

Commission

INTRODUCTION & MOTIVATION

Atmospheric methane (CH₄), with a lifetime of 8-10 years, is the second most important anthropogenic greenhouse gas (radiative forcing by human-linked greenhouse gases since 1750 is due to methane Atmospheric methane for the increase in radiative forcing by human-linked greenhouse gases since 1750 is due to methane (Nisbet et al., 2014). Identified emission sources include anthropogenic (coal mining, oil and gas exploitation, rice cultures, domestic ruminant animals, biomass burning, and waste management) and natural contributions (wetlands, termites, methane hydrates and ocean). Due to its sinks, methane plays a role of regulator of hydroxyl radical (OH) and of ozone precursor in the troposphere (Montzka et al., 2011) while in the stratosphere, it acts as a source of water vapor and as a sink for chlorine atoms which reduces the rate of ozone depletion. Atmosphere, it acts as a source of water vapor and as a sink for chlorine atoms which reduces the rate of ozone depletion. (WMO, 2016). In the last ten years, methane has been on the rise again at rates of ~0.3%/year (e.g., Bader et al., 2017), after a period of stabilization of about 5 years. This recent increase is not fully understood due to remaining uncertainties in the methane budget, influenced by numerous anthropogenic and natural emission sources. In order to examine the cause(s) of this increase, we focus on one of the two main methane isotopologues, i.e. CH₃D. Methane isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two main methane isotopologues are emitted in the atmosphere with different ratio depending on the emission processes involved (Rigby et al., 2012). As heavier isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two main methane isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two main methane isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two main methane isotopologues are emitted in the atmosphere with different ratio depending on the emission processes involved (Rigby et al., 2012). As heavier isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two mains are emitted in the emission processes involved (Rigby et al., 2012). As heavier isotopologues will react more slowly than ¹²CH₄, each isotopologue will be depleted from a solution of the two mains are emitted in the emission processes involved (Rigby et al., 2012). As heavier isotopologue will be depleted from a solution of the emission processes involved (Rigby et al., 2012). As heavier isotopologue will be depleted from a solution of the emission of the emis the atmosphere at a specific rate depending on the removal process (Saueressig et al., 2001). Methane isotopologues are therefore good tracers of the methane budget.

INSTRUMENTATION & DATASET

First development and optimization of the retrieval strategy for CH₃D from ground-based FTIR (Fourier Transform infrared) solar observations with the SFIT-4 algorithm.

Toronto, ON, Canada (Wiacek, et al., 2007)

Bomem DA8 Fourier Transform Spectrometer

~1430 days of observations since May 2002

Eureka, NU, Canada (Batchelor et al., 2009, Fu et al., 2007)

Bruker IFS-125 HR Fourier Transform Spectrometer

 \sim 760 days of observations since July 2006

Portable Atmospheric Research Interferometric Spectrometer for the InfraRed

~240 days of spring observations at Eureka since 2004

Jungfraujoch, Switzerland (Zander et al., 2008)

Bruker 120-HR : ~2590 days of observations since 1990



WATER VAPOR SPECIFIC WINDOW ?

(2941.65 – 2941.89 cm⁻¹, Bader et al., 2017) is tested as the fifth window.



Figure 4 - Correlation between CH_3D and H_2O retrieved total columns with (in red) and without the use of the "H₂O" window (in green). While combining the 4 windows

smoothing reduces and measurement errors and increases DOFS (see Fig. 3), overlapping of H₂O and CH₃D absorption lines in windows **C** and **D** (see Fig. 2), may result in a misinterpretation of the CH₃D total columns for humid days.

First retrievals of methane isotopologue, CH₃D, from FTIR ground-based observations

<u>*W. Bader*^{1,2}</u>, K. Strong¹, K. Walker¹, E. Buzan³

¹Department of Physics, University of Toronto, Toronto, ON, Canada

²Institute of Astrophysics and Geophysics, University of Liège, Liège, Belgium

³Department of Chemistry and Biochemistry, Old Dominion University, Norfolk, Virginia, USA

A PRIORI MIXING RATIO PROFILES

CH₃D vertical mixing ratio information is obtained from a Whole Atmosphere Community Climate Model (WACCM v.4, Marsh et al., 2013) simulation run as a standalone model with a resolution of 4×5° (latitude x longitude) and 66 vertical levels (Buzan et al., 2016). The model was run as a perpetual year 2000 for a total of 20 years: 17 years of spin-up time followed by 3 years that were averaged to produce a single a priori mixing ratio profile per station. Water vapor mixing ratio profiles from daily averages NCEP reanalysis and ozone monthly averaged profiles over the years 1980-2020 from a WACCM simulation (V6) are used as a priori.

SPECTROSCOPIC LINELISTS

In order to minimize residuals, the best combination of available spectroscopic linelists has been determined : H₄ and isotopologues

- Spectroscopic linelist for GGG2014, TCCON data archive developed by Toon et al. (2014) Water vapour
- Experimental linelist of water vapor line parameters developed by Loos et al. (2017a and 2017b) **Interfering species**
- The HITRAN 2008 compilation (Rothman et al., 2009), for other interfering species, including O₃

CH4 PROFILE RETRIEVAL ?

neously retrieve precise and accurate ¹²CH₄ total columns along with CH₃D. We observe to, averaged over 2010 and 2005 and with DOFS of 1.43 and 3.95, respectively. a systematic bias between the retrieved and the archived CH₄ columns of approximately 11.2 and 11.5 % respectively with a scaled or retrieved CH_4 profile. Retrieving the CH_4 profile increases smoothing, measurement and interference errors of about 0.02 %.



Figure 5 - CH₄ total column time series as retrieved from Eureka observations with (blue) or without CH₄ profile retrieval (red) along with CH₄ as archived on the NDACC website. www.ndacc.org

In order to better identify the water vapor interferences, a H₂O specific window To assess isotopic ratios of CH₄ through δ-D computation, it is necessary to simulta- Figure 6 illustrates averaging kernels, eigen vectors and associated eigen values for Eureka and Toron-

PItit 20

10





ERROR BUDGET AND INFORMATION CONTENT



REFERENCES

Science Postdoctoral Fellowship Award.

Bader, W. et al., 2017, Atmos. Chem. Phys., 10.5194/acp-17-2255-2017 Batchelor, R. et al., 2009, J. Atmos. Ocean. Technol., 10.1175/2009JTECHA1215.1 Buzan, E. M. et al., 2016, Atmos. Meas. Tech., 10.5194/amt-9-1095-2016 Fu, D. et al., 2007, J. Quant. Spectrosc. Radiat. Transf., 10.1016/j.jqsrt.2006.05.006 Loos J. et al., 2017a, J. Quant. Spectrosc. Radiat. Transf., 10.1016/j.jqsrt.2017.02.013 Loos J. et al., 2017b, J. Quant. Spectrosc. Radiat. Transf., 10.1016/j.jqsrt.2017.03.033 Marsh, D. R., et al., 2013, J. Clim., 10.1175/JCLI-D-12-00558.1 Montzka, S.A. et al., 2011, Science, 10.1126/science.1197640 Nisbet, E.G., et al., 2014, Science, 10.1126/science.1247828 Rigby, M., et al., 2012, J. Geophys. Res, 10.1029/2011JD017384 Rodgers, C.D., 1990, J. Geophys. Res., 10.1029/JD095iD05p05587 Rothman, L. S. et al., 2009, J. Quant. Spectrosc. Radiat. Transf., 10.1016/ j.jqsrt.2009.02.013 Saueressig, G. et al., 2011, J. Geophys. Res., 10.1029/2000JD000120 Steck, T. et al., 2002, Appl. Opt., 10.1364/A0.41.001788 Stocker, T., 2013, 5th IPCC Report, Working Group I Toon, G. C., 2014, doi:10. 14291/tccon.ggg2014.atm.R0/1221656 Wiacek, A., et al., 2007, J. Atmos. Ocean. Technol., 10.1175/JTECH1962.1 World Meteorological Organization, Greenhouse gas bulletin, N°12, October 2016 Zander, R., et al., 2008, Science of the Total Env., 10.1016/j.scitotenv.2007.10.018 Author contact : wbader@atmosp.physics.utoronto.ca Acknowledgments - W. Bader has received funding from the European Union's Horizon2020 research and innovation programme under the Marie Sklodowska-Curie grant agreement n°704951, and from the University of Toronto through a Faculty of Arts &