Methane in the South China Sea and the Western Philippine Sea

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1. Introduction

Methane (CH\textsubscript{4}), the most abundant hydrocarbon in the atmosphere, has a global warming potential of 25 times that of carbon dioxide (CO\textsubscript{2}), plays an important role in the atmospheric chemistry (Naqvi et al., 2010), and contributes approximately 75% of global oceanic CH\textsubscript{4} emissions (Bange et al., 1994). The global oceanic CH\textsubscript{4} emissions from these environments are probably higher due to the contribution from sedimentary sources in well-mixed coastal zones (Borges et al., 2016). The oversaturation of CH\textsubscript{4} in the oxygenated surface mixed layer has been widely known for more than four decades (Lamontagne et al., 1973; Scranton and Brewer, 1977; Forster et al., 2008). The CH\textsubscript{4} concentrations in near-surface waters throughout much of the world’s oceans are 5–75% oversaturated with respect to the atmospheric equilibrium (Karl et al., 2008). Previous research has revealed that CH\textsubscript{4} is released during zooplankton grazing (de Angelis and Lee, 1994) or is formed in anoxic microenvironments within zooplankton fecal pellets (Traganza et al., 1979; de Angelis and Lee, 1994; Karl and Tilbrook, 1994), which are mostly found in the euphotic zone. Recent impingement of human activities on oceanic CH\textsubscript{4} emissions, such as waste water discharge into the coastal areas, unlike on terrestrial emissions, is not well understood and has been poorly quantified (Naqvi et al., 2010). Especially, continental shelves and estuaries contribute approximately 75% of global oceanic CH\textsubscript{4} emissions (Bange et al., 1994), and CH\textsubscript{4} emissions from these environments are probably higher due to the contribution from sedimentary sources in well-mixed coastal zones (Borges et al., 2016).

The concentrations of both CH\textsubscript{4} and chlorophyll \textalpha were high in the 150 m surface layer of the wPS, but were not significantly correlated with each other. CH\textsubscript{4} concentrations generally declined with increasing depth below the euphotic zone but remained constant below 1,000 m, both in the SCS and the wPS. Some high CH\textsubscript{4} concentrations were observed at mid-depths and bottom waters in the SCS, and were most likely caused by the release of CH\textsubscript{4} from gas hydrates or gas seepage.

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research has also determined that the formation of CH$_4$ inoxic conditions can occur via methylphosphonate cycling in subtropical gyres, which are phosphate-depleted (Karl et al., 2008), and by dimethylsulphide (DMS) cycling in both polar and tropical oligotrophic waters (Damm et al., 2010; Zindler et al., 2013). Florez-Leiva et al. (2013) suggested that the CH$_4$ production may be induced by DMS cycling in the surface water of an upwelling ecosystem off central Chile in which nutrient depletion has never been observed even in winter (Morales and Anabalón, 2012). Overall, the enhancement of primary production may lead to increased CH$_4$ concentrations, which has led Bogart et al. (2014) to suggest the possibility of a relationship between CH$_4$ concentration and phytoplankton standing stocks (i.e. chlorophyll a).

Only recently has CH$_4$ in the South China Sea (SCS) and the western Philippine Sea (wPS) been studied and the focus has usually been on the surface water (Chen et al., 1994; Rehder and Suess, 2001), the bottom water and sediments (Chuang et al., 2006; Yang et al., 2006). Chen and Tseng (2006), Chen et al. (2008a, 2008b) and Zhou et al. (2009) reported the CH$_4$ distribution in the water column of the SCS but the study was only local. Here, for the first time we investigated the CH$_4$ distribution throughout the SCS and the wPS. Our results elucidate the spatial and vertical distributions of CH$_4$ in the water column of the SCS and wPS, as well as its sea-to-air fluxes. Further, we discuss differences between these seas and factors that influence the distribution and fluxes of CH$_4$.

2. Materials and methods

2.1. Study area

The SCS is a semi-enclosed marginal sea off the Asian continent in the West Pacific. It is the largest marginal sea in the world with an area of 3.5×10$^6$ km$^2$ and an average depth of 1350 m. The SCS properly characterized by either a tropical or subtropical climate has both deep basin and extensive shelf systems at its northern and southern boundaries, which are associated with large riverine inputs. At the southern edge of the SCS lies the Sunda Shelf, which connects the sea to the Straits of Malacca and has an average depth of approximately 50 m. The eastern part of the SCS is connected with the Sulu Sea through the Mindoro Strait, and the northern part of the SCS is connected with the East China Sea through the Taiwan Strait. The northeastern SCS is connected with the wPS via the Luzon Strait, which is around 2200 m in depth and has the deepest sill that connects the SCS with any adjacent body of water. The SCS also features dynamic exchange with the wPS via an upper water exchanges with the Kuroshio and inflow at depth (Chen et al., 2001; Dai et al., 2013; Du et al., 2013).

The Asian monsoon dominates climatic variations at the sea–air interface of the SCS. The southwest monsoon season runs from May to October and brings a large amount of rainfall. The northeast monsoon season runs from November to April and is characterized by the high wind speeds (Han, 1998).

The wPS is located in the western part of the North Pacific from 123°E to 135°E and from 10°N to 35°N. The mean depth of the wPS is about 5500 m. This study focuses on the area between 120.5°E and 130°E and between 21°N and 28°N, where the wPS connects with the SCS and water masses exchange frequently.

2.2. Water samples collections

Samples were taken during six cruises on board R/V Ocean Researchers I and III in the SCS (Fig. 1a; Table 1): ORIII-896 (August, 2003), ORI-695 (September, 2003), ORIII-983 (July, 2004), ORIII-1081 (July, 2005), ORI-802 (July, 2006) and ORI-837 (July, 2007).

In this study, four contrasting physical–biogeochemical domains (Table 1) are examined to provide a better understanding of the spatial variability of CH$_4$ distributions and fluxes in the SCS. Domain A is in the northeastern part of the SCS, close to the south of the Taiwan Strait. Samples were collected in domain A during the ORIII-896 and ORIII-983 cruises in August 2003 and July 2004, respectively. This area is affected by Kuroshio intrusions, which generate various mesoscale eddies, upwelling and internal waves (Yuan et al., 2008; Sheu et al., 2010; Chen et al., 2015; Huang et al., 2015). Cruises ORI-695 and ORIII-1081 were conducted in September 2003 and July 2005, respectively in domain B in the warm season when the Pearl River exhibited a large discharge (Chen et al., 2008a; Gan et al., 2009; Cao et al., 2011; Han et al., 2012). The Pearl River (Zhujiang) is the second largest river (after the Mekong River) that enters the SCS (Chen et al., 2008a). Domain C is located in the southwestern part of the SCS. In July 2006, samples were taken during the ORI-802 cruise from domain C and time-series station SCS1, which was located in domain B. Some sampling stations were on the Sunda Shelf and others were off the Sunda Shelf and on the continental slope, where the depth increases sharply. Domain D is located from the eastern part to the southern part of the SCS. The samples in domain D were taken during July 16–31 of 2007 during the ORI-837 cruise. Surface water samples were taken from the Luzon Strait and along the western coast of Luzon Island, Palawan Island and northwestern coast of Borneo Island. Discrete water samples were taken at various depths from the eastern part of the SCS to the southern part of the SCS.

Samples from the wPS were collected during three cruises (Fig. 1b). Discrete water samples were taken at various depths on ORI-725 (July, 2004) and ORIII-1149 (May, 2006). These two cruises were carried out along 22°N from 121°E to 126°E in July 2004 and from 121°E to 124°E in May 2006. Surface water samples were taken from the northeastern coast of Taiwan to the Luzon Strait on ORI-837 (July, 2007). The analysis herein also included a very limited CH$_4$ data-set which was obtained from 25 samples from surface waters collected exclusively in the wPS in various cruises in the years 1992–1996.

Water samples from various areas and depths were collected using a Rosette sampler that was fitted with 5 L or 10 L Niskin bottles, along with a conductivity–temperature–depth (CTD) probe (Sea-Bird 911, USA). Saturated HgCl$_2$ was added to all samples to inhibit microbial activity, except for those samples that were used to measure salinity. Temperature data were obtained from the CTD profiles, whereas the salinity values that were used in the analysis were determined from discrete AUTOSAL measurements.

Water samples for measuring CH$_4$ concentration were collected in 120 mL dark glass bottles. The bottles were rinsed three times with the sampled water. After 1–2 fold of bottle volume was allowed to overflow the bottle, 10 mL of the water sample was withdrawn from the bottle to create an air space, and 0.5 mL saturated HgCl$_2$ was then added. The sample bottles were then immediately sealed with a butyl rubber stopper and an aluminum cap. The samples were stored at 4°C in the dark. All the water samples were transferred to a laboratory and analyzed within three months of collection.

2.3. Atmospheric air samples collections

Atmospheric CH$_4$ samples were collected using modified version of the sampling method of Chang and Yang (1997). A 12 mL disposable plastic syringe with an 8 cm-long plastic tube was used to withdraw 12 mL of air four times from a 12 mL serum bottle to increase the exchange of the atmospheric gas into the 12 mL serum bottle. The serum bottle was then sealed with a butyl rubber stopper and an aluminum cap. While making CTD casting, gas samples were taken on the top deck to collect clean air, facing into the wind to prevent contamination by the emissions of the ship.

2.4. Chemical analysis

The salinity of the discrete samples was determined by measuring
conductivity using an AUTOSAL salinometer, which was calibrated with standard seawater (batch no. P141) from the International Association for the Physical Sciences of the Oceans (IAPSO). The precision of the measurements was ± 0.003 salinity unit. Chlorophyll $a$ samples were collected by filtration through a 0.45 µm diameter Millipore polycarbonate filter. A Turner Designs model 10-AU fluorometer (Varian Eclipse) was utilized to measure chlorophyll $a$ concentration following extraction by 90% acetone (Strickland and Parsons, 1972) with a precision of ± 5%.

The concentration of dissolved CH$_4$ was estimated by the modified head-space equilibrium method (Johnson et al., 1990) using a gas chromatograph (GC; HP 5890 Series II) that was equipped with a flame ionization detector (FID). The samples were placed in a water bath at a constant temperature of 25 °C, and allowed to equilibrate for at least three hours. Finally, 2 mL of the gas from the headspace was injected into the GC. The GC-FID had a 6 foot-long stainless steel column with a diameter of one-eighth of an inch, which was filled with a 60/80 mesh molecular sieve 5 A. The primary standard was NIST (National Institute of Standards and Technology) 1 ppm V CH$_4$ standard. Mixtures of CH$_4$ in N$_2$ at concentrations of 0 (Jing-Shang, Taiwan), 1

which were described in Wiesenburg and Guinasso (1979).

GC-FID. Atmospheric CH4 samples were placed at the room temperature (25 °C) for more than 2 h and 2 mL of the gas in the serum bottle was injected into the GC to analyze the CH4 concentration. The Bunsen coe

s. The means of the CH4 concentration in the water samples were obtained using the Bunsen coefficients to calculate the equilibrium solubility, which were described in Wiesenburg and Guinasso (1979).

Saturation, expressed in %, was calculated using R=(C_{eq}/C_{obs})\times 100 where C_{obs} represents the observed concentration of gas dissolved in the water, and C_{eq} is the expected equilibrium water concentration.

2.5. Saturation ratio

The concentrations of the CH4 in the water samples were obtained from the concentrations that were measured in the headspace by subtracting the in

fluence of atmospheric CH4 at the sampling sites and owing out plus the water piled up within the SCS. The SCS receives a large amount of nutrients, in the form of river discharge, atmospheric fallout, and intensive upwelling above the surface were obtained from the National Aeronautics and Space Administration (NASA) QuikSCAT satellite platform (https://winds.jpl.nasa.gov/missions/quikscat/). The average monthly wind speed was used to estimate the sea-to-air flux.

The seawater fluxes from the SCS to the wPS were calculated as described by Chen et al. (2001). The principle of conservation of water mass dictates that the water flowing into the SCS must be balanced by the water flowing out plus the water piled up within the SCS.

3. Result and discussion

3.1. Atmospheric CH4 concentration

The mean concentrations of CH4 in all atmospheric samples from the SCS and wPS were 1.81 ± 0.04 ppm V and 1.82 ± 0.04 ppmV, respectively. These values agreed closely with the mean monthly CH4 mixing ratios that were measured at stations of the Earth System Research Laboratory under the National Oceanic and Atmospheric Administration (NOAA-ESRL; http://www.esrl.noaa.gov/gmd/ccgg/ggrn.php) in situ program. The mean CH4 mixing ratios at Station BKT (Bukit Kotobabang; 0°12'S, 100°19'E), Station GMi (Marina Islands; 13°23'N, 144°39'E) and Station MID (Sand Island; 28°12'N, 144°39'E) of NOAA-ESRL between 2003 and 2007 were 1.807 ppm V (1.805–1.809 ppm V), 1.785 ppm V (1.777–1.798 ppm V) and 1.821 ppm V (1.817–1.827 ppm V), respectively.

Fluxes of CH4 across the air-water interface were estimated using

\( F = k(C_{w} - \beta C_{a}) \) where \( k \) (cm h\(^{-1}\)) is the gas exchange coefficient, \( C_{w} \), is the concentration of dissolved gas in the water (mol l\(^{-1}\)), \( \beta \) is the Bunsen solubility, and \( C_{a} \) is the atmospheric gas concentration. A positive flux indicates gas transfer from the water to the atmosphere.

The value of \( k \) is a specific function of properties of the gas, the temperature T (°C) and turbulence, and is frequently parameterized as a function of the wind speed \( u \) (m s\(^{-1}\)). Many equations are used for calculating the k/wind speed relationship (Liss and Merlivat, 1986; Wanninkhof, 1992; Erickson, 1993; Nightingale et al., 2000). As Wanninkhof (1992) has been widely used and has recently been updated (Wanninkhof, 2014), here, we calculated \( k \) by using the relationship established by Wanninkhof (2014). The updated relationship is expressed as \( k = 0.251u^{2}(Sc/660)^{-0.5} \), where \( k \) is the gas transfer velocity, \( u \) is the average squared wind speed, and \( Sc \) is the Schmidt number, which is the kinematic viscosity of water divided by the molecular diffusion coefficient of the gas. The new estimates of the Schmidt number (Wanninkhof, 2014) are provided, expanding the temperature range from 0 to 30 °C to −2 to 40 °C. Wind speeds at 10 m

above the surface were obtained from the National Aeronautics and Space Administration (NASA) QuikSCAT satellite platform (https://winds.jpl.nasa.gov/missions/quikscat/)

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The concentrations of the CH4 in the water samples were obtained from the concentrations that were measured in the headspace by subtracting the influence of atmospheric CH4 at the sampling sites and using the Bunsen coefficients to calculate the equilibrium solubility, which were described in Wiesenburg and Guinasso (1979).

Saturation, expressed in %, was calculated using $$R = \frac{C_{\text{eq}}}{C_{\text{obs}}} \times 100$$ where $$C_{\text{obs}}$$ represents the observed concentration of gas dissolved in the water, and $$C_{\text{eq}}$$ is the expected equilibrium water concentration.

2.6. Fluxes

Fluxes of CH4 across the air-water interface were estimated using $$F = k(C_{w} - \beta C_{a})$$ where $$k$$ (cm h\(^{-1}\)) is the gas exchange coefficient, $$C_{w}$$, is the concentration of dissolved gas in the water (mol l\(^{-1}\)), $$\beta$$ is the Bunsen solubility, and $$C_{a}$$ is the atmospheric gas concentration. A positive flux indicates gas transfer from the water to the atmosphere.

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Flux estimations are follows. (Chen and Huang, 1995; Chao et al., 1996). As a result, the surface water in the SCS contains more nutrients than that in the wPS (Lin et al., 2002; Chen et al., 2006). Higher nutrient contents lead to higher primary production. Although the so-called methane paradox, which concerns methanogenesis in an aerobic environment, has not yet been fully explained, substantial research indicated the possibility that biological processes are responsible for the formation of CH₄ (Traganza et al., 1979; de Angelis and Lee, 1994; Karl and Tilbrook, 1994). The riverine input brings not only nutrients into the SCS but also a high concentration of CH₄ (Chen et al., 2008a). Consequently, the surface CH₄ concentrations in the SCS were higher than those in the wPS.

The sea-to-air CH₄ flux in the wPS was 4.9 ± 4.9 μmol m⁻² d⁻¹ and in the SCS it was higher at 8.6 ± 6.4 μmol m⁻² d⁻¹. Most areas of the SCS were probably moderate to strong sources of atmospheric CH₄, with sea-to-air flux values of 9.9 ± 8.7 μmol m⁻² d⁻¹ in the northeastern SCS, 12.0 ± 7.4 μmol m⁻² d⁻¹ in the northern SCS and the Pearl River estuary region, 8.3 ± 4.1 μmol m⁻² d⁻¹ in the southwestern SCS and 4.3 ± 3.9 μmol m⁻² d⁻¹ in the eastern and southern SCS (Table 1).

In general, the observed saturations and fluxes of CH₄ in the coastal, shelf and marginal seas varied greatly within a range of 74–10500% and 0.0095–1200 μmol m⁻² d⁻¹ (Table 2 and references therein). The observed saturations and fluxes of CH₄ in the SCS and wPS in this study fell within these ranges. On a global scale, the mean surface CH₄ concentrations in the SCS were comparable to the North Atlantic, Aegean Sea, but higher than other areas such as the Arabian Sea, North Sea and the Yellow Sea (Table 2).

### Table 2

<table>
<thead>
<tr>
<th>Study area</th>
<th>Station</th>
<th>Date</th>
<th>Surface CH₄ (nmol L⁻¹)</th>
<th>Saturation (%)</th>
<th>Sea-to-air flux (μmol m⁻² d⁻¹)</th>
<th>Refs.</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>19</td>
<td>Jul.–Aug. 1995</td>
<td>13.1 ± 10.6</td>
<td>200 ± 74</td>
<td>5.02 ± 4.59</td>
<td></td>
</tr>
<tr>
<td></td>
<td>11</td>
<td>Apr.–May 1994</td>
<td>2.6–48</td>
<td>140 ± 37</td>
<td>0.03 ± 0.162</td>
<td>Jayakumar et al. (2001)</td>
</tr>
<tr>
<td>Baltic Sea</td>
<td>63</td>
<td>Feb. 1992</td>
<td>113 ± 5</td>
<td>395 ± 82</td>
<td>0.10–1200</td>
<td>Bange et al. (1994)</td>
</tr>
<tr>
<td>Bay of Bengal</td>
<td>−5</td>
<td>Jul. 1993</td>
<td>4.80 ± 0.31</td>
<td>231 ± 32</td>
<td>1.56°</td>
<td>Bange et al. (1996)</td>
</tr>
<tr>
<td>North Aegean Sea</td>
<td>−30</td>
<td>Jul. 1993</td>
<td>3.17 ± 0.45</td>
<td>149 ± 18</td>
<td>1.90°</td>
<td>Bange et al. (1996)</td>
</tr>
<tr>
<td>Yellow Sea</td>
<td>14</td>
<td>Mar.–Apr. 2001</td>
<td>3.43 ± 0.23</td>
<td>121 ± 5.4</td>
<td>0.81 ± 0.59</td>
<td>Zhang et al. (2004)</td>
</tr>
<tr>
<td></td>
<td>29</td>
<td>Apr. 2001</td>
<td>3.24 ± 0.59</td>
<td>141 ± 23.6</td>
<td>1.63 ± 1.67</td>
<td>Zhang et al. (2008)</td>
</tr>
<tr>
<td></td>
<td>21</td>
<td>Sep. 2003</td>
<td>9.49 ± 11.00</td>
<td>487 ± 555</td>
<td>20.9 ± 54.8</td>
<td>This study</td>
</tr>
<tr>
<td>Western Philippine Sea</td>
<td>68</td>
<td>Jul. 2004</td>
<td>3.0 ± 1.2</td>
<td>155 ± 62</td>
<td>4.9 ± 4.9</td>
<td>This study</td>
</tr>
<tr>
<td></td>
<td>May 2006</td>
<td>Jul. 2007</td>
<td>4.5 ± 3.6</td>
<td>230 ± 184</td>
<td>8.6 ± 6.4</td>
<td>This study</td>
</tr>
</tbody>
</table>

**3.3. Water and CH₄ exchange between the SCS and the wPS**

Since the Luzon Strait is 2200 m deep, surface and intermediate waters are exchanged unimpededly between the SCS and the wPS. Yet, waters that are deeper than 2200 m in the wPS cannot enter the SCS. Hence, waters below 2200 m in the SCS are relatively homogeneous, with hydrochemical properties similar to the water at 2200 m in the wPS (Chen et al., 2006). In the wet season, surface and intermediate waters are net-exported from the SCS to the wPS but deep water is net-imported (Chen et al., 2001). Chen et al. (2001) calculated the amount of surface water that flows out of the SCS through the Luzon Strait in the wet season as 13.9x10⁶ t·s⁻¹ and the amount of wPS water that flows into the SCS as 12.8x10⁶ t·s⁻¹. Based on our data, the CH₄ concentration above 350 m in the SCS (SCS surface water layer) is 4.8 ± 5.7 nmol L⁻¹; and in the wPS (wPS surface water layer), it is 3.9 ± 1.9 nmol L⁻¹ (Table 3). As a result, the net export of surface water from the SCS to the wPS carries 1.45x10⁶ mol d⁻¹ CH₄ in the wet season.

According to the observational data (Chen and Huang, 1996), the SCS Intermediate Water, defined by Chen and Huang (1996) as being at depths between 350 and 1350 m, is a mixture of the upwellled deep water and the surface water. The Intermediate Water flows out of the SCS at a rate of 1.8x10⁶ t·s⁻¹ (Chao et al., 1996; Chen and Huang, 1996). As the CH₄ concentration of the SCS Intermediate Water was 3.4 ± 3.0 nmol L⁻¹, the SCS Intermediate Water exported 0.53x10⁶ mol d⁻¹ CH₄ to the wPS. The deep water below 1350 m flows into the SCS year-round at a rate of 1.2x10⁶ t·s⁻¹ (Chao et al., 1996; Chen and Huang, 1996) and the CH₄ concentration was 1.5 ± 0.7 nmol L⁻¹. Consequently, the deep water from the wPS exported 0.16x10⁶ mol d⁻¹ CH₄ to the SCS. Briefly, the SCS exports 1.82x10⁶ mol d⁻¹ CH₄ to the wPS (net value) and emits 30.1x10⁶ mol d⁻¹ CH₄ to the atmosphere in the wet
3.4. Sources of CH$_4$ in the SCS and the wPS

3.4.1. CH$_4$ sources in the surface layer of the wPS

The concentrations of both CH$_4$ and chlorophyll $a$ were high in the 150 m surface layer of the wPS (Fig. 2a and b), however, no significant correlation between CH$_4$ and chlorophyll $a$ concentrations in the surface 150 m layer was identified herein. This result not only shows that the formation of CH$_4$ may not be directly associated with the photosynthetic process or phytoplankton biomass, as has been suggested in some studies (Zindler et al., 2013; Bogard et al., 2014), but it also reveals that other processes, such as the physical mixing of different water masses, may affect the distribution of CH$_4$ in the wPS.

3.4.2. CH$_4$ sources in the surface layer of the SCS

CH$_4$ maxima were observed in the subsurface layer at most stations (Fig. 2a) and the highest chlorophyll $a$ concentrations throughout the water column were found above 100 m (Fig. 2b). As for the wPS, no significant correlation between CH$_4$ and chlorophyll $a$ concentrations in the surface 100 m layer was identified.

Domain B is near the Pearl River estuary where the hydrogeochemistry is influenced by the fresh water inputs, especially in the wet season (Yin et al., 2000). Chen et al. (2008a) reported extremely high CH$_4$ concentrations from 23 to 2984 nmol L$^{-1}$ in the Pearl River, its tributaries, and estuary. Although the high CH$_4$ concentration quickly declines offshore, rivers export a large amount of CH$_4$ and organic matter into the coastal ocean. The riverine water brings large amounts of nutrients into the SCS (Dai et al., 2006, 2008; Chen et al., 2008a; Han et al., 2012), increasing the primary productivity and the chlorophyll $a$ concentration. The river inflow caused the surface water closer to shore to have lower salinity, higher chlorophyll $a$ and CH$_4$ concentrations (Fig. 3). In 2005, a very large amount of fresh water flowed into the SCS and greatly reduced the salinity of seawater near shore (Fig. 3d). This very large amount of the fresh water was brought by the Super-typhoon Haitang (category 5 on the hurricane scale) close to this area one week before the research cruise, causing intense precipitation and increased river discharge. Fig. 4 shows that the surface CH$_4$ concentration decreased with increasing salinity near the Pearl River estuary in 2005. Based on the linear equation (Fig. 4; $[CH_4]_s = -0.62$ salinity + 26, $n=19$, $r^2=0.34$), when the salinity is 0, CH$_4$ concentration of the Pearl River in the estuary should be around 26 ± 6.9 nmol L$^{-1}$, which falls within the range of 6.9–173.7 nmol L$^{-1}$ reported by Zhou et al. (2009). This result indicates that the high surface CH$_4$ concentration in Domain B was caused by fresh water inputs, as was also reported in other areas such as the North Sea (Scranton and McShane, 1991; Rehder et al., 1998; Upstill-Goddard et al., 2000).

Domain C is in the southwestern SCS and most of it is on the Sunda Shelf. The sampling stations that were closer to the Mekong River estuary exhibited higher surface CH$_4$ concentrations but lower salinity compared with other sampling stations in Domain C (Fig. 5). It revealed that although the Mekong River mouth is more than 150 km away from the sampling stations, it still affected our sampling area. According to the model simulations, the Mekong River sediment can transport to more than 250 km away from the Mekong River mouth (Xue et al., 2012). In addition, the sediment core and isotopic data showed that over the past 5500 yrs, tremendous amount of Mekong River sediment input has allowed the Mekong River Delta to prograde

### Table 3

CH$_4$ concentrations, water fluxes and CH$_4$ fluxes of various water masses in the SCS and wPS.

<table>
<thead>
<tr>
<th>Water Mass</th>
<th>SCS Avg. Concentration (nmol L$^{-1}$)</th>
<th>Water Flux ($\times 10^6$ ts$^{-1}$)</th>
<th>CH$_4$ Flux ($\times 10^6$ mol d$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Surface Water Layer</td>
<td>4.8 ± 5.7</td>
<td>-13.9 ± 1.8</td>
<td>-5.76</td>
</tr>
<tr>
<td>Intermediate Water Layer (350–1350 m)</td>
<td>3.4 ± 3.0</td>
<td>-1.8 ± 0.4</td>
<td>-0.53</td>
</tr>
<tr>
<td>Deep Water Layer (1350–2200 m)</td>
<td>1.5 ± 0.7</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Positive and negative numbers represent inflow and outflow, respectively.

![Fig. 2](image.jpg) Vertical distributions of (a) CH$_4$ (nmol L$^{-1}$) and (b) chlorophyll $a$ (µg L$^{-1}$) in the western Philippine Sea, obtained at 43 stations during three cruises from July 2004 to July 2007 and in the South China Sea, obtained at 122 stations during six cruises from August 2003 to July 2007.
more than 250 km to the southeast (Nguyen et al., 2000).

3.4.3. CH₄ sources in the intermediate and deep water layers

At most stations in the SCS and in the wPS (Fig. 2a), CH₄ concentrations generally declined with increasing depth beyond the euphotic zone and remained constant below 1,000 m. In the deep ocean, CH₄ is typically undersaturated relative to the present atmospheric mixing ratio because of lower atmospheric CH₄ concentrations at the time when deep water was formed (Rehder et al., 1999), and because of bacterial oxidation (Scranton and Brewer, 1978).

The average CH₄ concentration in the deep water layer (1350–2200 m) were 1.5 ± 0.7 nmol L⁻¹ in the wPS and 1.9 ± 1.7 nmol L⁻¹ in the SCS. However, there were some high CH₄ concentrations (4.2–14.5 nmol L⁻¹) at mid-depths (800–2300 m; Fig. 2a) in the SCS. High CH₄ concentrations were found in the bottom water in the area between latitude 20–22°N and longitude 118–120°E (in Domain A) and stations #19, #20 (Fig. 6; in Domain B). These signals may be related to the sediments on the upper continental slope, which are rich in organic matter and CH₄ may have been generated in the anoxic sediments. The same phenomenon has been observed in the East China Sea (ECS). For instance, Lin et al. (1992) found that the organic carbon concentrations in the bottom sediments increased across the shelf break, reaching their highest value in sediments at depths of 1000–1500 m in the ECS. Further, Zhang et al. (2008) found high concentrations of CH₄ (around 40 nmol L⁻¹) in bottom waters in the ECS, which may have been related to the production within and emission of CH₄ from the organic-rich sediments. Tseng et al. (2016) reported in the same sampling area, high N₂O concentrations which may have been produced in sediments on the continental slope.

Regions of CH₄ seepage were also discovered in Domain A with buildups of authigenic carbonate, named the Jiulong Methane Reef (Suess, 2005; Han et al., 2005, 2008). Previous research (Suess et al., 1985; Boetius et al., 2000; Boetius and Suess, 2004) has reported that the formation of authigenic carbonates at cold vent sites of continental margin has been associated with methane-rich fluid and the activity of chemoautotrophic communities.

Fig. 3. Surface distributions of (a) salinity (b) CH₄ (nmol L⁻¹) and (c) chlorophyll a (µg L⁻¹) in September 2003; surface distributions of (d) salinity (e) CH₄ (nmol L⁻¹) and (f) chlorophyll a (µg L⁻¹) in July 2005 near the Pearl River estuary, obtained at 41 stations during two cruises in September 2003 and July 2005.

Fig. 4. Surface CH₄ (nmol L⁻¹) versus salinity near the Pearl River estuary, obtained at 19 stations during one cruise in July 2005.
The depths of sampling stations #19 (19°30.975′N; 114°49.508′E) and #20 (19° 40.427′N; 114° 55.462′E) were 850 and 1500 m, respectively. Worth mentioning is that both sampling stations had high CH₄ concentrations throughout the water column (Fig. 7), which is not common in the deep sea domain. Zhou et al. (2009) reported high bottom CH₄ concentrations (between latitude 18°–20°N and longitude 113°–116°E) near these two sampling stations and possibly due to CH₄ seeps from seafloor sediments. Wu et al. (2009) revealed that a large amount of hydrocarbon may be accumulated in the continental shelf of the SCS. Seismic profiles of these areas also show that bottom simulating reflectors (BSR) are widespread under the sea floor, indicative of possible gassy sediments (Wu et al., 2005). Additional geochemical evidence also suggests the presence of submarine gas hydrate deposits around these area (Chen et al., 2005; Su et al., 2005). As a result, these high mid-depth CH₄ concentrations and especially the one that was 115 nmol L⁻¹ at a depth of 35 m - most likely originated from gas hydrates (Chen and Tseng, 2006) and oil gas beneath the sediment with subsequent vertical migration and advection of CH₄. The CH₄ that is released from the seafloor will migrate upward through the water column either as dissolved CH₄ or as bubbles (Zhang and Zhai, 2015).

Domain D includes the eastern and southern parts of the SCS, from the southern Luzon Strait southward along the western coast of Luzon, Palawan and Borneo. High surface CH₄ concentrations were measured along the coast of Brunei (Fig. 8a), in agreement with previous study (Rehder and Suess, 2001). The occurrence of pockmarks, indicating the recent or ancient seepage of gas or fluid from the bottom, as well as some gas seeps that have recently been active, have been identified off the shore of Brunei (Howland and Judd, 1988). Fig. 9 presents the cross-section between the eastern and the southern SCS in 2007 (ORI-
The water column of the section was well stratified in both salinity and temperature during the wet season (Fig. 9a and b). But surprisingly, from sampling station SL1–SL5 (near Brunei), the CH4 concentration did not decrease with the depth beyond euphotic zone but reaches the maxima in the bottom water (Fig. 9c). At all sampling stations near Brunei, high CH4 concentrations were found in the bottom water (Fig. 9c). At all sampling stations near Brunei, high CH4 concentrations were found in the water column close to the continental slope near the Luzon Strait. To the east of it, the water column of the section was well stratified in both salinity and temperature during the wet season (Fig. 9a and b). But surprisingly, from sampling station SL1–SL5 (near Brunei), the CH4 concentration did not decrease with the depth beyond euphotic zone but reaches the maxima in the bottom water (Fig. 9c). At all sampling stations near Brunei, high CH4 concentrations were found in the bottom water (Fig. 9c). At all sampling stations near Brunei, high CH4 concentrations were found in the water column close to Brunei may reflect seepage from the abundant fossil fuel and gas deposits (OPL, 2000).

Chen and Huang (1996) and Chen et al. (2015) reported that a mid-depth boundary between 350 m and 1350 m exists near 122°E above the continental slope near the Luzon Strait. To the east of it, the water mass belongs to the wPS, whereas to the west, it is mainly the SCS water. Fig. 10 displays the CH4 vertical distribution above 2000 m along 22°N from 121°E to 126°E. The CH4 vertical distribution in the intermediate water along 22°N reveals rather high CH4 concentrations west of 121.5°E. These high CH4 concentrations in the intermediate water were presumably exported from the SCS. However, determining whether the high CH4 concentration signals disappeared east of 121.5°E because of mixing with the wPS intermediate water, which has a lower CH4 concentration, or because of being transferred north by the Kuroshio Current, requires further investigation.

The vertical distributions of CH4 in the SCS and the wPS (Fig. 2a) show that many CH4 data were higher in the SCS than in the wPS. Those higher CH4 concentrations appeared above 2300 m in the SCS and might come from the riverine input in the surface water layer or be released from the sediment in the intermediate and deep water layers. In order to exclude the effects of sea-air exchange and the influence from the fresh water and the SCS, the correlation between CH4 concentrations and potential temperature below 100 m in the wPS east of 121.5°E was examined (Fig. 11). The CH4 concentrations correlate positively with potential temperature (CH4=0.12T+1.28, n=103, r²=0.44). As the water masses above 2200 m exchange freely between the wPS and the SCS, and waters deeper than about 2200 m originate in the 2200 m deep wPS water outside the Luzon Strait (Chen et al., 2001), the correlation of CH4 concentrations with potential temperature in the SCS should be similar with the one in the wPS. Therefore, the values above the linear regression plus 2 sigma were likely released from the sediment in the SCS (Fig. 11).

3.5. CH4 in the continental shelf of the SCS and future aspects

In the continental shelf and upper slope region (depth < 500 m) of the SCS, the average surface CH4 concentration was 5.1 ± 2.1 nmol L⁻¹ while at depths of greater than 500 m, it was 3.8 ± 1.6 nmol L⁻¹. Coastal regions receive more terrestrial nutrients and organic matter than deep seas, and the stronger coupling in such regions between sediments and surface waters favors high dissolved CH4 concentrations. During the wet season, the southwesterly monsoon induces upwelling along the coastal ocean over the continental shelf of the northern SCS (Shaw, 1992; Li, 1993; Su, 1998; Gan et al., 2009) and the Vietnamese coast (Ho et al., 2000; Kuo et al., 2000; Xie et al., 2003). During the dry season, the northeasterly monsoon causes coastal upwelling off northwest Borneo Island (Yan et al., 2015), resulting in the input of nutrients. Combining CH4 with wind speed data obtained using satellites yields estimated sea-to-air CH4 fluxes of 11.0 ± 7.4 μmol m⁻² d⁻¹ in the continental shelf and upper slope regions, and 6.1 ± 6.0 μmol m⁻² d⁻¹ in the deep-sea regions of the SCS. The continental shelf and upper slope regions cover around 60% of the SCS area and account for approximately 72% of the CH4 emission from the SCS.

Global warming has increased the stratification of the surface ocean, possibly expanding oxygen minimum zones (Stramma et al., 2008), and possibly increasing CH4 production (Naqvi et al., 2010). Additionally, coastal regions are subject to increasing terrestrial inputs of nutrients and organic matter (Chen et al., 2008b; Howarth, 2008; Conley et al., 2009), and decomposition of the settled organic matter causes hypoxia when the oxygen at the bottom is insufficiently replenished. Consequently, an increasing number of coastal ecosystems are reported to exhibit hypoxia (Diaz and Rosenberg, 2008; Lui et al., 2014). Changes in nutrient and carbon exported from rivers have been demonstrated to affect exchange of biogenic trace gases, such as CO2 (Gypens et al., 2009) and DMS (Gypens and Borges, 2014), with the atmosphere. As increasing amounts of nutrients and organic matter have been exported from the land and the hypoxic areas of coastal oceans have been increasing, more CH4 may have accumulated in the coastal regions and more CH4 emission may occur in the future.

4. Conclusions

In both the SCS and the wPS, CH4 in the surface water was oversaturated with respect to the atmospheric equilibrium, and CH4 concentrations generally declined with increasing depth below the euphotic zone, but remained constant below 1,000 m. In the SCS, some
high CH$_4$ concentrations at mid-depths were observed, and could most likely be attributed to the release of CH$_4$ from sediments or seepage from gas hydrates or gassy sediment.

The SCS influences the surface and intermediate waters in the wPS. This study has established that the SCS emits CH$_4$ to the atmosphere and also exports CH$_4$ to the wPS. The SCS emits more CH$_4$ to the atmosphere than transports to the wPS in the wet season.

The continental shelf and upper slope are responsible for approximately 72% of the emissions of CH$_4$ from the SCS. Since marginal seas represent a large percentage of coastal regions and are strongly affected by anthropogenic activities, the situation in marginal seas may have become more severe comparing with other sea areas. Importantly, as terrestrial input increases and coastal areas under hypoxia expand, more CH$_4$ may be emitted.

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