Electrical transport and percolation in magnetoresistive manganite/insulating oxide composites: Case of La_{0.7}Ca_{0.3}MnO₃/Mn₃O₄

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We report the results of electrical resistivity measurements carried out on well-sintered La_{0.7}Ca_{0.3}MnO₃/Mn₃O₄ composite samples with almost constant composition of the magnetoresistive manganite phase (La_{0.7}Ca_{0.3}MnO₃). A percolation threshold (Φ_c) occurs when the La_{0.7}Ca_{0.3}MnO₃ volume fraction is ~0.19. The dependence of the electrical resistivity ρ as a function of La_{0.7}Ca_{0.3}MnO₃ volume fraction (f_{LCMO}) can be described by percolationlike phenomenological equations. Fitting the conducting regime ($f_{LCMO} > \Phi_c$) by the percolation power law $\rho \propto (f_{LCMO} - \Phi_c)^{-t}$ returns a critical exponent *t* value of 2.0±0.2 at room temperature and 2.6±0.2 at 5 K. The increase of *t* is ascribed to the influence of the grain boundaries on the electrical conduction process at low temperature.

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The colossal magnetoresistance¹ in $La_{1-v}A_{v}MnO_{3-d}$ compounds (A=Ca,Sr) is strongly influenced by the microstructure.^{1–3} The presence of grain boundaries in polycrystalline materials leads to a significant magnetoresistance at low fields and all temperatures below the paramagnetic/ ferromagnetic transition temperature T_c .⁴ Several authors have tried to increase the low-field magnetoresistance by mixing a CMR manganite phase with an insulating oxide secondary phase.^{5–16} When manganite-insulator composites are sintered at high temperature to achieve densification, ionic diffusion usually results in a pronounced shift of the manganite composition.⁵⁻¹⁰ As a result, it is not possible to analyze the electrical properties as a function of the volume percentage of each phase, since the composition of each phase varies significantly throughout the series of samples. In order to prevent interdiffusion phenomena, some authors have prepared composites using very short heat treatments, 12-16 so that the composition of the manganite phase was not affected. However, the short sintering treatments result in large porosity of the samples and poor connectivity between the manganite grains; it is difficult to take these effects properly into account when parametrizing the system for a description of the electrical conductivity behavior.

In a previous work,¹⁷ we have shown that well-sintered $La_{0.7}Ca_{0.3}MnO_3/Mn_3O_4$ composite samples can be obtained with almost constant manganite composition throughout the series. A one-step spray drying synthesis^{17,18} followed by a long thermal treatment at 1300 °C results in the formation of dense samples containing Mn_3O_4 and a $La_{0.7}Ca_{0.3}MnO_3$ (LCMO) manganite phase. The stoichiometry of the LCMO phase is not significantly affected by the Mn_3O_4 secondary phase, as proved¹⁷ by comparing the properties of the two end members of the composite series (cell parameters, Curie temperature, and saturation magnetization).

In the present work, the dependence of the electrical resistivity properties as a function of the $La_{0.7}Ca_{0.3}MnO_3$ volume fraction is analyzed in the framework of percolation theory. The microstructure of the samples was characterized by scanning electron microscopy (Philips FEG ESEM XL-30) of polished cross sections. The volume resistivity of the high-resistivity samples was measured with a Keithley 617 electrometer across the thickness of disk-shaped pellets using silver paste contacts attached to the circular faces. Surface leakage currents were found to be negligible with respect to volume current, whence no guard ring was used. The resistance of the low-resistivity samples was measured with a HP 34420 nanovoltmeter using the conventional four-point technique on thin bar-shaped samples cut from the pellets using a wire saw. In the intermediate resistivity regime ($10^2 < \rho$ $< 10^6 \Omega$ cm), both geometries were used and results were compared to each other. Electrical resistance *vs.* temperature R(T) curves were recorded in a Quantum Design physical properties measurement system (PPMS).

Scanning electron micrographs (Fig. 1) revealed three types of microstructures as the Mn_3O_4 content increases: (i) Mn_3O_4 islands in a LCMO matrix, (ii) a labyrinthic pattern of the two phases, and (iii) LCMO islands in a Mn_3O_4 matrix. In Mn_3O_4 -rich samples, a significant porosity was observed. Density measurements by the Archimedes method²¹ were carried out in order to calculate the volume fractions of each phase, i.e., the LCMO phase, the Mn_3O_4 phase and the total (open+closed) porosity. The dimensionless volume fractions are defined in such a way that $f_{LCMO}+f_{Mn_3O_4}+f_{porosity}=1$. In LCMO-rich samples ($f_{LCMO}>0.3$), $f_{porosity}$ does not exceed 0.10 whereas in Mn_3O_4 -rich samples



FIG. 1. Electron micrographs of polished cross sections of samples with decreasing LCMO volume fractions (a) 0.83, (b) 0.40, and (c) 0.18. Light and dark gray regions correspond to LCMO and Mn_3O_4 , respectively.



FIG. 2. Main graph: Temperature dependence of the zero-field resistivity of samples containing manganite volume fractions f_{LCMO} ranging from 0.18 to 0.92. T_p of each sample is indicated by an arrow. Inset: Same data presented in linear scale for $f_{LCMO} \sim 0.20$.

 $(f_{LCMO} < 0.3)$, $f_{porosity}$ lies in the 0.20–0.25 range. A systematic overestimation of the absolute f_{LCMO} values cannot be excluded with this method¹⁹ and is estimated to be smaller than 0.03. The important point is the repeatability of the f_{LCMO} determination, which was within the $[f \pm 0.002]$ interval.

The electrical resistance of the composite materials depends on the sample porosity and on the respective volume fractions and resistivities of the LCMO and Mn_3O_4 phases. Resistivity values of pure LCMO samples prepared using the synthesis technique described above are typically on the order of $10^{-2} \Omega$ cm (for 5 < T < 300 K), whereas pure Mn_3O_4 samples display electrical resistivity values on the order of $10^8 \Omega$ cm at room temperature and higher values for T < 300 K. Air pores are insulating. The composite materials can thus be considered as made up of a high-conductivity part (LCMO) and a low-conductivity part (Mn_3O_4 +pores). The results will be discussed in terms of the conducting phase volume fraction (f_{LCMO}) as the unique parameter characterizing the different composite materials.

The temperature dependences of the zero-field resistivity of samples with f_{LCMO} = 0.18, 0.20, 0.40, 0.80, and 0.92 are shown in Fig. 2 with the resistivity axis on a logarithmic scale. All samples with $f_{LCMO} > \sim 0.19$ present a resistivity transition from a semiconductinglike behavior above the transition peak temperature T_p to a metalliclike behavior below T_p . The T_p value of the sample with $f_{LCMO} \sim 0.20$ is only \sim 20 K lower than the value for the manganite-pure sample. This is *much* smaller than the corresponding T_p drops observed in other insulator-manganite composite series densified at high temperature.^{7,9,10,12} The difference between T_p and the Curie temperature T_C (determined from magnetization measurements¹⁷) never exceeds 5 K, as usually observed in well-sintered La_{0.7}Ca_{0.3}MnO₃.^{1,2} When f_{LCMO} decreases from 0.92 toward 0.20, a low-temperature bump progressively grows and eventually becomes higher than the T_p maximum in the case of the sample with $f_{LCMO} \sim 0.20$ (see inset of Fig. 2). Such a behavior is usually considered as a



FIG. 3. Electrical resistivity at 300 K as a function of the LCMO volume fraction f_{LCMO} . Inset: Electrical resistivity at 5 K for $f_{LCMO} < \sim 0.19$. The plain lines correspond to fits by different equations (see text for details).

grain-boundary-related contribution in manganite polycrystalline materials.^{2,20} On the contrary, the samples with $f_{LCMO} < \sim 0.19$ do not display any resistive transition in the temperature range where the electrical resistance could be accurately measured with the electrometer ($R < 100 \text{ G}\Omega$).

The resistivity data of the samples in the semiconductinglike state can be fitted by a simple law $\rho = \rho_{\infty} \exp(E_0/kT)$.¹ The E_0 values for the samples presenting a resistive transition are in the 90–140 meV range. For the high-resistivity samples with $f_{LCMO} < \sim 0.19$, E_0 increases to values lying between 400 and 500 meV.

Figure 3 shows the resistivity of the composite samples at 300 K as a function of the LCMO volume fraction f_{LCMO} . The resistivity axis is in logarithmic scale. A sharp increase of four orders of magnitude in resistivity occurs at f_{LCMO} ~ 0.19 (dashed line) and clearly establishes the existence of a percolation threshold. The resistivity vs. f_{LCMO} data at T =5 K is shown in the inset of Fig. 3. For $f_{LCMO} > \sim 0.19$, the behavior is similar to the 300 K data, with a sharp resistivity increase occurring at the same percolation threshold f_{LCMO} ~0.19. The resistivity values for $f_{LCMO} < \sim 0.19$ were too high to be measured accurately at low temperatures. Experimentally, resistivity values determined on samples with almost identical f_{LCMO} were found to agree within a factor of 2. Such variations are observed even in manganite-pure samples and probably result from a small error in the geometrical factor estimation (mainly due to the finite size of voltage contacts) and the imperfect reproducibility of the pressing and sintering steps. In the immediate vicinity of the percolation threshold $[f_{LCMO} \sim (0.18 \pm 0.01)]$, there is also a size effect of the contacts. In the case of the pellet measurement, the contact area is ~ten times larger and the distance between contacts is ~five times smaller than in the case of bar-shaped samples. Therefore the probability of finding a conducting filamentary path is significantly larger in the case of the pellet configuration. This may result in overall resistivity values several orders of magnitude smaller for the pellet than for the bar-shaped samples. The bar-shaped configuration was found to lead to much better reproducibility than the pellet configuration and the resistivity data shown in Fig. 3 for these samples correspond to the average of four-point resistivity measurements performed on three bars cut from each pellet.

The 0.19 experimental percolation threshold found for our LCMO/Mn₃O₄ composites is similar to the values calculated for regular periodic lattices: it was first observed by Scher and Zallen²¹ that calculations for all usual three-dimensional lattice types predict critical volume fractions Φ_c in the range 0.16 ± 0.02 . However, theoretical predictions for regular lattices do not necessarily apply to continuum media and we should also compare the 0.19 percolation threshold to literature data for other manganite-insulator composite systems with no or little interdiffusion between the phases. Balcells et al.¹³ report a percolation threshold $f_{LSMO} = 0.20$ for a La2/3Sr1/3MnO3/CeO2 composite series, in good agreement with the value obtained here. However, most papers report experimental percolation thresholds far from the Scher and Zallen range: f_{LCMO} between 0.92 and 0.85 for $La_{0.67}Ca_{0.33}MnO_3/Al_2O_3$,^{14,16} $f_{LCMO} \sim 0.60$ for $La_{0.67}Ca_{0.33}MnO_3/SrTiO_3$,¹² and $f_{LSMO} \sim 0.50-0.55$ for $La_{0.7}Sr_{0.3}MnO_3/borosilicate glass$.¹⁵ Several reasons may explain the dispersion of these experimental values. First, the significant porosity is sometimes not taken into account when calculating the volume fraction. Second, the poor connectivity between neighboring manganite grains is likely to affect the Φ_c values since some of the neighboring conducting grains fail to make contact.²² Third, the significant difference in grain size between the conducting and the insulating phases in Refs. 12 and 15 results in a highly nonuniform distribution of the two phases in the system. Such mechanisms can yield critical volume fractions far from the 0.16 theoretical prediction.²³

In view of the above, the rather good agreement between the experimental percolation threshold of our composites and the Scher and Zallen theoretical range for lattice percolation suggests that the one-step spray drying synthesis leads to more or less random "auto-organization" of the two phases in the samples. Our experimental percolation threshold is also an indirect indication of the good connectivity between the LCMO grains in the composites. This is further supported by the very small temperature difference (<5 K) between the electrical transition temperature T_p and the Curie temperature T_C in the studied samples. For these reasons, the La_{0.7}Ca_{0.3}MnO₃/Mn₃O₄ system appears as a good benchmark to study the applicability of percolation laws to manganite-insulator composite systems.

The high value ($\sim 10^{10}$ at T=300 K) of the resistivity ratio between the "insulating" (Mn₃O₄) and the "conducting" (LCMO) phases allows us to attempt to describe the electrical resistivity ρ of the composites by percolation laws.²⁴ In the conducting regime ($f_{LCMO} > \Phi_c$),

$$\rho = \rho_1 (f_{LCMO} - \Phi_C)^{-t}; \tag{1}$$

in the insulating regime $(f_{LCMO} < \Phi_c)$,

$$\rho = \rho_2 (\Phi_C - f_{LCMO})^s \tag{2}$$

where Φ_c is the critical volume fraction of the conducting phase, *t* and *s* are critical exponents, and ρ_1 and ρ_2 are adjustable parameters.

We have used Eqs. (1) and (2) to fit the resistivity data in the range $f_{LCMO} > 0.19$ and $f_{LCMO} < 0.19$, respectively. The best fits are shown as plain lines in Fig. 3 (300 K) and inset (5 K). The experimental trend in the whole f_{LCMO} range is reasonably described by percolationlike laws with Φ_c =0.191. Although percolation theory is supposed to apply to the vicinity of the percolation threshold, it is in fact frequent that percolation laws can model data obtained far from the percolation threshold.²³ The fitting returns critical exponent tvalues of 2.0 at 300 K and 2.6 at 5 K. The error on t was estimated by observing that a 0.2 variation of t decreases the fit quality markedly. The fit of the high-resistivity regime at 300 K yields a critical exponent s of 1.4, but this value cannot be compared to results for biphasic insulator-conductor composites because the "high-resistivity phase" is actually made up of a mixture of Mn_3O_4 and porosity: when f_{LCMO} $<\Phi_c$, charge carriers travel in both LCMO and Mn₃O₄ but not in air pores and it would be necessary to consider three phases.

In three-dimensional continuum composites, the conductivity exponent t is often found in the 1.6–2.0 range^{23,24} but larger values of t are also frequently reported,²⁴ so the t values found here at 300 K (i.e., 2.0±0.2) and 5 K (i.e., 2.6 ± 0.2) are not abnormally high. The perceptible difference between the two t values may result from the contribution of the grain boundaries to the resistivity at low temperatures, as explained hereafter. For $f_{LCMO} > \Phi_c$, the electrical current flows through a continuous LCMO path by crossing the grain boundaries between neighboring LCMO grains. Above T_C (e.g., at 300 K), the LCMO phase is paramagnetic and Mn spins are disordered, both in the grains and at the grain boundaries. Below T_C (e.g., at 5 K), the LCMO phase is ferromagnetic and the main contribution to the electrical resistance results from the grain boundaries, where the parallel alignment of the Mn spins is disrupted.^{1,2} In the present work, the low-temperature bump in the $\rho(T)$ curve becomes more pronounced when f_{LCMO} approaches Φ_c , indicating an increasing contribution of the grain boundaries^{2,20} to the conduction process as f_{LCMO} is reduced from 1 to 0.19. As a result, below T_C , the "effective resistivity" of the LCMO phase cannot be considered as constant in the whole f_{LCMO} $>\Phi_c$ range. This increase of the LCMO effective resistivity reasonably explains the observed increase of the conductivity exponent t at 5 K with respect to the 300 K value.

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- ¹J. M. D. Coey, M. Viret, and S. von Molnar, Adv. Phys. **48**, 167 (1999).
- ²B. Vertruyen, R. Cloots, A. Rulmont, G. Dhalenne, M. Ausloos, and Ph. Vanderbemden, J. Appl. Phys. **90**, 5692 (2001).
- ³Ph. Vanderbemden, B. Vertruyen, A. Rulmont, R. Cloots, G. Dhalenne, and M. Ausloos, Phys. Rev. B **68**, 224418 (2003).
- ⁴M. Ziese, Rep. Prog. Phys. 65, 143 (2002).
- ⁵D. Das, A. Saha, S. E. Russek, R. Raj, and D. Bahadur, J. Appl. Phys. **93**, 8301 (2003).
- ⁶D. Das, P. Chowdhury, R. N. Das, C. M. Srivastava, A. K. Nigam, and D. J. Bahadur, J. Magn. Magn. Mater. **238**, 178 (2002).
- ⁷Z. C. Xia, S. L. Yuan, W. Feng, L. J. Zhang, G. H. Zhang, J. Tang, L. Liu, D. W. Liu, Q. H. Zheng, L. Chen, Z. H. Fang, S. Liu, and C. Q. Tang, Solid State Commun. **127**, 567 (2003).
- ⁸D. Das, C. M. Srivastava, D. Bahadur, A. K. Nigam, and S. K. Malik, J. Phys.: Condens. Matter 16, 4089 (2004).
- ⁹P. Kameli, H. Salamati, M. Eshraghi, and M. R. Mohammadizadeh, J. Appl. Phys. **98**, 043908 (2005).
- ¹⁰B. X. Huang, Y. H. Liu, R. Z. Zhang, X. Yuan, C. J. Wang, and L. M. Mei, J. Phys. D **36**, 1923 (2003).
- ¹¹O. A. Shlyakhtin, K. H. Shin, and Y. J. Oh, J. Appl. Phys. **91**, 7403 (2002).

- ¹²D. K. Petrov, L. Krusin-Elbaum, J. Z. Sun, C. Feild, and P. R. Duncombe, Appl. Phys. Lett. **75**, 995 (1999).
- ¹³L. Balcells, A. E. Carrillo, B. Martinez, and J. Fontcuberta, Appl. Phys. Lett. **74**, 4014 (1999).
- ¹⁴L. E. Hueso, J. Rivas, F. Rivadulla, and M. A. Lopez-Quintela, J. Appl. Phys. **89**, 1746 (2001).
- ¹⁵A. Gupta, R. Ranijt, C. Mitra, P. Raychaudhuri, and R. Pinto, Appl. Phys. Lett. **78**, 362 (2001).
- ¹⁶L. E. Hueso, J. Rivas, F. Rivadulla, and M. A. Lopez-Quintela, J. Non-Cryst. Solids **287**, 324 (2001).
- ¹⁷B. Vertruyen, J.-F. Fagnard, Ph. Vanderbemden, M. Ausloos, A. Rulmont, and R. Cloots, J. Eur. Ceram. Soc. (to be published).
- ¹⁸B. Vertruyen, A. Rulmont, R. Cloots, J.-F. Fagnard, M. Ausloos, I. Vandriessche, and S. Hoste, J. Mater. Sci. **40**, 117 (2005).
- ¹⁹M. W. Barsoum, *Fundamentals of Ceramics* (Institute of Physics, Bristol, 2003), p. 79.
- ²⁰J. E. Evetts, M. G. Blamire, N. D. Mathur, S. P. Isaac, B. S. Teo, L. F. Cohen, and J. L. MacManus-Driscoll, Philos. Trans. R. Soc. London, Ser. A **356**, 1593 (1998).
- ²¹H. Scher and R. Zallen, J. Chem. Phys. **20**, 325 (1970).
- ²²A. de Andres, M. Garcia-Hernandez, and J. L. Martinez, Phys. Rev. B **60**, 7328 (1999).
- ²³D. S. McLachlan, M. Blaszkiewicz, and R. E. Newnham, J. Am. Ceram. Soc. **73**, 2187 (1990).
- ²⁴C.-W. Nan, Prog. Mater. Sci. **37**, 1 (1993).