

Chapter 13

Fundamentals of Cold-Adapted Enzymes

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13.1 Introduction

As already discussed in the preceding chapters, life at low temperatures is abundant, diverse and widespread, with organisms from all three domains of life being represented. Bacteria and archaea at thermal equilibrium with their environment are found to be preponderant, and these cold-adapted, or psychrophilic, microorganisms have been found to not only endure, but to flourish under the harsh conditions of permanently low-temperatures. In fact, for some, this environment is not only optimal, but mandatory for sustained cell proliferation, with moderate to high temperatures (e.g., >12°C) being deleterious (Xu et al. 2003c).

Clearly, adaptations at all levels of organisation, both structural and physiological, have occurred which allow these psychrophilic microorganisms to overcome key obstacles inherent to life at low temperatures (D'Amico et al. 2006a). The current knowledge of the adaptive features characteristic to psychrophilic organisms are addressed in this book (see Chaps. 11–19) with the present chapter being focused on the enzymes produced by these and on their adaptation to the cold. Here, the challenges imposed by low temperatures on enzyme activity and stability will be briefly introduced, whereafter the general characteristics of the cold-adapted enzymes hitherto investigated will be presented and discussed.

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13.2 The low temperature challenge

Low temperatures have a strong negative effect on most biological reactions; reducing reaction rates, provoking alterations in the strength of enzyme-substrate interactions, inducing an increased viscosity of the solvent and an altered solubility of proteins, salts and gases and ultimately leading to protein cold-denaturation (Makhatadze and Privalov 1994; Georlette et al. 2004).

All chemical reactions, including enzymatic reactions, are strongly temperature dependent as described by the Arrhenius equation (Arrhenius 1889):

$$k_{cat} = Ae^{-Ea/RT} \quad (1)$$

From this equation, it can be clearly seen that any decrease in the temperature (T) leads to an exponential decrease in the enzymatic reaction rate (k_{cat}) with the magnitude of this decrease being dependent on the value of the activation energy for the reaction (Ea). Here, R is the universal gas constant ($8.314 \text{ J mol}^{-1} \text{ K}^{-1}$) and A is a frequency factor related to the frequency of collision of the reactants and to the probability of the reactants being in the appropriate orientation to react. In fact, the rate constant can also be expressed according to the transition state theory to give:

$$k_{cat} = \frac{\kappa k_B T}{h} e^{-\Delta G^*/RT} \quad (2)$$

From this and equation (1), one can also derive that (Garcia-Viloca et al. 2004; Collins et al. 2007):

$$A = \frac{\kappa k_B T}{h} e^{-\Delta S^*/R} e^1 \quad (3)$$

Thus, it can be seen that A is related to the activation entropy (ΔS^*). In the above equations, κ is the transmission coefficient expressing the probability that the activated state will not always give rise to the product, but will return to the ground state, k_B is the Boltzmann constant and h the Planck constant (Garcia-Viloca et al. 2004; Collins et al. 2007). The transmission coefficient κ can be negatively influenced by the viscosity of the medium through possible re-crossing and non-equilibrium effects so that the frequency factor A will also be dependent on the viscosity of the reaction medium, and the increase in this latter, concomitant to a temperature decrease, can indeed be quite significant. For example, the average cytoplasmic viscosity is augmented from a mean of 2.5 cP at 20°C (Mastro and Keith 1984) to approximately 5 cP at 0°C. This will obviously have a negative effect on enzyme activity, and such a viscosity increase has been shown to lead to an approximately 5-fold loss of activity for a lactate dehydrogenase at 25°C (Demchenko et al. 1989). Thus, it is important to keep this inverse relationship of viscosity with temperature and reaction rate in mind when monitoring temperature effects on enzyme activity. A model which takes this viscosity effect into account has even been recently proposed (Siddiqui et al. 2004).

The effect of low temperatures on enzyme–substrate binding is much less clear-cut than that on reaction rates, with both positive and negative outcomes being possible. In effect, electrostatic interactions are formed exothermically and are stabilized by low temperatures at least down to those temperatures where the dehydration of the individual ions can still be achieved. On the other hand, hydrophobic interactions are formed endothermically in the positive temperature range up to approximately 40°C and hence are destabilised at low temperatures. Thus, the effect of temperature (either detrimental or beneficial) is determined by the types and contributions of the bonds involved in substrate binding and hence by the particular enzyme and substrate concerned.

Finally, cold denaturation, while being a scantily studied phenomenon, is believed to be due to the hydration of charged, polar and non-polar amino acid residues that leads, in particular, to a weakening of critical hydrophobic forces. While this should typically occur at temperatures below the freezing point of water, it obviously poses a further predicament to enzymes produced by organisms inhabiting permanently low temperature environments (Makhatadze and Privalov 1994).

As a result of these diverse adverse effects, most biological reactions display an approximately 16- to 80-fold drop in activity on reducing the temperature from 37°C to 0°C. In contrast, psychrophilic microorganisms have been found to maintain relatively high metabolic fluxes at low temperatures, indicating that adaptation of the enzymatic repertoire has taken place to allow for appropriate reaction rates. Reported mechanisms of cold-adaptation include an increased enzyme production (Crawford and Powers 1992; De Vos et al. 1998), yet this is energetically expensive, and while an expression of specific isotypes adapted to different temperatures has also been reported (Lin and Somero 1995), this suffers the disadvantage of requiring the presence of multiple gene copies and is most apt for organisms requiring seasonal adaptation (e.g., fish and nematodes). Finally, the synthesis of enzymes specifically adapted to operate at permanently low temperatures is another putative cold-adaptation mechanism and this in fact appears to be the main physiological adaptation used by psychrophiles at the enzyme level.

We have attempted to tabulate all the currently known enzymes isolated from psychrophilic organisms, both prokaryotic and eukaryotic, and the listing of these can be found at <http://www.ulg.ac.be/biochlab> where both the identity of the host organism as well as the level of characterisation of the enzymes are also indicated. Only those enzymes purified from psychrophilic organisms inhabiting permanently low temperature environments have been listed and it currently appears that 108 prokaryotic and 51 eukaryotic enzymes have been isolated and studied. These originate from various sources and have been characterized to varying levels in terms of their physical, chemical, kinetic and structural properties. The majority of these have been found to have evolved strategies for overcoming the temperature challenge. In the subsequent sections, the principle results of the various studies of these enzymes and in particular the comparative studies of the activity, stability and flexibility will be presented and discussed.

13.3 Activity

In relation to activity, the dominant features of most cold-adapted enzymes are an increased specific activity (k_{cat}) or catalytic efficiency (k_{cat}/K_m) at low and moderate temperatures and a shift in the apparent optimal temperature towards low temperatures as compared to their mesophilic and thermophilic homologs (Fig. 13.1).

The apparent optimum temperature of an enzyme is obviously dependent on the experimental conditions used and in particular on the assay time. Nevertheless, analysis of the available data shows that the thermodependence of activity of psychrophilic enzymes is highly variable, with diverse apparent optimum temperatures for activity and both limited and broad temperature ranges being reported. For example, while apparent temperature optima around 30°C are preponderant, optima at approximately 15°C (Huston et al. 2000; Arnorsdottir et al. 2005) and as high as 64°C (Birolo et al. 2000) have also been reported. In fact, the defining feature of these cold-adapted enzymes is a shift in the apparent optimum temperature towards lower values and the retention of a high activity at low temperatures, in particular when compared to that of their mesophilic and thermophilic homologs (Bae and Phillips 2004). Examples include a psychrophilic xylanase (Collins et al. 2002b) and chitinase (Bendt et al. 2001) which retain, respectively, 60% and 40% of their

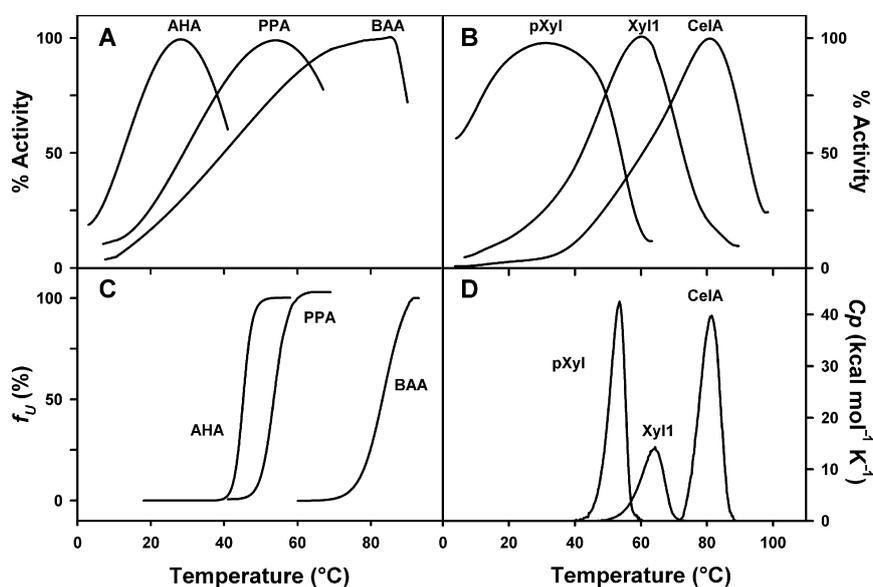


Fig. 13.1 Thermodependence of activity (A and B) and thermal stability as determined by fluorescence monitoring (C) and differential scanning calorimetry (D) of homologous α -amylases (A and C) and glycoside hydrolases (B and D). AHA, PPA and BAA are, respectively, psychrophilic, mesophilic and thermophilic α -amylases. pXyl, Xyl1 and CelA are, respectively, psychrophilic, mesophilic and thermophilic glycoside hydrolases. Taken from Georlette et al. (2004)

maximum activity at 5°C, and display apparent optimum temperature shifts towards lower temperatures of approximately 25°C. Indeed, this shift in the apparent optimum temperature for activity is related to the thermal lability of these enzymes (see Sect. 13.4), while the reduced temperature dependence and high activity at low temperatures is brought about by a reduction of the activation energy barrier for the reaction ΔG^* (i.e. the energy difference between the activated enzyme–substrate transition state and the ground state enzyme–substrate complex).

Extensive comparative thermodynamic analysis of the activity of psychrophilic, mesophilic and thermophilic enzymes (Lonhienne et al. 2000; Siddiqui and Cavicchioli 2006) indicates that, as compared to their higher temperature adapted homologs, the low ΔG^* of cold-adapted enzymes is characterized by a reduced activation enthalpy partially compensated for by a less favorable activation entropy. The decrease of the former reflects the lower dependence of these enzymes on temperature and indicates a reduction in the number or strength of weak interactions that are broken during activation. Accordingly, this favorable enthalpy change is the driving force for low temperature activity, but the magnitude of its effect is reduced by the unfavorable entropy change observed (i.e. $\Delta G^* = \Delta H^* - T\Delta S^*$). It is important to note that, while the entropy change on activation is often negative due to a more organized transition state, positive values are also encountered. Therefore, the unfavorable entropy change observed during the activation process in psychrophilic enzymes indicates, in the former case, the necessity for a greater change in order to reach the activated state (i.e. a higher negative entropy change as compared to mesophilic and thermophilic homologs) and, in the latter case, a lower degree of disordering (i.e. a lower positive entropy change). Both of these observations can be explained by a more disordered enzyme–substrate ground state complex for the cold-adapted enzyme as compared to its high temperature homologs. This is postulated to be the result of a more flexible enzyme structure leading to a broader distribution of conformational states for the ground state complex (Lonhienne et al. 2000).

In agreement with the apparent increased disorder of the enzyme–substrate ground state complex in psychrophilic enzymes, these have been found to be frequently characterized by an increased K_m in addition to a high k_{cat} (Collins et al. 2002a). Furthermore, this high K_m value frequently observed for psychrophilic enzymes may be conjectured as further proof of a highly flexible structure (Fields 2001), in particular at or around the active site, which perturbs the geometry and enzyme–substrate complementarity and leads to the observed decreased substrate affinity (i.e. increased K_m). Similarly, the frequently observed decreased affinity for various ions (Davail et al. 1994; D’Amico et al. 2006b), cofactors (Okubo et al. 1999) and inhibitors (Alvarez et al. 1998), and the reduced substrate specificity of some cold-adapted enzymes (Smalas et al. 2000), also points to an increased flexibility of the molecular edifice of these. Finally, and in a similar vein of thought, recent comparative studies of a psychrophilic α -amylase with its mesophilic counterpart has indicated that the former, in contrast to the latter, is capable of forming an enzyme–substrate–inhibitor ternary complex and has a higher activity on large macromolecular substrates but a decreased activity on small substrates (D’Amico et al. 2006b). These observations, in conjunction with the structural identity of the

catalytic cleft of these enzymes are also compatible with a more flexible active site in the psychrophilic enzyme.

For enzymes operating at high substrate concentrations, for example extracellular and in particular digestive enzymes (Narinx et al. 1997; Collins et al. 2002b), the high K_m value discussed above is of little importance as optimization of k_{cat} is the only relevant parameter. In contrast, for enzymes typically operating at low substrate concentrations, for example certain intracellular enzymes, both K_m and k_{cat} are important and indeed both of these parameters have been found to be optimized in a number of cases (Bentahir et al. 2000; Hoyoux et al. 2001; Lonhienne et al. 2001). Obviously, a different strategy must be employed by these proteins whereby the adaptive drift of the K_m is counteracted. For example, comparison of the kinetic parameters of a cold-adapted and mesophilic chitinase (Fig. 13.2) reveals that these enzymes have been fine tuned so as to enable optimization of their kinetic parameters in accordance with their physiological temperature (Lonhienne et al. 2001). In particular, the crossed K_m plots for these enzymes illustrate the improved substrate binding for the psychrophilic enzyme on reduction of temperature, and this is believed to be achieved by the subtle replacement of two tryptophan residues (endothermic interactions) involved in substrate binding in the mesophile by polar residues (exothermic interactions) in the psychrophile, which, as discussed above, improves the enzyme-substrate interactions at low temperatures.

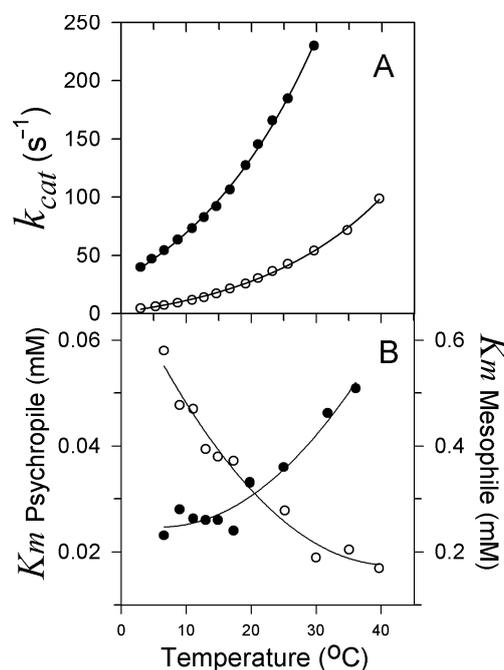


Fig. 13.2 Thermodependence of the k_{cat} (A) and K_m (B) for a psychrophilic (•) and mesophilic (o) chitinase. Note the different scales used. Taken from Lonhienne et al. (2001)

A further interesting observation of a number of comparative studies is that both the K_m and k_{cat} of psychrophilic and mesophilic enzymes are often highly similar at their respective physiological temperatures. This gives rise to the corresponding state hypothesis suggested by Somero (1995) whereby enzymes have evolved so as that both of these parameters are optimized at the environmental temperature of the enzyme. However, this kinetic optimization has not always been possible and the kinetic parameters of some psychrophilic enzymes at their environmental temperature are sometimes found to be inferior to that of their high temperature homologs (Feller et al. 1992; Xu et al. 2003a; Xu et al. 2003b), hence pointing to an 'incomplete' adaptation of these enzymes to their environment.

13.4 Stability

In light of the essential role of protein stability in thermal adaptation, extensive studies have been performed in order to compare the conformational stability of psychrophilic enzymes with their mesophilic and thermophilic counterparts, and have shown that cold-adapted enzymes are generally the most unstable. As demonstrated by techniques such as fluorescence spectroscopy and differential scanning calorimetry (Fig. 13.3) their unfolding typically occurs at lower temperatures and the calorimetric enthalpy, ΔH_{cal} , i.e., the enthalpy of disruption of bonds involved in maintaining the protein structure, is generally reduced. Furthermore, thermal unfolding curves for cold-adapted enzymes are typically sharp and symmetrical and thereby reflect a highly cooperative process without any stable intermediate, while in contrast more stable enzymes exhibit distinct thermodynamic domains of varying stability (Fig. 13.3).

13.4.1 Reversible and irreversible unfolding

While the majority of proteins unfold irreversibly, some have been found to show completely reversible thermal denaturation and thus allow calculation of the stabilization energy (ΔG) curves as presented in Fig. 13.4. These curves represent the energy difference between the free energy of the unfolded state and that of the native state, in other words, they represent the energy required to unfold the molecular structure at each temperature. It can be clearly seen that the increase of the melting points of the more thermostable proteins is obtained by an uplifting of the stability curves and that this is mainly achieved through an increase of the stabilization enthalpy, hence demonstrating the major involvement of enthalpic factors in the stabilization of protein structures (Makhatadze and Privalov 1995; Kumar et al. 2001). Furthermore, the environmental temperatures of mesophiles and thermophiles are found to be above those corresponding to the ΔG_{max} and, consequently, it can be suggested that these enzymes use the thermal dissipative forces

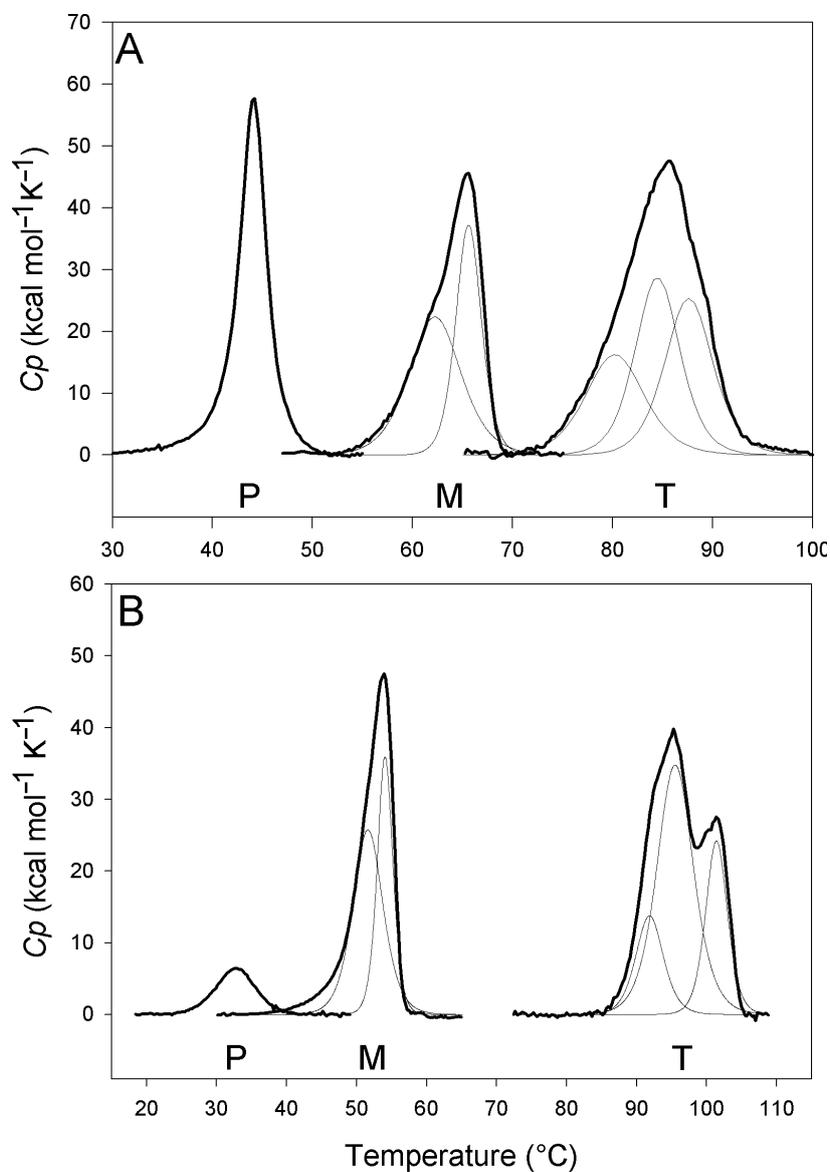


Fig. 13.3 Thermal unfolding of psychrophilic (*P*), mesophilic (*M*) and thermophilic (*T*) enzymes recorded by differential scanning calorimetry. **A** From left to right: α -amylases from *Pseudoalteromonas haloplanktis*, pig pancreas and *Bacillus amyloliquefaciens*. **B** From left to right: DNA ligases from *Pseudoalteromonas haloplanktis*, *Escherichia coli* and *Thermus scotoductus*. Deconvolution of the peaks into two or three domains for the mesophilic and thermophilic enzymes respectively are shown in thin lines. Taken with modification from D'Amico et al. (2001) and Georlette et al. (2003)

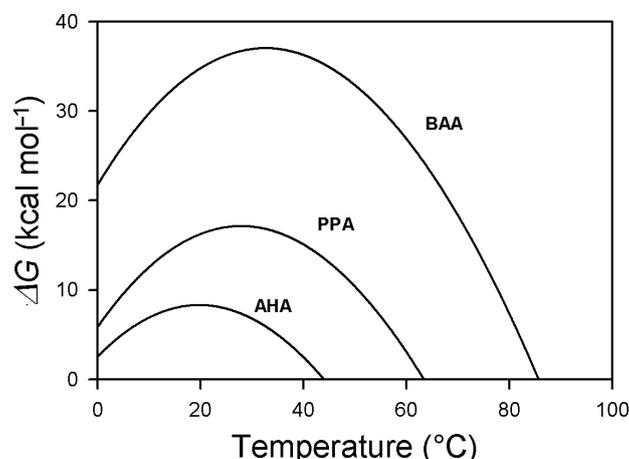


Fig. 13.4 Stabilization energy curves of three α -amylases: AHA, psychrophilic, PPA, mesophilic and BAA, thermophilic, as calculated from microcalorimetric data. Taken from D'Amico et al. (2003)

of high temperatures to promote the molecular motions needed for enzyme function. In contrast, for psychrophiles, the environmental temperatures are below that of the ΔG_{max} , indicating that molecular motions at low temperatures are gained from the hydration of ionic-, polar- and non-polar interactions, i.e. the factors ultimately leading to cold unfolding (Feller 2007). In fact, the stability curves shown in Fig. 13.4 also allow the prediction (by extrapolation) of a temperature of cold-unfolding and it can be seen that, contrary to what would be intuitively expected, cold-adapted enzymes are also the least resistant to cold denaturation. This poorly investigated observation must be kept in mind when considering the lower temperature limit of life and indeed further studies in this field are called for.

As has already been mentioned, the majority of proteins unfold irreversibly, thereby preventing extraction of thermodynamic equilibrium data but nevertheless allowing for determination of the activation parameters for these irreversible processes. Irreversible denaturation and/or inactivation have been found to be characterized by a low activation free energy value (ΔG^\ddagger) in cold-adapted enzymes (Collins et al. 2003; D'Amico et al. 2003; Georlette et al. 2003) and simply reflect the rapid unfolding of these proteins. The small differences observed in the ΔG^\ddagger arise from large differences in the enthalpic and entropic contributions and are the consequence of classical enthalpy–entropy compensations. The high activation enthalpy (ΔH^\ddagger) for psychrophilic enzymes, as a result of the high slope of the Arrhenius plots used to calculate these parameters, simply reflects a higher cooperativity of the denaturation and inactivation processes in these enzymes. Furthermore, the lower value of the activation entropy (ΔS^\ddagger) for the thermostable enzymes suggests that the activated state is reached with less disordering and thus reflects the higher resistance of the thermophilic enzyme to irreversible denaturation.

13.4.2 *Global versus local stability*

From an evolutionary point of view, several studies have discussed the possibility that the instability of psychrophilic enzymes is the result of a random genetic drift due to the lack of a selective pressure on stability (Miyazaki et al. 2000; Wintrode et al. 2001). The fact is, however, that enzymes displaying a high stability as well as a high specific activity at low temperatures are not apparently found in nature, even though this appears to be possible to obtain *in vitro*, with small size synthetic substrates and after several rounds of random mutagenesis where a selective pressure on stability is imposed (Wintrode et al. 2001). Furthermore, it is important to note that while psychrophilic enzymes found in nature are generally characterized by a reduced stability, this destabilization is not always found to involve the whole protein structure and, indeed, a number of proteins with localized zones of reduced stability have been reported. Obviously this observation questions the possibility of random genetic drift playing a part in the poor stability of psychrophilic enzymes as it is improbable that this could affect only a specific part of the protein structure.

In particular, this destabilization in cold-adapted proteins appears to be concentrated at or near the active site. In multi-domain enzymes, such as the psychrophilic chitobiase (Lonhienne et al. 2001), phosphoglycerate kinase (Bentahir et al. 2000) and FKBP 22 (Suzuki et al. 2005), as well as in the precursor of the cold adapted α -amylase (Claverie et al. 2003), the catalytic domain is always found to be the least stable domain whereas other non catalytic domains can be as stable as that of the mesophilic homologs. Furthermore, although the thermal unfolding of the mature α -amylase is highly cooperative, it has been shown using transverse urea gel electrophoresis that the active site is the least stable structural entity in the presence of urea (Siddiqui et al. 2005).

Further strong support for this concept of a reduced stability at or near the active site can be found on comparing enzyme activity and stability (Collins et al. 2003; D'Amico et al. 2003; Georgette et al. 2003). As can be seen from Fig. 13.1, the initial loss of activity for the mesophilic and thermophilic enzymes corresponds quite well with the thermal protein unfolding, whereas in the case of the cold-adapted enzymes the activity decreases well before any unfolding process as recorded by fluorescence spectroscopy or microcalorimetry. This simply means that here the active site or the enzyme–substrate complex is more heat-labile than the whole protein structure and as a consequence the enzyme is inactivated several degrees before the melting temperature. This again points to a localized adaptation in psychrophilic enzymes whereby the active site and/or its surroundings are destabilized and may again be proposed to be due to the requirement for an increased flexibility of these specific regions for cold-adaptation.

13.5 Flexibility

From the previous discussions of the activity and stability of psychrophilic enzymes it is obvious that a number of studies indirectly point to the high activity and low stability of these to being a consequence of an increased flexibility of the whole or

of a particular part of the molecular edifice. Indeed, it is believed that this high flexibility, which derives from the observed low stability of psychrophilic enzymes, would allow for the molecular motions necessary for activity at a low energy cost, i.e. at low temperatures.

In this concept of an activity–stability–flexibility relationship, stability and activity can be experimentally demonstrated quite easily. In contrast, the determination of protein flexibility, mobility or plasticity, as it is also known, is much more difficult and, in particular, if it is limited to a small part of the whole structure. Moreover, we must consider the molecular motions in terms of amplitude as well as time scale. One approach which has been used to evaluate this peculiar characteristic of proteins has been to carry out a comparative study of fluorescence quenching in extremophilic enzymes so as to assess the propensity of the whole molecular structure to being penetrated by a small quencher molecule (acrylamide being the most frequently used). This technique gives an idea of all conformational opening processes occurring in the whole protein and averaged over a large time scale, and it has been successfully used to compare the flexibility of homologous Ca^{2+} - Zn^{2+} proteases (Chessa et al. 2000), xylanases (Collins et al. 2003), DNA ligases (Georgette et al. 2003), α -amylases (D'Amico et al. 2003) and cellulases (G. Sonan, personal communication). These experiments unequivocally demonstrated the decrease in protein permeability on going from psychrophilic to thermophilic enzymes.

Another approach is neutron scattering and this has been used to compare the mean macromolecular dynamics of whole crude extracts obtained from bacteria adapted to different environmental temperatures (Tehei et al. 2004). Here, the authors showed that at a given temperature, the resilience (which is approximately equivalent to rigidity) increased from psychrophile to thermophile. However, at their respective environmental temperatures, the measured resiliencies were similar and therefore in perfect agreement with the “corresponding state” theory (Somero 1995). Other methods which have been used more or less successfully to evaluate protein flexibility include hydrogen/deuterium exchange (Zavodszky et al. 1998; Svingor et al. 2001) and tryptophan phosphorescence (Fischer et al. 2000; Gershenson et al. 2000).

Finally, molecular dynamics studies are becoming increasingly common (Brandsdal et al. 1999; Olufsen et al. 2005; Papaleo et al. 2006; Spiwok et al. 2007) and a recent study of an elastase family (Papaleo et al. 2006) showed that (1) loop regions clustered around the active site are characterized by an enhanced flexibility in the cold-adapted enzyme, and (2) some scattered regions distant from the functional sites are more rigid in the cold-adapted enzyme, thus in perfect agreement with the theory we developed previously in this review.

Thus, it can be seen that while still suffering from the lack of a widely accepted technique to unequivocally characterise protein dynamics, it appears that many psychrophilic enzymes are characterised by an increased flexibility of at least a part of their structure and it is this increased flexibility, originating from the low stability of crucial parts of the molecular structures, which allows for the high activity at low temperatures.

13.6 Structural adaptations

The crystal structures of 22 psychrophilic enzymes have been reported up to now (Table 13.1) and many of these have been subjected to comparative structural analyses with their mesophilic and/or thermophilic homologs. The first observation from these studies is that the overall fold of proteins adapted to different thermal environments are very similar, and many residues, in particular those involved in catalysis, are found to be conserved (Aghajari et al. 1998b). A closer comparative examination does, however, reveal subtle structural modifications, with cold-adapted proteins being most frequently characterized by a reduced number and/or strength of stabilizing interactions. Indeed, any of a large number or combination of a variety of structural adjustments, such as a reduction of salt-bridges, hydrogen

Table 13.1 Cold-adapted enzymes for which the crystal structure has been determined

| Enzyme | Host organism | Reference |
|--|---|----------------------------|
| Adenylate kinase | <i>Bacillus globisporus</i> | Bae and Phillips (2004) |
| Alkaline phosphatase | Bacterial strain TAB5 | Wang et al. (2007) |
| Alpha-amylase | <i>Pseudoalteromonas haloplanktis</i> | Aghajari et al. (1998a) |
| Aspartate carbamoyltransferase | <i>Moritella profunda</i> | De Vos et al. (2007) |
| Beta-galactosidase | <i>Arthrobacter</i> sp. C2-2 | Skalova et al. (2005) |
| Catalase | <i>Vibrio salmonicida</i> | Riise et al. (2007) |
| Cellulase | <i>Pseudoalteromonas haloplanktis</i> | Violot et al. (2005) |
| Citrate synthase | <i>Arthrobacter</i> sp. strain DS2-3R | Russell et al. (1998) |
| Lipase B | <i>Candida antarctica</i> | Uppenberg et al. (1994) |
| Malate dehydrogenase | <i>Aquaspirillum articum</i> | Kim et al. (1999) |
| Phosphatase (alkaline phosphatase) | <i>Pandalus borealis</i> | de Backer et al. (2002) |
| Phosphatase (protein-tyrosine phosphatase) | <i>Shewanella</i> sp. | Tsuruta et al. (2005) |
| Protease (alkaline metalloprotease) | <i>Pseudomonas</i> sp. strain TAC II 18 | Aghajari et al. (2003) |
| Protease (anionic trypsin) | <i>Salmon salar</i> | Helland et al. (1998) |
| Protease (elastase) | <i>Salmon salar</i> | Berglund et al. (1995) |
| Protease (pepsin Iib) | <i>Gadus morhua</i> | Karlsen et al. (1998) |
| Protease (proteinase K-like) | <i>Serratia</i> sp. | Helland et al. (2006) |
| Protease (subtilisin-like serine proteinase) | <i>Vibrio</i> sp. PA-44 | Arnorsdottir et al. (2005) |
| Protease (trypsin) | <i>Oncorhynchus ketav</i> | Toyota et al. (2002) |
| Triose phosphate isomerase | <i>Vibrio marinus</i> | Alvarez et al. (1998) |
| Uracil-DNA glycosylase | <i>Gadus morhua</i> | Leiros et al. (2003) |
| Xylanase | <i>Pseudoalteromonas</i> sp. strain TAH3a | Van Petegem et al. (2003) |

bonds, core hydrophobicity, aromatic interactions, arginine and/or proline content, density of charged surface residues, an increased surface hydrophobicity and an increased clustering of glycine residues etc., have been observed, with each enzyme adopting a specific strategy (Feller et al. 1997; Smalas et al. 2000). Indeed, these are the main structural modifications involved in cold-adaptation and it is obvious how such alterations can give rise to a decreased stability but also an increased flexibility of the molecular structure and hence allow for the molecular motions required for activity at low temperatures.

In addition to the above common observations a number of other more enzyme specific structural adaptations have also been reported. These include a larger and more accessible active site (Russell et al. 1998; Aghajari et al. 2003) an improved electrostatics of and in the vicinity of the reactive centre (Brandsdal 2001; Leiros et al. 2003), a long and highly flexible linker region (Violot et al. 2005) and a reorganization of the quaternary structure (Skalova et al. 2005)

13.7 Conclusions

It is important to remember that while only cold-adapted enzymes have been discussed in this chapter, life at low temperatures requires a multitude of adaptations at all levels within the cell, from single molecules up to supra-molecular structures.

Enzymes achieve cold adaptation by means of fine tuning the activity, flexibility, stability balance. They efficiently catalyze reactions at low temperatures as a result of an inherent increased flexibility, in particular at or near the catalytic site and this enables the molecular motions necessary for activity in this low energy environment. On the other hand, this increased low temperature activity and flexibility is achieved at the expense of stability and frequently also an increased K_m . Of course there is a limit to this adaptation process since the stability of a protein cannot decrease below a certain limit, i.e. it must maintain a properly folded form and this is probably one of the reasons why enzyme adaptation appears incomplete for some proteins. Finally, it appears that cold-adaptation is mainly achieved by means of quite discrete structural modifications and primarily by removal of stabilizing interactions, with each enzyme using a specific strategy to achieve its goal.

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