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## Catabolism of hydroxyacids and biotechnological production of lactones by *Yarrowia lipolytica*

Received: 26 August 2002 / Revised: 21 November 2002 / Accepted: 22 November 2002 / Published online: 28 January 2003  
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**Abstract** The  $\gamma$ - and  $\delta$ -lactones of less than 12 carbons constitute a group of compounds of great interest to the flavour industry. It is possible to produce some of these lactones through biotechnology. For instance,  $\gamma$ -decalactone can be obtained by biotransformation of methyl ricinoleate. Among the organisms used for this bioproduction, *Yarrowia lipolytica* is a yeast of choice. It is well adapted to growth on hydrophobic substrates, thanks to its efficient and numerous lipases, cytochrome P450, acyl-CoA oxidases and its ability to produce biosurfactants. Furthermore, genetic tools have been developed for its study. This review deals with the production of lactones by *Y. lipolytica* with special emphasis on the biotransformation of methyl ricinoleate to  $\gamma$ -decalactone. When appropriate, information from the lipid metabolism of other yeast species is presented.

### Introduction

The possibility of producing a lactone in a biotechnological way was discovered in the 1960s when the group of Okui (Okui et al. 1963a, 1963b, 1963c; Uchimaya et al. 1963) was studying the catabolism of hydroxyacids in various organisms. Although the animal cells and bacteria studied degraded ricinoleic acid to C16, C14 and C12 hydroxy acids, the yeast *Candida tropicalis* accumulated  $\gamma$ -decalactone, a lactone exhibiting fruity and oily notes important in the formulation of peach, apricot or strawberry aromas.

Since this observation, biotechnologists have devoted their efforts to the selection of yeast strains able to produce high amounts of lactones.

The yeast *Yarrowia lipolytica*, which is particularly adapted to hydrophobic environments (Barth and Gailardin 1996, 1997), has been successfully tested. The fact that there are genetic tools adapted to the study of this organism has helped this species to become a reference in research dealing with non-polar substrate metabolism. In the field of lipid-derived aroma compounds, the importance of *Y. lipolytica* is also high, as acknowledged by the many patents and papers mentioning it.

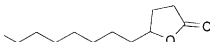
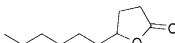
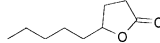

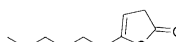

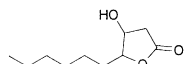
$\gamma$ -Decalactone is one of the most produced lactones. Its production through biotechnology results in a natural lactone. The annual market for  $\gamma$ -decalactone has been growing in recent years and was estimated in 1997 at about 10 t (Gatfield 1999). This high production resulted in a sharp decrease in prices: from U.S. \$ 12,000/kg in 1986, the price declined to U.S. \$ 500/kg in 1998 (Gatfield 1999). As a consequence, companies who just required a simple production process some years ago are getting more and more interested in lowering the manufacturing costs. As a result, research on this subject, which was mainly devoted to the screening of strains or conditions, is now getting closer to the biological mechanisms of the catabolism of hydroxyacids to such an extent that this very applied field now generates a non-negligible amount of basic results concerning yeast lipid metabolism.

In this Mini-Review, we present the production of lactones for which *Y. lipolytica* is employed; and the biological pathway to obtain  $\gamma$ -decalactone is described in detail. In the first part, processes are presented and the metabolic pathways for  $\gamma$ -decalactone production from methyl ricinoleate (or ricinoleic acid) are described. Then, some current fields of investigation are presented, such as the optimisation of production through the study of  $\beta$ -oxidation fluxes, better knowledge of lactone toxicity towards the producing yeast and preliminary results regarding the yeast surface properties to facilitate substrate uptake.

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**Table 1** Structure, name and odorous properties of the lactones that can be produced by the yeast *Yarrowia lipolytica*

Structure	Name	Odorant notes
	$\gamma$ -dodecalactone	peach, butter, fatty
	$\gamma$ -decalactone	peach, fatty, fruity
	$\gamma$ -nonalactone	coconut, fatty, fruity, aniseed
	$\delta$ -decalactone	peach, oily, creamy
	dec-3-en-4-olide	fruity, oily, fatty
	dec-2-en-4-olide	mushroom
	3-hydroxy- $\gamma$ -decalactone	no odour

## Production of lactones by biotransformation

$\gamma$ - and  $\delta$ -lactones (4- and 5-alkanolides) with odorous properties are naturally present in fruit and in some fermented food products (Tang and Jennings 1968; Maga 1976). These aroma compounds possess varied sensory properties with mainly fruity and fatty characteristics (Table 1), which make them interesting food additives. The flavour of lactones is influenced by the lactone cycle structure, the length of the lateral carbon chain, the presence of unsaturations and chirality (Maga 1976; Dufossé et al. 1993). Some fungi (Collins and Halim 1972; Sarris and Latrasse 1985; Berger et al. 1986) and some yeast species (Tahara et al. 1972) were first identified for their ability to produce small amounts of aroma compounds by de novo synthesis. As their extraction from fruits was not financially viable, this production by micro-organisms was progressively optimised, giving rise to industrial applications that enabled researchers to obtain lactones with a "natural" label at economically viable costs. The patents mentioning *Y. lipolytica* for the production of lactones are presented in Table 2. For instance  $\gamma$ -decalactone, which is currently one of the main products derived from aroma biotechnology, is generally obtained by biotransformation of a

**Table 2** Processes of production of lactones mentioning *Y. (Candida) lipolytica*

Species or strain	Substrate	Products	Yield	Description	Reference (and organisation)
<i>Y. lipolytica</i> , <i>Aspergillus orizae</i> , <i>G. klebahnii</i> and many <i>Candida</i> spp	Castor oil, castor oil + lipase, castor oil hydrolysate	$\gamma$ -Decalactone	0.69 g/l in up to 1 week	Agitated flasks	Farbood and Willis (1983) (F. Dodge and Olcott)
<i>Candida</i> spp	Castor oil, ricinoleic acid	Unsaturated lactones	High aeration and agitation		Farbood et al. (1990) (IFF)
Many yeast species, including <i>Candida</i> , <i>Pichia</i> and <i>Sporidiobolus</i> spp	Hydroxide or hydroperoxide of C18 acids	$\gamma$ -Dodecalactone, unsaturated lactones, $\delta$ -lactones		Utilisation of soy bean lipoxygenase	Cardillo et al. (1991a) (Pernod Ricard)
<i>Y. lipolytica</i> , <i>Aspergillus orizae</i> , <i>G. klebahnii</i> and <i>H. saturnus</i>	Ricinoleic esters	$\gamma$ -Decalactone	5.1 g/l after 70 h	Lactonisation at 60–120 °C and pH 2	Meyer (1993) (BASF)
<i>Yarrowia</i> and <i>Candida</i> spp	10-OH-stearic acid, Oleic acid	$\gamma$ -Dodecalactone, unsaturated lactones	3.5 g/l	Hydroxylated ( <i>Pseudomonas</i> spp) C18 fatty acids	Farbood et al. (1994) (IFF)
<i>Y. lipolytica</i> PO1d ( <i>ura3</i> auxotrophic strain derived from ATCC 24060)	Methyl ricinoleate	$\gamma$ -Decalactone	9.5 g/l after 75 h	First step: biomass production; second step: high density reactor with a non proliferating strain	Nicaud et al. (1996) (CNRS)
<i>Sporobolomyces</i> and <i>Yarrowia</i>	Residue derived from the distillation of fermentation products (cognac)	$\gamma$ -Decalactone and other products		Methyl ricinoleate is added as the precursor	Ambid et al.(1999)
<i>Y. lipolytica</i> HR145 (DSM12397)	Castor oil	$\gamma$ -Decalactone, hydroxylactone	11 g/l in less than 70 h	One-step reaction	Rabenhorst and Gatfield (2000) (Haarmann and Reimer)
<i>Saccharomyces</i> and <i>Candida</i> spp	5-Hydroxydecanoic acid	$\gamma$ -Nonalactone		$\alpha$ -Oxidation	Lange and Garbe (2000)

long-chain hydroxy fatty acid precursor by yeast cells. Ricinoleic acid is a convenient substrate since this hydroxy acid represents almost 90% of hydrolysed castor oil. *Y. lipolytica* is one of the yeast species that are able to produce  $\gamma$ -decalactone from ricinoleic acid or its methyl ester. It was initially mentioned in a process patented by Farbood and Willis (1983) and its use for lactones production has been patented several more times, concerning the production of  $\gamma$ -decalactone (Cardillo et al. 1991a, 1991b; Meyer 1993; Ambid et al. 1999; Rabenhorst and Gatfield 2000). As for the utilisation of other yeast species, the processes often lead to the production of the direct precursor of  $\gamma$ -decalactone, i.e. 4-hydroxydecanoic acid, which is then spontaneously lactonised under acid conditions (Meyer 1993; Dufossé 1993). All these processes report the production of lactones by *Y. lipolytica* at concentrations of several grams per litre; and examples of 9.5 g/l after 75 h (Nicaud et al. 1996) and 12 g/l (Rabenhorst and Gatfield 2000) were given. In the case of Nicaud et al. (1996), an uracil auxotrophic strain (*ura3*; the W29 (ATCC 24060) derived strain PO1d) was transferred at high cell concentrations in a medium containing poor amounts of uracil, which only allowed a very weak yeast proliferation. *Y. lipolytica* was also reported in patents dealing with the production of some other lactone aroma compounds:  $\gamma$ -dodecalactone (Cardillo et al. 1991a, 1991b; Farbood et al. 1994),  $\gamma$ -nonalactone from 5-hydroxydecanoic acid (Lange and Garbe 2000), unsaturated lactones (Farbood et al. 1990; Cardillo et al. 1991a, 1991b; Farbood et al. 1994) and  $\delta$ -lactones (Cardillo et al. 1991a, 1991b).

*Y. lipolytica* has also been described for its ability to produce high amounts of some other lactones during the biotransformation of methyl ricinoleate into  $\gamma$ -decalactone: 3-hydroxy- $\gamma$ -decalactone, dec-2-en-4-olide and dec-3-en-4-olide (Gatfield et al. 1993). These three compounds, which are derived from 4-hydroxydecanoic acid (or its CoA ester; Waché et al. 2001), are currently not used by the aroma industry: in fact, the dec-3-en-4-olide has been described as possessing a fruity, peach-like odour, more powerful than that of  $\gamma$ -decalactone, but its potential utilisation would depend on a cheap method to separate it from its isomer (dec-2-en-4-olide), which possesses a mushroom-like note.

The lactone concentrations obtained during the processes were achieved through optimisation of the yeast culture conditions, leading very often to biotransformation yields that were rather low. Further improvements were expected to be reached from the knowledge of the metabolic pathways leading to the bioproduction of the lactones; and these have been progressively elucidated.

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### Biological pathway

We focus here on the biological pathway from methyl ricinoleate to  $\gamma$ -decalactone, the main lactone produced by *Y. lipolytica*.

Although the implication of  $\beta$ -oxidation in this biotransformation pathway has been hypothesised from the first study (Okui et al. 1963b), many obscure points remain, corresponding to subjects of yeast lipid metabolism, which has been less studied. In particular, the entry of the substrate into the peroxisome, some aspects of the  $\beta$ -oxidation reactions (especially those involving auxiliary enzymes and the organisation of the fluxes) and the lactonisation/way-out for the product are not well known. The various studies on the subject have been carried out in different yeast species, each enabling more or less important steps forward in the understanding of the precise pathway. When appropriate, we present in this part data from studies dealing with yeasts other than *Y. lipolytica*.

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### Substrate entry into the cell

First in this process, the yeast cell is in the presence of the substrate, which can be castor oil, ricinoleic acid or, more often, methyl ricinoleate. Except in the case of the free fatty acid, the substrate has thus to be hydrolysed, a step which seems to take place in the extracellular medium. It can be mentioned here that most of the lactone-producing yeasts, including *Y. lipolytica* (Pignède et al. 2000a, 2000b), possess active extracellular lipases.

The second step concerns the adsorption of the fatty acid to the yeast surface and its entry into the cell. This step is rather unknown, although interactions of fatty acids with cell walls are currently being investigated and will be discussed later. Different mechanisms of entry have been described, mostly in bacteria, involving hydrophobic interactions or surfactant-mediated phenomena. They result in a mechanism of direct interfacial or surfactant-facilitated transport (Bouchez-Naitali et al. 1999, 2001). Some protrusion-forming structures resembling canals have been observed during the growth of *C. tropicalis* in alkane (Osumi et al. 1975; Meisel et al. 1976, 1977; Tanaka and Fukui 1989). They could constitute the transport structure but, to date, no demonstration has been proposed to explain the cell wall passage. However, the cell wall is not required in the lipid metabolism, since spheroplasts of *C. tropicalis* are able to metabolise oleic acid (Medvedeva et al. 1969) and those of *Y. lipolytica* are able to utilise methyl ricinoleate and to produce  $\gamma$ -decalactone (Aguedo et al. 2000).

The passage of fatty acids through the plasma membrane is, in all organisms, highly controversial. In yeast, investigations carried out by Kohlwein and Paltauf (1983) in *Saccharomyces cerevisiae* and *Saccharomycopsis lipolytica* (syn. *Y. lipolytica*) with lauric or oleic acid concluded that, beyond a threshold of 10  $\mu$ M, an energy-free transporter was required whereas, above this concentration, the fatty acids diffused freely.

## Cytoplasmic fate

Inside the cell, fatty acids can be either stocked or utilised. Fatty acids have been shown to accumulate in different structures, described as vacuoles or spherosomes (Gill et al. 1977; Brennan and Lösel 1978). These structures seem to be in contact with some organelles favouring the accession of fatty acids to their degradation site (Brennan and Lösel 1978). Feron et al. (1997) observed in *Sporidiobolus salmonicolor* that, in contrast with oleic acid, ricinoleic acid accumulated inside the cells. Moreover, ricinoleic acid was metabolised more slowly than oleic acid, although the cell density was significantly higher. They explained this difference by the accumulation of  $\gamma$ -decalactone during metabolism of the hydroxy acid. This product would be toxic (decreasing metabolism) through its capacity for permeabilising membranes (thus increasing the accumulation of ricinoleic acid inside the cell). The accumulation of methyl ricinoleate in *Y. lipolytica* has been investigated and the results suggest that this accumulation is closely dependent on the physiological state of the cells (Aguedo 2000; Aguedo et al., unpublished results).

To be catabolised, fatty acids have to be activated in their coenzyme A ester forms. The fatty acid transport within the cytoplasm to the right enzyme has not been elucidated and the rare works dealing with fatty acid-binding proteins (FABP) in yeasts have not been successful (Scholz et al. 1990; Smaczynska et al. 1994), apart from those of Dell'Angelica et al. (1992), who identified a FABP in *Y. lipolytica*. The activation is catalysed by an acyl-CoA synthetase (ACS). In contrast to FABP, ACSs have been observed in yeast cells for years: the acyl-CoA synthetase 1 of *C. lipolytica* was characterised in 1979 (Hosaka et al. 1979). These enzymes are localised in organelles. In *Pichia pastoris*, the very-long-chain-specific ACS is peroxisomal and the long-chain-specific one is mitochondrial (Kalish et al. 1995) whereas, in *Saccharomyces cerevisiae*, there are two peroxisomal ACS, one long-chain-specific and the other very-long-chain-specific, and both can activate C18 acids (Watkins et al. 1998).

Acyl-CoA can be transported by acyl-CoA-binding proteins. These proteins, initially discovered as an impurity in a FABP preparation (Mogensen et al. 1987), constitute a family that is well conserved over all eukaryotic species (Knudsen et al. 1999). In yeast, overexpression experiments have shown its role in acyl-CoA pool formation (Mandrup et al. 1993; Knudsen et al. 1994); and gene disruption strongly suggests a role in transport (Schjerling et al. 1996).

## Transport to the peroxisomes

Recent results (Dansen et al. 2000) and older works (Nicolay et al. 1987) agree that the peroxisome internal pH is different from the cytoplasmic one. This ATPase-maintained intraperoxisomal pH could contribute to the generation of a proton motive force (Douma et al. 1987),

which could play a role in transports. The peroxisomal membrane has been shown in vitro to be permeable to small compounds (such as sucrose, NAD<sup>+</sup>, CoA, ATP, carnitine) and this could be due to the presence of pore-forming proteins (Van Veldhoven et al. 1987). However, more recent works suggest that, in vivo, the membrane is rather impermeable to NAD(H) and acetyl-CoA (Van Roermund et al. 1995). ATP for its part has been shown to enter the peroxisomal lumen through the action of a transporter (Palmieri et al. 2001), in the absence of which medium-chain (but not long-chain) fatty acids could not be activated into acyl-CoA (Van Roermund et al. 2001).

## Peroxisomal $\beta$ -oxidation

The implication of  $\beta$ -oxidation has been deduced from the accumulation of metabolic intermediates. The compounds detected during the pathway were first those accumulating between two  $\beta$ -oxidation loops (Okui et al. 1963a), excluding the 4-hydroxydecanoic acid which was probably lactonising spontaneously before detection, as reported for other 4-hydroxy acids (Fantin et al. 2001). This very unstable acid was later detected (Feron et al. 1996) and the detection of other intermediates was improved with the use of biphasic media (Spinnler et al. 1996) or by processing the pathway in vitro (Blin-Perrin et al. 2000).

Apart from rare and older studies mentioning acyl-CoA dehydrogenase activities (Haffner and Tressl 1996) or other characteristics that are specific to mitochondrial  $\beta$ -oxidation (Blin-Perrin et al. 2000), this pathway is, in yeast, generally considered as being peroxisomal.  $\beta$ -Oxidation is a four-reaction sequence resulting in a two-carbon chain-shortening catalysed in peroxisomes by the following activities: acyl-CoA oxidase, 2-enoyl-CoA hydratase, 3-hydroxyacyl-CoA dehydrogenase and 3-ketoacyl-CoA thiolase (Fig. 1). The commonly accepted pathway from ricinoleyl-CoA to  $\gamma$ -decalactone is presented in Fig. 2: four  $\beta$ -oxidation cycles occur, yielding 4-hydroxy-decanoyl-CoA, which is then, after some still unknown steps, cyclised to  $\gamma$ -decalactone.

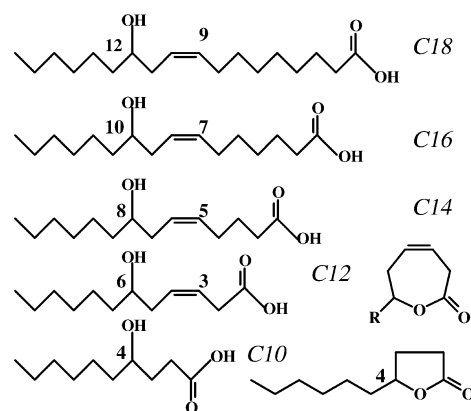
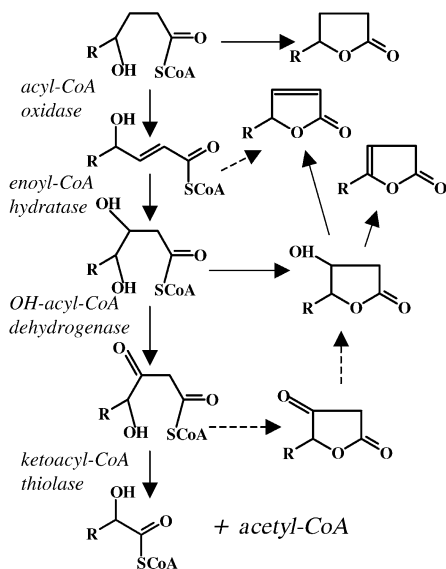


Fig. 1 Intermediates detected during the degradation of ricinoleic acid. R: C<sub>6</sub>H<sub>13</sub>





**Fig. 2** The four steps of  $\beta$ -oxidation and the catalysing enzymes involved. The substrate example given for this  $\beta$ -oxidation loop is 4-hydroxydecanoyl-CoA, the C10 intermediate of the  $\beta$ -oxidation of ricinoleyl-CoA. The lactones potentially formed at each step are shown.  $R$ :  $C_6H_{13}$

In *Y. lipolytica*, works dealing with the enzymes of  $\beta$ -oxidation have been carried out in our laboratories since the mid-1990s. This species possesses a family of five acyl-CoA oxidases (Aox1–5, encoded by *POX1*–5). Interest in this enzyme family came from our first experiments: disruption of *pox1* resulted in an increased  $\beta$ -oxidation activity but a decreased production of  $\gamma$ -decalactone, whereas overexpression of the thiolase did not modify the production (Pagot et al. 1998). Since then, the Aox family has been better characterised, using functional genetics (Waché et al. 1998; Wang et al. 1998; 1999a, 1999b, 2000). This enabled us to show that two Aox exhibited a high activity and a chain-length specificity, one being long-chain-specific (Aox2) and the other short-chain-specific (Aox3). The role of the other Aox was less evident, as Aox4 and Aox5 exhibited a weak activity on the whole spectrum of straight-chain acyl-CoA (from C4 to C18) and Aox1 did not exhibit any detectable activity. The disruption of the genes corresponding to these three Aox resulted in a 2- to 5-fold increase in the global Aox activity, suggesting a role in the regulation of their activity. The substrate specificity of Aox2 and Aox3 was later confirmed with the purified enzymes (Luo et al. 2000). Furthermore, Titorenko et al. (2002) showed that the global acyl-CoA oxidase was targeted to the peroxisomes as a heteropentamer containing each Aox.

Our investigations into lactone production by *POX* mutants suggested strongly that the short-chain-specific Aox was involved in the continuation of the  $\beta$ -oxidation flux beyond the decanoyl-CoA (the  $\gamma$ -decalactone precursor) level and/or in the degradation of newly synthesised lactone (Waché et al. 1998, 2000a).

What happens to the double-bond of ricinoleic acid during its  $\beta$ -oxidation is not surely known in *Y. lipolytica*. Data concerning this metabolism by the auxiliary enzymes of  $\beta$ -oxidation have been obtained from other yeasts. From the enzymes present in *S. cerevisiae*, Gurvitz et al. (1999) proposed three different possible pathways for enoyl-CoA possessing odd-double-bonds, an isomerase-dependent one (observed in the degradation of oleic acid) and two postulated di-isomerase pathways, including one involving a reductase activity (Fig. 3). This reductase pathway was observed in the degradation of ricinoleic acid by *Sporobolomyces odorus* (Haffner and Tressl 1996), whereas a 3-*cis*, 5-*cis* intermediate was isolated by Iacazio et al. (Iacazio et al. 2002) during the degradation of ricinoleic acid by *P. guilliermondii*. This suggests that at least four pathways are possible to resolve the odd-double-bond problem and that there is a certain diversity between yeast species. However, the observation by Gatfield et al. (1993) of dodec-3-en-6-olide accumulation during the degradation of ricinoleic acid by *Y. lipolytica* suggests that the pathway in this species goes through a dodec-3-enoyl-CoA intermediate, a compound that is present in the isomerase-dependent pathway.

Another problem concerns the stereochemistry of the carbon carrying the hydroxy group of ricinoleic acid ( $R$ ) which later becomes the asymmetric carbon of the lactone. In *Y. lipolytica*, the resulting lactone has been described as being 98% ( $R$ ) (Gatfield 1999), which corresponds to the results obtained with other yeasts, such as *Sporobolomyces* (Haffner and Tressl 1996; Dufossé et al. 1997). We can however mention that the stereochemistry of the lactone-generating reactions is an intrinsic parameter for each species. Ercoli et al. (1992) thus observed, when growing *Y. lipolytica* or *P. ohmeri* on the same racemic hydroxy acids, that the two species produced the opposite enantiomers.

The yeast *Y. lipolytica* accumulates other C10-lactones in the medium (Gatfield et al. 1993; Waché et al. 2001; Fig. 1). This point is relatively original, since the presence of these lactones has only rarely been reported with other yeasts (Farbood et al. 1990). The accumulation of dec-2-en-4-olide, dec-3-en-4-olide and 3-hydroxy- $\gamma$ -decalactone in *Y. lipolytica* culture media was first reported by the company Haarmann and Reimer in 1993 (Gatfield et al. 1993). As presented in Fig. 2, these lactones seem to appear following lactonisation of  $\beta$ -oxidation intermediates at the various steps of the C10- $\beta$ -oxidation cycle. Their structure is closely related to that of  $\gamma$ -decalactone, but their sensorial properties are quite different, as mentioned above.

The role of carnitine in yeast lipid metabolism is not completely established. Carnitine seems to be involved in the utilisation of acetyl units after peroxisomal  $\beta$ -oxidation. In *C. lipolytica* and *Saccharomyces cerevisiae*, there are a mitochondrial and a peroxisomal carnitine acyl-transferase (CAT) which are both encoded by the same gene, which possesses two different targeting sequences: an internal peroxisomal one and a C-terminal mitochondrial one (Elgersma et al. 1995). Although the role of



after delactonisation, an  $\alpha,\omega$ -dicarboxylic acid. The production of such diacids by cells with defects in  $\beta$ -oxidation has already been described (Picataggio et al. 1992; Fabritius et al. 1998). This topic is of great interest to biotechnologists, as it enables the production of macrocyclic ketones of interest to the fragrance industry. However, with cells exhibiting an intact  $\beta$ -oxidation, this reaction can occur on the dicarboxylic acid, degrading it. Although Endrizzi-Joran (1994) did not see any differences in the degradation between uninduced and  $\beta$ -oxidation-induced cells, the involvement of  $\beta$ -oxidation in the degradation is highly suggested by our results with acyl-CoA oxidase-modified mutants (Waché et al. 2001): the mutant exhibiting no more  $\beta$ -oxidation is the only one unable to degrade  $\gamma$ -decalactone.

To avoid lactone degradation, some processes have proposed subtracting the lactone from the catabolism of yeast cells. This is presented later.

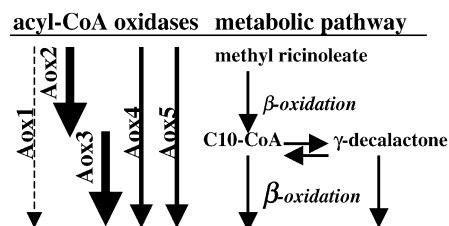
## Current developments

### Optimising the biotransformation

The main topic in the strategies to improve the  $\gamma$ -decalactone production yields concerns the  $\beta$ -oxidation pathway (Pagot et al. 1998; Waché et al. 1998, 2000a, 2001; Blin-Perrin et al. 2000).

As mentioned above, we modified the  $\beta$ -oxidation genotype in the yeast *Y. lipolytica*. Although the overexpression of the gene coding for thiolase did not modify  $\gamma$ -decalactone production (Pagot et al. 1998),  $\beta$ -oxidation enzymes appeared to be of the utmost importance in the biotransformation.

Two main problems have to be solved. First,  $\beta$ -oxidation can go on after the C10-level or can occur on  $\gamma$ -decalactone itself (Fig. 4). To resolve this point, works have been carried out to lower  $\beta$ -oxidation on the short-chain acyl-CoA (shorter than C10). In *Y. lipolytica*, the disruption of the short-chain-specific acyl-CoA oxidase (Aox3)-encoding gene (*pox3*) decreases lactone degradation during culture on C18 methyl ricinoleate substrate (Waché et al. 2000a, 2001). However, this disruption does not modify the degradation of C10 lactone substrate. This can be explained by the presence of two other non-chain-length-specific acyl-CoA oxidases (Aox4, Aox5). By disrupting the corresponding genes (*pox4*, *pox5*), one takes the risk of constructing a strain which is not very active, as the only active Aox remaining would be Aox2. We therefore constructed a strain disrupted for *pox2*, *pox3* and *pox5* (which still possesses *POX4*, encoding a weakly active Aox) and with *POX2* reincorporated in multicopies (Waché et al. 2002). The metabolism of this strain was slow, but no lactone degradation was observed. Since the Aox enzymes are targeted to peroxisomes as a heteropentamer possessing each Aox (Titorenko et al. 2002), the challenge would be to construct a strain possessing only the long-chain-specific Aox (Aox2) with good targeting.



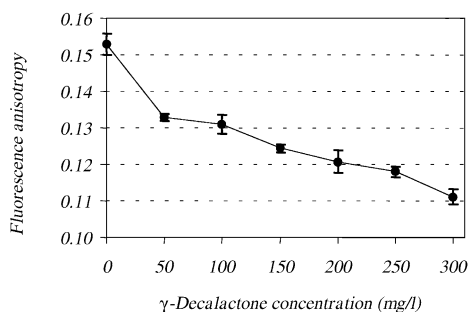
**Fig. 4** Involvement of the five acyl-CoA oxidases (Aox1–5) of *Y. lipolytica* in the  $\beta$ -oxidation of ricinoleic acid and the degradation of  $\gamma$ -decalactone

The second problem is the modification of  $\beta$ -oxidation fluxes which can enable the switch between production of  $\gamma$ -decalactone and production of hydroxy-related lactones (Fig. 1). The yeast *Y. lipolytica* seems to have a low activity at the multifunctional enzyme level and this results in the accumulation of the three hydroxy-related lactones. It is therefore possible to decrease the Aox activity so that the first  $\beta$ -oxidation enzyme could increase its control on the pathway and  $\gamma$ -decalactone could accumulate more. This can be achieved with  $\Delta$ *pox2* $\Delta$ *pox3* mutants, as the corresponding Aox2 Aox3 enzymes are the more active ones and are required in targeting the global acyl-CoA oxidase to the peroxisome (Titorenko et al. 2002). With such mutants, the production of hydroxylactone is close to zero, whereas that of  $\gamma$ -decalactone goes on all along the culture (Waché et al. 2001).

It is also possible to modify the environmental conditions to change  $\beta$ -oxidation fluxes. For this goal, a recent study utilised the reducing agent dithiothreitol with a *Sporidiobolus* strain. Although the effect of this compound on the cell was not investigated, it resulted in a modification of  $\gamma$ -decalactone yield (Wang et al. 2000). Other environmental parameters can modify the biotransformation, probably by perturbing the equilibrium of  $\beta$ -oxidation fluxes. For instance, we investigated the influence of the biotransformation medium pH. This can strongly influence lactone production by *Y. lipolytica*. It was observed that the  $\gamma$ -decalactone concentration in a medium containing phosphate buffer at pH 7 was 3-fold higher than that obtained in a non-buffered medium (Fig. 5). Gatfield et al. (1993) noted also a strong impact of the agitation parameter on the production of the various lactones.

### Lactone toxicity

The high lactone concentrations (up to several grams per litre) produced by yeast cells may also be a limiting factor in industrial applications, since the metabolites become toxic towards the producing yeast (Feron et al. 1997). The decline in cell viability has been clearly associated with the increase in lactone concentration within culture media during biotransformation (Feron et al. 1997; Dufossé et al. 1999). Moreover, the antimicrobial properties of some

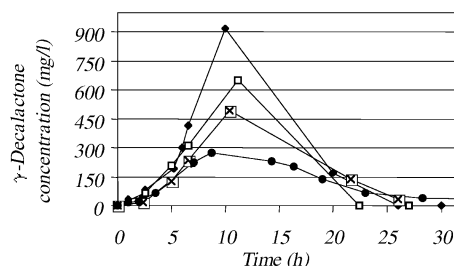


**Fig. 5** Fluorescence anisotropy of *Y. lipolytica* cells determined in situ with the fluorescent probe 1,6-diphenyl-1,3,5-hexatriene, at 27 °C in the presence of  $\gamma$ -decalactone (adapted with permission from Aguedo et al. 2002a)

other lactones have been reported. For example, 2-deceno- $\delta$ -lactone, at a concentration of 100 mg/l inhibits the growth of bacteria (Nago et al. 1993). 6-Pentyl- $\alpha$ -pyrone is toxic towards a producing strain of *Trichoderma viride* (Bonnamme et al. 1997); and  $\gamma$ -decalactone was reported to inhibit the growth of the producing yeast *S. salmonicolor* (Feron et al. 1996). Generally, microorganisms and animal cells (Adams et al. 1998) are able to withstand and, in many cases, to metabolise lactones up to a strain-dependent concentration threshold, beyond which these compounds become toxic (Feron et al. 1996).

Some mechanisms underlying lactone toxicity have been studied in our laboratory, using the yeast *Y. lipolytica* as a model. The growth of this yeast in a glucose-containing medium is inhibited by  $\gamma$ -decalactone at concentrations higher than 150 mg/l. Such concentrations lead to a dissipation of membrane potential, as a consequence of the interaction of hydrophobic lactone with cell membranes. Infrared spectroscopic measurements revealed that increasing concentrations of the compound lowered the phase transition temperature of dimyristoylphosphatidylcholine films (Aguedo et al. 2002a) and in vivo fluorescence anisotropy measurements showed that yeast membrane fluidity is strongly increased (Fig. 6; Aguedo et al. 2002a). Interestingly, the data indicated also that, even for non-toxic concentrations of  $\gamma$ -decalactone, the membrane properties of *Y. lipolytica* were modified, due to the diffusion of lactone inside the phospholipid bilayers.

Some strategies have been developed to reduce lactone toxicity within the fermentation medium, as for example the addition into the culture media of inert oils (hydrogenated coconut oil or a mixture of tripalmitine, tristearine, triolein) or hydrophobic porous sorbents (Dufossé et al. 1997, 1999; Souchon et al. 1998). Such techniques enable the improvement of yeast viability, by trapping the hydrophobic lactones and so by decreasing their concentration within the aqueous phase. However, these techniques lower the biotransformation yields, probably by adsorbing part of the substrate. Another way of protecting yeast cells from lactone toxicity is the use of alginate beads to immobilise the micro-organism. Such a process



**Fig. 6**  $\gamma$ -Decalactone production in a non-buffered methyl ricinoleate medium (10 g/l; black circles), or in medium containing phosphate buffer (0.1 M) at pH 7 (black diamonds), or pH 8 (white squares), or adjusted to pH 9 with NaOH (squares containing crosses; all data from Aguedo 2002)

was investigated with *Sporidiobolus* sp. but did not notably improve the yields (Lee et al. 1999). So, according to these recent data concerning the interaction of lactones with the yeast membranes, new strategies may be tested within the culture medium to decrease the fluidising action of the metabolites. It should be possible to obtain such an effect, for example by lowering the process temperature or by adding membrane-stabilising agents to the culture medium (Beney and Gervais 2001).

#### Yeast behaviour within the biphasic medium

The biotransformation medium generally contains an important concentration of the precursor fatty acids and, once shaken in the presence of a surfactant, it forms an oil-in-water emulsion. It was previously reported that the size of lipid droplets (Bakhuis and Bos 1969), the inoculum and the cell concentration in the medium all influence the characteristics of an emulsion and the rate of yeast growth (Prokop et al. 1972). In the case of  $\gamma$ -decalactone production by *Y. lipolytica*, preliminary studies have been made concerning these points, as it appears important to better understand the uptake of the non-polar precursor, i.e. methyl ricinoleate. Thus, it was observed that in a medium containing 5 g methyl ricinoleate/l and inoculated with *Y. lipolytica* [optical density at 600 nm ( $OD_{600}$ ) = 0.25], the cells alone were sufficient to rapidly decrease the size of the lipid droplets (Waché et al. 2000b). The hypothesis of biosurfactant production was put forward, as *Y. lipolytica* was previously reported to release a bioemulsifier called liposan when grown in a hexadecane-containing medium (Cirigliano and Carman 1985). However, such a compound appears in the medium after several hours of culture, which is inconsistent with the observed immediate effect of the cells on the emulsion. So, cell-surface activity appeared to be an important factor in the yeast cell behaviour within the biotransformation medium. When the cell concentration is higher ( $OD_{600}$  = 4.0) in a medium containing 10 g methyl ricinoleate/l, the cells alone do not lead to lipid emulsification and the presence of a surfactant is necessary to improve the formation of small



lipid droplets and thus lipid degradation by the cells. A direct contact occurs between the yeast cell surface and lipid droplets; and mainly numerous small-sized droplets adsorb onto the cell surface. So, the yeast surface properties are important in the interactions occurring between the emulsified substrate and the cells. These properties are currently being studied and it appears in fact that the adsorption of fatty globules to the yeast surface is linked to Lewis acid–base interactions, in part, rather than only to hydrophobic/hydrophilic interactions (Aguedo et al. 2003).

The cell release of biosurfactants and lipases into the medium may influence substrate assimilation by yeast cells. The excretion of such compounds during biotransformation is probable, although no data are available concerning these points. *Y. lipolytica* is able to excrete a biosurfactant (Cirigliano and Carman 1985) and also to produce high amounts of lipases (Pignède et al. 2000a, 2000b).

## Conclusion

Several points are still to be studied, but the most important ones are probably concerning  $\beta$ -oxidation fluxes which can, by small environmental or genetic changes, greatly modify production yields. With the aim of studying this, an effort has to be made on the modelling of this complex pathway. Genetic constructions can also be improved, especially to get strains biotransforming more rapidly. Another field to investigate is the production of other lactones, a domain which is closely linked to the hydroxylation of the substrate.

The review of these data concerning the production of lactones shows that this field has an important impact on increasing our knowledge of lipid metabolism in yeast. The yeast *Y. lipolytica*, the object of this study, is, due to its great adaptation to hydrophobic environments, a very good model for lactone production. One great interest in this species is that it has become a model for many steps of non-polar substrate metabolism, such as the action of lipases, cytochrome P450, acyl-CoA oxidases and the biology of peroxisomes. The data existing at all these levels and the genome knowledge to come (Casaregola et al. 2000) will help to develop new applications for this unconventional yeast.

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