

27th September 2016

# Occurrence of greenhouse gases (CO<sub>2</sub>, N<sub>2</sub>O and CH<sub>4</sub>) in groundwater of the Walloon Region (Belgium)

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# OUTLINE

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## 1. Introduction

Objective

## 2. Methodology

2.1 Study area

2.2 Sampling collection

## 3. Results

## 4. Assessment of the occurrence of GHGs

## 5. Conclusions



25-29<sup>th</sup>  
September 2016

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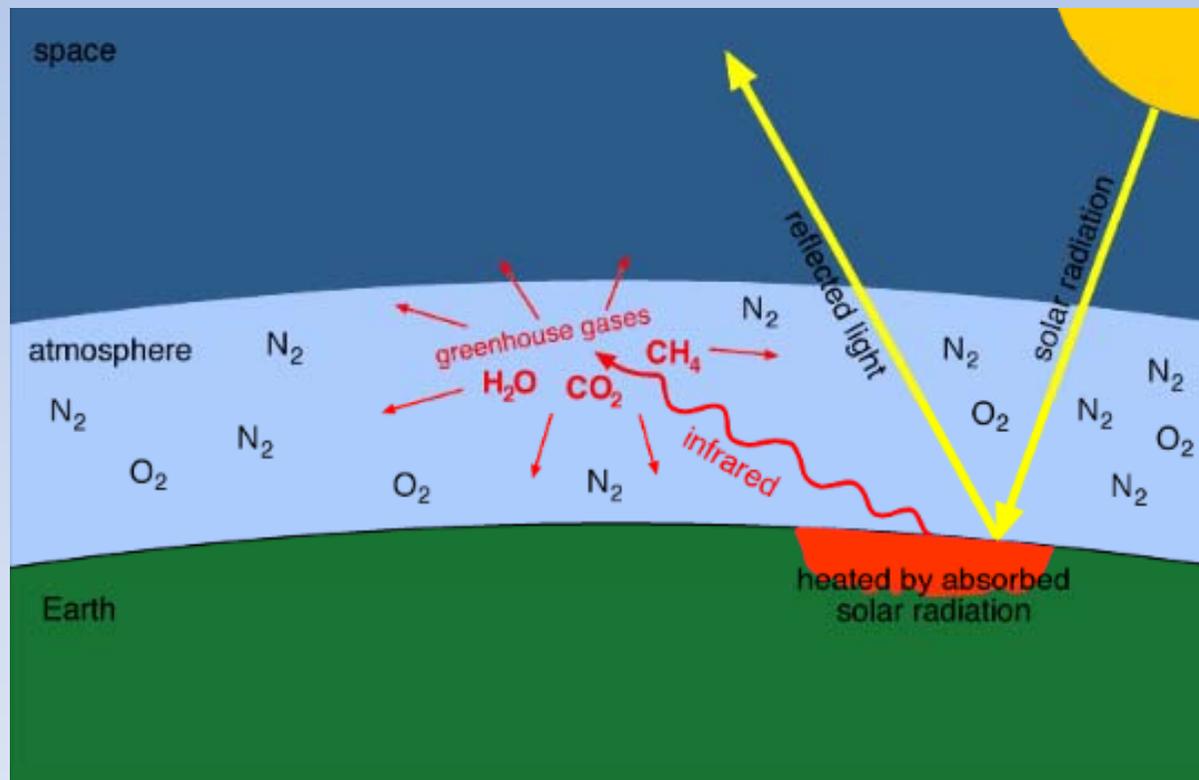
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# 1. Introduction

## Importance of GHGs

- ✓ GHGs are essential for life
- ✓ Without GHGs temperatures on Earth surface would fall from 15°C to -18 °C



Source: <http://www.ehso.com/climatechange/climatechange-causes-greenhouse-effect.php>

# 1. Introduction

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## WHAT IS THE PROBLEM?

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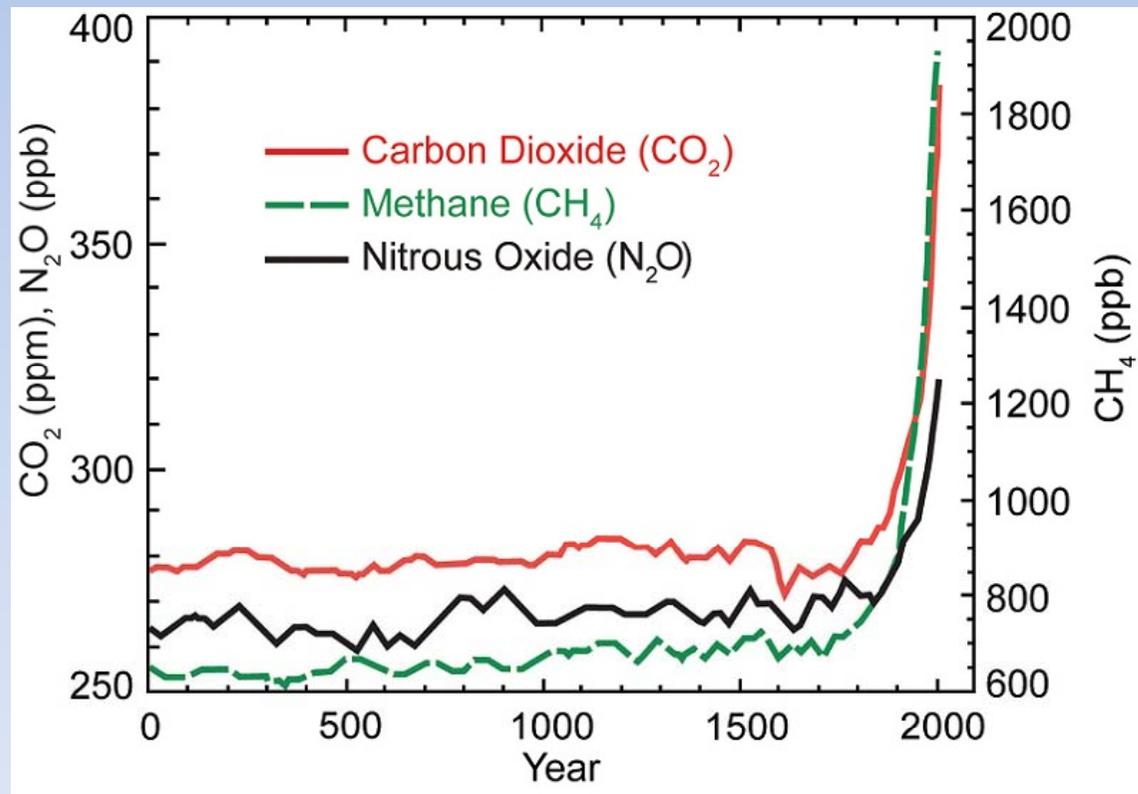


# 1. Introduction

## Importance of GHGs

- ✓ GHGs are essential for life
- ✓ Without GHGs temperatures on Earth surface would fall from 15°C to -18 °C

## WHAT IS THE PROBLEM?



Source: U.S. National Assessment (2014)



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# 1. Introduction

## Source of GHGs

### CARBON DIOXIDE (CO<sub>2</sub>)

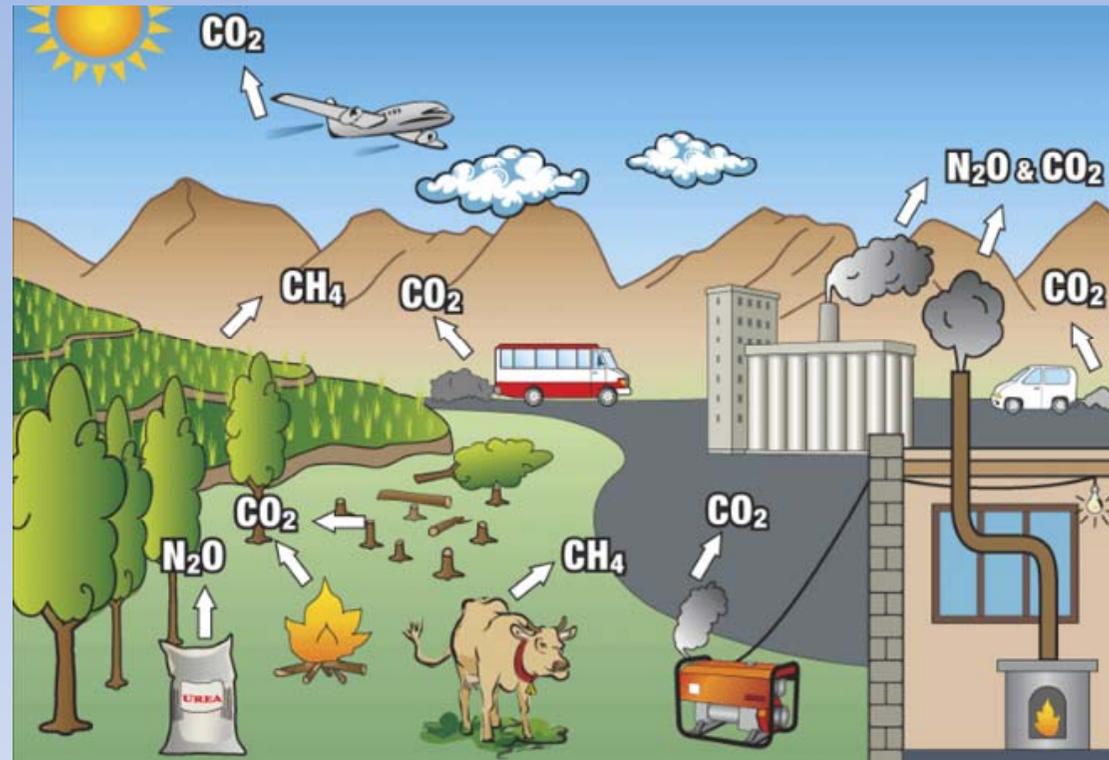
- ✓ Fossil fuel burning
- ✓ Changes in land use
- ✓ Industrial activities

### NITROUS OXIDE ( N<sub>2</sub>O)

- ✓ Agricultural activities
- ✓ Fossil fuel combustion and industrial processes
- ✓ Natural processes (i.e soils)

### METHANE (CH<sub>4</sub>)

- ✓ Fossil fuel production, distribution and use
- ✓ Livestock farming
- ✓ Landfills and waste
- ✓ Wetlands

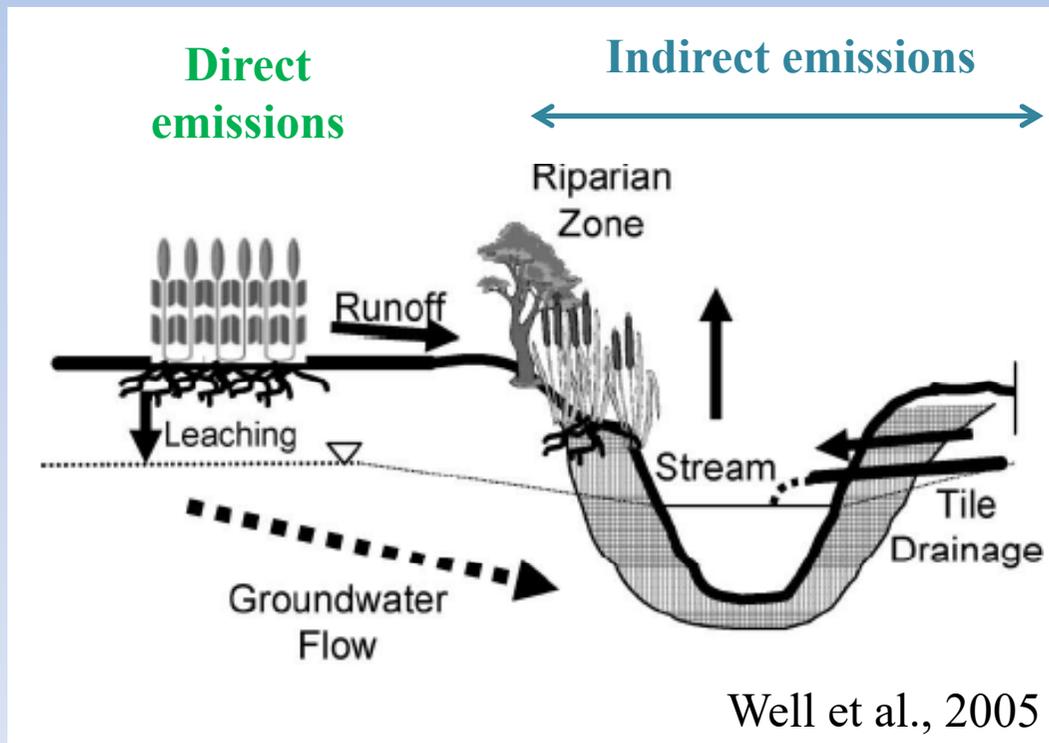


# 1. Introduction

## Groundwater as a source GHGs

Groundwater has been proposed as a potential indirect source of GHGs to the atmosphere.

**Agricultural areas** (Anderson et al., 2014; Jahangir et al., 2012; Minamikawa et al., 2011)



### Some examples...

Jahangir et al. (2012) → **Groundwater indirect N<sub>2</sub>O emissions via denitrification represented the 3–11% of total N<sub>2</sub>O emissions.**

Anderson et al. (2014) → **20 % of total N loss from a riparian area may be attributed to N<sub>2</sub>O emissions from shallow groundwater via denitrification.**

# 1. Introduction

## Groundwater as a source GHGs

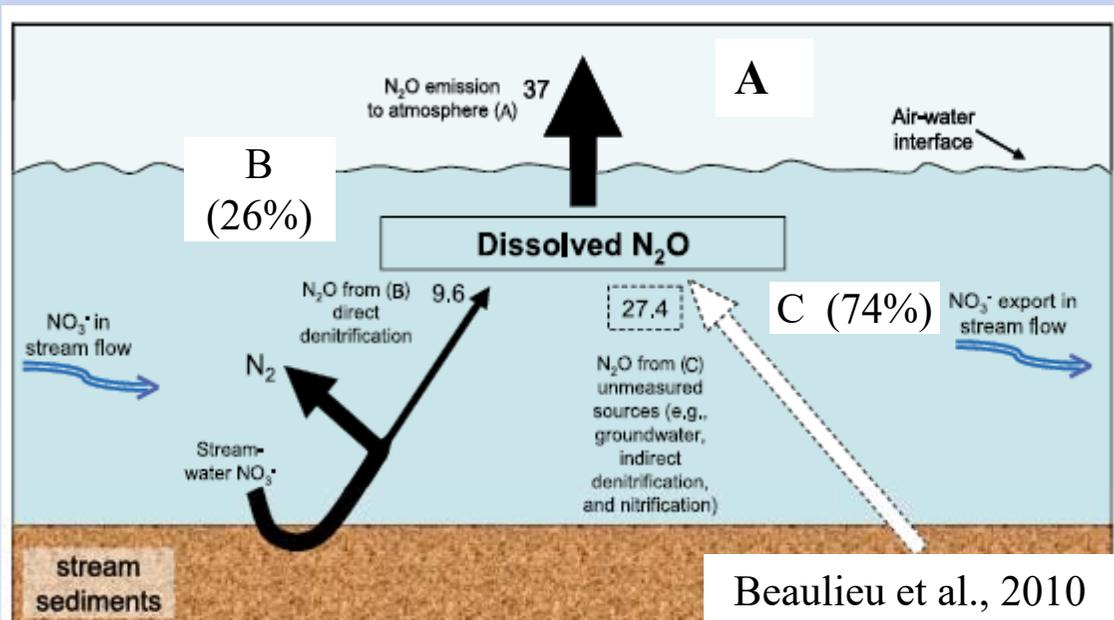
Groundwater has been proposed as a potential indirect source of GHGs to the atmosphere.

**Riverine areas** (Beaulieu et al., 2010, Hotchkiss et al., 2015 )

**10% of the global anthropogenic N<sub>2</sub>O emission rate** was due to microbial transformations **in river networks**

Average N<sub>2</sub>O fluxes estimated

$$A=B+C$$



(A) N<sub>2</sub>O produced in the stream temporarily resides in a pool of dissolved N<sub>2</sub>O before being emitted to the atmosphere.

(B) In-situ N<sub>2</sub>O produced by denitrification in stream

(C) Other **unmeasured sources** N<sub>2</sub>O sources:

- ✓ **Inputs supersaturated GW**
- ✓ **indirect Nitrif/Denitrif from soils**

# 1. Introduction

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## Aims of the research

Investigate the occurrence GHGs in groundwater of the Walloon  
Region (Belgium)

- ✓ Identification of the hydrogeological contexts and in situ conditions
- ✓ Identification of the geochemical processes

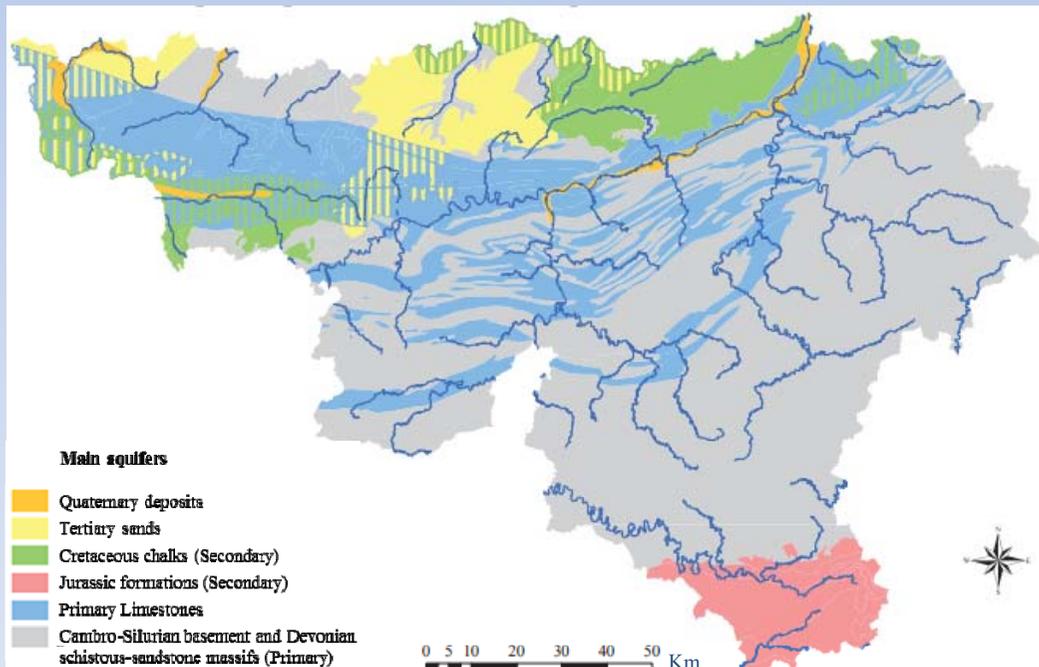
## 2. Methodology

### Walloon Region (Belgium)

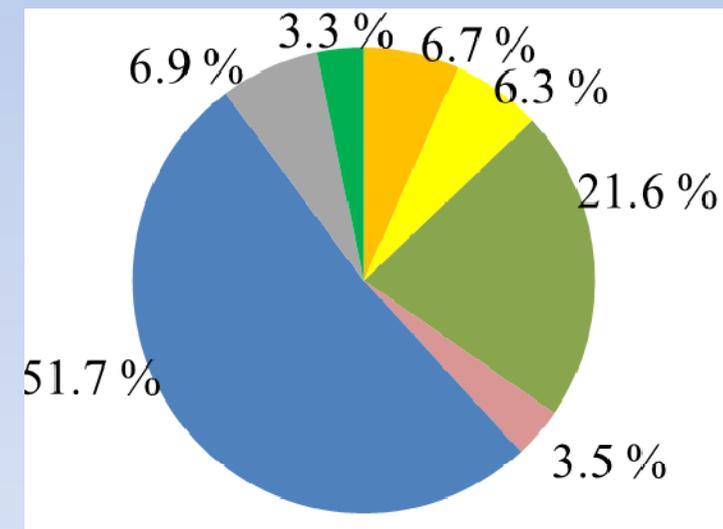


- Extension: 16844 Km<sup>2</sup>
- Land use: agricultural (51,8%) > forests (29,4%) > urban (14,3%)
- Groundwater represents the 78 % of water supply (297,5 million m<sup>3</sup>)

#### Main aquifers



#### Groundwater abstraction (% , 2012)



## 2. Methodology

### Sampling campaigns

#### GHGs (C1-C3)

- ✓ CO<sub>2</sub>, N<sub>2</sub>O, CH<sub>4</sub>

#### General analysis

- ✓ Minor and major elements (C1-C3)
- ✓ Metals (Fe/Mn) (C3)

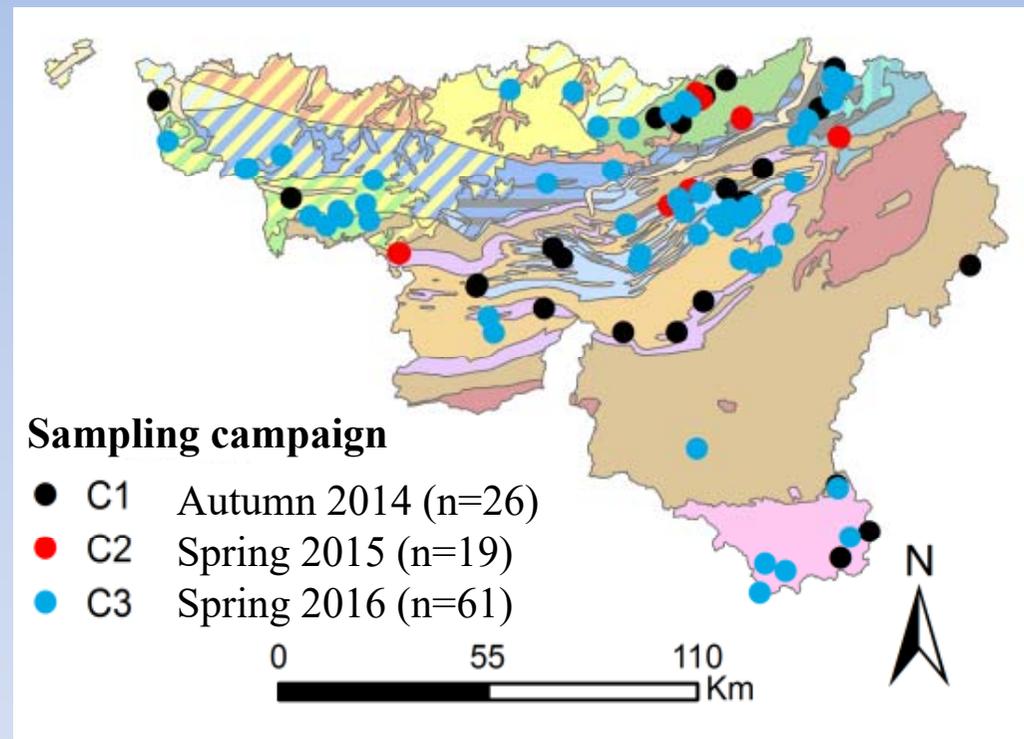
#### Environmental isotopes (C3)

- ✓ <sup>34</sup>S and <sup>18</sup>O from sulphate
- ✓ <sup>15</sup>N and <sup>18</sup>O from nitrate
- ✓ <sup>18</sup>O and D from water

#### In situ parameters (C1-C3)

- ✓ O<sub>2</sub>/EC/PH/Temp

### Spatial distribution of the sampling points

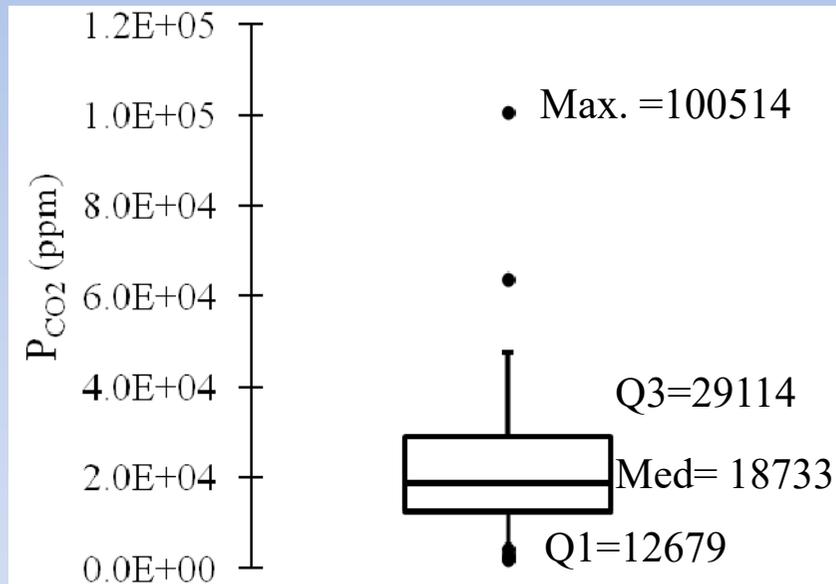


# 3. Results

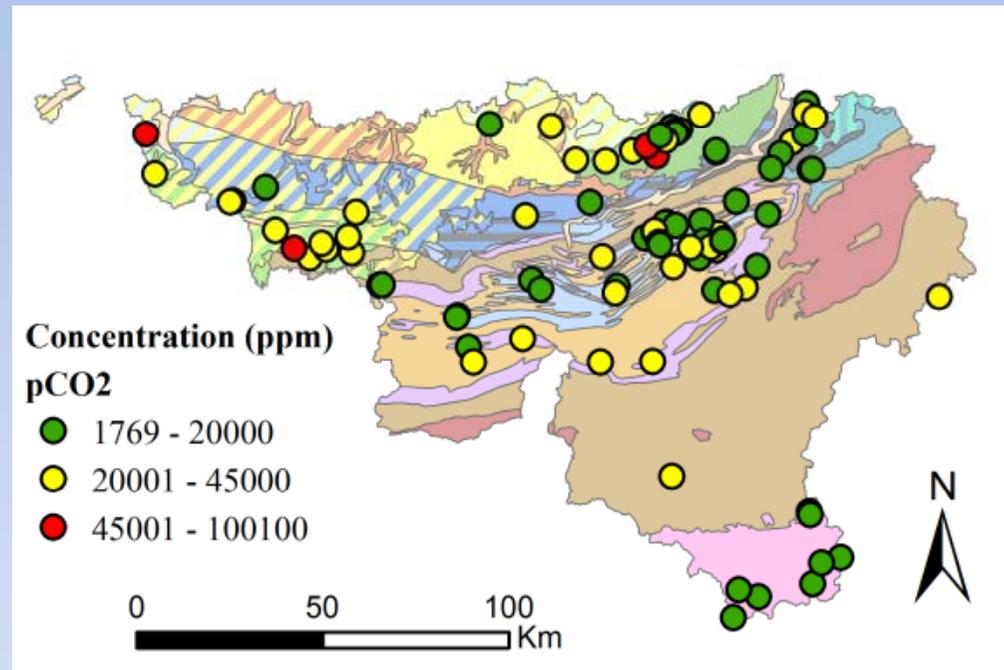
## Concentrations of Carbon dioxide (pCO<sub>2</sub>)

Range → 1769-100514 ppm

Average → 22003 ppm



## Spatial distribution of the sampling points

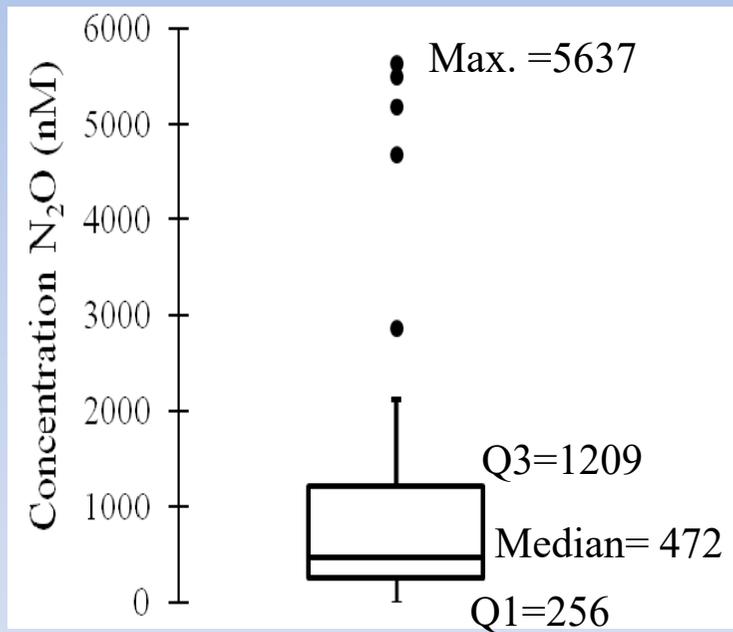


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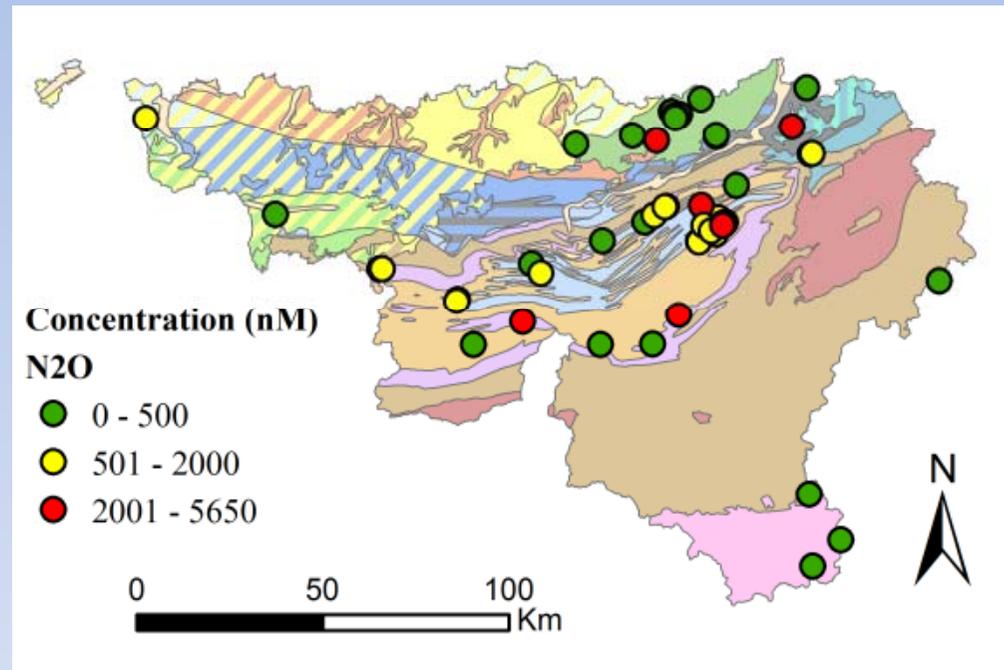
## Concentrations of Nitrous oxide (N<sub>2</sub>O)

Range → 1-5637 nM

Average → 996 nM



## Spatial distribution of the sampling points

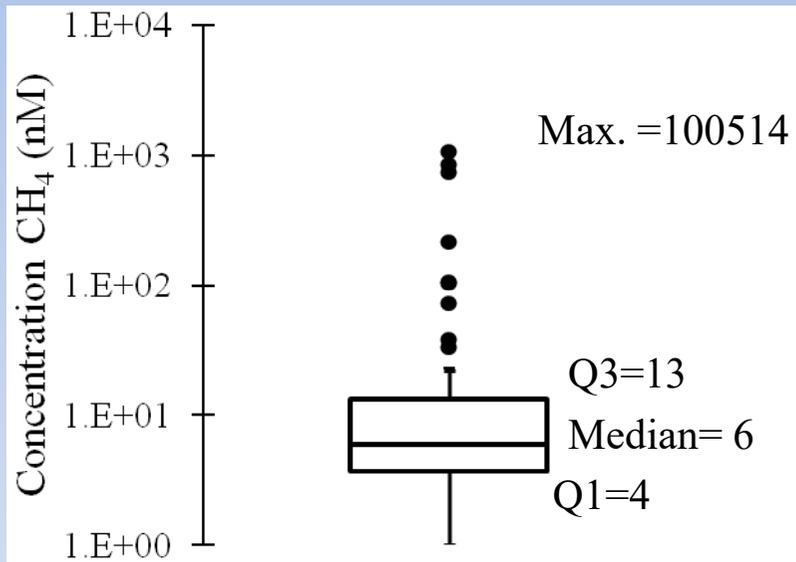


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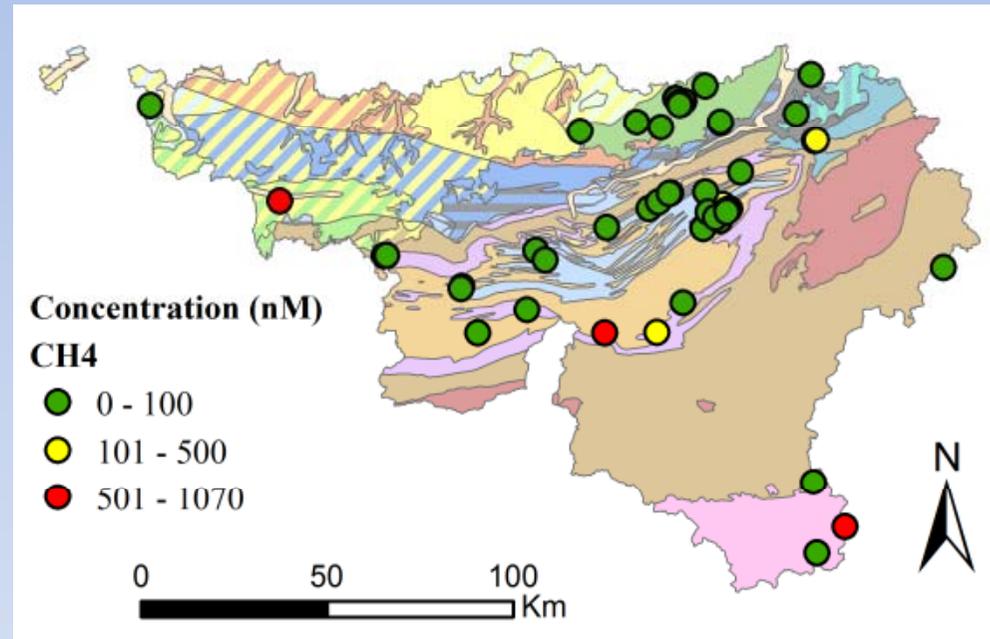
## Concentrations of Methane (CH<sub>4</sub>)

Range → 0-1064 nM

Average → 64 nM



## Spatial distribution of the sampling points



## 4. Discussion

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### Occurrence of N<sub>2</sub>O in groundwater

Denitrification:  $\text{NO}_3^- \rightarrow \text{NO}_2^- \rightarrow \text{NO} \rightarrow \text{N}_2\text{O} \rightarrow \text{N}_2$

$\text{N}_2\text{O}$

↑

Nitrification:  $\text{NH}_4^+ \rightarrow \text{NO}_2^- \rightarrow \text{NO}_3^-$



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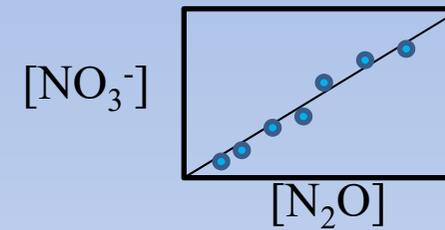
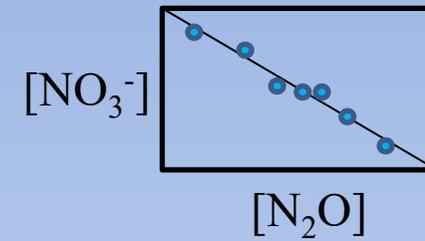
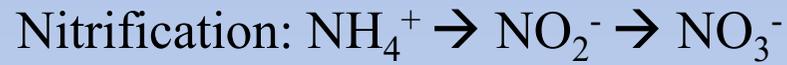
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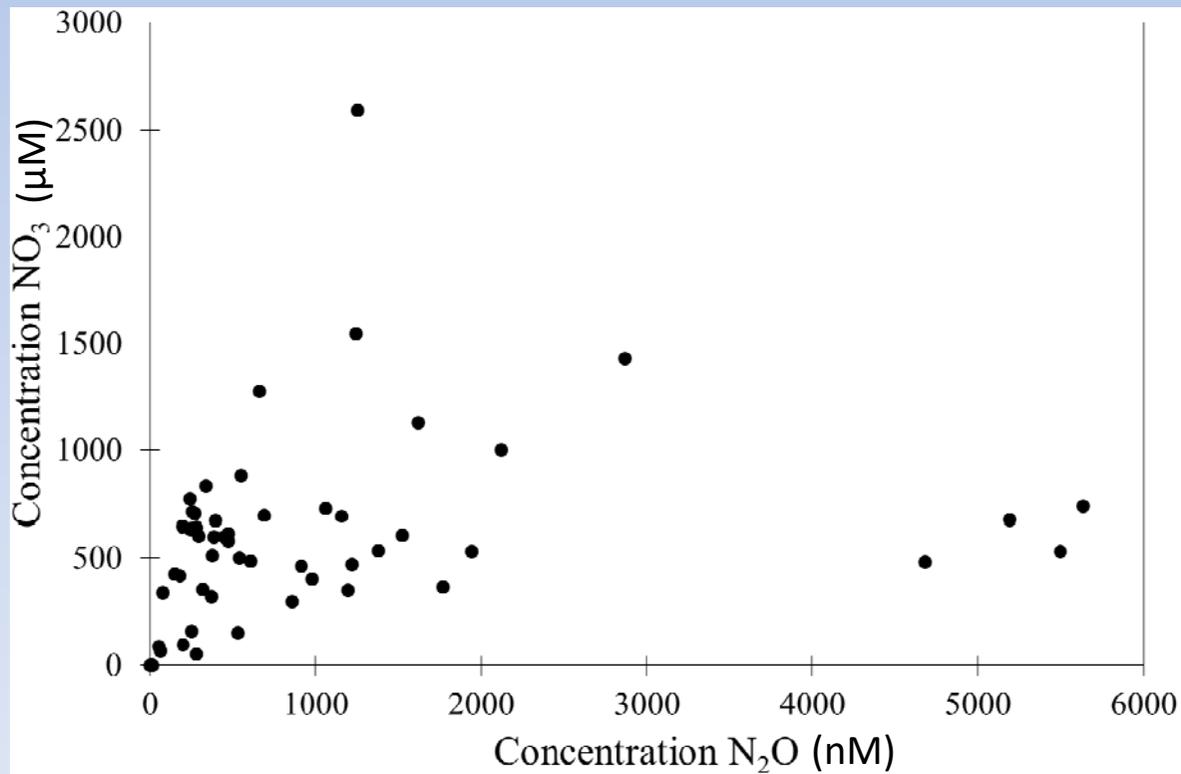
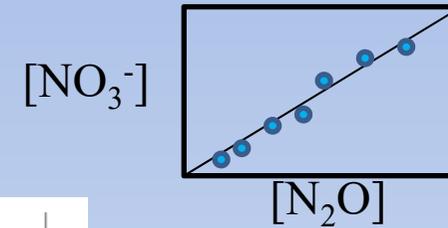
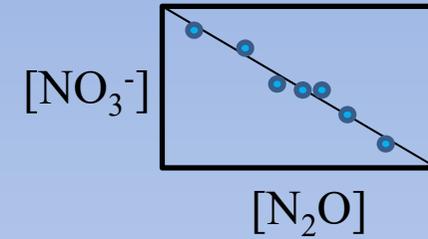
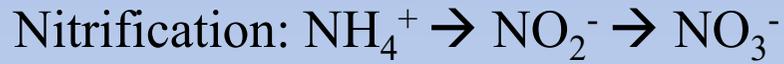
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## Occurrence of N<sub>2</sub>O in groundwater



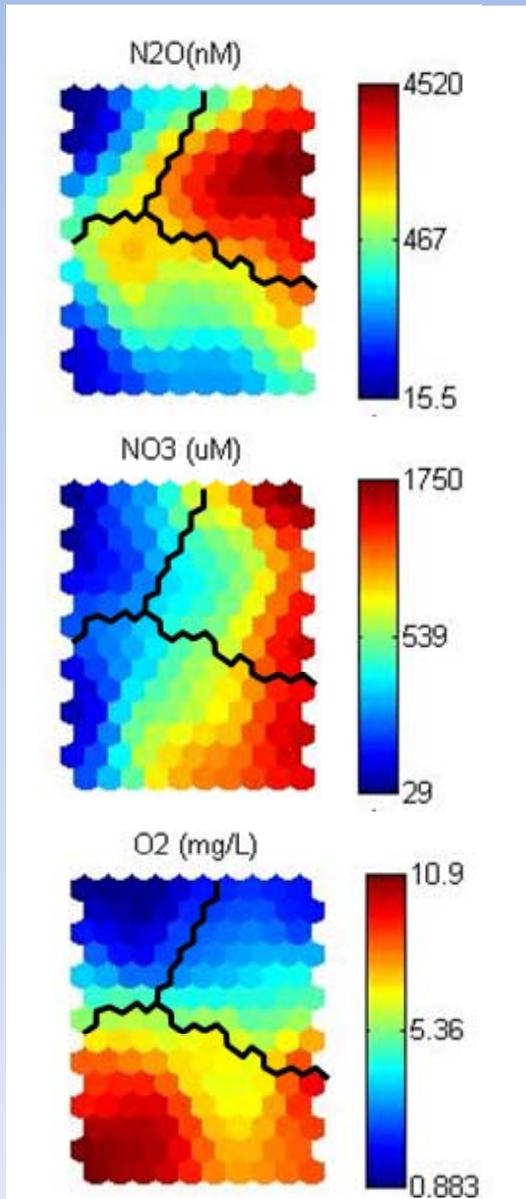
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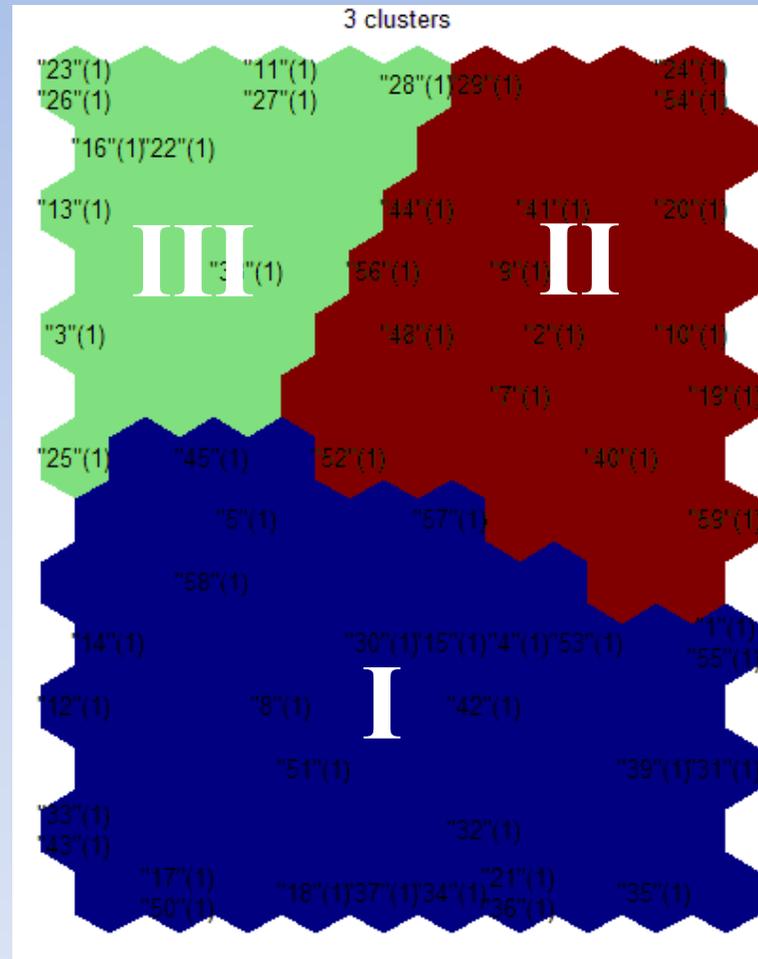


# 4. Discussion

## Occurrence of N<sub>2</sub>O in groundwater

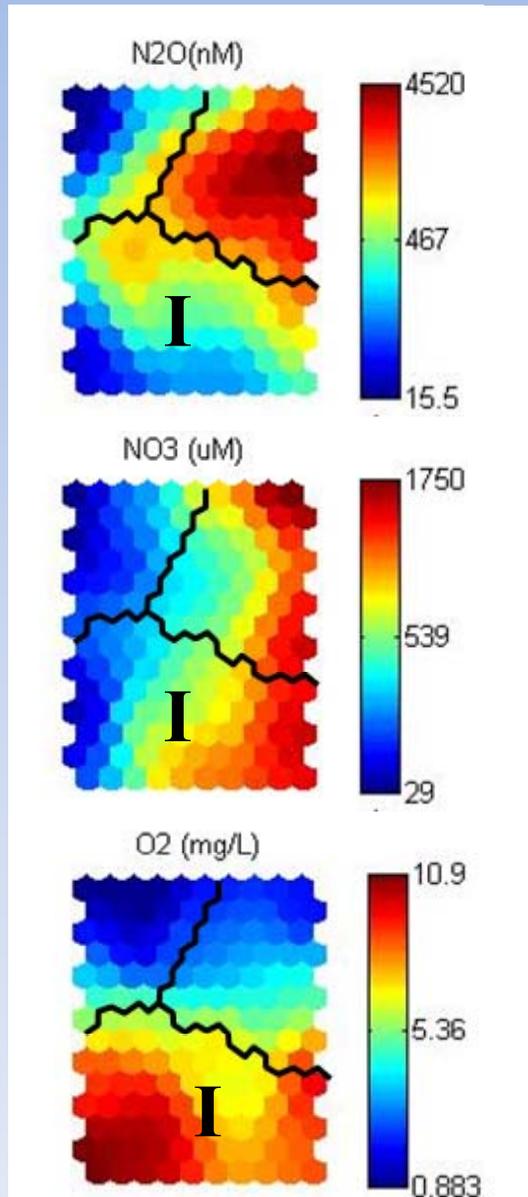


## Self-Organizing Maps (SOMs)



## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



#### Average concentrations

Group	N <sub>2</sub> O (nM)	O <sub>2</sub> (mg/L)	NO <sub>3</sub> (μM)
I	516.55	7.51	564.37
II	2377.31	3.64	831.28
III	188.15	1.51	215.04

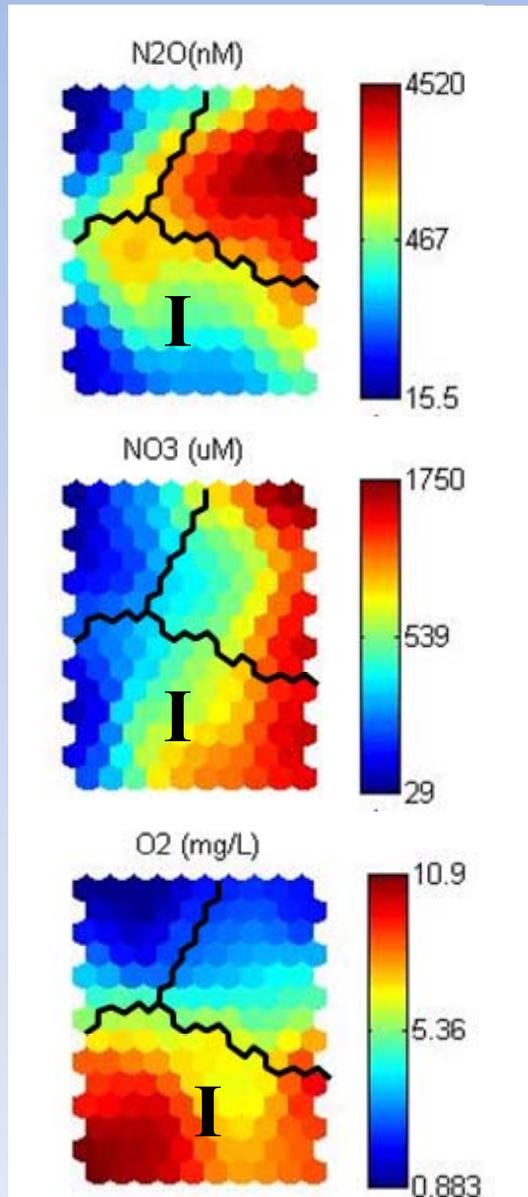
#### Group I

High concentrations of O<sub>2</sub>

Medium concentrations of N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>

## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



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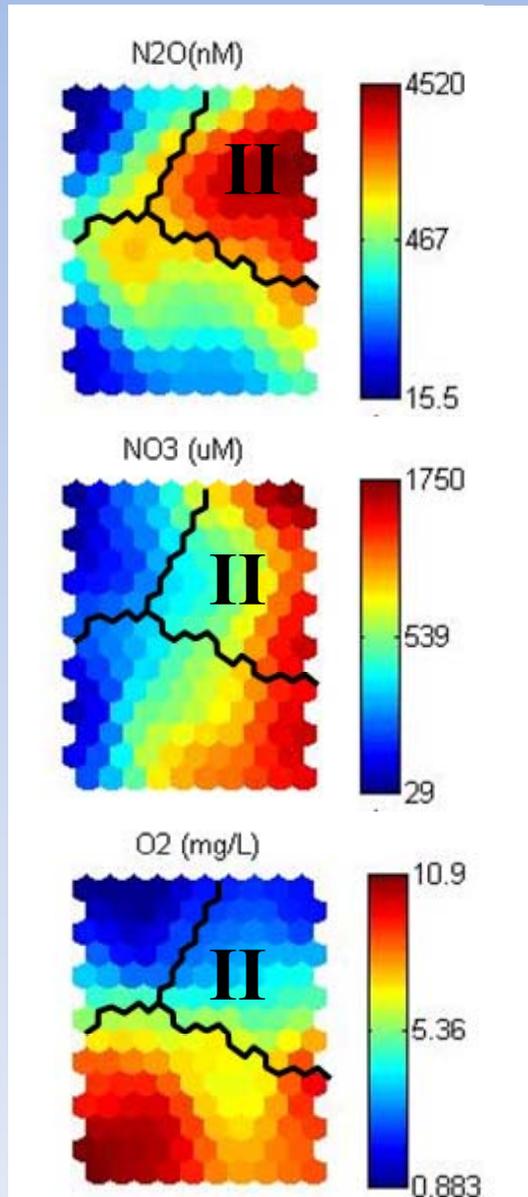
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Medium concentrations of N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>



NITRIFICATION

## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



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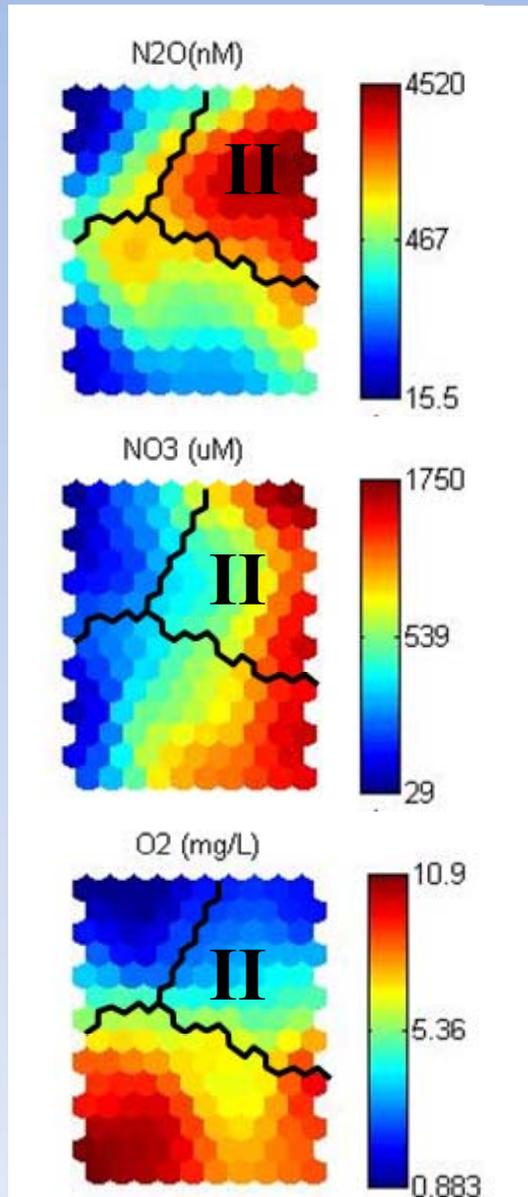
#### Group II

Moderate O<sub>2</sub>

High concentrations of N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>

## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



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High concentrations of N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>

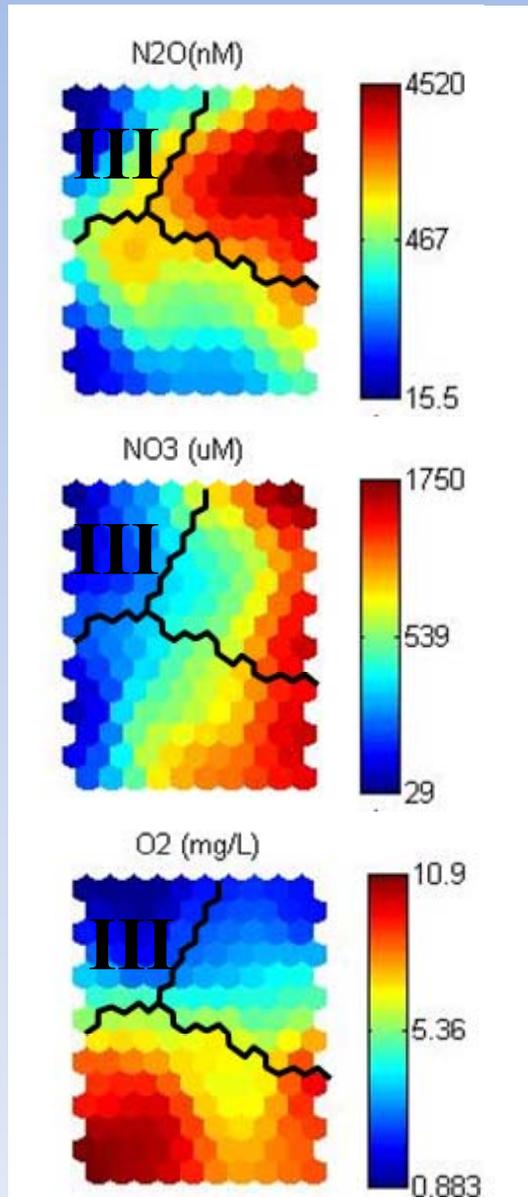


DENITRIFICATION



## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



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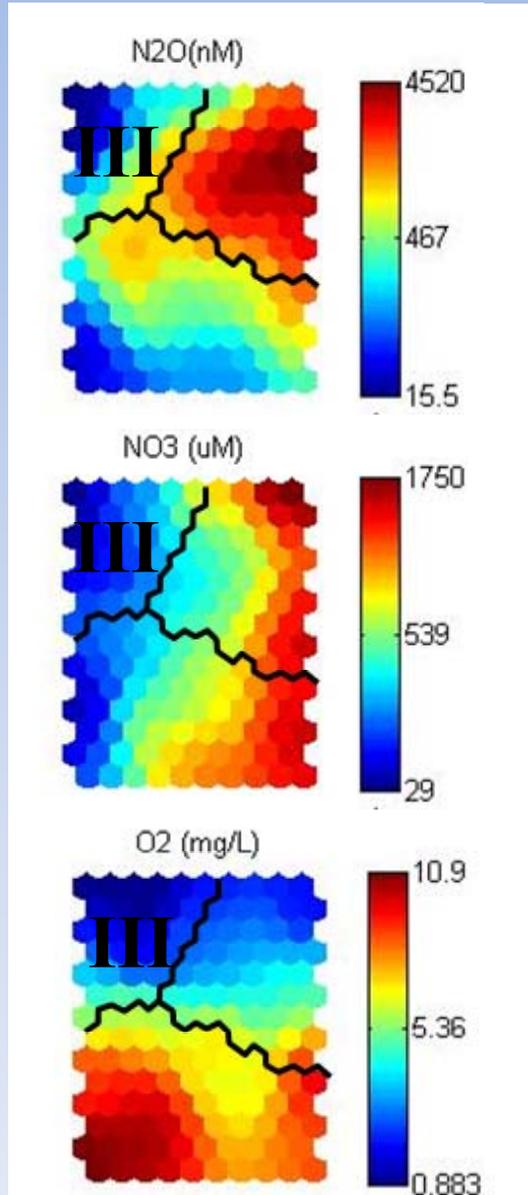
#### Group III

Low concentrations of O<sub>2</sub>

Low concentrations of N<sub>2</sub>O and NO<sub>3</sub><sup>-</sup>

## 4. Discussion

### Occurrence of N<sub>2</sub>O in groundwater



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DENITRIFICATION



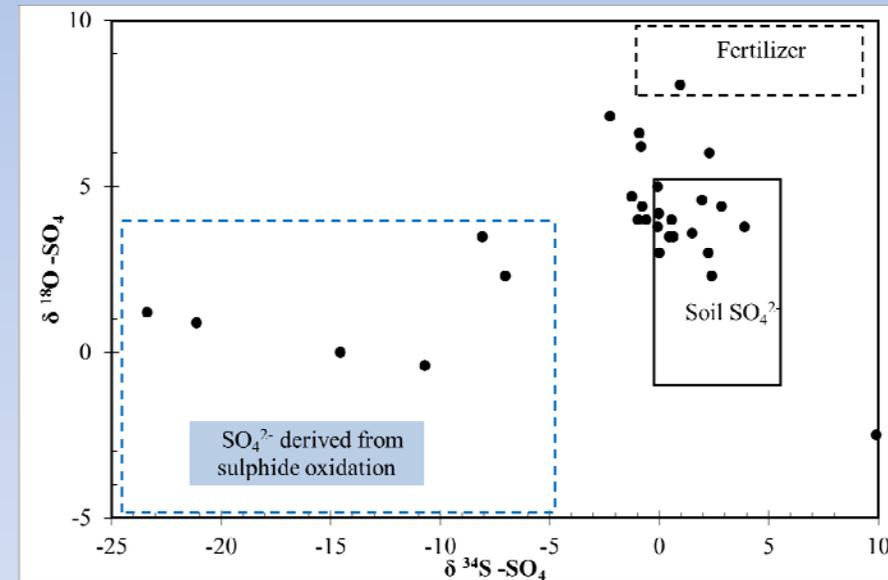
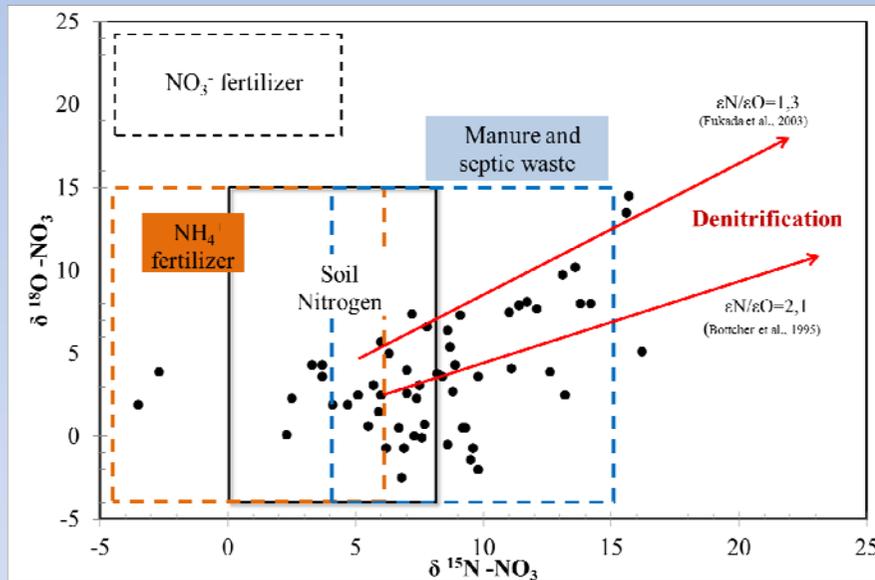
# 4. Discussion

## Occurrence of N<sub>2</sub>O in groundwater

Environmental isotopes coupled with hydrochemistry data

Identify origin of nitrogen

Processes that produce/consume N<sub>2</sub>O



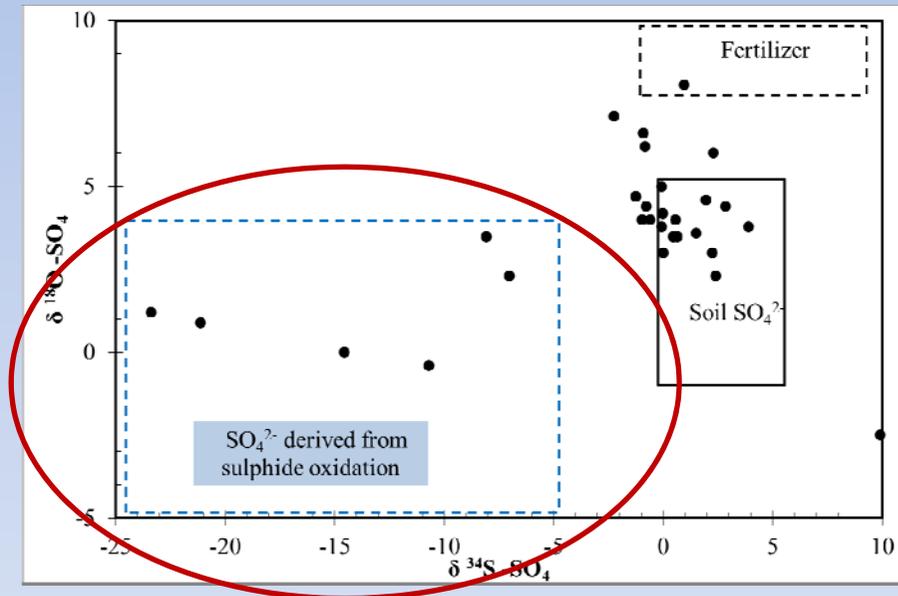
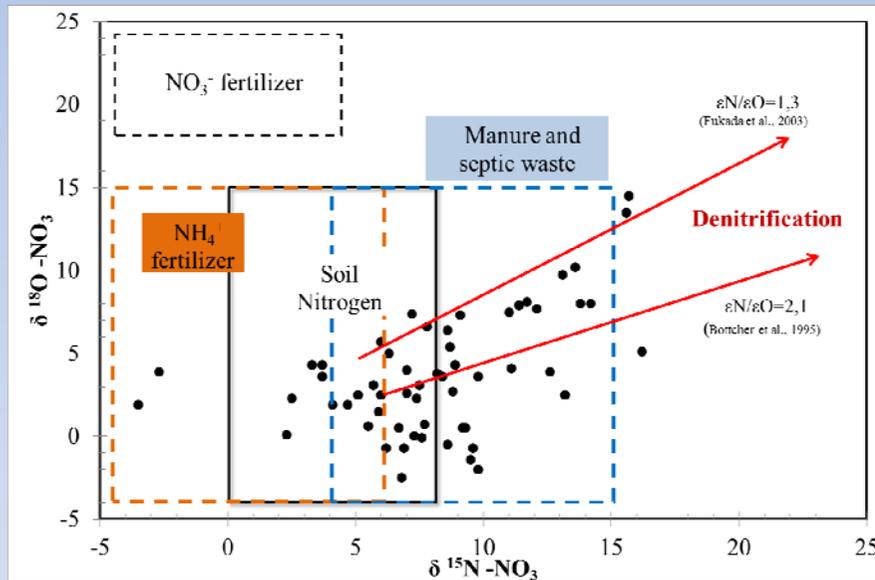
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**Autotrophic denitrification** (Pyrite as electron donor):



## 5. Conclusions

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1. Groundwater of Walloon Region is oversaturated in  $\text{CO}_2$  and  $\text{N}_2\text{O}$  relative to the atmospheric concentrations.
2. Preliminary results show that  $\text{N}_2\text{O}$  is produced by nitrification and denitrification and also consumed by denitrification.
3. Most favourable conditions for the accumulation of  $\text{N}_2\text{O}$  in groundwater seems to occur when  $\text{NO}_3^-$  is available and at medium oxygen concentrations.

### Future work :

- Integrate all the data available (environmental isotopes)
- Investigate the occurrence of  $\text{CO}_2$  and  $\text{CH}_4$

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# Thank you for your attention

**Acknowledgements.** A. Jurado gratefully acknowledge the financial support from the University of Liège and the EU through the Marie Curie BeIPD-COFUND postdoctoral fellowship programme (2015-2017 from FP7-MSCA-COFUND, 600405).



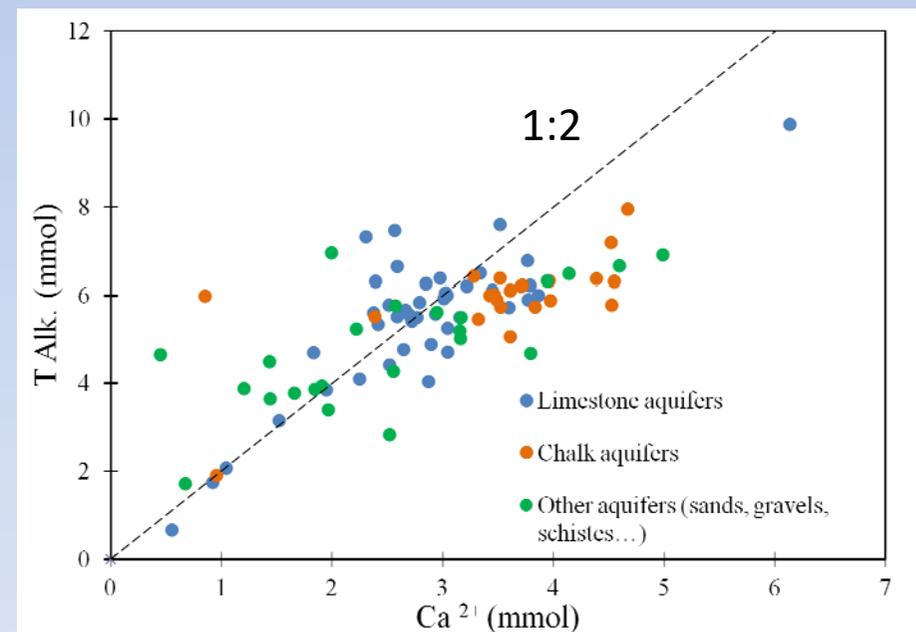
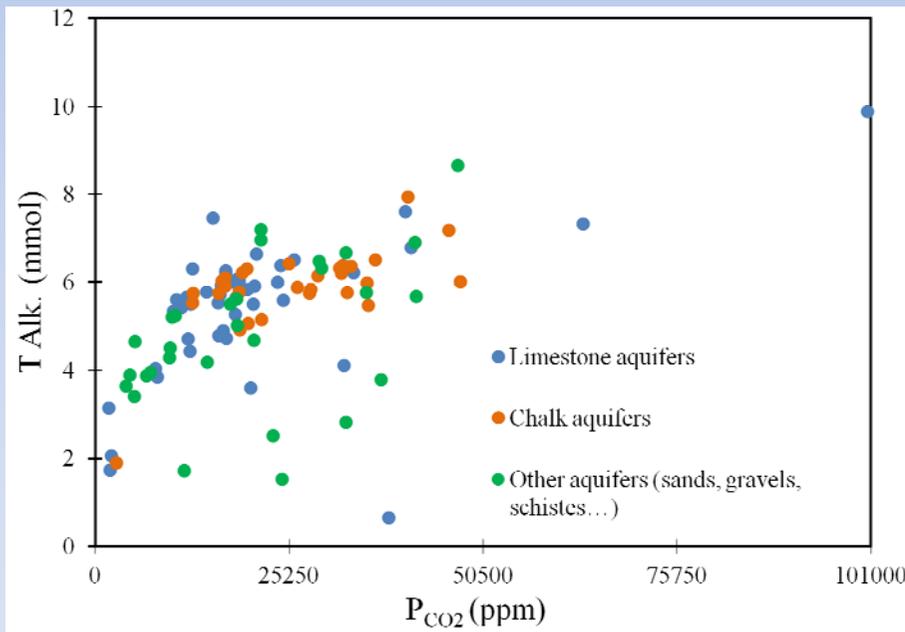
# 4. Discussion

## Occurrence of CO<sub>2</sub> in groundwater

High PCO<sub>2</sub> in groundwater

Water percolating through the soil becomes enriched in CO<sub>2</sub>

High PCO<sub>2</sub> can contribute to the dissolution of carbonate formations



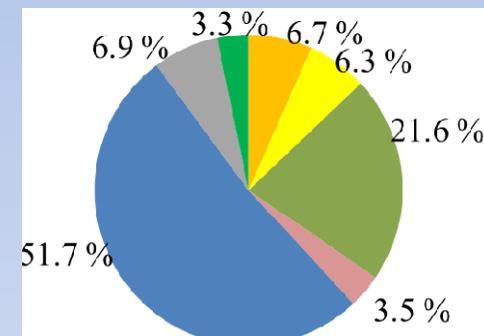
## 2. Methodology

### Walloon Region (Belgium)

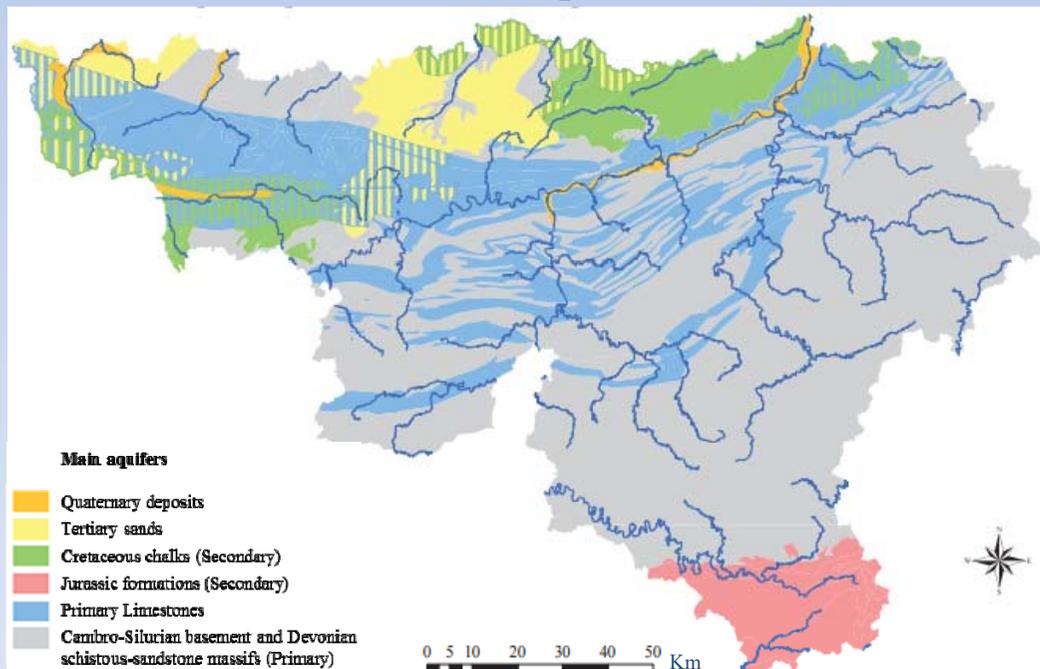


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- Groundwater represents the 78 % of water supply (297,5 million m<sup>3</sup>)
- Land use: agricultural (51,8%) > forests (29,4%) > urban (14,3%)

Groundwater abstraction (% , 2012)



Main aquifers



Land use

