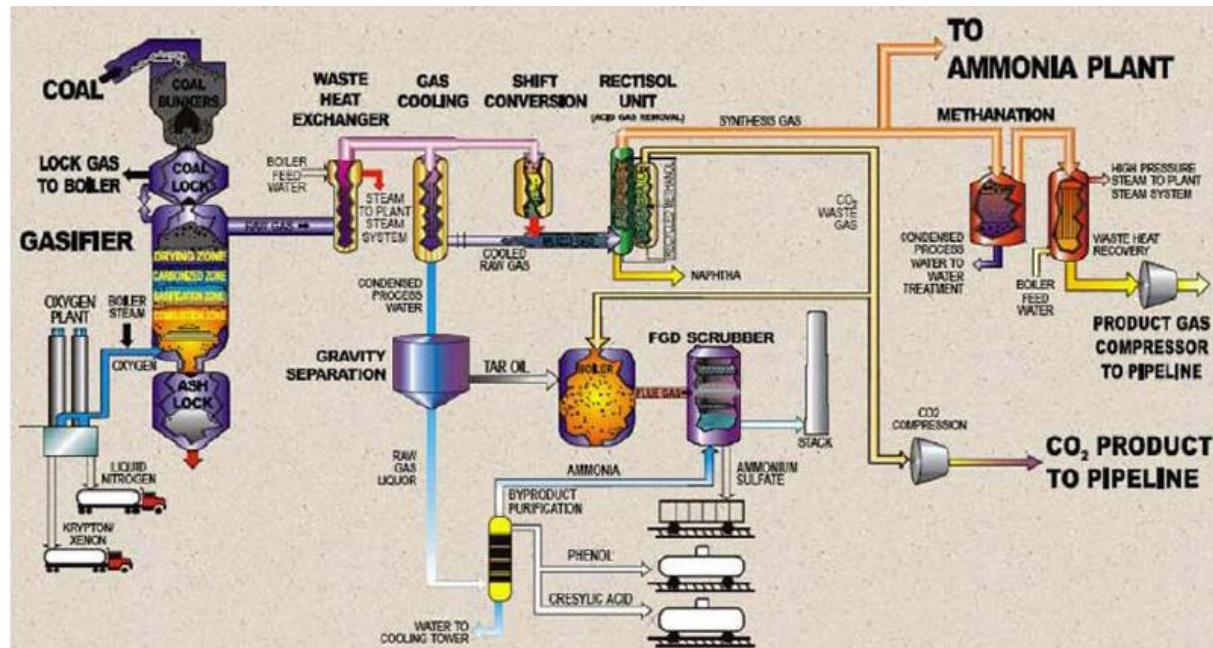
- This kinetic model is implemented in process models
  - => predict degradation depending on operating conditions
  - 81 g MEA/ton CO<sub>2</sub> < 284 g MEA/ton CO<sub>2</sub>



# Pre-combustion capture

## 3. Remove C from the solid fuel by gasification

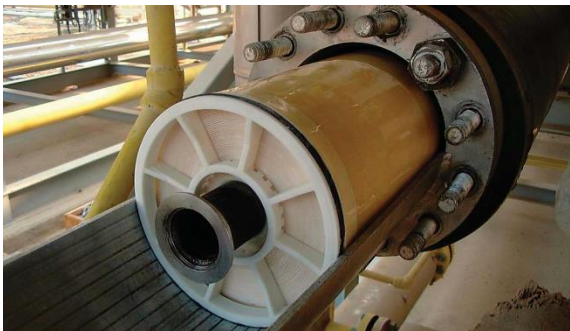
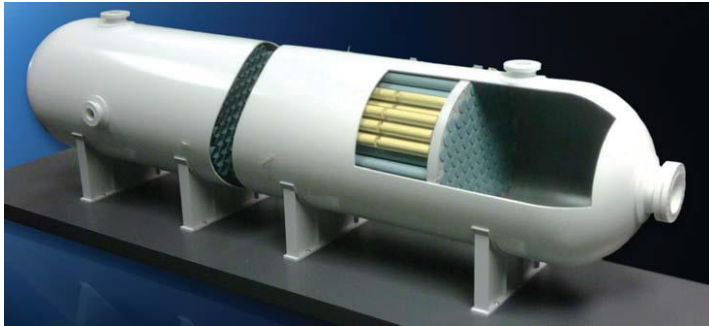
- ❑ Great Plains Synfuel Plant (US), 8 200 tCO<sub>2</sub>/day
- ❑ Rectisol process: physical absorption in cold methanol
  - Largest utility consumption and largest plant bottleneck



# Pre-combustion capture

## 3. Remove C from the fuel => Natural gas sweetening

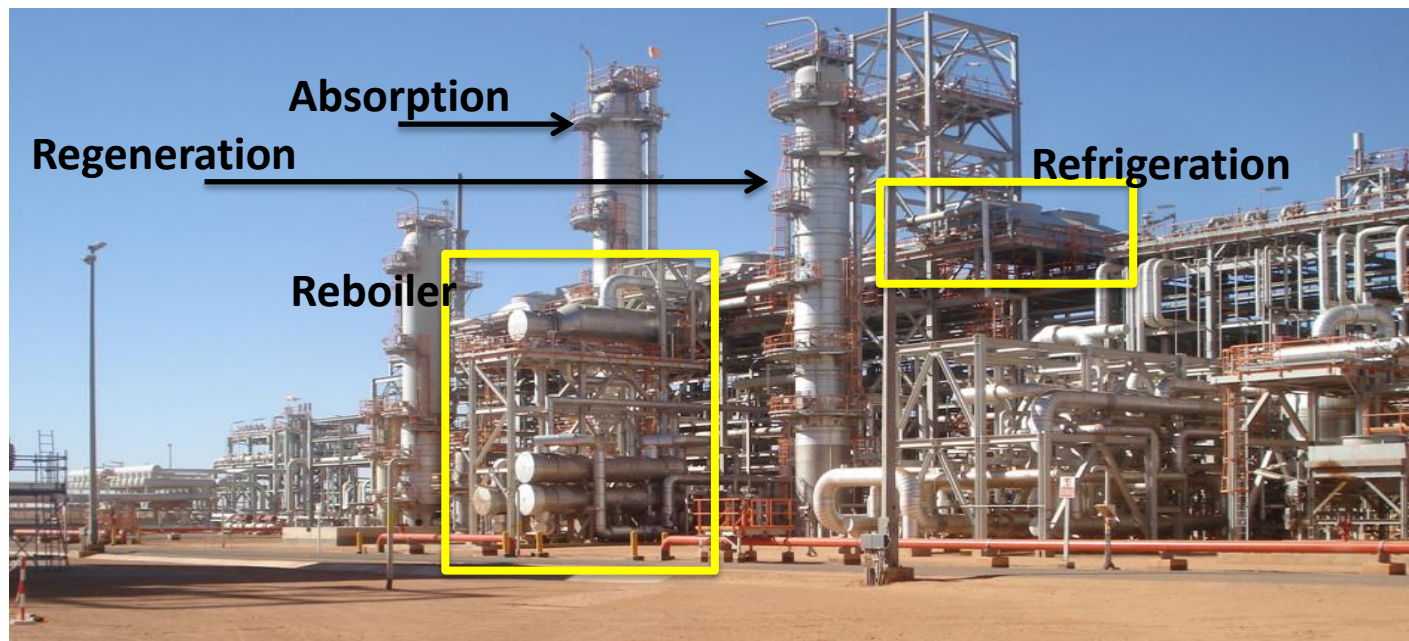
- Off-shore platforms: possibly use membranes



# Pre-combustion capture

## 3. Remove C from the fuel => Natural gas sweetening

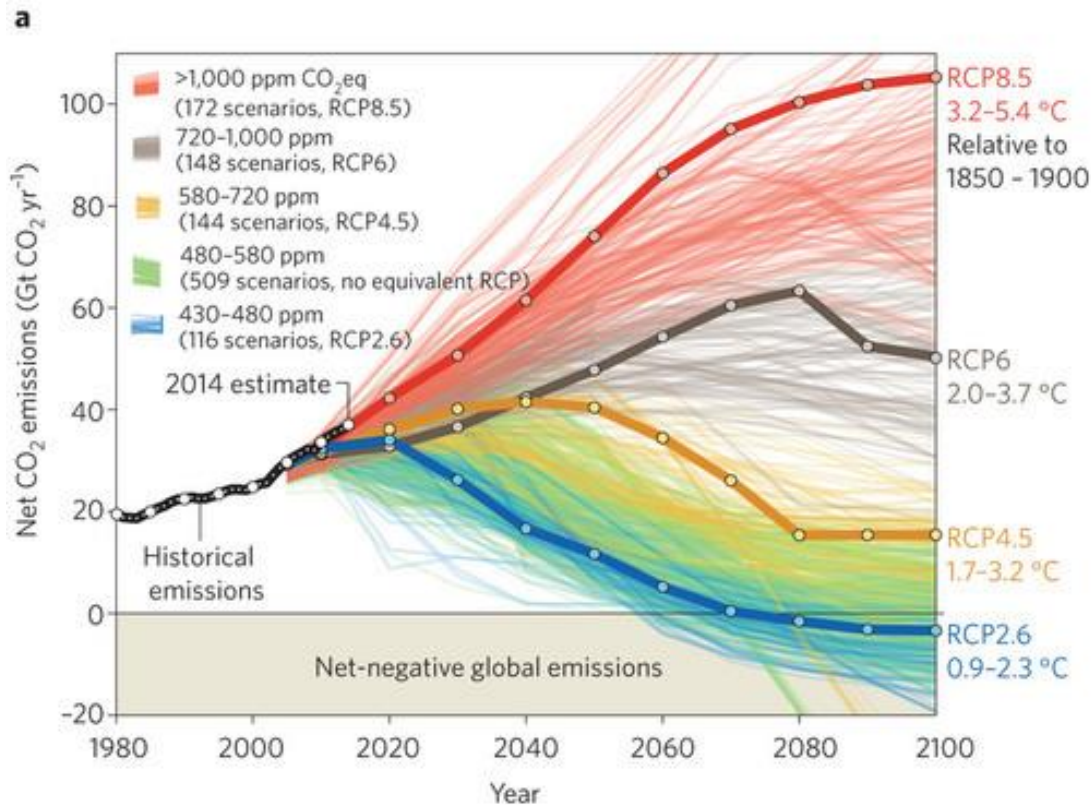
- ❑ Conventional process: absorption with solvents
- ❑ From 80 to 2 vol%; down to 50 PPM if liquefaction



- ❑ => Multi-objective optimization of sour natural gas sweetening processes

# Trends and challenges

- Negative CO<sub>2</sub> emissions
  - Biomass-enhanced CCS
  - Direct air capture



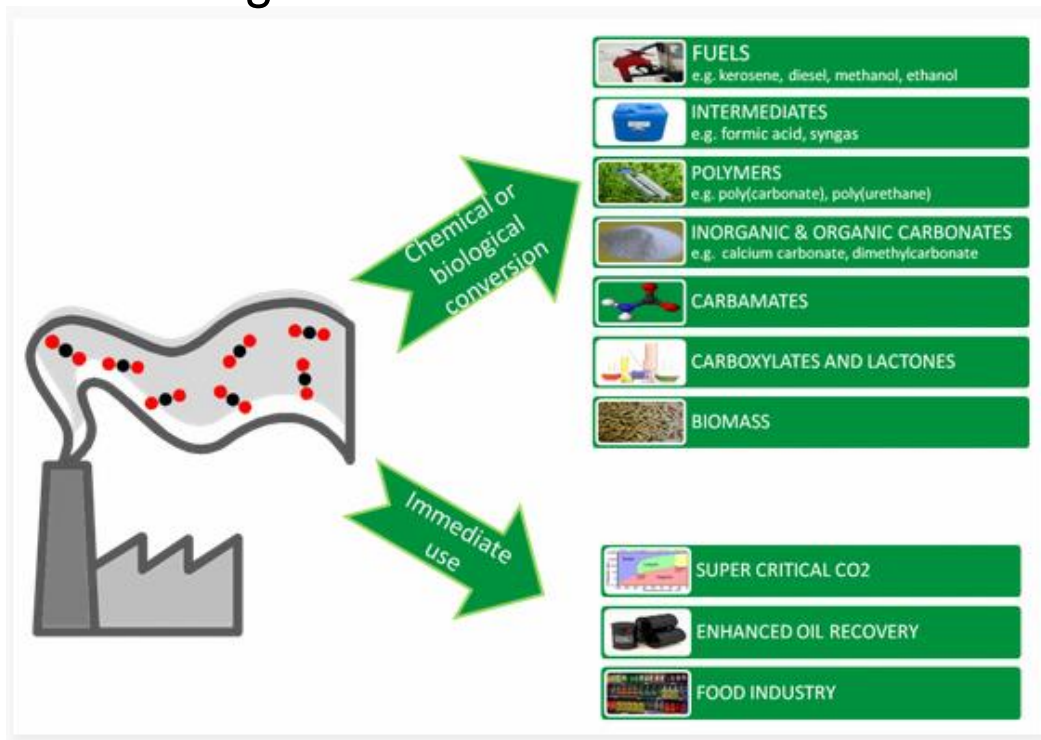
---

# 3. CO<sub>2</sub> re-use technologies

---

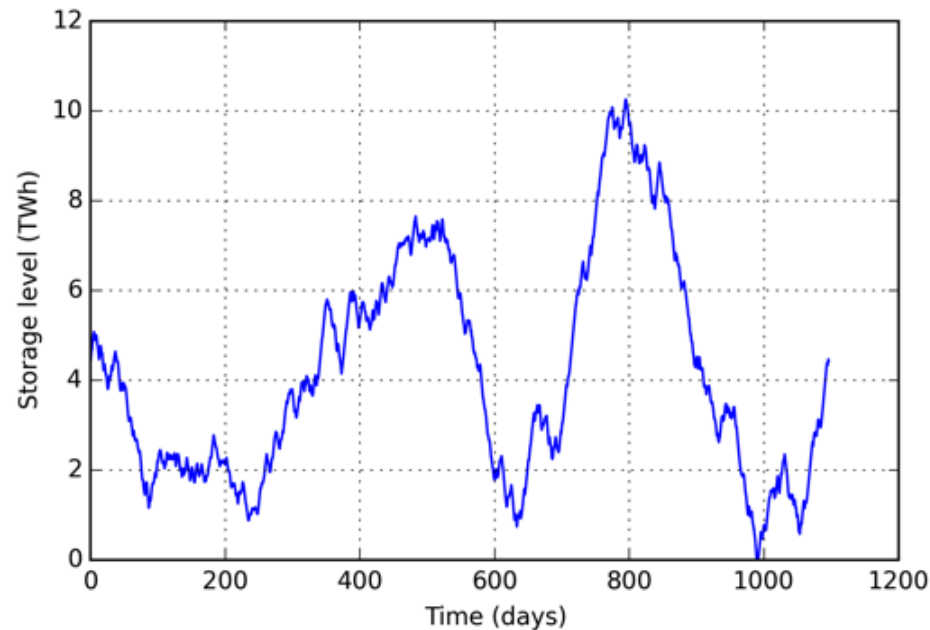
# Technologies and products

- Carbon is root of organic chemistry and materials
  - => Potential of applications for CO<sub>2</sub> is huge
    - New products
    - New technologies



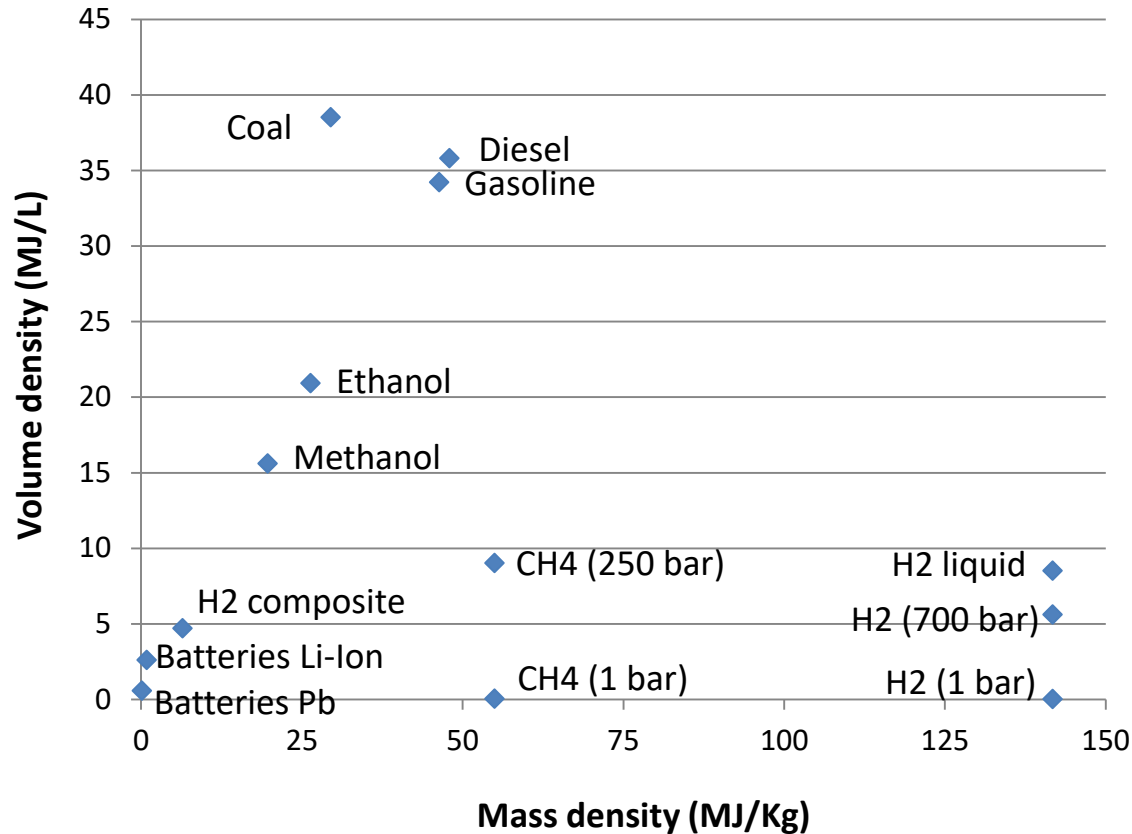
# Power-to-fuel

- Study with 100% variable renewables + storage for electricity grid:
  - Reasonable electricity cost (83.4 €/MWh)
  - Second and minute scale storage for frequency regulation
  - Inter-seasonal scale also needed!



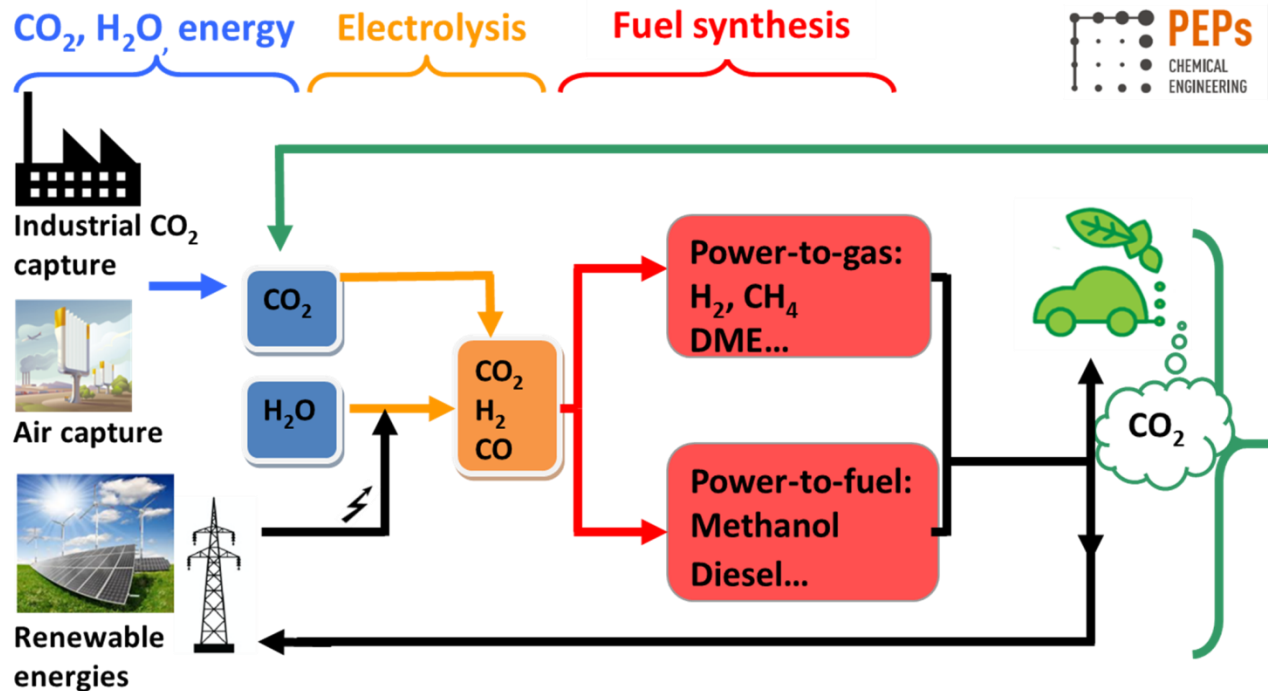
# Power-to-fuel

- High energy density is required for long-term storage



# Power-to-fuel

- CO<sub>2</sub> capture, water electrolysis, fuel synthesis



# Power-to-fuel

How to choose the right reactor?

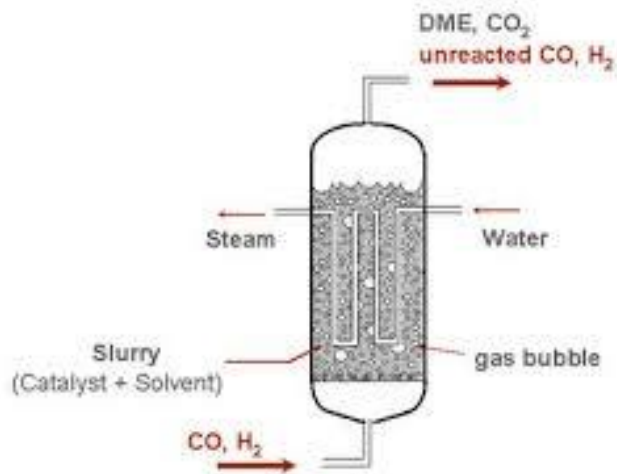
- High variety of reactions and operating conditions
- Rational analysis of processes to identify limiting factors

Criterion	Types of reactions
Phases	Homogeneous (1 phase) Heterogeneous (2, 3 or 4 phases)
Stoichiometry	Simple (1 reaction) Complex (multiple reactions, side reactions...)
Thermodynamics	Irreversible Equilibrium
Kinetics	Limiting factors are physical Limiting factors are chemical
Heat balance	(Strongly) Endothermic ( $\Delta H > 0$ ) (Strongly) Exothermic ( $\Delta H < 0$ )

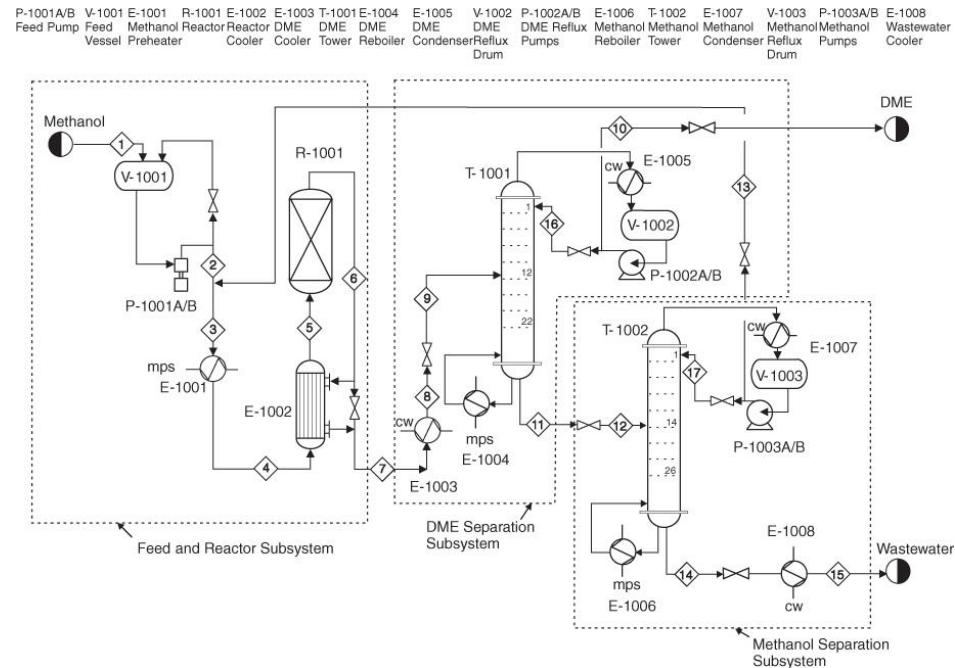
# Power-to-DME

## ■ DME ( $\text{CH}_3\text{OCH}_3$ )

- Directly from syngas => more exothermal => slurry
- From methanol => fixed bed gas reactor



Yagi et al., 2010. DOI: 10.2202/1542-6580.2267

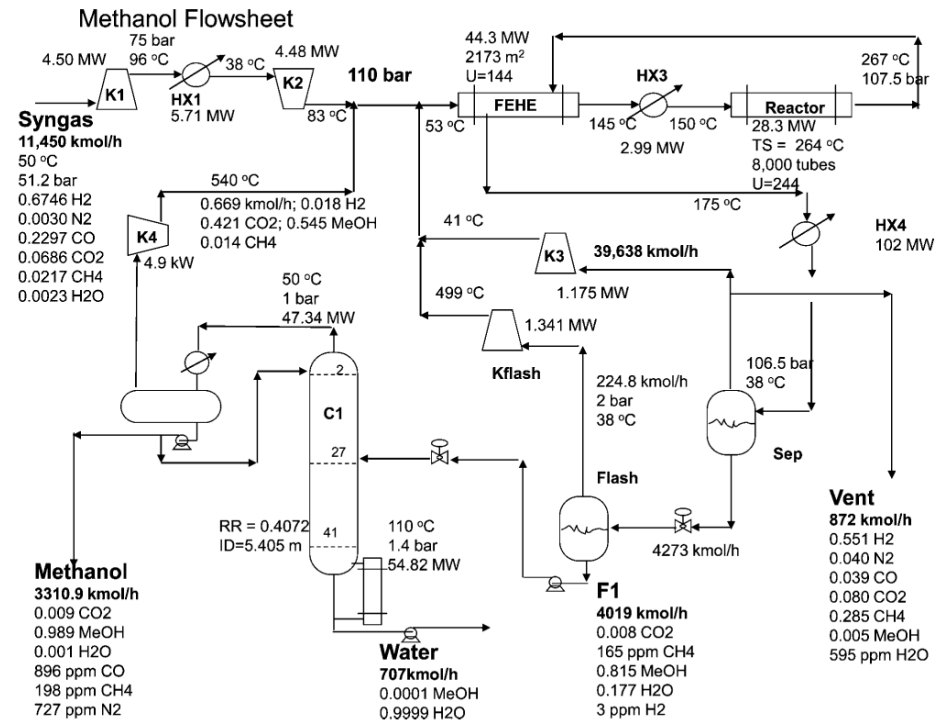


Turton et al., Prentice Hall, 2012

# Power-to-methanol

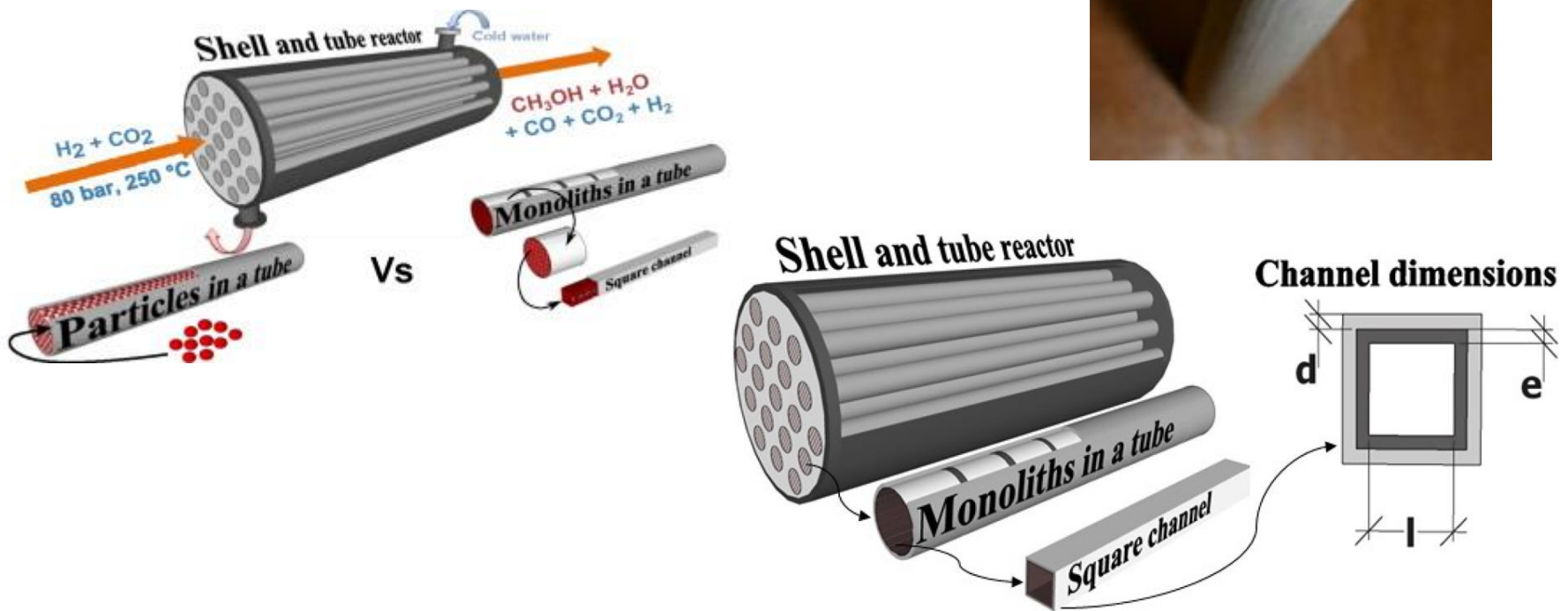
## ■ Conventional methanol synthesis

- ❑ Limiting step: thermodynamic equilibrium (25% H<sub>2</sub> conversion) + exothermal reaction
- ❑ => High P, Low T, large gas recycle
- ❑ => Shell & Tubes reactor



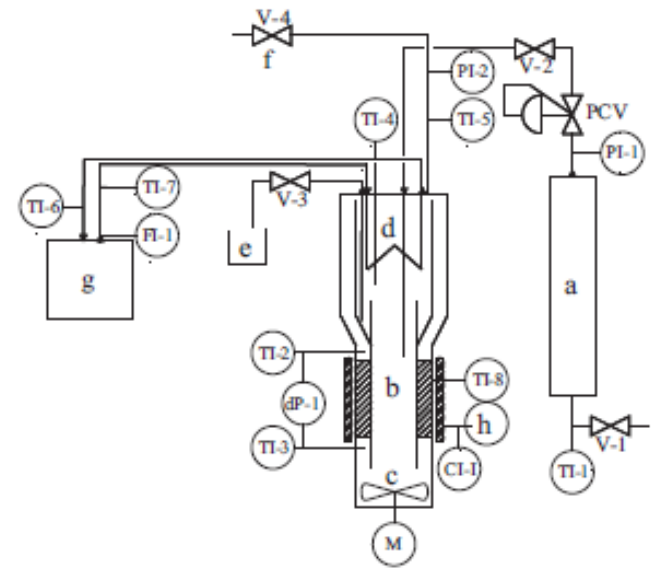
# Power-to-methanol

- Novel methanol reactor designs
  - Improve the heat management
  - Lower  $\Delta P$  at high flow rates



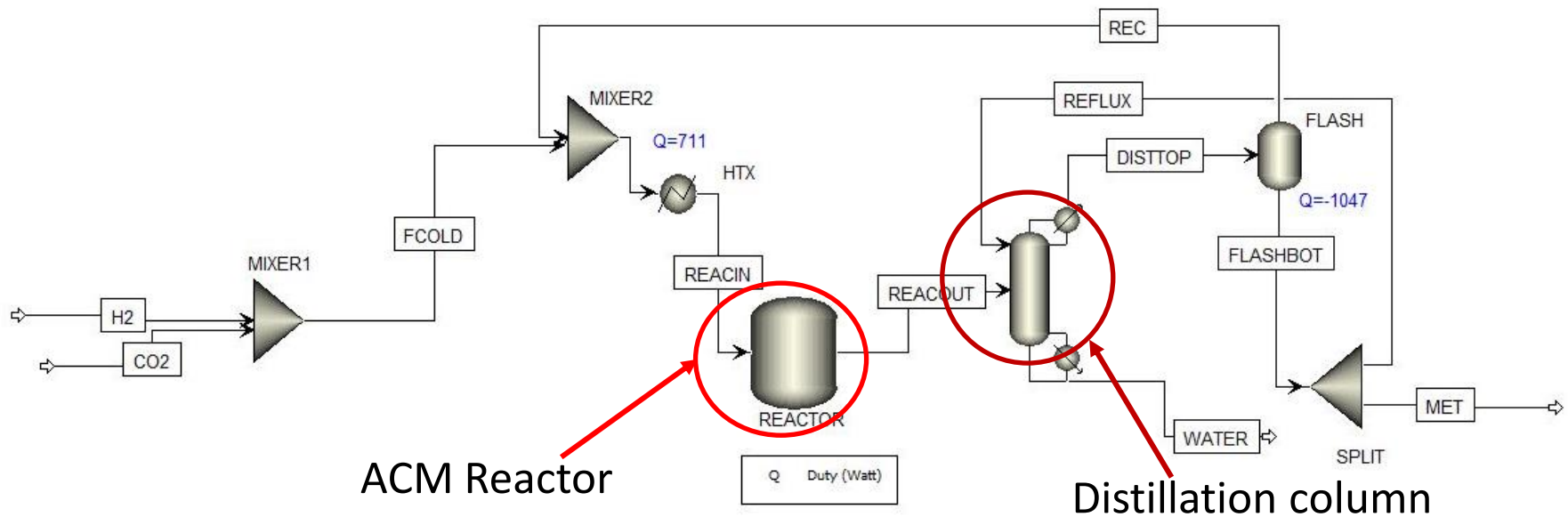
# Power-to-methanol

- Novel methanol reactor designs
  - Remove the thermodynamic limitation
  - Displace the equilibrium
  - Conversion reaches 99.9%!



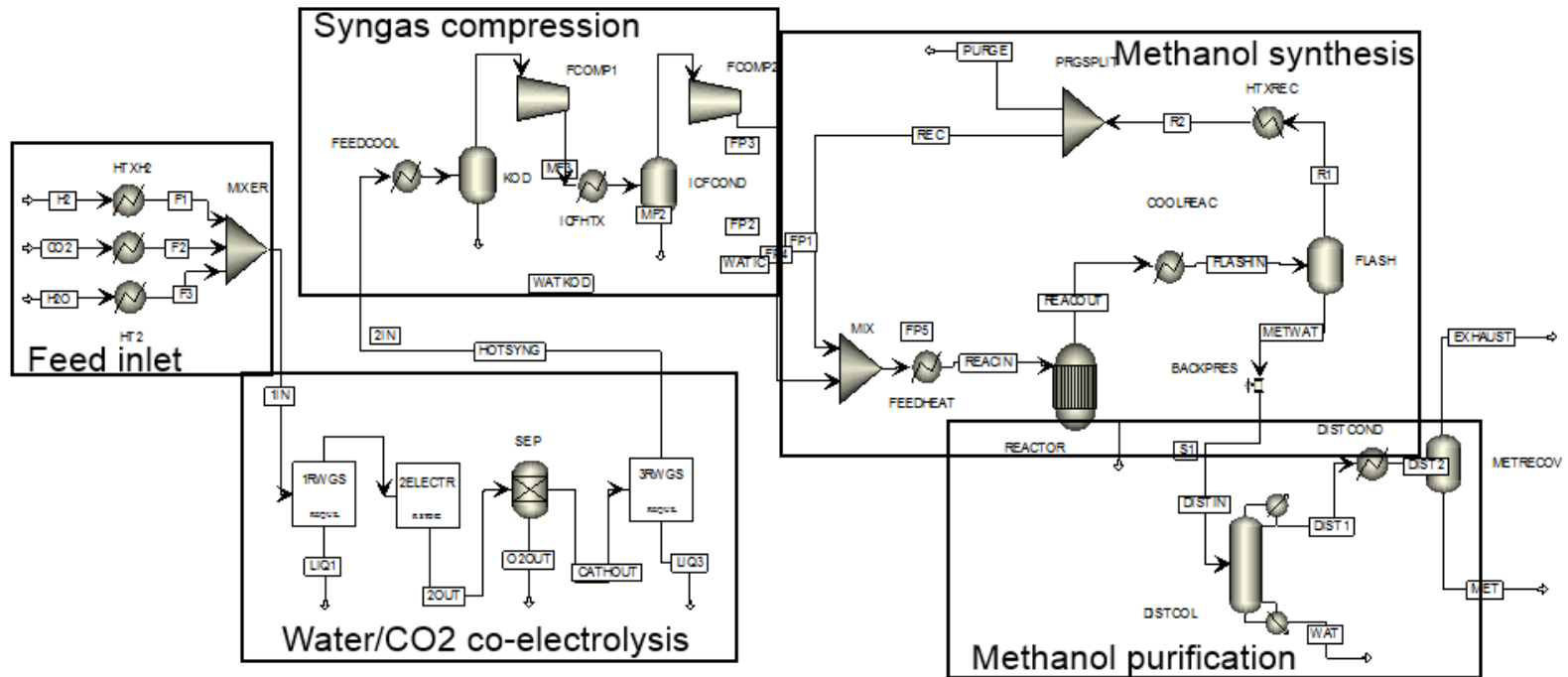
# Power-to-methanol

- Novel methanol reactor design
  - Intensification of synthesis reactor for CO<sub>2</sub> reduction to methanol



# Power-to-methanol

- Improved heat integration
  - Electricity to fuel efficiency increases from 40.1 to 53.0%



---

# 4. The CO<sub>2</sub>FUTURE Platform at ULiege

---

[www.chemeng.uliege.be/CO2Future](http://www.chemeng.uliege.be/CO2Future)

# CO<sub>2</sub>FUTURE

Chemical Transformation

Synthetic Fuels



Monomers & Polymers



Mineralization



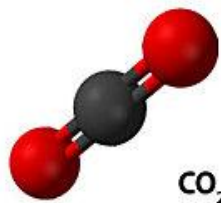
Pharmaceuticals & Cosmetology



Direct CO<sub>2</sub> use  
(solvent, foaming...)



Sourcing  
Capture & Purification



CO<sub>2</sub>

Process sustainability  
(LCA and economics)



Physical Use

Transversal  
W/P

PEPs

CHEMICAL  
ENGINEERING

- Industrial partner in Brazil
  - Produces sugar, ethanol and various other products
  - Convert CO<sub>2</sub> waste into new products => 400 tCO<sub>2</sub>/h
  - Looking for brazilian research partners



# A flavor of teaching

- Case study for grad students in Chemical Engineering

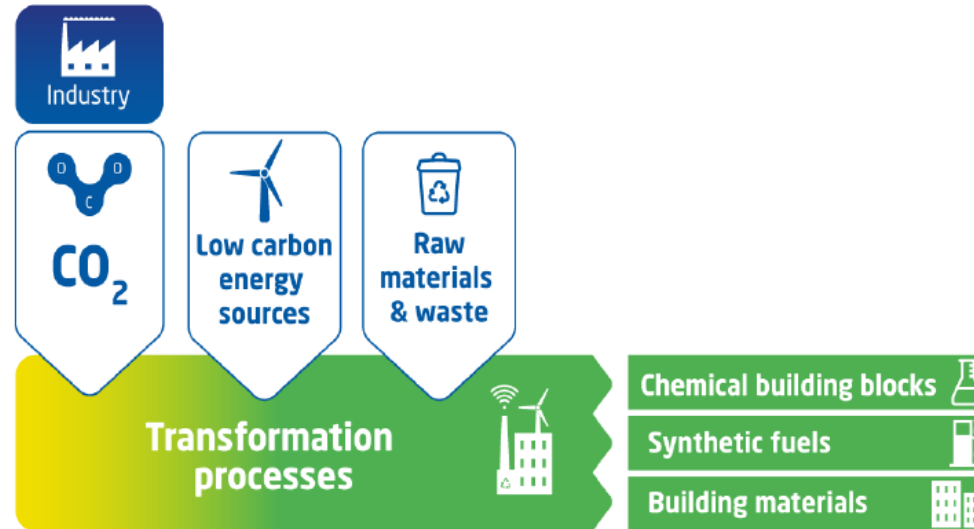
Process	Profit (USD/ton of CO <sub>2</sub> )	Profit (USD/ton of product)	Market Share of Product	TRL
Enhanced Oil Recovery	-10.79	N/A	4.276%	9
Concrete	9503.48	95.03	0.261%	5
Syngas Production	-81.45	-388.78	Naphta: 0.072% Diesel: 0.023%	Dry Reforming: 4 Fischer-Tropsch: 6
Microalgae Production	-452.02	-1674.15	0.275%	5
Methanol Production	-191.74	-333.46	1.932%	7
Polymer Production	1377.08	661.74	7769.067%	8
Urea Production	39.91	29.56	2.442%	9

=> Urea production process will be studied in Q2

# CO<sub>2</sub> Value Europe

- 30.11.2017: Creation of CO<sub>2</sub> Value Europe, an Association for promoting:

*“the development and market deployment of sustainable industrial solutions that convert CO<sub>2</sub> into valuable products, in order to contribute to the net reduction of global CO<sub>2</sub> emissions and to the diversification of the feedstock base.”*



---

Many thanks to the team...



# Thank you for your attention!

<http://klesbutterfly.com/2015/03/22/where-the-heck-is-liege/>



<https://vimeo.com/95988841>

**PEPs**

CHEMICAL  
ENGINEERING

# CO<sub>2</sub> Value Europe

## ■ Who's in?



+ ULiège!