

# Exploring The Scope of Transition Metal Complexes Bearing Imidazol(in)ium-2-dithiocarboxylate Ligands

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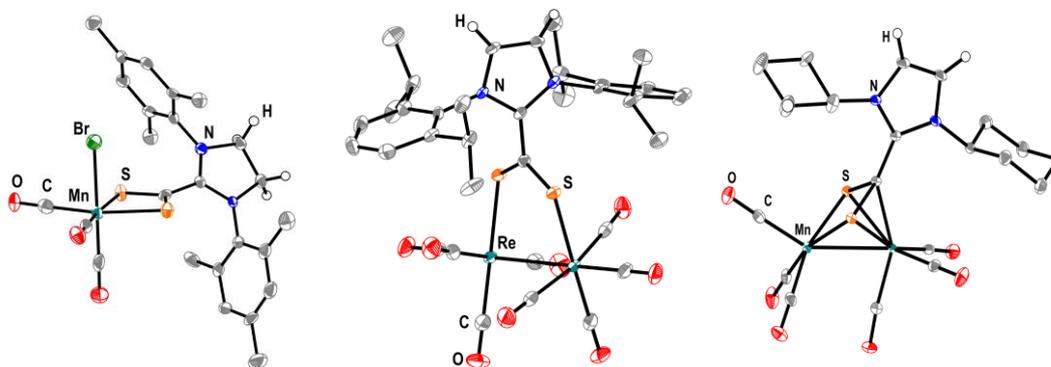
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Despite the widespread use of N-heterocyclic carbenes (NHCs) as neutral, two-electron donors in organic synthesis and in organometallic chemistry, only a modest number of reports have looked at their potential to generate other ligand systems. Yet, the facile reaction of NHCs with heteroallenes X=C=Y such as COS, CS<sub>2</sub>, or RNCS to afford the corresponding betaines in high yields and purities provides a convenient starting point to generate new heteroatom-based ligands with tailored binding modes.<sup>1</sup>

In particular, the reaction of preformed or *in situ* generated NHCs with carbon disulfide affords stable, crystalline adducts that can act as  $\kappa^2$ -S,S' chelating or bridging ligands toward various transition metals. Thus, in 2009 our group reported the formation of cationic ruthenium - arene complexes bearing imidazol(in)ium-2-dithiocarboxylate ligands.<sup>2</sup> In collaboration with James Wilton-Ely at Imperial College, the coordination chemistry of NHC·CS<sub>2</sub> zwitterions was further extended to various other complexes of ruthenium,<sup>3,4</sup> osmium,<sup>4</sup> palladium,<sup>5</sup> and gold.<sup>6</sup> We also demonstrated that NHC·CS<sub>2</sub> betaines were suitable ligands for the stabilization of gold nanoparticles.<sup>6</sup>

The potential of NHC·CS<sub>2</sub> zwitterions to generate new molecular architectures prompted us to further investigate the preparation of a large family of mono- and polynuclear complexes based on transition metals from groups 6 to 8 bearing these ligands. In this presentation, we shall discuss the coordination mode of NHC·CS<sub>2</sub> zwitterions in a series of new transition metal carbonyl complexes, on the basis of X-ray diffraction analysis and spectroscopic characterization techniques. We will also discuss their potential applications in catalysis.



**Figure 1.** ORTEP representations of [MnBr(CO)<sub>3</sub>(S<sub>2</sub>C·SIMes)] (left), [Re<sub>2</sub>(CO)<sub>8</sub>·(S<sub>2</sub>C·IDip)] (center) and [Mn<sub>2</sub>(CO)<sub>6</sub>(S<sub>2</sub>C·ICy)] (right).

## References

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