

THE DISSOCIATIVE IONIZATION OF N₂O BY LOW-ENERGY ELECTRON IMPACT THE N⁺ AND N₂⁺ DISSOCIATION CHANNELS

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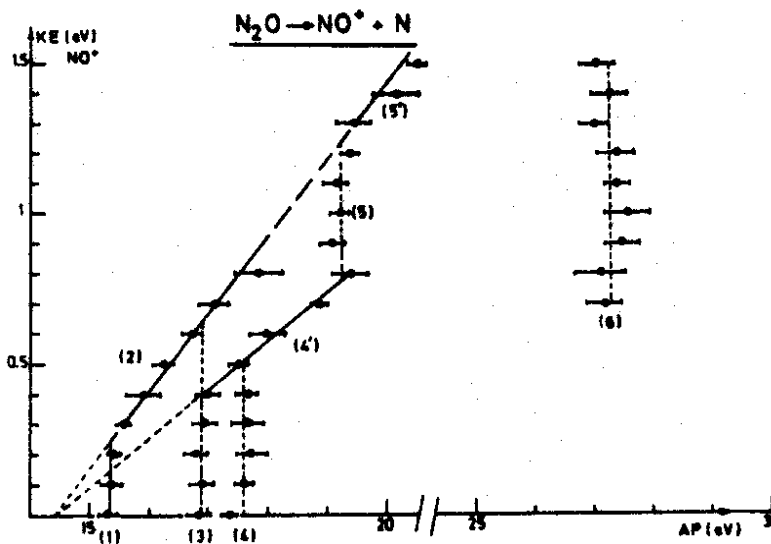
The N₂O molecule has been the subject of many investigations during the last few years. BERKOWITZ⁽¹⁾ made a detailed photoionization study of the four dissociation channels. BEAR⁽²⁾ and NENNER⁽³⁾ performed the threshold photoelectron-photoion coincidence study of N₂O. Few are the dissociative electroionization studies of this molecule.

In the present abstract a part of the dissociative electroionization work on N₂O, together with ion kinetic energy analysis, is reported.

The details of the instrumentation and the measuring techniques have been described elsewhere⁽⁴⁾. The result of the kinetic energy measurements and the threshold energy determinations for NO⁺ are summarized in fig. 1. Over the electron energy range investigated (15-40 eV) six different channels for NO⁺ formation have been evidenced.

Below 21 eV electron energy only one dissociation limit, i.e. N₂O + e⁻ → N(⁴S) + NO⁺(X¹Σ⁺) is involved and calculated at 14.2 eV⁽¹⁾ in agreement with earlier measurements⁽¹⁻³⁾. Experimentally this dissociation limit is measured at 14.1 eV and 14.4 eV by extrapolation of the least square fits (2) and (4') in fig. 1.

FIG. 1. Kinetic Energy-versus-electron energy plot for NO⁺ ions from NO.



Several threshold energies have been measured in the electro-ionization efficiency curve of NO⁺, i.e. at (15.3 ± 0.1) eV, (16.8 ± 0.1) eV, (17.6 ± 0.1) eV, (19.3 ± 0.2) eV and (27.3 ± 0.2) eV. They are mainly interpreted in terms of dissociative autoionization and/or predissociation of ionized N₂O⁺ states through the ⁴Σ⁻ ionic state. The NO⁺ fragment is mainly produced in vibrationally excited levels of its ground X¹Σ⁺ state.

In a similar way, the N₂⁺ dissociation channel has been investigated in detail and will also be reported at the meeting.

References

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