Metallo-Supramolecular Micellar Gels: A Structural Study

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^eBragg Institute, Building 87, Australian Nuclear Science and Technology Organisation, Locked Bag 2001, Kirrawee DC NSW 223, Australia The formation of metallo-supramolecular micellar gelsis investigated by SANS. The micelles consist of polystyrene-*block*-poly(*tert*-butylacrylate), PS-*b*-PtBA-[,(-[isterpyridine) block copolymers, dissolved in deuterared ethanol, which guarantees a significant contrast.

The hydrophobic polystyrene core is stabilized by the poly(*tert*-butylacrylate) corona. The influence of the copolymer concentration on the micellestructure and on its space organization has been investigated, as well as the significant rheological impact of the subsequent addition of three metal ions (Fe(II), Ni(II) and Zn(II)).

The form factor of the micelles is described using the Pedersen and Gerstenberg model[1], with a solvent-free spherical core and gaussian PtBA chains. The fitting parameters are the average radius of the core, the standard deviation of its size distribution, and the radius of gyration of the PtBA chains.

Upon increasing the copolymer concentration, the average distance between the micelles decreases and interparticle interferences lead to a structure factor peak at increasing q values. This behaviour has been taken into account through the Percus-Yevick hard sphere model[2]. Two new parameters are introduced: the hard sphere interaction distance and the hard sphere volume fraction in the solution.

The structural parameters characterizing both the micelles and the network built upon gelation have beendetermined upon increasing concentration and addition of several metal ions. Theresults arecorrelated with dynamic light scattering, electronic microscopy and rheology experiments.

- 1. Pedersen, J. S.; Gerstenberg, M. C. Macromolecules 1996, 29, 1363-1365.
- 2. Percus, J. K.; Yevick, G.; J. Phys. Rev1958, 110, 1-13.