

# Occurrence & controlling factors of inorganic (and organic) compounds in groundwater of urban and industrial areas

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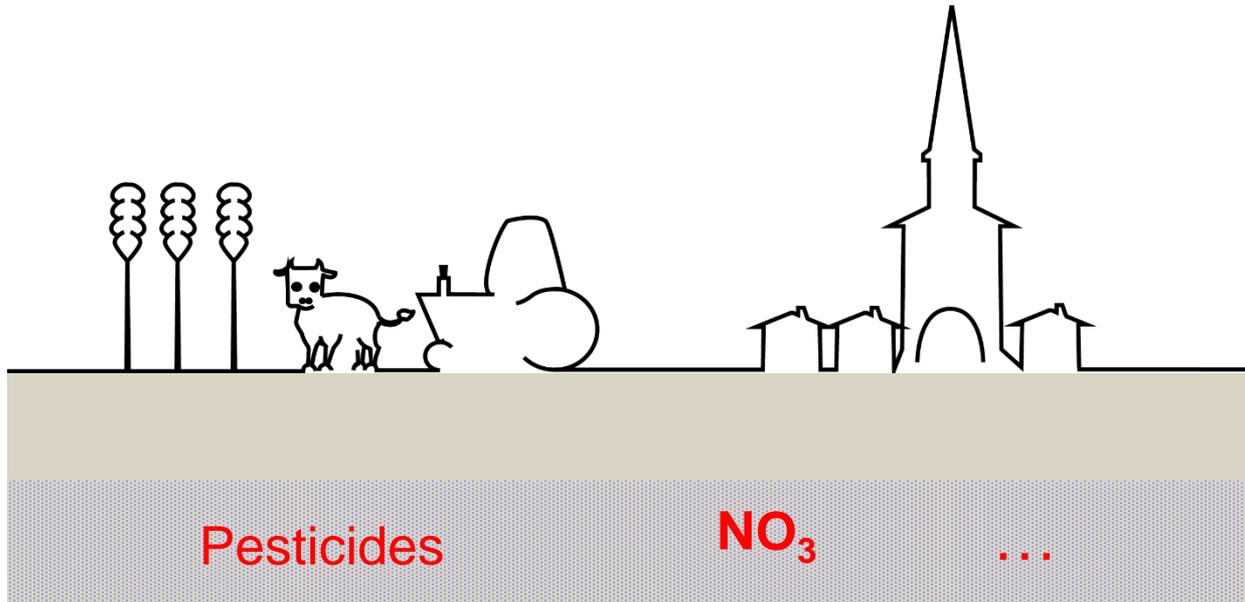
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# The WFD and the diffuse pollution of groundwater

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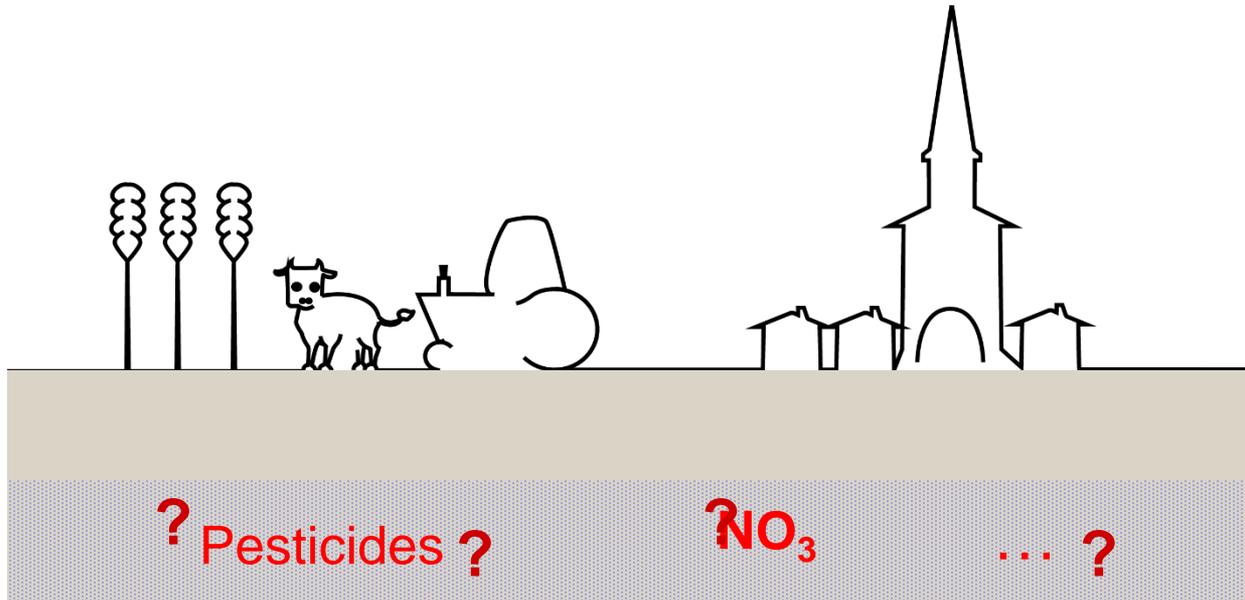
In rural zones, diffuse pollution of groundwater is relatively well established and studied by hydrogeologists and decision makers...



# The WFD and the diffuse pollution of groundwater

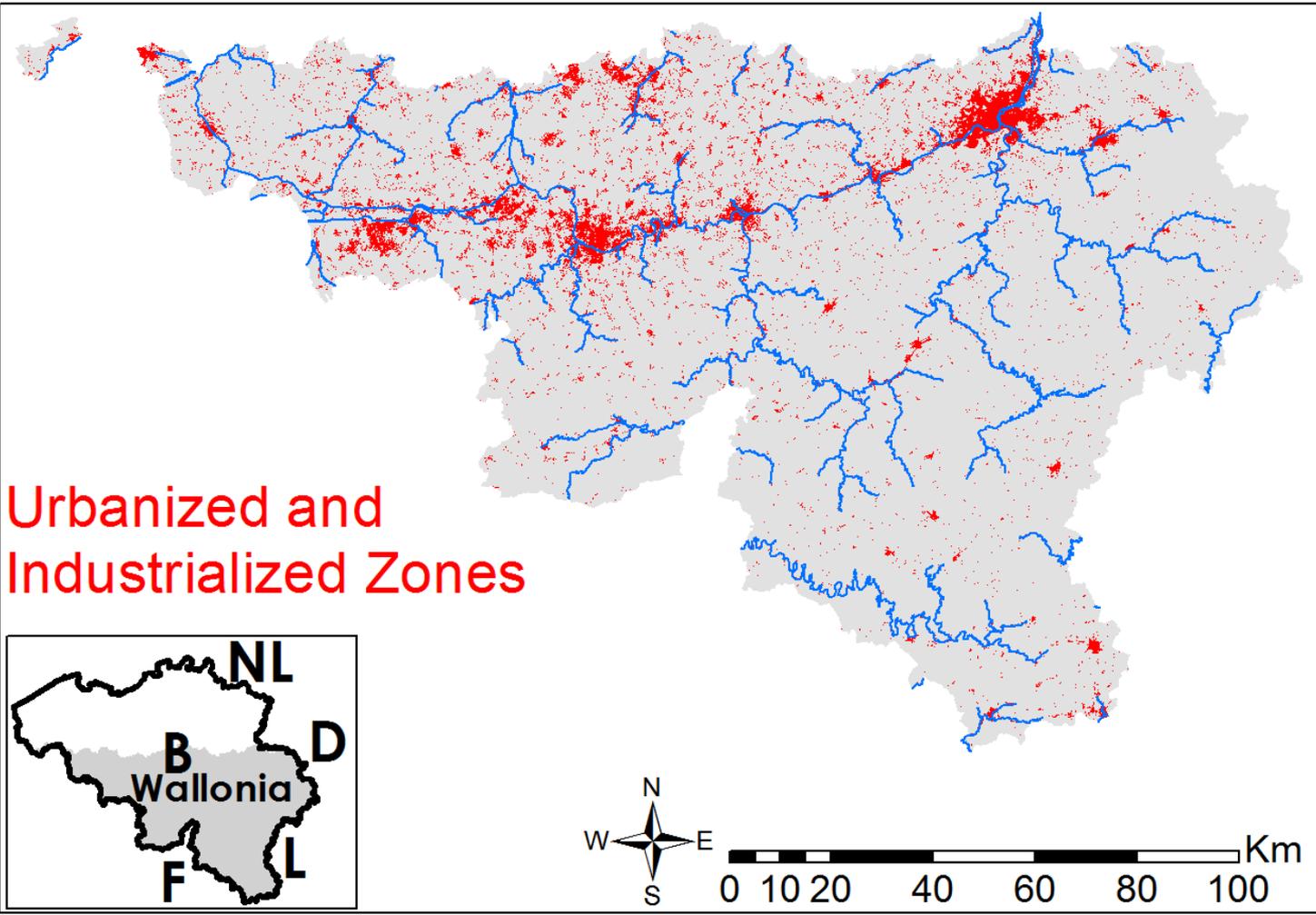
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In urbanised and industrialised zones (U&IZ), diffuse pollution of groundwater is usually less known...

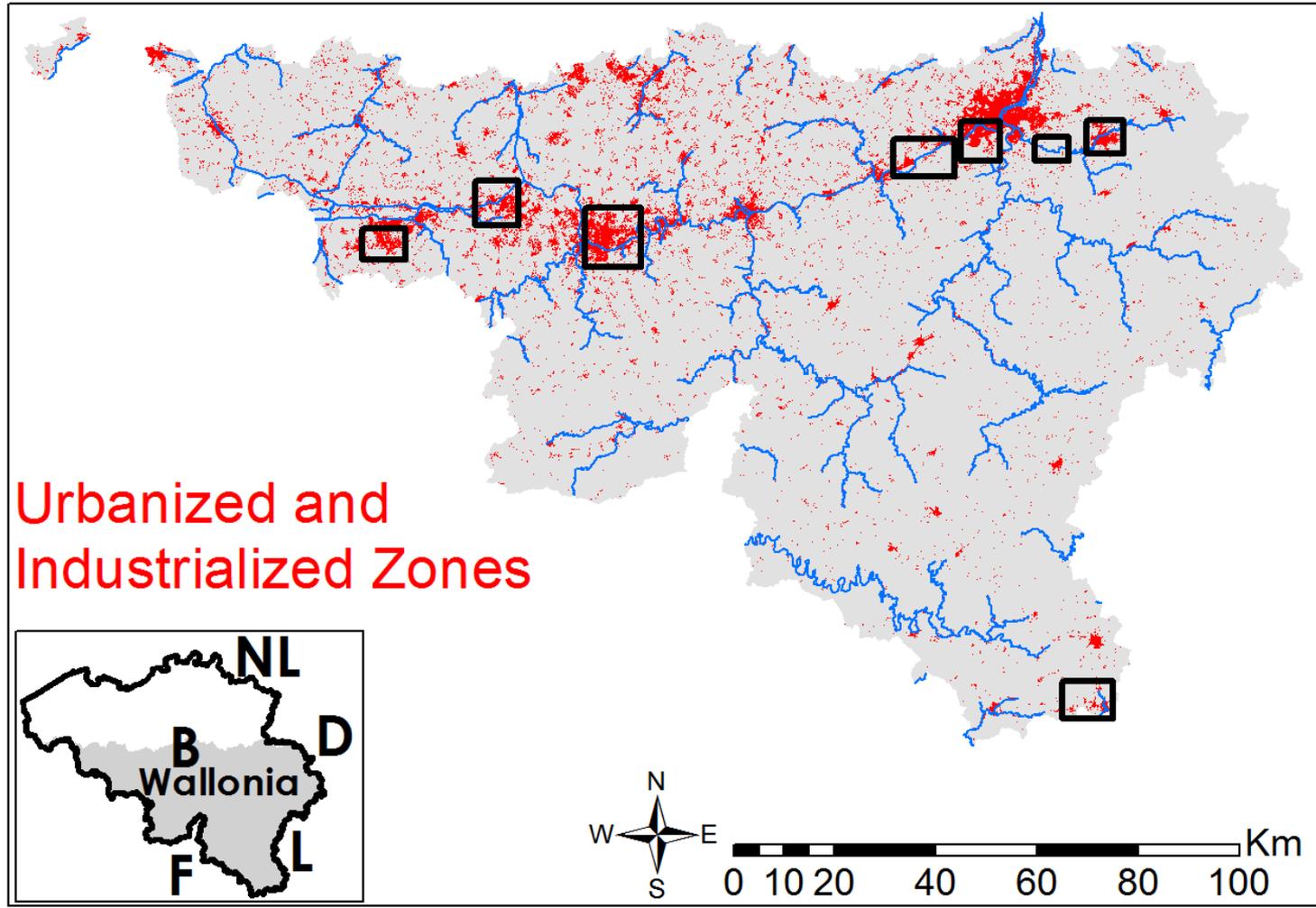


- What are the most frequent pollutants?
- Difference between U&IZs and rural areas in terms of backgrounds?
- Mechanisms of occurrence? Man-induced or Natural factors?
- Consequences in terms of groundwater management?

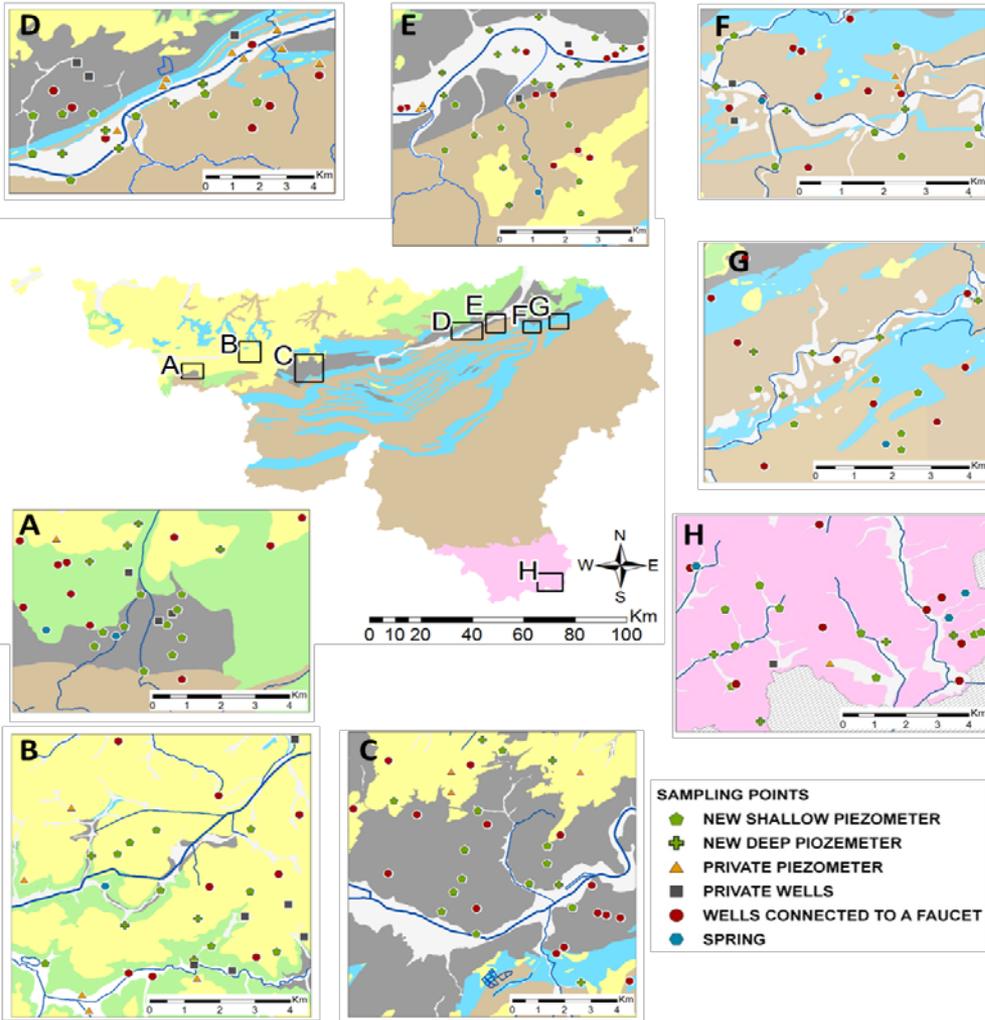
To obtain pieces of answer to these questions:  
regional scale investigations in several U&IZ



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regional scale investigations in several U&IZ



# Investigations of different geological contexts



**SAMPLING POINTS**

- ◆ NEW SHALLOW PIEZOMETER
- NEW DEEP PIEZOMETER
- ▲ PRIVATE PIEZOMETER
- PRIVATE WELLS
- WELLS CONNECTED TO A FAUCET
- SPRING

**Main Walloon aquifers**

□ Alluvial gravels (Neogene)	■ Chalk and marl (Cretaceous)	■ Shale and sandstone (Upper Carboniferous)
■ Sands (Paleogene and Neogene)	■ Sandstone and shale (Jurassic)	■ Limestone (Devonian and Carboniferous)
	■ Shale and sandstone (Cambrian to Devonian)	

Selection of groundwater sampling locations in all sectors:

- 79 existing pumping wells: Household, agricultural and industrial wells ;
- 40 existing piezometers:
- 9 springs
- 74 additional 2" shallow piezometers
- 58 additional 4" deeper piezometers



# Two sampling campaigns over two years

## 101 parameters analysed on all collected groundwater samples



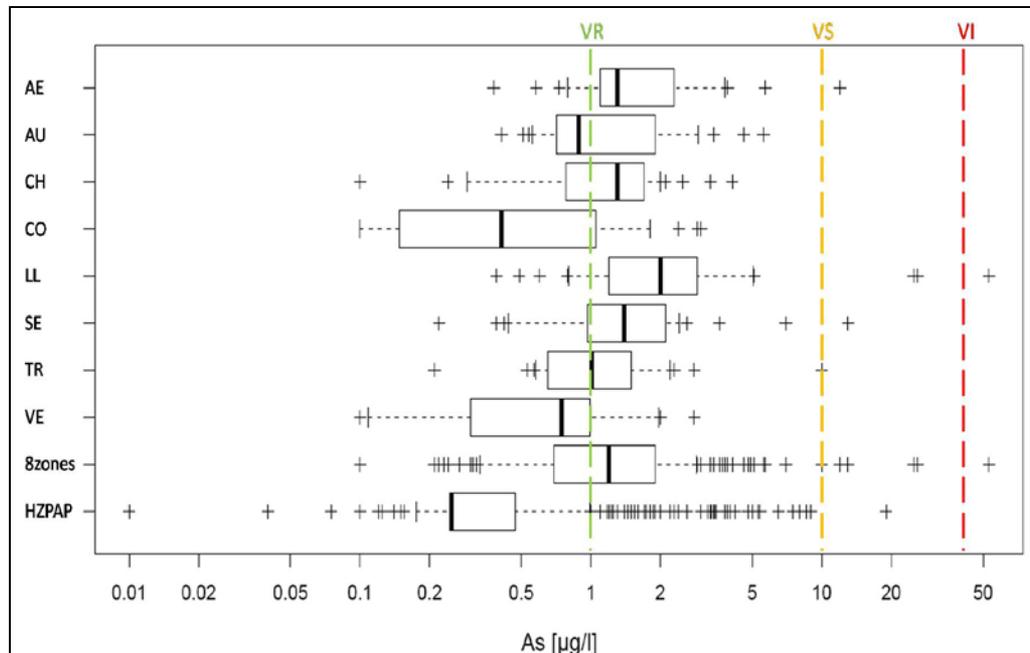
- **In-situ parameters**  
pH, temperature, CE, dissolved O<sub>2</sub>, Eh
- **inorganic trace elements**  
Ag B Ba Be Sb Mo Se Cr As Cd Co Cu Hg  
Ni Pb V Al Zn
- **Organic micro-pollutants**  
16 PAH, BTEX, TPH C10-C35, 17 HVOC,  
chlorobenzenes, phenol index
- **Major /minor inorganic elements**  
Ca Mg K Na SiO<sub>2</sub> Fe Mn NH<sub>4</sub> HCO<sub>3</sub> Cl  
SO<sub>4</sub> NO<sub>3</sub> CO<sub>3</sub> NO<sub>2</sub>

+ Compilation of existing data for the rest of the Walloon region (outside U&IZ's): **70870 records of inorganic compounds** on 7418 samples from 1965 sampling locations grouped into **1432 sampling sites**

# Statistical and graphical description of the data

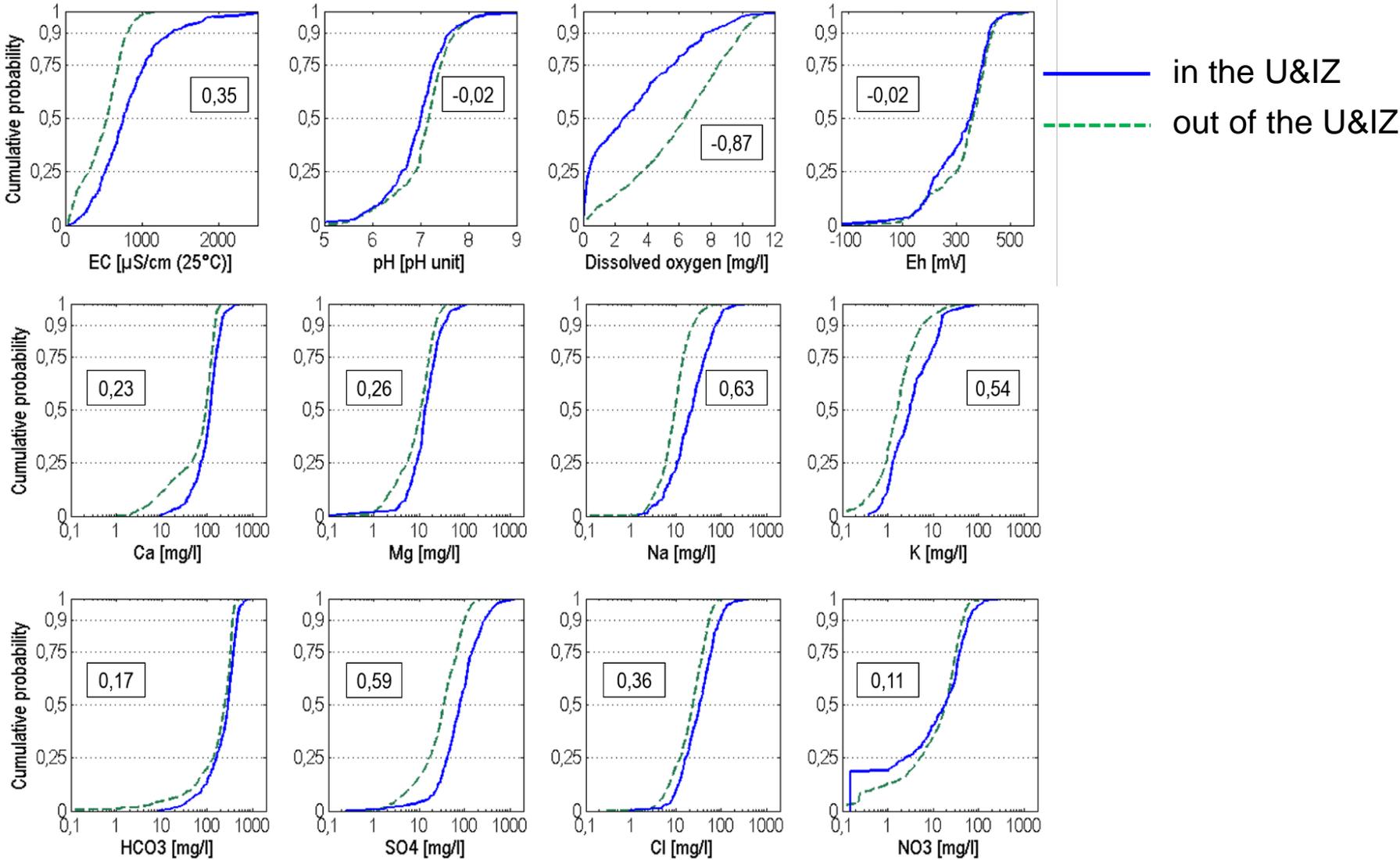
Dataset investigated using general statistical descriptors (FP6 BRIDGE procedure, Muller et al 2006)

- $C < LOQ \Rightarrow C = LOQ/2$
- Calculation of P25, P50, MEAN, P75, P90 percentiles
- Removal of atypical values  $C > P75 + \frac{1}{2} \times (P75 - P25)$
- Representation as Box-Plots & Cumulative Distribution Function Plots (CDFPs)



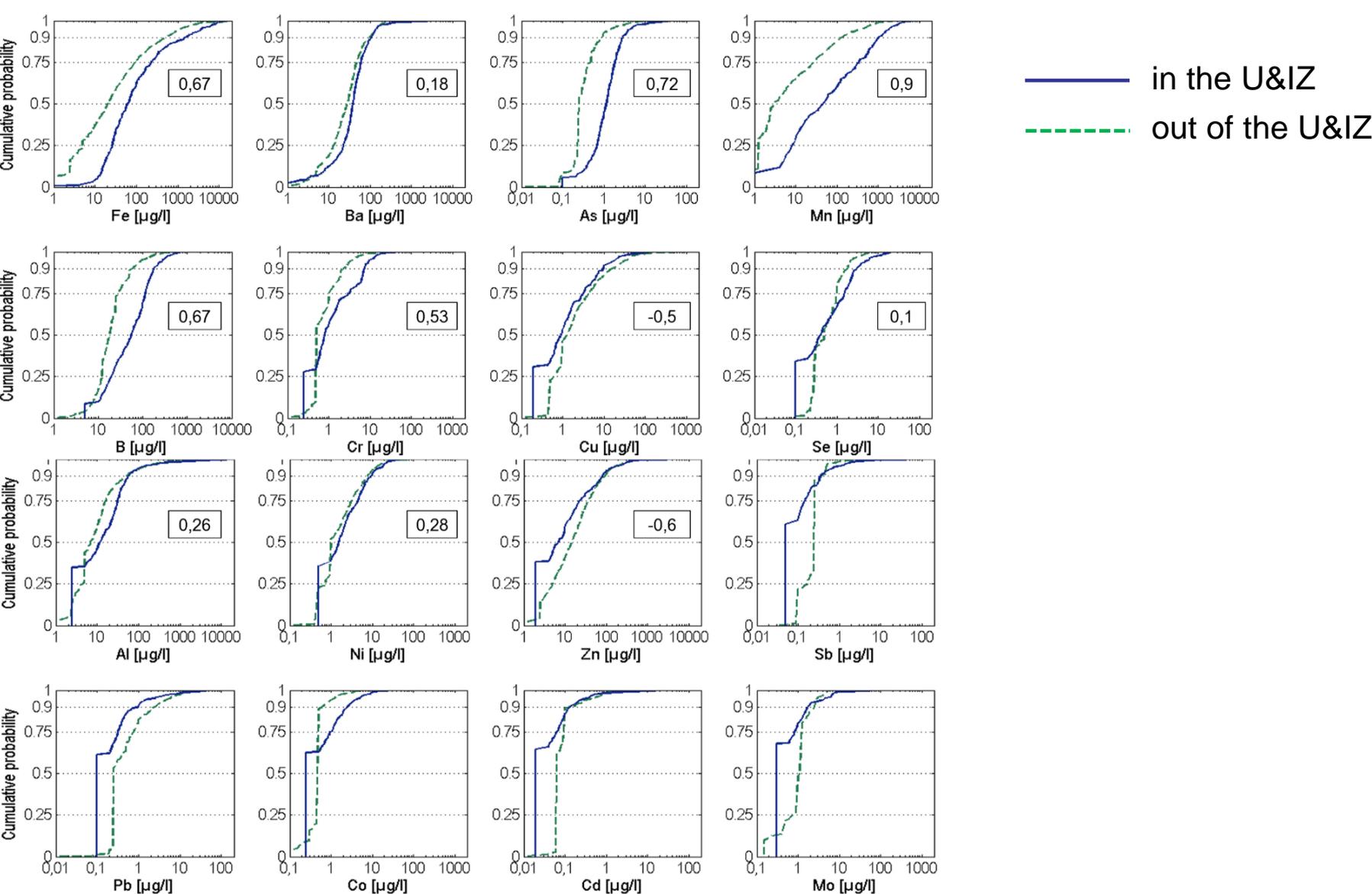
# Major elements and physicochemical parameters

(cumulative distribution function plots)



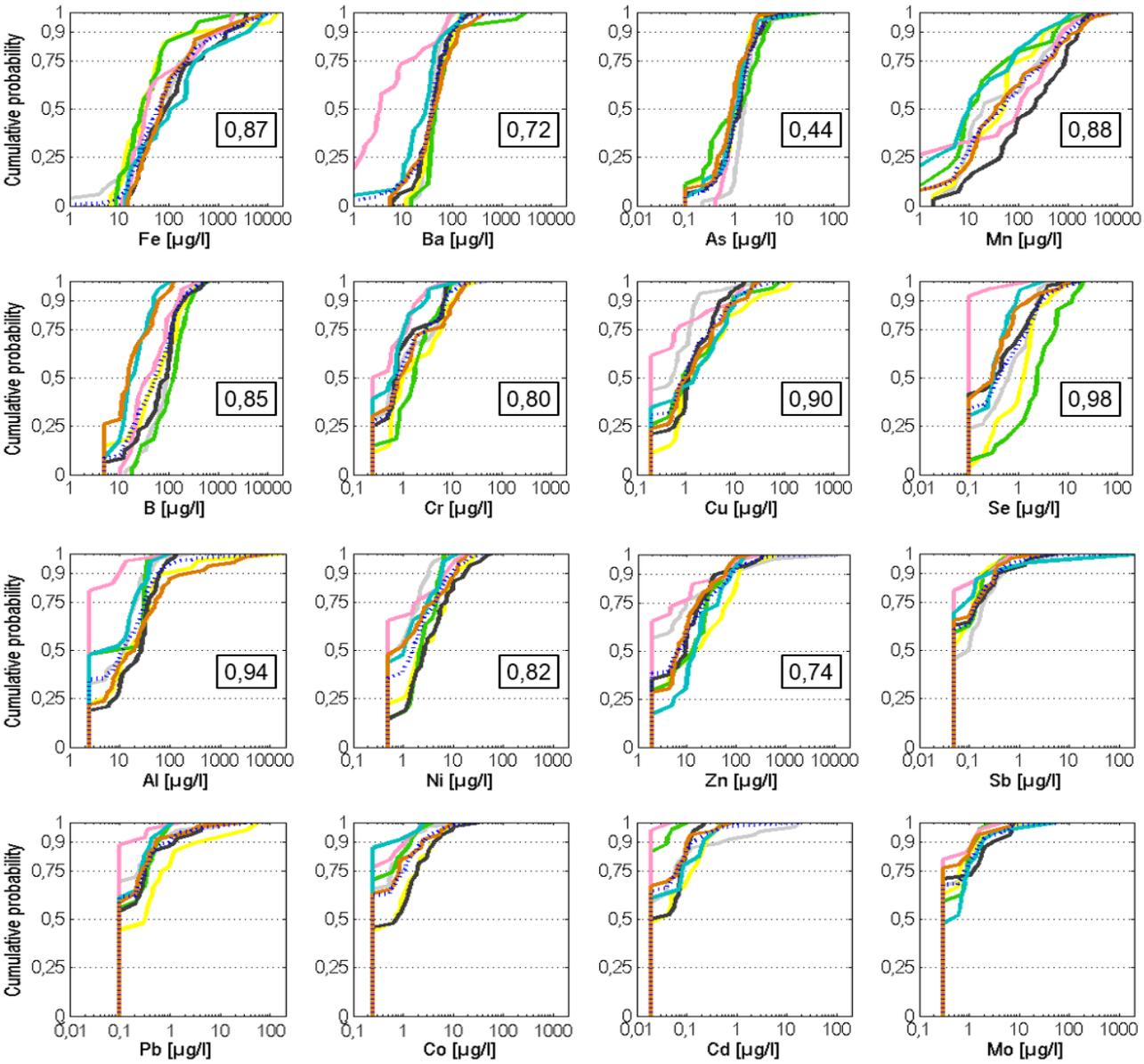
# Inorganic trace elements inside and outside de I&UZ

(cumulative distribution function plots)



# Inorganic trace elements in the different geologies

(cumulative distribution function plots)



- Alluvial gravels
- Sands
- Chalk and marl
- Sandstone and shale
- Shale and sandstone
- Limestone
- Shale and sandstone



# Cross-correlations between different parameters

Fe	Ba	As	Mn	B	Cr	Cu	Se	Al	Ni	Zn	Sb	Pb	Co	Cd	Mo
		0,24	0,33				-0,17						0,27		
	Ba						0,16	0,17					0,17		
		As	0,29	0,27							0,31	-0,17			0,29
			Mn				-0,17		0,22						
				B			0,21				0,16				
					Cr	0,16	0,16					0,18			
						Cu			0,18	0,45		0,44		0,21	
							Se				0,24				
								Al				0,27			
									Ni	0,19	0,17		0,44	0,34	
										Zn				0,36	
											Sb		0,18	0,21	
												Pb			
													Co	0,21	
														Cd	
															Mo

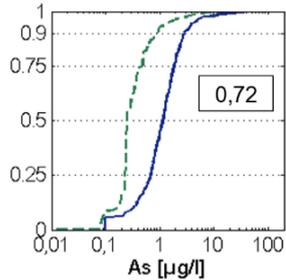
O <sub>2</sub> diss	CE	Eh	pH
-0,25		-0,26	
-0,20	0,29	-0,19	
-0,40		-0,27	-0,19
	0,37		
		0,19	
	0,21		
-0,21			-0,20
			-0,17

Ca	Mg	Na	K	NH <sub>4</sub>	Cl	SO <sub>4</sub>	NO <sub>3</sub>	HCO <sub>3</sub>	SiO <sub>2</sub>
							-0,34		
		0,21	0,16						
0,31	0,19	0,15	0,24			0,30		0,17	0,19
						0,16	-0,36		
0,31	0,24	0,39	0,45		0,23			0,32	
	-0,16						0,23		
0,21		0,30	0,24		0,16	0,26	0,45		
	0,17								
		0,15	0,16				0,18		0,17
			0,16					0,16	
				0,27					
								0,18	
		0,16				0,18	-0,16		
								-0,20	
			0,28						

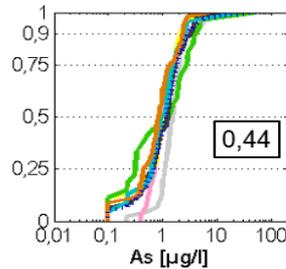
Kendall-tau, with significance level for  $p < 0.001$

# Explaining factors: Example 1 : Arsenic

— in U&IZ    - - - out U&IZ



As higher in U&IZ compared to non U&IZ



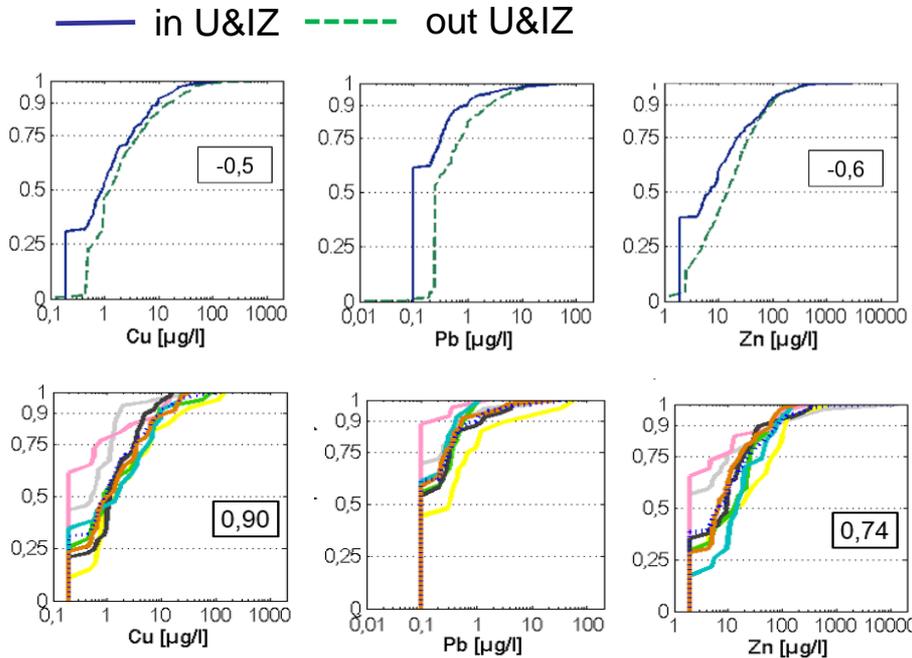
As not so geology dependent

As positively correlated with Fe, Mn, SO<sub>4</sub>...  
negatively correlated with dissolved O<sub>2</sub>, Eh

→ As occurrence in groundwater related to changes in redox conditions in the U&I zones

→ Pyrite oxydation + co-precipitation with Fe – Mn oxy-hydroxides, followed by the dissolution of these minerals in reducing environments (e.g. Pauwels et al. 2010)

# Explaining factors: Example 2 : Cu, Pb, Zn



Cu, Pb, Zn slightly lower in U&IZ compared to non U&IZ

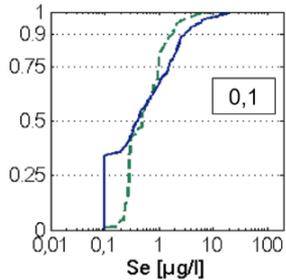
Cu, Pb, Zn relatively geology dependent

Cu, Pb, Zn positively correlated with each other  
Cu positively correlated with  $\text{NO}_3$ ,  
Pb, Zn as well but to a lesser extent

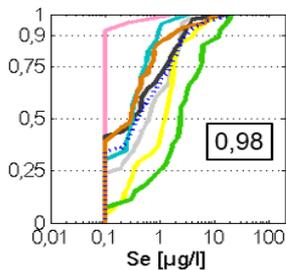
→ Slightly reducing (anaerobic) redox conditions favour the occurrence of Cu, Pb and Zn in groundwater

# Explaining factors: Example 3 : Selenium

— in U&IZ    - - - out U&IZ



Se not so different in and out of U&IZ



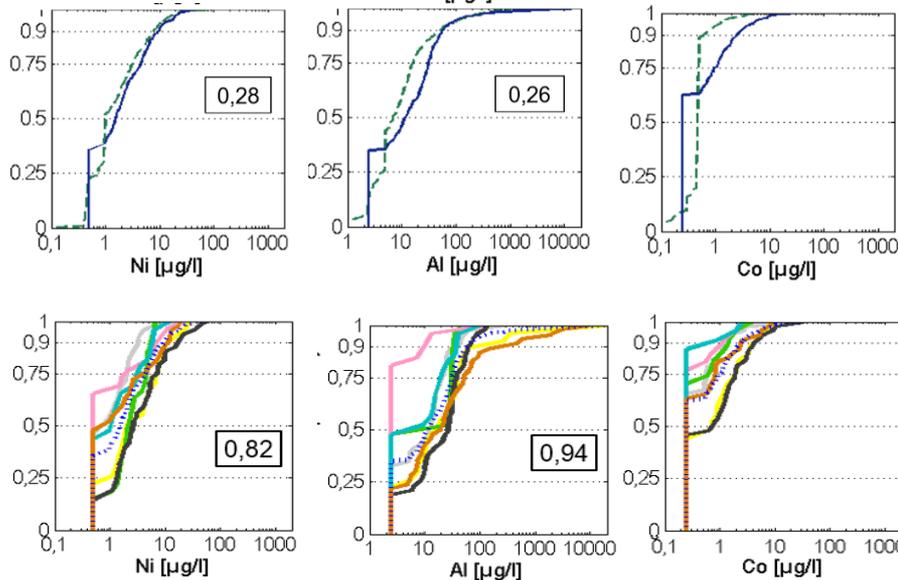
Se strongly geology dependent, with higher concentrations in chalk / sand aquifers

Se positively correlated with CE and major elements, particularly with nitrate

- Geogenic origin, with selenium released by oxidation of sulfide minerals
- Selenium also oxidized into selenate, remaining oxidized and dissolved because of high NO<sub>3</sub> keeping oxidizing anaerobic conditions (e.g. Wright 1999, Bailey et al. 2011)

# Explaining factors: Example 4 : Ni, Al, Co

— in U&IZ    - - - out U&IZ



Ni, Al, Co not very different in and out of U&Izs

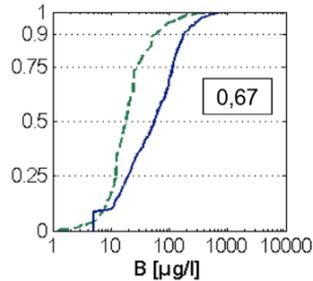
Ni, Al, Co relatively geology dependent (higher in Carboniferous shale)

Ni, Al, Co negatively correlated with pH  
Co correlated with Fe, Ni with Mn

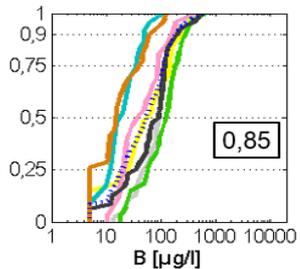
- Acidic conditions contribute to their solubility
- Source: sorption on Fe-Mn oxi-hydroxides and release when these mineral dissolve

# Explaining factors: Example 5 : Boron

— in U&IZ    - - - out U&IZ



Boron higher in U&IZ compared to non U&IZ



Boron geologically dependent

Boron not correlated with redox indicators  
correlated with CE and major elements,  
particularly to K, Cl and Na

- Boron occurrence in groundwater related to sewage waste waters
- correlation with CE & EM + geology dependence → geogenic sources as well

# Synthesis and conclusions

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In urbanised and industrialised zones (U&IZ), the occurrence of inorganic pollutants in groundwater can be related to different factors...

External inputs from pollution sources:  
no strong evidences, except for boron from waste waters

Geogenic sources:  
evidences for several compounds such as Fe, Mn, As, Se ...

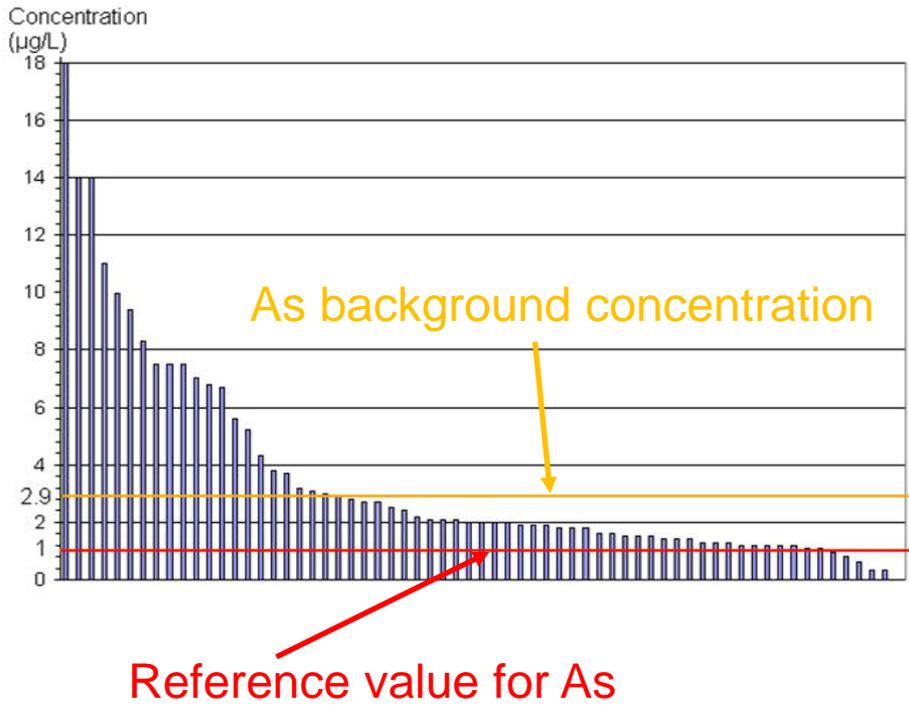
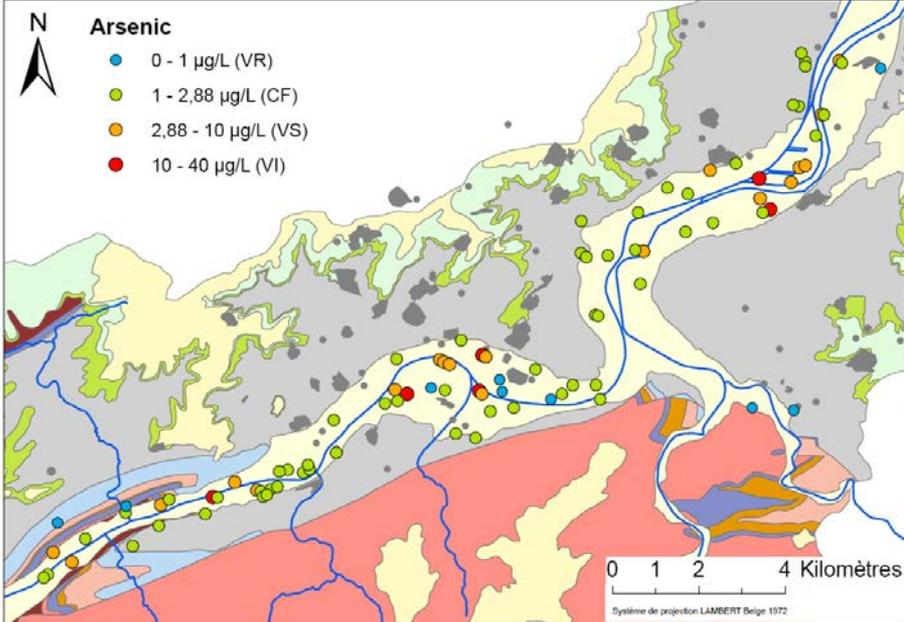
But among all, it seems that the most significant factor relates to changes in the environmental conditions in groundwater, especially **lower redox conditions, with anaerobic conditions** prevailing more often in the U&IZ than outside

# Synthesis and conclusions

In terms of decision-making related to groundwater management ...

Trigger values for pollutant concentrations in groundwater to be adapted to account for the generalized pollution in the U&IZ (e.g. remedial objectives)

e.g. Arsenic in the Meuse alluvial aquifer around Liège



# Synthesis and conclusions

In terms of decision-making related to groundwater management ...

Trigger values for pollutant concentrations in groundwater adapted to account for the generalized pollution in the U&IZ (e.g. remedial objectives) → **BCK\_CONC= P90**

	<i>Fe</i>	<i>Ba</i>	<i>As</i>	<i>Mn</i>	<i>B</i>	<i>Cr</i>	<i>Cu</i>	<i>Se</i>	<i>Al</i>	<i>Ni</i>	<i>Zn</i>	<i>Sb</i>	<i>Pb</i>	<i>Co</i>	<i>Cd</i>	<i>Mo</i>
<i>Reference Value</i>			1,0			2,5	15,0			10	90		2,5		0,25	
Alluvial gravels (Neogene)	1124	66,0	2,57	807	185	6,80	1,70	2,50	37,5	3,25	70,4	0,43	0,43	1,25	0,09	1,68
Sands (Paleogene and Neogene)	112	118	2,30	460	262	11,08	25,6	4,16	53,5	15,2	114	0,20	1,07	4,32	0,25	1,68
Chalk and marl (Cretaceous)	313	111	3,75	591	290	7,32	10,0	12,4	34,0	5,62	70,2	0,24	0,59	1,48	0,04	1,32
Sandstone and shale (Jurassic)	1306	67,5	2,95	875	180	2,20	9,70	0,20	10,0	8,70	72,5	0,26	0,21	1,15	0,04	1,25
Shale and sandstone (Upper Carboniferous)	1774	91,4	2,88	1710	206	7,20	5,75	2,86	64,3	19,0	47,6	0,35	0,46	3,90	0,12	3,25
Limestone (Devonian and Carboniferous)	4325	85,6	2,40	744	50,8	3,20	9,68	1,07	34,8	6,00	80,0	0,28	0,41	0,79	0,29	1,60
Shale and sandstone (Cambrian to Devonian)	886	118	2,40	1670	66,5	12,0	13,9	1,75	87,8	11,5	61,5	0,35	0,47	2,50	0,11	1,30

*Resulting background concentrations for the main aquifer contexts of the Walloon region (conc. In µg/L)*

# Synthesis and conclusions

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In terms of decision-making related to groundwater management ...

In the scope of the Water Framework Directive, **elevated background concentrations should also lead to revision of groundwater quality objectives and management plans** to account for these specific contexts (excessive costs, technical infeasibility of remedial options...)

# Any questions?

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