

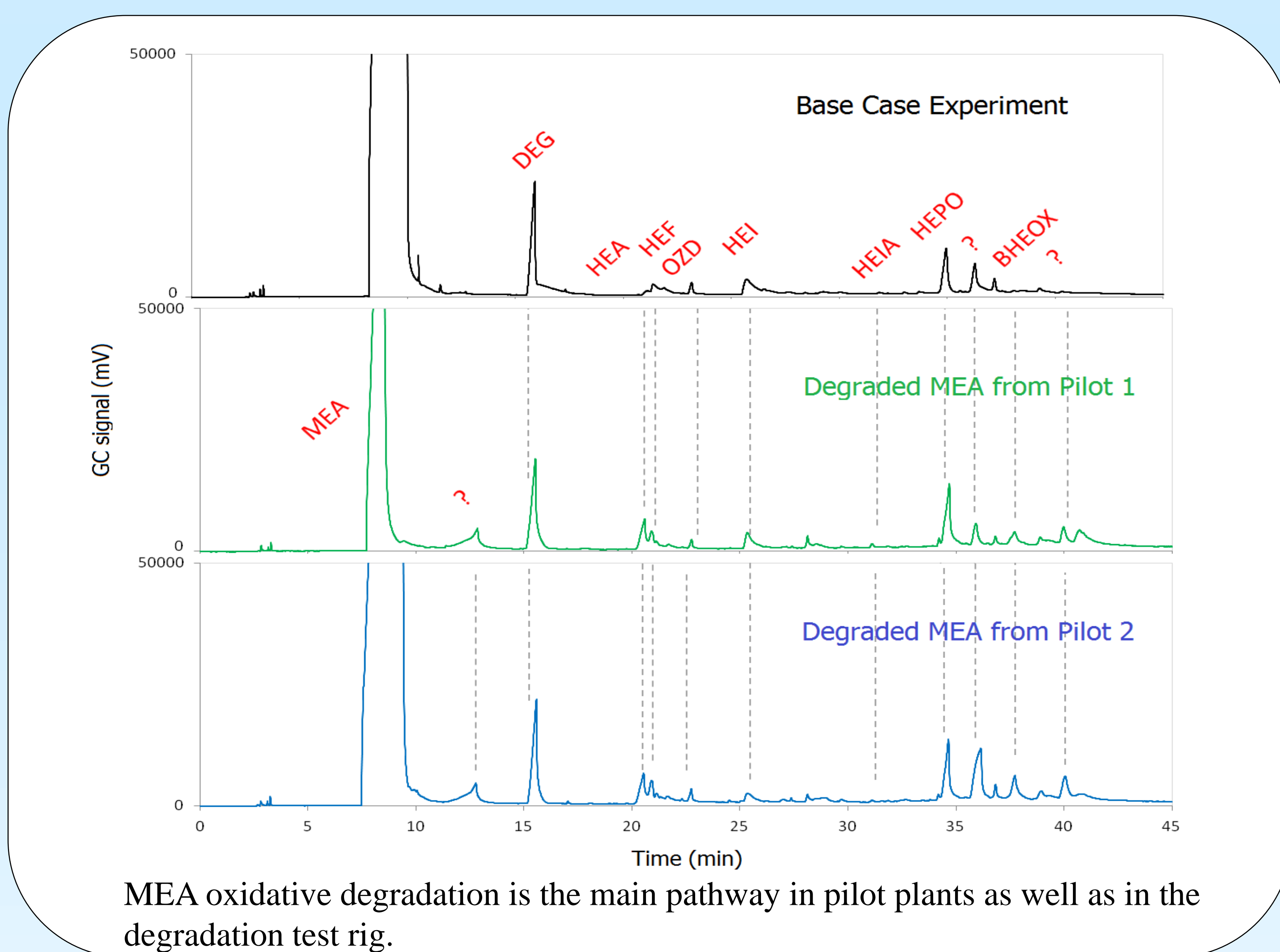
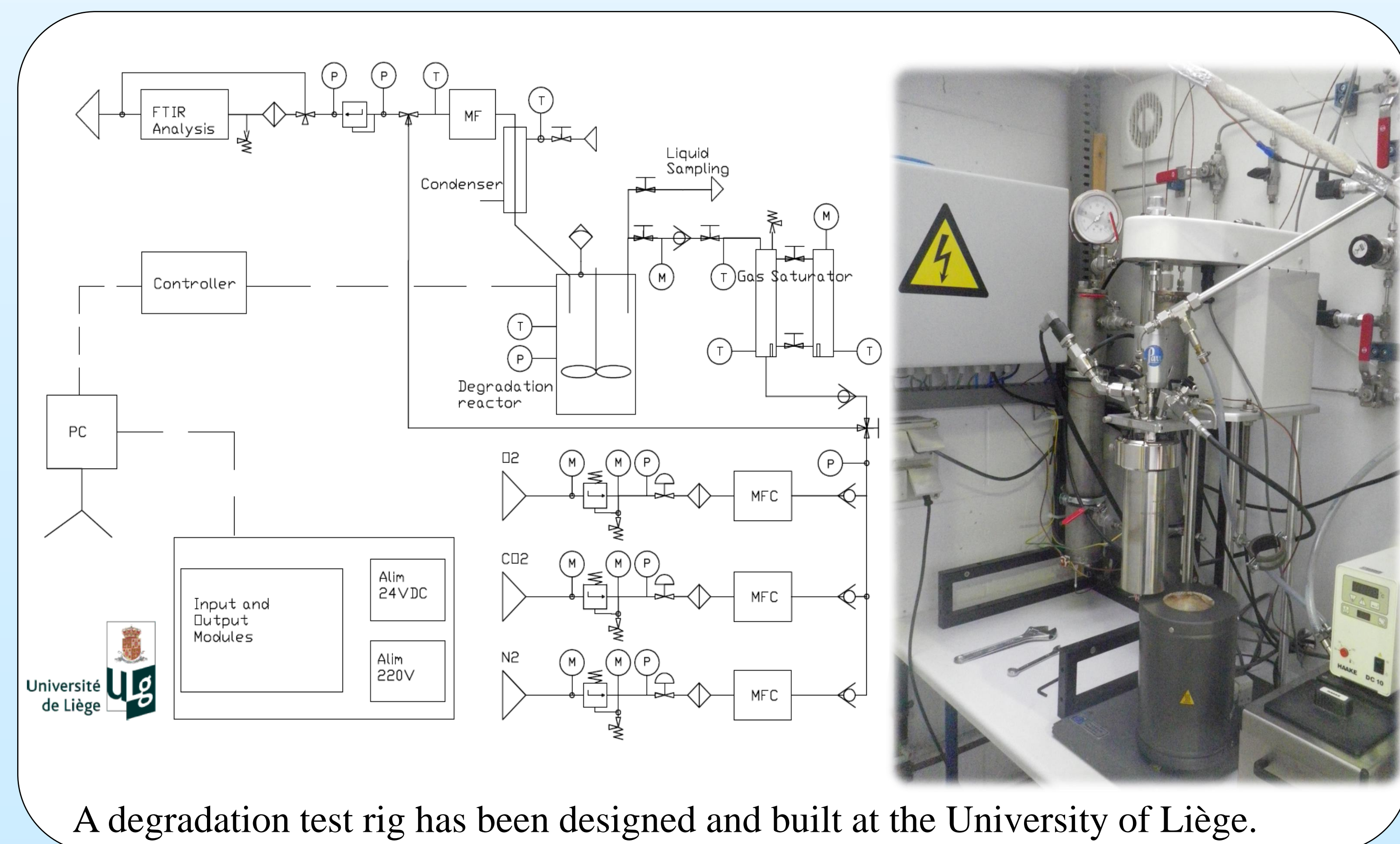
Abstract

Solvent degradation may be a major drawback for the large-scale implementation of post-combustion CO₂ capture due to amine consumption and emission of degradation products. A kinetics model describing solvent oxidative and thermal degradation has been developed based on experimental results. This model has been included into a global Aspen Plus model of the CO₂ capture process, so that optimal operating conditions can be identified to minimize both energy and environmental impacts of the process.

Solvent degradation

Solvent degradation reduces the process efficiency, increases corrosion, and implies additional costs for the treatment of degradation products. It may represent up to 22% of the CO₂ capture Opex^[1]!

The degradation of MEA is artificially accelerated in an experimental test rig. The nitrogen mass balance can be closed within the 10% range and repeatability is demonstrated with a deviation lower than 5%. Identified degradation pathways at lab-scale are similar to pathways observed in PCCC pilot plants.



Kinetics of the solvent degradation

The kinetics of MEA degradation has been determined by studying the influence of different operating parameters:

- Agitation rate
- Temperature
- Composition of the flue gas feed (N₂, O₂, CO₂)

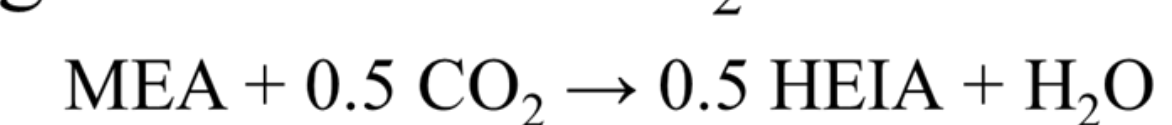
Two different degradation equations have been included in an Aspen Plus model of the CO₂ capture process:

MEA oxidative degradation:



$$-r_{\text{MEA}} = 5.35e^5 \cdot e^{-\frac{41730}{RT}} \cdot [\text{O}_2]^{1.46}$$

MEA thermal degradation with CO₂:



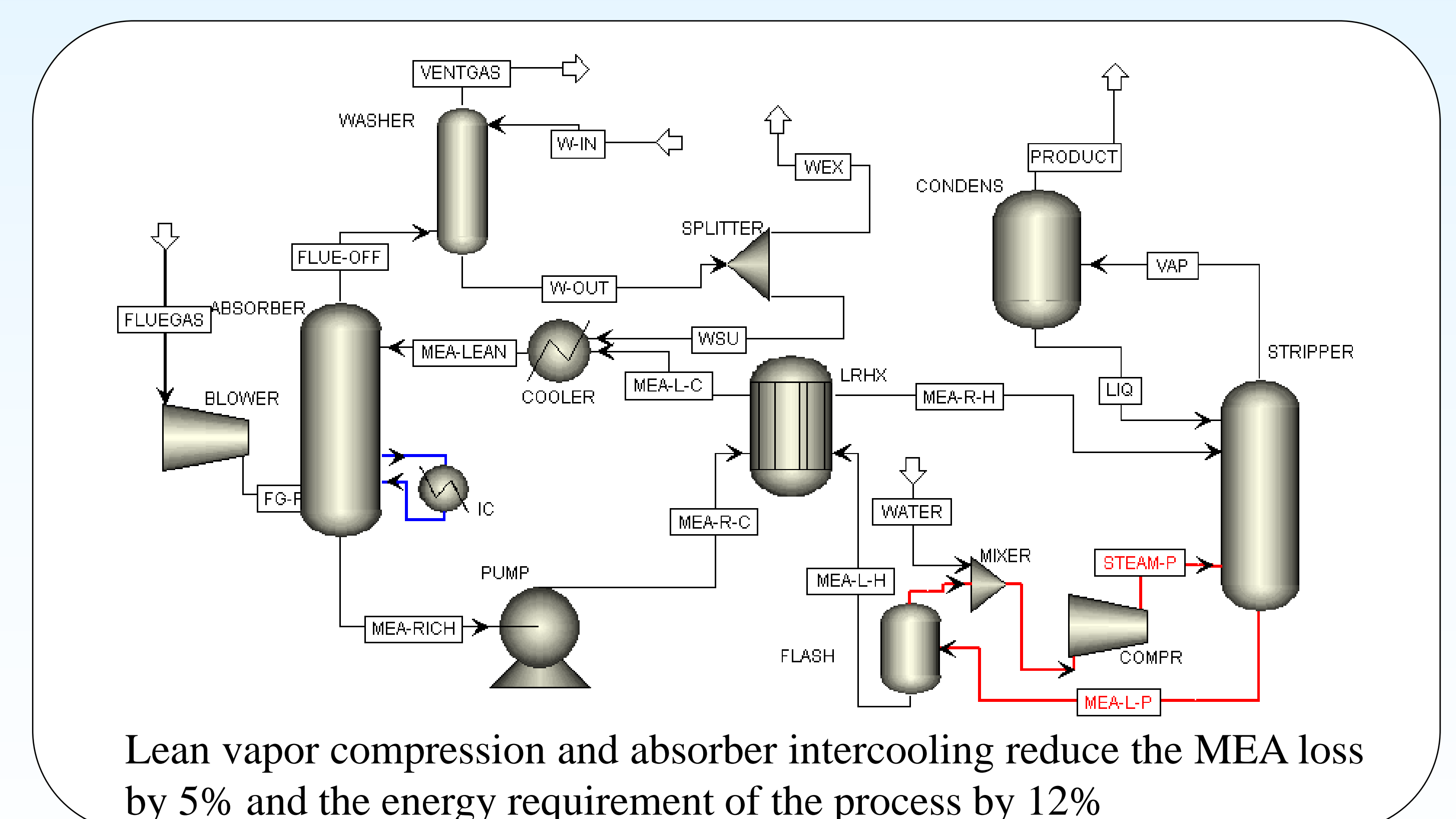
$$-r_{\text{MEA}} = 6.27e^{11} \cdot e^{-\frac{143106}{RT}} \cdot [\text{CO}_2]^{0.9}$$

Global process modeling

The global model of the CO₂ capture with assessment of solvent degradation may be used for the design of CO₂ capture plants to consider not only the process energy penalty, but also its environmental penalty which is particularly relevant for large-scale applications.

The optimized process configuration results into a MEA consumption of 78 g MEA/ton CO₂. This is lower than values reported in pilot plants, probably due to the modeling assumptions made in first approach. The model highlights that the MEA consumption is mainly related to oxidative degradation in the absorber, with ammonia as the major degradation product. The corresponding energy requirement of the CO₂ capture process is evaluated to 3.1 GJ/ton CO₂.

Finally, this model proposes an original approach for assessing solvent degradation within the modeling and design process of CO₂ capture plants.



Acknowledgements

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[1]: Abu Zahra M., 2009. Carbon dioxide capture from flue gas. PhD Thesis, TU Delft, The Netherlands.